# Routine Determination of Sulfonylurea, Imidazolinone, and Sulfonamide Herbicides at Nanogram-Per-Liter Concentrations by Solid-Phase Extraction and Liquid Chromatography/Mass Spectrometry

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## ABSTRACT

Sulfonylurea (SU), imidazolinone (IMI), and sulfonamide (SA) herbicides are new classes of lowapplication-rate herbicides increasingly used by farmers. Some of these herbicides affect both weed and crop species at low dosages and must be carefully used. Less is known about the effect of these compounds on noncrop plant species, but a concentration of 100 ng/L (nanograms per liter) in water has been proposed as the threshold for possible plant toxicity for most of these herbicides. Hence, analytical methods must be capable of detecting SUs, IMSs, and SAs at concentrations less than 100 ng/L in ambient water samples. The authors developed a two-cartridge, solid-phase extraction method for isolating 12 SU, 3 IMI, and 1 SA herbicides by using high-performance liquid chromatography/ electrospray ionization–mass spectrometry (HPLC/ESI–MS) to identify and quantify these herbicides to 10 ng/L. This method was used to analyze 196 surface- and ground-water samples collected from May to August 1998 throughout the Midwestern United States, and more than 100 quality-assurance and quality-control samples.

During the 16 weeks of the study, the HPLC/ESI-MS maintained excellent calibration linearity across the calibration range of 5 to 500 ng/L, with correlation coefficients of 0.9975 or greater. Continuing calibration verification standards at 100-ng/L concentration were analyzed throughout the study, and the mean measured concentrations for individual herbicides ranged from 93 to 101 ng/L. Mean recovery of herbicides from 27 reagent water samples spiked at 50 and 100 ng/L ranged from 39 to 92 percent and averaged 73 percent. The standard deviation of recoveries ranged from 14 to 26 percent and averaged 20 percent. This variability reflects multiple instruments, operators, and the use of automated and manual sample preparation. Spiked environmental water samples had similar recoveries, although for some herbicides, the sample matrix enhanced recoveries by as much as 200 percent above the spiked concentration. This matrix enhancement was sample and compound dependent. Concentrations of herbicides in unspiked duplicate environmental samples were typically within 25 percent of each other. The results demonstrate the usefulness of HPLC/ESI-MS for determining low-application-rate herbicides at ambient concentrations.

### INTRODUCTION

Sulfonylurea (SU), imidazolinone (IMI), and sulfonamide (SA) herbicides are new classes of herbicides that function by inhibiting the action of acetolactate synthase or acetohydroxyacid synthase, key enzymes in the biosynthesis of amino acids in plants. These herbicide classes are applied at rates that are typically much lower than those used for the triazine or acetanilide herbicides that are most commonly used in agriculture. In addition, SUs, IMIs, and SAs are much less toxic to mammals and other animals (Beyer and others, 1987; Meister, 1997). As a result, these herbicides are gaining in popularity among farmers. Crops that are treated with SUs, IMIs, and SAs include barley, corn, cotton, durum wheat, peanuts, rice, soybeans, spring wheat, and winter wheat.

The range in observed toxicity of SUs, IMIs, and SAs to crop and noncrop plants varies by a factor of greater than 1,000. Some of these herbicides are active on crop species at low dosages. Field studies indicate that some yield is lost in the most sensitive crop species when these herbicides are applied or present at between 1/100th and 1/500th of the typical application rate of 0.5 to 0.1 g (gram) active ingredient per ha (hectare)(Fletcher and others, 1993; Obrigawitch and others, 1998). Little is known about the effect of these compounds on noncrop plant species, but a concentration of 100 ng/L in water has been proposed as the concentration below which possible plant toxicity is minimized (Michael Barrett, U.S. Environmental Protection Agency, oral commun., 1997).

It may be difficult to detect SUs, IMIs, and SAs in many hydrologic settings because of their low application rates and low overall use amounts. Detections of SUs, IMIs, and SAs in samples collected from surface and ground water have been reported at nanogram-per-liter concentrations (Bergstrom, 1990; Michael and Neary, 1993). Commonly used analytical techniques might not have adequate sensitivity or selectivity to unambiguously detect and quantify SU, IMI, and SA herbicides, particularly in the presence of coextracted matrix interferences. Most of the SU, IMI, and SA herbicides are highly polar, water soluble, and labile, limiting their analysis by gas chromatography or gas chromatography/mass spectrometry. Highperformance liquid chromatography (HPLC) has been successfully used to separate SU, IMI, and SA herbicide mixtures, but standard optical detection methods have lacked sensitivity, particularly in the presence of the ultravioletabsorbing dissolved organic carbon present in natural-water samples.

Several investigators have used coupled HPLC/electrospray ionization-mass spectrometry (HPLC/ESI-MS) to separate, detect, and quantify SU, IMI, and SA herbicides in soils (D'Ascenzo and others, 1998a; Marek and Koskinen, 1996; Powley and de Bernard, 1998; Stout and others, 1998; Stout and others, 1997), plant materials (Chivanov and others, 1997; Stout and others, 1996a; Stout and others, 1996b), and water (Bossi and others, 1996; D'Ascenzo and others, 1998b; Di Corcia and others, 1997; Koppen and Spliid, 1998; Krynitsky, 1997; Rodriguez and Orescan, 1998). These studies have demonstrated that HPLC/ESI-MS is well suited to the determination of SU, IMI, and SA herbicides because these compounds are efficiently ionized under electrospray conditions. However, these studies were limited in the number of compounds or the types of water samples analyzed. To date (1999), routine application of HPLC/ESI-MS has not been demonstrated, particularly for surface- and ground-water samples collected from a range of hydrologic conditions and water chemistries.

In this paper the authors present an improved two-cartridge, solid-phase extraction (SPE) method for isolation of 12 SU, 3 IMI, and 1 SA herbicides (table 1), using HPLC/ESI-MS for detection, identification, and quantitation. This method was used to determine the concentrations of the 16 SU, IMI, and SA herbicides in about 200 surface-and ground-water samples collected throughout the Midwestern United States. More than 100 quality-assurance and quality-control (QA/QC) samples were concurrently analyzed. By using this data set, the authors demonstrate that SPE and HPLC/ESI-MS can be used to routinely identify and quantify SU, IMI, and SA herbicides at concentrations less than 10 ng/L in water samples of varying chemical composition.

 Table 1. Common names, herbicide classes, CAS numbers, and trade names of the sixteen herbicides

 measured in this study (data from Meister, 1997)

Common name	Herbicide class	CAS number	Trade name <sup>1</sup>	
Bensulfuron methyl	sulfonylurea	83055-99-6	Londax	
Chlorimuron ethyl	sulfonylurea	90982-32-4	Classic	
Chlorsulfuron	sulfonylurea	64902-72-3	Glean, Telar	
Flumetsulam	sulfonamide	98967-40-9	Broadstrike, Preside, Scorpion	
Halosulfuron methyl	sulfonylurea	100784-20-1	Battalion, Manage, Permit,	
			Sempra	
Imazapyr	imidazolinone	81334-34-1	Arsenal, Chopper	
Imazaquin	imidazolinone	81335-37-7	Scepter, Detail	
Imazethapyr	imidazolinone	81335-77-5	Pursuit	
Metsulfuron methyl	sulfonylurea	74223-64-6	Allie, Ally, Escort, Quit, Reform	
Nicosulfuron	sulfonylurea	111991-09-4	Accent	
Primisulfuron methyl	sulfonylurea	86209-51-0	Beacon, Tell	
Prosulfuron	sulfonylurea	94125-34-5	Peak	
Sulfometuron methyl	sulfonylurea	74222-97-2	Oust	
Thifensulfuron methyl	sulfonylurea	79277-27-3	Pinnacle	
Triasulfuron	sulfonylurea	82097-50-5	Amber	
Triflusulfuron methyl	sulfonylurea	126535-15-7	Upbeet	

[CAS, Chemical Abstracts Service]

<sup>1</sup>Use of brand, firm, and trade names in this article is for identification purposes only and does not constitute endorsement by the U.S. Geological Survey

## EXPERIMENTAL

More than 212 surface- and ground-water samples were collected from a range of surfaceand ground-water sites throughout the Midwestern United States. Most surface-water sites were sampled twice and ground-water sites were sampled once. The sampling design and trends in SU, IMI, and SA herbicide concentrations, as well as trends for triazine and acetanilide herbicides, are described in a separate paper in this volume (Battaglin and others, 1999).

All samples were filtered through a 0.7-um (micrometer) nominal pore size glass fiber filter and collected in Teflon-lined, screw capped 1-L amber glass bottles. Two 1-L bottles were collected for each sample. The bottles were baked at 425°C (degrees Celsius) for 8 hours prior to use to minimize the potential for organic contamination. After filtration and collection, the samples were shipped on ice by overnight express and kept at 2 to 4°C in the dark until extraction.

#### **Solid-Phase Extraction Preparation**

Ultrapure (distilled-in-glass or equivalent) reagent water and solvents were used in the

development of the method and in the determination of SU, IMI, and SA herbicides. For each sample extraction, two SPE cartridges were stacked in series. The first cartridge was a 6mL (milliliter) barrel, 500-mg (milligram) bed strong anion exchange resin cartridge (Jones Chromatography, Inc., Lakewood, Colorado). This cartridge removed much of the dissolved organic carbon (DOC) from the sample; for most environmental samples, the strong anion exchange resin bed was colored with DOC. The IMIs were retained on the strong anion exchange cartridge, while the SUs and SAs were not retained. The second cartridge was a 6-mL barrel, 1-g bed, RP-102 (Applied Separations, Allentown, Pennsylvania) styrene-divinyl benzene polymeric SPE cartridge. The RP-102 cartridge retained the SUs, IMIs, and any SAs that were not retained on the strong anion

exchange cartridge. Prior to analysis, the combined cartridge stack was conditioned by passing 60 mL of 1 percent acetic acid in acetone and 40 mL of 1 percent acetic acid in water through the cartridges. The stack was not allowed to dry after conditioning.

## **Sample Extraction**

The pH of the samples was measured prior to sample extraction, and the pH was adjusted to 3 or less by addition of acetic acid. The volume of acetic acid used ranged from 10 to 18 mL. The samples were then extracted using a Zymark AutoTrace SPE Workstation (Zymark Corporation, Hopkinton, Massachusetts). The sample extraction flowrate through the stacked SPE cartridges was 10 mL/min (milliliters per minute). The stack was then dried in-line with nitrogen for 1 minute to remove excess water. A 1.2-mL aliquot of methanol was then passed through each stack to remove interstitial water trapped in the SPE material. The stack was then dried in-line with nitrogen for 45 minutes to remove as much residual water as possible. Three 4-mL aliquots of 1 percent acetic acid in acetone were used to elute the compounds from the stacked SPE cartridges. The flowrate for elution was 3 mL/min. The 12-mL final volume was collected in graduated polyethylene test tubes.

## **Sample Concentration**

The eluted sample extracts were placed in a Zymark TurboEvap (Zymark Corporation, Hopkinton, Massachusetts) solvent reduction workstation for solvent removal. The nitrogen gas pressure was kept at 10 pounds per square inch and the water bath temperature maintained at 38°C. Sample reduction to just dryness took about 60 minutes because all residual water was not removed in the previous drying steps. A 1mL aliquot of acetonitrile was then added to each sample extract. The sample extract was dried again. An additional 1-mL aliquot of acetonitrile was again added and the sample dried for a third time. These two drying steps were used to ensure the removal of acetic acid. The sample extracts were then reconstituted with 100 •L (microliters) acetonitrile, followed by 900 •L of water in the

polyethylene test tubes. The sample extracts were then filtered through a 0.2-•m Teflon membrane syringe filter to remove any particulate material prior to analysis. The filtered extracts were put into amber, graduated 2-mL sample vials (Hewlett Packard, Palo Alto, California). The sample vials were capped and kept at 2 to 4°C until analysis.

## **Sample Analysis**

The sample extracts were placed in the autosampler of a Hewlett Packard Series 1100 HPLC/MS system (Hewlett Packard, Palo Alto, California). The HPLC and MS operational characteristics are listed in table 2. A Metasil Basic 3-•m pore size 2x150 mm (millimeter) analytical HPLC column and a Metasil Basic 3-•m pore size 2x10 mm (Metachem Technologies, Inc, Torrance, California) guard column were used for HPLC separation. Acetonitrile and a 10mM (millimolar) ammonium formate/formic acid buffer, pH 3.7 in water, were used as the two components to the HPLC gradient. The solvent compositions used for the HPLC gradient in this analysis are listed in table 3.

Full-scan electrospray ionization mass spectra were collected for each compound using a Hewlett Packard Series 1100 LC/MSD. The instrument was tuned for positive ions using horse skeletal muscle myoglobin. Ionization conditions were optimized so that at least three ions were formed for each herbicide. The relative abundances of the three ions were used to confirm identification, and the sum of the three ion signals was used for maximum detectability and quantitation. The selected ions monitored for each compound and the fragmentor voltages required to generate those ions are listed in table 4.

SU, IMI, and SA detections were verified in two ways. First, the chromatographic retention time observed for a peak in a sample chromatogram was compared to the retention of the compound observed in a standard chromatogram (fig.1). The qualitative identification was then confirmed by comparing the relative abundances of the three ions measured in the sample peak to known relative abundances of a library reference spectrum **Table 2.** High-performance liquid chromatography (HPLC) and mass spectrometry (MS) operating conditions for the combined analysis of sulfonylurea, imidizolinone, and sulfonamide herbicides

Flowrate	0.2 milliliter per minute		
Autosampler temperature	8 degrees Celsius		
Analysis time	45 minutes		
Column temperature	30 degrees Celsius		
Injection volume	50 microliters		
Ionization mode	electrospray		
MS Operating Conditions			
Polarity	positive		
Electron multiplier voltage	2,748		
Sheath (drying) gas volume	10 liters per minute		
Nebulizer pressure	55 pounds per square inch		
Gas temperature	325 degrees Celsius		
Spray chamber capillary voltage	4,000 volts		

**Table 3.** High-performance liquid chromatography binary gradient program for the combined analysis of sulfonylurea, imidizolinone, and sulfonamide herbicides

Time,	Acetonitrile,	Aqueous buffer,
in minutes	in percent	in percent
0.50	20	80
1.75	35	65
10.0	40	60
13.0	50	50
20.0	55	45
23.0	90	10
27.0	90	10
30.0	20	80
45.0	20	80

Compound	Ions monitored	Fragmentor voltage for SIM analysis	
Imazapyr	217, 234, 262	95	
Imazethapyr	177, 245, 290	115	
Flumetsulam	129, 262, 326	95	
Nicosulfuron	182, 213, 411	95	
Imazaquin	199, 267, 312	95	
Thifensulfuron methyl	167, 270, 388	80	
Metsulfuron methyl	167, 264, 382	80	
Chlorsulfuron	141, 167, 358	85	
Sulfometuron methyl	150, 199, 365	85	
Triasulfuron methyl	141, 167, 402	85	
Bensulfuron methyl	149, 182, 411	90	
Halosulfuron methyl	182, 403, 435	85	
Prosulfuron	141, 167, 420	85	
Chlorimuron ethyl	186, 213, 415	85	
Triflusulfuron methyl	264, 461, 493	90	
Primisulfuron methyl	199, 254, 469	90	

**Table 4.** Selected-ion monitoring (SIM) program for the combined analysis of sulfonylurea,

 imidizolinone, and sulfonamide herbicides



**Figure 1.** Chromatogram of a 100-ng/L (nanogram-per-liter; equivalent to 5 ng on-column) 16 component sulfonylurea-imidazolinone-sulfonamide standard using high-performance liquid chromatography/electrospray ionization–mass spectrometry with selected-ion monitoring.

produced from a pure standard. This comparison, however, might not be conclusive if unknown compounds interfere with the sample peak, either partially or completely. If comparison of the unknown peak spectrum with the library reference spectra yields dissimilar spectral curves, then the presence of the compound was notconfirmed, and so it was not identified. If a chromatographically resolvable peak was identified, with all three ions present, but one ion's relative abundance varied by more than 20 percent from the standard, then the compound concentration was qualified as an estimate. All detections less than 10 ng/L were qualified as estimates.

Concentrations were quantified by comparing the sum of the three integrated peak areas from an environmental sample to the sum of integrated peak areas from a calibration line. A five-point external calibration curve for a concentration range of 5 to 500 ng/L was produced for each compound. Calibration lines were linear and had correlation coefficients of 0.9975 or greater.

Samples were extracted and isolated in sets of twelve: ten environmental samples, a method blank, and a method spike in which a blank water sample was amended with all 16 compounds to make a concentration of 50 or 100 ng/L per compound. In addition, duplicate environmental samples were analyzed for quality-control purposes. Thirty-four duplicate samples were amended with a solution of all 16 compounds to 50 or 100 ng/L. These matrix spikes were analyzed in the same set as the unspiked environmental sample. Thirty-two samples were analyzed in duplicate without amendment. These replicates were analyzed within and between sets to estimate method variability over time.

A set of five different ca produced from a pure standard. This comparison, however, might not be conclusive if libration standards was analyzed with each set of processed samples. In addition, continuing calibration verification samples and instrumental blank samples were interspersed between the environmental samples and after the calibration standards to verify that the instrument remained within calibration limits, and that cross- contamination between injections did not occur.

#### **RESULTS AND DISCUSSION**

Recoveries for this method are shown in figure 2. Reagent water and South Platte surfacewater samples were spiked at 100 ng/L and analyzed in quadruplicate. Recoveries ranged from 56 to 86 percent in reagent water, averaging 76 percent. The variation in recovery for all 16 compounds averaged 6 percent. Recoveries are more variable in surface water, ranging from 46 to 163 percent and averaging 93 percent. The variation in recovery for all 16 compounds averages 7 percent. Enhanced recoveries were consistently observed for South Platte surface water (and also observed in ground water), particularly for thifensulfuron methyl, metsulfuron methyl, triasulfuron methyl, and bensulfuron methyl. The observed matrixenhanced recoveries ranged as high as 300 percent for a of water-soluble salts and DOC.sample fortified at 100 •/L. After minor modification of the method by introduction of a solvent-exchange step, the mean matrix enhancement was less than 200 percent, likely the result of improved removal of water-soluble salts and DOC.

The samples were analyzed in 27 separate sets, and instrumental and method QA/QC samples were analyzed as part of each set. Table 5 lists data for long-term calibration stability, as indicated by the performance of 75 continuing calibration verification samples analyzed over the course of the study. The continuing calibration verification sampleequivalent concentration for each compound was set at 100 ng/L. The mean measured concentration for individual SU, IMI, and SA herbicides ranged from 93 to 101 ng/L. The mean standard deviation for all SU, IMI, and SA herbicides measured in continuing calibration verification samples throughout the study was 6 ng/L. These data, along with linear calibration



**Figure 2.** Recoveries of 16 herbicides from reagent water (RW1) and South Platte River water (SW1) fortified at 100ng/L (nanograms per liter).

coefficients of 0.9975 or greater, indicate that instrument performance was highly stable throughout the course of the study.

Long-term method performance is indicated by the recovery calculated from the results of 27 set spikes, one analyzed with each set during the study. The recoveries for all individual SU, IMI, and SA herbicides are shown in figure 3. Mean recoveries for all 16 herbicides from 27 reagent water set spike samples, fortified at 50 and 100 ng/L, ranged from 39 to 92 percent and averaged 73 percent. The standard deviation of recoveries ranged from 14 to 26 percent and averaged 20 percent. This variability reflects multiple instruments, operators, and the use of automated and manual sample preparation. Chlorsulfuron, halosulfuron methyl, prosulfuron, triflusulfuron methyl, and primisulfuron methyl had the lowest and most variable recoveries. **Table 5.** Continuing calibration performance during the course of the study. The concentrationstatistics were calculated from determinations of 75 continuing calibration verification samples[All concentrations in nanograms per liter]

Compound	Compound mean	Compound median	Compound standard deviation	Minimum concentration	Maximum concentration
Imazapyr	100	100	6.6	85	128
Imazethapyr	100	101	6.4	84	117
Flumetsulam	96	96	5.2	84	114
Nicosulfuron	95	96	4.3	83	111
Imazaquin	101	101	6.6	85	126
Thifensulfuron methyl	93	93	7.0	60	120
Metsulfuron methyl	94	94	5.8	81	117
Chlorsulfuron	95	95	6.3	80	122
Sulfometuron methyl	96	96	5.9	80	113
Triasulfuron methyl	93	93	5.1	79	108
Bensulfuron methyl	93	94	5.3	78	106
Halosulfuron methyl	94	94	6.9	81	122
Prosulfuron	94	95	6.6	71	121
Chlorimuron ethyl	94	94	5.4	79	110
Triflusulfuron methyl	93	94	6.2	76	114
Primisulfuron methyl	96	96	6.6	81	118
Set mean concentration	95	96	6.0	79	117
Set median concentration	95	95	6.3	80	117
Set standard deviation	2.7	2.7	.8	6.2	6.3
Minimum concentration	93	93	4.3	60	106
Maximum concentration	101	101	7.0	85	128

Thirty-one unspiked duplicate samples were analyzed to estimate method precision at ambient herbicide concentrations and in the presence of interferences. These duplicates were analyzed within a single sample set (intraset replicates) or in two separate sets (interset) replicates. Intraand interset replicates were combined for data analysis. Box plots of the distributions of relative percentage differences calculated for the five herbicides detected in unspiked replicates more than three times are shown in figure 4. Except for chlorimuron ethyl, concentrations of herbicides in unspiked duplicate environmental samples were typically within 25 percent of each other.

In addition to unspiked replicates, 28 randomly selected sample duplicates were spiked at 100 ng/L and analyzed. This spike determined the magnitude of possible matrix enhancement during the study. These results are shown in figure 5.



**Figure 3.** Box plots of recoveries of sulfonylurea, imidazolinone, and sulfonamide herbicides from 27 reagent water samples processed and analyzed from September 1998 to January 1999. Reagent water samples were spiked at 50 or 100 ng/L (nanograms per liter). In this plot, hinges indicate median recoveries, gates indicate 25<sup>th</sup> and 75<sup>th</sup> percentiles, and whiskers indicate 5<sup>th</sup> and 95<sup>th</sup> percentiles. The small • indicates the compound mean recovery, and the small x indicates outlier recoveries.

The box plots (fig. 5) indicate that there was measurable matrix enhancement of recovery. This enhancement was variable and seemed to affect most of the herbicides measured. However, the results suggest that matrix enhancement would result in overestimation by no more than a factor of two, and more likely compound recoveries would be enhanced by a factor of 1.5 or less. Given other aspects of analysis and sample collection variability, and the typically low concentrations measured in environmental samples, this level of matrix enhancement is unlikely to alter the interpretation of surface- or ground-water data.



**Figure 4.** Relative percent differences (RPD) between unspiked duplicate natural-water samples. In this plot, hinges indicate median recoveries, gates indicate 25<sup>th</sup> and 75<sup>th</sup> percentiles, and whiskers indicate 5<sup>th</sup> and 95<sup>th</sup> percentiles. The small • indicates the compound mean recovery, and the small x indicates outlier recoveries. Thirty-one total samples were analyzed (n=number of samples with detections in both duplicates; C1 and C2 refer to the concentrations of the first and second duplicate, respectively).

A summary of field results from the analysis of 196 surface- and ground-water samples is presented in Battaglin and others (1999). The selected herbicides were detected at concentrations greater than the reporting limit in 83 percent of stream samples. In surface-water samples, the three most frequently detected herbicides were imazethapyr (71 percent), flumetsulam (63 percent), and nicosulfuron (52 percent). Average concentrations of these five herbicides ranged from 11 to 92 ng/L. Six other herbicides were measured in 2 to 24 percent of samples. The data suggest that several of the sixteen selected compounds are commonly detected in surface-water samples of the Midwestern United States, and that the average concentrations are less than the 100-ng/L level of concern. Battaglin and others (1999) provide a detailed statistical analysis of the field results.



**Figure 5.** Recovery of herbicides added to duplicate surface- and ground-water samples at a concentration of 100 ng/L (nanograms per liter). Recoveries corrected for ambient water herbicide concentrations. In this plot, hinges indicate median recoveries, gates indicate 25<sup>th</sup> and 75<sup>th</sup> percentiles, and whiskers indicate 5<sup>th</sup> and 95<sup>th</sup> percentiles. The small • indicates the compound mean recovery, and the small x indicates outlier recoveries.

In this study, the authors have demonstrated the usefulness of HPLC/MS for the routine determination of SU, IMI, and SA herbicides by using a tandem SPE isolation coupled to HPLC/ESI–MS. The analytical technologies used in this method are critical to adequate assessment of the presence and distribution of lowapplication-rate herbicide residues in surface and ground water. The long-term QA/QC data developed as part of this study show that HPLC/ESI–MS is a robust technique that can be used for the routine determination of herbicides at concentrations at or less than 10 ng/L. Extensive QA/QC data collection and interpretation are required to verify the performance of this method and to assess the magnitude of sample matrix enhancement of concentrations.

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