

**Bibliography of Work on the  
Heterogenous Photocatalytic  
Removal of Hazardous  
Compounds from Water and Air  
Update Number 2  
To October 1996**

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## Summary

The Solar Industrial Program, funded by the United States Department of Energy, has developed processes that destroy hazardous substances in or remove them from water and air. This work was done by the National Renewable Energy Laboratory in Golden, CO, and Sandia National Laboratory in Albuquerque, NM, and by their subcontractors at universities and small or large businesses. The processes of interest in this report are based on the application of heterogeneous photocatalysts, principally titanium dioxide or modifications thereof, but work on other heterogeneous catalysts is included in this compilation.

This report continues bibliographies that were published in May, 1994, and October, 1995. The author compiles published work on the photocatalytic oxidation and reduction of organic or inorganic compounds in air or water. The previous reports included 663 and 574 citations, respectively. This update contains an additional 518 references. These were published during the period from June 1995 to October 1996, or are references from prior years that were not included in the previous reports. The work generally focuses on removing hazardous contaminants from air or water to meet environmental or health regulations. This report also references work on properties of semiconductor photocatalysts and applications of photocatalytic chemistry in organic synthesis. This report follows the same organization as the previous publications. The first part provides citations for work done in a few broad categories that are generic to the process. Three tables provide references to work on specific substances. The first table lists organic compounds that are included in various lists of hazardous substances identified by the United States Environmental Protection Agency (EPA). The second table lists compounds not included in those categories, but which have been treated in a photocatalytic process. The third table covers inorganic compounds that are on EPA lists of hazardous materials or that have been treated by a photocatalytic process. A short update on companies that are active in providing products or services based on photocatalytic processes is provided.

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## 1.0 Introduction

This update in combination with the previous reports<sup>1,2</sup> provides a comprehensive bibliography of work available in the open literature for scientists and engineers interested in the use of heterogeneous photocatalytic oxidation or reduction processes in environmental remediation, process emission control, indoor air quality, or other applications. The combined bibliographies include more than 1700 citations to work published between 1970 and the third quarter of 1996. The literature cited includes United States and foreign patents. Information was compiled by manually scanning the literature and searching commercial databases. This update includes about 150 citations to work done prior to 1995 that were not included in the earlier reports. Some citations have doubtless been missed and topics covered in certain papers may not have been identified and covered in every appropriate category. The author is grateful to the many people who have sent references and reprints of their work. Coverage is limited to heterogeneous processes except in a few cases where review material for homogeneous processes is relevant to heterogeneous photocatalytic chemistry. The author invites readers to send references to relevant work that appeared before 1996 that has been missed to the mail or e-mail address included in Section 5.1 of this report.

The photocatalytic oxidation of organic compounds in water has received the most attention but there is a rapidly increasing amount of work on the oxidation of volatile organic or inorganic compounds in the gas phase. Photocatalytic reduction of both organic compounds and metal-containing ions is also receiving increasing attention.

It is widely observed that intermediates and by-products are often formed that persist in the treated stream. These can include acids, oxygenated compounds, and carbon monoxide. The nature of the intermediates is increasingly being used to reveal details of the chemical mechanism of the photocatalytic process. The net process involves oxidizing the organic compound to an intermediate stage of oxygen content or to

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<sup>1</sup>Blake, Daniel M. (1994). ***Bibliography of Work on the Photocatalytic Removal of Hazardous Compounds from Water and Air***. NREL/TP-430-6084. Golden, CO: National Renewable Energy Laboratory. 75 pp. [DE94006906] Available from the National Technical Information Service, Springfield, VA 22161.

<sup>2</sup>Blake, Daniel M. (1995). ***Bibliography of Work on the Photocatalytic Removal of Hazardous Compounds from Water and Air. Update Number 1, to June***, 1995. NREL/TP473-20300. Golden, CO: National Renewable Energy Laboratory. 102 pp. [DE95013148] Available from the National Technical Information Service, Springfield, VA 22161.

carbon dioxide, water, and a mineral acid (if a heteroatom such as nitrogen or chlorine is present). Other oxidizing agents may be substituted for oxygen. Certain metal ions can be modified or removed from water when the ions replace oxygen as the electron acceptor in the process and sacrificial electron donor compounds are provided.

Reference to test work on more than 270 compounds is included in the tables in Section 3. The most-studied compounds are phenol derivatives, BTEX (benzene, toluene, ethyl benzene, and xylene) components found in fuel spills, and chlorinated solvents such as trichloroethylene and chloroform. Test work reports that a significant number of pesticides, dye compounds, and surfactants can be completely oxidized (mineralized), and a variety of bacteria and viruses have been killed by irradiation with near-ultraviolet light in the presence of titanium dioxide. The breadth of work attests to the very wide range of applications that are being evaluated for the technology.

The section on companies active in the field notes only additions and changes that have occurred since the last report.<sup>2</sup> The high level of participation of Japanese companies in the Second International Conference on TiO<sub>2</sub> Photocatalytic Purification and Treatment of Water and Air<sup>3</sup> and the numerous patents assigned to Japanese companies indicate a high level of activity in that country. However, the author has little information on the individual companies that may be active in this field in Japan.

The following sections discuss reviews written on various aspects of the technology, work in developing and testing photocatalysts and oxidants, engineering issues, other topics, patents, and companies active in bringing photocatalytic processes or services to the marketplace. These sections, which include information that can apply to a range of applications, are followed by tables listing references to work performed on specific substances. References to work on systems in which the compound to be treated is carried in the gas phase are indicated by the suffix "g" in the citation number.

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<sup>3</sup>Al-Ekabi, Hussain, *Abstracts of the Second International Conference on TiO<sub>2</sub> Photocatalytic Purification and Treatment of Water and Air*, Cincinnati, OH, October 26-29, 1996, Science and Technology Integration, Inc., London, ON.

## 2.0 Generic Information

This section refers to work that pertains to the field of photocatalytic processes.

### 2.1 Reviews and Articles

Reviews describe many aspects of photocatalytic chemistry and technology. This section also includes reviews written in the years prior to 1995 that were not cited in the previous reports. Reviews covering the broad topics given can be found in the following:

Semiconductors and supports - 120, 362, 368, 369, 370, 371, 396, 495, and 506; methodology for comparing reaction systems - 396, 424, and 446; mechanisms/activity - 165, 173, 174, 179, 231, 277, 365, 366, 367, 368, 398, 425, 435, and 439; **nanocrystalline** semiconductors - 165, 231, and 233; photoelectrochemical systems - 113 and 115; photocatalysis - 46, 47, 104, 105, 106, 107, 113, 114, 115, 119, 121, 127, 132, 146, 157, 173, 231, 365, 366, 367, 368, 372, and 506; solar processes - 112, 355, and 461g; companies active in the field - 16; adsorption - 78; disinfection - 17, 167, 269, 330g, and 489; water treatment - 30, 269, and 297; indoor air quality - 17, 324, 326, 330g, 438g, and 450g; environmental applications - 112, 144, 145, 166, 173, 174, 179, 250, 386, 412, and 461g; energy storage - 193; preparation and applications of semiconductor thin films - 232, 233, and 250; organic synthesis - 372; intercalation compounds - 350; oxidation of phenols - 414 and 435; comparison of advanced oxidation processes - 109, 269, and 406; issues and questions in application of photocatalysis - 355, 412, and 494; reactors - 259; problems of immobilization of catalyst - 45; photocatalytic ammonia synthesis - 52 and 84; activation of oxygen - 103; charge injection into semiconductor particles - 110, 111, 116, 117, and 118; and activating  $\text{TiO}_2$  with visible light - 217.

### 2.2 Photocatalysts

The nature of the photocatalyst determines the rate and efficiency of the process. The anatase form of titanium dioxide has the desirable properties of being chemically stable, readily available, and active as a catalyst for oxidation processes. The 3.2 eV band gap matches the output of a wide variety of readily available lamps but is not ideal for solar applications. The photoefficiency for reaction of hazardous molecules is generally rather low, particularly for the aqueous-phase processes. Considerable work has

been directed toward modifying  $\text{TiO}_2$  and testing other semiconductors to identify ways to increase process efficiency and to improve the overlap of the absorption spectrum of the photocatalyst with the solar spectrum. This work is broken down into a few broad categories and covered in the references cited.

### **2.2.1 Modified Titanium Dioxide**

Titanium dioxide and modified forms, including different commercially available forms, heat treated materials, and materials prepared by a range of techniques:

sol-gel - 2, 3, 13, 14, 21, 91, 164, 212, 239, 241, 253, 254, 311, 316, 336, 343g, 388, 397g, 429, 492, and 508; controlled hydrolysis of  $\text{TiCl}_4$  - 423; heat treatment - 65, 164, and 252; colloidal or quantized particles - 51, 67, 68, 96, 102, 291, 337, 407, 419, 420, 421, and 453; xerogel/aerogel - 2, 3, 246, 317, and 470; flame synthesis - 77 and 381; or catalysts from different vendors - 35, 51, 285, 334, 389, 413, 424, and 446; spray pyrolysis - 44 and 77; impregnation - 44 and 67; co-precipitation - 44 and 67; microemulsion - 68; reduction - 183; rutile - 432g; potassium pillared layered titanate - 3 18 and 3 19; cysteine - 387; and surface modification - 462.

### **2.2.2 Hydrophobic Surface Treatment**

No new work was identified in this area

### **2.2.3 Dye Sensitized Titanium Dioxide**

Sensitizers have been used in conjunction with titanium dioxide to improve the response to visible light:

ruthenium complexes - 38, 99, 235, 364, and 476; vanadium(+5) complexes - 41; cobalt(+2) complexes - 3 1; nickel(+2) complexes - 31; copper(+2) complexes - 3 1; cobalt(+3) complexes - 40; metal phthalocyanine - 43 and 266; organic dyes - 38, 266, and 278; poly-(10-substituted-2-(9-enthral) ethylmethacrylates) - 122; polypyrrole - 507; or erythrosin B - 234.

### **2.2.4 Metal Ion Doping of Titanium Dioxide or Mixed Oxides**

Other metal ions have been introduced into the titanium dioxide lattice to modify the properties. They are covered in the following:



Al - 13; Si - 2, 3, 13, 14, 21, 33, 55, 139, and 329g; V - 39, 190, 407, and 423; Cr - 195, 200, 370, 407, and 423; Fe - 44, 51, 96, 340, 342, 407, and 423; Co - 39; Cu - 69; Ga - 182, 185, 187, and 474g; Zr - 139; Nb - 182, 185, 187, and 474g; Mo - 92; Ce - 55; or W - 92, 299, and 329g.

### **2.2.5 Metallized Titanium Dioxide**

Noble metals have been deposited on the titanium dioxide surface to enhance catalytic activity:

Ni - 73, 172, 180, 181, 378, and 380; Rh - 98, 172, 178, 180, and 181; Pd - 146, 156, 218, and 398; Pt - 22, 32, 73g, 74g, 76g, 89, 98, 140g, 172, 175, 176, 178, 180, 181, 191, 192, 194, 197, 201, 278, 287g, 298, 310, 316, 370, 371, 376, 378, 380, 398, 457, 486, 510, and 514; Ag - 198, 199, 226, and 251; Au - 32 and 198; Pt/Rh - 199; Ag/Rh - 199; or Pt/Pd - 199.

### **2.2.6 Other Semiconductors**

A wide range of other semiconductors and other materials have been tested for photocatalytic activity. In general they have been found to be less active than titanium dioxide. Relevant work is cited in the following:

$\text{Al}_2\text{O}_3$  - 60g and 237;  $\text{V}_2\text{O}_5$  - 196g and 375g;  $\text{Fe}_2\text{O}_3$  - 60g, 158, 159, 160, 200, and 402; ZnO - 9, 22, 99, 196g, 200, 202, 237, 375g, 393, 402, and 426; Zn - or CdS - 265, 402, and 486; Pt/CdS - 486; ZnTe - 298;  $\text{ZrO}_2$  - 133, 196g, 200, 339, and 375g;  $\text{ZrTiO}_4$  - 75, 339, and 241;  $\text{MoS}_2$  - 298;  $\text{SnO}_2$  - 33, 99, 122, 133, 196g, 230, 375g, and 475;  $\text{Sb}_2\text{O}_4$  - 196g, 200, and 375g;  $\text{Sn/SbO}_2$  - 375g;  $\text{CeO}_2$  - 196g, 200, and 375g;  $\text{WO}_3$  - 123, 196g, 200, 202, and 375g;  $\text{Nb}_2\text{O}_5$  - 26; fly ash - 160; furnace slag or core sands - 70; polyaniline - 59; poly(p-phenylene) - 472g; polythiocyanogen - 472g; muscovite - 160; or natural minerals - 134 and 158.

### **2.2.7 Immobilized Photocatalysts**

Most experimental work on aqueous systems has used fine particles of photocatalyst suspended in the liquid phase. In a waste treatment application it would be simpler if the catalyst were immobilized in the photoreactor so the material would not have to be separated from the treated fluid in a subsequent process step. Most work on the treatment of gases has used immobilized catalyst. Titanium dioxide has been affixed to a variety of surfaces:

glass (including fibers) - 18, 21, 57, 97, 120, 141, 148g, 239, 252, 254, 255g, 307, 308, 311, 316, 327, 328, 331, 357, 358, 359, 383, 397, 427, 428, 429, 467, 473g, 477, 492, 502, and 514; silica - 49, 67, 326g, 405g, 458g, 463, 471, 498, and 514; metal - 97, 163, 164, 331, 514, and 518; clays - 67 and 120; polymer - 6, 7, 15, 120, 262, 265, 331, 459g, 464, 482, and 483; thin films - 23, 120, 255g, 343g, 397g, 403, 441, 442, 443, and 493; internal light guide - 444; zeolite - 21, 53, 120, 254, 325, 458g, 471, 498, 501, 512, and 515; alumina - 458g and 498; carbon - 215, 216, 220g, 292, 458g, and 471; paper - 305 and 306g; Ambersorb - 514; PHOTOPERM™ - 39, 41, and 42; metal oxides and ceramics - 120, 147, 405g, and 459g; vesicles and micelles - 120; or silica coating on TiO<sub>2</sub> films - 441, 442, and 443.

## **2.3 Hydrogen Peroxide and Other Oxidants**

Oxygen is the most commonly used oxidant, but other oxidants have been found to improve rates of reaction with a variety of organic substrates under some conditions. This work is covered in the following:

hydrogen peroxide - 9, 29, 35, 48, 206, 364, 373, 511, and 517; superoxide - 373; peroxydisulfate - 35, 303, and 313; chlorite - 303; chlorate - 303; bromate - 303 and 312; periodate - 303; or ozone - 26, 48, 304, 379, and 499.

## **2.4 Engineering Issues**

In recent years the success of laboratory work has led to interest in applying the technology to environmental remediation and treatment of process waste streams. Literature now discusses scale-up of the process and resolution of engineering problems. Progress has been significant and many companies are now providing turn-key systems for treating contaminated water and air.

### **2.4.1 Reactor and System Design**

A number of papers address topics relevant to the design of reactors for photocatalytic processes:

photochemical reactors - 251, 252, 300, 301, and 310; non-concentrating reactor - 345 and 460g; parabolic trough - 50, 79, and 313; compound parabolic concentrator - 79; kinetic modeling - 300, 301, 405g, 497, and 513; fixed bed - 91 and 343; filtration - 491 and 508; controlled periodic illumination - 100, 416, 417, and 436; light scattering model - 4, 149; flat plate reactor - 28;

ceramic monolith - 405g; effectiveness factors for reactions in a planar membrane - 94; mass transfer limitations - 95 and 25 1; photocatalytic membrane reactor - 214; fiber optic cable reactor - 358 and 359; or field tests - 392g.

## 2.4.2 Systems Analysis

As the technology for the photocatalytic treatment of contaminated air or water has progressed, some work on economic evaluation has appeared in the literature. However, there was limited new information during the last year:

process waste water - 379; or photon costs from lamps and sunlight - 48.

## 2.5 Miscellaneous Topics

This category includes papers of interest that do not fall into the preceding headings:

actinometry - 210; adsorption - 22, 292, 302, 320, 344g, 405g, 426g, and 443g; combined photocatalytic and biological treatment - 48; effect of applied voltage bias on photocatalytic reactions - 477, 478, and 479; non-aqueous solvent systems - 131, 135, 137, 228, 234, 235, 377, 396, 431, 436, and 486; application in chemical analysis - 1, 262, 445, and 457; polymer aging - 6, 7, and 152; photo initiators - 40; “inert” ion effect - 29, 51, 285, and 290; dissolved metal ion effect - 56, 66, 159, 290, 463, 511, and 517; comparison with other advanced oxidation processes - 51, 81, 364, 366, 379, and 418; purification of water for recycle in semiconductor fabs - 63; purification of drinking water - 64; photoelectrode - 71; organic synthesis - 108, 114, 119, 124, 125, 126, 127, 128, 132, 136, 228, 356, 377, 396, 486, and 496; surface science of  $\text{TiO}_2$  - 150, 286g, and 287g; quantum yield as a function of particle size and light intensity - 15 1, 420, and 42 1; photoconductivity/electrical conductivity - 177, 184, 185, 186, 187, 196g, and 201; self cleaning surface - 357, 437, and 459; mixed waste - 418; or treatment of photographic processing effluents - 422.

## 2.6 Patents

The number of patents for aspects of photocatalytic technology has increased rapidly in the last decade. They cover a range of aqueous and gas-phase applications. The general topic of the patents are described in the following:

process for oxidation of hydrocarbons - 229; parts having photocatalytic function - 168, 169, 240, 241, 242, and 258; catalyst formulation - 15, 171, 238, 240, 241, 242, 244g, 258, 263, 268, 308, 327, 331, 336, 353, 360, 381, 399, 447, 449g, 454, 455, 467, 468, 470, 490, and 515; thin silica coating - 441; odor control - 335g; water treatment - 308, 347, 466, 505, and 514; water treatment system - 58, 142, 205, 270, and 447; painted bodies - 204 and 456; photocatalytic tiles - 222; photocatalyst on paper supports - 451 and 452; photocatalyst on flexible supports - 223; membrane for water purification - 279; regeneration of adsorbents - 24g and 449g; metals removal from water - 245; method for activity measurement - 348; removal of chlorinated compounds from air (includes scrubber) - 309g; removal of air pollutants (includes catalysts on exterior walls of buildings) - 24g, 25g, 141g, 143g, 148g, 215g, 238g, 242g, 258g, 308g, 449g, and 467; method for cleaning solid surface - 332; additive to enhance degradation of plastics in the environment - 352; deodorizing agents - 242 and 401; process for killing cells or disinfection - 93, 205, 354, and 448; methanol synthesis - 353; method for preparing hydrocarbons - 390; ethylene oxidation to preserve fruit and vegetables - 400; pigment additive - 55; slurry reactor configuration - 72; catalyst cleaning - 499; or plasma reactor - 500.

## **2.7. Companies Active in the Field**

Research and development (R&D) and market assessment activity is underway at many companies in the United States but there is little change in the list of companies that have started business lines that use photocatalytic processes?

The assets, in the field of photocatalytic oxidation, of IT Corporation, NEPCCO, and PHOTOX Corporation have been acquired by a new company, Zentox Corporation:

Zentox Corporation

Mr. J. Sandy Reese, President

2140 N.E. 36th Avenue

Ocala, FL 34470

Telephone: (352) 867-9706

Fax: (352) 867-1320

Products: Advanced oxidation, ozone treatment, electrocoagulation, low temperature distillation, and photocatalytic oxidation systems for groundwater remediation, industrial water treatment, and treatment of contaminated vapor streams for industrial, environmental, consumer, and medical applications.



Universal Air Technology, Inc.

Sanjeev K. Hingorani, Ph.D.

Sid Martin Biotechnology Development Institute

12085 Research Drive

Alachua, FL 32615

Telephone: (904)418-0661

Fax: (904) 462-0875

Products: Stand-alone and HVAC duct integrated products for indoor air disinfection using photocatalytic technology.

### 3.0 Compounds Studied

The tables in this section have the same format as in the first two reports. No compounds have been removed from the tables but new compounds have been added to the second table to incorporate new work. The list of the compounds included in various lists of priority pollutants, air toxics, and the toxic release inventory compiled by the EPA<sup>4</sup> provided the starting point for these tables. Table 1 lists compounds in the EPA categories; Table 2 lists organic compounds that are not in EPA lists; and Table 3 covers inorganic compounds in EPA lists or that have been treated by a photocatalytic process. The inorganic compounds are arranged by element unless a significant number of citations referred to work on a specific ion or compound. Formulas of compounds, when given, are not in the standard format because the software used to prepare the tables does not support subscripts. A few broad categories are included in Table 2 that reflect new applications: bacteria, algae, and virus; coal or carbon; adsorbable organic halides (AOX); color or chemical oxygen demand (COD); and oil or petroleum, for example. In these tables the citation suffix “g” indicates a gas-phase study. The treatability of compounds that have not been tested can usually be inferred from results for related compounds in the tables.

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<sup>4</sup>“Notice of the Second Priority List of Hazardous Substances Commonly Found at Super-fund Sites,” *Environmental Reporter*, October 28, 1988, 1255-1260.

Table 1. Organic Compounds in EPA Lists of Priority Pollutants, Air Toxics, or Toxic Release Inventory

Substance	Formula	Halo- gen	Het. Atom	Reference
1,1,1-Trichloroethane	CHCl <sub>2</sub> CH <sub>2</sub> Cl	Cl		206,392g
1,1,2,2-Tetrachloroethane	CHCl <sub>2</sub> CHCl <sub>2</sub>	Cl		206
1,1,2-Trichloroethane	CHCl <sub>2</sub> CH <sub>2</sub> Cl	Cl		206
1,1,2-Trichloro-1,2,2-trifluoroethane	CCl <sub>2</sub> FCFClF <sub>2</sub>	Cl,F		
1,1 -Dichloroethane	CH <sub>3</sub> CHCl <sub>2</sub>	Cl		206
1,1 -Dimethyl hydrazine	(CH <sub>3</sub> ) <sub>2</sub> NNH <sub>2</sub>		N	
1,2,3-Trichloropropane	CH <sub>2</sub> ClCHClCH <sub>2</sub> Cl	Cl		
1,2,4-Trichlorobenzene	C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub>	Cl		
1,2,4-Trimethylbenzene	C <sub>6</sub> H <sub>3</sub> (CH <sub>3</sub> ) <sub>3</sub>			
1,2-Butylene oxide	H <sub>2</sub> COCHCH <sub>2</sub> CH <sub>3</sub>			
1,2-Dibromoethane	BrCH <sub>2</sub> CH <sub>2</sub> Br	Br		
1,2-Dibromo-3-chloropropane (DBCP)	CH <sub>2</sub> BrCHBrCH <sub>2</sub> Cl	Br,Cl		
1,2-Dichlorobenzene	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	Cl		
1,2-Dichloroethane	ClCH <sub>2</sub> CH <sub>2</sub> Cl	Cl		206,257g
1,2-Dichloroethylene	ClHC:CHCl	Cl		206,392g
1,2-Dichloropropane	CH <sub>3</sub> CHClCH <sub>2</sub> Cl	Cl		
1,2-Dinitrotoluene	C <sub>6</sub> H <sub>3</sub> CH <sub>3</sub> (NO <sub>2</sub> ) <sub>2</sub>		N	
1,2-Diphenylhydrazine	C <sub>12</sub> H <sub>12</sub> N <sub>2</sub>		N	
1,2-Trans-dichloroethene	C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	Cl		
1,3,5-Trinitrobenzene	C <sub>6</sub> H <sub>3</sub> (NO <sub>2</sub> ) <sub>3</sub>		N	87, 408
1,3-Butadiene	H <sub>2</sub> C:CHHC:CH <sub>2</sub>			
1,3-Dichlorobenzene	C <sub>6</sub> H <sub>4</sub> Cl	Cl		
1,3-Dichloropropene	CHCl:CHCH <sub>2</sub> Cl	Cl		
1,4-Dichlorobenzene	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	Cl		292
1,4-Dioxane	OCH <sub>2</sub> CH <sub>2</sub> OCH <sub>2</sub> CH <sub>2</sub>			
1-Amino-2-methylanthraquinone	C <sub>6</sub> H <sub>4</sub> [C(O)] <sub>2</sub> C <sub>6</sub> H <sub>2</sub> NH <sub>2</sub> CH <sub>3</sub>			
1-Bromo-4-phenyloxybenzene	p-BrC <sub>6</sub> H <sub>4</sub> OC <sub>6</sub> H <sub>5</sub>	Br		
2,2,4-Trimethylpentane	(CH <sub>3</sub> ) <sub>3</sub> C <sub>5</sub> H <sub>9</sub>			
2,3,7,8-Tetrachlorodibenzo-p-dioxin	C <sub>12</sub> H <sub>4</sub> Cl <sub>4</sub> O <sub>2</sub>	Cl		
2,4,5-Trichlorophenoxyacetic acid	C <sub>6</sub> H <sub>2</sub> Cl <sub>3</sub> OCH <sub>2</sub> CO <sub>2</sub> H	Cl		333
2,4,5-TP acid (silvex)	Cl <sub>3</sub> C <sub>6</sub> H <sub>2</sub> OCH(CH <sub>3</sub> )COOH	Cl		
2,4,5-Trichlorophenol	C <sub>6</sub> H <sub>2</sub> Cl <sub>3</sub> OH	Cl		26,82,208
2,4,6-Trichlorophenol	C <sub>6</sub> H <sub>2</sub> Cl <sub>3</sub> OH	Cl		82, 208
2,4,6-Trinitrotoluene	CH <sub>3</sub> C <sub>6</sub> H <sub>2</sub> (NO <sub>2</sub> ) <sub>3</sub>			48,87,408,408,409,410,411
2,4 Diaminoanisole	(NH <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OCH <sub>3</sub>		N	
2,4-Dichlorophenoxyacetic acid (2,4-D)	Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OCH <sub>2</sub> COOH	Cl		81,294,333,364,389,440
2,4-Diaminoanisole sulfate	(NH <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OCH <sub>3</sub> .H <sub>2</sub> SO <sub>4</sub>		N	
2,4-Dichlorophenol	Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH	Cl		82, 247, 511
2,4-Dimethylphenol	(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH			465

Substance	Formula	Halo- g e n	Het. Atom	Reference
2,4-Dinitrophenol	C6H3OH(NO2)2		N	
2,4-Dinitrotoluene	C6H3CH3(NO2)2			87
2,4-Toluene diamine	CH3(NH2)2C6H3		N	
2,6-Dinitrotoluene	C6H3CH3(NO2)2		N	87
2,6-Xylidine	(CH3)2C6H3NH2		N	
2-Acetylaminofluorene	CH3C(O)NHC6H3CH2C6H4	F	N	
2-Aminoanthraquinone	C6H4(CO)2C6H3NH2		N	
2-Butanone	CH3COCH2CH3			
2-Chloroacetophenone	C6H5COCH2Cl	Cl		
2-Chloroethyl vinyl ether	CH2ClCH2OCHCH2	Cl		
2-Chlorophenol	C6H4OHCl	Cl		59,82,267
2-Ethoxyethanol	H3CCH2OCH2CH2CH2OH			
2-Methoxyethanol	MeOCH2CH2OH			
2-Methylnaphthalene	C10H7CH3			
2-Nitrophenol	NO2C6H4OH		N	413,414
2-Nitropropane	CH3CHNO2CH3		N	
2-Pentanone, 4-Methyl	CH3(CH2)2COCH3			
2-Phenylphenol	C6H5C6H4OH			
3,3'-Dichlorobenzidine	C6H3ClNH2C6H3ClNH2	Cl	N	
3,3'-Dimethoxybenzidine	[C6H3(OCH3)NH2]2		N	
3,3'-Dimethylbenzidine (o-Tolidine)	[C6H3(CH3)NH2]2		N	
4,4'-Dichlorodiphenyldichloroethylene	(ClC6H4)2CCCl2	Cl		
4,4'-Diaminodiphenyl ether	NH2(C6H4)2NH2		N	
4,4'-Isopropylidenediphenol	(CH3)2C(C6H4OH)2			
4,4'-Methylenebis(N,N-dimethyl benzenamine)	C17H22N2		N	
4,4'-Methylenedianiline	H2NC6H4CH2C6H4NH2		N	
4,4'-Methylene-bis-(2-chloroaniline)	CH2(C6H4ClNH2)2	Cl	N	
4,4'-Thiodianiline	C12H12N2S		S,N	
4,6-Dinitro-o-cresol	CH3C6H2(NO2)2OH		N	
4,6-Dinitro-2-methylphenol	C7H6N2O5		N	
4-Aminoazobenzene	C6H5NNC6H4NH2		N	
4-Aminobiphenyl	C6H5C6H4NH2		N	
4-Chloroaniline	ClC6H4NH2	Cl		
4-Chlorophenyl phenyl ether	p-ClC6H4OC6H5	Cl		
4-Dimethylaminoazobenzene	(CH3)2C6H3NH2		N	
4-Methylphenol	p-CH3C6H4OH			
4-Nitrobiphenyl	C6H5C6H4NO2		N	
4-Nitrophenol	NO2C6H4OH		N	85,86,299,413,414
5-Nitro-o-anisidine	NO2C6H3(NH2)(OCH3)		N	
Acenaphthene	C10H6(CH2)2			
Acenaphthylene	C12H8			
Acetaldehyde	CH3CHO			158,305g,306g,343g,344g,



Substance	Formula	Halo- g e n	Het. Atom	Reference
Acetamide	CH <sub>3</sub> CNOH <sub>2</sub>		N	397g,405g,432g,450g <b>293</b>
Acetone	CH <sub>3</sub> COCH <sub>3</sub>			282g,329g
Acetonitrile	CH <sub>3</sub> CN		N	282g,283,283g,284g
Acetophenone	CH <sub>3</sub> C(O)C <sub>6</sub> H <sub>5</sub>			
Acrolein	CH <sub>2</sub> CHCHO			
Acrylamide	CH <sub>2</sub> CHCONH <sub>2</sub>		N	
Acrylic acid	H <sub>2</sub> C:CHCOOH			
Acrylonitrile	H <sub>2</sub> C:CHCN		N	
Aldrin	C <sub>12</sub> H <sub>8</sub> Cl <sub>6</sub>	Cl		
Allyl chloride	H <sub>2</sub> C=CHCH <sub>2</sub> Cl	Cl		
Aniline	C <sub>6</sub> H <sub>5</sub> NH <sub>2</sub>		N	
Anthracene	C <sub>6</sub> H <sub>4</sub> (CH) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>			<b>134</b>
Aramite	(CH <sub>3</sub> ) <sub>3</sub> CC <sub>6</sub> H <sub>4</sub> OCH <sub>2</sub> CH(CH <sub>3</sub> )- SO <sub>3</sub> C <sub>2</sub> H <sub>4</sub> Cl	Cl	S	
Atrazine	C <sub>18</sub> H <sub>14</sub> ClN <sub>5</sub>	Cl	N	41,42,313,383
Benzal chloride	C <sub>6</sub> H <sub>5</sub> CHCl <sub>2</sub>	Cl		
Benzamide	C <sub>6</sub> H <sub>5</sub> CONH <sub>2</sub>		N	81
Benzene	C <sub>6</sub> H <sub>6</sub>			140g,224g,282g,429
Benzidine	NH <sub>2</sub> (C <sub>6</sub> H <sub>4</sub> ) <sub>2</sub> NH <sub>2</sub>		N	
Benzoic acid	C <sub>6</sub> H <sub>5</sub> COOH			<b>160</b>
Benzoic trichloride	C <sub>6</sub> H <sub>5</sub> CCl <sub>3</sub>	Cl		
Benzoyl chloride	C <sub>6</sub> H <sub>5</sub> COCl	Cl		
Benzoyl peroxide	(C <sub>6</sub> H <sub>5</sub> CO) <sub>2</sub> O <sub>2</sub>			
Benzo(a)anthracene	C <sub>22</sub> H <sub>14</sub>			
Benzo(a)pyrene	C <sub>20</sub> H <sub>12</sub>			
<b>Benzo(b)fluoranthene</b>	C <sub>20</sub> H <sub>12</sub>			
<b>Benzo(g,h,i) perylene</b>	C <sub>22</sub> H <sub>12</sub>			
Benzyl alcohol	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> OH			
Benzyl chloride	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Cl	Cl		
BHC (Benzenehexachloride)	C <sub>6</sub> H <sub>6</sub> Cl <sub>6</sub>	Cl		
Biphenyl	C <sub>6</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub>			
<b>Bis(2-Chloroethoxy)methane</b>	CH <sub>2</sub> (2-ClC <sub>2</sub> H <sub>5</sub> O) <sub>2</sub>	Cl		
<b>Bis(2-chloroethyl) ether</b>	ClCH <sub>2</sub> CH <sub>2</sub> OCH <sub>2</sub> CH <sub>2</sub> Cl	Cl		
Bis(2-chloro-1-methylethyl) ether	[ClCH <sub>2</sub> (CH <sub>3</sub> )CH] <sub>2</sub> O	Cl		
<b>Bis(2-ethylhexyl) adipate</b>	(C <sub>7</sub> H <sub>13</sub> ) <sub>2</sub> C <sub>4</sub> H <sub>8</sub> (CO <sub>2</sub> ) <sub>2</sub>			
<b>Bis(2-ethylhexyl)phthalate</b>	(C <sub>4</sub> H <sub>9</sub> CH(CH <sub>2</sub> )) <sub>2</sub> OOC			
Bis(chloromethyl)ether	(CH <sub>2</sub> Cl)O(CH <sub>2</sub> Cl)	Cl		
Bromochloromethane	BrCH <sub>2</sub> Cl	Br,Cl		
Bromodichloromethane	CHCl <sub>2</sub> Br	Cl,Br		
Bromoethane	C <sub>2</sub> H <sub>5</sub> Br	Br		
Bromoform (Tribromomethane)	CHBr <sub>3</sub>	Br		<b>43</b>
Bromomethane (Methyl bromide)	CH <sub>3</sub> Br	Br		

Substance	Formula	Halo- g e n	Het. Atom	Reference
Butyl acrylate	CH <sub>2</sub> :CHCOOC <sub>4</sub> H <sub>9</sub>			
Butylbenzyl phthalate	C <sub>4</sub> H <sub>9</sub> O <sub>2</sub> CC <sub>6</sub> H <sub>4</sub> COOC <sub>7</sub> H <sub>7</sub>			
Butyraldehyde	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CHO			
Calcium cyanamide	NCNCa		N	
Caprolactam	CH <sub>2</sub> (CH <sub>2</sub> ) <sub>4</sub> NHCO		N	
<b>Captan</b> (N-Trichloromethylmercapto- tetrahydrophthalimide)	C <sub>9</sub> H <sub>8</sub> Cl <sub>3</sub> NO <sub>2</sub> S	Cl	N,S	
Carbaryl [ 1 -Naphthalenol, methylcarbamate]	C <sub>10</sub> H <sub>7</sub> OOCNHCH <sub>3</sub>		N	
Carbon disulfide	cs <sub>2</sub>		S	
Carbon tetrachloride	ccl <sub>4</sub>	Cl		206,282g,404g,513
Carbonyl sulfide	c o s			
Catechol	C <sub>6</sub> H <sub>4</sub> (OH) <sub>2</sub>			302
Chloramben (Benzoic acid, 3-amino-2, 5-dichloro-)	C <sub>6</sub> H(CO <sub>2</sub> H)(NH <sub>2</sub> )Cl <sub>2</sub>	Cl	N	
Chlordane	C <sub>10</sub> H <sub>6</sub> Cl <sub>8</sub>	Cl		
Chloroacetic acid	CH <sub>2</sub> ClCOOH	Cl		101
Chlorobenzene	C <sub>6</sub> H <sub>5</sub> Cl	Cl		
Chlorobenzilate (Benzeneacetic acid, 4-chloro-alpha-(4-chlorophenyl)-)	(C <sub>6</sub> H <sub>4</sub> Cl) <sub>2</sub> C(OH)COOC <sub>2</sub> H <sub>5</sub>	Cl		
Chlorodibenzodioxins, various	Cl <sub>2</sub> O <sub>2</sub> H <sub>8</sub> -xCl <sub>x</sub>	Cl		
Chlorodibenzofurans	Cl <sub>2</sub> O <sub>2</sub> H <sub>8</sub> -xCl <sub>x</sub>	Cl		
Chlorodibromomethane	ClBr <sub>2</sub> CH <sub>2</sub>	Br,Cl		
Chlorodifluoromethane	CHClF <sub>2</sub>	Cl,F		
Chloroethane	C <sub>2</sub> H <sub>5</sub> Cl	Cl		
Chloroform	CHCl <sub>3</sub>	Cl		206,276,282g,300,301,404g
Chloromethane	CH <sub>3</sub> Cl	Cl		
Chloromethyl methyl ether	C <sub>2</sub> H <sub>5</sub> ClO	Cl		
Chloroprene	H <sub>2</sub> C:CHCCl:CH <sub>2</sub>	Cl		
<b>Chlorothalonil</b> (1,3-Benzenedicarbonyl, 2,4,5,6-tetrachloro-)	C <sub>6</sub> Cl <sub>4</sub> (CN) <sub>2</sub>		N	254
Chrysene	C <sub>18</sub> H <sub>12</sub>			
cis- 1,2-Dichloroethylene	ClHC:CHCl	Cl		
cis- 1,3-Dichloropropene	CHCl:CHCH <sub>2</sub> Cl	Cl		
o-,m-,p-Cresols	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> OH			
Cumene	C <sub>6</sub> H <sub>5</sub> CH(CH <sub>3</sub> ) <sub>2</sub>			
Cumene hydroperoxide	C <sub>6</sub> H <sub>5</sub> C(CH <sub>3</sub> ) <sub>2</sub> OOH			
Cupferron (Benzeneamine, N-hydroxy- N-nitrose, ammonium salt)	C <sub>6</sub> H <sub>5</sub> N(NO)ONH <sub>4</sub>		N	
Cyclohexane	C <sub>6</sub> H <sub>12</sub>			323,413
Cyclohexanone	C <sub>6</sub> H <sub>10</sub> O			
Cyclonite (RDX)	(CH <sub>2</sub> ) <sub>4</sub> (NNO <sub>2</sub> ) <sub>4</sub>		N	48
Decabromodiphenyl oxide	(C <sub>6</sub> Br <sub>5</sub> ) <sub>2</sub> O	Br		

Substance	Formula	Halo- gen	Het. Atom	Reference
<b>Dialate</b> [Carbamothioic acid, bis (1 -methylethyl)-, S-(2,3-dichloro-2-propenyl) ester]	(CH3)2CH]2NCOSCH2CClCHCl	Cl	N,S	
Diaminotoluene (mixed isomers)	C7H7N2		N	
<b>Diazomethane</b>	CH2N2		N	
Dibenzofuran	C12H8O			
<b>Dibenzo(a,h)anthracene</b>	C22H14			
Dibromochloropropane	CH2BrCHBrCH2Cl	Br,Cl		
Dibutyl phthalate	C16H22O4			
Dichlorobenzene (mixed isomers)	C6H4Cl2	Cl		
Dichlorobromomethane	CHBrCl2	Cl,Br		
Dichlorodifluoromethane	CCl2F2	Cl,F		
Dichlorvos (Phosphoric acid, 2 dichloroethenyl dimethyl ester)	(CH3O)2P(O)OCH:CCl2	Cl	P	427,428,517
<b>Dicofol</b> ,4,4'-Dichloro-alpha-trichloro-methylbenzhydrol	C14H9Cl5O	Cl		
<b>Dieldrin/aldrin</b>	C12H10OCl6	Cl	P	
Diepoxybutane	C4H6O2			
Diethanolamine	(HOCH2CH2)2NH		N	
Diethyl phthalate	C10H14O4			
Diethyl sulfate	(C2H5)2SO4		S	
<b>Dimethyl</b> aminoazobenzene	C12H15N3		N	
<b>Dimethyl</b> for-man-tide (DMF)	HCON(CH3)2		N	293
<b>Dimethyl</b> phthalate	C10H14O4			
<b>Dimethyl</b> sulfate	(CH3)2SO4		S	
Dimethylcarbanyl chloride	(CH3)2NCOCl	Cl	N	
Disulfoton	(C2H5O)2P(S)SCH2CH2SCH2CH3		P,S	
Di-n-butyl phthalate	C18H26O4			
Di-n-octyl phthalate	C26H42O4			
Di-(2-ethylhexy) phthalate (DEHP)	C24H38O4			
Endosulfan	C9H6Cl6O3S	Cl		
<b>Endrin</b> aldehyde/ endrin	(C12H8OCl)6	Cl		
Epichlorohydrin	CH2OCHCH2Cl	Cl		
Ethyl acrylate	CH2=CHCOOC2H5			
Ethyl chloroformate	ClCOOC2H5	Cl		
Ethylbenzene	C8H10			
Ethylene	H2C=CH2			493g
Ethylene glycol	CH2OHCH2OH			
Ethylene oxide	CH2CH2O			
Ethylene thiourea	NHCH2CH2NHCS		N,S	
Ethyleneimine (Aziridine)	CH2NHCH2		N	
Fluometuron [Urea, N,N-dimethyl-N'-[3-(trifluoromethyl)phenyl]-]	C10H11F3N2O	F	N	

Substance	Formula	Halo- g e n	Het. Atom	Reference
Fluoranthene	C <sub>16</sub> H <sub>10</sub>			
Fluorene	C <sub>16</sub> H <sub>10</sub>			
Fluorotrichloromethane	CCl <sub>3</sub> F	Cl,F		
Formaldehyde	HCHO			144g,405g
<b>Heptachlor/heptachlor epoxide</b>	C <sub>10</sub> H <sub>7</sub> Cl <sub>7</sub>	Cl		
Heptane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>5</sub> CH <sub>3</sub>			
Hexachlorobenzene	C <sub>6</sub> Cl <sub>6</sub>	Cl		
Hexachlorobutadiene	Cl <sub>2</sub> C=CClCCl=CCl <sub>2</sub>	Cl		
<b>Hexachlorocyclopentadiene</b>	C <sub>5</sub> Cl <sub>6</sub>	Cl		
<b>Hexachloroethane</b>	C <sub>2</sub> Cl <sub>6</sub>	Cl		
Hexachloronaphthalene	C <sub>10</sub> H <sub>2</sub> Cl <sub>6</sub>	Cl		
Hexamethylphosphoramide	[(N(CH <sub>3</sub> ) <sub>2</sub> ) <sub>3</sub> PO]		P,N	
Hexamethylene- 1,6-diisocyanate	O=CN(CH <sub>2</sub> ) <sub>6</sub> NCO		N	
Hexane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>			
<b>Hydroquinone</b>	C <sub>6</sub> H <sub>4</sub> (OH) <sub>2</sub>			57
Indeno( 1,2,3-cd)pyrene	C <sub>22</sub> H <sub>12</sub>			
Isophorone	C(O)CHC(CH <sub>3</sub> )CH <sub>2</sub> C(CH <sub>3</sub> ) <sub>2</sub> CH <sub>2</sub>			
Isopropyl alcohol	(CH <sub>3</sub> ) <sub>2</sub> CHOH			55,101,207,228,271,274g,282g 339,374,413
Lindane (gamma-BenzenehexacNoride)	C <sub>6</sub> H <sub>6</sub> Cl <sub>6</sub>	Cl		161,280
Malachite Green	C <sub>23</sub> H <sub>25</sub> ClN <sub>2</sub>	Cl	N	
Malathion	(CH <sub>3</sub> O) <sub>2</sub> P(S)SCH(CO <sub>2</sub> C <sub>2</sub> H <sub>5</sub> )- CH <sub>2</sub> CO <sub>2</sub> C <sub>2</sub> H <sub>5</sub>		P,S	
<b>Maleic anhydride</b>	C <sub>4</sub> H <sub>2</sub> O <sub>3</sub>			
Maneb (Carbamodithioic acid, 1,2-ethanediybis-,manga.nese complex)	SSCNCH <sub>2</sub> CH <sub>2</sub> NHCSS)Mn		N,S	
Mechlorethamine	CH <sub>3</sub> N(CH <sub>2</sub> CH <sub>2</sub> Cl) <sub>2</sub>	Cl	N	
Melamine	H <sub>2</sub> NCNC(NH <sub>2</sub> )NC(NH <sub>2</sub> )N		N	
Methanol	CH <sub>3</sub> OH			107,228,274g,282g,283,283g, 344g,374,376,378,380,482, 183,510
Methoxychlor	Cl <sub>3</sub> CCH(C <sub>6</sub> H <sub>4</sub> OCH <sub>3</sub> ) <sub>2</sub>	Cl		
Methyl acrylate	CH <sub>2</sub> =CHCOOCH <sub>3</sub>			
Methyl butyl ketone	CH <sub>3</sub> COC <sub>4</sub> H <sub>9</sub>			
Methyl ethyl ketone	CH <sub>3</sub> COCH <sub>2</sub> CH <sub>3</sub>			513
Methyl iodide	CH <sub>3</sub> I	I		
Methyl isobutyl ketone	(CH <sub>3</sub> ) <sub>2</sub> CHCOCH <sub>3</sub>			
Methyl isocyanate	CH <sub>3</sub> NCO		N	
Methyl methacrylate	CH <sub>2</sub> =C(CH <sub>3</sub> )COOCH <sub>3</sub>			
Methyl tert-butyl ether	(CH <sub>3</sub> ) <sub>3</sub> COCH <sub>3</sub>			
Methylene bromide	CH <sub>2</sub> Br <sub>2</sub>	Br		
Methylene chloride	CH <sub>2</sub> Cl <sub>2</sub>	Cl		206,282g,283,283g,404g
Methylenebis(phenylisocyanate) (MBI)	CH <sub>2</sub> (C <sub>6</sub> H <sub>4</sub> NCO) <sub>2</sub>		N	

Substance	Formula	Halo- gen	Het. Atom	Reference
Methylhydrazine	CH3NHNH2		N	
Michler's ketone	CO[C6H4N(CH3)2]2		N	
Mirex	C10Cl12	Cl		
Mustard gas	S(CH3CH2Cl)2	Cl	S	
<b>m-Nitroaniline</b>	NO2C6H4NH2		N	
<b>N,N-Dimethylaniline</b>	C6H5N(CH3)2		N	
<b>Naphthalene</b>	C10H8			158,160
Naphthylamine (alpha-, beta-)	C10H7NH2		N	
Nitrilotriacetic acid	N(CH2COOH)3		N	
Nitrobenzene	C6H5NO2		N	81,87,356
Nitrofen [Benzene, 2,4-dichloro-1-(4-nitrophenoxy)-]	C12H7Cl2NO3	Cl	N	
Nitrogen mustard (2-Chloro-N-(2-cNoroethyl)-N-methylethanamine)	(ClCH2CH2)2NCH3	Cl	N	
Nitroglycerin	CH2NO3CHNO3CH2NO3		N	329g
Nitrophenol	NO2C6H4OH		N	
n-Butyl alcohol	CH3(CH2)2CH2OH			
n-Dioctyl phthalate	(C8H17OOC)2C6H4			
N-Nitrosodiethylamine	C4H10N2O		N	
N-Nitrosodimethylamine	(CH3)2N2O		N	
N-Nitrosodiphenylamine	(C6H5)2NNO		N	
N-Nitrosodi-n-butylamine	ONN(n-C4H9)2		N	
N-Nitrosodi-n-propylamine	ONN(n-C3H7)2		N	
N-Nitrosomethylvinylamine	ONN(CH3)(C2H3)		N	
N-Nitrosomorpholine	ONNC4H8O		N	
N-Nitrosomicotine			N	
N-Nitrosopiperidine	C5H10NHNO		N	
N-Nitroso-N-ethylurea	C(O)(NH2)N(NO)C2H5		N	
N-Nitroso-N-methylurea	C(O)(NH2)N(NO)(CH3)		N	
n-Pentane	CH3(CH2)3CH3			
<b>Octachloronaphthalene</b>	C10Cl8	Cl		
Octane	CH3(CH2)6CH3			
Oxirane	H2COCH2			
o-Anisidine	CH3OC6H4NH2		N	
o-Anisidine hydrochloride	CH3OC6H4NH2.HCl	Cl	N	
o-Nitroaniline	NO2C6H4NH2		N	
o-Toluidine	CH3C6H4NH2		N	
o-Toluidine hydrochloride	CH3C6H4NH2.HCl	Cl		
Parathion (DNTP)	(C2H5O)2P(S)OC6H4NO2		P,S	427,428,517
PCBs (Aroclor 1260,1254,1248, and 1242)	C12ClxH10-x	Cl		70,209,351
<b>Pentachlorobenzene</b>	C6Cl5H	Cl		
Pentachlorophenol	C6Cl5OH	Cl		208,280,359
Peracetic acid	CH3COOOH			

Substance	Formula	Halo- gen	Het. Atom	Reference
Phenanthrene	C <sub>4</sub> H <sub>10</sub>			
Phenol	C <sub>6</sub> H <sub>5</sub> OH			28,29,56,57,67,68,79,91,102, 276,311,318,319,349,382,413, 414,424,446,464,488,492
Phenol,2-methyl	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> OH			
Phosgene	COCl <sub>2</sub>	Cl		225g,328g
Phthalic anhydride	C <sub>6</sub> H <sub>4</sub> (CO) <sub>2</sub> O			
Picric acid	C <sub>6</sub> H <sub>2</sub> (NO <sub>2</sub> ) <sub>3</sub> OH		N	
Polybrominatedbiphenyls	C <sub>12</sub> Br <sub>x</sub> H <sub>10-x</sub>	Br,Cl		
Propane sultone	C <sub>3</sub> H <sub>6</sub> SO <sub>2</sub>		S	
Propionaldehyde	C <sub>2</sub> H <sub>5</sub> CHO			458g
Propiolactone, beta-	OCH <sub>2</sub> CH <sub>2</sub> CO			
Propoxur [Phenol, 2-(1-methylethoxy)- methylcarbamate]	C <sub>1</sub>		N	294
Propylene oxide	CH <sub>2</sub> OCHCH <sub>3</sub>			
Propylene (Propene)	CH <sub>3</sub> CH:CH <sub>2</sub>			
Propyleneimine	CH <sub>3</sub> H <sub>2</sub> CNHCH <sub>2</sub>		N	
Pyrene	C <sub>16</sub> H <sub>10</sub>			
p-Anisidine	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>		N	10
p-Chloro-m-cresol	C <sub>6</sub> H <sub>3</sub> CH <sub>3</sub> OHC <sub>1</sub>	Cl		
p-Cresidine	CH <sub>3</sub> C <sub>6</sub> H <sub>3</sub> (NH <sub>2</sub> )OCH <sub>3</sub>		N	
p-Nitrosodiphenylamine	(C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> NNO		N	
p-Phenylenediamine	C <sub>6</sub> H <sub>4</sub> (NH <sub>2</sub> ) <sub>2</sub>		N	
Quinoline	C <sub>9</sub> H <sub>7</sub> N		N	61
Quinone	C <sub>6</sub> H <sub>4</sub> O <sub>2</sub>			
Quintozene (Pentachloronitrobenzene)	C <sub>6</sub> Cl <sub>5</sub> NO <sub>2</sub>	Cl	N	
Safrole	C <sub>3</sub> H <sub>5</sub> C <sub>6</sub> H <sub>3</sub> O <sub>2</sub> CH <sub>2</sub>			
set-Butyl alcohol	CH <sub>3</sub> CH <sub>2</sub> CHOHCH <sub>3</sub>			
Sevin (carbaryl)	C			
Sodium Alizarinsulfonate	SO <sub>3</sub> C <sub>6</sub> H <sub>3</sub> (CO) <sub>2</sub> C <sub>6</sub> H <sub>2</sub> (OH) <sub>2</sub> Na		N,S	
Styrene	C <sub>6</sub> H <sub>5</sub> CH:CH <sub>2</sub>			
Styrene oxide	C <sub>6</sub> H <sub>5</sub> CHOCH <sub>2</sub>			
Terephthalic acid	C <sub>6</sub> H <sub>4</sub> (COOH) <sub>2</sub>			
tert-Butyl alcohol	(CH <sub>3</sub> ) <sub>3</sub> COH			282g
tert-Butylformate	(CH <sub>3</sub> ) <sub>3</sub> COC(O)H			
Tetrachloroethylene	Cl <sub>2</sub> C:CCl <sub>2</sub>	Cl		162,206,282g,392g,463,504g,
Tetrachlorvinphos	C	Cl	P	248
Tetrahydrofuran	CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> O			
Thioacetamide	CH <sub>3</sub> CSNH <sub>2</sub>		S,N	
Thiourea	(NH <sub>2</sub> ) <sub>2</sub> CS		S,N	

Substance	Formula	Halo- g e n	Het. Atom	Reference
Toluene	C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub>			49g,53g,60g,224g,340,404g,
Toluene diisocyanate	CH <sub>3</sub> C <sub>6</sub> H <sub>3</sub> (NCO) <sub>2</sub>		N	
Total xylenes	C <sub>6</sub> H <sub>4</sub> (CH <sub>3</sub> ) <sub>2</sub>			
Toxaphene	C <sub>10</sub> H <sub>10</sub> Cl <sub>8</sub>	Cl		
Triaziquone	C <sub>12</sub> H <sub>13</sub> N <sub>3</sub> O <sub>2</sub>		N	
Trichlorfon	(CH <sub>3</sub> O) <sub>2</sub> P(O)CH(OH)CCl <sub>3</sub>	Cl	P	
Trichloroethylene	CHCl:CCl <sub>2</sub>	Cl		39g,41,42,206,213g,225g,255g 257g,275,276,282g,283,283g, 292,328g,392g,404g,463, 493g,504g,505,5 13
Triethylamine	N(C <sub>2</sub> H <sub>5</sub> ) <sub>3</sub>		N	
Trifluralin	F <sub>3</sub> C(NO <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>2</sub> N(C <sub>3</sub> H <sub>7</sub> ) <sub>2</sub>	F	N	
Trinitrophenylmethyl nitramine	(NO <sub>2</sub> ) <sub>3</sub> C <sub>6</sub> H <sub>2</sub> N(NO <sub>2</sub> )CH <sub>3</sub>		N	
Tris(2,3-dibromopropyl) phosphate	(CH <sub>2</sub> BrCHBrCH <sub>2</sub> O) <sub>3</sub> PO	Br	P	
Urethane (ethyl carbamate)	CO(NH <sub>2</sub> )OC <sub>2</sub> H <sub>5</sub>		N	
Vinyl acetate	CH <sub>3</sub> COOCH:CH <sub>2</sub>			
Vinyl bromide	CH <sub>2</sub> CHBr	Br		
Vinyl chloride	CH <sub>2</sub> :CHCl	Cl		
Vinylidene chloride	CH <sub>2</sub> :CCl <sub>2</sub>	Cl		
Xylene (mixed isomers)	C <sub>6</sub> H <sub>4</sub> (CH <sub>3</sub> ) <sub>2</sub>			49g,322g,496
Zineb	Zn(CS <sub>2</sub> NHCH <sub>2</sub> ) <sub>2</sub>		S,N	

Table 2. Other Organic Compounds Treated by a Photocatalytic Process

Substance	Formula	Halo- g e n	Het. Atom	Reference
1,1,1,2-Tetrachloroethane	Cl <sub>3</sub> CC <sub>2</sub> Cl	Cl		
1,3,5,7-Tetramethylcyclotetrasiloxane			Si	44 1,442,443
1,1,1-Trifluoro-2,2,2-trichloroethane	F <sub>3</sub> CCCL <sub>3</sub>	F,Cl		282g
1,1,1-Trifluorobromocloroethane	C <sub>2</sub> HF <sub>3</sub> ClBr	F		
1,1,3-Trichloropropene	Cl <sub>2</sub> CCHCH <sub>2</sub> Cl	Cl		404g
1,1-Difluoro-1,2,2-trichloroethane	ClF <sub>2</sub> CC <sub>2</sub> Cl <sub>2</sub>	F,Cl		
1,1-Difluoro-1,2-dichloroethane	FC <sub>2</sub> CC <sub>2</sub> Cl <sub>2</sub>	F,Cl		
1,1-Difluoroethylene	CH <sub>2</sub> CF <sub>2</sub>	F		
1,1-Dimethyl-3-phenylurea	C <sub>6</sub> H <sub>5</sub> NHC(O)N(CH <sub>3</sub> ) <sub>2</sub>		N	393
1,1-Diphenylethylene	(C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> C=CH <sub>2</sub>			125
1,2-Dimethoxybenzene	(CH <sub>3</sub> O) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>			8,9,12,373
1,2-Bis(2-chloroethoxy)ethane	(ClC <sub>2</sub> H <sub>4</sub> ) <sub>2</sub> C <sub>2</sub> H <sub>4</sub>	Cl		
1,2-, 1,3-, or 1,4-Dinitrobenzene	(NO <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>		N	87
1,2,4,5-Tetramethylbenzene	(CH <sub>3</sub> ) <sub>4</sub> C <sub>6</sub> H <sub>2</sub>			
1,3-Dimethoxybenzene	(CH <sub>3</sub> O) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>			12
1,3-Diphenylisobenzofuran	(C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>2</sub> O			237
1,4-Dinitrobenzene	(NO <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>		N	296
1,4-Naphthoquinone	C <sub>10</sub> H <sub>6</sub> O <sub>2</sub>			160
1,4-Pentanediole	CH <sub>3</sub> CH(OH)(CH <sub>2</sub> ) <sub>3</sub> OH			133
1-Benzylnicotinamide	(C <sub>6</sub> H <sub>5</sub> )CH <sub>2</sub> (C <sub>5</sub> H <sub>3</sub> N)C(O)NH <sub>2</sub>		N	
1-Bromodecane	BrC <sub>10</sub> H <sub>21</sub>	Br		
1-Bromododecane	BrC <sub>12</sub> H <sub>25</sub>	Br		
1-Butanol	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> OH			101,374
1-Dodecanol	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>11</sub> OH			
1-Hexene	C <sub>6</sub> H <sub>12</sub>			
1-(Methoxyphenyl)-2-propanol	(CH <sub>3</sub> O)C <sub>6</sub> H <sub>4</sub> (CH <sub>3</sub> )CHOH			
1-Phenylethanol	CH <sub>3</sub> CH(C <sub>6</sub> H <sub>5</sub> )OH			485
1-Propanol	n-C <sub>3</sub> H <sub>7</sub> OH			101,374,376,378,380
1-Nitronaphthalene	C <sub>10</sub> H <sub>7</sub> NO <sub>2</sub>		N	296
2,2-Dichloropropionic acid	CH <sub>3</sub> CCl <sub>2</sub> CO <sub>2</sub> H	Cl		59
2,3-, 2,4- or 3,4-Difluorophenol	F <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH	F		
Tris-(2,4-dichlorophenoxy)ethylphosphite	C <sub>2</sub> H <sub>5</sub> P[OC <sub>6</sub> H <sub>3</sub> Cl <sub>2</sub> ] <sub>3</sub>	Cl	P	
2,6-Dichlorophenol	C <sub>6</sub> H <sub>3</sub> Cl <sub>2</sub> OH	Cl		52.315
2,7-Dichlorodibenzo-p-dioxin	Cl <sub>2</sub> C <sub>12</sub> H <sub>6</sub> O <sub>2</sub>	Cl		
2-, 3-, or 4-Chlorobenzoic acid				<b>80</b>
2-, 3-, or 4-Fluorophenol	FC <sub>6</sub> H <sub>4</sub> OH	F		
2-,3-, or 4-Nitrotoluene	NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>		N	87



Substance	Formula	Halo- gen	Het. Atom	Reference
2,3-Benzofuran	C <sub>8</sub> H <sub>6</sub> O			11
2,3- and 2,5-Dichlorophenol	Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH	Cl		82,208,291,315
2,3-Dihydrobenzofuran	C <sub>8</sub> H <sub>8</sub> O			377
2,4-Dimethoxybenzene	(CH <sub>3</sub> O) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>			8,12
2,5-Dinitrophenol	(NO <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH	Cl	N	
2,6-Dichloroindophenol	C <sub>8</sub> H <sub>2</sub> N(OH)Cl <sub>2</sub>			
2,5-Furandimethanol	C <sub>4</sub> H <sub>2</sub> O(CH <sub>2</sub> OH) <sub>2</sub>			
2,4,6-Trinitrobenzoic acid	(NO <sub>2</sub> ) <sub>3</sub> C <sub>6</sub> H <sub>2</sub> CO <sub>2</sub> H		N	408
2-Chloroaniline	ClC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>	Cl	N	
2-Chlorobiphenyl	ClC <sub>6</sub> H <sub>4</sub> C <sub>6</sub> H <sub>5</sub>	Cl		513
2-Chlorodibenzo-p-dioxin	ClC <sub>12</sub> H <sub>7</sub> O <sub>2</sub>	Cl		
2-Chloroethylethylsulfide	(ClC <sub>2</sub> H <sub>4</sub> )C <sub>2</sub> H <sub>5</sub> S	Cl	S	130
2-Chloroethylmethylsulfide	(ClC <sub>2</sub> H <sub>4</sub> )CH <sub>3</sub> S	Cl	S	130
2-Coumaranone	C <sub>8</sub> H <sub>6</sub> O <sub>2</sub>			11
2-Furoic Acid	(CH <sub>2</sub> ) <sub>3</sub> CHOCO <sub>2</sub> H			
2-Hydroxypyridine	HOC <sub>5</sub> H <sub>4</sub> N		N	
2-Hydroxytetrahydropyran	HOC <sub>5</sub> H <sub>9</sub> O			
2-Methyl- 1,4-hydroquinone	CH <sub>3</sub> C <sub>6</sub> H <sub>3</sub> (OH) <sub>2</sub>			100
2-Naphthol	C <sub>10</sub> H <sub>7</sub> OH			160
2-Tolualdehyde	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> CHO			60g
2,2'-Dihydroxybiphenyl	C <sub>12</sub> H <sub>8</sub> (OH) <sub>2</sub>			54
3,3,3-Trifluoropropene	CH <sub>2</sub> CHCF <sub>3</sub>			
3,3'-Dichlorobiphenyl	(ClC <sub>6</sub> H <sub>4</sub> ) <sub>2</sub>	Cl		
3,4,5-Trichlorophenol	Cl <sub>3</sub> C <sub>6</sub> H <sub>2</sub> OH	Cl		82
3,4-Dichlorophenol	3,4-Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH	Cl		82
3-Aminoanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>		N	10
3-Chloroanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> Cl	Cl		10
3-Chlorophenol	m-ClC <sub>6</sub> H <sub>4</sub> OH	Cl		59,81,82
3-Chlorosalicylic acid	C <sub>7</sub> H <sub>5</sub> ClO <sub>3</sub>	Cl		
3-Chlorotoluene	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> Cl	Cl		496
3-Fluoroanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> F	F		10
3-Fluorotoluene	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> F	F		496
3-hydroxyanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> OH			10
3-Methyl-2-oxobutanoic acid	(CH <sub>3</sub> )CHC(O)CO <sub>2</sub> H			486
3-Nitroanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>			10
3-Nitrotoluene	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>	N		496
4-Acetobenzaldehyde	CH <sub>3</sub> C(O)C <sub>6</sub> H <sub>4</sub> CHO			228
4-Aminoanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> NH <sub>2</sub>		N	10
4-Chloroanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> Cl	Cl		10
4-Chlorotoluene	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> Cl	Cl		496
4-Fluoroanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> F	F		10
4-Fluorotoluene	CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> F	F		496
4-Hydroxyanisole	CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> OH			10

Substance	Formula	Halo- gen	Het. Atom	Reference
4-Nitroanisole	CH3OC6H4NO2		N	10
4-Bromophenol	BrC6H4OH	Br		
4-t-Butyltoluene	p-(t-C4H9)C6H4CH3			496
4-Chloro-3-nitro-benzotrifluoride	C6HCl(NO2)F3	F,Cl	N	
4-Chlorophenol	ClC6H4OH	Cl		59,81,82,85,157g,303,358,359, 434,435,478,485,487,498
4-Chlorophenylisocyanate	ClC6H4NCO	Cl	N	
4-Fluorophenol	FC6H4OH	F		
4-Hydroxyacetophenone	HOC6H4C(O)CH3			
4-Hydroxybenzyl Alcohol	p-HO(C6H4)CH2OH			
4-Hydroxybiphenyl	C12H9OH			54
4-Iodophenol	IC6H4OH			
4-Methoxybenzyl(3-phenylpropyl)- ether	CH3OC6H4CH2O(CH2)3C6H5			377
4-Methoxyphenol	CH3C6H4OH			
4-Nitroaniline	NO2C6H4NH2		N	
4-Nitrobenzaldehyde	NO2C6H4CHO		N	228
4-Nitrocatechol	(NO2)C6H3(OH)2		N	
4-Nitrophenylethylphosphinate	(NO2)C6H4(C2H5)PO2		N,P	
4-Nitrophenylisopropylphosphinate	(NO2C6H5)(C3H7)HPO3		N,P	
4-Nitrophenyldiethylphosphate	(NO2C6H5)(C2H5)2PO4		N,P	
4-Nitrotoluene	CH3C6H4NO2		N	296,496
4-Phenylbutylamine	C6H5(CH2)4NH2		N	128
4-Picoline	CH3C5H4N		N	340
4-Thiophenyl- 1 -butanol .	C6H5S(CH2)4OH		S	
4-Trifluoromethylphenol	CF3C6H4OH	F		
4,4'-Dihydroxybiphenyl	C12H8(OH)2			54
5-Bromouracil	BrC4H(NH)2(O)2	Br	N	5
5-Chlorouracil	ClC4H(NH)2(O)2	Cl	N	5
5-Fluorouracil	FC4H(NH)2(O)2	F		5
5-Hydroxypentanoic acid	HO(CH2)4CO2H			
6-Methoxytetralin	CH3OC8H5O			377
5-Methylresorcinol	CH3C6H5O2			382
12-Nitrododecanoic acid	NO2C11H22CO2H		N	296
12-phenyldodecanesulfonate, sodium salt	C6H5(CH2)12SO3H			
Acenaphthene	C10H16(CH2)2			
Acetic Acid	CH3CO2H			90,239,344g,405g
Acetophenone	CH3COC6H5			160,485,498
Acetylene	C2H2			
Acid orange 7	Na,O3SC6H4N2C10H6OH		N,S	477,478,479,481
Adipic acid	C5H11CO2H			
Aldicarb	CH3SC(CH3)2CHN(O)C(O) NHCH3		N,S	

Substance	Formula	Halo- g e n	Het. Atom	Reference
p-Alkylphenol (various)	R(C <sub>6</sub> H <sub>4</sub> )OH			
Allyl alcohol	C <sub>3</sub> H <sub>5</sub> OH			
Alochlor				
p-Aminophenol	NH <sub>2</sub> (C <sub>6</sub> H <sub>4</sub> )OH			
Anthraquinone-Zsulfonic acid	HO <sub>3</sub> SC <sub>14</sub> H <sub>7</sub> O <sub>2</sub>			
AOX or Haloform Precursors				
Asulam				304
Azobenzenes (various)	XC <sub>6</sub> H <sub>4</sub> NNC <sub>6</sub> H <sub>4</sub> X			
Azobisformamidoacetic acid				
Bacteria/Algae/Virus				36,147,167,273,281,305,307, 330g,347,430,473,489
Basagran				256
Benzaldehyde	C <sub>6</sub> H <sub>5</sub> C(O)H			160,228
Benzophenone	(C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> CO			
Benzoquinone	C <sub>6</sub> H <sub>4</sub> O <sub>2</sub>			
Benzyl dodecyldimethylammonium chloride	(C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> )(C <sub>12</sub> H <sub>25</sub> )(CH <sub>3</sub> ) <sub>2</sub> N,Cl	Cl		
Benzyl ethers	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> O-R R=CH <sub>3</sub> , n-C <sub>6</sub> H <sub>13</sub> , t-C <sub>4</sub> H <sub>9</sub> , (CH <sub>2</sub> ) <sub>3</sub> C <sub>6</sub> H <sub>5</sub> , CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>			377
Benzyltetradecyldimethylammonium chloride	(C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> )(C <sub>14</sub> H <sub>27</sub> )(CH <sub>3</sub> ) <sub>2</sub> N,Cl	Cl		
Biomass				
Biphthalate	(C <sub>6</sub> H <sub>4</sub> )(CO <sub>2</sub> H)CO <sub>2</sub> (-1)			
Bromobenzene	BrC <sub>6</sub> H <sub>5</sub>	Br		485
Butane	C <sub>4</sub> H <sub>10</sub>			
But-1-ene	CH <sub>2</sub> CHCH <sub>2</sub> CH <sub>3</sub>			19
But-2-ene, trans	CH <sub>3</sub> CHCHCH <sub>3</sub>			
Butylamine	n-C <sub>4</sub> H <sub>9</sub> NH <sub>2</sub>		N	62
tButylazine			N	415
n-Butyltin chloride	C <sub>4</sub> H <sub>9</sub> SnCl <sub>3</sub>		Sn	338
But-1-yne	CHCCH <sub>2</sub> CH <sub>3</sub>			19
Butyric acid	C <sub>3</sub> H <sub>7</sub> CO <sub>2</sub> H			
Carbetamide			N	361
Carbon dioxide (reduction)	CO <sub>2</sub>			18,20,31,156,271,298,320g, 466,503
Carbon monoxide	CO			175g, 182g, 186g, 188g, 189g, 474
Carbon tetrabromide	CBr <sub>4</sub>	Br		243g, 286g, 287g
Catechol	C <sub>6</sub> H <sub>4</sub> (OH) <sub>2</sub>			57
C 12-Betaine	(C <sub>12</sub> H <sub>25</sub> )(CH <sub>3</sub> ) <sub>2</sub> NCO <sub>2</sub>		N	202,203
C 12-Amidobetaine				202

Substance	Formula	Halo- gen	Het. Atom	Reference
Cetyldimethylbenzylammonium chloride	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>15</sub> (CH <sub>3</sub> ) <sub>2</sub> -(C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> )N,Cl		N	
Cetylpyridinium chloride	N-[CH <sub>3</sub> (CH <sub>2</sub> ) <sub>15</sub> ](C <sub>5</sub> H <sub>5</sub> N),Cl		N	92,334
Cetyltrimethylammonium bromide	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>15</sub> (CH <sub>3</sub> ) <sub>3</sub> N,Cl		N	202
<b>Chloroacetaldehyde</b>	CH <sub>2</sub> ClC(O)H	Cl		
Chlorobenzoic acids, o-, m-, or p-	Cl(C <sub>6</sub> H <sub>4</sub> )CO <sub>2</sub> H	Cl		
<b>Chlorofluorocarbons</b> , various		F,C		
<b>Chloral hydrate</b>	Cl <sub>3</sub> CO(OH) <sub>2</sub>	Cl		
Chloranil, o- and p-	C <sub>6</sub> Cl <sub>4</sub> O <sub>2</sub>	Cl		
<b>Chloroethylammonium chloride</b>	ClH <sub>3</sub> N,Cl	Cl	N	
<b>Chlorpyrifos</b>		Cl	S,N	
Ciba Orange RI				
Citric acid	HO <sub>2</sub> CCH <sub>2</sub> C(OH)(CO <sub>2</sub> H)CH <sub>2</sub> CO <sub>2</sub> H			48,101
Coal or Carbon				
Color and/or COD (in wastewater)				34
Congo Red	C <sub>32</sub> H <sub>22</sub> O <sub>6</sub> N <sub>6</sub> S <sub>2</sub> Na <sub>2</sub>			
Cresol violet	C <sub>16</sub> H <sub>8</sub> NO(NH <sub>2</sub> ),Cl		N	
Creosote phenolics				
<b>Cyanuric acid</b>	C <sub>3</sub> N <sub>3</sub> (OH) <sub>3</sub>		N	
Cyclododecanol	C <sub>12</sub> H <sub>23</sub> OH			
Cyclohexanedicarboxylic acids	C <sub>6</sub> H <sub>10</sub> (CO <sub>2</sub> H) <sub>2</sub>			
Cyclohexanol	C <sub>6</sub> H <sub>11</sub> OH			323
Cyclohexene	C <sub>6</sub> H <sub>10</sub>			
Cyclohexene oxide	C <sub>6</sub> H <sub>10</sub> O			
Cyclooctatetraene	C <sub>8</sub> H <sub>8</sub>			135
Cyclopentane	C <sub>5</sub> H <sub>10</sub>			73g,74g,370g
Cyclophosphamide	OPONHC <sub>3</sub> H <sub>6</sub> [N(C <sub>2</sub> H <sub>4</sub> Cl) <sub>2</sub> ]		P,N	
Cinnamyl alcohol	C <sub>6</sub> H <sub>4</sub> C <sub>2</sub> H <sub>2</sub> OH			
DDT	(ClC <sub>6</sub> H <sub>4</sub> ) <sub>2</sub> CHCCl <sub>3</sub>	Cl		333
<b>Decalin</b>	C <sub>10</sub> H <sub>18</sub>			
Decamethyltetrasiloxane	(CH <sub>3</sub> ) <sub>10</sub> Si <sub>4</sub> O <sub>3</sub>		Si	
Decanoic acid	C <sub>9</sub> H <sub>19</sub> CO <sub>2</sub> H			
Decanol	HOC <sub>10</sub> H <sub>21</sub>			
Desipramine	(C <sub>6</sub> H <sub>4</sub> ) <sub>2</sub> (CH <sub>2</sub> ) <sub>2</sub> N(CH <sub>2</sub> ) <sub>3</sub> NHCH <sub>3</sub>			
Diazinon	[(CH <sub>3</sub> ) <sub>2</sub> CHC <sub>4</sub> N <sub>2</sub> H(CH <sub>3</sub> )O]-PS(OC <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>		N,P,S	304
Dibenzo-p-dioxines, various	X <sub>n</sub> C <sub>12</sub> H <sub>(8-n)</sub> O <sub>2</sub>	Cl		516
Dibromomethane	CH <sub>2</sub> Br <sub>2</sub>	Br		
Dimethylacetamide	CH <sub>3</sub> C(O)N(CH <sub>3</sub> ) <sub>2</sub>		N	293
Dichloroacetic acid	Cl <sub>2</sub> CHCO <sub>2</sub> H	Cl		51,253,285,345,358
Dichloroacetyl Chloride	Cl <sub>2</sub> CHCOCl	Cl		225g,328g
Dimethylamine	(CH <sub>3</sub> ) <sub>2</sub> NH		N	
Dimethylphenols (Xylenols)	(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub> OH			465

Substance	Formula	Halo- gen	Het. Atom	Reference
Dimethylsulfide	CH <sub>3</sub> ) <sub>2</sub> S		S	
Dimethyl-2,2-dichlorovinyl phosphate	CH <sub>3</sub> ) <sub>2</sub> (Cl <sub>2</sub> CCH)PO <sub>4</sub>	Cl	P	
Diphenylacetylene	C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> C <sub>2</sub>			207
Diphenylmethane	C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> CH <sub>2</sub>			
Diphenylsulfide	C <sub>6</sub> H <sub>5</sub> ) <sub>2</sub> S		S	
Direct blue 1	(Na <sub>2</sub> O <sub>3</sub> S) <sub>2</sub> C <sub>16</sub> H <sub>6</sub> (NH <sub>2</sub> )(OH)- (OCH <sub>3</sub> ) <sub>2</sub> N <sub>2</sub> ) <sub>2</sub>		N,S	
Diquat				256
Disinfection by-products				394,514
Disperse red 74				509
Diuron	C <sub>6</sub> H <sub>3</sub> Cl <sub>2</sub> NHC(O)N(CH <sub>3</sub> ) <sub>2</sub>		N	256
Dodecane	C <sub>12</sub> H <sub>26</sub>			
Dodecyl sulfate	(C <sub>12</sub> H <sub>25</sub> ) <sub>2</sub> SO <sub>4</sub>			
Dodecylbenzenesulfonate	(C <sub>12</sub> H <sub>25</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>4</sub> SO <sub>3</sub> (-1)		S	92,202,334
Dodecyldecaoxyethylenephosphates			P	
Dodecylpyridinium chloride	(C <sub>12</sub> H <sub>25</sub> ) <sub>2</sub> C <sub>5</sub> H <sub>5</sub> NH <sub>2</sub> Cl	Cl		12,203
Doxycycline				
Dyes, unidentified				96,226,227
Eosin				
Ethambutol			N	
<b>Ethane</b>	<b>C<sub>2</sub>H<sub>6</sub></b>			
Ethanol	C <sub>2</sub> H <sub>5</sub> OH			2,101,228,264,274g,329g, 74,405g
Ethylacetate	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>			
Ethylenediaminetetraacetic acid	C <sub>10</sub> H <sub>16</sub> N <sub>2</sub> O <sub>8</sub>			8,101,342,484
2-, 3-, or 4-Ethylphenol	C <sub>9</sub> H <sub>10</sub> O			
Fenitrothion	C <sub>9</sub> H <sub>12</sub> NO <sub>3</sub> S		S	49
Fluorescein	C <sub>20</sub> H <sub>12</sub> O <sub>5</sub>			
Folicur			N	
Formamide	CH <sub>3</sub> (O)NH <sub>2</sub>		N	93
Formic Acid	CH <sub>2</sub> O <sub>2</sub> H			18,90,101,163,214,251,252, 144g,416,444
Fullerenes	C <sub>60</sub> , C <sub>70</sub> , and C <sub>84</sub>			36
Fulvic acid				
Glucose	C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>			18
Glycerol	C <sub>3</sub> H <sub>8</sub> (OH) <sub>3</sub>			
HCFC or HFC		Cl,F		
Heparin				
Heptanal	C <sub>7</sub> H <sub>14</sub> CHO			228
Hexafluorobenzene	C <sub>6</sub> F <sub>6</sub>	F		
Hexafluoropropene	C <sub>3</sub> F <sub>2</sub> CF <sub>3</sub>	F		
HMX	(CH <sub>2</sub> ) <sub>4</sub> (NNO <sub>2</sub> ) <sub>4</sub>		N	48
Hexanol	C <sub>6</sub> H <sub>14</sub> OH			

Substance	Formula	Halo- gen	Het. Atom	Reference
<b>Humic Acids</b>				34,37,480
Hydroxybenzoic acid (various)	HOC6H4(OH)CO2H			
Hydroxycarboxylic acids, alpha	RCH(OH)CO2H			
Hydroxyethylcellulose				
Imidazole	C3H4N2		N	346
<b>Indole</b>	C8H6NH		N	
Isobutane	C4H10			88g
Isobutanol	CH3CH(CH3)CH2OH			
Isobutene	C4H8			
Isobutyric Acid	CH3CH(CH3)CO2H			101
Isonicotinaldehyde	C5H4NCHO		N	228
Isoprene	CH2C(CH3)CHCH2			
Isosorbide dinitrate	C6O2H8(ONO2)2			
<b>Iso-octane</b>	(CH3)2CH(CH2)4CH3			282g
L-Lysine	NH2(CH2)4CH(NH2)CO2H			101
Lactic acid	C3H6O3			
Landfill leachate				35
Kraft lignin				
Malachite green oxalate	(C6H5)[C6H4N(CH3)2][C6H4N-(CH3)2]C.(C2O4H)2.H2C2O4		N	153,154
<b>Maleic acid</b>	HO2CCHCHCO2H			
<b>Malic acid</b>	HO2CCH2CH(OH)CO2H			97
<b>Malonic acid</b>	CH2(CO2H)2			219
Methane	CH4			
Methanethiol	CH3SH		S	450g
Methylcyclohexane	CH3C6H			
Methyl orange	Na.O3SC6H4N2C6H4N(CH3)2		N,S	
alpha-Methylstyrene	C6H5(CH3)CCH2			396
Methyl viologen	(CH3C5H4N)2.Cl2	Cl	N	
Methylene blue	(CH3)2NC6H3NSC6H3N(CH3)2.Cl	Cl	N,S	26,272
Methylvinylketone	CH3COC2H3			
Mixed waste				418
Molasses				402
Monocrotophos			P	428,428,517
Monuron	ClC6H4NHCON(CH3)2	Cl	N	27
Morpholine	C4H8ONH		N	90
m-Phenoxytoluene	m-C6H5O-C6H4CH3			
N-Benzylidiphenylamine	(C6H5CH2)(C6H5)2N		N	108,129
N-Dodecanoyl-N-(2-hydroxyethyl)amide	CH3(CH2)10C(O)NH(CH2)2OH		N	202,203
N-Dodecanoyl-N,N-bis(2-hydroxyethyl)-amide	CH3(CH2)10C(O)N[(CH2)2OH]2		N	203
N-Hydroxysuccinimide	C2H4(C(O))2NOH		N	346
N,N,N',N'-Tetraethyloxonine			N	



Substance	Formula	Halo- gen	Het. Atom	Reference
Naphthol	C10H7OH			
Naphthol blue black	(C6H5N2)C12H2(OH)(NH2)- (SO3Na)2		N,S	337,475,477
Nicotine	C5H4NC4H7NCH3		N	305
Nile Blue A	C16NO(NH2)N(C2H5)2,SO4		N,S	
Nitrocellulose			N	
p-Nitrotoluenesulfonic acid	(CH3)(NO2)C6H3SO3H		N,S	
Nitrotoluene, various	NO2C6H4CH3		N	
Nonylphenoethoxylate	C9H17C6H4OC2H5			
n-Octanol	C8H17OH			436,502
octanal	C7H15CHO			228
Octaphenylcyclotetrasiloxane	(C6H5)8(SiO2)4		Si	314
Oil/Petroleum				170
Oxalic acid	C2O4H2			101,159,200,342,358,423
PCB - polyhydroxy		Cl		315
Pendimethalin				195
Pentaethyleneglycol n-dodecyl ether	(HO)(CH2CH2O)5(C12H23)			202
Pentafluorophenol	C6F5OH	F		
n-Pentyl amine	n-C5H 11 NH2		N	
Permethrin				
Pesticides - unspecified				221
Phenacetylthioether	PhCH2SCH2C(C6H5)CH2		S	138
Phenosafrafin	C6H5N2C12H4(NH2)(CH3)2		N	
p-Phenylenediamine	H2NC6H4NH2		N	288,289,291
Phenylmercaptotetrazole				484
Phorate	(C2H5O)P(S)SCH2SC2H5		P,S	427,428,5 17
Phthaldialdehyde	C6H4(CHO)2			160
Phthalan				377
Phthalic acid	C6H4(CO2H)2			160
Picoline	CH3C5H4N		N	
Piperidene	C5H10NH		N	52
Polyethoxylene alkyl ethers	R2(OC2H4)n			
Polyethylene	(CH2CH2)n			
Poly(methylphenylsiloxane)	[(C6H5)(CH3)SiO]n		Si	314
Polypropylene	[(CH3)CHCH2]n			
Polyvinylalcohol	(C2H3OH)n			
Polyvinylchloride	(CH2CHCl)n	Cl		152
<b>Proline</b>	C4H8NCO2H		N	
Prometon				
Prometryn				
Propane	C3H8			194
Propionamide	CH3CH2C(O)NH2		N	293
Propionic acid	C2H5CO2H			101

Substance	Formula	Halo- gen	Het. Atom	Reference
Propylene	C3H6			375g
Propylene glycol dinitrate	CH3CH(NO3)CH2(NO3)		N	
Propyne	CH3CCH			
Propylamide			N	471
Pyridine	C5H5N		N	62,224g
Pyrocatechol	o-C6H4(OH)2			
Pyrrole	C4H5N		N	
Quinoline acid yellow				260
Reactive Dyes				
Red Dye 79			N,S	
Resorcinol	C6H6O2			382
Rhodamine B	CH3OC(O)(C6H4)C13H6O(NH2)2		N	
Rhodamine 6G	C2H5OC(O)C6H4C13H4(CH3)2-[N(C2H5)]2,Cl	Cl	N	14
Rhodamine 6ZH				
Rose Bengal	Na2,O2CC6Cl4C13H2O14O2	Cl,I		
S-Dodecyl thioether carboxylates			S	
S-Ethyl-N,N-dipropyl thiocarbamate (EPTC)	(C2H5)SC(O)N(C3H7)2		N,S	
S-Ethyl-N,N-diisopropylthiocarbamate (Butylate)	(C2H5)SC(O)N(i-C3H7)2		N,S	
S-Ethyl-4-hexahydro-1-H-azepine-1-carbothionate (molinate)			N,S	
S-Propyl-N-cyclohexyl thiocarbamate (cycloate)	(C3H7)SC(O)NH(C6H11)		N,S	
S-Propyl-N,N-dipropyl thiocarbamate (vernolate)	(C3H7)SC(O)(NC3H7)2		N,S	
Salicylaldehyde	C7H6O2			
Salicylic acid	C7H6O3			48,91,102,246,469
Simazine	(C2H5)Cl(NHC2H5)C3N3		N	
Sodium chloroacetate	CH3CO2Na	Cl		
Sodium dodecylbenzene sulfonate	C12H25C6H4SO3Na		S	
Stearic acid	CH3(CH2)16CO2H			429
S tilbene	C6H5CHCHC6H5			131
Succinic acid	HO2CCH2CH2CO2H			
Sucrose	C12H22O11			
Sulfones	RS(O)2R'		S	
Surfactants - unspecified				295
Tetrachlorvinphos	CHClCH(2,4,5-Cl3C6H2)(CH3)PO4	Cl	P	
Tetrafluoroethylene	C2F4	F		
Tetralin	C10H12			377
Tetramethylenediamine	NH2(CH2)4NH2		N	
Tetrabutylammonium phosphate	[(n-C4H9)4N]4,PO4		P	



Substance	Formula	Halo- gen	Het. Atom	Reference
Tetradecyldimethylbenzylammonium- chloride	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>13</sub> (CH <sub>3</sub> ) <sub>2</sub> (C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> )- NCl		N	202, 203
Theophylline	C <sub>7</sub> H <sub>8</sub> N <sub>4</sub> O <sub>2</sub> .H <sub>2</sub> O		N	
Thioethers	RSR'		S	
Thiobencarb				
Thymine	C <sub>5</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub>		N	
Thionine	C <sub>12</sub> NS(NH <sub>2</sub> ),O <sub>2</sub> CCCH <sub>3</sub>		S,N	
p-Toluenesulfonic acid	CH <sub>3</sub> (C <sub>6</sub> H <sub>4</sub> )SO <sub>3</sub> H		S	
s-Triazines			N	38
Trichloroacetic acid	Cl <sub>3</sub> CCO <sub>2</sub> H	Cl		433
Trietazine			N	
Triethanolamine	N(CH <sub>2</sub> CH <sub>2</sub> OH) <sub>3</sub>		N	
Trifluoroacetic acid	CF <sub>3</sub> CO <sub>2</sub> H	F		
Trihydrazinotriazine			N	
Trihydroxybenzene	(HO) <sub>3</sub> C <sub>6</sub> H <sub>3</sub>			
Trimethylamine	(CH <sub>3</sub> ) <sub>3</sub> N		N	
Trinitrophenol	(NO <sub>2</sub> ) <sub>3</sub> C <sub>6</sub> H <sub>2</sub> OH			
Triphenylacetic acid	(C <sub>6</sub> H <sub>5</sub> ) <sub>3</sub> CCO <sub>2</sub> H			
Triphenylbismuthine	(C <sub>6</sub> H <sub>5</sub> ) <sub>3</sub> Bi		Bi	431
Triphenylphosphine	(C <sub>6</sub> H <sub>5</sub> ) <sub>3</sub> P		P	431
Triphenylstibine	(C <sub>6</sub> H <sub>5</sub> ) <sub>3</sub> Sb		Sb	431
Tri-(p-Tolyl)arsine	(CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> ) <sub>3</sub> As		As	431
Triton X- 100	C <sub>8</sub> H <sub>17</sub> C <sub>6</sub> H <sub>4</sub> (OCH <sub>2</sub> CH <sub>2</sub> ) <sub>x</sub> OH			92, 334
Umbelliferone	C <sub>9</sub> H <sub>6</sub> O <sub>3</sub>			
Uracil	HNC(O)NHC(O)CHCH		N	5
Urea	C(O)(NH <sub>2</sub> ) <sub>2</sub>		N	

Table 3. Inorganic Substances Included in EPA Lists of Hazardous Substances and/or Treated by a Photocatalytic Process

Substance/Element	Formula/Symbol	Reference
Actinides	Th,Pa,U,Np,Pu	18
Aluminum (fume or dust)	Al	
Aluminum oxide	Al <sub>2</sub> O <sub>3</sub>	
Ammonia	NH <sub>3</sub>	161g,321g,472g
Ammonium nitrate (soln)	NH <sub>4</sub> NO <sub>3</sub>	
Ammonium sulfate (soln)	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	
Antimony	Sb	
Arsenic	As	
Asbestos	Mg,Si	
Azide ion	N <sub>3</sub> (-)	
Barium	Ba	
Beryllium	Be	
Bismuth	Bi	
Boron	B	
Cadmium	Cd	
Chlorine	Cl	
Chlorine dioxide	Cl <sub>2</sub> O <sub>2</sub>	
Chromium	Cr	154
Cobalt	Co	
Copper	Cu	101,197,198
Cyanide and Complexes	CN(- 1) and M(CN) <sub>x</sub>	2,3,218,384,385
Cyanate ion	CNO(- 1)	
Gold	Au	197
Halide ion	X( 1-), X = F, Cl, Br, or I	33,137,484
Hydrazine	H <sub>2</sub> NNH <sub>2</sub>	
Hydrogen sulfide	H <sub>2</sub> S	
Hypophosphorus acid	H <sub>2</sub> PO <sub>2</sub>	48
Iridium	Ir	197
Iron	Fe	
Lead	Pb	387
Manganese	Mn	
Mercury	Hg	
Molybdenum	Mo	
Nickel	Ni	197
Nitrates/nitrites	NO <sub>3</sub> (-1),NO <sub>2</sub> (-1)	388

Substance	Formula	Reference
Nitrogen oxides	NOx	18,25g,143g,216g,220g. 243g,244g,261g,262g. 449g,459g,460g,461g. 472,501,512
Nitrogen	N <sub>2</sub>	
Oxalate ion	C <sub>2</sub> O <sub>4</sub> ( <sup>2-</sup> )	
Oxygen	O <sub>2</sub>	
Ozone	O <sub>3</sub>	
Palladium	Pd	197,199,264
Phosphorus	P	
Platinum	Pt	98,197,199,264,444
Radium	Ra	
Radon	Rn	
Rhodium	Rh	98,197,199,264
Selenium	Se	
Silicon	Si	
Silver	Ag	197,198,199,211,290
Strontium	Sr	
Sulfate radical	SO <sub>4</sub> ( <sup>1-</sup> )	
Sulfite	SO <sub>3</sub> ( <sup>1-</sup> )	
Sulfur	S	
Sulfur oxides	SO <sub>x</sub>	449g
Sulfuric acid	H <sub>2</sub> SO <sub>4</sub>	
Thallium	Tl	
Thiocyanate	SCN( <sup>1-</sup> )	33
Thiosulfate	S <sub>2</sub> O <sub>3</sub> ( <sup>2-</sup> )	48,198,484
Thorium	Th	
Tin	Sn	
Tritium	H.(T)	
Tungsten	W	
Vanadium	V	
Zinc	Zn	

## 4.0 Conclusions

The level of activity in this field remains high. The potential to develop new technology for environmental remediation is still a major driving force for R&D activity. However, it is clear that many companies see potential markets for VOC removal in indoor air quality applications. The application of photocatalytic chemistry for the disinfection of water and air has been the subject of sporadic reports over the last 9 years. The potential impact of this application is very large and it is drawing increased attention. In the work to date no clear mechanism for the killing of micro-organisms has been established. The magnitude of the contribution of a photocatalytic effect versus the effect of heat, light, or extraneous agents such as hydrogen peroxide or transition metal ions is often difficult to determine.

In research, more attention is being paid to the detection and identification of intermediates and by-products that can be formed during the photocatalytic process, both in aqueous and gas phase systems. Defining intermediates helps researchers understand the chemical mechanisms of the processes and is necessary to insure that potentially harmful substances are not left in the processed stream. Key areas of work identified in the last report continue to be important. Few studies include mass balances for the reactions and kinetic models that can be used to size treatment systems are still rare. As systems are deployed in the field, it is increasingly important that the issues of catalyst lifetime and regeneration be addressed. Related to this is the need to identify those components of an air or water stream that can inhibit or kill activity. All of these are important to the design of efficient and economical treatment systems. Questions concerning the economic viability of photocatalytic processes are being raised more often. The significance of simple mineralization of one more organic compound without regard to the amount of energy or time that it took is questioned by those who want to see the process efficiency improved. This may be taken as a sign that the field is maturing. There are entrepreneurs and companies that would like to capitalize on the scientific foundation that has been developed for photocatalytic chemistry.

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