

In cooperation with NEW YORK STATE DEPARTMENT OF ENVIRONMENTAL CONSERVATION

Ground-Water Quality in the Mohawk River Basin, New York, 2006

By Elizabeth A. Nystrom

Open-File Report 2008-1086

U.S. Department of the Interior

U.S. Geological Survey





In cooperation with NEW YORK STATE DEPARTMENT OF ENVIRONMENTAL CONSERVATION

Ground-Water Quality in the Mohawk River Basin, New York, 2006

By Elizabeth A. Nystrom

Open-File Report 2008-1086

U.S. Department of the Interior

U.S. Geological Survey

U.S. Department of the Interior DIRK KEMPTHORNE, Secretary

U.S. Geological Survey

Mark D. Myers, Director

U.S. Geological Survey, Reston, Virginia 2008

For more information on the USGS—the Federal source for science about the Earth, its natural and living resources, natural hazards, and the environment:

World Wide Web: http://www.usgs.gov
Telephone: 1-888-ASK-USGS

Suggested citation:

Nystrom, E.A., 2008, Ground-water quality in the Mohawk River Basin, New York, 2006: U.S. Geological Survey Open-File Report 2008-1086, 33 p., online only.

Any use of trade, product, or firm names is for descriptive purposes only and does not imply endorsement by the U.S. Government.

Contents

Abstract	1
Introduction	2
Purpose and Scope	2
Hydrogeologic Setting	2
Methods of Investigation	4
Site Selection	8
Sampling Methods	9
Analytical Methods	10
Ground-Water Quality	11
Physical Properties	11
Major lons	11
Nutrients and Organic Carbon	
Trace Elements and Radionuclides	13
Pesticides	
Volatile Organic Compounds and Phenolic Compounds	
Bacteria	15
Summary	16
References Cited	
Appendix.	21
Figure 1. Hydrologic and geographic features and location of the Mohawk River Basin in centra Figure 2. Topography, physiographic regions, and locations of wells sampled in the Mohawk River Figure 3. London and Locations of wells sampled in the Mohawk River	3 ver Basin5
Figure 3. Land cover and locations of wells sampled in the Mohawk River Basin	
Figure 4. Generalized bedrock geology and locations of wells sampled in the Mohawk River Bas	31N <i>1</i>
Tables	
Table 1. Drinking water standards and summary statistics for concentrations of major ions in gr	
samples from the Mohawk River Basin, 2006.	
Table 2. Drinking water standards and summary statistics for concentrations of nutrients in groups and a frage the Mahasuk Bissay Basin, 2006	
samples from the Mohawk River Basin, 2006.	
Table 3. Drinking water standards and summary statistics for concentrations of trace elements	
222 in ground-water samples from the Mohawk River Basin, 2006.	
Table A1. Information on wells sampled in the Mohawk River Basin, 2006	
Table A2. Constituents analyzed for but not detected in ground-water samples from the Mohaw	
Table A3. Physical properties of ground-water samples from the Mohawk River Basin, 2006	
Table A3. Physical properties of ground-water samples from the Mohawk River Basin, 2000	
, אווסוט אידי סטוויסוונו מווועונס ובאידי סטווער וווען וווען וווען וווען וווען אווער אידי סטווער אידי סטווער אידי אידי אידי אידי אידי אידי אידי איד	, ೭೦೦೦20

Table A5. Concentrations of nutrients and organic carbon in ground-water samples from the M	ohawk River
Basin, 2006	27
Table A6. Concentrations of trace elements and radionuclides in ground-water samples from th	ne Mohawk
River Basin, 2006	28
Table A7. Concentrations of pesticides and caffeine detected in ground-water samples from the	e Mohawk
River Basin, 2006	31
Table A8. Concentrations of volatile organic compounds and phenolic compounds detected in q	ground-
vater samples from the Mohawk River Basin, 2006	32
Table A9. Bacteria in ground-water samples from the Mohawk River Basin, 2006	33

Conversion Factors, Datum, Abbreviated Water-Quality Units, and Acronyms

Multiply	Ву	To obtain
	Length	
centimeter (cm)	0.3937	inch (in.)
foot (ft)	0.3048	meter (m)
	Area	
acre	0.004047	square kilometer (km²)
square mile (mi²)	2.590	square kilometer (km²)
	Volume	
liter (L)	0.2642	gallon (gal)
gallon (gal)	3.785	liter (L)
	Flow rate	
gallon per minute (gal/min)	0.06309	liter per second (L/s)
inch per year (in/yr)	25.4	millimeter per year (mm/yr)
	Radioactivity	
picocurie per liter (pCi/L)	0.037	becquerel per liter (Bq/L)
<u> </u>		

Temperature in degrees Celsius (°C) may be converted to degrees Fahrenheit (°F) as follows: $^{\circ}F=(1.8\times^{\circ}C)+32$

Vertical coordinate information is referenced to the North American Vertical Datum of 1988 (NAVD 88). Horizontal coordinate information is referenced to the North American Datum of 1983 (NAD 83). Altitude, as used in this report, refers to distance above the vertical datum.

Abbreviated water-quality units used in this report:

micrograms per liter (µg/L) milligrams per liter (mg/L) microsiemens per centimeter at 25 degrees Celsius (µS/cm at 25 °C) platinum-cobalt units (Pt-Co units)

Acronyms used in this report

AMCL Alternative maximum contaminant level

CFU Colony-forming units

cICP-MS Collision/reaction cell inductively coupled plasma-mass spectrometry

ESA Ethanesulfonic acid

GC-MS Gas chromatography-mass spectrometry

GPS Global positioning system
GWSI Ground-Water Site-Inventory

HPLC-MS High-performance liquid chromatography-mass spectrometry ICP-AES Inductively coupled plasma-atomic emission spectrometry

ICP-MS Inductively coupled plasma-mass spectrometry

ICP-OES Inductively coupled plasma-optical emission spectrometry

LC-MS Liquid chromatography-mass spectrometry

MCL Maximum contaminant level MTBE Methyl *tert*-butyl ether

NWQL USGS National Water Quality Laboratory

NYSDEC New York State Department of Environmental Conservation

NYSDOH New York State Department of Health

OGRL USGS Organic Geochemistry Research Laboratory

PERC Tetrachloroethene

POC Principal organic contaminant

PVC Polyvinyl chloride

SDWS Secondary drinking water standards

THM Trihalomethane
TTHMs Total trihalomethanes

USEPA U.S. Environmental Protection Agency

USGS U.S. Geological Survey VOC Volatile organic compound

Ground-Water Quality in the Mohawk River Basin, New York, 2006

By Elizabeth A. Nystrom

Abstract

Water samples were collected from 27 wells from August through November 2006 to characterize ground-water quality in the Mohawk River Basin. The Mohawk River Basin covers 3,500 square miles in central New York; most of the basin is underlain by sedimentary bedrock, including shale, sandstone, and carbonates. Sand and gravel form the most productive aquifers in the basin. Samples were collected from 13 sand and gravel wells and 14 bedrock wells, including production and domestic wells. The samples were collected and processed through standard U.S. Geological Survey procedures and were analyzed for 226 physical properties and constituents, including physical properties, major ions, nutrients, trace elements, radon-222, pesticides, volatile organic compounds, and bacteria.

Many constituents were not detected in any sample, but concentrations of some constituents exceeded current or proposed Federal or New York State drinking-water quality standards, including color (1 sample), pH (2 samples), sodium (11 samples), chloride (2 samples), fluoride (1 sample), sulfate (1 sample), aluminum (2 samples), arsenic (2 samples), iron (10 samples), manganese (10 samples), radon-222 (12 samples), and bacteria (6 samples). Dissolved oxygen concentrations were greater in samples from sand and gravel wells (median 5.6 milligrams per liter [mg/L]) than from bedrock wells (median 0.2 mg/L). The pH was typically neutral or slightly basic (median 7.3); the median water temperature was 11°C. The ions with the highest concentrations were bicarbonate (median 276 mg/L), calcium (median 58.9 mg/L), and sodium (median 41.9 mg/L). Ground water in the basin is generally very hard (180 mg/L as CaCO₃ or greater), especially in the Mohawk Valley and areas with carbonate bedrock. Nitrate-plus-nitrite concentrations were generally higher samples from sand and gravel wells (median concentration 0.28 mg/L as N) than in samples from bedrock wells (median < 0.06 mg/L as N), although no concentrations exceeded established State or Federal drinkingwater standards of 10 mg/L as N for nitrate and 1 mg/L as N for nitrite. Ammonia concentrations were higher in samples from bedrock wells (median 0.349 mg/L as N) than in those from samples from sand and gravel wells (median 0.006 mg/L as N). The trace elements with the highest concentrations were strontium (median 549 micrograms per liter [µg/L]), iron (median 143 µg/L), boron (median 35 µg/L), and manganese (median 31.1 µg/L). Concentrations of several trace elements, including boron, copper, iron, manganese, and strontium, were higher in samples from bedrock wells than those from sand and gravel wells. The highest radon-222 activities were in samples from bedrock wells (maximum 1,360 pCi/L); 44 percent of all samples exceeded a proposed U.S. Environmental Protection Agency drinking water standard of 300 pCi/L. Nine pesticides and pesticide degradates were detected in six samples at concentrations of 0.42 µg/L or less; all were herbicides or their degradates, and most were degradates of alachlor, atrazine, and metolachlor. Six volatile organic compounds were detected in four samples at concentrations of 0.8 µg/L or less, including four trihalomethanes, tetrachloroethene, and toluene; most

detections were in sand and gravel wells and none of the concentrations exceeded drinking water standards. Coliform bacteria were detected in six samples but fecal coliform bacteria, including *Escherichia coli*, were not detected in any sample.

Introduction

The Federal Clean Water Act Amendments of 1977 require that States monitor and report biennially on the chemical quality of surface water and ground water within their boundaries (U.S. Environmental Protection Agency, 1997, Section 305(b)). In 2002, the U.S. Geological Survey (USGS), in cooperation with the New York State Department of Environmental Conservation (NYSDEC), developed a program to evaluate ground-water quality throughout the major river basins in New York State on a rotating basis. The work parallels the NYSDEC Rotating Intensive Basin Study program, which evaluates surface-water quality in two or three of the 14 major river basins in the State each year. The ground-water quality program began in 2002 with a pilot study in the Mohawk River Basin (Butch and others, 2003). Sampling was completed in the Chemung River Basin in 2003 (Hetcher-Aguila, 2005); the Lake Champlain and Susquehanna River Basins in 2004 (Nystrom, 2006; Hetcher-Aguila and Eckhardt, 2006); and the St. Lawrence (Nystrom, 2007a), Delaware (Nystrom, 2007b), and Genesee River Basins (Eckhardt and others, 2007) in 2005. In 2006, ground water was sampled in the Mohawk River Basin, Niagara River Basin, Allegheny River Basin, Lake Erie tributaries, and western Lake Ontario tributaries. The Mohawk River Basin study, the subject of this report, entailed collection of 13 samples from surficial deposits and 14 samples bedrock, from August through November of 2006.

Purpose and Scope

This report presents the results of the 2006 ground-water quality study in the Mohawk River Basin. It (1) describes the methods of site selection, sample collection, and chemical analysis, and (2) presents the analytical results for physical properties, major ions, nutrients, trace elements and radionuclides, pesticides, volatile organic compounds (VOCs), and bacteria. Information about the sampled wells and results of the analyses are presented in tables A1 through A9, by constituent type, at the end of the report.

Hydrogeologic Setting

The Mohawk River Basin covers about 3,500 mi² in central New York State (fig. 1). It encompasses parts of 14 counties, including all of Montgomery County, most of Schoharie and Schenectady Counties, parts of Herkimer, Hamilton, Fulton, Greene, Oneida, Saratoga, and Albany Counties, and small parts of Lewis, Madison, Otsego, and Delaware Counties. The Mohawk River is a major tributary to the Hudson River; the major tributaries to the Mohawk River are the Schoharie and West Canada Creeks (fig. 1). The Erie Canal was built along the Mohawk River as a major corridor for cargo transport; it was begun in 1817 and completed in 1825. Modernizations moving much of the canal into the river channel were completed in 1918 and the canal was renamed the New York State Barge Canal; currently the canal is used mostly for recreation. The Mohawk River Basin contains three major reservoirs (fig. 1): the Schoharie Reservoir, which diverts water out of the basin as part of New York City's water-supply system; Hinckley Reservoir, which provides drinking water to the Utica area, and Delta Reservoir.

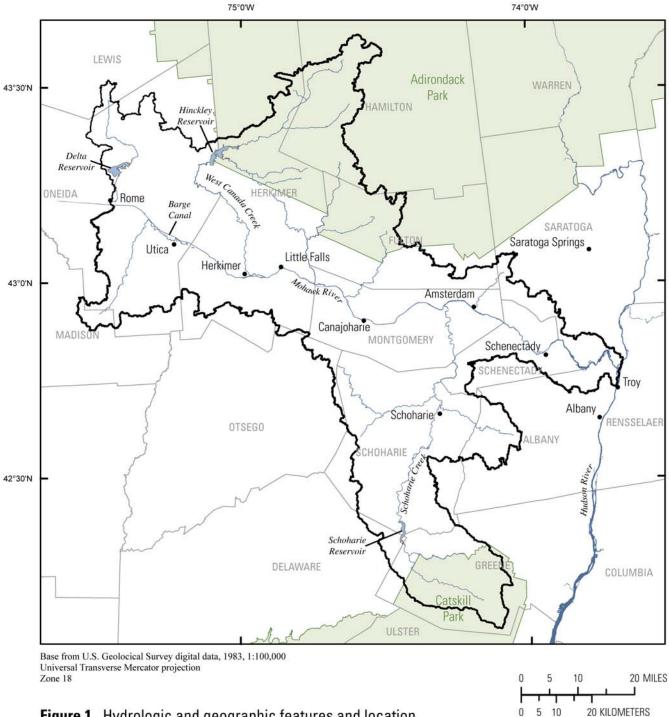


Figure 1. Hydrologic and geographic features and location of the Mohawk River Basin in Central New York.



The Mohawk River Basin contains three main physiographic regions (fig. 2); these are (1) the Mohawk Valley, which extends eastward from the Great Lakes Lowland to the Hudson Valley through the center of the basin, (2) the Adirondack Mountain Upland, which spans the northern part of the basin, and (3) the Appalachian Upland, which spans the southern portion of the basin and includes parts of the Catskill Mountains. The highest elevations in the northern part of the (about 3,600 ft above sea level) are in the Adirondack Mountains Upland and the highest elevations in the southern end of the Basin are in the Catskill Mountain area (about 4,000 ft). The areas with the highest elevations also receive the most precipitation; the northern, northwestern, and southern parts of the basin receive more than 50 in/yr (Randall, 1996). Land use (fig. 3) also corresponds to the physiography of the basin; agriculture is concentrated along the Mohawk River and the northern part of Schoharie Creek, and the largest urban centers in the basin, including Utica, Amsterdam, and Schenectady (fig. 1), are within the Mohawk Valley; smaller urban areas in the Appalachian Upland to the south are present in the valleys of the Catskills region. The upland areas of the basin are predominantly forested (Vogelmann and others, 2001). Parts of the Adirondack and Catskill State Parks lie within the basin. The Adirondack State Park, created in 1892, contains about 6 million acres; about 6 percent of which is in the Mohawk River Basin. The Catskill Park was created in 1904 and encompasses about 700,000 acres, about 15 percent of which is in the basin.

Most of the Mohawk River Basin is underlain by sedimentary bedrock (fig. 4). The Mohawk Valley is underlain by shale, sandstone, and carbonate rocks of Upper to Middle Ordovician age (Isachsen and others, 2000). Outcrops of Cambrian rock are present at isolated locations; for example, at Little Falls (fig. 1). The southern part of the basin is underlain by interlayered sandstone and shale of Middle to Upper Devonian age. The northern part of the Basin is underlain by crystalline metamorphic bedrock of Precambrian age consisting mainly of gneisses. Of the bedrock aquifers in the basin, carbonate rocks generally produce the highest yields, and the crystalline bedrock generally has the lowest; the sandstone and shale aquifers generally have low to moderate yields (Hammond and others, 1978).

The surficial material throughout the basin was deposited primarily during the Pleistocene epoch, when the Wisconsin glaciers covered most of the Northeast. Till of low permeability was deposited by glaciers over most of the basin; wells finished in till generally have low yields. Sand and gravel including alluvium, outwash, and ice-contact deposits, form the most productive aquifers in the basin; wells finished in these deposits may yield as much as 1,000 gallons per minute (Phillips and Hanchar, 1996).

Methods of Investigation

The methods used in this study, including well-selection criteria, sampling methods, and analytical methods, were designed to maximize data precision, accuracy, and comparability to other studies. Sample collection and processing were conducted in accordance with standard USGS procedures (U.S. Geological Survey, variously dated). Samples were analyzed at three laboratories--the USGS National Water Quality Laboratory (NWQL) in Denver, Colo., the USGS Organic Geochemistry Research Laboratory (OGRL) in Lawrence Kans., and a New York State Department of Health (NYSDOH)-certified laboratory. All procedures were done in accordance with documented methods.

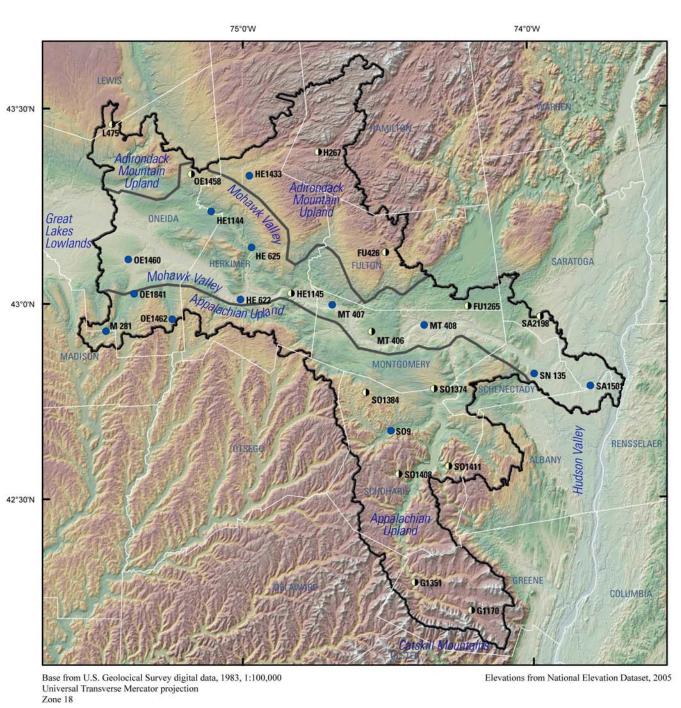
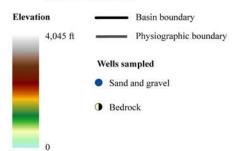


Figure 2. Topography, physiographic regions, and locations of wells sampled in the Mohawk River Basin.







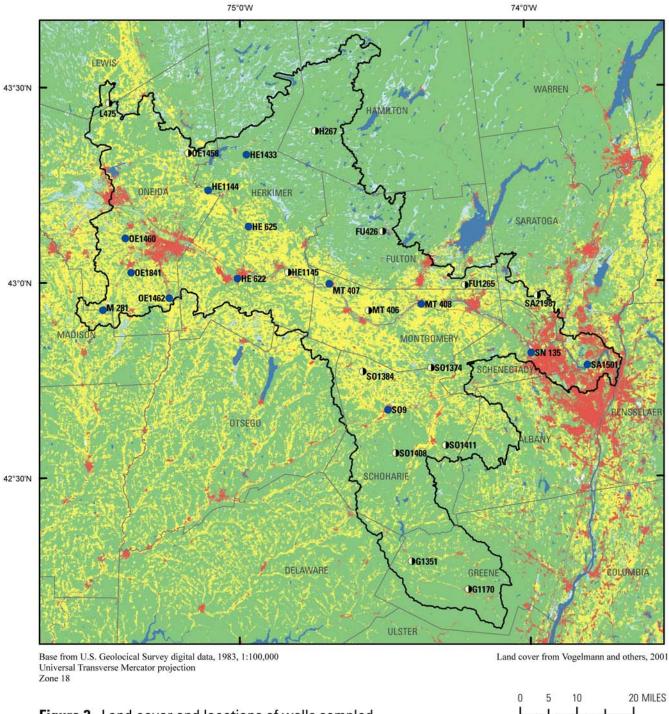
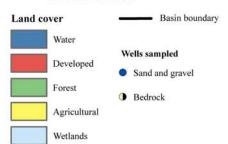


Figure 3. Land cover and locations of wells sampled in the Mohawk River Basin.







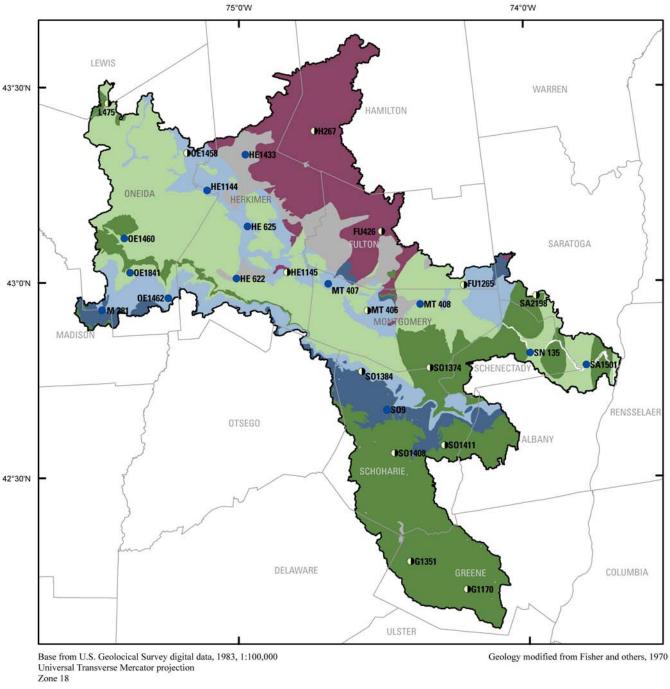
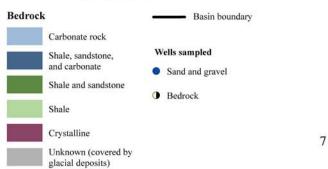


Figure 4. Generalized bedrock geology and locations of wells sampled in the Mohawk River Basin.







Site Selection

Wells were selected to provide sufficient spatial coverage of the basin, with emphasis on areas of greatest ground-water use. Selection also was based on the availability of well-construction data and hydrogeologic information on each well and its surrounding area. The study did not target specific municipalities, industries, or agricultural practices; rather, the 27 wells selected for sampling represented a combination of forested, rural, residential, and agricultural areas. Locations of the wells are shown in figures 2 and 3; the characteristics of the wells sampled and the types of land cover surrounding each well are listed in table A1 (at end of report). The depths and geologic units from which samples were collected, and the numbers of production and domestic wells are summarized below.

	Number of wells				
Material in which well is completed	Production	Domestic	Total		
Sand and gravel Depth range: 22.5 - 190 feet	10	3	13		
Bedrock	2	12	14		
Depth range: 75 - 815 feet					
Carbonate	1	3	4		
Shale	1	3	4		
Interlayered sandstone and shale		5	5		
Gneiss		1	1		
Total no. of wells	12	15	27		

Land use within a half-mile radius of most wells was typically forested but other types of land use prevailed at some wells; for example well MT406 represents agricultural land, and well S09 is in a developed are (table A1).

Selection of the 15 domestic wells was based on information from the New York State Department of Environmental Conservation (NYSDEC) Water Well program, which began in 2000. The program requires that licensed well drillers file a report with NYSDEC containing basic information about each well drilled—such as well and casing depth and diameter, yield, and a hydrogeologic log. Inspection of well-completion reports identified about 200 wells as potential sampling sites. The well owners were sent a letter that included a request for permission to sample the well, and a questionnaire about the well. Well owners who granted permission were contacted later by phone to verify well information and arrange a convenient time for sampling.

Of the 12 production wells that were sampled, 11 had been previously sampled in 2002 as part of the initial study for this project. The additional production well was identified through the USGS Ground-Water Site Inventory (GWSI) database. Town officials and (or) water managers were sent letters requesting permission to sample a well, and follow-up phone calls were made to arrange a time for sampling. Well information that was not already in the data base, such as depth, was provided by water managers. The type of aquifer tapped by production and domestic wells was verified through published geologic maps, such as Fisher and others (1970).

Sampling Methods

Ground-water samples were collected from August through November 2006 and processed through standard USGS protocols (U.S. Geological Survey, variously dated). The domestic-well samples were collected from a spigot between the well and pressure tank, where possible, and ahead of any water-treatment system so that they would be representative of the water quality within the aquifer. Production wells were sampled at the spigot or faucet used for collection of raw-water samples by water managers.

One or two wells were sampled per day. Typically, samples were collected through one or more 10-ft lengths of Teflon tubing attached to the garden-hose spigot closest to the well. After the tubing was connected, the well was purged by discharging to waste for at least 20 minutes, or until at least one well-casing volume of water had passed the sampling point. Most of the production wells were pumped for at least 1 hour before sampling, typically at pumping rates of about 100 gal/min. Domestic wells were purged at pumping rates ranging from about 5 to 10 gal/min; wells that had been recently used were purged of amounts less than three well-casing volumes. Notes about the well and surrounding land and land use were taken during well purging, and a global positioning system (GPS) measurement of latitude and longitude was recorded. After the well was purged, water was directed at 1 gal/min or less into a flow-through chamber that contained a meter with temperature, pH, specific conductance, and dissolved-oxygen sensors. Field values were then recorded at regular intervals; sampling began when the values of temperature, pH, specific conductance, and dissolved oxygen concentration had stabilized (U.S. Geological Survey, variously dated).

The flow rate for sampling was adjusted to about 0.5 gal/min or less. The Teflon sampling tube was then disconnected from the multi-probe meter and connected to a sampling chamber consisting of a polyvinyl chloride (PVC) frame and a clear-plastic chamber bag. The sampling chamber was placed on a table with a built-in drain constructed from a plastic box. The Teflon tubing and spigot-attachment equipment were cleaned in the laboratory before each day of sampling with a dilute phosphate-free detergent solution, followed by tap water and deionized water rinses. Equipment for filtration of pesticide samples was rinsed with methanol. A new sampling-chamber bag was used at each site. Samples were collected and preserved in the sampling chamber according to standard USGS methods. Sample bottles for nutrient, major-ion, and some trace-element analyses were filled with water filtered through disposable (one-time use) 0.45-um-pore-size polyether sulfone capsule filters that were preconditioned in the laboratory with deionized water on the day of sample collection. Sample bottles for pesticide analysis were filled with water filtered through baked 0.7-um-pore-size glass fiber filters. Acid preservation was required for trace element, VOC, and major ion analyses. Acid preservative was added after collection of other samples was completed to avoid the possibility of cross contamination by the acid preservative; for example, samples preserved with nitric acid were acidified after the collection of samples for nutrient analysis. Samples for radon analysis were collected through a septum chamber with a glass syringe according to standard USGS procedures. All samples analyzed by NYSDOHcertified laboratories were collected in bottles provided by the analyzing laboratory. All samples except those for radiochemical analysis were chilled to 4°C or less after collection and shipped by overnight delivery to the designated laboratories.

All but five sampling sites had easy access to a spigot; these five wells (HE622, MT406, SA1501, SN135, and SO9, fig. 2) were production wells and were sampled from taps or hydrants at which water-system personnel routinely collect raw-water samples. At these sites, physical properties were measured by the multisensor meter in a bucket; flow was adjusted to avoid air entrainment. The

syringe for radon-222 sample collection at these sites was inserted directly into the flowing water in the throat of the tap or hydrant to minimize sample exposure to the atmosphere.

Analytical Methods

Samples were analyzed for 226 physical properties and constituents, including inorganic constituents, nutrients, trace elements, radionuclides, pesticides and their degradates, VOCs, and bacteria. Physical properties, namely water temperature, pH, dissolved oxygen, and specific conductance were measured at the sampling site. Analyses for inorganic constituents, nutrients, trace elements, radon-222, pesticides and pesticide degradates, and VOCs were conducted at the USGS NWQL in Denver, Colo.; some additional pesticide and pesticide-degradate analyses were done at the USGS OGRL in Lawrence, Kans. The analyses for total organic carbon, phenolic compounds, and bacteria were done at Friend Laboratory in Waverly, N.Y., a NYSDOH-certified laboratory.

Anion concentrations were measured by ion-exchange chromatography; cation concentrations were measured by inductively coupled plasma-atomic emission spectrometry (ICP-AES), as described in Fishman (1993). Nutrients concentrations were measured through colorimetry, as described by Fishman (1993), and through Kjeldahl digestion with photometric finish, as described by Patton and Truitt (2000). Mercury concentrations were measured by cold vapor—atomic fluorescence spectrometry according to methods described by Garbarino and Damrau (2001). Arsenic, chromium, and nickel concentrations were measured through collision/reaction cell inductively coupled plasma-mass spectrometry (cICPMS) as described by Garbarino and others (2006). Remaining trace-element analyses were done by ICP-AES (Struzeski and others, 1996), inductively coupled plasma-optical emission spectrometry (ICP-OES), and inductively coupled plasma-mass spectrometry (ICP-MS) (Garbarino and Struzeski, 1998). In-bottle digestions for trace-element analyses were done as described by Hoffman and others (1996). Radon-222 concentrations were measured through liquid-scintillation counting (ASTM International, 2006).

Samples for pesticide analyses were processed as described by Wilde and others (2004). Pesticides and pesticide-degradates were analyzed at the NWQL through gas chromatography-mass spectrometry (GC-MS) and high-performance liquid chromatography-mass spectrometry (HPLC-MS) as described by Zaugg and others (1995), Sandstrom and others (2001), and Furlong and others (2001). Acetamide parent compounds and degradation-product analyses were conducted through liquid chromatography-mass spectrometry (LC-MS) at the USGS OGRL according to methods described by Lee and Strahan (2003). The VOC analyses were done through GC-MS by methods described by Connor and others (1998).

Total organic carbon concentration was measured by method SW-846 9060 (U.S. Environmental Protection Agency, 2004), and total phenolic compounds were measured by USEPA method 420.2 (U.S. Environmental Protection Agency, 1983). Bacterial samples were collected in accordance with NYSDEC and NYSDOH protocols, except that the tap from which each water sample was collected was not flame sterilized. Samples were tested for total coliform, fecal coliform, and *Escherichia coli* (*E. coli*) by Standard Methods 9222 B, D, and G (American Public Health Association, 1998). Results for fecal coliform analysis are reported as estimated because the samples could not reach the laboratory within the 6-hour holding time; the remaining samples were processed within holding times. A heterotrophic plate count test (SM 9215 B) also was conducted.

One blank sample and two sequential replicate samples were collected for quality assurance in addition to the 27 well-water samples. Nitrogen-purged VOC/pesticide-grade blank water and inorganic-grade blank water supplied by the USGS NWQL were used for an equipment blank; the water was run through a part of the Teflon tubing used for sampling, and water for filtered-water constituents was pumped through cleaned, pre-conditioned filters. Quality assurance samples were acidified in the same manner as environmental samples. No constituents were detected at concentrations above laboratory reporting levels in the blank. The percent-concentration differences from the sequential replicate sample were less than 5 percent for most of the constituents detected in the replicate samples; the largest percent differences were in trace elements whose concentrations were less than or near the reporting level.

Ground-Water Quality

More than half (159) of the 226 constituents or properties for which ground-water samples were analyzed were not detected in any sample at a concentration exceeding laboratory reporting levels (table A2). Concentrations of the 67 constituents and properties that were detected are reported in tables A3 through A9. The concentrations of some constituents are listed in these tables as "estimated" (preceded by letter E); estimated values are typically reported where the detected value is less than the established laboratory reporting levels, or when recovery of a compound has been shown to be highly variable (Childress and others, 1999). Concentrations of some constituents exceeded maximum contaminant levels (MCLs) or secondary drinking water standards (SDWS) set by the U.S. Environmental Protection Agency (USEPA) (U.S. Environmental Protection Agency, 2003) or NYSDOH (New York State Department of Health, 2007). MCLs are enforceable standards for finished water at public water supplies; although they are not enforceable for private homeowner wells, they are presented here as a standard for evaluation of the water-quality results. Secondary drinking water standards typically relate to corrosivity or aesthetic concerns such as taste, odor, or staining of plumbing fixtures and are not enforceable.

Physical Properties

Sample color ranged from < 1 to 20 Pt-Co units; the median was 2 Pt-Co units (table A3). The color of one sample, 20 Pt-Co units, exceeded the NYSDOH MCL and USEPA SDWS of 15 Pt-Co units. Dissolved oxygen concentration from < 0.1 to 12.2 mg/L and was generally greater in samples from sand and gravel wells (median 5.6 mg/L) than in samples from bedrock wells (median 0.2 mg/L). Sample pH ranged from 6.1 to 8.7; the median pH was 7.2 in samples from sand and gravel wells and 7.7 in samples from bedrock wells. The pH of two samples, 6.1 and 8.7, exceeded the USEPA SDWS range for pH, 6.5 to 8.5. Specific conductance ranged from 32 to 2,230 μ S/cm at 25°C with a median of 670 μ S/cm @ 25°C. Water temperature ranged from 8.5 to 14.5°C with a median temperature of 11.0°C.

Major Ions

The anion with the highest concentrations was bicarbonate (tables 1 and A4), with a median concentration of 276 mg/L. The cations with the highest concentrations were calcium (median 58.9 mg/L) and sodium (median 41.9 mg/L). The concentration of sodium in 11 samples exceeded the

USEPA nonregulatory drinking water advisory taste threshold of 60 mg/L; the maximum concentration of sodium was 662 mg/L. The concentration of chloride in two samples, 1,070 and 600 mg/L, exceeded the NYSDOH MCL of 250 mg/L. The concentrations of sodium and chloride were generally higher in samples from bedrock wells than in samples from sand and gravel wells, as shown in table 1, and were elevated in samples from three bedrock wells--G1170, G1351, and SO1374. The concentration of fluoride in one sample, 2.8 mg/L, exceeded the NYSDOH MCL of 2.2 mg/L but did not exceed the USEPA MCL of 4.0 mg/L. The concentration of sulfate in one sample, 451 mg/L, exceeded the USEPA SDWS and NYSDOH MCL of 250 mg/L.

Water hardness in the samples ranged from 23 to 510 mg/L as CaCO₃, and was typically very hard (180 mg/L or more as CaCO₃; Hem, 1985); the median hardness was 230 mg/L as CaCO₃. Nine of the 13 samples from the Mohawk Valley were very hard, whereas, all four samples from the Adirondack Uplands were soft (60 mg/L of CaCO₃ or less). Concentrations of calcium and magnesium (and therefore water hardness), were generally greater in samples from sand and gravel wells than from those from bedrock wells. Alkalinity ranged from 34 to 367 mg/L as CaCO₃, with a median of 226 mg/L as CaCO₃; it was highest in areas with carbonate bedrock and lowest in areas with crystalline bedrock. Residue on evaporation at 180°C, a measurement of total dissolved solids, ranged from 57 to 1,920 mg/L with a median of 409 mg/L; the highest values were in samples from bedrock wells.

Table 1. Drinking water standards and summary statistics for concentrations of major ions in groundwater samples from the Mohawk River Basin, 2006.

[All concentrations in milligrams per liter in filtered water except as noted; --, not applicable; <, less than]

		Concentration							
		Drinking	No. of Sand and gravel aquifers Bedrock aquifers (13 samples) (14 samples)						
		Water Standard	exceeding standard	Minimum	Median	Maximum	Minimum	Median	Maximum
	Calcium			15.8	63.9	132	5.87	24.8	123
ons	Magnesium			1.95	17.0	57.0	1.08	6.14	53.7
Cations	Potassium			.50	1.29	3.99	.56	1.52	7.38
	Sodium	60°	11	1.48	15.2	73.6	4.12	83.6	662
	Bicarbonate			55	288	394	41	245	448
Anions	Chloride	250 ^b	2	1.85	27.3	150	.31	65.1	1,070
Ani	Fluoride	2.2 ^b	1	<.1	.1	.5	.1	.4	2.8
·	Sulfate	250 ^{bc}	1	4.4	21.5	99.5	< .9	11.9	451
Hardness, unfiltered, mg/L as CaCO ₃		47	300	400	23	85	510		
Alk	alinity, filtered	d, mg/L as C	aCO,	45	236	323	34	201	367
Res	sidue on evapo	ration, filter	ed, mg/L	72	334	627	57	509	1,920

^a USEPA Drinking Water Advisory Taste Threshold

^b NYSDOH Maximum contaminant level

^c USEPA Secondary drinking water standard

Nutrients and Organic Carbon

Concentrations of nutrients varied with well depth and aquifer type. Concentrations of nitrate and organic carbon were generally higher in wells finished in sand and gravel than in those finished in bedrock, as shown below in table 2. Concentrations of nitrate ranged from < 0.06 to 2.6 mg/L as N (tables 2 and A5); the median concentration was 0.28 mg/L as N in samples from sand and gravel wells and < 0.06 mg/L as N in samples from bedrock wells. Nitrite was detected in only two sand and gravel wells with a maximum concentration of 0.003 mg/L as N. No concentrations of nitrate or nitrite exceeded established MCLs of the USEPA or NYSDOH (10 mg/L as N and 1 mg/L as N, respectively). Organic carbon was detected in samples from 7 of the 13 sand and gravel wells with a median concentration of 1 mg/L and a maximum of 1.6 mg/L; organic carbon was not detected in any samples from bedrock wells (reporting level 1 mg/L).

Concentrations of ammonia and orthophosphate were generally higher in samples from bedrock wells than in samples from sand and gravel wells. The concentration of ammonia ranged from < 0.010 to 1.88 mg/L as N; the median concentration was 0.349 mg/L as N in samples from bedrock wells and estimated 0.006 mg/L as N in samples from sand and gravel wells. Orthophosphate concentrations ranged from estimated 0.004 to 0.143 mg/L as P; the median concentration in samples from bedrock wells was 0.018 mg/L as P; the median in samples from sand and gravel wells was 0.007 mg/L as P.

Table 2. Drinking water standards and summary statistics for concentrations of nutrients in groundwater samples from the Mohawk River Basin, 2006.

[All concentrations	in filt	ered water	excent as	noted:	not applic	able. /	less thanl
TAIL CONCEINIANOUS		SIGU WAIGI	EXCEDI AS	HUIGU	ион аплин	anne. <.	icss main

					Concer	ntration			
	Drinking	No. of Samples		and grave 13 sample			Bedrock wells (14 samples)		
	Water Standard	exceeding standard	Minimum	Median	Maximum	Minimum	Median	Maximum	
Ammonia plus organic N, mg/L as N			< 0.10	0.06	1.2	< 0.1	0.36	2.1	
Ammonia, mg/L as N			< .010	E .006	1.02	.005	.349	1.88	
Nitrite plus nitrate, mg/L as N	10 ^{ab}	0	< .06	.28	2.6	< .06	< .06	.6	
Nitrite, mg/L as N	1^{ab}	0	< .002	< .002	.003	< .002	< .002	< .002	
Orthophosphate, mg/L as P			E .004	.007	.143	.006	.018	.123	
Total organic carbon, unfiltered sample, mg/L			< 1	1	1.6	< 1	< 1	< 1	

^a USEPA Maximum Contaminant Level

Trace Elements and Radionuclides

The trace elements present in the highest concentrations were strontium (median 549 μ g/L), iron (median 143 μ g/L in unfiltered water; 39 μ g/L in filtered water), boron (median 35 μ g/L), and manganese (median 31.1 μ g/L in unfiltered water; 24.0 μ g/L in filtered water) (tables 3 and A6). Median concentrations of boron, copper, iron, lithium, manganese and strontium were generally greater

^b NYSDOH Maximum contaminant level

in samples from wells finished in bedrock than in those from wells finished in sand and gravel (table 3). The concentration of aluminum in two samples, 348 and 59 μ g/L, exceeded the USEPA SDWS range of 50 to 200 μ g/L. The concentration of arsenic in two samples, 17.8 and 16.4 μ g/L, exceeded the USEPA MCL of 10 μ g/L. The concentration of iron in 10 unfiltered and 6 filtered samples exceeded the USEPA SDWS and NYSDOH MCL of 300 μ g/L; the maximum concentration of iron was 8,190 μ g/L in an unfiltered sample. The concentration of manganese exceeded the USEPA SDWS of 50 μ g/L in 10 unfiltered and 11 filtered samples and exceeded the NYSDOH MCL of 300 μ g/L in 3 filtered samples. No samples exceeded drinking water standards for antimony, barium, beryllium, cadmium, chromium, copper, lead, selenium, zinc, or uranium.

Radon-222 activities in the water samples ranged from 20 to 1,360 pCi/L, with a median of 250 pCi/L. The highest radon activities (600 pCi/L and greater) were in samples from bedrock wells. Radon concentration in drinking water is currently not regulated; however, the USEPA has proposed a two-part standard for drinking water: (1) a 300 pCi/L MCL for areas that do not implement an indoor air radon-mitigation program, and (2) an alternative MCL (AMCL) of 4,000 pCi/L for areas that do (U.S. Environmental Protection Agency, 1999). Activities in 12 of the samples (44 percent) exceeded the proposed MCL; none of the samples exceeded the proposed AMCL.

Pesticides

Nine pesticides were detected in six samples; all were herbicides or their degradates and most were degradates of the triazine and amide broadleaf herbicides atrazine, alachlor, and metolachlor (table A7). Pesticides were detected in samples from four sand and gravel wells and two bedrock wells. Most concentrations were on the order of hundredths or thousandths of micrograms per liter; the highest were metolachlor degradates (maximum $0.42~\mu g/L$). The most frequently detected pesticides were atrazine (4 samples), CIAT (2-chloro-4-isopropylamino-6-amino-s-triazine, 4 samples), and Metolachlor ESA (3 samples). No pesticide concentrations exceeded established drinking water standards; pesticide degradates are not currently regulated.

Volatile Organic Compounds and Phenolic Compounds

VOCs were detected in samples from four wells—three sand and gravel wells and one bedrock well (table A8). Six VOCs were detected, including, tetrachloroethene, toluene, and four trihalomethanes. Trihalomethanes (THMs) are disinfection byproducts that form when chlorine or bromine are used as disinfectants; they are also used as solvents. The THMs detected were bromodichloromethane, dibromochloromethane, tribromomethane (bromoform), and trichloromethane (chloroform). Three of the four THMs were detected in two samples each, the last was detected in three samples. The median concentration of detected THMs was 0.3 μ g/L, the maximum was 0.8 μ g/L. The USEPA and NYSDOH MCLs for total trihalomethanes (TTHMs) is 80 μ g/L; the maximum TTHMs in the samples was 2 μ g/L. Tetrachloroethene (PERC), a solvent sometimes used for dry cleaning, was detected in one sample at a concentration of 0.4 μ g/L; the USEPA and NYSDOH MCLs for tetrachloroethene are 5 μ g/L. Toluene, a component of gasoline, was detected in two wells at a concentration of 0.2 μ g/L; it was the only VOC detected in a sample from a bedrock well. The NYSDOH MCL for toluene is 5 μ g/L, and the USEPA MCL is 1,000 μ g/L. Phenolic compounds were detected in samples from three wells; the maximum concentration was 13 μ g/L.

Table 3. Drinking water standards and summary statistics for concentrations of trace elements and radon-222 in ground-water samples from the Mohawk River Basin, 2006.

[<, less than; M, constituent detected but not quantified; --, not applicable. All concentrations are in micrograms per liter except as noted]

		Concentration							
	Drinking	No. of Samples		Sand and gravel wells (13 samples)			Bedrock wells (14 samples)		
	Water	exceeding			•			•	
	Standard	standard	Minimum	Median	Maximum	Minimum	Median	Maximum	
Aluminum, unfiltered	50-200°	1-2	< 2	< 2	18	< 2	6	348	
Antimony, unfiltered	6^{ab}	0	< .2	< .2	< .2	< .2	< .2	.4	
Arsenic, unfiltered	10°	2	.12	.55	17.8	.06	.40	16.4	
Barium, unfiltered	2000 ^{ab}	0	M	58	916	4	100	475	
Beryllium, unfiltered	4 ^{ab}	0	< .06	< .06	.03	< .06	< .06	.04	
Boron, filtered			4.1	24	125	< 7.0	122	608	
Cadmium, unfiltered	5 ^{ab}	0	< .04	< .04	.05	< .04	< .04	.19	
Chromium, unfiltered	100 ^{ab}	0	< .60	< .60	.88	< .60	.30	14.8	
Cobalt, unfiltered			.028	.173	.742	.023	.108	.556	
Copper, unfiltered	1000°	0	0.6	1.2	4.7	.5	2.8	26.2	
Iron, filtered	300 ^{bc}	6	< 6	8	1,240	< 6	72	2,190	
Iron, unfiltered	300 ^{bc}	10	< 6	29	1,210	17	208	8,190	
Lead, unfiltered	15 ^d	0	< .06	.09	2.13	< .06	.12	2.21	
Lithium, unfiltered			1	5.7	75.7	< .6	43.7	824	
Manganese, filtered	50°	11	< .6	10.1	1,960	< .6	37.1	306	
Manganese, unfiltered	50°	10	< .6	18.1	2,020	.5	47.8	306	
Molybdenum, unfiltered			< .2	.4	4.5	< .2	2.3	10.4	
Nickel, unfiltered			.24	.76	2.29	.17	1.03	2.48	
Selenium, unfiltered	50 ^{ab}	0	< .08	.04	.71	< .08	< .08	.40	
Strontium, unfiltered			30.2	369	2,640	30.0	608	31,100	
Zinc, unfiltered	5000 ^{bc}	0	< 2	3	10	< 2	2	16	
Radon-222, unfiltered, pCi/L	300 ^e	12	30	420	530	20	130	1,360	
Uranium, unfiltered	30°	0	.151	.245	1.52	< .012	.118	1.80	

^a USEPA Maximum Contaminant Level

Bacteria

Coliform bacteria were detected in samples from six wells (table A9); three of whish were sand and gravel wells and three were bedrock wells. The smallest detection was 1 CFU/100 mL and the greatest was too numerous to count. Any detection of coliform bacteria in finished drinking water is considered to be above the MCL. The owners of these six wells were notified of the results upon receipt from the laboratory. Fecal coliform and *E. coli* were not detected in any sample. The

^b NYSDOH Maximum contaminant level

^c USEPA Secondary Drinking Water Standard

^d USEPA Treatment Technique

^e USEPA Proposed Maximum Contaminant Level

heterotrophic plate count ranged from < 1 CFU/mL to 940 CFU/mL with a median of 4 CFU/mL. The USEPA MCL for the heterotrophic plate count is 500 CFU/mL; this limit was exceeded in one sample (940 CFU/mL).

Summary

Ground-water samples were collected from 13 sand and gravel wells and 14 bedrock wells to characterize the ground-water quality in the Mohawk River Basin. Wells screened in sand and gravel ranged from 22.5 ft to 190 ft deep; the wells completed in bedrock were 75 ft to 815 ft deep and typically tapped shale and sandstone or carbonate rock. Of the 27 wells sampled, 12 were production wells, and 15 were domestic wells. Sample collection and analysis was done through standard USGS procedures and other documented methods. Samples were analyzed for physical properties, major ions, nutrients, trace elements, radon-222, pesticides, VOCs, and bacteria; many constituents were not detected in any sample.

The chemical quality of the samples was generally good, although the concentrations of certain constituents in some samples, including color, pH, sodium, chloride, fluoride, sulfate, aluminum, arsenic, iron, manganese, radon-222 and bacteria exceeded State or Federal drinking water quality standards. Constituents whose concentrations most often exceeded drinking water standards were radon-222 (12 samples with concentrations greater than 300 pCi/L), iron (10 samples with concentrations greater than 300 μ g/L), manganese (10 samples with concentrations greater than 50 μ g/L), and bacteria (6 samples with detections of coliform bacteria). Two samples had arsenic concentrations that exceeded the USEPA MCL of 10 μ g/L. No pesticide or VOC was detected at a concentration exceeding established drinking water standards. Drinking-water quality standards used for comparison included NYSDOH MCLs and USEPA MCLs, SDWS, and drinking water advisories.

Concentrations and frequencies of some constituents that were detected in samples from sand and gravel wells differed from those in samples from bedrock wells; for example, samples from sand and gravel wells had higher concentrations of dissolved oxygen than samples from bedrock wells. Concentrations of other constituents varied by physiographic region and by underlying bedrock type. Water in the Mohawk River Basin is generally hard to very hard, especially in the Mohawk Valley and areas underlain by carbonate bedrock. Three bedrock wells had elevated levels of sodium and chloride (maximum concentrations 662 mg/L and 1,070 mg/L respectively). Nitrate and nitrite concentrations were generally higher in samples from sand and gravel wells than in those from bedrock wells although no concentrations exceeded established drinking-water standards; conversely, ammonia concentrations were higher in samples from bedrock wells than in samples from sand and gravel wells. Concentrations of several trace elements, including boron, copper, iron, manganese, and strontium were higher in samples from bedrock wells than in those from sand and gravel wells. The highest Radon-222 activities were in samples from bedrock wells (maximum 1,360 pCi/L). Nine pesticides and pesticide degradates were detected in four samples from sand and gravel wells and two samples from bedrock wells; most were degradates of atrazine, alachlor, and metolachlor. Most VOC detections were in samples from sand and gravel wells; these included toluene, tetrachloroethene, and four trihalomethanes, none of which exceeded drinking water standards.

References Cited

- American Public Health Association, 1998, Standard methods for the examination of water and wastewater (20th ed.): Washington, D.C., American Public Health Association, American Water Works Association, and Water Environment Federation [variously paged].
- ASTM International, 2006, D5072-98(2006), Standard test method for radon in drinking water: ASTM International, accessed 12/28/06 at http://www.astm.org.
- Butch, G.K., Murray, P.M., Hebert, G.J., and Weigel J.F., 2003, Water Resources Data, New York, Water Year 2002: U.S. Geological Survey Water-Data Report, NY-02-1, p. 502-520.
- Childress, C.J.O., Foreman, W.T., Connor, B.F., and Maloney, T.J., 1999, New reporting procedures based on long-term method detection levels and some considerations for interpretations of water-quality data provided by the U.S. Geological Survey National Water Quality Laboratory: U.S. Geological Survey Open-File Report 99-193, 19 p.
- Connor, B.F., Rose, D.L., Noriega, M.C., Murtagh, L.K., and Abney, S.R., 1998, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of 86 volatile organic compounds in water by gas chromatography/mass spectrometry, including detections less than reporting limits: U.S. Geological Survey Open-File Report 97-829, 78 p.
- Eckhardt, D.A., Reddy, J.E., and Tamulonis, K.L., 2007, Ground-water quality in the Genesee River Basin, New York, 2005-06: U.S. Geological Survey Open-File Report 2007-1093, 26 p. (online only).
- Fisher, D.W., Isachsen, Y.W., and Rickard, L.V., 1970, Geologic Map of New York State: New York State Museum Geological Survey, Map and Chart Series no. 15, Adirondack sheet, scale 1:250,000.
- Fishman, M.J., ed., 1993, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of inorganic and organic constituents in water and fluvial sediments: U.S. Geological Survey Open-File Report 93-125, 217 p.
- Furlong, E.T., Anderson, B.D., Werner, S.L., Soliven, P.P., Coffey, L.J., and Burkhardt, M.R., 2001, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory— Determination of pesticides in water by graphitized carbon-based solid-phase extraction and high-performance liquid chromatography/mass spectrometry: U.S. Geological Survey Water-Resources Investigations Report 01-4134, 73 p.
- Garbarino, J.R. and Damrau, D.L., 2001, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of organic plus inorganic mercury in filtered and unfiltered natural water with cold vapor—atomic fluorescence spectrometry: U.S. Geological Survey Water-Resources Investigations Report 01-4132, 16 p.
- Garbarino, J.R., Kanagy, L.K., and Cree, M.E., 2006, Determination of elements in natural-water, biota, sediment and soil samples using collision/reaction cell inductively coupled plasma-mass spectrometry: U.S. Geological Survey Techniques and Methods, book 5, chap.1, sec. B, 88 p.

- Garbarino, J.R. and Struzeski, T.M., 1998, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of elements in whole-water digests using inductively coupled plasma-optical emission spectrometry and inductively coupled plasma-mass spectrometry: U.S. Geological Survey Open-File Report 98-165, 101 p.
- Hammond, D.S., Heath, R.C., and Waller, R.M., 1978, Ground-water data on the Hudson River Basin, New York: U. S. Geological Survey Open File Report 78-710, 18 p.
- Hem, J.D., 1985, Study and interpretation of the chemical characteristics of natural water: U.S. Geological Survey Water-Supply Paper 2254, 264 p.
- Hetcher-Aguila, K.K., 2005, Ground-water quality in the Chemung River Basin, New York, 2003: U.S. Geological Survey Open-File Report 2004-1329, 19 p. (online only).
- Hetcher-Aguila, K.K. and Eckhardt, D.A., 2006, Ground-water quality in the upper Susquehanna River Basin, New York, 2004: U.S. Geological Survey Open-File Report 2006-1161, 21 p. (online only).
- Hoffman, G.L., Fishman, M.J., and Garbarino, J.R., 1996, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—In-bottle acid digestion of whole-water samples: U.S. Geological Survey Open-File Report 96-225, 28 p.
- Isachsen, Y.W., Landing, Ed, Lauber, J.M., Rickard, L.V., and Rogers, W.B., eds., 2000, Geology of New York—A simplified account (2nd ed.): Albany, NY, New York State Museum/Geological Survey, 294 p.
- Lee, E.A. and Strahan, A.P., 2003, Methods of analysis by the U.S. Geological Survey Organic Geochemistry Research Group—Determination of acetamide herbicides and their degradations products in water using online solid-phase extraction and liquid chromatography/mass spectrometry: U.S. Geological Survey Open-File Report 03-173, 17 p.
- New York State Department of Health, 2007, NYCRR Title 10, Chapter 1, Part 5, Subpart 5-1 Public Water Systems: Tables: Albany, NY, accessed 12/31/07 at http://www.health.state.ny.us/environmental/water/drinking/part5/tables.htm.
- Nystrom, E.A., 2006, Ground-water quality in the Lake Champlain Basin, New York, 2004: U.S. Geological Survey Open-File Report 2006-1088, 22 p. (online only).
- Nystrom, E.A., 2007a, Ground-water quality in the St. Lawrence River Basin, New York, 2005-06: U.S. Geological Survey Open-File Report 2007-1066, 33 p. (online only).
- Nystrom, E.A., 2007b, Ground-water quality in the Delaware River Basin, New York, 2001 & 2005-06: U.S. Geological Survey Open-File Report 2007-1098, 36 p. (online only).
- Patton, C.J. and Truitt, E.P., 2000, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of ammonium plus organic nitrogen by a Kjeldahl digestion method and an automated photometric finish that includes digest cleanup by gas diffusion: U.S. Geological Survey Open-File Report 00-170, 31 p.

- Phillips, P.J. and Hanchar, D.W., 1996, Water-quality assessment of the Hudson River Basin in New York and adjacent States—Analysis of available nutrient, pesticide, volatile organic compound, and suspended-sediment data, 1970-90: U.S. Geological Survey Water-Resources Investigations Report 96-4065, 77 p.
- Randall, A.D., 1996, Mean annual runoff, precipitation, and evapotranspiration in the glaciated northeastern United States, 1951-80: U.S. Geological Survey Open-File Report 96-395, 2pl., 1:250,000.
- Sandstrom, M.W., Stroppel, M.E., Foreman, W.T., and Schroeder, M.P., 2001, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of moderate-use pesticides and selected degradates in water by C-18 solid-phase extraction and gas chromatography/mass spectrometry: U.S. Geological Survey Water-Resources Investigations Report 01-4098, 70 p.
- Struzeski, T.M., DeGiacomo, W.J., and Zayhowski, E.J., 1996, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of dissolved aluminum and boron in water by inductively coupled plasma-atomic emission spectrometry: U.S. Geological Survey Open-File Report 96-149, 17 p.
- U.S. Environmental Protection Agency, 1983, Methods for chemical analysis of water and wastes: Washington, D.C., U.S. Environmental Protection Agency, Environmental Monitoring and Support Laboratory Office of Research and Development, EPA 600/4-79-020 p. 420.2-1-5.
- U.S. Environmental Protection Agency, 1997, Guidelines for preparation of the comprehensive state water quality assessments (305(b) Reports) and electronic updates: Washington, D.C., U.S. Environmental Protection Agency, Office of Water, EPA 841-B-97-002A and EPA 841-B-97-002B, PL95-217, 271 p.
- U.S. Environmental Protection Agency, 1999, Proposed radon in drinking water rule: Washington, D.C., U.S. Environmental Protection Agency, Office of Water, EPA 815-F-99-006, 6 p.
- U.S. Environmental Protection Agency, 2003, National primary drinking water standards and national secondary drinking water standards: Washington, D.C., U.S. Environmental Protection Agency, Office of Water, EPA 816-F-03-016, 6 p.
- U.S. Environmental Protection Agency, 2004, Test Methods for Evaluating Solid Waste, Physical/Chemical Methods: EPA SW-846, p. 9060A1--5.
- U.S. Geological Survey, variously dated, National field manual for the collection of water-quality data: U.S. Geological Survey Techniques of Water-Resources Investigations, book 9, chaps. A1-A9, [variously paged].
- Vogelmann, J.E., Howard, S.M., Yang, L., Larson, C.R., Wylie, B.K., and Van Driel, J.N., 2001, Completion of the 1990's National Land Cover Data Set for the conterminous United States, Photogrammetric Engineering and Remote Sensing v. 67, p. 650-662.

- Wilde, F.D., Radtke, D.B., Gibs, Jacob, and Iwatsubo, R.T., eds., 2004, Processing of water samples (version 2.1): U.S. Geological Survey Techniques of Water-Resources Investigations, book 9, chap. A5, accessed 1/11/07 at http://pubs.water.usgs.gov/twri9A5/
- Zaugg, S.D., Sandstrom, M.W., Smith, S.G., and Fehlberg, K.M., 1995, Methods of analysis by the U.S. Geological Survey National Water Quality Laboratory—Determination of pesticides in water by C-18 solid-phase extraction and capillary-column gas chromatography/mass spectrometry with selected-ion monitoring: U.S. Geological Survey Open-File Report 95-181, 60 p.

Appendix.

Table A1. Information on wells sampled in the Mohawk River Basin, 2006.

[--, unknown; Pr, production; Dom, domestic; MHV, Mohawk-Hudson Valley, AKU, Adirondack upland; APU, Appalachian upland; D, developed; F, forested; A, agricultural; W, wetlands and open water; SS, sandstone]

Well	Date	Well depth, feet below	Casing depth, feet below	Well	Bedrock	Physiographic		cover², per e radius a		
number¹	sampled	land surface	land surface	type	type	region	D	F	Α	W
Sand and	gravel wells									
HE622	9/19/2006	52		Pr		MHV		71		7 15 6
HE625	9/25/2006	47		Pr		MHV	21	52		27
HE1144	8/21/2006	36	36	Dom		MHV		61		36 3
HE1433	8/16/2006	126	126	Dom		AKU		88		12
M281	9/20/2006	56		Pr		APU	1	58		41
MT407	8/31/2006	68		Pr		MHV	30	30		34 5
MT408	8/15/2006	190		Pr		MHV		60	12	18 10
OE1460	9/20/2006	28		Pr		MHV	28	39		28 5
OE1462	9/6/2006	87	75	Pr		APU	3	78		19
OE1841	8/24/2006	22.5	22.5	Dom		APU	9 ;	35	5	6
SA1501	9/18/2006	30		Pr		MHV	11 2	6 4	59)
SN135	8/9/2006	69	49	Pr		MHV	37		40	14 10
SO9	11/1/2006	40.25	35.25	Pr		APU		78		9 13
Bedrock v	vells									
FU426	8/15/2006	150	76	Dom	Carbonate	AKU		55		45
FU1265	8/9/2006	300	182		Carbonate	MHV	23		77	
G1170	8/14/2006	597	118.5		SS and shale	APU	26		74	1
G1351	8/14/2006	467	78	Dom	SS and shale	APU	1	69		29 1
H267	9/14/2006	130	130		Gneiss	AKU		99)	1
HE1145	8/16/2006	278	40	Dom	Shale	MHV	16	61		16 7
L475	8/17/2006	75	30	Dom	Shale	AKU	8 24		63	4
MT406	8/23/2006	815		Pr	Shale	MHV	2	98	3	
OE1458	8/22/2006	500		Pr	Carbonate	MHV	5 31		65	
SA2198	8/7/2006	150	18	Dom	Shale	MHV	11	73		15
SO1374	8/8/2006	500		Dom	SS and shale	APU		54	4	41 5
SO1384	8/10/2006	380	239		Carbonate	APU	4	7	5	53
SO1408	8/10/2006	200	58	Dom	SS and shale	APU	7	75		18
SO1411	8/8/2006	220	87	Dom	SS and shale	APU		93		7

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

² Land cover estimated from the National Land Cover Data set (1992).

³Totals may not equal 100 percent as a result of rounding.

Table A2. Constituents analyzed for but not detected in ground-water samples from the Mohawk River Basin, 2006.

USGS parameter code	Compound	Laboratory reporting level					
Trace Elements in unfiltered water, micrograms per liter							
71900	Mercury	0.01					
01077	Silver	.16					
01059	Thallium	.2					
Pesticides i	n filtered water, micrograms per li	ter					
50470	2,4-D methyl ester	.190					
39732	2,4-D	.04					
38746	2,4-DB	.02					
82660	2,6-Diethylaniline	.006					
04038	2-Chloro-6-ethylamino-4-amino- <i>s</i> -triazine	.08					
63781	2-Chloro- <i>N</i> -(2,6-diethylphenyl)acetamide	.02					
63782	2-Chloro- <i>N</i> -(2-ethyl-6-methylphenyl)acetamide	.02					
49308	3-Hydroxy carbofuran	.008					
61029	Acetochlor ethanesulfonic acid	.02					
61030	Acetochlor oxanilic acid	.02					
62847	Acetochlor sulfynilacetic acid	.02					
49260	Acetochlor	.006					
49315	Acifluorfen	.028					
50009	Alachlor ethanesulfonic acid	.02					
61031	Alachlor oxanilic acid	.02					
62848	Alachlor sulfynilacetic acid	.02					
46342	Alachlor	.005					
49313	Aldicarb sulfone	.02					
49314	Aldicarb sulfoxide	.100					
49312	Aldicarb	.15					
34253	alpha-HCH	.005					
82686	Azinphos-methyl	.050					
50299	Bendiocarb	.08					
82673	Benfluralin	.010					
50300	Benomyl	.022					
61693	Bensulfuron	.02					
38711	Bentazon	.02					
04029	Bromacil	.02					
49311	Bromoxynil	.04					
04028	Butylate	.004					
50305	Caffeine	.018					
49310	Carbaryl	.02					
82680	Carbaryl	.041					
49309	Carbofuran	.016					

USGS parameter code	Compound	Laboratory reporting level				
Pesticides in filtered water, micrograms per literContinued						
82674	Carbofuran	.020				
61188	Chloramben methyl ester	.02				
50306	Chlorimuron	.032				
04039	Chlorodiamino-s-triazine	.04				
38933	Chlorpyrifos	.005				
82687	cis-Permethrin	.006				
49305	Clopyralid	.07				
04041	Cyanazine	.018				
04031	Cycloate	.01				
49304	Daethal monoacid	.03				
82682	DCPA	.003				
63778	Dechloroacetochlor	.02				
63777	Dechloroalachlor	.02				
63779	Dechlorodimethenamid	.02				
63780	Dechlorometolachlor	.02				
62170	Desulfinyl fipronil	.012				
39572	Diazinon	.005				
38442	Dicamba	.04				
49302	Dichlorprop	.03				
39381	Dieldrin	.009				
61951	Dimethenamid ethanesulfonic acid	.02				
62482	Dimethenamid oxanilic acid	.02				
61588	Dimethenamid	.02				
49301	Dinoseb	.04				
04033	Diphenamid	.01				
82677	Disulfoton	.02				
49300	Diuron	.02				
82668	EPTC	.004				
82663	Ethalfluralin	.009				
82672	Ethoprop	.012				
49297	Fenuron	.10				
62169	Desulfinylfipronil amide	.029				
62167	Fipronil sulfide	.013				
62168	Fipronil sulfone	.024				
62166	Fipronil	.016				
61952	Flufenacet ethanesulfonic acid	.02				
62483	Flufenacet oxanilic acid	.02				
62481	Flufenacet	.02				
61694	Flumetsulam	.04				
38811	Fluometuron	.02				

USGS parameter code	Compound	Laboratory reporting level						
Pesticides i	Pesticides in filtered water, micrograms per literContinued							
04095	Fonofos	.005						
63784	Hydroxyacetochlor	.02						
63783	Hydroxyalachlor	.02						
64045	Hydroxydimethenamid	.02						
63785	Hydroxymetolachlor	.02						
50356	Imazaquin	.04						
50407	Imazethapyr	.04						
61695	Imidacloprid	.020						
39341	Lindane	.004						
38478	Linuron	.01						
82666	Linuron	.035						
39532	Malathion	.027						
38482	MCPA	.07						
38487	MCPB	.10						
50359	Metalaxyl	.03						
38501	Methiocarb	.034						
49296	Methomyl	.070						
82667	Methyl parathion	.015						
82630	Metribuzin	.028						
61697	Metsulfuron	.07						
82671	Molinate	.003						
61692	<i>N</i> -(4-Chlorophenyl)- <i>N</i> '-methylurea	.04						
82684	Napropamide	.007						
49294	Neburon	.01						
50364	Nicosulfuron	.04						
49293	Norflurazon	.02						
49292	Oryzalin	.02						
38866	Oxamyl	.05						
34653	p,p'-DDE	.003						
39542	Parathion	.010						
82669	Pebulate	.004						
82683	Pendimethalin	.022						
82664	Phorate	.055						
49291	Picloram	.03						
82676	Propyzamide	.004						
62766	Propachlor ethanesulfonic acid	.05						
62767	Propachlor oxanilic acid	.02						
04024	Propachlor	.010						
82679	Propanil	.011						
82685	Propargite	.02						
49236	Propham	.030						
50471	Propiconazole	.01						

USGS parameter code	Compound	Laboratory reporting level
Pesticides i	n filtered water, micrograms per	literContinued
38538	Propoxur	.008
38548	Siduron	.02
04035	Simazine	.005
50337	Sulfometuron	.090
82670	Tebuthiuron	.02
82665	Terbacil	.034
04032	Terbacil	.026
82675	Terbufos	.02
82681	Thiobencarb	.010
82678	Triallate	.006
49235	Triclopyr	.03
82661	Trifluralin	.009
Volatile org	anic compounds in unfiltered wa s per liter	ter,
34506	1,1,1-Trichloroethane	.1
77652	1,1,2-Trichloro-1,2,2- trifluoroethane	.1
34496	1,1-Dichloroethane	.1
34501	1,1-Dichloroethene	.1
34536	1,2-Dichlorobenzene	.1
32103	1,2-Dichloroethane	.2
34541	1,2-Dichloropropane	.1
34566	1,3-Dichlorobenzene	.1
34571	1,4-Dichlorobenzene	.1
34030	Benzene	.1
34301	Chlorobenzene	.1
77093	cis-1,2-Dichloroethene	.1
34668	Dichlorodifluoromethane	.2
34423	Dichloromethane	.2
81576	Diethyl ether	.2
81577	Diisopropyl ether	.2
34371	Ethylbenzene	.1
50005	Methyl tert-pentyl ether	.2
85795	m-Xylene plus p-xylene	.2
77135	o-Xylene	.1
77128	Styrene	.1
50004	tert-Butyl ethyl ether	.1
78032	Methyl tert-butyl ether	.2
32102	Tetrachloromethane	.2
34546	trans-1,2-Dichloroethene	.1
39180	Trichloroethene	.1
34488	Trichlorofluoromethane	.2
39175	Vinyl chloride	.2

Table A3. Physical properties of ground-water samples from the Mohawk River Basin, 2006.

[mg/L, milligrams per liter; μ S/cm @ 25°C, microsiemens per centimeter at 25 degrees Celsius; (00080), USGS National Water Information System parameter code; <, less than; bold values exceed one or more drinking water standards]

Well number¹	Color, platinum- cobalt units (00080)	Dissolved oxygen, mg/L (00300)	pH, standard units (00400)	Specific conductance, µS/cm @ 25°C (00095)	Water temperature, degrees Celsius (00010)
Sand and g	jravel wells				
HE622	2	7.9	7.2	1,040	9.9
HE625	2	6.9	7.4	366	10.1
HE1144	2	9.1	7.6	283	10.1
HE1433	2	8.6	8.1	106	9.5
M281	< 1	.3	7.2	547	8.9
MT407	8	5.9	7.2	519	9.8
MT408	5	.1	7.1	1,020	11.2
OE1460	2	5.6	7.1	910	13.6
OE1462	10	.3	7.3	749	9.6
OE1841	15	.1	7.3	768	12.3
SA1501	2	.6	7.1	455	12.4
SN135	2	12.2	7.4	443	12.4
SO9	8	.5	6.1	1,100	12.2
Bedrock w	ells				
FU426	20	2.4	7.9	32	9.8
FU1265	5	.1	7.7	315	11.1
G1170	< 1	.6	8.2	1,940	11.8
G1351	2	.2	8.7	2,230	11.5
H267	2	.2	8.1	84	8.5
HE1145	2	< .1	8.2	714	12.3
L475	8	.1	7.9	685	10.9
MT406	5	5.6	6.6	1,060	9.4
OE1458	2	.4	6.9	466	9.6
SA2198	2	7.8	7.1	475	12.2
SO1374	5	.1	7.1	800	14.5
SO1384	< 1	4.5	7.0	1,170	11.0
SO1408	< 1	.1	7.3	171	11.1
SO1411	2	.1	7.7	670	10.9

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A4. Concentrations of major ions in ground-water samples from the Mohawk River Basin, 2006.

[mg/L, milligrams per liter; CaCO₃, calcium carbonate; (00900), USGS National Water Information System parameter code; <, less than; E, estimated value; bold values exceed one or more drinking water standards]

Well number ¹	Hardness, filtered, mg/L as CaCO ₃ (00900)	Calcium, filtered, mg/L (00915)	Magnesium, filtered, mg/L (00925)	Potassium, filtered, mg/L (00935)	Sodium, filtered, mg/L (00930)	Acid neutralizing capacity, unfiltered, mg/L as CaCO ₃ (90410)	Alkalinity, filtered, fixed end point, lab, mg/L as CaCO ₃ (29801)
Sand and gr	ravel wells						
HE622	310	97.8	17.0	3.60	68.2	242	243
HE625	190	58.3	10.2	1.07	4.06	184	184
HE1144	140	44.5	7.81	.50	2.58	130	130
HE1433	47	15.8	1.95	.53	1.48	45	45
M281	300	63.5	35.3	1.29	3.89	276	278
MT407	250	63.9	23.2	1.09	11.8	224	236
MT408	340	87.7	29.0	2.84	73.6	284	283
OE1460	350	91.5	29.3	2.09	55.1	294	290
OE1462	330	86.3	28.4	1.18	24.7	199	198
OE1841	380	58.9	57.0	2.33	13.2	322	323
SA1501	230	72.6	11.2	2.38	15.2	210	212
SN135	180	54.8	9.33	1.22	19.6	169	170
SO9	400	132	16.2	3.99	70.4	289	288
Bedrock we	ells						
FU426	34	9.70	2.39	.89	4.12	34	34
FU1265	120	30.6	10.5	.70	23.4	161	160
G1170	81	26.6	3.67	.87	662	108	108
G1351	23	7.31	1.08	1.27	411	122	120
H267	34	9.21	2.76	.56	6.08	48	47
HE1145	34	5.87	4.67	2.92	139	331	331
L475	51	13.6	4.01	7.38	112	165	166
MT406	420	123	28.5	4.24	51.4	295	293
OE1458	240	72.1	15.7	1.78	5.82	226	226
SA2198	360	106	23.8	2.29	80.6	386	318
SO1374	510	115	53.7	3.73	139	369	367
SO1384	280	67.3	28.3	2.17	86.6	240	240
SO1408	89	23.0	7.61	.70	41.9	176	176
SO1411	71	21.4	4.15	.84	247	309	309

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A4. Concentrations of major ions in ground-water samples from the Mohawk River Basin, 2006.—Continued

[mg/L, milligrams per liter; CaCO₃, calcium carbonate; (00900), USGS National Water Information System parameter code; <, less than; E, estimated value; bold values exceed one or more drinking water standards]

Well number¹	Bicarbonate ² , filtered, fixed end point, lab, mg/L (29805)	Chloride, filtered, mg/L (00940)	Fluoride, filtered, mg/L (00950)	Silica, filtered, mg/L (00955)	Sulfate, filtered, mg/L (00945)	Residue on evaporation, filtered, mg/L (70300)
Sand and gr	avel wells					
HE622	296	130	< 0.1	7.22	35.3	531
HE625	224	5.23	E .1	8.03	8.7	201
HE1144	159	4.19	E .1	10.1	10.9	167
HE1433	55	1.85	.1	13.2	4.4	72
M281	339	4.46	.1	8.23	37.1	334
MT407	288	21.7	E .1	8.19	18.7	303
MT408	345	150	.5	15.3	21.5	552
OE1460	354	98.1	E .1	6.21	30.4	453
OE1462	242	65.7	.1	13.2	99.5	463
OE1841	394	35.0	E .1	11.6	54.2	433
SA1501	259	21.9	.2	11.8	19.5	283
SN135	207	27.3	E .1	6.67	18.4	252
SO9	351	141	.1	10.8	60.8	627
Bedrock we	lls					
FU426	41	.50	E .1	15.4	7.7	57
FU1265	195	2.20	.5	8.88	14.3	194
G1170	132	1,070	.5	6.42	2.4	1,920
G1351	146	600	.6	6.80	< .9	1,130
H267	57	.31	.6	16.5	2.1	71
HE1145	404	20.2	2.8	7.66	12.4	409
L475	203	110	E .1	9.17	8.5	361
MT406	357	110	.3	7.15	101	634
OE1458	276	6.94	.1	10.9	20.5	266
SA2198	388	132	.3	16.2	45.7	620
SO1374	448	13.0	.4	11.0	451	1,070
SO1384	293	150	.5	10.9	80.3	609
SO1408	215	1.00	.2	9.63	11.4	202
SO1411	377	235	.7	8.26	1.8	736

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

² Bicarbonate values calculated from alkalinity.

Table A5. Concentrations of nutrients and organic carbon in ground-water samples from the Mohawk River Basin, 2006.

[N, nitrogen; P, phosphorus; mg/L, milligrams per liter; (00623), USGS National Water Information System parameter code; <, less than; E, estimated value]

Well number¹	Ammonia plus organic-N, filtered, mg/L as N (00623)	Ammonia, filtered, mg/L as N (00608)	Nitrate plus nitrite, filtered, mg/L as N (00631)	Nitrite, filtered, mg/L as N (00613)	Ortho- phosphate, filtered, mg/L as P (00671)	Organic carbon, unfiltered, mg/L (00680)
Sand and g	ravel wells					
HE622	E 0.08	< 0.010	2.60	< 0.002	E 0.004	1
HE625	E .06	E.006	.43	< .002	E .005	< 1
HE1144	E .05	< .010	1.65	E .001	.007	< 1
HE1433	< .10	< .010	.43	< .002	.013	< 1
M281	< .10	E .010	.18	.003	.006	1.3
MT407	E .06	E .006	1.94	< .002	E .005	1
MT408	1.2	1.02	< .06	< .002	.014	< 1
OE1460	< .10	E.006	2.04	< .002	E .004	1.1
OE1462	E.10	.075	< .06	< .002	.007	< 1
OE1841	.17	.175	< .06	< .002	.008	< 1
SA1501	.48	.343	E .04	< .002	.007	1.6
SN135	< .10	< .010	.28	< .002	.143	1.1
SO9	.27	.227	< .06	< .002	E.005	1.1
Bedrock w	ells					
FU426	<.10	E .005	.15	< .002	.123	< 1
FU1265	.36	.377	< .06	< .002	.084	< 1
G1170	.22	.213	< .06	< .002	.019	< 1
G1351	.37	.351	< .06	< .002	.065	< 1
H267	< .10	.022	< .06	< .002	.039	< 1
HE1145	2.1	1.88	< .06	< .002	.014	< 1
L475	1.9	1.85	< .06	< .002	.007	< 1
MT406	.47	.409	< .06	< .002	.007	< 1
OE1458	.33	.163	< .06	< .002	.006	< 1
SA2198	.50	.479	E .03	< .002	.009	< 1
SO1374	1.6	1.67	< .06	< .002	.035	< 1
SO1384	E .08	.036	.60	< .002	.016	< 1
SO1408	< .10	.029	< .06	< .002	.007	< 1
SO1411	.37	.347	< .06	< .002	.031	< 1

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A6. Concentrations of trace elements and radionuclides in ground-water samples from the Mohawk River Basin, 2006.

[μ g/L, micrograms per liter; (01106), USGS National Water Information System parameter code; <, less than; E, estimated value; M, presence verified but not quantified; bold values exceed one or more drinking water standards]

Well number¹	Aluminum, unfiltered, µg/L (01105)	Antimony, unfiltered, µg/L (01097)	Arsenic, unfiltered, µg/L (01002)	Barium, unfiltered, µg/L (01007)	Beryllium, unfiltered, µg/L (01012)	Boron, filtered, µg/L (01020)	Cadmium, unfiltered, µg/L (01027)	Chromium, unfiltered, µg/L (01034)
Sand and g	ravel wells							
HE622	< 2	< 0.2	0.60	200	< 0.06	27	E 0.03	< 0.60
HE625	< 2	< .2	.15	20	< .06	9.8	< .04	E .50
HE1144	18	< .2	.46	3	< .06	E 6.2	< .04	.85
HE1433	4	< .2	.12	M	< .06	E 4.1	< .04	.88
M281	E 1	< .2	.66	58	< .06	20	< .04	< .60
MT407	< 2	< .2	.20	17	< .06	11	< .04	E .33
MT408	E 1	< .2	1.5	916	< .06	125	E .03	< .60
OE1460	< 2	< .2	.21	97	< .06	28	< .04	< .60
OE1462	< 2	< .2	17.8	69	E .03	24	< .04	< .60
OE1841	E 1	< .2	4.0	104	< .06	43	< .04	< .60
SA1501	< 2	< .2	.38	20	< .06	24	.05	E .45
SN135	< 2	< .2	.55	23	< .06	15	< .04	< .60
SO9	< 2	< .2	1.2	100	< .06	49	< .02	< .60
Bedrock we	ells							
FU426	348	<.2	.41	8	E .04	< 7.0	< .04	14.8
FU1265	49	< .2	4.8	70	< .06	35	< .04	< .60
G1170	59	< .4	16.4	475	< .12	104	< .08	E .30
G1351	5	<.2	6.5	261	< .06	264	E .03	< .60
H267	7	< .2	E .09	4	< .06	E 5.8	< .04	E .38
HE1145	5	< .2	E .07	118	< .06	608	< .04	E .57
L475	< 2	<.2	E .06	95	< .06	76	< .04	E .49
MT406	< 2	< .2	.91	105	< .06	139	< .04	< .60
OE1458	< 2	< .2	E .06	178	< .06	28	.19	E .36
SA2198	23	<.2	.79	171	< .06	76	< .02	< .60
SO1374	10	<.2	9.3	56	< .06	410	E .03	< .60
SO1384	E 2	.4	.20	51	< .06	345	< .04	E .56
SO1408	4	<.2	.39	56	< .06	329	< .04	E .31
SO1411	12	< .2	.14	147	< .06	399	< .04	< .60

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A6. Concentrations of trace elements and radionuclides in ground-water samples from the Mohawk River Basin, 2006.—Continued

[μ g/L, micrograms per liter; (01106), USGS National Water Information System parameter code; <, less than; E, estimated value; M, presence verified but not quantified; bold values exceed one or more drinking water standards]

Well number¹	Cobalt, unfiltered, µg/L (01037)	Copper, unfiltered, µg/L (01042)	Iron, filtered, µg/L (01046)	Iron, unfiltered, μg/L (01045)	Lead, unfiltered, µg/L (01051)	Lithium, unfiltered, µg/L (01132)	Manganese, filtered, μg/L (01056)	Manganese, unfiltered, µg/L (01055)
Sand and gi	ravel wells							
HE622	0.183	3.6	< 6	E 6	0.15	5.7	E 0.5	< 0.6
HE625	.114	.9	< 6	< 6	.09	1.3	< .6	< .6
HE1144	.113	4.7	< 6	580	2.13	1.9	1.6	18.1
HE1433	E .028	.6	8	150	.07	1.8	2.3	3.8
M281	.270	4.2	23	28	E .03	12.9	10.1	10.1
MT407	.163	1.6	8	29	.13	1.0	E.3	< .6
MT408	.173	2.3	1,230	1,170	.09	75.7	160	149
OE1460	.235	1.0	< 6	E 3	E .06	10.4	< .6	< .6
OE1462	.152	1.0	251	254	.34	10.4	33.0	31.1
OE1841	.240	1.2	1,240	1,210	< .06	18.1	41.8	38.4
SA1501	.742	1.2	E 4	7	.07	4.7	1,960	2,020
SN135	.150	3.2	E 4	E 4	.09	2.6	111	106
SO9	.418	E .8	1,140	1,180	< .06	7.6	349	390
Bedrock we	ells							
FU426	.389	26.2	81	8,190	.88	E .5	.8	93.7
FU1265	.138	E .5	229	326	.25	.8	50.2	49.9
G1170	.090	6.0	62	273	.46	111	306	306
G1351	E .027	1.2	18	52	E .03	352	12.0	12.1
H267	E .031	2.7	44	66	.13	< .6	23.2	22.0
HE1145	E .023	.7	< 6	19	< .06	824	< .6	E .5
L475	E .033	1.5	389	381	E .03	106	51.1	45.8
MT406	.260	3.0	154	143	< .06	14.5	19.9	18.3
OE1458	.106	3.8	88	84	E .03	9.1	24.0	20.0
SA2198	.556	3.3	< 6	980	2.21	28.0	220	279
SO1374	.378	9.7	2,190	2,150	.54	115	253	277
SO1384	.247	7.6	9	17	.13	38.7	5.3	5.3
SO1408	.109	E .6	39	49	.10	48.7	148	144
SO1411	.060	1.6	419	505	.12	608	97.6	97.5

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A6. Concentrations of trace elements and radionuclides in ground-water samples from the Mohawk River Basin, 2006.—Continued

[μ g/L, micrograms per liter; (01106), USGS National Water Information System parameter code; <, less than; E, estimated value; M, presence verified but not quantified; bold values exceed one or more drinking water standards]

Well number¹	Molybdenum, unfiltered, µg/L (01062)	Nickel, unfiltered, µg/L (01067)	Selenium, unfiltered, µg/L (01147)	Strontium, unfiltered, µg/L (01082)	Zinc, unfiltered, µg/L (01092)	Radon- 222, unfiltered, picoCuries per liter (82303)	Uranium, unfiltered, µg/L (28011)
Sand and g	ravel wells						
HE622	E 0.1	1.96	0.71	651	9	450	0.221
HE625	.3	.31	< .08	197	< 2	530	.210
HE1144	E.2	.76	.11	171	E 1	440	.245
HE1433	.2	.24	E .04	30.2	3	420	.166
M281	1.2	1.42	< .08	963	6	150	1.52
MT407	.2	.53	.23	178	3	270	.322
MT408	4.5	1.06	< .08	2,640	10	30	.364
OE1460	< .2	1.01	E .06	369	E 2	480	.151
OE1462	.9	.75	< .08	1,510	4	40	.579
OE1841	.5	.42	< .08	782	3	70	.314
SA1501	1.0	2.29	< .08	237	< 2	500	.236
SN135	.4	.89	.09	293	4	440	.242
SO9	1.0	.63	.28	549	2	80	.810
Bedrock we	ells						
FU426	.9	1.06	.40	34.2	11	670	.076
FU1265	4.4	.56	< .08	396	< 2	660	.107
G1170	2.5	2.48	E.10	601	16	770	.070
G1351	5.9	1.21	< .08	434	E 2	110	E.007
H267	3.8	.17	< .08	30.0	< 2	1,360	.104
HE1145	< .2	.38	.13	1,600	< 2	20	< .012
L475	E.1	.34	< .08	705	< 2	20	< .012
MT406	2.7	1.84	< .08	2,090	7	150	1.80
OE1458	2.1	1.40	< .08	614	3	60	.130
SA2198	.8	1.91	E .05	695	3	50	.723
SO1374	10.4	2.23	E .04	5,720	5	110	1.27
SO1384	.5	1.00	.08	31,100	E 2	250	.393
SO1408	.8	.44	< .08	437	E 1	1,020	.471
SO1411	6.3	.84	< .08	417	E 1	80	.201

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A7. Concentrations of pesticides and caffeine detected in ground-water samples from the Mohawk River Basin, 2006.

[µg/L, micrograms per liter; ESA, ethanesulfonic acid; SA, secondary amide; CIAT, 2-chloro-4-isopropylamino-6-amino-s-triazine; OIET, 2-Hydroxy-4-isopropylamino-6-ethylamino-s-triazine, OA, oxanilic acid; (04040), USGS National Water Information System parameter code; <, less than; E, estimated value; M, presence verified but not quantified; bold values indicate detections]

Well number¹	Acetochlor/ Metolachlor ESA SA, filtered, µg/L (62850)	CIAT, filtered, µg/L (04040)	OIET, filtered, µg/L (50355)	Alachlor ESA SA, filtered, µg/L (62849)	Atrazine, filtered, µg/L (39632)	Metolachlor ESA, filtered, µg/L (61043)	Metolachlor OA, filtered, µg/L (61044)	Metolachlor, filtered, µg/L (39415)	Prometon, filtered, µg/L (04037)
Sand and	gravel wells								
HE622	< 0.02	< 0.014	< 0.032	< 0.02	< 0.007	< 0.02	< 0.02	< 0.006	< 0.01
HE625	< .02	E .004	< .032	< .02	E .003	< .02	< .02	< .006	< .01
HE1144	< .02	< .014	< .032	< .02	< .007	.02	< .02	< .006	< .01
HE1433	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
M281	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
MT407	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
MT408	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
OE1460	< .02	E .033	< .032	< .02	.024	< .02	< .02	E .002	< .01
OE1462	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
OE1841	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
SA1501	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
SN135	< .02	E .006	E .005	< .02	.013	.12	< .02	E .003	M
SO9	< .02	< .014	< .080	< .02	< .007	< .02	< .02	< .010	< .01
Bedrock v	vells								
FU426	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
FU1265	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
G1170	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
G1351	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
H267	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
HE1145	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
L475	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
MT406	.05	< .014	< .032	.02	< .007	.42	.11	< .006	< .01
OE1458	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
SA2198	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
SO1374	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
SO1384	< .02	E .003	< .032	< .02	.008	< .02	< .02	< .006	< .01
SO1408	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01
SO1411	< .02	< .014	< .032	< .02	< .007	< .02	< .02	< .006	< .01

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A8. Concentrations of volatile organic compounds and phenolic compounds detected in ground-water samples from the Mohawk River Basin, 2006.

[μ g/L, micrograms per liter; (32730), USGS National Water Information System parameter code; <, less than; bold values indicate detections]

Well number¹	Total Phenolic Compounds, unfiltered, µg/L (32730)	Bromo- dichloro- methane, unfiltered, µg/L (32101)	Dibromo- chloro- methane, unfiltered, µg/L (32105)	Tetrachloro- ethene, unfiltered, µg/L (34475)	Toluene, unfiltered, µg/L (34010)	Tribromo- methane, unfiltered, µg/L (32104)	Trichloro- methane, unfiltered, µg/L (32106)
Sand and g	ravel wells						
HE622	< 4	0.3	0.7	< 0.1	< .1	0.8	0.2
HE625	< 4	< .1	< .2	< .1	< .1	< .2	< .1
HE1144	< 4	< .1	< .2	< .1	< .1	< .2	< .1
HE1433	< 4	< .1	< .2	< .1	< .1	< .2	< .1
M281	8	< .1	< .2	< .1	< .1	< .2	< .1
MT407	4	< .1	< .2	< .1	< .1	< .2	.1
MT408	< 4	< .1	< .2	< .1	< .1	< .2	< .1
OE1460	< 4	.1	.3	.4	.2	.5	.1
OE1462	< 4	< .1	< .2	< .1	< .1	< .2	< .1
OE1841	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SA1501	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SN135	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SO9	< 4	< .1	< .2	< .1	< .1	< .2	< .1
Bedrock we	ells						
FU426	< 4	< .1	< .2	< .1	< .1	< .2	< .1
FU1265	< 4	< .1	< .2	< .1	.2	< .2	< .1
G1170	< 4	< .1	< .2	< .1	< .1	< .2	< .1
G1351	< 4	< .1	< .2	< .1	< .1	< .2	< .1
H267	< 4	< .1	< .2	< .1	< .1	< .2	< .1
HE1145	< 4	< .1	< .2	< .1	< .1	< .2	< .1
L475	< 4	< .1	< .2	< .1	< .1	< .2	< .1
MT406	< 4	< .1	< .2	< .1	< .1	< .2	< .1
OE1458	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SA2198	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SO1374	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SO1384	< 4	< .1	< .2	< .1	< .1	< .2	< .1
SO1408	13	< .1	< .2	< .1	< .1	< .2	< .1
SO1411	< 4	< .1	< .2	< .1	< .1	< .2	< .1

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

Table A9. Bacteria in ground-water samples from the Mohawk River Basin, 2006.

[CFU, colony-forming unit; mL, milliliter; (78943), USGS National Water Information System parameter code; <, less than; E, estimated value; TNTC, too numerous to count; bold values exceed one or more drinking water standards]

Well number¹	Heterotrophic plate count, unfiltered, CFU/mL (78943)	Escherichia coli, unfiltered, Presence /Absence (50278)	Fecal coliform, unfiltered, CFU/100mL (31625)	Total coliform, unfiltered, CFU/100mL (31501)
Sand and gi	ravel wells			
HE622	< 1	Negative	E < 5	< 1
HE625	2	Negative	E < 5	3
HE1144	140	Negative	E < 5	< 1
HE1433	33	Negative	E < 5	< 1
M281	< 1	Negative	E < 5	< 1
MT407	< 1	Negative	E < 5	Positive*
MT408	35	Negative	E < 5	< 1
OE1460	13	Negative	E < 5	< 1
OE1462	1	Negative	E < 5	< 1
OE1841	23	Negative	E < 5	100
SA1501	< 1	Negative	E < 5	< 1
SN135	250	Negative	E < 5	< 1
SO9	< 1	Negative	E < 5	< 1
Bedrock we	ells			
FU426	4	Negative	E < 5	< 1
FU1265	< 1	Negative	E < 5	< 1
G1170	15	Negative	E < 5	< 1
G1351	46	Negative	E < 5	< 1
H267	3	Negative	E < 5	< 1
HE1145	77	Negative	E < 5	< 1
L475	70	Negative	E < 5	TNTC
MT406	2	Negative	E < 5	< 1
OE1458	2	Negative	E < 5	< 1
SA2198	110	Negative	< 2*	< 1
SO1374	17	Negative	E < 5	1
SO1384	< 1	Negative	E < 5	25
SO1408	< 1	Negative	E < 5	< 1
SO1411	E 940	Negative	E < 5	< 1

¹ FU, Fulton County; G, Green County; H, Hamilton County; HE, Herkimer County; L, Lewis County; M, Madison County; MT, Montgomery County; OE, Oneida County; SA, Saratoga County; SN, Schenectady County; SO, Schoharie County.

^{*} Processed with defined-substrate technology instead of membrane filtration.