

The **CAMEO**[®] Software System



CAMEO[®]

USER'S MANUAL

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U.S. ENVIRONMENTAL PROTECTION AGENCY

Office of Emergency Management

Washington, D.C.



**NATIONAL OCEANIC AND ATMOSPHERIC
ADMINISTRATION**

Office of Response and Restoration

Emergency Response Division

Seattle, Washington

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Welcome to CAMEO

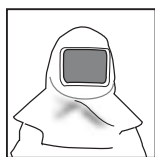
Welcome to CAMEO! This manual explains how to use the CAMEO® (Computer-Aided Management of Emergency Operations) software. It assumes that you already know how to use the Microsoft® Windows® or Apple® Macintosh® computer on which you're using your copy of CAMEO. This manual explains how to use CAMEO for various planning and response projects, including how to use the MARPLOT® and ALOHA® programs with other CAMEO components. However, to fully understand how MARPLOT and ALOHA work, you should read their separate manuals.

This chapter explains what CAMEO is, its basic functions, who designed it, and why and for whom it was developed. It explains how to get CAMEO and where to find technical support and training. It also explains about related software programs that might be useful to you.

CAMEO in brief

CAMEO is a suite of software programs you can use to plan for and respond to chemical emergencies. It was developed for chemical emergency planners and responders by the CAMEO team. That team includes the U.S. Environmental Protection Agency's Office of Emergency Management (EPA OEM, <http://www.epa.gov/emergencies>) and the National Oceanic and Atmospheric Administration's Office of Response and Restoration (NOAA OR&R, <http://response.restoration.noaa.gov>). CAMEO includes a set of databases (or modules), a hazard modeling program called ALOHA, and a mapping program called MARPLOT.

You can use CAMEO in two main ways:



RESPONSE

- To access, store, and evaluate information needed for emergency response to hazardous materials incidents. Firefighters, police officers, and other emergency workers know that a response can be hampered by lack of accurate information about the substance spilled and safe response actions. CAMEO is designed to help you quickly access just such information, just when you need it at the scene of a response.
- To develop hazardous materials emergency plans for your community. CAMEO is especially designed to help you meet the requirements of the Emergency Planning and Community Right-to-Know Act, or EPCRA (see “EPCRA” on page 261).



PLANNING

CAMEO was developed because NOAA recognized the need to assist first responders with easily accessible and accurate response information. Since 1988, EPA OEM and NOAA OR&R have collaborated to further the development of CAMEO. The U.S. Census Bureau and the U.S. Coast Guard also have worked with EPA and NOAA to enhance CAMEO.

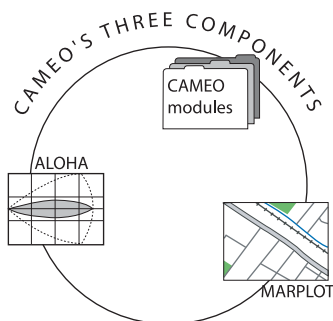
Who uses CAMEO?

Within the U.S., CAMEO's primary users include firefighters, State Emergency Response Commissions (SERCs), Tribal Emergency Response Commissions (TERCs), Local Emergency Planning Committees (LEPCs), industry, schools, environmental organizations, and police departments.

Outside of the U.S., CAMEO is in use in many countries and has been translated into French and Spanish (ALOHA has also been translated into Swedish and Korean). CAMEO was selected by the United Nations Environment Programme (UNEP, <http://www.unep.org>) as a tool for helping developing nations prepare for and respond to chemical accidents, and is part of the UNEP's Awareness and Preparedness for Emergencies at the Local Level (APELL) program. It has been demonstrated or taught in 50 countries as part of the APELL workshops on community preparedness for chemical accidents.

CAMEO's three components

CAMEO contains three integrated components:



The CAMEO modules. A set of databases (or modules) contains information for responders and planners. You can quickly search the Chemical Library module for information and response recommendations for more than 6,000 hazardous substances. The Chemical Library contains information and recommendations on fire and explosive hazards, health hazards, firefighting techniques, cleanup

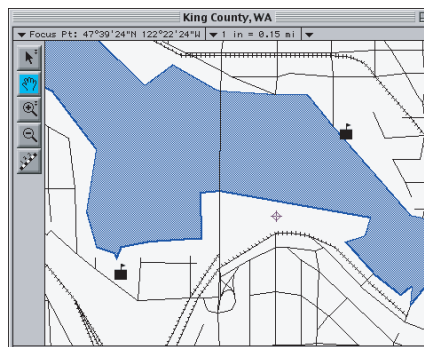
The screenshot shows a web-based interface for the "Chemical Library". The "Chemical Name" field contains "CHLORINE". Below this, there are two main sections: "Chemical Identification Information" and "Response Information Data Sheets". The "Response Information Data Sheets" section includes a grid of tabs for "Firefighting", "Fire Hazards", "Non-Fire Response", "Health Hazards", "Protective Clothing", "General Description", "Properties", "Reactivity", "Reactive Hazards", and "First Aid". The "General Description" tab is active, displaying the following text:

A greenish yellow gas with a pungent suffocating odor. Toxic by inhalation. Slightly soluble in water. Liquefies at 35°C and room pressure. Readily liquefied by pressure applied at room temperature. Density (as a liquid) 13.0 lb / gal. Contact with unconfined liquid can cause frostbite by evaporative cooling. Does not burn but, like oxygen, supports combustion. Long-term inhalation of low concentrations or short-term inhalation of high concentrations has ill effects. Vapors are much heavier than air and tend to settle in low areas. Contact CHEMTREC to activate chlorine response team 800-424-9300. Used to purify water, bleach wood pulp, and to make other chemicals.

Rate of onset: Immediate to hours
Persistence: Minutes to hours

procedures, and protective clothing recommendations (a view of a record in the Library is shown above). You can use other CAMEO modules to maintain records on facilities that store chemicals, the inventories of chemicals at those facilities ("Tier II data"), emergency planning resources and contacts, and special locations (such as schools and hospitals that you might need to contact quickly during an emergency). For descriptions of all of CAMEO's modules, see "[CAMEO's modules in brief](#)" on page 24.

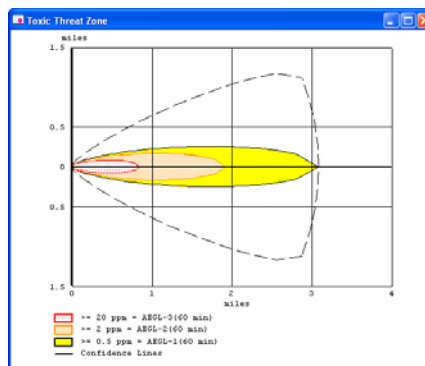
MARPLOT. Use MARPLOT (Mapping Application for Response, Planning, and Operational Tasks) to view and print maps of your community that show roads, facilities, schools, response assets, and other information useful for response and planning tasks (like the Seattle map at right). On your map, you also can overlay predicted hazard zones—areas that could be impacted by chemical releases.



MARPLOT maps are created from U.S. Census Bureau TIGER/Line® files, and can be downloaded at no cost (<http://www.epa.gov/osweroe1/content/cameo/marmaps/>). (To learn how CAMEO and MARPLOT work together, see “Using MARPLOT with CAMEO” on page 181; for a complete explanation of how to use MARPLOT, download the MARPLOT manual from <http://www.epa.gov/osweroe1/content/cameo/marplot.htm>.)

ALOHA. Use ALOHA (Areal Locations of Hazardous Atmospheres) to model key hazards—toxicity, flammability, thermal radiation (heat), and overpressure (explosion blast force)—related to chemical releases that result in toxic gas dispersions, fires, and/or explosions. (**Note:** In versions prior to ALOHA 5.4, only the toxic threat is modeled.)

ALOHA’s predictions are based on the characteristics of the released chemical, atmospheric conditions, and the circumstances of the release. ALOHA displays a threat zone plot (like the one shown at right) showing one or more areas where a hazard—toxicity, flammability, thermal radiation, or damaging overpressure—may exceed key Levels of Concern (LOCs) and pose a threat to people and property.



You can also plot ALOHA threat zones on a map in MARPLOT. On the same map, you also can check the locations of vulnerable locations (such as hospitals and schools) to see which ones could be affected by a chemical release. You then can extract information such as emergency phone numbers for these locations from CAMEO modules. (To learn how ALOHA works together with CAMEO and MARPLOT, see [“Using ALOHA with MARPLOT and CAMEO” on page 192](#); for a complete explanation of how to use ALOHA, download the ALOHA manual from <http://www.epa.gov/osweroe1/content/cameo/aloha.htm>.)

CAMEO supports EPCRA planning work

CAMEO was designed to help you to meet your responsibilities under the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA). EPCRA establishes requirements for Federal, state, and local governments, and for industry for emergency planning and “community right-to-know” reporting on hazardous and toxic chemicals (see [“EPCRA” on page 261](#)). You can use CAMEO for EPCRA-related tasks like:

- maintaining and reviewing Tier II information submitted by facilities in your local area (see [“Tier II information and CAMEO” on page 108](#)), or
- assessing the hazards to your community from accidental releases of locally stored hazardous chemicals (see [“Using Screening & Scenarios to assess hazards” on page 139](#)).

CAMEO and the Clean Air Act. Many CAMEO users, especially facility operators, also must meet certain responsibilities under Section 112(r) of the Clean Air Act (CAA) Amendments of 1990 (see [“CAA 112\(r\)” on page 270](#)). You may be able to use CAMEO for some of your CAA-related projects. For example, you can check a substance’s Chemical Library record to find out whether the substance is covered under Section 112(r), and, if it is, to see the threshold quantity designated for it under Section 112(r). However, note that you cannot use CAMEO’s Screening & Scenarios module for the offsite consequence analyses required under Section 112(r). For more on this topic, see [“What are the differences between Screening & Scenarios, ALOHA, and RMP endpoint distances?” on page 155](#).

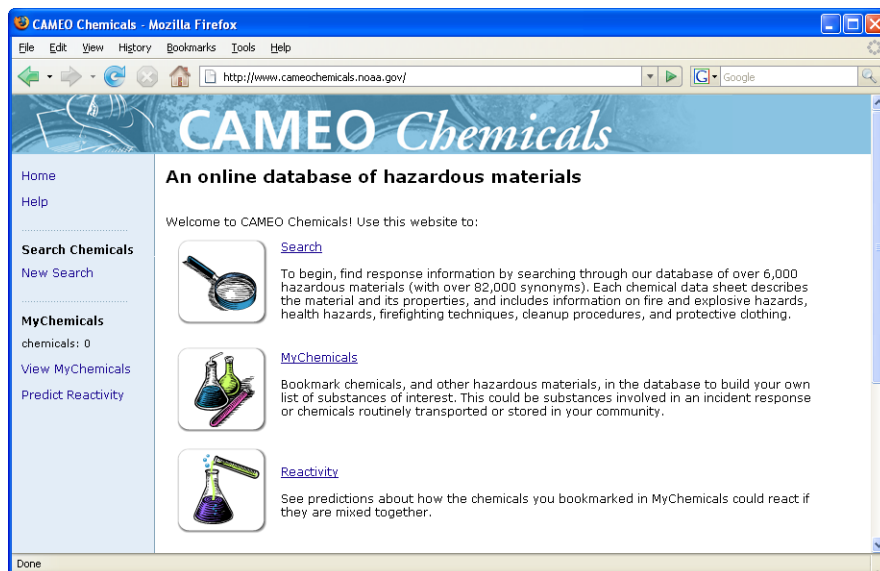
CAMEO and terrorism

CAMEO's Chemical Library and the ALOHA modeling program were designed to help people plan for and respond to accidents involving industrial chemicals, rather than deliberate releases of chemical or biological warfare agents. For people concerned with planning for or responding to terrorist attacks, the two programs can be most useful for planning for or responding to terrorist actions that result in releases of industrial chemicals. However, some substances that are considered as nerve agents rather than industrial chemicals are included in the Chemical Library and in ALOHA's built-in chemical database. Examples of nerve agents in the Chemical Library and/or ALOHA database include tabun, sarin, mustard gas, lewisite, hydrogen cyanide, cyanogen chloride, chlorine, phosgene, tear gas, and mace® (chloroacetophenone).

Not all common nerve agents are included in ALOHA. Included are only those expected to be volatile enough on their own (without use of explosives) to get into the air in high enough concentrations to pose a hazard to people. Most nerve agents, in contrast, are oily liquids with very low vapor pressures. If you're considering using ALOHA in response or planning for nerve agent attacks, we encourage you to carefully review the discussion of ALOHA's limitations in the ALOHA manual and online at <http://response.restoration.noaa.gov/aloha>. ALOHA can't model nerve agent scenarios that involve indoor releases, releases requiring an explosive propellant, mixtures of chemicals, or nerve agents that exhibit low volatility at ambient temperatures. If you need to do sophisticated modeling of realistic deliberate nerve agent release scenarios, we recommend that you contact chemical warfare specialists for help, rather than relying on CAMEO and ALOHA.

CAMEO Chemicals: An online version of CAMEO

Now you can access much of the CAMEO information online on the CAMEO Chemicals website: <http://cameochemicals.noaa.gov>. You can search the extensive CAMEO chemical database, view the response data sheets, find out how chemicals could react if they mixed, and print customized reports. **Note:** ALOHA, MARPLOT, and the EPCRA modules are not available online.



Getting training and staying informed

To learn CAMEO basics, work through “[A CAMEO guided tour](#)” on [page 36](#). Later, you may want to browse through the following sections of this manual that describe how to complete key CAMEO tasks. If you’re interested in

- reviewing CAMEO’s response recommendations and other information about substances of concern, see “[Searching for a Chemical Record](#)” on [page 80](#) and “[Understanding the information in the Chemical Library](#)” on [page 85](#).

- adding records describing chemical facilities and their chemical inventories, see “[Creating and editing facility records](#)” on page 122.
- importing or exporting data to or from CAMEO, including Tier II data importing, see [Chapter 6, “Importing and Exporting Data,”](#) beginning on page 197.
- planning how you’ll use your CAMEO system, on or off a network, see “[A CAMEO management plan](#)” on page 223.

CAMEO website

Check the CAMEO website (<http://www.epa.gov/osweroe1/content/cameo>) for more CAMEO-related resources. You can:

- view news items and download software updates and training materials.
- troubleshoot CAMEO, ALOHA, or MARPLOT problems, using the CAMEO Troubleshooter (<http://www.epa.gov/osweroe1/content/cameo/shoot.htm>).
- check CAMEO training and events calendars (<http://www.epa.gov/osweroe1/content/cameo/cam-evnt.htm>). Trainers certified by NOAA and EPA run frequent CAMEO and ALOHA training events around the U.S., and sometimes in other countries as well.

CAMEO News Service

We encourage you to subscribe to the CAMEO News Service, an email announcement list (one-way listserve) used by CAMEO managers and developers to keep the CAMEO community informed. Bulletins are sent to list subscribers on an approximately monthly basis. Subscribing to the news service is a way to get access to updated information. Check the CAMEO website to find out how to subscribe to the CAMEO News Service.

Three other programs for CAMEO users

Tier2 Submit

Each year, facilities covered by EPCRA must submit an emergency and hazardous chemical inventory form to their Local Emergency Planning Committee (LEPC), State Emergency Response Commission (SERC), and local fire department. Facilities provide either a Tier I or Tier II form. Most states require the Tier II form. Tier II forms require basic facility identification information, employee contact information, and information such as storage amounts, storage conditions, and locations for chemicals stored or used at the facility. (See “[Sections 311 and 312: community right-to-know requirements](#)” on page 266.)

Tier2 Submit™ is a free chemical inventory software program developed by the CAMEO development team for both reporting facilities and for state and local emergency responders:

- Facilities can use Tier2 Submit to manage, print, and send electronic Tier II reports.
- State and local emergency responders can import those facilities’ reports directly into their copies of CAMEO. By importing facility reports, responders can have current chemical inventory information available to them if they need to respond to an incident at a reporting facility.

Please check <http://www.epa.gov/osweroe1/content/epcra/tier2.htm> to find out whether your state is participating and/or to download a copy of Tier2 Submit.

LandView

LandView® incorporates a database management system along with the MARPLOT mapping program. You can use LandView not only to view maps of U.S. counties or other geographic areas but also to work with environmental and census data describing those counties, contained in databases from the U.S. Environmental Protection Agency, U.S. Census Bureau, and U.S. Geological Survey. For example, you can estimate the population and number of housing units within a given radius from a location such as a chemical facility. To learn more about LandView or to purchase a copy, visit the LandView website at <http://landview.census.gov>.

RMP*Comp

RMP*Comp is a free program you can use to complete the offsite consequence analyses (both worst case scenarios and alternative scenarios) required under the Risk Management Planning Rule, which implements Section 112(r) of the 1990 Clean Air Act (see “[CAA 112\(r\)](#)” on page 270). You can download RMP*Comp or obtain more information about Section 112(r) from http://www.epa.gov/osweroe1/content/rmp/rmp_comp.htm. Offsite consequence analyses are like the hazards analyses required under EPCRA, but they aren’t quite the same. If you’re confused by the apparent similarities and not sure what the differences are, see “[What are the differences between Screening & Scenarios, ALOHA, and RMP endpoint distances?](#)” on page 155.

Getting Started

This chapter contains an overview of the basic features of CAMEO and an explanation of how you use them. It also includes a Guided Tour for learning how to use CAMEO to complete basic planning and response tasks.

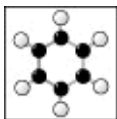
CAMEO's components

The complete CAMEO system consists of two kinds of components:

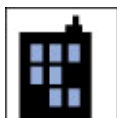
- *Modules* (or databases): the Chemical Library of response-related information about thousands of hazardous substances, and eight other modules you use for response and planning record-keeping.
- *Programs*: ALOHA (a hazard model) and MARPLOT (a mapping program).

To complete various tasks, you're likely to use two or three CAMEO components in sequence. For example, suppose you want to assess the potential hazard posed to your community by a particular chemical in the inventory of a local chemical processing plant. As part of your investigation, you might (1) find the record for that plant in CAMEO's Facilities module, then (2) view the location of the plant on a map in MARPLOT. You then might (3) check the record in CAMEO's Chemicals in Inventory module for the chemical of concern in the plant's inventory. You then might (4) use ALOHA and MARPLOT to predict the extent of the area that could be at risk, if that chemical were to be released.

CAMEO's modules in brief



The **Chemical Library** module contains records for more than 6,000 hazardous substances. Each record describes a chemical or substance, including its chemical name, trade names and other synonyms, identification numbers, regulatory information, and labeling conventions. For emergency responders and planners, the most important part of each chemical record is the Response Information Data Sheets (RIDS) section, which contains a general description of the chemical, its physical properties, fire and health hazards, and recommendations for firefighting and non-fire response, first aid, and protective clothing for response.



Use the **Facilities** module to store information about facilities where chemicals are maintained, including address, emergency contact information, and site plans showing the layout of the facility. You can link any record in this module to an electronic map so that you can quickly see where a facility is located in your community.



Use the **Chemicals in Inventory** module to maintain inventory records for stored chemicals, including descriptions of each chemical's physical state, storage conditions and locations, and quantities routinely on site.



Use the **Contacts** module as a telephone and address directory of important contacts, such as chemical experts, government agencies and organizations, contacts for facilities, response resources, and other people who can help you with emergency planning or response.



Use the **Incidents** module to track information about accidental spills of hazardous chemicals, either at fixed facilities or along transport routes. You can link records in this module to symbols shown on electronic maps, so that you can quickly find the location of an incident on a map.



Use the **Screening & Scenarios** module to make hazards analyses for your community, using procedures described in the guidebook, *Technical Guidance for Hazards Analysis* (see “[Bibliography](#)” on page 272). In such a hazards analysis, you estimate the extent of the area around a facility, or other potential spill site, that might be affected by the accidental release of a chemical of concern. You then can plot this area (called a *threat zone*) on an electronic map in MARPLOT.



Use the **Special Locations** module to keep descriptions of schools, nursing homes, hospitals, and other facilities that require special protection and consideration during emergencies. You can link records in this module to symbols shown on electronic maps, so that you can quickly find a special location on a map.



Use the **Routes** module to compile information about routes commonly used to transport chemicals. You also can maintain information about the particular chemicals transported along a given route, and you can link records in this module to routes shown on electronic maps, so that you can quickly find a route on a map.



Use the **Resources** module to maintain information about resources (such as supplies, experts, and contractors) that you may need to quickly access during a response, or that are useful to you in your planning activities. You can link Resources records to symbols on an electronic map so that you can quickly find resources on your map.

Basic CAMEO module operations

The following explanations of basic operations and features apply to the CAMEO modules only. ALOHA and MARPLOT look and work differently. Check their manuals to learn how to use them (download those manuals from the MARPLOT and ALOHA download pages at <http://www.epa.gov/osweroe1/content/cameo/request.htm>).

Anatomy of a CAMEO module. Each module in CAMEO is a **database**. A database is just an organized collection of information; CAMEO's modules are computer databases. Each CAMEO module includes the same standard components as other computer databases. If you've worked with any database program, you already know those components:

- **records.** A database is composed of records. Each record is a compilation of information about a particular entity. For example, in CAMEO's Facilities module, each record describes a particular facility (or division of a large facility).
- **data fields.** Each record is composed of data fields, and each data field contains a particular type of information. In the Facilities module, one data field contains a facility's address; another holds its zip code, and so on.

CAMEO's modules are interrelated. For example, in the Chemicals in Inventory module, you can keep track of the chemical inventories of facilities in your Facilities module. Likewise, in the Contacts module, you can keep information about contact people, such as representatives of:

- facilities (described on Facilities records).
- special locations (described on Special Locations records).
- response organizations (described on Resources records).

You can quickly access information in one module when you're working with a related record in another (the Guided Tour shows how to do this).

Home and navigation

CAMEO's Home is a navigation window containing buttons you use to quickly navigate among CAMEO's modules and programs, or to quickly complete basic CAMEO tasks, like searching the Chemical Library for a chemical(Figure 2-1).

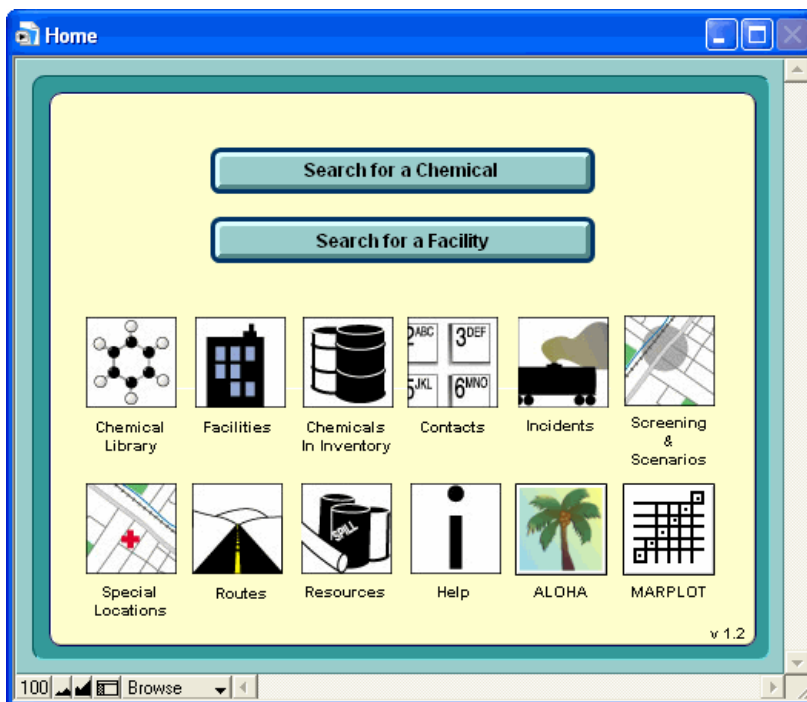



FIGURE 2-1. CAMEO's Home navigation window.



To display the Home navigation window, either select Go Home from the File menu, or click the Home button in the toolbar.


The Toolbar


CAMEO contains a simple toolbar that's visible when any CAMEO module is open (except when you're searching the module or editing a record). It contains:


- *Buttons for common operations.*

Click  to open the Home navigation window.

Click  or  to switch between **List view**, where you see a list of records in the module, and **Record view**, where you see the contents of a single record. See [“List and Record views” on page 29](#).


Click  to perform a new search within the open CAMEO module.


Click  when you need to edit the record you're currently viewing (see [“Editing records” on page 30](#)). **Note:** This button does not appear in the Chemical Library, because you can't edit the records in that module.


Click  to go to the list of Help topics.

- *Navigation buttons.* Click the four navigation buttons on the right side of the Toolbar to move forwards or backwards through the records in a module:

 Click to see the *first* record in the module.

 Click to see the *previous* record.

 Click to see the *next* record.



 Click to see the *last* record.

List and Record views

Name
Green Valley Water Facility (DEMO)
Adams Petroleum Refinery (DEMO)
General Hardware and Garden
M & S Chemicals (DEMO)
Abalone Press (DEMO)

Whenever you have a CAMEO module open, you're viewing that module in either List or Record view. When you first open a module, you're in List view, where you see a list of all the records in that module. The example at left shows part of the record list in the Facilities module. Records appear in the list in the order in which you create them, unless you sort them into a different order (see [“Sorting records in CAMEO modules” on page 251](#)). In List view, you click on a record name to select that record. Once a record is selected, a small black rectangle appears to the left of its name. In the example at left, General Hardware and Garden is selected.

In Record View, you see one or another of the records in that module. When you're in Record View, you can view the information contained in all the individual fields on that record.

You can go back and forth between the two views by clicking the  and  buttons in the toolbar. If you're in List view, you also can double-click on the name of any record in the list to see that record in Record view.


Searching and browsing

You can **browse** through the records in a CAMEO module in four ways:

- in List view, scan the list of records visually until you find the one you want. Use the scrollbar to review the entire list if it's too long to fit in the window.
- in Record view, click the navigation buttons on the right side of the toolbar to navigate from record to record.
- click the pages of the book icon in the upper left corner of the module window to page through the records in the module (the book icon is shown at left).
- from the Record menu, select Go To, then select either First, Previous, Next, or Last to navigate to the first, previous, next, or last record in the module.



If you can't find a record you need by browsing, **search** the module to find that record:

1. Click the  button or select Start Search from the Search menu.
2. For a quick, simple search, just type the word(s) or number(s) you're looking for—for example, the name of a chemical in the Chemical Library—then press Search.

To learn how to make more complex searches, see [“Searching” on page 246](#).

Using the Show Related command

You can associate records in your CAMEO modules with other records to which they're logically related. For example, you can create a record for a school in the Special Locations module, then create a Contacts record for that school's principal. That Contacts record for the principal is then automatically associated with the school's record in Special Locations.

When you're working with a record that's associated with one or more other records, select Show Related from the Record menu to quickly access the records associated with that record ([Figure 2-2 on page 31](#)). You'll also need to indicate the module of interest. For example, from the Facilities record for a local chemical processing plant, select Show Related, and then Contact Records, to view the Contacts records associated with the plant.

Editing records

You can edit records in CAMEO's modules in three ways:

- You can manually change records in any CAMEO module *except* the Chemical Library. You can't make any changes to the Chemical Library.
- You can import Tier II data (see [“Importing data from Tier2 Submit” on page 205](#)).
- You can import data from someone else's copy of CAMEO.

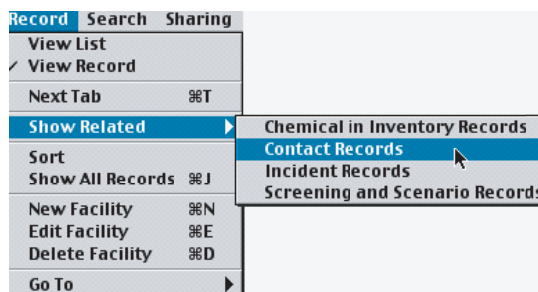



FIGURE 2-2. The Show Related command, used to find the Contact records associated with the selected Facilities record.

Whenever you start CAMEO, you're in Browse mode. To make changes to your CAMEO records, you must switch to Edit mode by pressing  in the toolbar.

On a CAMEO record in Edit mode, different data fields have different appearances: one may be a text box (in which you'd type, say, the amount of a stored chemical), while another might be a menu (from which you'd choose amount units: e.g., pounds, tons, or kilograms). To tell the difference, click within the data field: a menu will drop down so you can select from all the choices it contains; a text box will not (Figure 2-3).

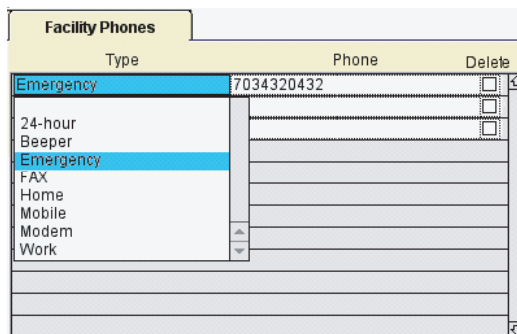
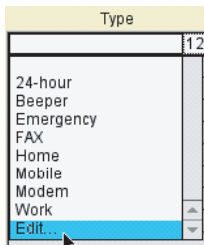


FIGURE 2-3. Type is a menu that drops down when you click within the data field (in Edit mode); Phone is a text box.



When you're in Edit mode, you can edit some of the menus that appear in CAMEO records. You can tell whether a menu can be edited by looking through the choices in the menu. In an editable menu, "Edit" is at the bottom of the list of menu choices. To edit a menu:

1. Click the Edit button in the toolbar.
2. Click within the menu data field to drop down the menu.
3. Select Edit from the menu (as at left). A list of all the items in the menu is displayed.
4. Add, remove, or edit items in the menu, then click OK.
5. Click Save Changes.

To delete a record from a CAMEO module, select that record, then select Delete <module name> from the Record menu (e.g., in the Special Locations menu, select Delete Special Location; in the Facilities module, select Delete Facility, as in Figure 2-4).

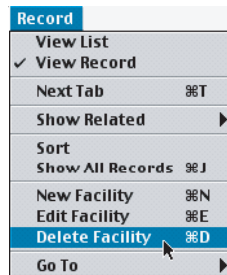
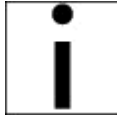


FIGURE 2-4. Deleting a Facility record.



Getting Help. Whenever CAMEO is running on your computer, you can access its help system, if you have installed a web browser (such as Internet Explorer® or Netscape Navigator®) on your computer. **Note:** You do not have to be online to access the help content.

To view the help, click the Help button (shown at left) in the Home navigation window, select CAMEO Help from the Help menu, or click the Help button in each module record or list view. Your web browser will start up and display the table of contents for the help. Click on any topic title to read about that topic. Whenever you want to return to the table of contents, click the TOC button in the upper right corner of the Help window. If you would rather look up a topic in the help index, click the Index button.

Things to ignore

There are some items in CAMEO's windows that you won't need to use directly to perform your work with CAMEO. However, these items are needed in order for CAMEO to function normally. Just ignore the following items—but please don't try to remove them:



- the Scripts menu.
- the button just above the book icon in the upper left corner of CAMEO's windows (shown at left).
- in Windows, the minimized window buttons that appear along the bottom of your screen. CAMEO needs to keep certain files open in order to complete its operations.

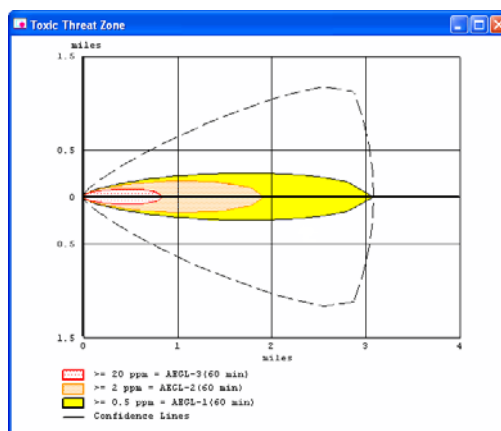
ALOHA and MARPLOT

ALOHA and MARPLOT aren't modules, but are programs included in CAMEO (you also can use either of them separately from CAMEO). To use them in CAMEO, you use the Sharing menu, which you'll find in the menu bar of all CAMEO components (all the modules, as well as ALOHA and MARPLOT).

ALOHA

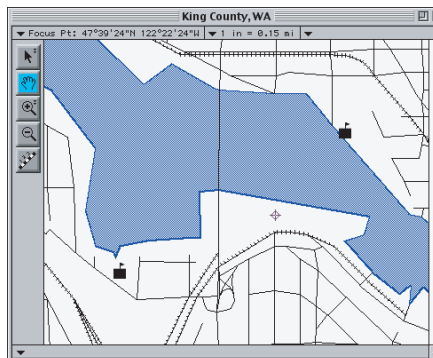
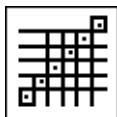


You use ALOHA to model key hazards—toxicity, flammability, thermal radiation (heat), and overpressure (explosion blast force)—related to chemical releases that result in toxic gas dispersions, fires, and/or explosions. (**Note:** In versions prior to ALOHA 5.4, only the toxic threat can be modeled.) Once you enter a description of a chemical release into the model, ALOHA displays a threat zone plot, like the one at right, representing the area(s) at risk. You can also plot ALOHA threat zones on a map in MARPLOT.



To use ALOHA with CAMEO, first find and open the record for a chemical in CAMEO's Chemical Library, then choose Select This Chemical in ALOHA from CAMEO's Sharing menu. If the chemical is in ALOHA's chemical library (less than 1,000 CAMEO chemicals have been judged toxic enough and volatile enough to be potential airborne hazards to be modeled in ALOHA), its property data will be loaded into ALOHA, and you then can begin to enter the details of a release scenario into ALOHA. Otherwise, you will be alerted that the chemical is not in ALOHA's library. For more details on using ALOHA with CAMEO and MARPLOT, see [“Using ALOHA with MARPLOT and CAMEO”](#) on page 192.

MARPLOT



Use MARPLOT to display and edit electronic maps (like the one at right). You can link CAMEO records for facilities, special locations, transport routes, and other items of interest to symbols representing those locations on maps in MARPLOT (see [“Linking map objects to CAMEO records” on page 187](#)). You also can plot threat zones from Screening & Scenarios or ALOHA on a map of your community.

To link a CAMEO record and map object, first create both the record and the map object (check your MARPLOT manual to learn how to create map objects). Next, in MARPLOT, select the object, then, in the Sharing menu, select CAMEOfm, then Link Object. In CAMEO, find and select the record, then, from the Link menu, select Link this Record.

To plot a Screenings & Scenarios threat zone in MARPLOT, open the Screening & Scenarios record in Record view, then click Show on Map. To plot an ALOHA threat zone on a MARPLOT map, while the threat zone is visible in ALOHA, click on the location of the release on the map in MARPLOT, then, in MARPLOT's Sharing menu, point to ALOHA, then click Set Source Point.

For more details on using MARPLOT with CAMEO, see [“Using MARPLOT with CAMEO” on page 181](#).

A CAMEO guided tour

Take this tour of a planning and response scenario for Prince William County, Virginia, to learn how to use CAMEO along with MARPLOT to complete common emergency planning and response tasks.¹ We've entered sample data into CAMEO for the purpose of this tour. When you've finished the tour, you can delete the sample data (we'll explain how to do this at the end of the tour).

Preparing for the tour

To complete the tour, you must already have installed CAMEO and MARPLOT. If you haven't already done that, follow the instructions on the CAMEO website (<http://www.epa.gov/osweroe1/content/cameo/cameo.htm>).

Note to LandView users: If you have installed LandView on your computer, you also have MARPLOT (which is a component of LandView). Do not install a second copy of MARPLOT.

You also need two MARPLOT maps, which are automatically installed when you install CAMEO and MARPLOT:

- a map of Prince William County, Virginia. This map is installed when you install MARPLOT. (If you have installed LandView, either use the disk with the Prince William County, Virginia, map or download a free copy from <http://www.epa.gov/osweroe1/content/cameo/marmaps>.)
- the CAMEO Map (when you install the CAMEO modules, this map is installed into the CAMEOMAP folder inside the CAMEO folder).

1. This tour does not cover ALOHA. To learn to use ALOHA, take the guided tour in the ALOHA manual (download it from <http://www.epa.gov/osweroe1/content/cameo/aloha.htm>).

A potential chemical hazard

Imagine that you are a member of the Local Emergency Planning Committee (LEPC) for Prince William County.² At the committee's most recent meeting, you were asked to complete two projects:

- evaluate the potential hazard to the county posed by the Green Valley Water Facility. You'll use the hazards analysis procedures outlined in the guidebook, *Technical Guidance for Hazards Analysis* (to find out how to obtain a copy, see [“Bibliography” on page 272](#)).
- update the records describing this facility in CAMEO.

At its site in Prince William County, the Green Valley Water Facility stores and uses chlorine in amounts exceeding the Threshold Planning Quantity (TPQ) of 100 pounds ([Figure 2-5 on page 38](#)). Chlorine is a designated Extremely Hazardous Substance (EHS) that is frequently shipped to the facility along James Madison Highway—a major route that passes through populated areas within the county. Because the facility stores an EHS in quantities greater than the TPQ, some of the requirements of EPCRA (Emergency Planning and Community Right-to-Know Act) come into play. In particular,

- the facility's operator must submit a chemical inventory report (called a **Tier II form**) annually to state and local emergency planning authorities and the fire department.
- the LEPC must evaluate the potential hazard posed by the facility, as part of its emergency response planning work.

2. An LEPC's role is to design, then regularly review and update a comprehensive emergency plan for a particular local emergency planning district. See [“Section 301-303: emergency planning” on page 263](#).

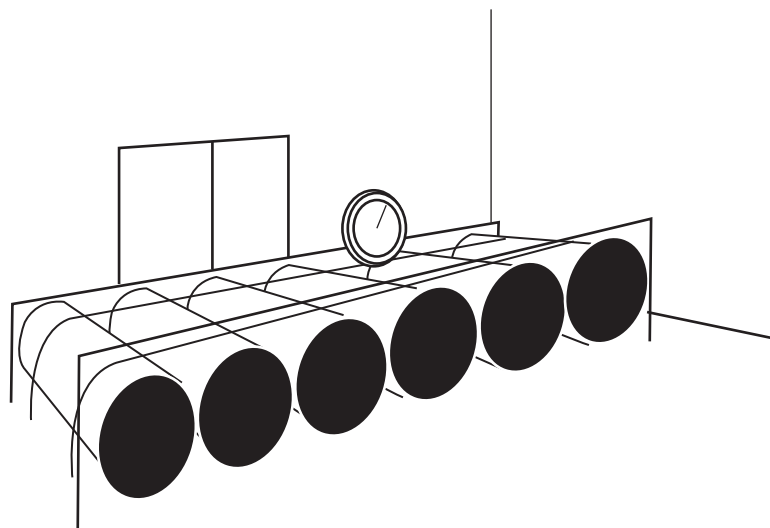


FIGURE 2-5. The chlorine storage room in the facility.

To begin your analysis,

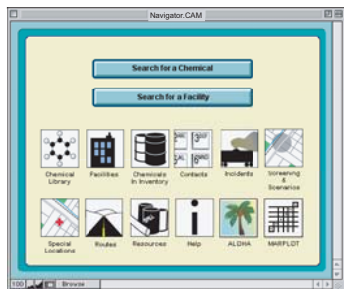
- you obtained the facility’s most recent Tier II chemical inventory form, indicating the amounts, locations, and storage conditions for the chlorine stored at the facility (for more on Tier II reporting, see “[Sections 311 and 312: community right-to-know requirements](#)” on page 266).
- you talked with the facility’s senior manager, Ms. Angela Wong, to familiarize yourself with daily operations at the plant.

Starting CAMEO

Begin the Guided Tour by starting CAMEO:

- in Windows, in the Start menu, point to Programs, then select CAMEO. CAMEO will start up, and open to the Home navigation window.
- on a Macintosh, open the CAMEO folder and double-click on CAMEOfm to start the program.

CAMEO’s Home (at left) is the starting point for common CAMEO tasks.




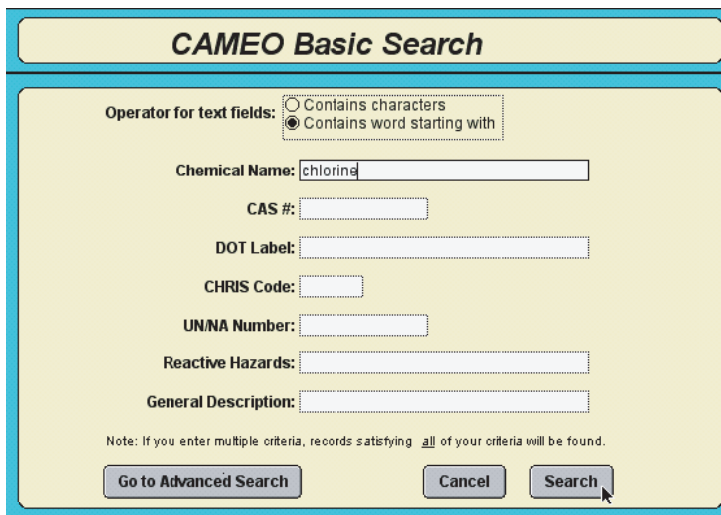
Assessing chlorine's hazards

The first step in your assessment is to familiarize yourself with the hazards posed by chlorine. The Chemical Library contains information about chlorine in a record for that chemical, so you'll search the library for this record.

You can search the Chemical Library for many kinds of information about a given substance: its chemical name (including trade names or synonyms), formula, United Nations accession number, Chemical Abstracts Service registry number, labeling requirements, or other identifying attributes.

To search the Chemical Library for the record for chlorine:

1. In the Home navigation window, click . (If the Home navigation window isn't visible, from the File menu, choose Go Home.)
2. In the Chemical Name box, type "chlorine" and then press Search (as shown below).



CAMEO Basic Search

Operator for text fields: Contains characters
 Contains word starting with

Chemical Name:

CAS #:

DOT Label:

CHRIS Code:

UN/NA Number:

Reactive Hazards:

General Description:

Note: If you enter multiple criteria, records satisfying all of your criteria will be found.

CAMEO finds all the records in the Chemical Library for substances that have "chlorine" in their name or in a name synonym. It displays an alphabetical list of those records.

Tip: This particular search will work whether you choose to search for a name that **contains the characters** “chlorine” or a name that **contains a word starting with** those characters. (Searches for a word *starting with* certain characters will typically find fewer records.)

Note: This list of records is an example of a **found set**. A found set is created in a CAMEO module when you search that module. It includes all the records meeting the criteria of your search (in this case, your criterion was “has ‘chlorine’ in the name or synonym”). While there’s a found set in place within a module, you can see just those records. To see all the records, you need to clear the found set by selecting either Clear Search from the Search menu or Show All Records from the Record menu.

3. Find and double-click “CHLORINE” in the list.

You’re now viewing the record for chlorine (shown below). It contains information about chlorine under two main tabs:

Chemical Identification Information—this section contains (a) information used to identify the chemical and (b) regulatory information.

Response Information Data Sheets (RIDS)—the RIDS section contains emergency response recommendations and information for responders and planners.

Home View List New Search Help

Chemical Library

Chemical Name
CHLORINE

Chemical Identification Information **Response Information Data Sheets**

Firefighting Fire Hazards Non-Fire Response Health Hazards Protective Clothing

General Description Properties Reactivity Reactive Hazards First Aid

A greenish yellow gas with a pungent suffocating odor. Toxic by inhalation. Slightly soluble in water. Liquefies at -35°C and room pressure. Readily liquefied by pressure applied at room temperature. Density (as a liquid) 13.0 lb./gal. Contact with unconfined liquid can cause frostbite by evaporative cooling. Does not burn but, like oxygen, supports combustion. Long-term inhalation of low concentrations or short-term inhalation of high concentrations has ill effects. Vapors are much heavier than air and tend to settle in low areas. Contact CHEMTREC to activate chlorine response team 800-424-9300. Used to purify water, bleach wood pulp, and to make other chemicals.

Rate of onset: Immediate to hours

Persistence: Minutes to hours

- Under the Chemical Identification Information tab, click the smaller tabs to review five kinds of identification and regulatory information about chlorine:

Chemical Identification: common ID numbers, labels, and codes for chlorine, as well as its formula.

Synonyms: common name synonyms for chlorine, in English and other languages.

NFPA Codes: codes representing the hazards posed by chlorine (0 indicates little or no hazard; 4 indicates highest hazard).

Regulatory Information: designations and thresholds established for chlorine under Federal laws.

Screening and Scenarios: information used for hazards analysis under the *Technical Guidance*.

- Click the Response Information Data Sheets tab, then click the smaller tabs in that section to see information about each of 10 emergency response-related topics.

Each piece of information in RIDS is notated to indicate its source.

These notations appear as abbreviations in parentheses (see “[Sources of CAMEO’s chemical data](#)” on page 103 for more information about each source).

In the example below, the general description of chlorine is from the Association of American Railroads (AAR).

Chemical Identification Information			Response Information Data Sheets	
Firefighting	Fire Hazards	Non-Fire Response	Health Hazards	Protective Clothing
General Description	Properties	Reactivity	Reactive Hazards	First Aid
<p>A greenish yellow gas with a pungent suffocating odor. Toxic by inhalation. Slightly soluble in water. Liquefies at -35°C and room pressure. Readily liquefied by pressure applied at room temperature. Density (as a liquid) 13.0 lb / gal. Contact with unconfined liquid can cause frostbite by evaporative cooling. Does not burn but, like oxygen, supports combustion. Long-term inhalation of low concentrations or short-term inhalation of high concentrations has ill effects. Vapors are much heavier than air and tend to settle in low areas. Contact CHEMTREC to activate chlorine response team 800-424-9300. Used to purify water, bleach wood pulp, and to make other chemicals (© AAR, 1999).</p>				

As you review the RIDS information about chlorine, consider the following questions, and note where you would check to find the kinds of information you would need quickly during a response:


What are the main hazards of chlorine? Could it become an airborne toxic gas hazard? Is it flammable? Is it reactive?

Getting information about a facility

You can keep a record in the Facilities module for each facility that reports Tier II chemical inventory information to the LEPC. You then can cross-reference any facility record to:

- a symbol representing its location on a MARPLOT map.
- records in the Chemicals in Inventory, Contacts, Screening & Scenarios, and Incidents modules, where you can maintain various kinds of information about facilities, including their chemical inventories, emergency contacts, hazards analysis results, and details of past incidents at given facilities.

You can search the Facilities module just as you searched the Chemical Library. The Green Valley Water Facility is in the town of Haymarket. To open the Facilities module and search for all the facilities in the town of Haymarket:

1. From the File menu, select Open, then Facilities. The module will open in List View (in List View, the names of all the records are shown in a list).
2. Click the  button, or select Start Search from the Search menu.

3. Type “haymarket” in the City box, then click Search, as shown below (it doesn’t matter whether you type letters in upper or lower-case).

You’ll see a list of the three facilities located in Haymarket, including the Green Valley Water Facility. You’ll work with the record for this facility.

Found Facilities		
Results of search for: City Contains characters haymarket		
Name	Dept	Street Address
Green Valley Water Facility (DEMO)	Chlorination	U.S. Highway 15
Adams Petroleum Refinery (DEMO)		Merovan Industrial Park A-9
M & S Chemicals (DEMO)	Woods Lake Plant	U.S. Highway 15

- In the list, double-click “Green Valley Water Facility (DEMO)” to open this record in Record View. In Record View, you can view the entire record and all the information in it.

The screenshot shows a web-based interface for viewing a facility record. At the top, there is a navigation bar with buttons for Home, View List, New Search, Edit, and Help, along with navigation arrows. Below this is a header for the 'Facilities' record, indicating it was last modified on 10/14/2008. The main form area is divided into several sections:

- Shipper Section:** Includes a checkbox for 'Shipper', a 'Report Year' dropdown set to '2000', and fields for 'Facility Name' (Green Valley Water Facility (DEMO)), 'Department' (Chlorination Division), and 'Site' (1 of 1).
- Navigation Tabs:** A row of tabs for 'ID Codes', 'State Fields', 'Map Data', 'Site Plan', and 'Notes'. Below these are more specific tabs: 'Address', 'Facility Phones', 'Contacts', 'Chemical Inventory', and 'Checklist'.
- Address Section:** Labeled 'Street Address', it contains a text field with 'U.S. Highway 15'. Below are dropdowns for 'City' (Haymarket), 'State' (VA), and 'Zip' (87530). There are also fields for 'County' (North), 'Fire District', and 'Cross Street'.
- Mailing Address Section:** Labeled 'Mailing Address (if different from street address)', it contains a text field with 'Town Hall'. Below are dropdowns for 'City' (Haymarket), 'State' (VA), and 'Zip' (87530). There is also a 'Country' field.
- Email Section:** A single text field for an email address.

- Click the tabs on the record to review the information about this facility. Check [Table 4-1 on page 115](#) if you'd like to see definitions of all the kinds of information on a Facilities record. Later, you'll navigate to related records in the Chemicals in Inventory and Contact modules to look more closely at the chemical inventory and contacts information for this facility.
- From the File menu, select Print Report. You're about to create a printable report describing this facility.
- Click the Current Record button if it isn't already selected.

8. Either (a) click Select All to include all the information from the record in the print report, or (b) click the box for each of the kinds of information you would like to include in the report.

Facilities Report

Records to include in the report:

Current Record (Green Valley Water Facility (DEMO))

Found Set (3 records)

Select data to include:

Address

Phones

Contacts

Chemical Inventory

Checklist

ID Codes

State Fields

Incidents

Map Data

Screening & Scenarios

Location

Notes

9. Click Print Report. The Print Setup window will be displayed. Adjust any settings you'd like to change.

Note: If you'd like to see what your report will look like before you print it, click on Preview Report instead. You'll see a view of the printable report and can choose to print or cancel the report from there.

10. Click OK to display the Print window. Adjust any settings you'd like to change, then click OK to print the report.
11. Click Cancel to return to the Report setup window, then click Cancel again to return to the Facilities record for the Green Valley Water Facility.

Adding emergency contact information

This facility record does not include any emergency contact information, so you'll add information about Angela Wong, the senior plant manager at the Green Valley Water Facility. To add that information, you'll need to enter **Edit Mode**, in which you can make changes to the record. Right now, you're in **Browse Mode**: you can view records, but cannot change the information that you see.

Follow these steps to add the information about Ms. Wong:

1. Click the Contacts tab.
2. Click the Edit button in the toolbar.
3. Click Add Contact.

Edit Facilities Last Modified 1/21/2007

Shipper Report Year 2000

Facility Name Green Valley Water Facility (DEMO)

Department Chlorination Division Site 1 of 1

ID Codes State Fields Map Data Site Plan Notes

Address Facility Phones **Contacts** Chemical Inventory Checklist

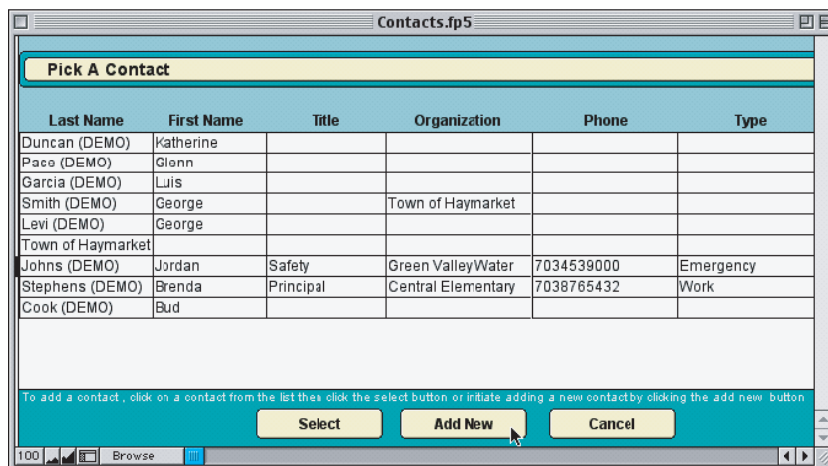
Last Name	First	Title	Organization	Delete
Johns (DEMO)	Jordan	Safety Coordinator	Green Valley Water Facility	<input type="checkbox"/>

Add Contact Edit Contact

Save Changes Cancel

You'll see a list of contacts in CAMEO, but the list doesn't yet include a record for Ms. Wong.

4. Click Add New.



5. On the new, blank record, type the following information about Angela Wong:
 - (1) click in the “First Name” box, then type “Angela.”
 - (2) press TAB to move your cursor to the “Last Name” box, then type “Wong (DEMO).”
 - (3) press TAB again to move to the “Organization” box, then type “Green Valley Water Facility.”
 - (4) TAB to the “Title” box, then type “Senior Plant Manager.” Next, press TAB to drop down the first Contact Type menu, then choose “Corporate Emergency Contact” from the menu.
6. Fill out the items under the Address tab as follows (press TAB to move between boxes):
 - a. Type “U.S. Highway 15” in the top street address box.
 - b. Type “Haymarket” in the City box.
 - c. Select “VA” from the State menu.
 - d. Type “87530” in the Zip box.
 - e. Type “Prince William” in the County box.
 - f. Type “Haymarket” in the Fire District box.
 - g. Type “awong@greenvalley.com” in the Email box.

7. Click the Phones tab, then click in the top Type box to drop down a menu of phone types. From the menu, select Work.
 - a. In the next box, type “703-232-5667.”
 - b. Press TAB twice to drop down the next Type menu, select Emergency from the menu, then type “703-232-7556” in the next box.

The new Contacts record displayed on your screen should look like the one below.

The screenshot shows a web application window titled "Edit Contacts" with a "Last Modified" timestamp of "4/25/2002". The form contains the following fields:

- First Name: Angela
- Last Name: Wong (DEMO)
- Organization: Green Valley Water Facility
- Title: Senior Plant Manager
- Contact Type: Corporate Emergency Contact

Below these fields are three tabs: "Address", "Phones", and "Notes". The "Phones" tab is active, displaying a table with the following data:

Type	Phone	Delete
Work	703-232-5667	<input type="checkbox"/>
Emergency	703-232-7558	<input type="checkbox"/>
		<input type="checkbox"/>

At the bottom of the form are two buttons: "Save Changes" and "Cancel".

- Press Save Changes. You'll see the list of contacts for the water facility, which now includes Ms. Wong.

The screenshot shows the 'Edit Facilities' window with the following details:

- Shipper:** (unchecked)
- Report Year:** 2000
- Facility Name:** Green Valley Water Facility (DEMO)
- Department:** Chlorination Division
- Site:** 1 of 1

Address		Facility Phones		Contacts	Chemical Inventory	Checklist
Last Name	First	Title	Organization	Delete		
Johns (DEMO)	Jordan	Safety Coordinator	Green Valley Water Facility	<input type="checkbox"/>		
Wong (DEMO)	Angela	Senior Plant Manager	Green Valley Water Facility	<input type="checkbox"/>		

Buttons at the bottom: Add Contact, Edit Contact, **Save Changes** (highlighted), Cancel.

- Press Save Changes again to save the new record.

Reviewing the facility's chemical inventory

CAMEO's Chemicals in Inventory module is one of several modules in which records related to facilities can be kept. In the Chemicals in Inventory module, you can keep records on the hazardous substances stored or used at facilities, or transported along routes in your community. From a Facilities record for a facility, you can quickly jump to the Chemicals in Inventory records for chemicals in that facility.

Under the Emergency Planning and Community Right-to-Know Act (EPCRA) of 1986, **Tier II forms** must be submitted each year by operators of facilities that maintain certain hazardous materials in more than minimum threshold amounts. On a Tier II form, the facility provides information about its inventory of hazardous materials (e.g., storage locations and amounts). Tier II forms are submitted to each state's State

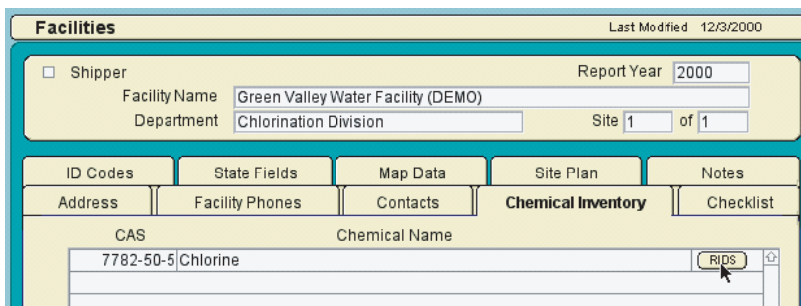
Emergency Response Commission (SERC). Many SERCs then share the Tier II data with county- and local-level response and planning organizations, such as fire departments and LEPCs.

You can keep Tier II information in CAMEO's Chemicals in Inventory module. A Chemicals in Inventory record is similar to the actual Tier II form, with some exceptions. In particular, you can describe only one chemical or mixture on each Chemicals in Inventory record.

Now, you'll take a closer look at the chemical inventory information for Green Valley Water Facility:

1. Click the Chemical Inventory tab on the Facilities record for Green Valley Water Facility. You'll see a space where stored chemicals are listed. Just chlorine is listed on the Water Facility's record, because it's the only hazardous substance stored at the facility.

When you see the name of a substance in the Chemical Inventory list on a facility's record, that means there's a corresponding record in the Chemicals in Inventory module.



The screenshot shows a web-based interface for a facility record. At the top, it says "Facilities" and "Last Modified 12/3/2000". Below that, there are fields for "Shipper" (unchecked), "Report Year" (2000), "Facility Name" (Green Valley Water Facility (DEMO)), and "Department" (Chlorination Division). There are also "Site 1 of 1" indicators. A navigation bar contains tabs for "ID Codes", "State Fields", "Map Data", "Site Plan", "Notes", "Address", "Facility Phones", "Contacts", "Chemical Inventory" (which is highlighted), and "Checklist". Below the tabs, there is a table with columns for "CAS" and "Chemical Name". The first row contains "7782-50-5" and "Chlorine". To the right of the "Chlorine" entry, there is a button labeled "RIDS" and a small icon.

2. Click the RIDS button to the right of "Chlorine" in the list. The Response Information Data Sheet section of chlorine's Chemical Library record is displayed. This is the same set of response recommendations you reviewed earlier. As you can see, clicking the RIDS button for a substance in a facility's inventory is a quick way to access the response recommendations for that substance.
3. From the File menu, select Close. The Green Valley Water Facility record is visible again.

- Double-click on the name “chlorine” (or on its CAS number). The Chemicals in Inventory record for chlorine at Green Valley Water Facility should open (as shown below).

The screenshot shows the 'Chemicals in Inventory' record for Chlorine. The record is titled 'Chemicals in Inventory' and was last modified on 9/2/2003. The facility is 'Green Valley Water Facility (DEMO)' and the report year is '2000'. The department is 'Chlorination Division', the city is 'Haymarket', and the state is 'VA'. The CAS number is '7782-50-5'. The chemical name is 'Chlorine'. There are buttons for 'View RIDS' and 'Adjust Link'. The record is checked as 'In Inventory', 'EHS Substance', and 'MSDS'. The record is divided into several tabs: 'Location', 'Physical State & Quantity', 'Components', 'Dates', 'State Fields', and 'Notes'. The 'Location' tab is selected, showing a table with columns for Amount, Unit, Type, Press, Temp, and Location. The table contains four rows of data:

Amount	Unit	Type	Press	Temp	Location
	Pounds	A	2	4	Chlorination Building #1
	Pounds	A	2	4	Chlorination Building #2
	Pounds	A	2	4	Chlorination Building #3
	Pounds	A	2	4	Chlorination Building #4

- Click the tabs on this record to review the information it contains about the storage amounts, storage locations, and hazards of chlorine at the water facility. Refer to [Table 4-3 on page 134](#) to see explanations of all the information items on the record.

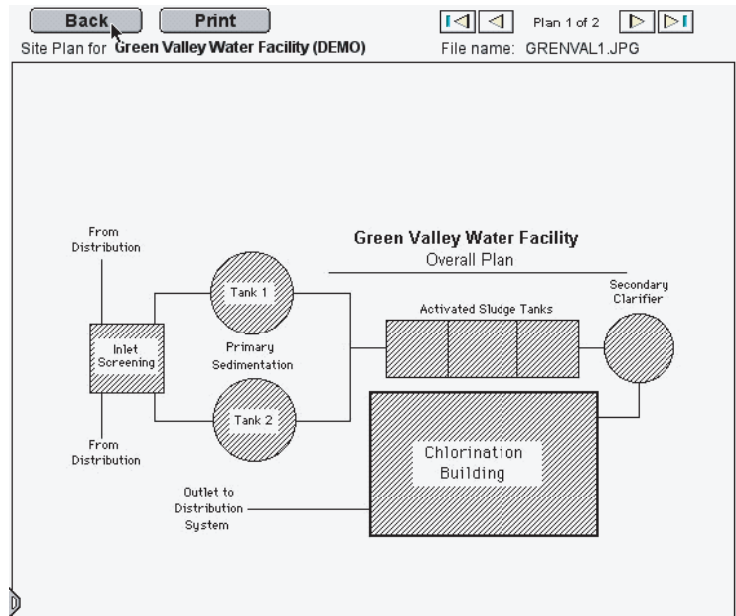
Viewing the site plan

You can store facility site plans in CAMEO along with other information about facilities. To see the site plan for the Green Valley Water Facility,

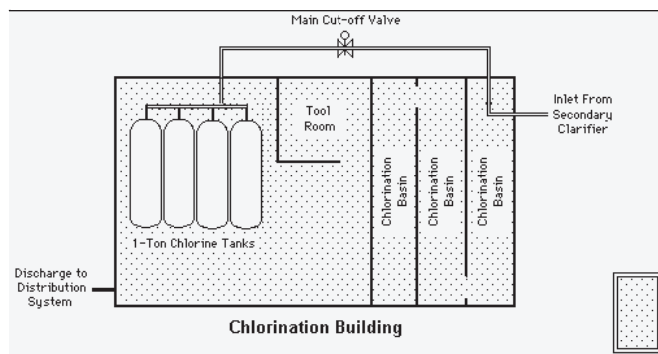
- From the Record menu in the Chemicals in Inventory record, select Show Related, then Facility/Route. The Facility record for Green Valley Water Facility will be displayed.

Note: Use Show Related when you need to quickly navigate among related records in different modules.

- Click the Site Plan tab. You'll see a list of the two site plans for this facility.
- Double-click `grenval1.jpg`. The site plan will be displayed (as below).



- Click Back to return to the Facilities record.
- Double-click `grenval2.jpg`, to see a plan of the chlorination building, then Back to return to the record.



You can create site plans like these in any graphics program, save them in common graphic file formats, and then include them with your Facilities records (see [“Adding and editing site plans” on page 124](#)).

Working with a map

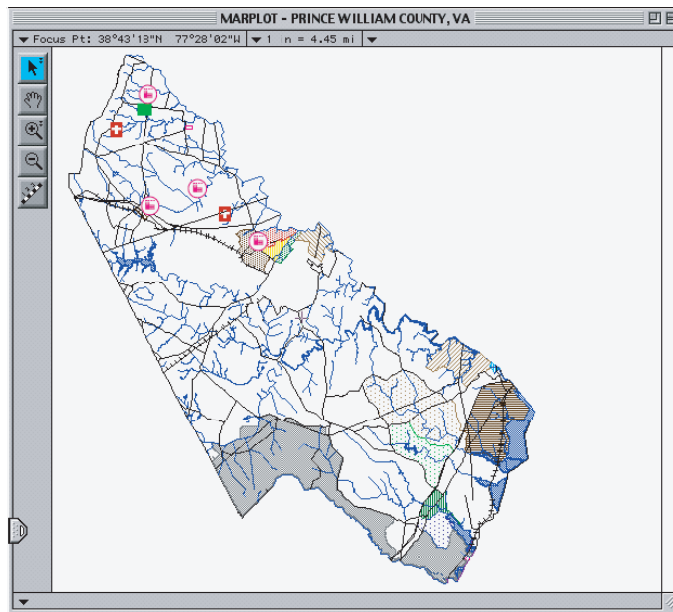
MARPLOT is the mapping program included in CAMEO. In this section of the tour, you'll learn how to display a map in MARPLOT, search for a particular location on a map, and link symbols on a map to records in CAMEO modules.

First, you'll open the map of Prince William County in MARPLOT, and then find and view the location of the Green Valley Water Facility:

1. To start MARPLOT, from CAMEO's Sharing menu, select MARPLOT, then select Go to MARPLOT.

MARPLOT will start up and be displayed. You might first be asked to locate MARPLOT:

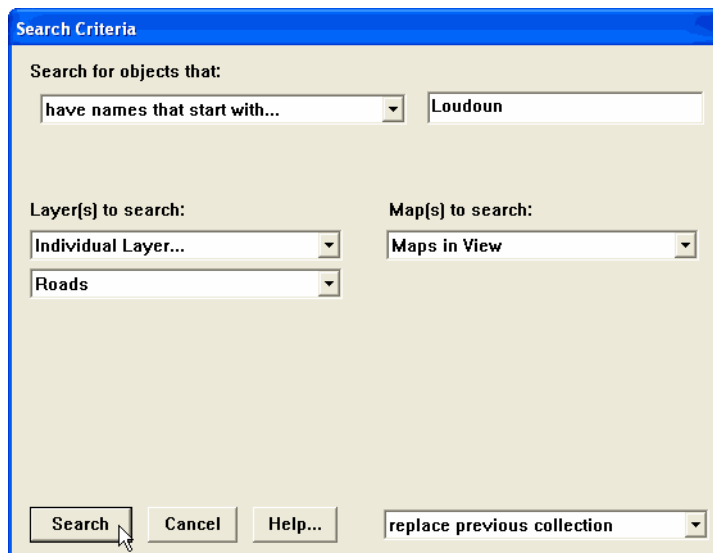
- Unless you have LandView installed, navigate through the files on your hard drive to locate the MARPLOT folder. Open the folder and select the “MARPLOT.EXE” file (in Windows) or the “MARPLOT” file (on a Macintosh) inside the MARPLOT folder.
 - If you have LandView installed, locate and open the LandView folder, then select the “MARPLOT.EXE” file.
2. MARPLOT displays a greeting window. Click OK (in Windows), or just click within the window (on a Macintosh). A map showing Prince William County is then displayed, as shown on the next page. (If you've used this map before, the view you see may differ from the shown view; however, the difference will not cause problems.)



Next, you need to set up and run a search for the water facility, which is represented on the map by a symbol. In MARPLOT, you can search for roads, intersections, street address ranges, symbols that represent facilities or other locations, geographical features, and other kinds of map objects. The Green Valley Water Facility is located near the intersection of Loudoun Drive and James Madison Highway (also called U.S. Highway 15). Here's one way to search for this location:

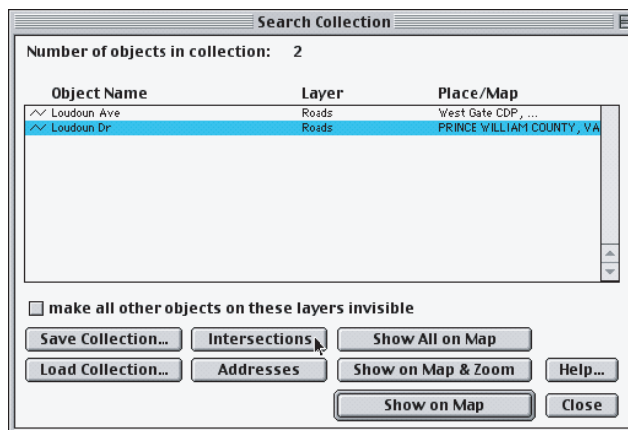
3. To begin your search, choose Search from the List menu.
4. To set up the search,
 - a. Under the "Search for objects that:" heading, be sure that "have names that start with..." is selected in the search types menu.
 - b. In the box to the right of this menu, type in "Loudoun."
 - c. Be sure that Individual Layer is selected in the "Layer(s) to search" menu.
 - d. Immediately below, select Roads from the menu of layers.

- e. Be sure that Maps in View is selected in the “Map(s) to search:” menu.
- f. When your search criteria look like those shown below, click Search.



Both “Loudoun Ave” and “Loudoun Dr” appear in the list of found items.

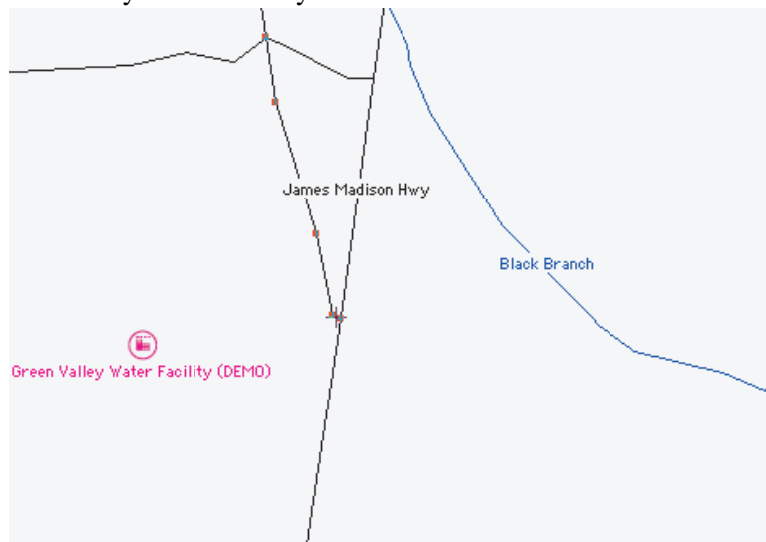
- 5. Click just once on “Loudoun Dr” to select it, then click Intersections (as shown below). MARPLOT will search for all intersections along Loudoun Drive.



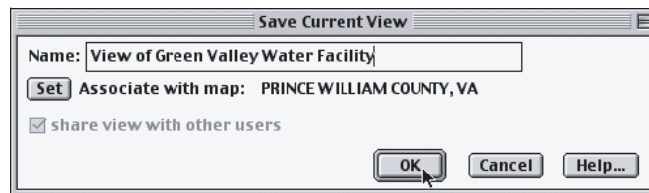
There are seven intersections along Loudoun Drive; the Green Valley Water Facility is located near the intersection with James Madison Highway (U.S. Highway 15).

6. Click just once on “James Madison Highway” to select it, then click Show on Map & Zoom.

You’ll see a view of the area around the intersection on your map (shown below; names may not appear in your view). The intersection is in the center of the view. To its left, you’ll see a symbol object, representing the Green Valley Water Facility.



7. To give this view a name and save it, so that you can easily return to it later, choose Save Current View... from the View menu.
8. In the “Save Current View” dialog box, type “View of Green Valley Water Facility” and then click OK.



Later, whenever you want to return to this or any other saved view, you would select Go to View from the View menu, select the name of the

view (e.g., View of Green Valley Water Facility) from the list of saved views, then click Go to View.

Next, you'll link the Green Valley Water Facility symbol to the Green Valley Water Facility's record in the Facilities module:



9. To create a link, first check to be sure that the arrow tool is selected in MARPLOT's tool palette, as it is in the example at left. If this tool is not selected, click on it once.

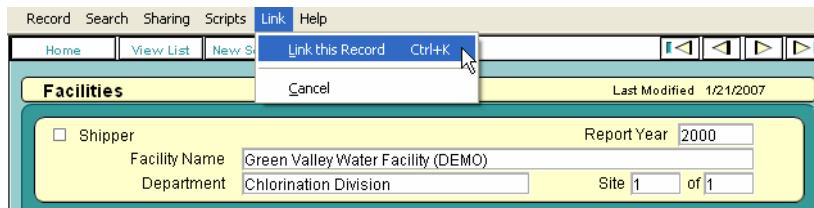
10. Click on the symbol for the Green Valley Water Facility, to select it.



When you select an object, you can see four small boxes around the object, indicating that it is selected (as at left). MARPLOT displays some basic information about the object along the lower map margin. Also, the Focus Point, a small, flashing, target-shaped icon that marks the most recent point of interest on the map, moves to the point of your click.

11. While the symbol is still selected (is still surrounded by the four boxes), from MARPLOT's Sharing menu, select CAMEOfm, then Link Object. You'll be taken back to the Green Valley Water Facility's record in the Facilities module (if not, find and open that record).

12. While that record is visible, from the Link menu, select Link this Record (as shown below).



Once you have created a link between a symbol and record, you can easily move back and forth between them:

13. To go from the Green Valley Water Facility record to the map symbol, from the Sharing menu, select MARPLOT, then Show on Map.

You'll be taken to MARPLOT. The Green Valley Water Facility symbol will be selected, and the Focus Point will appear centered on the symbol.



Tip: If the map is more zoomed-in than you'd like, click the zoom-out button in MARPLOT's toolbar (shown at left), then click several times on the water facility's symbol.

When you select a linked symbol on a map, you can also quickly go to the CAMEO record to which it is linked:

14. While the symbol remains selected, from MARPLOT's Sharing menu, select CAMEOfm, then Get Info. (If the symbol has become unselected, click on the arrow tool button, then click in the symbol to reselect it.)

You're taken back to the Green Valley Water Facility record in CAMEO's Facilities module.

Working with Screening and Scenarios

Next, you'll try out using the Screening & Scenarios module to perform the **hazards analyses** explained in the *Technical Guidance for Hazards Analysis* (see "[Bibliography](#)" on page 272).

In a hazards analysis, you estimate the size of the area around a facility or other potential spill site that could be affected by the accidental release of a chemical of concern. That area is called the **threat zone**. You can use MARPLOT to display threat zones on an electronic map of your community. Once you have an estimate of the radius of the threat zone, you can plot the zone on a map in MARPLOT. You can choose to estimate the size of the threat zone around a facility or along a transportation route by:

- using EPA's "credible worst case" assumptions to obtain a worst-case threat zone estimate (a **screening** estimate); or
- entering weather conditions and other information that you believe to be more typical of the region and facility (a **scenario** estimate).

Now, you'll review the results of a screening for the Green Valley Water Facility:

- 1.** Return to CAMEO if you're still working in MARPLOT. The Facilities record for the Green Valley Water Facility should be visible on your screen.

2. Click the Chemical Inventory tab. You'll see that chlorine is the only chemical included in this facility's inventory.
3. From the Record menu, select Show Related, then Screening and Scenario Records. Use Show Related whenever you want to see records in other CAMEO modules that are related to the record you're working with. The Screening & Scenarios module will open in List View. You'll see the names of two records: "Screening" and "Scenario #1."
4. Double-click on the "Screening" record in the list to open it in Record View.

The screenshot shows the 'Screening & Scenarios' record view. At the top, it displays 'Last Modified 11/19/2003'. The main form contains the following fields and sections:

- Facility / Route Name:** Green Valley Water Facility (DEMO)
- Chemical:** Chlorine (CAS: 7782-50-5)
- Screening Name:** Screening (with a 'View RDS' button)
- Location:** In Inventory, In Transit, Shipper
- Screening Description:**
 - Amount Released: 2000 pounds
 - Concentration: 100 weight %
 - Release Duration: 10 minutes
 - If stored in container with a dike, enter surface area within dike: [] sq ft
 - Atmospheric Concentration Level of Concern: 0.073 gm/m³
 - LOC Description: Greenbook LOC
- Physical State:** Gas, Liquid, Solid
- Weather Information:**
 - Wind Speed: 3.35 mph
 - Ground Roughness: Open Country
 - Wind From: [] in degrees measured clockwise from 0 N. (for example: 015, 315, 270)
 - Stability Class: F
- Risk Assessment:**
 - Risk: Low (Probability of described accident occurring)
 - Consequences: Medium (Severity of consequence to people)
 - Overall Risk: High (Combination of probability and severity of consequence)
- Threat Zone Radius:** > 10 miles (with a 'Show on Map' button)

This record shows the results of a "credible worst-case" screening calculation for chlorine at the Green Valley Water Facility, made according to the procedures described in the *Technical Guidance*. In a worst-case screening, the entire contents of one of the facility's chlorine tanks are assumed to escape into the atmosphere over 10 minutes, forming a gas cloud that would drift away from the plant in whatever direction the wind is blowing. A worst-case screening assumes a low wind speed, very little atmospheric turbulence to dilute the chlorine cloud, and a conservative estimate of the concentration of chlorine that might cause adverse health effects in people.

In the Threat Zone Radius box, you'll see the estimate of the threat zone radius. You'll see that under these worst-case conditions, the threat zone—the area potentially at risk—is predicted to extend more than 10 miles downwind. Depending on the wind direction, chlorine concentrations within the zone could reach or exceed the level of concern shown on the Screening & Scenarios record. Check the *Technical Guidance* or see [“Using Screening & Scenarios to assess hazards” on page 139](#) to learn more about making screening calculations.



5. Click the right arrow button in the upper right corner of the “Screening” record (shown at left), to view the next record in the module. This record shows the threat zone estimate for a more likely scenario.

Screening & Scenarios Last Modified 11/19/2003

Facility / Route Name

Chemical CAS

Scenario Name

In Inventory In Transit Shipper

Scenario Description	Notes
Amount Released <input type="text" value="2000"/> pounds	Physical State <input checked="" type="radio"/> Gas
Concentration <input type="text" value="100"/> weight %	<input type="radio"/> Liquid
Release Duration <input type="text" value="30"/> minutes	<input type="radio"/> Solid
If stored in container with a dike, enter surface area within dike: <input type="text" value="0"/> sq ft	
Atmospheric Concentration Level of Concern <input type="text" value="0.073"/> gm/m ³	
LOC Description <input type="text" value="Greenbook LOC"/>	

Weather Information

Wind Speed mph Ground Roughness

Wind From in degrees measured clockwise from 0 N. Stability Class
(for example: 015, 315, 270)

Risk Assessment

Risk Probability of described accident occurring

Consequences Severity of consequence to people

Overall Risk Combination of probability and severity of consequence

Threat Zone Radius miles

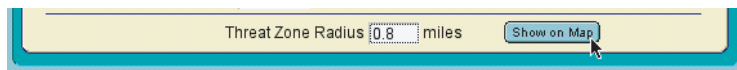
The highest-risk operation at the plant is the changing of chlorine tanks, done only during daytime hours. This record shows the results of a hazards analysis calculation for a release of chlorine through the valve of a single 1-ton chlorine cylinder, which is sheared off while the cylinder is being transferred by forklift from a transport vehicle to the facility.

In this scenario, the amount released is the same as before—the contents of one tank—but the chlorine is expected to escape over 30 minutes rather than 10 minutes. Local meteorologists were able to provide a description of the most common weather conditions observed at Haymarket: instead of “worst case” weather conditions, wind speed is higher and the atmosphere is more turbulent (so that air is more quickly mixed into the chlorine cloud, diluting it to below hazardous concentrations). This time, the threat zone is predicted to extend only 0.8 miles from the accident site.

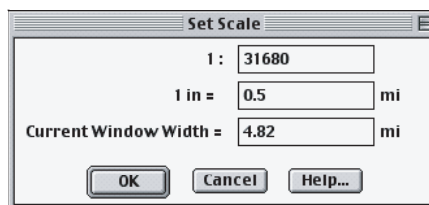
Working with a threat zone on a map

To plot this threat zone on the map,

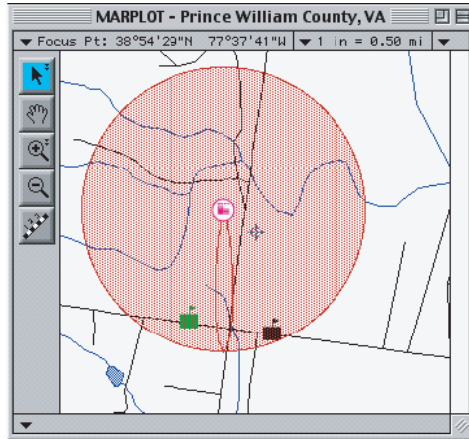
1. Click Show on Map. MARPLOT will come forward, and the zone will automatically be plotted on the map, centered on the Green Valley Water Facility symbol.



2. However, at the current map scale, it's not possible to see the entire threat zone at once. To adjust the map scale, choose Set Scale from MARPLOT's View menu.
3. In the “Set Scale” dialog box, next to the “1 in =” heading, type “0.50” to change the map scale to 1 inch = 0.5 mile (as below). (Your “Current Window Width” might not match the value shown in the diagram.)



4. Click OK. You'll see the entire threat zone on your screen (as shown below).



Once you've plotted a threat zone on a map, you can check to see whether any symbol objects representing especially vulnerable populations, such as schools or hospitals, are within the threat zone.

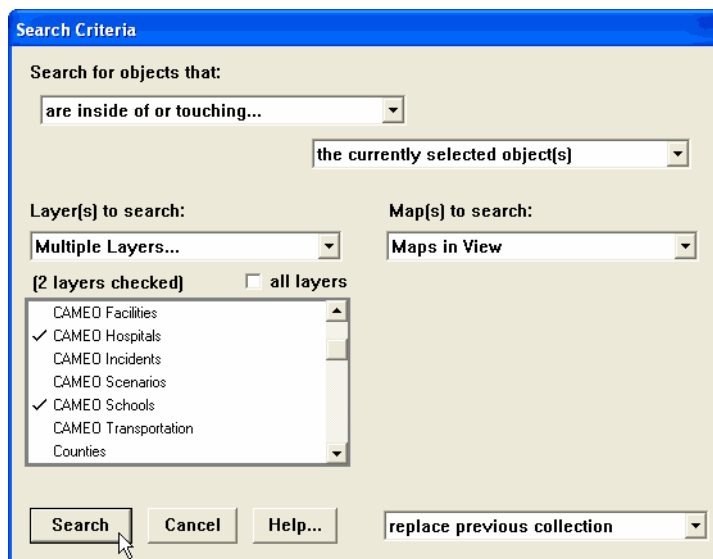
Tip: You can use the same procedure to check for special locations within an ALOHA threat zone that you have plotted on a map. See [“Checking for special locations within a threat zone” on page 163](#).

To check for special locations in the threat zone,

1. While the threat zone object is still selected, select Search from the List menu. (If it has become unselected, click on the zone's outer boundary to reselect it, then select Search.)
2. To set up the search,
 - a. Under the “Search for objects that:” heading, select **are inside of or touching...** from the menu of object types.
 - b. Check to be sure that **the currently selected object(s)** is selected in the next menu.
 - c. Under the “Layer(s) to search:” heading, choose **Multiple layers...** from the popup menu.

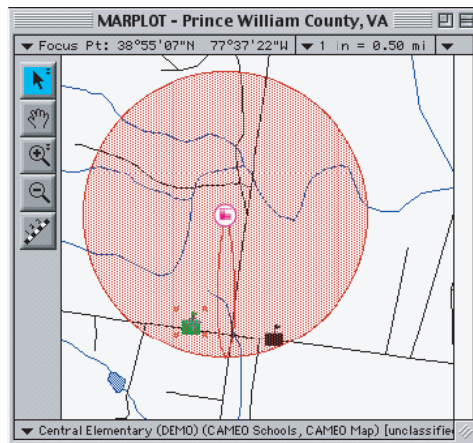
- d. In the list of layers, click on the “CAMEO Hospitals” and “CAMEO Schools” layers to place checkmarks next to the names of those layers (the checkmarks indicate that MARPLOT will search these layers).
- e. Click the names of any other layers that have checkmarks next to them, to remove the checkmarks (MARPLOT will then not search these layers).

When your search criteria look like the ones below, click Search.



- 3. The symbol object representing Central Elementary School is inside the threat zone, so it appears in the search collection. While its name is highlighted, click Show on Map.

Once the map is redrawn, the symbol for Central Elementary will be selected, and the Focus Point will be centered on it (as below).



Note: If more than one school and or hospital symbol was within the threat zone, you could click Show All on Map to see all the symbols selected on the map.

4. While the symbol is still selected, from MARPLOT's Sharing menu, select CAMEOfm, then Get Info. (If the symbol becomes unselected, click on it to reselect it.)

This symbol was previously linked to a record describing the school in CAMEO's Special Locations module. This record will be displayed on your screen. (If more than one linked symbol had been selected, the records linked to all those symbols would be shown to you in a list.)

You can use the Special Locations module to keep information describing vulnerable locations in your community, including emergency phone numbers you could use to quickly contact the location during the response to an incident.

In a real emergency, you might need to call a contact person at the Central Elementary School. In CAMEO, there are two ways to quickly find an emergency phone number for the school, as follows:

1. First, click the Phones tab to see the emergency phone number for Central Elementary School (shown below).

The screenshot shows the 'Special Locations' form. At the top, it says 'Special Locations' and 'Last Modified 3/24/2004'. Below that, there are input fields for 'Location Name' (Central Elementary (DEMO)), 'Location Type' (Elementary School), and 'Building Type' (Low Rise (fewer)). A tabbed interface below has tabs for 'Address', 'Population', 'Phones', 'Contacts', 'Map Data', 'Site Plan', and 'Notes'. The 'Phones' tab is active, displaying a table with the following data:

Type	Phone
Emergency	(703) 342-0049

Tip: In an emergency, you can quickly print out a list of phone numbers for one or more special locations to hand to a dispatcher. To do this, from the File menu, you would select Print Report. You then would click the Phones checkbox and the checkboxes for any other items you'd like included, then Print Report.

2. Second, from the Record menu, select Show Related, then Contact Records. A list of the Contacts records for the school, containing just one name, will be displayed. Double-click that name (or click the View Record button in the toolbar) to open the Contacts record for Brenda Stephens, the Central Elementary School Principal. Click the Phones tab to see her telephone number (as below).

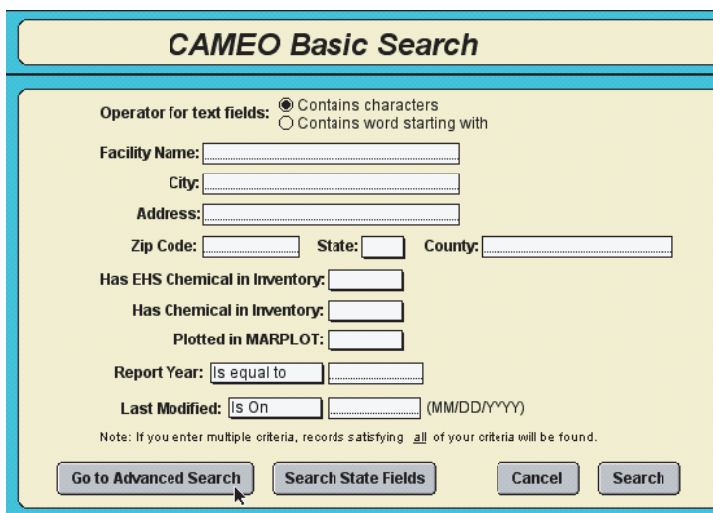
The screenshot shows the 'Contacts' form. At the top, it says 'Contacts' and 'Last Modified 11/30/2000'. Below that, there are input fields for 'First Name' (Brenda), 'Last Name' (Stephens (DEMO)), 'Organization' (Central Elementary (DEMO)), 'Title' (Principal), and 'Contact Type' (Other). A tabbed interface below has tabs for 'Address', 'Phones', and 'Notes'. The 'Phones' tab is active, displaying a table with the following data:

Type	Phone
Work	(703) 876-5432

Making an advanced search

Having evaluated the hazard posed by the Green Valley Water Facility, you'll next check to see which other facilities in Haymarket store or use chlorine. To find that out, you'll run an advanced search of the Facilities module.

1. From the File menu, select Open, then Facilities. Note that you can open any CAMEO module via the Open command.
2. From the Search menu, select Start Search, then click Go to Advanced Search (as shown below).



CAMEO Basic Search

Operator for text fields: Contains characters
 Contains word starting with

Facility Name:

City:

Address:

Zip Code: State: County:

Has EHS Chemical in Inventory:

Has Chemical in Inventory:

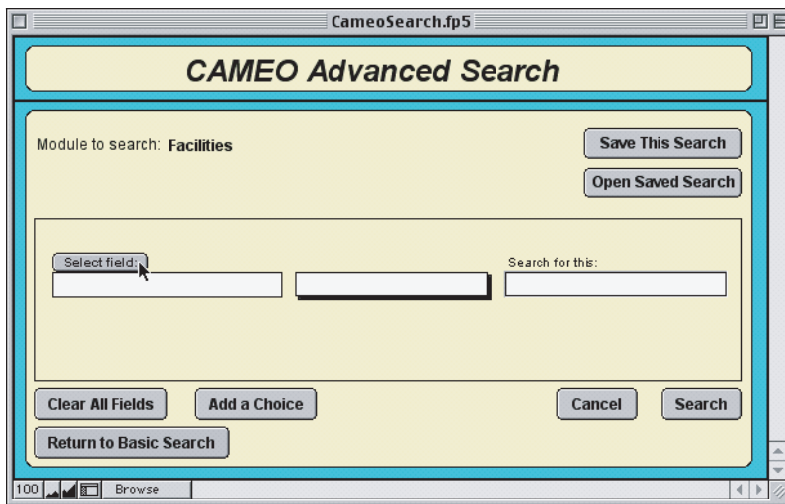
Plotted in MARPLOT:

Report Year: is equal to

Last Modified: is On (MM/DD/YYYY)

Note: If you enter multiple criteria, records satisfying all of your criteria will be found.

3. Click Select Field (as shown below).



In the next steps, you'll indicate

- a. which module you want to search—you're offered a choice whenever you're searching for facility records because information about facilities is stored in more than one module in CAMEO.
- b. which data field you want to search, within that module.

4. Make sure the Facilities button is highlighted, then, in the list of data fields, click City to select that data field. Click Select (as shown below).

Select a Field

Module to search: **Facilities**
Note: you can choose to search either this module, or any of the other sources listed below, which contain information related to this module.

Click on a field name:

- Facilities
- Chemicals in Inventory
- Contacts
- Facility IDs
- Incidents
- Mixture Components
- Screening and Scenarios
- Storage Locations

- City
- Clean Air Act
- Country
- County
- Cross Street
- Date Modified
- Date Received
- Date Signed
- EHS
- Email Address

Select

Cancel

5. The Advanced Search window is displayed again; leave **contains characters** selected in the popup menu of search types, then type “Haymarket” in the text box to the right of the menu.

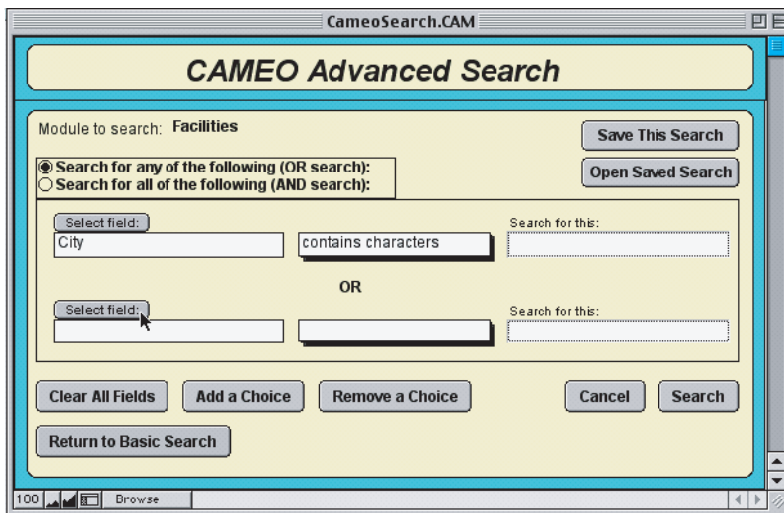
Select field: City

Search type: contains characters

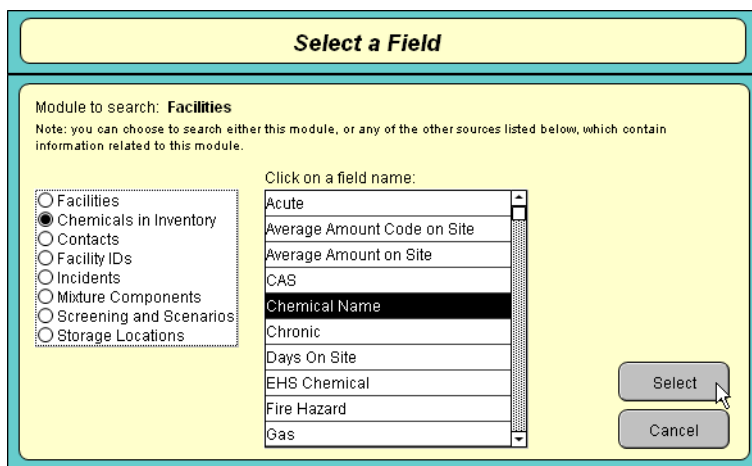
Search for this: Haymarket

6. Click Add a Choice. You need to add a second choice because you’re searching for two *criteria*: facilities that (1) *are located in Haymarket* and (2) *have chlorine in their inventories*.

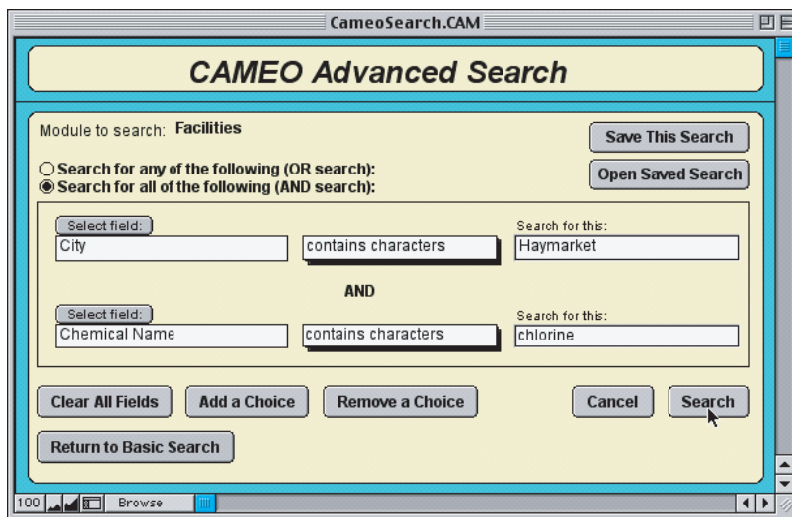
- Click the Select Field button just above the lower, blank box (as shown below).



- In the left-hand list, click the Chemicals in Inventory button.
- In the list of data fields, click Chemical Name, then click Select (as shown below) to return to the Advanced Search window.



10. Leave the lower menu set to **contains characters**, and to its right, type “Chlorine” in the empty text box.
11. Click the **Search for all of the following** button (it’s in the upper left part of the window).
12. When your window looks like the one below, click Search.



You’ll see that two facilities located in Haymarket store or process chlorine: M & S Chemicals and the Green Valley Water Facility.

Note: Whether you click **Search for any of the following (OR search)** or **Search for all of the following (AND search)** when you set up a search for two or more criteria makes a big difference. When you click “Search for all of the following (AND search),” you’re setting up a search for all facilities that are located in the town of Haymarket *and* that also maintain chlorine on site. When you click “Search for any of the following (OR search),” you’re setting up a search for all facilities that are located in the town of Haymarket *or* that maintain chlorine on site. If you select Start Search from the Search menu again, type the same search criteria again, then click “Search for any of the following (OR search),” then click Search, you’ll discover that the search collection contains three records instead of two: Adams Petroleum Refinery now is included in the list because it meets one of your criteria—it’s located in Haymarket. But it doesn’t meet the other criterion: it doesn’t include chlorine in its inventory.

Predicting chemical reactivity

Not only are you a member of Prince William County's LEPC; you're also a member of Haymarket's volunteer fire department. In that role, you receive a call to head out along James Madison Highway to the scene of a new incident:

A train has collided with a truck at an intersection in the industrial section of Haymarket. A tankcar and a flatbed car filled with containers have derailed. Now the tankcar is lying in a ditch alongside the tracks, surrounded by some containers that have broken loose from the flatbed car (Figure 2-6).

- The tankcar is labeled "hydrogen fluoride."
- The UN/NA number shown on all the containers is 1806.

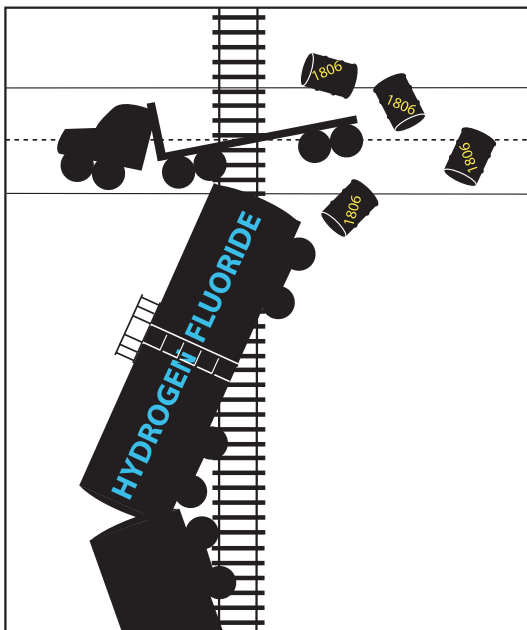


FIGURE 2-6. Schematic overhead view of the accident scene.

You want to know: What problems could occur if these chemicals were to accidentally mix together when crews right the tankcar and collect the containers?

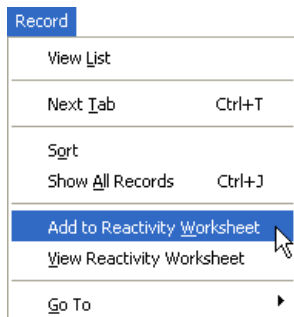
You can use CAMEO to answer this question. You'll "mix" the two chemicals by finding their Chemical Library records and then adding both chemicals to the Reactivity Worksheet. CAMEO will then predict the mixture's potential reactivity.

Working the problem. First, you'll need to search the Chemical Library for the two chemicals you want to add:

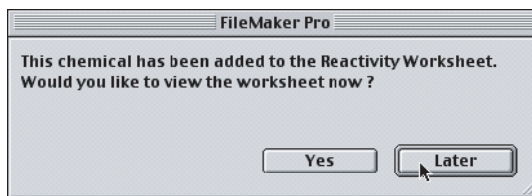
1. Click the Home button in the toolbar, then click Search for a Chemical.
2. Type "hydrogen fluoride" in the Chemical Name box, then press Search (searching for either a name that "contains characters" or a "contains word starting with" will work for this chemical).

In the list of found records, you should find "HYDROGEN FLUORIDE, ANHYDROUS." If it's not there, repeat your search, this time checking your spelling carefully: in "fluoride," "u" comes before "o."

3. In the list, double-click on HYDROGEN FLUORIDE, ANHYDROUS to open that record, then click the Response Information Data Sheets tab.
4. Check the reactivity characteristics of hydrogen fluoride as follows:
 - a. Click the Reactive Hazards tab to see the especially acute hazards of hydrogen fluoride. You'll see that hydrogen fluoride is both water- and air-reactive.
 - b. Click the Reactivity tab to learn more about its reactions with air and/or water and to view its Chemical Profile to review more details about the reactivity of hydrogen fluoride.



- From the Record menu, select Add to Reactivity Worksheet (as at left).
- When asked whether you want to view the Reactivity Worksheet, click Later (as shown below).



- Perform a search for chemicals with the UN/NA number of 1806: click New Search in the toolbar; type “1806” in the UN/NA Number box, then press Search (searching for either a word that “contains characters” or “contains word starting with” 1806 will work).³
- You'll find just one chemical, PHOSPHORUS PENTACHLORIDE. Double-click on PHOSPHORUS PENTACHLORIDE to open the record for this chemical. Again, click the Response Information Data Sheets tab, then click the Reactive Hazards and Reactivity tabs to check its characteristics, and then select Add to Reactivity Worksheet from the Record menu. This time, when asked whether you want to view the Reactivity Worksheet, click Yes.

Note: In many cases, more than one chemical have the same UN/NA number. In such cases, you'll need more information before you can be sure of the identity of a given chemical. Even when you find just one chemical, as in this case, carefully review the information in the Response Information Data Sheets to verify that you've correctly identified the chemical at hand. In particular, check the General Description. (It says that phosphorus pentachloride is a “greenish-yellow crystalline solid with an irritating odor.”) If that description doesn't fit the chemical you find on scene, check for more information to help you be sure of its identity.

3. To CAMEO, a “word” includes a series of letters or numbers, so an identification number is a “word.”

9. Check the Reactivity Worksheet to see statements about the mixture's potential reactivity (shown below):

The screenshot shows a software window titled "Reactivity Worksheet" with a yellow background. At the top, there are tabs for "Glossary" and "Help". Below the title bar, the text "Chemicals in the Mixture:" is followed by a note: "(Double click chemical to see chemical specific reactivity information)". A list box contains two entries: "HYDROGEN FLUORIDE, ANHYDROUS" and "PHOSPHORUS PENTACHLORIDE". Below the list box are three buttons: "Clear All", "Clear Selected Chemical", and "Add Reactive Groups". A small instruction reads: "To add a new chemical, click 'Done', then click 'New Search'". The next section is titled "Hazards from mixing the reactive groups for the chemicals listed above" and contains a text area with the following text: "- Heat generated from chemical reaction may initiate an explosion." and "- Combination liberates gaseous products, including both flammable and toxic gases. May cause pressurization." Below this is the heading "Special Hazards of Each Chemical" followed by a numbered list: "1) HYDROGEN FLUORIDE, ANHYDROUS" with sub-entry "- Water-Reactive, Air-Reactive" and "2) PHOSPHORUS PENTACHLORIDE". At the bottom of the window are three buttons: "Preview Report", "Print Report", and "Done".

These statements tell you that the response team would need to be prepared for a possible explosion and release of a toxic gas if these two chemicals were to accidentally mix. The toxic gas that would be liberated in this reaction would be hydrogen chloride, but CAMEO doesn't report this.

10. When you've finished reviewing the worksheet, click Done.

Finishing up and moving on

You can choose either to keep the sample records, map symbols, and map used in this Guided Tour, or to delete them. Keeping them will not interfere with your other uses of CAMEO.

To delete the sample records, symbols, and map, follow the steps below.

To delete the sample records:

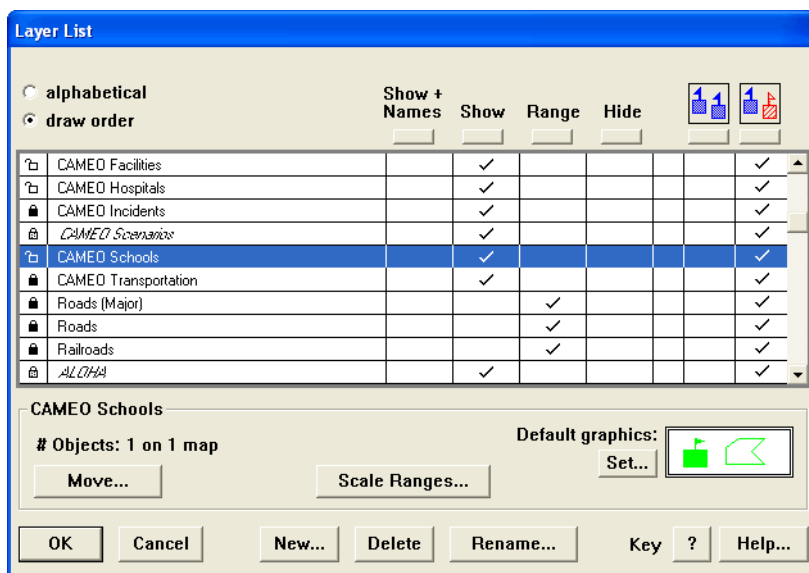
1. Click the Home button in the toolbar, then click the Facilities button. Click the View List button in the toolbar if you're not in List view.
2. From the Search menu, select Clear Search. This module still displays just the found set of records that match your most recent search; clearing the search allows you to access and work with all the records.
3. From the Record menu, select Delete All Records. The five sample records in this module are deleted, along with all records related to the deleted facilities in the Chemicals in Inventory, Screening & Scenarios, and Contacts modules.

Important: Select Delete All Records in this and other modules only if you have not yet entered any other data into the module that you'd like to keep. If you have entered data, select each sample record (all have "DEMO" in their names), then select Delete Facility from the Record menu.

4. Click the Home button in the toolbar, then click the Contacts button.
5. From the Record menu, select Delete All Records to delete the sample records in this module (or, if you have entered data you want to keep, select each sample record, then select Delete Contact from the Record menu).
6. Click the Home button in the toolbar, then click the Special Locations button.
7. From the Record menu, select Delete Special Location to delete the single sample record in this module.
8. Click the Home button in the toolbar, then click the Routes button.
9. From the Record menu, select Delete Route to delete the single sample record in this module.

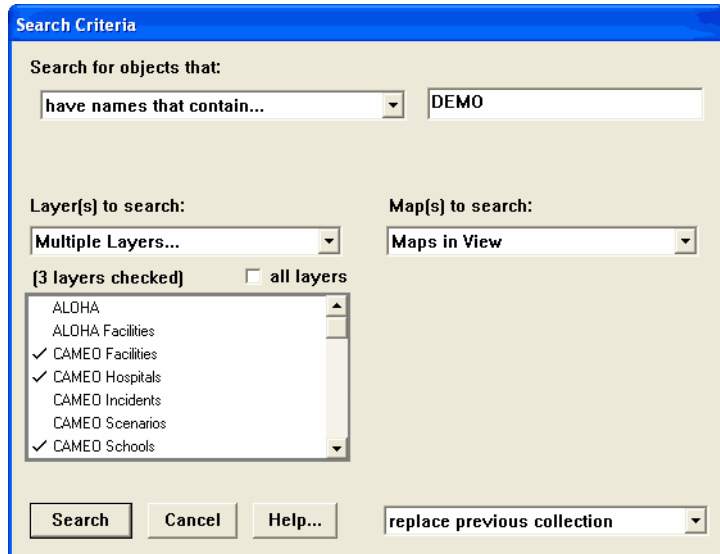
To delete the sample map symbols and map: Follow these steps to delete the sample map symbols (which are located on your CAMEO Map) and the sample Prince William County map from your hard drive.

1. Click the Home button in CAMEO's toolbar, then click the MARPLOT button to go to MARPLOT.
2. From MARPLOT's List menu, select Layer List.
3. Click the lock icons next to CAMEO Facilities, CAMEO Hospitals, and CAMEO Schools to unlock these layers (when clicked, the icons change to look like unlocked padlocks). The layer list should look like the one below. Click OK.

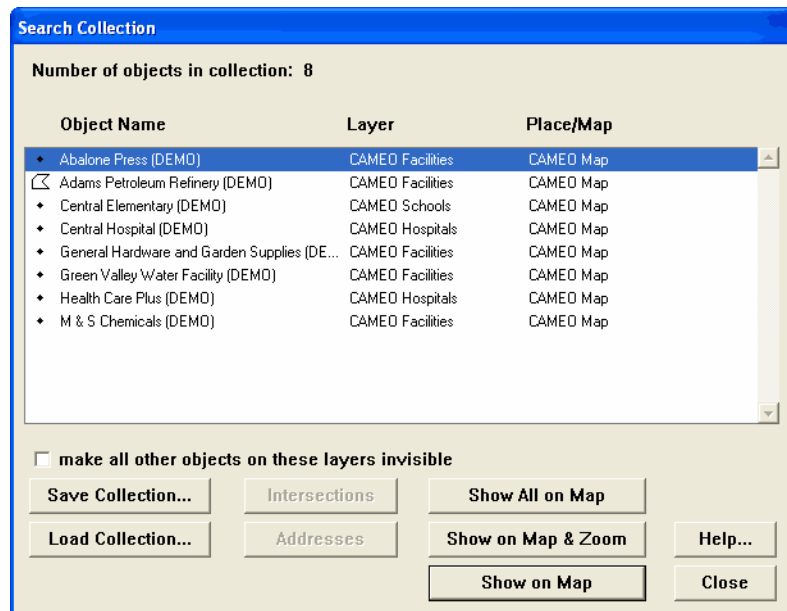


4. From the List menu, select Search.
5. Under "Search for objects that:" select "have names that contain." In the text box to the right, type "DEMO."

6. Under “Layers to search,” select Multiple Layers.
7. Click on “CAMEO Facilities,” “CAMEO Hospitals,” and “CAMEO Schools” to place checkmarks next to these layers. Click on any other layers that are checked, to uncheck them. When you’ve finished, your search criteria should look like the ones below.



8. Click Search. The eight sample symbols should appear in the search collection (as shown below).



9. Click Show All on Map. All six symbols will appear selected on the map.



10. From the Edit menu, select Clear (as at left). The sample objects are deleted from your CAMEO map.

11. Quit (Exit) from MARPLOT.

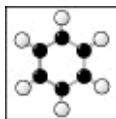
12. On your hard drive, find and open the "MARPLOT" folder.

13. Move the PWC_MAP folder to the Recycle Bin (in Windows) or Trash (on a Macintosh).

Working with Chemical Records

In this chapter, you'll find explanations of how to search for information about chemicals in the Chemical Library and how to predict the potential reactivity between chemicals, along with an explanation of the information about chemicals included in the Library.

The Chemical Library



The Chemical Library module contains more than 6,000 records. Each record describes a substance or mixture of substances, including its chemical name, trade names and other synonyms, identification numbers, regulatory information, and labeling conventions. For emergency responders and planners, the most important part of each chemical record is the **Response Information Data Sheet (RIDS)**, which contains a general description of the chemical, its physical properties, fire and health hazards, and recommendations for fire fighting and non-fire response, first aid, and protective clothing for response. RIDS information was compiled from a variety of source documents and databases (see [“Sources of CAMEO’s chemical data” on page 103](#)).

Searching for a Chemical Record


To view information about a chemical of concern, you first find the Chemical Library record describing that chemical. You can make either a **basic search** or a more **advanced search** for a record.

Making a basic search for a chemical record


Make a basic search when you just need to look up a chemical by its name, CHRIS code, UN number, CAS number, or other straightforward piece of identifying information. Here's how:

1. Begin the search in the following ways:

- a. While you're working in the Chemical Library, click

the  toolbar button or select Start Search from the Search menu (as at left).



- b. In the Home navigation window, click . (To access the Home navigation window, click the Home toolbar button or, from the File menu, select Go Home.)

2. In the Basic Search dialog, fill out the information you have about the chemical you're looking for. Check [Table 3-1 on page 82](#) to see explanations of all the choices in the dialog. If you fill in two or more criteria, CAMEO will search for chemicals that match all criteria. For example, if you type "slimicide" in the chemical name box and select "poison" from the DOT Label pull-down menu (as shown in [Figure 3-1 on page 81](#)), the record for "ACROLEIN, INHIBITED" will be found. Although two other chemicals also share the synonym "slimicide," and almost 900 other CAMEO chemicals also are placarded as "poison" when they are transported, only acrolein meets both criteria.
3. If you've typed text into a box (e.g., if you type "slimicide" in the chemical name box), then click the button for either:
 - a. Contains characters—to search for part of a word, phrase, code, or number.
 - b. Contains word starting with—to search either for the first part of or for an entire word, phrase, code, or number.

FIGURE 3-1. Searching for the poison, “slimicide.”

Note: For many searches, either choice will work, but sometimes the choice makes a big difference. For example, if you’re searching for 1-bromo 3-chloropropane, and you type “propane” in the chemical name box, you’ll find the chemical only if you select “Contains characters,” because this name contains the characters “propane,” but not at the beginning of a word.

4. Click Search to run your search. CAMEO will run the search and then display either a list of the chemicals that match your criteria, or a message, “No records found,” if no chemicals in the Chemical Library match your criteria.
5. Double-click the name of any chemical in the Found Chemicals list to view that chemical’s record. Whenever you want to return to the Found Chemicals list, press the View List button in the toolbar. If you’d like to view the list of all CAMEO chemicals again, in the Search menu, select Clear Search.

TABLE 3-1. Criteria you can use in a basic search.

Name	Description
Operator for text fields	Choose “Contains characters” to search for part of a word, phrase, code, or number; or “Contains word starting with” to search either for the first part of or for an entire word, phrase, code, or number.
Chemical Name	Type all or part of the name, synonym, or trade name for the chemical.
CAS #	Type all or the first part of the Chemical Abstracts Service number for the chemical, including all hyphens.
DOT Label	Select the required DOT label for the chemical from the pull-down menu (for example, “Flammable” or “Explosive”).
CHRIS Code	Type all or part of the U.S. Coast Guard’s three-letter code for a chemical (as used in the CHRIS guide to chemical hazards, http://www.chrismanual.com/).
UN/NA Number	Type all or part of a chemical’s UN Number.
Reactive Hazards	Select a special reactive hazard of the chemical (e.g., “Water-Reactive” or “Strong Oxidizing Agent”) from the pull-down menu.
General Description	Type a word or phrase describing the appearance, behavior, or other attribute of the chemical, which you think might be included in its general description. For example, a search for “blue-green crystal” finds several chemicals with that appearance.

Making an advanced search for a chemical record

You can make an advanced search whenever you need to use other criteria than the ones available for a basic search. Here’s how:

1. Begin your search just as you would start a basic search: either
 - a. click Search for a Chemical in the Home navigation window, or
 - b. while you’re working in the Chemical Library, click New Search in the toolbar or select Start Search from the Search menu.

2. Click Go to Advanced Search.
3. Click Select field to indicate which data field to search.
4. Click the radio button for either
 - a. Chemical ID and RIDS (to search for identifying information or response recommendations), or
 - b. Properties (if you want to search for a physical property, such as boiling point or lower explosive limit).
5. In the list of searchable fields, click the name of a field, then click Select.
6. Choose an operator from the popup menu (e.g., “contains characters,” “is equal to,” or “is greater than”). The operator to choose depends on the search you’re making.
7. Type the word, phrase, number, or code to search for in the box. **Note:** Leave the box empty if you don’t need to type something in—for example, when you choose an operator like “is empty” or “is not empty.”
8. Click Search to run your search.

Some advanced search examples:

- EPCRA EHS Chemical (is) “YES” finds all Extremely Hazardous Substances in the Chemical Library.
- Boiling Point is less than or equal to 60 finds chemicals that are gases when unconfined at temperatures above 60°F.
- NFPA Flammability contains the character “4” finds all the substances that pose the maximum flammability hazard under the NFPA system.¹

Adding more choices. You can make an advanced search for records that match more than one criterion. To add a second criterion to your search,

1. While you’re working in the Advanced Search dialog, press Add a Choice to add a second criterion to search for.
2. Follow steps 3 through 7 above to set up the search for that criterion.

1. Under the NFPA system, a chemical posing the highest possible flammability, reactivity, or health hazard is given a ranking of 4, on a 0 to 4 scale. (See “[Under the NFPA 704 Codes tab:](#)” on page 89.)

3. Indicate whether to search either for
 - a. records that meet *both* your criteria (click “Search for all of the following”), or
 - b. records that meet *either* criterion (click “Search for any of the following”).

Here’s an example of the difference between these two kinds of searches: To find out how many chemicals in the Chemical Library pose the most severe flammability, reactivity, *and* health hazard, you would click “Search for all of the following (AND search)” to search for values equal to “4” in the NFPA Flammability, NFPA Health Hazard, AND NFPA Reactivity boxes, as shown below.

The screenshot shows the 'CAMEO Advanced Search' window. At the top, it says 'Module to search: Chemical Library'. There are two radio buttons: 'Search for any of the following (OR search):' (unselected) and 'Search for all of the following (AND search):' (selected). To the right are buttons for 'Save This Search' and 'Open Saved Search'. Below this, there are three search criteria stacked vertically, each separated by 'AND'. Each criterion has a 'Select field:' dropdown, a search operator dropdown, and a 'Search for this:' text box. The first criterion is 'NFPA Flammability' with operator 'is equal to' and value '4'. The second is 'NFPA Health Hazard' with operator 'is equal to' and value '4'. The third is 'NFPA Reactivity' with operator 'is equal to' and value '4'. At the bottom, there are buttons for 'Clear All Fields', 'Add a Choice', 'Remove a Choice', 'Cancel', 'Search', and 'Return to Basic Search'.

You would find just four records that meet your criteria.

If, instead, you wanted to search for the chemicals that pose the most severe of *any one* of those hazards, you would click “Search for any of the following (OR search)” to search for values equal to “4” in the NFPA Flammability, NFPA Health Hazard, OR NFPA Reactivity boxes, as shown on the next page.

CAMEO Advanced Search

Module to search: **Chemical Library** Save This Search

Search for any of the following (OR search): Open Saved Search

Search for all of the following (AND search):

Select field: NFPA Flammability is equal to Search for this: 4

OR

Select field: NFPA Health Hazard is equal to Search for this: 4

OR

Select field: NFPA Reactivity is equal to Search for this: 4

Clear All Fields Add a Choice Remove a Choice Cancel Search

Return to Basic Search

You would find about 200 records meeting these criteria.

To add another choice, click Add a Choice again. You can add up to four choices. If you need to search for more than four criteria, first run a search for the first four of your criteria, then choose Append Search (to widen your search) or Subset Search (to narrow your search) from the Search menu to add additional criteria (see [“Append searches and subset searches” on page 250](#)).

You can save any complex advanced search to reuse later. For instructions, see [“Saving searches” on page 250](#).

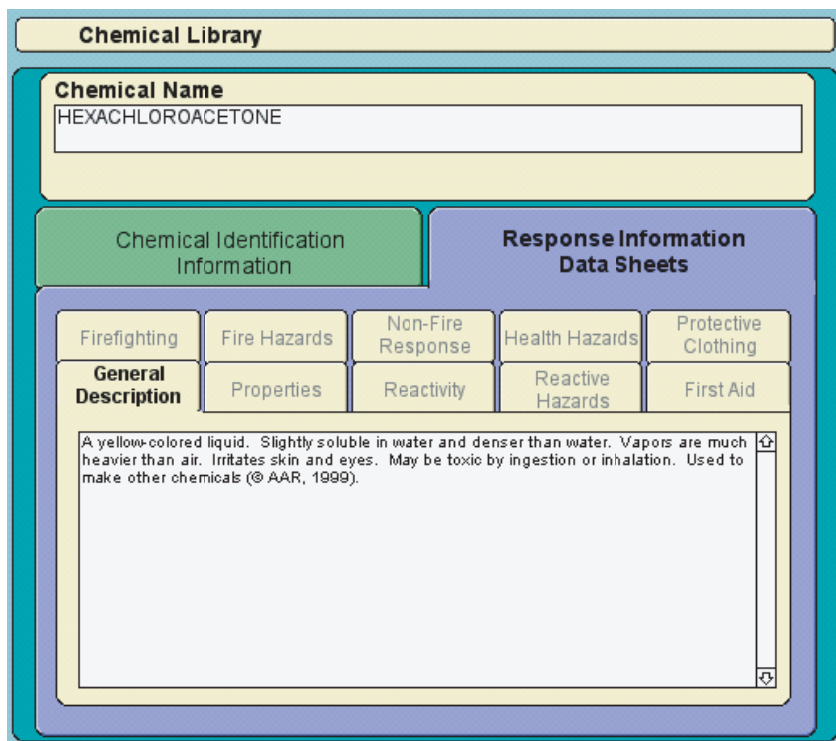
Understanding the information in the Chemical Library

A chemical record contains two basic kinds of information, each represented by a main tab within the record: **Chemical Identification Information** ([Figure 3-2 on page 86](#)) and a **Response Information Data Sheets** tab ([Figure 3-3 on page 87](#)).

The screenshot displays a web-based interface for a Chemical Library. At the top, a yellow header bar contains the text "Chemical Library". Below this, a white box labeled "Chemical Name" contains the text "HEXACHLOROACETONE". The main content area is divided into two sections: "Chemical Identification Information" (green background) and "Response Information Data Sheets" (purple background). Under "Chemical Identification Information", there are five tabs: "Chemical Identification" (selected), "Synonyms", "NFPA Codes", "Regulatory Information", and "Screening and Scenarios". The "Chemical Identification" tab is active, showing several input fields: "Formula:" with the value "C3Cl6O", "DOT Label:" with the value "POISON", "CAS #:" with the value "116-16-5", "UN/NA Number:" with the value "2661", "STCC #:" with the value "4925108", and "CHRIS Code:" which is empty. Each of the CAS, UN/NA, STCC, and CHRIS fields has a small dropdown arrow icon to its right.

FIGURE 3-2. View of the Chemical Information section of the record for Hexachloroacetone.

Check the Chemical Identification information to make sure that you're working with the record for the correct chemical, and to quickly see what legislation regulates uses of this chemical. [Table 3-2 on page 88](#) includes explanations of each kind of chemical identification information in Chemical Library records.



Chemical Library

Chemical Name
HEXACHLOROACETONE

Chemical Identification Information		Response Information Data Sheets		
Firefighting	Fire Hazards	Non-Fire Response	Health Hazards	Protective Clothing
General Description	Properties	Reactivity	Reactive Hazards	First Aid
<p>A yellow-colored liquid. Slightly soluble in water and denser than water. Vapors are much heavier than air. Irritates skin and eyes. May be toxic by ingestion or inhalation. Used to make other chemicals (© AAR, 1999).</p>				

FIGURE 3-3. View of the Response Information Data Sheet for Hexachloroacetone.

TABLE 3-2. Chemical identification information in a Chemical Library record.

Item	Description
Chemical Name	Name of the substance. Most often, this is the name used for this substance in the DOT Emergency Response Guidebook (USDOT 2000; see “ Bibliography ” on page 272). Otherwise, it’s usually the name for this substance listed in the Title III List of Lists (USEPA 2001; see “ Bibliography ” on page 272), or the name assigned by the U.S. Department of Transportation (as listed in 49 CFR 172—Subpart B—Table of Hazardous Materials and Special Provisions). In a few other cases, names were taken from other sources (see “ Sources of CAMEO’s chemical data ” on page 103).
Under the Chemical Information tab:	
Formula	Chemical formula used by the American Chemical Society.
DOT Label	Required DOT label for the chemical. This label must be displayed on shipped packages, railroad tank cars, and tank trucks according to specifications described in 49 CFR § 172.
CAS #	Chemical Abstracts Service registry number. Identification number assigned to this chemical by the American Chemical Society.
UN/NA Number	The United Nations-North America number. This numbering system was developed by the U.S. Department of Transportation, and then became the UN standard system for classifying hazardous materials.
STCC #	Standard Transportation Commodity Code. Seven-digit chemical identification code used by the Association of American Railroads.
CHRIS Code	Three-letter code used by the U.S. Coast Guard to identify individual chemicals included in its CHRIS guide to chemical hazards (http://www.chrismanual.com/).
Under the Synonyms tab:	
Synonyms	Names for this chemical, including trade names and other synonyms.

TABLE 3-2. Chemical identification information in a Chemical Library record. (Continued)

Under the NFPA 704 Codes tab:	
<p>NFPA codes represent hazards posed by a chemical. A ranking of “0” in one of the fields listed below represents little hazard to health, low flammability, or little reactivity. Higher rankings represent increased hazard. A “4” represents the highest hazard in all three categories.</p>	
Flammability	NFPA code 0 - 4, representing flammability hazard of the chemical.
Health	NFPA code 0 - 4, representing health hazard of the chemical.
Reactivity	NFPA code 0 - 4, representing the chemical’s reactivity.
Special	Notations that alert you to especially significant hazards posed by a few chemicals: “No water” means that the chemical is unusually reactive with water and “Oxidizer” means that the material is a strong oxidizer.
Under the Regulatory Information tab:	
Names	Names under which this substance is regulated under Federal laws such as CERCLA, EPCRA, RCRA, and the Clean Air Act of 1990.
CAA Section 112(r) chemical	Checkbox. If checked, this chemical has been listed as a toxic, flammable, or explosive hazardous substance under Section 112(r) of the Clean Air Act.
CERCLA Chemical	Checkbox. If checked, this chemical is listed as a hazardous substance under CERCLA (40 CFR § 302).
EPCRA EHS Chemical	Checkbox. If checked, this is an Extremely Hazardous Substance (EHS), identified by the U.S. Environmental Protection Agency as an acute inhalation toxic threat, and listed under EPCRA.
ECPRA Section 313 chemical	Checkbox. If checked, this chemical has been included in a list of toxic chemicals covered under Section 313 of EPCRA. Facilities storing or using Section 313 chemicals may be required to submit annual reports of the amount of each chemical released into the environment, either routinely or by accident (40 CFR § 372).
CAA Threshold Quantity	Threshold Quantity (TQ) listed in the Clean Air Act. When stored quantities of this substance exceed this amount, a facility becomes subject to the accidental release prevention provisions of the Act.

TABLE 3-2. Chemical identification information in a Chemical Library record. (Continued)

CERCLA Reportable Quantity	Reportable Quantity (RQ) listed in CERCLA. Threshold release size established by the U.S. Environmental Protection Agency. A facility must report spilled quantities in excess of the RQ to Federal, state, and local governments.
EHS Threshold Planning Quantity	Extremely Hazardous Substance Threshold Planning Quantity, listed in 40 CFR § 355. When storage quantities of a hazardous chemical at a facility exceed this threshold, the reporting, community right-to-know, and emergency planning requirements of EPCRA must be met.
RCRA Chemical Code	Four-character identification code assigned to this substance under the Resource Conservation and Recovery Act of 1976 (RCRA).
Under the Screening and Scenarios tab:	
Liquid Factor (Ambient)	A factor required to calculate evaporation rate of a liquid at ambient temperature, using methods described in <i>Technical Guidance for Hazards Analysis</i> (USEPA, FEMA, and USDOT 1987; see “Bibliography” on page 272).
Liquid Factor (Boiling)	A factor required to calculate evaporation rate of a liquid at or above its boiling point, using methods in the <i>Technical Guidance</i> .
Liquid Factor (Molten)	A factor required to calculate rate of release of a molten solid, using methods in the <i>Technical Guidance</i> .
Level of Concern	LOC. The concentration of the chemical, in grams per cubic meter, above which there may be serious, irreversible health effects as a result of a single exposure for a relatively short period of time. Defined in the <i>Technical Guidance</i> . Local authorities may establish additional LOCs for their hazards analyses.
Ambient State	The physical state (solid, liquid, or gas) of the chemical at 68°F.

Refer to a chemical’s RIDS information when you need to see recommendations for response to releases of that chemical. [Table 3-3 on page 91](#) includes explanations of each of the kinds of information under the Response Information Data Sheets tab in each chemical record. [Table 3-4 on page 92](#) shows the full names of protective materials listed under the Protective Clothing tab; these names are shown abbreviated in CAMEO.

TABLE 3-3. Response Information Data Sheet information in a Chemical Library record.

Tab Title	Tab Contents
General Description	General appearance, behavior, and toxicity of the chemical.
Properties	Physical properties, flammability limits, and toxic thresholds (toxic exposure limits). To view a definition of each physical property, see “Glossary of Terms” on page 317 .
Reactivity	Description of the potential reactivity between the chemical and other chemicals, as well as its reactivity with air and water, and any other intrinsic reactive hazards (e.g., polymerizability, peroxidizability). Also includes a list of the reactive groups to which this chemical belongs.
Reactive Hazards	Special reactivity alerts for the chemical.
First Aid	Response recommendations.
Firefighting	Response recommendations for fire incidents in which the chemical is involved.
Fire Hazards	Description of the chemical’s flammability, byproducts that may evolve if the chemical is burned, and risk of explosion.
Non-Fire Response	Response recommendations for incidents not involving fire, in which the chemical is involved.
Health Hazards	Description of the health hazards of the chemical, such as toxicity, flammability, and corrosivity.
Protective Clothing	Recommendations and table of breakthrough times for protective materials. See Table 3-4 on page 92 for the full names of listed materials (which are shown abbreviated on Chemical Library records).

TABLE 3-4. Full names of protective materials listed under the Protective Clothing tab on Chemical Library records.^a

Abbreviation displayed in CAMEO	Full name of the protective material
ACRYLIC	Same
AL PE	Aluminized Polyethylene
AL PE/PET	Aluminized Polyethylene/Polyethylene Terephthalate
AL PET	Aluminized Polyethylene Terephthalate
AL PET/PVC	Aluminized Polyethylene Terephthalate/Polyvinyl Chloride
AL PVF	Aluminized Polyvinyl Fluoride
BARRICADE	Plastic Laminate (Dupont Company)
BLUE MAX	Plastic Laminate (MSA Company)
BUTYL	Butyl Rubber
BUTYL/NAT RUB	Butyl Rubber/Natural Rubber
BUTYL/NEOP	Butyl Rubber/Chloroprene Rubber
CELLULOSE ACETATE	Same
CELLULOSE PROPIONATE	Same
CHECKMATE	Plastic Laminate (Lakeland Company)
CHEMREL	Plastic Laminate (Chemron Company)
CHEMREL MAX	Plastic Laminate (Chemron Company)
CHEMTUFF	Plastic Laminate (Chemron Company)
CHLOROBUTYL	Chlorobutyl Rubber
COMFORT-GARD II	Comfort-Gard II Fabric
CPE	Chlorinated Polyethylene
CPE/SARAN	Chlorinated Polyethylene/Polyvinylidene Chloride
CPE+HYPALON+PVC	Chlorinated Polyethylene + Hypalon + Polyvinyl Chloride

TABLE 3-4. Full names of protective materials listed under the Protective Clothing tab on Chemical Library records.^a (Continued)

Abbreviation displayed in CAMEO	Full name of the protective material
CPF III	Plastic Laminate (Kappler Company)
CR 39	CR 39
ECO/BUTYL	Epichlorohydrin/butyl Rubber
EMA	Ethylene - Methyl Acrylate
EPDM/BUTYL	Ethylenepropylene/butyl Rubber
EPDM+NAT RUB	Ethylenepropylene + Natural Rubber
EVAC	Ethylene Vinyl Acetate
FEP TEFLON	Fluorinated Ethylene Propylene
FEP/PTFE	Fluorinated Ethylene Propylene/Polytetrafluoroethylene
GORE-GARD	Gore-Gard Fabric
GORE-TEX	Gore-Tex Fabric
HYPALON	Same
HYPALON/NEOP	Hypalon/Chloroprene Rubber
INTERCEPTOR	Plastic Laminate (Lakeland Company)
MET PVF	Metallized Polyvinylfluoride
NAT RUB	Natural Rubber
NAT RUB+BAYPRENE	Natural Rubber + Bayprene
NAT RUB+NEOP	Natural Rubber + Chloroprene Rubber
NAT RUB+NEOP+NBR	Natural Rubber + Chloroprene Rubber + Nitrile-Butadiene Rubber
NAT RUB+NITRILE	Natural Rubber + Nitrile Rubber
NAT RUB+NITRILE+SBR	Natural Rubber + Nitrile Rubber + Styrenebutadiene Rubber

TABLE 3-4. Full names of protective materials listed under the Protective Clothing tab on Chemical Library records.^a (Continued)

Abbreviation displayed in CAMEO	Full name of the protective material
NEOP	Chloroprene Rubber
NEOP/BUTYL	Chloroprene Rubber/Butyl Rubber
NEOP/NAT RUB	Chloroprene Rubber/Natural Rubber
NEOP+NAT RUB/NITRILE	Chloroprene Rubber + Natural Rubber/nitrile Rubber
NEOP+PVC	Chloroprene Rubber + Polyvinyl Chloride
NEOP+PVC	Chloroprene Rubber + Polyvinyl Chloride
NEOP+SBR	Chloroprene Rubber + Styrenebutadiene Rubber
NITRILE	Nitrile Rubber
NITRILE/NAT RUB	Nitrile Rubber/Natural Rubber
NITRILE+PVC	Nitrile Rubber + Polyvinyl Chloride
NONWOVEN FABRIC	Same
PCTFE	Polychlorinated trifluoroethylene
PE	Polyethylene
PE/EVAL/PE	Polyethylene/Ethylene Vinyl Alcohol/Polyethylene
PET	Polyethylene Terephthalate
PETG	Polyethylene Terephthalate, Modified
POLYCARBONATE	Same
POLYIMIDE	Same
POLYISOPRENE	Same
POLYSULFONE	Same
POLYURETHANE	Polyurethane Rubber
POLYURETHANE/NAT RUB	Polyurethane Rubber/Natural Rubber

TABLE 3-4. Full names of protective materials listed under the Protective Clothing tab on Chemical Library records.^a (Continued)

Abbreviation displayed in CAMEO	Full name of the protective material
POLYURETHANE/NIT+PVC	Polyurethane Rubber/Nitrile Rubber + Polyvinyl Chloride
POLYURETHANE+PVC	Polyurethane Rubber + Polyvinyl Chloride
POLYVINYL CHLORIDE	Same
PP	Polypropylene
PTFE TEFLON	Polytetrafluoroethylene Teflon
PVAC	Polyvinyl Acetate
PVAL	Polyvinyl Alcohol
PVAL/PE	Polyvinyl Alcohol/Polyethylene
PVC	Polyvinyl Chloride
PVDC	Polyvinylidene Chloride
RESPONDER	Plastic Laminate (Life-Guard Company)
SARANEX23P	Polyethylene/Polyvinylidene Chloride/Polyethylene/Tyvek
SBR	Styrenebutadiene Rubber
SBR/NEOP	Styrenebutadiene Rubber/Chloroprene Rubber
SILICONE RUB	Silicone Rubber
SILVER SHIELD	Same
TRELLCHEM HPS	Plastic Laminate (Trellchem Company)
TYVEK QC SUPRA	Polyethylene/Tyvek
VELOSTAT	Microporous Polyolefin
VITON	Fluoroelastomer
VITON/BUTYL	Fluoroelastomer/Butyl Rubber
VITON/BUTYL/NEOP	Fluoroelastomer/Butyl Rubber/Chloroprene Rubber

TABLE 3-4. Full names of protective materials listed under the Protective Clothing tab on Chemical Library records.^a (Continued)

Abbreviation displayed in CAMEO	Full name of the protective material
VITON/CHLOROBUTYL	Fluoroelastomer/Chlorobutyl Rubber
VITON/NEOP	Fluoroelastomer/Chloroprene Rubber
VITON/NITRILE	Fluoroelastomer/Nitrile Rubber

a. In abbreviations, “+” indicates a blend of two or more materials; “/” indicates that two or more materials are layered.

Predicting potential chemical reactivity

Reactivity is the tendency of substances to undergo chemical change. You can use CAMEO to find out about the reactivity of substances or predict the reactivity of mixtures of substances. You can

- Check any chemical’s intrinsic reactive properties, such as peroxidizability, polymerizability, and radioactivity.
- “Virtually mix” chemicals to find out what dangers could arise if they were accidentally mixed together. To make reactivity predictions, you select chemicals from the Chemical Library module, and add them to a “mixture.” CAMEO then predicts the reactivity of this mixture.

How CAMEO predicts mixture reactivity

Each substance in the Chemical Library was assigned to one or more reactive groups, based on the known chemistry of that substance.² Reactive

2. CAMEO’s reactivity prediction method was inspired by earlier work by the Hazardous Materials Management Section of the California Department of Health Services, and by the U.S. Coast Guard. To develop CAMEO’s method, the CAMEO team was substantially aided by Dr. Wade Freeman and Dr. Mike Krumpolc of the University of Illinois.

groups are categories of chemicals that react in similar ways because they are similar in their chemical structure.³

To find out which reactive groups a particular chemical belongs to, find that chemical's record in the Chemical Library, click the Response Information Data Sheets tab, and then click the Reactivity tab. Under the REACTIVE GROUPS heading, you'll see the reactive group(s) to which the chemical belongs (as shown in Figure 3-4).

To see detailed descriptions of all the reactive groups used in CAMEO, check [“The Reactive Groups” on page 275](#).

To predict the reactivity of a mixture of chemicals, CAMEO first identifies the reactive groups to which each of the chemicals belongs, and then predicts the kinds of chemical reactions likely to occur when members of these groups are mixed together.

Reactivity predictions are pairwise. CAMEO only predicts the reactivity between two chemicals at a time. If you virtually “mix” three or more chemicals, it will predict the reactivity between each of the possible pairings of those chemicals.

Be aware that:

- sometimes, three or more chemicals can react in ways CAMEO can't predict. For example, sulfuric acid, nitric acid, and glycerine react to produce nitroglycerine. CAMEO recognizes that this mixture would be

3. Each substance in the Chemical Library has been assigned to one or more of the following reactive groups (these are structurally distinctive chemical groups, within which the members behave similarly): acid halides; acids, inorganic nonoxidizing; acids, inorganic oxidizing; alcohols; aldehydes; amides; amines; anhydrides; azo, diazo, azido; bases, CFCs, HCFCs; carbamates; chlorosilanes; cyanides, inorganic; epoxides; esters; ethers; halogenated organics; halogenating agents, strong; hydrocarbons, aliphatic saturated and unsaturated; hydrocarbons aromatic; inorganic compounds/neither oxidizing or reducing; inorganic oxidizing agents; inorganic reducing agents; isocyanates; ketones; metal hydrides; metals, alkali; metals, elemental; metals, less reactive; nitrides; phosphides; carbides; silicides; nitriles; nitrites; nitrates; organometallics; peroxides; phenols; phosphates, thiophosphates; acidic salts; basic salts; inorganic sulfides; organic sulfides, thiocarbamate salts, dithiocarbamate salts.

Chemical Library

Chemical Name
ACROLEIN, INHIBITED

Chemical Identification Information

Response Information Data Sheets

Firefighting	Fire Hazards	Non-Fire Response	Health Hazards	Protective Clothing
General Description	Properties	Reactivity	Reactive Hazards	First Aid

ACROLEIN, [INHIBITED] can react violently with oxidizing agents. Polymerizes exothermically on contact with small amounts of acids (including sulfur dioxide), alkalis, volatile amines and pyridines, salts, thiourea, oxidizing agents (air) and on exposure to light and heat. Polymerization initiated by amines and pyridines occurs after a deceptive induction period. Watersolutions of mineral acids and metal ions can initiate polymerization. The inhibitor (usually hydroquinone) greatly reduces tendency to polymerize. Undergoes Diels-Alder reaction with itself to give acrolein dimer. This can become a runaway reaction at 90°C [Kirk-Othmer, 4th Ed, Vol. 1]. Mixing in equal molar portions with any of the following substances in a closed container caused the temperature and pressure to increase: 2-aminoethanol, ammonium hydroxide, chlorosulfonic acid, ethylenediamine, ethyleneimine [NFPA 1991].

REACTIVE GROUPS:
Aldehydes (Reactivity, 2001*)

FIGURE 3-4. Acrolein belongs to the “Aldehydes” reactive group.

very reactive, but does not predict that nitroglycerine would be produced.

- sometimes, one chemical can catalyze (speed up) the reaction between other chemicals. For example, nickel carbonyl catalyzes many polymerization reactions and other kinds of synthetic organic reactions. CAMEO cannot predict when a reaction between two chemicals could be sped up by another chemical.

However, reactions among more than two chemicals are relatively uncommon in nature (except for the catalyzed reactions that are common components of the metabolic processes of living organisms).

Making a reactivity prediction

Here are the steps for predicting the potential chemical reactions between two or more chemicals:

1. Find and open the Chemical Library record for one of the chemicals (see [“Searching for a Chemical Record” on page 80](#)).
2. From the Record menu, select Add to Reactivity Worksheet.
3. When asked whether you’d like to view the Worksheet, click Later.
4. Find and open the record for one of the other chemicals.
5. From the Record menu, select Add to Reactivity Worksheet.
6. When asked whether you’d like to view the Worksheet, click either
 - a. Later, if you have more chemicals to add. In this case, repeat steps 1 and 2 until you’ve added all the chemicals, then click Yes.
 - b. Yes, if you have finished adding chemicals.

The Reactivity Worksheet will be displayed. In the Worksheet, you’ll see a list of the chemicals you’ve “mixed,” followed by a listing of the potential hazards from reactivity between those chemicals (as shown in Figure 3-5). If you have mixed more than two chemicals, you’ll see a summary of the hazards of all possible pairings of the chemicals, followed by a list of the hazards from each possible pairing.

7. Click Print Report if you’d like to print your results. Once you’ve completed your work, you can click Clear All to clear all the chemicals from the Reactivity Worksheet (the Worksheet will be cleared when you quit from CAMEO).

Note: In CAMEO, you can “mix” not only individual chemicals, but reactive groups along with chemicals. You might want to do this if you know the chemical class of a substance, but not its exact name or CAS number, or if you know a substance’s chemical class (e.g., its MSDS may state its class), and the substance is not in the Chemical Library. To see detailed descriptions of all the reactive groups used in CAMEO, check [“The Reactive Groups” on page 275](#).

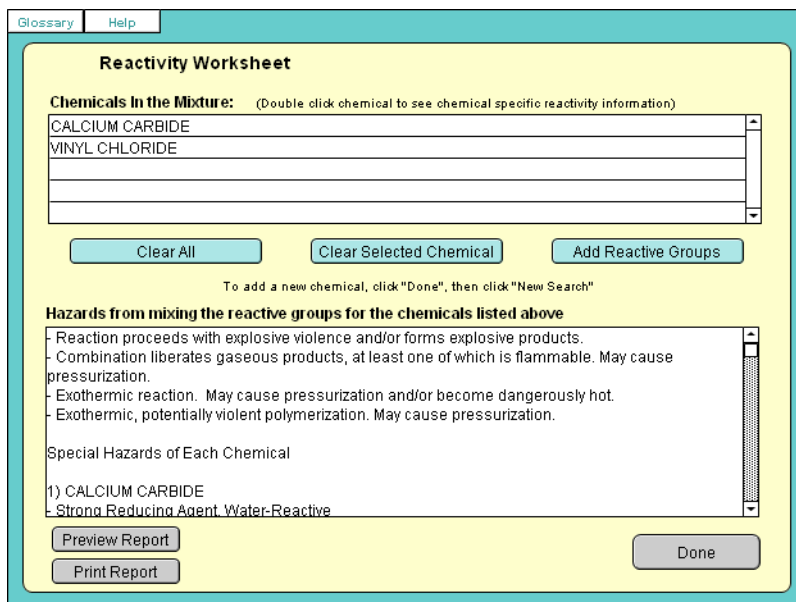


FIGURE 3-5. The Reactivity Worksheet.

Significant reactive hazards of some chemicals

Substances that pose certain special or acute reactivity-related hazards are categorized not only into reactivity groups, but also into nine **Reactive Hazard** classes shown in [Table 3-5 on page 101](#). To check whether a chemical poses any of these acute reactive hazards,

1. Find that chemical's record in the Chemical Library.
2. Click the Response Information Data Sheets tab, then click the Reactive Hazards tab to see the acute reactive hazards of that chemical (as in [Figure 3-6 on page 101](#)).

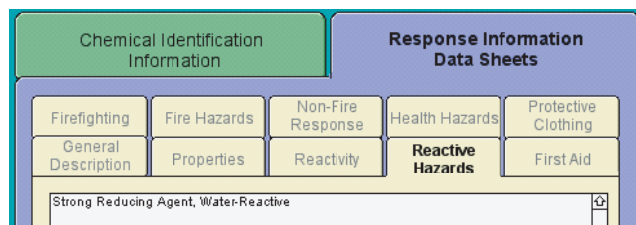


FIGURE 3-6. The acute reactive hazards of calcium carbide.

You also can search the Chemical Library to find all the chemicals that pose a particular reactive hazard. For example, to find all the water-reactive chemicals in the Library,

1. From the Search menu, select Start Search.
2. From the Reactive Hazards pull-down menu, select “Water-Reactive.”
3. Click Search. You’ll see a list of the more than 500 water-reactive chemicals in the Chemical Library.

TABLE 3-5. CAMEO’s Reactive Hazard classes.

Reactive Hazard class	Definition
Highly flammable	Substances having a flash point of less than 100°F and mixtures that include substances with flash points of less than 100°F.
Explosive	A material synthesized or mixed deliberately to allow the very rapid release of chemical energy. Also, a chemical substance that is intrinsically unstable and liable to detonate under conditions that might reasonably be encountered.
Polymerizable	Capable of undergoing self-reactions that release energy. Some polymerization reactions generate a great deal of heat. The products of polymerization reactions are generally less reactive than the starting materials.

TABLE 3-5. CAMEO's Reactive Hazard classes. (Continued)

Reactive Hazard class	Definition
Strong oxidizing agent	Oxidizing agents gain electrons from other substances and are themselves thereby chemically reduced. Strong oxidizing agents accept electrons particularly well from a large range of other substances. The ensuing oxidation-reduction reactions may be vigorous or violent and may release new substances that may take part in further additional reactions. Strong oxidizing agents should be kept well separated from strong reducing agents. In some cases, the presence of a strong oxidizing agent can greatly enhance the progress of a fire.
Strong reducing agent	Reducing agents give up electrons to other substances. They are themselves thereby oxidized. Strong reducing agents donate electrons particularly well to a large range of other substances. The ensuing oxidation-reduction reactions may be vigorous or violent and may generate new substances that take part in further additional reactions.
Water-reactive	Substances that may react rapidly or violently with liquid water and steam, producing heat (or fire) and often toxic reaction products.
Air-reactive	Likely to react rapidly or violently with dry air or moist air. May generate toxic and corrosive fumes upon exposure to air, or may catch fire.
Peroxidizable Compound	Apt to undergo spontaneous reaction with oxygen (a component of air) at room temperature, to form peroxides and other products. Most such autooxidations are accelerated by light or by trace impurities. Many peroxides are explosive, which makes peroxidizable compounds a particular hazard. Ethers and aldehydes are particularly subject to peroxide formation (the peroxides generally form slowly after evaporation of the solvent in which a peroxidizable material had been stored).
Radioactive Material	Spontaneously and continuously emitting ions or ionizing radiation. Radioactivity is not a chemical property, but an additional hazard that exists in addition to the chemical properties of a material.

Sources of CAMEO's chemical data

Information displayed in the Chemical Library was compiled from a variety of documents and databases, each prepared by a different organization, such as the American Association of Railroads or the EPA. Information sources are identified in two ways:

- Following each piece of information displayed in a Response Information Data Sheets field is the abbreviation for the source from which that information was taken and that source's date of release, shown in parentheses: for example, "(USCG, 1999)."⁴
- [Table 3-6 on page 105](#) shows the sources of the chemical identification information included in each record in the Chemical Library, and [Table 3-7 on page 106](#) shows the sources of the RIDS information for each chemical.

For a description of each source document or database, see "[Bibliography](#)" [on page 272](#). In the Bibliography, a notation in brackets (such as "[EPA]") appearing within a citation indicates that the cited reference is one of the sources of CAMEO's chemical data.

In both [Table 3-6](#) and [Table 3-7](#), the numbers in the table cells indicate the priority of the various databases used for each data field in a Chemical Library record. Smaller numbers indicate higher priority: "1" in a table cell indicates the highest-priority source for a particular piece of information. For example, [Table 3-6](#) shows how the names for the chemicals in the Chemical Library were chosen. When different names for a particular chemical appeared in different databases, our preference was to adopt the name for that chemical used in the *Emergency Response Guidebook* (U.S. Department of Transportation 2000), our second preference was to use the name for that chemical listed in the *Title III List of Lists* (EPA 1999), our third preference was to use the name shown in 49 Code of Federal Regulations, and so on. In the case of synonyms for each chemical name,

4. The sole exception to this rule is Reactive Hazards; this information was generated by NOAA during the development of the Chemical Reactivity Worksheet.

we included all the synonyms for a given chemical from the indicated sources.

Some of the information in the Chemical Library was edited by the CAMEO Team after it was extracted from the original source. In [Table 3-6](#), chemical names shown in the Name field were taken from the sources shown and then further edited. Information in the Formula, CAS Registry Number, and Label fields was extracted from the sources shown, then lightly edited. In [Table 3-7](#), texts shown in the General Description field were taken from the sources shown and then further edited; molecular weights were extracted from the sources shown, then lightly edited.

TABLE 3-6. Sources of chemical identification information in Chemical Library records. (See [“Bibliography”](#) on page 272.)

	49 CFR	DOT	AAR	USCG	CAS	NIOSH	Title III	NTP	EPA	TECH
Name	3	1	6	4			2	5		
UN/NA Number	1	2	4					3		
Formula					1			2		
CAS #				5	1	4	3	2		
STCC #			1							
DOT Label	1							2		
CHRIS Code				1						
Synonyms		1		1	1	1	1	1	1	
Regulatory Information^a							1			
Screening & Scenarios^b										1

- a. Items under the Regulatory Information tab, including regulated names, threshold quantities, and other information established by one or more Federal laws: the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA), the Resource Conservation and Recovery Act (RCRA), EPCRA (Sections 302 and 313), and the Clean Air Act (Section 112r).
- b. Items under the Screening & Scenarios tab, including liquid factors, level of concern, and physical state at ambient temperature. All these items are used for hazards analysis. See [“Using Screening & Scenarios to assess hazards”](#) on page 139.

TABLE 3-7. Sources of RIDS data in Chemical Library records.

	NFPA	AAR	EPA	USCG	DOT	Little	NIOSH	NTP	React
General Description		1	2	3			5	4	
Fire Hazards			1	2	3			4	
Firefighting		2	1	3	4			5	
Protective Clothing summary text		4	2	3			1	5	
Protective Clothing ratings table						1		2	
Non-fire Response		1	2	3	4			5	
Health Hazards			1	2	3		4	5	
First Aid		6	1	4	5		3	2	
Flash Point		5	1	3			4	2	
Lower Explosive Limit (LEL)			1	3			4	2	
Upper Explosive Limit (UEL)			1	3			4	2	
Autoignition Temperature		3		1				2	
Melting Point		5	1	3			4	2	
Vapor Pressure		5	1	3			4	2	
Vapor Density			1	3				2	
Specific Gravity		4	1	2			5	3	
Boiling Point		5	1	3			4	2	
Molecular Weight			2	4			5	3	1
Water Solubility							2	1	

Working with Facility Records

This chapter explains how to use CAMEO to keep track of information about facilities where hazardous materials are maintained. It explains how to access and use facility information already stored in CAMEO, how to create new CAMEO records for facilities, and how to use CAMEO to maintain Tier II information submitted by facilities in your area.

Where facility information is kept

You keep information about facilities in five related CAMEO modules. You can access all the information about a given facility from the facility's record in the Facilities module, which acts as a sort of central hub for information about that facility.

- In the Facilities module, you create a record containing the basic information about a given facility, such as its location, its owner, and the type of business it is. If you have created a site plan for the facility in a graphics program, you can link that plan to this facility record. You can also keep track of reports relating to each facility that have been submitted under the reporting requirements of the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA). And you can

link any facility's record to a symbol for that facility that you place on a MARPLOT map, so that you can quickly see the facility's location in your community or use Screening & Scenarios or ALOHA to assess the potential hazard to the community posed by chemicals at the facility (see [“Using Screening & Scenarios to assess hazards” on page 139](#) or [“Using ALOHA with MARPLOT and CAMEO” on page 192](#)).

- In the Chemicals in Inventory module, you can create a record for each of the hazardous substances stored at a facility (mixtures as well as pure substances). Each Chemicals in Inventory record contains information such as the quantity of the substance stored at the facility, the conditions of storage, and the location in the facility where the substance is stored.
- In the Contacts module, you can create a record for each of the contact people for the facility, such as the emergency manager and owner.
- In the Screening & Scenarios module, you can use simple, automated methods to assess the hazards to the surrounding community from accidental releases from that facility, following the procedures described in the guidebook, *Technical Guidance for Hazards Analysis* (see [“Bibliography” on page 272](#) to learn how to obtain a copy).
- In the Incidents module, you can keep records describing past accidental releases of hazardous substances from that facility. This information can be used for risk analysis.

Tier II information and CAMEO

You can store information submitted on Tier II reports in the Facilities, Chemicals in Inventory, and Contacts modules.¹ If you're familiar with Tier II forms, you'll notice that many of the data fields in the Facilities, Chemicals in Inventory, and Contacts modules are identical to data fields on Tier II forms.

1. Under the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA), facilities that maintain more than threshold amounts of certain hazardous materials must submit an annual chemical inventory report, called a Tier II form, to their state's State Emergency Response Commission (SERC), Local Emergency Planning Committee (LEPC), and local fire department. See [“Sections 311 and 312: community right-to-know requirements” on page 266](#).

You can put Tier II data in CAMEO in either of two ways:

Manually entering data from print Tier II forms into CAMEO

To do this,

1. Create a new Facilities record for each submitting facility (see [“Creating and editing facility records” on page 122](#)).
2. Create a new Chemicals in Inventory record for each chemical or mixture maintained at that facility (see [“Keeping track of the chemicals in a facility’s inventory” on page 125](#)). [Figure 4-1 on page 110](#) shows the process.
3. Create new Contacts records for the contacts listed for each facility.
4. For each annual update, either:
 - a. Edit the facility, inventory, and contacts records to update them as needed (to edit a CAMEO record, open the record, click the Edit button in the toolbar, make the needed changes, then press Save Changes), or
 - b. Add new records for the new reporting year, if you would rather keep a set of records for each reporting year.

Importing a Tier2 Submit data file

To import a Tier2 Submit file, follow the procedure described in [“Importing data from Tier2 Submit” on page 205](#). When you import a Tier2 Submit file, new Chemicals in Inventory records are created automatically for all the chemicals in each facility’s inventory. Each such record contains all the imported data about the given chemical. New Contacts records are created automatically for the contacts listed in the Tier II form submitted by each facility.

Note: It isn’t possible for a facility manager to maintain records for that facility’s Tier II chemicals in CAMEO, and then, at reporting time, export that information to Tier2 Submit in order to generate the submittable data file.

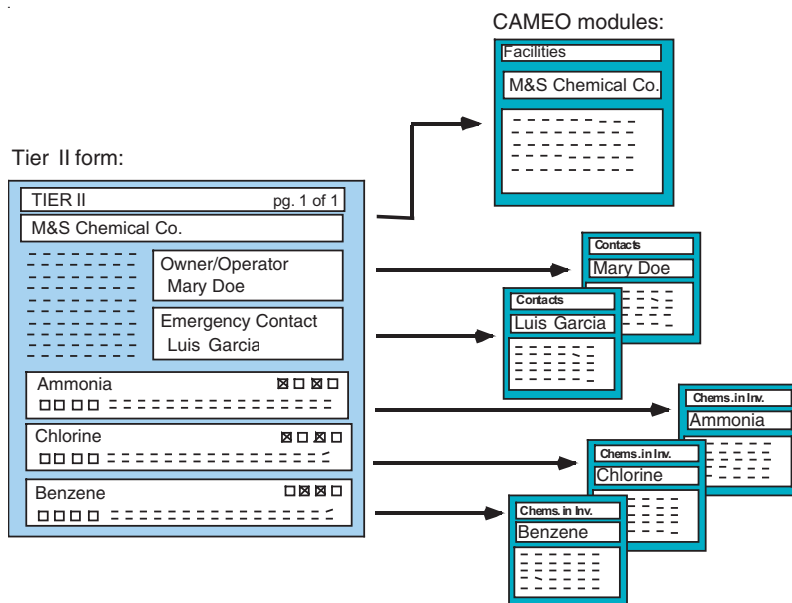


FIGURE 4-1. From a printed Tier II form, facility information is entered into a Facilities record; each contact person for the facility is represented by a Contacts record, and each chemical in the facility's inventory is represented by a Chemicals in Inventory record.

State fields in the Facilities module. Records in both the Facilities and Chemicals in Inventory modules contain a State Fields tab. This section is designed to support the emergency planning work of U.S. states that have developed their own reporting requirements in addition to the Tier II requirements specified in EPCRA, which apply to all 50 states. In these states, Tier II forms contain additional data fields, called **state fields** in CAMEO. CAMEO's state fields, located under the State Fields tab, represent those state-required fields, providing a place where people working in these states can maintain their reporting information.

Accessing information about a facility


To see the information in CAMEO on a particular facility, you first search the Facilities module to find the record describing that facility. Once you've found that record, you then can review:

- that record itself, for basic information about the facility.
- records in the related modules, for additional information about the facility.

You can make either a basic search or a more advanced search to find a facility record.

Making a basic search for a facility record

Make a basic search when you just need to look up a facility by its name, address, or other straightforward piece of identifying information. Here's how:

1. Start the search in either of two ways:
 - a. While you're working in the Facilities module, click the  button or select Start Search from the Search menu.
 - b. In the Home navigation window, click Search for a Facility. (To access the Home navigation window, click the Home toolbar button or, from the File menu, select Go Home.)
2. In the Basic Search dialog, fill out the information you have about the facility you're looking for. For your "Operator for text fields," choose "Contains characters" to search for part of a word, phrase, code, or number; or "Contains word starting with" to search either for the first part or all of a word, phrase, code, or number. Check [Table 4-1 on page 115](#) to see explanations of the other choices in the dialog.

Note: If you fill in two or more criteria, CAMEO will search for facilities that match all criteria. For example, if you type "Haymarket" in the City field and choose "Yes" from the Has EHS Chemical in Inventory menu (as shown in [Figure 4-2 on page 112](#)), you'll see a list of all the facilities in Haymarket that store or use Extremely Hazardous Substances (EHSs).

FIGURE 4-2. Searching for the facilities in Haymarket that maintain EHSs.


3. Click Search to start your search. CAMEO will run the search and then display either a list of the facilities that match your criteria or a message, “No records found,” if no facilities match your criteria.
4. Double-click the name of any facility in the Found Facilities list to view that facility’s record in Record view. Whenever you want to return to the Found Facilities list, press the View List button in the toolbar. If you’d like to view the list of all facilities again (as opposed to just the ones that met your search criteria), from the Search menu, select Clear Search.

Some basic search examples:

- To find the most recent records for facilities in a particular zip code, type that zip code in the Zip Code box, then for Report Year, choose the “is equal to” operator, and type the most recent report year in the box.
- To find the facilities in a particular county that store EHS chemicals (Extremely Hazardous Substances), from the Has EHS Chemical in Inventory menu, choose “Yes,” and then type all or part of the name of the county in the County box.

Making an advanced search for a facility record

You can make an advanced search whenever you need to use other criteria than the ones available for a basic search. Here's how:

1. Start your search just as you would start a basic search, by either
 - a. clicking Search for a Facility in the Home navigation window, or
 - b. while you're working in the Facilities module, clicking the  button or selecting Start Search from the Search menu.
2. Click Go to Advanced Search.
3. Click Select field.
4. Click the radio button for:
 - Facilities—to search for a facility by basic information about it, such as its name or address.
 - Chemicals in Inventory—to search for a facility that stores or uses a particular chemical.
 - Contacts—to search for a facility for which a particular person is a contact (e.g., an emergency contact or the facility's owner).
 - Incidents—to search for a facility where a particular incident occurred.
 - Mixture Components—to search for a facility storing a mixture made up of particular components.
 - Screening and Scenarios—to search for a facility for which you have prepared hazards analysis screenings or scenarios, using the Screening & Scenarios module.
 - Storage Locations—to search for a facility where a chemical is stored at a particular location (e.g., "Warehouse 43").
5. In the list of searchable data fields, click the name of a field, then click Select.
6. Choose an operator from the popup menu (e.g., "contains characters," "is equal to," or "is greater than"). Your choices for operator depend on the kind of data you're searching for: whether it's text, a number, or a date, or either/or information (for which your choices are "yes" or "no").

7. Type the word, phrase, number, or code to search for in the box. Leave the box empty if you don't need to type something in—for example, when you choose an operator like “is empty” or “is not empty.”
8. Press Search to start your search.

Adding more choices. You can make an advanced search for records that match more than one criterion. To add a second criterion to your search,

1. While you're working in the Advanced Search dialog, press Add a Choice to add a second criterion to search for.
2. Follow steps 3 through 7 in the previous section to set up the search for the second criterion.
3. Indicate whether to search either for
 - a. records that meet *both* your criteria (click “Search for all of the following”), or
 - b. records that meet *either* criterion (click “Search for any of the following”).
4. If you want to add another choice, press Add a Choice again. You can add up to three more choices, for a total of four choices. If you need to search for more than four criteria, first run a search for the first four of your criteria, then choose either Append Search (for a search for *any* of the criteria) or Subset Search (for a search for *all* your criteria) from the Search menu to add additional criteria (see “[Append searches and subset searches](#)” on page 250).
5. Press Search to start your search.

You can save any set of advanced search criteria to reuse later. For instructions, see “[Saving searches](#)” on page 250.

Some advanced search examples:

- To find all the facilities that store or use chemicals that pose a fire hazard, search Chemicals in Inventory for Fire Hazard (is) “YES.”
- To find all the facilities in Haymarket that store or use chlorine, search Facilities for City contains characters “Haymarket” AND search Chemicals in Inventory for Chemical Name contains characters “chlorine.”

- To find facilities owned by a particular person, search Contacts for Contact Type contains characters “owner” AND Last Name contains characters “<last name of that person>” (and if necessary) AND First Name contains characters “<first name of that person>.”
- To find all the facilities in Haymarket that store chlorine in quantities averaging 500 pounds or more, search Facilities for City contains characters “Haymarket” AND search Chemicals in Inventory for Chemical Name contains characters “chlorine” AND search Chemicals in Inventory for Average Amount on Site is greater than or equal to “500.”

Understanding the information in facility records

Once you’ve found the record for a facility in the Facilities module, you can view the information about it in either of two ways:

- Click tabs on the facility’s record to review most of the information about that facility.
- Select Show Related from the Record menu to access Chemicals in Inventory, Contacts, Incidents, and Screening & Scenarios records for the facility.

Check Table 4-1 for definitions of the various kinds of information shown on each Facilities record and to see how to access each kind of information about a facility.

TABLE 4-1. Items in Facilities records.

Item	Description
Shipper	Check this box if this facility is a shipper of chemicals.
Report Year	The reporting year for which the Tier II information contained in the record was submitted by the facility. See “Sections 311 and 312: community right-to-know requirements” on page 266.
Facility Name	Name of the facility.

TABLE 4-1. Items in Facilities records. (Continued)

Item	Description
Department	Complete if the record is for a particular department of a facility, rather than for the entire facility.
Site	Complete these boxes if the facility has more than one physical site location.
Under the Address tab:	
Street Address	Street address of the facility, as well as the county (or borough or parish), fire district, and country where it's located, and the closest cross street to the facility.
Mailing Address	Mailing address for the facility.
Email	Email address for the facility or a contact person for the facility.
Under the Facility Phones tab:	
Type	Type of phone number (e.g., 24-hour, emergency, office). Pull-down menu in Edit mode.
Phone	Phone number.
Under the Contacts tab:	
Last/First Name	Name(s) of the contact.
Title	Title of the contact.
Organization	Organization represented by the contact (if different from the facility).
Under the Chemical Inventory tab:	
CAS	CAS number of the chemical in the facility's inventory.
Chemical Name	Name of a chemical in the facility's inventory.

TABLE 4-1. Items in Facilities records. (Continued)

Item	Description
RIDS	Click this button to see the response recommendations in the matching Chemical Library record, <i>if</i> either the name for the substance in the Chemical Name box or the CAS number in the CAS box matches the name or CAS number of a chemical in the Chemical Library, or you've linked the name of this chemical to a Chemical Library record. (If the name or CAS number matches multiple chemicals, you'll see a list of matching chemicals to choose from; double-click any name to see this chemical's record.)
Under the Checklist tab:	
Submitted general site plan...	Click this box if the facility has submitted a general site plan to state and local emergency planners (see "Section 301-303: emergency planning" on page 263).
MSDS received with Tier II forms	Click this box if a Material Safety Data Sheet (MSDS) has been received from this facility along with a Tier II form.
Is facility subject to the Clean Air Act (112r)?	Click if facility is subject to CAA 112(r). See "CAA 112(r)" on page 270.
Are dikes or other safeguard measures employed?	Click if passive safeguards such as dikes, berms, enclosures, or drains are used to mitigate accidental hazardous releases in the facility.
Site coordinate abbreviations submitted	Click if a list of site coordinate abbreviations were submitted that correspond to buildings, lots, areas, etc., located throughout the facility.
Date Tier II signed	Date when the Tier II form submitted by this facility was signed. See "Sections 311 and 312: community right-to-know requirements" on page 266.
Date Tier II received	Date when the Tier II form submitted by this facility was received. See "Sections 311 and 312: community right-to-know requirements" on page 266.
Signature	Name of owner/operator or authorized representative who signed the Tier II Chemical Inventory Form.

TABLE 4-1. Items in Facilities records. (Continued)

Item	Description
Checked automatically if items present in CAMEO	Boxes are automatically checked if the facility uses or stores EHS chemicals or if there are records for this facility in the Chemicals in Inventory, Screening & Scenarios, and/or Incidents modules, and/or if the record for this facility is linked to a symbol on a MARPLOT map.
Under the ID Codes tab:	
Number of employees on site	The maximum number of employees on site at any time.
State Fees Total	Total fees (if any) collected by the state for Tier II reporting.
Type	Type of ID Code (e.g., Dun and Bradstreet, SIC, NAICS). Pull-down menu in Edit mode.
ID	ID code number. (In Edit mode, either type a number or choose a code number from the ID menu in the cases of codes such as SIC, for which a specific set of code numbers has been defined.)
Description	Description of the kind of facility represented by this ID code. (In Edit mode, type a description or, if this is a SIC or NAICS code, select a description from the pull-down menu.)
Under the State Fields tab:	
State Fields	On annual Tier II reports submitted by facilities, some states require additional data fields beyond those fields required on all Tier II forms. Once you have entered the 2-letter abbreviation for a state under the Address tab, CAMEO automatically displays all fields required by that state, as well as any optional fields also used on Tier II reports in that state. See “State fields in the Facilities module” on page 110.
Under the Map Data tab:	
Latitude/Longitude	Latitude and longitude of the facility, expressed as decimal values.

TABLE 4-1. Items in Facilities records. (Continued)

Item	Description
Method for determining latitude and longitude	Code and description of the method by which latitude and longitude were measured. Pull-down menu in Edit mode.
Description of location identified by latitude and longitude	Code and description of the location for which latitude and longitude were measured. Pull-down menu in Edit mode.
Record is linked to MARPLOT object	This box is checked if this record is linked to an object on a MARPLOT map (see “Linking map objects to CAMEO records” on page 187).
Under the Site Plan tab:	
Site Plan names	A list of the names of site plan files associated with this facility. Double-click the name of any site plan file to display that site plan in another view. (To add or edit site plans, see “Adding and editing site plans” on page 124 .)
Under the Notes tab:	
Notes	Keep your own notes about this facility here.

Viewing a facility’s map location. A Facilities record in CAMEO can be linked to a symbol that represents the facility’s location on a MARPLOT map (for instructions, see [“Linking map objects to CAMEO records” on page 187](#)). To find out whether a facility record has been linked to a map symbol, click the Map Data tab, and check whether the Record is linked to MARPLOT object box is checked (as at left). If it is checked, the record is linked.

Record is linked to MARPLOT object

If a facility record has been linked to a map symbol, you can view that symbol and map, as follows:

1. Either find and select the facility’s record or open the record in Record view.

2. From the Sharing menu, select MARPLOT, then Show on Map. MARPLOT will come forward, and will display the map, centered on the symbol (a facility symbol on a map is shown in Figure 4-3).

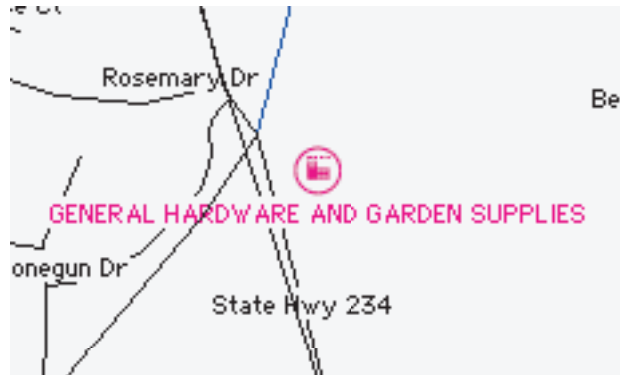


FIGURE 4-3. A facility symbol on a MARPLOT map.

Making and printing reports on facilities

You can make three kinds of print reports from the Facilities module:

- a report on a single facility.
- a report on multiple facilities.
- a set of mailing labels for a given set of facilities.

Making a report for a single facility

To create and print a report for just one facility:

1. Working in the Facilities module, either
 - a. in the list of facilities in List view, click on the name of the facility, or
 - b. double-click on that facility name to open its record in Record view.
2. From the File menu, select Print Report.

3. Check to be sure that the Current Record button is selected.
4. Click the checkbox for each kind of data to be included in the report.
5. Click Print Report. The Print Setup window will be displayed. Adjust any settings you'd like to change. **Note:** If you'd like to see what your report will look like before you print it, click on Preview Report instead. You'll see a view of the printable report and can choose to print or cancel the report from there.
6. Click OK to display the Print window. Adjust any settings you'd like to change, then click OK to print the report.
7. Click Print or Cancel to return to the Report setup window.

Making and printing a report for multiple facilities

To include more than a single facility in your report:

1. First, open the Facilities module and then choose either option below:
 - a. To include all your Facilities records in the report, choose Show All Records from the Record menu to clear any existing found set.
 - b. To include just certain records in the report, run a search, choosing your search criteria so that just the records you want in your report will be included in the found set. For more information about searching for facility records, see [“Accessing information about a facility”](#) on page 111.
2. From the File menu, select Print Report.
3. Check to be sure that the Found Set button is selected.
4. Click the checkbox for each kind of data to be included in the report. (Some of the information about a facility is in records in related modules. For example, if you click the Contacts checkbox, information from records in the Contacts module for this facility will be included in the report.)
5. Click Print Report. The Print Setup window will be displayed. Adjust any settings you'd like to change. **Note:** If you'd like to see what your report will look like before you print it, click on Preview Report instead. You'll see a view of the printable report and can choose to print or cancel the report from there.

6. Click OK to display the Print window. Adjust any settings you'd like to change, then click OK to print the report.
7. Click Print or Cancel to return to the Report setup window.

Making and printing a set of mailing labels

To make a set of 1-inch by 2 5/8-inch mailing labels for some or all of the facilities in your Facilities module:

1. First, follow steps 1 through 3 in the previous section.
2. Click Mailing Labels. You'll see a preview of your mailing labels. From the File menu, select Print Report to print the labels.

Creating and editing facility records

As you add information about a facility to CAMEO, you can create records in the five facility-related modules: Facilities, Chemicals in Inventory, Contacts, Incidents, and Screening & Scenarios. Below are the procedures for adding and editing records in these modules.

Adding a new facility record and data. You add a new facility record in any of three ways:

- creating the new facility record manually, working in the Facilities module and following the instructions below.
- importing a Tier2 Submit data file. When you import a Tier2 Submit data file, CAMEO automatically creates a new record for each facility in that file, along with new Chemicals in Inventory and Contacts records for that facility. For more details and instructions for importing a file, see [“Importing data from Tier2 Submit” on page 205](#).
- transferring data from someone else's copy of CAMEO. See [“Transferring data between different copies of CAMEO” on page 210](#).

To create a new facility record manually:

1. In the Facilities module, from the Record menu, select New Facility. You'll see a new, blank Facilities record, automatically in Edit mode. Fill out the record with the information you have on the facility. See [Table 4-1 on page 115](#) for the definitions of all the data fields in a Facilities record.
 - *To add a contact for the facility*, follow the steps in [“Adding contact information for a facility” on page 138](#).
 - *To add a Chemical in Inventory record*, follow the steps in [“Keeping track of the chemicals in a facility’s inventory” on page 125](#).
 - *To add a Screening & Scenarios record for a chemical in the facility’s inventory*, follow the instructions in [“Using Screening & Scenarios to assess hazards” on page 139](#).
 - *To add a phone number for the facility*, click the Facility Phones tab (click Edit if you aren’t in Edit mode), then click within the uppermost blank table row directly below the “Type” heading. In the menu of phone number types, click to select a type. Type the phone number to the right of the menu in the same table row.
 - *To add an Incidents record for an incident at the facility*, follow the steps in [“Adding and editing Incidents records” on page 175](#).
2. When you’ve added all your information about the facility, click Save Changes.

You can edit the facility record, or any related record in the Contacts, Chemicals in Inventory, Screening & Scenarios, or Incidents modules later, whenever you need to, by either selecting the record in List view or opening it in Record View, then clicking the Edit button in the toolbar. Make the needed changes, then click Save Changes.

Linking a facility record to a map symbol. To link a facility record to a symbol representing that facility on a map in MARPLOT, follow the steps in [“Linking map objects to CAMEO records” on page 187](#).

Adding and editing site plans. If you have created a site plan for a facility in a graphics program, you can link that plan to the record for that facility in the Facilities module, as follows:

1. Draw the site plan in a graphics program of your choice, then save it in a standard graphic file format (for example, JPEG, TIFF, GIF, BMP, or PNG). **Note:** You must save the file with a 3-character extension (for example, .jpg *not* .jpeg).
2. Place the site plan file in the SitePlans folder on your hard drive (site plan files must be kept in this folder). The SitePlans folder is inside of the CAMEO folder.
3. On your Facilities record, click the Site Plan tab, then click the Edit button in the toolbar.
4. Type or paste the file name of the site plan file into the box to the left of the Add button. Include the 3-character file name extension.
5. Press Add. You'll see the name of the plan in the list of site plans under the Site Plan tab.
6. Click Save Changes. Now you can open and view the site plan whenever you need to, by double-clicking its name.

To edit an existing site plan, reopen it in the graphics program you used to create it, make the needed changes, then save the file under the same name you used before.

To delete a site plan from a Facilities record,

1. Open the record in Record view, click the Site Plan tab, click the Edit button in the toolbar,
2. Click on the name of that site plan, click Remove, then click Save Changes.
3. Delete the site plan file from your SitePlans folder if you have no more need of the file. Deleting a site plan from a record does not delete the site plan file itself; it remains within the SitePlans folder inside the CAMEO folder unless you remove it.

Tip: You can keep photos of facilities in the SitePlans folder, as well as site plan files. Photos should be in JPEG format.

Keeping track of the chemicals in a facility's inventory



Keep records describing the hazardous chemicals stored or used at a given facility in the Chemicals in Inventory module. A Chemicals in Inventory record can include descriptions of each chemical's physical state, storage conditions and locations, and quantities routinely on site. You can create records both for single substances or for mixtures. (You also can create Chemicals in Inventory records for routes along which hazardous materials are transported, see [“Keeping track of chemicals transported along a route” on page 172.](#))

When you follow the steps in this section to add a chemical to a facility's inventory record, you're actually adding a new record to the Chemicals in Inventory module; information about that chemical also will be displayed in the Facilities record.

Chemicals in Inventory and Tier II data. You can use the Chemicals in Inventory module to maintain information about hazardous substances in facility inventories that was submitted to you on Tier II forms. See [“Tier II information and CAMEO” on page 108.](#)

Manually creating a Chemicals in Inventory record

Whenever you create a new Chemicals in Inventory record for a chemical, you need to decide whether to use a name or synonym² for the chemical that matches a name or synonym shown on either a record in the Chemical Library or on one or more other Chemicals in Inventory records.

-
2. CAMEO distinguishes between the name for a chemical shown in the Chemical Name box—which is unique for each substance in the Chemical Library—and the synonyms for that name listed under the Synonyms tab on each Chemical Library record. Many synonyms are shared by multiple chemicals.

To check for a match, you can use the **Look Up Chemical** button. In the Name box of your new record, you type either a whole name or synonym, or as many of the first few characters of the name or synonym as you're sure of (more details are explained in [“Guidelines for looking up chemicals” on page 129](#)). You then click Look Up Chemical, and you see three lists, each under a separate tab ([Figure 4-4 on page 127](#)). One shows the names and synonyms on Chemical Library records that match what you've typed. The second list shows matching names and synonyms in your existing Chemicals in Inventory records. The third list shows matching mixture components in your existing Chemicals in Inventory records. If you select any of the names or synonyms in the lists, it will replace what you've typed, and you can be sure that you have a match. (You can click Cancel to avoid replacing what you've typed.) You can use the Look Up Component button the same way when you add mixture components to your Chemicals in Inventory records.

Generally, we recommend that you use Look Up Chemical or Look Up Component to ensure that the names or synonyms you use match Chemical Library records. For one thing, using matching names or synonyms ensures that you'll be able to quickly access the response recommendations for inventory chemicals when you need to (via the RIDS or View RIDS buttons, discussed below).

However, you might not always want to use matching names or synonyms, and CAMEO doesn't require you to. If you simply don't click Look Up Chemical (or Look Up Component in the case of a mixture component), CAMEO will accept whatever name you type on your new record. You might want to avoid looking up a chemical if you're creating a Chemicals in Inventory record to contain information from a Tier II form. If you want the name shown on that record to match the name shown on the Tier II form, even if the name on the form is misspelled or is an unusual synonym or trade name that doesn't appear in the Chemical Library, don't click Look Up Chemical.

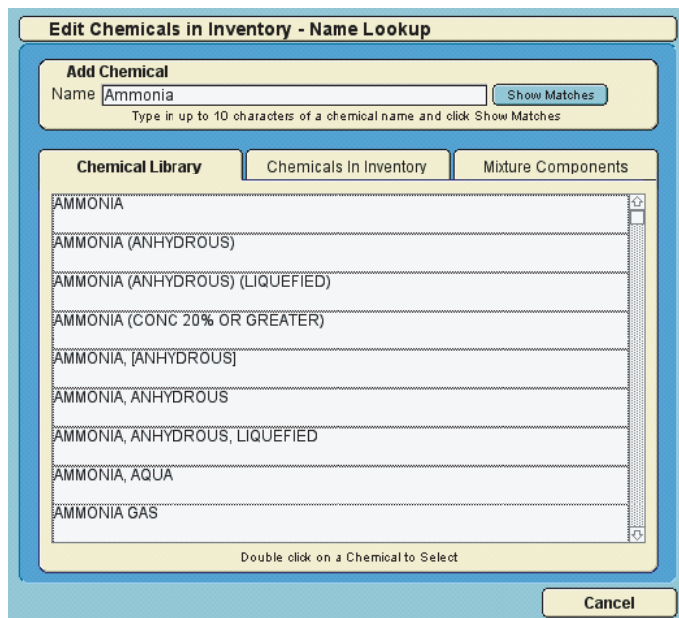


FIGURE 4-4. List of chemical names from Chemical Library records that match “Ammonia.” (Clicking the Chemicals in Inventory or Mixture Components tab displays matching names shown in Chemicals in Inventory records.)

To manually create a new Chemicals in Inventory record for a substance in a facility's inventory:

1. In the Facilities module, open the record for the facility in Record view.
2. Click the Chemical Inventory tab, then click the Edit button.
3. Click New Chemical in Inventory.
4. *If you're adding a single substance (as opposed to a mixture), then*
 - a. Type either the chemical's name or its CAS number in the Chemical Name or CAS # box. When you type a CAS number, you must type the entire CAS number (either with or without hyphens). **Note:** If you plan to check for a name or synonym match, then type either the whole name (or synonym) or, if you aren't sure of the exact spelling, as many of the first few letters of the name as you're sure of. Check [“Guidelines for looking up chemicals”](#) on page 129 for tips.

- b. If you want to check for a name match, click Look Up Chemical. Otherwise, skip down to step c. You'll see a list of all the names and name synonyms in the Chemical Library that match what you've typed ([Figure 4-4 on page 127](#)). For example, if you type "ammonia," the list will include "ammonia," "ammonia, anhydrous," "ammonia, aqua," and all other names and synonyms that start with the word "ammonia." Under separate tabs, you'll also see all matching names for substances and mixture components in other Chemicals in Inventory records. Double-click the name that matches your chemical. The name that you select replaces any text you typed in the Chemical Name box (the CAS # box is unaffected). Click Cancel to avoid replacing what you've typed in the box
 - c. Click the EHS Substance checkbox if this is an Extremely Hazardous Substance (EHS) and the checkbox is not checked. (The checkbox is automatically checked only if you type the CAS number of an EHS, click Look Up Chemical, and then select the chemical's name.)
- 5. If you're adding *a mixture of two or more substances*, then
 - a. Type the name of the mixture in the Chemical Name box.
 - b. Click the Components tab.
 - c. In the topmost blank row under the Components tab, type either the first component's name or its entire CAS number. If you plan to check for a name or synonym match, then type either the whole name (or synonym) or, if you aren't sure of the exact spelling, as many of the first few letters of the name as you're sure of. Check "[Guidelines for looking up chemicals](#)" on page 129) for tips on entering names.
 - d. If you want to check for a name or synonym match, click Look Up Component. Otherwise, skip this step. You'll see a list of the names and name synonyms in the Chemical Library that match what you've typed. Under separate tabs, you'll also see all matching names of substances and mixture components in other Chemicals in Inventory records. Double-click the name that is the best match for your chemical. The name that you select replaces any text you typed in the Chemical Name box (the CAS # box is unaffected). Click Cancel to avoid replacing what you've typed in the box.

- e. Click the EHS checkbox if this is an Extremely Hazardous Substance (EHS) and this checkbox is not checked. The checkbox is automatically checked only if you type the CAS number of an EHS, click Look Up Component, and then select the chemical's name.
 - f. In the % box, type either the percentage by weight or by volume of this component in the mixture, then, from the percentage units menu (under the "Wt/Vol" heading), choose either Weight or Volume.
 - g. Repeat steps c through g for each component of the mixture.
6. Fill out the information about the storage conditions and locations of the substance or mixture. Refer to [Table 4-3 on page 134](#) for the definitions of each of the fields on a Chemicals in Inventory record.
 7. Click Save Changes to save your changes to the Chemicals in Inventory record. You'll be returned to the Facilities record that you started from.
 8. Click Save Changes again, to save the changes to the Facilities record.

Once you've created the record, if you'd like to create a link from the name and/or CAS number on the new Chemicals in Inventory record to a specific Chemical Library record, follow the directions in ["Using the Adjust Link button" on page 130](#).

Guidelines for looking up chemicals. When you click Look Up Chemical or Look Up Component, CAMEO checks through the Chemical Library records to find chemical names matching the text you enter. It also checks through the names in all your existing Chemicals in Inventory records. However,

- Only the first 10 characters are considered, both in the text you enter and in the chemical names. Entering "PROPYLTRICHLOROSILANE" will provide the same matches as entering "PROPYLTRIC".
- The lookup attempts to find only chemical names that begin with the text you enter. To find "BORIC ACID" you cannot enter "ACID"; instead, enter "BORIC" or "BORIC ACID".

- If your search text contains characters like “,” or “)” or spaces, names may be displayed that do not exactly match in the position of those characters. Suppose you want to find “1,2,4-TRIMETHYLBENZENE”. If you enter “1,2,4”, the list will include “12418 RED”. To avoid this, type a longer string of characters like “1,2,4-TRI”.
- On the other hand, if your lookup fails to locate the desired name, try entering fewer characters. Entering “1-BETA-D2” results in no matches, but “1-BETA-D” results in several, including “1-BETA-D-2'-DEOXYRIBOFURANOSYL-5-FLUOROURACIL”, possibly the desired name.
- If all else fails, click Cancel to leave the Chemicals In Inventory editing window, open the Chemical Library, and search for the chemical name, specifying “Contains characters.” Once you find the desired name (in the Synonyms tab), note precisely its first 10 characters. Return to Chemicals In Inventory, click Edit, enter those 10 characters, then click Look Up Chemical.

Using the Adjust Link button. This button is visible on a Chemicals in Inventory record once you have saved your editing changes. You can use it to link a particular Chemical Library record with either (a) a chemical name or (b) a combination of a chemical name and a CAS number that you've typed in one or more of your Chemicals in Inventory records. You would use this button only when you're sure that a particular Chemical Library record matches the name or combination of name-and-CAS number on a Chemicals in Inventory record. Generally, a link is useful in either of two circumstances:

- You want to be sure that you'll be able to readily access the response recommendations and other information about a chemical from its Chemicals in Inventory record (you would do this by clicking the View RIDS button on that record) or from a Facilities or Routes record (you would click the RIDS button next to the name of the chemical in the inventory).

- You want to use the Screening & Scenarios module to predict a hazard zone for the chemical shown on the Chemicals in Inventory record (see [“Using Screening & Scenarios to assess hazards” on page 139](#)). You would *not* need to make a link if the name shown on the Chemicals in Inventory record matches just one Chemical Library record. That name must match either (a) the name shown in the Chemical Name box on *only one* Chemical Library record, or (b) an official EHS name (in a few cases, this name differs from the name in the Chemical Name box). To find out the number of matching Chemical Library records, click the View RIDS button on the Chemicals in Inventory record of concern. If there are multiple matching Chemical Library records, then you *would* need to make a link. (Also, the Chemical Library record must be for an EHS. If the EPCRA EHS Chemical box is checked, it is.)

Important: Because you make a link not between two records, but between either a name or name/CAS number and a record, the links you make can affect multiple Chemicals in Inventory records. For example, if you link “Nitric Acid” on one of your Chemicals in Inventory records to the Chemical Library record for “NITRIC ACID, FUMING,” then any other Chemicals in Inventory records for “Nitric Acid” also will automatically be linked to the Chemical Library record for “NITRIC ACID, FUMING”.

Inadvertently linking to the wrong Chemical Library record associates the wrong response recommendations with a chemical in an inventory. To avoid inadvertent links, when you plan to link names to Chemical Library records, choose relatively specific chemical names, such as “Nitric Acid, 40%,” rather than more general names like “Nitric Acid.”³ We also recommend not making a link if you aren’t sure that a particular Chemical Library record matches the chemical name or name and CAS number combination that you’ve included on your Chemicals in Inventory record(s).

3. You also could choose to use a unique name, such as “Nitric Acid in Process 7,” on the linked Chemicals in Inventory record to ensure that the link will not affect other Chemicals in Inventory records for nitric acid.

An alternative to using Adjust Link is to use Look Up Chemical to be sure that you've chosen a correctly spelled name for the chemical. Once you've done that, clicking View RIDS will display a list of all the Chemical Library records that match that chemical name.

Before you make a link, click View RIDS on the Chemicals in Inventory record (or the RIDS button next to the chemical's name under the Chemical Inventory tab on the related Facilities record). There is no need to make a link if you're satisfied with the results you obtain. (Review [Table 4-2 on page 133](#) for more help in making this determination.)

Otherwise, to make a link,

1. Open the Chemicals in Inventory record in Record mode.
2. Click Adjust Link.
3. When asked whether you wish to create a link, click Yes. The Chemical Library module will be displayed.
4. Find and select the Chemical Library record you want to link to. For instructions for searching the Chemical Library for a record, see [“Searching for a Chemical Record” on page 80](#).
5. From the Link menu, select Link this Record.

Tip: You can first search the Chemical Library to find the record you want to link to, and then follow the steps above to create the link.

To remove a link that you made in error, click Adjust Link, then click Remove Link.

Whether you want to use Look Up Chemical and/or Adjust Link in a particular situation depends on your project at hand and your goals for it. [Table 4-2 on page 133](#) is a quick-reference decision aid showing when you might want to use one or the other of the buttons, and when you might not.

TABLE 4-2. Decision table for looking up names and linking to records.

If you need to...	Then do this:
Make sure a chemical name matches a name or synonym in the Chemical Library.	Type the name, then click Look Up Chemical to check for names and synonyms in the Chemical Library that match what you've typed.
Retain a name on a Chemicals in Inventory record that <i>doesn't</i> match CAMEO's name for the chemical (for example, a name on a submitted Tier II form).	Type the name and <i>don't</i> click Look Up Chemical. (You can use Adjust Link if you know which Chemical Library record matches the name you're using.)
Use Screening & Scenarios to plot hazard zones for a chemical.	Use Adjust Link unless the name on the Chemicals in Inventory record matches just one Chemical Library record (either the name in the Chemical Name box or the official EHS name). The Chemical Library record must be for an EHS (if the EPCRA EHS Chemical box is checked, it is).
Ensure you can quickly access the right response recommendations during an incident response.	Use Adjust Link unless the name and/or CAS number on the Chemicals in Inventory record matches only the Chemical Library record(s) that correctly describe the chemical (in some cases—for example, a chemical sometimes stored at different solution strengths—you might want more than one Chemical Library record to match the Chemicals in Inventory record).

Items in Chemicals in Inventory record

Table 4-3 explains the items shown on Chemicals in Inventory record.

TABLE 4-3. Items on a Chemicals in Inventory record.

Item	Description
Facility/Route	Filled in automatically with either the name of the facility that maintains this substance or the name of the transportation route along which this substance is transported.
Report Year	Filled in automatically with the report year shown on the related Facilities, if this is an inventory record for a facility. Should be the year for which this chemical inventory information applies (typically, the reporting year in which the Tier II form or data file was submitted). Not editable.
Dept.	Filled in automatically with the department or division of a facility.
City	Filled in automatically with the city where the facility or route is located.
State	Filled in automatically with the state where the facility or route is located.
CAS #	Chemical Abstracts Service number for the substance.
View RIDS	Click this button to see the RIDS record for the chemical shown on this record. If you do not see a RIDS record, see “Manually creating a Chemicals in Inventory record” on page 125.
Chemical Name	The name of the chemical.
Adjust Link	Click this button to make or change the link between a Chemicals in Inventory record and a Chemical Library record. See “Using the Adjust Link button” on page 130.
In Inventory	Automatically checked if this is an inventory record for a facility.
In Transit	Automatically checked if this record is for a chemical associated with a transportation route.

TABLE 4-3. Items on a Chemicals in Inventory record. (Continued)

Item	Description
EHS Substance	Click this checkbox if the chemical (or one of the components of the mixture) is one of the Extremely Hazardous Substances identified by EPA.
Trade Secret	Click this checkbox if the formula for this substance is a trade secret.
MSDS	MSDS number for chemical (provided by manufacturer).
Under the Location tab:	
Amount	The amount of the substance stored or transported.
Unit	Amount units (mass or volume). Pull-down menu in Edit view.
Type	The letter code for type of storage (click Type to see a list of type code definitions). Pull-down menu in Edit view.
Press	The number code for storage pressure (click Press to see a list of pressure code definitions). Pull-down menu in Edit view.
Temp	The number code for storage temperature (click Temp to see a list of temperature code definitions). Pull-down menu in Edit view.
Location	Short description of the location of the stored (or transported) substance.
Under the Physical State & Quantity tab:	
Pure/Mixture	Click Pure if the substance is in pure form; Mixture if it is a mixture.
Solid/Liquid/Gas	Click these checkboxes to indicate the physical state(s) of the stored or transported substance.
Fire	Click this checkbox if the substance is a fire hazard (e.g., flammables, combustible liquids, and oxidizers).

TABLE 4-3. Items on a Chemicals in Inventory record. (Continued)

Item	Description
Pressure	Click this checkbox if the substance is a “sudden release of pressure” hazard (e.g., explosives and compressed gases).
Reactive	Click this checkbox if the substance is a reactive hazard (e.g., water reactives, unstable reactives, and organic peroxides).
Acute/Chronic	Click one or both of these checkboxes if the substance poses either acute (immediate) or chronic (delayed) health risks. Examples of acute health hazards include toxics, corrosives, irritants, and sensitizers; chronic health hazards include carcinogens.
Max Daily Amount/Max Code	Type the maximum amount of the substance stored or transported, then select the appropriate code for maximum amount. This is the same as the reporting range code used on Tier II forms.
Average Daily Amount/Ave Code	Type the average amount of the substance stored or transported, then select the appropriate code for average amount. This is the same as the reporting range code used on Tier II forms.
Max amount in largest container	Type the maximum amount (in pounds) of the substance stored or transported in a single container. or in interconnected vessels.
<p>Under the Components tab: Each component of a mixture is described in a table row under this tab. All rows should be blank in the case of a pure substance.</p>	
EHS/CAS/Component/%/ Wt/Vol	Working from left to right in Edit mode, filling out one row for each mixture component: click the EHS checkbox if the component is an Extremely Hazardous Substance; type the component’s CAS number; type the name of the component (click Look Up Component to check for a name match with a Chemical Library record or other Chemicals in Inventory record); type the percentage by weight or volume of the component within the mixture, then select either Weight or Volume as percentage units.

TABLE 4-3. Items on a Chemicals in Inventory record. (Continued)

Item	Description
Under the Dates tab:	
Report year from January 1 to December 31	Filled in automatically with the report year shown on the related Facilities record, if this is an inventory record for a facility. Should be the year for which this chemical inventory information applies (typically, the reporting year in which the Tier II form or data file was submitted). Not editable.
Chemical identical to previous year	Click this checkbox if the information about this substance is identical to the information submitted during the previous year.
Days On Site	The number of days during the reporting year that the substance was found at this site.
Date Tier II Signed	Date when the Tier II form was signed by the facility owner or operator. Filled in automatically with the date shown on the related Facilities record; not editable.
Date Tier II Received	Date when the Tier II form was received. Filled in automatically with the date shown on the related Facilities record; not editable.
Under the State Fields tab:	
State Fields	On annual reports submitted by facilities, some states require additional data fields beyond those fields required on all Tier II forms. CAMEO automatically displays all state-required fields under this tab, along with any optional fields used on annual facility reports in that state.
Under the Notes tab:	
Notes	Keep your own notes about this substance here.

Adding contact information for a facility



You can create records in the Contacts module that contain contact information for people representing a facility or other kind of organization or business. When you add a contact for a facility, you're actually adding a new record to the Contacts module. Information about that contact will also be displayed in the facility's record in the Facilities module.

To add a record for a contact for a facility (to add other kinds of contacts to the Contacts module, see [“Contacts” on page 165](#)):

1. Find and open the Facilities record for the facility for which you want to add a contact record.
2. Click the Contacts tab, then click the Edit button in the toolbar.
3. Click Add Contact. A list of all contacts in the Contacts module appears.
4. Either
 - a. Click on the name of a contact already in the list, then click Select. You would do this if a contact for the facility is already in the list, because you already added a Contacts record for that person (perhaps because he or she is a contact for more than one facility). **Note:** You can associate the same contact record with more than one facility by repeating steps 1 through 4 for each of the facilities for which this person is a contact.
 - b. Click Add New to add a new Contacts record. You would do this if the contact is not already in the list (because you haven't previously created a Contacts record for that person). You'll see a new, blank contacts record. Fill in the information about the new contact (refer to [Table 5-2 on page 167](#) for the definitions of each of the fields on a Contacts record), then click Save Changes. **Note:** A new Contacts record that you create by clicking New Contact from a Facilities record will be associated with that Facilities record, even if you don't type the name of the facility, or other identifying information about the facility, on the contact record.

Note: New Contacts records for facility contacts will automatically be added to your copy of CAMEO whenever you import a Tier2 Submit data file that contains contact information. See [“Importing data from Tier2 Submit” on page 205](#).

Adding records for incidents at a facility



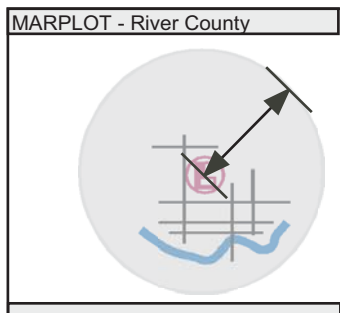
Use the Incidents module to record incidents, such as accidental spills and releases, at given facilities (or routes). To create a record in the Incidents module that's associated with a particular facility, follow the steps in [“Adding and editing Incidents records” on page 175](#).

Using Screening & Scenarios to assess hazards



You can use CAMEO's Screening & Scenarios module to assess the hazards to your community from accidental releases of locally stored hazardous chemicals. Such an assessment is called a **hazards analysis**. You can use the results of your hazards analysis to prepare emergency response plans for your community. The basic procedures for hazards analysis are described in *Technical Guidance for Hazards Analysis: Emergency Planning for Extremely Hazardous Substances* (1987). This guidebook was prepared by the U.S. Environmental Protection Agency (EPA), the Federal Emergency Management Agency (FEMA), and the U.S. Department of Transportation (DOT). The procedures described in the *Technical Guidance* were developed to help community planners, especially members of Local Emergency Planning Committees (LEPCs), meet the provisions of the Emergency Planning and Community Right-to-Know Act of 1986 (see [“EPCRA” on page 261](#)).

The *Technical Guidance* describes calculations you can make to estimate the size of the area around a chemical storage facility or along a transportation route that could be affected by an accidental release of a hazardous chemical. This area is called the **threat zone**. It may also be called the **vulnerable zone** or the **screening zone**, to reflect the particular type of hazards analysis you perform (as discussed below).



Screening & Scenarios is essentially a calculator you use to estimate the **radius** of a threat zone (a threat zone radius is shown at left). To do this, you enter a set of simple conditions to describe an accidental chemical release at a chemical facility or along a transportation route. Screening & Scenarios then estimates the radius of the threat zone around the facility or along the route, using the *Technical Guidance* calculations. If you have a MARPLOT map of your area, you also can plot threat zones on your map.

You can estimate a threat zone radius only for Extremely Hazardous Substances (EHSs), chemicals that have been identified by the U.S. Environmental Protection Agency as acute inhalation toxic threats and that have been listed under EPCRA. There are 356 EHSs. To check whether a chemical of concern is an EHS, find its record in the Chemical Library, then click the Regulatory Information tab. If the EHS checkbox is checked, the chemical is an EHS (Figure 4-5 on page 141).⁴

Important: Never use Screening & Scenarios as an emergency response tool or for meeting the requirements of Section 112(r) of the 1990 Clean Air Act. See “[What are the differences between Screening & Scenarios, ALOHA, and RMP endpoint distances?](#)” on page 155.

Choosing between *screening* and *scenarios* calculations

When you use Screening & Scenarios, you need to choose whether to estimate the threat zone using either:

1. EPA’s “credible worst case assumptions” to make *screening* calculations, or
2. atmospheric and chemical information that you judge to be more typical of the region and facility, in order to make *scenarios* calculations.

4. There are more than 356 records for EHSs in the Chemical Library, because there are multiple records for some EHSs that occur in different solution strengths or mixtures.

About screening calculations

When you use EPA's worst-case assumptions (shown in [Table 4-4 on page 142](#)), you're performing a screening, and the threat zone is typically called a screening zone. In a screening, you estimate screening zones for all EHSs that are either (a) stored above their specified Threshold Planning Quantity (TPQ) at chemical facilities within a community, or (b) transported along a local route in quantities greater than the TPQ.

To see the TPQ for any EHS, find the Chemical Library record for that EHS, click the Regulatory Information tab, then check the EHS Threshold Planning Quantity box (Figure 4-5). If a facility in your area stores more than the TPQ quantity of an EHS, it should be included in your screening project.

The screenshot shows a web interface for a Chemical Library record. At the top, the title is "Chemical Library". Below it, the "Chemical Name" field contains "ACROLEIN, INHIBITED". The interface is divided into two main sections: "Chemical Identification Information" (green background) and "Response Information Data Sheets" (purple background). Under "Chemical Identification Information", there are five tabs: "Chemical Identification", "Synonyms", "NFPA Codes", "Regulatory Information" (which is selected), and "Screening and Scenarios". The "Regulatory Information" tab displays the following data:

Names:	
2-PROPENAL	
ACROLEIN	

CAA Section 112(i) chemical: <input checked="" type="checkbox"/>	CAA Threshold Quantity: 5000 pounds
CERCLA chemical: <input checked="" type="checkbox"/>	CERCLA Reportable Quantity: 1 pounds
EPCRA EHS chemical: <input checked="" type="checkbox"/>	EHS Threshold Planning Quantity: 500 pounds
EPCRA Section 313 chemical: <input checked="" type="checkbox"/>	RCRA chemical code: P003

FIGURE 4-5. This Chemical Library record shows that acrolein is an Extremely Hazardous Substance (EHS), with a Threshold Planning Quantity (TPQ) of 500 pounds.

TABLE 4-4. Worst-case assumptions used for screening zone estimation.

Atmospheric stability class = F	Indicates very stable, nighttime atmospheric conditions. See Table 4-6 on page 154 .
Wind speed = 3.35 miles per hour	On average, a chemical cloud can travel the farthest downwind at this relatively low wind speed.
Wind direction = any direction	Wind direction can't be predicted in advance. Because an escaping chemical cloud could potentially travel in any direction away from its point of release, the screening zone forms a circle around the potential release point. The zone therefore does not represent the area that could be affected during a release; the part of the zone that is affected would depend on the wind direction during the release.
Ground roughness = Open Country	The chemical cloud travels across flat, rural terrain that presents no obstacles to air movement.
Level of concern = the value listed in the Technical Guidance for the EHS of concern	This is a conservative estimate of the chemical concentration that might cause adverse health effects. A screening zone encompasses the area around the potential release point within which ground-level concentrations of pollutant could reach or exceed your level of concern.
Amount released = maximum quantity in vessel or interconnected vessels	The maximum quantity of a chemical that can be contained in a single storage vessel or in a group of interconnected vessels within the facility under examination. Release is from ground level.
Release duration = 10 minutes, or depends on evaporation or volatilization rate	Depends on the chemical's state. Solids in powder or solution form and gases are expected to be released within 10 minutes. The duration of a liquid or molten solid's release depends on its rate of evaporation or volatilization.

Once you have estimated your screening zones, the *Technical Guidance* describes in detail the remaining steps of a hazards analysis. Briefly, those steps are:

1. Use your estimated screening zones as an aid in identifying the populations and critical facilities that might be placed at risk by a worst-case, accidental release. This gives you a way to estimate the greatest possible severity of the consequences of a release.
2. Estimate the likelihood of a given release.
3. Use your estimates of the greatest possible severity of release consequences and the likelihood that the release will occur to estimate the overall risk associated with a particular chemical for each facility and transportation route.
4. Differentiate between high-priority and low-priority risks to your community. You might choose to focus first on high-priority risks when preparing emergency plans.

About scenarios calculations

If you change at least one of the screening assumptions, you are making a *scenario calculation* instead of a screening calculation. You might choose to make a scenario calculation if you thought that one or more of the worst-case assumptions for weather, release amount, level of concern, and/or other circumstances of a release did not represent likely release circumstances for a facility. For example, you might change wind speed and stability class to fit historical weather data. You can then find the size of a threat zone under more typical weather conditions rather than the worst-case conditions. When you make scenarios instead of screening calculations, the threat zone is typically termed the *vulnerable zone* to distinguish it from a screening zone.

You can use both the screening and scenarios features in Screening & Scenarios. Use the screening assumptions to compare the relative risks to your community from the worst possible releases of all chemicals of concern, both at fixed facilities and along transportation routes. Once you have completed your initial screening, you can construct scenarios to explore how changing your release assumptions might influence the size of the vulnerable zone. (You also could use ALOHA to further investigate potential release scenarios; see [“Using ALOHA with MARPLOT and CAMEO” on page 192.](#))

Before estimating a threat zone

Before you begin, you need to make the following preparations:

1. Create a Facilities or Routes record, and link it to a map object, following the instructions in either [“Linking symbol objects to records” on page 187](#) for a Facilities record, or [“Linking transportation routes to CAMEO records” on page 189](#) for a Routes record. Any Screening & Scenarios records you create for the facility or route will become indirectly linked to the same map object, and you’ll then be able to plot a threat zone for that facility or route on the map.
2. Create a Chemicals in Inventory record for a chemical that’s either maintained at the facility (follow the steps in [“Keeping track of the chemicals in a facility’s inventory” on page 125](#)), or routinely transported along the route (follow the steps in [“Keeping track of chemicals transported along a route” on page 172](#)). To run a screening calculation, you must enter a value for Max amount in largest container.
3. Check that the chemical is an EHS, if you don’t already know: open its Chemical Library record in Record view, click the Regulatory Information tab, then check whether the EPCRA EHS Chemical box is checked. If it is checked, the chemical is an EHS.

Estimating a screening zone

Once you've made the preparations shown above, follow the steps below to estimate the radius of a screening zone for a particular EHS at a facility or along a route:

1. Select the Chemicals in Inventory record for the chemical for which you want to estimate a threat zone, then select New Screening/Scenario Record from the Record menu. If you are then told that
 - the record is not linked to a Chemical Library record, then link the chemical name on the Chemicals in Inventory record to the Chemical Library record for that chemical (see [“Manually creating a Chemicals in Inventory record” on page 125](#)).
 - you first must enter the maximum amount in the largest container, then click OK, open the Chemicals in Inventory record in Record view, click the Edit toolbar button, click the Physical State & Quantity tab, type an amount in pounds in the Max amount in largest container box, then click Save Changes.
 - there is not enough information on the Chemical Library record, then the chemical is not an EHS, and you can't create a Screening & Scenarios record for it.
2. Click Screening when asked whether you want to create a screening record or alternative scenario. Information about the stored chemical and the facility or route where it's stored or transported will be copied onto a new Screening & Scenarios record, which will be displayed in Edit mode.⁵ **Note:** For each Chemicals in Inventory record, you can create just one screening record (which you can edit whenever you like). You can create as many alternate scenario records as you like.
3. Type a name for the screening in the Screening Name box.

-
5. On the new record, the Amount Released box will be filled in with the maximum amount in largest container shown on the Chemicals in Inventory record. Concentration is set to 100 percent, and physical state at 68°F and LOC (level of concern) are filled in with the information shown under the Screening & Scenarios tab on the Chemical Library record for the substance. Weather conditions and release duration are set to the values shown in [Table 4-4 on page 142](#).

4. The chemical's physical state at 68°F—solid, liquid, or gas—is automatically filled in. If the chemical is a liquid stored at or above its boiling point, select Near Boiling from menu to the right of the Liquid button. If it is a liquid stored below its boiling point, select Ambient from the menu. If it is a solid, select Solution if it's in solution, Powder if it's in powdered form, or Molten if it's in molten form. Your choice influences how CAMEO makes threat zone calculations (see the *Technical Guidance* for more details about how the calculations are made).
5. Type a diked area in square feet if the chemical's container is surrounded by a dike.
6. Click Estimate Threat Zone Radius. The radius estimate will be displayed. You also can fill in the risks and consequences of the release. Consult the *Technical Guidance* for help with this step.
7. Once you have completed your radius calculation, click Save Changes to save this record. For the directions for plotting the screening zone on a map, see [“Plotting a threat zone on a map” on page 151](#).

Making scenarios calculations

To make scenarios calculations rather than screening calculations,

1. Select the Chemicals in Inventory record for the chemical for which you want to estimate a threat zone, then select New Screening/Scenario Record from the Record menu.
2. Click Scenario if you are asked whether you want to create a screening record or alternative scenario. If you have already created a Screening record for this chemical, a new Scenario record will automatically be created for you, because there can be only a single Screening record for each chemical in an inventory.
3. Type a name for the screening in the Screening Name box.

4. The chemical's physical state at 68°F—solid, liquid, or gas—is automatically filled in. If the chemical is a liquid stored at or above its boiling point, select Near Boiling from the menu to the right of the Liquid button. If it is a liquid stored below its boiling point, select Ambient from the menu. If it is a solid, select Solution if it's in solution, Powder if it's in powdered form, or Molten if it's in molten form. Your choice influences how CAMEO makes threat zone calculations (see the *Technical Guidance* for more details about how the calculations are made).
5. Adjust any storage and release conditions that need to be changed. Refer to [Table 4-5 on page 148](#) to make your choices.
6. Click Estimate Threat Zone Radius. The radius estimate will be displayed. You also can fill in the risks and consequences of the release. Consult the *Technical Guidance* for help with this step.
7. Once you've completed your radius calculation, click Save Changes to save this record. For the directions for plotting the threat zone on a map, see ["Plotting a threat zone on a map" on page 151](#).

TABLE 4-5. Items in Screening & Scenarios records^a.

Item	Description
Facility/Route Name	The name of the facility or route where this chemical is maintained or transported. Not editable.
Chemical	Name of the chemical. Not editable.
CAS	Chemical Abstracts Service (CAS) number for this chemical. Not editable.
Screening Name (Scenario Name)	Name of this screening or scenario.
In Inventory	Checked automatically if this Screening & Scenarios record is associated with a facility. Not editable.
In Transit	Checked automatically if this if this Screening & Scenarios record is associated with a transportation route. Not editable.
Shipper	Checked automatically if the facility that maintains this chemical is a shipper (as indicated on its Facilities record). Not editable.
Under the Screening (Scenario) Description tab:	
Amount Released	Amount of chemical released, in pounds. For screening calculations, this is the maximum amount stored in one vessel or in connected vessels; you can use other values for alternative scenarios.
Concentration	Concentration of the chemical, in weight percent.
Release Duration	Release duration in minutes.
Physical State	Chemical's physical state (solid, liquid, or gas) at 68°F. Automatically filled in when you select a chemical from the list of EHS chemicals.

TABLE 4-5. Items in Screening & Scenarios records^a. (Continued)

Item	Description
If stored in a container with a dike, enter surface area within dike:	If the container in which the chemical is stored is diked, type the area within the dike, in square feet.
Atmospheric Concentration Level of Concern	Or LOC. Atmospheric concentration of the chemical on which to base threat zone calculations, in grams per cubic meter. The value listed in the <i>Technical Guidance</i> is automatically filled in; you can type a different value if you're running a scenario rather than a screening.
LOC Description	Either "Greenbook LOC," if this is the <i>Technical Guidance</i> value, or "Other," if this is a different LOC.
Wind Speed	Wind speed in miles per hour (mph). For screening calculations, wind speed is set to 3.4 miles per hour. Your value for wind speed must be appropriate for the stability class that you selected (Table 4-6 on page 154).
Wind From	Direction from which the wind is blowing, in degrees true north. This field remains blank when you perform screening calculations, because you can't predict wind direction in advance of a release.
Ground Roughness	A measure of the size of the obstacles on the ground that a dispersing chemical cloud must pass over. For screening calculations, ground roughness is set to Open Country. For a scenario, choose either Open Country (relatively smaller and fewer obstacles) or Urban or Forest (relatively more and larger obstacles) from this menu.
Stability Class	Atmospheric stability category that depends on wind speed and cloud cover. For screening calculations, stability class is set to F. For scenarios, review Table 4-6 on page 154 to choose a class.

TABLE 4-5. Items in Screening & Scenarios records^a. (Continued)

Item	Description
Risk/Consequences/Overall Risk	Using the information you entered for the Screening/Scenarios calculations, you can rank the likelihood, consequences, and overall risk of a release of a chemical. Select “high,” “medium,” or “low” from each menu. (Refer to the <i>Technical Guidance</i> for more on risk assessment.)
Threat Zone Radius	Distance from the release point beyond which the predicted concentration of the airborne pollutant is expected to be below the LOC. At locations closer to the release point and directly downwind, concentrations are predicted to exceed the LOC. Because wind direction is not taken into account when this distance is calculated, a threat zone around a release point always forms a circle.
Estimate Threat Zone Radius	Visible only in Edit mode. Click this button to estimate the threat zone radius.
Under the Notes tab:	
Notes	Keep your own notes about a screening or scenario here.

a. You’ll notice that there is no data field for ambient temperature. For all Screening & Scenarios calculations, temperature is assumed to be 68°F (20°C).

Plotting a threat zone on a map

To plot a threat zone on a MARPLOT map from a Screening & Scenarios record, click **Show on Map** (or, from the Sharing menu, select MARPLOT, then Show on Map). If you've linked the facility or route record for which you're making this threat zone to a map object, MARPLOT will start up and will display the map, with the threat zone on it.

In the case of a facility, the threat zone will appear as a shaded circle around the facility, with a radius equal to the threat zone radius shown on the Screening & Scenarios record (Figure 4-6). A threat zone for a scenario also includes a small oval area, representing the area that could potentially be affected if the wind blows from the direction you indicated when you entered the information about that scenario on its Screening & Scenarios record. Threat zones for screenings don't include an oval area because wind direction isn't specified in screenings.

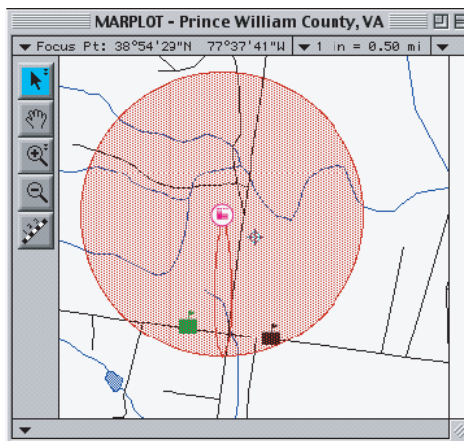


FIGURE 4-6. A threat zone plotted around a facility map symbol.

In the case of a route, the threat zone will appear as a shaded corridor along the full length of the route, twice as wide at every point along the route as the calculated threat zone radius (Figure 4-7 on page 152).

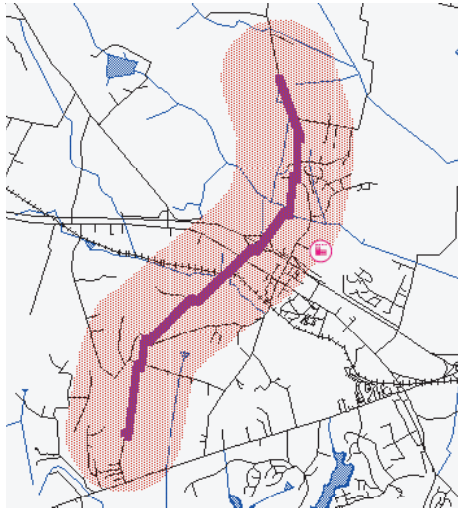


FIGURE 4-7. A threat zone plotted along a route.

Working with threat zones on maps

You can find out which special locations could be at risk during a potential incident by using Special Locations along with MARPLOT and Screening & Scenarios. For instructions, see [“Checking for special locations within a threat zone” on page 163.](#)

All your threat zones will automatically be removed from the map when you quit (exit) MARPLOT. To delete a threat zone from your map without quitting (exiting) from MARPLOT, return to the Screening & Scenarios record from which you plotted the threat zone, and select MARPLOT, then Delete Scenario Object from CAMEO’s Sharing menu. If you’ve plotted multiple threat zones on your map and would like to remove them all, while Screening & Scenarios is open, select MARPLOT, then Delete All Scenario Objects from the Sharing menu.

Choosing a stability class

Atmospheric stability class is one of the conditions you can adjust when making scenarios rather than screening calculations. The atmosphere is more or less turbulent at any given time, depending on the amount of incoming solar radiation as well as other factors. Meteorologists have defined six atmospheric stability classes, each representing a different degree of turbulence in the atmosphere. When moderate to strong incoming solar radiation heats air near the ground, causing it to rise and generating large eddies, the atmosphere is considered “unstable,” or relatively turbulent. Unstable conditions are associated with atmospheric stability classes A and B. When solar radiation is relatively weak, air near the surface has less of a tendency to rise and less turbulence develops. In this case, the atmosphere is considered “stable,” or less turbulent, the wind is weak, and the stability class would be E or F. Stability classes D and C represent conditions of more neutral stability, or moderate turbulence. Neutral conditions are associated with relatively strong wind speeds and moderate solar radiation.

Use [Table 4-6 on page 154](#) to choose the stability class that best fits a given combination of wind speed and solar radiation strength:

TABLE 4-6. Stability class choices for day and nighttime (adapted from Turner 1994).

Surface Wind Speed		DAYTIME			NIGHTTIME*	
		Incoming solar radiation:			Cloud cover:	
		Strong	Moderate	Slight	>5/10	<5/10
Miles per hour	Meters per second					
< 4	< 2	A	A-B	B	E	F
4 to 7	2 to 3	A-B	B	C	E	F
7 to 11	3 to 5	B	B-C	C	D	E
11 to 13	5 to 6	C	C-D	D	D	D
> 13	> 6	C	D	D	D	D
Choose D for completely overcast conditions during day or night.						
*Nighttime is the period from 1 hour before sunset to 1 hour after sunrise.						

What are the differences between Screening & Scenarios, ALOHA, and RMP endpoint distances?

Three kinds of toxic hazard predictions can seem very similar on first encounter:

- the *threat zone* you estimate using Screening & Scenarios.
- the *threat zone* produced by ALOHA, CAMEO's hazard model. ALOHA's toxic threat zone is defined as the area downwind of an accidental release where chemical concentrations in the air at about ground level may be high enough to be of concern.
- the *endpoint distance* you estimate when you perform an offsite consequence analysis under the Risk Management Planning Rule (or RMP Rule; see [“CAA 112\(r\)” on page 270](#)). An endpoint distance represents the distance to a particular level of hazard, such as a toxic gas concentration or heat radiation intensity. You may have used the RMP*Comp program (see [“RMP*Comp” on page 22](#)) or similar software to estimate endpoint distances.

First, what's the difference between a threat zone from Screening & Scenarios and ALOHA's threat zone? Screening & Scenarios makes the simplified threat zone calculations described in the *Technical Guidance*. ALOHA makes a toxic threat zone estimate by taking into account many factors—such as additional properties of the chemical, weather conditions, and the specific characteristics of the release source—that are not included in Screening & Scenarios calculations. ALOHA calculations are more complex, and often may more accurately predict the effects of a release.

Second, an endpoint distance for an offsite consequence analysis is similar in concept to a Screenings & Scenarios threat zone radius—so how are they different? At first glance, the hazard analysis procedures described in the *Technical Guidance* and implemented in Screening and Scenarios look similar to the RMP consequence analysis procedures specified by the RMP Rule. For example, both the RMP Rule and the *Technical Guidance* specify F stability and 3.4 miles per hour (1.5 meters per second) wind speed conditions for worst-case scenarios. But there are key differences between the two procedures:

- For many substances, the toxic endpoints specified in the RMP Rule differ from the LOCs (“Levels of Concern”) specified in the *Technical Guidance*.
- Screening & Scenarios makes only Gaussian dispersion calculations, but the RMP Rule specifies that you must “appropriately account for gas density.” That is, you must use another technique to model heavy gases (Gaussian calculations are designed for neutrally buoyant gases).
- While the *Technical Guidance* mandates rural terrain for all worst-case scenarios, the RMP Rule specifies that you should choose either urban or rural terrain conditions, depending on which choice best describes the terrain around the facility.
- Both the *Technical Guidance* and the RMP Rule specify that toxic gases should be assumed to be released over a 10-minute period. But the *Technical Guidance* specifies that the dispersion distance then be estimated by treating the release as a steady-state (infinite-duration) emission, at the rate that would result in the release of the entire quantity of the substance in 10 minutes. In contrast, the RMP Rule specifies that the distance be estimated by treating the release as an emission lasting only 10 minutes. This difference in method may seem obscure, but it can make a big difference. Distances estimated using the *Technical Guidance* method can be substantially longer than distances estimated using the RMP Rule method.

Here are the key points to remember:

- For emergency response, use ALOHA, not Screening & Scenarios.
- Don’t use Screening & Scenarios to estimate endpoint distances for the offsite consequence analyses required under the RMP Rule. You can use ALOHA or RMP*Comp for this task.
- Use Screening & Scenarios only for the hazards analyses described in the *Technical Guidance*, to meet requirements of EPCRA.

Working With Other CAMEO Components

In this chapter, you'll find descriptions of the Special Locations, Contacts, Resources, Routes, and Incidents modules, and explanations of how you can use the ALOHA and MARPLOT programs along with CAMEO's modules.

Special Locations



Use the Special Locations module to keep track of the locations of high-density, confined, or sensitive populations that might need special attention during an emergency (e.g., schools, nursing homes, hospitals, medical clinics, and community locations where people congregate). Create a Special Locations record describing each special location in your area of jurisdiction or concern. If you have a map of your area, you also can link each Special Locations record to a symbol marking its location on the map.

Adding and editing special location records

To create a new record,

1. Open the Special Locations module.
2. Select New Special Location from the Record menu. A new, blank record will be created.
3. Fill out the new Special Locations record. Check [Table 5-1 on page 159](#) for explanations of all the items. To add phone numbers or information about contact people, follow the steps in [“Adding contact information for special locations” on page 161](#).
4. Click Save Changes to save your new record.

To edit an existing Special Locations record, click the Edit button in the toolbar (see [“Editing records” on page 30](#)).

Adding and editing site plans

If you have created a site plan for a location in a graphics program, you can link that plan to the record for that location in the Special Location module, as follows:

1. Draw the site plan in a graphics program of your choice, then save it in a standard graphic file format (for example, JPEG, TIFF, GIF, BMP, or PNG). **Note:** You must save the file with a 3-character extension (for example, .jpg *not* .jpeg).
1. Place the site plan file in the SitePlans folder on your hard drive (site plan files must be kept in this folder). The SitePlans folder is inside of the CAMEO folder.
2. On your Special Location record, click the Site Plan tab, then click the Edit button in the toolbar.
3. Type or paste the file name of the site plan file into the box to the left of the Add button. Include the 3-character file name extension.
4. Press Add. You’ll see the name of the plan in the list of site plans under the Site Plan tab.
5. Click Save Changes. Now you can open and view the site plan whenever you need to, by double-clicking its name.

To edit an existing site plan, reopen it in the graphics program you used to create it, make the needed changes, then save the file under the same name you used before.

To delete a site plan from a Special Location record,

1. Open the record in Record view, click the Site Plan tab, click the Edit button in the toolbar,
2. Click on the name of that site plan, click Remove, then click Save Changes.
3. Delete the site plan file from your SitePlans folder if you have no more need of the file. Deleting a site plan from a record does not delete the site plan file itself; it remains within the SitePlans folder inside the CAMEO folder unless you remove it.

Tip: You can keep photos of locations in the SitePlans folder, as well as site plan files. Photos should be in JPEG format.

Mapping a special location

To link your new record to a symbol on a MARPLOT map, follow the steps in [“Linking map objects to CAMEO records” on page 187](#).

TABLE 5-1. Items in Special Locations records.

Item	Description
Location Name	Name of the special location.
Location Type	Editable menu of special location types (e.g., Elementary School).
Building Type	Editable menu of building types (e.g., Single Family, Tower).
Under the Address tab:	
Street Address	Street address for this special location, as well as the county (or borough or parish), fire district, and country where it's located.
Mailing Address	Mailing address for this special location
Email	Email address for this special location.

TABLE 5-1. Items in Special Locations records. (Continued)

Item	Description
Under the Population tab:	
Hours of Operation	Time periods during the day when people are present at this location (e.g., “8:00 am - 5:00 pm”).
Peak Season	Season of the year when the largest number of people are present at this location (e.g., “School Year” for an elementary school; “Summer” for a resort hotel).
Average Population	Average number of people at this location during its hours of operation on typical days.
Average Age	Average age of the people at this location.
Daily Population, Min/Max	Minimum and maximum number of people at this location during the course of a day.
Seasonal Population, Min/Max	Minimum and maximum number of people at this location during a season. (Minimum is the number present during the slowest season; maximum the number during the peak season.)
Under the Phones tab:	
Type	Type of phone number (e.g., 24-hour, emergency, office). Pull-down menu in Edit mode.
Phone	Phone number.
Under the Contacts tab: Double-click any contact to see the record for that contact.	
Last/First Name	Name(s) of the contact.
Title	Title of the contact.
Organization	Organization represented by the contact.
Under the Map Data tab:	
Latitude/Longitude	Latitude and longitude of the special location, expressed as decimal values.

TABLE 5-1. Items in Special Locations records. (Continued)

Item	Description
Method for determining lat/long	Code and description of the method by which latitude and longitude were measured.
Description of location identified by lat/long	Code and description of the location for which latitude and longitude were measured.
Record is linked to MARPLOT object	This box is checked if this record is linked to a symbol object on a MARPLOT map (see “Linking map objects to CAMEO records” on page 187).
Under the Notes tab:	
Notes	Keep your own notes about this special location here.

Adding contact information for special locations

You can add contact information for special locations in two ways:

- Including telephone numbers in Special Locations records themselves (e.g., emergency phone number for a school). Generally speaking, you should do this if you foresee referring to CAMEO in order to contact special locations during an emergency response, to be able to most quickly access the phone numbers. For an example of how you might use phone numbers in Special Locations records during a response, see [“Checking for special locations within a threat zone” on page 163](#).
- Adding Contacts records containing contact information for people representing particular special locations (such as a school principal or a facility emergency manager). In Contacts records, you can keep addresses and phone numbers for your contacts as well as notes about each contact, such as notes from meetings or phone conversations. When you add a contact for a special location, you’re actually adding a new record to the Contacts module. Information about that contact will also be displayed under that Contacts tab in the related Special Locations record.

To add a phone number to a Special Locations record,

1. Open the record in Record View.
2. Click the Phones tab, then click the Edit button in the toolbar.
3. Click within the first blank table row below the “Type” heading. In the menu of phone number types, click to select a type (e.g., 24-hour, emergency). If you need to add or modify a type, select Edit from the menu.
4. Type the phone number to the right of the menu in the same table row.
5. Click Save Changes.

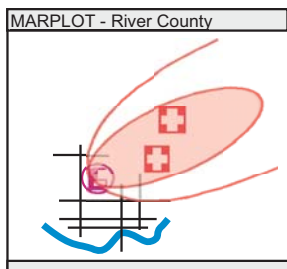
To add a record for a contact person for a special location,

1. Select the Special Locations record for that special location; open the record in Record View.
2. Click the Contacts tab, then click the Edit button in the toolbar.
3. Click Add Contact. You’ll see a list of all contacts in the Contacts module.
4. Either
 - a. Click on the name of a contact already in the list, then click Select. You would do this if a person representing the special location is already in your Contacts module (perhaps because they are a contact for more than one special location). **Note:** You can associate the same contact record with more than one special location by repeating steps 1 through 4 for each special location for which this person is a contact.
 - b. Click Add New to add a new Contacts record. You would do this if the contact is not already in the list. You’ll see a new, blank Contacts record. Fill in the information about the new contact (refer to [Table 5-2 on page 167](#) for the definitions of the items on a Contacts record), then click Save Changes. **Note:** A new Contacts record that you create by clicking New Contact from a Special Locations record will be associated with that Special Locations record, even if you don’t type the name of the special location or other identifying information about the special location, on the contact record.

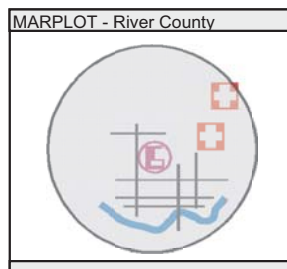
Checking for special locations within a threat zone

Special locations...

...within an ALOHA threat zone



...within an S&S threat zone



To find out which special locations could be at risk during a real or planning incident, you can use Special Locations along with MARPLOTT and either

- **ALOHA**, during an emergency response to a toxic gas release (or for planning projects). In this case, you would be most concerned about special locations located within ALOHA's threat zone (like the ones represented by the hospital symbols at left). The threat zone represents the area where gas concentrations are predicted to be high enough to be hazardous (see the discussion of ALOHA in [“CAMEO's three components”](#) on page 15).
- **Screening & Scenarios**, when you're planning for emergencies. In this case, you'd be most concerned about special locations located within a threat zone you plotted from Screening & Scenarios (like the hospital symbols at left). The threat zone represents the area that could be affected by an accidental release of a hazardous chemical (for an explanation of Screening & Scenarios threat zones and how they differ from ALOHA threat zones, see [“Using Screening & Scenarios to assess hazards”](#) on page 139).

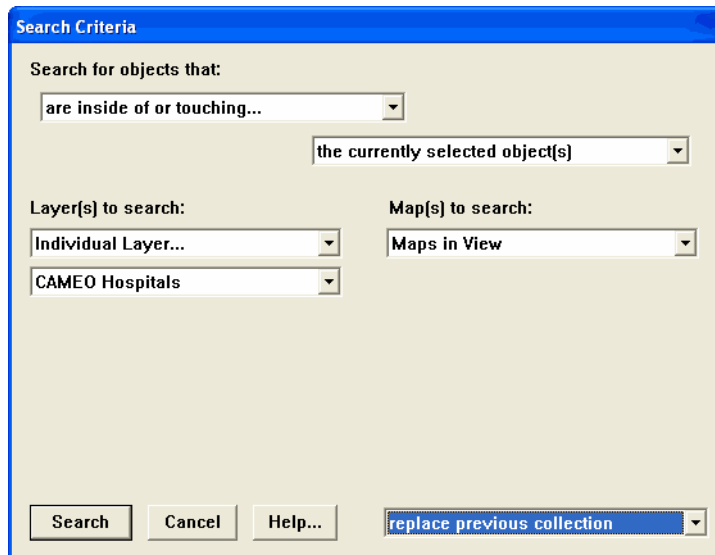
You first need to make the following preparations:

1. Create a Special Locations record for each special location of concern to you. To create a new record, open Special Locations, then select New Special Location from the Record menu; fill in the information about the special location, then click Save Changes to save your new record.
2. In MARPLOTT, place a symbol on the CAMEO Map at the location of each of these special locations, then link each special location symbol to the corresponding record. (To create and link symbols, follow the steps in [“Linking map objects to CAMEO records”](#) on page 187.)

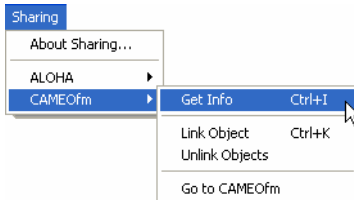
Once you've made these necessary preparations, then whenever you use ALOHA or Screening & Scenarios to create a threat zone on the map, you can quickly check for the special locations located inside the threat zone.

Here are the steps:

1. In MARPLOT, click directly on the threat zone object to select it. Check to be sure that you've selected only that object.
2. From MARPLOT's List menu, select Search.
3. Set up a search for objects that are "inside of or touching" the currently selected object(s) on the map layer(s) where you have placed your special location symbols, then press Search (check the MARPLOT manual for more on setting up searches; you can download the manual from <http://www.epa.gov/osweroe1/content/cameo/marplot.htm>). Below is an example of such a search for hospitals shown on a MARPLOT map.



4. Click Show All on Map. The map will be displayed at a scale that makes all the symbols on the layer that are within the threat zone visible; all the symbols will be selected.
5. From MARPLOT's Sharing menu, select CAMEOfm, then Get Info (as at left). The Special Locations module will be displayed, showing a list of all the records linked to the selected map objects (if there are multiple records, they're placed in a found set, just as though you had found them by running a search for them). Double-click any record in the list to view that record in Record View.



Once you have a set of Special Locations records representing the special locations within a threat zone, you can make and print a report containing the contact phone numbers for those locations:

1. From the File menu in the Special Locations module, select Print Report.
2. Click the Found Set button, so that the report will include information about all the special locations within the threat zone (each is represented by a record in the set of found records).
3. Click the Phones checkbox, as well as any other checkboxes you'd like to include. Clicking Phones includes the phone numbers for all the special locations in the report.
4. Click Print Report and OK on the following screens. You or a dispatcher now can use the report to call the locations to alert them to the emergency.

Note: You can use variations of this method for other planning and response purposes. For example, you could draw a circle on your MARPLOT map to represent the area within hearing range of an emergency siren, then search within it to find those special populations that would be able to hear the siren. In MARPLOT, it also is possible to search for special locations that are outside of the circle (and not able to hear the siren).

Contacts



Use the Contacts module as a directory of people and organizations involved in hazardous materials emergency response and/or planning. You can associate a given Contacts record with the record for a specific facility, special location, incident, or response resource (such as a media outlet or contractor).

Check [Table 5-2 on page 167](#) to see explanations of the various kinds of information shown on a Contacts record.

Adding and editing Contacts records

When you add a new record to the Contacts module, you can choose whether to associate it with a facility, special location, incident, or response

resource. The advantage of associating a Contacts record with a facility, special location, incident, or resource is that you then can use the Show Related command in the Record menu to quickly move between the related records (see [“Using the Show Related command” on page 30](#)).

- To add a record for a contact person for a particular **facility**, follow the instructions in [“Adding contact information for a facility” on page 138](#).
- To add a record for a contact person for a particular **special location**, follow the instructions in [“Adding contact information for special locations” on page 161](#).
- To add a record for a contact person for a particular **incident** (either the reporter of the incident or the discharger), follow the instructions in [“Adding Reporter/Discharger contact information” on page 176](#).
- To add a record for a contact person representing a **response resource**, follow the instructions in [“Adding and editing Resources records” on page 168](#).

To add a new contact record not associated with a facility, special location, or other record elsewhere,

1. Open the Contacts module.
2. Choose New Contact from the Record menu. A new, blank record will be created.
3. Fill out the new Contacts record. Refer to [Table 5-2 on page 167](#) for explanations of all the items.
4. Click Save Changes to save the new record.

To add a phone number to a Contacts record,

1. Open the record in Record View.
2. Click the Phones tab, then click the Edit button in the toolbar.
3. Click within the first blank table row below the “Type” heading. In the menu of phone number types, click to select a type (e.g., 24-hour, emergency). If you need to add or modify a type, select Edit from the menu.
4. Type the phone number to the right of the menu in the same table row, then click Save Changes.

To edit any Contacts record, open that record, then click the Edit button in the toolbar (see “[Editing records](#)” on page 30).

TABLE 5-2. Items in Contacts records.

Item	Description
First/Last Name	First and last names of the contact person.
Organization	Organization with which contact is affiliated.
Title	The contact’s job title or position.
Contact Type	Choose items from one or more of the four identical contact type menus to describe the primary functions of this contact person.
Under the Address tab:	
Street Address	Street address of this contact person, as well as the county (or borough or parish), fire district, and country where he or she is located.
Mailing Address	Mailing address of this contact person.
Email	Email address of this contact person.
Under the Phones tab:	
Type	Type of phone number (e.g., 24-hour, emergency, work). Pull-down menu in Edit view.
Phone	Phone number.
Under the Notes tab:	
Notes	Keep your own notes about this contact person here.



Resources

Use the Resources module to keep information about companies, people, or organizations that supply resources for emergency response or planning. Examples of resource suppliers you might want to track in the Resources module include media outlets, cleanup contractors, medical clinics and hospitals, heavy equipment operators, and equipment rental companies.

Adding and editing Resources records

To add a new Resources record,

1. Open the Resources module.
2. Choose New Resource from the Record menu. A new, blank record will be created.
3. Fill out the new Resources record. Refer to [Table 5-3 on page 169](#) for explanations of all the items.
4. Click Save Changes to save the new record.

To add a phone number to a Resources record,

1. Open the record in Record View.
2. Click the Supplier Phones tab, then click the Edit button in the toolbar.
3. Click within the first blank table row below the “Type” heading. In the menu of phone number types, click to select a type (e.g., 24-hour, emergency). If you need to add or modify a type, select Edit from the menu.
4. Type the phone number to the right of the menu in the same table row, then click Save Changes.

To edit a Resources record, click the Edit button in the toolbar (see [“Editing records” on page 30](#)).

Adding contact information for resource suppliers

To add a record for a contact person for a resource,

1. Open the Resources record in Record View.
2. Click the Contacts tab, then click the Edit button in the toolbar.
3. Click Add Contact. You'll see a list of all contacts in the Contacts module.
4. Either
 - a. Click on the name of a contact already in the list, then click Select. You would do this if a person representing the resource is already in your Contacts module (perhaps because they are a contact for more than one resource). **Note:** You can associate the same contact record with more than one resource by repeating steps 1 through 4 for each resource for which this person is a contact.
 - b. Click Add New to add a new Contacts record. You would do this if the contact is not already in the list. You'll see a new, blank Contacts record. Fill in the information about the new contact (refer to [Table 5-2 on page 167](#) for the definitions of the items on a Contacts record), then click Save Changes. **Note:** A new Contacts record that you create by clicking New Contact from a Resources record will be associated with that Resources record, even if you don't type the name of the resource, or other identifying information about the resource, on the contact record.

TABLE 5-3. Items in Resources records.

Item	Description
Supplier Name	Name of the company, person, or organization that can supply response resources.
Type	Brief description of the supplier (e.g., 911 Center, Bomb Disposal). Editable pull-down menu.
Under the Address tab:	
Street Address	Street address of this supplier, as well as the county (or borough or parish), fire district, and country where it's located.

TABLE 5-3. Items in Resources records. (Continued)

Item	Description
Mailing Address	Mailing address for this supplier.
Email	Email address for this supplier.
Under the Supplier Phones tab:	
Type	Type of phone number (e.g., 24-hour, emergency, work). Pull-down menu in Edit view.
Phone	Phone number.
Under the Items tab:	
ID	Model number or other ID number for an item supplied by this resource.
Item	Name or short description of the item supplied.
Amount	Number or quantity of items available (e.g., “4” for number of respirators, or “30” for gallons of a chemical neutralizer).
Under the Contacts tab: Double-click on any contact to see the record for that contact.	
First/Last Name	Name(s) of this contact person.
Title	Job title or position of this contact person.
Organization	Organization with which this contact person is affiliated.
Under the Map Data tab:	
Latitude/Longitude	Latitude and longitude of the location of the resource.
Method for determining latitude and longitude	Code and description of the method by which latitude and longitude were measured. Pull-down menu in Edit view.

TABLE 5-3. Items in Resources records. (Continued)

Item	Description
Description of location identified by latitude and longitude	Code and description of the location for which latitude and longitude were measured.
Record is linked to MARPLOT object	This box is checked if this record is linked to an object on a MARPLOT map (see “Linking map objects to CAMEO records” on page 187).
Under the Notes tab:	
Notes	Keep your own notes about this resource supplier here.

Routes



Use the Routes module to maintain information about routes, such as railroads or highways, used to transport hazardous materials. Route types can include air, water, or land routes. You can link records in this module to routes depicted on a map in MARPLOT.

Adding and editing Routes records

To add a new route record,

1. Open the Routes module.
2. Choose New Route from the Record menu. A new, blank record will be created.
3. Fill out the new route record. Refer to [Table 5-4 on page 173](#) for explanations of all the items.
4. Click Save Changes to save the new record.

To edit a route record, open that record, then click the Edit button in the toolbar (see [“Editing records”](#) on page 30).

Mapping a route

To link any Routes record to a line object on a MARPLOT map representing that route, follow the procedure described in [“Linking transportation routes to CAMEO records” on page 189](#).

Keeping track of chemicals transported along a route

You can add Chemicals in Inventory records for a route record. You would do this if you need to track hazardous materials that are regularly transported along the route (such as regular shipments of a particular hazardous material from a facility located along the route). To create a Chemicals in Inventory record associated with the record for a particular route,

1. In the Routes module, open the route record in Record View.
2. Turn to [“Manually creating a Chemicals in Inventory record” on page 125](#). Beginning at step 2 in that section, follow the procedure to create the new Chemicals in Inventory record.

Investigating hazards along a route

Once you have created a Chemicals in Inventory record for a hazardous material shipped along a particular route, you then can use the Screening & Scenarios module to investigate potential hazards posed by an accidental release of that substance. Follow the instructions in [“Using Screening & Scenarios to assess hazards” on page 139](#).

Tracking incidents on a route

Use the Incidents module to record incidents, such as accidental spills and releases, that have occurred along a particular route. To create a record in the Incidents module that’s associated with a particular route, follow the steps in [“Adding and editing Incidents records” on page 175](#).

TABLE 5-4. Items in the Routes module.

Item	Description
Route Name	Name of this route.
Route Type	County road, railroad, river, canal, or other type. Pull-down menu in Edit view.
Types of Vehicles	Types of vehicles common on this route or used to transport hazardous materials along this route.
Evacuation	Check if this is a designated route for evacuation during chemical emergencies.
Snow	Check if this is a designated snow emergency route.
School	Check if this is a designated route for school-related transportation.
HAZMAT	Check if this is a designated route for vehicles carrying hazardous materials (HAZMAT).
Mass Transit	Check if this is a designated mass transit route (e.g., a bus route).
Start/End Point	Starting and ending points of the route or route segment of concern.
County	County or other political unit through which the route passes.
Fire District	Fire district through which the route passes.
Vehicles Per Day	Average number of vehicles traveling the route each day.
Route plotted in MARPLOT	This box is automatically checked if this record is linked to a line object on a MARPLOT map (see “Linking map objects to CAMEO records” on page 187). Not editable.
Under the Intersections tab:	
Order	Use this box to keep track of the order of the intersections along a route (e.g., assign the order number “1” to the first intersection in a sequence, “2” to the second intersection, and so on).

TABLE 5-4. Items in the Routes module. (Continued)

Item	Description
Intersection	An intersection name or milepost number identifying a specific location on a highway that is of particular concern—either because of congestion and accident risks, or because of high volumes of hazardous materials moving through the area.
Under the Chemical Inventory tab:	
CAS	CAS number of the chemical transported along the route.
Chemical Name	Name of the transported chemical.
RIDS	Click this button to see the RIDS record for the chemical shown on this record. If you do not see a RIDS record, see “Manually creating a Chemicals in Inventory record” on page 125.
Under the Notes tab:	
Notes	Keep your own notes about this route here.

Incidents



You can use the Incidents module to keep track of releases of chemicals at facilities described in the Facilities records or along transport routes described in Routes records. You can link Incidents records to symbols marking release locations on maps in MARPLOT.

The Incidents module contains data fields (boxes) for information that, under Section 304 of EPCRA, facility operators must report for accidental releases of

- “Extremely Hazardous Substances” (also known as EHS chemicals, and defined in EPCRA).
- “Hazardous Substances” (also known as CERCLA chemicals, and defined in CERCLA).

For more details about these reporting requirements, see “[Section 304: emergency notification](#)” on page 265.

Also included on Incidents records are some of the data fields recommended for use in reporting incidents by the National Fire Information Council (NFIC) and the Emergency Response Notification System (ERNS).

A source of historical incident data

You can find information about past incidents at facilities in your area, which you can include on Incidents records, by querying the National Response Center’s incident database at <http://www.nrc.uscg.mil/foia.html>.

Adding and editing Incidents records

When you create a new Incidents record, you can choose to associate it with a particular facility or route where the incident occurred. To create a new record for an incident,

1. Skip to step 2 if you do not plan to associate the new record with a facility or route. Otherwise—if you haven’t already done so—create either a Facilities record describing the facility or a Routes record describing the route where the incident occurred.
2. Open the Incidents module and choose New Incident from the Record menu.
3. When asked whether you’d like to associate the new record with a route or facility, click Route to associate it with a route, or Facility to associate it with a facility. If you don’t want to associate the new record with a facility or route, click Neither and skip to step 5 (note that you can’t later associate the record with a facility or route record).
4. In the list of routes or facilities, click on the name of a route or facility, then click Select. A new Incidents record will be created for this route or facility.
5. Fill out the new Incidents record (check [Table 5-5 on page 177](#) for explanations of all the items), then click Save Changes to save the record.

To edit an Incidents record, click the Edit button in the toolbar (see [“Editing records” on page 30](#)).

Adding Reporter/Discharger contact information

You can create records in the Contacts module that contain contact information for incident reporters or dischargers who are associated with incidents. When you add a contact for an incident, you’re actually adding a new record to the Contacts module, but information about that contact will also be displayed in the Reporter/Discharger section under the Notification tab in the record for that incident.

To add a record for a reporter or discharger associated with a particular incident,

1. Open the incident record in Record View.
2. Click the Notification tab, then click the Edit button in the toolbar.
3. Click Add Contact. You’ll then see a list of all contacts currently in the Contacts module.
4. Either
 - a. Click on the name of a contact already in the list, then click Select. You would do this if the reporter or discharger is already in your Contacts module (perhaps because they are associated with more than one incident). **Note:** You can associate the same contact record with more than one incident by repeating steps 1 through 4 for each of the incidents for which this person is a contact.
 - b. Click Add New to add a new Contacts record. You would do this if the reporter/discharger is not already in the list. You’ll see a new, blank contacts record. Fill in the record (refer to [Table 5-2 on page 167](#) for explanations of all the items), then click Save Changes. **Note:** A new Contacts record that you create by clicking New Contact from an Incidents record will be associated with that Incidents record, even if you don’t type the name of the incident or other identifying information about the incident on the contact record.

Mapping an incident

You can link any Incident record to a symbol on a MARPLOT map representing the location of that incident. To make a link, follow the procedure described in [“Linking map objects to CAMEO records”](#) on page 187.

TABLE 5-5. Items on an Incidents record.

Item	Description
Name of facility or route	Name of facility or route where the incident took place. If this record is not associated with a facility or route, this box is blank. Not editable.
Route type or facility department	Type of route or department/division of facility where the incident took place. If this record is not associated with a facility or route, this box is blank. Not editable.
Incident	Name of the incident.
Under the Location/Date tab:	
Location	Description of the location of an incident, including the county and fire district, route or waterway, and milepost where the incident occurred.
Date/Time Spilled	Date and time when the incident happened.
Date/Time Discovered	Date and time when the incident was discovered.
Date/Time Reported	Date and time when the incident was reported to the authorities.
Under the Source tab:	
Source checkboxes	Click the checkbox for any of the 11 source categories to indicate the source(s) of the spill.
Vehicle ID	Identification or license number, if a vehicle was the source of the spill.
# of Tanks	Number of tanks involved in the release.

TABLE 5-5. Items on an Incidents record. (Continued)

Item	Description
Tank Capacity/ Units	Capacity, in weight or volume, of each tank, and a pull-down menu of capacity units.
Material Type	Click any of the four checkboxes to indicate the type of material that has been released.
Chemical	The name of the spilled chemical. If a mixture spilled, use a separate line to describe each hazardous component of the mixture.
Quantity/Units	The amount of the chemical that was released into the environment, and a pull-down menu of amount units.
Qty in Water/Units	The amount of the chemical that was released into water, and a pull-down menu of amount units.
Under the Cause tab:	
Medium	Click any of the eight checkboxes to indicate the medium or media into which the chemical was released.
Cause	Click any of the seven checkboxes to indicate the cause(s) of the incident.
Cause Description	Short description of the cause(s) of the incident.
Action Taken	Short description of the initial action taken to respond to the incident.
Under the Notification tab:	
Notified	Click any of the six checkboxes to indicate the agencies notified and the means of notification used. Click Multiple Reports if more than one agency was notified.
Agency Notified	If you clicked the Other checkbox (under the Notified heading), type the name of the agency notified about this incident.
Reporter/ Discharger	Names and titles of the people who either reported this incident or are responsible for the discharge. Double-click on a contact name or title to see the full Contacts record for that person.

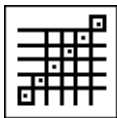
TABLE 5-5. Items on an Incidents record. (Continued)

Item	Description
Confidentiality	Click this checkbox if the reporter information in this record or in the associated Contacts record is confidential.
Through NRC	Click this checkbox if this incident was reported to the National Response Center.
Location/Report	Click this checkbox if a known location was reported.
SSI Report	Click this checkbox if a report on a statistically significant increase (SSI) in a continuous release has been filed.
Incident/Discharge	Click this checkbox if this is a single incident or discharge rather than a continuous release.
Case ID/Regional Case ID	Case ID number assigned to this incident by the regional EPA or U.S. Coast Guard office.
Case ID/NRC	Case ID number assigned to this incident by the National Response Center.
Case ID/Discharger	Case ID number assigned to this incident by the discharger.
Case ID/CR	Case ID number assigned to this incident by the EPA, if this is a continuous release.
Under the Response & Evaluation tab:	
Response Agency/Type	List of the agencies involved in the response to this incident, and the type of each agency.
Evacuation Required	Click this checkbox if people were evacuated in response to this incident.
Followup Required	Click this checkbox if a report describing this spill is required to be submitted by the discharger to the U.S. Department of Transportation, EPA, or the Chemical Safety Board.
Followup Received	Click this checkbox if a followup report describing this incident has been submitted by the discharger to the U.S. Department of Transportation, EPA, or the Chemical Safety Board.

TABLE 5-5. Items on an Incidents record. (Continued)

Item	Description
Response & Evaluation	Brief description of the response to this incident, and the evaluation of the incident.
Injuries	Number of people injured by this incident.
Deaths	Number of people killed by this incident.
Property Damage > \$50,000	Click this checkbox if the incident caused more than \$50,000 in property damages.
Under the Map Data tab:	
Latitude/ Longitude	Latitude and longitude of the incident location.
Method for determining latitude and longitude	Code and description of the method by which latitude and longitude were measured. Pull-down menu in Edit view.
Description of location identified by latitude and longitude	Code and description of the location for which latitude and longitude were measured.
Record is linked to MARPLOT object	This box is checked if this record is linked to an object on a MARPLOT map (see “Linking map objects to CAMEO records” on page 187).
Under the Notes tab:	
Notes	Keep your own notes about this incident in this space.

Using MARPLOT with CAMEO



You use MARPLOT, the mapping program, whenever you use a map with CAMEO. You use a map when you link the CAMEO record for a chemical facility, location of a hazardous chemical incident, special location, or transportation route to its symbol on a map in MARPLOT, and whenever you plot a Screening & Scenarios threat zone. You also use MARPLOT and ALOHA together when you plot an ALOHA threat zone on a map.

MARPLOT was primarily designed to display maps of U.S. counties generated from TIGER/Line files developed by the U.S. Census Bureau (download MARPLOT maps from <http://www.epa.gov/osweroe1/content/cameo/marmaps/> or learn more about TIGER/Line files at <http://www.census.gov/geo/www/tiger/>). This section explains the procedures for using CAMEO and MARPLOT together for some basic projects, but doesn't provide detailed explanations of TIGER/Line files, and doesn't fully explain how to use MARPLOT. You'll find all that information in the MARPLOT manual (download it from <http://www.epa.gov/osweroe1/content/cameo/marplot.htm>).

A point of potential confusion is that, although MARPLOT is the mapping component of CAMEO, MARPLOT and CAMEO's collection of modules are actually two separate programs communicating with each other. For the two programs to share information, they must both be running on your computer at the same time. Whenever you do something in a CAMEO module that requires communication with MARPLOT, or vice versa, if one of the two programs isn't yet running, the other will automatically start it.

The CAMEO Map

The CAMEO Map is a special MARPLOT map that you use as an overlay on other maps, such as the sample map of Prince William County that is included with your copy of MARPLOT. Use the CAMEO Map, not maps like the Prince William County map, as the location for all CAMEO-related map objects that you create (such as symbols marking the locations of facilities and special locations).

The CAMEO Map comes with CAMEO and is automatically installed when you install CAMEO. It is itself invisible, but on it, you can include layers on which you can place different kinds of visible, CAMEO-related map objects, such as symbols for facilities and special locations.

To check whether the CAMEO Map is installed in your copy of MARPLOT, open MARPLOT and choose Map List from the List menu. If the CAMEO Map is installed, you'll see it in the list of maps in use. You also can select Layer List from the List menu to see the various layers that are used in the CAMEO Map (as in Figure 5-1), along with map layers of other maps in view. The names of layers used on the CAMEO Map start with the word “CAMEO”—such as CAMEO Facilities, CAMEO Hospitals, and CAMEO Schools.

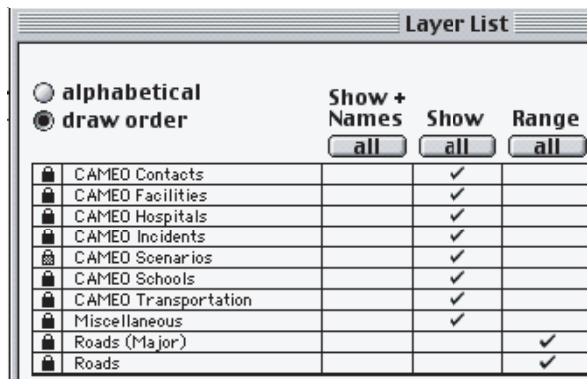


FIGURE 5-1. The seven standard layers used in the CAMEO Map appear at the top of this list of MARPLOT map layers.

You can add a new layer to the CAMEO map, as follows:

1. Choose Layer List from MARPLOT's List menu, then click New.
2. Type a name for the layer that starts with “CAMEO” (e.g., CAMEO Community Centers).
3. Check that the create layer button is selected, then click OK.

Map objects and links

You can place **map objects** on layers of the CAMEO Map and then link them to CAMEO records, as explained below. A link is a two-way connection between a particular map object and a related CAMEO record (an example is shown in Figure 5-2). Once you link a record to an object, you can use the link to quickly move back and forth between a map in MARPLOT and record(s) in CAMEO in the following ways:

- If you're working with one or more records for locations or routes that are linked to map object(s), you can view those locations or routes on a map.
- If you're working with a map in MARPLOT and have selected one or more object(s) on it, you can see the records linked to those objects.

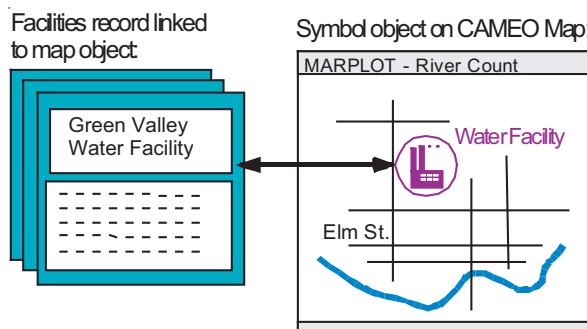


FIGURE 5-2. The link between a record in the Facilities module and an object on the CAMEO Map is represented by an arrow in this diagram.

Two types of map objects are most commonly linked to CAMEO records (Figure 5-3 on page 184):

- **symbol objects** that mark the locations of facilities (such as chemical processing plants), special locations (such as schools or hospitals), response/planning resources (such as equipment stockpiles), and past incidents.

- **polyline objects** that represent routes (along roads, rivers, railroads, or combinations of these) used to transport hazardous substances.

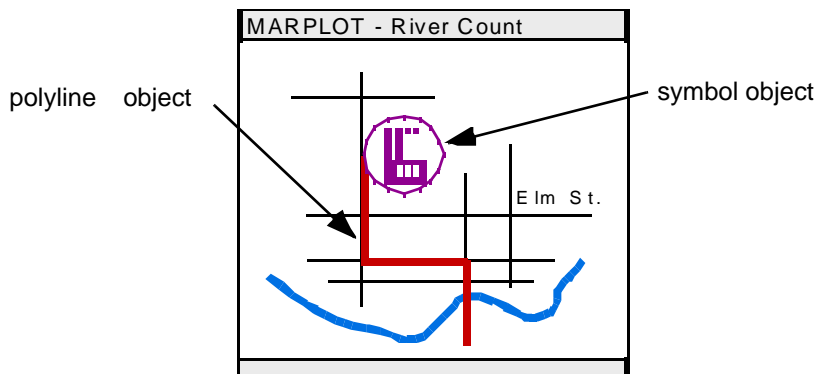


FIGURE 5-3. Polyline and symbol objects on a map.

You also can create polygon objects on maps to represent very large facilities, and link those objects to facility records. Check the MARPLOT manual for more details about symbol, polyline, and polygon objects.

CAMEO modules and linking. In terms of linking to MARPLOT map objects, there are three kinds of CAMEO modules:

1. Records in the Facilities, Special Locations, Routes, Incidents, and Resources modules can be linked directly to map objects.
2. Records in the Chemical Library, Contacts, and Chemicals in Inventory modules can't be linked to map objects.
3. Screening & Scenarios records can't be linked directly to map objects. But you can plot threat zones on maps from any Screening & Scenarios records that you have indirectly linked to map objects by creating a linked Facilities or Routes record, a Chemicals in Inventory record, and a Screening & Scenarios record, following the steps in [“Plotting a threat zone on a map” on page 151.](#)

Using links

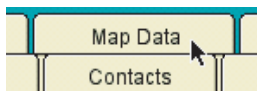
You use links to move back and forth between CAMEO and MARPLOT, as follows.

Checking whether a record is linked. To tell whether a CAMEO record is linked to a map object, either

- With the record selected or in Record view, from CAMEO’s Sharing menu, select MARPLOT, then Show on Map (you’ll be alerted if the record is not linked).

or

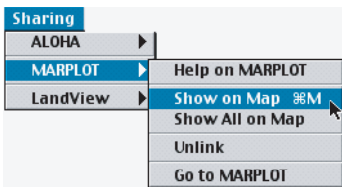
1. Open the record in Record view.
2. Click the Map Data tab (as at left). **Note:** There is no Map Data tab in the Routes module.



Record is linked to MARPLOT object

The “Record is linked to MARPLOT object” box is checked if the record is linked to a map object (as at left). Records in modules that can’t be linked to map objects don’t contain this checkbox.

When you start from CAMEO... If you’re viewing a record for a facility, special location, incident, resource, or route, to see that location or route on a map,



- From CAMEO’s Sharing menu, select MARPLOT, then Show on Map (as at left). MARPLOT will come forward, and the map object linked to the record will be displayed. You’ll be alerted if the record is not linked to a map object; follow the steps in [“Linking map objects to CAMEO records” on page 187](#) if you’d like to create a map object and link.

If you’re working with a Screening & Scenarios record and want to view the threat zone on a map, follow the instructions in [“Plotting a threat zone on a map” on page 151](#).

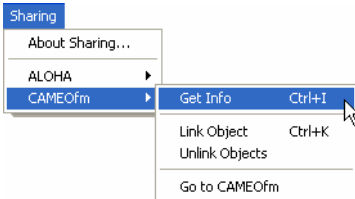
If you have run a search in a CAMEO module and have found multiple records linked to map objects, to see all those objects on the map:

- While that module is open, from CAMEO's Sharing menu, select MARPLOT, then Show All on Map. MARPLOT will come forward, and all the objects will be displayed (they also will be selected).

When you start from MARPLOT.. While you're working with a MARPLOT map, to see the record linked to a map object,



1. Click on the map object to select it. When an object is selected, four small squares surround it (like the squares around the facility symbol at left).
2. From MARPLOT's Sharing menu, select CAMEOfm, then Get Info (as at left). CAMEO will come forward, and will display the linked record.



While you're working with a map, to see multiple records in CAMEO that are linked to multiple objects on the map:

1. Select the linked objects, either by shift-clicking on them or by making a search for them (check the MARPLOT manual to learn how to search for map objects).
2. From MARPLOT's Sharing menu, select CAMEOfm, then Get Info.
3. CAMEO comes forward and puts the linked records into a found set¹ (just as if you had found the linked records by making a search in CAMEO). Use the navigation buttons in the toolbar to browse through the linked records in the collection. **Note:** Select Show All Records from the Record menu when you're ready to clear the found set and return to normal browsing.

1. Or more than one found set, if the map objects are linked to records in more than one CAMEO module. In this case, a found set is created in each of the modules; when you select Get Info, you'll be asked which module you would like to view. After viewing the linked records in one module, to pick a different module to view, return to MARPLOT, and choose Get Info again.

Linking map objects to CAMEO records

You need to run CAMEO and MARPLOT simultaneously in order to establish map links. You can confirm that the programs are communicating correctly with each other in either of two ways:

- In MARPLOT, from the Sharing menu, selecting CAMEOfm, then Go to CAMEOfm (this should transfer you from MARPLOT to CAMEO).
- In CAMEO's Home navigation window, clicking the MARPLOT button (this should take you from CAMEO to MARPLOT).

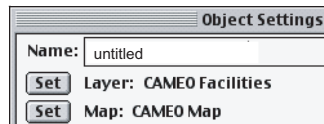
To link or not to link. Consider the uses you plan to make of your map and your CAMEO records as you choose whether to link a given CAMEO record to a map object. Don't create a map object and link unless you know that it will be useful to you to move back and forth between MARPLOT and CAMEO to see the map object representing a CAMEO record and/or vice versa to see the record linked to a map object. You're likely to find that the advantage of keeping your map uncluttered outweighs the advantage of linking some of your CAMEO records to map objects. On the other hand, other linked map objects can be very handy. For example, see [“Checking for special locations within a threat zone” on page 163](#).

Linking symbol objects to records. To link a symbol object on a MARPLOT map to a record in the Facilities, Special Locations, Incidents, or Resources module,

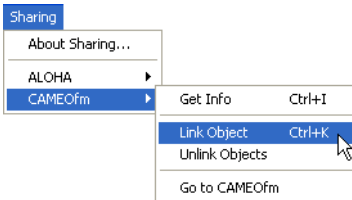
1. Create and save the record in CAMEO, if you haven't already.
2. Open your map in MARPLOT so that the location where you'd like to place the symbol is visible.
3. In MARPLOT, from the List menu, select Layer List. You're about to place a new symbol on a map layer. You should place any symbol that you'll link to a CAMEO record on one of the CAMEO layers (which are on the CAMEO Map). For example, place symbols for facilities on the CAMEO Facilities layer, hospital symbols on the CAMEO Hospitals layer, and so on. You can create new CAMEO layers if needed; check the MARPLOT manual for instructions.



4. Click on the lock icon for the CAMEO layer on which you want to place the symbol. This unlocks the layer (at left, the CAMEO Facilities layer has just been unlocked). Click OK.
5. In MARPLOT's toolbar, click the symbol tool to select it (as at left).
6. On the map, click on the location where you'd like to place the symbol. The Object Settings dialog box will appear.
7. Check that the new symbol object has been placed on the unlocked CAMEO layer of the CAMEO Map (placement is correct in the example below). If it has been placed on a different layer and/or map, click Set to move the object to the correct layer and/or map. No need to name the object; it will automatically be given the name of the record you're linking it to.



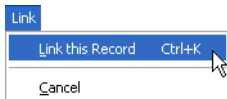
8. Adjust the symbol's color and other characteristics as you'd like. Click OK.
9. From the List menu, select Layer List, then click again on the lock icon for the CAMEO layer. This relocks the layer so you won't accidentally move or delete the symbol. Click OK.
10. In MARPLOT, find and select the new symbol object (it's selected when four small squares surround it).



11. In the Sharing menu, select CAMEOfm, then Link Object (as at left). CAMEO will come forward.

12. In CAMEO, find and select the record you want to link to the object.

13. From CAMEO's Link menu, select Link this Record (as at left). To check that the link has been made, click the Map Data tab. If the link was established, the "Record is linked to MARPLOT object" box is checked (and the Link menu will disappear).



Linking transportation routes to CAMEO records. Roads, rail lines, and waterways are represented on MARPLOT maps by polylines, like the polylines on the map shown in Figure 5-4, which represent neighborhood roads and a creek.

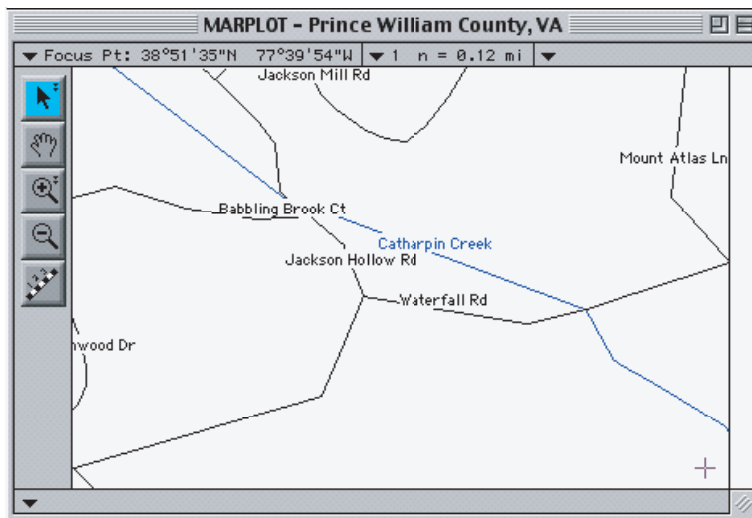


FIGURE 5-4. On this MARPLOT map, roads are shown as black polylines (water bodies such as Catharpin Creek are blue polylines).

It's likely to seem logical at first to link records in the Routes module directly to polyline objects on the Roads, Railroads, or Water layers of your MARPLOT map. However, there are drawbacks to doing this:

- Only rarely will a transportation route of concern to you correspond to a single Road, Railroad, or Water object in MARPLOT. That's because routes on MARPLOT maps typically include either pieces of several roads, or just part of a railroad or river object.
- Objects on the Roads, Railroads, and Water layers, which are derived from the nationwide TIGER/Line database, may exhibit idiosyncrasies, such as missing or misnamed segments.

- For purposes of using maps with CAMEO, it's easiest to always link CAMEO records to objects on the CAMEO Map, because then you'll always know where to find those objects. Other kinds of objects, such as Roads, Railroads, and Water objects, are on particular county maps, not on the CAMEO Map.

To avoid these drawbacks, when you want to link a record in the Routes module to a map object, follow the steps below to create a polyline object on the CAMEO Map that overlays the route of interest. You then link this object to your CAMEO record. Figure 5-5 shows an example of such a route object.

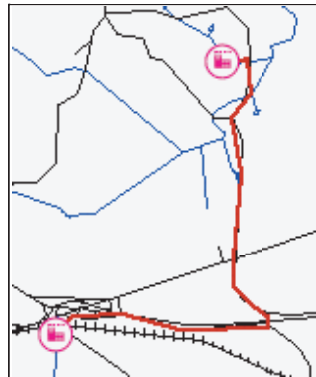
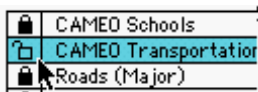


FIGURE 5-5. The red polyline object, tracing a hazmat transportation route between two chemical facilities, can be selected and linked to a Routes record.

To create a route object in MARPLOT and link it to a Routes record,



1. Open your map in MARPLOT so that the entire route of interest is visible (zoom out if necessary).

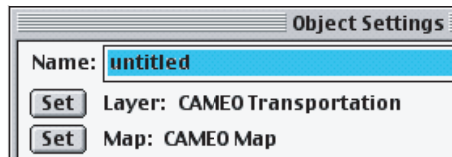
2. In MARPLOT, from the List menu, select Layer List.

3. Click on the lock icon for the CAMEO Transportation layer, to unlock this layer (as at left), then click OK.

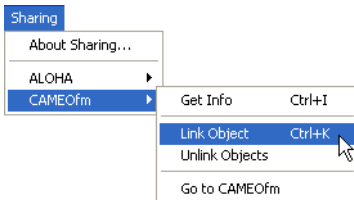
4. In MARPLOT's toolbar, click the polyline tool to select it (as at left).



5. Use the tool to create a polyline that traces the path of a road, railroad, or waterway that is itself an object on the underlying map. You'll place the polyline on the CAMEO Transportation layer of the CAMEO Map.
6. Click on one endpoint of the route, then trace the route by clicking on each vertex (that is, on each place where the route bends), double-clicking on the final endpoint. Once you've double-clicked the final endpoint, the Object Settings dialog box will be displayed.
7. Check that the new route object has been placed on the CAMEO Transportation layer of the CAMEO Map (placement is correct in the example below). If it has not, click Set to move the object to the correct layer or map. No need to name the object; it will be given the name of the record it's linked to.



8. Adjust the route's line width, color, and other characteristics as you'd like. Click OK.
9. From the List menu, select Layer List, then click again on the lock icon for the CAMEO Transportation layer (to relock this layer so you won't accidentally move or delete the route object). Click OK.
10. From MARPLOT's Sharing menu, select CAMEOfm, then Link Object (as at left). CAMEO will come forward.
11. Find and select the Routes record that you want to link to your new route object.
12. From CAMEO's Link menu, select Link this Record to make the link.

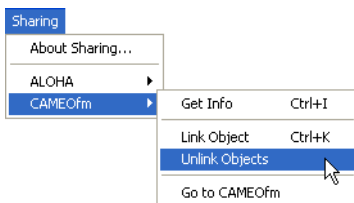


Deleting a map link

For any of various reasons, you might decide to delete, or “break” a link between a CAMEO record and a map object. For instance, you might discover that you've linked a record to the wrong object. Or you might want to remove an object from your map or a record from a CAMEO module altogether.

To delete a link between a CAMEO record and an object on a map,

1. Open MARPLOT, then find and select the linked object.
2. From MARPLOT's Sharing menu, select CAMEOfm, then Unlink Objects. All the information about the link will be deleted from both MARPLOT and CAMEO.



To delete a record linked to an object on a map, select Delete <Module Name> from the Record menu in CAMEO (that is, select “Delete Special Location” in the Special Locations module, “Delete Facility” in the Facilities module, and so on). There’s no need to delete the link before deleting the record; it is automatically deleted when you delete the record. The map object, now unlinked, will remain on the map unless you also delete it.

Screening & Scenarios threat zones are automatically removed from the map when you quit (exit) MARPLOT. To delete a threat zone from your map without quitting from MARPLOT, return to the Screening & Scenarios record from which you plotted the threat zone, and select MARPLOT, then Delete Scenario Object from CAMEO’s Sharing menu (or Delete All Scenario Objects to delete multiple threat zones).

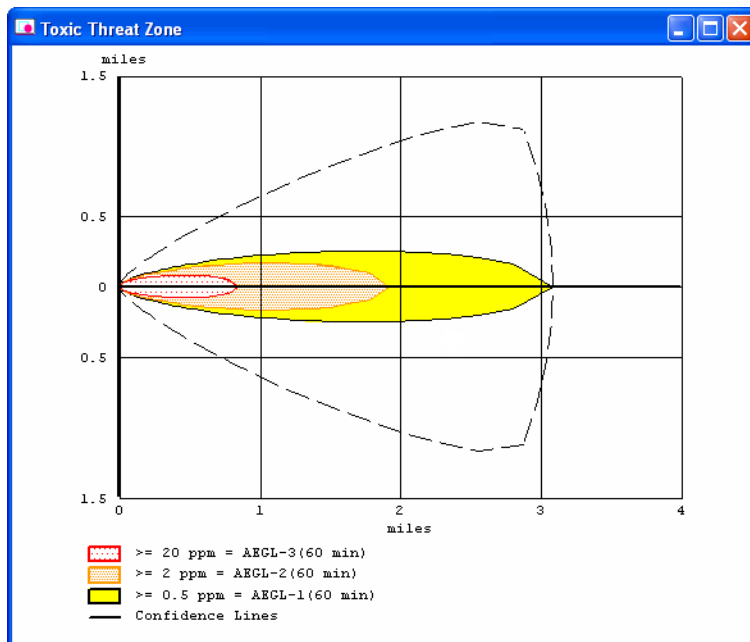
Using ALOHA with MARPLOT and CAMEO



You use ALOHA, CAMEO’s hazard model, to model key hazards— toxicity, flammability, thermal radiation (heat), and overpressure (explosion blast force)—related to chemical releases that result in toxic gas dispersions, fires, and/or explosions.

Note: In versions prior to ALOHA 5.4, only the toxic threat is modeled.

Once you enter a description of that accidental chemical release into ALOHA, it displays a threat zone plot (like the one shown at below) showing one or more areas where a hazard—toxicity, flammability, thermal radiation, or damaging overpressure—may exceed key Levels of Concern (LOCs) and pose a threat to people and property. You can plot ALOHA threat zones on maps in MARPLOT.



This manual explains only how you can use ALOHA along with CAMEO and MARPLOT. For a full explanation of how to use ALOHA and a “crash course” in the basics of air dispersion modeling, download the ALOHA manual from <http://www.epa.gov/osweroe1/content/cameo/aloha.htm>.

You can use ALOHA on its own, without using CAMEO and MARPLOT along with it. But, like MARPLOT, ALOHA is a component of CAMEO that can communicate both with CAMEO modules and MARPLOT. For ALOHA, MARPLOT, and CAMEO to share information, they must be running on your computer at the same time. Whenever you do something in CAMEO or MARPLOT that requires communication with ALOHA, or vice versa, if one of the two programs isn't yet running, the other will automatically start it (as long as there is enough free memory on your computer).

ALOHA contains a hazardous chemicals database, which includes a subset of the chemicals in CAMEO's Chemical Library. Less than 1,000 of the CAMEO chemicals have been judged toxic enough and volatile enough to be potential airborne hazards. These are the chemicals you'll find in ALOHA's database. To quickly find out whether a particular CAMEO chemical is in ALOHA, open the record for that chemical in the Chemical Library, then choose Select This Chemical in ALOHA from CAMEO's Sharing menu. If the chemical is in ALOHA's chemical database, ALOHA will start up, the chemical's property data will be loaded into ALOHA, and you then can begin to enter the details of a release scenario into ALOHA. If the chemical is not in ALOHA's database, you will be alerted that the chemical does not match any of the chemicals in ALOHA's chemical library.

Note: Many people wonder about the difference between the threat zones that ALOHA plots and the threat zones plotted by the Screening & Scenarios module. For an explanation, see [“What are the differences between Screening & Scenarios, ALOHA, and RMP endpoint distances?”](#) on page 155.

What ALOHA, CAMEO, and MARPLOT can accomplish together

Using ALOHA with MARPLOT and/or CAMEO modules makes it possible to accomplish some tasks that you couldn't do otherwise. For example,

- You can create a threat zone plot in ALOHA, then place the threat zone on a map in MARPLOT, so that you can see the area that might be affected by the release.
- You can create Chemicals in Inventory records in CAMEO for chemicals stored in the inventories of local facilities. Later, from any of those records, you can start ALOHA with the chemical selected so that you can plot a threat zone for a potential release of that chemical from that facility.

Tip: To simplify your work with ALOHA, keep notes in your chemical inventory records on storage container dimensions, dike dimensions, storage temperature, and other information you'd need for ALOHA scenarios. You then can refer to those notes next time you need to run ALOHA. To find out just what information you'd need, either check the ALOHA manual or just experiment with ALOHA.

An ALOHA modeling example. Imagine that a local facility in your area stores a chemical such as chlorine or ammonia, which would pose a serious toxic gas hazard if accidentally released. For an emergency planning project, you might want to investigate the possible hazard to your community from a potential release of that chemical at the facility. Here's how you could use the three CAMEO components to accomplish this planning task:

1. Create a Facilities record describing that facility, and link it to a symbol object representing the facility on a MARPLOT map of your community. Also create a Chemical Inventory record describing the hazardous chemical at that facility. (Follow the procedures in [Chapter 4, "Working with Facility Records,"](#) beginning on page 107, to create these records and link.)

2. Create Special Locations records for each of the locations in your community (such as schools, hospitals, and community centers) that would need special attention in the event of a toxic gas release. Link each of those records to symbols representing those special locations on your MARPLOT map. (Follow the procedures in [“Special Locations” on page 157](#) to create these records and links.)
3. Open the Chemical Inventory record, then, from the Sharing menu, select ALOHA, then Select this Chemical in ALOHA. This starts ALOHA with the chemical selected.
4. Run a release scenario for this chemical in ALOHA and plot a toxic threat zone in ALOHA. (Refer to your ALOHA manual to learn how to do this.) The threat zone represents the area where the toxic gas might pose a hazard to people, given the circumstances in your release scenario.
5. In MARPLOT, find the location of the release on the map of your community, click on that location with the arrow tool, then, from the Sharing menu, select ALOHA, then Set Source Point. The threat zone will be drawn on the map.
6. Search within the threat zone for the symbols for special locations, such as schools and hospitals. Use the map links to quickly collect together all Special Locations records linked to those symbols. During a real incident, you could quickly make and print out a list of contact phone numbers for those locations (for detailed instructions, see [“Checking for special locations within a threat zone” on page 163](#)).

Importing and Exporting Data

In this chapter, you'll find explanations of the kinds of data transfers that can be made to or from CAMEO, and instructions for completing each kind of transfer. You'll also find background information you'll need to know to avoid problems when you transfer data.

Avoiding trouble

To avoid trouble when you transfer data to and from CAMEO, we recommend that you

- Always back up your existing data before importing new data into CAMEO. It's easy to do: just export all the data from the affected module(s) into an archive file (see [“Exporting data from all CAMEO modules” on page 211](#) and [“Exporting data from all records in a CAMEO module” on page 212](#)).
- Follow the **CAMEO Data Ownership Rule**.

The Ownership Rule says that the CAMEO user with the most intimate knowledge of the data on a CAMEO record should be considered the **owner** of that record. The owner of a record should be the only person who makes changes to it (or directs others to change it). To implement the Ownership Rule, first, establish who owns which data. Then, import data only from its owner, and export data only if you own it.

Here's an example of how the Ownership Rule works in practice: Suppose that you work for a fire district that has a mutual aid agreement with neighboring districts. Also suppose that each district has agreed to be responsible for collecting information about the facilities in their district and creating CAMEO records describing those facilities.

To prepare for mutual aid calls, you might want to import the other districts' facility records into your copy of CAMEO. The other districts likewise might ask you to export your facility records to them, for use in their copies of CAMEO. You can avoid causing problems for yourselves by agreeing to the following rules:

- Each district is the owner of the data describing the facilities in its own fire district.
- Each district will export only the records that it owns.
- Each district will import records only from their owner.

To export your facility records to the other districts, you would follow the steps in [“Exporting data from a subset of the records in a module” on page 213](#) to export just the data for the facilities in your district.

What if your fire district, and the neighboring districts, instead receive your facility data from the SERC, which compiled that data from submitted Tier II reports? You would then consider the SERC to be the owner of the facility data for all your districts. Each district would import data only from the SERC. Districts would never export facility records.

Why following the Ownership Rule can save you time and trouble has to do with the way CAMEO keeps track of records. Review [“How CAMEO assigns and uses record IDs” on page 202](#) for details.

For suggested strategies for avoiding inadvertent record duplication, see [“Strategies for avoiding duplicate records” on page 218](#).

Important: Do not delete either the ImportFiles or ExportFiles folder from your CAMEO folder. Without them, CAMEO cannot import or export data.

Possible data transfers

You can transfer data to and from CAMEO in three ways:

- from Tier2 Submit to CAMEO. Tier2 Submit is a program used by facilities for preparing and submitting Tier II chemical inventory reports. See [“Importing data from Tier2 Submit” on page 205](#).
- between two copies of CAMEO. See [“Transferring data between different copies of CAMEO” on page 210](#).
- from CAMEO to a data file that can be imported into a spreadsheet, database, or other program. See [“Exporting data to other programs” on page 222](#).

If your facility data comes from Tier2 Submit and you have linked the data to MARPLOT maps and site plans, you have the option on importing current data to copy the map links and site plan links to the new data from the previous year’s records.

Difficult or impossible data transfers

Two kinds of data transfers can't be made, or can be made only with difficulty:

- Data can't be exported from CAMEO to Tier2 Submit.¹
- It would be difficult, though not impossible, to import data created in non-CAMEO programs into CAMEO. A file using CAMEO's field order and format could be imported into CAMEO, but you would need to carefully consider assignment of record IDs (see [“How CAMEO assigns and uses record IDs” on page 202](#)).

Note: If you want to try creating an import file, first open and inspect it in a database or spreadsheet program a file exported from any CAMEO module, so that you can see the field order and format used by CAMEO. The first record within any file exported from CAMEO provides data field names and field order; import files must be in zipped merge format (see [“Zipped merge files: CAMEO's standard file format” on page 201](#)).

Things to know before transferring data

Before beginning data transfers, it's helpful to understand

- The format of the zipped **merge files** CAMEO uses in data import and export operations.
- The behind-the-scenes **record ID** method CAMEO uses to keep track of relationships among records in related modules.

1. Data can be transferred between different copies of Tier2 Submit. See the Tier2 Submit manual for details.

Zipped merge files: CAMEO's standard file format

To import and export data, CAMEO uses a standard type of data file called a merge file. When you export data from CAMEO, a merge file is created for each module from which data is exported (merge files are also created for behind-the-scenes supporting files). Merge files exported from CAMEO contain the data exported from the corresponding modules and/or supporting files.

A merge file is a text file in which data fields are separated by commas, records are separated by carriage returns, and the contents of each data field are enclosed in quotation marks. A merge file has a file extension of .mer (for example, "export.mer").

With each data export, the merge files exported from CAMEO are automatically compressed together into a single zipped file, which has a file extension of .zip (for example, "export.zip"). Depending on the nature of the data you're exporting from CAMEO, the zipped file created by CAMEO may contain a single merge file or multiple merge files.

Once you've exported merge files from CAMEO and have uncompressed (unzipped) them, you can open and work with them in standard database and spreadsheet programs, and in text editing programs.

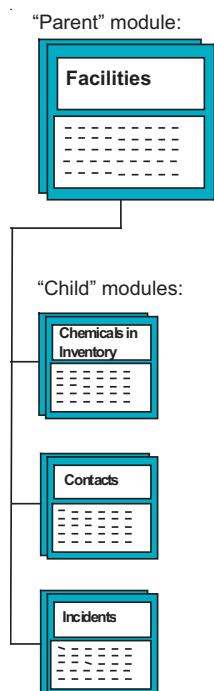
The format of Tier2 Submit files. You might receive Tier2 Submit data files either directly from facilities reporting their chemical inventory data or from colleagues. Tier2 Submit data files contain the Tier II information submitted by those facilities, and can be imported into CAMEO (see ["Importing data from Tier2 Submit" on page 205](#)). A data file generated from Tier2 Submit is a collection of individual merge files compressed together into a single zipped file. CAMEO can automatically uncompress a Tier2 Submit file and import the data from the merge files.

How CAMEO assigns and uses record IDs

CAMEO uses **record ID numbers**, or record IDs, to keep track of the records in its modules and the relationships between records.

Each new record is assigned a unique 20-character record ID at the moment it's created. In the case of CAMEO's facility records, a record ID is assigned to a given facility in either of two ways:

- When you first create a facility record in CAMEO, a record ID is automatically generated for that record.
- When a file is created for a facility in Tier2 Submit, a record ID is created for that facility. The file exported from Tier2 Submit contains that record ID, which is imported into your copy of CAMEO when you import that Tier2 Submit file into CAMEO.



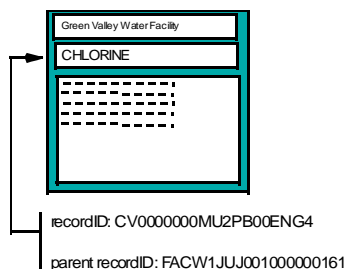
Parent and child modules and record IDs. Record IDs are used in CAMEO to track relationships among records stored in different modules, particularly relationships between records in “parent” and “child” modules. In CAMEO, as with other kinds of databases, some modules are related to each other in somewhat the same way that parents and children are related: child modules are subordinate to parent modules. For purposes of tracking relationships among records in CAMEO,

- The Facilities and Routes modules are parent modules, as are the Special Locations, Chemical Library, and Resources modules.
- The Chemicals in Inventory, Contacts, and Incidents modules are child modules, because the main purpose of these modules is to store information that helps to describe facilities, routes, and special locations.

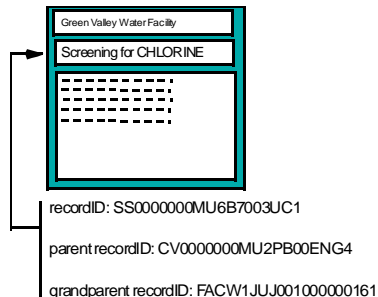
For example:

- Chemicals in Inventory records describe either chemicals in a facility's inventory or chemicals routinely transported along a given route.
- Incidents records typically describe incidents that have happened at facilities or along routes.

- Contacts records describe contact people for facilities and special locations. Other Contacts records may describe people involved in incidents, such as the discharger or the notified agency, and/or contact people for companies that provide response resources (records on such companies can be kept in the Resources module).



In the Facilities and Routes modules, each new facility or route record is assigned a record ID when it is created. From then on, when a new Chemicals in Inventory record is created to describe a chemical (or mixture) stored at a given facility or transported on a given route, the new record is given its own unique record ID and also a **parent record ID** that matches the facility or route's record ID, as in the example Chemicals in Inventory record at left. CAMEO uses the parent record ID to track the facilities or routes associated with the Chemicals in Inventory record.



Screening & Scenarios can be considered a “grandchild” module because it is subordinate to the Chemicals in Inventory module, just as the Chemicals in Inventory module is subordinate to the Facilities and Routes modules. For that reason, any Screening & Scenarios record contains (a) its own unique record ID, (b) the parent record ID for the related Chemicals in Inventory record, and (c) the **grandparent record ID** for the facility or route record that is related to the Chemicals in Inventory record, as in the example Screening & Scenarios record at left. CAMEO uses the parent record ID to associate the Screening & Scenarios record with the Chemicals in Inventory record, allowing information about the chemical to be used in hazard zone estimation. It uses the grandparent record ID when you choose to plot a hazard zone on a map (the hazard zone is plotted either around the facility or along the route that has a record ID matching the grandparent record ID).

Data transfers and record IDs. For each record in a data file exported from one copy of CAMEO into another copy of CAMEO, one of two things happens:

- A new record is created in the target copy of CAMEO to hold the newly imported data.
- An existing record is updated.

What determines the outcome—whether a new record is created or an existing record is updated—is the record ID for that record, which is included in the file and imported along with the rest of the data in that record. The record ID for each imported record is compared with the record IDs for the existing records in a given target module:

- If no match is found, a new record containing the imported data is added to the target module, and the record ID for the new record is set to match the record ID of the imported record.
- If a match is found, the corresponding record in the target module is updated with the imported data. During the update, *all* the data in the target record is replaced by the data in the matching imported record.

Records in CAMEO, especially records describing facilities, can be duplicated when you might not necessarily expect it, because of how CAMEO uses record IDs. For example,

- A duplicate record for a facility is created if a record for that facility exists in your Facilities module and you import a Tier2 Submit file containing information about that facility. That’s because both CAMEO and Tier2 Submit assigned different record IDs to that facility.
- Records can be duplicated when you transfer data between two copies of CAMEO, if the source and target modules each contain a record for the same entity, such as a facility, and if those records came from different sources, and hence have different record IDs.

The best way to avoid unintended record duplication is to follow the CAMEO Data Ownership Rule. For other strategies for avoiding or troubleshooting unintended record duplication, see [“Strategies for avoiding duplicate records” on page 218](#).

Viewing record IDs. Record ID numbers are not displayed directly in CAMEO’s windows, and you generally don’t need to see the record IDs for the records in your copy of CAMEO, but you can view them if you need to. You might want to view the record IDs in a particular module if you suspect that the module contains unwanted duplicates of some records.

To see the record IDs in any of your CAMEO modules,

1. Export the data from that module, following the steps in [“Exporting data from all records in a CAMEO module” on page 212](#).
2. Open the exported file in a spreadsheet program, following the steps in [“Exporting data to other programs” on page 222](#).

In the exported file, you can recognize record IDs by their distinctive format. Each is 20 characters long, and the record IDs for each CAMEO module can be identified by their unique two-letter prefix. For example, the first two characters of any facility record ID are “FA.”

Importing data from Tier2 Submit

Under the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA), facilities that maintain more than threshold amounts of certain hazardous materials must submit an annual chemical inventory report, called a Tier II form, to their state’s State Emergency Response Commission (SERC), Local Emergency Planning Committee (LEPC), and local fire department. Operators of facilities covered by EPCRA who must submit an annual Tier II report can use Tier2 Submit to prepare and submit that report as an electronic data file (see [“Tier2 Submit” on page 21](#)).

You might receive such Tier2 Submit files if you are

- A member of a SERC, LEPC, or local fire department to whom facilities submit their reports.
- A local responder or planner who receives, from your SERC or LEPC, a Tier2 Submit file of data for facilities in your area.

You can import Tier2 Submit data files into CAMEO. When you import a Tier2 Submit file into CAMEO, data from that file will be imported into the Facilities, Chemicals in Inventory, and Contacts modules, as follows ([Figure 6-1 on page 206](#)):

- A new Facilities record will be created for each facility described in the Tier2 Submit file. It will be assigned the record ID generated for that facility by Tier2 Submit.

- A new Chemicals in Inventory record will be created for each chemical (or mixture) in a given facility's inventory that's listed in the Tier2 Submit data file. The parent record ID assigned to this record will match the facility's record ID.
- A new Contacts record will be created for each contact person listed in the Tier2 Submit data file. The parent record ID assigned to this record will match the facility's record ID.

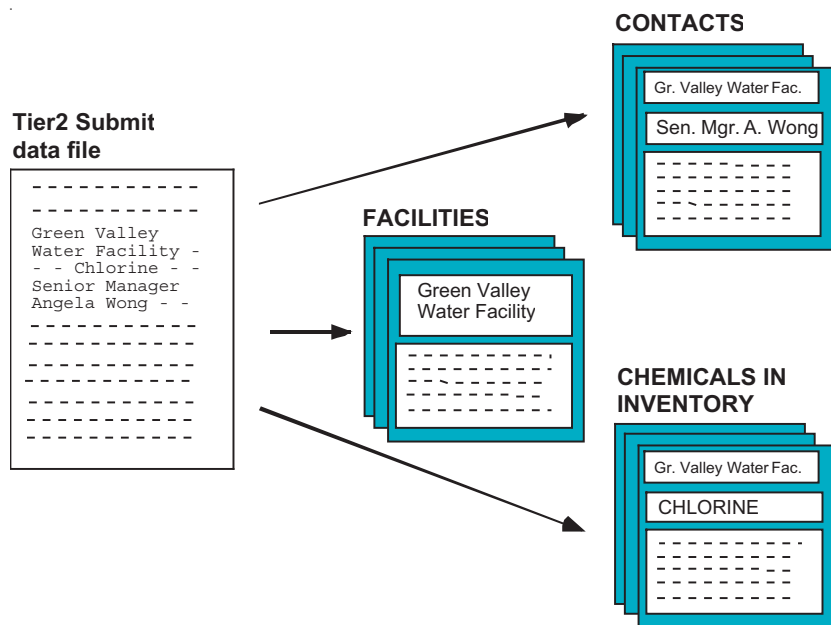


FIGURE 6-1. Destinations of Tier II data in CAMEO. Data from a Tier2 Submit file is imported into the Facilities, Chemicals in Inventory, and Contacts modules.

Note: The procedure for importing a Tier2 Submit file is unaffected by how many facilities are represented in the file. A Tier2 Submit file submitted by the operator of a reporting facility would generally contain information for just a single facility. In contrast, a Tier2 Submit file generated by a SERC to share with local planners and responders would contain data on multiple facilities.

Things to know before importing Tier II data

If your copy of CAMEO already contains records for some of the facilities, chemicals in inventory, and/or facility contacts included in the Tier2 Submit file (either because you created those records manually or because you imported a Tier2 Submit data file in a previous reporting year), new records for all those facilities, chemicals, and contacts will be created when you import the Tier2 Submit file. The existing records will not be updated. That's because Tier2 Submit generates its own record IDs when someone enters information in that program. Those record IDs are imported into CAMEO when the Tier2 Submit file is imported, and they will not match the record IDs that CAMEO has assigned to the corresponding Facilities, Chemicals in Inventory, or Contacts records.

Given CAMEO's record-keeping method, there are two ways for you to manage your Tier II data in CAMEO, each with advantages and drawbacks:

First, you can simply import new Tier2 Submit data file(s) for the facilities of interest to you each year, following the steps in [“Importing a Tier2 Submit data file” on page 210](#). Then, for each reporting facility, *for each reporting year*, there will be one record in your Facilities module, one Chemicals in Inventory record for each chemical (or mixture) in the facility's inventory, and a new Contacts record for each facility contact. Note that there will be multiple Contacts records for the same contact person if multiple facilities have reported the same contact person.

Because reporting year is shown in the Report Year box on records in the Facilities and Chemicals in Inventory modules, you can differentiate among records that were created in different reporting years in these modules.

If you have facility records from multiple reporting years, and want to work with your Facilities records for just the current reporting year,

1. Open the Facilities module and then click the button or select Start Search from the Search menu.
2. Type the current reporting year in the Report Year box, then click Search. The found set for this search will contain just the Facilities records for the current reporting year.

You can use the same procedure in the Chemicals in Inventory module to find just the Chemicals in Inventory records for the current reporting year.

If you want to see the record for a particular facility for the current reporting year,

1. Open the Facilities module and then click the button or select Start Search from the Search menu.
2. Type the facility's name in the Facility Name box.
3. Type the current reporting year in the Report Year box, then click Search. Only the record for the given facility for the current reporting year will be found.

If instead you want to view the data for a particular facility from all consecutive reporting years,

1. Open the Facilities module and then click the button or select Start Search from the Search menu.
2. Type the name of the facility in the Facility Name box, then click Search. The found set for this search will contain only the records for the given facility, for all reporting years.
3. To sort these records by date, follow the instructions in [“Sorting records by a single criterion” on page 255](#) to sort the found set by Record Year.

Reporting year is not shown on Contacts records. To access the most recent Contacts record for a facility contact person, begin in the Facilities module rather than the Contacts module:

1. Open the most recent Facilities record for that facility in Record view.
2. Click the Contacts tab.
3. Double-click on the name of the contact person. The most recent Contacts record for that contact will be displayed.

Note: An alternative procedure is to select or open the most recent Facilities record for the facility, then, from the Record menu, select Show Related, then Contact Records.

An alternative way to manage your Tier II data is as follows: Rather than keeping data from multiple reporting years in your copy of CAMEO, you could delete all the records for previous reporting years and then import the current year's Tier2 Submit file(s). When you then import Tier2 Submit data file(s) for facilities of interest submitted in the current reporting year, then for each reporting facility, there would be just one record in your Facilities module, just one Chemicals in Inventory record for each chemical (or mixture) in the facility's inventory, and just one Contacts record for each contact person for that facility (unless multiple facilities have listed the same contact person).

Important: Do *not* choose this approach if you foresee needing facility-related data from previous reporting years in any of the following modules: Facilities, Contacts, Chemicals in Inventory, Incidents, and Screening & Scenarios. For example, if you want to keep Incidents records for accidents at a particular facility over a period of years, avoid this approach.

If you choose this approach and want to back up your Tier II data from previous year(s) and then delete those records from CAMEO, before you import Tier2 Submit data file(s) for the current reporting year:

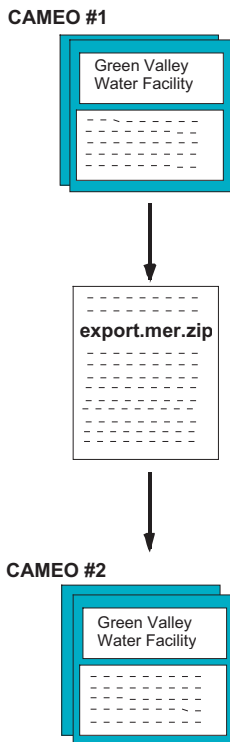
1. Start CAMEO if it isn't already running, and open the Facilities module.
2. From the File menu, select Import\Export, then click Export.
3. Click Export Data for Entire Module. When you export data from the Facilities module, data also is exported from related records in Contacts, Chemicals in Inventory, Incidents, and Screening & Scenarios. The result of the export is a single zipped file containing a set of merge files.
4. When prompted, choose a location for the export file. A zipped file containing your exported data will be created at that location. We recommend that you make a second copy of the archive file on another storage device.
5. While the Facilities module is still open, select Show All Records from the Record menu.

6. From the Record menu, select Delete All Records. This deletes all records from the Facilities module. Records related to the deleted facilities in the Chemicals in Inventory, Contacts, Incidents, and Screening & Scenarios modules are also deleted (except for any Contacts records that are also related to other entities that are not deleted, such as Resources records).

Importing a Tier2 Submit data file

Follow the steps below to import a Tier2 Submit file:

1. Start CAMEO.
2. Back up your facilities-related data, following the procedure in [“Exporting data from all records in a CAMEO module” on page 212](#).
3. From the File menu, select Import\Export.
4. Click Import, then click Import again when reminded to back up data.
5. Find and open the Tier2 Submit file. The import will start immediately.



Note: If you accidentally import the same Tier2 Submit data file more than once, don't worry. Each of the newly imported records will simply be overwritten, rather than duplicated.

Transferring data between different copies of CAMEO

In some situations, you might need to transfer data between your own copy of CAMEO—the **source** copy—and someone else's—the **target** copy.

To transfer data from your copy of CAMEO to someone else's, you export the data from your own records into an export file. Your colleague then imports the data from that file into his or her copy of CAMEO, as at left. If instead you receive data from your colleague, you would import his or her export file.

Note: You would follow the same procedure if you wanted to transfer data from one of the two previous versions of CAMEO into the latest version.

You can export data

- **from all your CAMEO modules at once, except for the Chemical Library.** It's the only CAMEO module from which you can't export data; you also can't import data into the Chemical Library.
- **from a single CAMEO module.** Data also will be exported from related records in child and grandchild modules. For example, when you export Facilities records, related records in Chemicals in Inventory, Contacts, and Screening & Scenarios also are exported.
- **from a found set of records in a single CAMEO module** (that is, a set of records found during a search). Data also will be exported from related records in child and grandchild modules (for example, from Chemicals in Inventory, Contacts, and Screening & Scenarios along with Facilities).
- **from a single CAMEO record** (as well as from related records in child and grandchild modules).

When you transfer data between two copies of CAMEO, it's possible to unintentionally duplicate records. Before transferring data, review [“Data transfers and record IDs” on page 203](#).

Exporting data from all CAMEO modules

You sometimes might want to export all the data from all your CAMEO modules into a single export file. This procedure is an easy way to create a backup archive file of all your CAMEO data.

To export all the data from all your CAMEO modules *except* the Chemical Library, follow the steps below:

1. From the File menu, select Import\Export, then click Export.
2. Click Export ALL CAMEO data.
3. Select a location for your export file, then click Save. All the data from all the records in all your CAMEO modules (except the Chemical Library) will be exported to a single zipped file at that location.

If some of the exported records are linked to map objects that you want to export, follow the procedure in [“Transferring map links and map objects” on page 215](#).

If some of the exported Facilities records are linked to site plans that you want to export, follow the procedure in [“Transferring site plans” on page 216](#).

Exporting data from all records in a CAMEO module

Sometimes, you might need to transfer all the data from just one module, along with data from related records in any child or grandchild modules. You might want to do this, for example, if you represent a fire district that has just entered into a mutual aid agreement with neighboring districts. If each district has in its copy of CAMEO only records for facilities in that district, then each district could export all its facility data, and share that export file with neighboring districts. Under the CAMEO Data Ownership Rule, each district would “own” the data for facilities in that district. Note, though, that if each district has records for facilities not only in its own district but also in neighboring districts, each district would instead export just the subset of its records containing only the facilities in its own district (see [“Exporting data from a subset of the records in a module” on page 213](#)).

To export all the data from a CAMEO module:

1. Open the module from which you want to export data. If you want to export facility data, open the Facilities module. To export transportation routes data, open Routes. Whenever you export data from either of these modules, related data in the Chemicals in Inventory, Contacts, Incidents, and Screening & Scenarios modules also is exported.
2. From the File menu, select Import/Export, then click Export.
3. Click Export Data for Entire Module.
4. Select a location for your export file, then click Save. All the data from all the records in the open module, along with all data from related records in child and grandchild modules, will be exported to a single zipped file at that location.

If some of the exported records are linked to map objects that you want to export, follow the procedure in [“Transferring map links and map objects” on page 215](#).

If you are exporting Facilities records that are linked to site plans that you also want to export, follow the procedure in [“Transferring site plans” on page 216](#).

Exporting data from a subset of the records in a module

Sometimes, you might need to transfer data from only certain records in a CAMEO module, along with data from related records in any child or grandchild modules. Here’s an example: Tier II data is particularly useful to emergency responders. So if you’re a member of a SERC that has received Tier II data, you might want to share that data with county- and tribal-level response organizations and/or with fire departments. You might want to export only certain records to each receiving organization: for example, just the records for King County to that county’s emergency management organizations.

Exporting from a subset of records can be a way to honor the CAMEO Data Ownership Rule. When you need to export records, you can search a module to create a found set containing just the records you “own” (for example, just your fire district’s records, rather than all your facility records). You then export just the found set.

To export only a subset of records from a CAMEO module:

1. In the module, create a found set containing just the records you want to export.

To do this, search the module in such a way that only the records you want will be found. For example, to create a found set in the Facilities module that contains just the records for facilities in Johnson County, choose Start Search from the Search menu, type “Johnson” in the County box, then press Search. (For more details about setting up searches, see [“Searching” on page 246](#))

If you want to export facility data, open the Facilities module. To export transportation routes data, open Routes. Whenever you export data from either of these modules, related data in the Chemicals in Inventory, Contacts, Incidents, and Screening & Scenarios modules also is exported.

2. From the File menu, select Import\Export, then click Export.
3. Click Export Data for Found Set.
4. Select a location for your export file, then click Save. All the data from all the records in the found set, along with all data from related records in child and grandchild module, will be exported to a single zipped file at that location.

If some of the exported records are linked to map objects that you want to export, follow the procedure in [“Transferring map links and map objects” on page 215](#).

If some of the exported records are Facilities records that are linked to site plans that you also want to export, follow the procedure in [“Transferring site plans” on page 216](#).

Exporting data from a single CAMEO record

You can export the data from just a single CAMEO record, along with data from related records in child and grandchild modules:

1. Select the record from which you want to export data.
2. From the File menu, select Import/Export, then click Export.
3. Click Export Data for Current Record.
4. Select a location for your export file, then click Save. The data from the record, along with all data from any related records in child and grandchild modules, will be exported to a single zipped file at that location.

Transferring map links and map objects

Sometimes, you might need to export records that are linked to objects on your CAMEO map. You might need to share with a colleague not only the data from those records, but the map objects as well.

To export all the records, map links, and map objects linked to records in a CAMEO module,

1. Export the records, following the instructions in [“Exporting data from all records in a CAMEO module” on page 212](#). The exported file contains the data from your records, including the map links from linked records, but not the linked map objects themselves.
2. Open the module (if it isn’t already).
3. From the Sharing menu, select MARPLOT, then Show All on Map.
4. The map objects linked to records in this module will be selected and displayed.
5. While the objects remain selected, from MARPLOT’s File menu, select Export.
6. Set up the export by clicking “selected objects” and “MARPLOT Import/Export (MIE),” then click Export. An MIE file containing all the selected objects is created. MIE files are a kind of text file that MARPLOT uses for importing and exporting data.
7. Give your colleague both the file you exported from CAMEO and the MIE file.

If you need to export just a subset of the records in a CAMEO module (for example, just the records you own under the CAMEO Data Ownership Rule, not the records you obtained from someone else) along with related map links and map objects, then

1. Follow the steps in [“Exporting data from a subset of the records in a module” on page 213](#).
2. While the found set is still in place, from the Sharing menu, select MARPLOT, then Show All on Map. Map objects linked to the found records will be selected in MARPLOT.
3. Follow steps 4 to 7 of the preceding procedure.

To import your data, your colleague should do the following:

1. Back up the data in his or her copy of CAMEO, following the instructions in either [“Exporting data from all CAMEO modules” on page 211](#), [“Exporting data from all records in a CAMEO module” on page 212](#), or [“Exporting data from a subset of the records in a module” on page 213](#). The exported file serves as an archive file.
2. Put the CAMEO file and MIE file that you exported anywhere on his or her hard drive.
3. From CAMEO’s File menu, select Import\Export, then Import.
4. Click on the name of the exported CAMEO file, then click Open. The data from the file will be imported into his or her copy of CAMEO.
5. Start MARPLOT.
6. In MARPLOT, from the File menu, select Import.
7. Find the MIE file, click on it to select it, then click Open. The map objects are then added to his or her CAMEO Map (if he or she does not have the map layer for the objects, a layer will be created). Map links between the imported records and map objects will now be functional in his or her copy of CAMEO.

Important: Follow the CAMEO Data Ownership Rule when you transfer map objects (see [“Avoiding trouble” on page 197](#)).

Transferring site plans

To share Facilities or Special Location records that contain links to site plans with a colleague, along with the site plans themselves,

1. Export the records, following the instructions in either [“Exporting data from all records in a CAMEO module” on page 212](#) or [“Exporting data from a subset of the records in a module” on page 213](#). The exported file contains the data from your records, including the links to site plans, but not the site plans themselves. The site plans are files stored inside the SitePlans folder.
2. Give your colleague both the exported file of facility and special locations data and your site plan files.

To import your data and site plans, your colleague should do the following:

1. Back up his or her facilities and special locations related data, following the instructions in either [“Exporting data from all records in a CAMEO module” on page 212](#) or [“Exporting data from a subset of the records in a module” on page 213](#). The exported file serves as an archive file.
2. Back up the SitePlans folder (in case any of the new site plan files have the same names as files already in the folder).
3. Put the exported CAMEO file anywhere on his or her hard drive.
4. From CAMEO’s File menu, select Import\Export, then Import.
5. Click on the name of the exported file, then click Open. The file then will be imported into his or her copy of CAMEO.
6. Place your site plan files inside his or her SitePlans folder. The links from Facilities and Special Location records to site plans will now be functional.

Important: Follow the CAMEO Data Ownership Rule when you transfer site plan files (see [“Avoiding trouble” on page 197](#)). One possible way to keep track of the owner of a particular site plan file is to include a code in the file name to indicate the owner. For example, you could use “FD1_” as the prefix in the names of all site plan files owned by Fire District 1.

Importing data from another copy of CAMEO

To import a data file exported from someone else’s copy of CAMEO, first check to be sure that the CAMEO Data Ownership Rule is being followed (see [“Avoiding trouble” on page 197](#)). That is, be sure that the exporter of the data is its “owner.” Then,

1. Back up the data in all of the CAMEO modules into which you will import data: if you are importing data into multiple CAMEO modules, follow the instructions in [“Exporting data from all CAMEO modules” on page 211](#). To back up a single module, follow the instructions in [“Exporting data from all records in a CAMEO module” on page 212](#). Your exported file serves as an archive file.
2. Place the export file anywhere on your hard drive.
3. From CAMEO’s File menu, select Import\Export, then Import.

4. Click on the name of the export file, then click Open. The file then will be imported into your copy of CAMEO.

If you need to import map objects, follow the procedure described for the data recipient in [“Transferring map links and map objects” on page 215](#).

If you need to import site plans, follow the procedure described for the data recipient in [“Transferring site plans” on page 216](#).

Strategies for avoiding duplicate records

The key to successfully exchanging data between two copies of CAMEO—so that all parties obtain just the data they want and no data they don’t want—is to honor the CAMEO Data Ownership Rule (see [“Avoiding trouble” on page 197](#)). Below are additional troubleshooting suggestions, in case a procedure goes awry despite your best efforts at honoring the rule.

Things to do before importing data. The following two strategies, which you can adopt *before* you import data, can help you to avoid unintended duplicate records.

You sometimes might know or suspect that duplicate records will be generated when you import data from a colleague. For example, you might have obtained facility data from one colleague, but later realize that a different colleague is the actual owner of that data. Your records and the owner’s records for the same facilities then are likely to have different record IDs, so if you import the owner’s data for those facilities, you’re likely to obtain duplicate records (see [“How CAMEO assigns and uses record IDs” on page 202](#)).

A solution that you might choose is to delete all the records from your module before you import the owner’s data. Do this *only* if it’s acceptable to you to replace the entire contents of one of your CAMEO modules (along with related records in child and grandchild modules) with new, imported data. When you import the new file, new records will automatically be created.

Important: Do *not* choose this approach if you foresee needing any of your existing facility-related data in any of the following modules: Facilities, Contacts, Chemicals in Inventory, Incidents, and Screening & Scenarios. For example, if you have created Incidents records for accidents at a particular facility over a period of years, avoid this approach.

To delete all the records from a CAMEO module in preparation for importing data,

1. Follow the directions in [“Exporting data from all records in a CAMEO module” on page 212](#) to back up the data in that module as an export file. We recommend that you also make a copy of the export file on another storage device.
2. From the Record menu, select Delete All Records to delete all the records in the open module (along with related records in child and grandchild modules).

Once you’ve completed these steps, you’re ready to import the new data. Follow the procedure in [“Importing data from another copy of CAMEO” on page 217](#).

Sometimes, you might need to replace only certain records in a module, rather than all the records in a module. For example, you might have obtained facility records for Fire District 20 from a colleague, but then realized that another colleague is the actual owner of the facility data for Fire District 20. Your records and the owner’s records for the facilities in Fire District 20 then are likely to have different record IDs, so if you import the owner’s data for those facilities, you’re likely to obtain duplicate records.

A solution that you might choose *only* if it’s acceptable to you to replace all your records for facilities in Fire District 20 with new, imported data is to delete those records from your module before you import the owner’s data. When you import the new file, new records will automatically be created.

Important: Do *not* choose this approach if you foresee needing any of your existing facility-related data in any of the following modules: Facilities, Contacts, Chemicals in Inventory, Incidents, and Screening & Scenarios.

You could simply browse through your Facilities records, deleting those that are to be replaced (related records in child and grandchild modules also will be deleted). To delete a record, select it or open it in Record view, then select Delete Record from the Record menu.

However, whenever you can, run a search that finds just the records that you need to delete, and then delete them, as follows:

1. Open the module in which you want to delete and then replace records. If you need to delete and replace facilities data, open the Facilities module.
2. From the Search menu, select Start Search. Set up and run a search that will find only the records to be replaced. In the case of the example above, you would make an Advanced Search for facility records with “20” in the Fire District field. For help with setting up and running searches, see [“Searching” on page 246](#).
3. Check over the set of found records to be sure that it contains *only* the records to be replaced. When a found set has been created, you can browse through only the records in that found set. So if you see any records that you don’t intend to delete, you’ll know that you need to refine your search.
4. From the File menu, select Import\Export, then click Export.
5. Click Export Data for Found Set. Before deleting the found records, you’ll back up the data from them into an archive file.
6. Choose a location for the new archive file. A zipped file of your exported data will be created at that location. We recommend that you also make a copy of the archive file on another storage device.
7. From the Record menu, select Delete All Found Records. Selecting this command while a found set exists in the module deletes only the found records, leaving the other records in the module untouched.

Once you’ve deleted the records to be replaced, you’re ready to import your new data. Follow the procedure in [“Importing data from another copy of CAMEO” on page 217](#).

Things to do after importing data. If you've already imported data into a CAMEO module, and now realize that you've inadvertently duplicated some records, you may be able to remove unwanted duplicates in one of the following ways. These methods will work *only* if you can tell by visual inspection which duplicates to keep and which to delete.

- Follow the instructions in [“Sorting records in CAMEO modules” on page 251](#) to sort the records in the module so that duplicate records appear next to each other in List view. For example, sort the Facilities module alphabetically by the “FacilityName” data field in order for duplicate records for the same facility to appear adjacent to each other in the list. Browse through the list to find duplicates. Inspect the duplicate records for a given facility or other entity to identify the record to keep. To delete an unwanted duplicate record, select it, then select Delete Record from the Record menu.
- If possible, before you look for duplicates, first run a search to create a found set of just the records likely to be duplicates. For example, you may have realized that you've inadvertently duplicated just the records for contact people in Haymarket. In this case, you would search the Contacts module for records in which the City field contains “Haymarket.” Once you've completed the search, from the Record menu, sort the found records as above, then select View List. You then can most quickly browse through the names of all the records found in the search to identify duplicates.

Important: Before beginning either procedure, back up the data in the module by following the procedure in [“Exporting data from all records in a CAMEO module” on page 212](#).

Exporting data to other programs

You can open and view a data file exported from CAMEO in any spreadsheet or database program that supports merge files.

To do this,

1. Uncompress (unzip) the export file. This file contains one or more merge files (one for each CAMEO module from which data was exported, along with a merge file for any supporting files from which data also was exported).
2. Open any of the merge files in your spreadsheet or database program. In merge files, data fields are separated by commas, records are separated by carriage returns, and the contents of each data field are enclosed in quotation marks. A merge file has a file extension of .mer (for example, “export.mer”).

For more information about CAMEO’s export file format, see [“Zipped merge files: CAMEO’s standard file format” on page 201](#).

Tip: To open a merge file in some database or spreadsheet programs, you may need to rename it so that its file extension is .txt rather than .mer.

Managing Your CAMEO System

This chapter describes how to create a management plan for your CAMEO system, and proposes elements to include in that plan. It also explains how to implement CAMEO on a network, how to protect your CAMEO files, and what to do when multiple people use the same CAMEO system.

A CAMEO management plan

Before you begin to use CAMEO, consider how you intend to manage and use your new system. In this section, we describe some of the issues to consider and guidelines to follow as you prepare a CAMEO management plan for your workplace.

As you plan how you'll manage your CAMEO system, consider your intended uses of all three CAMEO components: the CAMEO modules, MARPLOT, and ALOHA.

Why make a CAMEO management plan?

It's important to plan how you intend to manage and use your CAMEO system before you dedicate resources to implementing your CAMEO, ALOHA, and MARPLOT programs, because

- A good management plan can help you to address issues before they become a problem.
- Your plan can prepare you for requesting additional resources as you require them, developing a training program, and handling future software and hardware developments.
- You can use your plan as a guide for implementing CAMEO.

Elements of a good CAMEO management plan

An effective CAMEO implementation goes beyond installing the programs, flipping through the manuals, searching for a few chemical records, and placing a few symbol objects on a MARPLOT map. To effectively implement CAMEO, you'll need to plan in advance your procedures for entering, maintaining, and using information, protecting and backing up both data and programs, training personnel, and customizing your CAMEO system. Your management plan should also address your equipment and personnel needs. An effective management plan can make CAMEO and your computer more useful tools.

Address the topics discussed below as you draft your CAMEO management plan.

Proposed uses of CAMEO. Describe your planning and response duties, and any other intended uses of your computer system.

Hardware and software. Describe your hardware inventory, including computer models, operating systems (e.g., Windows 2000, Macintosh OS X, and so on), hard drive sizes and available memory (RAM), and other hardware, such as modems, storage devices, and printers. Describe your local network if you have one.

Describe your software inventory. List the word-processing, graphics, spreadsheets, database and other programs that you intend to use with your CAMEO system.

Describe your inventory of MARPLOT maps, including areas of coverage.

If you're connected to the Internet, list the addresses of websites where you can obtain resources and information you plan to use with your CAMEO system (such as the MARPLOT maps page at <http://www.epa.gov/osweroe1/content/cameo/marmaps/>).

CAMEO system manager. Describe the position and responsibilities of the system manager. Include his or her name, telephone number(s), and email address.

CAMEO information management. Outline the procedures you'll use to enter information into CAMEO and onto the maps you'll use with CAMEO. Describe procedures for making data linkages both within CAMEO and between CAMEO and MARPLOT, and information-tracking procedures. Propose data element standards, such as acceptable common abbreviations (for example, "St." for Street).

Describe your strategies for maintaining, tracking, and updating information. Describe your method for backing up your CAMEO system, and include a schedule (see "[Implementing a CAMEO backup program](#)" on page 229).

If you plan to import Tier II data into your CAMEO system, see "[Importing data from Tier2 Submit](#)" on page 205.

Training. Describe who will be trained and the purposes and goals of your CAMEO training program. Discuss word processing, graphics, and other skills required of trainees. Outline your proposed training program and list the subjects that you intend to cover. Describe the resources that you need to conduct your program. Include a training schedule.

CAMEO changes. Describe any modifications that you propose to make to your CAMEO system. For example, you might add items to some of the editable pull-down menus in CAMEO modules. Plan how you'll keep track of the changes you make to CAMEO, so you'll be able to update future versions of the program.

Program security. Describe your method of ensuring program security and any procedures that you intend to use to restrict access to the program.

The CAMEO system manager

An important decision affecting your CAMEO system will be selection of the system manager. Designate one person as the system manager. This person should be experienced in using CAMEO as well as computer operating system and office software.

The responsibilities of the system manager include:

- receiving and installing program updates,
- troubleshooting,
- directing the training program,
- ensuring that backups are made,
- supervising modifications to CAMEO,
- directing data entry,
- ensuring program security, and
- developing data element standards.

Some components of these tasks may be delegated to others. In essence, the system manager should be responsible for maintaining your CAMEO system and ensuring that it is used as effectively as possible. The importance of this position cannot be over-emphasized. If your system manager is not in tune with the CAMEO system, its needs, and your needs, then CAMEO may never prove to be a useful tool for you.

Preparing a plan for entering and updating data

The quality of your CAMEO system depends on the quality of the data that you enter into your system. Before you enter data, develop a data entry plan. As you develop your plan, try to answer the following questions:

1. What information do you intend to enter into CAMEO? Identify the information that can help you to perform your emergency planning and response duties.
2. How will you keep track of the information that you enter? We recommend that you bear the CAMEO Data Ownership Rule in mind as you consider this topic (see [“Avoiding trouble” on page 197](#)).
3. Do you need to develop data standards? You may want to standardize information such as facility identification numbers, abbreviations, the symbols placed in MARPLOT, and other information.
4. Do you need to develop step-by-step procedures for entering information into CAMEO? An advantage of preparing written, step-by-step procedures for entering and maintaining information is that you then can refer to these procedures during training sessions and when entering, maintaining, and verifying CAMEO information. The more you can standardize your data entry system, the better the quality of your CAMEO data, and the lower the risk of errors associated with entering multiple records for the same facility, organization, or other entity.
5. Who will be responsible for entering information? Assign data entry and maintenance responsibilities only to trained and responsible staff members. They should follow the procedures that you develop. If several people will enter and maintain your CAMEO data, be sure to develop and use an effective tracking system to minimize errors.
6. How and how often do you plan to verify or update the information in your CAMEO system? Plan to verify and update your CAMEO information regularly. Establish a schedule for verifying and updating data and stick to it (you might want to schedule data verification and updating tasks around regular fire department inspections or important EPCRA-related dates, such as reporting deadlines).

If you plan to import data into CAMEO (including Tier II data) or transfer data between copies of CAMEO, review [Chapter 6, “Importing and Exporting Data,”](#) beginning on page 197.

Finally, consider how you might enhance CAMEO to better meet your data entry and management needs.

Program security

The success of your CAMEO program will depend on your ability to ensure the security of your CAMEO system.

Your CAMEO system, and the information that it contains, can easily be modified or damaged. When program security is inadequate, a novice user or failed hard drive can jeopardize your entire CAMEO system. Only well-trained, authorized users should have full access to CAMEO. You may want to limit access by other users. Such access restrictions can help to prevent unacceptable CAMEO system modifications.

There are two kinds of security concerns:

- program access, and
- program backup.

As you address program security in your management plan, ask yourself:

- Who should have access to my CAMEO system?
- What methods for restricting access to CAMEO should I use (if any)?
- How will I back up my system? Who will perform backups, how, and how often? (Backing up your CAMEO data is easy and quick; see [“Implementing a CAMEO backup program”](#) on page 229.)

Password protection

One way to restrict access to your CAMEO files is to implement password protection.

All of the CAMEO files are already password-protected, but because the passwords for these files are set as defaults, users are not asked for a password when they work with CAMEO. You can change CAMEO's passwording so that people would need to enter a password to work with your CAMEO modules. To do this, you would need to purchase a copy of FileMaker® Pro 8 or FileMaker® Pro 8 Advanced.

Especially before you share your CAMEO files over a network, we recommend that you change the password(s) for your CAMEO files. Change the passwords both for the administrator and for normal users, and delete the default passwords. Once you change the password(s), people would need to enter a password to use the passworded CAMEO files. To learn how to modify passwords and delete default passwords in FileMaker files, including CAMEO, check the FileMaker Pro manual.

Once you make this change, any CAMEO user would need to enter a password before being able to access the data or change CAMEO records.

Other ways to use passwording to restrict access to CAMEO include:

- Requiring a password to log onto the computer where CAMEO is installed.
- Using a password screensaver on that computer.

Implementing a CAMEO backup program

You can quickly and easily create backup archive files containing all your CAMEO data (except for the Chemical Library data), or just data from selected modules. Follow the procedures for creating export files in [“Transferring data between different copies of CAMEO” on page 210](#): the files that you export serve as your backup files.

We recommend that you:

- Export all your CAMEO data into an archive file on a regular basis.
- Keep at least the two most recent archive files on physically separate storage media. Consider keeping older backups in case problems remain undiscovered until after one or more backup cycles.

Installing and running CAMEO on a network

You can install and run the CAMEO modules on a network. You might want to do so if multiple people in your workplace need to use a single copy of CAMEO as a reference.

We recommend that MARPLOT and ALOHA be installed and run on individual users' local computers, rather than on the network. However, MARPLOT map files can be shared over a network. Check your MARPLOT manual to learn how to do this.

You can choose to network CAMEO in either of two main ways:

- From a dedicated server computer, to which every user connects over the network. This is your best choice if multiple people need to routinely use the same copy of CAMEO.
- From one user's computer, to which others on the network connect from time to time.

While we've completed some testing of CAMEO on networks, we could not emulate all possible network configurations. If you encounter problems installing or running CAMEO on your particular network, please check the CAMEO website (<http://www.epa.gov/osweroel/content/cameo/>) for troubleshooting help and resources. More user support resources are available on the FileMaker website (<http://www.filemaker.com/support/>). The comments below are provided as guidelines only.

Ensure security first

Carefully consider file security before you install CAMEO on a network and especially before you share sensitive CAMEO data, such as Tier II data, over a network. Above all, check to be sure that your network is secure, so that unauthorized people can't access your data. Consider password-protecting your CAMEO files to prevent unwanted access (see "[Password protection](#)" on page 228).

Possible configurations

You can network CAMEO and support multiple users in three ways. Which is most appropriate for you depends mainly on the number of concurrent CAMEO users you need to support. You can

- Install the runtime version of CAMEO on a network (this is the version of CAMEO that you can download and install). This configuration allows you to support just one CAMEO user at a time.
- Purchase and install a copy of FileMaker Pro 8 or FileMaker Pro 8 Advanced, which you use to host the CAMEO files. This configuration allows you to support a limited number of concurrent users, each of whom also must have a licensed copy of FileMaker Pro. The number of users who can work with CAMEO concurrently depends on how many files each has open at the same time; fewer people can use the program when each has more files open. With this configuration, up to five people can use CAMEO if each has all or nearly all CAMEO files open at the same time.
- Use FileMaker Server 8 or FileMaker Server 8 Advanced for the fastest-possible CAMEO networking. This configuration allows you to support up to 250 concurrent CAMEO users, each of whom must have a licensed copy of FileMaker Pro.

Installing the runtime version on a network

You can install the runtime version of CAMEO either on a network server or in a shared directory on a networked workstation without purchasing and installing any additional software. Just one person at a time can work with the runtime version of CAMEO over a network.

Using FileMaker Pro to host runtime files

The CAMEO files can be modified so that multiple people can work with them concurrently.

When FileMaker databases (such as the CAMEO modules) are set to be shared over a network, whoever first opens a particular database file (such as a CAMEO module) becomes the **host** of that file, and their computer becomes the server of the file. Other users can modify the file by connecting to the host computer.

From the CAMEO perspective, the main advantage of hosting files is that you can serve multiple concurrent users. The runtime version of CAMEO cannot host files, but you can use FileMaker Pro 8 or FileMaker Pro 8 Advanced to host the CAMEO files on your network so that multiple people can work with them. To accomplish this, you would install both FileMaker Pro and the CAMEO modules either in a shared directory on a networked workstation or on your network server computer. Others on the network who also have installed FileMaker Pro then can work concurrently with the shared copy of CAMEO (they do not need to have installed CAMEO on their computers).

In our own tests of FileMaker Pro hosting of CAMEO on local networks, we observed the following:

- Multiple users can view the same record in Browse mode at the same time.
- Multiple people also can open the same record in Edit mode. However, as soon as one person clicks his or her cursor within an editable field, the others are locked out of Edit mode; they see an explanatory message when this happens.
- Multiple people can perform basic searches of the same CAMEO module at the same time. Multiple people also can view the advanced search dialog at the same time. However, as soon as one person selects a field to use as a search criterion, saves a set of search criteria, or opens a saved set of search criteria, the others are locked out (and are shown an explanatory message).

Important: A key to trouble-free networking of CAMEO is to be sure that only one computer hosts all the CAMEO files. We recommend that you use a dedicated computer to host CAMEO, and that you keep the CAMEO files open on that computer all the time, to prevent other users from accidentally hosting any of those files.

Follow all the procedures below to set up CAMEO for FileMaker Pro 8 or FileMaker Pro 8 Advanced hosting.

Relocate plug-in files and indicate the directory path. Once you have installed CAMEO and FileMaker Pro on the computer that will host CAMEO, you must move some files from the locations where they were installed when you ran the CAMEO installer. You also must create a base directory file that FileMaker Pro will reference, as follows:

In Windows: Copy the following files from the Extensions folder inside the CAMEO folder into the Extensions folder inside the folder containing FileMaker (e.g., C:\Program Files\FileMaker\FileMaker Pro 8\Extensions):

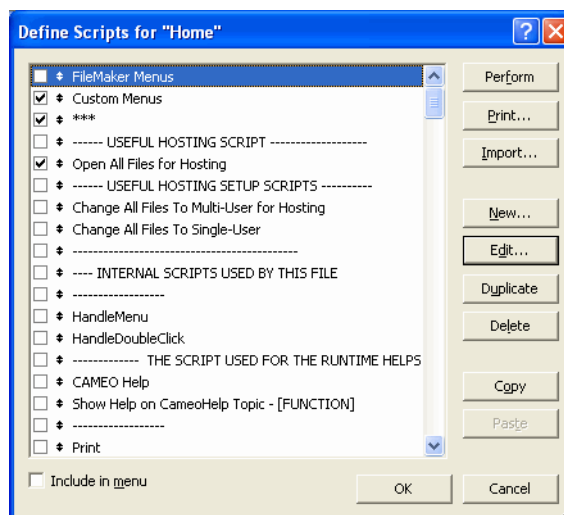
- Cameo.fmx
- Marplot.fmx
- Menu.fmx

On a Macintosh: Copy the following files from the Extensions folder inside the CAMEO folder into the Extensions folder inside the folder containing FileMaker (e.g., Macintosh HD:Applications:FileMaker Pro 8 Folder:Extensions):

- Cameo.fmplugin
- Marplot.fmplugin
- Menu.fmplugin

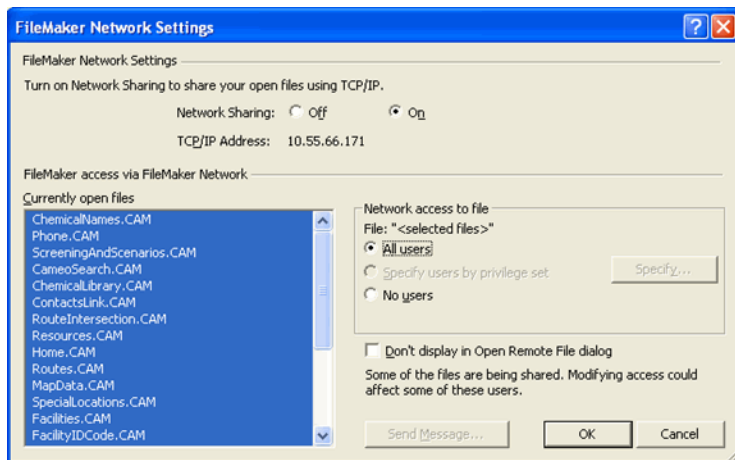
Prepare the CAMEO files to be shared. Once you have installed the software, then follow the steps below to prepare your CAMEO files to be shared by multiple, networked users. You'll only need to complete these steps once.

1. From FileMaker Pro's File menu, select Open.
2. If you're working in Windows, select "All Files" from the "Files of Type" list.
3. Hold down either the Shift key in Windows or the Option key on a Macintosh as you select Home.CAM (inside the CAMEO folder) and click Open.
4. When prompted for an account name and password, type **Admin** (case sensitive) in both places and then click OK.
5. From the Scripts menu, select ScriptMaker.
6. In the list of scripts, highlight (but do not check) the "Change All Files to Multi-User for Hosting," then click Perform to run the script.
7. From the Scripts menu, select ScriptMaker.
8. Click in the box next to "Open All Files for Hosting" (to check this box).



9. Highlight the FileMaker Menus line, then click Perform to run the script.
10. From the Edit menu, select Sharing, then FileMaker Network.

11. In the FileMaker Network Settings dialog box, turn Network Sharing on. Select all of the currently open files in the list and change the Network access to “All users,” then click OK.



12. Quit or Exit from FileMaker.

Prepare users' computers. You'll need to copy supporting files to the computer of each colleague who will work with the hosted copy of CAMEO.

In Windows: Copy the following files into the Extensions folder inside the folder containing FileMaker (e.g., C:\Program Files\FileMaker\FileMaker Pro 8\Extensions):

- Cameo.fmx
- Marplot.fmx
- Menu.fmx

On a Macintosh: Copy the following files into the Extensions folder inside the folder containing FileMaker (e.g., Macintosh HD:Applications:FileMaker Pro 8 Folder:Extensions):

- Cameo.fmplugin
- Marplot.fmplugin
- Menu.fmplugin

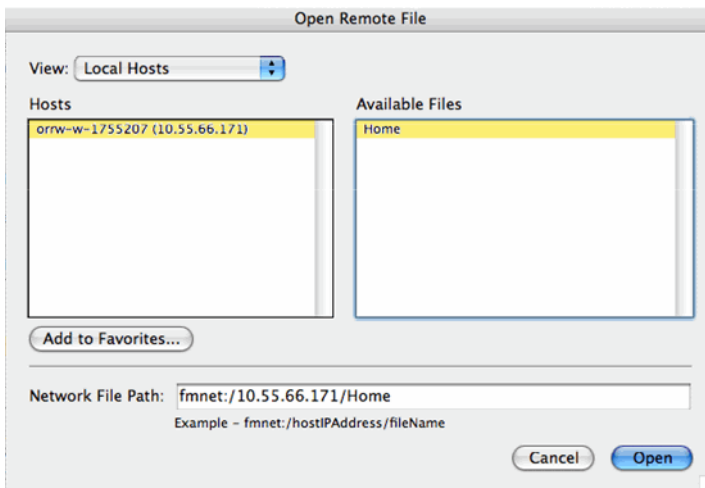
Begin hosting CAMEO. To begin hosting the CAMEO files over your network, follow the steps below. You must complete these steps each time you want to start hosting the CAMEO files from a computer:

1. On the computer that will be used to host the CAMEO files, start FileMaker Pro.
2. From the File menu, select Open.
3. Click Remote. The Open Remote File dialog box appears.
4. Review the list to be sure that Home does not appear in the list of available files for any of the local hosts. If it appears in the list, another computer is hosting this file. If so, ask that user to quit from FileMaker.
5. Click Cancel on the Open Remote File dialog box.
6. If you're working in Windows, select "All Files" from the "Files of Type" list.
7. Open Home.CAM.
8. From the Scripts menu, select "Open All Files for Hosting."

Accessing the hosted CAMEO. Once the CAMEO files are being hosted by a computer on your network, those colleagues on the network now can access CAMEO, as follows:

1. Start FileMaker Pro (on their computer).
2. From the File menu, select Open Remote. The Open Remote File dialog box appears.

3. Set the view to Local Hosts and highlight the host computer in the Hosts box. In the Available Files box, highlight Home.



4. Click Open.

Important: Your colleagues should open *only* the copy of Home on the host computer. If they see that Home is being hosted on any other computer, you or they should ask that user to quit or exit from FileMaker Pro.

Using FileMaker Server

You can use FileMaker Server 8 or FileMaker Server 8 Advanced rather than FileMaker Pro to host your CAMEO files. The advantages of this choice are that you can support many more concurrent users—up to 250 at a time—and CAMEO operations will proceed substantially faster.

If you choose this approach, you will need to install FileMaker Server and the CAMEO files on a networked workstation or on a dedicated server computer. Others on the network who also have installed FileMaker Pro (but not CAMEO) then can work with the shared copy of CAMEO, concurrently. For more details on FileMaker Server, check <http://www.filemaker.com>.

Note: We did not specifically test CAMEO hosting via FileMaker Server, but our best understanding is that if you choose this approach, you will need to prepare your files as explained in [“Using FileMaker Pro to host runtime files” on page 232](#).

Managing multiple users

We recommend that you plan your CAMEO projects so that multiple people do not routinely use CAMEO on your network at the same time. We especially recommend that you manage your workflow to avoid having multiple people working concurrently over a network to edit records and/or perform CAMEO operations that require the most computer power, such as searching the Chemical Library or adding records. The program needs time to complete scripts and respond to user commands, and CAMEO operations are slower over a network.

We recommend that you avoid importing and exporting data to or from a hosted copy of CAMEO via client computers over your network. Instead, whenever you or your colleagues need to import or export data, work either on the host computer or on a copy of CAMEO that is not networked.

Managing data entry by multiple people. You sometimes may need to have multiple people entering CAMEO data. For example, if you receive Tier II data from many facilities, you may need to give a stack of Tier II forms to each of several data entry operators who then would create new Facilities records to contain that Tier II data.

When you need to have multiple people entering data concurrently, we suggest that each person be given a copy of the runtime version of CAMEO,¹ into which he or she adds records, rather than having everyone making editing changes to a single, networked copy of CAMEO.

1. That is, a copy of CAMEO that has not been modified for multi-user hosting.

When each data entry operator has completed adding records, he or she should export a file containing those records, by following the procedure in either “[Exporting data from all records in a CAMEO module](#)” on page 212 if records have been added to just one module, or “[Exporting data from all CAMEO modules](#)” on page 211 if records have been added to multiple modules.

The data file exported by each data entry operator can then be imported into a central, networked copy of CAMEO (as shown in Figure 7-1), by following the procedure in “[Importing data from another copy of CAMEO](#)” on page 217.

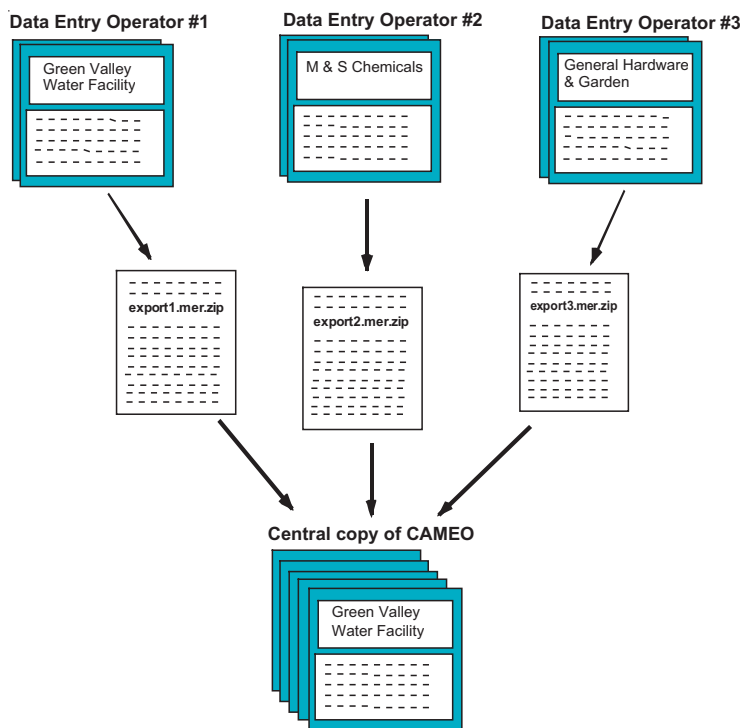


FIGURE 7-1. Managing data entry by multiple people: each operator exports a file of data that is imported into a central copy of CAMEO.

Using CAMEO if you already have FileMaker

When you follow the standard installation instructions (that is, not the network instructions), you install the runtime version of CAMEO. We recommend that as you work with CAMEO, you use the runtime version of CAMEO even if you already have a copy of FileMaker Pro 8 or FileMaker Pro 8 Advanced, *unless* you are hosting CAMEO on a network (see [“Installing and running CAMEO on a network” on page 230](#)). If you aren’t hosting the CAMEO files, there is no advantage to running CAMEO in FileMaker Pro rather than as a runtime program, because all of CAMEO’s features are available in the runtime version.

Reference Information

This chapter contains basic reference information about CAMEO: explanations of the functions of CAMEO’s menu commands, explanations of how to search for and sort records in CAMEO modules, an overview of the legislation on which CAMEO is based, and a bibliography.

CAMEO’s menus

Below are explanations of the functions of each of CAMEO’s menu commands.

TABLE 8-1. CAMEO’s Menus and Menu Commands.

File menu commands:	
Go Home	Opens the Home navigation window (which contains buttons to open CAMEO’s modules and perform common tasks). See “Home and navigation” on page 27 .
Open	Select Open , then point to any module name to open that module.

TABLE 8-1. CAMEO's Menus and Menu Commands. (Continued)

Close	Close the current window.
Import\Export	Displays controls you use to import or export data to or from CAMEO. See “Importing and Exporting Data” on page 197.
Print Report	Displays controls you use to create and print reports.
Quit (Macintosh) or Exit (Windows)	Quits or exits from CAMEO.
<p>Edit menu commands: The Edit menu contains standard FileMaker commands, most of which work in some but not all places in CAMEO (e.g., text-editing commands work within text boxes in CAMEO, but not within lists of records).</p>	
Undo	Undoes your most recent text-editing action.
Cut	Removes selected text from a text box, and places it in the Clipboard.
Copy	Copies selected text into the Clipboard.
Paste	Pastes the contents of the Clipboard into the currently selected text box.
Clear	Deletes selected text from a text box.
Select All	Selects all the text with in a text box in which you have clicked.
Spelling	Checks the spelling of text in text boxes and Notes fields: select the text, then select Spelling, then Check Selection, from the Edit menu. You can check spelling in either Browse or Edit mode, but you can replace misspelled words only in Edit mode.
Preferences	In Windows, you can use a checkbox on the General tab to set a larger window size for increased readability. You should not adjust any other preferences settings.
<p>Record menu commands:</p>	
View List	Displays a list of the records in the current module, in List view. See “List and Record views” on page 29. Available in Record view.

TABLE 8-1. CAMEO's Menus and Menu Commands. (Continued)

View Record	Displays the selected record in Record view (in List view, click on a record to select it; you'll then see a small black rectangle next to its name). See "List and Record views" on page 29 . Available in List view.
Next Tab	Brings the next tab forward. Available in Record view only.
Show Related	Displays a submenu listing all modules that could be related to the current record. Choose a module name to view all the records in that module that are related to the current record, in List view. See "Using the Show Related command" on page 30 . Available in all modules except the Chemical Library.
Sort	Choose to sort the records in the current module (or found set) alphabetically, numerically, or chronologically, by one or more data fields. See "Sorting records in CAMEO modules" on page 251 .
Show All Records	Choose to view all the records in the current module, rather than just the found set of records collected during your most recent search or other activity. You would choose this command after you had finished working with a search.
New <module name>	Creates a new record in the current module (e.g., in the Facilities module, choosing New Facility creates a new Facility record). Available in all modules except the Chemical Library.
Edit <module name>	Allows you to edit the current record (e.g., while you're viewing a record in the Incidents module, choose Edit Incident in order to edit that record). Available in all modules except the Chemical Library.
Delete <module name>	Deletes the current record, along with subordinate ("child") records in other modules. (For example, when a Facility record is the current record, choosing Delete Facility deletes not only that record but also all Chemicals in Inventory, Screening and Scenarios, and Incidents records for that facility, as well as all Contacts records for that facility that aren't also associated with another facility or organization.) Available in all modules except the Chemical Library.

TABLE 8-1. CAMEO's Menus and Menu Commands. (Continued)

Delete All Records	Deletes all the records from the current module. Available in all modules except the Chemical Library. Available in List view only.
Delete All Found Records	Deletes all the records from the current found set. Available in all modules except the Chemical Library, when a found set exists. Available in List view only.
Add to Reactivity Worksheet	Adds the substance from the current Chemical Library record to the Reactivity Worksheet. See “Predicting potential chemical reactivity” on page 96 . Available only in the Chemical Library.
View Reactivity Worksheet	Displays the Reactivity Worksheet. See “Predicting potential chemical reactivity” on page 96 . Available only in the Chemical Library.
Go to	Displays a submenu of navigation commands: choose First to view the first record in the current module (or set of found records), Previous to view the record preceding the current one, Next to view the record immediately after the current record, and Last to view the last record in the module.
Search menu commands:	
Start Search	Initiates a search of the current module (e.g., in the Chemical Library, choose Start Search to find the record for a particular chemical). See “Searching” on page 246 .
Append Search	Choose when you want to broaden a search; that is, to search for more records than you found in a previous search. Any records found in an append search that aren't in the existing set of records found in your previous search will be added to that set of found records. See “Append searches and subset searches” on page 250 .
Subset Search	Choose when you want to narrow your search; that is, to search through a set of records you found in a previous search, in order to find those that meet more restrictive criteria. See “Append searches and subset searches” on page 250 .
Clear Search	Allows you to view all the records in the current module, rather than just the found set of records collected during your most recent search.

TABLE 8-1. CAMEO's Menus and Menu Commands. (Continued)

Sharing menu commands:	
ALOHA/Help on ALOHA	Displays an explanation of ALOHA and how it works with CAMEO. For more on ALOHA, see “Using ALOHA with MARPLOT and CAMEO” on page 192.
ALOHA/Select this Chemical in ALOHA	Available only in the Chemical Library. Starts ALOHA if it isn't already running, then loads the chemical from the current record into ALOHA if the chemical is in ALOHA's chemical library.
ALOHA/Go to ALOHA	Starts ALOHA if it isn't already running, and brings it forward.
MARPLOT/Help on MARPLOT	Displays an explanation of MARPLOT and how it works with CAMEO. For more on MARPLOT, see “Using MARPLOT with CAMEO” on page 181.
MARPLOT/Show on Map	If the current record is linked to a map object, brings MARPLOT forward with that object visible and selected. In Screening & Scenarios, displays the threat zone (if the associated facility or route is linked to a map object; see “Plotting a threat zone on a map” on page 151).
MARPLOT/Show All on Map	If multiple records in the current module or found set are linked to map objects, brings MARPLOT forward with those objects visible and selected.
MARPLOT/Unlink	If the current record is linked to a map object, unlinks that record and object. Always unlinks just one record at a time, even if multiple records have been found in a search. See “Deleting a map link” on page 191.
MARPLOT/Delete Scenario Object	Available only in Screening & Scenarios. Deletes the threat zone map object linked to the current or selected Screening and Scenarios record. See “Working with threat zones on maps” on page 152.
MARPLOT/Delete All Scenario Objects	Available only in Screening & Scenarios. Deletes all objects on the Scenarios layer in MARPLOT. See “Working with threat zones on maps” on page 152.
MARPLOT/Go to MARPLOT	Starts MARPLOT if it isn't already running, and brings it forward. The current or most recent map view is displayed.

TABLE 8-1. CAMEO's Menus and Menu Commands. (Continued)

LandView/Help on LandView	Choose this submenu command to read an explanation of LandView and how it works with CAMEO. See “LandView” on page 22 .
LandView/Go to LandView	Starts LandView if it's installed on your computer and isn't already running, and brings it forward.
Scripts menu commands:	
Custom Menus	Restores CAMEO's normal menus if they fail to display properly.
***	This command must appear in the menu for CAMEO to function normally, but choosing it doesn't initiate any actions.
Help menu commands:	
CAMEO Help	Opens and displays the CAMEO help table of contents in your web browser.
About CAMEO	Displays version information.
Link menu commands: This is a temporary menu that is only visible when you have (a) begun linking a MARPLOT map object to a CAMEO record (see “Linking map objects to CAMEO records” on page 187), or (b) chosen to adjust the link from a chemical inventory name to a Chemical Library record (see “Using the Adjust Link button” on page 130).	
Link this Record	Links the selected map object to the current CAMEO record.
Cancel	Cancels the link in progress.


Searching

You can make either a basic search or a more advanced search for the records you want to see in any CAMEO module, as explained below. You also can save the criteria you used for particularly complex searches so that you can rerun them again later.

You'll find more specific instructions about searching the Chemical Library in ["Searching for a Chemical Record" on page 80](#), and about searching for records on particular facilities in ["Accessing information about a facility" on page 111](#).

Making a basic search

Choose to make a basic search when you just need to look up records by name, address, or other straightforward piece of identifying information. Here's how to do a basic search:

1. To start your search, first open the module you want to search, then click the  toolbar button or select Start Search from the Search menu.
2. In the Basic Search dialog, fill out the information you have about the record you're looking for. For your "Operator for text fields," choose "Contains characters" to search for part of a word, phrase, or code; or "Contains word starting with" to search either for the first part of a word, phrase, or code, or for an entire word, phrase, or code. **Note:** If you fill in multiple criteria, CAMEO will search for records that match all your criteria.
3. Click Search to start your search. CAMEO will run the search and then display either a list of the records that match your criteria or a message, "No records found," if no records match your criteria.
4. Double-click the name of any record in the list to view that record. Whenever you want to return to the list, click the View List button in the toolbar. If you'd like to view the list of all records in the module again, from the Search menu, select Clear Search.

Searching State Fields

Both Facilities and Chemicals in Inventory records contain a State Fields tab to support the emergency planning work of U.S. states that have developed their own reporting requirements in addition to the Tier II requirements specified in EPCRA (which apply to all 50 states). Tier II forms submitted in those states contain additional data fields, which are called “State Fields” in CAMEO. CAMEO’s State Fields provide a place where people working in these states can maintain their reporting information.

You can search for records for facilities or chemicals in inventories that contain particular information in one or more State Fields. For example, the State of Arizona requires facilities to report whether they are located on Tribal Land. To search State Fields,

1. Open the module you want to search, then click the [New Search](#) toolbar button or select Start Search from the Search menu.
2. Click Search State Fields. You’ll see the window for basic state field searches.
3. If the report year and state shown in the window are different from the report year and state you want to search for, click Choose State and Report Year, change the report year and/or the state, then click Select. **Note:** You can only search for records with a particular report year and state at one time. If you need to search for records for more than one report year (e.g., for facilities on Tribal lands for which records were created in report years 2004, 2005, and 2006), run a search for one of those years, then run successive Append searches, one for each of the other years. If you need to search for records for more than one state, run a search for one of those states, then successive Append searches, one for each of the other states.
4. Fill out the information you have about the records you’re looking for. For example, if you’re searching for Arizona facilities located on Tribal Lands, search for records for which “Is the facility on Tribal Land?” is “Yes.” **Note:** If you fill in multiple criteria, CAMEO will search for records that match all your criteria.
5. Click Search to run the search.

Making an advanced search

You can make an advanced search whenever you need to use other criteria than the ones available for a basic search. Here's how:

1. Start your search just as you would start a basic search: in the module you want to search, either select Start Search from the Search menu or click the New Search toolbar button.
2. Click Go to Advanced Search.
3. Click Select field to indicate which data field to search. In some modules, you'll need to click a button to indicate what kind of information you're searching for.
4. In the list of searchable fields, click the name of a field, then click Select.
5. Choose an operator from the popup menu (e.g., "contains characters," "is equal to," or "is greater than"). The operator to choose depends on the search you're making.
6. Type the word, phrase, number, or code to search for in the box. **Note:** Leave the box empty if you don't need to type something in—for example, when you choose an operator like "is empty" or "is not empty."
7. Click Search to start your search.

Adding more choices. You can make an advanced search for records that match more than one criterion. To add a second criterion to your search,

1. While you're working in the Advanced Search dialog, click Add a Choice.
2. Follow steps 3 through 6 in the previous section to set up the search for that criterion.
3. Indicate whether to search either for
 - a. records that meet *both* your criteria—click "Search for all of the following (AND search)."
 - b. records that meet *either* criterion—click "Search for any of the following (OR search)."

Saving searches

You can save any advanced search to reuse later. To save a search, first fill out the Advanced Search dialog with the criteria choices you'd like to save, and then click Save This Search. Type a name for the saved search, then click Save This Search.

If you've just set up and run an advanced search, your criteria will be saved in the Advanced Search dialog. You then can save that search which you've just completed by returning to the Advanced Search dialog and clicking Save This Search. (To return to the Advanced Search dialog at any time, select Start Search from the Search menu, then click Go to Advanced Search.)

To open and reuse a saved search, in the Advanced Search dialog, click Open Saved Search. From the popup menu of search names, choose a search name, then click Select This Search. The Advanced Search dialog will be filled in with the criteria for this search. You then can click Search to run this reopened search.

Append searches and subset searches

Two commands in CAMEO's Search menu allow you to refine either a basic or advanced search. Once you have run a search and collected a set of found records, there are two ways to continue your search.

In the Search menu, run a second search by selecting either:

- **Append Search.** This *widens* your search. Records found during an append search will be added to the existing set of found records. You might choose to append a search if your first search did not turn up all the records you needed to work with. For example, you might have run a search of the Chemical Library for chemicals with a particular CAS number, then realized that you also wanted to work with the records for chemicals with another CAS number. You would run an append search for the second CAS number in order to obtain a set of records for chemicals with either CAS number.

- **Subset Search.** This *narrows* your search. Only records that were found in the first search and that also are found during your second search will be included in the set of records found in any subset search. You might choose a subset search if your first search turned up too many records, and you needed to narrow down the number of found records. For example, if you search the Chemical Library for chemicals that have “Poison” on their DOT placard label, you would find more than 900 records. If you then ran a subset search for chemicals that also have “Oxidizer” on their placard label, you would find about 40 records, all for chemicals that have both “Poison” and “Oxidizer” on their placard.

Sorting records in CAMEO modules

By default, in the Chemical Library, records are shown in List view in alphabetical order by chemical name. In all other CAMEO modules, records are listed in the order in which they were added to the module.

You can change the order of the records in any module by sorting them into a new order. When you do this, the new sort order is retained until you resort the module or unsort it to put the records back in default order (or perform an action such as a search).

You can sort either all the records in a module or just a found set of records (those records that you found by running a search of a module). When you sort just a found set, the sort order of those records is lost when you quit (exit) from CAMEO.

Choosing sort criteria

To sort records, you must choose one or more sort criteria. For example, to sort your Facilities records alphabetically by facility name, you would use facility name as your sort criterion.

You set up a sort and choose your criteria using the Sort dialog (Figure 8-1). In the Sort dialog, you see a list of the names of data fields that you can use as sort criteria. These names generally are similar to the data field names shown on your CAMEO records, although they don't correspond exactly, because the names you see in the Sort dialog are internal names used by CAMEO. Many names begin with a one- or two-letter prefix indicating the module they are in (for example, "CoFireDistrict" is the Fire District field in Contacts records). Field names in gray text can't be used as sort criteria.

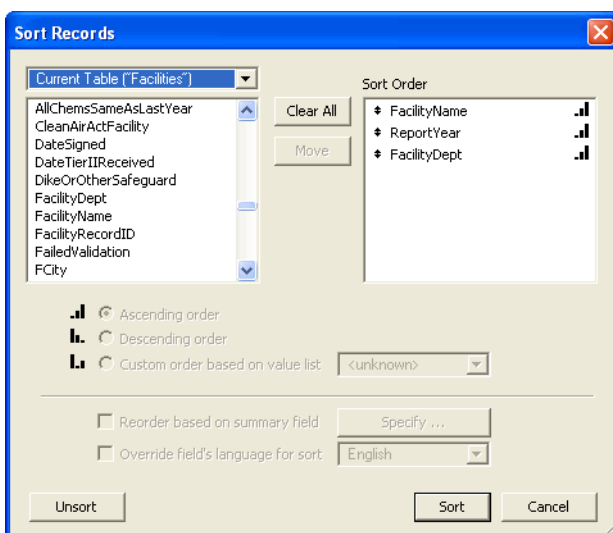


FIGURE 8-1. CAMEO's Sort dialog. From the sort criteria list, you select and move the desired criteria into the Sort Order box.

[Table 8-2 on page 253](#) shows the names of some of the sort criteria you may find most useful for sorting each CAMEO module.

Note that choosing some sort criteria may not produce the results you would expect. Here are two examples:

- A given Chemical Library record may contain multiple CAS numbers, stored as a return-delimited list (a list of numbers separated by return characters). Sorting the Chemical Library by CAS number will sort on the first number in the list, which may not be a meaningful operation.

- Physical properties are stored in the Chemical Library as a combination of several data fields. For example, melting point information is kept in three fields: MeltingPointNote, MeltingPointRange, and MeltingPointValue. For some chemicals, a specific value for the melting point is stored in the MeltingPointValue data field, while for others, a range is stored in the MeltingPointRange field. If you sort the Chemical Library by MeltingPointValue, making an “ascending” sort so that the smallest values appear first, all records with no MeltingPointValue will appear at the beginning of your sorted list. Among these records would be some for chemicals that have high melting points stored as ranges.

TABLE 8-2. Some CAMEO sort criteria.

Item to sort	Name of the corresponding sort criterion
Chemical Library items:	
Chemical Name	OfficialChemicalName
DOT Label	DOTLabel
EPCRA EHS Chemical	EpcraEhsChemical
Chemicals in Inventory items:	
Facility/Route Name	_CFacilityRouteName
Chemical Name	EnteredChemName
Facility Department	_CFacilityDepartment
Contacts items:	
Last Name	LastName
First Name	FirstName
Organization	Organization
Facilities items:	
Facility Name	FacilityName

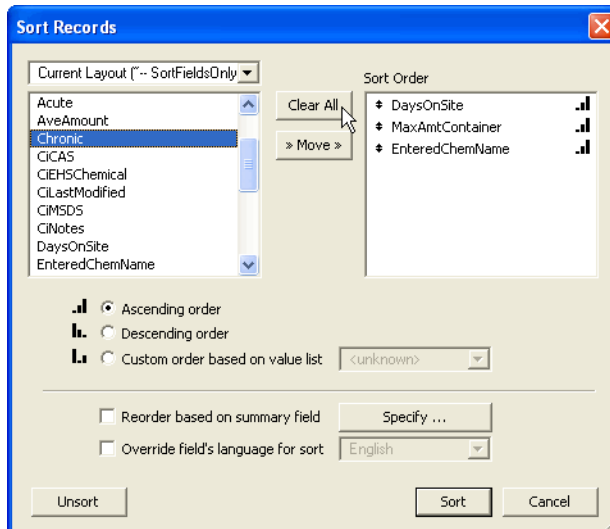
TABLE 8-2. Some CAMEO sort criteria. (Continued)

Item to sort	Name of the corresponding sort criterion
Report Year	ReportYear
Department	FacilityDept
Incidents items:	
Facility/Route Name	_FacRouteName
Incident Name	IncidentName
Spilled Date	SpillDate
Resources items:	
Supplier Name	ResourceName
Resource Type	ResourceType
Routes items:	
Route Name	RouteName
Route Type	RouteType
Screening & Scenarios items:	
Facility/Route Name	_scFacilityName
Chemical	EnteredChemName
Threat Zone Radius	_RadiusValue
Screening	Screening
Special Locations items:	
Location Name	LocationName
Location Type	LocationType

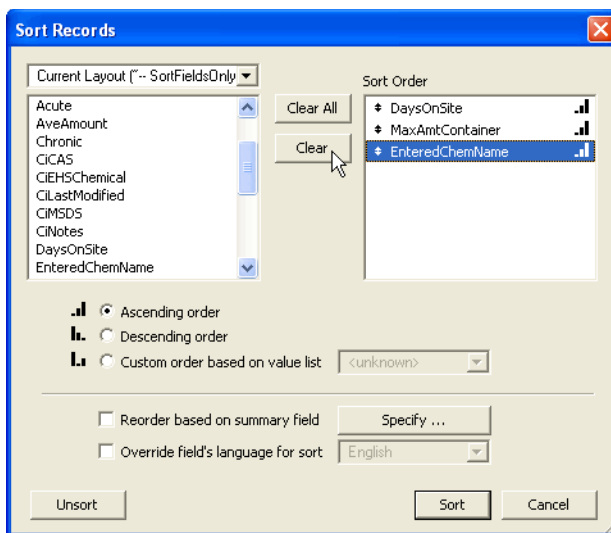
Sorting records by a single criterion

To sort all the records in a module using a single data field—such as chemical name or reporting year—as your sort criterion,

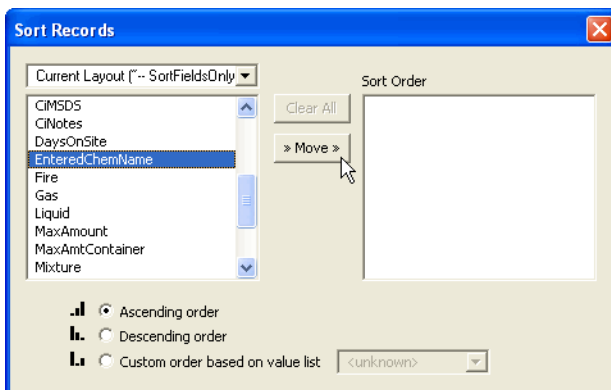
1. Open the module to be sorted.
2. From the Record menu, select Show All Records to clear any existing found set (there will be a found set if you ran a search in the module, unless you cleared your search results).
3. From the Record menu, select Sort. The Sort dialog will be displayed.
4. Choose the data field to use as your sort criterion, as follows:
 - a. The names of some of the data fields in the module may be listed as sort criteria in the Sort Order box (the box on the right side of the dialog). If you don't want to sort by any of these, click Clear All. In the example below, three Chemicals in Inventory criteria are being cleared.



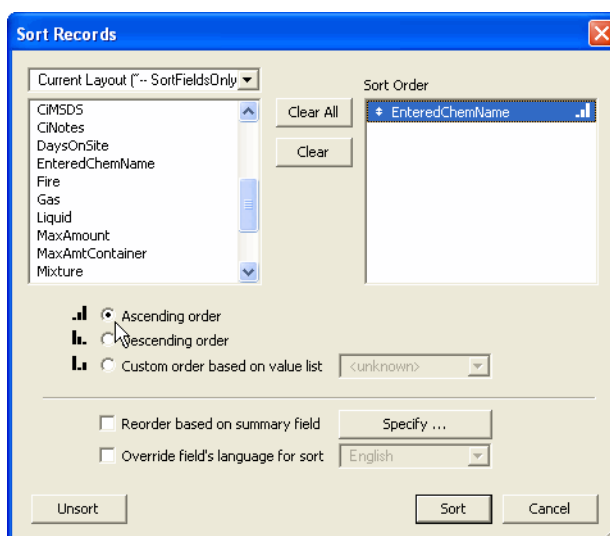
- b. If one of the criteria in the Sort Order box is the criterion you want to sort by, then click on the name of each other criterion, then click Clear, until just the criterion you want is listed in the box.



- c. If the criterion you want to sort by does not appear in the Sort Order box, then, in the list of data field names on the left, click on its name, then click Move to move it into the Sort Order box.



5. Once the Sort Order box contains just the criterion of your choice, then click on the name of your criterion in the Sort Order box to select it and then click either
 - a. Ascending order, to sort the records alphabetically from A to Z, in numerical order from smallest to largest number, or in date or time order from earliest to latest date or time. In the example below, the Chemicals in Inventory module is being sorted alphabetically by chemical name (“EnteredChemName”).



- b. Descending order, to sort the records alphabetically from Z to A, in numerical order from largest to smallest number, or in date or time order from latest to earliest date or time.
6. Click Sort to sort the records (or Cancel if you decide not to sort the records).

Unsorting records

You can unsort the records in any CAMEO module to return them to their original, default order¹, as follows:

1. Open the module to be unsorted.
2. From the Record menu, select Sort. The Sort dialog will be displayed.
3. Click Unsort (or Cancel if you decide not to unsort the records).

Sorting a subset of records in a module

Sometimes, you might want to sort just some of the records in a module. For example, in the Facilities module, you might need to sort just the records for facilities in the town of Haymarket without sorting the records for facilities elsewhere.

To sort just some records, first search the module to find just the records you want to sort, then sort those records, as follows:

1. From the Search menu, select Start Search.
2. Choose the criteria for your search, making an Advanced Search if necessary, then click Search. (For help setting up searches, see [“Searching” on page 246.](#)) When you run a search, a found set of records that meet your criteria is established within the module you’ve searched. You can access and sort only those records until you clear your search.
3. From the Record menu, select Sort.
4. Choose the criteria for your sort, then click Sort. Only the records in the found set will be sorted. In List view, you’ll see just the names of those records, sorted in the order you chose.
5. When you’ve finished working with your sorted records, if you want to clear your search so you can work with all the records in the module, select Clear Search from the Search menu (or Show All Records from the Record menu). The found set will be cleared, and the found records will revert to their original, default order among the other records in the module.

1. In the Chemical Library, the default record order is alphabetical by chemical name. In all other modules, the default is the order in which records were created.

Sorting by multiple criteria


Sometimes, you might find it useful to use more than one criterion to sort a module. For example, you might want to sort the records in the Contacts module alphabetically by last name. If some of your contacts have the same last names, you also might want to sort the records alphabetically by first name. If you sort this way, “Abe Johnson” would appear in your sorted list ahead of “Sarah Johnson.”

When you use more than one criterion in a sort, arrange your criteria so that the most important criterion appears first in the Sort Order box. In the example above, last name would appear before first name. In that case, your Contacts records will be sorted first by last name and then, within each last name, by first name.

To sort the records in a module using more than one criterion:

1. Open the module to be sorted.
2. From the Record menu, select Show All Records to clear any existing found set (unless you want to sort the found set; a found set exists if you ran a search in the module, and haven’t cleared your search results).
3. From the Record menu, select Sort. The Sort dialog will be displayed.
4. Choose the data fields to use as your sort criteria, as follows:
 - a. The names of some of the data fields in the module may be listed as sort criteria in the Sort Order box (the box on the right side of the dialog). If you don’t want to sort by *any* of these, click Clear All.
 - b. If one or more of the criteria in the Sort Order box are criteria you want to sort by, then click on the name of each other criterion, then click Clear, until just the criteria you want to use are listed in the box.
 - c. If any of the criteria you want to sort by does not appear in the Sort Order box, then, in the list of data field names on the left, click on the name of each of those criteria, then click Move to move it into the Sort Order box.

Note: In the Sort dialog, names of data fields generally are similar to the data field names shown on your CAMEO records, although they don’t correspond exactly.

5. Order your sort criteria by their importance: move the most important criterion to the top of the Sort Order list (in the example above, last name is the most important criterion), the next most important to the position just below it, and so on. To move a criterion, click the move button, , to the left of the criterion's name in the list, then drag the button up or down to move the criterion up or down in the list. For example, you might sort your Contacts records first by Organization (the most important criterion) and then by last name (the less-important criterion). Your contacts then will be sorted by organization, with contacts for the same organization listed alphabetically.
6. Once the Sort Order box contains just the criteria of your choice, then click on the name of *each* criterion in the Sort Order box to select it, and then click either
 - a. Ascending order, to sort the records by that criterion alphabetically from A to Z, in numerical order from smallest to largest number, or in date or time order from earliest to latest date or time. In the example above, Contacts records will be sorted alphabetically by last name.
 - b. Descending order, to sort the records by that criterion alphabetically from Z to A, in numerical order from largest to smallest number, or in date or time order from latest to earliest date or time.

For example, you might sort your Facilities records (a) in ascending order by city, which orders the list alphabetically by city, (b) in ascending order by facility name, which orders records within cities alphabetically by facility name, and then (c) in descending order by report year, which orders the records for any given facility with the most recent records listed before less recent records.
7. Click Sort to sort the records (or Cancel if you decide not to sort the records).

Some example sorts

- To sort your Screening & Scenarios records alphabetically by facility/ route name, choose “_scFacilityName” as your sort criterion, and sort in ascending order.
- To sort the Chemical Library records so that records for EHSs appear first in the list, with EHSs with the smallest Levels of Concern (LOCs) listed first (a smaller LOC indicates greater toxicity, so the most toxic EHSs would then be listed first): sort first by EpcraEhsChemical, in descending order, and then by LevelofConcern in ascending order.
- To sort the Chemical Library so that you see just the records for EHSs, with those with smaller LOCs listed first: make an advanced search of the Chemical Library for “EPCRA EHS Chemical” is “Yes,” then sort this found set by LevelofConcern in ascending order.

Overview of legislation

CAMEO is designed not only to be an emergency response tool, but to be useful for people who must meet the provisions of the Emergency Planning and Community Right-to-Know Act (EPCRA) of 1986.

EPCRA

EPCRA, also known as Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA), establishes requirements for federal, state, and local governments and for industry for emergency planning and “community right-to-know” reporting on hazardous and toxic chemicals. The community right-to-know provisions of EPCRA are intended to increase public knowledge of and access to information about the presence of hazardous chemicals in their communities and releases of these chemicals into the environment.

EPCRA includes four major sections: emergency planning (Section 301-303), emergency release notification (Section 304), community right-to-know reporting requirements (Sections 311 and 312), and toxic release inventory requirements (Section 313). Under EPCRA, manufacturers and other facilities are required to report accidental releases of hazardous chemicals to state and local officials, and to local fire departments. State and local officials are required to prepare contingency plans for communities where hazardous materials are manufactured, stored, or processed. Records of chemical inventories at facilities, and of accidental and routine chemical spills must be readily available to interested citizens, local and state governments, and local fire departments.

You can learn more about the reporting and planning requirements of EPCRA at <http://www.epa.gov/oswer/e1/content/epcra/> or by contacting the nearest EPA regional office. You can also review parts of the Code of Federal Regulations (CFR) describing these requirements. The relevant parts are 40 CFR § 300 and 355 (describing Sections 301 to 303 and Section 304 of EPCRA); 40 CFR § 370 (describing Sections 311 to 312); and 40 CFR § 372 (describing Section 313).

TABLE 8-3. A snapshot of EPCRA’s major sections.

Sections	Summary
Sections 301 to 303: Emergency Planning	Local governments are required to prepare chemical emergency response plans, and to review plans at least annually. State governments are required to oversee and coordinate local planning efforts. Facilities that maintain EHSs (Extremely Hazardous Substances) on site in quantities greater than corresponding Threshold Planning Quantities (TPQs) must cooperate in emergency plan preparation.
Section 304: Emergency Notification	Facilities must immediately report accidental releases of EHS chemicals and “hazardous substances” in quantities greater than corresponding Reportable Quantities (RQs) defined under CERCLA (the Comprehensive Environmental Response, Compensation, and Liability Act) to state and local officials. Information about accidental chemical releases must be available to the public.

TABLE 8-3. A snapshot of EPCRA’s major sections.

Sections	Summary
Sections 311 and 312: Community Right-to-Know Requirements	Facilities manufacturing, processing, or storing designated hazardous chemicals must make Material Safety Data Sheets (MSDSs) describing the properties and health effects of these chemicals available to state and local officials and local fire departments. Facilities must also report, to state and local officials and local fire departments, inventories of all on-site chemicals for which MSDSs exist. Information about chemical inventories at facilities and MSDSs must be available to the public.
Section 313: Toxic Chemical Release Reporting	EPA must maintain records of toxic chemical emissions from certain facilities, and must use this information in research and in preparation of guidelines and regulations. This EPA module is called the Toxic Release Inventory (TRI). Facilities subject to these reporting requirements must submit annual reports of total amounts of chemicals released to the environment (both routinely and accidentally) to EPA and state officials.

Section 301-303: emergency planning. The emergency planning sections of the law are designed to develop state and federal emergency response and preparedness capabilities through improved coordination and planning, especially within local communities. EPCRA, when enacted, required the governor of each state to designate a State Emergency Response Commission or SERC. Each SERC then designated local emergency planning districts and appointed a Local Emergency Planning Committee, or LEPC, for each district. Each SERC is responsible for supervising and coordinating the activities of LEPCs, establishing procedures for receiving and processing public requests for information collected in order to meet other provisions of EPCRA, and reviewing local emergency plans.

Each LEPC includes, at a minimum, elected state and local officials, police, fire, civil defense, and public health professionals, environment, hospital, and transportation officials, as well as representatives of facilities subject to the reporting requirements, community groups, and the media. As soon as facilities become subject to the emergency planning requirements, they must designate representatives to participate in the EPCRA planning process.

The LEPC's primary responsibility is to develop a chemical emergency response plan, and to review it at least annually thereafter. In developing a response plan, each LEPC evaluates available resources for preparing for and responding to a potential chemical accident.

Each plan:

- identifies facilities and transportation routes of extremely hazardous substances.
- describes on- and off-site emergency response procedures.
- designates a community emergency response coordinator and facility coordinator(s) to implement the plan.
- outlines emergency procedures.
- describes methods for determining the occurrence of a release and the possible affected area and population.
- describes community and industry emergency equipment and facilities and identifies the people responsible for them.
- outlines evacuation plans.
- describes a training program (including schedules) for emergency response personnel.
- presents methods and schedules for exercising emergency response plans.

To assist LEPCs in preparing and reviewing plans, the National Response Team (NRT), composed of 16 federal agencies with emergency response responsibilities, published guidance on emergency response planning. This guidance, the *Hazardous Materials Emergency Planning Guide*, was published by the NRT in March 1987.

Each emergency response plan must initially be reviewed by the SERC and then must be reviewed at least annually by the LEPC. Regional Response Teams (RRTs), composed of federal regional officials and state government representatives, may also review the plans and provide assistance to LEPCs upon request. Planning activities of LEPCs and facilities were to be initially focused on, but not limited to, the 360 extremely hazardous substances (also known as EHS chemicals, or EHSs). However, plans were to be comprehensive, addressing all hazardous materials of concern within each community and covering transportation routes as well as fixed facilities.

Any facility where any of the extremely hazardous substances (EHSs) are present in quantities equal to or greater than threshold planning quantities (TPQs) are subject to these emergency planning requirements. In addition, each state governor or SERC, after a period of public comment, can designate other facilities to be subject to these requirements. A facility must notify the appropriate SERC and LEPC that it has become subject to EPCRA planning requirements within 60 days after beginning to keep on site any EHS chemical in quantities equal to or greater than its TPQ. In addition, the SERC must notify the nearest EPA regional office of all facilities subject to these emergency planning requirements, including any facilities designated by either the SERC or governor.

Section 304: emergency notification. Facilities must immediately notify any LEPCs and SERCs likely to be affected by a release into the environment of a listed hazardous substance in an amount exceeding the reportable quantity (RQ) of that substance. Substances subject to this requirement include both the 360 extremely hazardous substances (listed in 40 CFR § 355), or the 725 hazardous substances subject to the emergency notification provisions of Section 103(a) of the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA) (listed in 40 CFR § 302.4). Initial notification can be made by telephone, radio, or in person. Emergency notification requirements for transportation incidents can be met by dialing the 911 emergency number, if possible, or otherwise by calling the operator. Releases of CERCLA hazardous substances must also be reported to the National Response Center (800-424-8802).

Sections 311 and 312: community right-to-know requirements.

Two community right-to-know requirements are included in EPCRA.

Section 311 requires each facility that must prepare material safety data sheets (MSDSs) under Occupational Safety and Health Administration (OSHA) regulations to submit either copies of its MSDSs or a list of all on-site MSDS chemicals to the LEPC, the SERC, and the local fire department with jurisdiction over the facility.

If the facility owner or operator chooses to submit a list of MSDS chemicals, that list must include the common or chemical name of each substance and must identify the applicable hazard categories into which each substance falls. These categories are:

- immediate (acute) health hazard,
- delayed (chronic) health hazard,
- fire hazard,
- sudden release of pressure hazard, and
- reactive hazard.

If a list is submitted, the facility must submit a copy of the MSDS for any chemical on the list if requested by the LEPC or SERC.

Also, EPA has established threshold quantities for hazardous chemicals. A facility is not required to report hazardous chemicals existing on site in quantities below the appropriate threshold. The current Section 311 thresholds are:

- **For extremely hazardous substances:** either 500 pounds, or the TPQ, whichever is lower.
- **For all other hazardous chemicals:** 10,000 pounds.

OSHA regulations require all facility owner/operators to maintain or prepare MSDSs for on-site hazardous chemicals, and therefore all existing facilities with hazardous chemicals on site should by now have submitted either lists or MSDSs. Under EPCRA, facilities not previously covered by the OSHA regulations must submit either MSDSs or a list of MSDS chemicals within three months after becoming covered. That is, whenever a facility becomes subject to the right-to-know requirements, it must submit either MSDSs or a list within three months after being required to prepare or have available MSDSs, under OSHA regulations.

Facilities must submit either an MSDS or revised list once they store new, previously unreported hazardous chemicals on site in quantities at or above the established threshold levels. Whenever significant new information about a hazardous chemical becomes available, if MSDSs rather than a list were submitted, an updated MSDS must be submitted.

Under Section 312, each subject facility must submit an annual emergency and hazardous chemical inventory form to the LEPC, the SERC, and the local fire department with jurisdiction over the facility. Hazardous chemicals covered by Section 312 are those for which facilities are required to prepare or have available an MSDS, under OSHA's Hazard Communication Standard, and that were present at the facility in quantities equaling or exceeding threshold quantities at any time during the previous year. These are the same chemicals covered under Section 311.

EPA has established threshold quantities for hazardous chemicals covered under Section 312. A facility is not required to report hazardous chemicals on site in quantities below the thresholds established for those chemicals. The current Section 312 thresholds, which are identical to the Section 311 threshold quantities, are:

- **For extremely hazardous substances:** either 500 pounds or the TPQ, whichever is lower.
- **For all other hazardous chemicals:** 10,000 pounds.

The inventory form incorporates a “two-tier” approach. Under Tier I, facilities must submit the following aggregate information about chemicals within each applicable hazard category:

- An estimate (in ranges) of the maximum amount of chemicals in each category present at the facility at any time during the preceding calendar year.
- An estimate (in ranges) of the average daily amount of hazardous chemicals in each category present on site during the preceding calendar year.
- The approximate location of hazardous chemicals in each category.

If requested by a LEPC, SERC, or local fire department, a facility also must provide the following Tier II information for each requested substance:

- The common or chemical name of the substance, as indicated on the MSDS.
- An estimate (in ranges) of the maximum amount of the chemical present at the facility at any time during the preceding calendar year.
- A brief description of the manner in which the chemical is stored.
- The location of the chemical at the facility.
- An indication of whether the owner/operator elects to withhold location information from public disclosure.

Since many states impose additional requirements or have incorporated the federal reporting provisions on their own forms, Tier I/II forms should be obtained from SERCs.

Facilities may choose to submit Tier II rather than Tier I forms, and some states require that Tier II forms be submitted. Individual citizens may also request copies of completed Tier II forms from either a SERC or LEPC. In such a case, if a Tier II form is not available, it must be requested from the facility. Information submitted by facilities under Sections 311 and 312 generally must be available for public inspection during normal working hours. Facilities must provide Section 311 and 312 information whenever requested, regardless of whether on-site quantities of hazardous chemicals

equal or exceed threshold quantities. Completed forms must be submitted by subject facilities on or before March 1 of each year.

Section 313: toxic chemical release reporting. Section 313 of EPCRA requires EPA to establish and maintain an inventory of toxic chemical emissions from certain facilities. Facilities subject to this reporting requirement must complete a Toxic Chemical Release Inventory Form (commonly known as Form R) for specified chemicals. Toxic Chemical Release Inventory Forms must be submitted to EPA and to state officials designated by the governor on or before July 1 of each year.

These reports should include descriptions of both routine and accidental releases of hazardous chemicals from subject facilities during the preceding calendar year. The purpose of this reporting requirement is to inform the public about routine releases of toxic chemicals to the environment. The information is also intended for use in research and in the development of regulations, guidelines, and standards.

The Section 313 reporting requirement applies to owners and operators of facilities with at least ten full-time employees, that are in Standard Industrial Classification (SIC) codes 20 through 39, and that manufacture, import, process, or otherwise use at least one listed toxic chemical in excess of specified threshold quantities.

Facilities that manufacture or process any listed toxic chemical in quantities exceeding 25,000 pounds must submit a completed Form R by July 1 each year, describing all releases that took place during the previous calendar year. Facilities using listed toxic chemicals in any other way in quantities greater than 10,000 pounds during a calendar year must submit inventory forms by July 1 of the following calendar year. EPA has the authority to revise these threshold quantities and to change the SIC codes covered by Section 313.

More than 700 chemicals in more than 20 categories are now included on the list. EPA, through its rule-making authority, can add or remove chemicals from this list.

Inventory reports from facilities must be submitted both to EPA and to designated state agencies. EPA has established and maintains a national toxic chemical inventory, called the Toxic Release Inventory (TRI) database, containing the submitted inventory information. Individuals may access this national module electronically, or may obtain included information by several other means.

Other EPCRA provisions. Section 322 of EPCRA addresses the effect of trade secrets on emergency planning, community right-to-know, and toxic chemical release reporting.

Section 325 of EPCRA describes the penalties for failure to comply with the requirements of this law. Civil and administrative penalties ranging from up to \$10,000 to \$75,000 per violation or per day per violation can be assessed facilities that fail to comply with the emergency planning (Section 302), emergency notification (Section 304), community right-to-know (Sections 311 and 312), toxic chemical release (Section 313), and trade secret (Sections 322 and 323) reporting requirements. Criminal penalties of up to \$50,000 or five years in prison may also be assessed any individual who knowingly and willingly fails to provide emergency notification of a hazardous chemical release. Penalties of up to \$20,000 and/or up to one year in prison may be assessed any person who knowingly and willfully discloses any information entitled to protection as a trade secret.

In addition, Section 326 includes provisions allowing citizens to initiate civil actions against EPA, SERCs, or the owner or operator of a facility for failure to meet the requirements of the emergency planning and community right-to-know provisions of EPCRA. A SERC, LEPC, or state or local government may initiate actions against facility owners or operators for failure to comply with EPCRA requirements. Finally, states may file suit against EPA for failure to provide trade secret information.

CAA 112(r)

Section 112(r) of the Clean Air Act (CAA) Amendments of 1990 requires EPA to publish regulations focusing on preventing chemical accidents. These new regulations build on the chemical safety work begun under EPCRA, as well as the Process Safety Management regulations of the

Occupational Safety and Health Administration (OSHA). While EPCRA requires communities to develop emergency response plans using information on hazardous chemicals provided by local industries, under CAA 112(r), facilities must identify and assess their chemical hazards and carry out certain activities designed to reduce the likelihood and severity of accidental chemical releases. Once information about chemicals is shared among industry, government, and the community, those entities can work to reduce the risks to public health and the environment.

In 1996, EPA published the final rule for CAA 112(r), called the Risk Management Plan Rule (“RMP Rule”) (40 CFR Part 68). A facility owner or operator is subject to the RMP Rule if, in any process at the facility, there is more than a threshold quantity of any of the more than 100 hazardous substances regulated under the rule. If a facility is subject to the RMP Rule, it must perform an offsite consequence analysis to check whether its process puts nearby populations at risk. If it does, the facility must take some steps to manage that risk (these steps are described in the rule).

You can view the text of CAA 112(r) and the RMP Rule, see the list of regulated substances and their threshold quantities, and obtain factsheets and other information related to CAA 112(r), at <http://www.epa.gov/osweroe1/content/rmp>.

The similarity between the offsite consequence analyses required by the RMP Rule and the hazards analysis procedures required under EPCRA has confused some CAMEO users. The main point to understand is that CAMEO’s Screening & Scenarios module is designed to implement the hazards analysis procedures described in an EPCRA guidance document (*Technical Guidance for Hazards Analysis: Emergency Planning for Extremely Hazardous Substances*), and *not* for offsite consequence analyses (see “[What are the differences between Screening & Scenarios, ALOHA, and RMP endpoint distances?](#)” on page 155). For offsite consequence analyses, you can use the RMP*Comp software program, which can be downloaded at no cost from http://www.epa.gov/osweroe1/content/rmp/rmp_comp.htm.

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Note: In this Bibliography, a notation in brackets (such as [ACGIH]) appearing within a citation indicates that the cited reference served as one of the sources of the chemical data in CAMEO's Chemical Library (see "[Sources of CAMEO's chemical data](#)" on page 103).

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The Reactive Groups

Each substance in the Chemical Library was assigned to one or more reactive groups, based on the known chemistry of that substance. Reactive groups are categories of chemicals that react in similar ways because they are similar in their chemical structure.

To predict the reactivity of a mixture of chemicals, CAMEO first identifies the reactive groups to which each of the chemicals belongs, and then predicts the kinds of chemical reactions likely to occur when members of these groups are mixed together. (See [“How CAMEO predicts mixture reactivity” on page 96](#) to learn how to make reactivity predictions.)

This appendix section contains descriptions of all the reactive groups used in CAMEO. To find out which reactive groups a particular chemical belongs to, find that chemical’s record in the Chemical Library, click the Response Information Data Sheets tab, and then click the Reactivity tab. Under the REACTIVE GROUPS heading, you’ll see the reactive group(s) to which the chemical belongs.

CAMEO's Reactive Groups.

- Acid Halides
- Acids, Inorganic Non-oxidizing
- Acids, Inorganic Oxidizing
- Alcohols and Polyols
- Aldehydes
- Amides and Imides, Organic
- Amines
- Anhydrides
- Azo, Diazo, and Azido Compounds and Organic Hydrazines, Organic
- Bases
- Carbamates and Esters
- Carboxylic Acids
- CFCs and HCFCs (chlorofluorocarbons and chlorofluorohydrocarbons)
- Chlorosilanes
- Cyanides, Inorganic
- Epoxides
- Esters
- Ethers
- Halogenated Organic Compounds
- Halogenating Agents, Strong
- Hydrocarbons, Aliphatic Saturated
- Hydrocarbons, Aliphatic Unsaturated
- Hydrocarbons, Aromatics
- Inorganic Compounds/Neither Reducing nor Oxidizing
- Inorganic Oxidizing Agents
- Inorganic Reducing Agents
- Insufficient Information for Classification
- Isocyanates and Isothiocyanates, Organic

- Ketones
- Metal Hydrides, Alkyls and Aryls
- Metals, Alkali, Very Active
- Metals, Elemental & Powder, Active
- Metals, Less Reactive
- Nitrides, Phosphides, Carbides, and Silicides, Inorganic
- Nitriles
- Nitrites, Nitrates, and Nitro Compounds, Organic
- Not Chemically Reactive
- Organometallics
- Peroxides, Organic
- Phenols, Cresols
- Phosphates and Thiophosphates, Organic
- Salts, Acidic Inorganic/Organic
- Salts, Basic Inorganic/Organic
- Sulfides, Inorganic
- Sulfides, Organic
- Thiocarbamate Esters and Salt/Dithiocarbamate Esters and Salts

Acid Halides

Flammability. These materials are flammable and pose a vapor cloud explosion threat. Vapors may flashback to source and cause a violent rupture of a closed container. Combustion of acid chlorides produces hydrochloric acid and phosgene; toxic and irritating gases are also generated from the combustion of acid bromides and iodides.

Reactivity. Materials in this group are water reactive; some are violently reactive. They are incompatible with strong oxidizing agents, alcohols, amines, alkali.

Toxicity. Extremely corrosive and irritating to mucous membranes, eyes, and the respiratory tract. Upon direct contact with skin, they will cause severe burns. Fumes are a serious inhalation hazard.

Other characteristics. Compounds in this group have a halogen atom (chlorine, bromine, or iodine) attached to a carbonyl (C=O) group that is also attached to an organic group. These compounds are used in industrial synthesis of organic compounds, especially acetyl compounds, and to eliminate all traces of water from organic liquids.

Examples. Acetyl bromide, acetyl chloride, acetyl iodide, benzoyl chloride, fumaryl chloride, isobutyryl chloride, methacryloyl chloride, phenacetyl chloride.

Acids, Inorganic Non-oxidizing

Flammability. Compounds in this group are generally nonflammable.

Reactivity. Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pHs of less than 7.0. Acids neutralize chemical bases (for example: amines and inorganic hydroxides) to form salts. Neutralization occurs as the base accepts hydrogen ions that the acid donates. Neutralization can generate dangerously large amounts of heat in small spaces. The dissolution of acids in water or the dilution of their concentrated solutions with additional water may generate significant heat; the addition of water often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting “bumping” spatters the acid. These materials react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas. They can initiate the polymerization of certain classes of organic compounds. They react with cyanide compounds to release gaseous hydrogen cyanide. They often generate flammable and/or toxic gases in contact with dithiocarbamates, isocyanates, mercaptans, nitrides, nitriles, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H₂S and SO₃), dithionites (SO₂), and even carbonates: the carbon dioxide gas from the last is non-

toxic but the heat and spattering from a rapid reaction can be troublesome. Acids often catalyze (increase the rate) of chemical reactions.

Toxicity. Corrosive to tissue. Acid fumes irritate sensitive tissues (such as the eyes and respiratory system) especially severely.

Other characteristics. Acids have a sour taste; they turn blue litmus red.

Examples. Hydrochloric acid, hydroiodic acid, hydrobromic acid, hydrofluoric acid, phosphoric acid.

Acids, Inorganic Oxidizing

Flammability. Materials in this group are generally nonflammable. They may accelerate the combustion of other materials by providing oxygen to the combustion site, that is, by serving as oxidizing agents.

Reactivity. Materials in this group are generally soluble in water with the release of hydrogen ions. The resulting solutions have pHs of less than 7.0. Materials in this group react with chemical bases (for example: amines and inorganic hydroxides) to form salts. These neutralization reactions occur as the base accepts hydrogen ions that the acid donates. Neutralizations can generate dangerously large amounts of heat in small spaces. The dissolution of acids in water or the dilution of their concentrated solutions with water may generate significant heat. The addition of water to acids often generates sufficient heat in the small region of mixing to boil some of the water explosively. The resulting “bumping” spatters acid widely. These materials have significant ability as oxidizing agents, but that ability varies (for example, from high for nitric acid to low for sulfuric acid and most sulfonic acids). They can react with active metals, including iron and aluminum, and also many less active metals, to dissolve the metal and liberate hydrogen and/or toxic gases. Like other acids, materials in this group can initiate polymerization in certain classes of organic compounds. Their reactions with cyanide salts and compounds release gaseous hydrogen cyanide. Flammable and/or toxic gases are also often generated by their reactions

with dithiocarbamates, isocyanates, mercaptans, nitrides, nitriles, sulfides, and weak or strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H₂S and SO₃), dithionites (SO₂), and even carbonates: the carbon dioxide gas from the last is non-toxic but the heat and spattering from the reaction can be troublesome. Acids often catalyze (increase the rate) of chemical reactions.

Toxicity. Corrosive to tissue. Acid fumes irritate sensitive tissues (such as the eyes and respiratory system) especially severely.

Other characteristics. Acids have a sour taste; they turn blue litmus red.

Examples. Sulfuric acid, nitric acid, perchloric acid, arsenic acid, chlorosulfonic acid, chromic acid, fluorosulfonic acid, chloric acid, nitrosulfuric acid, selenic acid, sulfurous acid.

Alcohols and Polyols

Flammability. Many alcohols are highly flammable (with flash points below 100 degrees F). Especially dangerous are methanol and ethyl alcohol, because of their wide flammability limits. Polyols are generally combustible. Their general low volatility means that they are poorly flammable.

Reactivity. Flammable and/or toxic gases are generated by the combination of these materials with alkali metals, nitrides, and strong reducing agents. They react with oxoacids and carboxylic acids to form esters plus water. Oxidizing agents convert them to aldehydes or ketones. They exhibit both weak acid and weak base behavior. They may initiate the polymerization of isocyanates and epoxides.

Toxicity. Alcohols and polyols vary widely in toxicity. Among the most toxic are methyl and allyl alcohol, which act as nervous system depressants.

Other characteristics. These materials are organic compounds containing one or more hydroxyl (-OH) groups linked to hydrocarbon

groups. Alcohols may have straight-chain, branched-chain or ring structures. Polyols contain two or more hydroxyl groups; glycols (diols) contain exactly two hydroxyl groups, each attached to a different carbon atom.

Examples. Ethyl alcohol, methanol, propanol, butanol, ethylene glycol, hexanol, allyl alcohol, amyl alcohol, benzyl alcohol, cyclopentanol, glycerol, isopropyl alcohol, isobutyl alcohol.

Aldehydes

Flammability. Many aldehydes are either gases or volatile liquids, and are highly flammable (with flash points below 100 degrees F). Especially dangerous are formaldehyde, acetaldehyde, acrolein, crotonaldehyde, and other aldehydes that have wide flammability limits.

Reactivity. Aldehydes are frequently involved in self-condensation or polymerization reactions. These reactions are exothermic; they are often catalyzed by acid. Aldehydes are readily oxidized to give carboxylic acids. Flammable and/or toxic gases are generated by the combination of aldehydes with azo, diazo compounds, dithiocarbamates, nitrides, and strong reducing agents. Aldehydes can react with air to give first peroxy acids, and ultimately carboxylic acids. These autoxidation reactions are activated by light, catalyzed by salts of transition metals, and are autocatalytic (catalyzed by the products of the reaction). The addition of stabilizers (anti-oxidants) to shipments of aldehydes retards autoxidation.

Toxicity. Variable. Aldehydes act by inhalation, by ingestion, and by contact with the skin. Their exact metabolic effect depends upon the route of exposure. Among the most toxic are acrolein and crotonaldehyde, which are known carcinogens. Aldehydes are known sensitizers for small populations of humans and serve to cause chemically induced allergic reactions. The effects of these reactions can be dramatic at rather low concentrations. The effects of formaldehyde and acetaldehyde have been well documented because these aldehydes are found in many synthetic products, such as clothing, insulation, and building products. These two

aldehydes are also naturally occurring as combustion products (e.g., of forest fires).

Other characteristics. Aldehydes contain a carbonyl group (C=O group) to which one hydrogen atom and either an aliphatic or aromatic organic group are attached. Compounds of this group are formed by the dehydration of alcohols.

Examples. Formaldehyde, butyraldehyde, acetaldehyde, benzaldehyde, crotonaldehyde, acrolein, propionaldehyde, pentanal, valeraldehyde.

Amides and Imides, Organic

Flammability. Compounds in this group are combustible. The products of their combustion include noxious NO_x.

Reactivity. Organic amides react with azo and diazo compounds to generate toxic gases. Flammable gases are formed by the reaction of organic amides with strong reducing agents. Amides are very weak bases (weaker than water). Imides are less basic yet and in fact react with strong bases to form salts. That is, they can react as acids. Mixing amides with dehydrating agents such as P₂O₅ or SOCl₂ generates the corresponding nitrile. The combustion of these compounds generate mixed oxides of nitrogen (NO_x).

Toxicity. Low to moderate. They act by ingestion and by contact with the skin.

Other characteristics. Organic amides contain the structural group RCONH₂ (where R is any organic radical). In some amides the hydrogen atoms are replaced by other R groups. Amides derive chemically from carboxylic acids by the replacement of the -OH portion of the grouping RCOOH with an -NH₂ group. Imides contain the grouping (RCO)₂NH. Most amides are solids at room temperature (exception: formamide). Amides containing five carbon atoms or fewer are soluble in water. Amides are versatile chemicals that enter into many useful reactions, especially in the fertilizer industry. Polyamides are polymers in which a succession of

amide linkages link hold together a molecular chain of high molecular weight. Proteins in foods are naturally occurring polyamides; Nylon is a synthetic polymer.

Examples. Urea, acetamide, acrylamide, benzamide, caprolactam, formamide, dimethylformamide, phenylurea.

Amines

Flammability. Amines are generally high-boiling liquids or solids at room temperature and are not highly flammable. Methylamine is an exception, because it is a gas with a wide flammability range. Amines are combustible. The combustion of amines yields noxious NO_x.

Reactivity. Amines are chemical bases. They neutralize acids to form salts plus water. These acid-base reactions are exothermic. The amount of heat that is evolved per mole of amine in a neutralization is largely independent of the strength of the amine as a base. Amines may be incompatible with isocyanates, halogenated organics, peroxides, phenols (acidic), epoxides, anhydrides, and acid halides. Flammable gaseous hydrogen is generated by amines in combination with strong reducing agents, such as hydrides.

Toxicity. Variable. Some are very poisonous; others are only slightly toxic. Many amines are skin irritants. Some amines are known sensitizers for small populations of humans and serve to cause chemically induced allergic reactions. The effects of such reactions can be dramatic at rather low concentrations.

Other characteristics. Amines are organic compounds derived from ammonia (NH₃) by the replacement of one or more of the three hydrogen atoms of NH₃ with an organic group R. The R group may be either aromatic or aliphatic. Aromatic amines contain at least one aromatic R group; aliphatic amines contain all aliphatic groups. Many amines have unpleasantly fishy or putrid odors.

Examples. Methylamine, aniline, triethanolamine, hexamethylenetetramine, cyclohexylamine, xylylidine, ethylaniline, pyridine, anisidine, diphenylamine, ethylenediamine.

Anhydrides

Flammability. Except for acetic anhydride, these materials are not flammable but are combustible. The products of combustion are noxious.

Reactivity. Organic anhydrides react exothermically with water. The reactions are sometimes slow, but can become violent when local heating accelerates their rate. The rate of reaction with water is also accelerated by acids. Compounds from this group are incompatible with acids, strong oxidizing agents, alcohols, amines, and bases.

Toxicity. Extremely corrosive and irritating to mucous membranes, eyes, respiratory tract. Burns the skin. Fumes are serious inhalation hazard.

Other characteristics. These substances are formed by the condensation reaction of two molecules of an organic acid with the loss of water. They have the form $R_1(CO)O(CO)R_2$, where R_1 and R_2 may be any of a number of different organic compounds. Compounds in this group are used in organic synthesis and as dehydrating agents in nitration, sulfonation, and other reactions.

Examples. Acetic anhydride, butyric anhydride, isobutyric anhydride, maleic anhydride, methacrylic anhydride, phthalic anhydride, propionic anhydride

Azo, Diazo, and Azido Compounds and Organic Hydrazines, Organic

Flammability. Many Azo, Diazo, and Azido compounds are highly flammable/explosive (with flash points below 100 degrees F). Especially

dangerous is diazomethane, which explodes at slightly elevated temperatures because of its instability. Methylhydrazine is especially dangerous because of its low flash points and wide flammability limits. All compounds in this group give products that include noxious NO_x. Many azides, especially the heavy metal azides, easily detonate by friction, shock, or heating. Azo dyes have been known to explode as finely divided dust dispersed in the air.

Reactivity. Some materials in this group can detonate. This applies in particular to organic azides that have been sensitized by the addition of metal salts or strong acids. Toxic gases are formed by mixing materials of this class with acids, aldehydes, amides, carbamates, cyanides, inorganic fluorides, halogenated organics, isocyanates, ketones, metals, nitrides, peroxides, phenols, epoxides, acyl halides, and strong oxidizing or reducing agents. Flammable gases are formed by mixing materials in this group with alkali metals. Explosive combination can occur with strong oxidizing agents, metal salts, peroxides, and sulfides. Azo dyes can be explosive when suspended in air at specific concentrations.

Toxicity. Acutely toxic by all routes of exposure. Some of these compounds bind strongly to essential enzymes, inactivating them.

Other characteristics. Azo compounds (RN=NR) arise from the oxidation of substituted hydrazines RHN-NHR. For example, azomethane (H₃CN=NCH₃) is prepared from dimethylhydrazine. The compounds are not basic in aqueous solution (because the presence of the N=N bond reduces the availability of the unshared pairs of electrons in the nitrogen atoms). They are readily reduced back to the hydrazine and further to primary amines. That is, they are good oxidizing agents. Diazo compounds contain the characteristic group >CN₂. Thus, diazomethane has the formulas CH₂N₂. The organic azides are derivatives of hydrazoic acid, HN₃. Alkyl azides, such as methyl azide, are prepared by treating sodium azide with the appropriate alkyl sulfate. Acyl azides have the formula RCON₃.

Examples. Azidoacetaldehyde, azidoacetic acid, azidoacetone, aminophenylazobenzene, azobenzene, diazomethane, diazoacetone nitrile, diazoacetaldehyde.

Bases

Flammability. Materials in this group are mostly nonflammable and non-combustible. Exceptions include the alcoholic solutions of sodium methylate (in which the solvent is flammable) and other mixtures that contain organic components. Also, ammonia, which is considered to be nonflammable, actually does burn with difficulty and in confined spaces has exploded.

Reactivity. Compounds in this group are chemically similar to sodium hydroxide (NaOH) or sodium oxide (Na₂O). They neutralize acids exothermically to form salts plus water. When soluble in water they give solutions having a pH greater than 7.0. Mixing these materials with water can generate troublesome amounts of heat as the base is dissolved or diluted. Bases react with certain metals (such as aluminum and zinc) to form oxides or hydroxides of the metal and generate gaseous hydrogen. Bases may initiate polymerization reactions in polymerizable organic compounds, especially epoxides. They may generate flammable and/or toxic gases with ammonium salts, nitrides, halogenated organics, various metals, peroxides, and hydroperoxides. Materials of this group often serve as catalysts.

Toxicity. Corrosive to tissue.

Other characteristics. Solutions of bases have a bitter taste and a slippery feel; they turn red litmus blue.

Examples. Sodium hydroxide, potassium hydroxide, ammonium hydroxide, barium hydroxide, barium oxide, beryllium hydroxide, calcium hydroxide, calcium oxide.

Carbamates and Esters

Flammability. Compounds in this group are high-boiling liquids or solids. They are generally nonflammable, but are combustible. Combustion products include noxious NO_x and carbon monoxide.

Reactivity. Materials in this group are chemically similar to, but more reactive than amides. Like amides they form polymers such as polyurethane resins. Carbamates are incompatible with strong acids and bases, and especially incompatible with strong reducing agents such as hydrides. Flammable gaseous hydrogen is produced by the combination of active metals or nitrides with carbamates. Strongly oxidizing acids, peroxides, and hydroperoxides are incompatible with carbamates.

Toxicity. Many carbamates are used as pesticides. These materials can be extremely toxic. They act by blocking the function of cholinesterase, an enzyme that is essential to the transmission of nerve impulses. Oral exposure is the principal concern. Absorption through the skin is slow, but dermal exposure must still be avoided because of the high toxicity.

Other characteristics. Compounds in this group are derivatives of carbamic acid (NH_2COOH). The urethanes are alkyl carbamates, that is, esters of carbamic acid.

Examples. Carbamic acid, ammonium carbamate, carbaryl, oxamyl, propoxur, urethane.

Carboxylic Acids

Flammability. Many low molecular weight carboxylic acids (C1-C4) have flash points between 100 and 150 degrees F, and reasonably wide flammability limits. They are therefore considered a moderate fire hazard.

Reactivity. These organic compounds donate hydrogen ions if a base is present to accept them. They react in this way with all bases, both organic (for example, the amines) and inorganic. Their reactions with bases, called “neutralizations,” are accompanied by the evolution of substantial amounts of heat. Neutralization between an acid and a base produces water plus a salt. Carboxylic acids with six or fewer carbon atoms are freely or moderately soluble in water; those with more than six carbons are slightly soluble in water. Soluble carboxylic acid dissociate to an extent in water to yield hydrogen ions. The pH of solutions of carboxylic acids is therefore

less than 7.0. Many insoluble carboxylic acids react rapidly with aqueous solutions containing a chemical base and dissolve as the neutralization generates a soluble salt. Carboxylic acids in aqueous solution and liquid or molten carboxylic acids can react with active metals to form gaseous hydrogen and a metal salt. Such reactions occur in principle for solid carboxylic acids as well, but are slow if the solid acid remains dry. Even “insoluble” carboxylic acids may absorb enough water from the air and dissolve sufficiently in it to corrode or dissolve iron, steel, and aluminum parts and containers. Carboxylic acids, like other acids, react with cyanide salts to generate gaseous hydrogen cyanide. The reaction is slower for dry, solid carboxylic acids. Insoluble carboxylic acids react with solutions of cyanides to cause the release of gaseous hydrogen cyanide. Flammable and/or toxic gases and heat are generated by the reaction of carboxylic acids with diazo compounds, dithiocarbamates, isocyanates, mercaptans, nitrides, and sulfides. Carboxylic acids, especially in aqueous solution, also react with sulfites, nitrites, thiosulfates (to give H₂S and SO₃), dithionites (SO₂), to generate flammable and/or toxic gases and heat. Their reaction with carbonates and bicarbonates generates a harmless gas (carbon dioxide) but still heat. Like other organic compounds, carboxylic acids can be oxidized by strong oxidizing agents and reduced by strong reducing agents. These reactions generate heat. A wide variety of products is possible. Like other acids, carboxylic acids may initiate polymerization reactions; like other acids, they often catalyze (increase the rate of) chemical reactions.

Toxicity. Corrosive to tissue.

Other characteristics. Carboxylic acids contain the carboxylic acid functional group (the -COOH group) linked to a hydrocarbon chain. This chain may be a simple alkyl group (containing carbon and hydrogen only and all single bonds) but may also include any of a large number of other organic functional groups, including additional carboxylic acid groups. Carboxylic acids have a sour taste and turn blue litmus red. Most carboxylic acids are solids at room temperature; formic, acetic, propanoic, and butanoic acids are liquids.

Examples. Formic acid, acetic acid, oxalic acid, propionic acid, trichlorophenoxyacetic acid, benzoic acid, citric acid, fumaric acid, the fatty acids.

CFCs and HCFCs (chlorofluorocarbons and chlorofluorohydrocarbons)

Flammability. These compounds are generally not flammable; some are combustible, but usually only poorly.

Reactivity. These materials are chemically inert in many situations, but can react violently with strong reducing agents such as the very active metals and the active metals. They suffer oxidation with strong oxidizing agents and under extremes of temperature.

Toxicity. Generally low.

Other characteristics. In these compounds chlorine and fluorine replaces some hydrogen in a hydrocarbon. The presence of fluorine lowers the reactivity relative to unfluorinated compounds. This category also includes compounds in which bromine replaces some of the chlorine atoms. All compounds are gases under standard conditions.

Examples. 1,1,2-trichloro-1,2,2-trifluoroethane, 1,2-dichloro-1,1,2,3,3-pentafluoropropane. 1-chloro-1,1-difluoroethane, 1-chloro-1,1,2,2-tetrafluoroethane, 1-chloro-2,2,2-trifluoroethane, 1,1-dichloro-1,2,2-trifluoroethane.

Chlorosilanes

Flammability. Flammable or pyrophoric.

Reactivity. Chlorosilanes are compounds in which silicon is bonded to from one to four chlorine atoms with other bonds to hydrogen and/or alkyl groups. Chlorosilanes react with water, moist air, or steam to produce heat and toxic, corrosive fumes of hydrogen chloride. They may also produce flammable gaseous H₂. They can serve as chlorination agents.

Chlorosilanes react vigorously both organic and inorganic acids and with bases to generate toxic or flammable gases.

Toxicity. Poisonous by ingestion and inhalation and a poisonous irritant to the skin, eyes, and mucous membranes. The toxicity is based on that of HCl, which forms by the reaction of chlorosilanes with water.

Examples. Allyltrichlorosilane, methylchlorosilane, silicon tetrachloride, vinyl trichlorosilane, trimethylchlorosilane, trichlorosilane, propyltrichlorosilane, phenyltrichlorosilane, methyl trichlorosilane.

Cyanides, Inorganic

Flammability. Materials in this group are generally noncombustible, except hydrogen cyanide (HCN), which is flammable and dangerous because of its wide flammability limits. Cyanides are often flammable by chemical reaction with moisture or acid; they react readily to evolve flammable HCN. Heating inorganic cyanides to decomposition liberates toxic gases and flammable gases that include HCN if traces of moisture are present.

Reactivity. Materials in this group react slowly with water to evolve gaseous hydrogen cyanide (HCN). Acids cause the rapid evolution of HCN; carbon dioxide from the air is sufficiently acidic to liberate HCN from solutions of cyanides. Inorganic cyanides are incompatible with isocyanates, nitrides, and peroxides. Cyanides have been known to initiate polymerization reactions of epoxides. Cyanides form compounds with metal salts; heat and hydrogen production may accompany these reactions. Some cyanides can detonate when exposed to shock, heat, or friction.

Toxicity. Many of these materials are corrosive to tissue and are extremely toxic by all routes of exposure. Cyanides bind hemoglobin in the blood, and many enzymes are inactivated in the presence of cyanide. Cyanogen chloride and bromide are lachrymators.

Other characteristics. Compounds in this group contain the characteristic -CN group. Cyanides are used as pesticides and in chemical synthesis.

Examples. Barium cyanide, calcium cyanide, copper cyanide, cyanogen chloride, hydrogen cyanide, lead cyanide, potassium cyanide, silver cyanide.

Epoxides

Flammability. Epoxides present serious fire and explosion hazards. Vapors are heavier than air, so flashback to source is a danger, along with rupture of a closed container.

Reactivity. Materials in this group are highly reactive. They polymerize in the presence of catalysts or when heated. These polymerization reactions can be violent. Compounds in this group react with acids, bases, and oxidizing and reducing agents. They react, possibly violently with water in the presence of acid and other catalysts.

Toxicity. Corrosive and irritating to mucous membranes, eyes, respiratory tract. Direct contact with skin will cause irritation. Toxic when absorbed through the skin. Fumes are serious inhalation hazard. Certain people have a particular sensitivity to epoxides and the resins they form.

Other characteristics. Compounds in this group contain a three-membered ring consisting of two carbon atoms and one oxygen atom. They are used in organic synthesis and are polymerized to form epoxy resins. Also used as solvents in synthetic reactions such as gums, cellulose esters, paints, varnishes, enamels, and lacquers.

Examples. Butylene oxide, ethylene oxide, propylene oxide, diepoxy butane, endrin, epibromohydrin, styrene oxide, heptachlor epoxide.

Esters

Flammability. Many esters are flammable or highly flammable. Low molecular weight esters like methyl formate have low flash points and wide flammability limits, making them dangerous flammability hazards. All are

capable of being combustible and yield toxic gases such as carbon monoxide when burned. Some may spontaneously heat and ignite if stored wet and hot.

Reactivity. Materials in this group react with acids to liberate heat along with alcohols and acids. Strong oxidizing acids may cause a vigorous reaction that is sufficiently exothermic to ignite the reaction products. Heat is also generated by the interaction of esters with caustic solutions. Flammable hydrogen is generated by mixing esters with alkali metals and hydrides.

Toxicity. Inhalation of the fumes from some esters irritates the mucous membranes. Esters have low to moderate toxicity via dermal and oral exposure. Some esters are used as flavoring agents in foodstuffs.

Other characteristics. Compounds in this group are formed by replacing the acidic hydrogen of a carboxylic acid with an organic group that is usually derived from an alcohol. They have the general formula RCOOR' . Their names are derived from the names of the acid and alcohol from which they are synthesized. Esters include the edible fats and oils, which are mixed esters between the triol glycerol and fatty acids. Esters are prominent in the perfumery and flavoring industries. They are used to manufacture synthetic fibers such as polyester and plastics.

Examples. Ethyl acetate, methyl acrylate, cyclohexyl acrylate, butyl propionate, butyl lactate, butyl formate, butyl acetate, benzyl acetate, amyl butyrate, amyl acetate, allyl acetate.

Ethers

Flammability. Low-molecular-weight ethers are flammable gases (methyl ether) or liquids (methyl ethyl ether or ethyl ether) that can produce explosive mixtures when they evaporate and mix with air. Ethers form peroxides if exposed to oxygen or air during storage; the ether peroxides can detonate with friction, shock, or heat, releasing enough energy to start a secondary fire in the unperoxidized ether.

Reactivity. Ethers tend to form unstable peroxides when exposed to oxygen. Ethyl, isobutyl, ethyl tert-butyl, and ethyl tert-pentyl ether are particularly hazardous in this respect. Ether peroxides can sometimes be observed as clear crystals deposited on containers or along the surface of the liquid. Ethers can act as bases. They form salts with strong acids and addition complexes with Lewis acids. The complex between diethyl ether and boron trifluoride is an example. Ethers may react violently with strong oxidizing agents. In other reactions, which typically involve the breaking of the carbon-oxygen bond, ethers are relatively inert.

Toxicity. As a group, ethers are moderately toxic. The lower-molecular weight ethers are powerful rapid-acting narcotics by inhalation. Toxic effects from these compounds are usually acute not chronic. The swallowing of one to two ounces of a liquid ether may be fatal.

Other characteristics. Ethers contain an oxygen atom bonded to two organic groups, that is, they have general formula $R-O-R'$, where the R and R' represent organic groups. Ethers are prepared by the Williamson synthesis (the reaction of a metallic alkoxide with an alkyl halide) and also by the reaction of alkyl hydrogen sulfates with alcohols. This amounts to the mixing of alcohols with sulfuric acid and heating. This process gives ethers as co-products in the preparation of alcohols from olefins in the presence of H_2SO_4 .

Examples. Phenyl ether, ethyl ether, allyl ethyl ether, anisidine, anisole, butyl methyl ether, butyl vinyl ether, diallyl ether, isopropyl ether, divinyl ether, methyl ether, isobutyl.

Halogenated Organic Compounds

Flammability. The flammability of these materials is variable. The low-molecular-weight materials (gases and liquids) are generally dangerously flammable. However, many halogenated organic compounds of high molecular weight, in particular those containing several halogen atoms per molecule, are nonflammable; some are in fact used as fire retardants.

Reactivity. Simple aromatic halogenated organic compounds are very unreactive; halogenated aliphatic compounds are moderately or very reactive. For both subgroups, reactivity generally decreases with increased degree of substitution of halogen for hydrogen atoms. Halogenated acetylene compounds are unstable and should be treated as explosives. Low molecular weight haloalkanes are highly flammable and can react with some metals to form dangerous products. Low molecular weight haloalkenes are highly flammable, peroxidizable and may polymerize violently. They may react violently with aluminum. Materials in this group are incompatible with strong oxidizing and reducing agents. Also, they are incompatible with many amines, nitrides, azo/diazo compounds, alkali metals, and epoxides.

Toxicity. Many halogenated hydrocarbons have moderate to high toxicity by inhalation. The brominated materials tend to be particularly toxic. Much of the toxicity is due to the fact that these substances are not metabolized, but persist and accumulate in fatty tissues (they tend to be fat-soluble). The combustion of chlorinated organic compounds may produce poisonous phosgene gas (COCl_2). Other materials formed by incomplete combustion are classes of chlorinated organic compounds, chlorodibenzodioxins and chlorodibenzofurans. These compounds cause cancer in laboratory tests.

Other characteristics. This class of compounds is extremely important in industry in the production of polymers, pesticides, and fire retardants.

Examples. Vinyl chloride, vinylidene chloride, chlorobenzene, trichlorobenzene, polychlorinated biphenyls (PCBs), methylene chloride, chloroform, chloroisoprene, trichlorobenzene.

Halogenating Agents, Strong

Flammability. These compounds are technically nonflammable, but are combustible. They can support the combustion of other materials.

Reactivity. Strong halogenating agents force the addition of one or more halogen atoms to the compound with which they are reacting. Fluorinating

agents can react violently with halide salts. Many of these compounds are water-reactive and air-reactive.

Toxicity. Generally toxic.

Examples. Antimony pentafluoride, molybdenum pentachloride, nitrogen fluoride oxide, antimony pentachloride, tungsten hexafluoride, tellurium hexafluoride, sulfur tetrafluoride, sulfur monochloride, silicon tetrafluoride, phosphorus pentafluoride.

Hydrocarbons, Aliphatic Saturated

Flammability. The lowest-molecular-weight compounds pose significant vapor cloud explosion hazards. Ethane, propane, and butane are gases that have been involved in many explosions. All aliphatic saturated hydrocarbons are combustible.

Reactivity. Materials in this group may be incompatible with strong oxidizing agents like nitric acid. Charring of the hydrocarbon may occur followed by ignition of unreacted hydrocarbon and other nearby combustibles. In other settings, aliphatic saturated hydrocarbons are mostly unreactive. They are not affected by aqueous solutions of acids, alkalis, most oxidizing agents, and most reducing agents. When heated sufficiently or when ignited in the presence of air, oxygen or strong oxidizing agents, they burn exothermically to produce carbon dioxide and water.

Toxicity. Compounds in this group have low toxicity. They act as asphyxiants.

Other characteristics. Compounds in this group are characterized by straight or branched carbon chains with the generic formula $C(n)H(2n+2)$. They are also known as alkanes or paraffins. Their physical form varies with increasing molecular weight from gaseous (methane) to solid. The solids are waxy and soft.

Examples. Propane, butane, pentane, cyclobutane, cycloheptane, hexane, cyclohexane, isopentane, cyclopentane, decane, dimethylpropane, ethane, heptane, isobutane, dodecane, isohexane, methane, isododecane.

Hydrocarbons, Aliphatic Unsaturated

Flammability. The lightest-molecular weight substances in this group are highly flammable and pose significant vapor cloud explosion hazards (examples are ethylene and propylene). All are combustible.

Reactivity. The unsaturated aliphatic hydrocarbons are generally much more reactive than the alkanes, which are saturated aliphatic hydrocarbons. Strong oxidizers may react vigorously with them. Reducing agents can react exothermically to release gaseous hydrogen gas. In the presence of various catalysts (such as acids) or initiators, compounds in this class can undergo very exothermic addition polymerization reactions. Many of these compounds undergo autoxidation upon exposure to the air to form explosive peroxides (this process generally occurs slowly). These peroxide and polyperoxide substances are usually extremely unstable and liable to detonation. The peroxidation of butadiene has been involved in several serious industrial explosion accidents.

Toxicity. Unsaturated aliphatic hydrocarbons have low toxicity, but act as asphyxiants.

Other characteristics. Unsaturated aliphatic hydrocarbons are characterized by straight or branched carbon chains containing double and triple bonds between the carbon atoms. They are also known as alkenes (or olefins) and alkynes. Their physical state at room conditions changes with increasing molecular weight from gaseous to waxy solid. They are used in making rubber and plastics and in organic synthesis.

Examples. Butadiene, pentadiene, butyne, butene, pentene, acetylene, amylenes, cycloheptatriene, cycloheptane, cyclohexane, cyclooctadiene, isobutylene, dipentene, ethylene, hexene, isohexene, isoprene, methyl pentadiene.

Hydrocarbons, Aromatics

Flammability. Aromatic hydrocarbons are all combustible. The lightest members of this class, benzene and substituted benzenes, have high enough vapor pressures and low enough flash points to be vapor explosion hazards.

Reactivity. Vigorous reactions, sometimes amounting to explosions, can result from the contact between these materials and strong oxidizing agents. They can react exothermically with bases and with diazo compounds. Substitution at the benzene nucleus occurs by halogenation (acid catalyst), nitration, sulfonation, and the Friedel-Crafts reaction.

Toxicity. Many aromatic hydrocarbons are poisons by inhalation and irritants to the skin and eyes. Some are poisons by skin contact. Aromatic hydrocarbons are very often carcinogenic (benzene is an example).

Other characteristics. Compounds in this group contain the benzene nucleus. In benzene six carbon atoms form a ring maintained by alternating single and double bonds; each carbon atom is also bonded to a hydrogen atom. Aromatic hydrocarbons possess a single benzene nucleus with side-groups replacing one or more of the hydrogen atoms or in fused sets of two or more rings. They are common air contaminants.

Examples. Benzene, toluene, naphthalene, pyrene, anthracene, chrysene, acenaphthylene, xylene, benzo[a]pyrene, biphenyl, cumene, fluorene, phenanthrene.

Inorganic Compounds/Neither Reducing nor Oxidizing

Flammability. These compounds are nearly all nonflammable. Some are combustible; the majority are not. Those that are combustible are usually only poorly combustible.

Reactivity. These materials have weak oxidizing or reducing powers. Redox reactions can however still occur. For example, CO₂, which is often

regarded as chemically inert, vigorously oxidizes the strong reducing agent Mg if the two are heated together. The majority of compounds in this class are slightly soluble or insoluble in water. If soluble in water, then the solutions are usually neither strongly acidic nor strongly basic. These compounds are not water-reactive. Some do react with acids: carbonates generate carbon dioxide and heat when treated with acids; fluorides, sulfites and sulfides generate toxic gases (hydrogen fluoride, sulfur dioxide and hydrogen sulfide, respectively) when treated with acids.

Toxicity. Most are toxic by ingestion; degree varies widely. Arsenates and arsenites are often quite toxic by skin contact, and inhalation of dust.

Examples. Iron(III) oxide, sodium chloride, lead sulfate, lead arsenate, aluminum fluoride, aluminum oxide, asbestos, barium chloride, beryllium sulfate, cadmium bromide, cadmium stearate, calcium chloride.

Inorganic Oxidizing Agents

Flammability. Materials in this group do not burn in themselves, but enhance the combustion of other substances.

Reactivity. Inorganic oxidizing agents can react with reducing agents to generate heat and products that may be gaseous (causing pressurization of closed containers). The products may themselves be capable of further reactions (such as combustion in the air). The chemical reduction of materials in this group can be rapid or even explosive, but often requires initiation (heat, spark, catalyst, addition of a solvent). Explosive mixtures of inorganic oxidizing agents with reducing agents often persist unchanged for long periods if initiation is prevented. Such systems are typically mixtures of solids, but may involve any combination of physical states. Some inorganic oxidizing agents are salts of metals that are soluble in water; dissolution dilutes but does not nullify the oxidizing power of such materials. Organic compounds in general have some reducing power and can in principle react with compounds in this class. Actual reactivity varies greatly with the identity of the organic compound. Inorganic oxidizing agents can react violently with active metals, cyanides, esters, and

thiocyanates. Explosives often consist of an inorganic oxidizing agent mixed in intimate contact with a reducing agent. Gunpowder is such a mixture. Other examples are a mixture of sugar (an organic compound) plus sodium chlorate and magnesium (an inorganic reducing agent) plus barium peroxide. Compounds that inherently contain a group that is a reducing agent and an oxidizing agent are classed in both Inorganic Oxidizing Agents and in Inorganic Reducing Agents; for example, ammonium nitrate. The strongly oxidizing elements oxygen and fluorine are classified here. Inorganic oxidizing agents that are also acids (such as nitric and perchloric acids) are not included in this group. They are classed in Acids, Inorganic Oxidizing.

Toxicity. Most are toxic by ingestion; degree varies widely.

Other characteristics. Inorganic oxidizing agents include nitrates, nitrites, permanganates chromates, bromates, iodates, chlorates, perchlorates, chlorites, hypochlorites, inorganic peroxides, and certain oxides. Although, in principle, inorganic sulfates are oxidizing agents, they are not included in this group because the reactions tend to be slow. The most dangerous oxidizing agents are the chlorites, chlorates, and perchlorates, in that order. Acidic media favor reactions in which these substances act as oxidizing agents.

Examples. Ammonium dichromate, ammonium nitrate, ammonium perchlorate, ammonium permanganate, barium bromate, barium chlorate, barium peroxide, cadmium chlorate, calcium chlorate, calcium chromate, calcium perchlorate, chromium nitrate, cobalt nitrate.

Inorganic Reducing Agents

Flammability. Nonflammable.

Reactivity. Compounds in this class react with oxidizing agents to generate heat and products that may be flammable, combustible, or otherwise reactive. Their reactions with oxidizing agents may be violent. Sulfites and hydrosulfites (dithionites) can react explosively with strong

oxidizing agents. Sulfites generate gaseous sulfur dioxide in contact with oxidizing acids and non-oxidizing acids.

Toxicity. Most are toxic by ingestion; degree varies widely.

Other characteristics. Organometallic compounds are reducing agents, but are not included in this group; they are regarded as organic and separated in Organometallics. Chlorosilanes are excluded from this group and put in Chlorosilanes. Elemental metals are classified in the various Metal groups. However, sulfur and other non-metallic, elemental reducing agents appear here.

Examples. Ammonium bisulfite, ammonium sulfite, ammonium thiosulfate, arsenic sulfide, arsenic trisulfide, calcium dithionite, chromous chloride, ferrous chloride, ferrous oxalate.

Insufficient Information for Classification

These substances are assigned to hazard classes in accordance with the label information. They may undergo chemical reactions in addition to those implied by these assignments, but insufficient information is available to allow assignment to a reactivity group.

Isocyanates and Isothiocyanates, Organic

Flammability. These materials usually have low vapor pressures and are non-flammable. All are combustible. They evolve poisonous gases such as HCN, NO_x, SO_x, and CO when they burn.

Reactivity. Isocyanates and thioisocyanates are incompatible with many classes of compounds, reacting exothermically to release toxic gases. Reactions with amines, aldehydes, alcohols, alkali metals, ketones, mercaptans, strong oxidizers, hydrides, phenols, and peroxides can cause vigorous releases of heat. Acids and bases initiate polymerization reactions in these materials. Some isocyanates react with water to form amines and

liberate carbon dioxide. Polyurethanes are formed by the condensation reaction of diisocyanates with, for example, ethyl glycol.

Toxicity. Upon direct exposure members of this chemical class have moderate toxicity. Vapors irritate mucous membranes. Organic isocyanates and isothiocyanates are known sensitizers for a small percentage of people, causing chemically induced allergic reactions. The effects of these reactions can be dramatic at rather low concentrations.

Other characteristics. Compounds in this group have the general formulas $R-N=C=O$ and $R-N=C=S$ where R represents an organic group. Isocyanates are isomers of cyanates, which have the general formula $R-O=C=N$. No organic cyanates exist in the monomeric state. They immediately trimerize to alkyl cyanurates, which are also in this reactivity group.

Examples. Phenylene diisocyanate, cyclohexane diisocyanate, naphthalene diisocyanate, cyclohexyl isocyanate, ethyl isocyanate, isobutyl isocyanate, isopropyl isocyanate, methyl isocyanate, n-butyl isocyanate, phenyl isocyanate, propyl isocyanate.

Ketones

Flammability. Many low-molecular-weight ketones (such as acetone and methyl ethyl ketone) are highly flammable. Most ketones are liquids with relatively high vapor pressures, capable of forming explosive mixtures with air.

Reactivity. Materials in this group are reactive with many acids and bases liberating heat and flammable gases (e.g., H_2). The amount of heat may be sufficient to start a fire in the unreacted portion of the ketone. Ketones react with reducing agents such as hydrides, alkali metals, and nitrides to produce flammable gas (H_2) and heat. Ketones are incompatible with isocyanates, aldehydes, cyanides, peroxides, and anhydrides. They react violently with aldehydes, HNO_3 , $HNO_3 + H_2O_2$, and $HClO_4$.

Toxicity. Varies very widely. Some ketones are highly volatile and may have narcotic or anesthetic effects. Entry into the body occurs by absorption through the skin as well as inhalation and ingestion.

Other characteristics. Compounds in this group are characterized by a carbonyl attached to two organic groups. These groups may be alkyl (paraffins) or aryl (aromatic). Reactions of this group are very similar in their behavior to that of aldehydes, because of their similar structure. These materials are generally used as solvents in the paint, textiles, plastics, and lacquer industries.

Examples. Acetone, acetophenone, cyclohexanone, diethylketone, diisobutylketone, dipropylketone, isophorone, methyl acetone, methyl amyl ketone, methyl butanone, methyl ethyl ketone.

Metal Hydrides, Alkyls and Aryls

Flammability. Many of these compounds are pyrophoric (flammable by spontaneous chemical reaction in air). They produce highly flammable gases in the presence of acid. The metal hydrides are combustible. Combustion of all compounds in this class produces irritating and toxic gases.

Reactivity. Materials in this group are reducing agents and react rapidly and dangerously with oxygen and with other oxidizing agents, even weak ones. Thus, they are likely to ignite on contact with alcohols. Hydrides are incompatible with acids, alcohols, amines, and aldehydes.

Toxicity. Extremely corrosive and irritating to mucous membranes, eyes, and the respiratory tract. Fumes are a serious inhalation hazard. Some of these materials are gases and are extremely poisonous by inhalation.

Other characteristics. Compounds in this group are generally used as reducing agents in industrial chemical synthesis and may act as catalysts.

Examples. Aluminum borohydride, aluminum hydride, amyl trichlorosilane, calcium hydride, diborane, lithium aluminum hydride, lithium hydride, magnesium hydride, potassium borohydride, sodium borohydride, sodium hydride.

Metals, Alkali, Very Active

Flammability. The alkali metals are nonflammable, but they are combustible. They may also burn in carbon dioxide and in nitrogen. Their reaction with water is violently rapid and quite exothermic. It produces gaseous hydrogen and other products. The heat that is generated is sufficient to melt the unreacted metal, ignite the liberated hydrogen and ignite the metal itself.

Reactivity. Materials in this group react, usually vigorously, with any substance having active hydrogen atoms to liberate gaseous hydrogen. This includes alcohols and acids, and most importantly, water. They react with sulfides, any oxidizing agent, aldehydes, and cyanides. They corrode rapidly on contact with the air, becoming coated with white oxides and peroxides. As a practical matter, most other organic compounds, including ethers, are wet enough (carry enough water as an impurity) to cause heating and liberation of gases in contact with alkali metals. Alkali metals and alloys containing alkali metals are all powerful reducing agents.

Toxicity. All the materials in this group react with moisture to generate caustic products. Contact with these metals with the skin can cause chemical burns by the action of these products and thermal burns from the heat of the reaction.

Other characteristics. Materials in this group are the elements in the Group IA of the periodic table and their alloys.

Examples. Cesium, lithium, potassium, sodium.

Metals, Elemental & Powder, Active

Flammability. Materials in this group are combustible, especially as finely divided powders. Powdered iron has exploded in the presence of air, moisture, and small amounts of organic impurities.

Reactivity. All of these materials are reducing agents and tend to react with oxidizing agents. Their reactivity is strongly influenced by their state of subdivision: in bulk they often resist chemical combination; in powdered form they may react very rapidly. Thus, bulk aluminum is used as a structural metal, but finely divided aluminum is pyrophoric. Many of these materials react exothermically with compounds that have active hydrogen atoms (such as acids and water) to form flammable hydrogen gas and caustic products. The reactions are less vigorous than the similar reactions of alkali metals, but the released heat can still ignite the released hydrogen. Materials in this group may react with azo/diazo compounds to form explosive products. These metals and the products of their corrosion by air and water can catalyze polymerization reactions in several classes of organic compounds; these polymerizations sometimes proceed rapidly or even explosively. Some metals in this group form explosive products with halogenated hydrocarbons.

Toxicity. Caustic products formed from the corrosion of some of these metals by air or water can cause chemical burns. Inhalation of metal powders, dusts and fumes can cause serious symptoms. Even metals that are less toxic by inhalation (such as zinc) often contain toxic impurities (cadmium, antimony, arsenic, and lead). Iron dust causes conjunctivitis and lodges in the lungs to induce cancers.

Other characteristics. Many alloys of these metals exist and have structural uses. The metals are used chemically as reducing agents. The metals and their compounds are used as catalysts in the synthesis of plastics, rubber, pharmaceuticals, and pesticides.

Examples. Aluminum, aluminum ferrosilicon, antimony, beryllium, calcium silicon, cerium, cobalt, gallium, hafnium, iron, magnesium alloy,

nickel catalyst, selenium, silicon, silver, strontium, titanium, zinc, zirconium.

Metals, Less Reactive

Flammability. Nonflammable. Not reactive enough to burn in air, even if finely divided.

Reactivity. Metals in this group can react exothermically with oxidizing acids to form noxious gases. Many catalyze polymerization and other reactions, particularly when finely divided. Metals in this group have been known to react with halogenated hydrocarbons, sometimes forming explosive compounds (for example, copper dissolves when heated in carbon tetrachloride). They are less reactive in massive form (sheet, rod, or drop) than when finely divided.

Toxicity. Fumes from the hot metals are toxic by inhalation.

Other characteristics. This reactive group includes items (such as certain munitions) that employ less reactive metals or their alloys for housings, supports, tanks, and other structural purposes.

Examples. Copper, lead, silver, gold.

Nitrides, Phosphides, Carbides, and Silicides, Inorganic

Flammability. These compounds react readily with oxygen or air. They are often pyrophoric, especially if finely divided. The transition metal nitrides are often explosively unstable.

Reactivity. Materials in this group are reducing agents. They generate flammable or noxious gases in contact with water. Many nitrides react with moisture to evolve gaseous ammonia, which is toxic and combustible.

Phosphides react quickly upon contact with moisture or acids to give the very toxic gas phosphine; phosphides also can react vigorously with oxidizing materials. Inorganic amides react with moisture to evolve ammonia. Inorganic carbides react with water to generate flammable gases. The transition metal nitrides react violently with water forming the metal hydroxides and ammonia. In general, materials in this group are incompatible with oxidizers such as atmospheric oxygen. They are violently incompatible with acids, particularly oxidizing acids.

Toxicity. Tend to react with water to give corrosive or toxic products. Corrosive to skin and mucous membranes.

Other characteristics. Compounds in this group are nominally salts of the following anions: N^{3-} , P^{3-} , NH_2^- , C^{2-} , and Si^{4-} .

Examples. Sodium amide, calcium carbide, magnesium silicide, lithium nitride, calcium nitride, chromium nitride, aluminum carbide, aluminum phosphide, calcium phosphide, calcium silicide, magnesium aluminum phosphide, magnesium phosphide, potassium phosphide.

Nitriles

Flammability. Many of the compounds of this group are highly flammable, especially substances such as acrylonitrile and acetonitrile that have low flash points and relatively wide flammability concentration limits. All are combustible. The incomplete combustion of these materials produces significant amounts of gaseous hydrogen cyanide. When heated to decomposition, nitriles emit highly toxic fumes.

Reactivity. Nitriles may polymerize in the presence of metals and some metal compounds. They are incompatible with acids; mixing nitriles with strong oxidizing acids can lead to extremely violent reactions. Nitriles are generally incompatible with other oxidizing agents such as peroxides and epoxides. The combination of bases and nitriles can produce hydrogen cyanide. Nitriles are hydrolyzed in both aqueous acid and base to give carboxylic acids (or salts of carboxylic acids). These reactions generate

heat. Peroxides convert nitriles to amides. Nitriles can react vigorously with reducing agents. Acetonitrile and propionitrile are soluble in water, but nitriles higher than propionitrile have low aqueous solubility. They are also insoluble in aqueous acids.

Toxicity. Pure alkyl cyanides are said to be only slightly toxic. Usually however they are contaminated with highly toxic isocyanides and should be treated as serious poisons. They are absorbed by inhalation, ingestion and by contact with the skin. Other nitriles vary in toxicity, with some approaching the cyanides in toxicity.

Other characteristics. Compounds in this group are organic compounds containing the nitrile group (-CN). This group is also called the cyanide group. Probably the most important use of these materials is in the production of nitrile rubber, which is a copolymer of acetonitrile with butadiene. Nitriles such as cyanogen are used in organic synthesis, fumigants, and rocket propellant materials.

Examples. Pentenenitrile, acetonitrile, acrylonitrile, benzonitrile, butyronitrile, cyanoacetic acid, isobutyronitrile, lactonitrile, methacrylonitrile, propionitrile.

Nitrites, Nitrates, and Nitro Compounds, Organic

Flammability. Most materials in this group are technically of low flammability. However, they are often chemically unstable and subject, in widely varying degree, to explosive decomposition. As oxidizing agents, they are often used in combination with reducing agents in explosive mixtures. Or, as in the primary explosive nitroglycerine, the oxidizing and reducing behavior is present internally in the same molecule. Because many of these substances are extremely sensitive to shock, friction, and heat, they are mixed with unreactive material to make them less dangerous.

Reactivity. Materials in this group range from slight to strong oxidizing agents. If mixed with reducing agents, including hydrides, sulfides and

nitrides, they may begin a vigorous reaction that culminates in a detonation. The aromatic nitro compounds may explode in the presence of a base such as sodium hydroxide or potassium hydroxide even in the presence of water or organic solvents. The explosive tendencies of aromatic nitro compounds are increased by the presence of multiple nitro groups. Nitroalkanes are milder oxidizing agents, but still react violently with reducing agents at higher temperatures and pressures. Nitroalkanes react with inorganic bases to form explosive salts. The presence of metal oxides increases the thermal sensitivity of nitroalkanes. Nitroalkanes with more than one nitro group are generally explosive. Azonitrates are extremely unstable and can decompose explosively. Nitromethane is somewhat soluble in water; the higher nitroalkanes are insoluble in water.

Toxicity. Many of the compounds in this group are extremely toxic. Chronic dermal and inhalation exposure causes hepatitis.

Other characteristics. This group includes many subclasses, such as the metal nitrophenoxides, nitroalkanes, nitroalkenes, and nitroalkyl peroxonitrates. These substances can be generally signified by RNO_x , the R being the organic portion of the molecule. Many are used in organic synthesis, but the biggest use of compounds in this group is in military and commercial explosives. Nitromethane is somewhat soluble in water; the higher nitroalkanes are insoluble in water.

Examples. Nitropyrene, nitroglycerine, dinitrophenol, trinitrotoluene, dinitrotoluene, nitropropane, amyl nitrate, nitrobenzene, dinitrocresol, dinitroaniline, ethyl nitrate, glyceryl trinitrate, dinitrobenzene, nitrocellulose, nitroanisole, nitrocyclohexane, nitroethane.

Not Chemically Reactive

These substances undergo chemical reactions only under relatively severe circumstances. They are resistant to ignition, although they may become flammable at very high temperatures. They may be resistant to oxidation and reduction, except in the most severe conditions. These materials may be nontoxic. Some can asphyxiate.

Organometallics

Flammability. Many organometallic compounds are flammable or highly flammable; sometimes they are pyrophoric. They are generally combustible.

Reactivity. Strongly reactive with many other groups. Incompatible with acids and bases. Organometallics are good reducing agents and therefore incompatible with oxidizing agents. Often reactive with water to generate toxic or flammable gases. Organometallics containing halogens (fluorine, chlorine, bromine, iodine) bonded to the metal typically will generate gaseous hydrohalic acids (HF, HCl, HBr, HI) with water, with heat released.

Toxicity. Generally highly toxic. Often react on contact with tissues to give toxic products. Irritating to skin, eyes, and mucous membranes. Can damage lung tissue and the liver.

Other characteristics. These compounds contain carbon bonded directly to a metal. This means that metallic carbonates are excluded and also metallic salts of carboxylic acids and other organic acids. Many chlorosilanes can be regarded as organometallic. However, these compounds are listed in the Chlorosilanes group.

Examples. Bis(1,5-cyclooctadiene)nickel, bis(cyclopentadienyl)chromium, bis(cyclopentadienyl)vanadium chloride, bis(cyclopentadienyl)vanadium dichloride, bis(tributyltin)oxide, chromium carbonyl, cobalt carbonyl, cobaltocene, dibutyltin diacetate, ethyl aluminum chloride.

Peroxides, Organic

Flammability. Generally, materials in this group are readily combustible. They behave as strong oxidants and therefore accelerate the combustion of other materials by providing molecular oxygen to the combustion site. Many peroxides are unstable and subject to explosive decomposition when

shocked, heated, or rubbed. Explosions of peroxides have caused many fatal accidents.

Reactivity. Peroxides are good oxidizing agents. Organic compounds can ignite on contact with concentrated peroxides. Strongly reduced material such as sulfides, nitrides, and hydrides may react explosively with peroxides. There are few chemical classes that do not at least produce heat when mixed with peroxides. Many produce explosions or generate gases (toxic and nontoxic). Generally, dilute solutions of peroxides (<70%) are safe, but the presence of a catalyst (often a transition metal such as cobalt, iron, manganese, nickel, or vanadium) as an impurity may even then cause rapid decomposition, a build-up of heat, and even an explosion. Solutions of peroxides often become explosive when evaporated to dryness or near-dryness.

Toxicity. Often highly toxic and irritating to the skin, eyes, and mucous membranes.

Other characteristics. The structure of the simplest peroxide H_2O_2 is H-O-O-H. Organic peroxides derive by the replacement of one or both of the H atoms in this compound by organic groups. Hydroperoxides result from the replacement of just one of the H atoms by an organic group. Thus, ethyl hydroperoxide has the formula $C_2H_5-O-O-H$. Some organic compounds form dangerous levels of explosive peroxides by autoxidation when exposed to the air during storage. These peroxides generally form slowly and are explosive when taken to dryness. Many peroxide solutions are inhibited to prevent decomposition to give oxygen and other products. Such reactions are often catalyzed by impurities in the absence of inhibition. When it occurs, such decompositions release very reactive free radicals that can initiate other reactions: peroxides are widely used as polymerization initiators. Peroxides are used in the paper industry (for bleaching), in making textiles (for bleaching), as oxidizers in organic synthesis reactions, as blowing agents, and in propellant formulations.

Examples. Benzoyl peroxide, peroxybenzoic acid, acetyl peroxide, cumene hydroperoxide, cyclohexanone peroxide, octanyl peroxide, diacetone alcohol peroxide.

Phenols, Cresols

Flammability. Materials in this group are combustible. Some nitro and diazo derivatives of phenols explode when heated.

Reactivity. Compounds in this group do not behave as organic alcohols, as one might guess from the presence of a hydroxyl (-OH) group in their structure. Instead, they react as weak organic acids. Phenols and cresols are much weaker as acids than common carboxylic acids (phenol has $K_a = 1.3 \times 10^{-10}$). These materials are incompatible with strong reducing substances such as hydrides, nitrides, alkali metals, and sulfides. Flammable gas (H_2) is often generated, and the heat of the reaction may ignite the gas. Heat is also generated by the acid-base reaction between phenols and bases. Such heating may initiate polymerization of the organic compound. Phenols are sulfonated very readily (for example, by concentrated sulfuric acid at room temperature). The reactions generate heat. Phenols are also nitrated very rapidly, even by dilute nitric acid. Nitrated phenols often explode when heated. Many of them form metal salts that tend toward detonation by rather mild shock.

Toxicity. Ranges from moderate to high toxicity. Some of the more toxic members of this class of compounds are the chlorinated and nitro-substituted phenols that are used as pesticides and antibacterials (dinoseb, creosote, chlorinated phenol, p-nitrophenol). Phenol is strongly irritating to the skin.

Other characteristics. Phenols are used in many organic syntheses, and the making of phenol/formaldehyde resins and polycarbonate resins.

Examples. Phenol, cresol, dinitrophenol, phenylphenol, dinitrocresol, alkyl phenol, aminochlorophenol, aminophenol, amyl phenol, butyl phenol, catechol, creosote, picric acid, resorcinol, trichlorophenol, dichlorocresol.

Phosphates and Thiophosphates, Organic

Flammability. These materials are combustible. Burning them generates highly toxic fumes of phosphorus oxides and (in the case of the thiophosphates) sulfur oxides. Many fire-retardant foams are organophosphate-based and act to lay down a protective layer that excludes oxygen.

Reactivity. Materials in this group are susceptible to formation of highly toxic and flammable phosphine gas in the presence of strong reducing agents such as hydrides. Partial oxidation by oxidizing agents may result in the release of toxic phosphorus oxides.

Toxicity. Variable. The pyrophosphate esters (such as tetraethyl pyrophosphate) are highly toxic. These derivatives form the basis of a group of insecticides. They interfere with normal nerve transmission by inhibiting cholinesterase. These pesticides should be handled with great care (when released to the environment they are rapidly degraded). Other organophosphates behave similarly.

Other characteristics. These compounds are derivatives of phosphoric acid and phosphonic acid in which S may substitute for O and organic groups may substitute for H. They include phospholipids or phosphatides, which are distributed widely in nature in the form of lecithin, certain proteins, and nucleic acids; esters of phosphinic and phosphonic acid, which are used as plasticizers, insecticides, resin ingredients, and flame retardants; pyrophosphates (or diphosphates) which form when phosphate groups condense with the elimination of a molecule of water; phosphate esters of polyols, which are used as components of fertilizer mixtures.

Examples. Acephate, chlormephos, chlorfenvinfos, coumaphos, demeton, diazinon, dimefox, dimethoate, ethoprophos, fenamiphos, fonofos, hexaethyl tetraphosphate, isofenphos, isofluorphate, leptophos, merphos, mevinphos, monocrotophos, profenofos, propetamphos.

Salts, Acidic Inorganic/Organic

Flammability. None of these materials are highly flammable. Inorganic salts are generally noncombustible as well. Salts containing organic groups are in principle combustible, although they may burn with difficulty.

Reactivity. Materials in this group are generally soluble in water. The resulting solutions contain moderate concentrations of hydrogen ions and have pHs of less than 7.0. They react as acids to neutralize bases. These neutralizations generate heat, but less or far less than is generated by neutralization of inorganic acids, inorganic oxoacids, and carboxylic acid. They usually do not react as either oxidizing agents or reducing agents but such behavior is not impossible. Many of these compounds catalyze organic reactions.

Toxicity. Widely variable. Solutions of these materials are generally corrosive to skin and irritating to mucous membranes.

Examples. Ferric chloride, aluminum sulfate, ammonium bromide, antimony tribromide, cupric oxalate, ferric sulfate.

Salts, Basic Inorganic/Organic

Flammability. None of these materials are highly flammable. Inorganic salts are generally noncombustible as well. Salts containing organic groups are in principle combustible, although they may burn with difficulty.

Reactivity. Materials in this group are generally soluble in water. The resulting solutions contain moderate concentrations of hydroxide ions and have pHs greater than 7.0. They react as bases to neutralize acids. These neutralizations generate heat, but less or far less than is generated by neutralization of the bases in the Bases reactivity group and the neutralization of amines. They usually do not react as either oxidizing agents or reducing agents but such behavior is not impossible.

Toxicity. Variable. Solutions of these materials can be corrosive to skin and irritating to mucous membranes.

Examples. Aluminum phosphate, aluminum carbonate, ammonium citrate, ammonium formate, ammonium oxalate, cadmium stearate, calcium phosphate, copper formate, lead phosphate, mercurous acetate, nickel acetate, nickel formate, potassium oxalate.

Sulfides, Inorganic

Flammability. The only highly flammable member of this group is hydrogen sulfide. It poses a particular danger because of its relatively wide flammability limits. However, some other inorganic sulfides can heat spontaneously and even ignite if exposed to moisture.

Reactivity. Materials in this group are generally basic and therefore incompatible with acids. Many of these compounds are reducing agents and therefore react vigorously with oxidizing agents, including inorganic oxoacids, organic peroxides and epoxides. Simple salts of sulfides (such as sodium, potassium, and ammonium sulfide) react vigorously with acids to release hydrogen sulfide gas.

Toxicity. Many of the members of this group behave as strong bases. Therefore, direct dermal exposure by these materials may cause severe burns. However, numerous inorganic sulfides are so insoluble (for example, mercury(II) sulfide and cadmium sulfide) that they are not caustic. This property depends upon solubility. Hydrogen sulfide is a toxic gas that irritates the eyes and mucous membranes at concentrations of around 50 ppm (sensitivity varies considerably from person to person). Most of the symptoms of hydrogen sulfide poisoning are reversible if the victim is quickly exposed to fresh air. The gas causes death at concentrations of around 1,000 ppm.

Other characteristics. Compounds in this group tend to have very unpleasant odors.

Examples. Ammonium hydrosulfide, sodium hydrosulfide, ammonium sulfide, antimony sulfide, arsenic sulfide, arsenic trisulfide, lead sulfide, hydrogen sulfide, mercuric sulfide, phosphorus heptasulfide, potassium hydrosulfite, potassium sulfide, selenium disulfide, sodium sulfide.

Sulfides, Organic

Flammability. The low-molecular-weight gaseous and liquid sulfides (such as methyl mercaptan or ethyl mercaptan) are extremely flammable, but flammability diminishes with increasing molecular weight. The products of combustion include sulfur dioxide.

Reactivity. Materials in this group are incompatible with acids, diazo and azo compounds, halocarbons, isocyanates, aldehydes, alkali metals, nitrides, hydrides, and other strong reducing agents. Reactions with these materials generate heat and in many cases hydrogen gas. Many of these compounds may liberate hydrogen sulfide upon decomposition or reaction with an acid.

Toxicity. Most organic sulfides are toxic by inhalation. In general, they are stenchers. Thus methyl mercaptan is added to natural gas (which is odorless) to assist in the detection of leaks.

Other characteristics. Compounds in this group are characterized by an organic group attached to a sulfur atom R-SH; disulfides have the formula R-S-S-R. Their reactions resemble those of alcohols. These chemicals are used in industrial synthesis of chemicals, and are important as catalysts, as additives in petroleum and natural gas, and as pesticides.

Examples. Amyl mercaptan, butyl mercaptan, diamyl sulfide, dimethyl sulfide, ethyl mercaptan, propyl mercaptan, methyl mercaptan.

Thiocarbamate Esters and Salt/ Dithiocarbamate Esters and Salts

Flammability. Materials in this group are combustible. Poisonous gases are generated by the thermal decomposition of compounds of this class, including carbon disulfide, oxides of sulfur, oxides of nitrogen, hydrogen sulfide, ammonia, and methylamine.

Reactivity. Many materials in this group slowly decompose in aqueous solution to form carbon disulfide and methylamine or other amines. Such decompositions are accelerated by acids. Flammable gases are generated by the combination of thiocarbamates and dithiocarbamates with aldehydes, nitrides, and hydrides. Thiocarbamates and dithiocarbamates are incompatible with acids, peroxides, and acid halides.

Toxicity. The slow release of poisonous gases from hydrolysis of many compounds in this group requires the use of respirators during handling. Poisoning can also occur by ingestion and absorption through the skin.

Other characteristics. Compounds in this group are derivatives of thiocarbamic acid (NH_2CSOH) or dithiocarbamic acid ($\text{NH}_2\text{CS}_2\text{H}$). In salts, a metal ion replaces the acidic H atom; in esters, an organic group replaces it. Additional compounds are derived by replacing one or both of the hydrogen atoms on the N with organic groups. These chemicals are used in the synthesis of polymers and in agriculture as pesticides, soil fumigants, and seed disinfectants.

Examples. Cycloate, ethyl dithiocarbamate, ferbam, maneb, ethylenebisdithiocarbamate, metham sodium, nabam, thiram, sodium dimethyldithiocarbamate, thiobencarb, zineb, zinc diethyldithiocarbamate.

Glossary of Terms

A

AAR. Association of American Railroads (<http://www.aar.org>). Trade organization representing the major freight railroads of North America.

Acaricide. A chemical agent used to control tick populations.

Acid. A substance that donates hydrogen ions that can be accepted by a base. Acids have a pH below 7. *Contrast with* **Base**.

Acute. Acute health effects usually develop rapidly after a short-term exposure to hazardous chemicals. Acute effects have a short duration. *Contrast with* **Chronic**.

AEGL. Acute Exposure Guideline Level, a level of concern representing the adverse health effects of a hazardous substance on members of the general public. The AEGLs are developed by the National Research Council's National Advisory Committee

on AEGLs (<http://www.epa.gov/oppt/aegl>). Find out more about AEGLs on the NOAA OR&R website (<http://response.restoration.noaa.gov/aegls>).

Aerobic. Living, growing, and reproducing in an environment with air or oxygen present (e.g., aerobic bacteria). *Contrast with* **Anaerobic**.

Aerosol. Fine liquid droplets (or solid particles) suspended in a gas.

Air-reactive. Likely to react rapidly or violently with dry air or moist air; may generate toxic and corrosive fumes upon exposure to air, or may catch fire.

Alkali. A general description given to strong bases, including the hydroxides and carbonates of the "alkali metal" or group 1A of the Periodic Table. Examples include sodium hydroxide and potassium hydroxide. A synonym for the strong alkalis is **caustics**.

Alkali metal. The univalent metals of Group 1A of the periodic table (lithium, sodium, potassium, rubidium, cesium, and francium), which are all strongly corrosive in the presence of moisture.

Alkaline. Having a high concentration of hydroxide ions in aqueous solution (i.e., a pH greater than 7). Strongly alkaline aqueous solutions are very corrosive to skin, metal, and other materials. *See also* **Base**.

Alkaline-earth. Barium, calcium, and strontium.

Alkane. An organic compound that contains only hydrogen and carbon and has only single bonds.

Alkene. A hydrocarbon that contains at least one carbon-carbon double bond. *See also* **Hydrocarbon**.

Alkyne. Hydrocarbons that contain at least one carbon-carbon triple bond. *See also* **Hydrocarbon**.

Allotrope. An element that can exist in different physical forms, depending on the conditions under which it forms. For instance, oxygen is an allotrope that can be either diatomic (O₂) or triatomic (O₃). The triatomic form requires high energy. Carbon is also an allotrope that forms diamond under intense pressure and heat, but in less energetic environments forms graphite and carbon black.

Alloy. Metals formed as mixtures or solutions (either liquid or solid). Some alloys include nonmetal substances in order to improve particular physical characteristics of that alloy. For instance, hardened steel includes a small percentage of carbon.

ALOHA. Computer model that predicts the area potentially affected by a toxic gas release, fire, or explosion; ALOHA is a component of the CAMEO software suite. Find out more about ALOHA on the NOAA OR&R website (<http://response.restoration.noaa.gov/aloha>).

Amalgam. An alloy of mercury, usually in combination with another metal (e.g., zinc), but sometimes with a nonmetal. Amalgams are used very commonly in dentistry.

Ambient temperature and pressure. Typical outdoor temperature and pressure.

Amorphous. *Lit.* “without shape.” A substance that lacks structure. Generally, amorphous solids lack three-dimensional crystal structure. Most liquids are amorphous, with exceptions such as liquid water, which is highly structured.

Anaerobic. Living, growing, and reproducing in an environment without air or oxygen (e.g., anaerobic bacteria). *Contrast with* **Aerobic**.

ANFO. Ammonium nitrate-fuel oil mixtures used in commercial explosives.

Anhydrous. *Lit.* “without water.” A chemical shipped or stored without water, rather than in solution, is in anhydrous form (anhydrous ammonia is a common example).

Anion. An atom or group of atoms having a negative net charge. *Contrast with* **Cation**.

Antioxidant. A chemical substance that reduces the tendency of oxygen to chemically combine with hydrocarbons in commercial products such as vegetable oils, rubber, petroleum products, and animal fats. These antioxidants are organic compounds added to mixtures in low concentrations.

Aphidicide. A chemical agent used to control aphid populations.

Aqueous. In an aqueous solution, a substance is dissolved in water.

Aromatic compound. An unsaturated organic compound containing one or more rings of carbon atoms in which single and double bonds alternate.

Asphalt. Thick, viscous, petroleum-based mixture used to surface roads, for roofing, and in other products.

Asphyxiant. Can cause unconsciousness or death by displacing oxygen from the air; especially dangerous in confined spaces.

ATSDR. Agency for Toxic Substances and Disease Registry (<http://www.atsdr.cdc.gov>). Federal public health agency that provides health information related to toxic substances.

Autoignition temperature. Minimum temperature at which a substance ignites when no ignition source (e.g., spark or flame) is present.

Autoxidation. The tendency of oxygen to chemically combine with hydrocarbons in commercial products such as vegetable oils, rubber, petroleum products, and animal fats.

Azo dye. A group of dye stuffs widely used for dyeing natural and synthetic fibers as well as nontextile materials. The dye molecule always contains double-bonded nitrogen atoms in the color-imparting portion of the molecule.

B

Base. A substance that accepts hydrogen ions donated by an acid. Bases have a pH greater than 7. *Contrast with Acid.*

Binary reaction. A reaction between two chemicals. CAMEO predicts the results of binary reactions only. When more than two chemicals are mixed, CAMEO predicts reactions between all possible pairs of those chemicals (but not the reaction of all chemicals together).

BLEVE. Boiling Liquid Expanding Vapor Explosion. BLEVEs typically occur in closed storage tanks that contain a liquefied gas, usually a gas that has been liquefied under pressure. A common BLEVE scenario happens when a container of liquefied gas is heated by fire, increasing the pressure within the container until the tank ruptures and fails. When the container fails, the chemical is released in an explosion. If the chemical is above its boiling point when the container fails, some or all of the liquid will flash-boil—that is, instantaneously become a gas. If the chemical is flammable, a burning gas cloud called a fireball may occur if a significant amount of the chemical flash-boils. Potential BLEVE hazards include thermal radiation, overpressure, hazardous fragments, smoke, and toxic byproducts from the fire.

Boiling point. The maximum temperature at which a substance's liquid phase can exist in equilibrium with its vapor phase. Above the boiling point a liquid vaporizes completely. The boiling point depends on a chemical's composition and the applied pressure. As pressure increases, the boiling point of a substance also increases. The boiling point is also the temperature at which the vapor pressure of a liquid is equal to the applied atmospheric pressure.

Brass. An alloy containing copper and zinc in various proportions.

Bronze. An alloy containing copper and tin in various proportions. Various bronze mixtures also include small percentages of aluminum, silicon, and phosphorus.

C

CAMEO. A suite of software programs developed by NOAA and EPA. CAMEO supports a number of information management functions, such as retrieval of chemical-specific information to support emergency response activities, threat zone calculation and plotting for risk assessment, organization and management of EPCRA information, and storage and computer display of area maps. Find out more about CAMEO on the NOAA OR&R website (<http://response.restoration.noaa.gov/cameo>). See also **ALOHA**. See also **MARPLOT**.

CAMEO Chemicals. An online version of CAMEO's chemical library, response data sheets, and reactivity predictions. **Note:** ALOHA, MARPLOT, and the EPCRA modules are not available in the online version of CAMEO. The CAMEO Chemicals website is <http://cameochemicals.noaa.gov>.

CAMEO Map. A MARPLOT map designed to hold all CAMEO-related objects, such as (a) symbol objects representing chemical facilities or special locations, and (b) polyline objects representing routes along which hazardous materials are transported.

Carbohydrate. Organic compounds (composed of carbon, hydrogen, and oxygen) that make up the basic components of plant tissues.

Carbon black. An amorphous form of carbon.

Carcinogen. Capable of causing cancer.

CAS number. Or CAS #. Chemical Abstracts Service Registry number. This chemical identification number, in the format XXX-XX-X, is assigned by the American Chemical Society (<http://www.cas.org/EO/regsys.html>).

Catalyst. A substance that speeds up (catalyzes) a chemical reaction between other substances without itself being chemically changed or consumed. Catalysts are widely used in the chemical industry. For example, an iron/aluminum catalyst is used to synthesize ammonia and a platinum catalyst is used to manufacture nitric acid. *Compare with* **Initiator**. *Contrast with* **Inhibitor**.

Catalyze. To act as a catalyst.

Cation. An atom or group of atoms having a positive net charge. *Contrast with* **Anion**.

Caustic. Strongly basic, with high pH. Very corrosive. See also **Base**.

CDC. Centers for Disease Control and Prevention (<http://www.cdc.gov>). Federal agency that conducts research and provides information about environmental health and other public health threats.

Cellophane. A form of the polymer cellulose, altered to have a clear appearance, and used in packaging.

Cellulose. A carbohydrate polymer that is the abundant cellular matrix found in all plant tissues. Cellulose is a complex polysaccharide.

Ceramics. Any of various substances derived from earthen materials such as clays, silicates, and sand. Ceramics are used in great quantity in industrial applications, such as household structural products, pottery, and so on.

CERCLA. ("Superfund") The Comprehensive Environmental Response, Compensation, and Liability Act of 1980 (<http://www.epa.gov/superfund/policy/cercla.htm>). Creates a framework for federal involvement in response and cleanup activities following hazardous substance releases.

Chelate. The organic portion of a complex composed of a metal atom bonded to nonmetal atoms on the organic portion. Such complexes are extremely important because they form the essential active site in enzymatic or catalytic reactions in nature and in synthetic reactions.

Child module. A module that is below another, related module in CAMEO's file hierarchy. Chemicals in Inventory is a child module relative to the Facilities and Routes modules.

CHRIS code. 3-letter code used by the U.S. Coast Guard to identify individual chemicals in the CHRIS guide to chemical hazards (<http://www.chrismanual.com>).

Chronic. Of long duration, or frequently recurring. Chronic health effects become apparent and/or continue for some time after exposure to hazardous chemicals. *See also Delayed. Contrast with Acute.*

Combustible. Can be ignited and burned. Combustible liquids have flash points between 100 - 200°F (37.8 - 93.3°C).

Combustion. Also burning. A chemical reaction between oxygen and another element or compound that is rapid and exothermic enough to generate heat and light.

Commission. The State Emergency Response Commission, or the Governor if there is no commission, for the State in which a facility is located. *See State Emergency Response Commission (SERC).*

Committee. The Local Emergency Planning Committee (LEPC) for the emergency planning district in which a facility is located. *See Local Emergency Planning Committee (LEPC).*

Compound. The combination of two or more elements into a distinct chemical material.

Concentration. The amount of a chemical present in a given weight or volume of air. Concentration of a gas in air may be expressed in units such as parts per million (by volume) or milligrams per cubic meter.

Condensation. A reaction in which two molecules combine to form a larger molecule as a small molecule is split out. The condensation polymerization of amino acids into proteins occurs with the splitting out of water. Also, the change of water or other substances from vapor to liquid phase.

Confirmed animal carcinogen. Increases the risk of cancer in animals that are given a relatively high dose, but scientific evidence doesn't indicate that it increases the risk of cancer in people.

Confirmed human carcinogen. Known to increase the risk of cancer in people, based on scientific studies.

Copolymer. Or polyblend. A polymer generated when two or more different monomers join to form a macromolecule. The copolymerization process is often employed to produce elastomers with very specific physical properties.

Corrosive. Liquid or solid that can destroy human skin or lung tissue or corrode metals.

Cryogenic. Very low temperature.

D

Database. An organized collection of information, usually in an electronic format. Each of CAMEO's modules is a database. Each record in a database describes an important element of information (e.g., each record in the Chemical Library describes a particular substance). Each database record is composed of data fields, each of which stores a particular kind of information (e.g., Chemical Name

is a data field on a Chemical Library record). *See also* **Module**.

Data field. Or **Field.** An area within a record in a CAMEO module where a specific kind of information is stored (e.g., the name of a facility, or a phone number). *See also* **Database**.

Deflagration. Rapid, sharp combustion with sudden evolution of flame. The flame front travels relatively slowly (subsonic speeds), as compared to a supersonic detonation. *Contrast with* **Detonation**.

Delayed. A hazard category that includes carcinogens and other hazardous chemicals, defined in 29 CFR § 1910.1200, that adversely affect a target organ. Such effects generally result from long-term exposure and are of long duration. *See also* **Chronic**.

Deliquescent. The property of becoming liquid by absorbing moisture from the air.

Denature. The addition of a toxic material such as benzene or methyl alcohol (a denaturant) to ethyl alcohol to inhibit or prevent its use for human consumption.

Density. The ratio of the mass (weight) of a substance to the volume it occupies. For example, if 1 cubic foot of a substance weighs 10 pounds, its density is 10 pounds per cubic foot.

Detergent. A natural or synthetic agent that suspends emulsified oils, greases, and fats in solution and, by doing this, acts as a cleaning agent. Examples of detergents include soaps and various alcohols, sulfonated organics like dodecylbenzene, and various alkylates.

Detonation. An explosion where the flame front travels at supersonic speeds as a shock wave. *Contrast with* **Deflagration**.

Dispersion. The movement of molecules or finely divided particles through a gaseous or liquid medium (e.g., the distribution of a toxic chemical cloud in the atmosphere).

Distillation. The process of separating a mixture of materials by heating and then condensing the resulting vapors. The separation is made possible by the inherent differences in boiling point among the individual mixture components (each can be distilled out of the mixture at a particular temperature).

DOT. U.S. Department of Transportation (<http://www.dot.gov>).

DOT label. Required DOT hazard warning label for the chemical (e.g., “Flammable Liquid,” “Corrosive”).

E

Efflorescent. Tending to lose moisture to the air as shown by the formation of a powder on the surface.

EHS. Extremely Hazardous Substance. EHS chemicals have been identified by the U.S. Environmental Protection Agency as particular toxic threats. They are listed under EPCRA in the appendices to 40 CFR 355, Emergency Planning and Notification.

Elastomer. Any of a group of rubber-like polymers that have essentially the same properties as natural rubbers.

Electrolyte. In a battery, the material that acts as a conductor of electric current between the dissimilar metal electrodes. These are ionically-bonded, inorganic salts that form ions in solution and thus increase electrical conductivity.

Emulsification. Generally, either the dispersion of oil particles in water or water in oil. The emulsification process is relatively permanent, and may proceed with the help of an emulsification agent.

Endothermic. A chemical reaction that must absorb heat from its surroundings in order to proceed. *Contrast with Exothermic*, in which heat is produced by the forward reaction.

Enzyme. Any of a number of complex proteins that act in extremely specific chemical reactions to accelerate reactions or catalyze a specific reaction. The rates of reaction are often many orders of magnitude faster than the rates for the same reaction at standard temperature and pressure conditions. All metabolic processes in living organisms are governed by enzymes.

EPA. U.S. Environmental Protection Agency (<http://www.epa.gov>).

EPCRA. The Emergency Planning and Community Right-to-Know Act (“Title III of SARA”). Defines 360 chemicals as Extremely Hazardous Substances (EHSs) and requires planning for and reporting of EHS releases from facilities. See “EPCRA” on [page 261](#).

ERPG. Emergency Response Planning Guideline, a level of concern representing the adverse health effects of a hazardous substance on members of the general public. The ERPGs were developed by the ERPG committee of the American Industrial Hygiene Association (<http://www.aiha.org>). Find out more about ERPGs on the NOAA OR&R website (<http://response.restoration.noaa.gov/erpgs>).

Exothermic. A term used to describe the generation of heat from a chemical reaction. *Contrast with Endothermic*, in which heat is taken in from the surroundings.

Explosion. A chemical or physical process in which the rate at which energy is being generated exceeds its ability to escape to its surrounding environment. The hazards of an explosion can include a shock wave, especially near the point of explosion, and projectiles generated by the shock wave.

Explosive. A material synthesized or mixed deliberately to allow the very rapid release of chemical energy (i.e., an explosion). Also, a chemical substance that is intrinsically unstable and liable to detonate under conditions that might reasonably be encountered.

Explosive limits. Or **flammability limits.** The lower explosive limit is the lowest concentration of a vapor in air that catches fire when exposed to a source of ignition. The upper explosive limit is the highest concentration of the vapor in air that catches fire when exposed to a source of ignition. The flash point and explosive limits of a substance are often used as measures of its flammability hazard in a given situation.

Extremely Hazardous Substances (EHS). EHS chemicals have been identified by the U.S. Environmental Protection Agency as particular toxic threats. They are listed under EPCRA in the appendices to 40 CFR § 355, Emergency Planning and Notification.

F

Facility. Defined in Section 302 of EPCRA as all buildings, equipment, structures, and other stationary items located on a single site or on contiguous or adjacent sites and which are owned or operated by the same person (or by any person who controls, is controlled by, or under common control with, such person). For purposes of emergency release notification, the term also includes motor vehicles, rolling stock, and aircraft.

Field. *See* **Data field.**

Fire hazard. A hazard category that includes chemicals described as flammable, combustible liquid, pyrophoric, or oxidizers (as defined by 29 CFR § 1910.1200).

Flammability. A substance's tendency to ignite or take flame. Flammability is closely related to the volatility of a substance (and is not its relative ability to burn in the presence of oxygen with the evolution of heat). *See* **Flammability limits.** *See also* **Flash point.**

Flammability limits. Or **explosive limits** The lower flammability limit is the lowest concentration of a vapor in air that catches fire when exposed to a source of ignition. The upper flammability limit is the highest concentration of the vapor in air that catches fire when exposed to a source of ignition. The flash point and flammability limits of a substance are often used as measures of its flammability hazard in a given situation.

Flammable. Easy to ignite and burns readily. The U.S. Department of Transportation (<http://www.dot.gov>) defines flammable liquids as those liquids that have a flash point below 100°F (37.8°C).

Flammable range. Range of concentration (of a gas in air) between the Lower and Upper Explosive Limits (LEL and UEL). Gas in the flammable range will burn rapidly or explode when ignited because it has been premixed to the right mixture of fuel and air for burning to occur. If the concentration falls below the LEL, then there is not enough fuel in the air to sustain a fire or an explosion—it is too lean. If the concentration rises above the UEL, then there is not enough oxygen to sustain a fire or an explosion—it is too rich (much like an engine that cannot start because it has been flooded with gasoline).

Flash point. The lowest temperature at which a liquid gives off enough vapor to be ignited at its surface.

Found set. A found set is created in a CAMEO module in one of three ways: (1) when you search the module, records that meet your search criteria are placed in a type of found set also called a **search collection**; (2) when you choose a command from the Show Related submenu of the Record menu, the records related to the current record are placed in a found set in the related module; (3) when you select map objects, then select Get Info, records linked to those objects are placed in a found set. Whenever a found set exists in a CAMEO module, only the records in the found set are shown in List view. To clear a found set, select Show All Records from the Record menu. *See also* **Search collection.**

Free radical. A molecule in which a portion is highly reactive because chemical bonds have been ruptured by thermal radiation from combustion or ionizing radiation. Conceptually, a free radical molecule contains sites with an unpaired electron available to attack other molecules. Free radicals often are involved in chain reaction-type mechanisms, such as combustion processes and the industrial processes used to make polymers.

Freezing point. Also **melting point.** The temperature at which the solid and liquid phases of a substance exist in equilibrium. The freezing point depends on the chemical composition and the applied pressure. The “normal freezing point” is defined at a pressure of 1 atmosphere. For example, the normal freezing point of water is 0°C (32°F).

Fumes. Dense, smoke-like vapors given off by fuming materials such as very reactive liquids, gases, or molten metals (for example, concentrated hydrochloric acid or sulfur monochloride).

Fumigant. A liquid, gas, or solid pesticide applied as a vapor or gas indoors or in a restricted area. Some typical fumigants are sulfur dioxide (protects grains), chlorine dioxide (kills anthrax), and hydrogen cyanide (protects tree crops).

Fungicide. Kills fungi or inhibits growth.

G

Glacial. A very pure form of some acids (e.g., acetic or phosphoric acid) with a freezing point just below room temperature.

H

Hazard category. Five categories of hazardous chemicals are defined in 29 CFR § 1910.1200. They include immediate (acute) and delayed (chronic) health hazards, as well as fire, sudden release of pressure, and reactive hazards. CAMEO's Chemicals in Inventory records contain checkboxes for all of these hazard categories.

Hazard class. One of 9 categories of hazardous materials used in DOT placards (DOT hazard label). The hazard class indicates the most important hazard of a given material (e.g., Explosives or Poison Gas). While some materials meet the criteria for more than one class, each material is assigned just one class.

Hazardous chemical. Any chemical that is a physical or health hazard as defined in 29 CFR § 1910.1200(c).

Hazardous material. Any substance or material in a quantity or form that may be harmful to humans, animals, crops, water systems, or other elements of the environment, if accidentally released. Hazardous materials include: explosives, gases (compressed, liquefied, or dissolved), flammable and combustible liquids, flammable solids or substances, oxidizing

substances, poisonous and infectious substances, radioactive materials, and corrosives.

Hazardous Materials Emergency Planning Guide.

Informally known as the "orange book" or "NRT-1." A publication of the National Response Team; describes a comprehensive approach to planning for hazardous material incidents. See "[Bibliography](#)" on page 272.

Hazardous substances. Substances designated as hazardous under CERCLA. CERCLA includes substances listed under the Clean Water Act, the Clean Air Act, the Resource Conservation and Recovery Act (RCRA), and the Toxic Substances Control Act (TSCA) Section 7. *See also CERCLA.*

Hazards analysis. A systematic method for evaluating the potential hazard to a community from accidental airborne releases of hazardous chemicals stored, transported, manufactured, or used within that community. *See also Technical Guidance.*

Herbicide. A chemical agent used to control weeds or nuisance plants.

High explosive. Or primary explosive. An explosive that is readily detonated by heat, friction, or shock. High explosives vary widely in their sensitivity to these forms of initiating energy.

Highly flammable. Designation applying to (a) substances with flash points below 100°F, and (b) mixtures that include substances with flash points below 100°F. Materials designated as highly flammable include pyrophoric solids and substances that present an exceptional flammability hazard, in that they may suddenly and dangerously increase the intensity of a fire.

Humidity. At a given temperature, the ratio of water vapor in the air to the maximum amount that can be held in the air at that temperature.

Hydration. Or solvation. The association of molecules of water with inorganic substances, usually salts, to form hydrates. Also refers to the strong affinity of water molecules for ions in aqueous solution.

Hydrocarbon. An organic compound containing only hydrogen and carbon.

Hydrogen ion. A hydrogen atom that has lost its single electron and consequently has a positive charge. Water (H₂O) can be thought of as a combination of hydrogen ion (H⁺) and hydroxide ion (OH⁻).

Hydrolysis. A chemical reaction in which a bond is broken by the agency of water. A hydrogen ion and hydroxide ion from the water become independently attached to the two atoms previously linked.

Hydrolyze. To break down by means of reaction with water.

Hydronium ion. When a hydrogen ion is transferred from one water molecule to another, a hydronium ion (H₃O⁺) is formed.

Hydrophilic. *Lit.* “Water-loving.” Refers to substances that attract and retain water, and to wettable solids whose surfaces readily attract water.

Hydrophobic. *Lit.* “Water hating.” Refers to substances that repel water and are not easily wetted or emulsified (fats, waxes, oils, metal powders, and many inorganic compounds).

Hydroxide ion. An ion with an overall negative charge, composed of a single oxygen and a single hydrogen atom. Water (H₂O) can be thought of as a combination of hydrogen ion (H⁺) and hydroxide ion (OH⁻).

Hygroscopic. Capable of readily absorbing moisture from the air, so as to swell up, contract in length, or change form or consistency.

Hypergolic. Igniting spontaneously on contact with another substance, which is typically an oxidizing agent.

I

IDLH. Immediately Dangerous to Life and Health, a level of concern for adult workers; estimate of the highest concentration from which escape is possible without permanent injury. The IDLHs were established by the National Institute for Occupational Safety and Health (NIOSH; <http://www.cdc.gov/niosh>). Find out more about IDLHs on the NOAA OR&R website (<http://response.restoration.noaa.gov/idlhs>).

Ignition point. Minimum temperature at which a substance ignites when no ignition source (e.g., spark or flame) is present.

Immediate (acute) hazard. A hazard category that includes highly toxic, toxic, irritant, sensitizer, corrosive (as defined by 29 CFR § 1910.1200), and other hazardous chemicals that cause an adverse effect to a target organ. Such effects usually develop rapidly, result from short-term exposure, and are of short duration.

Immiscible. Substances of the same phase (liquid or solid) that cannot be mixed in any proportion with one another. *Contrast with Miscible.*

Impurity. Usually, a very small amount of a foreign or extraneous substance that naturally or by accident exists in a substance. Many cases have been documented in which unintended impurities caused severe consequences during a chemical reaction process (often, processes such as polymerization reactions in which a small amount of impurity acted as a catalyst to initiate a runaway reaction).

Incompatible substances. Substances that react together, if mixed.

Inert. Not chemically reactive (e.g., neon).

Ingestion. Taking into the body by the mouth.

Inhalation. A route of chemical exposure in which a toxic material is brought into the bloodstream as particles or volatile compounds breathed into the lungs and nasal passages.

Inhibitor. Substance added to a polymerizable substance to slow down or prevent a polymerization reaction. The inhibitor level in a substance can drop over time, making a reaction possible. *Contrast with Catalyst.*

Initiator. A substance that can induce chemical reactions in other substances. *Compare with Catalyst.*

Inorganic. Compounds that do not contain the element carbon, as well as the following carbon-containing compounds: the carbon oxides, the carbides, carbon disulfide, phosgene, cyanides of metals, and carbonyl sulfide. *Contrast with Organic.*

Insecticide. Substance that kills insects.

Inventory form. The Tier I and Tier II emergency and hazardous chemical inventory forms described in 40 CFR § 370.4. (See [“Sections 311 and 312: community right-to-know requirements” on page 266.](#)) Data fields on records in CAMEO’s Chemicals in Inventory module resemble the data fields on Tier II forms. *See also Tier II.*

Ion. An atom or group of atoms that has an overall non-zero electric charge (which may be either negative or positive).

Isomer. Compounds sharing the same molecular weight and molecular formula, but having different chemical structures. The difference in structure often causes differences in chemical and physical properties such as boiling point, melting point, and

water solubility. Property differences among isomers produce differences in their reactivity and toxicity.

Isotope. Elements that are otherwise identical but that have different weights because they have different numbers of neutrons in their nuclei. The simplest example is the isotopes of hydrogen, including H¹, normal hydrogen, and H², or deuterium, which has an additional neutron.

L

Lachrymator. Substance that causes the eyes to water and may temporarily impair vision. Tear gas is a lachrymator.

Lacquer. A fast-drying, organic coating material.

LC50. Lethal Concentration 50. Measure of acute toxicity of a chemical in air. The concentration of a chemical in air that kills 50% of a group of test animals in a given time period. The smaller the LC50, the more toxic the chemical.

LD50. Lethal Dose 50. Measure of a substance’s acute toxicity. The amount of a substance that causes the death of 50% of a group of test animals (usually mice or rats). The smaller the LD50, the more toxic the chemical.

LEL. Lower Explosive Limit or Lower Flammability Limit. Lowest concentration of a flammable vapor in air at which explosion or combustion can occur.

LEPC. Local Emergency Planning Committee. A committee appointed by the State Emergency Response Commission (SERC), as required by EPCRA, first to design, then to regularly review and update a comprehensive emergency plan for an emergency planning district. See [“Section 301-303: emergency planning” on page 263.](#)

Level of Concern (LOC). “Threshold concentration” of an airborne pollutant, usually the concentration above which a hazard to people is believed to exist. Find out more about LOCs on the NOAA OR&R website (<http://response.restoration.noaa.gov/locs>).

Lignin. An amorphous organic polymer that helps bind cellulose fibers together in plant tissue.

Link. Either a connection between a CAMEO record and a related object on a MARPLOT map, or a connection between a chemical name shown on Chemicals in Inventory record(s) and a Chemical Library record.

List view. View of a CAMEO module in which records in the module are displayed as a list. Double-clicking any record in the list displays that record in Record view. *Contrast with Record view.*

Litmus. A substance or material that, on contact with another substance, indicates that substance’s pH or hydrogen ion concentration. Examples include litmus paper and pH indicator solutions. *See also pH.*

LNG. Liquefied natural gas.

LOC. Level of Concern. “Threshold concentration” of an airborne pollutant, usually the concentration above which a hazard to people is believed to exist. Find out more about LOCs on the NOAA OR&R website (<http://response.restoration.noaa.gov/locs>).

Local Emergency Planning Committee (LEPC). A committee appointed by the State Emergency Response Commission (SERC), as required by EPCRA, first to design, then to regularly review and update a comprehensive emergency plan for an emergency planning district. See “[Section 301-303: emergency planning](#)” on page 263.

Lower Explosive Limit (LEL). Or Lower Flammability Limit. Lowest concentration of a flammable vapor in air at which explosion or combustion can occur.

LPG. Liquefied petroleum gas.

M

MARPLOT. Electronic mapping program included in CAMEO. Find out more about MARPLOT on the NOAA OR&R website (<http://response.restoration.noaa.gov/marplot>).

Mass. Mass is a physical property related to weight. Mass is a measure of the amount of a substance that occupies a given space. While the weight of a given amount of a substance is a measure of the force by which it is attracted by gravity (and is less on the moon than on the earth), the substance’s mass is independent of gravity.

Material Safety Data Sheet (MSDS). Product data sheet prepared under 29 CFR § 1910.1200 by the manufacturer or marketer of a hazardous material; describes product, its hazards, and safe handling and response procedures.

Melting point. Also **freezing point.** The temperature at which the solid and liquid phases of a substance exist in equilibrium. Depends on chemical composition and applied pressure.

Metallurgy. The scientific study of the process of obtaining metals (mining), and the properties and uses of metals.

Mineral. The inorganic constituents of the earth’s crust. Most minerals have a definite internal structure and chemical composition.

Miscible. Completely soluble. *Contrast with Immiscible.*

Mixture. A uniform or nonuniform blend of two or more substances. Examples include blood, milk, petroleum products, and alloys.

Module. Any of the databases included in CAMEO (e.g., Facilities, Chemical Library, Contacts). *See also Database.*

Mole. Amount of a substance containing 6.02×10^{23} molecules. The molecular weight of a chemical is the mass of 1 mole of that chemical.

Molecular Weight. The sum of the weights of all the atoms in a molecule.

Molecule. A chemical entity composed of one or more elements in the form of atoms.

Molluscicide. A chemical agent used to control snail populations.

Monomer. One of the molecules that link together to make a polymer. For example, the monomer of natural rubber is isoprene. Monomers may be naturally occurring or synthetic. *See also Polymerization.*

MSDS. Material Safety Data Sheet. Product data sheet prepared under 29 CFR § 1910.1200 by the manufacturer or marketer of a hazardous material; describes product, its hazards, and safe handling and response procedures.

Mutagen. A chemical or radiation source that alters an organism's DNA, affecting transmission of inherited characteristics from one generation to the next.

N

N.O.S. Not otherwise specified.

NAICS. North American Industry Classification System. An industry classification system. *Compare with SIC.*

National Response Center (NRC). The central federal clearinghouse for information on hazardous chemical spills and other oil or hazardous substance releases. Responsible parties should contact the NRC (<http://www.nrc.uscg.mil>) in order to fulfill reporting requirements for spills of oil and hazardous substances (hotline: 1-800-424-8802).

National Response Team (NRT). A planning, policy, and coordinating body consisting of representatives from 16 federal agencies with interest and expertise in aspects of emergency response to pollution incidents. The NRT (<http://www.nrt.org>) provides national level policy guidance prior to an incident and can provide assistance during an incident.

Nematicide. A pesticide employed to kill nematodes (a type of worm common in soil and water).

Neutralization/Neutralize. The reaction between an acid and base to form a salt and water. Neutralization is often rapid, vigorous, and exothermic (heat-producing).

NFPA. National Fire Protection Association (<http://www.nfpa.org>), a private, non-profit organization that produces technical data related to fire protection and prevention, including the widely used "NFPA diamond" containing codes representing chemical hazards.

NFPA codes. Numbers representing the flammability, health, reactivity (instability), and other hazards of individual chemicals; shown on the "NFPA diamond."

NFPA diamond. Standard placard used to identify the level of chemical hazard at a fixed location or in a transported container.

NIOSH. National Institute for Occupational Safety and Health (<http://www.cdc.gov/niosh>). The federal agency responsible for conducting research and making recommendations for the prevention of work-related disease and injury. NIOSH is part of the Centers for Disease Control and Prevention (CDC).

NOAA. National Oceanic and Atmospheric Administration (<http://www.noaa.gov>).

Noble gases. A group of six elements (helium, neon, argon, krypton, xenon, and radon) exhibiting little or no chemical reactivity.

Non-combustible. Non-reactive with air, even at very high temperatures.

Nonflammable. Difficult to ignite.

Nonoxidizing. Completely or nearly lacking the ability to oxidize (to transfer oxygen to other groups or lose electrons to other groups).

NOS. Not otherwise specified.

NO_x. The oxides of nitrogen, taken as a group. Nitrogen forms several distinct compounds with oxygen; some of these compounds convert back and forth readily under ordinary conditions.

Noxious. Irritating; may cause temporary impairment.

NRC. The central federal clearinghouse for information on hazardous chemical spills and other oil or hazardous substance releases. Responsible parties should contact the NRC (<http://www.nrc.uscg.mil>) in order to fulfill reporting requirements for spills of oil and hazardous substances (hotline: 1-800-424-8802).

NRT. A planning, policy, and coordinating body consisting of representatives from 16 federal agencies with interest and expertise in aspects of emergency response to pollution incidents. The NRT (<http://www.nrt.org>) provides national level policy guidance prior to an incident and can provide assistance during an incident.

O

Office of Response and Restoration. NOAA's Office of Response and Restoration (<http://response.restoration.noaa.gov>). OR&R co-developed the CAMEO program with the Environmental Protection Agency's Office of Emergency Management (<http://www.epa.gov/emergencies>).

Olefin. An alkene. *See also Alkene.*

Oleum. Or fuming sulfuric acid. The mixture of sulfuric acid and sulfur trioxide.

OR&R. NOAA's Office of Response and Restoration (<http://response.restoration.noaa.gov>). OR&R co-developed the CAMEO program with the Environmental Protection Agency's Office of Emergency Management (<http://www.epa.gov/emergencies>).

Organic. Generally, compounds that contain the element carbon, except for some carbon-containing compounds that are considered to be inorganic (carbon oxides, carbides, carbon disulfide, phosgene, the cyanides of the metals, and carbonyl sulfide). *Contrast with Inorganic.*

OSHA. Occupational Safety and Health Administration (<http://www.osha.gov>). Federal agency within the U.S. Department of Labor with the responsibility of ensuring worker safety and health.

Overpressurization. Increase of pressure within a container beyond the pressure the container is designed to contain; can lead to explosion of the container.

Oxidation. A chemical reaction in which oxygen bonds to an element or compound. By extension, a reaction in which one element or compound rises to a higher oxidation state while another drops to a lower oxidation state (the term is used in this sense even when no oxygen whatever is present). *Contrast with Reduction.*

Oxidizer. Or **Oxidizing agent.** Substance that yields oxygen readily to support a fire.

Oxidizing agent. Or **Oxidizer.** Substance that yields oxygen readily to support a fire. *Contrast with Reducing agent.*

P

Parent module. A CAMEO module located above another, related module in CAMEO's file hierarchy. Facilities and Routes are parent modules relative to the Chemicals in Inventory module.

Partial pressure. At equilibrium, molecules move from the liquid to the gas phase at the same rate as they return to the liquid from the gas phase. In liquids that are mixtures of chemicals, each component vaporizes (evaporates) at its own characteristic rate. The partial pressure of a chemical in a mixture is a measure of the concentration of that chemical's gas-phase molecules in the air directly in contact with the liquid. In most actual cases equilibrium is not reached so the partial pressure represents the limiting value for chemical concentration in the air.

Peroxidizable. Can slowly react with oxygen in the air or with oxidizers (oxidizing agents) to eventually form potentially explosive, white crystalline peroxides.

Person. In the context of EPCRA, any individual, trust, firm, joint stock company, corporation (including a government corporation), partnership, association, State, municipality, commission, political subdivision of State, or interstate body.

pH. Measure of the acidity or alkalinity of a solution. At room temperature, a solution having a pH of 7 is neutral (neither acidic nor basic). Solutions with pHs greater than 7 are basic (alkaline); solutions with pHs below 7 are acidic.

Polymer. The product of polymerization. Proteins, starches, cellulose and natural rubber are naturally occurring polymers; polystyrene, nylon, Teflon®, and synthetic rubber are synthetic polymers. *See also Polymerization.*

Polymerizable. Can react with itself. Polymerization reactions typically generate heat and could cause container to overpressurize, possibly leading to a fire or explosion.

Polymerization. Chemical reaction in which small molecules join to form larger molecules; polymerization reactions typically release heat, and can cause containers to overpressurize.

ppb. Parts per billion. Units used to express the concentration of a gas or vapor in air (as molecules of chemical per billion molecules of air).

ppm. Parts per million. Units used to express the concentration of a gas or vapor in air (as molecules of chemical per million molecules of air).

Pressurization. Build up of pressure within a closed container. May cause container to explode.

Prill. Small spherical or cylindrical pellets used in the fertilizer and explosive industry because they are convenient to handle (e.g., ammonium nitrate prills, often coated with wax because this salt tends to cake when hydrated).

PSI. Pounds per square inch; a unit of pressure measurement.

PSIG. Pounds per square inch gauge; pressure relative to atmospheric pressure.

Pyrophoric. Any substance that ignites in the presence of air at or below ambient temperatures. Many pyrophoric materials react with moisture in the air to generate flammable hydrogen gas and enough heat to ignite the hydrogen. These are extremely dangerous fire hazards that are generally stored under an inert atmosphere or in a solvent like ether or kerosene that excludes air.

Pyrotechnics. The manufacture of fireworks, signal flares, and so on, involving the mixture of different chemicals to achieve various visual and auditory effects. Chemicals used in pyrotechnics include many explosive inorganic compounds such as potassium nitrate, metal perchlorates, dichromate, powdered metals, and phosphorus.

Q

Quaternary ammonium salt. Ammonia derivatives in which four alkyl or aryl groups are attached to a nitrogen atom. When ionized in aqueous solution, acts as an antibacterial, antistatic agent, and an accelerant in photographic development.

R

Radiation. Electromagnetic energy or light, depending on the wavelength, which imparts energy to molecules and atoms. Radiation absorption causes ionization and bond-breaking.

Radical. In inorganic chemistry, refers to an aqueous, dissociated ionized group that acts as a fragmented, highly reactive, short-lived substance. Free radical generation is a very important reaction that can initiate polymerization and other kinds of reactions.

Radioactive. Spontaneously and continuously emitting ions or ionizing radiation. Radioactivity is not a chemical property, but an additional hazard apart from other properties of a material.

Rapid reaction. A reaction in which chemical change becomes evident within minutes to hours after the mixing of incompatible chemicals. That evidence can be dramatic change such as fire or explosion, or more subtle effects, such as chemical heat production, evolution of gases, or deposition/disappearance of solids.

Rare. Can refer either to the noble gases (helium, neon, and argon), the rare-earth elements (or lanthanide series, starting with lanthanum), or rare metals such as gold, silver, cadmium, and platinum.

RCRA chemical. A hazardous waste regulated under the Resource Conservation and Recovery Act of 1976 (RCRA). Four-character identification codes are assigned to regulated hazardous wastes.

Reaction. The change that takes place when two or more substances interact to form new substances.

Reactive. Readily reacts with other chemicals (described in 29 CFR § 1910.1200).

Reactive group. Reactive groups are categories of chemicals that react in similar ways because they are similar in their chemical structure. For purposes of predicting reactivity between mixed chemicals, each substance in the Chemical Library has been assigned to one or more reactive groups, based on the known chemistry of that substance. See [“How CAMEO predicts mixture reactivity” on page 96.](#)

Reactivity. The tendency of a substance to undergo chemical change. The reactivity of most substances depends on the temperature and pressure of the surroundings, and on the chemicals with which it comes in contact. Under ordinary conditions, innately reactive substances are those that react rapidly with water, air, and other common components of the environment, as well as substances that self-react (decompose or polymerize).

Reagent. Any chemical substance used in chemical analysis.

Record. In any CAMEO module, each record describes an important entity or element of information. For example, each record in the Chemical Library describes a particular substance; each Facilities record describes a single facility (or department within a large facility). Each record contains datafields, each of which stores a particular kind of information (e.g., Chemical Name is a data field on a Chemical Library record). *See also Database.*

Record view. View of a single record within a CAMEO module, in which you can access all the information contained in that record (typically, you clicks tabs to see different kinds of information in a record). *Contrast with List view.*

Redox. A contraction for “oxidation-reduction.” Redox reactions, in which electrons are transferred from a reducing agent to an oxidizing agent, are a major category of chemical change.

Reducing agent. Substance that can react strongly or explosively with oxidizers. *Contrast with Oxidizing agent.*

Reduction. A reaction in which either oxygen is removed from a substance or, in a more general sense, one or more electrons is accepted from another substance. (Can also mean a process in metallurgy by which materials are separated into their wanted and unwanted components from an ore body.) *Contrast with Oxidation.*

Reportable quantity (RQ). The quantity of a hazardous substance or extremely hazardous substance that, if released, must be reported to the National Response Center, the State Emergency Response Commission, and the community emergency coordinator for areas likely to be affected by the release. See “[Section 304: emergency notification](#)” on page 265.

Resins. Naturally-occurring “essential oils” that are water-insoluble and extracted from natural sources, such as trees and shrubs. Also, synthetic polymers such as polystyrene or polyethylene, even though such materials are not complex mixtures of compounds like other resins, but instead are long, complex polymers.

Response Information Data Sheets (RIDS). In a Chemical Library record, a set of detailed descriptions of chemical properties, hazards, and emergency response information.

Risk analysis. A systematic method of assessing the damage that could be caused to a community by a hazardous substance release.

RQ. Reportable Quantity (RQ). The quantity of a hazardous substance or extremely hazardous substance that, if released, must be reported to the National Response Center, the State Emergency Response Commission, and the community emergency coordinator for areas likely to be affected by the release. See “[Section 304: emergency notification](#)” on page 265.

S

Salt. An ionic compound derived from the combination of cations (positively-charged ions) with anions (negatively-charged ions). Sodium chloride (common table salt) is an example.

SARA. Title III of SARA is also known as the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA). *See also* **EPCRA**.

Saturated. Refers to organic compounds that contain the maximum possible number of hydrogen atoms per carbon atom. Also used to describe a solution of a solute that is at its solubility limit in a given solvent system at a specified temperature. *Contrast with* **Unsaturated**.

Screening zone. Threat zone for screening calculations.

Search collection. A type of **found set**. When you perform a search or request information on linked MARPLOT objects, CAMEO collects together the records in the given module that meet your criteria. You then can browse through the collected records. While a search collection exists, you can move among only the collected records. Before you can work with other records in the module, you must clear the search collection (choose Clear Search from the Search menu). *See also* **Found set**.

Search criteria. The criteria, or conditions, that you set when you search a CAMEO module for a specific piece of information, such as the record for a particular chemical in the Chemical Library.

Sensitizer. Substance that can cause an allergic reaction in some people; the reaction can be severe, and breathing can be obstructed.

SERC. State Emergency Response Commission (SERC). A commission appointed by each State governor under EPCRA. A SERC designates emergency planning districts, appoints local emergency planning committees (LEPCs), supervises and coordinates the activities of planning committees, reviews emergency plans, receives chemical release notifications, and establishes procedures for receiving and processing requests from the public for information. *See also* **Local Emergency Planning Committee (LEPC)**.

Sharing. The process by which CAMEO, ALOHA, and MARPLOT share information in order to establish and use links between module records, map objects, and ALOHA threat zones.

Shock wave. A pressure wave generated by an explosion.

SIC. Standard Industrial Classification. SIC codes are assigned to U.S. industries. Facilities with SIC codes 20 to 39 (manufacturers) may be subject to the hazardous chemical inventory reporting requirements of EPCRA.

Skin absorption. Chemical exposure through the skin. Because the skin does not act as a reliable barrier to hazardous chemicals, it can be a route of acute poisoning. Compounds such as dimethyl sulfoxide are known to be directly absorbed into the bloodstream through the skin.

Sludge. A thick or viscous mixture of solids in aqueous solution, such as sewage sludge.

Solubility. A measure of a chemical's ability to dissolve in water. If a chemical is highly soluble, it will dissolve easily into water.

Solute/Solvent. A solute is a gas, liquid, or solid substance that is uniformly dispersed in a liquid solvent substance, forming a solution. The solvent molecules act to break the solute molecules' attraction for one another, and also the solvent's natural structure. For instance, water is a highly-structured substance, in the absence of any solutes. *See also* **Solution**.

Solution. Mixtures of chemicals in which the components are interspersed uniformly at the molecular level. *See also* **Solute/Solvent**.

Special locations. Locations of people who may be more susceptible to the toxic effects of an accidental release than the general population, because of pre-existing health conditions, age, or other factors. Schools, day-care centers, and nursing homes are examples of special locations.

Specific gravity. The ratio of the density of a substance at a given temperature to the density of water at the same temperature. A substance with a specific gravity greater than 1.0 will sink rather than float in water. *See also* **Density**.

State. Any State of the United States, as well as the District of Columbia, the Commonwealth of Puerto Rico, Guam, American Samoa, the United States Virgin Islands, the Northern Mariana Islands, and any other territory or possession over which the United States has jurisdiction.

State Emergency Response Commission (SERC). A commission appointed by each State governor under EPCRA. A SERC designates emergency planning districts, appoints local emergency planning committees (LEPCs), supervises and coordinates the activities of planning committees, reviews emergency plans, receives chemical release notifications, and establishes procedures for receiving and processing requests from the public

for information. *See also* **Local Emergency Planning Committee (LEPC)**.

State Fields. Records in the Facilities and Chemicals in Inventory modules contain a State Fields section, designed to support the emergency planning work of U.S. states that have developed their own reporting requirements, in addition to the Tier II requirements specified in EPCRA (which apply to all 50 states). In these states, Tier II forms contain additional data fields, called "State Fields" in CAMEO. CAMEO's State Fields, located under the State Fields tab, provide a place where people working in these states can maintain their reporting information.

STCC number. Sounds like "Stick number." Standard Transportation Commodity Code. Seven-digit chemical identification code used by the Association of American Railroads (<http://www.aar.org>). Numbers beginning with 49- indicate hazardous materials.

Strong oxidizing agent. Oxidizing agents gain electrons from other substances and are themselves thereby chemically reduced. Strong oxidizing agents accept electrons particularly readily from many other substances. The ensuing redox reactions may be vigorous or violent and may release new substances that may take part in further additional reactions. Keep strong oxidizing agents well separated from strong reducing agents.

Strong reducing agent. Reducing agents give up electrons to other substances, and are themselves thereby oxidized. Strong reducing agents donate electrons particularly readily to many other substances. The ensuing redox reactions may be vigorous or violent and may generate new substances that take part in further additional reactions. Keep strong reducing agents well away from strong oxidizing agents.

Sublime. A frozen substance sublimates when it passes directly into the gas phase without first becoming liquid. Substances that sublime include solid carbon dioxide, sulfur, camphor, and naphthalene.

Sudden release of pressure. A hazard category that includes explosives and compressed gases (described in 29 CFR § 1910.1200).

Superfund. The trust fund established under CERCLA to provide funding for cleanup of hazardous substances.

Superfund Amendments and Reauthorization Act of 1986 (SARA). Title III of SARA is also known as the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA). *See also* **EPCRA**.

Surfactant. A substance that reduces surface tension of liquids. Nearly synonymous with detergent (which reduces the surface tension of water), wetting agent, and emulsifier.

Suspected human carcinogen. Suspected to increase the risk of cancer in people, but not enough scientific data is available to be sure.

Suspension. A uniformly-dispersed mixture of fine particles in a liquid.

Synonyms. In CAMEO, a single chemical may have many names. The synonyms by which a chemical is named can include common names in different languages, trade names, and other names.

T

Technical grade. Term applied to substances that are unrefined, impure, and/or less than 100% pure grade.

Technical Guidance. *Technical Guidance for Hazards Analysis: Emergency Planning for Extremely Hazardous Substances.* A guidebook for use by members of Local Emergency Planning Committees (LEPCs) in assessing the potential risks to a community from accidental airborne releases of extremely hazardous substances. *See* “Bibliography” on page 272.

TEELs. Temporary Emergency Exposure Limit, a level of concern representing the adverse health effects of a hazardous substance on members of the general public. The TEELs are defined by the U.S. Department of Energy for use when ERPGs or AEGLs aren’t available (see http://www.hss.energy.gov/HealthSafety/WSHP/chem_safety/teel.html). Find out more about TEELs on the NOAA OR&R website (<http://response.restoration.noaa.gov/teels>).

Threat zone. A threat zone encloses the area around the location of a hazardous chemical release, within which concentrations of the chemical could reach or exceed a specified LOC (level of concern). *See also* **LOC**.

Threshold Planning Quantity (TPQ). The amount of an extremely hazardous substance present at a facility, above which the facility’s owner/operator must give emergency planning notification to both the appropriate State Emergency Response Commission and Local Emergency Planning Committee. TPQs are defined in 40 CFR § 355. *See* “Section 301-303: emergency planning” on page 263.

Tier II. Facilities covered by EPCRA must submit a hazardous chemical inventory form to the Local Emergency Planning Committee, the State Emergency Response Commission, and the local fire department annually. Facilities provide either a Tier I or Tier II form. Most States require the Tier II form. See “[Sections 311 and 312: community right-to-know requirements](#)” on page 266.

Title III. Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA Title III) is also known as the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA). *See also EPCRA.*

TLV. Threshold Limit Value, a level of concern representing the adverse health effects of a hazardous substance on healthy adult workers. Find out more about TLVs on the NOAA OR&R website (<http://response.restoration.noaa.gov/tlvs>).

Toxic. Poisonous; can injure or kill people or other organisms.

TPQ. Threshold Planning Quantity (TPQ). The amount of an extremely hazardous substance present at a facility, above which the facility’s owner/operator must give emergency planning notification to both the appropriate State Emergency Response Commission and Local Emergency Planning Committee. TPQs are defined in 40 CFR § 355. See “[Section 301-303: emergency planning](#)” on page 263.

U

UEL. **Upper Explosive Limit** or Upper Flammability Limit. Highest concentration of a flammable vapor in air at which explosion or combustion can occur.

Unlink. A command which breaks a link previously established between a CAMEO record and an object on a MARPLOT map. *Contrast with Link.*

UN/NA number. United Nations-North America number. (Also UN Number or DOT Number.) Four-digit number identifying an individual chemical or group of chemicals with similar characteristics. Required on shipping papers; often shown on placards or labels. Often preceded by “UN”: e.g., “UN1219.”

Unsaturated. Refers to organic compounds that contain fewer than the maximum possible number of hydrogen atoms per carbon atom. *Contrast with Saturated.*

Upper Explosive Limit (UEL). Or Upper Flammability Limit. Highest concentration of a flammable vapor in air at which explosion or combustion can occur.

UV. Abbreviation for ultraviolet radiation.

V

Vapor. The gas given off by a substance that is liquid (or solid) at normal temperatures and pressures.

Vapor density. The ratio of the density of a gas to the density of air at the same temperature. A gas with a vapor density greater than 1.0 is heavier than air and will sink in the atmosphere. *See also Density.*

Vapor pressure. A measure of a substance’s tendency to evaporate; liquids with higher vapor pressures evaporate faster.

Viscosity. Measure of a liquid’s resistance to flow.

Volatile. Evaporates readily.

Volatility. Readiness to evaporate. The volatility of a given substance increases with temperature.

Vulcanization. The industrial process by which rubber is allowed to “cross-link” to form disulfide bonds between molecules. Vulcanization improves the durability of the rubber product.

Vulnerable zone. Threat zone for scenarios calculations. *See also* **Threat zone**.

W

Water-reactive. Can react vigorously when mixed with water or steam or when spilled into water; the reaction may generate hazardous gas or overpressurize container.

Water of crystallization. Water chemically combined within a crystalline substance.

Wettable powder. A solid material whose surface is hydrophilic, and therefore attracts water.

WMD. Weapon of Mass Destruction.

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