

## Mixing Ratio

From the user, an air temperature ( $T$ ), a dewpoint temperature ( $T_d$ ), and a station pressure ( $p_{sta}$ ) are given. The calculation of the mixing ratio is quite involved. In order to calculate the mixing ratio, a saturated vapor pressure ( $e_s$ ) must be computed for values of air temperature, and an actual vapor pressure ( $e$ ) must be computed for values of dewpoint temperature. But before the vapor pressures can be computed, the air temperature and/or dewpoint temperature must be converted to degrees Celsius ( $^{\circ}C$ ). To see how to convert temperatures, see the link below:

<http://www.wrh.noaa.gov/Saltlake/projects/wxcalc/formulas/tempConvert.pdf>

Then, using the values of air temperature and/or dewpoint temperature the vapor pressure(s) can be computed. To see how to calculate the vapor pressure(s), see the link below:

<http://www.wrh.noaa.gov/Saltlake/projects/wxcalc/formulas/vaporPressure.pdf>

Next, the station pressure ( $p_{sta}$ ) must be converted to millibars ( $mb$ ) or hectoPascals ( $hPa$ ). To see how to convert the station pressure, see the link below:

<http://www.wrh.noaa.gov/Saltlake/projects/wxcalc/formulas/pressureConversion.pdf>

Finally, the actual mixing ratio ( $w$ ) and/or saturated mixing ratio ( $w_s$ ) can be calculated using the formula below:

$$w = 621.97 \times \frac{e}{p_{sta} - e} \qquad w_s = 621.97 \times \frac{e_s}{p_{sta} - e_s}$$

For a bonus answer, the relative humidity ( $rh$ ) can be calculated using the answers from the actual mixing ratio and the saturated mixing ratio by using the formula below:

$$rh = \frac{w}{w_s} \times 100$$