

SOUTHERN RESEARCH
I N S T I T U T E

Mercury Control with Calcium-Based Sorbents and Oxidizing Agents

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Acknowledgements

- Barbara Carney -- DOE Project Manager

- Babcock Power -- Na₂S₄ Technology

- We would also like to thank EPA and EPRI for funding and allowing collaboration with a fundamental mercury speciation program, also being conducted at Southern Research Inst.

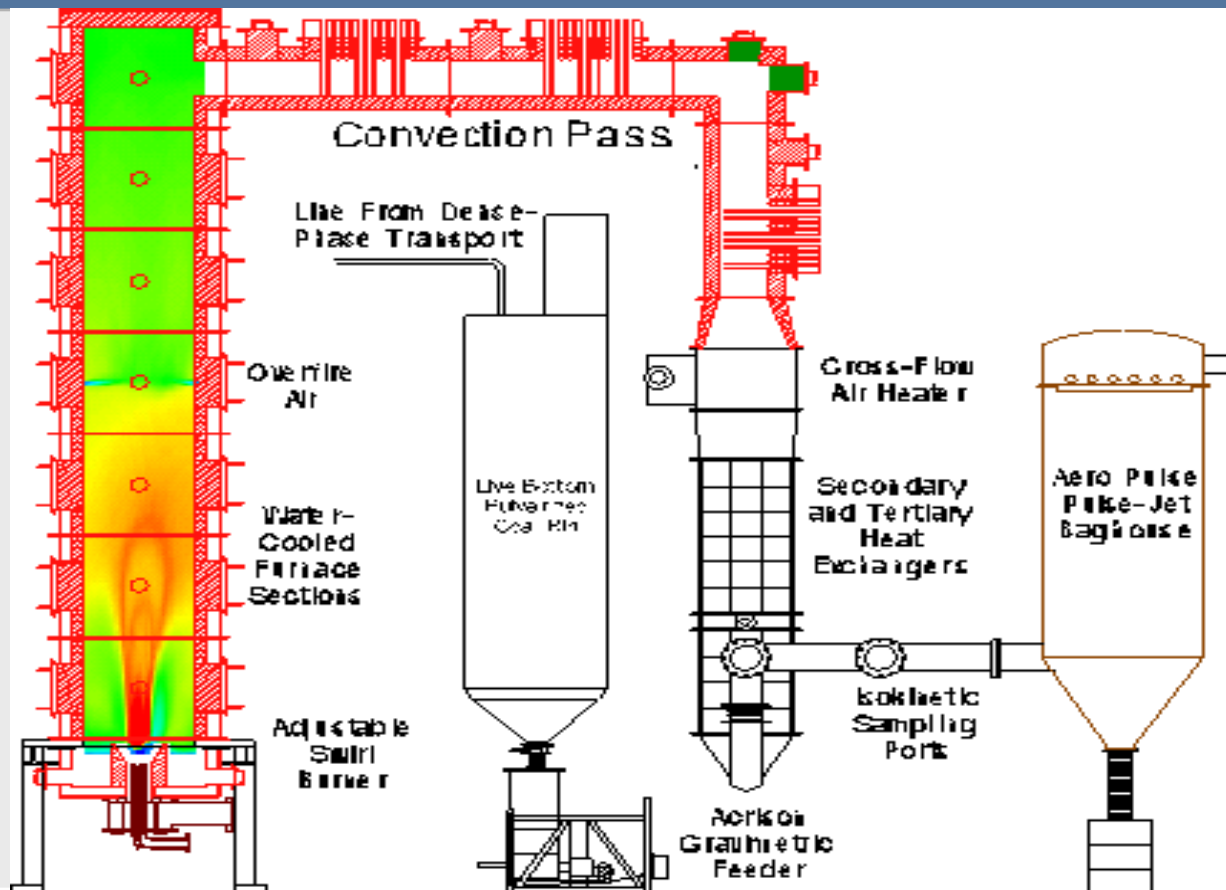


Outline

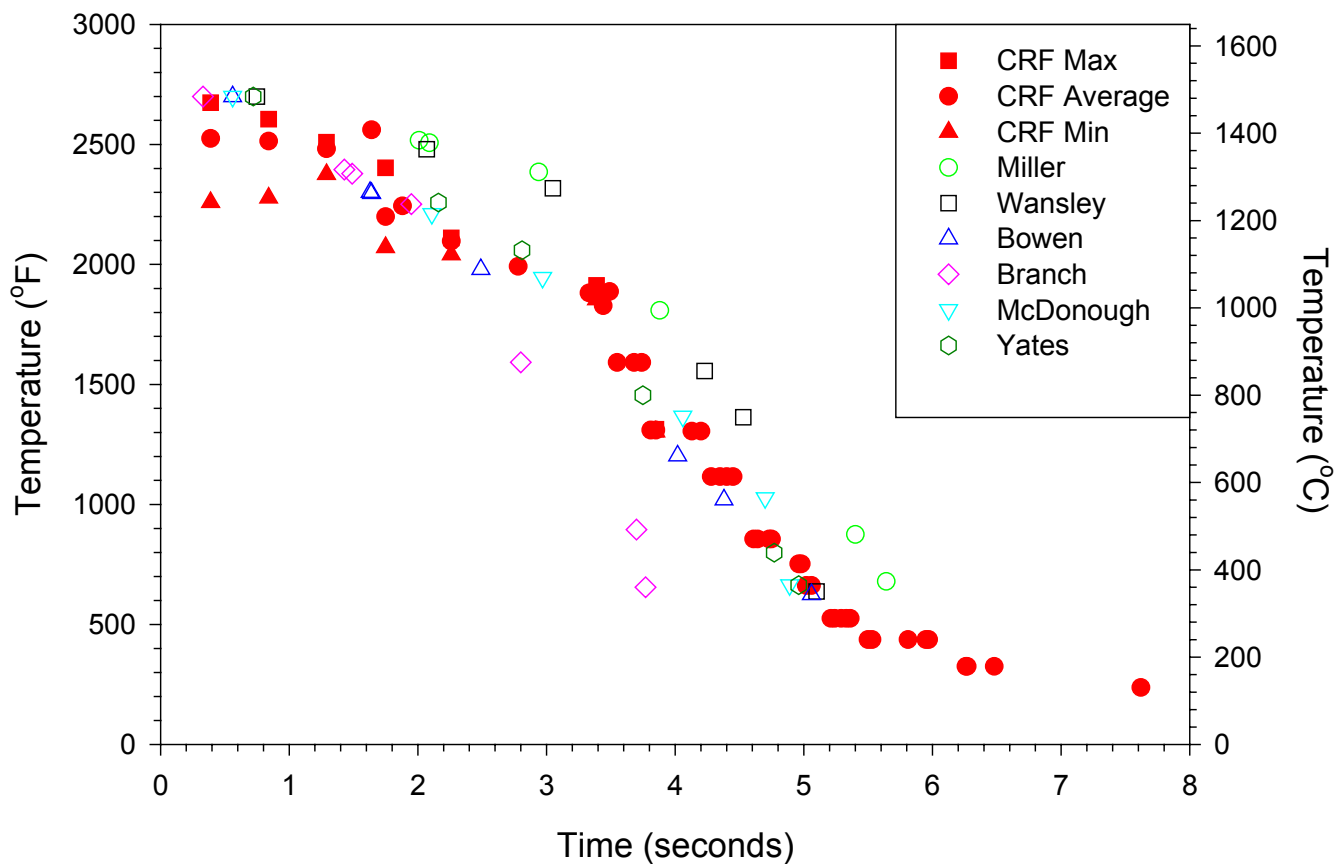
- Experimental
- Comparison of Hg-Oxidation and Hg-Capture -- PRB and Bituminous coals
 - Chlorine
 - General Flue Gas Components
 - Catalytic Material
- Na_2S_4 Injection
 - Effects of Chlorine, Temperature, Coal Type.



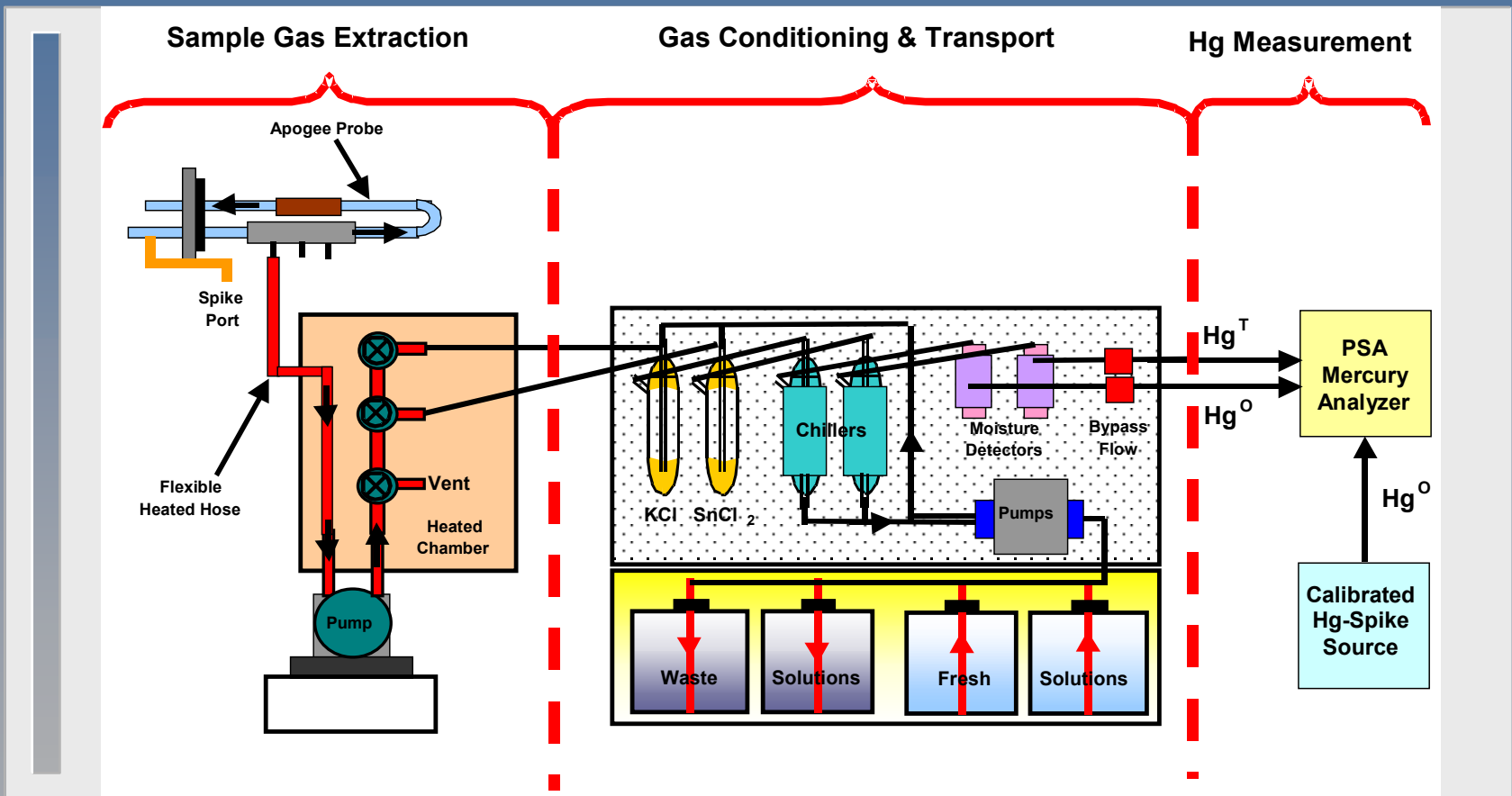
Combustion Research Facility



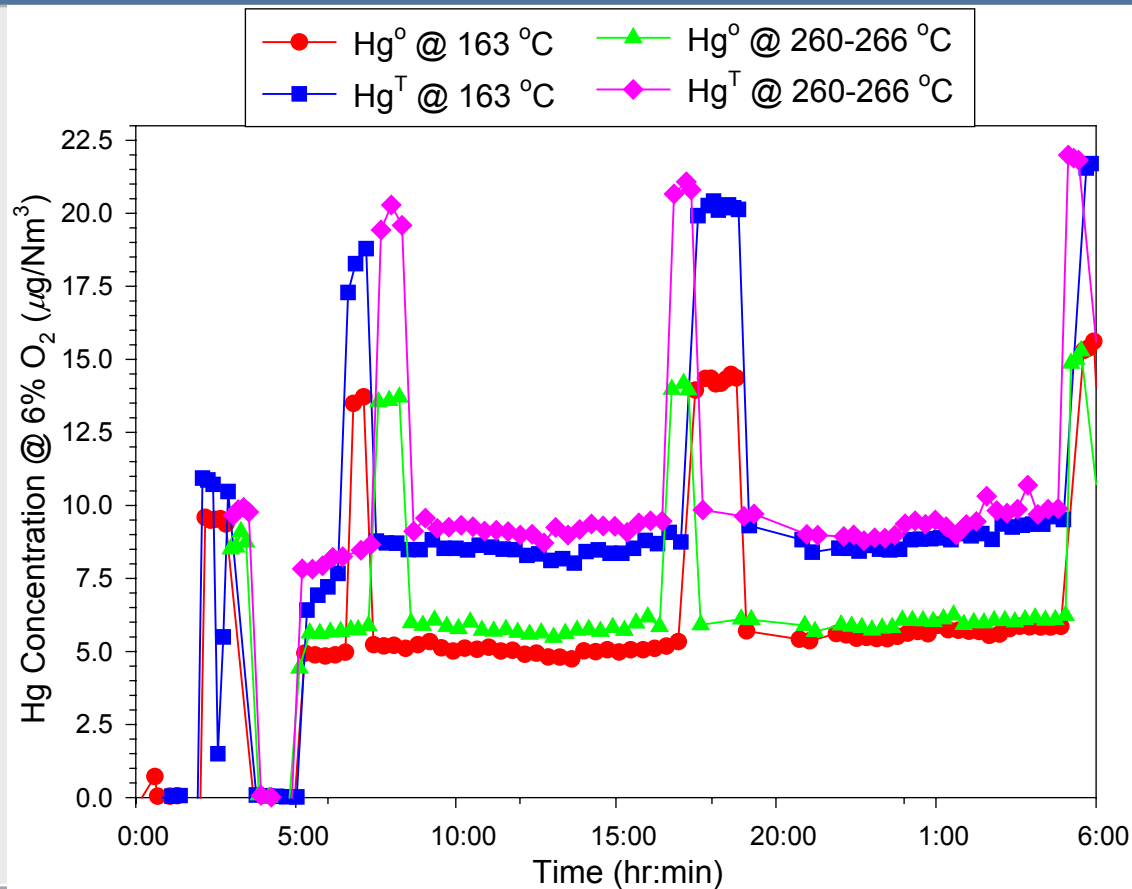
CRF dT/dt Compared to Full-Scale



Mercury Monitoring System Including Spike and Recovery



Example of Data from Monitor Using Spike and Recovery



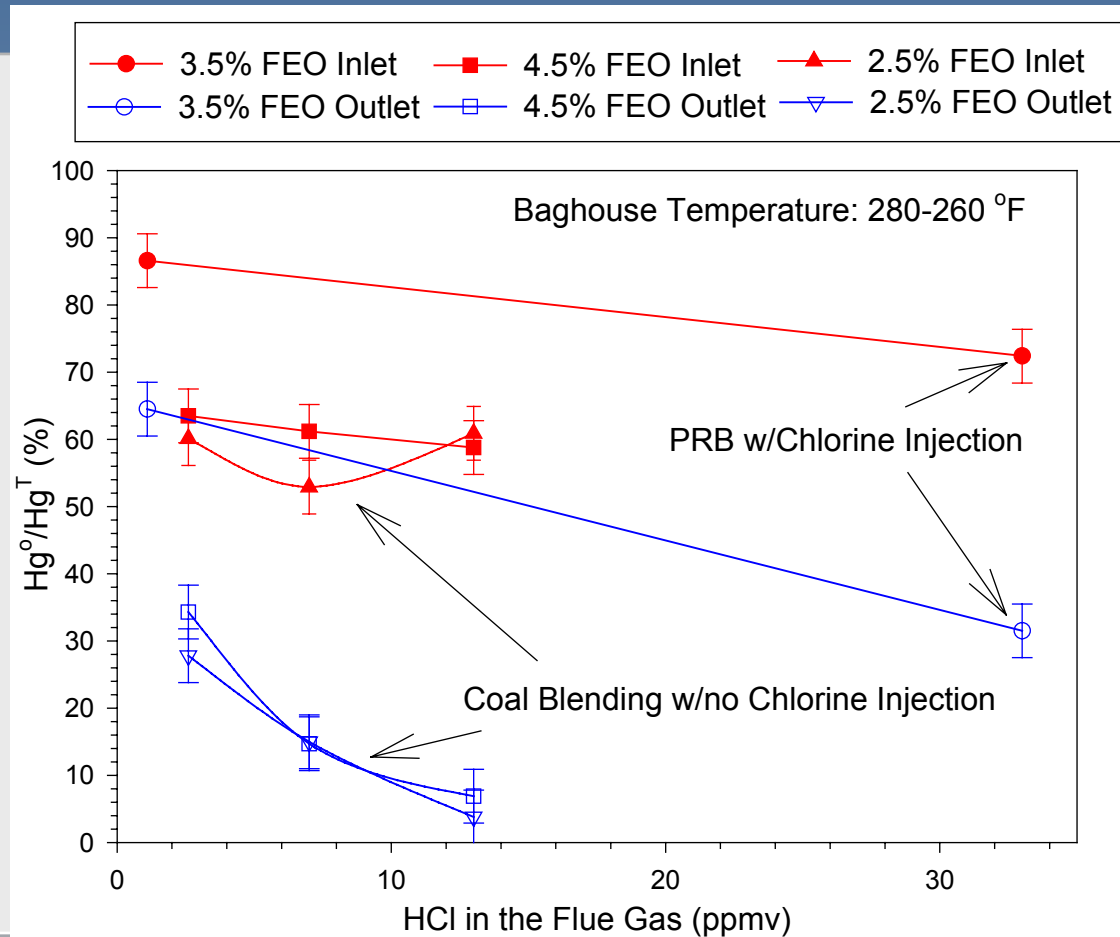
PRB/Bituminous Component Comparison Regarding Hg-Oxidation and Capture

- PRB Coal *compared with*
Bituminous Coal
- via coal blending

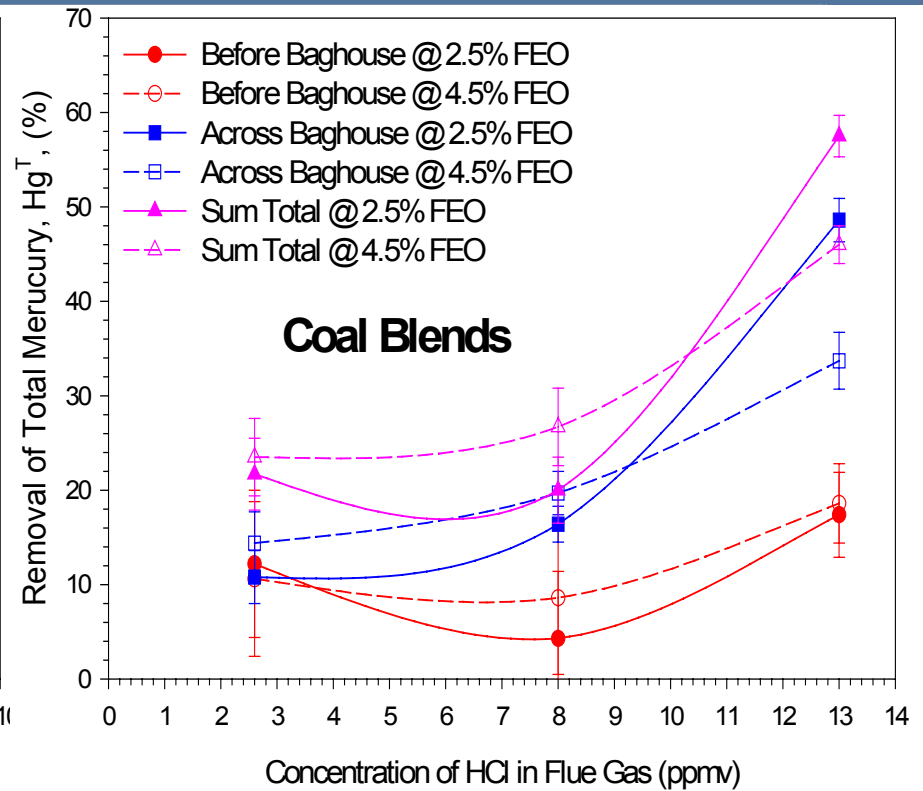
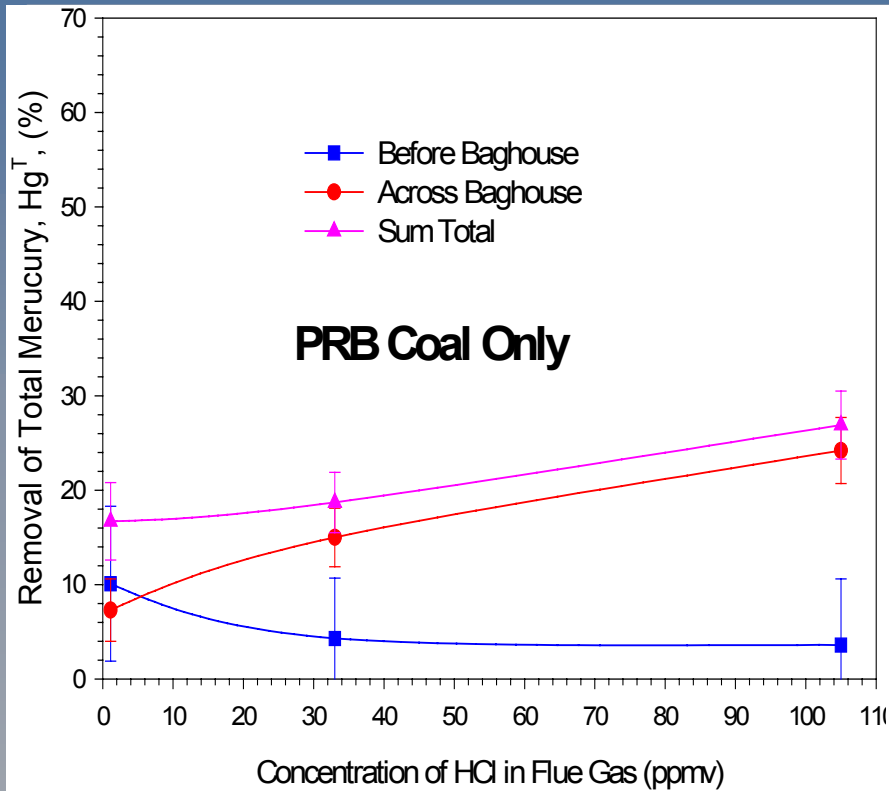
- Hg-Oxidation
- Hg-Capture



Effect of Flue-Gas Chlorine on Hg-Oxidation



Effect of Chlorine on Hg-Removal

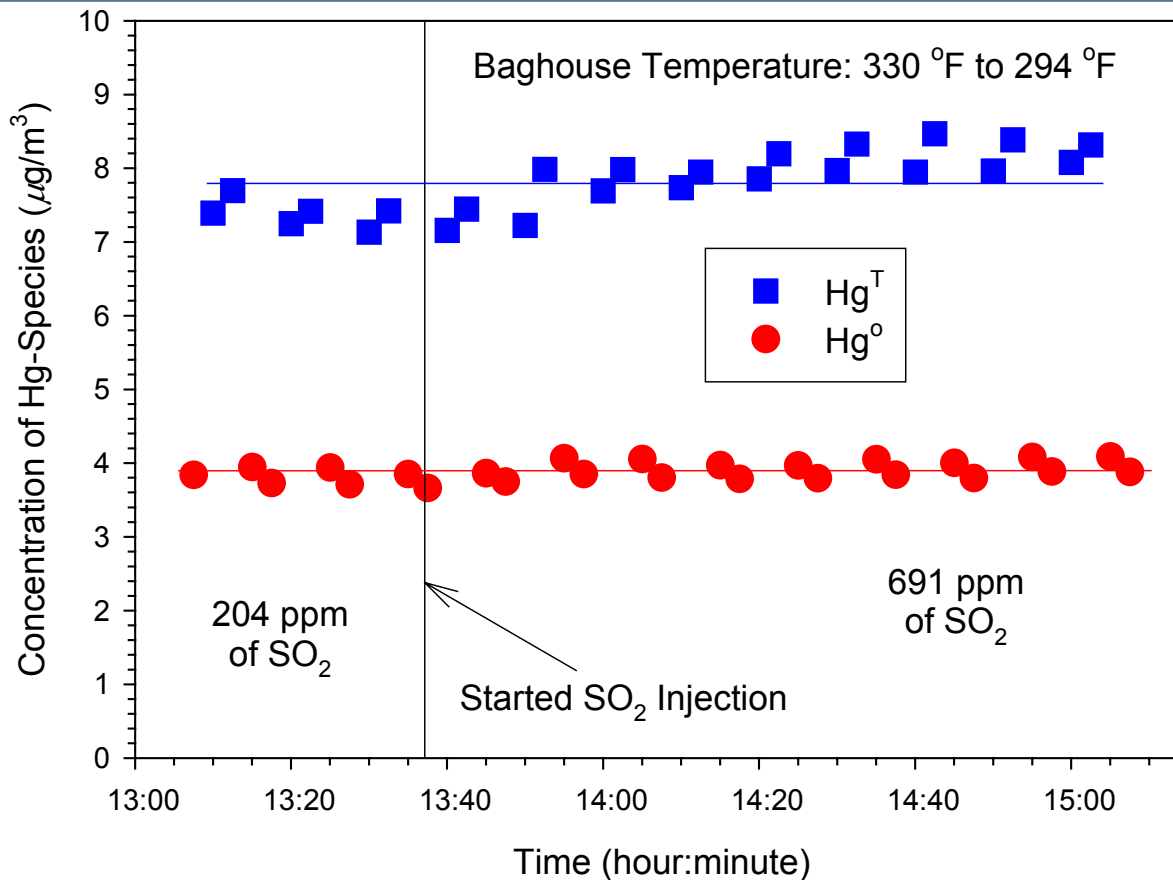


Isolated and Non-Correlated Parameters

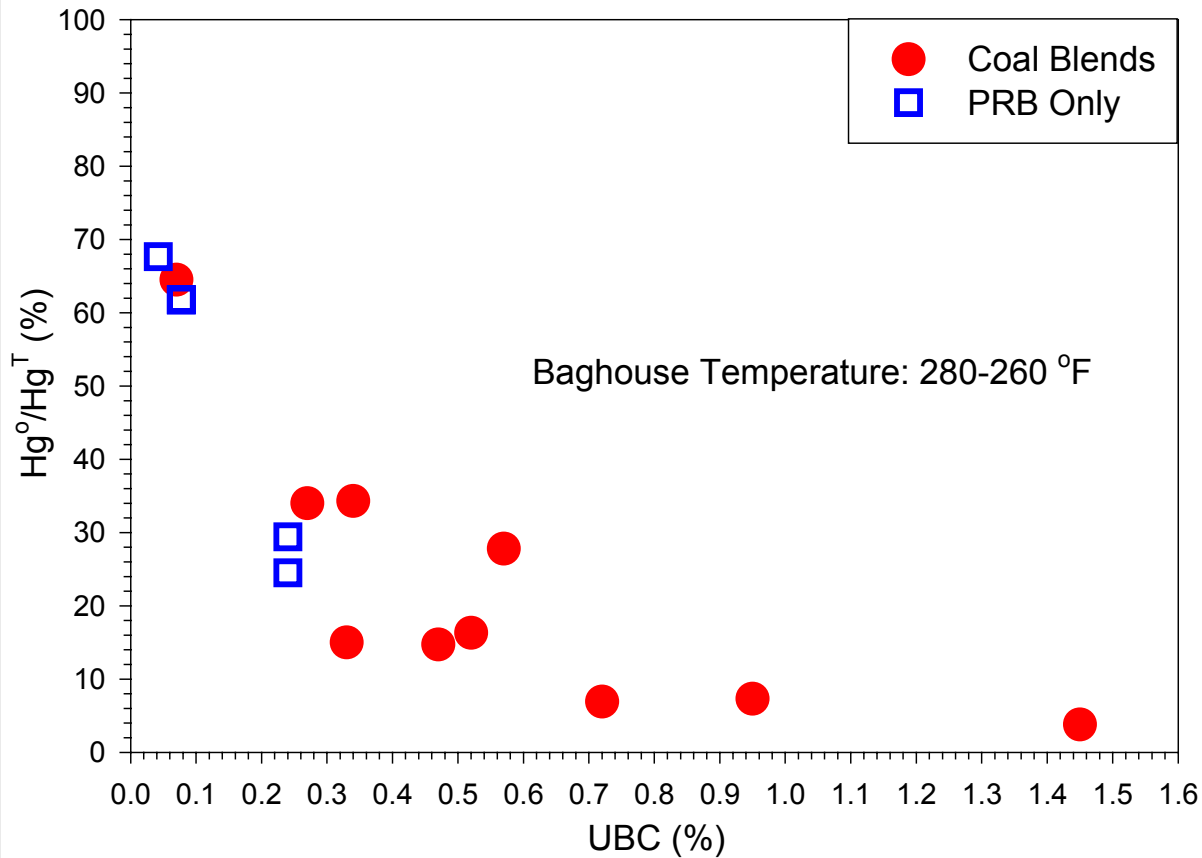
- NO Correlation with CO_2 , CO , O_2
- NO Correlation with NO_x or H_2O
- Through Isolation -- NOT SO_2
- What's Left? UBC and Coal Minerals



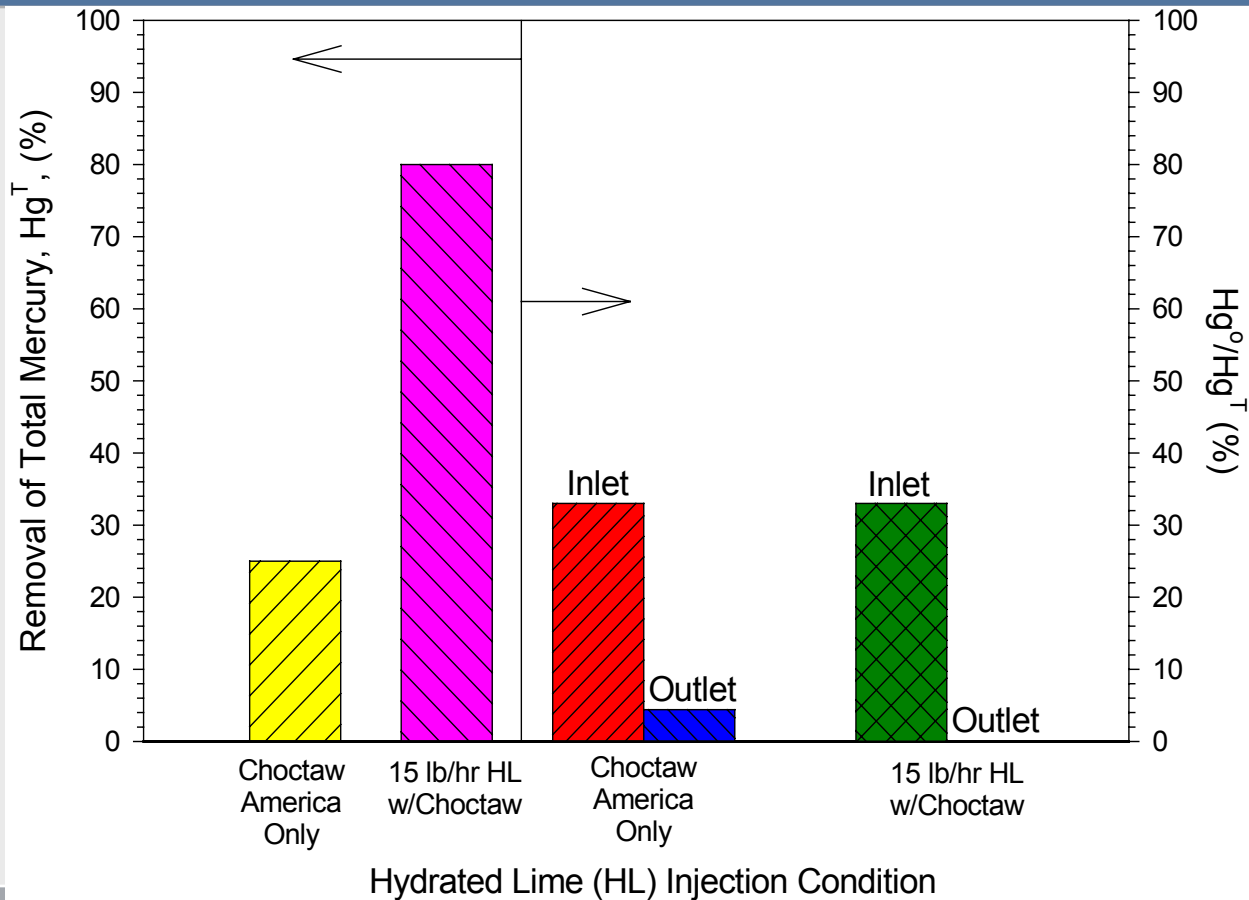
Effect of SO₂ on Hg-Speciation



ANSWER: Unburned Carbon



Hydrated Lime w/Catalyst is Effective



Conclusions Regarding Hg-Oxidation

■ FOR LOW UBC CONDITIONS

- Total chlorine content, injected through the burner or inherent in the coal, tends to increase Hg-oxidation prior to and across the baghouse.
- The catalytic material in coal ash is a more important factor in determining Hg-oxidation than total chlorine content.
- The primary parameter responsible for enhancement of Hg-oxidation for blends of PRB and bituminous coal is the UBC in bituminous ash.



Conclusions Regarding Hg-Removal

FOR LOW UBC CONDITIONS

- Total chlorine content, injected through the burner or inherent in the coal, has little effect on total mercury removal.
- Hydrated lime and even high-calcium ashes such as PRB can be effective sorbents, if they are mixed with a catalyst.
- The primary parameter responsible for enhancement of Hg-capture for the blends of PRB and bituminous coal in this investigation was the UBC in bituminous ash.
- Most effective Hg-removal was observed for high-calcium and high UBC concentrations in the ash.

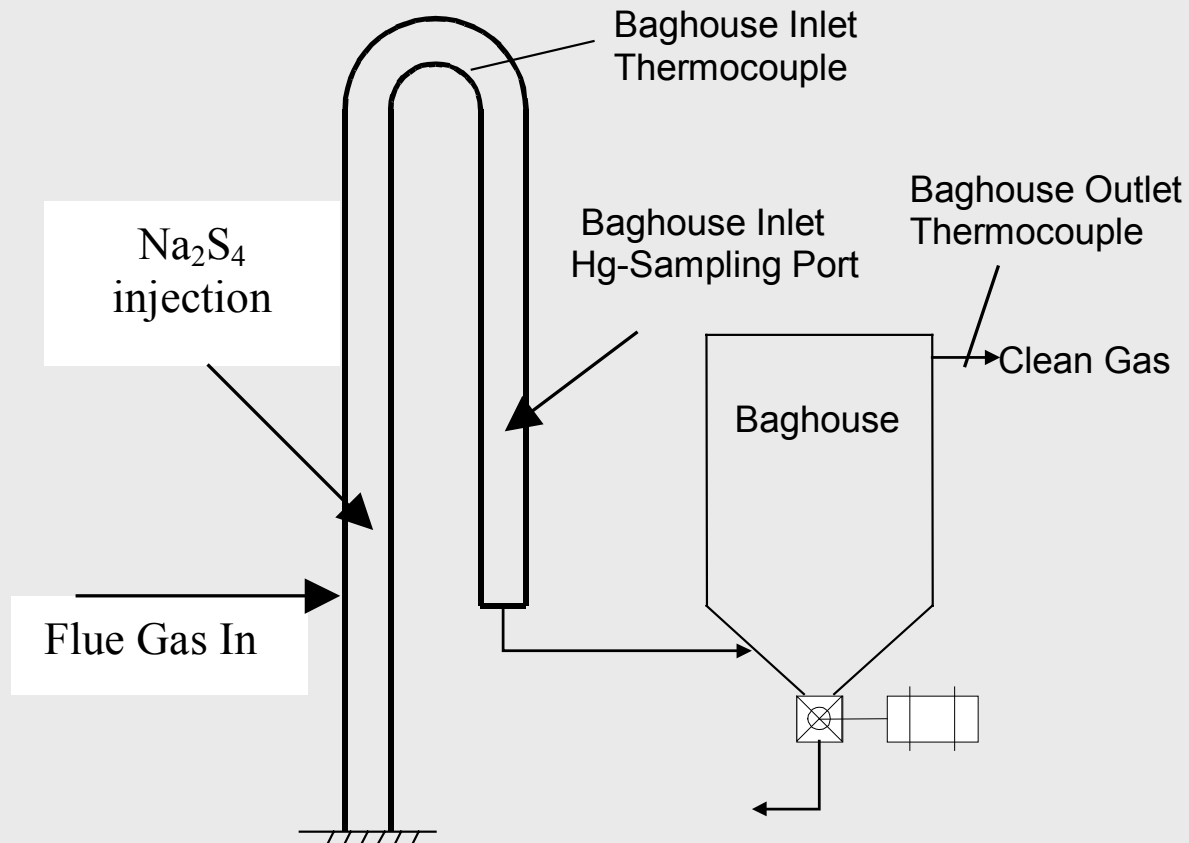


Na₂S₄-Injection

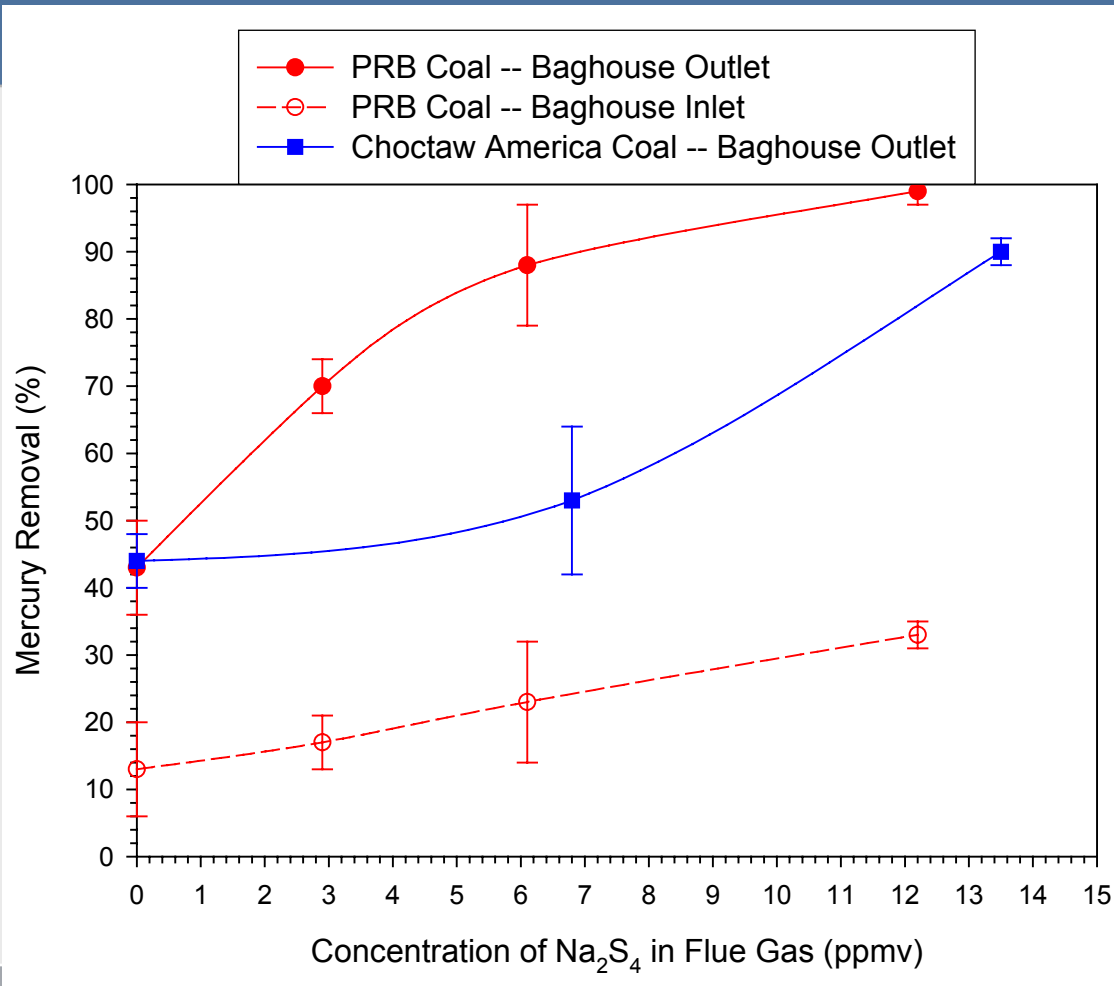
- Fine-spray injection of Sodium Tetrasulfide into flue gas before baghouse or ESP.
- Na₂S₄ decomposes into elemental S⁰ and ionic sulfur S⁻²
- Hg⁰ is captured by S⁰ and HgCl₂ by S⁻² to form HgS.
- HgS is the most stable and benign form of mercury in the environment.
- Results of injection ~2.0 seconds before baghouse.



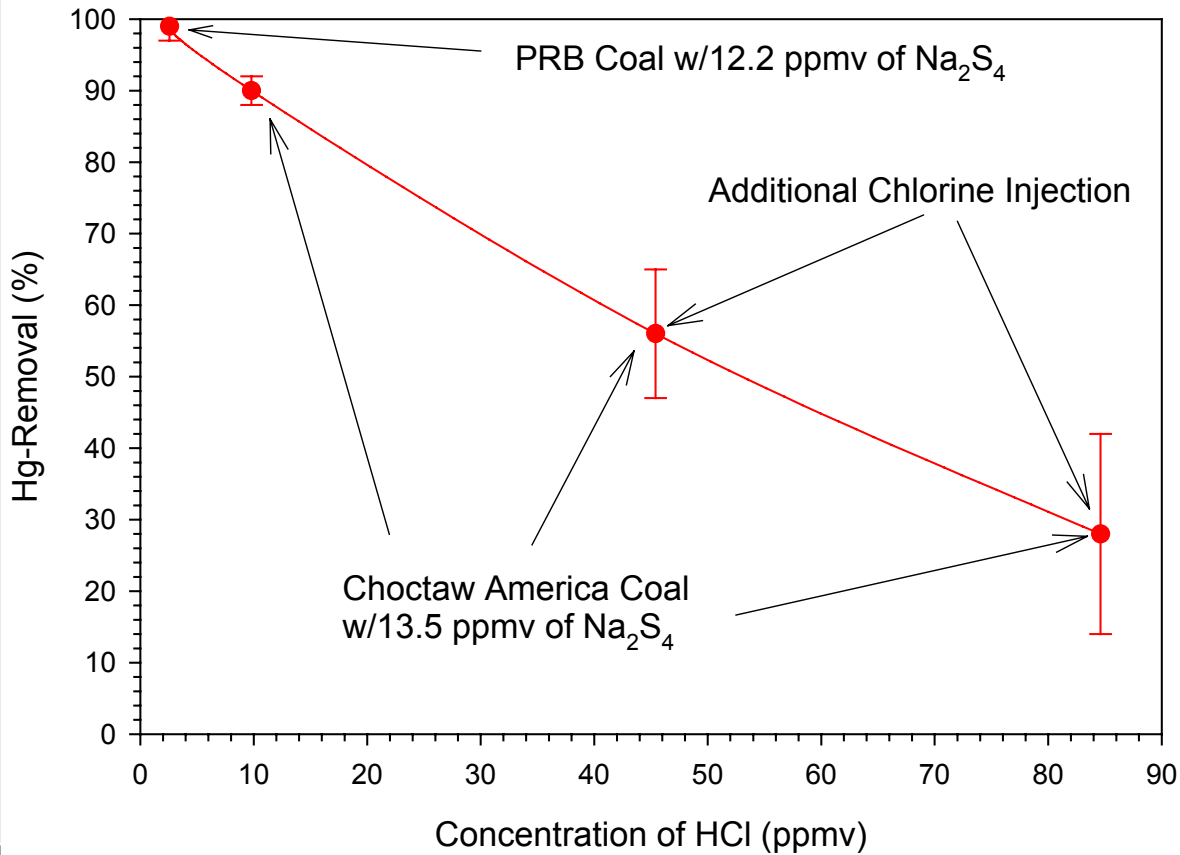
Na_2S_4 -Injection Location



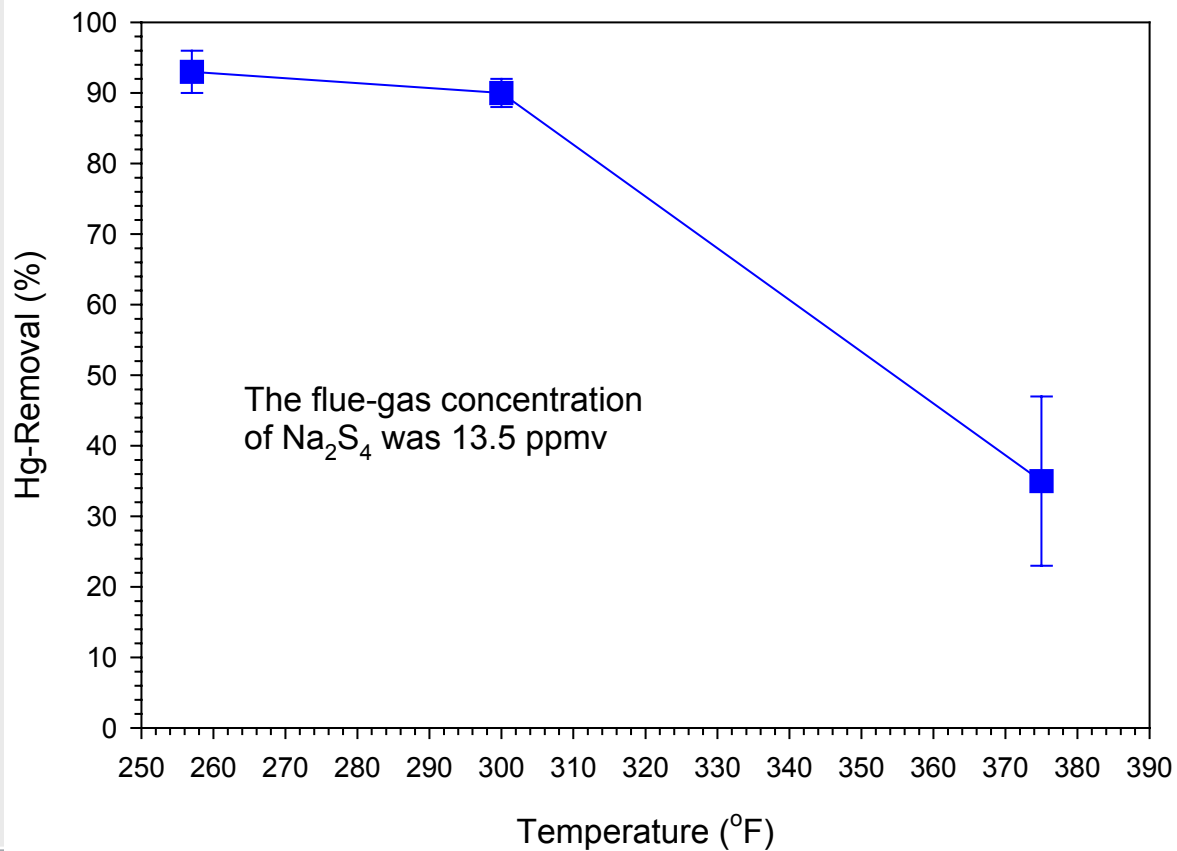
Gas-Phase Hg-Removal by Na_2S_4 -Injection



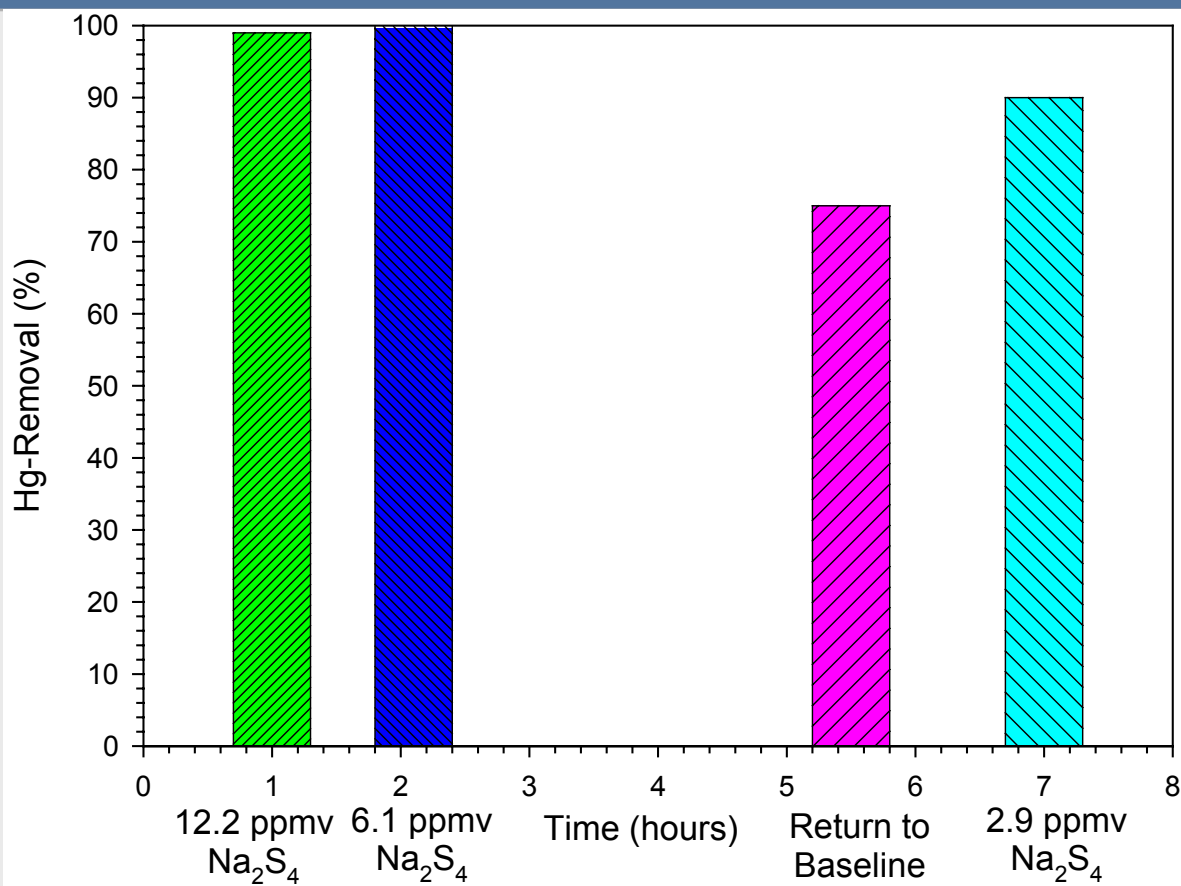
Effect of Chlorine



Temperature Effect on Na_2S_4 -Injection



Residual Effect of Na_2S_4 -Injection



Conclusions

Sodium tetrasulfide injection (~ 10 ppmv Na_2S_4 in flue gas), approximately 2.0 seconds ahead of a baghouse, is sufficient to removal 100% \pm 2% of flue-gas *vapor-phase* mercury, while burning a relatively low chlorine coal.

Injection temperatures above 350 °F are detrimental to the effectiveness of Na_2S_4 -injection technology, while injection temperatures as low as 250 °F appear favorable.

Chlorine in the flue gas reduced the effectiveness of Na_2S_4 -injection for Hg-capture directly proportional to the concentration of chlorine in the flue gas. However, previous work has shown this technology to be successful for use in high-chlorine waste-incineration flue gas.

Other than chlorine content, Na_2S_4 -injection technology was unaffected by differences in coal type or flue-gas composition in this investigation.

While Na_2S_4 -injection technology may be effective in the disperse phase (i.e., ESP applications), Na_2S_4 -injection in front of a baghouse benefits from a residual effect, probably associated with the baghouse filter cake. Hence, Na_2S_4 -injection in front of a baghouse may only require intermittent injection, and thus operational costs may be lower.



Future Work

- Calcium-based sorbent development and optimization will continue, utilizing information obtained on catalytic enhancement of Hg-oxidation.
- An optimized sorbent will be tested to observe the ability of this designer sorbent to remove mercury in the disperse phase (ESP) and in a baghouse.
- Sorbents specifically designed to remove SO₂ in a semi-dry recirculating system will be examined for their effectiveness as a Multi-Pollutant Control Technology for removing both SO₂ and Hg from the flue gas.
- Sodium tetrasulfide injection will be tested for its ability to remove mercury across an ESP.
- Finally, field-testing options will be explored for promising technologies.

