Carbon Sequestration Science

Focus Area Overview Presentation

- Mission and Scope
- Program Relationships
- Scientific Challenges
- Research Plans
- Facility Plans

July 2001





Carbon Sequestration Science Focus Area New Projects Contribute to Sequestration Science

Systems Integration Virtual Simulation of CO₂ Capture Technologies 3 km Coal C/ 20 °C, 20 atm Liquid CO₂, 100 tons Other ~1 kg CO₂ / s ² 5 MW **Fuels** 2 inch tube Flue gas Oxygen CO. CO -CO2 CO, Membrane CO, Combustor Oil Power 🕀 well MEA CO₂ Capture Facility plant Coal Other **Fuels** Gas CO, Stream ,CH₄ CO,/N, Sequestration Coal uur bed 🗲 СН Cleanup Aquiclude H₂O Oil field CO, H_O -H.O→ Aquifer CO₂ Oxygen Membrane Gasification Water 4 Rock

Carbon Sequestration Science Focus Area

Mission

 Provide scientific basis for carbon sequestration options for large stationary sources of CO₂

Goal

 Produce information that leads to development of economic and environmentally sound CO₂ capture and sequestration options

Scope

- Address:
 - Low-cost capture
 - Long-term storage
 - Beneficial utilization
- Sequestration options include:
 - Geologic (coal seams, deep brine fields, oil and gas reservoirs)
 - Ocean
 - Revolutionary, innovative approaches
- Support for policy development
 - Validation and verification

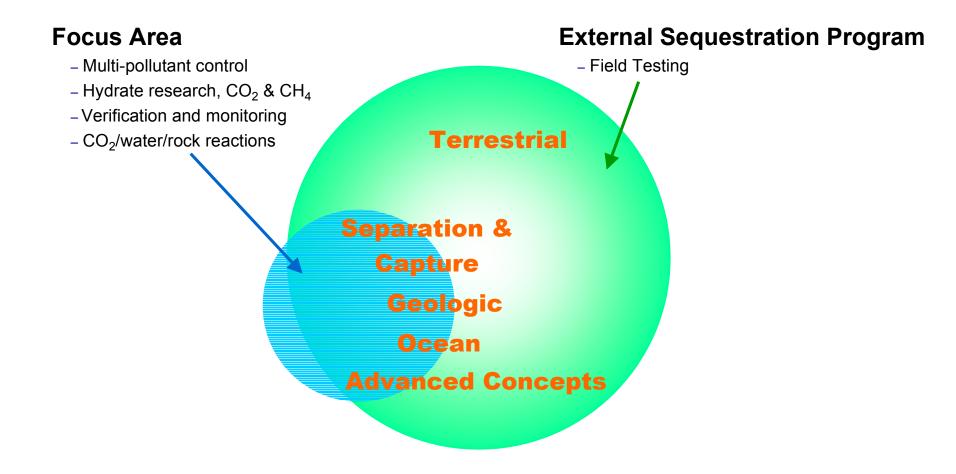


Carbon Sequestration Science Focus Area Tasks

- 1 Separation and Capture
- 2 Oceanic Sequestration
- **3** Chemical Sequestration
- 4 Geological Sequestration
- **5** Geological Sequestration Modeling
- 6 Process Modeling and Economic Assessment



Relationship to Office of Fossil Energy/NETL Sequestration Program





Focus Area Planning and Stakeholder Input

Research partnerships and potential collaborations

AES Corporation Albany Research Center Alberta Research Council Battelle Columbus Dravo Fluent, Inc. IMC Chemicals Los Alamos National Laboratory Monterey Bay Aquarium Research Institute Office of Science Center United States Geological Survey Sud Chemie

Arizona State University, Clarkson University, Pennsylvania State University, University of Akron, University of Texas, Regional Universities – Carnegie Mellon University, West Virginia University, University of Pittsburgh, and Duquesne University Burlington Resources Chung-Ang University Colorado School of Mines Dakota Gasification Energy Information Agency GoreTex National Research Council, Canada OPHIR Corporporation Selexol



Scientific Challenges and Research Plans – Capture and Separation – Task 1

• Scientific issues

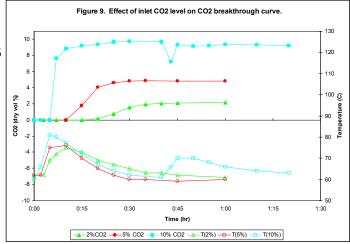
- Energy penalties in chemical scrubbing processes
- Effect of contaminants in gas streams
- Current processes not suitable for all applications
- Regenerability of sorbents
- Degradation of amines

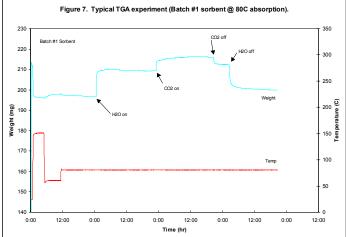
Research Plan

- Evolutionary and Revolutionary approaches
- Investigate novel solvents and sorbents
 - Physical solvents for IGCC and ammonia-based solvents for flue gas
 - Dry, regenerable sorbents
 - High and low temperature applications
- Study multipollutant control
 - CO₂, SO₂, NO_x, PM, Hg

Collaborators

- IMC Chemicals, AES Corp, Sud Chemie, Regional
- Universities





• Scientific issues

- Physical and chemical behavior of CO₂ in deep ocean
- Impact of ice-like CO₂ clathrate hydrate
 - Hydrate particle density
 - Hydrate coating effect on buoyancy and dissolution of CO₂
 - Promoter effects

Approach

- High-pressure, lab-scale research
- Mathematical simulations
- Water Tunnel Facility
 - Low pressure facility for design optimization
 - High pressure facility for simulation of oceanic water column

Collaborations

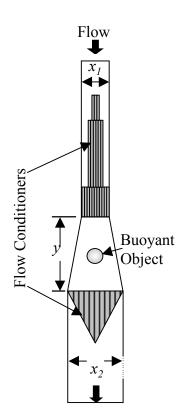
University of Pittsburgh, MBARI



Air bubble stabilized in downward flow of water in low pressure water tunnel



Development of a Water Tunnel Facility



Oblong Windows

Circular Windows

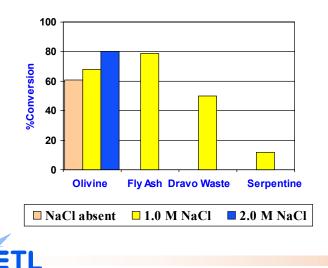


High-Pressure Water Tunnel Facility in newly renovated laboratory



- Kinetics heat, pressure, stirring requirements for mineral carbonation reactions
- Mineral carbonation of Mg-rich mineral silicates
- Scientific issues
 - Kinetics are slow, requires elevated temperatures and pressures as well as fine grinding

Effect of Solution Chemistry & Reactants on the Extent of Mineral Carbonation



Summary of Carbonation Experiments

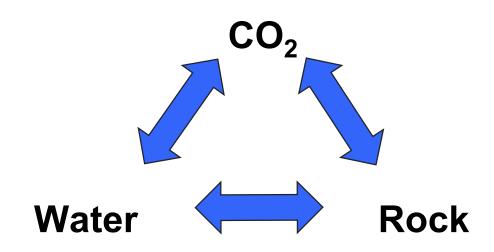
Reactant	HCO ₃ ⁻ Solution	Pct.Conv., MgCO ₃
Coal Fly Ash	Na ₂ CO ₃ /NaHCO ₃ /NaCl	80
Olivine	Na ₂ CO ₃ /NaHCO ₃ /NaCl	70
Dravo Waste	Na ₂ CO ₃ /NaHCO ₃ /NaCl	50
Serpentine	Na ₂ CO ₃ /NaHCO ₃ /NaCl	12

• Approach

- Find other sources of Ca/Mg rich feedstocks that are waste materials and explore the possibility of using them
- Study carbonation of mixtures of waste matter and mineral Mg silicates
- Modify solution chemistry of the mineral carbonation reaction by adding bicarbonate and Group I hydroxides
- Increase mineral porosity using steam and/or NaOH activation

Collaborators

• Regional Universities, LANL, Albany Research Center, Arizona State University





 Unmineable coal seams, deep saline aquifers, active and depleted oil and gas fields, including natural gas hydrates

• Scientific issues

- Understanding viscous fingering, gravitational segregation, permeability heterogeneity
- Understanding-multiphase flow
- Mineral dissolution and precipitation and their effects on permeability
- Better defining the kinetics of CO₂/water/rock reactions and the factors that influence it

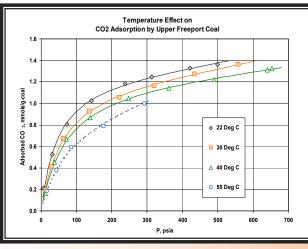




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Scientific issues

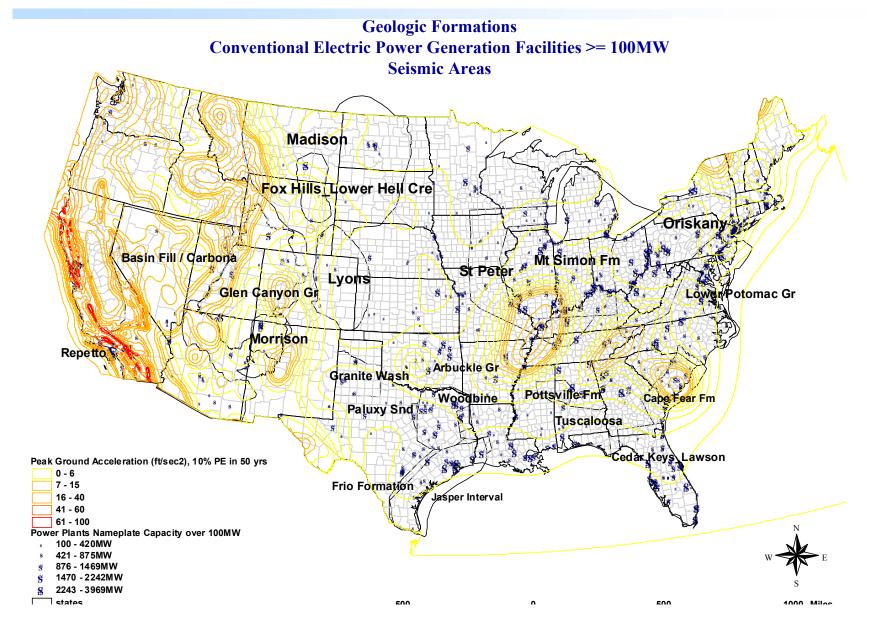
- Better understanding of those factors that affect the capacity of a coal seam to absorb CO_2 and the stability of the formation
- Defining factors that determine the optimum conditions for sequestration in each geological formation
- Developing means to verify and monitor the integrity of sequestration sites
- Gain increased understanding of the structure of CO₂ hydrates



Coal	Rank	Heat of Adsorption
Upper Freeport	Med. Vol. Bit.	5.65 +/- 0.46 Kcal/mol
Illinois No. 6	High Vol. Bit.	3.47 +/- 0.20 Kcal/mol

Summary of Coals Studied to Date

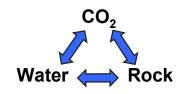
• Conclusion: The heat of adsorption of CO₂ is about the same as that of a hydrogen bond. The CO_2 is considered to be physically adsorbed. In the pressure range studied, the mechanism of adsorption appears to be physical adsorption into both micro- and meso-pores.

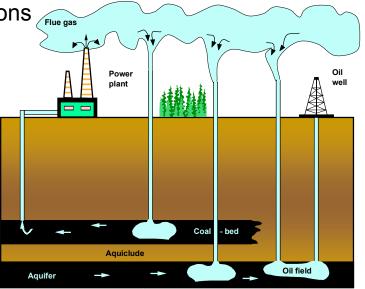




• Approach

- Use phase/volume measurements and high pressure flow calorimetry to study phase behavior and heats of mixing
- Experiment and model two-phase flow through porous media measure flow patterns
- Begin detailed experimental and modeling study of CO₂/water/rock chemistry
- Measure and model adsorption isotherms and heats of sorption/desorption for Argonne coals under various conditions
- Begin developing ground based monitoring systems for tracers and geological gases







Potential Collaborators

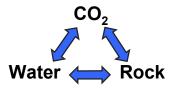
Clarkson University, Pennsylvania State University, Colorado School of Mines, University of Texas, Chung-Ang University

Regional Universities – Carnegie Mellon University, West Virginia University, University of Pittsburgh, and Duquesne University

Alberta Research Council

United States Geological Survey

Battelle Columbus Burlington Resources Fluent, Inc. GoreTex OPHIR Corporporation



Los Alamos National Laboratory

National Research Council, Canada



 Unmineable coal seams, deep saline aquifers, active and depleted oil and gas fields, including natural gas hydrates

Scientific issues

- Understanding viscous fingering, gravitational segregation, permeability heterogeneity
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- Better defining the kinetics of CO₂/water/rock reactions and the factors that influence it
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Water 📥 Rock

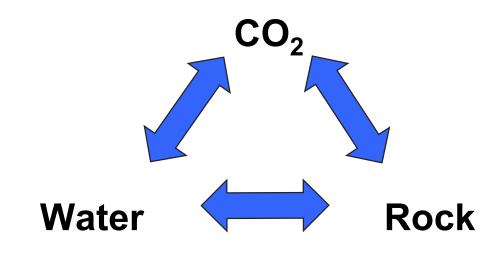
• Approach

- Combine experiments and modeling with traditional methods of petroleum engineering
 - An Iterative interaction between computer simulation and lab experiments
- Obtain existing models and codes to identify desirable traits that make them good candidates for revision to include sequestration of CO₂
 - University of Texas reservoir simulator developed by Gary Pope
 - SOMINEQ USGS code that models CO₂/water/rock interactions
 - EQ6 code that models CO₂/water/rock interactions
 - ACS model for complex ionic equilibria
 - Western Research Institute model for enhanced methane recovery from coal seams
 - FRACGEN and NFFLOW geological and fluid flow models
 - EIA gas reservoir models
 - PROSIM CO₂/Brine reaction simulator



Approach continued

- Estimate manpower/resources to amend the codes to include a representation of sequestration in gas, oil, brine and coal fields
- Begin an iterative set of lab/modeling experiments
- Revise several codes to include CO₂ sequestration





Scientific Challenges and Research Plans -Geological Sequestration Modeling – Task 5 Potential Collaborations

Clarkson University, Pennsylvania State University, University of Texas, Chung-Ang University Battelle Columbus Western Research Institute Fluent, Inc.

Regional Universities – Carnegie Mellon University

Alberta Research Council

United States Geological Survey

Energy Information Agency



Scientific/Technical Issues

- What are the economics of existing CO₂ capture processes
- What are the most inefficient steps and process points in each existing capture technology
- What new technological innovations could be used to improve these inefficient steps
- How much of an impact will each incorporated innovation make







• Approach

- Using ASPEN and variations thereof, model the existing capture plants
 - AES Corp. Cumberland, MD
 - Rectisol Dakota Gasification
- Convene group of experts from industry, academia and government incorporate newest technological innovations into the models, determine effects





Potential Collaborations

IMC Chemicals

AES Corp.

Dakota Gasification

Selexol





Facility Plans

• Capture research

- Modular facility
 - Advanced instrumentation and diagnostics

Geologic sequestration simulation

- Multipurpose facility for simulating various geologic formations
 - Highly instrumented, advanced imaging

Oceanic sequestration research

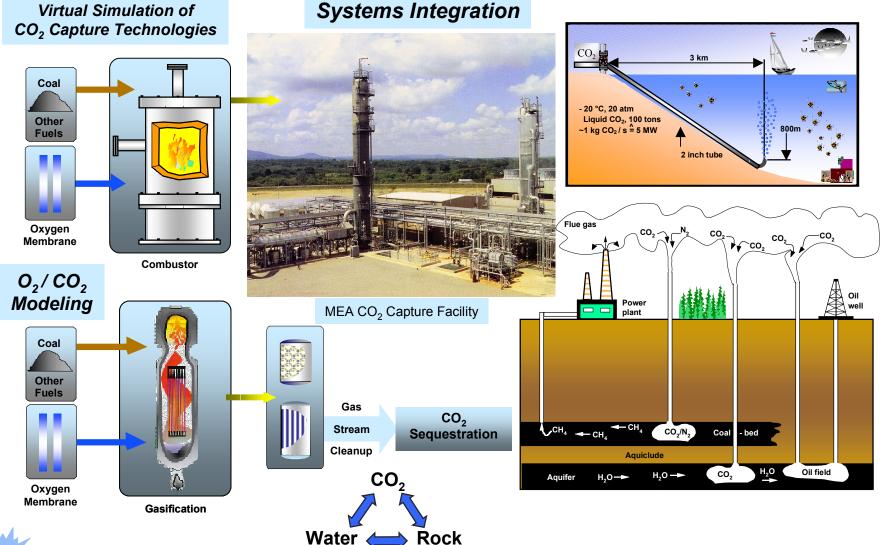
- High pressure water tunnel
 - Under construction
 - Machine vision, advanced imaging
- Integrated carbon sequestration research facility
 - In initial planning stage



High-Pressure Water Tunnel Facility



Carbon Sequestration Science Focus Area New Projects Contribute to Sequestration Science



ETL