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SAMPLING AND ANALYSIS PLAN FOR ISSAQUAH CREEK VALLEY GROUNDWATER MANAGEMENT AREA

Prepared for:

King County Department of Natural Resources Seattle, Washington

Submitted by:

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1. INTRODUCTION

1.1 Project Description

Between 1989 and 1996, the Seattle - King County Department of Public Health, under Washington Department of Ecology guidance (according to WAC 173-100), developed Ground Water Management Plans (GWMPs) for five Ground Water Management Areas (GWMAs) in King County, including the Issaquah Creek Valley (ICV) GWMA (Figure 1). Each GWMP included proposals for data collection and management, with the purpose of monitoring both the groundwater quality and quantity.

Water quality in the ICV GWMA is generally excellent. The emphasis of the GWMP is therefore to protect existing water quality. The GWMP proposed eighteen goals to address groundwater quality and quantity, including hazardous materials management, infrastructure (e.g. sewage treatment, underground storage tanks and pipelines, and landfills), pesticides, golf courses, well construction, and sand and gravel mining (Issaquah Creek Valley Ground Water Advisory Committee, 1999).

The study served by this SAP is intended to provide an overall description of groundwater conditions in the ICV GWMA and, in combination with sampling in the other GWMAs, throughout King County as a whole. Under this plan, sampling will be conducted at wells previously sampled during development of the ICV GWMP. (In 1996, after the data collection and analysis had been completed, a portion of the Sammamish Plateau in the East King County GWMA was moved to the ICV GWMA; some of the wells to be sampled for the present study were actually inherited from the earlier East King County study.) Comparison of water quality results from the current study to previous sampling events will permit evaluation of long term changes in water quality. To this end the selection of wells, sampling techniques, and analytical methods for this present study have been made as compatible as possible with the previous GWMP sampling. Variability among replicate samples for a given parameter during the original GWMP sampling should provide an indication of the significance of any observed changes during the present sampling rounds. Water levels recorded at the time of the present rounds of sampling will be compared to earlier measurements to evaluate changes in groundwater quantity.

1.2 Purpose

This Sampling and Analysis Plan (SAP) has been developed to guide the initial groundwater sampling in the Issaquah Creek Valley GWMA. Initial efforts under this SAP will begin in 2001. Following receipt and interpretation of the results of this effort, the data collection program will be revised in conjunction with the ICV Ground Water Management Committee, according to the new information available or identification of new areas of concern. This SAP will be revised to cover subsequent sampling efforts.

1.3 Study Area Description

The Issaquah Creek Valley Groundwater Management Area (GWMA) is an area approximately 94 square miles in size, located about 15 miles east of Seattle. It includes the Issaquah Creek and Tibbetts Creek drainage basins and the Sammamish Plateau. The boundary is based on the assumption that groundwater divides coincide with surface water basin boundaries and accommodates water district service areas. The GWMA forms the southern and eastern portion

of the larger Lake Sammamish watershed (Issaquah Creek Valley Ground Water Advisory Committee, 1999).

1.4 Sampling Plan Objectives

Data collected under this SAP is intended for use by both the public (including Management Committees) and technical people to assess groundwater quality and develop future sampling needs. This data will also be used for Environmental Benchmarks under the Growth Management Act. This SAP outlines field procedures and quality control procedures that will be followed to ensure a complete and accurate data set. Although groundwater will be analyzed for drinking water parameters, analysis need not be by drinking water methods or be performed by a lab certified to perform drinking water analyses. Drinking Water Analysis Reporting Forms also do not need to be used. However, method detection limits (MDLs) below maximum contaminant levels (MCLs) will be attained.

1.5 Sampling Locations and Frequency

Approximately twenty-four wells in the ICV GWMA will be sampled. The locations of these wells are shown in Figure 2. Groundwater sampling under this Plan will be conducted semi-annually. The first round of sampling will be conducted in March 2001. Another round of sampling will be conducted in summer 2001. As such, samples will be collected during both wet and dry seasons.

1.6 Project Organization and Responsibilities

Activities outlined in this SAP will be conducted by personnel from both King County Department of Natural Resources (KCDNR) and Golder Associates (Golder). Ken Johnson (KCDNR) and Bob Anderson (Golder) will assume the roles of Project Managers in their respective organizations. Contact information is provided below:

Ken Johnson King County Department of Natural Resources Water and Land Resources Division Regional Water Resources Services Unit 201 S. Jackson St. (King Street Center), Suite 600 Seattle, WA 98104-3855 (206) 296-8323

Bob Anderson Golder Associates 18300 NE Union Hill Road, Suite 200 Redmond, WA 98052-3333 (425) 883-0777

1.6.1 King County DNR Responsibilities

Chris Hughes will be in charge of the groundwater sampling program under the direction of Ken Johnson. Prior to sampling, King County personnel will arrange site access with all landowners and for the provision of sample bottles and coolers from the King County Environmental

Laboratory (KCEL). KCDNR personnel will conduct all groundwater sampling, transmit samples to KCEL and process and manage water quality results.

1.6.2 Golder Associates Responsibilities

Golder is responsible for preparation of this sampling and analysis plan (SAP). Golder personnel will provide groundwater sampling assistance upon request.

2. DATA QUALITY OBJECTIVES

2.1 Overview

As stated previously, the goal of groundwater sampling under this SAP is to evaluate groundwater quality and quantity in the ICV GWMA. The wells selected for monitoring provide good geographic coverage of the study area. During well selection, representation from major geologic units was considered, but the well selection criteria included the potential or previous detection of contamination. These wells should be more likely than others in the area to show problems and thus would be an "early warning" system for contamination in the GWMA.

Because the goal of this study is to evaluate groundwater quality on a regional scale, the primary focus of this study is the identification of widespread water quality problems. Well selection included an evaluation of susceptible areas based on land use, though no attempt has been made to sample any areas of possible small scale or point source impacts.

2.2 Target Analytes

Samples will be analyzed for the constituents listed in Table 1. Field parameters (pH, conductivity, dissolved oxygen, turbidity, and temperature) will be monitored at all wells.

3. SAMPLING AND ANALYSIS PROCEDURES

3.1 Field Data Collection

3.1.1 Sample Collection

The wells chosen for this study are primarily domestic wells. Figure 3 illustrates a typical configuration of a domestic well house. Because well configurations will vary from site to site, guidelines will be provided for the point in the distribution system where samples should be collected and the volume of water that should be purged prior to sample collection. At each site, professional judgement will be exercised in implementing these guidelines. To ensure comparability of the data between sample rounds, well-specific sampling protocols will be documented.

The goal of groundwater sampling is to collect a sample that is representative of subsurface conditions. For domestic water wells, water samples should therefore be collected from a hose bib in the well distribution system as close to the wellhead as possible. Three well volumes should be purged prior to sample collection. For some wells, purging three full well volumes may not be possible due to slow pumping rates. In these cases, a minimum of one well volume will be purged and then samples will be collected once field parameters (pH, conductivity, temperature, dissolved oxygen and turbidity) have stabilized. If feasible, samples should be collected upstream of any holding tank (e.g., cistern, water heater, pressure tank). If samples are collected downstream of a holding tank, the full volume of the holding tank should be purged prior to sample collection. In systems where this would amount to large volumes of water, samplers should consult the Project Manager. Samples should be collected ahead of any water treatment such as chlorination, fluoridation, or softening.

Sampling procedures specific to springs are outlined in Section 3.1.2. Two-inch monitoring wells may be sampled using low flow procedures as described in Section 3.1.3.

A photograph (or sketch) will be taken to document each systems configuration. At each well, a well sampling field form (Attachment A) will be completed. Prior to sampling, the water level will be recorded, provided the well head is accessible. Well details and purge volumes will be recorded on the well sampling form. At each well, field parameters (pH, conductivity, dissolved oxygen, turbidity, and temperature) will be recorded during purging. Purge water will be directed away from the well house using a garden hose. Field parameter samples will be collected at the end of this hose. Water samples will be collected once field parameters have stabilized and turbidity is less than 5 NTU, as outlined in Golder Technical Procedure TP 1.2-20. Whenever possible, samples will be collected directly from the hose bib. A complete set of field parameters will be measured on a sample collected directly from the hose bib (as opposed to at the end of the garden hose) to ensure transmission through the garden hose has not affected field parameters.

In cases where samples cannot be collected directly from the hose bib (e.g., high pressures, potential flooding of the well house) polyethylene or tygon tubing will be used. This tubing will be attached to the hose bib with a hose clamp. Tubing will be decontaminated by the KCEL prior to sampling. The lab will verify cleanliness of the tubing for metals sampling before sending it out in the field. No de-con of sampling collection apparatus (with the exception of field parameter sample bottles) will occur in the field. Separate tubing will be used for metals and bacteriology samples. Sterile tubing will be used to collect bacteriological and all other samples

with the exception of metals. Metals will be collected using a second acid-washed tube. Acid-washed tubing must not be used to collect conventional parameters.

Groundwater sampling will be conducted in accordance with Golder Technical procedure TP 1.2-20, "Collection of Groundwater Quality Samples." This procedure describes sample purging, sample collection, and measurements to ensure representative groundwater samples are obtained using bailers, a variety of portable pumps or with dedicated pumps.

Table 2 provides sample volumes, preservation and holding times for groundwater samples to be collected at each well. One inch of headspace will be left in all bottles, if possible. If a bottle is overfilled, excess water will not be decanted from the bottle. The metals collection bottle will be rinsed three times with sample water prior to filling. Samples for all other parameters will be collected directly into the appropriate sample bottles. A water sample collected for bacteriology must not contact any surface that is not sterile prior to transfer to the bacteriology bottle.

Because of the low detection limits required for the metals tests, precautions will be taken during sampling to minimize contamination. Sample containers for metals samples will be sealed in zip lock bags before and after sample collection. Sample collection personnel will wear clean, powder free vinyl gloves while handling the zip lock bags and the sample containers. Gloves will be changed frequently. The field blank, a sample bottle containing reagent water from the lab, will be handled in the same manner as the samples. This includes removing the bottle from the zip lock bag, opening the bottle, exposing the blank water to the ambient conditions at a typical sampling site for the duration of sample collection, and returning the bottle to the zip lock bag.

All samples will be placed in the appropriate sample containers and labeled. Golder Chain of Custody procedure TP-1.2-23 will be followed during storage and shipment of all samples for analysis. Samples will be shipped or dropped off at the laboratory for analysis at the end of each day. Short holding times for nitrate, nitrite, and coliform necessitate this. At the request of KCEL, sample collection will not be conducted on Fridays. This will ensure that all analyses are conducted within the appropriate holding times.

Field analysis of total nitrate, iron, and manganese will be conducted with HACH kits at the discretion of the Project Manager. HACH kits allow rapid and easy determination of nitrate, iron and manganese concentrations using colorimetric methods. The method detection limits of these analyses are 1 mg/L (nitrate), 0.1 mg/L (iron), and 0.02 mg/L (manganese). Concentrations are determined by comparison of the color of a groundwater solution following treatment with reagents to a color chart. These analyses are therefore appropriate as "screening" tools. The precision and accuracy of these results are considered secondary to laboratory determined values. Field-tests to evaluate biological activity (BART tests) may also be employed at the direction of the Project Manager. BART tests provide an easy way to detect specific bacterial groups and algae in water. Bacteria may impart offensive taste and odor in potable water or result in plugging of wells. The following types of bacteria can be evaluated using BART tests: iron-related, nitrifying, sulfate reducing, slime-forming, and total aerobic. Personnel are referred to the BART manual for specific instructions regarding each test.

3.1.2 Sample Collection from Holding Tanks

Spring sampling may require sample collection from a tank or reservoir. Changes to standard procedures may also result in sampling from a well's holding tank. When sampling directly from a holding tank or other reservoir, the sampler shall wear new disposable gloves. A sterilized

bailer will be used to collect the bacteriology and conventional samples. A separate, acid-washed bailer will be used to collect the metals samples. All samples collected with the bailers will be transferred directly into the correct lab containers. If the holding tank is part of a potable water system, extreme care must be taken to prevent contamination of the water source with the sampling equipment. Bailers will be provided by KCEL.

3.1.3 Low Flow Purging and Sampling

At the discretion of the KCDNR Project Manager, two-inch monitoring wells included in the groundwater survey may be sampled using low flow sampling procedures. Low flow (low stress) sampling techniques described in this document follow procedures developed by the U.S. Environmental Protection Agency (EPA) as described in "USEPA Region II Low Stress (Low Flow) Ground Water Sampling Standard Operating Procedure" (USEPA, 1998).

The objective of low flow sampling is to pump in a manner such that stress (drawdown) to the system is minimized to a practical extent. Typically, flow rates on the order of 100 to 500 mL are used (as outlined in the procedure below); however, flow rates will ultimately be dependent on site-specific hydrogeology. The benefits of low flow sampling are 1) minimization of sediment disturbance at the bottom of the well thereby lowering turbidity; and 2) reduction in purge volumes.

EPA states that their procedure is applicable to monitoring wells with a diameter of 2 inches or greater and a maximum screened interval of 10 feet.

Low flow sampling should be conducted as follows:

- 1. Record water level in well.
- 2. Slowly lower the pump into the well positioning the pump intake at the middle or slightly above the middle of the screened interval. The pump intake should be kept at least two feet above the bottom of the well to prevent resuspension and entrainment of any sediment present in the bottom of the well. Lowering the pump slowly into the well should minimize disturbance of water in the well column. Most of the need for purging has been found to be due to passing the sampling device through the overlying casing water which causes mixing of these stagnant waters and the dynamic waters within the screened interval (Puls and Barcelona, 1995).
- 3. Record the pump depth.
- 4. Record the water level with the pump now in position. Leave the water level measuring device in the well.
- 5. Purge the well at 200 to 500 mL/min while ensuring the water level drawdown does not exceed 0.3 feet. In some cases, the well yield may be insufficient to maintain a drawdown of less than 0.3 feet while continuously pumping at the minimal purge rate. In these instances, pumping should be temporarily stopped to allow well recovery before continuing well purging. Pumping rates should be recorded on the well sample form. Wells are considered sufficiently purged when field parameters remained stable for 3 consecutive measurements taken approximately 5 minutes apart. EPA defines stabilization as follows:
 - +/- 0.1 unit for pH;

- +/- 3% for conductivity; and
- +/- 10% for dissolved oxygen and turbidity.

The EPA criterion for conductivity is more stringent than that defined in Golder Technical Procedure TP 1.2-20 "Collection of Groundwater Quality Samples" in which stabilization is defined as a variation of less than 10%. For ICV sampling, a 10% criterion for conductivity will be applied.

Once field parameters have stabilized, the pumping rate should be reduced to 100 to 250 mL/min and the samples collected. Sample collection will follow the procedures outlined above for domestic well sampling.

3.1.4 Sample Designation

Each sample will be clearly labeled with a unique identification number, which will also be used on the Chain of Custody, and field logbooks for identification and tracking and for use in the database that will be developed. The sample identification code format will consist of a unique number and increment sequentially for each new sample. The following information will be collected on the sample integrity sheets or a field book for entry into the database to further describe samples:

- 1. Sample Location (including LIMS Locator and GIS Coordinates);
- 2. Filtered/Unfiltered: Filtered = F; Unfiltered = U;
- 3. Filter Size:
- 4. QC Type: (Blank = Bk, Duplicate = Dp);
- 5. Date: Mo/Da/Yr;
- 6. Time: Hour and Minutes HH:MM; and
- 7. Sampler's initials.

3.1.5 Equipment

During all sampling activities, disposable equipment will be used whenever possible. All non-dedicated sampling equipment (in contact with sample) shall be thoroughly cleaned prior to each sampling event to prevent cross-contamination between samples and to ensure accurate representation of analytes of interest in each sample. No field cleaning procedures shall be performed for any bacteriology sampling; all sample containers and sampling equipment shall be sterilized and transported to the field under conditions to preserve its sterility. No field cleaning procedures shall be performed for metals sampling. Sampling equipment for metals samples will be prepared at the lab prior to each sampling event and transported to the field in plastic bags to preserve cleanliness. The analytical laboratory as part of their agreement shall provide all sample containers, container preparation services, preservatives, gloves, and field blanks.

Non-dedicated pumps and tubing used to purge two-inch monitoring wells will be decontaminated between wells following procedures outlined in Technical Procedure 1.2-20.

Field meters (pH, dissolved oxygen, conductivity, temperature, and turbidity) will be thoroughly rinsed with distilled or deionized water before and after each reading. Personnel performing

decontamination of field testing equipment shall wear gloves, eye protection, and such other safety equipment as needed. Equipment decontamination procedures are described in Golder Technical Procedure TP 1.2-20.

3.2 Laboratory Analysis

Groundwater samples will be submitted to KCEL for analysis. KCEL will arrange for a subcontract with AmTest, Inc., located in Redmond, Washington for the analysis of Total Organic Halides, Silica and Fluoride.

3.2.1 Analytes and Method Detection Limits

Table 1 lists the parameters to be analyzed at each well. Groundwater will be analyzed for drinking water parameters using method detection limits (MDLs) below the maximum contaminant levels (MCLs). Table 3 lists the MDL that will be achieved for each parameter based on the chosen analytical method. This table also provides Washington State Drinking Water Limits and Groundwater Limits for comparison. Except for total mercury, which will be determined by cold vapor atomic absorption spectroscopy, all metals testing will be conducted by ICP-OES and ICP-MS. Based upon expected sample values, total calcium, magnesium, sodium, potassium and iron will be reported by ICP-OES and total arsenic, barium, cadmium, chromium, copper, lead, manganese, selenium, silver and zinc by ICP-MS.

3.2.2 Bacteriology Follow-up Samples

In the instances where bacteriology results suggest that resampling of the potable water is indicated, the laboratory project manager will contact the Project Manager to provide verbal notification of the results. If resampling is ordered, the samples will be collected from the appropriate potable tap, using appropriate sampling containers and technique. The sample should be delivered to the lab clearly labeled as a follow-up POTABLE water for Total Coliform analysis. Samples labeled as Potable Water will receive confirmation for Fecal Coliform as part of the Total Coliform analysis under Drinking Water criteria.

4. QUALITY ASSURANCE/QUALITY CONTROL

4.1 Field Equipment Calibration and Maintenance

Calibration of all measuring and test equipment, whether in existing inventory or purchased for this investigation, shall be controlled as required by Golder procedure QP-11.1, "Calibration and Maintenance of Measuring and Test Equipment." Lease equipment shall require certifications or other documentation demonstrating acceptable calibration status for the entire period of use for this project. Field calibration requirements shall be in compliance with the technical procedure describing the instruments use and/or with the manufacturer's instructions issued with the equipment.

4.2 Field Measurement OC

To assess the precision of field measurements, duplicate measurements will be obtained at a frequency of 1 in 10 measurements or 1 set per day, whichever is greater. Duplicate measurements should be made using a fresh portion of sample relative to the original measurement and after rinsing the meter using the same procedures used between wells. A copy of the field results for pH, conductivity and turbidity will be left with the laboratory at sample delivery.

4.3 Field Replicates and Blanks

To assess the precision of field sampling procedures and the variability of the sample source, field replicates will be collected at a frequency of 1 in 10 samples or 1 per day, whichever is greater. Each field replicate is to be collected after performing the decontamination and purging routines used for normal sample collection. One field blank, provided by KCEL, will also be collected each day for metals analysis. The information obtained by collecting field replicates and blanks will be taken into account by the data user when making decisions based on data generated under this QA plan.

Prior to collection of the field replicate, field parameters (pH, conductivity, turbidity, dissolved oxygen and temperature) will be measured to ensure that they are still stable and turbidity is less than 5 NTU. If field parameters are consistent with the original sample, the replicate will be collected. If field parameters are no longer stable (stability as defined in TP 1.2-20), the well will continue to be purged until parameters re-stabilize. Upon stabilization, the replicate sample will be collected.

4.4 Laboratory QA/QC

4.4.1 Precision

Laboratory precision will be assessed using laboratory duplicates. When both sample results exceed the RDL (reporting detection limit) the RPD (relative percent difference) should be less than 25% or the current lab acceptance limit, whichever is lower. For microbiology measurements, the RPD should meet the method requirements. No criteria are presented for duplicate results that are below the RDL, but above the MDL, as these RPD are provided for informational purposes only.

4.4.2 Accuracy

For this project, laboratory control samples, or blank spikes, whichever are available, will be used to assess accuracy. Results should be within 20% of the true value or within the criteria provided with the purchase of the control sample.

Accuracy will also be assessed by the evaluation of method blank data. Analytical results for method blanks should be less than the MDL (method detection limit). Note that some common organics laboratory contaminants may exceed the reported MDL. Sample results that are less than 10 times the concentration detected in the method blank will be qualified with a "B" flag to indicate the sample results may be biased.

The use of matrix spike recovery data will provide additional information regarding method performance on actual samples. The laboratory will use professional judgment regarding assessment of data quality and any subsequent action taken as a result of matrix spike recoveries.

4.4.3 Representativeness

Representative samples will be obtained through the following practices:

- The use of generally accepted sampling procedures will allow for the collection of representative samples.
- Subsampling within KCEL will be conducted according to lab standard operating procedures. These procedures are designed to obtain representative subsamples.

4.4.4 Comparability

Data comparability will be obtained through the use of standard sampling procedures and trained personnel as described in this plan, and through standard analytical methods used by the laboratory. Additionally, adherence to the procedures and QC approach contained in this QA Plan will provide for comparable data throughout the duration of this project.

4.4.5 Completeness

Completeness will be evaluated by the following criteria:

- The number of usable data points compared to the projected data points as detailed in this plan.
- Compliance with the data quality criteria as presented in this section.
- Compliance with required holding times.

The goal for the above criteria is to obtain 100% data completeness. However, where data are not complete, decisions regarding resampling and/or reanalysis will be made by a collaborative process involving both data users and data generators. These decisions will take into account the project data quality objectives as presented above.

5. DATA MANAGEMENT AND REPORTING

5.1 Field Reports

Personnel will record daily sampling activities in a field book or field reporting form. A well sampling form (Attachment A) will be completed at each site. At least one photograph (or sketch) will be taken to document each well.

5.2 Laboratory Reports

Data reduction, review, and reporting will be performed under KCEL's standard operating procedures. Data will be provided to data recipients within 30 days of receipt of the last sample for a sampling event. Data will be reported in the standard laboratory reporting format. This includes an analytical result, MDL (method detection limit), and RDL (reporting detection limit).

All analytical data packages submitted by the analytical laboratory shall include the following:

- Sample receipt, chain-of-custody and shipping documentation, including identification of field sampling personnel, shipping personnel (or organization);
- Analytical results for each sample containing the reduced results for all analytes/constituents requested in the chain of custody, request for analysis or purchase order; and
- Sample results will be available through LIMS or in an electronic version (Excel) of the hardcopy report (Comprehensive Report). QC summaries are available in hardcopy form only.

5.3 Database Management

Prior to entry into the database, all analytical data will be validated. The Project Manager will dictate the level of data validation. Procedures outlined in USEPA Contract Laboratory Program National Functional Guidelines for Inorganic Data Review (EPA, 1994) may be used as a reference during data validation.

Each sampling event for a specific well will be assigned a unique numerical identifier for use in the database. Water quality data will be entered into EQUIS or Access.

5.4 Annual Water Quality Report

At the completion of one year of sampling (two rounds), water quality data will be summarized in an annual report. This report will include:

- Tabulation of all water quality data;
- Identification of exceedences to drinking water and groundwater standards;
- Water quality trend analysis;
- Graphical representation of water quality types (stiff or piper diagrams); and
- Mapping of water quality data to show geographical distributions.

This report should also include an evaluation of the QA/QC protocols. This evaluation may include evaluation of field replicates and field blanks. Charge balances may also be performed on individual water analyses as a tool in assessing analytical accuracy.

6. REFERENCES

- EPA, 1994. USEPA Contract Laboratory Program National Functional Guidelines for Inorganic Data Review. EPA-540/R-54-013.
- EPA, 1998. Final USEPA Region II Low Stress (Low Flow) Ground Water Sampling, March 16, 1998.
- Issaquah Creek Valley Ground Water Advisory Committee, 1998. Issaquah Creek Ground Water Management Plan: Management Strategies. Final Issaquah Creek Valley Ground Water Management Plan Supplement 1 Area Characterization, Submitted March 1998.
- Puls, Robert W., and Michael J. Barcelona, 1995. EPA Ground Water Issue, Low-flow (minimal drawdown) ground-water sampling procedures, EPA/540/S-95/504.

TABLES

Analyte List for Groundwater Sampling

Total Metals	
Arsenic	
Cadmium	
Lead	
Selenium	
Barium	
Calcium	
Chromium	
Copper	
Iron	
Magnesium	
Manganese	
Potassium	
Silver	
Sodium	
Zinc	
Mercury	
Cyanide	
Total Phosphorus	
Conventionals	
Total Dissolved Solids	
Total Alkalinity	
Chloride	
Nitrate + Nitrite	
Sulfate	
Subcontracted Analyses	
Silica	
Fluoride	
Special Study/Requested	
Microbiology	
Total Coliform	
Fecal Coliform	
Special Study/Requested	
Organic-Chlorinated Herbicia	les
2,4,5-T	
2,4,5-TP(Slvex)	
2,4-D	
2,4-DB	
Dalapon	
Dicamba	
Dichloroprop	
Dinoseb	
MCPA	
MCPP	
·	

Organic-BNA 1,2,4-Trichlorobenzene 1,2-Dichlorobenzene 1,2-Diphenylhydrazine 1,3-Dichlorobenzene 1,4-Dichlorobenzene 1,4-Dichlorobenzene 2,4,5-Trichlorophenol 2,4-Dichlorophenol 2,4-Dinitrophenol 2,4-Dinitrophenol 2,4-Dinitrotoluene 2,6-Dinitrotoluene 2-Chloronaphthalene 2-Chlorophenol 2-Methylnaphthalene 2-Methylphenol 3,3'-Dichlorobenzidine 3-Methylphenol 3-Nitroaniline 4,6-Dinitro-O-Cresol 4-Bromophenyl Phenyl Ethel 4-Chloro-3-Methylphenol
1,2,4-Trichlorobenzene 1,2-Dichlorobenzene 1,2-Diphenylhydrazine 1,3-Dichlorobenzene 1,4-Dichlorobenzene 2,4,5-Trichlorophenol 2,4,6-Trichlorophenol 2,4-Dinitrophenol 2,4-Dinitrophenol 2,4-Dinitrotoluene 2,6-Dinitrotoluene 2-Chloronaphthalene 2-Chlorophenol 2-Methylnaphthalene 2-Methylphenol 3,3'-Dichlorobenzidine 3-Methylphenol 3-Nitroaniline 4,6-Dinitro-O-Cresol 4-Bromophenyl Phenyl Bhe 4-Chloro-3-Methylphenol 4-Chloroaniline
1,2-Dichlorobenzene 1,2-Diphenylhydrazine 1,3-Dichlorobenzene 1,4-Dichlorobenzene 2,4,5-Trichlorophenol 2,4,6-Trichlorophenol 2,4-Dinitrophenol 2,4-Dinitrophenol 2,4-Dinitrophenol 2,4-Dinitrotoluene 2,6-Dinitrotoluene 2-Chloronaphthalene 2-Chlorophenol 2-Methylphenol 2-Methylphenol 3,3'-Dichlorobenzidine 3-Methylphenol 3-Nitroaniline 4,6-Dinitro-O-Cresol 4-Bromophenyl Phenyl Ehe 4-Chloro-3-Methylphenol 4-Chloroaniline
1,3-Dichlorobenzene 1,4-Dichlorobenzene 2,4,5-Trichlorophenol 2,4,6-Trichlorophenol 2,4-Dichlorophenol 2,4-Dinitrophenol 2,4-Dinitrophenol 2,4-Dinitrotoluene 2,6-Dinitrotoluene 2-Chlorophenol 2-Methylnaphthalene 2-Methylphenol 2-Nitroaniline 2-Nitrophenol 3,3'-Dichlorobenzidine 3-Methylphenol 3-Nitroaniline 4,6-Dinitro-O-Cresol 4-Bromophenyl Phenyl Ehe 4-Chloro-3-Methylphenol 4-Chloroaniline
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4-Chloro-3-Methylphenol 4-Chloroaniline
4-Chloroaniline
4-Chlorophenyl Phenyl Ethe
4-Methylphenol
4-Nitroaniline
4-Nitrophenol
Acenaphthene
Acenaphthylene
Aniline
Anthracene
Atrazine
Benzo(a)anthracene
Benzo(a)pyrene
Benzo(b)fluoranthene
Benzo(g,h,i)perylene
Benzo(k)fluoranthene

BNA list continued Benzoic Acid Benzyl Alcohol Benzyl Butyl Phthalate Bis(2-Chloroethoxy)Methane Bis(2-Chloroethyl)Ether Bis(2-Chloroisopropyl)Ether Bis(2-Ethylhexyl)Phthalate Caffeine Carbazole Chrysene Coprostanol Diazinon Dibenzo(a,h)anthracene Dibenzofuran Diethyl Phthalate Dimethyl Phthalate Di-N-Butyl Phthalate Di-N-Octyl Phthalate Huoranthene Huorene Hexachlorobenzene Hexachlorobetadiene Hexachloroethane Indeno(1,2,3-Cd)Pyrene Isophorone Naphthalene Nitrosodimethylamine N-Nitrosodi-N-Propylamine N-Nitrosodi-N-Propylamine N-Nitrosodi-N-Propylamine Pentachlorophenol Phenanthrene Phenol Pyrene Pyridine Smazine		
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Smazine	Pyridine	
	Smazine	

1,1,1-Trichloroethane 1,1,2,2-Tetrachloroethane 1,1,2-Trichloroethylene 1,1,2-Trichloroethylene 1,1-Dichloroethylene 1,1-Dichloroethylene 1,2-Dichloroethylene 1,2-Dichloropropane 2-Butanone (MEK) 2-Hexanone 4-Methyl-2-Pentanone (MIBIA) Acetone Acrylonitrile Benzene Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Tirans-1,2-Dichloroethylene	Special Study/Requested
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1,2-Dichloropropane 2-Butanone (MEK) 2-Hexanone 4-Methyl-2-Pentanone (MIB) Acetone Acrylonitrile Benzene Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	1,1-Dichloroethylene
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2-Hexanone 4-Methyl-2-Pentanone (MIBIACetone Acrylonitrile Benzene Bromodichloromethane Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
4-Methyl-2-Pentanone (MIB) Acetone Acrylonitrile Benzene Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chlorodethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Acetone Acrylonitrile Benzene Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Acrylonitrile Benzene Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Styrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Benzene Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Bromodichloromethane Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Bromoform Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Bromomethane Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	Bromodichloromethane
Carbon Disulfide Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Carbon Tetrachloride Chlorobenzene Chlorodibromomethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetrachloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Chlorobenzene Chlorodibromomethane Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Styrene Tetra chloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Chlorodibromomethane Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Styrene Tetra chloroethylene Total Xylenes Trans-1,2-Dichloroethylene	
Chloroethane Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Styrene Tetra chloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Chloroform Chloromethane Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetra chloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Chloromethane Cis-1,3-Dichloropropene Ehylbenzene Methylene Chloride Syrene Tetra chloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Cis-1,3-Dichloropropene Ethylbenzene Methylene Chloride Syrene Tetra chloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Ethylbenzene Methylene Chloride Styrene Tetrachloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Methylene Chloride Styrene Tetra chloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Syrene Tetra chloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Tetrachloroethylene Toluene Total Xylenes Trans-1,2-Dichloroethylene	
Toluene Total Xylenes Trans-1,2-Dichloroethylene	Styrene
Total Xylenes Trans-1,2-Dichloroethylene	
Trans-1,2-Dichloroethylene	
Trans-1,2-Dichloroethylene Trans-1 3-Dichloropropene	Total Xylenes
Trans-1.3-Dichloropropene	Trans-1,2-Dichloroethylene
"and ",o Diomoropropone	Trans-1,3-Dichloropropene
Trichlorofluoromethane	Trichlorofluoromethane
Vinyl Acetate	
Vinyl Chloride	Vinyl Chloride

Sample Volumes, Preservation, and Holding Times

Conventionals				
TDS	KCEL	1 L polyethylene	Cool 4°C +/- 2	7 days
Alkalinity (total)	KCEL	500 mL polyethylene	as above	14 days
Chloride, Sulfate	KCEL	250 mL polyethylene	as above	28 days
Nitrate, Nitrite	KCEL	125 mL polyethylene	as above	48 hours
Subcontracted Analyses				
Fluoride, Silica	AmTest	250 mL polyethylene	Cool 4oC +/- 2	28 days
Analyte:				
Microbiology				
		500 mL HDPE, sterile,		
Total and Fecal Coliform	KCEL	leave one inch of headspace	Cool 4oC +/- 2	30 hours
Organics				
				14 days before
Volatile Organic Acids (VOA)	KCEL	40mL glass (4)	Cool 4oC +/- 2	extraction
				14 days before
Chlorinated Herbicides	KCEL	1L Amber glass (1)	Cool 4oC +/- 2	extraction
				14 days before
Base Neutral Acids (BNA)	KCEL	1L Amber glass (3)	Cool 4oC +/- 2	extraction

KCEL - King County Environmental Laboratory. AmTest - AmTest Laboratories, Inc.

Analyte Analytical Techniques and Method Detection Limits

Parameter	Method	Analytical Technique	MDL (mg/L)	Method	Analytical Technique	MDL (mg/L)	WAC 246- 290-310 Drinking Water Standards (mg/L)	WAC 173- 200-050 Ground Water Quality Criteria (mg/L)
Total Metals			(IIIg/L)			(IIIg/ L)	(mg/ L)	(III g/ L)
Arsenic	EPA 200.8	ICP-MS	0.0005	EPA 200.7	ICP-OES	0.05	0.05	0.05
Cadmium	EPA 200.8	ICP-MS	0.0003	EPA 200.7	ICP-OES	0.003	0.005	0.01
Lead	EPA 200.8	ICP-MS	0.0002	EPA 200.7	ICP-OES	0.003	0.015 ^a	0.05
Selenium	EPA 200.8	ICP-MS	0.0015	EPA 200.7	ICP-OES	0.05	0.05	0.01
Barium	EPA 200.8	ICP-MS	0.0002	EPA 200.7	ICP-OES	0.001	2.0	1.0
Calcium	EPA 200.8	ICP-MS	0.02	EPA 200.7	ICP-OES	0.05	-	-
Chromium	EPA 200.8	ICP-MS	0.0004	EPA 200.7	ICP-OES	0.005	0.1	0.05
Copper	EPA 200.8	ICP-MS	0.0004	EPA 200.7	ICP-OES	0.004	1.3ª	1.0
Iron	N/A	N/A	N/A	EPA 200.7	ICP-OES	0.05	0.3 (S)	0.3
Magnesium	EPA 200.8	ICP-MS	0.02	EPA 200.7	ICP-OES	0.03	-	-
Manganese	EPA 200.8	ICP-MS	0.0002	EPA 200.7	ICP-OES	0.002	0.05 (S)	0.05
Potassium	N/A	N/A	NA/	EPA 200.7	ICP-OES	2	-	_
Silver	EPA 200.8	ICP-MS	0.0002	EPA 200.7	ICP-OES	0.004	0.1 (S)	0.05
Sodium	N/A	N/A	N/A	EPA 200.7	ICP-OES	0.5	20 ^b	_
Zinc	EPA 200.8	ICP-MS	0.0005	EPA 200.7	ICP-OES	0.005	5.0 (S)	5.0
Mercury	EPA 245.2	CVAA	0.0002				0.002	0.002
Cyanide	EPA 335.2 ?	?	?				0.2	0.2
Total Phosphorus	SW 9010 ?	?	?				-	-
Conventionals								
Total Dissolved								
Solids	SM 2540-C	Gravimetric	5				500	500
Total Alkalinity	SM2320-B	Titrim etric	0.2				-	-
Chloride	SM 4110B	IC	0.05				250 (S)	250
Nitrate + Nitrite	SM 4500	Colorimetric	0.02				10 as N	10 as N
Sulfate	SM 4110B	IC	0.1				250	250
Subcontracted A	•							
Silica	SM 4500-SI-E	Colorimetric	0.04				-	-
		Alpchem						
Fluoride	EPA 340.2	Autoanalyzer	0.1				2.0 (S)	4
Special Study/Ro	equested							
Microbiology								
		Membrane	1 CFU					
Total Coliform	SM9222-B	Filtration	per100mL					
		Membrane	1 CFU					
Fecal Coliform	SM9222-D	Filtration	per100mL					
Organic								
770.4	77D 4 - 22 4		varies per				varies per	varies per
VOA	EPA 624	GC/MS	analyte				analyte	analyte
Chlorinated	GM 04.54	GCMS	varies per				varies per	varies per
Herbicides	SM 8151	MODIFIED	analyte varies per				analyte varies per	analyte varies per
BNA	SM 3520C/8270C	GC/MS	analyte				analyte	analyte

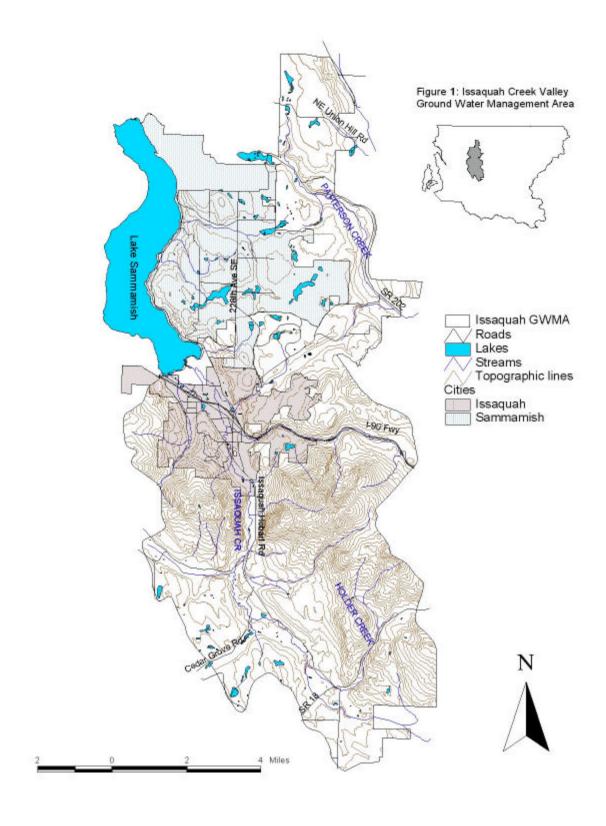
^a Action Level

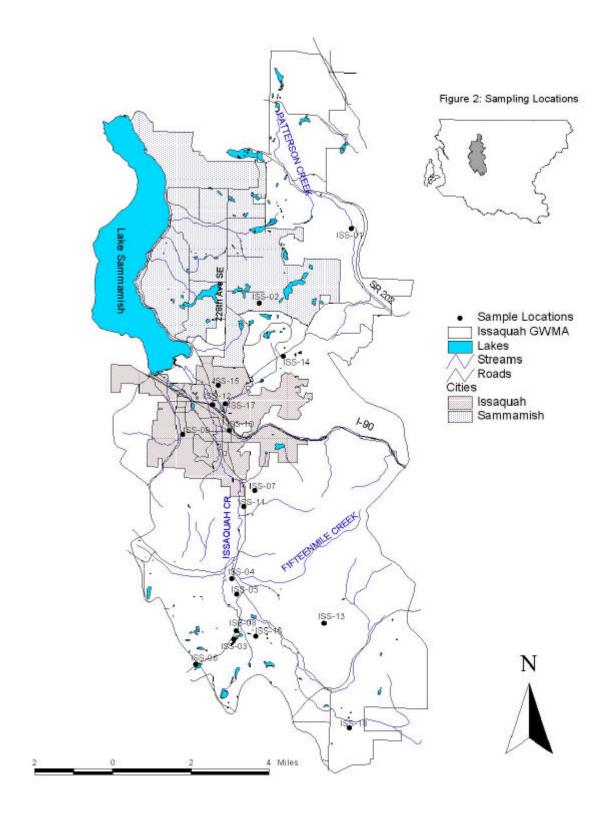
⁽S) Secondary MCL

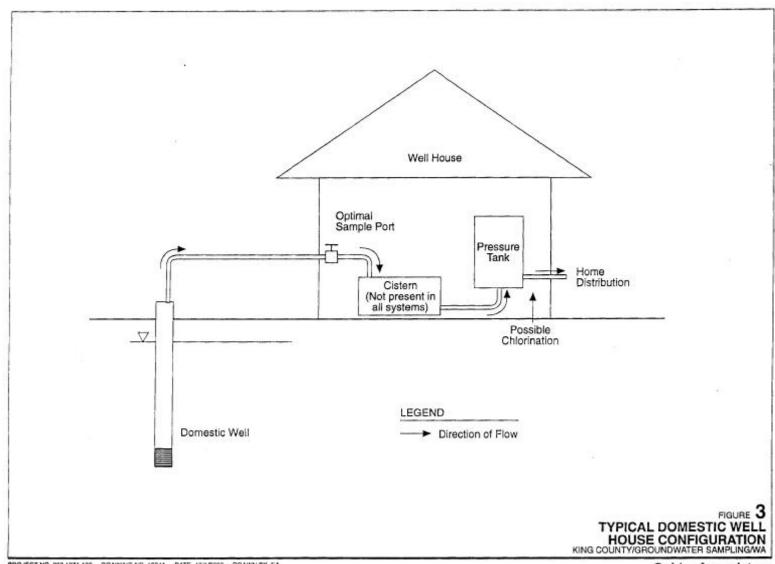
^b Recommended Level

^{&#}x27;-' parameter not regulated.

FIGURES







PROJECT NO. 003 1071,100 DRAWING NO. 10941 DATE 12/1/2000 DRAWN BY EA

Golder Associates

ATTACHMENT A WELL SAMPLING FIELD FORM

Project:	Well Identification:	
Site:	Sampling Method:	
Job Number:	Purge Method:	

Well		System Configuration
Specifications		
Hole Diameter d _h =		
Well Casing		
Inside Diam d _w ID =		
Water Level H =		
Depth of Well TD =		·
		
Purge Volume	¥	
Calculations		
Well Volume (ft ³)= $V_c = \pi (d_w ID/2)^2 (TD-H) =$		
$Vt = Vc * 7.482 \text{ gal/ft}^3 \text{ or } Vt = (TD-H) * wcgf$		
Minimum Purge Volume (gallons)=		
Vp = Vt x 3 =		
Holding Tank Volume =		

			Water Cumulative Wate Water Characteristics								
1	Tir	ne	Purged	Pu	irged .		Conductivity	Turbidity	DO	Temperature	
Date	Begin	Finish	(gal)	(gal)	Well Vol	рН	(µ mhos/cm)	(NTU)	(mgdo/L)	(° Celsius)	Comments
										-	

pampiing vate:	
Sample Number:	
Analyses to be Performed:	
No. and Type of Sample Containers/Prese	rve:

Chain of Custody Seal Number: Analytical Laboratory: Date Shipped:

Carrier:

well casing diameter to gallons per foot of head: wcgf = 8 " = 2.637 6" = 1.468 4" = 0.653 3" = 0.36

3'' = 0.367

2'' = 0.163

TECHNICAL PROCEDURES

TP 1.2-20	Collection of Groundwater Quality Samples
TP-1.2-23	Chain of Custody Procedure
OP-11.1	Calibration and Maintenance of Measuring and Test Equipment

Note: Printed copies of the Technical Procedures section are available by request.

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