

Microwave Spectra of Molecules of Astrophysical Interest

X. Isocyanic Acid

G. Winnewisser and W. H. Hocking

Max-Planck-Institut für Radioastronomie, D-53 Bonn, Germany

and

Physikalisch-Chemisches Institut, Justus Liebig Universität, D-63 Giessen, Germany

and

M. C. L. Gerry

Dept. of Chemistry, University of British Columbia, Vancouver, B.C., Canada

The available data on the microwave spectrum of isocyanic acid are critically reviewed for information applicable to radio astronomy. Molecular data such as rotational constants, centrifugal distortion parameters, dipole moments, hyperfine coupling constants, and structural parameters are tabulated. Detailed centrifugal distortion calculations have been carried out for all isotopic forms of this molecule, including DNCO. Transitions have been predicted for the parent molecule for the frequency range 160 MHz–300 GHz. All predicted transitions include error limits. The quoted uncertainties represent one standard deviation. A 95 percent confidence limit is obtained by using approximately twice the calculated standard deviation. Estimated error limits for the measured transitions are discussed. References are given for all data included.

Key words: Isocyanic acid; interstellar molecules; microwave spectra; molecular parameters; rotational transitions; radio astronomy.

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1. Introduction

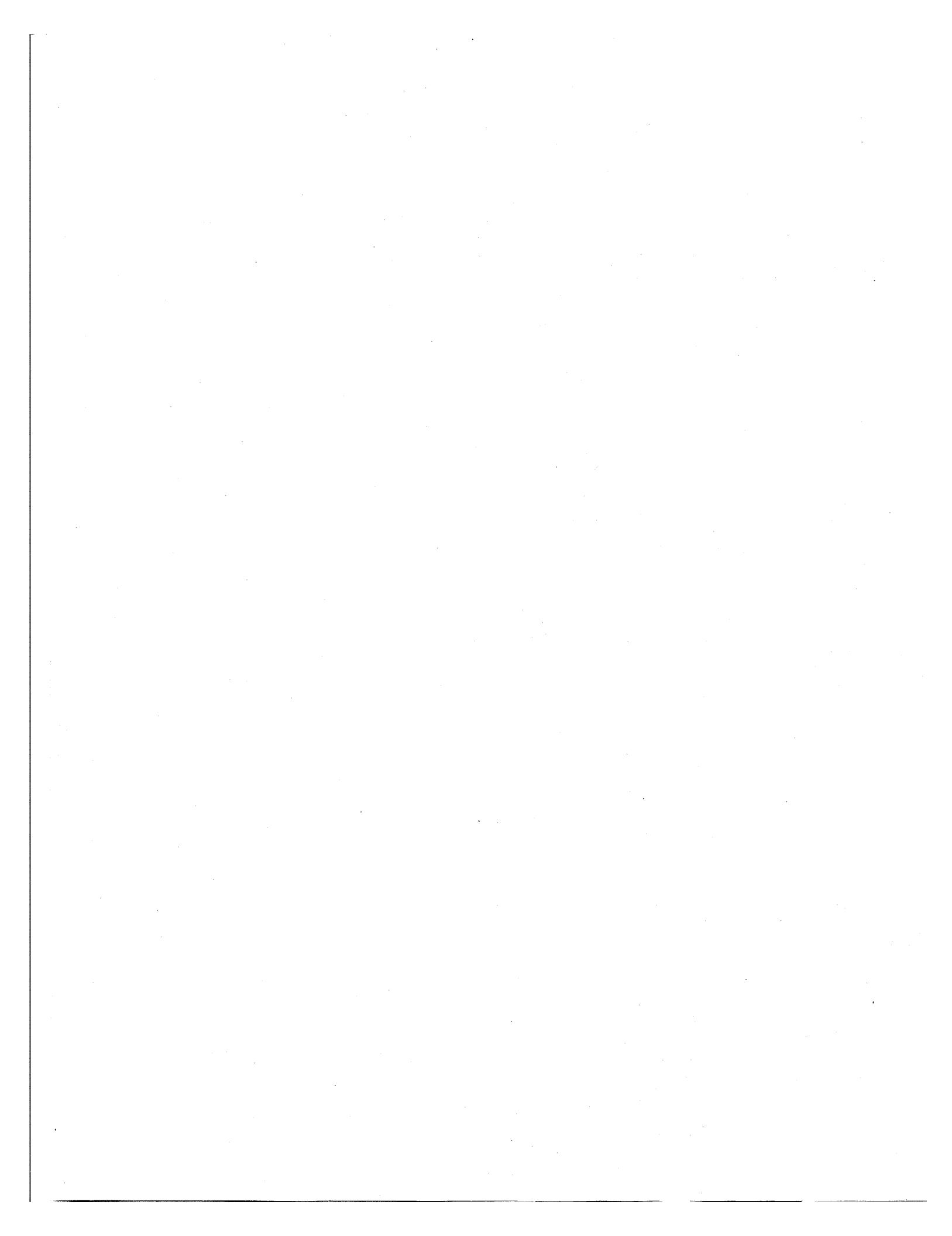
The present tables were prepared in response to the needs of the rapidly progressing field of molecular radio astronomy and are intended to update and revise the existing tabulated literature on molecules already identified in interstellar observations [1].¹ The spectral information reported includes predicted and observed transitions between 160 MHz and 300 GHz for five iso-

¹ Figures in brackets indicate literature references in section 1.4.

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topic species of isocyanic acid: $H^{14}N^{12}C^{16}O$, $D^{14}N^{12}C^{16}O$, $H^{15}N^{12}C^{16}O$, $H^{14}N^{13}C^{16}O$ and $H^{14}N^{12}C^{18}O$. The frequency predictions for $H^{14}N^{12}C^{16}O$ and $D^{14}N^{12}C^{16}O$ are extensive and include transitions whose lower energy states are below 750 cm^{-1} and 250 cm^{-1} , respectively.² For the remaining isotopic species transitions of particular astrophysical interest are predicted. The deuterated species has been included in these tables, since it is conceivable that interstellar deutero-isocyanic acid may be observable.

² In keeping with the commonly accepted convention in molecular spectroscopy, the fundamental frequencies and vibrational energies are frequently expressed in their wavenumber (cm^{-1}) equivalents.



It is felt that these limits are generous enough to allow for the presentation of all transitions which might be observed by existing telescopes, or by those likely to be developed in the next several years.

1.1. Molecular Parameter Tables

The rotational constants and centrifugal distortion constants presented in tables 1, 2, and 3 were obtained from a least-squares analysis of the observed spectral lines with a computer program which includes centrifugal distortion terms, in addition to the basic rigid asymmetric rotor energy matrix. The very high rotational energies and large centrifugal distortion made it necessary to include high order terms (up to the twelfth power) in the angular momentum in the Hamiltonian. However, the small number of observable branches of isocyanic acid in the microwave and millimeter-wave regions meant that some low order distortion constants, in particular D_K and H_K , were indeterminate. The A_0 rotational constants given in tables 1-3 are thus effective constants which contain substantial contributions from the indeterminate distortion constants. All of the effective A_0 values were corrected for the largest distortion term ($D_K - H_K$) using the far infrared data [2] before calculation of the moments of inertia [3]. Details of the centrifugal distortion calculation and the statistical analysis used in this review have been discussed by Hocking, Gerry, and Winnewisser [3] as well as by Helminger, Cook, and de Lucia [4]. This formulation is similar to those discussed by Kirchoff [5] and by Steenbeckeliers [6]. As pointed out in earlier parts of this series, it is necessary to retain more significant figures in the spectral constants than indicated by the statistical error limits, if the constants are to reproduce the observed spectra to within experimental error.

1.2. Microwave Spectral Tables

Tables 4 through 9 contain the results of the statistical analysis of the rotational spectrum of $\text{H}^{14}\text{N}^{12}\text{C}^{16}\text{O}$, $\text{D}^{14}\text{N}^{12}\text{C}^{16}\text{O}$, $\text{H}^{14}\text{N}^{13}\text{C}^{16}\text{O}$, $\text{H}^{15}\text{N}^{12}\text{C}^{16}\text{O}$, and $\text{H}^{14}\text{N}^{12}\text{C}^{18}\text{O}$, respectively. For each spectral line in tables 4 and 5 the first column contains the upper state and lower state quantum number in the form JK_aK_c for a rigid asymmetric rotor plus the total angular momentum quantum number $F=J+I_1, J+I_1-1, \dots, J-I_1$, where I_1 is the nuclear spin angular momentum quantum number for the ^{14}N nucleus, with $I_1=1$. The quantum numbers are followed by the observed unsplit line frequency. The estimated experimental uncertainty is quoted in the footnote at the end of each table. The third column contains the calculated frequency and estimated uncertainty in MHz. The calculated uncertainties represent one standard deviation obtained from the least squares analysis. A 95 percent confidence limit on the predictions is obtained by using approximately twice (this varies slightly with the number of data included in the calculation) standard deviation of the calculated un-

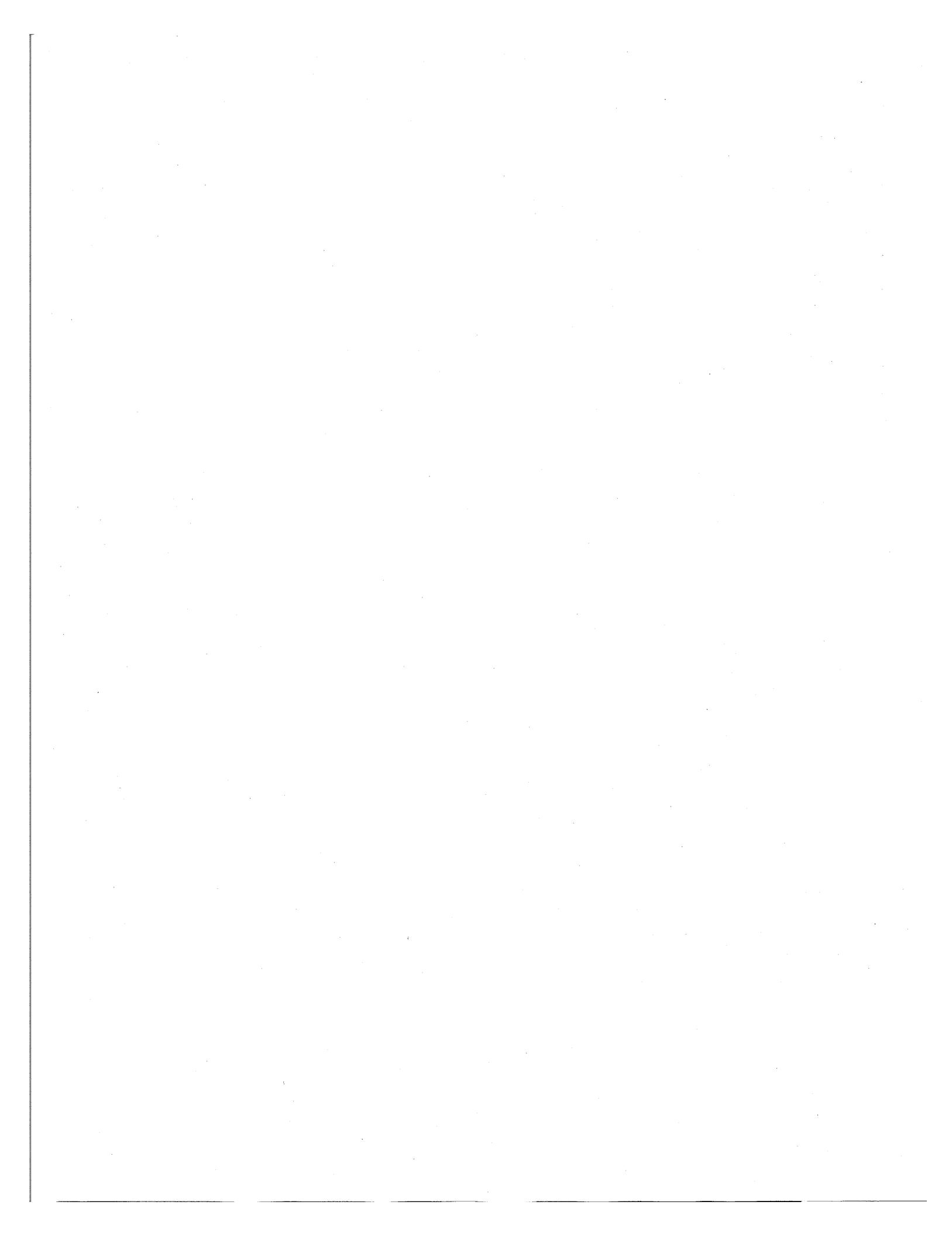
certainties. Underneath each rotational transition the F quantum numbers are given for all transitions whose total quadrupole splitting is larger than 50 kHz. Opposite the F quantum numbers the calculated splittings due to the nuclear electric quadrupole interaction are listed along with their estimated uncertainties in MHz. The calculated uncertainties represent one standard deviation. The actual transition frequencies can be obtained by adding the hyperfine splittings to the unsplit frequency, and the estimated error of each is then the root mean square of the individual estimated uncertainties.

Tables 7 and 8 contain essentially only the measured spectra of ^{13}C , ^{15}N , and ^{18}O labelled isocyanic acid together with the results of the statistical analysis and a selected number of predicted transitions. Further information is available from the authors on request. The line strengths for the unsplit rotational transitions are given in column 4. These line strengths, denoted by $^xS(J'_{K_a, K_c}; J''_{K_a, K_c})$ are defined in this review as:

$$^xS(J'_{K_a, K_c}; J''_{K_a, K_c}) = \frac{(2J'+1)|\mu_{J' \leftarrow J''}|^2}{\mu_x^2},$$

where the superscript x refers to one of the principal axes of the molecule ($x=a$, b , or c); $|\mu_{J' \leftarrow J''}|$ is the dipole moment matrix element connecting the upper J'_{K_a, K_c} and lower, J''_{K_a, K_c} , rotational levels involved in the transition, and μ_x is the magnitude of the component of μ along the x -axis. Thus, the line strength as defined is independent of the absolute magnitude of the dipole moment.

The total rotational energy of each rotational level was calculated using all distortion constants which were used in the analysis and quoted in tables 1, 2, and 3. These are given in columns 5 and 6 in cm^{-1} . The estimated accuracy of the calculated values for the energy levels are not better than: for $K_a=0$ about two decimal places, for $K_a=2$ about one decimal place, for $K_a=3$ about $\pm 1 \text{ cm}^{-1}$, $K_a=4$ about $\pm 3 \text{ cm}^{-1}$, and for $K_a=5$ about $\pm 7 \text{ cm}^{-1}$. No levels with $K \geq 6$ have been included, due to their general uncertainty, and they are also not likely to be observed astrophysically. Although the b -type transitions have been observed in the laboratory in the microwave and millimeter-wave regions, they are likely not to be detected astrophysically, due to their very high J quantum numbers. For completeness, they have, however, been included in the tables. If any further information is needed on these transitions, the authors of this review possess the necessary programs. The laboratory measurements are taken from ref. [75], unless quoted differently in the last column. As a convenience to the user, the calculated unsplit transition frequencies from table 4 for $\text{H}^{14}\text{N}^{12}\text{C}^{16}\text{O}$ have been listed according to increasing frequency in table 10. We have included in this table some low J high frequency b -type transitions. They may



become of interest for the developing field of far infrared astronomy.

1.3. List of Symbols and Conversion Factors

a. Symbols

A, B, C	Rotational constants (MHz). $A \geq B \geq C$.
D, δ_J, R_6	Quartic centrifugal distortion constants.
H	Sextic centrifugal distortion constants.
L	Octic centrifugal distortion constants.
S	Dectic centrifugal distortion constants.
T	12th order centrifugal distortion constants.
a, b, c	Principal axes corresponding to A, B, C .
$\mu, \mu_x(x=a, b, c)$	Dipole moment and components of the dipole moment along the principal axes.
eqQ_{aa}, \dots	Nuclear electric quadrupole coupling constant along indicated principal axis (MHz).
X_{ij}	Elements of the quadrupole coupling tensor (MHz).
$r(X-Y)$	Distance between centers of mass of atom X and Y (\AA).
$\chi(X, Y, Z)$	Angle formed by atoms X, Y, and Z (degrees).
$I_{a, b, c}$	Moments of inertia of whole molecule with respect to the indicated principal axis.
F	Total angular momentum quantum number.
J	Total rotational angular momentum quantum number.
K_a	Projection of J on the symmetry axis in the limiting prolate symmetric top.
K_c	Projection of J on the symmetry axis in the limiting oblate symmetric top.
(. . .)	Parentheses in the numerical listings contain measured or estimated uncertainties. These should be interpreted as: $1.532(30) = 1.532(0.030) = 1.532 \pm 0.030$.

b. Conversion Factors

The following conversion factors have been used:

$$A, B, C \text{ (MHz)} = \frac{5.05375 \times 10^5}{I_{a, b, c}(\text{amu} \text{\AA}^2)}$$

$$1 \text{ cm}^{-1} = 29,979.2456 \text{ MHz.}$$

c. Acknowledgements

In an attempt to increase the usefulness of this series, the format of the earlier contributions of this series by various authors, notably by William H. Kirchhoff, Donald R. Johnson, and Frank J. Lovas of the National Bureau of Standards, has been followed whenever

possible. We acknowledge gratefully the support of the Deutsche Forschungsgemeinschaft, the National Research Council of Canada, and the North Atlantic Treaty Organization.

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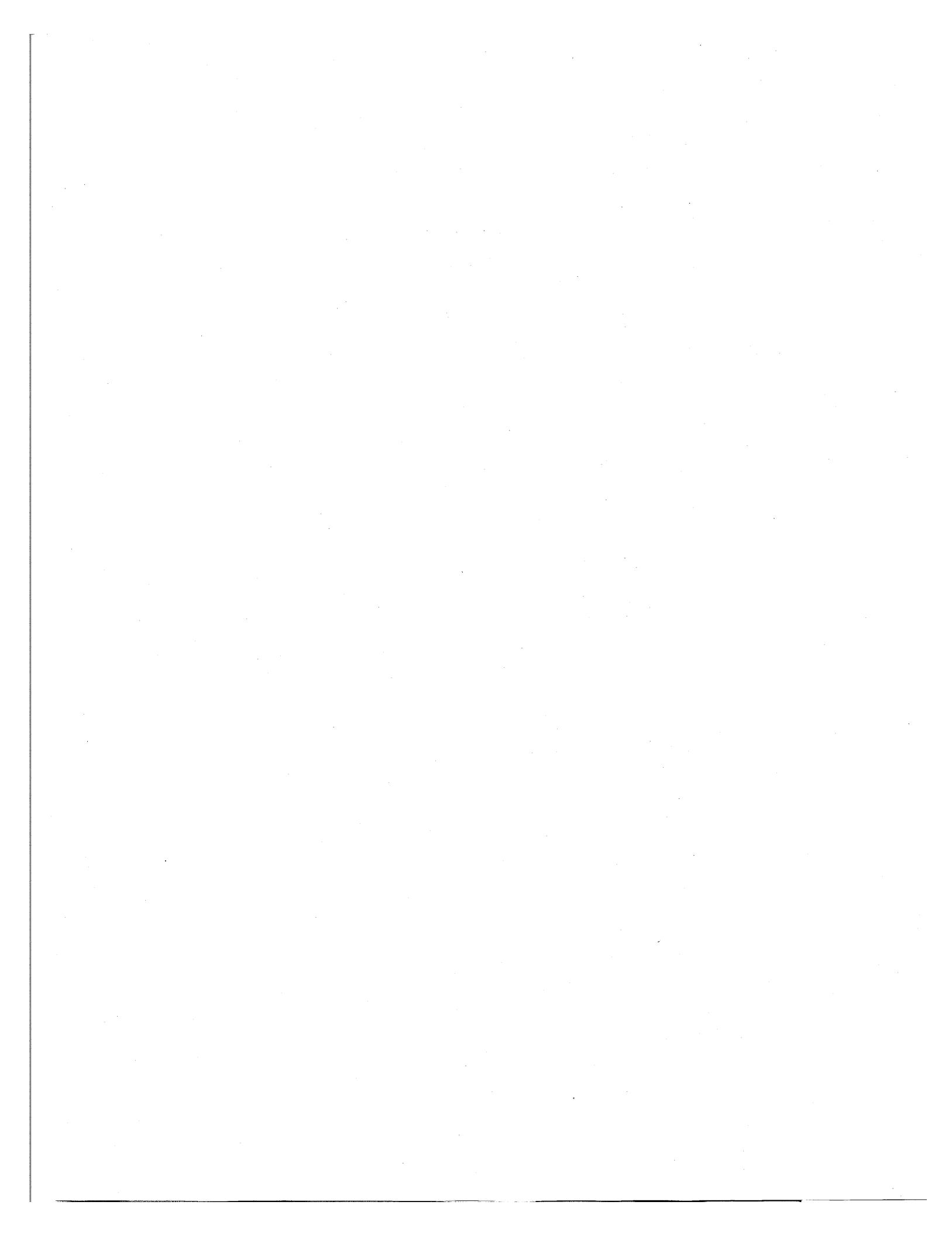
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2. Isocyanic Acid Spectral Tables

TABLE 1. Molecular Parameters for Isocyanic Acid

$\text{H}^{14}\text{N}^{12}\text{C}^{16}\text{O}$		
Rotational Constants ^a (MHz)		Ref. [75]
A_o	912.712.288	(136)
B_o	11.071.01027	(62)
C_o	10.910.57748	(64)
Distortion Constants ^a (MHz)		Ref. [75]
D_J	-3.516609(3000) $\times 10^{-3}$	
D_{JK}	0.93376	(64)
δ_J	7.30161 (890) $\times 10^{-5}$	
R_G	-2.264 (230) $\times 10^{-5}$	
H_J		
H_{JK}	1.569 (150) $\times 10^{-6}$	
H_{KJ}	3.32 (32) $\times 10^{-2}$	
L_{JK}		
L_{KJ}	-3.315 (49) $\times 10^{-3}$	
S_{KJ}	1.529 (29) $\times 10^{-4}$	
T_{KJ}	-2.650 (54) $\times 10^{-6}$	
$D_K - H_K$	4990 (400)	Ref. [68B]
Dipole Moment (Debye)		Ref. [75]
μ_a	1.575 \pm 0.005	
μ_b	1.35 \pm 0.10	

TABLE 1. Molecular Parameters for Isocyanic Acid -Continued

$\text{H}^{14}\text{N}^{12}\text{C}^{16}\text{O}$		
Nitrogen Quadrupole Coupling Constants (MHz)		
χ_{aa}	2.0527(10)	Ref. [71]
χ_{bb}	-0.473(7)	Ref. [75]
χ_{cc}	-1.583(7)	Ref. [75]
Moments of Inertia (amu \AA^2)		
I_a^o	= 0.55070	
I_b^o	= 45.648499	
I_c^o	= 46.319730	
Δ	= $I_c^o - I_b^o - I_a^o = 0.12053$	
Structure		Ref. [75]
$r(\text{N-H})$	= 0.986 \AA ;	
$r(\text{N-C})$	= 1.209 \AA ;	
$r(\text{C-O})$	= 1.166 \AA ;	
$\angle(\text{HNC})$	= 128.0°	
$\angle(\text{NCO})$	= 180°	

^a The numbers in parenthesis are standard errors in units of the last significant figures. See section 1.3A for interpretation of these standard deviations.

TABLE 2. Molecular Parameters for Isocyanic Acid

	$D^{14}N^{12}C^{16}O$	$H^{15}N^{12}C^{16}O$
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Rotational Constants^a (MHz)

A_o	510 971.6806 (400)	902 803.34 (82.50)
B_o	10 313.71371 (63)	10 737.8611(110)
C_o	10 079.67647 (65)	10 585.4961(110)

Ref. [75]

Distortion Constants^a (MHz)

D_J	3.29278(240) $\times 10^{-3}$	4.149 (260) $\times 10^{-3}$
D_{JK}	-0.24510 (25)	1.0559 (93)
δ_J	2.04319(360) $\times 10^{-4}$	6.2971(200) $\times 10^{-5}$
R_6	-3.66 (23) $\times 10^{-5}$	
H_J	-2.398 (170) $\times 10^{-8}$	2.74 (59) $\times 10^{-6}$
H_{JK}	1.846 (230) $\times 10^{-6}$	8.73 (1.90) $\times 10^{-4}$
H_{KJ}	-7.3431 (430) $\times 10^{-3}$	3.74 (19) $\times 10^{-2}$
L_{JK}		-8.99 (2.00) $\times 10^{-5}$
L_{KJ}	8.838 (290) $\times 10^{-5}$	-1.884 (130) $\times 10^{-3}$
S_{KJ}	-9.60 (77) $\times 10^{-7}$	
T_{KJ}	4.70 (70) $\times 10^{-9}$	
$D_K - H_K$	2700 (200)	4990 (400)

Ref. [75]

Nitrogen Quadrupole Coupling Constants^a (MHz)

χ_{aa}	2.1230(10)	Ref. [71]
χ_{bb}	-0.540 (12)	Ref. [75]
χ_{cc}	-1.570 (12)	Ref. [75]

Ref. [68B]

Deuterium Quadrupole Coupling Constant^a (kHz)

χ_{aa}	53.6 (2)	Ref. [71]
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TABLE 3. Molecular Parameters for Isocyanic Acid

	$H^{14}N^{13}C^{16}O$	$H^{14}N^{12}C^{18}O$
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Rotational Constants^a (MHz)

Ref. [75]

A_o	910 498.79 (1.21)	912 623.326 (400)
B_o	11 071.48157(260)	10 470.8950 (12)
C_o	10 910.73242(260)	10 327.2418 (12)

Distortion Constants^a (MHz)

Ref. [75]

D_J	3.5387 (93) $\times 10^{-3}$	3.1588(55) $\times 10^{-3}$
D_{JK}	0.9293 (31)	0.8196(11)
δ_J	7.320 (30) $\times 10^{-5}$	6.1215(93) $\times 10^{-5}$
R_6	-3.315(1000) $\times 10^{-5}$	
H_J		-7.638(1.100) $\times 10^{-6}$
H_{JK}		2.653 (51) $\times 10^{-2}$
H_{KJ}	3.391 (130) $\times 10^{-2}$	7.991(960) $\times 10^{-6}$
L_{JK}		-2.252 (67) $\times 10^{-3}$
L_{KJ}	-3.409 (190) $\times 10^{-3}$	6.300(250) $\times 10^{-5}$
S_{KJ}	1.55 (11) $\times 10^{-4}$	
T_{KJ}	-2.64 (20) $\times 10^{-6}$	
$D_K - H_K$	4990 (400)	4990(400)

Nitrogen Quadrupole Coupling Constants (MHz)

Ref. [75]

χ_{aa}	2.067 (20)	2.060 (13)
χ_{bb}	-0.402 (11)	-0.472 (8)
χ_{cc}	-1.585 (11)	-1.588 (8)

^a The numbers in parenthesis are standard errors in units of the last significant figures.

^a The numbers in parenthesis are standard errors in units of the last significant figures.

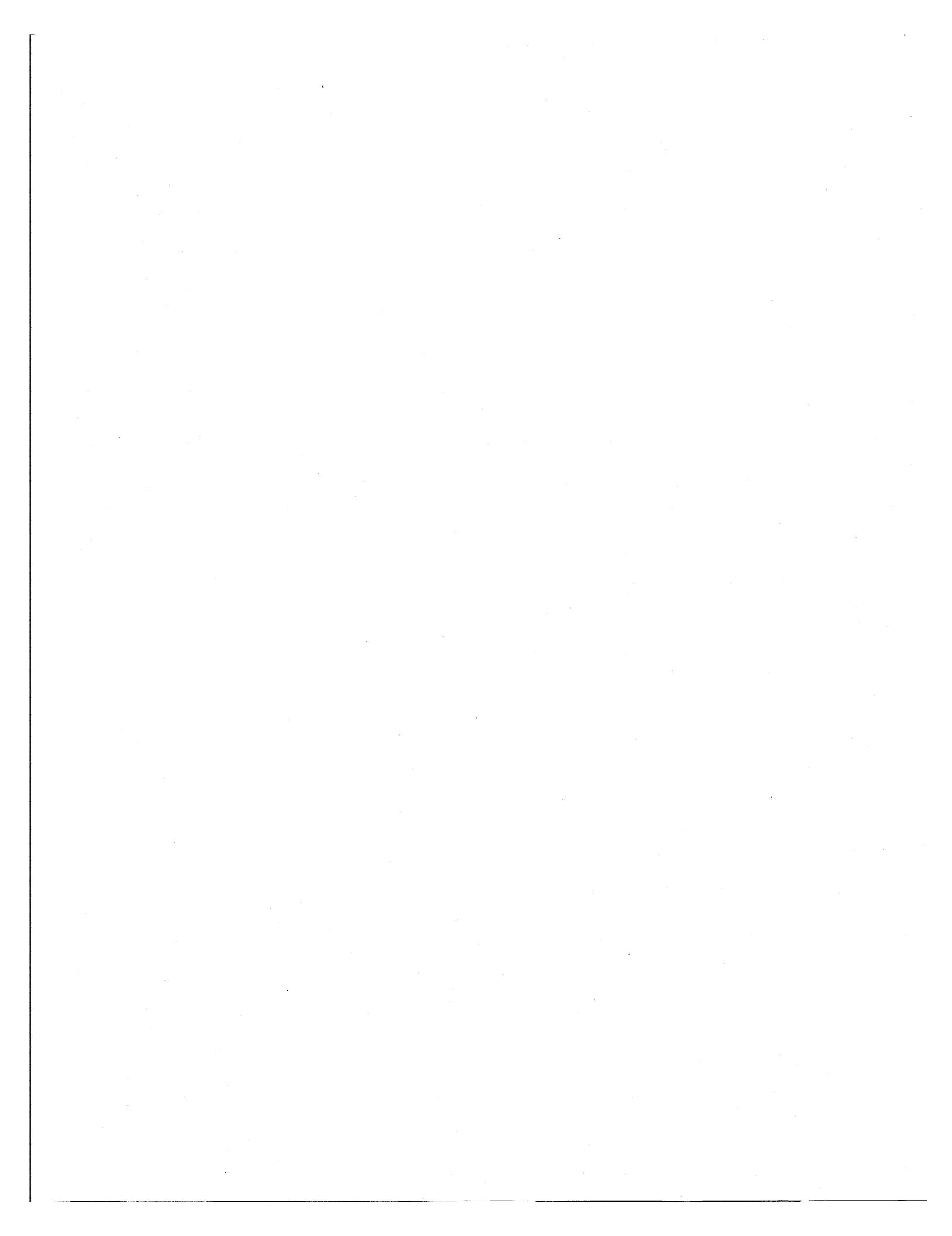


TABLE 4. The Microwave Spectrum of $H^{14}N^{12}C^{16}O$ (MHz)

Transition J, K_a, K_c	(1)			(2)			(3)			(4)			(5)			(6)	
	J''	K_a''	K_c''	Observed (Est. Uncert. σ_c)			Calculated + Quadrupole shifts (standard dev.) ^d			Line strength + Rel. Intensity of Quadrupole comp.			Energy levels ^a in cm^{-1}			Ref. ^b	
1 0 $F'' = 2$	1	0	0	21981.47055			21981.574(1)	-0.103(1)		1.000			0.733	0.0		71	
	1	1	1	21982.08535				0.514(3)		0.556						71	
	0	1	1	21980.54533				-1.028(6)		0.333						71	
2 0 $F'' = 3$	2	0	1	43963.000			43963.042(2)	-0.044(0)		2.000						71	
	2	2	2	43963.626				0.617(3)		0.467						71	
	2	1	1	43963.000				0.000(0)		0.083						71	
1 1 $F'' = 1$	1	1	1	43962.007				-1.028(6)		0.250						71	
	1	0	0	43963.626				0.514(3)		0.083						71	
	1	1	1	44119.879				-0.910(4)		0.111						71	
2 1 $F'' = 2$	2	1	1	44119.757			44119.879(2)	-0.113(1)		1.500						71	
	2	2	2	44120.159				-0.039(2)		0.467						71	
	2	1	1	44120.390				0.514(3)		0.083						71	
1 0 $F'' = 3$	1	1	1	44119.903				-0.278(2)		0.250						71	
	1	0	0	44118.967				-0.632(4)		0.083						71	
	1	1	0	43799.423			43799.019(2)	-0.137(1)		0.900						71	
2 1 $F'' = 3$	2	1	1	43799.423				0.372(2)		0.467						71	
	2	2	1	43799.533				0.514(3)		0.083						71	
	2	1	1	43798.360				-0.278(2)		0.250						71	
1 0 $F'' = 4$	1	0	0	65944.301			65944.298(4)	-0.632(4)		0.900						71	
	1	3	2	65944.301				-0.025(0)		0.429						71	
	1	2	2	65698.439				0.000(0)		0.296						71	
3 1 $F'' = 4$	3	1	2	65698.439			65698.308(3)	-0.062(0)		0.429						71	
	3	2	2	65698.439				-0.024(0)		0.200						71	
	3	3	2	65698.439				0.129(1)		0.296						71	
3 2 $F'' = 4$	3	2	2	66179.531			66179.590(3)	-0.049(0)		0.429						71	
	3	3	2	66179.729				0.129(1)		0.296						71	
	2	1	1	66179.531				-0.079(0)		0.200						71	
3 2 $F'' = 4$	3	2	2	65924.118			65924.118(3)	-0.147(1)		1.667						71	
	3	4	2	65294.015				-0.147(1)		0.429						71	

MICROWAVE SPECTRUM OF ISOCYANIC ACID

TABLE 4. Continued

(1)	(2)	(3)	(4)	(5)
3	2	2	0.296	0.296
2	1	1	0.200	0.200
3	2	2	65924.679	65924.679
$F' =$	4	3	65923.626	65923.626
4	0	3	65924.015	65924.015
$F' =$	5	4	65924.679	65924.679
2	1	1	65923.626	65923.626
4	4	3	87925.252	87925.238(4)
$F' =$	5	4	-0.147(1)	1.667
3	3	3	0.514(3)	-0.514(3)
$F'' =$	4	3	0.514(3)	0.514(3)
4	1	3	87597.342	87597.333(3)
$F' =$	5	4	-0.035(0)	4.000
4	1	3	3.750	0.407
$F' =$	4	3	0.051(0)	0.407
3	3	2	0.044(0)	0.313
4	1	3	88239.036	88239.027(3)
$F' =$	5	4	-0.026(0)	4.000
4	1	3	3.750	0.407
$F' =$	4	3	0.051(0)	0.313
3	3	2	0.008(0)	0.238
4	2	3	87898.341	87898.416(4)
$F' =$	5	4	-0.075(0)	3.000
4	2	3	3.000	0.407
$F' =$	4	3	0.206(1)	0.313
3	3	2	0.147(1)	0.238
4	2	3	87898.341	87898.620(4)
$F' =$	5	4	-0.075(0)	3.000
4	2	3	3.000	0.407
$F' =$	4	3	0.206(1)	0.313
3	3	2	0.147(1)	0.238
4	2	3	87898.565	87898.620(4)
$F' =$	5	4	-0.075(0)	3.000
4	2	3	3.000	0.407
$F' =$	4	3	0.206(1)	0.313
3	3	2	0.147(1)	0.238
4	3	2	87898.565	87898.620(4)
$F' =$	4	3	-0.075(0)	3.000
4	3	2	3.000	0.407
$F' =$	5	4	0.206(1)	0.313
3	3	2	0.147(1)	0.238
4	3	2	87867.132	87867.280(5)
$F' =$	4	3	-0.149(1)	1.750
4	3	2	0.463(3)	1.750
$F' =$	5	4	0.463(3)	1.750
3	3	2	-0.386(2)	1.750
4	4	0	87867.732	87867.280(5)
$F' =$	5	1	-0.044(0)	2.6865
4	4	0	87866.892	87866.892
$F' =$	5	1	0.103(1)	2.6865
5	0	5	109905.758	109905.753(5)
$F' =$	5	1	109496.008	109496.007(4)
5	1	4	110298.080	110298.098(4)
$F' =$	5	1	109872.337	109872.366(5)
5	2	4	109872.765	109872.773(5)
$F' =$	5	2	-0.044(0)	1.25.82
5	2	3	0.103(1)	1.25.82
$F' =$	5	3	-0.059(0)	1.25.82

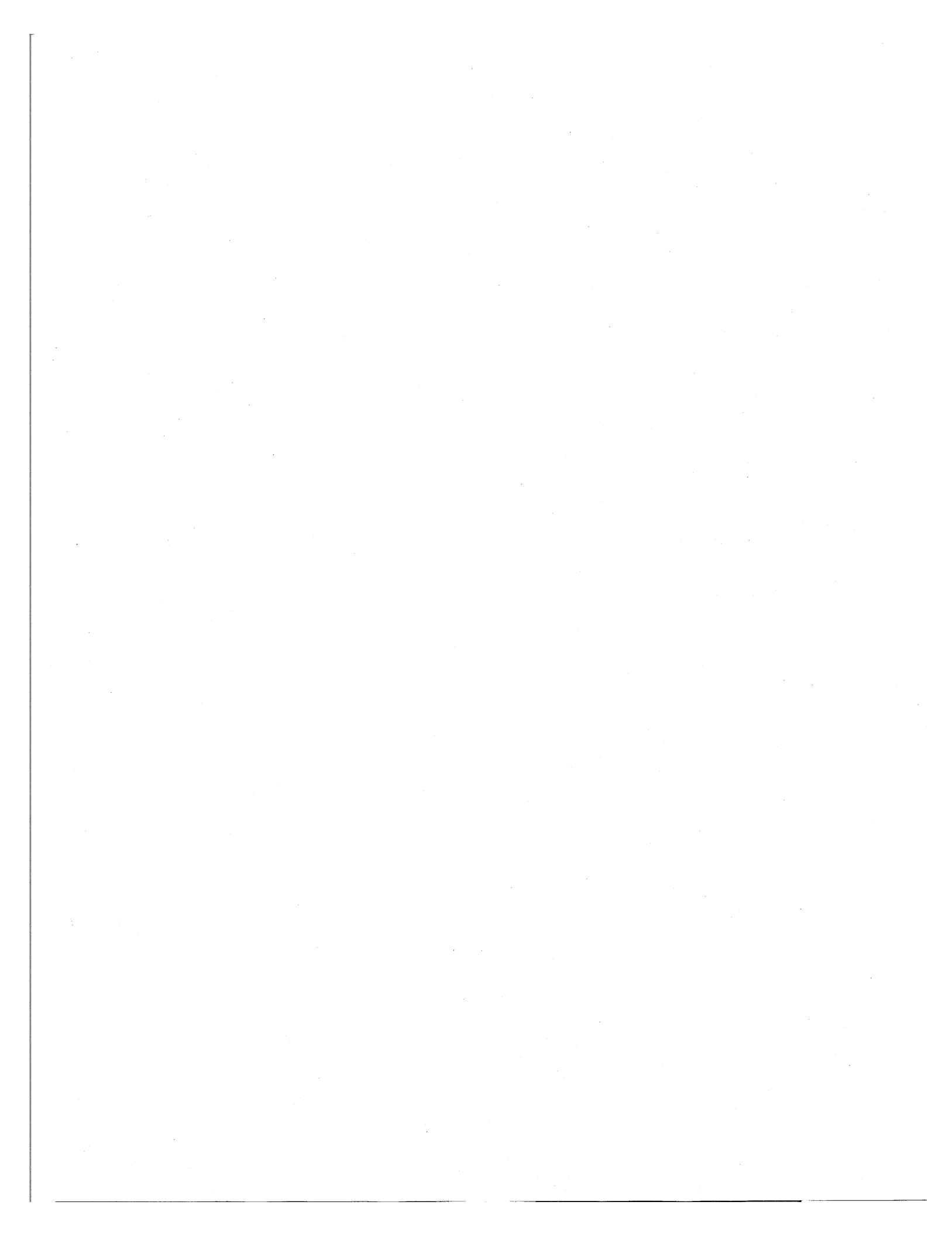


TABLE 4. Continued

(1)	(2)	(3)	(4)	(5)	(6)
5 3 3 4 4 4	109833.489(6) 109833.489(6) -0.085(O) +0.231(1) -0.163(O)	3.200 3.200 0.394 0.320 0.259	272.31 272.31	268.65 268.65	
5 3 2 4 4 4	109833.391 109833.737 109833.391				
F' = F" =					
5 4 2 4 4 4	109778.542 109779.125	1.800 1.800	469.19 469.19	465.52 465.52	
F' = F" =					
5 4 1 4 4 4	109778.700(7) 109778.700(7) -0.143(1) -0.308(2) 0.411(2)	0.394 0.259 0.320	10.998 15.398 15.398	10.998 45.42 45.53	
F' = F" =					
5 4 0 6 5 5	131885.740(6) 131394.241(5) 132356.711(5) 131845.880(5) 131846.590(6)	6.000 5.833 5.833 5.333 5.333	41.04 41.12 129.49 129.49		
F' = F" =					
6 1 6 5 5 5	-0.028(O) 0.059(O) -0.028(O)	0.385 0.324 0.273	133.89 133.89 133.89		
F' = F" =					
6 1 5 5 5 5	131799.292(7) 131799.292(7) -0.054(O) 0.132(1) -0.063(O)	4.500 4.500 0.385 0.324 0.273	276.71 276.71 473.58 473.58	272.31 272.31	
F' = F" =					
6 2 5 5 5 5	131733.534(7) 131733.534(7) -0.069(O) 0.235(1) -0.159(1)	3.333 3.333 0.385 0.324 0.273	469.19 469.19 473.58 473.58		
F' = F" =					
6 2 4 5 5 5	131640.747(9) 131640.747(9) -0.135(1) 0.367(2) -0.257(1)	1.833 1.833 0.385 0.324 0.273	741.47 741.47 737.07 737.07		
F' = F" =					
6 5 2 5 5 5	153865.08 153291.946(6) 154414.776(6) 153818.869(6) -0.019(O) 0.037(O) -0.015(O)	7.000 6.857 6.857 6.429 0.378 0.327 0.282	20.530 50.53 50.68 45.42 45.53		
F' = F" =					
7 0 7 6 6 6	153819.98 -0.019(O) 0.037(O) -0.015(O)	6.429 0.378 0.327 0.282	133.89		
F' = F" =					
7 1 7 6 6 6					
F' = F" =					
7 2 6 6 6 6					
F' = F" =					
7 2 5 6 6 6					
F' = F" =					
7 2 5 6 6 6					
F' = F" =					

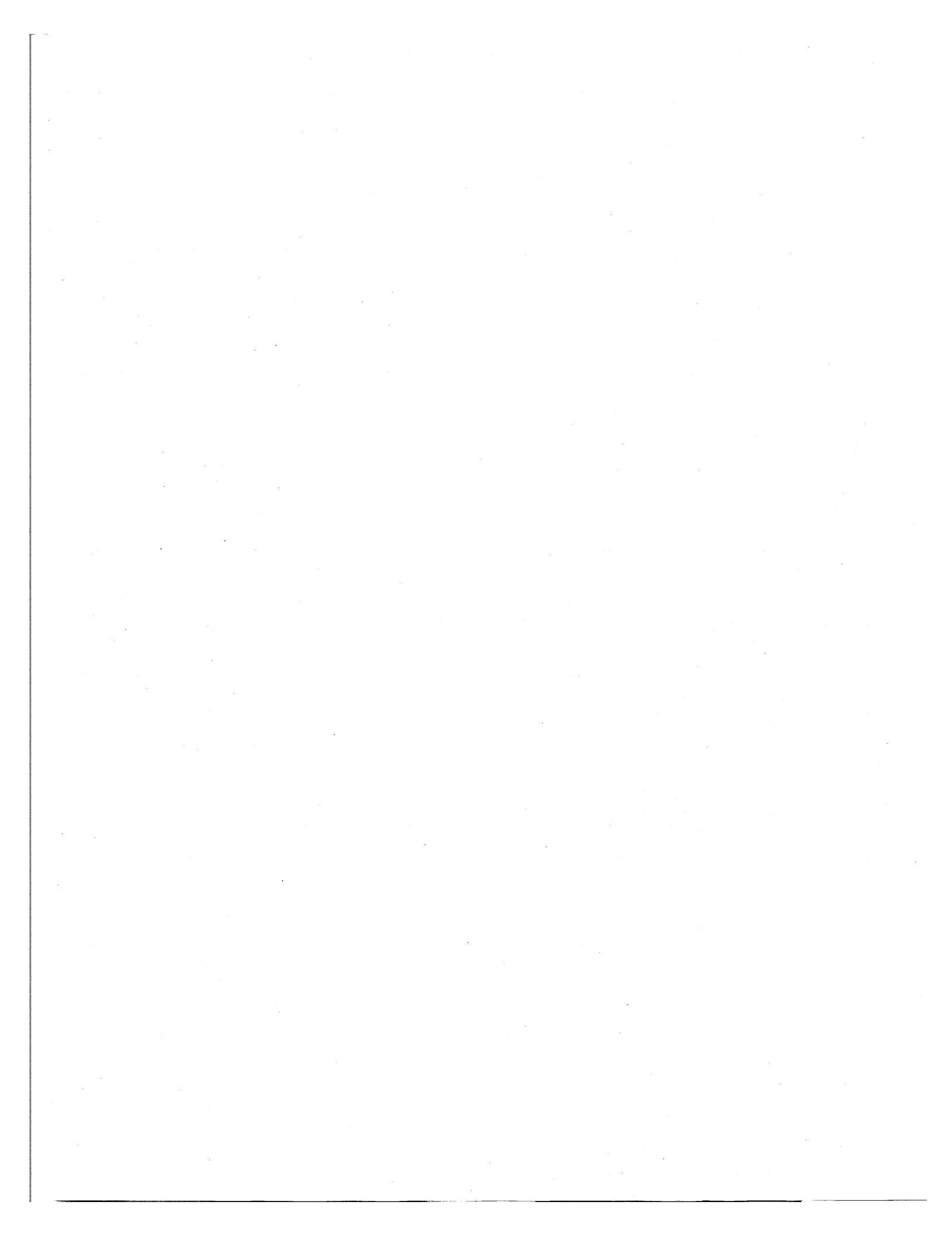


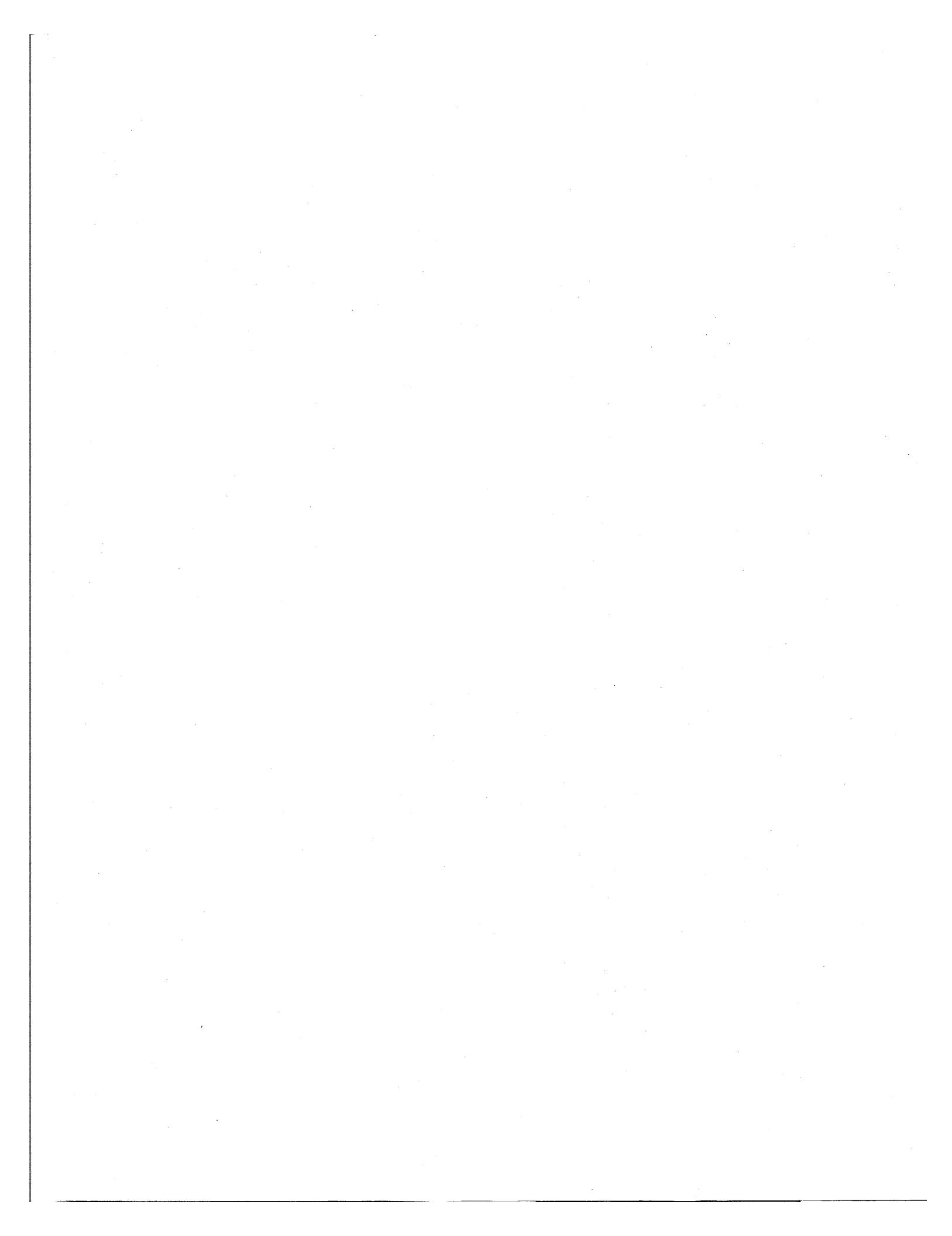
TABLE 4. Continued

(1)		(2)		(3)		(4)		(5)		(6)	
7	3	5	6	3	4	153764.59	153764.606(8)	5.714	281.84	276.71	
7	3	4	6	3	3	153764.59	153764.606(8)	5.714	281.84	276.71	
F' =	8	F" =	7	F' =	7	-0.036(0)	-0.036(0)	0.378			
7	4	5	2	6	4	153687.87	153687.873(7)	4.714	478.71	473.58	
F' =	8	F" =	7	F' =	7	-0.060(0)	-0.060(0)	0.378	478.71	473.58	
7	4	4	6	4	3	153687.87	153687.873(7)	4.714	478.71	473.58	
F' =	7	F" =	6	F' =	6	0.147(1)	0.147(1)	0.327			
6	6	6	6	5	5	-0.092(0)	-0.092(0)	0.282			
7	5	3	6	5	2	153579.62	153579.619(9)	3.429	746.59	741.46	
7	5	2	6	5	1	153579.62	153579.619(9)	3.429	746.59	741.46	
F' =	8	F" =	7	F' =	7	-0.090(0)	-0.090(0)	0.378			
7	5	2	6	5	5	0.230(1)	0.230(1)	0.327			
6	6	6	6	5	5	-0.150(1)	-0.150(1)	0.282			
8	0	8	7	0	7	175843.701	175843.703(7)	8.000	26.395	20.530	
8	1	8	7	1	7	175189.041	175189.037(8)	7.875	56.38	50.53	
8	1	7	7	1	6	176472.204	176472.199(8)	7.875	56.57	50.68	
8	2	7	7	2	6	175791.267	175791.248(7)	7.500	144.88	139.02	
8	2	6	6	2	5	175792.957	175792.954(8)	7.500	144.88	139.02	
8	3	6	7	3	5	175729.350	175729.350(9)	6.875	287.70	281.84	
8	3	5	7	3	4	175729.350	175729.350(9)	6.875	287.70	281.84	
F' =	9	F" =	8	F' =	8	-0.026(0)	-0.026(0)	0.373			
8	8	9	7	7	6	0.055(0)	0.055(0)	0.328			
F' =	8	F" =	7	F' =	7	-0.029(0)	-0.029(0)	0.289			
8	4	5	7	4	4	175641.671	175641.637(8)	6.000	484.6	478.7	
8	4	4	7	4	3	175641.671	175641.637(8)	6.000	484.6	478.7	
F' =	9	F" =	8	F' =	8	-0.042(0)	-0.042(0)	0.373			
8	5	3	7	5	2	175517.910	175517.913(8)	4.875	752.4	746.6	
F' =	9	F" =	8	F' =	8	-0.063(0)	-0.063(0)	0.328			
8	5	4	7	5	3	0.153(1)	0.153(1)	0.288			
8	5	3	7	5	2	-0.058(0)	-0.058(0)	0.289			
9	0	9	8	0	8	197821.469(7)	197821.469(7)	9.000	32.994	26.395	
9	1	9	8	1	8	197085.30	197085.424(10)	8.889	62.95	56.38	
9	1	8	8	1	7	198529.03	198528.892(10)	8.889	63.19	56.57	
9	2	8	8	2	7	197762.90	197762.928(9)	8.556	151.48	144.88	
9	2	7	8	2	6	197765.30	197765.366(10)	8.556	151.48	144.88	
9	3	7	8	3	6	197693.45	197693.444(10)	8.000	294.29	287.70	
9	3	6	8	3	5	197693.45	197693.444(10)	8.000	294.29	287.70	
9	4	6	8	4	5	197594.74	197594.741(9)	7.222	491.2	484.6	
9	4	5	6	4	4	197594.74	197594.741(9)	7.222			



TABLE 4. Continued

(1)	(2)	(3)	(4)	(5)	(6)
10 0 10 9 0 9	219798.32 218981.17 218981.019(12)	219798.282(8) 219733.824(11)	10.000 9.900	40.325 70.25	32.994 62.95
10 0 10 9 1 9	219798.32 218981.17 218981.019(12)	219798.282(8) 219733.824(11)	10.000 9.900	40.325 70.25	32.994 62.95
10 1 9 9 1 8	219798.32 218981.17 218981.019(12)	219798.282(8) 219733.824(11)	10.000 9.900	40.325 70.25	32.994 62.95
10 1 9 9 1 8	219798.32 218981.17 218981.019(12)	219798.282(8) 219733.824(11)	10.000 9.900	40.325 70.25	32.994 62.95
10 2 9 9 2 8	219798.32 218981.17 218981.019(12)	219798.282(8) 219733.824(11)	10.000 9.900	40.325 70.25	32.994 62.95
10 2 8 9 2 7	219798.32 218981.17 218981.019(12)	219798.282(8) 219737.175(13)	10.000 9.600	40.325 70.25	32.994 62.95
10 3 8 9 3 7	219798.32 218981.17 218981.019(12)	219798.282(8) 219656.805(13)	10.000 9.100	40.325 70.25	32.994 62.95
10 3 7 9 3 6	219798.32 218981.17 218981.019(12)	219798.282(8) 219656.805(13)	10.000 9.100	40.325 70.25	32.994 62.95
10 4 7 9 4 6	219798.32 218981.17 218981.019(12)	219798.282(8) 219547.105(11)	10.000 8.400	40.325 70.25	32.994 62.95
10 4 6 9 4 5	219798.32 218981.17 218981.019(12)	219798.282(8) 219547.105(11)	10.000 8.400	40.325 70.25	32.994 62.95
11 0 11 10 0 9	241774.037(10)	11.000	48.390	40.326	32.994
11 0 11 10 1 9	240875.735(16)	10.909	78.28	70.25	32.994
11 1 10 10 1 9	242639.717(16)	10.909	78.64	70.55	32.994
11 1 10 10 2 9	241703.846(15)	10.636	166.87	158.81	32.994
11 1 2 10 2 8	241708.315(17)	10.636	166.87	158.81	32.994
11 3 9 10 3 8	241619.351(16)	10.182	309.68	301.62	32.994
11 3 8 10 3 7	241619.353(16)	10.182	309.68	301.62	32.994
11 3 4 8 10 4 7	241498.644(15)	9.545	506.5	498.5	32.994
11 4 7 10 4 6	241498.644(15)	9.545	506.5	498.5	32.994
12 0 12 11 0 9	263748.630(13)	12.000	57.188	48.390	32.994
12 1 12 11 1 9	262769.484(20)	11.917	87.05	78.28	32.994
12 1 11 11 1 10	264693.665(20)	11.917	87.47	78.64	32.994
12 2 11 11 2 10	263672.909(20)	11.667	175.67	166.87	32.994
12 2 10 11 2 9	263678.717(22)	11.667	175.67	166.87	32.994
12 3 10 11 3 9	263581.003(21)	11.250	318.47	309.68	32.994
12 3 9 11 3 8	263581.006(21)	11.250	318.47	309.68	32.994
12 4 9 11 4 8	263449.276(20)	10.667	515.3	506.5	32.994
12 4 8 11 4 7	263449.276(20)	10.667	515.3	506.5	32.994
13 0 13 12 0 9	285721.952(17)	13.000	66.719	57.188	32.994
13 1 13 12 1 12	284662.177(25)	12.923	96.55	87.05	32.994
13 1 12 12 1 11	286746.515(25)	12.923	97.04	87.47	32.994
13 2 12 12 2 11	285640.924(26)	12.692	185.19	175.67	32.994
13 2 11 12 2 10	285648.316(29)	12.692	185.20	175.67	32.994
13 3 11 12 3 10	285541.677(27)	12.308	328.00	318.47	32.994
13 3 10 12 3 9	285541.681(27)	12.308	328.00	318.47	32.994
13 4 10 12 4 9	285398.921(27)	11.769	524.8	515.3	32.994
13 4 9 12 4 8	285398.921(27)	11.769	524.8	515.3	32.994
1 1 0 1 1 0	F'' = 1	1 1 1 0	1.500	30.814	32.286
1 1 1 1 1 0	F'' = 1	0 1 2 1	0.111	30.809	32.270
1 1 2 2 2 0	F'' = 1	1 1 2 1	0.139		
2 2 0 0 0 0	F'' = 1	2 1 1 1	-0.419(2)		
2 2 1 1 1 0	F'' = 1	2 1 1 1	-0.278(2)		
2 2 2 2 2 0	F'' = 1	1 1 1 1	0.083		
2 2 1 1 1 0	F'' = 1	2 1 1 1	0.056(0)		
2 2 0 0 0 0	F'' = 1	1 1 1 1	0.417		
2 2 1 1 1 0	F'' = 1	2 1 1 1	0.139		
2 2 0 0 0 0	F'' = 1	1 1 1 1	0.111		
1 1 0 0 0 0	F'' = 1	1 1 1 1	0.910(4)		
1 1 1 1 1 0	F'' = 1	2 1 1 1	0.833		
1 1 2 2 2 0	F'' = 1	0 1 2 1	0.231		

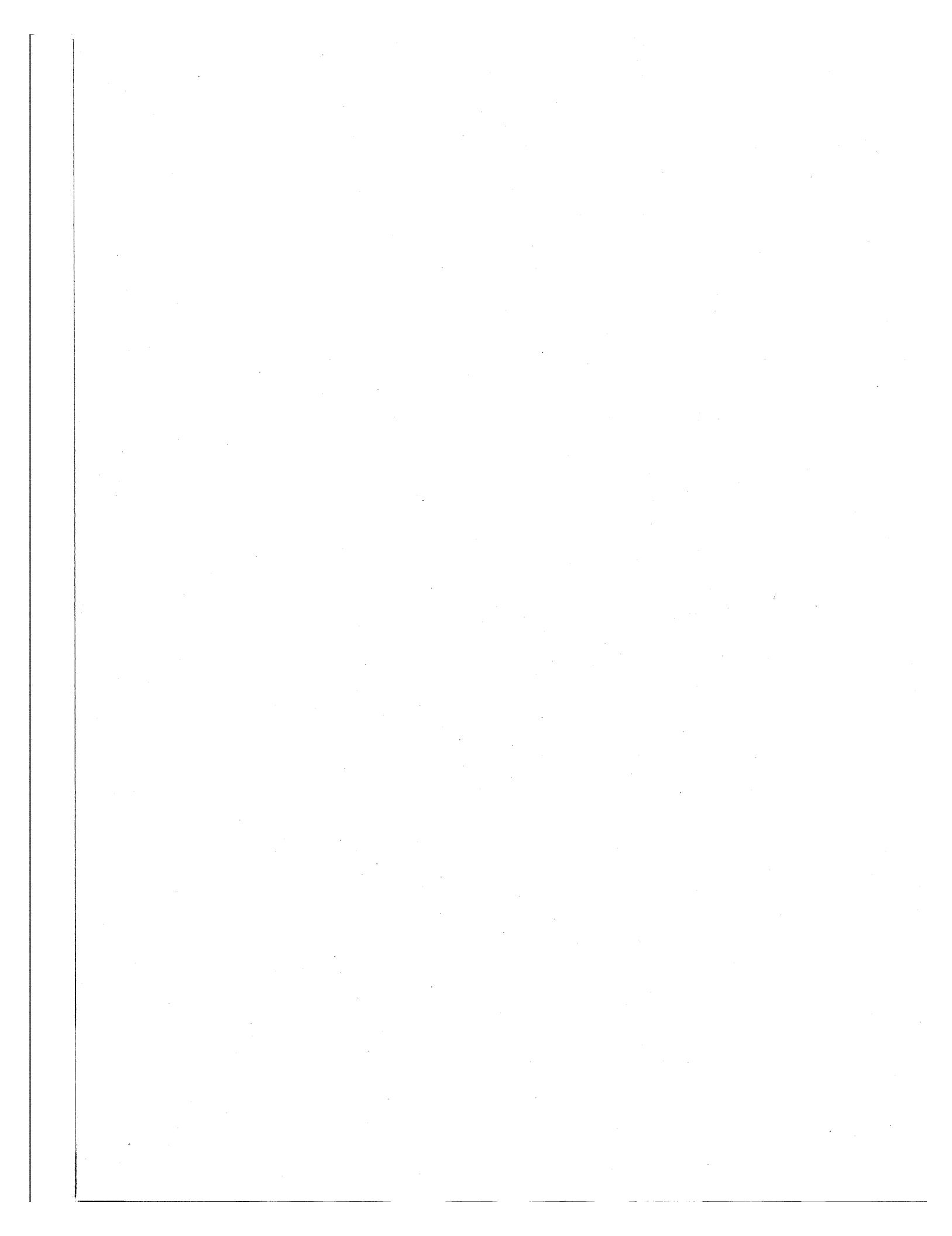


MICROWAVE SPECTRUM OF ISOCYANIC ACID

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TABLE 4. Continued

	(1)	(2)	(3)	(4)	(5)	(6)
	3	3	0.079(1)	0.415		
	1	1	0.278(2)	0.150		
3	2	3	962.576(1)	0.583		
$F' =$	3	1	-0.278(3)	0.280		
	4	4	0.093(1)	0.402		
	2	2	0.222(2)	0.212		
4	1	3	1604.270(1)	0.450		
$F' =$	4	4	-0.278(2)	0.301		
	5	5	0.101(1)	0.391		
	3	3	0.198(1)	0.243		
5	1	4	2406.361(2)	0.367		
$F' =$	5	5	-0.278(2)	0.311		
	6	6	0.107(1)	0.383		
	4	4	0.185(1)	0.262		
6	1	5	3368.830(2)	0.310		
$F' =$	5	6	-0.278(2)	0.318		
	7	7	0.111(1)	0.377		
	5	5	0.177(1)	0.274		
7	1	5	4491.659(3)	0.310		
$F' =$	7	7	-0.276(2)	0.322		
	3	8	0.114(1)	0.372		
	5	6	0.171(1)	0.283		
8	1	7	5774.822(4)	0.236		
$F' =$	8	8	-0.278(2)	0.324		
	9	9	0.117(1)	0.368		
	7	7	0.167(1)	0.290		
9	1	8	7218.291(4)	0.211		
$F' =$	9	9	-0.278(2)	0.326		
	10	10	0.119(1)	0.365		
	8	8	0.163(1)	0.295		
10	1	9	8822.034(5)	0.191		
$F' =$	11	10	0.121(1)	0.362		
	10	10	-0.278(2)	0.327		
	9	9	0.161(1)	0.299		
11	1	10	10586.015(5)	0.174		
$F' =$	12	11	0.122(1)	0.360		
	11	11	-0.278(2)	0.328		
	10	10	0.159(1)	0.302		
12	1	11	12510.196(6)	0.160		
			87.471	87.053		
			70.548	70.254		
			78.642	78.288		



WINNEWISSE, HOCKING, AND GERRY

TABLE 4. Continued

	(1)	(2)	(3)	(4)	(5)	(6)
$F' = 13$	$F'' = 13$	12510.35	0.123(1)	0.358		
12	12	12509.92	-0.278(2)	0.329		
11	11	12510.35	0.157(1)	0.305		
$F' = 14$	13	1	14594.535(6)	0.148		
13	12	13	14594.67	0.124(1)	0.356	
$F'' = 14$	14	14	14594.25	-0.278(2)	0.330	
13	13	13	14594.67	0.155(1)	0.307	
12	12	12				
$F' = 14$	13	14	1	16838.985(6)	0.138	
14	13	14	14	16839.12	0.125(1)	0.355
$F'' = 15$	15	15	16838.71	-0.278(2)	0.330	
14	14	14	16839.12	0.154(1)	0.309	
13	13	13				
$F' = 15$	14	15	1	19243.496(6)	0.129	
14	14	15	15	19243.61	0.126(1)	0.353
$F'' = 16$	16	16	15	19243.20	-0.278(2)	0.331
15	15	15	14	19243.61	0.153(1)	0.310
14	14	14				
$F' = 16$	15	16	1	21808.016(6)	0.121	
15	15	16	16	21808.16	0.127(1)	0.352
$F'' = 17$	16	16	16	21807.75	-0.278(2)	0.331
14	14	15	15	21808.16	0.152(1)	0.312
15	15	15				
$F' = 17$	16	17	1	24532.487(6)	0.114	
16	16	17	17	24532.62	0.128(1)	0.351
$F'' = 18$	17	17	17	24532.20	-0.278(2)	0.331
15	15	16	16	24532.62	0.151(1)	0.313
16	16	16				
$F' = 18$	17	18	1	24532.487(6)	0.114	
17	17	18	18	24532.62	0.128(1)	0.351
$F'' = 19$	18	18	18	24532.20	-0.278(2)	0.331
16	16	17	17	24532.62	0.151(1)	0.313
17	17	17				
$F' = 19$	18	19	1	27416.848(6)	0.108	
18	18	19	19	27417.00	0.128(1)	0.350
$F'' = 20$	19	19	18	27416.58	-0.278(2)	0.332
17	17	17	17	27417.00	0.151(1)	0.314
18	18	18				
$F' = 20$	19	20	1	30461.036(7)	0.103	
19	19	20	20	30461.19	0.129(1)	0.350
$F'' = 21$	20	20	19	30460.77	-0.278(2)	0.332
18	18	19	18	30461.19	0.150(1)	0.315
19	19	20				
$F' = 21$	20	21	1	33664.981(9)	0.098	
20	20	21	21	33665.12	0.129(1)	0.349
$F'' = 22$	21	21	20	33664.70	-0.278(2)	0.332
21	21	21	19	33665.12	0.149(1)	0.316
20	20	21				
$F' = 22$	21	21	1	37028.612(12)	0.093	
21	21	21	22	37028.77	0.130(1)	0.348
$F'' = 23$	22	22	21	37028.34	-0.278(2)	0.332
20	20	21	20	37028.77	0.149(1)	0.317
21	21	21				

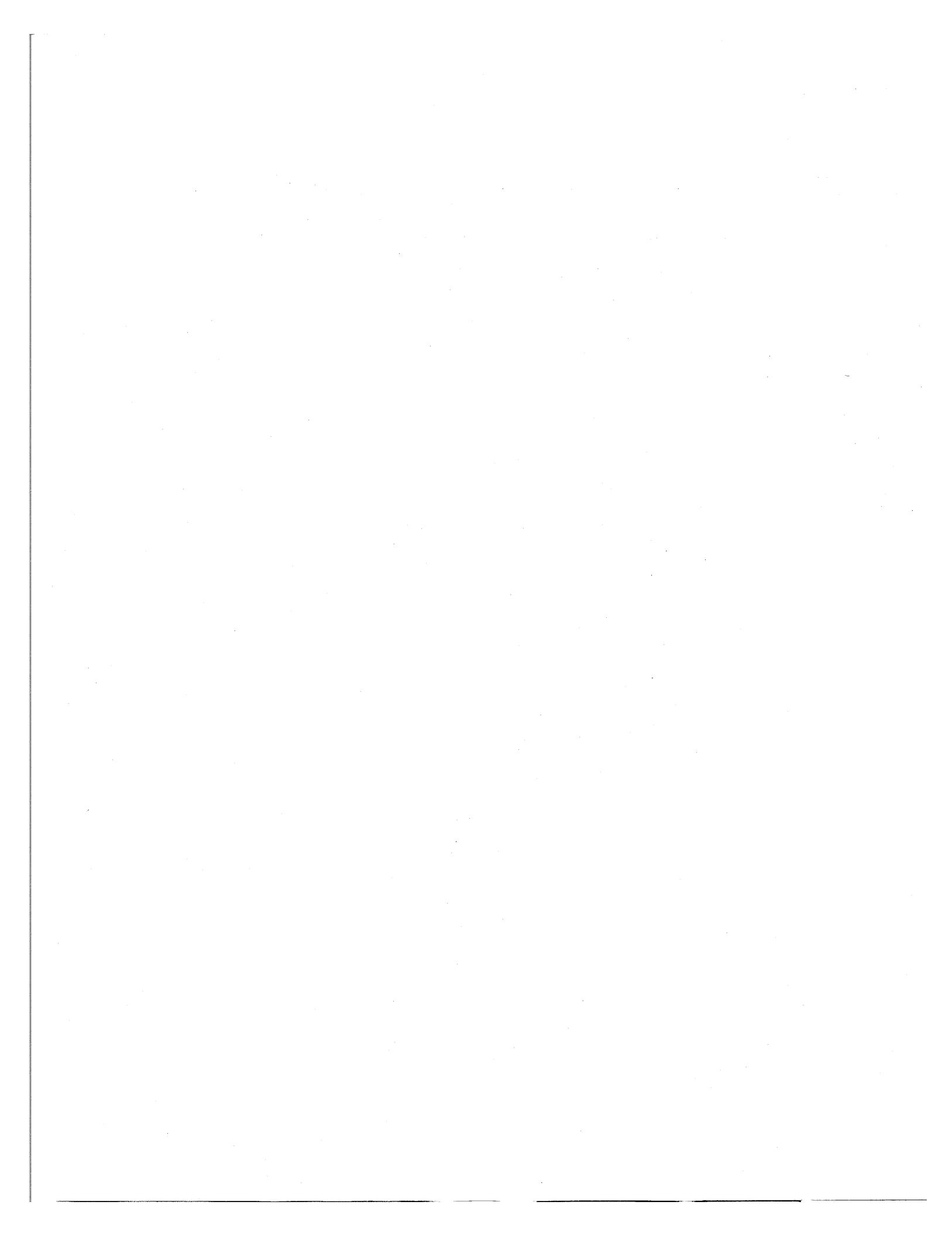


TABLE 4. Continued

MICROWAVE SPECTRUM OF ISOCYANIC ACID

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	(1)	(2)	(3)	(4)	(5)	(6)
29	1 29	30 0	30	207631.364	207631.365(12)	14.819
30	1 30	31 0	31	183318.380	183318.365(10)	15.352
31	1 31	32 0	32		158934.883(8)	15.886
32	1 32	33 0	33		134481.571(7)	16.423
33	1 33	34 0	34	109959.087	109959.100(6)	16.961
34	1 34	35 0	35	85369.134	85368.158(6)	17.502
35	1 35	36 0	36		60709.448(6)	18.046
F' = 36	F'' = 37			60709.339	-0.063(0)	0.342
35		36		60709.597	+0.131(1)	0.333
34		35		60709.399	-0.069(1)	0.324
36	1 36	37 0	37	35983.71	35983.693(6)	18.592
37	1 37	38 0	38	11191.64	11191.630(7)	19.140
39	0 39	38 1	38	13666.02	13665.987(7)	19.691
40	0 40	39 1	39		38588.382(7)	20.245
41	0 41	40 1	40		63574.773(6)	
F' = 42	F'' = 41			63574.810	0.062(0)	0.342
41		40		63574.618	-0.130(1)	0.333
40		39		63574.810	0.068(1)	0.325
42	0 42	41 1	41	88624.372	88624.352(6)	
43	0 43	42 1	42	113736.305	113736.298(7)	
44	0 44	43 1	43	138909.751	138909.772(8)	
45	0 45	44 1	44		164143.924(10)	
46	0 46	45 1	45	189437.889	189437.882(13)	
47	0 47	46 1	46	214790.736	214790.761(18)	

a The estimated accuracy of the height of the energy levels are as follows: for $K_a = 0$ about 2 decimal places; for $K_a = 1$, $K_a = 2$ about 1 decimal place; for $K_a = 3$ about $\pm 1 \text{ cm}^{-1}$; for $K_a = 4$ about $\pm 3 \text{ cm}^{-1}$; for $K_a = 5$ about $\pm 7 \text{ cm}^{-1}$.

b If there is no entry under References, then the data are taken from Ref. (75).

c The estimated experimental uncertainty in the measured transitions are: for the microwave transitions quoted to two decimal places $\leq \pm 0.100 \text{ MHz}$, for millimeter wave transitions above 37 GHz quoted to two decimal places $< \pm 0.150 \text{ MHz}$, and for those quoted to three decimal places $< \pm 0.030 \text{ MHz}$.

d Note that one standard deviation is presented for the calculated uncertainty. These values should be multiplied by a factor of 2 to obtain $\sim 95\%$ confidence level.

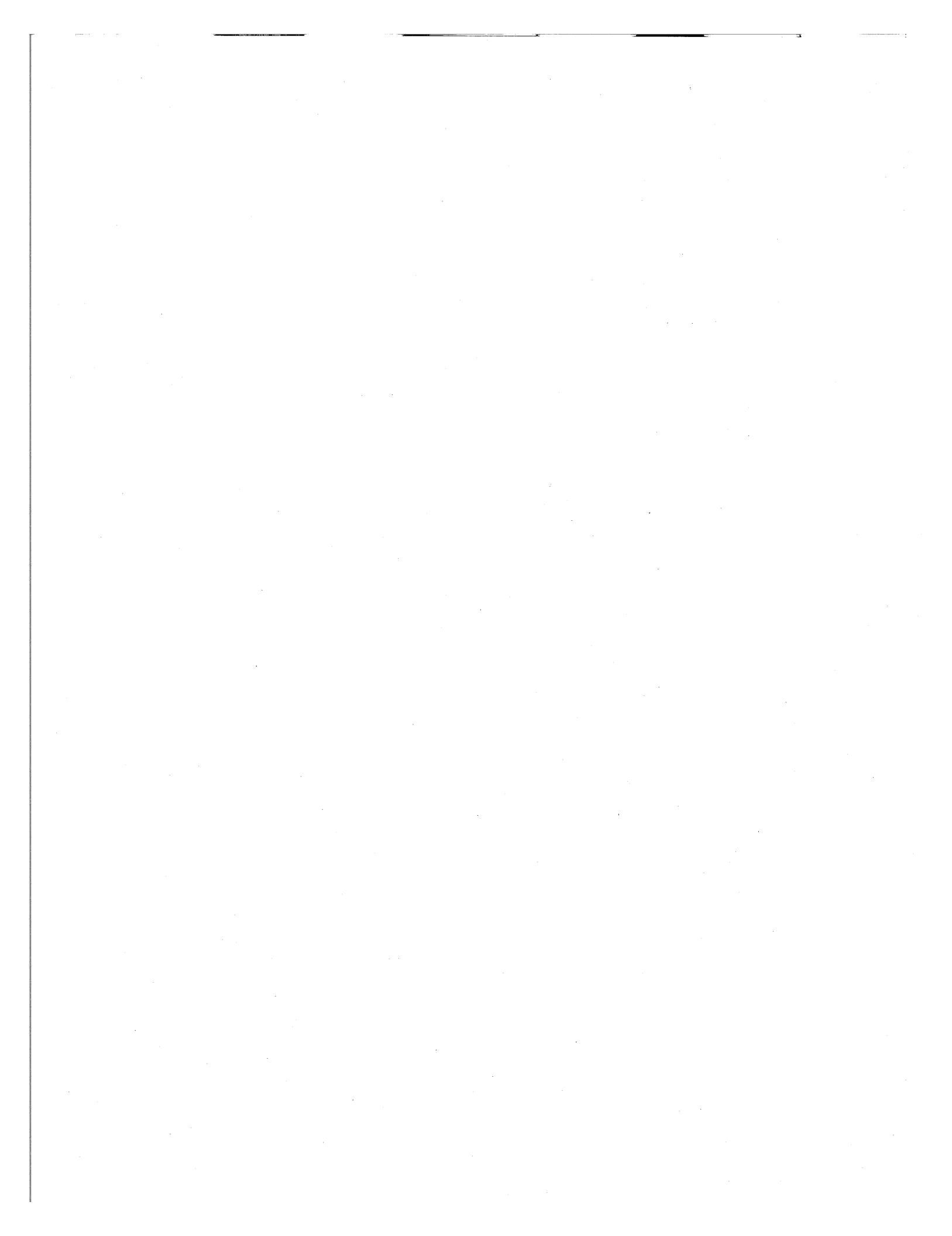


TABLE 5. The Microwave Spectrum of Isocyanic Acid-d. (MHz) ($D^{14}N^{12}C^{16}O$)

(1)				(2)				(3)				(4)				(5)			
Transition J'	K_a'	K_c'	J''	K_a''	K_c''	Observed (Est. Uncert.) ^c	Calculated + Quadrupole shifts (standard dev.) ^d	Line Strength + Rel. Intensity of Quadrupole comp.	Energy levels ^a in cm ⁻¹ upper state	Energy levels ^a in cm ⁻¹ lower state	Ref.								
1 0	1	0	0	1	0	20393.30	20393.377(1)	1.000	0.680	0.680	0.0								
			$F'' =$			20393.91	-0.106(1)	0.556											
						20392.40	0.528(4)	0.333											
							-1.055(9)	0.111											
2 0	2	1	0	1	2	40786.568	40786.593(2)	2.000	2.041	0.680	0.0								
			$F'' =$			40787.199	-0.045(0)	0.467											
						40786.568	0.633(5)	0.083											
						40785.534	0.000(0)	0.250											
						40787.199	-1.055(9)	0.083											
							0.527(4)	0.111											
2 1	1	1	1	0	2	41021.524	41021.658(2)	1.500	18.757	17.388									
			$F'' =$			41021.918	-0.117(1)	0.467											
						41022.186	0.056(3)	0.083											
						41021.704	0.528(4)	0.250											
						41020.699	0.258(4)	0.083											
							-0.920(7)	0.111											
2 1	2	1	1	1	2	40553.456	40553.596(2)	1.500	18.733	17.380									
			$F'' =$			40553.971	-0.139(1)	0.467											
						40554.144	0.366(3)	0.083											
						40552.917	0.528(4)	0.250											
							-0.258(4)	0.083											
3 0	3	2	0	2	1	61179.505	61179.488(3)	2.999	4.081	2.041									
			$F'' =$			60830.158(3)	60830.158(3)	2.667	20.762	18.733									
						60830.097	-0.063(0)	0.429											
						60830.291	0.132(1)	0.296											
						60830.097	-0.027(1)	0.200											
3 1	2	2	1	1	3	61532.156	61532.226(3)	2.667	20.809	18.757									
			$F'' =$			61532.360	-0.050(0)	0.429											
						61532.156	0.132(1)	0.296											
							-0.079(1)	0.200											
3 2	2	2	2	1	3	61184.850	61185.030(2)	1.667	69.91	67.87									
			$F'' =$			61185.547	-0.151(1)	0.429											
						61184.479	0.528(4)	0.296											
							-0.528(4)	0.200											
3 2	1	2	2	0		61185.351(2)		1.667	69.91	67.87									

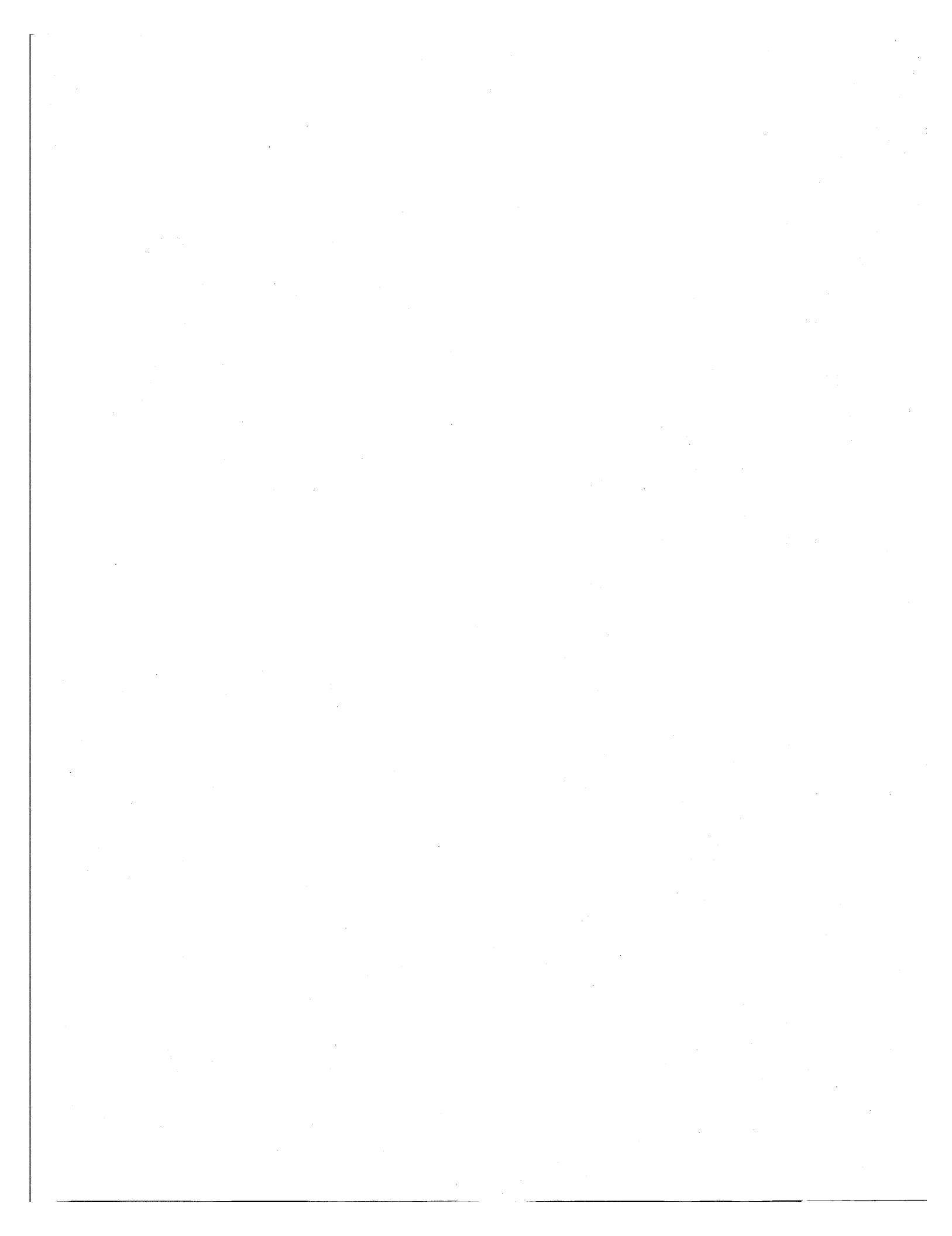


TABLE 5. Continued

(1)	(2)	(3)	(4)	(5)	(6)
$F' = 4$	$F'' = 3$	61185.188 61185.869 61184.850	0.151(1) 0.527(4) -0.528(4)	0.429 0.296 0.200	4.061 20.762 20.809 69.91
4	0	81571.913 81106.420 82042.469	81571.899(4) 81106.436(3) 82042.480(3)	3.998 3.750 2.750	6.802 23.468 23.546 72.63
4	1	81579.486 $F'' = 4$	81579.612(2) -0.077(1)	3.000 0.407	20.809 69.91
4	2	81579.812 $F'' = 3$	0.211(2) -0.151(1)	0.313 0.238	23.546 72.63
4	3	81579.486 $F'' = 2$	81580.415(3) -0.077(1)	3.000 0.407	20.762 69.91
4	4	81580.283 $F'' = 1$	81580.614 81580.283	0.313 0.238	4.061 20.762 20.809 69.91
4	5	$F' = 4$	81586.072 81586.686 81585.810	1.750 0.407 0.313 0.238	149.31 149.31 149.31 149.31
4	3	81586.243(4) 81586.243(4)	81586.243(4) -0.153(1)	1.750 0.407	149.31 149.31
4	4	$F'' = 3$	0.475(4) -0.396(3)	0.313 0.238	149.31 149.31
4	5	$F'' = 4$	81586.072 81586.686 81585.810	1.750 0.407 0.313 0.238	149.31 149.31 149.31 149.31
5	0	101963.694 $F' = 5$	101963.666(5) 101382.342 102552.335 101973.809 101975.407 101975.433(3) 101982.290(5)	5.000 4.800 4.800 4.200 4.200 4.200 3.200	6.802 23.467 23.546 72.635 72.635 72.635 152.03
5	1	101963.694 $F'' = 4$	101963.666(5) 101382.342 102552.335 101973.809 101975.407 101975.433(3) 101982.290(5)	5.000 4.800 4.800 4.200 4.200 4.200 3.200	6.802 23.467 23.546 72.635 72.635 72.635 152.03
5	2	101963.694 $F'' = 3$	101963.666(5) 101382.342 102552.335 101973.809 101975.407 101975.433(3) 101982.290(5)	5.000 4.800 4.800 4.200 4.200 4.200 3.200	6.802 23.467 23.546 72.635 72.635 72.635 152.03
5	3	101963.694 $F'' = 2$	101963.666(5) 101382.342 102552.335 101973.809 101975.407 101975.433(3) 101982.290(5)	5.000 4.800 4.800 4.200 4.200 4.200 3.200	6.802 23.467 23.546 72.635 72.635 72.635 152.03
5	4	101963.694 $F'' = 1$	101963.666(5) 101382.342 102552.335 101973.809 101975.407 101975.433(3) 101982.290(5)	5.000 4.800 4.800 4.200 4.200 4.200 3.200	6.802 23.467 23.546 72.635 72.635 72.635 152.03
5	5	$F' = 6$	101988.742 101989.369	1.800 0.394 0.422(3) -0.317(3)	264.61 264.61 264.61 264.61
5	4	2	4	1	261.21
5	4	1	4	0	261.21
5	6	$F' = 5$	101988.742 101989.369	1.800 0.394 0.422(3) -0.317(3)	264.61 264.61 264.61 264.61
6	0	6	5	0	14.285
6	1	6	5	1	30.907
6	2	5	5	1	31.071
6	3	4	5	2	80.12
6	4	3	4	3	80.12
6	5	3	3	3	76.04
6	6	3	2	4	159.51
6	7	$F' = 6$	6	5	155.43
6	6	$F' = 5$	5	4	155.43
6	5	$F' = 4$	5	4	155.43
6	4	$F' = 3$	5	4	155.43
6	3	$F' = 2$	5	4	155.43
6	2	$F' = 1$	5	4	155.43
6	1	$F' = 0$	5	4	155.43
6	0	$F' = -1$	5	4	155.43
6	1	$F' = -2$	5	4	155.43
6	2	$F' = -3$	5	4	155.43
6	3	$F' = -4$	5	4	155.43
6	4	$F' = -5$	5	4	155.43
6	5	$F' = -6$	5	4	155.43
6	6	$F' = -7$	5	4	155.43

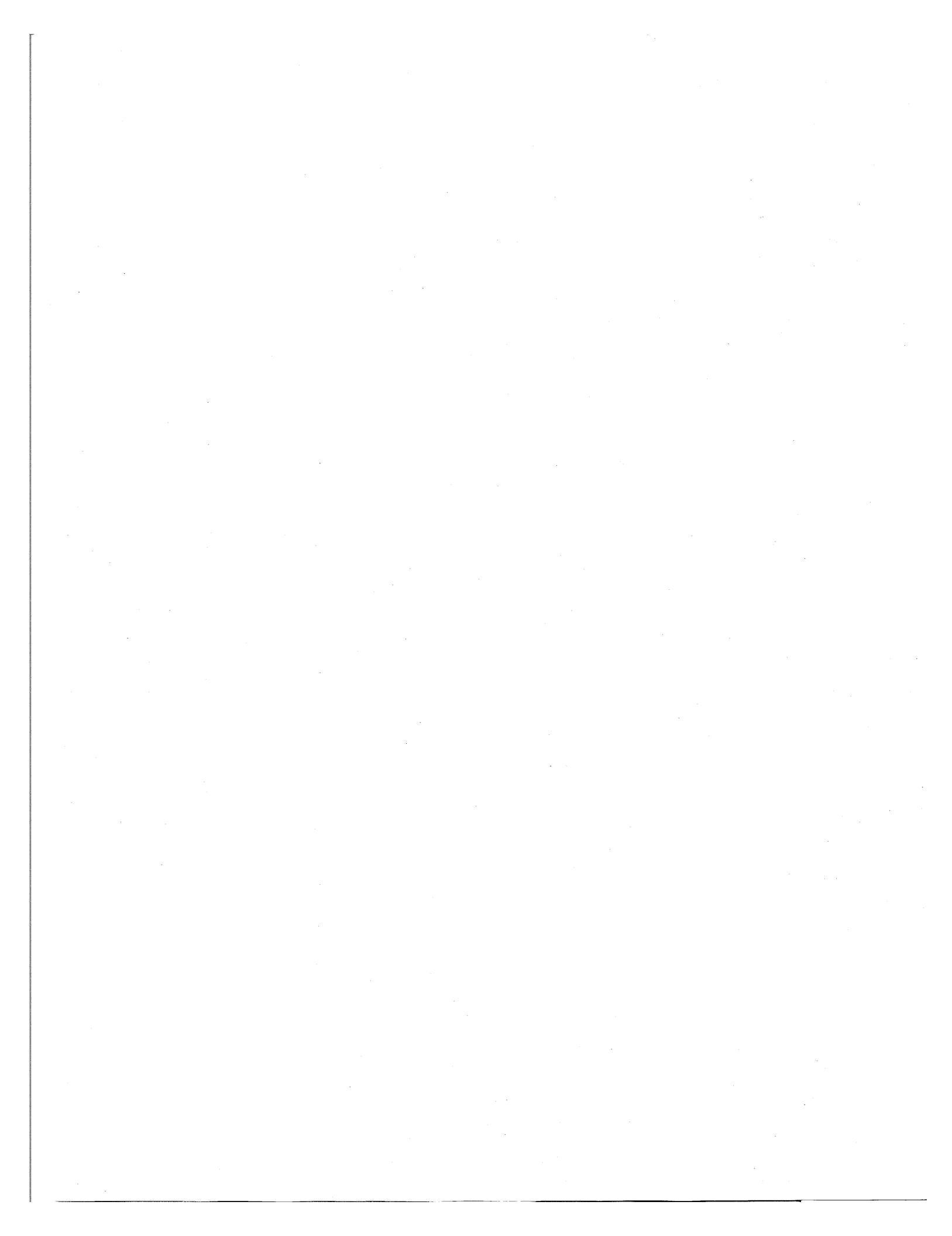


TABLE 5. Continued

(1)	(2)	(3)	(4)	(5)
6 4 3	4 5 2	122385.949 (7)	3.333	264.61
6 4 4	4 5 1	122385.949 (7)	3.333	264.61
F' = 7	F" = 6	-0.092 (1)	0.385	
		0.241 (2)	0.324	
		-0.163 (1)	0.273	
7 7 0	7 6 0	142744.619	19.046	14.285
7 7 1	6 6 1	141932.599	6.857	35.642
7 7 1	6 6 1	143570.318 (4)	6.857	30.907
7 7 2	6 6 2	142760.781	6.429	31.071
7 7 2	5 6 2	142765.263	6.429	80.118
7 7 3	5 6 3	142773.280	142773.290 (6)	80.118
7 7 3	4 6 3	142773.280	5.714	84.88
7 7 3	4 6 3	142773.292 (6)	5.714	164.27
7 7 4	4 6 4	142782.475	4.714	159.51
7 7 4	3 6 4	142782.502 (8)	4.714	268.69
7 7 4	3 6 4	142782.502 (8)	4.714	268.69
8 8 0	8 7 0	163133.522	8.000	24.488
8 8 1	8 7 1	162206.809	7.875	19.046
8 8 1	7 7 1	164078.251	7.875	41.052
8 8 2	6 7 2	163153.355 (6)	7.500	35.642
8 8 2	6 7 2	163160.093 (7)	7.500	41.333
8 8 3	6 7 3	163168.149	6.875	90.32
8 8 3	5 7 3	163168.149	6.875	84.88
8 8 3	5 7 3	163168.107 (7)	6.875	164.27
8 8 4	5 7 4	163178.543 (9)	6.000	164.27
8 8 4	4 7 4	163178.543 (9)	6.000	273.45
8 8 4	4 7 4	163178.543 (9)	6.000	273.45
9 9 0	9 8 0	183521.086	9.000	30.609
9 9 1	8 8 1	182480.240	8.889	47.139
9 9 1	8 8 1	184585.370	8.889	41.052
9 9 2	8 8 2	183545.171	8.559	41.333
9 9 2	7 8 2	183545.186 (8)	8.559	47.491
9 9 2	7 8 2	183554.791	8.559	90.32
9 9 3	7 8 3	183562.372	8.000	96.44
9 9 3	6 8 3	183562.372	8.000	169.72
9 9 3	6 8 3	183574.022	7.222	169.72
9 9 4	6 8 4	183574.022	7.222	278.90
9 9 4	5 8 4	183574.022	7.222	278.90
10 10 0	10 9 0	203907.199 (6)	10.000	37.411
10 10 1	10 9 1	202752.842	9.900	30.609
10 10 1	9 9 1	205091.551	9.900	53.902
10 10 2	9 9 2	203936.218	9.600	47.139
10 10 2	8 9 2	203949.419 (12)	9.600	54.331
10 10 3	8 9 3	203956.028	9.100	47.491
10 10 3	7 9 3	203956.028	9.100	103.25
10 10 4	7 9 4	203968.799 (11)	8.400	96.44
10 10 4	6 9 4	203968.799 (11)	8.400	103.24
11 11 0	11 10 0	224291.760	11.000	103.24
				175.84
				182.64
				175.84
				182.64
				291.82
				285.02
				291.82
				285.02
				44.893
				37.411

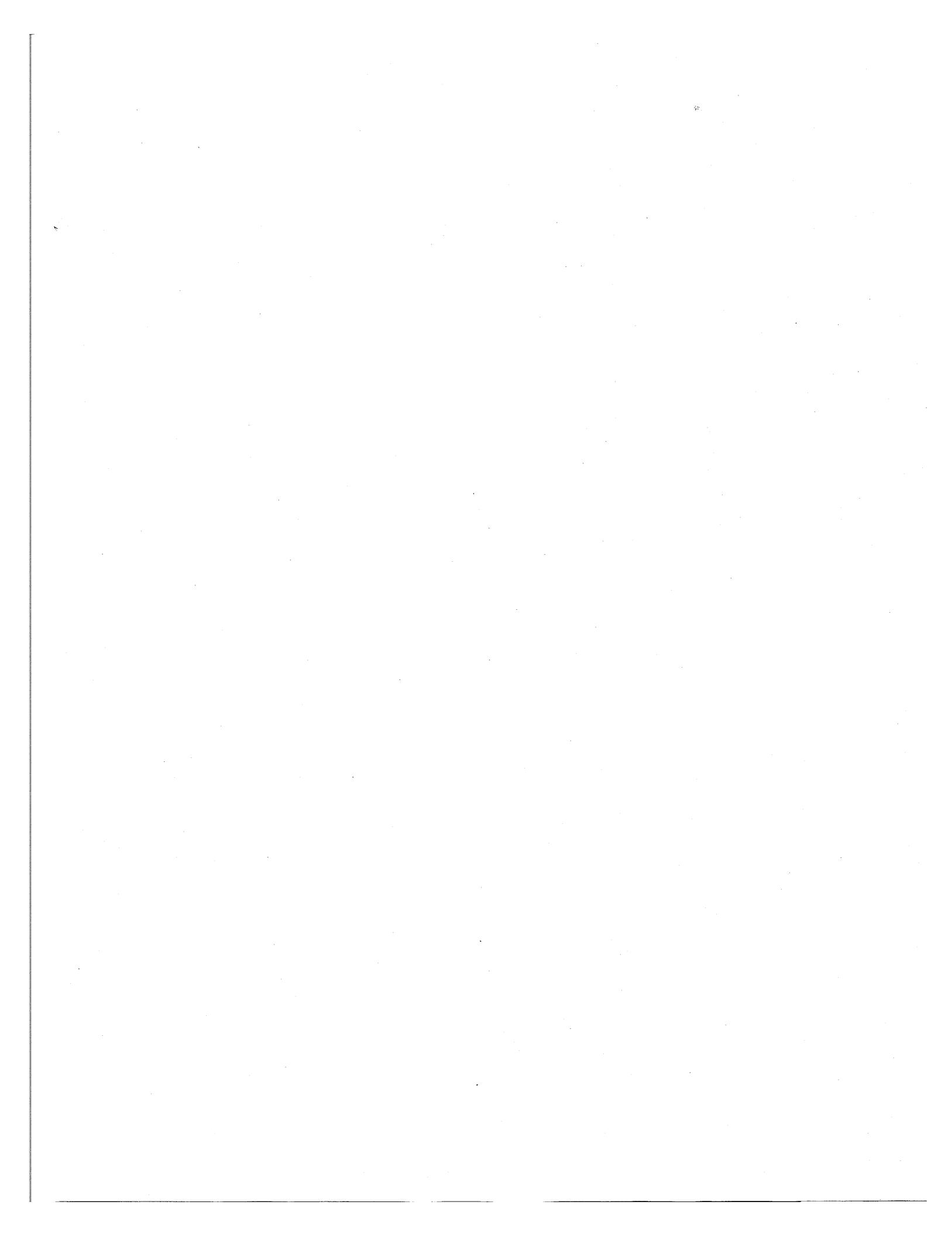


TABLE 5. Continued

(1)	(2)	(3)	(4)	(5)	(6)
11 1 11 10 1 10	223024.484(6)	10.909	61.342	53.902	
11 1 10 1 1 9	225596.697(6)	10.909	61.857	54.331	
11 2 10 10 2 9	224326.273(14)	10.636	110.73	103.25	
11 1 2 9 10 2 8	224343.909(16)	10.636	110.73	103.24	
11 1 3 9 10 3 8	224348.969(11)	10.182	190.13	182.64	
11 1 3 8 10 3 7	224348.990(11)	10.182	190.13	182.64	
11 1 4 8 10 4 7	224362.883	299.31	291.82		
11 1 4 7 10 4 6	224362.846	9.545	291.82		
12 0 12 11 0 11	244674.487	12.000	53.054	44.892	
12 1 12 11 1 11	243295.122	11.917	69.457	61.342	
12 1 11 11 1 10	246100.655	11.917	70.066	61.657	
12 2 11 11 2 10	244715.460	11.667	118.89	110.73	
12 2 10 11 2 9	244738.382	11.667	118.90	110.73	
12 3 10 11 3 9	244741.168(14)	11.250	198.29	190.13	
12 3 9 11 3 8	244741.201(14)	11.250	198.29	190.13	
12 4 9 11 4 8	244756.128(17)	10.667	307.47	299.31	
12 4 8 11 4 7	244756.128(17)	10.667	307.47	299.31	
13 0 13 12 0 12	265055.285(8)	13.000	61.895	53.054	
13 1 13 12 1 12	263564.546(11)	12.923	78.249	69.457	
13 1 12 12 1 11	266603.390(10)	12.923	78.959	70.066	
13 2 12 12 2 11	265103.316(23)	12.692	127.74	118.89	
13 3 11 12 3 10	265132.539(17)	12.308	207.13	198.29	
13 3 10 12 3 9	265132.588(17)	12.308	207.13	198.29	
13 4 10 12 4 9	265148.509(22)	11.769	316.32	307.47	
13 4 9 12 4 8	265148.510(22)	11.769	316.32	307.47	
13 2 11 12 2 10	265132.477(26)	12.692	127.74	118.90	
1 1 0 1 1 F"=C	234.035(0)	1.500	17.39	17.39	
F"=1	-0.662(7)	0.111			
F"=1	-0.420(3)	0.139			
F"=1	-0.258(4)	0.083			
F"=1	0.052(1)	0.417			
F"=2	0.213(4)	0.139			
F"=2	0.920(7)	0.111			
2 1 F"=2	2 1 F"=2	0.833	18.76	18.73	
3 1 F"=3	3 1 F"=3	0.231			
3 1 F"=4	4 2	0.074(1)			
4 1 F"=4	4 1 F"=4	0.415			
		0.258(4)			
		1404.165(2)	0.583		
		0.258(4)	0.280		
		0.086(1)	0.402		
		0.206(4)	0.212		
		2340.208(3)	0.450	23.55	23.47
		-0.258(4)	0.301		
		0.094(2)	0.39		



TABLE 5. Continued

	(1)	(2)	(3)	(4)	(5)	(6)
3	3	0.184(3)	0.243			
5	4	5	5	0.367		
$F' =$	5	1	$F'' =$	0.311		
6	6	5	5	0.383		
4	4	6	6	0.262		
6	1	5	6	4914.060(5)	0.310	26.85
$F' =$	6	6	$F'' =$	-0.258(4)	0.318	
7	7	6	6	0.103(2)	0.377	
5	5	7	7	0.164(3)	0.274	
7	1	6	7	6551.758(7)	0.268	
$F' =$	7	7	$F'' =$	-0.258(4)	0.322	
8	1	7	8	0.106(2)	0.372	
$F' =$	9	8	$F'' =$	0.159(3)	0.283	
7	7	7	7	8423.215(8)	0.236	
8	1	7	8	8423.34	0.109(2)	35.64
$F' =$	9	8	$F'' =$	8422.96	-0.258(4)	
7	7	7	7	8423.34	0.154(3)	
9	1	8	9	10528.48	0.211	
$F' =$	10	9	$F'' =$	10528.11	0.111(2)	41.052
8	8	8	8	10528.48	-0.258(4)	
10	1	9	10	10528.48	0.152(3)	
$F' =$	11	10	$F'' =$	10528.48	0.295	
9	9	9	9	10528.48	41.333	
10	1	9	10	12867.076(9)	0.191	
$F' =$	11	10	$F'' =$	12867.19	0.112(2)	47.139
9	9	9	9	12866.80	-0.258(4)	
11	1	10	11	12867.19	0.149(3)	
$F' =$	12	11	$F'' =$	12867.19	0.295	
10	10	11	11	12867.19	54.332	53.902
11	1	10	11	15439.289(9)	0.174	
$F' =$	12	11	$F'' =$	15439.41	0.113(2)	
10	10	11	11	15439.03	0.360	
12	1	11	12	15439.41	-0.258(4)	
$F' =$	13	12	$F'' =$	15439.41	0.147(3)	
11	11	12	12	18244.878(9)	0.174	
13	1	12	13	18245.02	0.113(2)	
$F' =$	14	12	$F'' =$	18244.60	0.358	78.959
12	12	12	12	18245.02	-0.258(4)	78.249
11	11	12	12	18245.02	0.146(2)	
13	1	12	13	21283.722(8)	0.160	
$F' =$	14	13	$F'' =$	21283.85	0.116(2)	87.716
12	12	13	13	21283.47	-0.258(4)	
11	11	12	12	21283.85	0.144(2)	
14	1	13	14	24555.687(9)	0.133	
$F' =$	15	14	$F'' =$	24555.85	0.116(2)	88.535



MICROWAVE SPECTRUM OF ISOCYANIC ACID

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TABLE 5. Continued

(1)	(2)	(3)	(4)	(5)	(6)
1.4	1.4	24555.42	-0.258(4)	0.330	
1.3	1.3	24555.85	0.143(2)	0.309	
1.5	1.4	1.5	1.5	28060.627(11)	0.129
F' = 1.6	F" = 1.6			0.117(2)	0.353
1.5	1.5			-0.258(4)	0.331
1.4	1.4			0.142(2)	0.310
1.6	1.5	1.6	1.6	31798.387(17)	0.121
F' = 1.7	F" = 1.7			0.118(2)	0.352
1.6	1.6			-0.258(4)	0.331
1.5	1.5			0.141(2)	0.312
1.3	1.3	1.4	0	204827.236	6.591
1.4	1.4	1.5	0	182849.512	182849.547(12)
1.5	1.5	1.6	0	160764.694(11)	7.635
1.6	1.6	1.7	0	138573.857	138573.843(10)
1.7	1.7	1.8	0	116278.174	115278.212(9)
1.8	1.8	1.9	0	93879.120	93879.084(10)
1.9	1.9	2.0	0	71377.807(10)	9.757
2.0	1.20	2.1	0	43775.793(11)	10.296
2.1	1.21	2.2	0	26074.517(12)	10.839
2.2	1.22	2.3	0	3275.520(14)	11.386
2.4	0.24	2.3	1	19619.58	19619.597(15)
2.5	0.25	2.4	1	42609.168(16)	12.493
2.6	0.26	2.5	1	65691.469(17)	13.053
2.7	0.27	2.6	1	88864.56	88864.708(17)
2.8	0.28	2.7	1	112127.039(16)	13.617
2.9	0.29	2.8	1	135476.553(15)	14.186
3.0	0.30	2.9	1	153911.284(14)	14.760
3.1	0.31	3.0	1	182429.187	15.339
3.2	0.32	3.1	1	206028.254	15.924
3.3	0.33	3.2	1	229706.240(25)	16.514
					17.109

a The estimated accuracy of the height of the energy levels are as follows: for $K_a = 0$ about 2 decimal places; for $K_a = 1$, $K_a = 2$ about 1 decimal place; for $K_a = 3$ about $\pm 1 \text{ cm}^{-1}$; for $K_a = 4$ about $\pm 3 \text{ cm}^{-1}$; for $K_a = 5$ about $\pm 7 \text{ cm}^{-1}$.

b If there is no entry under References, then the data are taken from Ref. (75).

c The estimated experimental uncertainty in the measured transitions are: for the microwave transitions quoted to two decimal places $\leq \pm 0.100 \text{ MHz}$, for millimeter wave transitions above 37 GHz quoted to two decimal places $< \pm 0.150 \text{ MHz}$, and for those quoted to three decimal places $< \pm 0.030 \text{ MHz}$.

d Note that one standard deviation is presented for the calculated uncertainty. These values should be multiplied by a factor of 2 to obtain $\approx 95\%$ confidence level.

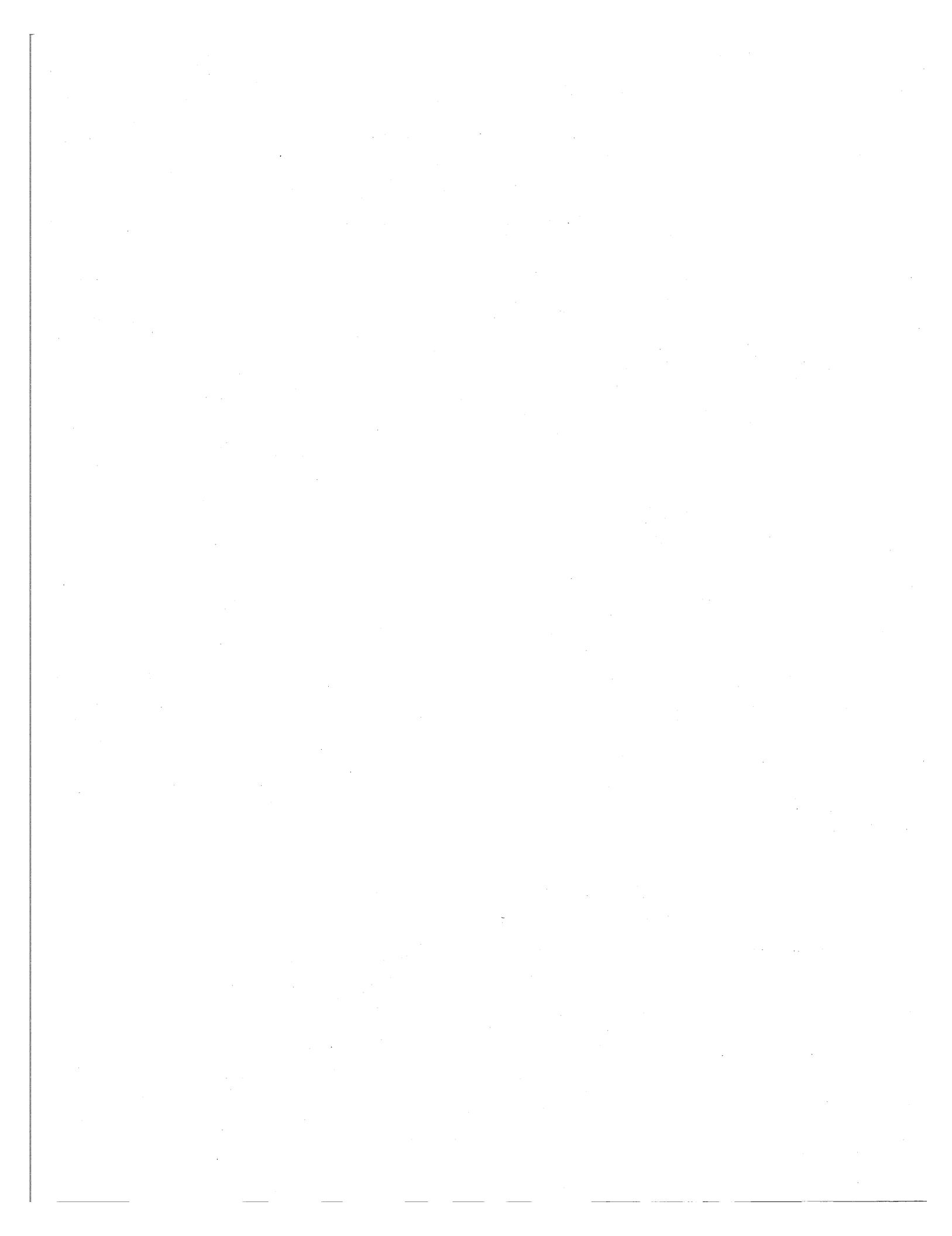


TABLE 6. Molecular Beam Measurements on $D^{14}N^{12}C^{16}O$

Transition	F' N	F' D	Frequency (kHz)	Ref.
$^1_{01} - ^0_{00}$	0		20 392314.60(50)	71
	2	3	20 393270.18(50)	71
	2	2	20 393281.93(50)	71
	1	1	20 393901.83(50)	71
	1	2	20 393909.20(50)	71
	1	0	20 393921.63(50)	71
eqQ_{D-N}	= 345(2) kHz			71
$\text{eqQ}({}^1_{01})$	= 2123.0 (1.0) kHz			71

TABLE 7. The Microwave Spectrum of Isocyanic Acid
 $H^{14}N^{13}C^{16}O$ (MHz)^c

Transition	(1) J' K' a' c'	(2) J'' K'' a'' c'' (Est. Uncert. ^a)	(3) Observed Calculated + Quadrupole shifts (Standard dev.) ^b
1 0 1 0 0 0			21982.200(5) -0.100(1)
r' = 0 F'' = 1		21982.00	
1 1 1 1 1 1		21982.70	0.517(5)
0 0 1 1 0 0		21981.15	-1.034(10)
2 0 2 1 0 1		43964.293(10)	
2 1 2 1 1 1		43799.974(5)	
2 1 1 1 1 0		44121.467(5)	
3 0 3 2 0 2		65946.174(15)	
3 1 3 2 1 2		65699.739(7)	
3 1 2 2 1 1		66181.971(7)	
3 2 2 2 2 1		65926.127(10)	
3 2 1 2 2 0		65926.207(10)	
4 0 4 3 0 3		87927.64	87927.736(19)
4 1 4 3 1 3		87599.12	87599.239(9)
4 1 3 3 1 2		88242.23	88242.199(9)
4 2 3 3 2 2			87901.094(13)
4 2 2 3 2 1			87901.293(13)
4 3 2 3 3 1			87870.058(17)
4 3 1 3 3 0			87870.058(17)
5 0 5 4 0 4		109908.95	109908.871(22)
5 1 5 4 1 4		109498.34	109498.384(11)
5 1 4 4 1 3		110301.98	110302.057(11)
5 2 4 4 2 3		109875.73	109875.709(16)
5 2 3 4 2 2		109876.08	109876.108(15)
5 3 4 3 0 3		109837.02	109836.955(20)
6 0 6 5 0 5			131889.474(24)
6 1 5 5 1 5			131397.087(13)
6 1 5 5 1 4			132361.456(13)
7 0 7 6 0 6		153869.52	152869.438(25)
7 1 7 6 1 6		153295.22	153295.259(16)
7 1 6 6 1 5		154420.46	154420.301(16)
7 2 6 6 2 5		153823.56	153823.536(24)
7 2 5 6 2 4		153824.66	153824.654(23)
7 3 6 3 0 3		153769.39	153769.435(27)
8 0 8 7 0 7		175848.57	175848.657(26)
8 1 8 7 1 7		175192.81	175192.811(21)
8 1 7 7 1 6		176478.51	176478.503(21)
8 2 6 7 2 5		175734.83	175734.852(30)
8 3 7 7 3 2		175646.34	175646.340(56)
8 4 7 7 4 2		175796.53	175796.573(30)
9 0 9 8 0 8		197827.07	197827.025(26)
9 1 9 8 1 8		197089.60	197089.653(27)
9 1 8 8 1 7		198536.00	198535.969(27)
9 2 8 8 2 7		197760.93	197760.900(38)
9 2 7 8 2 6		197771.34	197771.301(39)
9 3 7 8 3 3		197699.63	197699.611(34)
9 4 8 8 4 3			197600.003(63)
9 5 8 8 5 3		197459.20	197459.200(56)
10 0 10 9 0 9			219804.434(25)
10 1 10 9 1 9			219805.700(26)
10 1 9 9 1 8			220592.604(35)
11 0 11 10 0 10			241780.779(26)
11 1 11 10 1 10			240880.861(45)
11 1 10 10 1 9			242648.320(45)
1 1 0 1 1 1		160.749(0)	
F' = 1 F'' = 0		-0.637(8)	
1 1 1 1 1 1		-0.420(3)	
1 2 2 2 2 2		-0.276(3)	
2 2 1 1 1 1		0.055(1)	
0 0 1 1 1 1		0.200(3)	
		0.913(8)	
2 1 1 2 1 2		482.242(1)	
F' = 2 F'' = 2		-0.276(3)	
3 3 3 3 3 3		0.079(1)	
1 1 1 1 1 1		0.276(3)	
1 2 3 3 3 3		964.474(2)	
F' = 3 F'' = 3		-0.276(3)	
4 4 4 4 4 4		0.092(1)	
2 2 2 2 2 2		0.221(2)	
4 1 3 4 1 4		1607.433(4)	
F' = 4 F'' = 4		-0.276(3)	
5 5 5 5 5 5		0.100(1)	
3 3 3 3 3 3		0.197(2)	
5 1 4 5 1 5		2411.106(6)	
F' = 5 F'' = 5		-0.276(3)	

TABLE 7. Continued

	(1)	(2)	(3)
	6 1 5 6 1 6	6 1 5 6 1 6	0.106(1)
	4	4	0.184(2)
	6 1 6 7 1 7	F' = 6 F'' = 6	3375.474(8)
	7	7	-0.276(3)
	5	5	0.110(1)
	7 1 6 7 1 7	F' = 7 F'' = 7	0.176(2)
	8 1 7 8 1 8	F' = 8 F'' = 8	4500.517(10)
	9	9	-0.276(3)
	7	7	0.114(1)
	6	6	0.170(2)
	8 1 7 8 1 8	F' = 8 F'' = 8	5786.210(12)
	9	9	-0.276(3)
	8	8	0.116(1)
	7	7	0.162(2)
	9 1 8 9 1 9	F' = 9 F'' = 9	7232.525(14)
	10	10	-0.276(3)
	8	8	0.118(1)
	9	9	0.162(2)
	10 1 10 1 10	F' = 11 F'' = 11	8839.545(16)
	10	10	-0.276(3)
	9	9	0.160(1)
	11 1 10 1 11	F' = 12 F'' = 12	10606.890(18)
	11	11	-0.276(3)
	10	10	0.158(1)
	12 1 11 12 1 12	F' = 13 F'' = 13	12535.01
	12	12	-0.276(3)
	11	11	0.156(1)
	13 1 12 13 1 13	F' = 14 F'' = 14	14623.34
	13	13	-0.276(3)
	12	12	0.154(1)
	14 1 13 14 1 14	F' = 15 F'' = 15	16872.189(20)
	14	14	-0.276(3)
	13	13	0.153(1)
	15 1 14 15 1 15	F' = 16 F'' = 16	19281.57
	15	15	-0.276(3)
	14	14	0.152(1)
	16 1 15 16 1 16	F' = 17 F'' = 17	21851.019(20)
	16	16	-0.276(3)
	15	15	0.151(1)
	17 1 16 17 1 17	F' = 18 F'' = 18	24580.863(19)
	17	17	-0.276(3)
	16	16	0.150(1)
	18 1 17 18 1 18	F' = 19 F'' = 19	27471.05
	18	18	-0.276(3)
	17	17	0.150(1)
	19 1 18 19 1 19	F' = 20 F'' = 20	30521.25
	19	19	-0.276(3)
	18	18	0.149(1)
	20 1 19 20 1 20	F' = 21 F'' = 21	33731.53
	20	20	-0.276(3)
	19	19	0.149(1)
	21 1 20 21 1 21	F' = 22 F'' = 22	37101.635(39)
	21	21	-0.276(3)
	20	20	0.148(1)
	36 1 36 37 0 37	33658.44	33658.455(50)
	37 1 37 38 0 38	8861.31	8861.288(32)
	39 0 39 38 1 38	16001.52	16001.513(51)

^a The estimated experimental uncertainty in the measured transitions are: for the microwave transitions quoted to two decimal places $\pm \pm 0.100$ MHz, for millimeter wave transitions above 37 GHz quoted to two decimal places $\pm \pm 0.150$ MHz.

^b Note that one standard deviation is presented for the calculated uncertainty. These values should be multiplied by a factor of 2 to obtain $\sim 95\%$ confidence level.

^c The calculated hyperfine quadrupole shifts of all transitions of $HN^{13}C^{16}O$ agree to within ≤ 6 kHz with those calculated for $H^{14}N^{12}C^{16}O$ quoted in Table 4. Since this difference lies well within the accuracy of astronomical observations, quadrupole shifts are reproduced for certain transitions only, where splittings have been observed in the laboratory. Otherwise the reader is referred to Table 4.

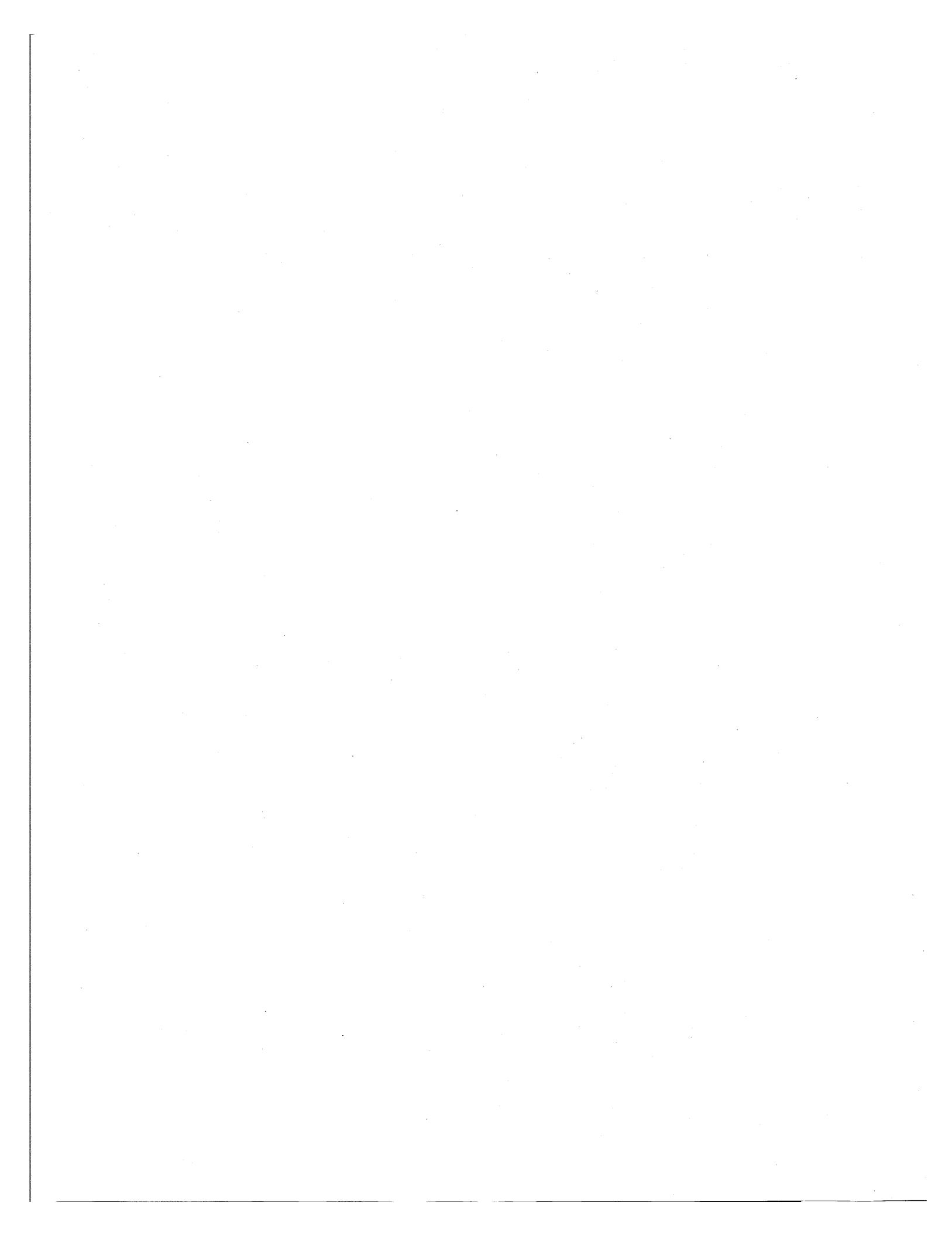


TABLE 8. The Microwave Spectrum of Isocyanic Acid
 $\text{H}^{15}\text{N}^{12}\text{C}^{16}\text{O}$ (MHz)

Transition J'	K_a'	K_c'	J''	K_a''	K_c''	Observed (Est. Uncert. ^a)	Calculated + Quadrupole shifts (Standard Dev.) ^b
1	0	1	0	0	0	21323.35	21323.341(20)
2	0	2	1	0	1	42646.563(35)	
2	1	2	1	1	1	42490.163(25)	
2	1	1	1	1	0	42794.889(25)	
3	0	3	2	0	2	63969.549(38)	
3	1	3	2	1	2	63735.038(28)	
3	1	2	2	1	1	64192.119(28)	
3	2	2	2	2	1	63947.377(49)	
3	2	1	2	2	0	63947.455(49)	
4	0	4	3	0	3	85292.129	85292.189(30)
4	1	4	3	1	3	84979.725	84979.670(23)
4	1	3	3	1	2	85589.122	85589.097(23)
4	2	3	3	2	2	85263.095	85262.932(37)
4	2	2	3	2	1	85263.095	85263.127(37)
4	3	3	3	3	3	85229.812	85229.807(39)
5	0	5	4	0	4	106614.42	106614.374(35)
5	1	5	4	1	4	106223.92	106223.986(24)
5	1	4	4	1	3	106985.75	106985.748(24)
5	2	4	4	2	3	106578.32	106578.294(37)
5	2	3	4	2	2	106578.684(37)	
5	3	4	3			106536.664(39)	
5	4	4	4			106459.960(1826)	
1	1	0	1	1	1	152.365(0)	
2	1	1	2	1	2	457.090(1)	
3	1	2	3	1	3	914.172(2)	
4	1	3	4	1	4	1523.599(3)	
5	1	4	5	1	5	2285.361(4)	
6	1	5	6	1	6	3199.443(5)	
7	1	6	7	1	7	4265.824(7)	
8	1	7	8	1	8	5484.485(9)	
9	1	8	9	1	9	6855.403(10)	
10	1	9	10	1	10	8378.53	8378.547(11)
11	1	10	11	1	11	10053.88	10053.889(13)
12	1	11	12	1	12	11881.40	11881.396(14)
13	1	12	13	1	13	13861.02	13861.028(14)
14	1	13	14	1	14	15992.74	15992.749(14)
15	1	14	15	1	15	18276.54	18276.514(14)
16	1	15	16	1	16	20712.25	20712.276(14)
17	1	16	17	1	17	23300.01	23299.988(13)
18	1	17	18	1	18	26039.61	26039.595(13)
19	1	18	19	1	19	28931.02	28031.042(15)
20	1	19	20	1	20	31974.31	31974.271(20)
21	1	20	21	1	21	35169.19	35169.218(27)
37	1	37	38	0	38	29522.18	29522.181(39)
38	1	38	39	0	39	5452.468(55)	
40	0	40	39	1	39	18684.61	18684.609(39)

^a The estimated experimental uncertainty in the measured transitions are: for the microwave transitions quoted to two decimal places ± 0.100 MHz, for millimeter wave transitions above 37 GHz quoted to two decimal places $< \pm 0.150$ MHz, and for those quoted to three decimal places $< \pm 0.030$ MHz.

^b Note that one standard deviation is presented for the calculated uncertainty. These values should be multiplied by a factor of 2 to obtain $\sim 95\%$ confidence level.

TABLE 9. The Microwave Spectrum of Isocyanic Acid
 $\text{H}^{14}\text{N}^{12}\text{C}^{18}\text{O}$ (MHz)

Transition J'	K_a'	K_c'	J''	K_a''	K_c''	(1)	(2)	(3)
						Observed Est. Uncert. ^a	Calculated + Quadrupole shifts (Standard dev.) ^b	
1	0	1	0	0	0	20798.03	20798.124(2)	
			$F' = 2$	$F'' = 1$		20798.64	-0.103(1)	
			1	1		20797.11	0.515(3)	
			0	1			-1.030(6)	
2	0	2	1	0	1	41596.132	41596.156(5)	
			$F' = 3$	$F'' = 2$		41596.725	-0.044(0)	
			2	2			0.618(4)	
			2	1		41596.132	0.000(0)	
			1	1		41595.142	-1.030(6)	
			1	0		41596.725	0.515(3)	
2	1	2	1	1	1	41449.198	41449.341(3)	
			$F' = 3$	$F'' = 2$		41449.840	-0.137(1)	
			2	2			0.374(2)	
			2	1		41449.840	0.515(3)	
			1	1			-0.279(3)	
			1	0		41448.666	-0.633(5)	
2	1	1	1	1	0	41736.575	41736.643(3)	
			$F' = 3$	$F'' = 2$		41736.757	-0.113(1)	
			3	2			0.038(2)	
			2	1		41737.193	0.515(3)	
			1	1			0.279(3)	
			1	0		41735.763	-0.912(5)	
3	0	3	2	0	2	62393.997	62394.001(7)	
			$F' = 4$	$F'' = 3$		62394.750	-0.025(0)	
			3	2			0.000(0)	
			2	1		62376.750	0.103(1)	
3	1	3	2	1	2	62173.751	62173.814(4)	
			$F' = 4$	$F'' = 3$		62173.927	-0.062(0)	
			3	2			0.129(1)	
			2	1		62173.751	-0.024(0)	
3	1	2	2	1	1	62604.693	62604.761(4)	
			$F' = 4$	$F'' = 3$		62604.872	-0.049(0)	
			3	2			0.129(1)	
			2	1		62604.693	-0.079(0)	
3	2	1	2	2	1	62376.095	62376.189(10)	
			$F' = 4$	$F'' = 3$		62376.750	-0.147(1)	
			3	2			0.515(3)	
			2	1		62375.710	-0.515(3)	
4	0	4	3	0	3	83191.568	83191.567(9)	
			4	1	3	82898.038	82898.052(5)	
			4	1	3	83472.641	83472.633(5)	
			4	2	3		83167.895(12)	
			4	2	2	83168.067(12)		
			4	3	2	83140.275(14)		
			4	3	0	83140.181	-0.149(1)	
			4	3	1	83140.765	0.464(3)	
			3	2		83139.922	-0.386(2)	
5	0	5	4	0	4	103988.774	103988.761(10)	
			5	1	4	103621.974	103621.975(7)	
			5	1	4	104340.185	104340.179(7)	
			5	2	4	103959.293	103959.290(14)	
			5	2	3	103959.620	103959.639(14)	
			5	3	3	103924.895	103924.895(16)	
			5	3	2	103924.751		
			5	4	4	103925.087	-0.085(1)	
			4	3	3	103924.751	0.322(1)	
5	4	4	4	4	4	103875.730	-0.163(1)	
			5	4	4	103876.331		
			4	3	3	103875.730	0.412(3)	
10	0	10	9	0	9	207966.18	207965.993(14)	
			10	1	10	207233.95	207234.123(19)	
			10	1	9	208670.16	208670.164(19)	
			10	2	9	207909.04	207909.050(14)	
			10	2	8	207911.90	207911.879(14)	



TABLE 9—Continued

(1)	(2)	(3)
1 1 0 1 1 1 F' = 1 F'' = 0		143.653(0) -0.633(5) -0.421(2) -0.279(3) 0.056(1) 0.197(2) 0.912(5)
1 1 2 1 2 F' = 2 F'' = 2		430.956(0) -0.279(3) 0.080(1) 0.279(3)
3 1 3 1 3 F' = 3 F'' = 3		861.901(1) -0.279(1) 0.093(1) 0.223(2)
4 1 3 4 1 4 F' = 4 F'' = 4		1436.482(1) -0.279(3) 0.102(1) 0.200(2)
5 1 4 5 1 5 F' = 5 F'' = 5		2154.688(2) -0.279(3) 0.107(1) 0.186(2)
6 1 5 6 1 6 F' = 6 F'' = 6		3016.501(3) -0.279(3) 0.112(1) 0.178(2)
7 1 6 7 1 7 F' = 7 F'' = 7		4021.905(4) -0.279(3) 0.115(1) 0.172(2)
8 1 7 8 1 8 F' = 8 F'' = 8		5170.879(5) -0.279(3) 0.118(1) 0.168(2)
9 1 8 9 1 9 F' = 9 F'' = 9		6463.400(5) -0.279(3) 0.120(1) 0.164(2)
10 1 9 10 1 10 F' = 10 F'' = 10		7899.442(6) -0.279(3) 0.121(1) 0.162(2)
11 1 10 11 1 11 F' = 12 F'' = 12	9479.12	9478.974(7) 0.123(1)
11 11 11	9478.71	-0.279(3)
10 10 10	9479.12	0.160(2)
12 1 11 12 1 12 F' = 13 F'' = 13	11202.11	11201.963(8) 0.124(1)
12 12 12	11201.69	-0.279(3)
11 11 11	11202.11	0.158(2)
13 1 12 13 1 13 F' = 14 F'' = 14		13068.375(8) 0.125(1)
13 13 13		-0.279(3)
12 12 12		0.156(2)
14 1 13 14 1 14 F' = 15 F'' = 15	15078.32	15078.170(8) 0.126(1)
14 14 14	15077.91	-0.279(3)
13 13 13	15078.32	0.155(2)
15 1 14 15 1 15 F' = 16 F'' = 16	17231.45	17231.307(8) 0.127(1)
15 15 15	17231.04	-0.279(3)
14 14 14	17321.45	0.154(2)
16 1 15 16 1 16 F' = 17 F'' = 17	19527.87	19527.742(8) 0.128(1)
16 16 16	19527.43	-0.279(3)
15 15 15	19527.87	0.153(2)
17 1 16 17 1 17 F' = 18 F'' = 18	21967.58	21967.425(8) 0.128(1)
17 17 17	21967.15	-0.279(3)
16 16 16	21967.58	0.152(2)
18 1 17 18 1 18 F' = 19 F'' = 19	24550.307(7)	0.129(1)
18 18 18		-0.279(3)
17 17 17		0.152(2)
19 1 18 19 1 19 F' = 20 F'' = 20	27276.46	27276.333(7) 0.129(1)
19 19 19	27276.03	-0.279(3)
18 18 18	27276.46	0.151(2)
20 1 19 20 1 20 F' = 21 F'' = 21	30145.62	30145.445(8) 0.130(1)
20 20 20	30145.18	-0.279(3)
19 19 19	30145.62	0.150(2)
21 1 20 21 1 21 F' = 22 F'' = 22	33157.74	33157.584(11) 0.130(1)
21 21 21	33157.32	-0.279(3)
20 20 20	33157.74	0.150(2)
22 1 21 22 1 22 F' = 23 F'' = 23	36312.81	36312.684(14) 0.131(1)
22 22 22	36312.41	-0.279(3)
21 21 21	36312.81	0.149(2)

TABLE 9—Continued

(1)	(2)	(3)
35 1 35 36 0 36	108908.082	108908.092(18)
37 1 37 38 0 38	62311.483	62311.474(11)
39 1 39 40 0 40	15475.45	15475.415(11)
41 0 41 40 1 40	8030.85	8030.824(11)
42 0 42 41 1 41	31595.01	31594.997(10)
44 0 44 43 1 43	78894.446	78894.452(11)
45 0 45 44 1 44	102628.354	102628.355(17)

^a The estimated experimental uncertainty in the measured transitions are: for the microwave transitions quoted to two decimal places ± 0.100 MHz, for millimeter wave transitions above 37 GHz quoted to two decimal places ± 0.150 MHz, and for those quoted to three decimal places ± 0.030 MHz.

^b Note that one standard deviation is presented for the calculated uncertainty. These values should be multiplied by a factor of 2 to obtain $\sim 95\%$ confidence level.

TABLE 10. Microwave transitions of $H^{14}N^{12}C^{16}O$
in order of frequency

Transition	Frequency (MHz)
1 1 0 1 1 1	160.432
2 1 1 2 1 2	481.293
3 1 2 3 1 3	962.576
4 1 3 4 1 4	1604.270
5 1 4 5 1 5	2406.361
6 1 5 6 1 6	3368.830
7 1 6 7 1 7	4491.659
8 1 7 8 1 8	5774.822
9 1 8 9 1 9	7218.291
10 1 9 10 1 10	8822.034
11 1 10 11 1 11	10586.015
12 1 11 12 1 12	11191.630
13 1 12 13 1 13	12510.196
14 1 13 14 1 14	13665.987
15 1 14 15 1 15	14594.535
16 1 15 16 1 16	16838.985
17 1 16 17 1 17	19243.496
18 1 17 18 1 18	21808.016
19 1 18 19 1 19	21981.574
20 1 19 20 1 20	24532.487
21 1 20 21 1 21	27416.848
22 1 21 22 1 22	30461.036
23 1 22 23 1 23	33664.981
24 1 23 24 1 24	35983.693
25 1 24 25 1 25	37028.612
26 0 40 39 1 39	38588.383
27 1 21 22 1 22	40551.853
28 1 22 23 1 23	43799.019
29 0 2 1 0 1	43963.042
30 1 21 22 1 22	44119.879
31 1 22 23 1 23	44234.625
32 1 23 24 1 24	48076.844
33 1 24 25 1 25	52078.422
34 1 35 36 0 36	60709.448
35 0 41 40 1 40	63574.773
36 1 36 37 0 37	65698.308
37 1 20 21 1 21	65924.118
38 2 21 22 1 22	65924.199
39 0 3 2 0 2	65944.298
40 1 2 2 1 1	66179.590
41 1 34 35 0 35	85368.158
42 0 42 41 1 41	87597.333
43 3 1 3 3 0	87867.280
44 2 3 2 3 1	87867.280
45 2 3 3 2 2	87898.416
46 2 2 3 3 2	87898.620
47 0 4 3 0 3	87925.238
48 1 3 3 1 2	88239.027
49 1 5 4 4 1	88624.352
50 1 4 4 4 1	109496.007
51 4 2 4 4 1	109778.700
52 4 1 4 4 0	109778.700
53 3 2 4 3 1	109833.489
54 3 3 4 3 2	109833.489

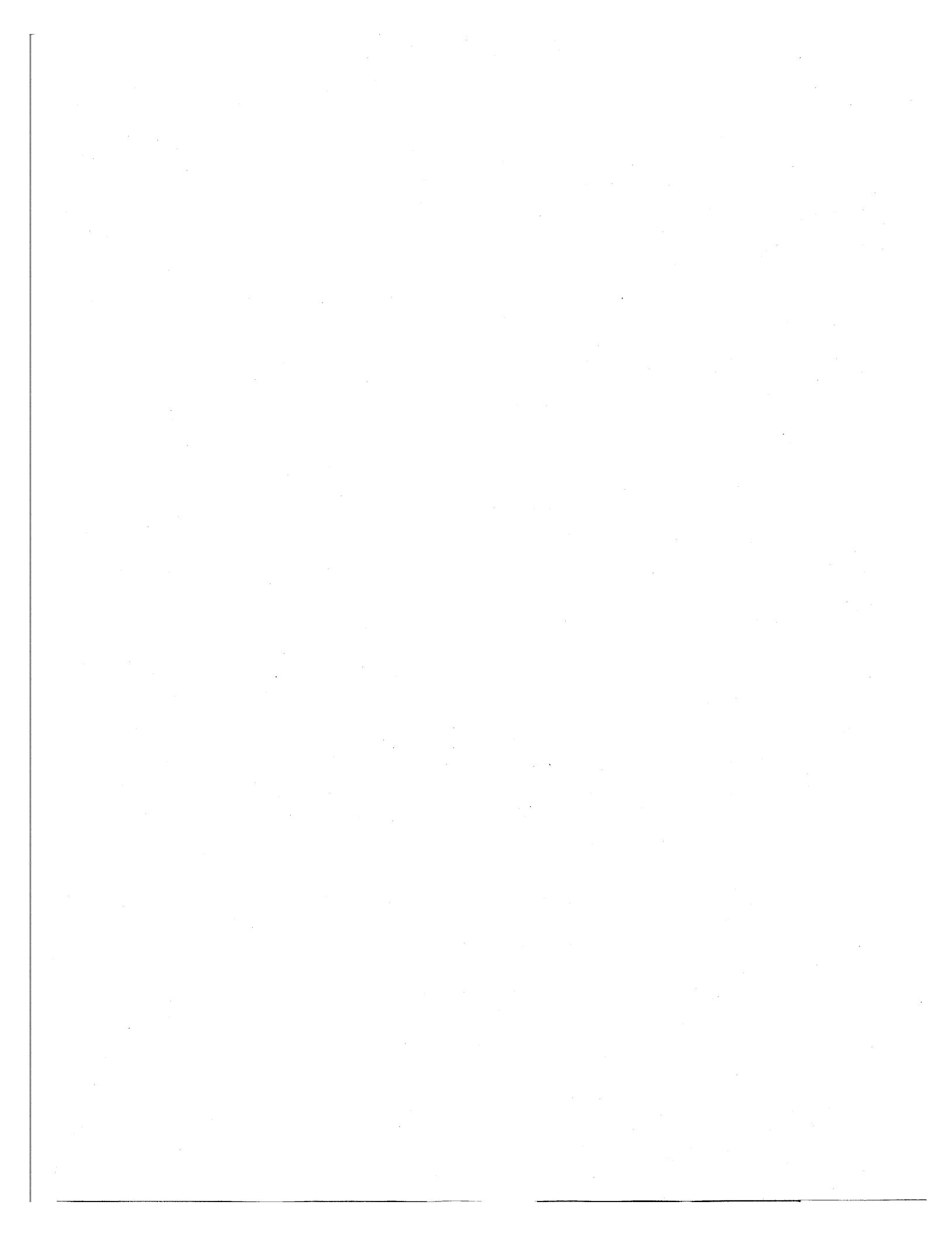


TABLE 10—Continued

5	2	4	4	2	3	109872.366
5	2	3	4	2	2	109872.773
5	0	5	4	0	4	109905.753
33	1	33	34	0	34	109959.100
5	1	4	4	1	3	110298.098
43	0	43	42	1	42	113736.298
6	1	6	5	1	5	131394.241
6	5	2	5	5	1	131640.747
6	5	1	5	5	0	131640.747
6	4	2	5	4	1	131733.534
6	4	3	5	4	2	131733.534
6	3	3	5	3	2	131799.292
6	3	4	5	3	3	131799.292
6	2	5	5	2	4	131845.880
6	2	4	5	2	3	131846.590
6	0	6	5	0	5	131885.740
6	1	5	5	1	4	132356.711
32	1	32	33	0	33	134481.571
44	0	44	43	1	43	138909.772
7	1	7	6	1	6	153291.946
7	5	3	6	5	2	153579.619
7	5	2	6	5	1	153579.619
7	4	3	6	4	2	153687.873
7	4	4	6	4	3	153687.873
7	3	4	6	3	3	153764.606
7	3	5	6	3	4	153764.606
7	2	6	6	2	5	153818.869
7	2	5	6	2	4	153820.007
7	0	7	6	0	6	153865.092
7	1	6	6	1	5	154414.776
31	1	31	32	0	32	158934.883
48	0	45	44	1	44	164143.924
8	1	8	7	1	7	175189.037
8	5	4	7	5	3	175517.913
8	5	3	7	5	2	175517.913
8	4	4	7	4	3	175641.637
8	4	5	7	4	4	175641.637
8	3	6	7	3	5	175729.350
8	3	5	7	3	4	175729.351
8	2	7	7	2	6	175791.248
8	2	6	7	2	5	175792.954
8	0	8	7	0	7	175843.703
8	1	7	7	1	6	176472.199
30	1	30	31	0	31	183319.365
46	0	46	45	1	45	189437.882
9	1	9	8	1	8	197085.424
9	5	5	0	5	4	197455.547
9	5	4	8	5	3	197455.547
9	4	5	8	4	4	197594.741
9	4	6	8	4	5	197594.741
9	3	7	8	3	6	197693.444
9	3	6	8	3	5	197693.444

TABLE 10—Continued

9	2	8	8	2	7	197762.928
9	2	7	8	2	6	197765.366
9	0	9	8	0	8	197821.469
9	1	8	8	1	7	198528.892
29	1	29	30	0	30	207631.365
47	0	47	46	1	46	214790.761
10	1	10	9	1	9	218981.019
10	5	6	9	5	5	219392.437
10	5	5	9	5	4	219392.437
10	4	6	9	4	5	219547.105
10	4	7	9	4	6	219547.105
10	3	8	9	3	7	219656.805
10	3	7	9	3	6	219656.805
10	2	9	9	2	8	219733.824
10	2	8	9	2	7	219737.175
10	0	10	9	0	9	219798.282
10	1	9	9	1	8	220584.762
28	1	28	29	0	29	231873.247
11	1	11	10	1	10	240875.735
11	3	9	10	3	8	241619.351
11	3	8	10	3	7	241619.353
11	2	10	10	2	9	241703.846
11	2	9	10	2	8	241708.315
11	0	11	10	0	10	241774.037
11	1	10	10	1	9	242639.717
27	1	27	28	0	28	256043.392
12	1	12	11	1	11	262769.484
12	3	10	11	3	9	263581.003
12	3	9	11	3	8	263581.006
12	2	11	11	2	10	263672.909
12	2	10	11	2	9	263678.717
12	0	12	11	0	11	263748.630
12	1	11	11	1	10	264693.665
26	1	26	27	0	27	280141.198
13	1	13	12	1	12	284662.177
13	3	11	12	3	10	285541.677
13	3	10	12	3	9	285541.681
13	2	12	12	2	11	285640.924
13	2	11	12	2	10	285648.316
13	0	13	12	0	12	285721.952
13	1	12	12	1	11	286746.515
1	1	0	1	0	1	901799.903(0.135)
2	1	1	2	0	2	901956.740(0.133)
3	1	2	3	0	3	902192.032(0.131)
4	1	3	4	0	4	902505.821(0.128)
5	1	4	5	0	5	902898.166(0.125)
6	1	5	6	0	6	903369.137(0.121)
7	1	6	7	0	7	903918.820(0.116)
8	1	7	8	0	8	904547.316(0.111)
9	1	8	9	0	9	905254.739(0.105)
10	1	9	10	0	10	906041.219(0.099)