

Estimating Solid-Liquid Phase Change Enthalpies and Entropies

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A group additivity method based on molecular structure is described that can be used to estimate solid-liquid total phase change entropy ($\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$) and enthalpy ($\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$) of organic molecules. The estimation of these phase changes is described and numerous examples are provided to guide the user in evaluating these properties for a broad range of organic structures. A total of 1858 compounds were used in deriving the group values and these values are tested on a database of 260 additional compounds. The absolute average and relative errors between experimental and calculated values for these 1858 compounds are $9.9 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ and $3.52 \text{ kJ}\cdot\text{mol}^{-1}$, and 0.154 and 0.17 for $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$, respectively. For the 260 test compounds, standard deviations of $\pm 13.0 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ ($\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$) and $\pm 4.88 \text{ kJ}\cdot\text{mol}^{-1}$ ($\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$) between experimental and calculated values were obtained. Estimations are provided for both databases. Fusion enthalpies for some additional compounds not included in the statistics are also included in the tabulation. © 1999 American Institute of Physics and American Chemical Society. [S0047-2689(99)00106-3]

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1. Introduction

1.1. Fusion Enthalpies

Fusion enthalpy is an important physical property of the solid state. The magnitude of the fusion enthalpy influences solute solubility in both an absolute sense and in its temperature dependence. This property plays an important factor in determining molecular packing in crystals and can be useful in correcting thermochemical data to a standard state when combined with other thermodynamic properties.

The discrepancy in numbers between the many new organic solids prepared and the few thermochemical measurements reported annually has encouraged the development of empirical relationships that can be used to estimate properties such as fusion enthalpy. We have found that techniques for estimating fusion enthalpies can play several useful roles.¹⁻³ Perhaps most importantly, they provide a numerical value that can be used in cases when there are no experimental data. Estimations are also useful in selecting the most probable experimental value in cases where two or more values are in significant disagreement. Given the choice between an estimated or experimental value, selection of the experimental value is clearly preferable. However, large discrepancies between estimated and calculated values can also identify systems exhibiting dynamic or associative properties. Some molecular systems exhibit phase transitions that occur in the solid state that are related to the onset of mo-

lecular motion. Others, such as liquid crystals exhibit nonisotropic molecular motion in the liquid phase.⁴ Both have associated with these phenomena, additional phase transitions that attenuate the enthalpy and entropy associated with fusion. A large positive discrepancy in the difference between estimated and experimentally measured fusion enthalpy is a good indication of this behavior.

1.2. Fusion Entropies

Very few general techniques have been developed for directly estimating fusion enthalpies, in part, as a consequence of the complex phase behavior exhibited by some compounds. Fusion enthalpies have been most frequently calculated from fusion entropies and the experimental melting temperature of the solid T_{fus} . One of the earliest estimation techniques is the use of Walden's Rule.⁵ The application of Walden's Rule provides a remarkably good approximation of $\Delta_{\text{fus}}H_m$, if one considers that the estimation is independent of molecular structure and based on only two parameters. Recent modifications of this rule have also been reported.^{6,7} Walden's Rule:

$$\Delta_{\text{fus}}H_m(T_{\text{fus}})/T_{\text{fus}} \approx 13 \text{ cal} \cdot \text{K}^{-1} \cdot \text{mol}^{-1} \\ = 54.4 \text{ J mol}^{-1} \text{K}^{-1}. \quad (1)$$

A general method for estimating fusion entropies based on the principles of group additivity has been reported recently.⁸⁻¹⁰ This method has been developed from the assumption that unlike fusion enthalpy and entropy, the total phase change entropy associated in going from a rigid solid at 0 K to an isotropic liquid at the melting point, T_{fus} , is a group property and that this property can be estimated by standard group additivity techniques. The total phase change entropy has been defined as the sum of the entropy associated with all the phase changes occurring in the condensed phase prior to and including melting. The assumption that the total phase change entropy is a more reliable group property than fusion entropy is readily apparent from an examination of these two properties as a function of the number of methylene groups for the *n*-alkanes. This is illustrated in Figs. 1 and 2. Many alkanes have additional phase transitions with significant entropy components that influence the magnitude of the fusion entropy. This leads to the nonlinear behavior illustrated in Fig. 1. When these components are added together, the total phase change entropy shows a much better linear correlation. Some odd-even alternation as a function of the number of carbon atoms is evident similar to what is observed in the melting points of these compounds

2. Estimation of Total Phase Change Entropy and Enthalpy

2.1. Derivation of Group Values

Initial group values for a methyl and methylene group were derived from the intercept (one half the intercept) and

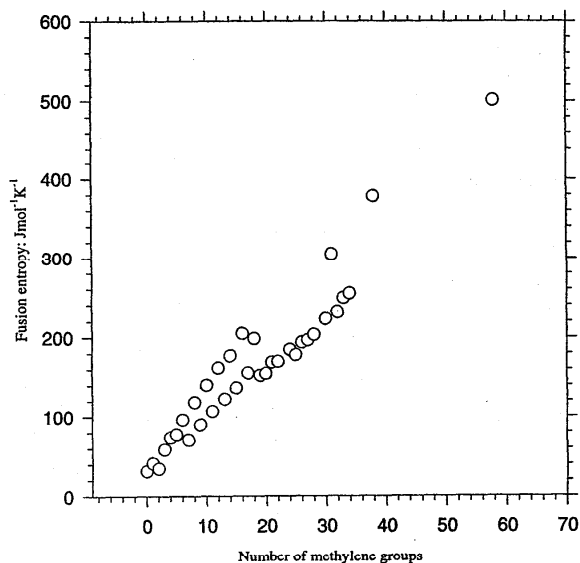


FIG. 1. Fusion entropy of the *n*-alkanes as a function of the number of methylene groups.

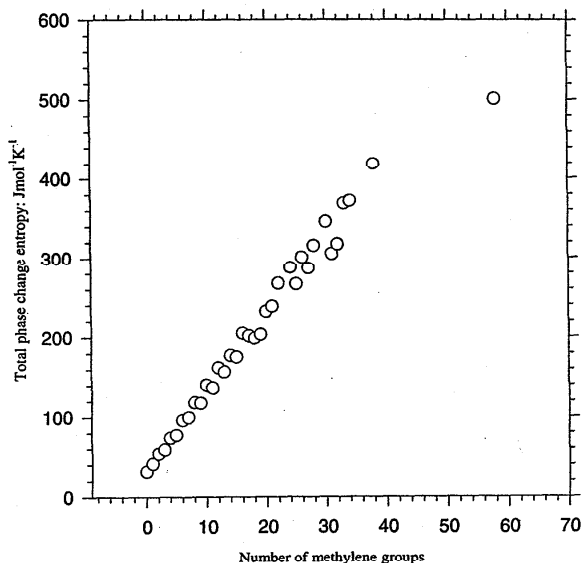


FIG. 2. Total phase change entropies of the *n*-alkanes as a function of the number of methylene groups.

slope of the line of Fig. 2, respectively. Group values for carbon in other common environments were initially derived from experimental data of compounds with appropriate structures using these two group values. Subsequent refinements were possible as additional experimental data became available. Once values were assigned for most carbon groups, these values were allowed to vary until the value of the function:

$$\sum_{i=1}^n [\Delta_0^{T_{\text{fus}}} S(\text{expt}) - \Delta_0^{T_{\text{fus}}} S(\text{calcd})]^2$$

did not change significantly upon successive iterations. Group values for the functional groups were derived in a similar fashion. Using group values for carbon established from the hydrocarbons, values for the functional groups in Tables 1 and 2 were derived. Once initial values for these groups were established, a similar least squares minimization of all the values were performed.

The total phase change entropy, $\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$, in most cases provides a good estimate of the entropy of fusion, $\Delta_{\text{fus}} S_m(T_{\text{fus}})$. If there are no additional solid phase transitions then $\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ becomes numerically equal to $\Delta_{\text{fus}} S_m(T_{\text{fus}})$. From the experimental melting point and $\Delta_{\text{fus}} S_m(T_{\text{fus}})$, it is possible to approximate the total phase change enthalpy, $\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$. Similarly, if there are no additional phase transitions then the total phase change enthalpy, $\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$, becomes numerically equivalent to the fusion enthalpy, $\Delta_{\text{fus}} H_m(T_{\text{fus}})$.

A listing of the group parameters that can be used to estimate these phase change properties is presented in Tables 1 and 2. The group values in these tables have been updated from previous versions by the inclusion of new experimental data in the parameterizations.^{8,9} Before describing the appli-

cation of these parameters in the estimation of $\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$, the conventions used to describe these group values need to be defined. Primary, secondary, tertiary, and quaternary centers, as found on atoms of carbon, silicon, and their congeners, are defined solely on the basis of the number of hydrogens attached to the central atom, 3, 2, 1, 0, respectively. It should be noted that the experimental melting point along with an estimated value of $\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ is necessary to estimate the fusion enthalpy of a compound. In addition, compounds whose liquid phase is not isotropic at the melting point are not modeled properly by these estimations. Those compounds forming liquid crystal or cholesteric phases as well amphiphilic compounds are currently overestimated by these parameters. A large discrepancy between the estimated total phase change enthalpy and experimental fusion enthalpy is a good indication of undetected solid-solid phase transitions or anisotropic liquid behavior.

The parameters used for estimating $\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ of hydrocarbons and the hydrocarbon portions of more complex molecules are listed in Table 1. The group value, G_i , associated with a molecular fragment is identified in the third column of the table. The group coefficients, C_i , are listed in column 4 of the table. These group coefficients are used to modify G_i whenever a functional group is attached to the carbon in question. Functional groups are defined in Table 2. Group values reported in parenthesis are based on only a limited database (arbitrarily chosen as less than seven entries) and should be considered as tentative assignments. All values of C_i and C_k that are not specifically defined in Tables 1 and 2 are to be assumed equal to 1.0. The group coefficient for a methylene group in Table 1, C_{CH_2} , is applied differently from the rest and its application is discussed below. Introduction of this coefficient is new and differentiates this pro-

to be used from earlier versions. The application of this group coefficient as well as the entire protocol is illustrated in the examples given in Tables 3 and 4.

3. Estimations of Hydrocarbons

3.1. Acyclic and Aromatic Hydrocarbons

Estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for acyclic and aromatic hydrocarbons (*aah*) can be achieved by summing the group values consistent with the structure of the molecule as illustrated by the following equation:

$$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(aah) = \sum_i n_i G_i + n_{\text{CH}_2} C_{\text{CH}_2} G_{\text{CH}_2};$$

$$C_{\text{CH}_2} = 1.31 \text{ when } n_{\text{CH}_2} \geq \sum n_i;$$

$$i \neq \text{CH}_2 \text{ otherwise } C_{\text{CH}_2} = 1. \quad (2)$$

The group coefficient for a methylene group C_{CH_2} is used whenever the total number of consecutive methylene groups in a molecule n_{CH_2} equals or exceeds the sum of the other remaining groups $\sum n_i$. This applies to both hydrocarbons and all derivatives. In oligomers, and polymers, the decision as to whether to include this group coefficient should be based on the structure of the repeating unit. Some examples illustrating the use of both the groups in Table 1(a) and Eq. (2) are given in Table 3 and additional discussion regarding the use of C_{CH_2} is provided in the discussion that pertains to polymers below. Entries for each estimation in Table 3 include the melting point T_{fus} and all transition temperatures T_i for which there is a substantial enthalpy change. The estimated and experimental (in parentheses) phase change entropies follow. Similarly, the total phase change enthalpy calculated as the product of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and T_{fus} is followed by the experimental total phase change enthalpy (or fusion enthalpy). Finally, details in estimating $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for each compound are included as the last entry for each compound.

3.1.1. Styrene

The estimation of the fusion entropy of styrene is an example of an estimation of a typical aromatic hydrocarbon. Identification of the appropriate groups in Table 1(a) results in an entropy of fusion of $52.5 \text{ J} \cdot \text{mol}^{-1} \cdot \text{K}^{-1}$ and together with the experimental melting point, an enthalpy of fusion of $12.6 \text{ kJ} \cdot \text{mol}^{-1}$ is estimated. This can be compared to the experimental value of $11.0 \text{ kJ} \cdot \text{mol}^{-1}$. It should be pointed out that the group values for aromatic molecules are purely additive while the group values for other cyclic sp^2 atoms, summarized in Table 1(b), are treated as corrections to the ring equation. This will be discussed in more detail below.

3.1.2. 1-Heptene

The fusion entropy of 1-heptene is obtained in a similar fashion. In this case, the number of consecutive methylene groups in the molecule exceeds the sum of the remaining

terms in the estimation and this necessitates the use of the group coefficient C_{CH_2} of 1.31. For a molecule such as 3-heptene (estimation not shown), the group coefficient of 1.31 would not be applied. For a molecule such as 3-decene (also not shown), the group coefficient of 1.31 would be applied only to the five consecutive methylene groups. The remaining methylene group at carbon 2 would be treated normally ($C_{\text{CH}_2} = 1.0$) and would not be counted in $\sum n_i$.

3.1.3. Perylene

Estimation of the phase change entropy of perylene provides an example of a molecule containing both peripheral and internal quaternary sp^2 carbon atoms adjacent to an sp^2 atom. The carbon atoms in graphite are another example of internal quaternary sp^2 carbon atoms. In the application of these group values to obtain the phase change properties of other aromatic molecules, it is important to remember that the aromatic portion of a molecule is defined in these estimations as molecules containing only benzenoid carbons and the corresponding nitrogen heterocycles. While a molecule like 1,2-benzacenaphthene (fluoranthene) would be considered aromatic, the five membered ring in acenaphthylene, according to this definition is not. Estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for acenaphthylene will be illustrated below.

3.2. Nonaromatic Cyclic and Polycyclic Hydrocarbons

The protocol established for estimating $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of unsubstituted cyclic hydrocarbons uses Eq. (3) to evaluate this term for the parent cycloalkane, $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring})$. For substituted and polycyclic cycloalkanes,

$$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring}) = [33.4] + [3.7][n - 3];$$

$$n = \text{number of ring atoms}, \quad (3)$$

$$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring}) = [33.4]N + [3.7][R - 3N];$$

$$R = \text{total number of ring atoms}; N = \text{number of rings}, \quad (4)$$

the results of Eqs. (3) or (4), respectively, are then corrected for the presence of substitution and hybridization patterns in the ring that differ from the standard cyclic secondary sp^3 pattern found in the parent monocyclic alkanes, $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr})$. These correction terms can be found in Table 1(b). Once these corrections are included in the estimation, any additional acyclic groups present as substituents on the ring are added to the results of Eqs. (3) or (4) and $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr})$. These additional acyclic and/or aromatic terms [$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(aah)$] are added according to the protocol discussed above in the use of Eq. (2). The following ex-

amples of Table 3 illustrate the use of Eq. (3) and (4) according to Eq. (5) to estimate the total phase change entropy of cyclic molecules $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{total})$:

$$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{total}) = \Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring}) + \Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr}) + \Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{aah}) \quad (5)$$

3.2.1. 10,10,13,13-Tetramethyl-1,5-cyclohexadecadiyne

The estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for 10,10,13,13-tetramethyl-1,5-cyclohexadecadiyne illustrates the use of Eq. (5) for a monocyclic alkyne. Once the hexadecane ring is estimated ($[33.4] + 13[3.7]$), correcting for the presence of two cyclic quaternary sp^3 carbon atoms ($2[-34.6]$), four cyclic sp carbon atoms ($4[-4.7]$) and four methyl groups ($4[17.6]$) completes this estimation.

3.2.2. Bullvalene

Bullvalene, a tricyclic hydrocarbon, provides an example of the use of Eqs. (4) and (5). The minimum number of bonds that need to be broken to form a completely acyclic molecule is used to determine the number of rings. In this case it is three. Application of Eq. (4) to bullvalene [$3[33.4] + 3.7[10-9]$] provides $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring})$. Addition of the contributions of the four cyclic tertiary sp^3 carbons and the six tertiary sp^2 carbons to the results of Eq. (4), $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr})$, completes the estimation.

3.2.3. Acenaphthylene

Estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ for acenaphthylene completes this section on cyclic hydrocarbons. Molecules that contain rings fused to aromatic rings but are not completely aromatic, according to the definition provided above, are estimated by first calculating $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring})$ for the contributions of the nonaromatic ring according to Eqs. (3) or (4). The atoms of the nonaromatic ring should be selected on the basis of the smallest number of ring atoms that account for all the nonaromatic carbons. This is then followed by addition of the adjustments for the nonsecondary sp^3 ring carbons, the contributions of the remaining aromatic groups and any other substituents that may be present. In acenaphthylene, the contribution of the five membered ring $\{\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring}): [33.4] + 2[3.7]\}$ is first adjusted for each nonsecondary sp^3 carbon atom in the ring $\{\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr}): +2[-1.6] + 3[-12.3]\}$, and then the remainder of the aromatic portion of the molecule is added $\{\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{aah}): [-7.5] + 6[7.4]\}$. In a molecule such as [2,2]-meta-cyclophane (estimation not shown), the acyclic ring the chosen to contain the fewest ring atoms, ten carbons in this instance. The six aromatic ring atoms that make up a portion of the ten membered ring are considered as cyclic sp^2 carbon atoms (four quaternary sp^2 and two tertiary sp^2

carbons). Addition of the contributions of the six remaining aromatic tertiary carbon atoms not included in the aliphatic ring completes this estimation.

4. Estimations of Hydrocarbon Derivatives

Estimations involving derivatives of hydrocarbons are performed in a fashion similar to hydrocarbons. The estimation consists of three parts: the contribution of the hydrocarbon component, that of the carbon(s) bearing the functional group(s), $\sum_i n_i C_i G_i$, and the contribution of the functional group(s) $\sum_k n_k C_j G_k$. The symbols n_i , n_k refer to the number of groups of type i and k . Acyclic and cyclic compounds are treated separately as before. For acyclic and aromatic molecules, the hydrocarbon portion is estimated using Eq. (2); cyclic or polycyclic molecules are estimated using Eqs. (3) and (4), respectively. Similarly, the contribution of the carbon(s) bearing the functional group(s) is evaluated from Tables 1 (a) or 1(b) modified by the appropriate group coefficient C_i as will be illustrated below. The group values of the functional groups G_k are listed in Tables 2(a) and 2(b). The corresponding group coefficient C_j is equal to one for all functional groups except those identified otherwise in Table 2(a). Selection of the appropriate value of C_j from Table 2(a) is based on the total number of functional groups and is discussed below. Functional groups that make up a portion of a ring are listed in Table 2(b). The use of these values in estimations will be illustrated separately. Equations (6) and (7) summarize the protocol developed to estimate $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{total})$ for acyclic and aromatic derivatives and for cyclic and polycyclic hydrocarbon derivatives, respectively,

$$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{total}) = \Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{aah}) + \sum_i n_i C_i G_i + \sum_k n_k C_j G_k, \quad (6)$$

$$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{total}) = \Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{ring}) + \Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr}) + \sum_i n_i C_i G_i + \sum_k n_k C_j G_k, \quad (7)$$

where

$$C_j = \sum_k n_k.$$

In view of the large number of group values listed in Tables 2(a) and 2(b), selection of the appropriate functional group(s) is particularly important. Four functional groups in Table 2(a), chlorine, the hydroxyl and carboxyl group, and tri-substituted amides are dependent on the total substitution pattern in the molecule. Coefficients for these four groups C_j are available for molecules containing up to six functional groups. Selection of the appropriate value of C_j for one of these four functional groups is based on the total number of functional groups in the molecule. Estimations of the fusion entropy of polymers suggests that the group coefficient for C_6 in Table 2(b), is adequate for molecules containing more than a total of six functional groups.¹⁹

4.1. Acyclic and Aromatic Hydrocarbon Derivatives

The estimations for decachlorobiphenyl, N-acetyl-L-alanine amide, 2,2,2-trifluoroacetonitrile, and isoquinoline, shown in Table 4(a), illustrate the estimations of substituted aromatic and acyclic hydrocarbon derivatives.

4.1.1. Decachlorobiphenyl

Decachlorobiphenyl is an example of an estimation of a polysubstituted aromatic molecule. Selection of the value for a quaternary aromatic sp^2 carbon from Table 1(a) depends on the nature of the functional group attached to carbon. If the functional group at the point of attachment is sp^2 hybridized or contains nonbonding electrons, the value for a "peripheral aromatic sp^2 carbon adjacent to an sp^2 atom" is selected. Otherwise a "peripheral aromatic sp^2 carbon adjacent to an sp^3 atom" is used. The remainder of the estimation follows the guidelines outlined above with the exception that chlorine is one of the four functional groups whose group coefficient C_j depends on the degree of substitution (C_6 is used in this example).

4.1.2. N-acetyl-L-alanine amide

The estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for N-acetyl-L-alanine amide follows directly from Eq. (6). The molecule contains both a primary and secondary amide linkage. The asymmetric center is a tertiary carbon that contains two functional groups attached to it and as such its contribution is attenuated by the group coefficient for a tertiary carbon. Addition of the contributions of the two methyl groups completes the estimation.

4.1.3. Trifluoroacetonitrile

The estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for trifluoroacetonitrile illustrates an example of a molecule containing fluorine. The group value for a fluorine on a trifluoromethyl group in Table 2(a) is given per fluorine atom. The contribution of the quaternary carbon atom when attached to functional groups is attenuated by the group coefficient C_i . Inclusion of the group value for a thiol completes this estimation. When fluorine is combined with the functional groups listed in Table 2(b), the group coefficient chosen should be based on the presence of fluorine as a single functional group, regardless of the number of fluorine atoms present. For example, a molecule such as trifluoromethanol would be considered to contain two functional groups.

4.1.4. Isoquinoline

The estimation of isoquinoline illustrates an example of another aromatic molecule. The only exception in this case is the need to substitute the group value for a heterocyclic aromatic amine. Otherwise the same protocol is followed as in the estimation of naphthalene (not shown).

4.2. Cyclic and Polycyclic Hydrocarbon Derivatives

The protocol for estimating the total phase change properties of cyclic and polycyclic molecules also follows from the procedure described above for the corresponding cyclic hydrocarbons. In cyclic molecules, the substituent or functional group may be attached to the ring or it may be part of the ring. If the functional group is part of the ring, the group values listed in Table 2(b) are to be used. The procedure first involves estimating $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for the corresponding hydrocarbon ring, then correcting for the heterocyclic component(s), and if necessary, correcting the ring carbons attached to the cyclic functional group by the appropriate group coefficients. This is illustrated in Table 4(b) by the following examples.

4.2.1. 2-Chlorodibenzodioxin

The dioxane ring of 2-chlorodibenzodioxin is treated as being a derivative of cyclohexane. According to Eq. (7), the ring equation is first used to estimate the contributions of the cyclohexane ring. This ring contains two cyclic ether oxygens and four quaternary cyclic sp^2 carbon atoms and must be modified accordingly. The remaining eight carbon atoms are treated as aromatic carbons and values appropriate to their substitution pattern are chosen. The addition of the contribution of the chlorine completes the estimation.

4.2.2. 6,8,9-Trimethyladenine

6,8,9-Trimethyladenine is estimated in a similar fashion. The ring equation [Eq. (3)] is used first to generate the contribution of the five membered heterocyclic ring. In this instance the ring has been modified by the addition of a cyclic sp^2 hybridized nitrogen atom and a nitrogen which comprises part of a cyclic tertiary amine. Both ring substitutions require appropriate corrections. The hybridization and substitution of the remaining three cyclic carbon atoms of the five membered ring have likewise been changed from the pattern found in cyclopentane and appropriate changes must also be included in $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}(\text{corr})$. The remaining four ring atoms comprise a portion of an aromatic ring; their contributions can be added directly. The two nitrogen atoms make up a portion of the heterocyclic aromatic ring along with a quaternary and tertiary aromatic sp^2 carbon atom. The quaternary aromatic sp^2 carbon atom is attached to an exocyclic nitrogen atom with a lone pair of electrons and consequently, the quaternary aromatic carbon is treated as being adjacent to an sp^2 center. The contributions of the tertiary aromatic sp^2 carbon atom, the methyl groups, and the acyclic secondary amine complete the estimation.

4.2.3. Lenacil

Estimations of Lenacil (3-cyclohexyl-6,7-dihydro-1H-cyclopentapyrimidine-2,4-(3H,5H)-dione) require some thought in properly identifying the functional groups in the molecule. The functional group that makes up a portion of the pyrimidine-2,4-dione ring in this molecule cannot be

found directly in Table 2(b). It must therefore be simplified and this simplification can be accommodated in various ways. The ring can be considered to be a combination of either an adjacent cyclic imide ($-\text{CONRCO}-$) and cyclic amide nitrogen ($-\text{NH}-$), a cyclic urea ($-\text{NRCONH}-$) and amide carbonyl ($-\text{CO}-$), or a cyclic secondary and tertiary amide. An examination of the available groups in Table 2(b) will reveal that although group values for cyclic imides are available ($-\text{NRCONH}-$, $-\text{NRCONR}-$), there is no appropriate group available for an N-substituted cyclic nitrogen of an amide. Similarly, group values for a cyclic urea and amide carbonyl are not available. The most appropriate group values that are available are for cyclic amides. Once the appropriate group is identified, the procedure follows the same protocol as established for other polycyclic molecules.

4.2.4. Cortisone

The estimation of the fusion enthalpy of cortisone illustrates an example of an estimation of a complex polycyclic compound. This tetracyclic 17 atom ring system contains three cyclic quaternary centers ($3[-34.6]$), three cyclic tertiary sp^3 centers, ($3[-14.7]$), a cyclic tertiary sp^2 center which is attached to a functional group [1.92][-1.6], a quaternary sp^2 center ($[-12.3]$) as well as two cyclic carbonyl group ($2[-1.4]$). Addition of these modifications to the ring equation ($4[33.4]+5[3.7]$) estimates the contributions of the ring. Addition of the contributions of the substituents which include three hydroxyls ($(3)(12.1)[1.7]$), two methyls ($2[17.6]$), a methylene ($[7.1]$), and a carbonyl group of an acyclic ketone ($[4.6]$) completes the estimation. The molecule contains five functional groups, hence C_5 for a hydroxyl group is used.

4.3. Polymers

In addition to the estimation of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of small molecules, the parameters of Tables 1 and 2 can be used to predict $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ (when the experimental melting point is known) of crystalline oligomers and linear polymers. Since the parameters in Tables 1 and 2 differ slightly from those reported previously,¹⁹ the predictions of Eqs. (2)–(6) likewise produce slightly different results. However a similar overall correlation (slightly improved) between experimental and calculated results is obtained using the updated parameters. The protocol used to evaluate $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of polymers is exactly the same as outlined above. In this instance, the entropic value is calculated on the basis of the structure of the repeat unit of the polymer. Best correlations are obtained when the group coefficient C_k chosen is based on the number of functional groups present on the repeat unit and on the two nearest neighbors. The polymer $(\text{CH}_2\text{O})_n$, is treated as an infinite chain with $n_0 = n_{\text{CH}_2}$. For a molecule such as $\text{CH}_3\text{O}(\text{CH}_2\text{CH}_2\text{O})_n\text{CH}_3$, the number of methylene groups in the repeat unit exceeds the number of oxygens and therefore the group coefficient for a methylene group should be used. As n becomes smaller, a point will be reached when the molecule no longer represents an oligomer. In this in-

stance the group coefficient for a methylene group should be dropped. This should occur when n becomes less than the number of other groups that make up the remainder of the molecule. In the case just described, this would occur when n becomes less than three.

The column entries in Tables 6 and 7 are identical (these data were not used in generating the group values of Tables 1 and 2) and are described below. Calculated and experimental values of $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ for a series of linear polymers are provided in Table 6.

5. The Group Coefficient in Cycloalkyl Derivatives

The protocol in determining whether to use the group coefficient C_{CH_2} depends on whether the number of consecutive methylene groups exceeds the sum of the remaining groups excluding other methylene groups in the count. In an estimation of a cyclic derivative, the contribution of the ring is determined by Eq. (3) or (4) along with other terms necessary to correct for substitution and hybridization changes. This will vary depending on the nature of the ring and its substitution patterns. Fewer terms are necessary to estimate the total phase change entropy of ethylcyclohexane than ethylcyclohexadiene, even though in principle, both contain the same number of groups. To avoid any ambiguity in determining when to use this group coefficient, the number of groups associated with a ring structure should be determined by the size of the ring and the number of substituents or functional groups attached to the ring. For example, a molecule such as 2,5-di- n -undecyloxy-1,4-benzoquinone, contains 10 adjacent methylene groups. These methylene groups should be compared to the total number of other groups on the molecule. This includes two carbonyls, two methyl groups, two ether oxygens, and four sp^2 hybridized carbon atoms, adding up to a total of 10. Since these two numbers are equal, the group coefficient should be applied to both undecyl groups.

6. Polymorphism

In some cases, particularly with some pharmaceuticals, different fusion enthalpies and melting points have been reported for the same material. For example, fusion enthalpies of 18 284 (428.2 K)²⁰ and 23 810 J mol⁻¹ (430.3 K)²¹ have been reported for codeine. While one of these values may be in error, the two values may represent accurate physical properties of different crystalline modifications of codeine. The value estimated by the group additivity approach described above generally gives total phase change entropies and enthalpies associated with the most stable modification at the melting point. A recent review article summarizes pharmaceuticals known to exhibit polymorphism.²²

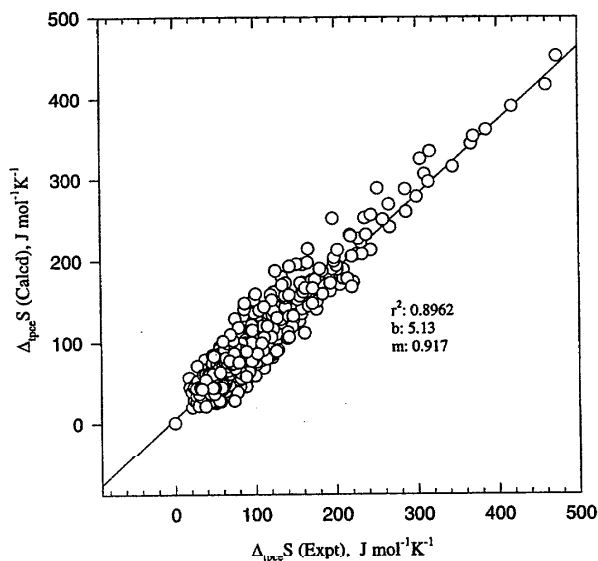


FIG. 3. A comparison of calculated and experimental $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of 1858 database compounds.

7. Statistics of the Correlation

7.1. Database Compounds

The group values included in Tables 1 and 2 were generated from the fusion entropies of a total of 1858 compounds. Melting and transition temperatures (column 1), experimental enthalpies associated with all solid–solid and solid–liquid phase changes (ΔH_{pce} , column 2), the corresponding phase change entropies (ΔS_{pce} , column 3), the total experimental phase change entropy (column 4) and enthalpy (columns 6), and the corresponding values estimated from the group values of Tables 1 and 4 (columns 5 and 7) for each of these compounds are given in Table 5. A summary of each calculation is also included in the form of the alphanumeric terms

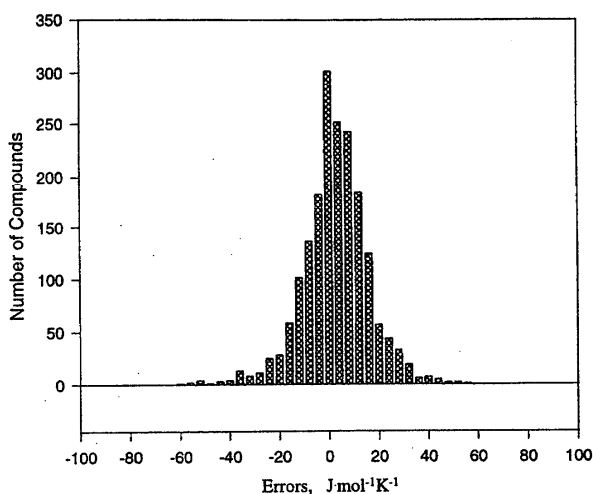


FIG. 4. A histogram illustrating the distribution of errors in estimating $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of the database compounds.

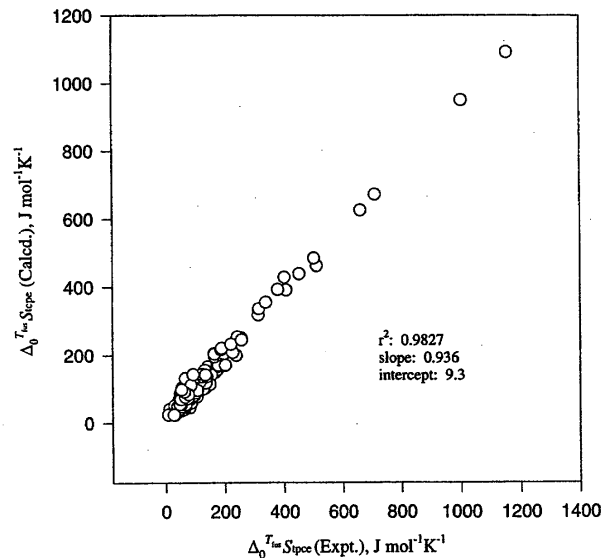


FIG. 5. A comparison of calculated and experimental $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of 260 test compounds.

used in each calculation. These alphanumeric terms are defined in Tables 1 and 2 for each group (in parenthesis). Table 5 also includes a number of compounds that were not included in deriving either the statistics or the group values. Reasons for this are noted in the table. An asterisk following the molecular formula in the table identifies these materials. Experimental and calculated total phase change entropies for the database are compared in Fig. 3. The correlation was characterized by the slope m , intercept b , and correlation coefficient (r^2) given in the figure. A histogram of the errors associated with this correlation is shown in Fig. 4. The absolute average and relative errors between experimental and

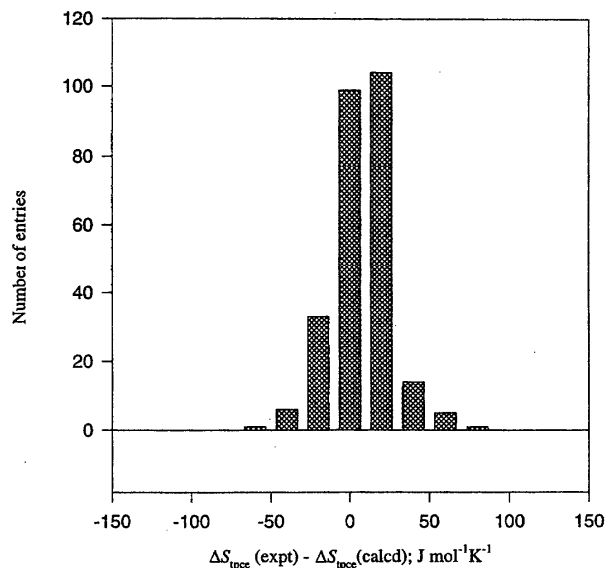


FIG. 6. A histogram illustrating the distribution of errors in estimating $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ of 260 test compounds.

calculated $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ values for these 1858 compounds are $9.9 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ and $3.52 \text{ kJ}\cdot\text{mol}^{-1}$, and 0.154 and 0.17, respectively. The standard deviations between experimental and calculated values for $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ are $\pm 13.0 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ and $\pm 4.88 \text{ kJ}\cdot\text{mol}^{-1}$, respectively. An additional 60 compounds exhibited errors exceeding 3 s.d. and were excluded from the correlations and from Figs. 3 and 4. These compounds are included in Tables 5 and 7.

7.2. Test Compounds

In addition to the 1858 compounds that make up the database, an additional 260 compounds have been used as test materials to provide an unbiased evaluation of the reliability of the group values given in Tables 1 and 2. These fusion enthalpies include compounds obtained from more recent

searches of the literature and are reported in Table 7. The data included in Table 7 are in the same format as the data in Table 5. The correlation between experimental and calculated values for the test compounds is shown in Fig. 5. The standard deviations between experimental and calculated values for $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ were $\pm 18.4 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ and $\pm 7.2 \text{ kJ}\cdot\text{mol}^{-1}$, respectively. The absolute average and relative errors between experimental and calculated $\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ values for these 260 compounds were $13.9 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ and $5.28 \text{ kJ}\cdot\text{mol}^{-1}$, and 0.181 and 0.194, respectively. In addition to these 260 compounds, some recently acquired data are also included in Table 7. As before, compounds not included in the correlations are identified by an asterisk following their molecular formula (see Tables 5, 6, and 7). References to Tables 5, 6, and 7 are listed in Table 8.

TABLE 1. (a) Contributions of the hydrocarbon portion of acyclic and aromatic molecules

Acyclic and aromatic carbon groups	Group value ^a G_i ($\text{J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$)	Group coefficients ^a C_i
primary sp^3	CH_3- 17.6 (A1)	
secondary sp^3	$>\text{CH}_2$ 7.1 (A2)	1.31 ^b (B2)
tertiary sp^3	$-\text{CH}<$ -16.4 (A3)	0.60 (B3)
quaternary sp^3	$>\text{C}<$ -34.8 (A4)	0.66 (B4)
secondary sp^2	$=\text{CH}_2$ 17.3 (A5)	
tertiary sp^2	$=\text{CH}-$ 5.3 (A6)	0.75 (B6)
quaternary sp^2	$=\text{C}(\text{R})-$ -10.7 (A7)	
tertiary sp	$\text{H}-\text{C}\equiv$ 14.9 (A8)	
quaternary sp	$-\text{C}\equiv$ -2.8 (A9)	
aromatic tertiary sp^2	$=\text{C}_a\text{H}-$ 7.4 (A10)	
quaternary aromatic sp^2 carbon adjacent to an sp^3 atom	$=\text{C}_a(\text{R})-$ -9.6 (A11)	
peripheral quaternary aromatic sp^2 carbon adjacent to an sp^2 atom	$=\text{C}_a(\text{R})-$ -7.5 (A12)	
internal quaternary aromatic sp^2 carbon adjacent to an sp^2 atom	$=\text{C}_a(\text{R})-$ -0.7 (A13)	

^aThe alphanumeric terms, A1, A2, B2, ... are a device used to identify each group value in the estimations provided in Tables 7, 8, and 9.

^bThe group coefficient of 1.31 for C_{CH_2} is applied only when the number of consecutive methylene groups equals or exceeds the sum of the remaining groups; see Eq. 2 in text.

TABLE 1. (b) Contributions of the cyclic hydrocarbon portions of the molecule

Contributions of cyclic carbons	Group value (G_i) ($\text{J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$)	Group coefficient C_i
Ring equations for nonaromatic cyclic compounds		
	$\Delta S_{\text{ring}} = [33.4(A14)] + [3.7(A15)][n-3]$; n = number of ring atoms	
Ring equation for nonaromatic polycyclic compounds		
	$\Delta S_{\text{ring}} = [33.4(A14)]N + [3.7(A15)][R-3N]$; R = total number of ring atoms; N = number of rings	
cyclic tertiary sp^3	$>\text{C}_c\text{H}(\text{R})$ -14.7 (A16)	
cyclic quaternary sp^3	$>\text{C}_c(\text{R})_2$ -34.6 (A17)	
cyclic tertiary sp^2	$=\text{C}_c\text{H}-$ -1.6 (A18)	1.92 (B18)
cyclic quaternary sp^2	$=\text{C}_c(\text{R})-$ -12.3 (A19)	
cyclic quaternary sp	$=\text{C}_c=$; $\text{R}-\text{C}_c\equiv$ -4.7 (A20)	

TABLE 2. (a) Contributions of the functional group portion of the molecule

Functional groups ^a	Group value (G_k) ^a J/(mol K)	Group coefficient (C_k) ^b						
		k						
		2	3	4	5	6		
bromine	R-Br	17.5	(A21)					
chlorine	R-Cl	10.8	(A22)	1.5 (B22)	1.5 (C22)	1.5 (D22)	1.5 (E22)	1.5 (F22)
fluorine on an sp^2 carbon	=CRF	19.5	(A23)					
fluorine on an aromatic carbon	=CF-	16.6	(A24)					
3-fluorines on an sp^3 carbon	CF ₃ -R	13.3	(A25)					
2-fluorines on an sp^3 carbon	R-CF ₂ -R	16.4	(A26)					
1-fluorine on an sp^3 carbon	R-CF-(R) ₂	12.7	(A27)					
fluorine on a ring carbon	-CHF-	[17.5]	(A28)					
	-CF ₂ -	[17.5]	(A28)					
iodine	R-I	19.4	(A29)					
hydroxyl group	R-OH	1.7	(A30)	10.4 (B30)	9.7 (C30)	13.1 (D30)	12.1 (E30)	13.1 (F30)
phenol	=C-(OH)-	20.3	(A31)					
ether	R-O-R	4.71	(A32)					
peroxide, 1	R-O-O-R	[10.6]	(A33)					
aldehyde	R-CH(=O)	21.5	(A34)					
ketone	R-C(=O)-(R)	4.6	(A35)					
carboxylic acid	R-C(=O)OH	13.4	(A36)	1.21 (B36)	2.25 (C36)	2.25 (D36)	2.25 (E36)	2.25 (F36)
formate ester	R-OCH(=O)	[4.2]	(A37)					
ester	R-C(=O)O-R	7.7	(A38)					
anhydride	R-C(=O)OC(=O)-R	[10.0]	(A39)					
acyl chloride	R-C(=O)Cl	[25.8]	(A40)					
aromatic heterocyclic amine	=N-	[10.9]	(A41)					
acyclic sp^2 nitrogen	=N-	[-1.8]	(A42)					
tertiary amine	R-N(R) ₂	-22.2	(A43)					
secondary amine	R-NH-R	-5.3	(A44)					
primary amine	R-NH ₂	21.4	(A45)					
azide	R-N ₃	[-32.5]	(A46)					
tertiary amine <i>N</i> -nitro	R ₂ -N-(NO ₂)	5.39	(A47)					
aliphatic secondary amine <i>N</i> -nitro	R-NH-(NO ₂)	[-4.59]	(A48)					
aromatic tertiary amine <i>N</i> -nitro	R-NH-(NO ₂)	[-41.7]	(A49)					
nitro group	R-NO ₂	17.7	(A50)					
<i>N</i> -nitro	>N-(NO ₂)	39.8	(A51)					
<i>N</i> -nitroso	>N-N=O	[28.6]	(A52)					
oxime	=N-OH	[13.6]	(A53)					
azoxy nitrogen	N=N(→O)-	[6.8]	(A54)					
nitrate ester	R-ONO ₂	[24.4]	(A55)					
nitrile	R-C≡N	17.7	(A56)					
isocyanide	R-NC	[17.5]	(A57)					
isocyanate	R-N=C=O	[23.1]	(A58)					
tertiary amides	R-C(=O)NR ₂	-11.2	(A59)					
secondary amides	R-C(=O)NH-R	1.5	(A60)					
primary amide	R-CONH ₂	27.9	(A61)					
<i>N,N</i> -dialkylformamide, 1	HC(=O)NR ₂	[6.9]	(A62)					
tetra substituted urea	R ₂ NC(=O)NR ₂	[-19.3]	(A63)					
1,1,3-trisubst urea	R ₂ NC(=O)NH-R	[0.2]	(A64)	-12.8 (B64)	-24 (C64)	6 (D64)		
1,1-disubstituted urea	R ₂ NC(=O)NH ₂	[19.5]	(A65)					
1,3-disubstituted urea	RNHC(=O)NH-R	[1.5]	(A66)					
mono substituted urea	R-NHC(=O)NH ₂	[22.5]	(A67)					
<i>N,N</i> -disubstituted carbamate	R-OC(=O)NR ₂	-23.12	(A68)					
<i>N</i> -substituted carbamate	R-OC(=O)NH-R	10.6	(A69)					
carbamate	R-OC(=O)NH ₂	[27.9]	(A70)					
imide	R-C(=O)NHC(=O)-R	[7.7]	(A71)					
phosphine	R ₃ -P	[-20.7]	(A72)					
phosphine oxide	R ₃ -P=O	[-32.7]	(A73)					
phosphate ester	P(=O)(O-R) ₄	[-10.0]	(A74)					
phosphonate ester	R-P(=O)(O-R) ₂	[-14.0]	(A75)					
phosphonic acid	R-P=O(OH) ₂	[7.7]	(A76)					
phosphonyl halide	R-P(=O)X ₂	[4.8]	(A77)					

TABLE 2. (a) Contributions of the functional group portion of the molecule—Continued

Functional groups ^a	Group value (G_k) ^a J/(mol K)	Group coefficient (C_k) ^b				
		k				
		2	3	4	5	6
phosphoramidate ester	(R-O) ₂ P(=O)NH-R	[-0.7]	(A78)			
phosphorothioate ester	(R-O) ₂ P(=S)	1.1	(A79)			
phosphorodithioate ester	R-S-P(=S)(O-R) ₂	-9.6	(A80)			
phosphonothioate ester	R-P(=S)(O-R) ₂	[5.2]	(A81)			
phosphoramidothioate ester	R-NHP(=S)(O-R) ₂	[16.0]	(A82)			
phosphoroamidodithioate ester	NH ₂ P(=S)(S-R)(O-R)	[6.9]	(A83)			
sulfides	R-S-R	2.1	(A84)			
disulfides	R-SS-R	9.6	(A85)			
thiols	R-SH	23.0	(A86)			
sulfoxide	R-S(→O)-R	[14.1]	(A87)			
sulfones	R-S(→O) ₂ -R	0.3	A88)			
sulfonate ester	R-S(→O) ₂ O-R	[7.9]	(A89)			
1,2-disubstituted thiourea	R-NHC(=S)NH-R	[14.4]	(A90)			
monosubst thiourea	R-NHC(=S)+NH ₂	[23.1]	(A91)			
thioamide	R-C(=S)NH ₂	[30.0]	(A92)			
N,N disubstituted thiocarbamate	R-S(C=O)N-R ₂	[5.6]	(A93)			
N,N-disubstituted sulfonamide	R-S(→O) ₂ N-R ₂	[-11.3]	(A94)			
N-substituted sulfonamide	R-S(→O) ₂ NH-R	6.3	(A95)			
sulfonic acid	R-S(→O) ₂ OH	[1.8]	(A145)			
sulfonamide	R-S(→O) ₂ NH ₂	[28.4]	(A96)			
trisubstituted aluminum	R ₃ -Al	[-24.7]	(A97)			
trisubstituted arsenic	R ₃ -As	[-6.5]	(A98)			
trisubstituted boron	R ₃ -B	[-17.2]	(A99)			
trisubstituted bismuth	R ₃ -Bi	[-14.5]	(A100)			
trisubstituted gallium	R ₃ -Ga	[-11.9]	(A101)			
tetrasubstituted germanium	R ₄ -Ge	[-35.2]	(A102)			
disubstituted germanium	R ₂ GeH ₂	[-14.7]	(A103)			
disubstituted mercury	R ₂ -Hg	[8.4]	(A104)			
trisubstituted indium	R ₃ -In	[-19.3]	(A105)			
tetrasubstituted lead	R ₄ -Pb	[-30.2]	(A106)			
trisubstituted antimony	R ₃ -Sb	[-12.7]	(A107)			
disubstituted selenium	R ₂ -Se	[6.0]	(A108)			
quaternary silicon	R ₄ -Si	-27.1	(A109)			
quaternary tin	R ₄ -Sn	-24.2	(A110)			
disubstituted zinc	R ₂ -Zn	[11.1]	(A111)			
disubstituted telluride	R ₂ -Te	[-2.2]	(A140)			
trisubstituted germanium	R ₃ -GeH	[-27.8]	(A141)			
disubstituted arsenic acid	R ₂ -AsO ₂ H	[-24]	(A142)			
trisubstituted thallium	R ₃ -Th	[1]	(A143)			
disubstituted cadmium	R ₂ -Cd	[-2]	(A144)			

^aR: any alkyl or aryl group unless specified otherwise; X: any halogen; units: J mol⁻¹ K⁻¹.

^bUnassigned values beneath each of the group coefficients; C_k can be assumed to be 1.



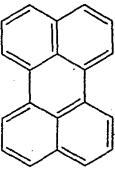



TABLE 2. (b) Contributions of functional groups as part of a ring

Heteroatoms and functional groups comprising a portion of a ring ^b		Group value (G_k) ^a	
cyclic ether	$R-O-R$	1.2	(A112)
cyclic peroxide	$R-OO-R$	[27.7]	(A113)
cyclic ketone	$R-C(=O)-R$	-1.4	(A114)
cyclic ester	$R-C(=O)O-R$	3.1	(A115)
cyclic carbonate	$R-OC(=O)O-R$	[1.3]	(A116)
cyclic anhydride	$R-C(=O)-O-C(=O)-R$	2.3	(A117)
cyclic sp^2 nitrogen	$R=N-R$	0.5	(A118)
cyclic tertiary amine	$R_2>N-R$	-19.3	(A119)
cyclic tertiary amine-N-nitro,	$R_2>N-(NO_2)-R$	-27.1	(A120)
cyclic tertiary amine-N-nitroso	$R_2>N-(N=O)-R$	-27.1	(A120)
cyclic secondary amine	$R_2>NH$	2.2	(A121)
cyclic tertiary amine N-oxide	$R_2>N(-O)-R$	[-22.2]	(A122)
cyclic azoxy group	$R=N(-O)-R$	[2.9]	(A123)
cyclic sec amide	$R-C(=O)NH-R$	2.7	(A124)
cyclic tertiary amide	$R-C(=O)N<RR$	-21.7	(A125)
cyclic tertiary amide	$R-C(=O)N<R_2$	[-9]	(A146)
cyclic carbamate	$R-OC(=O)N-RR$	[-5.2]	(A126)
cyclic carbamate	$R-OC(=O)N-HR$	[19.7]	(A125)
cyclic urea	$R-NC(=O)N<RR$	[-40.6]	(A127)
N-substituted cyclic imide	$R-C(=O)N(R)C(=O)-R$	[1.1]	(A128)
cyclic imide	$R-C(=O)N(H)C(=O)-R$	[1.4]	(A129)
cyclic phosphorothioate	$R-O-P(=S)<(OR)(OR)$	[-15.6]	(A130)
cyclic sulfide	$R-S-R$	2.9	(A131)
cyclic disulfide	$R-SS-R$	[-6.4]	(A132)
cyclic disulfide S-oxide	$R-SS(-O)-R$	[1.9]	(A133)
cyclic sulphone	$R-S(-O)_2-R$	[-10.4]	(A134)
cyclic thiocarbonate	$R-OC(=O)S-R$	[14.2]	(A135)
cyclic sulfate	$R-OS(-O)_2O-R$	0.9	(A136)
cyclic N-substituted sulphonamide	$R-S(-O)_2NH-R$	[-0.4]	(A137)
cyclic thiocarbamate	$R-S-(C=O)NHR$	[13.9]	(A138)
cyclic quaternary silicon	$R_2>Si<R_2$	-34.7	(A139)

^aR: any alkyl or aryl group unless specified otherwise; values in brackets are tentative assignments; units: $J mol^{-1} K^{-1}$.

^bThe R groups that are a part of the ring structure are designated by italics.

TABLE 3. Estimations of total phase change entropies and enthalpies of hydrocarbons^a

<p>C₈H₈ styrene^b</p>  <p>T_{fus}°: 242.3 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 52.2 (45.2) $\Delta_0^{fus} H_{tpce}^{\circ}$: 12.6 (11.0) $\Delta_0^{fus} S_{tpce}^{\circ}$: {5[7.4]+[-7.5]+[5.3]+[17.3]}</p>		<p>C₇H₁₄ 1-heptene^b</p>  <p>T_{fus}°: 154.3 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 77.5 (81.) $\Delta_0^{fus} H_{tpce}^{\circ}$: 12.0 (12.6) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[17.3]+[5.3]+4[1.31][7.1]+17.6]}</p>	
<p>C₂₀H₁₂ perylene^b</p>  <p>T_{fus}°: 551 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 42.4 (57.9) $\Delta_0^{fus} H_{tpce}^{\circ}$: 23.4 (31.9) $\Delta_0^{fus} S_{tpce}^{\circ}$: {12[7.4]+6[-7.5]+2[-0.7]}</p>		<p>C₂₀H₃₂ 10,10,13,13-tetramethyl-1,5-cyclohexadecadiene^c</p>  <p>T_{fus}°: 323 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 63.8 (58.3) $\Delta_0^{fus} H_{tpce}^{\circ}$: 20.6 (18.8) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[33.4]+13[3.7]+4[17.6]+2[-34.6]-4[-4.7]}</p>	
<p>C₁₀H₁₀ bullvalene^b</p>  <p>T_{fus}°: 366.5 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 35.3 (41.6) $\Delta_0^{fus} H_{tpce}^{\circ}$: 12.9 (15.3) $\Delta_0^{fus} S_{tpce}^{\circ}$: {3[33.4]+[3.7]-6[-1.6]+4[-14.7]}</p>		<p>C₁₂H₈ acenaphthylene^{b,d}</p>  <p>T_i°: 116.6; 127.1 K T_{fus}°: 362.6 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 37.6 (12.1+19.1) $\Delta_0^{fus} H_{tpce}^{\circ}$: 13.6 (1.5+6.9) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[33.4]+2[3.7]+[-7.5]+6[7.4]+3[-12.3]+2[-1.6]}</p>	

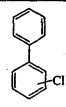



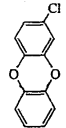
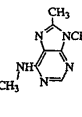
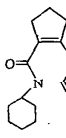
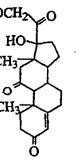
^aUnits for $\Delta_0^{fus} S_{tpce}^{\circ}$ and $\Delta_0^{fus} H_{tpce}^{\circ}$ are $J \cdot mol^{-1} \cdot K^{-1}$ and $kJ \cdot mol^{-1}$, respectively; experimental values are included in parentheses following the calculated value (in cases where additional solid-solid transitions are involved, the first term given is the total property associated with the transition(s) and the second term represents the fusion property).

^bReference 11.

^cReference 12.

^dReference 13.

TABLE 4. Estimations of total phase change entropies and enthalpies

A. Substituted Aromatic and Aliphatic Molecules ^a	
<p>C₁₂Cl₁₀ decachlorobiphenyl^b</p>  <p>T_{fus}°: 577.7 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 72.1 (68.1) $\Delta_0^{fus} H_{tpce}^{\circ}$: 41.6 (39.3) $\Delta_0^{fus} S_{tpce}^{\circ}$: {12[-7.5]+10[1.5][10.8]}</p>	<p>C₅H₁₀N₂O₂ N-acetyl-L-alanine amide^c</p>  <p>T_{fus}°: 431 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 54.9 (50.4) $\Delta_0^{fus} H_{tpce}^{\circ}$: 23.7 (21.7) $\Delta_0^{fus} S_{tpce}^{\circ}$: {2[17.0]+[27.9]+0.6[-16.4]+[1.5]}</p>
<p>C₂F₃N 2,2,2-trifluoroacetone^c</p>  <p>T_{fus}°: 128.7 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 34.6 (38.6) $\Delta_0^{fus} H_{tpce}^{\circ}$: 4.5 (5.0) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[[-34.8][0.66]+3[13.3]+[17.7]}</p>	<p>C₉H₇N isoquinoline^c</p>  <p>T_{fus}°: 299.6 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 47.9 (52.1) $\Delta_0^{fus} H_{tpce}^{\circ}$: 14.3 (13.5) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[10.9]+7[7.4]+2[-7.5]}</p>
B. Substituted Cyclic Molecules ^a	
<p>C₁₂H₇ClO₂ 2-chlorodibenzodioxin^d</p>  <p>T_{fus}°: 362.2 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 58.5 (63.8) $\Delta_0^{fus} H_{tpce}^{\circ}$: 21.2 (23.1) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[33.4]+3[3.7]+2[1.2]+4[-12.3]+7[7.4]+[-7.5]+[1.5][10.8]}</p>	<p>C₈H₁₁N₅ 6,8,9-trimethyladenine^e</p>  <p>T_{fus}°: 438 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 54.3 (52.7) $\Delta_0^{fus} H_{tpce}^{\circ}$: 23.8 (23.1) $\Delta_0^{fus} S_{tpce}^{\circ}$: {[33.4]+2[3.7]+3[17.6]+2[10.9]+[0.5]+[-19.3]+[-5.3]+[-7.5]+[7.4]+3[-12.3]}</p>
<p>C₁₂H₁₈N₂O₂ Lenacil^f</p>  <p>T_{fus}°: 584.3 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 64.0 (72.4) $\Delta_0^{fus} H_{tpce}^{\circ}$: 37.4 (42.3) $\Delta_0^{fus} S_{tpce}^{\circ}$: {3[33.4]+6[3.7]+2[-12.3]+[-14.7]+[-21.7][2.7]}</p>	<p>C₂₁H₃₀O₆ cortisone^f</p>  <p>T_{fus}°: 495 K $\Delta_0^{fus} S_{tpce}^{\circ}$: 75.2 (74.5) $\Delta_0^{fus} H_{tpce}^{\circ}$: 37.2 (36.9) $\Delta_0^{fus} S_{tpce}^{\circ}$: {4[33.4]+5[3.7]+[4.6]+2[17.6]+[7.1]+[-12.3]+[-1.6][1.92]+2[-1.4]+3[-14.7]+3[-14.7]+3[-34.6]+2[1.7][12.1]}</p>

^aUnits for $\Delta_0^{fus} S_{tpce}^{\circ}$ and $\Delta_0^{fus} H_{tpce}^{\circ}$ are $J \cdot mol^{-1} \cdot K^{-1}$ and $kJ \cdot mol^{-1}$, respectively; experimental values are given in parentheses.

^bReference 14.

^cReference 11.

^dReference 15.

^eReference 16.

^fReference 17.

^gReference 18.

TABLE 5. Experimental and calculated total phase change enthalpy and entropy of database^a

	T(K)	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{pce}$ (expt)	$\Delta_0^{T_{fus}} S_{pce}$ (calcd)	$\Delta_0^{T_{fus}} H_{pce}$ (expt)	$\Delta_0^{T_{fus}} H_{pce}$ (calcd)
CBrCl ₃		bromotrichloromethane					
	238.2	4.62	19.4				
	259.3	0.53	2.03				
	267.9	2.03	7.58	29.0	43.2	7.2	11.6
		A4*B4+A21+3*A22*D22					
CBr ₄		carbon tetrabromide					
	320	5.94	18.58				
	363.2	3.95	10.88	29.46	47.3	9.9	17.2
		A4*B4+4*A21					
CCl ₃ F		fluorotrichloromethane					
	162.7	6.9	0	42.38	38.4	6.9	6.2
		A4*B4+3*A22*D22+A27					
CCl ₄		carbon tetrachloride					
	224.6	4.6	20.49				
	249	2.69	10.82	31.31	41.9	7.3	10.4
	225.4	4.58	20.3				
	250.3	2.52	10.1	30.4		7.1	
	225.7	4.63	20.5				
	250.5	2.56	10.2	30.7		7.2	
		4*A22*D22+A4*B4					
CF ₄		carbon tetrafluoride					
	76.27	1.71	22.43				
	89.56	0.71	7.95	30.38	30.1	2.42	2.7
	76.1	1.73	21.4				
	88.4	0.69	7.7	29.1		2.4	
	76.1	1.46	19.2				
89.5	0.71	7.9	27.1		2.2		
		4*A25+A4*B4					
CHClF ₂		chlorodifluoromethane					
	59	0.07	1.13				
	115.7	4.12	35.65	36.78	39.3	4.19	4.5
		2*A26+A3*B3+A22*B22					
CHCl ₃		trichloromethane					
	209.6	8.8	0	41.98	38.9	8.8	8.2
		A3*B3+3*A22*C22					
CHF ₃		trifluoromethane					
	118.0	4.06	0	34.85	30.5	4.06	3.6
		3*A25+A3*B3					
CHF ₃ S		trifluoromethanethiol					
	116.0	4.93	0	42.44	39.9	4.93	4.6
		3*A25+A4*B4+A86					
CH ₂ Cl ₂		dichloromethane					
	178.2	6.16	0	34.56	39.5	6.16	7.0
		A2+2*A22*B22					
CH ₂ N ₂		cyanamide					
	317.2	8.76	0	27.62	39.1	8.76	12.4
	318.7	7.27	0	22.8		7.27	
		A56+A45					
CH ₂ N ₄		tetrazole					
	432.1	17.7	0	40.96	41.5	17.7	17.9
	430.7	18.4	0	42.7		18.4	
	242.5	0.014	0.06				
	430	18.0	41.9	42.0		18.14	
		A14+2*A15+A121+3*A118+A18*B18					
CH ₃ Br		bromomethane					
	173.8	0.47	2.72				
	179.5	5.98	3.33	36.02	35.1	6.45	6.3
		A21+A1					
CH ₃ Cl		methyl chloride					
	174.5	6.43	0	36.82	28.4	6.42	5.0
		A1+A22					
CH ₃ CIFOP		methylphosphonyl chlorofluoride					
	250.7	11.85	0	47.28	51.2	11.85	12.8
		A1+A22*C22+A27+A77					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}}S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}}S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}}H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}}H_{tpcc}$ (calcd)		
CH ₃ Cl ₂ OP	306.1	methylphosphonyl dichloride		59.05	18.08	16.8		
		A1+2*A22*C22+A77	0				54.8	18.08
CH ₃ Cl ₃ Si	197.4	trichloromethylsilane		45.32	8.95	7.7		
		3*A22*D22+A1+A109	0				39.1	8.95
CH ₃ F ₂ OP	236.3	methylphosphonyl difluoride		50.27	11.88	13.0		
		A1+2*A26+A77	0				55.1	11.8
CH ₃ NO	275.7	formamide		28.94	7.98	7.7		
		7.98	0				27.9	7.98
		8.67	0				31.5	8.67
CH ₃ NO ₂	244.8	nitromethane		39.62	9.7	8.64		
		A61	0				35.3	9.7
CH ₃ NO ₃	190.2	methyl nitrate		43.33	8.24	8.0		
		A1+A50	0				42.0	8.24
CH ₄ O	161.1	methanol		21.8	3.77	3.4		
		0.59	3.7				19.3	3.77
		3.18	18.1				19.3	3.77
		0.64	4.0				22.3	3.86
		3.22	18.3				22.3	3.86
CH ₄ N ₄ O ₄	371	N,N'-dinitro-diaminomethane		96.63	35.85	28.7		
		A1+A30	0				77.5	35.85
CH ₄ S	137.6	methanethiol		40.92	8.1	6.1		
		2.2	1.59				40.6	8.1
		5.9	39.33				40.6	8.1
CH ₅ N	179.7	methylamine		34.14	6.13	7.0		
		A1+A86	0				38.9	6.13
CH ₆ N ₂	220.8	methylhydrazine		47.19	10.42	7.4		
		A1+A45	0				33.7	10.42
C ₂ Br ₂ F ₂	162.8	dibromodifluoroethylene		43.22	7.04	8.6		
		2*A21+2*A23+2*A7	0				52.6	7.04
C ₂ Br ₂ F ₄	162.8	1,2-dibromotetrafluoroethane		43.24	7.04	8.9		
		2*A44*B4+2*A21+4*A26	0				54.9	7.04
C ₂ ClF ₃	115	chlorotrifluoroethylene		48.28	5.55	6.1		
		2*A7+3*A23+A22*B22	0				53.2	5.55
C ₂ ClF ₅	80.24	pentafluorochloroethane		43.56	4.51	7.5		
		2.63	32.76				42.9	4.51
		1.88	10.79				42.9	4.51
C ₂ Cl ₂ F ₄	173.7	1,2-dichloro-tetrafluoroethane		39.0	5.35	9.4		
		2*A26+2*A44*B4+A22*B22+3*A25	0				52.1	5.35
		1.21	11.1				52.1	5.35
		2.63	19.52				52.1	5.35
C ₂ Cl ₃ F ₃	109.3	1,1,2-trifluoro-1,2,2-trichloroethane		48.28	3.3	11.4		
		0.83	10.08				48.2	3.3
		2.47	10.42				48.2	3.3
C ₂ Cl ₄	210	tetrachloroethene		47.28	11.7	10.9		
		3*A22*D22+2*A26+2*A44*B4+A27	0				43.3	11.7
C ₂ Cl ₄ F ₂	130	1,1,2,2-tetrachlorodifluoroethane		18.42	4.49	13.3		
		4*A22*D22+2*A7	0				44.3	4.49
C ₂ Cl ₄ F ₂	299.7	1,1,2,2-tetrachlorodifluoroethane		18.42	4.49	13.3		
		0.79	6.08				44.3	4.49
C ₂ Cl ₄ F ₂	299.7	1,1,2,2-tetrachlorodifluoroethane		18.42	4.49	13.3		
		3.7	12.35				44.3	4.49

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}S_{pcc}}$ (expt)	$\Delta_0^{T_{fus}S_{pcc}}$ (calcd)	$\Delta_0^{T_{fus}H_{pcc}}$ (expt)	$\Delta_0^{T_{fus}H_{pcc}}$ (calcd)
C_2N_2	245.3	4*A22*E22+2*A27+2*A4*B4 cyanogen 8.11	0	33.05	35.5	8.11	8.7 [215,158]
C_2Cl_6	318	2*A56 hexachloroethane 2.57	8.07				
	345	8.22	23.83				
	458	9.75	21.29	53.18	51.4	20.54	23.5 [216]
C_2F_3N	128.7	6*A22*F22+2*A4*B4 trifluoroacetone 4.97	0	38.62	34.6	4.97	4.5 [216]
C_2F_4	142	3*A25+A4*B4+A56 tetrafluoroethylene 7.71	0	54.31	56.5	7.71	8.0 [216]
C_2F_6	104.0	4*A23+2*A7 hexafluoroethane 3.74	35.9				
	173.1	2.69	15.5	51.4	33.8	6.0	6.0
$C_2HBrClF_3$	154.7	6*A25+2*A4*B4 2-bromo-2-chloro-1,1,1-trifluoroethane 4.84	0	31.29	41.0	4.84	6.3 [216]
$C_2HBrClF_3$	146.2	A22*C22+A21+3*A25+A4*B4+A3*B3 1-bromo-2-chloro-1,1,2-trifluoroethane 4.38	0	29.96	46.6	4.38	6.8 [216]
C_2HCl_3	188.5	A22*C22+A21+2*A26+A4*B4+A3*B3+A27 trichloroethylene 8.45	0	44.83	41.8	8.45	7.9 [216]
$C_2HCl_3O_2$	330.7	3*A22*C22+A6*B6+A7 trichloroacetic acid 5.88	0	17.78	55.7	5.88	18.4 [215]
$C_2H_2Br_2F_2$	206.3	3*A22*D22+A36*D36+A4*B4 1,2-dibromo-1,1-difluoroethane 8.3	0	40.23	52.1	8.3	10.8 [215]
$C_2H_2Cl_2$	150.9	2*A21+2*A26+A2+A4*B4 1,1-dichloroethane 6.51	0	43.26	39.0	6.51	5.9 [216]
$C_2H_2Cl_2F_2$	163.0	2*B22*A22+A7+A5 1,2-difluoro-2,2-dichloroethane 8.19	0	50.26	42.0	8.19	6.8 [216]
$C_2H_2Cl_2O_2$	286.5	2*A22*C22+A4*B4+2*A27+A2 dichloroacetic acid 12.34	0	43.08	52.8	12.34	15.1 [216]
$C_2H_2Cl_4$	207.3	2*A22*C22+A3*B3+A36*C36 1,1,2,2-tetrachloroethane 0.54	2.62				
	230.8	9.17	39.74	42.38	45.4	9.72	10.5
	204.8	0.36	1.74				
	230.3	9.52	41.5	43.2		9.88	
$C_2H_3Br_3$	244	2*A3*B3+4*A22*D22 1,1,2-tribromoethane 9.11	0	37.34	50.1	9.11	12.2 [215]
C_2H_3Cl	119.3	A2+A3*B3+3*A21 vinyl chloride 4.92	0	41.21	32.0	4.92	3.8 [216]
$C_2H_3ClF_2$	142.4	A5+A6*B6+A22 1,1-difluoro-1-chloroethane 2.69	0	18.86	43.6	2.69	6.2 [216]
$C_2H_3ClO_2$	334.3	2*A26+A22*B22+A1+A4*B4 (α form) chloroacetic acid 16.3	0	48.74	39.4	16.3	13.2 [216]
$C_2H_3ClO_2$	329.2	A22*B22+A2+A36*B36 (β form) chloroacetic acid 13.93	0	42.33	39.4	13.93	13.0 [216]
$C_2H_3Cl_3$	237.1	A22*B22+A2+A36*B36 1,1,2-trichloroethane 11.38	0	48	46.0	11.38	10.9
	237.9	10.9	0	45.7		10.9	
$C_2H_3Cl_3$		3*A22*C22+A2+A3*B3 1,1,1-trichloroethane					7.4 [74]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}S_{pcc}}$ (expt)	$\Delta_0^{T_{fus}S_{pcc}}$ (calcd)	$\Delta_0^{T_{fus}H_{pcc}}$ (expt)	$\Delta_0^{T_{fus}H_{pcc}}$ (calcd)	
205	0.21	1.02					
223.6	7.45	33.31					
240.1	1.88	7.84	42.17	43.3	9.54	10.4	
224.2	7.47	33.3					
240.2	1.88	7.8	41.1		9.4		
224.8	7.49	33.3					
243.1	2.35	9.67	43.0		9.8		
$C_2H_3F_3$	3*A22*C22+A4*B4+A1						[216]
	1,1,1-trifluoroethane						
	161.9	6.19	0	38.23	34.5	6.19	5.6
156.4	0.3	1.9					
161.8	6.19	38.3	40.2		6.39		
C_2H_3N	3*A25+A4*B4+A1						[215]
	acetone						
	216.9	0.9	4.14				
229.3	8.17	35.61	39.75	35.3	9.06	8.1	
$C_2H_3N_3$	A1+A56						[216]
	1,2,4-triazole						
393.5	16.1	0	40.91	37.8	16.1	14.9	
C_4H_4	A14+2*A15+2*A118+A121+2*A18*B18						[216]
	ethylene						
104.0	3.35	0	32.24	34.7	3.35	3.6	
C_2H_4BrCl	2*A5						[216]
	1-bromo-2-chloroethane						
	182	3.1	17.15				
256.4	9.62	37.53	54.69	48.0	12.72	12.3	
$C_2H_4Br_2$	2*A2+A21+A22*B22						[216]
	1,2-dibromoethane						
	249.5	1.94	7.78				
283	10.94	38.66	46.44	49.4	12.88	14.0	
$C_2H_4Cl_2$	2*A21+2*A2						[216]
	1,1-dichloroethane						
176.2	7.87	0	44.77	40.3	7.87	7.1	
$C_2H_4Cl_2$	2*B22*A22+A1+A3*B3						[215]
	1,2-dichloroethane						
	237.2	8.83	0	37.24	46.6	8.83	11.1
175	2.85	16.2					
237.6	8.75	36.8	53.0		11.6		
$C_2H_4D_2O_2$	2*A22*B22+2*A2						[216]
	dihydroxyethane-d ₂						
258.8	9.75	0	37.67	50.5	9.75	13.1	
$C_2H_4N_4$	2*A2+2*A30*B30						[55]
	1H-1,2,4-triazol-3-amine						
428.3	21.93	0	51.2	52.2	21.93	22.4	
$C_2H_4N_4$	A14+2*A15+2*A121+2*A118+A18*B18+A19+A45						[221]
	1-methyltetrazole						
315	15.7	0	49.85	37.5	15.7	11.8	
$C_2H_4N_4$	A14+2*A15+A119+3*A118+A1+A18*B18						[174]
	2-methyltetrazole						
286	12.37	0	43.25	37.5	12.37	10.7	
$C_2H_4N_4$	A14+2*A15+A119+3*A118+A1+A18*B18						[174]
	5-methyltetrazole						
418	16	0	38.28	49.9	16	20.8	
C_2H_4O	A14+2*A15+3*A118+A121+A1+A19						[174]
	ethylene oxide						
160.7	5.17	0	32.22	34.6	5.17	5.6	
C_2H_4O	A14+A112						[216]
	acetaldehyde						
149.8	2.31	15.42					
242.9	1.72	7.06	22.49	39.1	4.03	9.5	
$C_2H_4O_2$	A1+A34						[216]
	ethanoic acid						
298.7	11.72	0	39.24	31.0	11.72	9.2	
C_2H_5Cl	A36+A1						[216]
	chloroethane						
134.8	4.45	0	33.01	35.5	4.45	4.8	
$C_2H_5Cl_3Si$	A22+A2+A1						[215]
	ethyltrichlorosilane						
165.3	6.96	0	42.1	46.2	6.96	7.6	
C_2H_5NO	A1+A2+3*A22*C22+A109						[216]
	acetamide						

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
353	15.6	0	44.19	45.5	15.6	16.1
354	15.5	0	43.8		15.5	
$C_2H_5NO_2$	A1+A61 nitroethane					[271,216]
183.7	9.85	0	53.64	42.4	9.85	7.8
$C_2H_5NO_2$	A1+A2+A50 methyl carbamate					[216]
328.6	16.7	0	50.82	45.5	16.7	14.9
$C_2H_5NO_3$	A1+A70 ethyl nitrate					[216]
178.6	8.53	0	47.74	49.1	8.53	8.8
C_2H_5NS	A1+A2+A55 ethanethioamide					[126]
385.7	18.36	0	47.59	47.6	18.36	18.4
C_2H_6	A1+A92 ethane					[221]
89.5	2.79	0	31.21	35.2	2.79	3.2
89.9	2.86	0	31.8		2.86	
$C_2H_6ClO_3P$	2*A1 2-chloroethylphosphonic acid					[216]
347.9	14.79	0	42.51	42.6	14.79	14.8
$C_2H_6Cl_2Si$	2*A2+A22*B22+A76 dimethyldichlorosilane					[221]
199.0	8.83	0	44.36	40.5	8.83	8.1
$C_2H_6N_2O$	2*A1+2*A22*C22+A109 N-methylurea					[216]
378.1	14.06	0	37.19	40.09	14.06	15.16
373.8	15.75		42.1		15.75	
$C_2H_6N_2O_2$	A1+A67 N-nitro-N-methylaminomethane					[138,216]
327	37.66	0	115.16	80.3	37.66	26.3
$C_2H_6N_4O_4$	2*A1+A51+A47 N,N'-dinitroethanediamine					[225]
450	29.5	0	65.55	84.6	29.5	38.07
C_2H_6O	2*A2+2*A48+2*A51 ethanol					[225]
111.4	3.14	28.16				
158.8	4.64	29.25	57.4	26.46	7.78	4.2
127.5	0.66	5.2				
159	4.93	31.0	36.2		5.6	
C_2H_6O	A1+A2+A30 dimethyl ether					[216]
131.7	4.94	0	37.5	39.9	4.94	5.3
C_2H_6OS	2*A1+A32 dimethyl sulfoxide					[216]
291.7	14.37	0	49.26	49.3	14.37	14.4
$C_2H_6O_2$	2*A1+A87 dihydroxyethane					[216]
260.6	9.96	0	38.21	50.5	9.96	13.2
260.8	11.6	0	44.6		11.6	
$C_2H_6O_2S$	2*A2+2*A30*B30 dimethylsulfone					[216]
382	18.28	0	47.91	35.4	18.30	13.5
C_2H_6S	2*A1+A88 ethyl mercaptan					[216]
195.3	4.97	0	25.48	47.7	4.97	9.3
C_2H_6S	A1+A2+A86 dimethyl sulfide					[216]
174.9	7.98	0	45.66	37.3	7.98	6.5
$C_2H_6S_2$	2*A1+A84 dimethyldisulfide					[216]
188.4	9.19	0	48.78	44.7	9.19	8.4
C_2H_6Se	2*A1+A85 dimethylselenium					[216]
185.1	8.5	0	45.91	41.1	8.5	7.6
$C_2H_6Se_2$	2*A1+A108 dimethyldiselenium					[170]
190.8	8.55	0	44.78	47.1	8.55	9.0
	2*A1+2*A108					[170]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
C_2H_6Zn	210.3	dimethyl zinc 1.06	5.05				
	230.1	6.83	29.68	34.73	46.2	7.89	10.6 [216]
$C_2H_7AsO_2$	470.9	2*A1+A111 hydroxydimethyl arsine 24.46	0	51.93	46.8	24.46	22.0 [221]
		2*A1+A98+A30*B30 dimethyl amine 5.94	0	29.68	29.9	5.94	5.4 [216]
$C_2H_8NOPS_2$	316.8	2*A1+A44 O,S-dimethyl phosphoramidothioate 13.34	0	42.1	42.1	13.34	13.3 [221]
		2*A1+A83 diaminoethane 0.49	2.57				
$C_2H_8N_2$	189.0	22.58	79.43	82.05	57.0	23.07	16.2 [201]
	284.2	2*A45+2*A2 N,N-dimethylhydrazine 10.07	0	46.64	34.3	10.07	7.4 [216]
$C_2H_8N_2$	264.3	2*A1+A45+A43 N,N'-dimethylhydrazine 13.64	0	51.6	24.6	13.64	6.5 [216]
		2*A1+2*A44 hexachlorocyclopropane 18.6	49.47	49.47	26.8	18.6	10.1 [216]
C_3Cl_6	376	6*A22*F22+3*A17+A14 hexafluoroacetone 8.38	0	56.74	38.3	8.38	5.7 [216]
		6*A25+2*A4*B4+A35 octafluoropropane 3.56	35.77				
C_3F_8	99.4	0.48	3.81	39.58	43.6	4.03	5.5 [216]
	125.5	2*A26+3*A4*B4+6*A25 3-chloro-1,1,1,3,3-pentafluoropropane 10.47	0	63.3	50.07	10.47	8.28 [216]
$C_3H_2ClF_5$	165.4	2*A26+A22*B22+2*A4*B4+A2+3*A25 1,1,1-trichloro-3,3,3-trifluoropropane 14.07	0	60.46	49.71	14.07	11.57 [215]
		3*A22*D22+3*A25+A2+2*A4*B4 dicyanomethane 10.8	0	35.4	42.59	10.8	12.99 [216]
$C_3H_2Cl_2F_3$	167.7	0.2	1.21				
	182.2	10.13	55.65	56.86	46.7	10.33	8.5 [216]
C_3H_3N	162.5	3*A25+A4*B4+A3*B3+A2+2*A22C22 acrylonitrile 1.19	7.32				
	189.6	6.23	32.84	40.17	39.0	7.42	7.4 [216]
C_3H_3NS	239.4	A5+B6*A6+A56 thiazole 9.58	40.08	40.04	35.0	9.58	8.5 [59,61]
		A14+2*A15+A131+A118+3*A18*B18 s-triazine 0.07	0.37				
$C_3H_3N_3$	197.7	14.56	41.2	41.57	55.0	14.63	19.4 [215]
	353.4	3*A10+3*A41 1,1,1-trifluoro-3-chloropropane 4.49	26.44				
$C_3H_4ClF_3$	169.8	5.05	28.2	54.6	47.3	9.54	8.5 [216]
	179.3	3*A25+A4*B4+2*A2+A22*B22 β -trichlorosilylpropionitrile 21.24	0	68.99	53.5	21.24	16.5 [103]
$C_3H_4Cl_4$	219.9	2.2	10.03				
	237.7	10.49	44.13	54.16	56.2	12.69	13.4 [216]
$C_3H_4N_2$	361.9	4*A22*D22+A4*B4+2*A2 imidazole 12.8	0	35.37	37.3	12.8	13.5

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpce}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpce}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{tpce}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{tpce}}}$ (calcd)
	A14+2*A15+2*A18*B18+A118+A121					[216]
$\text{C}_3\text{H}_4\text{N}_2$	pyrazole					
343.2	14.2	0	41.38	37.3	14.2	12.8
	A14+2*A15+2*A18*B18+A118+A121					[216]
$\text{C}_3\text{H}_4\text{N}_2\text{O}$	cyanoacetamide					
346.5	1.2	3.46				
387.3	21.7	56.03	59.49	52.8	22.9	20.4
	A2+A56+A61*B61					[216]
$\text{C}_3\text{H}_4\text{O}_2$	acrylic acid					
285.5	11.16	0	39.09	34.6	11.16	9.9
	A5+A6*B6+A36					[215]
$\text{C}_3\text{H}_4\text{O}_2$	β -propiolactone					
239.9	9.41	0	39.22	40.2	9.41	9.7
	A14+A15+A115					[32]
$\text{C}_3\text{H}_4\text{O}_3$	ethylene carbonate					
309.5	13.3	0	42.96	42.1	13.3	13.0
	A14+2*A15+A116					[52]
$\text{C}_3\text{H}_3\text{Br}_3$	1,2,3-tribromopropane					
289.4	23.78	0	82.17	57.3	23.78	16.6
	2*A2+A3*B3+3*A21					[215]
$\text{C}_3\text{H}_3\text{N}$	propionitrile					
177.0	17.07	9.67				
180.4	5.03	27.91	37.57	42.4	22.1	7.7
	A1+A2+A56					[216]
$\text{C}_3\text{H}_3\text{NO}$	acrylamide					
358	15.33	0	42.82	49.2	15.33	17.6
	A5+A6*B6+A61					[216]
$\text{C}_3\text{H}_3\text{N}_3\text{O}_9$	trinitroglycerine					
285.5	21.87	0	76.6	77.8	21.87	22.2
	2*A2+A3*B3+3*A55					[215]
C_3H_6	propene					
88.2	2.93	0	33.3	40.2	2.93	3.5
87.85	3.0		34.18		3.0	
	A1+A5+A6					[216]
C_3H_6	cyclopropane					
145.6	5.44	0	37.4	33.4	5.44	4.9
	A14					[216]
$\text{C}_3\text{H}_6\text{Br}_2$	1,3-dibromopropane					
238.6	14.64	0	61.5	63.1	14.64	15.1
	2*A21+3*A2*B2					[216]
$\text{C}_3\text{H}_6\text{ClNO}_2$	2-chloro-2-nitropropane					
213.8	9.54	44.62				
261.6	1.34	5.1	49.72	46.2	10.88	12.1
	2*A1+A4*B4+A50+A22*B22					[216]
$\text{C}_3\text{H}_6\text{Cl}_2$	1,2-dichloropropane					
172.7	6.4	0	37.06	47.4	6.4	8.2
	A1+A2+A3*B3+2*A22*B22					[215]
$\text{C}_3\text{H}_6\text{Cl}_2$	2,2-dichloropropane					
188	5.98	31.8				
239.3	2.34	9.62	41.42	44.7	8.32	10.8
	2*B22*A22+2*A1+A4*B4					[216]
$\text{C}_3\text{H}_6\text{N}_2\text{O}_2$	malonamide					
393	1.9	4.83				
443	35.8	80.81	85.65	63.0	37.7	27.9
	2*A61+A2					[292]
$\text{C}_3\text{H}_6\text{N}_2\text{O}_4$	2,2-dinitropropane					
267.7	11.28	42.13				
259.7	1.87	7.2				
324.5	2.64	8.12	57.15	47.8	15.78	15.5
	2*A1+A4*B4+2*A50					[216]
$\text{C}_3\text{H}_6\text{N}_4$	1,5-dimethyltetrazole					
349	14.7	0	42.12	46.0	14.7	16.0
	A14+2*A15+3*A118+A119+A19+2*A1					[174]
$\text{C}_3\text{H}_6\text{N}_4$	2,5-dimethyltetrazole					
256.4	13.5	0	52.65	46.0	13.5	11.8
	A14+2*A15+3*A118+A119+A19+2*A1					[174]
$\text{C}_3\text{H}_6\text{N}_4\text{O}_4$	1,3-dinitro-1,3-diazacyclopentane					
410	25.1	0	62.3	66.2	25.1	27.1
	A14+2*A15+2*A120+2*A51					[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{pcc}$ (expt)	$\Delta_0^{T_{fus}} H_{pcc}$ (calcd)	
$C_3H_6N_6O_3$	367	1,3,5-trinitroso-1,3,5-triazacyclohexane						
	376	17.78	48.45					
$C_3H_6N_6O_5$	446	$A14+3*A15+3*A120+3*A52$	3.77	10.02	58.47	49.0	21.55	18.4
		1,3-dinitro-5-nitroso-1,3,5-triazacyclohexane	25.97	0	58.24	71.4	25.97	31.8
$C_3H_6N_6O_6$	478.2	$A14+A15*3+3*A120+2*A51+A52$						[225]
		1,3,5-trinitro-1,3,5-triazacyclohexane	37.66	0	78.75	82.6	37.66	39.5
C_3H_6O	176.6	$A14+A15*3+3*A120+3*A51$						[216]
		acetone	5.72	0	32.34	39.7	5.72	7.0
C_3H_6O	161.3	$2*A1+A35$						[216]
		propylene oxide	6.57	0	40.75	37.5	6.57	6.0
C_3H_6O	161.2		6.53	0	40.52		6.53	
		$A1+A14+A112+A16$						[216]
C_3H_6O	171.3	propanal	8.59	0	50.14	46.3	8.59	7.9
		$A1+A2+A34$						[216]
$C_3H_6O_2$	252.7	propionic acid	10.66	0	42.2	38.1	10.66	9.6
		$A1+A2+A36$						[216]
$C_3H_6O_2$	142.4	1,3-dioxolane	2.68	18.8				
			6.57	37.33	56.13	43.2	9.24	7.6
$C_3H_6O_2$	175.9	$A14+2*A15+2*A112$						[216]
		methyl acetate	7.49	0	42.82	42.8	7.49	7.5
$C_3H_6O_2S$	291.9	$2*A1+A38$						[1]
		β -thiolactic acid	16.97	0	58.15	53.4	16.97	15.6
$C_3H_7O_3$	289.9	$2*A2+A36+A86$						[216]
		DL lactic acid	11.34	0	39.12	42.2	11.34	12.2
$C_3H_6O_3$	333.4	$A1+A3*B3+B30*A30+A36*B36$						[216]
		1,3,5-trioxane	15.11	0	45.3	48.2	15.11	16.1
C_3H_6S	176.7	$A14+3*A15+3*A112$						[215]
		thiacyclobutane	0.67	3.77				
C_3H_7Br	184.1		8.24	41.25	45.02	40.0	8.91	8.0
		$A14+A15+A131$						[216]
C_3H_7Cl	156	2-bromopropane	6.53	0	35.5	43.1	6.53	7.9
		$2*A1+A3*B3+A21$						[216]
C_3H_7N	237.8	2-chloropropane	7.39	0	47.37	36.3	7.39	5.7
		$A3*B3+2*A1+A22$						[215]
C_3H_7NO	212.9	cyclopropylamine	13.18	0	55.44	40.0	13.18	9.5
		$A14+A45+A16$						[215]
C_3H_7NO	303.8	N,N-dimethylformamide	8.95	0	42.05	42.1	8.95	9.0
		$2*A1+A62$						[216]
$C_3H_7NO_2$	321.9	N-methylacetamide	9.73	0	32.01	36.6	9.73	11.1
		$2*A1+A60$						[270]
$C_3H_7NO_2$	321.7	ethyl carbamate	15.23	0	47.31	52.6	15.23	16.9
			20.9	0	64.8		20.9	
$C_3H_7NO_3$	190.9		16.8	0	52.3		16.8	
		$A1+A2+A70$						[215, 216]
$C_3H_7NO_3$	190.9	isopropyl nitrate	10.1	0	52.9	49.9	10.1	9.5
		$2*A1+A3*B3+A55$						[173]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}}S_{pcc}$ (expt)	$\Delta_0^{T_{fus}}S_{pcc}$ (calcd)	$\Delta_0^{T_{fus}}H_{pcc}$ (expt)	$\Delta_0^{T_{fus}}H_{pcc}$ (calcd)
C_3H_8	85.5	propane					
		3.52	0	41.24	42.3	3.52	3.6
$C_3H_8N_2O$	365.1 367.8	2*A1+A2					[215]
		N-ethylurea					
		14.39 13.9	0 0	39.41 37.9	47.2	14.39 13.9	17.2
$C_3H_8N_2O$	454	A1+A2+A67					[138]
		1,1-dimethylurea					
$C_3H_8N_2O$	379.5 301.2 161.3	2*A1+A65					[215]
		1,3-dimethylurea					
C_3H_8O	148.8	13	34.26				
		0.08 0.32	0.26 1.97	36.48	36.6	13.4	5.91
C_3H_8O	185.2 184.7	2*A1+A66					[124, 138]
		1-propanol					
C_3H_8O	185.2 184.7	5.37	0	36.12	33.6	5.37	5.0
		A1+2*A2+A30					
$C_3H_8O_2$	168.0	2-propanol					
		5.41 5.37	0 0	29.21 29.1	27.3	5.41 5.37	5.1
$C_3H_8O_2$	168.0	2*A1+A3*B3+A30					[216]
		dimethoxymethane					
$C_3H_8O_3$	293 291	8.33	0	49.59	51.7	8.33	8.7
		2*A1+A2+2*A32					
C_3H_8S	167.2	1,2,3-trihydroxypropane					
		18.28 18.28	0 0	62.34 62.8	55.6	18.28 18.28	16.3
C_3H_8S	142.1 160	2*A2+A3*B3+3*A30*C30					[215]
		ethyl methyl sulfide					
C_3H_8S	142.1 160	9.76	0	58.37	44.4	9.76	7.4
		2*A1+A2+A84					
C_3H_8S	112.5 142.6	1-propanethiol					
		3.97 5.48	27.95 34.23	62.17	54.9	9.45	8.8
C_3H_8S	112.5 142.6	A1+2*A2+A86					[216]
		2-propanethiol					
$C_3H_8SO_2$	307.7	0.05 5.74	0.46 40.21	40.67	48.5	5.78	6.9
		2*A1+A3*B3+A86					
C_3H_9Al	288.4	ethylmethylsulfone					
		11.3	0	36.71	42.6	11.3	13.1
C_3H_9Al	288.4	2*A1+A2+A88					[276]
		trimethylaluminum					
C_3H_9As	186.6	8.79	0	30.48	28.1	8.79	8.1
		3*A1+A97					
C_3H_9B	113.2	trimethylarsine					
		8.96	0	48.03	46.3	8.96	8.6
C_3H_9B	113.2	3*A1+A98					[171]
		trimethylborane					
C_3H_9ClSi	185.1 218.0	3.25	0	28.68	35.6	3.25	4.0
		3*A1+A99					
C_3H_9Ga	257.9 244.5 257.8	chlorotrimethylsilane					
		0.7 9.68	3.75 44.42	48.17	41.8	10.38	9.1
C_3H_9N	188.4 188.4	3*A1+A22*B22+A109					[216]
		trimethylgallium					
C_3H_9N	188.4 188.4	11.05 0.33 10.6	0 1.4 41.1	42.83 42.5	40.8	11.05 11.0	10.5
		3*A1+A101					
C_3H_9N	188.4 188.4	1-aminopropane					
		10.97 10.63	0 0	58.24 56.4	53.2	10.97 10.63	10.0
C_3H_9N	178	A1+2*A2+A45					[215, 216]
		2-aminopropane					
C_3H_9N	156.1	7.33	0	41.17	46.9	7.33	8.3
		2*A1+A3*B3+A45					
C_3H_9N	156.1	trimethylamine					
		6.54	0	41.92	30.5	6.54	4.8
		3*A1+A43					[215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_3H_{10}N_2$	222	1,2-diaminopropane 0.07	0.3			
	236.5	18.42	77.89	78.19	57.8	18.49
$C_3H_{10}N_2$	201.2	$A2+A3*B3+A1+2*A45$ trimethylhydrazine 9.49	0	47.13	25.2	9.49
		$3*A1+A44*B44+A43$ maleic anhydride 12.26	0	37.65	36.9	12.26
$C_4H_2O_3$	325.7	$A14+2*A15+2*A18*B18+A117$ 2-bromothiophene 0.01	0.25			
		7.9	38.43	38.7	42.7	7.91
C_4H_3BrS	55.3 205.3	$A14+2*A15+A21+A131+2*A18+A18*B18+A19$ 2-chlorothiophene 8.97	0	44.56	41.3	8.97
		$A14+2*A15+A22*B22+A131+2*A18+A18*B18+A19$ α -(trifluoromethoxy)- α , α -difluoromethyl acetate 8.51	0	50.84	52.0	8.51
$C_4H_4N_2$	328.2	$3*A25+2*A26+A1+A38+2*A44*B4$ pyrazine 12.95	0	39.46	51.5	12.95
		$4*A10+2*A41$ succinonitrile 6.2	26.57			
$C_4H_4N_2$	233.3 331.2	3.7	11.21	37.78	49.7	9.9
		$2*A56+2*A2$ N-nitro-bis(N,N-cyanomethyl) amine 38.66	0	105.34	94.9	38.66
$C_4H_4N_4O_2$	367	$2*A2+2*A56*C56+A51*C51+A47*C47$ furan 2.05	13.64			
		3.8	20.29	33.93	32.6	5.85
C_4H_4O	150.0 187.6	$A14+2*A15+A112+2*A18*B18+2*A18$ 1,4-dioxane-2,5-dione 1.81	5.82			
		14.8	41.55	47.36	50.8	16.61
$C_4H_4O_4$	312.1 356.2	$3*A15+A14+2*A115$ ethylene oxalate 13.4	0	32.29	50.8	13.4
		$A14+3*A15+2*A115$ thiophene 1.21	7.11			
C_4H_4S	171.1 233.7	4.97	21.34	28.45	34.3	6.18
		171.6 235.0	0.64	3.7	25.4	5.7
$C_4H_5ClO_2$	333.7	$A14+2*A15+2*A18*B18+A131+2*A18$ <i>cis</i> -3-chloro-2-butenic acid 13.81	0	41.42	43.1	13.81
		$A36*B36+A1+A7+A6*B6+B22*A22$ Z-3-chloro-2-butenic acid 20.71	0	56.48	43.09	20.71
$C_4H_5ClO_2$	366.8	$A36*B36+A1+A6*B6+A7+A22*B22$ E-3-chloro-2-butenic acid 13.81	0	41.38	43.1	13.81
		$A36*B36+A1+A6*B6+A7+A22*B22$ pyrrole 7.91	0	31.66	33.6	7.91
C_4H_5N	249.7	$2*A18+A121+A14+2*A15+2*A18*B18$ succinimide 17.0	0	42.5	42.2	17.0
		$A14+2*A15+A129$ 2-methylthiazole 12.16	43.44	48.91	43.3	12.16
$C_4H_5NO_2$	400	$A14+2*A15+A131+A118+A1+2*A18*B18+A19$ 4-methylthiazole 8.9	0	38.85	43.3	8.9
		$A14+2*A15+A131+A118+A1+2*A18*B18+A19$ 5-methylthiazole				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
C_4H_6	232.8	7.65	0	32.86	43.3	7.65	10.1
		A14+2*A15+A131+A118+A1+2*A18*B18+A19 1,3-butadiene					[61]
C_4H_6	164.2	7.98	0	48.62	45.2	7.98	7.4
		2*A5+2*A6					[215]
C_4H_6	136.9	6.96	0	50.8	37.4	6.96	5.1
		1,2-butadiene					[216]
C_4H_6	240.9	9.25	0	38.38	29.6	9.25	7.1
		A1+A5+A9+A6 2-butyne					[215]
C_4H_6	147.4	6.03	0	40.9	36.9	6.03	5.4
		2*A1+2*A9 1-butyne					[216]
$C_4H_6N_6O_8$	430	29.37	0	68.31	70.8	29.37	30.4
		A14+3*A15+2*A120+2*A51+2*A50+A17 1,3,5,5-tetranitro-1,3-diazacyclohexane					[225,193]
$C_4H_6O_2$	197.5	9.73	0	49.26	46.5	9.73	9.2
		A1+A5+A6*B6+A38 methyl acrylate					[216]
$C_4H_6O_2$	287.5	8.06	0	28.04	37.6	8.06	10.8
		A1+A36+A7+A5 α -methylacrylic acid					[216]
$C_4H_6O_2$	344.4	12.57	0	36.49	40.1	12.57	13.8
		A1+A6+A36+A6*B6 <i>cis</i> -crotonic acid					[215]
$C_4H_6O_2$	230	9.57	0	41.84	43.9	9.57	10.1
		A14+2*A15+A115 γ -butyrolactone					[32]
$C_4H_6O_3$	218.2	9.62	0	44.07	44.9	9.62	9.8
		2*A15+A14+A1+A16*B16+A116 propylene carbonate					[89]
$C_4H_6O_4$	327.6	21.07	0	64.32	50.5	21.07	16.5
		2*A1+2*A38*B38 dimethyl oxalate					[215]
$C_4H_6O_4$	457	32.95	0	72.1	46.5	32.95	21.3
		2*A36*B36+2*A2 succinic acid					[340]
$C_4H_6O_5$	402	33.52	0	83.39	74.5	33.52	30.0
		A2+2*C36*A36+A3*B3+A30*C30 (<i>dl</i>) malic acid I					[216]
$C_4H_6O_5$	396	30.17	0	76.19	74.5	30.17	29.5
		A2+2*C36*A36+A3*B3+A30*C30 (<i>dl</i>) malic acid II					[216]
$C_4H_6O_5$	376	23.01	0	61.2	74.5	23.01	28.0
		A3*B3+2*C36*A36+A2+A30*C30 (<i>d</i>) malic acid					[273]
C_4H_7NO	299	13.92	0	46.56	43.5	13.92	13.0
		A14+2*A15+A124 2-pyrrolidone					[216]
C_4H_8	385.1	15	0	38.95	52.1	15	20.1
		A1+A7+A5+A61 methacrylamide					[216]
C_4H_8	145.7	5.71	39.17				
	182.4	1.09	5.96	45.13	37.1	6.79	6.8
C_4H_8	87.8	3.85	0	43.84	47.3	3.85	4.2
		A14+A15 1-butene					[216]
C_4H_8	134.3	7.31	0	54.43	45.7	7.31	6.1
		A1+A2+A5+A6 <i>cis</i> -2-butene					[216]
C_4H_8	167.6	9.76	0	58.22	45.7	9.76	7.7
		2*A1+2*A6 <i>trans</i> -2-butene					[216]
C_4H_8	132.4	5.92	0	44.72	41.8	5.92	5.5
		isobutene					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
	A7+2*A1+A5					[216]
C ₄ H ₈ Br ₂ O ₂	(dl) 2,3-dibromo-1,4-butanediol					
363.2	29.29	0	80.64	75.9	29.29	27.6
	2*A21+2*A2+2*A3*B3+2*A30*D30					[226]
C ₄ H ₈ Br ₂ O ₂	(d) 2,3-dibromo-1,4-butanediol					
388.2	33.89	0	87.3	75.9	33.89	29.5
	2*A21+2*A2+2*A3*B3+2*A30*D30					[226]
C ₄ H ₈ Cl ₂ O	1,5-dichloro-3-oxapentane					
226.5	8.39	0	37.02	65.6	8.39	14.9
	4*A2+2*A22*C22+A32					[216]
C ₄ H ₈ Cl ₃ O ₄ P	dimethyl (2,2,2-trichloro-1-hydroxyethyl)phosphonate					
351.0	20.37	0	58.03	58.3	20.37	20.5
357	22.4	0	62.75	58.3	22.4	20.8
384	25	0	65.1	58.3	25.0	22.4
	3*A22*E22+A4*B4+A3*B3+A30*E30+2*A1+A75					[216]
C ₄ H ₈ N ₂ O ₂	N-acetylglycine amide					
408.2	25.6	0	62.71	54.1	25.6	22.1
	A1+A2+A61+A60					[216]
C ₄ H ₈ N ₄ O ₄	1,3-dinitro-1,3-diazacyclohexane					
343	15.8	46.06				
354	2.97	8.39	54.45	69.9	18.77	24.7
	A14+3*A15+2*A120+2*A51*C51					[147]
C ₄ H ₈ N ₆ O ₅	1,5-dinitro-3-nitroso-1,3,5-triazacycloheptane					
404	25.7	63.61				
440	2.9	6.59	70.2	75.1	28.6	33.1
	A14+4*A15+3*A120+2*A51+A52					[147]
C ₄ H ₈ N ₆ O ₈	1,3,5,7-tetranitro-1,3,5,7-tetrazocine					
553.2	69.87	0	126.3	102.7	69.87	56.8
	A14+5*A15+4*A120+4*A51					[216]
C ₄ H ₈ N ₁₂ O ₆	1,7-diazido-2,4,6-trinitro-2,4,6-triazaheptane					
406	40.17	0	98.93	99.0	40.17	40.2
	4*A2+3*A51+2*A46+3*A47					[225]
C ₄ H ₈ O	2-butanone					
186.5	8.39	0	45.27	46.9	8.39	8.7
	2*A1+A2+A35					[341]
C ₄ H ₈ O	butanal					
176.8	11.09	61.09	62.8	53.5	11.09	9.4
	A1+2*A2+A34					[216, 84]
C ₄ H ₈ O	tetrahydrofuran					
164.8	8.54	0	51.88	42.0	8.54	6.9
	A14+2*A15+A112					[215]
C ₄ H ₈ O ₂	butanoic acid					
264.7	11.07	0	41.82	45.2	11.07	12.0
	A1+A2+A36+A2					[215]
C ₄ H ₈ O ₂	ethyl acetate					
189.3	10.48	0	55.35	50.0	10.48	9.5
	2*A1+A38+A2					[215]
C ₄ H ₈ O ₂	1,4-dioxane					
272.9	2.35	8.79				
284.1	12.84	45.19	53.97	46.9	15.2	13.3
	A14+3*A15+2*A112					[216]
C ₄ H ₈ O ₂ S	tetramethylene sulfone					
288.6	5.35	18.55				
301.6	1.43	4.73	23.29	30.4	6.78	9.2
	A14+2*A15+A134					[202]
C ₄ H ₈ O ₄	tetroxane					
385	22.6	0	58.58	56.8	22.6	21.9
	5*A15+A14+4*A112					[216]
C ₄ H ₈ S	thiacyclopentane					
177.0	7.35	0	41.55	43.7	7.35	7.7
	A14+2*A15+A131					[136]
C ₄ H ₈ S ₂	1,3-dithiane					
316.4	0.8	2.53				
327.2	14.4	44.01	46.54	50.3	15.2	16.5
	A14+3*A15+2*A131					[216]
C ₄ H ₈ S ₂	1,4-dithiane					
384.6	21.6	0	56.16	50.3	21.6	19.3

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)	
C_4H_9Br	160.4	A14+3*A15+2*A131					[216]	
		1-bromobutane	9.23	57.57	57.57	63.1	9.23	10.1
C_4H_9Br	208.6 231.5 256.1	3*A2*B2+A21+A1					[216]	
		<i>tert</i> -butyl bromide	5.65	27.08				
			1.05	4.52				
			1.97	7.68	39.33	47.4	8.66	12.2
C_4H_9Br	160.3	3*A1+A21+A4*B4					[216]	
		2-bromobutane	6.88	0	42.92	50.2	6.88	8.0
C_4H_9Cl	182.9 219.3 248.1	2*A1+A2+A3*B3+A21					[215]	
		<i>tert</i> -butyl chloride	1.87	10.25				
			5.88	26.82				
			1.97	7.95	45.02	40.7	9.82	10.1
C_4H_9N	207.1 215.3	3*A1+A4*B4+A22					[136]	
		pyrrolidine	0.54	2.61				
			8.58	39.84	42.44	43.0	9.12	9.3
$C_4H_9NO_2$	352.9 353.7 384.1	A121+A14+2*A15					[216]	
		2-amino-2-methylpropanediol	5	14.17				
			18.46	52.19				
			2.78	7.24	73.6	64.4	26.24	24.7
$C_4H_9NO_3$	310 361	2*A2+A4*B4+2*A30*C30+A45+A1					[274]	
		2-methyl-2-nitro-1-propanol	17.2	55.47				
			3.74	10.35	65.82	55.3	20.93	20.0
$C_4H_9NO_3$	352 424	2*A1+A4*B4+A2+A30*B30+A50					[216]	
		2-methyl-2-nitro-1,3-propanediol	25.72	73.08				
			3.84	9.07	82.14	60.7	29.57	25.7
C_4H_{10}	107.6 134.9	A1+A4*B4+2*A2+2*A30*C30+A50					[216]	
		butane	2.07	19.06				
			4.66	34.56	53.62	49.4	6.73	6.7
C_4H_{10}	113.7	2*A1+2*A2					[216]	
		isobutane	4.54	0	40.11	36.4	4.54	4.1
$C_4H_{10}Cl_2Si$	174.1	3*A1+A3					[216]	
		dichlorodiethylsilane	8.96	0	51.45	54.7	8.96	9.5
$C_4H_{10}Hg$	181.5	2*A22*C22+2*A1+2*A2+A109					[216]	
		diethyl mercury	10.5	0	57.87	57.8	10.5	10.5
$C_4H_{10}N_2O$	381	2*A1+A104+2*A2					[216]	
		<i>N</i> -propylurea	14.63	0	38.4	54.4	14.63	20.7
$C_4H_{10}N_2O$	429 375.5 280.8	2*A2+A1+A67					[215]	
		<i>N</i> -isopropylurea	17.5	40.79				
			2.31	6.15				
			1.41	5.02	51.97	48.0	21.22	13.5
$C_4H_{10}N_2O$	344.4	2*A1+A3*B3+A67					[138]	
		1,1,3-trimethylurea	14.3	0	41.52	52.9	14.3	18.2
$C_4H_{10}N_4O_4$	410	3*A1+A64					[215]	
		<i>N,N'</i> -dimethyl- <i>N,N'</i> -dinitro-1,2-ethanediamine	60.32	0	147.13	139.7	60.32	57.3
$C_4H_{10}O$	183.9	2*A1+2*A2+2*A51+2*A47					[225]	
		butyl alcohol	9.28	0	50.46	47.3	9.28	8.7
$C_4H_{10}O$	184.7	3*A2*B2+A1+A30					[215]	
		2-butanol	5.97	0	32.33	34.4	5.97	6.4
$C_4H_{10}O$	286.1	2*A1+A2+A3*B3+A30					[76]	
		<i>tert</i> -butyl alcohol	0.83	2.9				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
294.5	0.49	1.66				
299.0	6.7	22.42	26.98	31.6	8.02	9.5
$C_4H_{10}O$	3*A1+A4*B4+A30 (+)-butanol					[216]
177.4	6	0	33.82	34.4	6	6.1
$C_4H_{10}O$	2*A1+A2+A3*B3+A30					[76]
171.2	6.32	0	36.93	27.7	6.32	4.7
$C_4H_{10}O$	2*A1+A2+A3+A30					[73]
127.3	5.85	0	45.73	47.8	5.85	6.1
$C_4H_{10}O$	3*A1+A32+A3*B3					[216]
156.9	7.19	0	45.81	54.1	7.19	8.5
$C_4H_{10}O$	2*A1+2*A2+A32					[75]
134.0	7.67	0	57.24	54.1	7.67	7.3
$C_4H_{10}O_2$	A32+2*A1+2*A2					[216]
293.6	18.7	0	63.7	73.6	18.7	21.6
$C_4H_{10}O_4$	4*A2*B2+2*A30*B30					[216]
396	42.36	0	106.97	86.6	42.36	34.3
$C_4H_{10}S$	2*A2+2*A3*B3+4*A30*D30					[216]
169.2	11.9	0	70.47	51.5	11.9	8.7
$C_4H_{10}S$	2*A1+2*A2+A84					[216]
160.2	9.91	0	61.88	51.5	9.91	8.7
$C_4H_{10}S$	2*A1+2*A2+A84					[216]
171.7	9.36	0	54.5	56.3	9.36	9.7
$C_4H_{10}S$	3*A1+A3*B3+A84					[216]
128.3	4.98	0	38.83	48.9	4.98	6.3
$C_4H_{10}S$	2*A1+A3+A2+A86					[216]
157.5	10.46	0	66.44	68.6	10.46	10.8
$C_4H_{10}S$	A1+3*A2*B2+A86					[216]
151.6	4.07	26.83				
157	0.65	4.13				
199.4	0.97	4.87				
274.4	2.48	9.04	44.87	52.9	8.17	14.5
$C_4H_{10}S$	3*A1+A4*B4+A86					[216]
133.0	6.48	0	48.7	55.5	6.48	7.4
$C_4H_{10}S_2$	2*A1+A2+A3*B3+A86					[216]
171.6	9.4	0	54.77	59.0	9.4	10.1
$C_4H_{10}Zn$	2*A1+2*A2+A85					[216]
148.4	0.28	1.86				
237.0	16.63	70.19	72.05	60.5	17.52	14.3
$C_4H_{11}N$	2*A1+2*A2+A111					[216, 96]
91.3	0.11	1.24				
202.3	6.05	29.92				
206.2	0.88	4.28	35.44	51.3	7.05	4.7
$C_4H_{11}NO_2$	3*A1+A4*B4+A45					[126]
352	25.21	71.61				
384	2.99	7.79	79.39	64.4	5.4	24.7
$C_4H_{11}NO_3$	2*A2+A4*B4+A1+2*A30*C30+A45					[216]
443.6	2.41	5.43				
407.5	33.42	82.01	87.45	88.7	40.87	36.1
$C_4H_{12}Ge$	3*A2+A4*B4+3*A30*D30+A45					[34]
184.4	7.45	0	40.4	35.1	7.45	6.5
	4*A1+A102					[54]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_4H_{12}N_2$	237.5	1,2-diamino-2-methylpropane 15.46	65.11			
	256.1	2.23	8.71	73.81	62.2	15.9 [50]
$C_4H_{12}Pb$	242.9	2*A45+2*A1+A2+A4*B4 tetramethyllead 10.8	0	44.45	40.2	10.8 9.8 [216]
		4*A1+A106				
$C_4H_{12}Si$	174.0	teramethylsilane 6.74	0	38.73	43.2	6.74 7.5 [216]
		4*A1+A109				
$C_4H_{12}Sn$	218.2	tetramethyltin 9.23	0	42.32	46.1	9.23 10.1 [166, 125]
		4*A1+A110				
$C_5F_{11}N$	161	perfluoropiperidine 6.63	41.17			
	171.9	1.84	10.71			
	274.1	2.82	10.25	62.13	44.5	11.28 12.2 [216]
$C_5F_{13}N$	149.7	A14+3*A15+5*A17+A119+11*A28 perfluoromethyldiethylamine 7.16	0	47.83	49.3	7.16 7.2 [216]
		4*A26+5*A4*B4+A43+9*A25				
$C_5H_2Cl_3O$	448.1	3,5,6-trichloro-2-pyridinol 25.97	0	57.55	57.2	25.79 25.6 [215]
		3*A22*E22+A31+A41+A10+4*A12				
$C_5H_3F_7O_2$	191.4	methyl perfluorobutanoate 11.7	0	61.49	62.3	11.77 11.8 [216]
		A1+A38+4*A26+3*A25+3*A4*B4				
$C_5H_4O_2$	235.1	furfural 14.37	0	61.11	45.0	14.37 10.6 [216]
		A14+2*A15+2*A18+A18*B18+A19+A34+A112				
$C_5H_5F_3O_2$	232.4	trifluoromethyl (2-hydroxy-1-propenyl)ketone 8.45	0	36.36	53.4	8.45 12.4 [216]
		A4*B4+3*A25+A1+A6*B6+A7+A30*E30+A35				
C_5H_5N	231.5	pyridine 8.28	0	35.75	48.0	8.28 11.1 [216]
		5*A10+A41				
C_5H_6	176.6	cyclopentadiene 8.01	0	45.36	34.3	8.01 6.1 [216]
		A14+2*A15+4*A18				
$C_5H_6N_2$	244.2	1,3-dicyanopropane 12.59	0	51.55	63.5	12.59 15.5 [216]
		2*A56+3*A2*B2				
$C_5H_6N_2$	302.6	2,2-dicyanopropane 9.87	32.59			
	307.5	4.05	13.18	45.17	47.8	13.92 14.7 [216]
		2*A56+A4*B4+2*A1				
$C_5H_6N_2$	429.9	4-aminopyridine 20.07	0	46.68	54.5	20.07 23.4 [221]
		4*A10+A12+A41+A45				
$C_5H_6N_2O_2$	321.3	thymine 17.51	0	54.5	52.1	17.51 16.7 [216]
		A14+3*A15+2*A124+A18*B18+A19+A1				
C_5H_6O	181.9	2-methylfuran 8.55	0	47.03	41.0	8.55 7.5 [106]
		A1+A14+2*A15+2*A18+A19+A112+A18*B18				
$C_5H_6O_2$	258.6	furfuryl alcohol 13.1	0	50.75	48.7	13.1 12.6 [216]
		A2+A14+2*A15+2*A18+A19+A112+A18*B18+A30*B30				
C_5H_6S	207.8	2-methylthiophene 9.47	0	45.57	42.7	9.47 8.9 [275]
		A14+2*A15+A131+2*A18+A19+A1+A18*B18				
C_5H_6S	204.2	3-methylthiophene 10.54	0	51.62	41.2	10.54 8.4 [136]
		A14+2*A15+A131+A1+A19+2*A18*B18+A18				
C_5H_7N	216.9	N-methylpyrrole 7.82	0	36.07	29.6	7.82 6.4 [216]
		A14+2*A15+A1+2*A18+2*A18*B18+A119				
$C_5H_7NO_2$	246.8	ethyl cyanoacetate 11.78	0	47.73	57.3	11.78 14.1

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus} S_{tpcc}}$ (expt)	$\Delta_0^{T_{fus} S_{tpcc}}$ (calcd)	$\Delta_0^{T_{fus} H_{tpcc}}$ (expt)	$\Delta_0^{T_{fus} H_{tpcc}}$ (calcd)
C_5H_8	166.1					[216]
		2*A2+A1+A38+A56				
		spiropentane				
		6.43	0	38.7	28.5	6.43
		2*A14+A17-A15				[216]
C_5H_8	132.4					
		1-cis-3-pentadiene				
		5.64	0	42.61	50.7	5.64
		A1+A5+3*A6				[216]
C_5H_8	185.7					
		trans-1,3-pentadiene				
		7.14	0	38.46	50.7	7.14
		A1+A5+3*A6				[216]
C_5H_8	124.3					
		1,4-pentadiene				
		6.14	0	49.41	52.3	6.14
		A2+2*A5+2*A6				[216]
C_5H_8	127.3					
		2-methyl-1,3-butadiene				
		4.92	0	38.68	34.7	4.92
		A1+A7+A5+2*A6				[216]
C_5H_8	159.5					
		3-methyl-1,2-butadiene				
		7.95	0	49.84	39.0	7.95
		2*A1+A9+A5+A7				[216]
C_5H_8	147.5					
		2,3-pentadiene				
		6.13	0	44.82	42.9	6.13
		2*A1+2*A6+A9				[216]
C_5H_8	135.9					
		1,2-pentadiene				
		7.56	0	55.73	44.5	7.56
		A1+A2+A5+A6+A9				[216]
C_5H_8	87.07					
		cyclopentene				
		0.48	5.51			
	138.1					
		3.36	24.32	29.83	37.6	3.84
		A14+2*A15+2*A18				[216]
C_5H_8	138.5					
		methylene cyclobutane				
		5.86	0	42.31	42.2	5.86
		A14+A15+A5+A19				[216]
$C_5H_8Br_4$	433.5					
		pentaerythrityl tetrabromide				
		27.97	0	64.52	63.9	27.97
		4*A2+A4+4*A21				[216]
$C_5H_8Cl_2O$	292.2					
		3,3-bis-(chloromethyl)oxacyclobutane				
		16.95	0	58	50.4	16.95
		2*A2+2*A22*C22+A15+A14+A17+A112				[216]
$C_5H_8F_4$	367.4					
		pentaerythrityl tetrafluoride				
		5.14	13.97			
	249.4					
		13.21	53.14	67.11	44.3	18.35
		4*A2+4*A27+A4				[216]
$C_5H_8O_2$	118					
		δ -valerolactone				
		0.46	3.88			
	135					
		0.3	2.2			
	200					
		0.2	0.9			
	263					
		10.53	40.04	43.1	47.6	11.29
		3*A15+A14+A115				[32]
$C_5H_8O_2$	225					
		methyl methacrylate				
		12.24	0	54.4	49.5	12.24
		2*A1+A38+A7+A5				[216]
$C_5H_8O_2$	254.8					
		acetylacetone enol				
		14.5	0	56.91	66.0	14.5
		2*A1+A35*B35+A30*B30+A6*B6+A7				[60]
$C_5H_8O_3$	306.2					
		levulinic acid				
		9.22	0	30.11	52.5	9.22
		A1+2*A2+A35*B35+A36*B36				[215]
$C_5H_8O_4$	348.5					
		glutaric acid				
		2.46	7.07			
	371					
		20.9	56.33	63.4	60.2	23.36
		3*A2*B2+2*A36*B36				[216]
C_5H_9Cl	169.4					
		chlorocyclopentane				
		7.63	45.05			
	180					
		0.64	3.54	48.59	36.8	8.27
		A14+2*A15+A16+A22				[35]
C_5H_9N	213					
		2-cyano-2-methylpropane				
		0.23	1.09			
	232.7					
		1.91	7.78			
	292.1					
		9.29	31.8	40.67	47.6	11.43
						13.9

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{fus} S_{tpce}$ (expt)	$\Delta_0^{fus} S_{tpce}$ (calcd)	$\Delta_0^{fus} H_{tpce}$ (expt)	$\Delta_0^{fus} H_{tpce}$ (calcd)
$C_5H_{10}ClNO$	351.3	26.05	0	74.15	26.05	17.6
						[216]
C_5H_9NO	342.3	16.1	0	47.02	16.1	16.2
						[221]
C_5H_{10}	122	4.9	40.13			
	138	0.34	2.49			
	179.7	0.6	3.35	45.96	5.84	7.3
						[216]
C_5H_{10}	121.8	7.11	0	58.39	7.11	6.4
						[215]
C_5H_{10}	133.0	8.35	0	62.82	8.35	7.0
						[215]
C_5H_{10}	107.9	5.81	55	53.82	5.81	5.9
						[215]
C_5H_{10}	104.7	5.36	0	51.19	5.36	4.3
						[216]
C_5H_{10}	139.4	7.60	0	54.47	7.60	8.3
						[216]
C_5H_{10}	135.6	7.91	0	58.34	7.91	6.6
						[216]
C_5H_{10}	138.6	5.76	0	41.56	5.76	5.5
						[215]
$C_5H_{10}N_2O_2$	431	21.7	0	50.35	21.7	23.7
						[216]
$C_5H_{10}N_2O_2S$	352.7	21.73	0	61.61	21.73	18.7
						[216]
$C_5H_{10}N_2O_3$	508.0	56.6	0	111.43	56.6	34.4
						[221]
$C_5H_{10}N_4O_4$	369	21.8	59.08			
	374	2.8	7.49	66.57	24.6	27.5
						[147]
$C_5H_{10}O$	158.5	0.5	3.15			
	183.9	4.81	26.15			
	272.1	2.52	9.26	38.56	7.83	14.0
						[163]
$C_5H_{10}O$	118.5	0.11	0.96			
	180	0.01	0.04			
	234.2	11.59	49.5	50.5	11.71	12.7
						[341]
$C_5H_{10}O$	110	2.09	2.18			
	196.3	10.63	54.14	56.32	12.72	10.6
						[341]
$C_5H_{10}O$	202.8	3.71	18.28			
	257.4	1.54	5.98	24.27	5.24	7.2
						[216]
$C_5H_{10}O$	180.0	9.34	0	51.9	9.34	8.5
						[216]
$C_5H_{10}O_2$	239.5	14.16	0	59.14	14.16	14.1
						[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_5H_{10}O_2$	278.3	2,2-dimethylpropanoic acid (pivalic acid)				
	309.1	8.18	29.39			
$C_5H_{10}O_4$	426	2.27	7.34	36.74	43.3	10.45
	468	38.5	90.37			
$C_5H_{10}O_5$	334	21.9	0	65.6	65.4	21.9
		7.68				
$C_5H_{10}S$	172.4	8.87	0	51.48	46.5	8.87
		10.37				
$C_5H_{10}S$	201.4	1.1	5.44			
	240.0	7.77	32.38			
$C_5H_{10}S$	292.3	2.45	8.37	46.19	47.4	11.32
		7.83				
$C_5H_{10}S$	155.4	7.83	0	50.38	49.1	7.83
		14.37				
$C_5H_{11}Br$	185.1	14.37	77.61	77.61	72.5	14.37
		0.48				
$C_5H_{11}N$	184.5	8.31	43.65	46.23	47.4	8.79
	190.4					
$C_5H_{11}NO$	262.1	14.85	0	56.64	46.7	14.85
		18.8				
$C_5H_{11}NO_2$	457.4	24.68	70.26			
		2.73	7.12	77.38	64.4	27.41
$C_5H_{11}NO_2$	351.3	27.12	0	71.01	47.6	27.12
	383.6					
$C_5H_{11}NO_3S$	382.0	8.4	0	58.58	63.2	8.4
		5.13				
C_5H_{12}	143.5	3.26	12.69	31.1	35.5	5.83
C_5H_{12}	113.4	18.41	18.41			
C_5H_{12}	140	2.58	18.41			
	256.5	3.26	12.69	31.1	35.5	5.83
$C_5H_{12}NO_3PS$	321.0	20.49	0	63.85	51.7	20.49
$C_5H_{12}N_2O$	313.1	7.02	22.42			
	344.9	0.88	2.55			
$C_5H_{12}N_2O$	369.3	14.55	39.4	64.37	68.1	22.45
$C_5H_{12}N_2O$	249	0.1	0.41			
	449.8	33.13	73.65	74.06	52.4	33.23
$C_5H_{12}N_2O$	197.3	2.07	10.49			
	342.3	16.78	49.02	59.51	68.9	18.85

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
$C_5H_{12}N_2O$	339.4	1,3-diethylurea 1.87	5.51				
	383.4	12.46 $2*A1 + A66 + 2*A2$	32.5	38.01	50.9	14.33	19.5 [215, 124, 138]
$C_5H_{12}N_2O$	272.2	tetramethylurea 13.4	0	49.23	51.0	13.4	13.9 [216]
		$4*A1 + A63$					
$C_5H_{12}N_2O_2$	331	N-methyl-N-nitrobutanamine 37.56	0	113.46	101.7	37.56	33.7 [225]
		$2*A1 + 3*A2 + A51 + A47$					
$C_5H_{12}O$	146	2,2-dimethyl-1-propanol 1.96	13.43				
	213	0.17	0.79				
	264	4.46 $3*A1 + A4 + A7 + A30$	16.88	31.1	26.8	6.59	7.1 [277]
$C_5H_{12}O$	195.6	1-pentanol 10.5	0	53.7	56.7	10.5	11.1 [216]
		$A1 + 4*A2*B2 + A30$					
$C_5H_{12}O$	164.6	methyl <i>tert</i> -butyl ether 7.6	0	46.19	52.2	7.6	8.6 [216]
		$4*A1 + A4*B4 + A32$					
$C_5H_{12}O$	145.7	ethyl propyl ether 8.39	0	57.61	61.3	8.39	8.9 [216]
		$2*A1 + 3*A2 + A32$					
$C_5H_{12}O$	157.5	methyl <i>n</i> -butyl ether 10.85	0	68.9	61.3	10.85	9.7 [216]
		$2*A1 + 3*A2 + A32$					
$C_5H_{12}O_2$	248	1,5-pentanediol 15.72	0	63.6	82.9	15.72	20.6 [216]
		$5*A2*B2 + 2*A30*B30$					
$C_5H_{12}O_2$	146	2-methyl-2-butanol 1.96	13.44				
	213	0.17	0.78				
	264	4.46 $3*A1 + A4*B4 + A2 + A30$	16.88	31.1	38.8	6.59	10.2 [216]
$C_5H_{12}O_2$	315.2	2,2-dimethyl-1,3-propanediol 13.8	43.78				
	403.2	4.6	11.41	55.19	50.8	18.4	20.5 [90]
		$2*A1 + 2*A2 + A4 + 2*A30*B30$					
$C_5H_{12}O_3$	354	2-hydroxymethyl-2-methyl-1,3-propanediol 23.17	65.46				
	470	5.38 $A1 + 3*A2 + A4 + 3*A30*C30$	11.44	76.91	55.1	28.55	25.9 [216]
$C_5H_{12}O_4$	460.4	pentaerythritol 43.93	95.4				
	538.7	7.11 $A4 + 4*A2 + 4*A30*D30$	13.2	108.78	85.4	51.04	46.0 [216]
$C_5H_{12}O_5$	374.7	1,2,3,4,5-pentahydroxypentane (Ribitol) 37.6	0	100.35	90.4	37.6	39.9 [216]
		$2*A2 + 3*A3*B3 + 5*A30*E30$					
$C_5H_{12}O_5$	365.7	1,2,3,4,5-pentahydroxypentane (Xylitol) 37.4	0	102.27	90.4	37.4	33.1 [216]
		$2*A2 + 3*A3*B3 + 5*A30*E30$					
$C_5H_{12}O_5$	379.4	1,2,3,4,5-pentahydroxypentane (D-Arabitol) 38.9	0	102.53	90.4	38.9	34.3 [216]
		$2*A2 + 3*A3*B3 + 5*A30*E30$					
$C_5H_{12}S$	190.8	methyl <i>tert</i> -butyl sulfide 8.41	0	44.1	49.6	8.41	9.5 [216]
		$4*A1 + A4*B4 + A84$					
$C_5H_{12}S$	156.1	ethyl propyl sulfide 10.58	0	67.8	58.7	10.58	9.2 [216]
		$2*A1 + 3*A2 + A84$					
$C_5H_{12}S$	175.6	methyl butyl sulfide 12.45	0	70.9	58.7	12.45	10.3 [136]
		$2*A1 + 3*A2 + A84$					
$C_5H_{12}S$	139.6	3-methyl-1-butanethiol 7.41	0	53.05	56.1	7.41	7.8 [216]
		$2*A1 + A3 + 2*A2 + A86$					
$C_5H_{12}S$	197.5	1-pentanethiol 17.53	0	88.78	77.9	17.53	15.4

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus} S_{tpce}}$ (expt)	$\Delta_0^{T_{fus} S_{tpce}}$ (calcd)	$\Delta_0^{T_{fus} H_{tpce}}$ (expt)	$\Delta_0^{T_{fus} H_{tpce}}$ (calcd)
$C_3H_{12}S$	144.5 146.1	7.06 0.61	48.87 4.15	53.01 60.0	7.67	8.8 [216]
$C_3H_{12}S$	144.5 146.1	7.06 0.61	48.89 4.16	53.05 49.7	7.67	8.8 [105]
$C_3H_{12}SO_2$	357.6	24.69	0	69.03 47.7	24.69	17.1 [216]
$C_3H_{12}S_4$	296.4 318.7 338.7	6.11 7.61 4.14	20.5 23.85 12.13	56.48	55.9 14.78	18.9 [216]
$C_3H_{12}Si$	155.5	6.76	42.87	43.48	37.6	6.76 5.8 [153]
$C_3H_{12}Si$	141.7	7.66	0	54.06	46.9	7.66 6.6 [216]
$C_3H_{14}N_2$	194.4	12.38	0	63.7	55.7	12.38 10.8 [216]
C_6ClF_5	191 245 257.5	3.64 0.98 8.36	19.04 4.01 32.45	55.5	54.5 12.36	14.0 [216]
$C_6Cl_3F_3$	335.0	19.83	0	59.2	53.6	19.83 17.9 [215]
$C_6Cl_4O_2$	567.2	30.87	0	54.43	57.4	30.87 32.6 [215]
$C_6Cl_5NO_2$	418	18.41	0	44.04	53.8	18.41 22.5 [215]
C_6Cl_6	505	23.85	0	47.23	52.2	23.85 26.4 [215]
$C_6F_5NO_2$	250.5	11.81	0	47.13	56.0	11.81 14.0 [216]
C_6F_6	278.3	11.59	0	41.67	54.9	11.59 15.3 [216]
C_6F_{14}	103 185	0.97 6.84	10 36.82	46.82	74.0 7.8	13.7 [216, 67]
$C_6F_{15}N$	146.4 156.2	1.56 5.56	10.67 35.61	46.28	59.1 7.12	9.2 [216]
C_6N_4	472.2	24.92	0	52.77	49.5	24.92 23.4 [3]
C_6HBr_5O	441.5 502	11.29 19.14	25.57 38.13	63.7	63.1 30.43	31.7 [191]
$C_6HCl_4NO_2$	373.3	19.46	0	52.13	52.5	19.46 19.6 [215]
C_6HCl_5	357.7	20.6	0	57.59	50.9	20.6 18.2 [215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
C_6HCl_5O	462.5	pentachlorophenol 17.15	0	37.08	17.15	26.0 [215, 191]
C_6HF_5	225.7	$5*A22*F22+A31+6*A12$ pentafluorobenzene 10.88	0	48.24	10.88	12.0 [215]
C_6HF_5O	287 310.6	$5*A24+5*A12+A10$ pentafluorophenol 1.16 16.41	4.04 52.83	56.87	17.57	18.2 [72]
$C_6H_2Br_4$	306.8 453.1	$6*A12+A31+5*A24$ 1,2,4,5-tetrabromobenzene 0.34 27.88	1.09 61.53	62.62	28.22	25.0 [216]
$C_6H_2Cl_4$	320	$4*A21+4*A12+2*A10$ 1,2,3,4-tetrachlorobenzene 17	0	53.13	17	15.9 [215]
$C_6H_2Cl_4$	421.2	$4*A12+2*A10+4*A22*D22$ 1,2,4,5-tetrachlorobenzene 24.1	0	57.22	24.1	20.9 [215]
$C_6H_2Cl_4$	323.9	$4*A12+2*A10+4*A22*D22$ 1,2,3,5-tetrachlorobenzene 19	0	58.66	19	16.1 [215]
$C_6H_2Cl_5N$	505.8	$4*A12+2*A10+4*A22*D22$ pentachloroaniline 18.7	0	36.97	18.7	29.0 [215]
$C_6H_2F_4$	233.3	$6*A12+5*A22*F22+A45$ 1,2,3,4-tetrafluorobenzene 10.93	0	46.85	10.93	12.0 [65]
$C_6H_2F_4$	226.9	$4*A12+2*A10+4*A24$ 1,2,3,5-tetrafluorobenzene 10.67	0	47.01	10.67	11.7 [65]
$C_6H_2F_4$	277	$4*A12+2*A10+4*A24$ 1,2,4,5-tetrafluorobenzene 15.05	0	54.31	15.05	14.3 [65]
$C_6H_2F_5N$	287.4 306.8	pentafuoroaniline 3.94 14.27	13.71 46.51	60.22	18.21	18.3 [216]
$C_6H_3BrCl_2O$	343.4	$5*A24+A45+6*A12$ 4-bromo-2,5-dichlorophenol 22.11	0	64.39	22.11	18.9 [221]
$C_6H_3Br_3O$	366.2	$2*A10+4*A12+2*A22*D22+A21+A31$ 2,4,6-tribromophenol 18.52	0	50.57	18.52	21.2 [215]
$C_6H_3Cl_3$	326.9	$4*A12+2*A10+3*A21+A31$ 1,2,3-trichlorobenzene 20.5	0	62.71	20.5	15.8 [215]
$C_6H_3Cl_3$	336.7	$3*A10+3*A12+3*A22*C22$ 1,3,5-trichlorobenzene 18.2	0	54.05	18.2	16.3 [215]
$C_6H_3Cl_3O$	340.3	$3*A10+3*A12+3*A22*C22$ 2,4,5-trichlorophenol 21.59	0	63.44	21.59	18.3 [221]
$C_6H_3Cl_4N$	337.2	$4*A12+2*A10+3*A22*D22+A31$ 2-chloro-6-(trichloromethyl)pyridine 20.3	0	60.2	20.3	19.6 [216]
$C_6H_3N_3O_6$	370 380.3	$3*A10+A41+A11+A12+A4*B4+4*A22*E22$ 1,3,5-trinitrobenzene 1.9 14.8	5.13 38.95	44.08	16.71	20.2 [216]
$C_6H_3N_3O_7$	394.1	$3*A10+3*A12+3*A50*C50$ picric acid 17.1	0	43.39	17.1	23.0 [216]
$C_6H_3N_3O_8$	454.8	$2*A10+4*A12+3*A50+A31$ 2,4,6-trinitroresorcinol 33.5	0	73.66	33.5	29.0 [216]
C_6H_4BrCl		$A10+5*A12+2*A31+3*A50$ 1,2-bromochlorobenzene				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{pcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{pcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{pcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{pcc}}$ (calcd)
260.6	12.37	0	47.47	48.4	12.37	12.6
$\text{C}_6\text{H}_4\text{BrCl}$	$A22*B22+A21+2*A12+4*A10$ 1,3-bromochlorobenzene					[216]
252.0	12.29	0	48.77	48.4	12.29	12.2
$\text{C}_6\text{H}_4\text{BrCl}$	$A22*B22+A21+2*A12+4*A10$ 1,4-bromochlorobenzene					[216]
337.8	18.76	0	55.54	48.4	18.76	16.4
$\text{C}_6\text{H}_4\text{BrI}$	$A22*B22+A21+2*A12+4*A10$ 1,2-bromiodobenzene					[216]
294.2	14.42	0	49.01	51.6	14.42	15.2
$\text{C}_6\text{H}_4\text{BrI}$	$4*A10+2*A12+A21+A29$ 1,3-bromiodobenzene					[215]
282.5	12.16	0	43.04	51.6	12.16	14.6
$\text{C}_6\text{H}_4\text{BrI}$	$4*A10+2*A12+A21+A29$ 1,4-bromiodobenzene					[215]
363.3	19.13	0	52.66	51.6	19.13	18.8
$\text{C}_6\text{H}_4\text{Br}_2$	$4*A10+2*A12+A21+A29$ 1,2-dibromobenzene					[215]
275	12.61	0	45.58	49.8	12.61	13.7
$\text{C}_6\text{H}_4\text{Br}_2$	$4*A10+2*A12+2*A21$ 1,3-dibromobenzene					[215]
266.3	13.21	0	49.61	49.8	13.21	13.3
$\text{C}_6\text{H}_4\text{Br}_2$	$4*A10+2*A12+2*A21$ 1,4-dibromobenzene					[215]
360.1	20.04	0	55.65	49.8	20.04	17.9
$\text{C}_6\text{H}_4\text{Br}_2\text{O}$	$2*A21+4*A10+2*A12$ 2,4-dibromophenol					[215]
313	14.64	0	46.79	55.2	14.64	17.3
$\text{C}_6\text{H}_4\text{ClNO}_2$	$3*A10+3*A12+2*A21+A31$ 1,2-chloronitrobenzene					[215]
308.2	19.08	0	61.9	48.6	19.08	15.0
$\text{C}_6\text{H}_4\text{ClNO}_2$	$4*A10+2*A12+A22*B22+A50$ 1,4-nitrochlorobenzene					[228]
354.6	11.85	0	33.42	48.6	11.85	17.2
$\text{C}_6\text{H}_4\text{ClNO}_2$	$4*A10+2*A12+A50+A22*B22$ 1,3-nitrochlorobenzene					[216]
317.6	19.37	0	60.99	48.6	19.37	15.4
$\text{C}_6\text{H}_4\text{Cl}_2$	$A22*B22+A50+4*A10+2*A12$ 1,2-dichlorobenzene					[215]
256.5	12.93	0	50.41	47.1	12.93	12.1
$\text{C}_6\text{H}_4\text{Cl}_2$	$4*A10+2*A12+2*A22*B22$ 1,3-dichlorobenzene					[215]
248.4	12.64	0	50.89	47.1	12.64	11.7
$\text{C}_6\text{H}_4\text{Cl}_2$	$4*A10+2*A12+2*A22*B22$ 1,4-dichlorobenzene					[215]
326	18.16	0	55.65	47.1	18.16	15.3
$\text{C}_6\text{H}_4\text{Cl}_2\text{N}_2\text{O}_2$	$2*A22*B22+4*A10+2*A12$ 2,6-dichloro-4-nitroaniline					[215]
466.8	32.64	0	69.92	56.4	32.64	26.3
$\text{C}_6\text{H}_4\text{Cl}_2\text{O}$	$4*A12+2*A10+A45+2*A22*D22+A50$ 2,3-dichlorophenol					[215]
330	21.36	0	64.73	52.4	21.36	17.3
$\text{C}_6\text{H}_4\text{Cl}_2\text{O}$	$3*A10+3*A12+A31+2*A22*C22$ 2,4-dichlorophenol					[215]
318	20.09	0	63.18	52.4	20.09	16.7
$\text{C}_6\text{H}_4\text{Cl}_2\text{O}$	$3*A10+3*A12+A31+2*A22*C22$ 2,5-dichlorophenol					[216]
331	22.43	0	67.76	52.4	22.43	17.4
$\text{C}_6\text{H}_4\text{Cl}_2\text{O}$	$3*A10+3*A12+A31+2*A22*C22$ 2,6-dichlorophenol					[216]
340	22.14	0	65.12	52.4	22.14	17.8
$\text{C}_6\text{H}_4\text{Cl}_2\text{O}$	$3*A10+3*A12+A31+2*A22*C22$ 3,4-dichlorophenol					[216]
341	20.93	0	61.38	52.4	20.93	17.9
$\text{C}_6\text{H}_4\text{Cl}_2\text{O}$	$3*A10+3*A12+A31+2*A22*C22$ 3,5-dichlorophenol					[216]
341	20.51	0	60.15	52.4	20.51	17.9
$\text{C}_6\text{H}_4\text{F}_2$	$3*A10+3*A12+A31+2*A22*C22$ 1,2-difluorobenzene					[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$\text{C}_6\text{H}_4\text{F}_2$	226	11.05 4*A10+2*A12+2*A24 1,3-difluorobenzene	0	48.95	48.0	11.05	10.8 [216]
	186.8	0.83	4.43				
	204.0	8.58	42.05	46.48	48.0	9.4	9.8 [216]
$\text{C}_6\text{H}_4\text{I}_2$		2*A24+2*A12+4*A10 1,2-diiodobenzene					
	296.6	14.01	0	47.24	53.5	14.01	15.9 [215]
$\text{C}_6\text{H}_4\text{I}_2$		2*A29+2*A12+4*A10 1,3-diiodobenzene					
	307.4	15.93	0	51.82	53.5	15.93	16.4 [215]
$\text{C}_6\text{H}_4\text{I}_2$		2*A29+2*A12+4*A10 1,4-diiodobenzene					
	402	22.37	0	55.65	53.5	22.37	21.5 [215]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_4$		2*A29+2*A12+4*A10 1,2-dinitrobenzene					
	396.1	22.84	0	57.66	50.2	22.84	19.9 [216]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_4$		4*A10+2*A12+2*A50 1,3-dinitrobenzene					
	363.2	17.36	0	47.82	50.2	17.36	18.2 [215]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_4$		4*A10+2*A12+2*A50 1,4-dinitrobenzene					
	446.7	28.12	0	62.93	50.2	28.12	22.4 [215]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_5$		4*A10+2*A12+2*A50 2,3-dinitrophenol					
	417	26.24	0	62.93	55.5	26.24	23.2 [216]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_5$		3*A10+3*A12+2*A50+A31 2,4-dinitrophenol					
	388	24.17	0	62.29	55.5	24.17	21.6 [216]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_5$		3*A10+3*A12+2*A50+A31 2,5-dinitrophenol					
	381	23.73	0	62.28	55.5	23.73	21.2 [216]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_5$		3*A10+3*A12+2*A50+A31 2,6-dinitrophenol					
	336	19.58	0	58.27	55.5	19.58	18.7 [216]
$\text{C}_6\text{H}_4\text{N}_2\text{O}_5$		3*A10+3*A12+2*A50+A31 3,4-dinitrophenol					
	407	25.37	0	62.33	55.5	25.37	22.6 [216]
$\text{C}_6\text{H}_4\text{O}_2$		3*A10+3*A12+2*A50+A31 <i>p</i> -benzoquinone					
	388	18.45	0	47.56	-29.4	18.45	11.4 [215]
$\text{C}_6\text{H}_5\text{Br}$		3*A15+A14+4*A18*B18+2*A114 bromobenzene					
	242.4	10.7	0	44.2	47.1	10.7	11.4 [216]
$\text{C}_6\text{H}_5\text{BrO}$		5*A10+A12+A21 4-bromophenol					
	336	16.57	0	49.32	52.5	16.57	17.6 [216]
$\text{C}_6\text{H}_5\text{Cl}$		A21+4*A10+2*A12+A31 chlorobenzene					
	227.9	9.55	0	41.92	40.4	9.55	9.2 [216]
$\text{C}_6\text{H}_5\text{ClO}$		5*A10+A22+A12 2-chlorophenol					
	276	0.09	0.33				
$\text{C}_6\text{H}_5\text{ClO}$	283	12.52	44.24	44.57	51.2	12.61	14.5 [215]
		4*A10+2*A12+A22*B22+A31 3-chlorophenol					
$\text{C}_6\text{H}_5\text{ClO}$	305.8	14.91	0	48.76	51.2	14.91	15.6 [215]
		4*A10+2*A12+A22*B22+A31 4-chlorophenol					
$\text{C}_6\text{H}_5\text{ClO}$	315.9	14.07	0	44.54	51.2	14.07	16.2 [215]
		4*A10+2*A12+A22*B22+A31 2,6-dichloro-4-benzenamine					
$\text{C}_6\text{H}_5\text{Cl}_2\text{N}$	467.2	29.48	0	63.11	53.5	29.48	25.0 [221]
		3*A10+3*A12+2*A22*C22+A45 phenyltrichlorosilane					
$\text{C}_6\text{H}_5\text{Cl}_3\text{Si}$	233.4	11.66	0	49.96	48.9	11.66	11.4 [216]
		5*A10+A11+3*A22*D22+A109 fluorobenzene					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}S_{tpcc}}$ (expt)	$\Delta_0^{T_{fus}S_{tpcc}}$ (calcd)	$\Delta_0^{T_{fus}H_{tpcc}}$ (expt)	$\Delta_0^{T_{fus}H_{tpcc}}$ (calcd)
230.9	11.31	0	48.95	46.2	11.31	10.7
C_6H_5I	5*A10+A12+A24 iodobenzene					[250]
241.8	9.75	0	40.31	49.0	9.75	11.8
$C_6H_5NO_2$	5*A10+A12+A29 picolinic acid					[216]
411	30	0	72.99	49.2	30	20.2
$C_6H_5NO_2$	4*A10+A12+A41+A36*B36 nicotinic acid					[216]
452	0.78	1.73				
510	26.7	52.35	54.08	49.2	27.48	25.1
$C_6H_5NO_2$	4*A10+A12+A41+A36*B36 isonicotinic acid					[182]
593	135	0	227.66	49.2	135	29.2
$C_6H_5NO_2$	4*A10+A12+A41+A36*B36 nitrobenzene					[216]
278.8	12.12	0	43.5	47.3	12.12	13.2
$C_6H_5NO_3$	5*A10+A12+A50 <i>o</i> -nitrophenol					[216]
318.2	17.45	0	54.83	52.7	17.45	16.8
$C_6H_5NO_3$	4*A10+2*A12+A50+A31 <i>m</i> -nitrophenol					[215,188]
371.2	19.19	0	51.7	52.7	19.19	19.6
$C_6H_5NO_3$	4*A10+2*A12+A50+A31 <i>p</i> -nitrophenol					[215,188]
388.2	18.25	0	47.02	52.7	18.25	20.5
C_6H_6	4*A10+2*A12+A50+A31 benzene					[216,188]
278.7	9.87	0	35.4	44.5	9.87	12.4
$C_6H_6Cl_6$	6*A10 1 α ,2 α ,3 β ,4 α ,5 α ,6 β -hexachlorocyclohexane					[216]
386.8	22.13	0	57.23	53.2	22.13	20.6
$C_6H_6Cl_6$	A14+3*A15+6*A16+6*A22*F22 1 α ,2 α ,3 β ,4 α ,5 α ,6 β -hexachlorocyclohexane (lindane)					[221]
388.9	15.9	0	40.88	53.2	15.9	20.7
$C_6H_6N_2O_2$	A14+3*A15+6*A16+6*A22*F22 2-nitroaniline					[221]
342.5	16.11	0	47.0	53.8	16.11	18.5
$C_6H_6N_2O_2$	4*A10+A45+A50+2*A12 3-nitroaniline					[216]
387	23.68	0	61.16	53.8	23.68	20.8
$C_6H_6N_2O_2$	4*A10+A45+A50+2*A12 4-nitroaniline					[216]
420.7	21.09	0	50.1	53.8	21.09	22.6
$C_6H_6N_6O_{14}$	4*A10+A45+A50+2*A12 2,2,2-trinitroethyl 4,4,4-trinitrobutyrate					[216]
362.7	25.94	71.52				
366.5	6.69	18.27	89.79	89.7	32.64	32.9
C_6H_6O	3*A2+2*A4*B4+A38+6*A50 phenol					[122]
314	11.51	0	36.82	49.9	11.51	15.7
$C_6H_6O_2$	5*A10+A31+A12 1,4-dihydroxybenzene					[216]
445	26.48	0	59.5	59.5	26.48	25.0
$C_6H_6O_2$	4*A10+2*A31+2*A12 1,2-dihydroxybenzene					[199]
376.9	22.01	0	58.39	55.2	22.01	20.8
$C_6H_6O_2$	4*A10+2*A31+2*A12 1,3-dihydroxybenzene					[199]
366.8	1.2	3.27				
382.6	18.9	49.41	52.64	55.2	20.1	21.1
$C_6H_6O_3$	4*A10+2*A31+2*A12 1,2,3-trihydroxybenzene					[199]
407.2	18.55	0	45.56	60.6	18.55	24.7
C_6H_6S	3*A10+3*A12+3*A31 thiophenol					[4]
258.3	11.4	0	44.3	52.6	11.3	13.6
C_6H_7N	5*A10+A12+A86 aniline					[281]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
267.1	10.54	0	39.45	51.0	10.54	14.1
$\text{C}_6\text{H}_7\text{N}$	5*A10+A12+A45					[216]
	2-methylpyridine					
206.5	9.72	0	47.1	48.5	9.72	10.0
$\text{C}_6\text{H}_7\text{N}$	4*A10+A11+A1+A41					[216]
	3-methylpyridine					
255	14.18	0	55.62	48.5	14.18	12.4
$\text{C}_6\text{H}_7\text{NO}$	4*A10+A11+A1+A41					[216]
	<i>o</i> -aminophenol					
447.4	34	0	75.99	56.3	34	25.2
$\text{C}_6\text{H}_7\text{NO}$	4*A10+2*A12+A31+A45					[216]
	<i>m</i> -aminophenol					
399	22.98	0	57.59	56.3	22.98	22.5
$\text{C}_6\text{H}_7\text{NO}$	4*A10+2*A12+A31+A45					[139]
	<i>p</i> -aminophenol					
459.5	31.2	0	67.9	56.3	31.2	25.9
462.5	26	0	56.22	56.3	26.0	26.0
C_6H_8	4*A10+2*A12+A31+A45					[216,223]
	1,3-cyclohexadiene					
161	4.2	0	26.1	38.0	4.2	6.1
C_6H_8	A14+3*A15+4*A18					[216]
	1,4-cyclohexadiene					
192	0.82	4.25				
224	5.72	25.51	29.76	38.0	6.53	8.5
$\text{C}_6\text{H}_8\text{N}_2$	A14+3*A15+4*A18					[216]
	<i>o</i> -phenylenediamine					
373.9	23.1	0	61.78	57.43	23.1	21.47
$\text{C}_6\text{H}_8\text{N}_2$	4*A10+2*A12+2*A45					[216]
	<i>m</i> -phenylenediamine					
335.5	15.4	0	45.9	57.43	15.4	19.27
$\text{C}_6\text{H}_8\text{N}_2$	4*A10+2*A12+2*A45					[216]
	<i>p</i> -phenylenediamine					
412.3	21.7	0	52.63	57.43	21.7	23.68
$\text{C}_6\text{H}_8\text{N}_2$	4*A10+2*A12+2*A45					[216]
	phenylhydrazine					
292.8	16.43	0	56.11	45.7	16.43	13.4
$\text{C}_6\text{H}_8\text{N}_2\text{O}_2$	5*A10+A12+A45+A44					[215]
	N-acetylglycine amide					
408.2	25.6	0	62.71	54.1	25.6	22.1
$\text{C}_6\text{H}_8\text{N}_2\text{O}_2$	A1+A60+A2+A61					[278]
	1,3-dimethyluracil					
398	14.6	0	36.68	30.0	14.6	11.9
$\text{C}_6\text{H}_8\text{N}_4\text{O}_2$	2*A1+A14+3*A15+2*A125+2*A18*B18					[292]
	bis(2-cyanoethyl)-N-nitroamine					
327	44.99	0	137.57	109.1	44.99	35.7
$\text{C}_6\text{H}_8\text{O}_2$	4*A2+2*A56+A51+A47					[225]
	1,4-cyclohexanedione					
322.2	6.15	19.09				
339.2	0.96	2.83				
348.2	10.04	28.84	50.76	41.8	17.15	14.5
$\text{C}_6\text{H}_8\text{O}_4$	3*A15+A14+2*A114					[114]
	dimethyl maleate					
254	14.64	0	57.74	58.4	14.64	14.8
$\text{C}_6\text{H}_8\text{O}_4$	2*A1+2*B6*A6+2*A38					[216]
	dimethyl fumarate					
375	35.15	0	93.72	58.4	35.15	21.9
$\text{C}_6\text{H}_8\text{O}_4$	2*A1+2*A38+2*B6*A6					[216]
	DL 3,6-dimethyl-1,4-dioxane-2,5-dione					
397.5	24.7	0	62.14	49.1	24.7	19.5
$\text{C}_6\text{H}_8\text{S}$	A14+A15+2*A115+2*A16+2*A1					[216]
	2,5-dimethylthiophene					
210.6	8.91	0	42.31	51.1	8.91	10.8
$\text{C}_6\text{H}_8\text{ClO}_2$	2*A1+A131+A14+2*A15+2*A19+2*A18					[216]
	chloroethyl methacrylate					
235.1	17	0	72.31	62.3	17	14.7
$\text{C}_6\text{H}_9\text{N}$	2*A2+A22*B22+A5+A7+A1+A38					[216]
	2,4-dimethylpyrrole					
268.5	9.6	0	35.75	48.9	9.6	13.1

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (calcd)	
	A14+2*A15+A18+2*A1+A121+A18*B18+2*A19					[70]	
$\text{C}_6\text{H}_9\text{N}$	2,5-dimethylpyrrole						
280.9	9.3	0	33.09	50.4	9.3	14.2	
	A14+2*A15+2*A18+2*A1+A121+2*A19					[216]	
$\text{C}_6\text{H}_9\text{NS}$	2,4,5-trimethylthiazole						
240.7	9	0	37.39	60.2	9	14.5	
	A14+2*A15+3*A19+3*A1+A118+A131					[61]	
C_6H_{10}	cyclohexane						
138.7	4.23	30.5					
169.7	3.28	19.35	49.85	41.3	7.51	7.0	
	A14+3*A15+2*A18					[216]	
$\text{C}_6\text{H}_{10}\text{N}_2\text{O}$	2,3-diazabicyclo[2.2.2]oct-2-ene N-oxide						
359.3	5.02	13.97					
399.3	8.05	20.16					
438	3.84	8.77	42.9	47.6	16.91	20.8	
	2*A14+2*A15+2*A16+A123					[42]	
$\text{C}_6\text{H}_{10}\text{O}$	cyclohexanone						
220.8	8.66	39.22					
245.2	1.33	5.42	44.64	43.1	9.99	10.6	
	A14+3*A15+A114					[156]	
$\text{C}_6\text{H}_{10}\text{O}$	cyclohexene oxide						
193.1	9.54	49.38					
238.1	1.06	4.47	53.85	42.2	10.6	10.1	
	2*A14+A15+A112+2*A16					[156]	
$\text{C}_6\text{H}_{10}\text{O}_2$	ϵ -caprolactone						
272	13.82	50.81	50.79	51.3	13.82	14.0	
	A14+4*A15+A115					[32]	
$\text{C}_6\text{H}_{10}\text{O}_3$	2,2-dimethyltrimethylene carbonate						
324.1	10.3	31.78					
387.2	5.62	14.52	46.31	46.3	15.92	17.9	
	A14+3*A15+A116+2*A1+A17					[200]	
$\text{C}_6\text{H}_{10}\text{O}_4$	adipic acid						
426.4	34.85	0	81.73	69.6	34.85	29.7	
	4*A2*B2+2*A36*B36					[340]	
$\text{C}_6\text{H}_{10}\text{O}_6$	(dl) dimethyl tartrate						
360.2	26.94	0	74.81	76.7	26.94	27.6	
	2*A38+2*A3*B3+2*A1+2*A30*D30					[220]	
$\text{C}_6\text{H}_{10}\text{O}_6$	(d) dimethyl tartrate						
322.2	17.36	0	53.89	76.7	17.36	24.7	
	2*A38+2*A3*B3+2*A1+2*A30*D30					[220]	
$\text{C}_6\text{H}_{11}\text{Br}$	bromocyclohexane						
216.9	10.79	0	49.75	47.3	10.79	10.3	
	A14+3*A15+A21+A16					[190]	
$\text{C}_6\text{H}_{11}\text{Cl}$	chlorocyclohexane						
120	0.05	0.42					
220.4	8.01	36.35					
229.3	2.04	8.91	45.67	40.5	10.1	9.3	
	A14+3*A15+A16+A22					[229]	
$\text{C}_6\text{H}_{11}\text{NO}$	cyclohexanone oxime						
240.8	0.01	0.06					
362.6	12.7	35.02					
273.4	0.09	0.34	35.43	45.8	12.81	12.5	
	A14+3*A15+A19+A53					[5]	
$\text{C}_6\text{H}_{11}\text{NO}$	ϵ -caprolactam						
343.3	16.1	0	46.89	50.9	16.1	17.5	
	A14+4*A15+A124					[6]	
$\text{C}_6\text{H}_{11}\text{NO}_3$	N-dimethylaminosuccinamic acid						
431.4	36.97	0	85.71	54.4	36.97	23.5	
	2*A1+2*A2+A36*B36+A59					[221]	
$\text{C}_6\text{H}_{11}\text{N}_2\text{O}_3\text{PS}_2$	S-2,3-dihydro-5-methoxy-2-oxo-1,3,4-thiadiazol-3-ylmethyl O,O-dimethyl phosphorodithioate						
315.1	28.54	0	90.59	90.7	28.54	28.6	
	3*A1+A14+2*A15+A138+A19+A118+A32+A80					[221]	
C_6H_{12}	cyclohexane						
186.1	6.74	36.2					
279.8	2.68	9.57	45.77	44.5	9.41	12.5	
	A14+3*A15					[216]	
C_6H_{12}	methylcyclopentane						
130.7	6.93	0	53.01	43.6	6.93	5.7	
	A14+A16+A1+2*A15					[216]	

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (calcd)
C_6H_{12}	133.4	1-hexene 9.35	0	70.1	61.6	9.37	8.2 [216]
C_6H_{12}	196.8 198.9	$A1+3*A2+A5+A6$ 2,3-dimethyl-2-butene 3.53 6.44	17.94 32.39	50.34	48.9	9.97	9.6 [216]
C_6H_{12}	124.9 158.4	$4*A1+2*A7$ 3,3-dimethyl-1-butene 4.35 1.09	34.84 6.87	41.71	40.5	5.44	5.1 [216]
C_6H_{12}	132	$A4+3*A1+A5+A6$ <i>cis</i> -2-hexene 8.88	0	67.27	59.9	8.88	7.9 [165]
$\text{C}_6\text{H}_{12}\text{N}_2$	351.1 433	$2*A1+2*A2+2*A6$ 1,4-diazabicyclo[2.2.2]octane 10.54 7.45	30.08 17.15	47.24	35.6	18.0	15.4 [216]
$\text{C}_6\text{H}_{12}\text{N}_2\text{O}_3$	480.1	$2*A14+2*A15+2*A119$ β -alanyl- β -alanine 58.3	0	121.45	81.4	58.3	39.1 [216]
$\text{C}_6\text{H}_{12}\text{N}_2\text{O}_3$	483.2	$4*A2+A45+A36*C36+A60$ α -alanyl- α -alanine (DL) 33.2	0	68.72	68.4	33.2	33.0 [216]
$\text{C}_6\text{H}_{12}\text{O}$	265.5 299.1	$2*A1+A45+A36*C36+A60+2*A3*B3$ cyclohexanol 8.8 1.8	33.3 6.0	39.3	31.5	9.9	9.4 [81]
$\text{C}_6\text{H}_{12}\text{O}$	310.2	$A14+3*A15+A16+A30$ 1-methylcyclopentanol 8.41	0	27.11	25.5	8.41	7.9 [230]
$\text{C}_6\text{H}_{12}\text{O}$	214.9 243.2	$A14+2*A15+A1+A17+A30$ hexanal 13.3 0.34	61.89 1.38	63.27	76.4	13.64	18.6 [128, 168]
$\text{C}_6\text{H}_{12}\text{O}$	221.7	$A14+4*A2*B2+A34$ 3,3-dimethyl-2-butanone 11.34	0	51.04	52.0	11.34	11.5 [216]
$\text{C}_6\text{H}_{12}\text{O}$	145 217.7	$4*A1+A4*B4+A35$ 3-hexanone 0.68 13.47	4.7 61.89	66.61	61.1	14.15	13.3 [216]
$\text{C}_6\text{H}_{12}\text{O}$	217.7	$2*A1+3*A2+A35$ 2-hexanone 14.9	68.42	68.41	61.1	14.9	13.3 [216]
$\text{C}_6\text{H}_{12}\text{O}_2$	229.6	$2*A1+3*A2+A35$ 2,2-dimethyl-1,3-dioxane 12.1	0	52.7	47.5	12.1	10.9 [47]
$\text{C}_6\text{H}_{12}\text{O}_2$	360.4 371.6	$A14+3*A15+2*A1+A17+2*A112$ <i>cis</i> -1,2-cyclohexanediol 19.89 3.32	55.19 8.93	64.12	51.2	23.21	19.0 [204]
$\text{C}_6\text{H}_{12}\text{O}_2$	372.3	$A14+3*A15+2*A30*B30+2*A16$ <i>trans</i> -1,2-cyclohexanediol 18.51	0	49.72	51.2	18.51	19.1 [204]
$\text{C}_6\text{H}_{12}\text{O}_3$	142.7 147.5 285.7	$A14+3*A15+2*A30*B30+2*A16$ 2,4,6-trimethyl-1,3,5-trioxane 0.26 0.77 13.52	1.81 5.24 47.32	54.37	56.7	14.55	16.2 [216]
$\text{C}_6\text{H}_{12}\text{O}_6$	414	$3*A1+3*A16+A14+3*A15+3*A112$ α -D-glucose 31.42	0	75.9	93.0	31.42	38.5 [216]
$\text{C}_6\text{H}_{12}\text{O}_6$	496.9	$A14+3*A15+5*A30*F30+A2+5*A16+A112$ myo-inositol 47.9	0	96.4	92.7	47.9	46.1 [216]
$\text{C}_6\text{H}_{12}\text{S}$	165 169.9	$A14+3*A15+6*A16+6*A30*F30$ cyclopentyl methyl sulfide 0.9 9.2	5.44 54.31	59.75	45.7	10.1	7.8 [105]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_6H_{12}S$	189.6	cyclohexanethiol 10 A14+3*A15+A86+A16	0	52.72	52.8	10.0	10.0 [341]
$C_6H_{13}Br$	188.1	1-bromohexane 18.05 A1+5*A2*B2+A21	0	95.98	81.8	18.05	15.4 [216]
$C_6H_{13}N$	269.4	2-methylpiperidine 18.58 A14+3*A15+A121+A1+A16	0	68.99	49.5	18.58	13.3 [216]
$C_6H_{13}NO$	374	hexanamide 25.1 4*A2*B2+A1+A61	0	67.12	82.8	25.1	31.0 [279]
C_6H_{14}	177.8	<i>n</i> -hexane 13.08 2*A1+4*A2*B2	0	73.22	72.5	13.08	12.9 [216]
C_6H_{14}	136.1 107 145.2	2,3-dimethylbutane 6.43 2.37 0.79 4*A1+2*A3	47.22 22.13 5.47	52.96	37.6	9.59	5.1 [216]
C_6H_{14}	110.3	3-methylpentane 5.31 2*A2+3*A1+A3	0	48.17	50.6	5.31	5.6 [216]
C_6H_{14}	119.6	2-methylpentane 6.27 2*A2+3*A1+A3	0	52.43	50.6	6.27	6.05 [216]
C_6H_{14}	126.8 140.8 174.3	2,2-dimethylbutane 5.4 0.28 0.58 4*A1+A2+A4	42.57 2.02 3.31	45.88	42.6	6.26	7.4 [216]
$C_6H_{14}O$	225.8	1-hexanol 15.48 A1+5*A2*B2+A30	0	68.56	66.0	15.48	14.9 [216]
$C_6H_{14}O$	187.8	isopropyl ether 12.05 4*A1+2*A3*B3+A32	0	64.02	55.7	12.05	10.5 [66]
$C_6H_{14}O$	158.4	4-oxaheptane 10.77 2*A1+4*A2+A32	0	67.99	68.4	10.77	10.8 [216]
$C_6H_{14}O_2$	316.2	2,3-dimethyl-2,3-butanediol 14.7 4*A1+2*A4*B4+2*B30*A30	0	46.49	60.8	14.7	19.2 [231]
$C_6H_{14}O_2$	320.6	1,6-hexanediol 25.52 2*A30*B30+6*A2*B2	0	79.6	92.2	25.52	29.0 [216]
$C_6H_{14}O_3$	209.1	2,5,8-trioxanonane 17.8 2*A1+4*A2+3*A32	0	85.1	77.8	17.8	16.3 [216]
$C_6H_{14}O_6$	366.5	D sorbitol 30.2 2*A2+4*A3*B3+6*A30*F30	0	82.4	111.7	30.2	40.9 [216]
$C_6H_{14}O_6$	460.3	dulcitol 65.1 2*A2+4*A3*B3+6*A30*F30	0	141.4	111.7	65.1	51.4 [216]
$C_6H_{14}O_6$	439.1 438.7	D mannitol 56.1 50.6 2*A2+4*A3*B3+6*A30*F30	0 0	127.8 115.3	111.7 111.7	56.1 50.6	49.0 49.0 [216, 394]
$C_6H_{14}S$	195.1	diisopropyl sulfide 10.42 4*A1+2*A3*B3+A84	0	53.39	52.8	10.42	10.4 [341]
$C_6H_{14}S$	170.4	dipropyl sulfide 12.14 2*A1+4*A2+A84	0	71.25	65.8	12.14	11.2 [216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}}S_{tpce}$ (expt)	$\Delta_0^{T_{fus}}S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}}H_{tpce}$ (expt)	$\Delta_0^{T_{fus}}H_{tpce}$ (calcd)	
$C_6H_{14}S$	butyl ethyl sulfide 12.39	0	69.57	65.8	12.39	11.7 [136]	
$C_6H_{14}S$	2*A1 + 4*A2 + A84 1-hexanethiol 18.03	0	93.51	87.2	18.03	16.8 [216]	
$C_6H_{14}S_2$	5*A2*B2 + A1 + A86 dipropyl disulfide 13.81	0	73.55	73.3	13.81	13.8 [216]	
$C_6H_{15}Al$	2*A1 + 4*A2 + A85 triethylaluminum 10.6	0	47.11	49.5	10.6	11.1 [216]	
$C_6H_{15}As$	3*A1 + 3*A2 + A97 triethylarsine 11.06	0	60.83	67.7	11.06	12.3 [216]	
$C_6H_{15}B$	3*A1 + 3*A2 + A98 triethylborane 11.85	0	65.7	57.0	11.85	10.3 [216]	
$C_6H_{15}Bi$	3*A1 + 3*A2 + A99 triethylbismuth 8.7	0	59.64	59.7	8.7	8.7 [167]	
$C_6H_{15}Ga$	3*A1 + 3*A2 + A100 triethylgallium 11.64	0	60.18	62.2	11.64	12.0 [216]	
$C_6H_{15}In$	3*A1 + 3*A2 + A101 triethylindium 13.01	0	54.76	54.8	13.01	13.0 [216]	
$C_6H_{15}Sb$	3*A1 + 3*A2 + A105 triethylantimony 9.45	0	61.42	61.4	9.45	9.5 [216]	
$C_6H_{16}Si_2$	3*A1 + 3*A2 + A107 1,1,3,3-tetramethyl-1,3-disilacyclobutane 10.26	0	38.57	38.0	10.26	10.1 [216]	
$C_6H_{18}OSi_2$	A14 + A15 + 2*A139 + 4*A1 hexamethyldisiloxane 11.92	0	58.17	56.0	11.92	11.5 [216]	
$C_6H_{18}O_3Si_3$	6*A1 + A32*B32 + 2*A109 hexamethylcyclotrisiloxane 16.61	0	49.55	49.6	16.61	16.6 [216,121]	
$C_6H_{18}Si_2$	A14 + 3*A15 + 6*A1 + 3*A112 + 3*A139 hexamethyldisilane 9.75	43.95					
	287.7	3.02	10.49	54.44	51.3	12.77	14.8 [216]
$C_6H_{21}N_3Si_3$	6*A1 + 2*A109 hexamethylcyclotrisilazane 15.17	0	59.63	52.4	15.17	13.3 [216]	
C_7F_8	A14 + 3*A15 + 3*A139 + 6*A1 + 3*A121 perfluorotoluene 11.49	0	55.23	53.0	11.49	11.0 [216]	
C_7F_{16}	5*A12 + A11 + A4*B4 + 5*A24 + 3*A25 perfluoroheptane 6.67	36.97					
	180.4	6.95	31.31	68.28	83.1	13.62	18.5 [216,67]
$C_7H_5Br_2NO$	7*A4*B4 + 6*A25 + 10*A26 3,5-dibromo-4-hydroxybenzotrile 32.03	0	69.03	58.0	32.03	26.9 [221]	
$C_7H_5F_5$	2*A21 + A56 + A31 + 4*A12 + 2*A10 2,3,4,5,6-pentafluorotoluene 13.28	0	54.48	53.7	13.28	13.1 [216,77]	
$C_7H_3Cl_2N$	5*A12 + A11 + A1 + 5*A24 2,6-dichlorobenzotrile 26.17	0	62.73	49.9	26.17	20.8 [215]	
$C_7H_3Cl_3O_2$	2*A22*C22 + A56 + 3*A12 + 3*A10 2,3,6-trichlorobenzoic acid 23.85	0	59.23	63.5	23.85	25.6 [215]	
$C_7H_3I_2NO$	4*A12 + 2*A10 + 3*A22*D22 + A36*D36 4-hydroxy-3,5-diiodobenzotrile 33.63	0	69.64	61.7	33.63	29.8 [221]	
$C_7H_3I_3O_2$	4*A12 + 2*A10 + 2*A29 + A56 + A31 2,3,5-triiodobenzoic acid						

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (calcd)
503.8	32.23	0	63.97	73.1	32.23	36.8
$\text{C}_7\text{H}_4\text{Cl}_2\text{O}_2$	3*A29+A36*D36+4*A12+2*A10 3,5-dichlorobenzoic acid					[215]
459.3	22.97	0	50.01	62.2	22.97	28.6
$\text{C}_7\text{H}_4\text{Cl}_3\text{NO}_3$	3*A12+3*A10+2*A22*C22+A36*C36 3,5,6-trichloro-2-pyridinyloxyacetic acid					[215]
423.3	31.17	0	73.63	78.8	31.17	33.4
$\text{C}_7\text{H}_4\text{N}_2\text{O}_6$	A10+4*A12+3*A22*F22+A41+A32+A36*F36+A2 3,5-dinitrobenzoic acid					[221]
480.4	22.8	0	47.47	65.3	22.8	31.4
$\text{C}_7\text{H}_5\text{Cl}_3\text{N}_2\text{O}_2$	3*A12+2*A50+A36*C36+3*A10 methyl 4-amino-3,5,6-trichloro-2-picolinate					[280]
394.3	26.78	0	67.91	68.7	26.78	27.1
$\text{C}_7\text{H}_5\text{ClO}_2$	5*A12+A41+A38+3*A22*F22+A45+A1 2-chlorobenzoic acid					[232]
413.4	25.73	0	62.34	47.0	25.73	19.4
$\text{C}_7\text{H}_5\text{ClO}_2$	4*A10+A36*B36+2*A12+A22*B22 3-chlorobenzoic acid					[215]
427.4	23.85	0	55.65	47.0	23.85	20.1
$\text{C}_7\text{H}_5\text{ClO}_2$	4*A10+A36*B36+2*A12+A22*B22 4-chlorobenzoic acid					[215]
512.9	32.26	0	62.76	47.0	32.26	24.1
$\text{C}_7\text{H}_5\text{Cl}_2\text{NO}_2$	4*A10+A36*B36+2*A12+A22*B22 3-amino-2,5-dichlorobenzoic acid					[215]
475.6	37.42	0	78.68	68.7	37.42	32.7
$\text{C}_7\text{H}_5\text{Cl}_3$	2*A22*D22+A45+A36*D36+4*A12+2*A10 benzotrifluoride					[215]
236.0	13.95	0	59.11	53.2	13.95	12.6
$\text{C}_7\text{H}_5\text{F}_3$	5*A10+A11+3*A22*C22+A4*B4 benzotrifluoride					[216]
244.1	13.78	0	56.45	44.7	13.77	10.9
$\text{C}_7\text{H}_5\text{IO}_2$	5*A10+A11+A4*B4+3*A25 2-iodobenzoic acid					[216]
435.1	21.38	0	49.14	50.2	21.38	21.8
$\text{C}_7\text{H}_5\text{IO}_2$	4*A10+2*A12+A36*B36+A29 3-iodobenzoic acid					[7]
460.4	28.7	0	62.34	50.2	28.7	23.1
$\text{C}_7\text{H}_5\text{IO}_2$	4*A10+2*A12+A36*B36+A29 4-iodobenzoic acid					[7]
543.8	35.24	0	64.8	50.2	35.24	27.3
$\text{C}_7\text{H}_5\text{N}$	4*A10+2*A12+A36*B36+A29 benzotrifluoride					[7]
260.3	10.98	0	42.18	47.3	10.98	12.3
$\text{C}_7\text{H}_5\text{NO}$	5*A10+A12+A56 benzoxazole					[134]
247	0.02	0.07				
302.5	16.78	55.48	55.56	44.6	16.8	13.5
$\text{C}_7\text{H}_5\text{NO}_4$	A14+2*A15+2*A19+A18*B18+A112+A118+4*A10 <i>o</i> -nitrobenzoic acid					[216]
419	27.99	0	66.8	48.5	27.99	20.3
$\text{C}_7\text{H}_5\text{NO}_4$	4*A10+2*A12+A36*B36+A50 <i>m</i> -nitrobenzoic acid					[216]
414.3	19.33	0	46.66	48.5	19.33	20.1
$\text{C}_7\text{H}_5\text{NO}_4$	4*A10+2*A12+A50+A36*B36 <i>p</i> -nitrobenzoic acid					[216]
512.4	36.9	0	72.02	48.5	36.9	24.9
$\text{C}_7\text{H}_5\text{NS}$	4*A10+2*A12+A36*B36+A50 benzothiazole					[215]
275.6	12.8	0	46.36	46.2	12.8	12.7
$\text{C}_7\text{H}_5\text{N}_3\text{O}_6$	4*A10+A14+2*A15+A118*+2*A19+A131+A18*B18 2,4,5-trinitrotoluene					[216]
376.2	24.7	0	66.0	53.5	24.7	20.1
$\text{C}_7\text{H}_5\text{N}_3\text{O}_6$	2*A10+3*A12+3*A50*C50+A1+A11 2,4,6-trinitrotoluene					[216]
352.2	23.43	0	66.52	53.5	23.43	18.9
$\text{C}_7\text{H}_5\text{N}_3\text{O}_8$	2*A10+3*A12+3*A50*C50+A1+A11 N methyl 2,4,6,N tetranitroaniline					[217]
402.6	25.86	0	64.23	100.8	25.86	40.6
	4*A12+3*A50+A51+A1+2*A10+A47					[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_7H_6N_2$	benzimidazole 19.25	0	43.43	45.5	19.25	20.2
	$4*A10+A118+A121+A14+2*A15+2*A19+A18*B18$					[282]
$C_7H_6N_2O_4$	2,4-dinitrotoluene 20.12	0	58.61	50.7	20.12	17.4
	$3*A10+A11+2*A12+2*A50+A1$					[215]
$C_7H_6N_2O_4$	2,6-dinitrotoluene 23.85	0	72.82	50.7	23.85	16.6
	$3*A10+A11+2*A12+2*A50+A1$					[217]
$C_7H_6N_2O_4$	2,3-dinitrotoluene 17.57	0	53.27	50.7	17.57	16.7
	$3*A10+A11+2*A12+2*A50+A1$					[217]
$C_7H_6N_2O_4$	3,4-dinitrotoluene 18.83	0	57.15	50.7	18.83	16.7
	$3*A10+A11+2*A12+2*A50+A1$					[217]
$C_7H_6N_2O_5$	2-methyl-4,6-dinitrophenol 19.41	0	54.02	56.1	19.41	20.2
	$3*A12+A11+2*A10+A1+A31+2*A50$					[215]
C_7H_6O	benzaldehyde 9.32	0	43.1	51.1	9.32	11.0
	$5*A10+A12+A34$					[216]
$C_7H_6O_2$	benzoic acid 18.0	0	45.45	43.0	18.0	17.0
	$5*A10+A12+A36$					[282]
$C_7H_6O_3$	2-hydroxybenzoic acid 24.6	0	56.97	51.1	24.6	22.1
	$4*A10+2*A12+A31+A36*B36$					[216,8]
$C_7H_6O_3$	3-hydroxybenzoic acid 26.2	0	55.15	51.1	26.2	24.3
	$4*A10+2*A12+A31+A36*B36$					[216,8]
$C_7H_6O_3$	4-hydroxybenzoic acid 30.9	0	63.31	51.1	30.9	24.9
	$4*A10+2*A12+A31+A36*B36$					[233]
C_7H_7Br	benzylbromide 13.2	0	48.57	52.2	13.2	14.2
	$5*A10+A11+A2+A21$					[49]
C_7H_7Br	4-bromotoluene 15.13	0	50.2	47.1	15.1	14.3
	$4*A10+A11+A12+A1+A21$					[234]
C_7H_7Cl	<i>p</i> -chlorotoluene 13.55	0	48.29	40.9	13.55	11.5
	$A1+A12+A11+4*A10+A22$					[234]
$C_7H_7ClN_2S$	1-(<i>o</i> -chlorophenyl)thiourea 22.29	0	53.91	53.9	22.29	22.3
	$4*A10+2*A12+A22*B22+A91$					[221]
C_7H_7F	2-fluorotoluene 9.8	0	46.51	46.8	9.8	9.9
	$4*A10+A11+A12+A1+A24$					[216]
C_7H_7F	3-fluorotoluene 8.3	0	45.11	46.8	8.3	8.6
	$4*A10+A11+A12+A1+A24$					[216]
C_7H_7F	4-fluorotoluene 9.35	0	43.18	46.8	9.35	10.13
	$4*A10+A11+A1+A12+A24$					[216]
C_7H_7I	benzyl iodide 13.2	0	44.07	54.0	13.2	16.2
	$5*A10+A11+A2+A29$					[46]
C_7H_7I	<i>p</i> -iodotoluene 14.96	0	48.79	49.5	14.96	15.2
	$4*A10+A11+A12+A29+A1$					[234]
C_7H_7NO	benzamide 18.49	0	45.96	57.5	18.49	23.1
	$5*A10+A12+A61$					[215]
$C_7H_7NO_2$	3-nitrotoluene 19.2	0	51.88	47.9	19.2	17.7
	$4*A10+A1+A11+A12+A50$					[235]
$C_7H_7NO_2$	4-nitrotoluene					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
318	16.9	53.1				
324.8	16.81	0	51.76	47.9	16.81	15.6
$C_7H_7NO_2$	4*A10+A12+A11+A1+A50 2-aminobenzoic acid					[215]
417.8	20.5	0	49.07	52.2	20.5	21.8
$C_7H_7NO_2$	4*A10+2*A12+A36*B36+A45 3-aminobenzoic acid					[215]
452.9	21.84	0	48.12	52.2	21.84	23.6
$C_7H_7NO_2$	4*A10+2*A12+A36*B36+A45 4-aminobenzoic acid					[215]
461.4	20.92	0	45.19	52.2	20.92	24.1
$C_7H_7NO_3$	4*A10+2*A12+A36*B36+A45 4-nitro-5-methylphenol					[215]
401	27.4	0	68.33	53.3	27.4	21.4
$C_7H_7NO_3$	3*A10+A1+A11+2*A12+A31+A50 2-nitro-5-methylphenol					[215]
302.8	20.79	0	68.66	53.3	20.79	16.1
$C_7H_7N_3O_2$	3*A10+A1+A11+2*A12+A31+A50 N-acetyl-pyrazinamide					[215]
366.7	23.6	0	64.36	61.9	23.6	22.7
C_7H_8	3*A10+A12+2*A41+A71+A1 toluene					[9]
178.0	6.62	0	37.15	45.0	6.62	8.0
C_7H_8	5*A10+A1+A11 cycloheptatriene					[216]
154.0	2.35	15.24				
198.0	1.16	5.86	21.11	38.5	3.51	7.6
C_7H_8	A14+4*A15+6*A18 tetracyclo[3.2.0.0(2,7).0(4,6)]heptane					[216]
180	7.2	40				
228	1.09	4.8	44.8	26.6	8.29	6.1
$C_7H_8N_2O$	4*A14-5*A15+6*A16 phenylurca					[216]
420.6	23.68	0	56.3	52.1	23.68	21.9
$C_7H_8N_4O_2$	5*A10+A12+A67 theophylline					[215]
544	28.2	0	51.84	44.6	28.2	24.3
546.1	28.27	0	51.76	44.6	28.3	24.3
C_7H_8O	2*A14+3*A15+2*A125+A118+A121+2*A1+2*A19+A18*B18 benzyl alcohol					[236,205]
257.6	8.79	0	34.11	36.4	8.79	9.4
C_7H_8O	5*A10+A11+A2+A30 o-hydroxytoluene					[215]
304.2	15.82	0	52.01	50.4	15.82	15.3
C_7H_8O	A31+A1+A12+4*A10+A11 m-hydroxytoluene					[216]
285.4	10.71	0	37.53	50.42	10.71	14.39
C_7H_8O	A31+A1+A12+4*A10+A11 p-hydroxytoluene					[216]
307.9	12.72	0	41.25	50.42	12.72	15.52
C_7H_8O	A31+A1+A12+4*A10+A11 methoxybenzene					[216]
268.7	12.9	0	48.0	51.9	12.9	13.9
C_7H_8S	5*A10+A12+A1+A32 methylphenylsulfide					[216]
256.4	14.85	0	57.86	49.3	14.85	12.63
$C_7H_9Cl_3NO_3PS$	5*A10+A12+A1+A84 O,O-dimethyl-O-(3,5,6-trichloro-2-pyridyl)phosphorothioate					[105]
318.7	25.92	0	81.32	73.2	25.92	23.3
$C_9H_{13}Cl_3NO_4P$	4*A12+A10+A41+3*A22*E22+2*A1+A79 O,O-diethyl-O-(3,5,6-trichloro-2-pyridyl)phosphate					[221]
312.5	15.61	0	49.97	76.4	15.61	23.9
C_7H_9N	4*A12+A10+A11+3*A22*E22+2*A1+A74+2*A2 m-toluidine					[221]
241.7	8.8	0	36.41	51.5	8.8	12.5
C_7H_9N	A45+4*A10+A12+A1+A11 p-toluidine					[215]
316.5	17.89	0	56.52	51.5	17.89	16.3
	A45+4*A10+A12+A1+A11					[215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpce}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpce}}$ (calcd)
C_7H_9N	<i>o</i> -toluidine					
249.6	8.08	0	32.37	51.5	8.08	12.9 [215]
	$A45+4*A10+A12+A1+A11$					
C_7H_9N	2-methylaniline					
258	11.66	0	45.1	51.5	11.66	14.8 [30]
	$4*A10+A11+A12+A1+A45$					
C_7H_9N	2,3-dimethylpyridine					
258.6	13.48	0	52.13	49.1	13.48	12.7 [69]
	$2*A1+2*A11+3*A10+A41$					
C_7H_9N	2,4-dimethylpyridine					
209.4	8.82	0	42.12	49.1	8.82	10.3 [69]
	$2*A1+2*A11+3*A10+A41$					
C_7H_9N	2,5-dimethylpyridine					
259.1	14.64	0	56.5	49.1	14.64	12.7 [69]
	$2*A1+2*A11+3*A10+A41$					
C_7H_9N	2,6-dimethylpyridine					
267.1	13.04	0	48.82	49.1	13.04	13.1 [69]
	$2*A1+2*A11+3*A10+A41$					
C_7H_9N	3,4-dimethylpyridine					
262.7	14.7	0	55.96	49.1	14.7	12.9 [69]
	$2*A1+2*A11+3*A10+A41$					
C_7H_9N	3,5-dimethylpyridine					
266.9	13.11	0	49.12	49.1	13.11	13.1 [69]
	$2*A1+2*A11+3*A10+A41$					
$C_7H_9N_2O_{12}$	2,2,7-trinitroethyl 4,4-dinitropentanoate					
363.8	20.08	55.2				
366.7	6.69	18.26	73.46	89.6	26.78	32.9 [122]
	$2*A4*B4+3*A2+A1+5*A50+A38$					
$C_7H_9N_3O_{12}$	2,2-dinitropropyl-4,4,4-trinitrobutyrate					
284.2	25.94	91.28				
335.5	20.92	62.35				
368.2	6.69	18.18	171.8	89.6	53.56	33.0 [122]
	$2*A4*B4+3*A2+A1+5*A50+A38$					
C_7H_{10}	bicyclo[2.2.1]hept-2-ene					
130.3	4.27	32.77				
319.5	3.48	10.89	43.66	37.8	7.75	12.1 [129,349]
	$2*A14+A15+2*A16+2*A18$					
$C_7H_{10}O$	2-norbomanone					
368.7	3.39	0	9.19	39.7	3.39	14.6 [217]
	$2*A14+A15+2*A16+A114$					
$C_7H_{10}N_2O$	6,7-diazatricyclo[3.2.2.0 2,4]non-6-ene-N-oxide					
372.6	15.8	42.4				
411.4	2.6	6.32	48.72	44.1	18.4	18.1 [42]
	$3*A14+2*A16+2*A16+A123$					
$C_7H_{10}N_2O_2$	N-acetyl-L-alanine amide					
431	21.7	0	50.35	54.9	21.7	23.7 [278]
	$2*A1+A60+A3*B3+A61$					
$C_7H_{10}N_2O_2$	1,3,6-trimethyluracil					
384.5	21.2	0	55.14	38.4	21.2	14.8 [216]
	$A14+3*A15+2*A125+3*A1+A18*B18+A19$					
$C_7H_{10}O_3$	3,3-dimethylpentanedioic anhydride					
396.2	17.99	0	45.41	47.4	17.99	18.8 [237]
	$A14+3*A15+2*A1+A17+A117$					
$C_7H_{11}N$	cyanocyclohexane					
215	7.43	34.53				
285.1	3.64	12.75	47.28	47.5	11.06	13.5 [216]
	$A14+3*A15+A16+A56$					
$C_7H_{11}N$	isocyanocyclohexane					
192.6	6.18	32.07				
279.6	4.23	15.12	47.19	47.2	10.4	13.2 [216]
	$A14+3*A15+A16+A57$					
C_7H_{12}	4-methylcyclohex-1-ene					
153.6	6.63	0	43.16	44.1	6.63	6.77 [161]
	$A14+3*A15+A1+A16+2*A18$					
C_7H_{12}	cycloheptene					
154	5.28	34.29				
210	0.71	3.38				
217	0.97	4.47	42.14	45.0	6.96	9.8 [216,161]
	$A14+4*A15+2*A18$					
$C_7H_{12}ClN_5$	6-chloro-N,N'-diethyl-1,3,5-triazine-2,4-diamine					
502.5	47.35	0	94.23	65.3	47.35	32.8

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$C_7H_{12}O_2$	209.5	0	82.61	67.9	17.31	14.2
						[221]
$C_7H_{12}O_4$	377.5	0	73.17	78.9	27.62	29.8
						[216]
$C_7H_{12}O_4S_2$	429	0	91.68	86.9	39.33	37.3
						[273]
$C_7H_{12}O_4S_2$	355	0	63.64	86.9	22.59	30.8
						[273]
$C_7H_{13}N$	196	26.53				
	430	13.81	40.33	40.1	11.09	17.3
						[216]
$C_7H_{13}NO$	310.3	0	44.39	54.6	13.78	16.9
						[216]
$C_7H_{13}N_3O_3S$	372.2	0	81.04	59.4	30.17	22.1
						[221]
C_7H_{14}	134.8	36.94				
	198.2	1.46				
	212.4	2.11				
	265.1	7.1	47.6	48.2	7.6	12.8
						[216]
C_7H_{14}	146.8	44.18				
	203.7	5.34	49.52	41.4	7.57	8.4
						[216]
C_7H_{14}	141.5	47.01				
	219.4	7.55	54.57	46.5	8.31	10.2
						[216]
C_7H_{14}	139.5	0	53.09	46.5	7.4	6.5
						[216]
C_7H_{14}	134.7	0	51.0	50.8	6.87	6.8
						[216]
C_7H_{14}	146.6	0	46.1	47.3	6.75	6.9
						[216]
C_7H_{14}	154.3	0	82.5	77.5	12.66	12.0
						[216]
$C_7H_{14}NO_3P$	326.9	0	68.4	55.0	22.36	18.0
						[221]
$C_7H_{14}N_2O_2S$	374.0	0	60.73	62.3	22.71	23.3
						[221]
$C_7H_{13}N_3O_3S$	372.2	0	81.05	59.4	30.17	22.1
						[221]
$C_7H_{14}O$	229.3	0	99.83	85.8	22.89	19.7
						[43]
$C_7H_{14}O$	204.8	0	54.6	55.3	11.2	11.3
						[216]
$C_7H_{14}O$	299.2	0	36.34	29.2	10.87	8.8
						[230]
$C_7H_{14}O$	172.2	16.98				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
	227.3	0.55	2.44				
	258.4	0.88	3.39				
	280.3	1.6	5.72	28.53	35.2	5.96	9.9
$C_7H_{14}O_2$		A14+4*A15+A30+A16					[216]
		heptanoic acid					
	224.8	2.04	9.08				
	265.8	15.44	58.07	67.15	77.6	17.48	20.6
		5*A2*B2+A1+A36					[216,143]
$C_7H_{15}Cl_2N_2O_2P$		2-[bis(2-chloroethyl)amino]tetrahydro-2H-1,3,2-oxazophosphorine-2-oxide					
	322.6	33.13	0	102.7	102.7	33.13	33.1
		A14+3*A15+A144+4*A2+2*A22*C22					[221]
C_7H_{16}		heptane					
	182.6	14.04	0	76.9	81.8	14.04	14.9
		2*A1+5*A2*B2					[216]
C_7H_{16}		2,4-dimethylpentane					
	154.0	6.85	0	44.46	44.7	6.85	6.9
		4*A1+A2+2*A3					[216]
C_7H_{16}		3-ethylpentane					
	154.6	9.55	0	61.77	57.8	9.55	8.9
		3*A1+3*A2+A3					[216]
C_7H_{16}		2 methylhexane					
	154.9	9.18	0	59.29	57.8	9.18	9.0
		3*A1+3*A2+A3					[216]
C_7H_{16}		3,3-dimethylpentane					
	138.2	7.07	0	51.16	49.7	7.07	6.9
		4*A1+A4+2*A2					[216]
C_7H_{16}		2,2,3-trimethylbutane					
	121	2.38	19.64				
	247.7	2.2	8.88	28.53	36.7	4.58	4.4
		5*A1+A3+A4					[216]
C_7H_{16}		2,2-dimethylpentane					
	148.1	5.86	0	39.55	49.74	5.86	7.37
		4*A1+A4+2*A2					[215]
$C_7H_{16}O$		1-heptanol					
	240.4	18.16	0	75.53	75.31	18.16	18.1
		A1+6*A2*B2+A30					[216]
$C_7H_{16}O_2$		1,7-heptanediol					
	295.2	21.3	0	72.15	101.6	21.3	30.0
		7*A2*B2+2*A30*B30					[215]
$C_7H_{16}S$		1-heptanethiol					
	229.9	25.4	0	110.4	96.6	25.4	22.2
		A1+6*A2*B2+A86					[216]
$C_7H_{17}NSi$		N-(β -trimethylsilyl)ethyl)ethylenimine					
	176.5	10.62	0	60.17	54.0	10.62	9.5
		3*A1+2*A2+A14+A119+A109					[216]
$C_7H_{20}Si_2$		hexamethyldisilylmethane					
	140.7	11.11	0	78.98	58.5	11.11	8.2
		6*A1+2*A109+A2					[216]
$C_8Cl_4N_2$		2,4,5,6-tetrachloro-1,3-benzenedicarbonitile					
	526.2	30	0	57.01	55.3	30	29.1
		4*A22*F22+2*A56+6*A12					[221]
C_8F_{18}		perfluorooctane					
	176.5	3.14	17.79				
	254.2	9.58	37.69	55.48	93.8	12.72	23.8
		8*A4*B4+12*A26+6*A25					[67]
$C_8H_3NO_5$		3-nitrophthalic anhydride					
	436.2	18.4	0	42.18	51.0	18.4	22.3
		A14+2*A15+A117+2*A19+A12+3*A10+A50					[179]
$C_8H_3NO_5$		4-nitrophthalic anhydride					
	388.2	17.14	0	44.15	51.0	17.14	19.8
		A14+2*A15+A117+2*A19+A12+3*A10+A50					[179]
$C_8H_4Cl_2O_2$		terephthalyl dichloride					
	337.3	2.34	6.92				
	356.1	21.1	59.25	66.18	66.2	23.44	23.6
		4*A10+2*A12+2*A40					[216]
$C_8H_4N_2$		1,2-dicyanobenzene					
	414.1	20	0	48.3	50.2	20	20.8
		4*A10+2*A12+2*A56					[71]
$C_8H_4O_3$		phthalic anhydride					
	403.3	23.09	0	57.25	48.2	23.09	19.4
		A14+2*A15+2*A19+A117+4*A10					[215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}G}_{tpcc}$ (expt)	$\Delta_0^{T_{fus}G}_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}H}_{tpcc}$ (expt)	$\Delta_0^{T_{fus}H}_{tpcc}$ (calcd)
$C_8H_5Cl_3O_2$	2,3,6-trichlorophenylacetic acid 22.43	0	51.89	68.5	22.43	29.6
	$2*A10+3*A12+A11+A36*D36+3*A22*D22+A2$					[215]
$C_8H_5Cl_3O_3$	(2,4,5-trichlorophenoxy)acetic acid 38	0	88.13	75.3	38	32.5
	$3*A22*E22+A32+A36*E36+4*A12+2*A10+A2$					[215]
$C_8H_5Cl_3O_4$	3,6-dichloro-5-hydroxy-2-methoxybenzoic acid 28.98	0	70.7	75.0	28.98	30.7
	$2*A22*E22+A31+A32+A36*E36+5*A12+A10+A1$					[215]
C_8H_6	phenylacetylene 9.46	0	41.49	41.7	9.46	9.5
	$5*A10+A12+A8+A9$					[132]
$C_8H_6Cl_2O_3$	(2,4-dichlorophenoxy)acetic acid 35.33	0	85.64	74.0	35.33	30.5
	$3*A10+3*A12+A2+2*A22*D22+A36*D36+A32*D32$					[221]
$C_8H_6Cl_2O_3$	3,6-dichloro-2-methoxybenzoic acid 22.9	0	59.23	69.6	22.9	26.9
	$4*A12+2*A10+2*A22*D22+A36*D36+A32*D32+A1$					[215]
$C_8H_6Cl_2O_3$	2,4-dichloro-2-methoxybenzoic acid 35.33	0	85.65	69.6	35.33	28.7
	$4*A12+2*A10+2*A22*D22+A36*D36+A32*D32+A1$					[215]
$C_8H_6Cl_2O_4$	3,6-dichloro-5-hydroxy-2-methoxybenzoic acid 28.98	0	70.71	75.0	28.98	30.7
	$A10+5*A12+2*A22*E22+A36*E36+A1+A32+A31$					[215]
$C_8H_6Cl_4$	tetrachloro- <i>o</i> -xylene 21.46	0	59.74	50.8	21.46	18.2
	$4*A12+2*A11+2*A1+4*A22*D22$					[215]
$C_8H_6Cl_4$	tetrachloro- <i>p</i> -xylene 22.59	0	61.35	50.8	22.59	18.7
	$4*A12+2*A11+2*A1+4*A22*D22$					[215]
$C_8H_6Cl_4O_4$	methyl tetrachloroterephthalic acid ester 16.89	0	38.03	92.7	16.89	41.2
	$6*A12+2*A1+A38+4*A22*F22+A36$					[221]
$C_8H_6N_2$	phthalazine 13.32	0	36.54	51.4	13.32	18.7
	$6*A10+2*A12+2*A41$					[29]
$C_8H_6N_2$	quinazoline 16.95	0	52.82	51.4	16.95	16.5
	$6*A10+2*A12+2*A41$					[29]
$C_8H_6N_2$	quinoxaline 11.80	0	38.61	51.4	11.80	15.7
	$6*A10+2*A12+2*A41$					[29]
$C_8H_6N_2OS_2$	6-methyl-1,3-dithiol[4,5- <i>b</i>]quinoxalin-2-one 29.92	0	67.49	67.5	29.92	29.9
	$A14+2*A15+A135+2*A19+2*A41+A1+A11$ $+3*A10+2*A12$					[221]
C_8H_6S	benzothiophene 11.82	0	38.82	44.1	11.82	13.4
	$4*A10+A14+2*A15+A19+A18+A131+A19+A18*B18$					[95]
$C_8H_7ClO_3$	(<i>d</i>) <i>o</i> -chloromandelic acid 24.69	0	62.89	66.1	24.69	25.9
	$4*A10+A11+A12+A3*B3+A30*C30+A36*C36+A22*C22$					[220]
$C_8H_7ClO_3$	(<i>dl</i>) <i>o</i> -chloromandelic acid 20.08	0	56.02	66.0	20.08	23.6
	$4*A10+A11+A12+A3*B3+A30*C30+A36*C36+A22*C22$					[220]
$C_8H_7ClO_3$	(<i>dl</i>) <i>p</i> -chloromandelic acid 27.2	0	69.03	66.0	27.2	26.0
	$4*A10+A12+A11+A3*B3+A30*C30+A36*C36+A22*C22$					[220]
$C_8H_7ClO_3$	(<i>d</i>) <i>p</i> -chloromandelic acid 23.01	0	58.41	66.0	23.01	26.0
	$4*A10+A11+A12+A3*B3+A30*C30+A36*C36+A22*C22$					[220]
$C_8H_7Cl_2NO$	methyl-3,4-dichlorophenylcarbamate 23.19	0	60.8	60.4	23.19	23.0
	$A1+3*A10+3*A12+2*A22*C22+A69$					[221]
$C_8H_7FO_3$	(<i>dl</i>) <i>m</i> -fluoromandelic acid 24.69	0	66.72	66.4	24.69	24.6

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{fus}S_{tpce}$ (expt)	$\Delta_0^{fus}S_{tpce}$ (calcd)	$\Delta_0^{fus}H_{tpce}$ (expt)	$\Delta_0^{fus}H_{tpce}$ (calcd)	
$C_8H_7FO_3$	4*A10+A11+A12+A3*B3+A36*C36+A30*C30+A24 (d) <i>m</i> -fluoromandelic acid	24.27	0	61.59	66.4	24.27	[220,187]
394							
$C_8H_7FO_3$	4*A10+A11+A12+A3*B3+A36*C36+A30*C30+A24 (dl) <i>o</i> -fluoromandelic acid	30.12	0	77.24	66.4	30.12	[220,187]
390							
$C_8H_7FO_3$	4*A10+A11+A12+A3*B3+A36*C36+A30*C30+A24 (d) <i>o</i> -fluoromandelic acid	20.92	0	57.63	66.4	20.92	[220,187]
363							
$C_8H_7FO_3$	4*A10+A11+A12+A3*B3+A36*C36+A30*C30+A24 (dl) <i>p</i> -fluoromandelic acid	29.29	0	72.67	66.4	29.29	[220,187]
403							
$C_8H_7FO_3$	4*A10+A11+A12+A3*B3+A36*C36+A30*C30+A24 (d) <i>p</i> -fluoromandelic acid	30.54	0	71.7	66.4	30.54	[220,187]
426							
$C_8H_7ClO_3$	4*A10+A11+A12+A3*B3+A36*C36+A30*C30+A24 4-chlorophenoxyacetic acid	36.27	0	84.42	72.8	36.27	[220,187]
429.6							
$C_8H_7N_5O_6$	4*A10+2*A12+A22*C22+A36*C36+A2+A32 2,4,6-trinitro-1,3-dimethylbenzene	38.49	0	84.52	54.1	38.49	[221]
455.4							
$C_8H_7N_5O_8$	A10+2*A11+3*A12+2*A1+3*A50 2,4,6,N-tetranitro-N-methyltoluidene	19.33	0	51.46	54.3	19.33	[197]
375.6							
$C_8H_7N_5O_8$	4*A12+A10+A11+3*A50+2*A1+A51+A49 2,4,6-N-tetranitroethylaniiline	23.51	0	63.6	60.8	23.51	[216]
369							
C_8H_8	A51+4*A12+2*A10+A1+A2+3*A50+A49 cubane	5.94	15.08				
394							
404.9		8.7	21.49	36.56	23.2	14.64	9.4
C_8H_8	5*A14-7*A15+8*A16 styrene	10.95	0	45.16	52.2	10.95	[150]
242.3							
C_8H_8	5*A10+A5+A6+A12 cyclooctatetraene	11.27	0	41.49	39.0	11.27	[216]
268.5							
$C_8H_8BrCl_2O_3PS$	A14+5*A15+8*A18 O-(4-bromo-2,5-dichlorophenyl)O,O-dimethyl phosphorothioate	31.15	0	95.74	71.1	31.15	[216]
325.3							
$C_8H_8Br_2$	2*A10+4*A12+2*A1+2*A22*D22+A21+A79 α, α' -dibromo- <i>o</i> -xylene	26.78	0	72.73	59.8	26.78	[221]
368.2							
$C_8H_8Br_2$	4*A10+2*A11+2*A2+2*A21*B21 α, α' -dibromo- <i>m</i> -xylene	23.69	0	67.65	59.8	23.69	[215]
350.2							
$C_8H_8ClNO_2$	4*A10+2*A11+2*A2+2*A21*B21 N-methyl-2-chlorophenylcarbamic acid ester	21.81	0	60.12	59.1	21.81	[215]
362.7							
$C_8H_8Cl_2$	A1+4*A10+2*A12+A69+A22*B22 α, α' -dichloro- <i>o</i> -xylene	21.26	0	64.78	57.1	21.26	[221]
328.2							
$C_8H_8Cl_2$	4*A10+2*A11+2*A2+2*A22*B22 α, α' -dichloro- <i>m</i> -xylene	19.51	0	63.51	57.1	19.51	[215]
307.2							
$C_8H_8Cl_2$	4*A10+2*A11+2*A2+2*A22*B22 α, α' -dichloro- <i>p</i> -xylene	23.97	0	64.23	57.1	23.97	[215]
373.2							
$C_8H_8Cl_2O_2$	4*A10+2*A11+2*A2+2*A22*B22 1,4-dichloro-2,5-dimethoxybenzene	27.56	0	68.23	61.8	27.56	[215]
403.9							
$C_8H_8Cl_2O_3$	4*A12+2*A10+2*A22*D22+2*A32*D32+2*A1 methyl 3,6-dichloro-2-methoxybenzoate	18.49	0	60.72	64.8	18.49	[215]
304.6							
$C_8H_8Cl_3O_3PS$	2*A10+4*A12+2*A1+2*A22*D22+A38*D38+A32*D32 O,O-dimethyl O-(2,4,5-trichlorophenyl) phosphorothioate						[221]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}}S_{tpce}$ (expt)	$\Delta_0^{T_{fus}}S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}}H_{tpce}$ (expt)	$\Delta_0^{T_{fus}}H_{tpce}$ (calcd)
313.0	18.94	0	60.51	69.7	18.94	21.8
$C_8H_8O_2$	2*A10+4*A12+3*A22*D22+2*A1+A79 <i>o</i> -toluic acid					[221]
376.9	20.17	0	53.51	43.5	20.17	16.4
$C_8H_8O_2$	4*A10+A12+A11+A1+A36 <i>m</i> -toluic acid					[215]
381.9	15.73	0	41.19	43.5	15.73	16.6
$C_8H_8O_2$	4*A10+A12+A11+A1+A36 <i>p</i> -toluic acid					[215]
452.8	22.72	0	50.17	43.5	22.72	19.7
$C_8H_8O_2$	4*A10+A12+A11+A1+A36 phenylacetic acid					[215]
349.9	14.49	0	41.41	48.0	14.49	16.8
$C_8H_8O_2$	5*A10+A11+A2+A36 methyl benzoate					[215]
261	13.9	0	53.26	54.8	13.9	14.3
$C_8H_8O_2S$	5*A10+A1+A38+A12 phenyl vinyl sulfone					[247]
343.4	11.72	0	34.12	52.5	11.72	18.0
$C_8H_8O_3$	5*A10+A5+A6+A12+A88 methyl 4-hydroxybenzoate					[238]
398.5	24.31	0	61	60.2	24.31	24.0
$C_8H_8O_3$	4*A10+2*A12+A1+A31+A38*B38 (<i>dl</i>) mandelic acid					[239]
392	25.52	0	65.11	51.9	25.52	20.3
$C_8H_8O_3$	5*A10+A3*B3+B30*A30+A36*B36+A11 (<i>d</i>) mandelic acid					[220]
406	23.36	0	64.92	51.9	26.36	21.1
$C_8H_8O_3$	5*A10+A3*B3+B30*A30+A36*B36+A11 4-hydroxyphenylacetic acid					[220]
423.6	28.4	0	67.04	56.1	28.4	23.9
$C_8H_8O_3$	4*A10+A11+A12+A2+A36*B36+A31 4-methoxybenzoic acid					[215]
457.8	28.4	0	62.04	53.1	28.4	24.3
455.4	27.8	0	61.1	53.1	27.8	23.9
$C_8H_9ClO_3$	4*A10+2*A12+A1+A36*B36+A32 (4-chloro-2-methylphenoxy)acetic acid					[215,394]
392.9	29.98	0	76.31	73.3	29.98	28.8
C_8H_9NO	3*A10+2*A12+A11+A22*C22+A36*C36+A32+A2+A1 4-aminoacetophenone					[221]
379.2	38	0	100.2	58.2	38	22.1
C_8H_9NO	4*A10+2*A12+A35+A45+A1 3-aminoacetophenone					[280]
371.2	29	0	78.12	58.2	29	21.6
C_8H_9NO	4*A10+2*A12+A35+A45+A1 acetanilide					[280]
387.5	21.65	0	55.87	48.6	21.6	18.8
$C_8H_9NO_2$	5*A10+A12+A1+A60 methyl 4-aminobenzoate					[216]
385.1	22.55	0	58.56	61.3	22.55	23.6
$C_8H_9NO_2$	4*A10+2*A12+A1+A45+A38 <i>o</i> -hydroxyacetanilide					[239]
364.5	21.25	0	58.3	54.0	21.25	23.8
$C_8H_9NO_2$	4*A10+A60+A1+2*A12+A31 <i>p</i> -hydroxyacetanilide					[216]
441.2	26.02	0	58.99	54.0	26.02	23.8
441.7	27.0	0	60.9	54.0	27.0	23.8
$C_8H_9NO_2$	4*A10+A60+A1+2*A12+A31 methyl <i>N</i> -phenylcarbamate					[239,394]
325	14.56	0	44.77	57.8	14.56	18.8
$C_8H_9ClNO_2PS$	5*A10+A12+A1+A69 O-(2-chloro-4-nitrophenyl)O,O-dimethyl phosphorothioate					[216]
323.9	29.08	0	89.78	70.0	29.08	22.7
$C_8H_9O_3PS$	2*A1+3*A10+3*A12+A22*C22+A50+A79 2-methoxy-4H-1,3,2-benzodioxaphosphorin 2-sulfide					[221]
327.86	16.92	0	51.61	51.6	16.92	16.9

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
	A14+3*A15+A19+A19+4*A10+A130+A1					[221]
C_8H_{10}	<i>o</i> -xylene					
247.8	13.6	0	54.9	45.6	13.6	11.3
	2*A1+4*A10+2*A11					[216]
C_8H_{10}	<i>m</i> -xylene					
225.3	11.57	0	51.4	45.6	11.57	10.3
	2*A1+4*A10+2*A11					[216]
C_8H_{10}	<i>p</i> -xylene					
286.3	17.11	0	59.77	45.6	17.11	13.1
	2*A1+4*A10+2*A11					[216]
C_8H_{10}	ethylbenzene					
178.2	9.16	0	51.43	52.2	9.16	9.3
	A1+A2+5*A10+A11					[216]
$\text{C}_8\text{H}_{10}\text{N}_4\text{O}_2$	caffeine					
426	0.94	2.21				
512	23.43	45.76	47.97	40.7	24.37	20.9
510.1	18.3	35.88	38.1	40.7	19.4	20.9
	2*A14+3*A15+2*A125+3*A1+2*A19+A18*B18+A119+A118					[227,395]
$\text{C}_8\text{H}_{10}\text{NO}_2\text{PS}$	O,O-dimethyl O-4-nitrophenyl phospho hioate					
308.2	20.07	0	65.12	68.7	20.07	21.2
	4*A10+2*A12+2*A1+A50+A79					[215]
$\text{C}_8\text{H}_{10}\text{O}$	2,3-dimethylphenol					
346	21.02	0	60.75	51.0	21.02	17.6
	3*A10+2*A11+A12+2*A1+A31					[215]
$\text{C}_8\text{H}_{10}\text{O}$	2,5-dimethylphenol					
348	23.38	0	67.18	51.0	23.38	17.7
	3*A10+2*A11+A12+2*A1+A31					[215]
$\text{C}_8\text{H}_{10}\text{O}$	2,6-dimethylphenol					
318.9	18.9	0	59.27	51.0	18.9	16.3
	3*A10+2*A11+A12+2*A1+A31					[215]
$\text{C}_8\text{H}_{10}\text{O}$	3,4-dimethylphenol					
334	18.13	0	54.28	51.0	18.13	17.0
	3*A10+2*A11+A12+2*A1+A31					[215]
$\text{C}_8\text{H}_{10}\text{O}$	3,5-dimethylphenol					
336.8	18	0	53.44	51.0	18.0	17.2
	3*A10+2*A11+A12+2*A1+A31					[215]
$\text{C}_8\text{H}_{10}\text{O}_2\text{S}$	benzylmethylsulfone					
400.5	25.52	0	63.73	52.5	25.52	21.0
	5*A10+A11+A1+A2+A88					[276]
$\text{C}_8\text{H}_{11}\text{N}$	2,5-dimethylaniline					
279	13.7	0	49.1	52.1	13.7	14.5
	3*A10+2*A11+A12+2*A1+A45					[51]
$\text{C}_8\text{H}_{11}\text{N}$	N,N-dimethylaniline					
275.6	11.56	0	46.28	42.5	11.56	11.7
	5*A10+A12+A43+2*A1					[51]
$\text{C}_8\text{H}_{11}\text{N}$	exo-2-cyanobicyclo[2.2.1]heptane					
237.7	7.93	33.4				
298.8	2.94	9.83	43.2	44.0	10.87	13.1
	2*A14+A15+2*A16+A16+A56					[216]
$\text{C}_8\text{H}_{11}\text{N}$	endo-2-cyanobicyclo[2.2.1]heptane					
177.3	2.25	12.7				
331.2	2.96	8.94	21.6	44.0	7.18	14.6
	2*A14+A15+2*A16+A56+A16					[216]
$\text{C}_8\text{H}_{11}\text{N}$	2,4,6-trimethylpyridine					
229.0	9.54	0	41.64	50.0	9.54	11.4
	3*A1+3*A11+2*A10+A41					[216]
$\text{C}_8\text{H}_{11}\text{N}_5$	6,8,9-trimethyladenine					
438	23.1	0	52.74	54.4	23.1	23.8
	A14+2*A15+3*A19+3*A1+A118+A119+2*A41+A10+A12+A44					[240]
$\text{C}_8\text{H}_{11}\text{N}_5\text{O}_2$	2-amino-9-[(2-hydroxyethoxy)methyl]-9H-purine					
462.2	42.2	0	91.3	86.5	42.2	40.0
	A14+2*A15+2*A19+A18*B18+2*A41+A45+A118+A119+3*A2+A30*F30+A32					[203]
$\text{C}_8\text{H}_{11}\text{N}_5\text{O}_3$	2-amino-9-[(2-hydroxyethoxy)methyl]-1,9-dihydro-6H-purin-6-one					
528.2	30.44	0	57.63	92.7	30.44	48.9

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)	
	2*A14+3*A15+3*A19+A18*B18+A124+2*A118+A45+A119+3*A2+A32+A30*F30						[203]
C ₈ H ₁₂	2-bicyclo[2.2.2]octene						
110.5	0.19	1.7					
176.5	5.65	32					
389.8	5.4	13.85	47.55	41.5	11.23	16.2	
	2*A14+2*A15+2*A16+2*A18						[100]
C ₈ H ₁₂	cycloocta-1,5-diene						
194.4	-0.38						
204	9.83	0	48.2	45.4	9.83	9.3	
	A14+5*A15+4*A18						[216]
C ₈ H ₁₂ NO ₃ PS ₂	O,O-dimethyl O-(4-aminosulfonylphenyl)phosphorodithioate						
344.2	26.21	0	76.13	79.4	26.21	27.3	
	2*A1+4*A10+2*A12+A96+A79						[221]
C ₈ H ₁₂ N ₂	tetramethylsuccinonitrile						
345	18.1	52.48					
442	7.15	16.17	68.64	60.1	25.25	26.6	
	4*A1+2*A4*B4+2*A56						[216]
C ₈ H ₁₂ N ₂ O ₂	1,6-hexamethylene diisocyanate						
206.1	18.64	0	90.46	102.2	18.64	21.1	
	6*A2*B2+2*A58						[216]
C ₈ H ₁₂ N ₂ O ₃	barbitol						
462.6	24.98	0	54	63.1	24.98	29.2	
	A14+3*A15+A128+A124+2*A1+2*A2+A17						[241]
C ₈ H ₁₂ N ₄ O ₁₀	2,2-dinitropropyl 4,4-dinitropentanoate						
330.6	23.01	69.61					
370.8	6.28	16.93	86.53	89.4	29.29	33.2	
	2*A4*B4+3*A2+2*A1+4*A50+A38						[122]
C ₈ H ₁₂ N ₄ O ₁₀	2-methyl-2-nitropropyl 4,4,4-trinitrobutyrate						
346.1	24.69	71.33					
349.4	5.27	15.09	86.41	89.4	29.96	31.3	
	2*A1+2*A4*B4+3*A2+4*A50+A38						[122]
C ₈ H ₁₂ O ₂	1,4-cyclooctanedione						
341.2	11.92	0	34.95	49.2	11.92	16.8	
	A14+5*A15+2*A114						[114]
C ₈ H ₁₃ ClN ₂ O ₂	5-chloro-3-(1,1-dimethylethyl)-6-methyl-2,4(1H,3H)-pyrimidinedione						
448	12.51	0	27.92	64.6	12.51	28.9	
	A14+3*A15+2*A19+A124+A125+4*A1+A4*B4+A22*C22						[221]
C ₈ H ₁₄	<i>endo</i> -2-methylbicyclo[2.2.1]heptane						
152.4	4.71	30.9					
278.3	1.62	5.82	36.72	43.8	6.35	12.2	
	2*A14+A15+A1+3*A16						[216]
C ₈ H ₁₄	<i>exo</i> -2-methylbicyclo[2.2.1]heptane						
164.1	8.38	0	51.0	43.8	8.38	7.2	
	2*A14+A15+A1+3*A16						[216]
C ₈ H ₁₄	bicyclo[2.2.2]octane						
164.3	4.6	28.01					
447.5	8.37	18.7	46.71	44.7	12.97	20.0	
	2*A14+2*A16+2*A15						[215]
C ₈ H ₁₄	cyclooctene						
190.1	9.8	51.55					
259.2	1.81	6.98	58.53	48.7	11.61	12.6	
	A14+5*A15+2*A18						[161]
C ₈ H ₁₄ N ₄ OS	4-amino-6-(1,1-dimethylethyl)-3-(methylthio)1,2,4-triazin-5(4H)-one						
399.4	18	0	45.06	58.2	18	23.3	
	A14+3*A15+2*A19+4*A1+A4+A125+A84+A45+2*A118						[215]
C ₈ H ₁₄ N ₅ Cl	6-chloro-N-ethyl-N'-(isopropyl)-1,3,5-triazine-2,4-diamine						
449.7	38.15	0	84.84	65.9	38.15	29.7	
	3*A41+3*A12+2*A44+A22*F22+3*A1+A2+A3*B3						[221]
C ₈ H ₁₄ N ₆ O ₁₀	1,7-diacetoxy-2,4,6-trinitro-2,4,6-triazaheptane						
422.5	38.49	0	91.11	214.5	38.49	90.6	
	2*A1+4*A2+2*A38+3*A51+3*A47						[216]
C ₈ H ₁₄ O	3-oxabicyclo[3.2.2]nonane						
208.5	7.02	33.65					
448.4	6.75	15.06	48.71	49.6	13.77	22.3	
	2*A14+3*A15+2*A16+A112						[216]
C ₈ H ₁₄ O ₄	suberic acid						
415.3	28.82	0	69.4	88.2	28.82	36.6	
	6*A2*B2+2*A36*B36						[340]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_8H_{14}O_4$	383	tetramethylsuccinic acid					
	464	13.43	35.07				
$C_8H_{15}N$	297.8	4* A_1 +2* A_4 * B_4 +2* A_{36} * B_{36}	6.47	13.95	49.02	56.8	19.9
	466.6	3-azabicyclo[3.2.2]nonane	14.55	48.87			26.4
$C_8H_{15}NO_2$	237.7	2* A_{14} +3* A_{15} +2* A_{16} + A_{121}	6.92	14.82	63.69	50.6	21.47
		dimethylaminoethyl methacrylate	16.85	0	70.9	59.1	16.85
C_8H_{16}	166.5	3* A_1 +2* A_2 + A_5 + A_7 + A_{38} + A_{43}					
	183.8	cyclooctane	6.32	37.94			
C_8H_{16}	288	5* A_{15} + A_{14}	0.48	2.6	48.89	51.9	9.2
			2.41	8.35			14.9
C_8H_{16}	153.2	1,1-dimethylcyclohexane					
	239.8	5.98	39.05				
C_8H_{16}	155.8	A_{14} + A_{17} +2* A_1 +3* A_{15}	2.01	8.37	47.43	45.1	7.99
		propylcyclopentane	10.04	0	64.45	57.9	10.04
C_8H_{16}	185	A_{14} + A_{16} + A_1 +2* A_2 +2* A_{15}					
		<i>trans</i> -1,2-dimethylcyclohexane	10.5	0	56.77	50.2	10.5
C_8H_{16}	172.5	A_{14} +3* A_{15} +2* A_1 +2* A_{16}					
	223.3	<i>cis</i> -1,2-dimethylcyclohexane	8.26	47.86			
C_8H_{16}	183.1	A_{14} +3* A_{15} +2* A_1 +2* A_{16}	1.64	7.36	55.22	50.2	9.9
		<i>trans</i> -1,3-dimethylcyclohexane	9.87	0	53.93	50.2	9.87
C_8H_{16}	197.6	A_{14} +3* A_{15} +2* A_1 +2* A_{16}					
		<i>cis</i> -1,3-dimethylcyclohexane	10.82	0	54.77	50.2	10.82
C_8H_{16}	236.2	A_{14} +3* A_{15} +2* A_1 +2* A_{16}					
		<i>trans</i> -1,4-dimethylcyclohexane	12.34	0	52.26	50.2	12.34
C_8H_{16}	185.7	A_{14} +3* A_{15} +2* A_1 +2* A_{16}					
		<i>cis</i> -1,4-dimethylcyclohexane	9.31	0	50.11	50.2	9.31
C_8H_{16}	161.4	A_{14} +3* A_{15} +2* A_1 +2* A_{16}					
		ethylcyclohexane	8.28	0	51.3	54.5	8.28
C_8H_{16}	171.5	A_{14} + A_{16} + A_1 + A_2 +3* A_{15}					
		1-octene	15.31	0	89.29	86.8	15.31
C_8H_{16}	178.9	A_1 +5* A_2 * B_2 + A_5 + A_6					
		2,4,4-trimethyl-1-pentene	8.77	0	49.0	49.2	8.77
C_8H_{16}	166	4* A_1 + A_2 + A_5 + A_7 + A_4					
		2,4,4-trimethyl-2-pentene	6.8	0	40.9	47.6	6.78
$C_8H_{16}N_2O_2$	404	5* A_1 + A_4 + A_7 + A_6					
		N-acetyl-D-leucine amide	20.2	0	50	63.2	20.2
$C_8H_{16}N_6$	378.8	3* A_1 + A_3 + A_3 * B_3 + A_2 + A_{61} + A_{60}					
		1-(methylamino)-3,5-bis(dimethylamino)-s-triazine	22.34	0	58.98	48.4	22.34
$C_8H_{16}N_6O$	381.5	3* A_{41} +3* A_{12} +2* A_{43} + A_{44} +5* A_1					
		1-(hydroxylamino)-3,5-bis(dimethylamino)-s-triazine	30.67	0	80.39	53.6	30.67
$C_8H_{16}O$	288.2	4* A_1 +3* A_{41} +3* A_{12} +2* A_{43} + A_{30} * F_{30} + A_{44}					
		octanal	25.86	0	89.73	95.1	25.86
$C_8H_{16}O$	252.86	6* A_2 * B_2 + A_1 + A_{34}					
		2-octanone	24.42	0	96.57	86.4	24.42

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}S_{pcc}}$ (expt)	$\Delta_0^{T_{fus}S_{pcc}}$ (calcd)	$\Delta_0^{T_{fus}H_{pcc}}$ (expt)	$\Delta_0^{T_{fus}H_{pcc}}$ (calcd)
$C_8H_{16}O_2$	289.7	0	73.8	86.9	21.35	25.2
						[216]
$C_8H_{16}O_2$	250.6	0	43.5	48.1	10.9	12.1
						[47]
$C_8H_{16}O_2$	181.68	0	82.18	78.5	14.93	14.3
						[216]
$C_8H_{16}O_2$	212.1	0	93.49	89.5	19.83	19.0
						[216]
C_8H_{18}	216.38	0	95.86	91.1	20.74	19.7
						[216]
C_8H_{18}	165.3	0	54.7	43.8	9.04	7.3
						[216]
C_8H_{18}	165.8	0	55.52	43.8	9.2	7.3
						[216]
C_8H_{18}	163.6	0	56.65	38.8	9.27	6.4
						[216]
C_8H_{18}	152.6	0	76.6	64.9	11.7	9.9
						[216]
C_8H_{18}	164.2	0	72.62	64.9	11.92	10.7
						[216]
C_8H_{18}	152.5	13.11	33.28	35.8	9.54	13.4
	373.9	20.16				[216]
C_8H_{18}	152.2	0	71.22	64.9	10.84	9.9
						[215]
$C_8H_{18}Cl_2Sn$	316.2	0	71.95	86.1	22.75	27.2
						[130]
$C_8H_{18}N_2$	242.6	20.16	59.91	56.3	15.17	14.6
	258.6	39.75				[42]
$C_8H_{18}N_2O$	268	31.12	71.06	64.9	19.86	18.7
	288.4	39.94				[42]
$C_8H_{18}N_2O_2$	405	0	63.95	80.2	25.9	32.5
						[216]
$C_8H_{18}N_4O_4$	331	0	186.35	113.0	61.68	37.4
						[225]
$C_8H_{18}O_2$	332.8	0	108.47	111.0	36.1	36.9
						[215]
$C_8H_{18}O_4$	229.3	0	103.34	96.8	23.71	22.2
						[216]
$C_8H_{18}S$	198.1	0	93.85	80.1	19.41	15.9
						[216]
$C_8H_{18}S$	224	0	108.35	105.9	4.27	23.7
						[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)	
$C_8H_{19}NSi$	199.4	N-(β -trimethylsilylethyl)trimethylenimine 12.9	0	64.68	57.7	12.9	11.5 [216]
$C_8H_{20}Ge$	180.3	$A14+A15+A119+3*A1+2*A2+A109$ tetraethylgermane 12.31	0	68.29	63.7	12.31	11.5 [216]
$C_8H_{20}O_4Si$	187.7	$4*A1+4*A2+A102$ tetraethoxysilane 13.2	70.32				
	191.0	11.14	58.33	128.66	90.6	24.34	17.3 [10]
$C_8H_{20}Pb$	141.4	$4*A1+4*A2+4*A32+A109$ tetraethyllead 9.11	0	64.43	68.7	9.11	9.7 [216]
$C_8H_{20}Si$	189.4	$4*A1+4*A2+A106$ tetraethylsilane 13.01	0	68.72	71.8	13.01	13.6 [227]
$C_8H_{20}Sn$	142.1	$4*A1+4*A2+A109$ tetraethyltin 9.15	0	64.35	74.6	9.15	10.6 [216]
$C_8H_{24}O_4Si_4$	258	$4*A1+4*A2+A110$ octamethylcyclotetrasiloxane 4.87	18.86				
	290.5	23.77	81.81	100.67	58.6	28.63	17.0 [216,98,121]
$C_8H_{28}N_4Si_4$	367.7	$8*A1+A14+5*A15+4*A139+4*A112$ octamethylcyclotetrasilazane 25.05	0	68.13	62.5	25.05	23.0 [216]
$C_9H_4Cl_3NO_2S$	454.2	$8*A1+A14+5*A15+4*A139+4*A121$ 2-[(trichloromethyl)thiol]-1H-isoindole-1,3(2H)-dione 35.49	0	78.14	74.8	35.49	34.0 [215]
$C_9H_4Cl_4O_4$	444.3	$A14+2*A15+A128+2*A19+4*A10+A4*B4+A84+3*A22*E22$ methyl tetrachloroterephthalic acid ester 16.89	0	38.01	75.1	16.89	33.4 [221]
$C_9H_4Cl_8O$	395.4	$4*A2*F22+A38+A36*F36+6*A12+A1$ 1,2,4,5,6,7,8,8-octachloro-2,3,3a,4,7,7a-hexahydro-4,7-methano-isobenzofuran 25.94	0	65.61	47.4	25.94	18.7 [232]
$C_9H_4O_5$	385	$3*A14+A15+3*A17+2*A19+4*A16+A112+8*A22*G22$ trimellitic anhydride 1,2,4-benzenetricarboxylic acid 10.46	0	27.18	33.3	10.46	12.8 [216]
$C_9H_5N_4Cl_3$	431.0	$A14+2*A15+3*A10+A12+A117+2*A19$ 4,6-dichloro-N-(2-chlorophenyl)-1,3,5-triazin-2-amine 31.48	0	73.04	68.2	31.48	29.4 [221]
$C_9H_6Cl_2N_2O_3$	396.3	$4*A10+5*A12+3*A22*G22+3*A41+A44$ 2-(3,4-dichlorophenyl)-4-methyl-1,2,4-oxadiazolidine-3,5-dione 29.5	0	74.42	63.6	29.5	25.2 [221]
$C_9H_6Cl_6O_4S$	419.7	$A14+2*A15+A125+2*A22*E22+3*A10+3*A12+A1+A126$ 6,7,8,9,10,10-hexachloro-6,9-methano-2,4,3-benzodioxathiapin-3,3-dioxide 21.66	0	51.6	51.6	21.66	21.7 [221]
$C_9H_6O_2$	342.1	$3*A14+3*A15+6*A22*D22+3*A17+2*A19+A136+2*A16$ coumarin 19.14	0	55.95	48.0	19.14	16.4 [82]
$C_9H_6O_2$	330.3	$A14+3*A15+A115+A18+A18+A19+A19+4*A10$ chromone 17.31	0	52.41	43.3	17.31	14.3 [215,216]
$C_9H_7Cl_3O_3$	450.6	$A14+3*A15+4*A10+A114+A112+2*A18*B18+2*A19$ 2-(2,4,5-trichlorophenoxy)propanoic acid 39.58	0	87.83	76.0	39.58	34.2 [221]
$C_9H_7Cl_3O_3$	361.9	$2*A10+4*A12+3*A22*E22+A36*E36+A32+A3*B3+A1$ methyl 2-(2,4,5-trichlorophenoxy)acetate 30.46	0	84.18	70.5	30.46	25.5 [221]
C_9H_7N	220	$2*A10+4*A12+A2+A32+A38+3*A22*E22+A1$ quinoline 0.07	0.31				
	258.4	10.66	41.27	41.58	47.9	10.73	12.4 [216]
C_9H_7N	299.6	$7*A10+2*A12+A41$ isoquinoline 13.54	0	45.21	47.9	13.54	14.3 [216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

T(K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$\text{C}_9\text{H}_7\text{N}_3\text{S}$	5-methyl-1,2,4-triazolo[3,4- <i>b</i>]benzothiazole 24.07	0	52.3	49.1	24.07	22.6
	$2^*A14+2^*A15+A1+3^*A10+3^*A19+A11+A119+2^*A118+A131+A18^*B18$					[221]
C_9H_8	indene 10.2	0	37.54	42.7	10.2	11.6
	$4^*A10+2^*A15+A14+2^*A18+2^*A19$					[216]
$\text{C}_9\text{H}_8\text{Cl}_2\text{O}_3$	methyl 3,6-dichloro-2-methoxybenzoate 18.49	0	60.7	64.8	18.49	19.7
	$4^*A12+2^*A10+2^*A1+A38^*D38+2^*A22^*D22+A32$					[215]
$\text{C}_9\text{H}_8\text{Cl}_2\text{O}_3$	2-(2,4-dichlorophenoxy)propanoic acid 30.43	0	78.18	74.8	30.43	29.1
	$3^*A10+3^*A12+2^*A22^*D22+A36^*D36+A32+A1+A3^*B3$					[215]
$\text{C}_9\text{H}_8\text{Cl}_2\text{O}_3$	methyl 2,4-dichlorophenoxyacetate 25.1	0	79.59	69.3	25.1	21.8
	$2^*A22^*D22+A1+3^*A12+3^*A10+A38+A32+A2$					[232]
$\text{C}_9\text{H}_8\text{O}_2$	cinnamic acid 22.63	0	55.71	52.1	22.63	21.2
	$5^*A10+A12+A6+A36+A6^*B6$					[215]
$\text{C}_9\text{H}_8\text{O}_2$	allocinnamic acid 16.95	0	49.68	52.1	16.95	17.8
	$5^*A10+A12+A6+A36+A6^*B6$					[215]
$\text{C}_9\text{H}_9\text{BrO}_3$	(<i>d</i>) 2-(<i>p</i> -bromophenoxy)propanoic acid 31.8	0	82.59	74.9	31.8	28.8
	$4^*A10+2^*A12+A1+A3^*B3+A36^*C36+A21+A32$					[220]
$\text{C}_9\text{H}_9\text{BrO}_3$	(<i>d</i>) 2-(<i>p</i> -bromophenoxy)propanoic acid 27.61	0	72.67	74.9	27.61	28.5
	$4^*A10+2^*A12+A1+A3^*B3+A36^*C36+A21+A32$					[220]
$\text{C}_9\text{H}_9\text{BrO}_3$	(<i>d</i>) 3-(<i>m</i> -bromophenyl)-3-hydroxypropanoic acid 26.78	0	76.73	74.6	26.78	26.1
	$4^*A10+A12+A11+C30^*A30+A36^*C36+A21+A2+A3^*B3$					[220]
$\text{C}_9\text{H}_9\text{BrO}_3$	(<i>d</i>) 3-(<i>m</i> -bromophenyl)-3-hydroxypropanoic acid 23.85	0	68.14	74.6	23.9	26.1
	$4^*A10+A12+A11+C30^*A30+A36^*C36+A21+A3^*B3+A2$					[220]
$\text{C}_9\text{H}_9\text{BrO}_3$	(<i>d</i>) 3-(<i>p</i> -bromophenyl)-3-hydroxypropanoic acid 28.87	0	77.82	74.6	28.9	27.7
	$4^*A10+A12+A11+A21+C30^*A30+A36^*C36+A3^*B3+A2$					[220]
$\text{C}_9\text{H}_9\text{BrO}_3$	(<i>d</i>) 3-(<i>p</i> -bromophenyl)-3-hydroxypropanoic acid 35.56	0	89.36	74.6	35.56	29.7
	$4^*A10+A12+A11+A21+C30^*A30+A36^*C36+A3^*B3+A2$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 2-(<i>o</i> -chlorophenoxy)propanoic acid 32.22	0	83.03	73.4	32.22	28.5
	$4^*A10+2^*A12+A1+A3^*B3+A36^*C36+A22^*C22+A32$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 2-(<i>o</i> -chlorophenoxy)propanoic acid 26.78	0	72.57	73.4	26.78	27.1
	$4^*A10+2^*A12+A1+A3^*B3+A36^*C36+A22^*C22+A32$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 2-(<i>m</i> -chlorophenoxy)propanoic acid 33.05	0	85.63	73.4	33.05	28.3
	$4^*A10+2^*A12+A1+A3^*B3+A36^*C36+A22^*C22+A32$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 2-(<i>m</i> -chlorophenoxy)propanoic acid 29.71	0	80.83	73.4	29.71	27.0
	$4^*A10+2^*A12+A1+A3^*B3+A36^*C36+A22^*C22+A32$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 3-(<i>p</i> -chlorophenyl)-3-hydroxypropanoic acid 29.71	0	83.21	73.1	29.71	26.1
	$4^*A10+A12+A22^*C22+C30^*A30+A36^*C36+A2+A3^*B3+A11$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 3-(<i>p</i> -chlorophenyl)-3-hydroxypropanoic acid 28.03	0	72.81	73.1	28.03	28.1
	$4^*A10+A12+A11+A22^*C22+C30^*A30+A36^*C36+A3^*B3+A2$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 3-(<i>m</i> -chlorophenyl)-3-hydroxypropanoic acid 23.85	0	70.14	73.1	23.85	24.9
	$4^*A10+A12+A11+A22^*C22+C30^*A30+A36^*C36+A2+A3^*B3$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(<i>d</i>) 3-(<i>m</i> -chlorophenyl)-3-hydroxypropanoic acid 28.03	0	76.18	73.1	28.03	26.9
	$4^*A10+A12+A11+A22^*C22+C30^*A30+A36^*C36+A3^*B3+A2$					[220]
$\text{C}_9\text{H}_9\text{ClO}_3$	(4-chloro- <i>o</i> -tolylloxy)acetic acid 29.98	0	76.3	73.3	29.98	28.8
	$3^*A10+2^*A12+A11+A2+A1+A22^*C22+A32+A36^*C36$					[215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_9H_9Cl_2NO$	363.7	18.26	0	50.22	18.26	21.2
		3',4'-dichloropropionanilide				[215]
		A1+A2+A60+3*A12+3*A10+2*A22*C22				
$C_9H_9FO_3$	290	20.5	0	70.7	20.5	21.3
		(dl) 3-(<i>m</i> -fluorophenyl)-3-hydroxypropanoic acid				[220]
		4*A10+A12+A11+A24+C30*A30+A36*C36+A2+A3*B3				
$C_9H_9FO_3$	311	24.27	0	78.03	24.27	22.9
		(d) 3-(<i>m</i> -fluorophenyl)-3-hydroxypropanoic acid				[220]
		4*A10+A12+A11+A24+C30*A30+A36*C36+A3*B3+A2				
$C_9H_9FO_3$	342	27.2	0	79.52	27.2	25.2
		(d) 3-(<i>o</i> -fluorophenyl)-3-hydroxypropanoic acid				[220]
		4*A10+A11+A12+A24+C30*A30+A36*C36+A2+A3*B3				
$C_9H_9FO_3$	348	22.59	0	64.92	22.59	25.6
		(d) 3-(<i>o</i> -fluorophenyl)-3-hydroxypropanoic acid				[220]
		4*A10+A11+A12+A24+C30*A30+A36*C36+A3*B3+A2				
$C_9H_9FO_3$	362	27.61	0	76.28	27.61	26.6
		(dl) 3-(<i>p</i> -fluorophenyl)-3-hydroxypropanoic acid				[220]
		4*A10+A11+A12+A24+C30*A30+A36*C36+A2+A3*B3				
$C_9H_9FO_3$	381	30.96	0	81.26	30.96	28.0
		(d) 3-(<i>p</i> -fluorophenyl)-3-hydroxypropanoic acid				[220]
		4*A10+A11+A12+A24+C30*A30+A36*C36+A3*B3+A2				
$C_9H_9NO_4$	416.9	31.46	0	75.46	31.46	30.4
		[(benzoylamino)oxy] acetic acid				[215]
		5*A10+A12+A60+A32+A2+A36*C36				
$C_9H_9NO_5$	411.4	32.22	0	78.31	32.22	30.8
		(dl) 2-(<i>p</i> -nitrophenoxy)propanoic acid				[220]
		4*A10+2*A12+A1+A3*B3+A36*C36+A32+A50				
$C_9H_9NO_5$	362	20.92	0	57.79	20.92	27.1
		(d) 2-(<i>p</i> -nitrophenoxy)propanoic acid				[220]
		4*A10+2*A12+A1+A3*B3+A36*C36+A32+A50				
C_9H_{10}	221.8	8.6	0	38.77	8.6	10.2
		indane				[216]
		4*A10+2*A19+A14+2*A15				
C_9H_{10}	250.8	11.92	0	47.55	11.92	13.5
		α -methylstyrene				[216]
		5*A10+A12+A1+A5+A7				
$C_9H_{10}Cl_2N_2O$	429.7	33.89	0	78.87	33.89	27.84
		3-(3,4-dichlorophenyl)-1,1-dimethylurea				[232]
		2*A1+2*A22*C22+3*A12+3*A10+A64*B64				
$C_9H_{10}BrClN_2O_2$	369.8	26.54	0	71.79	26.54	25.4
		3 (4 bromo 3 chlorophenyl)-1-methoxy-1-methylurea				[221]
		2*A1+3*A10+3*A12+A32+A22*D22+A21+A64				
$C_9H_{10}Cl_2N_2O_2$	365.8	26.56	0	72.61	26.56	24.6
		<i>N'</i> -(3,4-dichlorophenyl)- <i>N</i> -methoxy- <i>N</i> -methylurea				[215]
		2*A22*D22+A32+A64+2*A1+3*A12+3*A10				
$C_9H_{10}O$	269.8	16.26	0	60.24	16.26	13.7
		chroman				[216]
		A14+3*A15+A19+A112+4*A10				
$C_9H_{10}O$	277.5	16.75	0	60.35	16.75	14.1
		isochroman				[216]
		A14+3*A15+2*A19+A112+4*A10				
$C_9H_{10}O$	308.2	15.73	0	51.04	15.73	15.1
		cinnamyl alcohol				[220]
		5*A10+A12+2*A6+A2+A30				
$C_9H_{10}O$	386.2	14.06	0	36.4	14.06	19.6
		4-ethylbenzoic acid				[220]
		4*A10+A11+A12+A1+A2+A36				
$C_9H_{10}O_2$	321.2	17.68	0	55.04	17.68	17.7
		hydrocinnamic acid				[215]
		5*A10+A11+2*A2+A36				
$C_9H_{10}O_2$	279.8	17.32	0	61.9	17.32	17.2
		phenyl glycidyl ether				[135]
		A14+A112+A32*B32+16+A2+5*A10+A12				
$C_9H_{10}O_2S$	340.4	10.88	0	31.96	10.88	18.0
		tolyl vinyl sulfone				[238]
		4*A10+A11+A1+A5+A6+A88+A12				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_9H_{10}O_3$ 366	(dl) 3-phenyl-3-hydroxypropanoic acid 29.71 5*A10+A11+A2+A3*B3+B30*A30+A36*B36	0	81.17	59.2	29.71	21.7 [220]
$C_9H_{10}O_3$ 391	(d) 3-phenyl-3-hydroxypropanoic acid 32.64 5*A10+A11+A2+A3*B3+A30*B30*+B36*A36	0	83.47	59.2	32.64	23.1 [220]
$C_9H_{10}O_3$ 388	(dl) 2-phenoxypropionic acid 33.05 5*A10+A12+A1+A3*B3+A36*B36+A32*B32	0	85.19	58.2	33.05	22.6 [220]
$C_9H_{10}O_3$ 359	(d) 2-phenoxypropionic acid 22.59 5*A10+A12+A1+A3*B3+A36*B36+A32*B32	0	62.93	58.2	22.59	20.9 [220]
$C_9H_{10}O_3$ 358.1	4-methoxyphenylacetic acid 21.8 4*A10+A11+A12+A1+A2+A36*B36+A32*B32	0	60.88	58.1	21.8	20.8 [215]
$C_9H_{10}O_3$ 402.5	4-hydroxyphenylpropionic acid 28.9 4*A10+A11+A12+2*A2+A36*B36+A31	0	71.8	63.2	28.9	25.5 [215]
$C_9H_{10}O_3$ 472.8	4-ethoxybenzoic acid 29.4 4*A10+2*A12+A1+A2+A36*B36+A32*B32	0	62.18	60.2	29.4	28.5 [215]
$C_9H_{10}O_4$ 395	(dl) erythro phenylglyceric acid 31.38 5*A10+A11+2*A3*B3+2*C30*A30+A36*C36	0	79.44	71.9	31.38	28.4 [220]
$C_9H_{10}O_4$ 371.5	(d) erythro phenylglyceric acid 23.43 5*A10+A11+2*A3*B3+2*C30*A30+A36*C36	0	63.07	71.9	23.43	26.7 [220]
$C_9H_{11}BrN_2O$ 368.3	N'-(4-bromophenyl)-N-methoxy-N-methyl urea 24.44 2*A1+A32*C32+4*A10+2*A12+A64*C64+A21	0	66.36	67.4	24.44	24.8 [215]
$C_9H_{11}ClN_2O_2$ 353.4	N'-(4-chlorophenyl)-N-methoxy-N-methyl urea 22.54 2*A1+4*A10+2*A12+A22*C22+A64*C64+A32	0	63.78	66.0	22.54	23.3 [215]
$C_9H_{11}ClN_2O$ 447.6	3-(4-chlorophenyl)-1,1-dimethyl urea 29.46 2*A1+A64*B64+A22*B22+2*A12+4*A10	0	65.82	66.0	29.46	29.6 [215]
$C_9H_{11}ClO_3$ 366.2	2-(4-chloro-2-methylphenoxy)propanoic acid 26.43 3*A10+2*A12+A11+A22*C22+A36*C36+A32*C32+A3*B3+2*A1	0	72.16	73.9	26.43	27.1 [221]
$C_9H_{11}Cl_3NO_3PS$ 315	O,O-diethyl-O-(3,5,6-trichloro-2-pyridyl)phosphorothioate 24.53 4*A12+A10+A41+3*A22*E22+2*A1+2*A2+A79	0	77.86	87.5	24.53	27.6 [215]
$C_9H_{11}N$ 290	1,2,3,4-tetrahydroquinoline 11.81 A14+3*A15+A121+A19+4*A10+A19	0	40.73	51.8	11.81	15.0 [215]
$C_9H_{11}N$ 222.7	5,6,7,8-tetrahydroquinoline 9.08 A14+3*A15+3*A10+A41+A19+A19	0	40.75	53.1	9.08	11.8 [215]
$C_9H_{11}NO_2$ 326	ethyl phenyl carbamate 16.27 5*A10+A12+A1+A2+A69	0	49.79	64.9	16.27	21.2 [102]
$C_9H_{11}NO_2$ 362.8	ethyl 4-aminobenzoate 23.56 4*A10+2*A12+A1+A2+A38+A5	0	64.94	68.5	23.56	24.8
$C_9H_{11}NO_2$ 363.2	22.0 4*A10+2*A12+A1+A2+A38+A5	0	60.6	68.5	22.0	24.8 [215,395]
$C_9H_{11}NO_2$ 400.3	p-methoxyacetanilide 27.82 2*A1+4*A10+2*A12+A32+A60	0	69.51	56.0	27.82	22.4 [239]
C_9H_{12} 218.7	1,2,3-trimethylbenzene 0.66	3				
C_9H_{12} 230.3	1.33	5.8				
C_9H_{12} 247.8	8.18 3*A1+3*A10+3*A11	33.01	41.81	46.2	10.17	11.4 [216]
C_9H_{12} 229.3	1,2,4-trimethylbenzene 13.19 3*A1+3*A10+3*A11	0	57.53	46.2	13.19	10.59 [216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
C_9H_{12}	1,3,5-trimethylbenzene					
228.4	9.51	0	41.66	46.2	9.51	10.6 [216.3]
	$3*A1 + 3*A10 + 3*A11$					
C_9H_{12}	isopropylbenzene					
177.1	7.32	0	41.34	46.3	7.32	8.2 [92]
	$2*A1 + 5*A10 + A3 + A11$					
C_9H_{12}	<i>n</i> -propylbenzene					
173.6	9.27	0	53.39	59.3	9.27	10.3 [215]
	$5*A10 + A1 + 2*A2 + A11$					
$C_9H_{12}ClN_5$	6-chloro- <i>N</i> -cyclopropyl- <i>N'</i> -(1-methylethyl)-1,3,5-triazine-2,4-diamine					
441.6	28.76	0	65.13	60.0	28.76	26.5 [221]
	$A14 + A16 + 2*A1 + A3*B3 + A22*F22 + 3*A41 + 3*A12 + 2*A44$					
$C_9H_{12}N_2O$	1,1,-dimethyl-3-phenylurea					
404.8	22.81	0	56.35	64.9	22.81	26.3 [215]
	$2*A1 + 5*A10 + A12 + A64$					
$C_9H_{12}N_4O_2$	8-ethyltheophylline					
545.3	37.2	0	68.22	60.2	37.2	32.8 [236]
	$2*A14 + 3*A15 + 2*A125 + A118 + A121 + 3*A1 + 3*A19 + A2$					
$C_9H_{13}BrN_2O_2$	5-bromo-6-methyl-3-(1-methylpropyl)-2,4(1H,3H)-pyrimidinedione					
428.3	22.02	0	51.41	68.5	22.02	29.3 [215]
	$A14 + 3*A15 + A124 + A125 + 3*A1 + A2 + A3*B3 + A21 + 2*A19$					
$C_9H_{13}ClN_6$	2-[[4-chloro-6-(ethylamino)-1,3,5-triazin-2-yl]amino]-2methylpropanenitrile					
437.9	41.96	0	95.81	70.6	41.96	30.9 [215]
	$3*A41 + A22 + 2*A44 + 3*A1 + A2 + A4*B4 + A56 + 3*A12$					
$C_9H_{13}N_5$	6,9-dimethyl-8-ethyladenine					
436.8	29.8	0	68.22	61.5	29.8	26.9 [240]
	$A14 + 2*A15 + 3*A19 + 3*A1 + A118 + A119 + 2*A41 + A10 + A12 + A44 + A2$					
$C_9H_{14}ClN_5$	2-chloro-4,6-bis(isopropylamino)-1,3,5-triazine					
490.3	41.87	0	85.39	66.9	41.87	32.8 [215]
	$A22*F22 + 2*A44 + 3*A41 + 3*A12 + 4*A1 + 2*A3*B3$					
$C_9H_{14}O_6$	glyceryl triacetate					
275.3	25.8	0	93.73	80.2	25.8	22.1 [216]
	$2*A2 + 3*A1 + A3*B3 + 3*A38$					
$C_9H_{15}N_3O_8$	neopentyl-4,4,4-trinitrobutyrate					
333.5	22.59	0	67.75	77.3	22.59	25.8 [122]
	$3*A1 + 3*A2 + A4 + A4*B4 + 3*A50 + A38$					
C_9H_{16}	<i>trans</i> -hexahydroindane					
213.9	10.9	0	50.98	48.4	10.9	10.4 [184]
	$2*A14 + 3*A15 + 2*A16$					
C_9H_{16}	<i>cis</i> -hexahydroindane					
182.3	8.26	45.33				
184.5	0.39	2.13				
236.5	1.4	5.91	53.37	48.4	10.05	11.5 [184]
	$2*A14 + 3*A15 + 2*A16$					
$C_9H_{16}ClN_5$	6-chloro- <i>N</i> -(1,1-dimethylethyl)- <i>N'</i> -ethyl-1,3,5-triazine-2,4-diamine					
448.6	33.57	0	74.84	70.5	33.57	31.6 [221]
	$4*A1 + A2 + A4*B4 + 2*A44 + A22*F22 + 3*A41 + 3*A12$					
$C_9H_{16}N_4OS$	<i>N</i> -[5-(1,1-dimethylethyl)-1,3,4-thiadiazol-2-yl]- <i>N,N'</i> -dimethylurea					
435.3	29.48	0	67.72	68.5	29.48	29.8 [215]
	$5*A1 + A4 + A14 + 2*A15 + A131 + 2*A118 + 2*A19 + A64$					
$C_9H_{16}O_4$	azelaic acid					
380	32.67	0	85.97	97.5	32.67	37.1 [215]
	$2*A36*B36 + 7*A2*B2$					
$C_9H_{17}N$	<i>trans</i> -(<i>R,S</i>)-decahydroquinoline					
321.4	25.72	0	80.02	54.3	25.72	17.5 [215]
	$2*A14 + 4*A15 + A16 + A16 + A121$					
C_9H_{18}	1-mopene					
191.6	19.37	0	104.23	96.1	19.97	18.4 [165]
	$A1 + 6*A2*B2 + A5 + A6$					
C_9H_{18}	<i>n</i> -butylcyclopentane					
165.2	11.31	0	68.49	65.0	11.31	10.7 [216]
	$A14 + A16 + 3*A2 + 2*A15$					
C_9H_{18}	<i>n</i> -propylcyclohexane					
178.3	10.37	0	58.19	61.6	10.37	11.0 [215]
	$A14 + A1 + A16 + 2*A2 + 3*A15$					
$C_9H_{18}N_2O_2S$	3,3-dimethyl-1-(methylthio)-2-butanone <i>O</i> -methylcarbamoyloxime					
330.2	19.83	0	60.04	60.4	19.83	20.0

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}S_{tpcc}}$ (expt)	$\Delta_0^{T_{fus}S_{tpcc}}$ (calcd)	$\Delta_0^{T_{fus}H_{tpcc}}$ (expt)	$\Delta_0^{T_{fus}H_{tpcc}}$ (calcd)
$C_9H_{18}N_6$	444.4	5*A1+A4+A2+A7+A42+A84+A69 1,3,5-tris(dimethylamino)-s-triazine 23.01	0	51.78	23.01	21.8 [215]
$C_9H_{18}N_6$	333	3*A41+3*A12+3*A43+6*A1 1-(ethylamino)-3,5-bis(dimethylamino)-s-triazine 16.74	0	50.27	16.74	18.5 [215]
$C_9H_{18}O$	253.9	3*A41+3*A12+2*A43+A44+5*A1+A2 nonanal 29.6	0	116.6	29.6	26.5 [43]
$C_9H_{18}O$	269.3	A1+7*A2*B2+A34 5-nonanone 24.94	92.59			
	451.8	11.27	24.94	117.5	36.2	37.3 [215]
$C_9H_{18}O_2$	268	2*A1+6*A2+A35 nonanoic acid 5.61	20.92			
	285.5	20.31	71.13	92.05	25.91	27.5 [215]
C_9H_{20}	217.2	7*A2*B2+A1+A36 nonane 6.28	28.91			
	219.7	15.48	70.29	99.2	21.76	22.1 [215, 216]
C_9H_{20}	174.5	2,2,3,3-tetramethylpentane 7.33	42.0			
	263.4	2.33	8.9	50.9	9.66	11.3 [216]
C_9H_{20}	206.7	6*A1+A2+2*A4 2,2,4,4-tetramethylpentane 9.75	0	47.17	42.9	8.9 [216]
C_9H_{20}	208.3	3,3-diethylpentane 0.48	2.32			
	210.4	0.81	3.85			
	240.1	10.09	42.02	48.2	11.38	15.4 [216]
$C_9H_{20}N_2O$	311.5	4*A1+4*A2+A4 1,3-dibutylurea 11.1	35.63			
	346.9	14.87	42.87	78.5	25.97	27.6 [216]
$C_9H_{20}N_2O$	253	2*A1+A66+6*A2 1,1,3,3-tetraethylurea 20.55	0	81.23	79.5	20.1 [169, 124]
$C_9H_{20}O$	263	4*A1+4*A2+A63 2,2,4,4-tetramethylpentan-3-ol 1.9	7.22			
	322	7.3	22.67	29.9	9.2	6.8 [216]
$C_9H_{20}O_2$	319.6	6*A1+2*A4+A3*B3+A30 1,9-nonanediol 36.4	0	113.89	36.4	38.4 [215]
$C_9H_{20}O_2S$	290.8	9*A2*B2+2*A30*B30 3(<i>n</i> -hexylthio)-1,2-propanediol 48.5	0	166.78	109.3	48.5 [217]
$C_9H_{20}O_3$	272.9	A1+5*A2*B2+A84+2*A30*C30+A3*B3+2*A2*B2 3(<i>n</i> -hexyloxy)-1,2-propanediol 10.2	0	37.38	111.9	10.2 [243]
$C_9H_{20}S$	267.7	A1+5*A2*B2+A32+2*A30*C30+A3*B3+2*A2 1-nonanethiol 33.5	0	125.14	33.5	30.9 [136]
$C_9H_{24}Si_2$	223.7	A1+8*A2*B2+A86 1,3-hexamethyldisilylpropane 16.05	0	71.75	72.7	16.3 [216]
$C_9H_{24}Si_3$	269.3	6*A1+2*A109+3*A2 1,1,3,3,5,5-hexamethyl-1,3,5-trisilacyclocyclohexane 16.5	0	61.26	45.9	16.5 [216]
$C_{10}F_{14}$	200	A14+3*A15+3*A139+6*A1 perfluorobicyclo[4.4.0]deca-1,6-diene 0.75	3.77			
	233	1.11	4.77			
	264	10.47	39.66	49.2	73.7	12.9 [216]
$C_{10}F_{18}$		2*A14+4*A15+6*A17+12*A28+2*A19+2*A23+2*A19 <i>cis</i> -perfluorodecalin				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (calcd)
232.5	4.24	18.2				
266.7	10.3	38.62	56.82	50.3	14.54	13.4
$\text{C}_{10}\text{F}_{18}$	$2^*A14+4^*A15+10^*A17+18^*A28$ <i>trans</i> -perfluorodecalin					[216]
	17.96	0	60.96	50.3	17.96	14.8
$\text{C}_{10}\text{H}_2\text{O}_6$	$2^*A14+4^*A15+10^*A17+18^*A28$ pyromellitic dianhydride	1,2,5,6-benzenetetracarboxylic acid				[216]
	15.82	0	28.38	51.9	15.82	28.9
$\text{C}_{10}\text{H}_4\text{Cl}_2\text{O}_2$	$2^*A14+4^*A15+4^*A19+2^*A117+2^*A10$ 2,3-dichloro-1,4-naphthalenedione					[216]
	28.53	0	60.83	54.7	28.53	25.7
$\text{C}_{10}\text{H}_5\text{Cl}_4\text{NO}_2\text{S}$	$A14+3^*A15+4^*A19+2^*A114+2^*A22^*D22+4^*A10$ $3\alpha,4,7,7\alpha$ -tetrahydro-2-[(1,1,2,2-tetrachloroethyl)thio]-1H-isoindole-1,3(2H)-dione					[215]
	40.22	0	92.96	81.4	40.22	35.2
$\text{C}_{10}\text{H}_5\text{Cl}_7$	$A14+2^*A15+2^*A19+A128+4^*A10+A4^*B4+A3^*B3+4^*A22+A84$ 1,4,5,6,7,8,8-heptachloro- $3\alpha,4,7,7\alpha$ -tetrahydro-4,7-endo-methanoindene					[221]
	23.4	65.33				
$\text{C}_{10}\text{H}_5\text{Cl}_{17}\text{O}$	2.09	5.63	70.96	41.5	25.49	15.4
	$3^*A14+A15+3^*A17+7^*A19+7^*A22^*G22+2^*A18+3^*A116$ 1,4,5,6,7,8,8-heptachloro-2,3-epoxy- $3\alpha,4,7,7\alpha$ -tetrahydro-4,7-endo-methanoindan					[222]
385.2	18.9	49.07				
434.9	2.85	6.55	55.62	42.4	21.75	18.5
$\text{C}_{10}\text{H}_6\text{Cl}_8$	$4^*A14-A15+3^*A17+2^*A19+7^*A22^*G22+5^*A16+A112$ 1,2,4,5,6,7,8,8-octachloro-2,3,3,4,4,7,7-hexahydro-4,7-methano-1H-indene					[222]
	23.15	0	60.94	46.1	23.15	17.5
$\text{C}_{10}\text{H}_6\text{Cl}_4\text{O}_4$	$3^*A14+A15+3^*A17+2^*A19+2^*A16+2^*A16+8^*A22^*G22$ dimethyl-2,3,5,6-tetrachloro-1,4-benzenedicarboxylate					[221]
	30.23	0	70.01	70.3	30.23	30.4
$\text{C}_{10}\text{H}_6\text{OS}_2$	$6^*A12+4^*A22^*F22+2^*A38+2^*A1$ naphthalene 1,8-disulfide S-oxide					[215]
	3.2	8.8				
421.2	23.3	0	55.32	42.9	23.3	18.1
$\text{C}_{10}\text{H}_6\text{S}_2$	$A14+2^*A15+2^*A19+A19+6^*A10+A12+A133$ naphthalene disulfide					[176]
	13	0	32.93	34.6	13	13.7
$\text{C}_{10}\text{H}_7\text{Br}$	$A14+2^*A15+3^*A19+6^*A10+A12+A132$ 1-bromonaphthalene					[44]
	15.16	0	55.86	47.0	15.16	12.8
$\text{C}_{10}\text{H}_7\text{Br}$	$7^*A10+3^*A12+A21$ 2-bromonaphthalene					[83]
	5.77	18.09				
319	14.40	43.76	61.85	47.0	20.17	15.5
329						
$\text{C}_{10}\text{H}_7\text{Cl}$	$7^*A10+2^*A12+A12+A21$ 1-chloronaphthalene					[216]
	12.9	0	47.65	40.2	12.9	10.9
$\text{C}_{10}\text{H}_7\text{Cl}$	$7^*A10+3^*A12+A22$ 2-chloronaphthalene					[83]
	14.7	0	44.28	40.2	14.7	13.4
$\text{C}_{10}\text{H}_7\text{Cl}_5\text{O}$	$7^*A10+3^*A12+A22$ 2-(3,5-dichlorophenyl)-2(2,2,2-trichloroethyl)oxirane					[83]
	18.54	0	59.17	62.9	18.54	19.7
$\text{C}_{10}\text{H}_7\text{I}$	$A14+2^*A12+A11+3^*A10+A17+A112+5^*A22^*F22+A2+A4^*B4$ 1-iodonaphthalene					[221]
	15.91	0	56.82	48.8	15.91	13.7
$\text{C}_{10}\text{H}_7\text{I}$	$7^*A10+3^*A12+A29$ 2-iodonaphthalene					[215]
	16.04	0	48.96	48.8	16.04	16.0
$\text{C}_{10}\text{H}_7\text{NO}_2$	$7^*A10+3^*A12+A29$ 1-nitronaphthalene					[215]
	18.43	0	55.87	47.2	18.43	15.6
C_{10}H_8	$7^*A10+3^*A12+A50$ naphthalene					[215]
	19.1	0	53.75	44.4	19.1	15.7
C_{10}H_8	8^*A10+2^*A12 azulene					[215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{pcc}$ (expt)	$\Delta_0^{T_{fus}} H_{pcc}$ (calcd)
373.5	17.53	0	46.9	44.2	17.53	16.5 [244]
$C_{10}H_8ClN_3O$	2*A14+4*A15+2*A19+8*A18 5-amino-4-chloro-2-phenyl-3(2H)-pyridazinone					
479.2	26.75	0	55.83	62.8	26.75	30.1 [215]
$C_{10}H_8ClN_3O_2$	A14+3*A15+2*A19+A18*B18+A45+A22*D22+A125 +5*A10+A12+A118 4-(2-chlorophenylhydrazono)-3-methyl-5-isoxazolone					
440.4	28.04	0	63.66	61.3	28.04	27.0 [221]
$C_{10}H_8O$	α -naphthol					
369	23.01	0	62.34	49.7	23.01	18.4 [215]
$C_{10}H_8O$	7*A10+2*A12+A31+A12 β -naphthol					
393.6	18.79	0	47.7	49.7	18.79	19.6 [215]
$C_{10}H_8O_3$	7*A10+2*A12+A31+A12 4-methyl-7-hydroxycoumarin					
460.7	29.14	0	63.25	60.3	29.14	27.8 [216]
$C_{10}H_9Cl_2NO$	A14+3*A15+A115+A31+3*A19+A18*B18+3*A10+A12+A1 N-(3,4-dichlorophenyl)-2-methyl-2-propenamide					
395.5	32.04	0	81.0	57.8	32.04	22.9 [215]
$C_{10}H_9Cl_3O_3$	A1+A5+A7+3*A10+3*A12+2*A22*C22+A60 methyl 2-(2,4,5-trichlorophenoxy)propionate					
360.6	31.95	0	88.59	71.3	31.95	25.7 [215]
$C_{10}H_9Cl_3O_3$	2*A10+4*A12+2*A1+A3*B3+A32+A38+3*A22*E22 4-(2,4,5-trichlorophenoxy)butanoic acid					
386.7	30.28	0	78.3	89.6	30.28	34.6 [215]
$C_{10}H_9Cl_4NO_2S$	2*A10+4*A12+3*A22*E22+A36*E36+A32+3*A2 N-[(1,1,2,2-tetrachloroethyl)thio]-4-cyclohexene-1,2-dicarboximide					
432	43.1	0	99.76	80.5	43.1	34.8 [232]
$C_{10}H_9N$	2*A14+3*A15+2*A16+2*A18+A128+A4*B4+A3*B3+4*A22*G22+A84 1-naphthylamine					
323.2	15.53	0	48.05	50.8	15.53	16.4 [215]
$C_{10}H_9N$	7*A10+2*A12+A45+A12 2-naphthylamine					
386.2	23.33	0	60.38	50.8	23.33	19.6 [215]
$C_{10}H_9NO_2$	7*A10+2*A12+A45+A12 4-methyl-7-aminocoumarin					
499.9	32.09	0	64.19	61.4	32.09	30.7 [216]
$C_{10}H_{10}$	A14+3*A15+A115+A45+3*A19+A18*B18+3*A10+A12+A1 bullvalene					
366.5	15.25	0	41.61	35.3	15.25	12.9 [216]
$C_{10}H_{10}Cl_2O_3$	3*A14+A15+4*A16+6*A18 4-(2,4-dichlorophenoxy)butyric acid					
391.4	38.42	0	98.16	88.3	38.42	34.6 [215]
$C_{10}H_{10}O_3$	3*A10+3*A12+3*A2+A36*D36+2*A22*D22+A32 2,3-dihydro-2,2-dimethyl-7-benzofuranol-3-one					
440.6	21.79	0	49.47	49.6	21.79	21.9 [221]
$C_{10}H_{10}O_4$	A14+2*A15+2*A19+A17+A112+A114+2*A1+A31+3*A10+A11 1,2-dicarbomethoxybenzene					
274.2	16.95	0	61.92	65.2	16.95	17.9 [217]
$C_{10}H_{10}O_4$	4*A10+2*A38+2*A12+2*A1 1,3-dicarbomethoxybenzene					
341.2	25.3	0	74.15	65.2	25.3	22.3 [217]
$C_{10}H_{10}O_4$	4*A10+2*A38+2*A12+2*A1 1,4-dicarbomethoxybenzene					
413.8	32.09	0	77.55	65.2	32.09	27.0 [217]
$C_{10}H_{11}ClO_3$	4*A10+2*A38+2*A12+2*A1 (dl) 2-(2-chloro-3-methylphenoxy)propionic acid					
391.5	30.54	0	78.02	73.9	30.54	29.0 [273]
$C_{10}H_{11}ClO_3$	3*A10+2*A1+A3*B3+2*A12+A11+A32+A36*C36+A22*C22 (D) 2-(2-chloro-3-methylphenoxy)propionic acid					
359.5	22.18	0	61.68	73.9	22.18	26.6 [273]
$C_{10}H_{11}F_3N_2O$	3*A10+2*A1+A3*B3+2*A12+A11+A32+A36*C36+A22*C22 N,N-dimethyl-N'-[3-(trifluoromethyl)-phenyl]urea					
434.1	29.82	0	68.69	65.0	29.82	28.2 [215]
	2*A1+A11+A12+A4*B4+3*A25+4*A10+A64*B64					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$\text{C}_{10}\text{H}_{11}\text{F}_3\text{N}_2\text{O}_3\text{S}$ 455.7	N-[4-methyl-3-[[trifluoromethyl]sulfonyl]amino]phenyl]acetamide 40.47	0	88.81	57.9	40.47	20.4 [221]
$\text{C}_{10}\text{H}_{11}\text{NO}_3$ 408	2*A1 + 3*A10 + A11 + 2*A12 + A4*B4 + 3*A25 + A95 + A60 N-salicylidene- β -alanine 28.5	0	69.85	81.4	28.5	33.2 [216]
$\text{C}_{10}\text{H}_{12}$ 237.4	2*A2 + A36*C36 + 4*A10 + 2*A12 + A31 + A6*B6 + A42 1,2,3,4-tetrahydronaphthalene 12.45	0	52.44	49.6	12.45	11.8 [216]
$\text{C}_{10}\text{H}_{12}$ 216 304.8	A14 + 3*A15 + 4*A10 + 2*A19 <i>endo</i> -dicyclopentadiene 9.66	44.72	52.01	38.5	11.88	11.7 [216]
$\text{C}_{10}\text{H}_{12}\text{ClNO}_2$ 313.9	2.22 3*A14 + A15 + 4*A16 + 4*A18 isopropyl-3-chlorophenylcarbamate 17.75	7.28	56.55	67.0	17.75	21.0 [215]
$\text{C}_{10}\text{H}_{12}\text{ClN}_3\text{O}_2$ 431.6	2*A1 + A3*B3 + 4*A10 + 2*A12 + A69 + A22*B22 5-chloro-6-[[[(methylamino)carbonyl]oxy]imino]bicyclo[2.2.1]heptane-2-carbonitrile 26.07	0	60.4	59.6	26.07	25.7 [221]
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{O}_3$ 442.6	2*A14 + A15 + A56 + A22*D22 + 2*A16 + 2*A16 + A19 + A42 + A69 + A1 allobarbitol 32.31	0	73	73.4	32.31	32.5 [241]
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{O}_3$ 405	A14 + 3*A15 + A129 + A124 + A17 + 2*A2 + 2*A5 + 2*A6 2-ethoxyisnitrosoacetanilide 23	0	56.79	63.0	23	25.5 [216]
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{O}_3$ 490	A1 + A2 + 4*A10 + 2*A12 + A32 + A60 + A53 + A6*B6 4-ethoxyisnitrosoacetanilide 7.6	0	15.51	63.0	7.6	30.9 [216]
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{O}_3\text{S}$ 412.5	A1 + A2 + 4*A10 + 2*A12 + A32 + A60 + A53 + A6*B6 3-(1-methylethyl)-(1H)-2,1,3-benzothiadiazin-4(3H)-one 2,2-dioxide 21.77	0	52.76	53.0	21.77	21.9 [221]
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{O}_5$ 313.7	A14 + 3*A15 + 2*A19 + A125 + 4*A10 + 2*A1 + A3*B3 + A137 2-sec-butyl-4,6-dinitrophenol 21.81	0	69.54	64.4	21.81	20.2 [221]
$\text{C}_{10}\text{H}_{12}\text{N}_2\text{S}$ 375	2*A10 + A11 + 3*A12 + 2*A1 + A2 + A3 + 2*A50 + A31 N-allyl-N-phenylthiourea 27.61	0	73.64	73.7	27.61	27.6 [216]
$\text{C}_{10}\text{H}_{12}\text{N}_3\text{O}_3\text{PS}_2$ 345.3	5*A10 + A12 + A2 + A5 + A6 + A90 S-(3,4-dihydro-4-oxobenzo[d][1,2,3]-triazin-3-ylmethyl) O,O-dimethylphosphorodithioate 27.76	0	80.4	61.6	27.76	21.3 [215]
$\text{C}_{10}\text{H}_{12}\text{O}_2$ 301 422	A14 + 3*A15 + 2*A118 + 2*A19 + 4*A10 + A125 + A2 + 2*A1 + A80 4-propylbenzoic acid 3.4	11.3	66.51	57.8	26.7	24.4 [177]
$\text{C}_{10}\text{H}_{12}\text{O}_3$ 330	23.3 A1 + 2*A2 + A11 + A12 + 4*A10 + A36 (dl) 3-hydroxy-3-phenylbutyric acid 19.66	0	59.59	63.6	19.66	21.0 [220]
$\text{C}_{10}\text{H}_{12}\text{O}_3$ 357	A1 + A2 + A4*B4 + 5*A10 + A11 + A36*B36 + B30*A30 (d) 3-hydroxy-3-phenylbutyric acid 22.59	0	63.29	63.6	22.59	22.7 [220]
$\text{C}_{10}\text{H}_{12}\text{O}_3$ 360.2	A1 + A2 + A4*B4 + 5*A10 + A11 + B36*A36 + B30*A30 4-ethoxyphenylacetic acid 23	0	63.85	65.2	23	23.5 [215]
$\text{C}_{10}\text{H}_{12}\text{O}_3$ 376.9	4*A10 + A11 + A12 + 2*A2 + A1 + A36*B36 + A32 4-methoxyphenylpropionic acid 28.5	0	75.62	65.2	28.5	24.6 [215]
$\text{C}_{10}\text{H}_{12}\text{O}_3$ 369.2 369.8	4*A10 + A11 + A12 + 2*A2 + A1 + A36*B36 + A32 propyl 4-hydroxybenzoate 27.99	0	75.82	74.5	27.99	27.5
$\text{C}_{10}\text{H}_{13}\text{ClN}_2\text{O}_2$ 399.2	26.7 A1 + 2*A2 + 4*A10 + 2*A12 + A31 + A38 N'-(3-chloro-4-methoxyphenyl)-N,N-dimethylurea 27.48	0	68.86	68.7	27.48	27.5 [221]
$\text{C}_{10}\text{H}_{13}\text{ClN}_6$ 438.5	3*A1 + 3*A10 + 3*A12 + A22*C22 + A32 + A64*C64 2-((4-chloro-6-(cyclopropylamino)-1,3,5-triazin-2-yl)amino)-2-methylpropanenitrile 22.51	0	51.34	64.6	22.51	28.3 [221]
	A14 + A16 + 3*A41 + 3*A12 + A22*F22 + 2*A44 + 2*A1 + A56 + A4*B4					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{pcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{pcc}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{pcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{pcc}}}$ (calcd)
$\text{C}_{10}\text{H}_{13}\text{ClO}_3$	4-(4-chloro-2-methylphenoxy)butanoic acid 32.02	0	85.73	87.6	32.02	32.7 [221]
$\text{C}_{10}\text{H}_{13}\text{NO}_2$	propyl 4-aminobenzoate 20.54	64.61	59.18	75.6	20.54	26.2 [215]
$\text{C}_{10}\text{H}_{13}\text{NO}_2$	4*A10+2*A12+A1+A45+A38+2*A2 methyl <i>p</i> -N,N-dimethylaminobenzoate 26.07	0	70.12	52.9	26.07	19.7 [215]
$\text{C}_{10}\text{H}_{13}\text{NO}_2$	3*A1+A38+A43+4*A10+2*A12 <i>p</i> -ethoxyacetanilide 31.25	0	76.75	63.1	31.25	25.7
	30.83	0	75.5	63.1	30.8	25.7 [239, 395]
$\text{C}_{10}\text{H}_{13}\text{NO}_2$	2*A1+4*A10+2*A12+A32+A60+A2 propyl <i>N</i> -phenyl carbamate 21.08	0	63.68	72.0	21.08	23.9 [102]
$\text{C}_{10}\text{H}_{13}\text{NO}_2$	2*A2+A1+5*A10+A12+A69 isopropyl phenylcarbamate 19.37	0	53.88	65.7	19.37	23.62 [215]
$\text{C}_{10}\text{H}_{13}\text{NO}_2$	5*A10+A12+2*A1+A3*B3+A69 3,4-dimethylphenyl methylcarbamate 24.97	0	71.17	58.9	24.97	20.7 [215]
$\text{C}_{10}\text{H}_{13}\text{NO}_4$	3*A1+2*A11+A12+3*A10+A69 2-(1,3-dioxolan-2-yl)phenyl methylcarbamate 23.82	0	61.51	69.3	23.82	26.8 [221]
$\text{C}_{10}\text{H}_{13}\text{N}_5\text{O}_3$	A14+2*A15+A16+2*A112+4*A10+A11+A12+A69+A1 2-acetyl-amino-9-[(2-hydroxyethoxy)methyl]-9H-purine 54.92	0	120.9	84.0	54.92	38.2 [203]
$\text{C}_{10}\text{H}_{13}\text{N}_5\text{O}_3$	A14+2*A15+2*A19+A18*B18+2*A41+A118+A119 +3*A2+A30*F30+A32+A60+A1+A10+A12 9-[(2-acetoxyethoxy)methyl]-2-amino-9H-purine 42.69	0	104.58	88.9	42.69	36.3 [203]
$\text{C}_{10}\text{H}_{13}\text{N}_5\text{O}_4$	A14+2*A15+2*A19+A18*B18+2*A41+A118+A119+ 3*A2+A32+A38+A1+A10+A12+A45 2-acetyl-amino-9-[(2-hydroxyethoxy)methyl]-1,9-dihydro-6H-purin-6-one 53.83	0	109.81	90.3	53.83	44.3 [203]
$\text{C}_{10}\text{H}_{13}\text{N}_5\text{O}_4$	2*A14+3*A15+3*A19+A18*B18+2*A118+A119+ A124+A60+3*A2+A30*F30+A32+A1 2-amino-9-[(2-acetoxyethoxy)methyl]-1,9-dihydro-6H-purin-6-one 49.9	0	96.86	95.1	49.9	49.0 [203]
$\text{C}_{10}\text{H}_{14}$	<i>tert</i> -butylbenzene 8.40	0	39.1	45.4	8.40	9.8 [216]
$\text{C}_{10}\text{H}_{14}$	3*A1+A4+5*A10+A11 1,2,3,4-tetramethylbenzene 11.23	0	42.31	46.7	11.23	12.4 [216]
$\text{C}_{10}\text{H}_{14}$	4*A1+2*A10+4*A11 1,2,3,5-tetramethylbenzene 12.93	0	52.01	46.7	12.93	11.6 [216]
$\text{C}_{10}\text{H}_{14}$	4*A1+2*A10+4*A11 1,2,4,5-tetramethylbenzene 20.88	0	59.25	46.7	20.88	16.5 [216]
$\text{C}_{10}\text{H}_{14}$	4*A1+2*A10+4*A11 1-isopropyl-4-methylbenzene 9.67	0	47.33	46.8	9.67	9.6 [216]
$\text{C}_{10}\text{H}_{14}$	3*A1+A3+4*A10+2*A11 <i>n</i> -butylbenzene 11.22	0	60.56	66.5	11.22	12.3 [216]
$\text{C}_{10}\text{H}_{14}\text{Cl}_2\text{NO}_2\text{PS}$	5*A10+A1+3*A2+A11 O-(2,4-dichlorophenyl) O-methyl-(1-methylethyl) phosphoramidothioate 29.25	0	90.99	91.1	29.25	29.3 [221]
$\text{C}_{10}\text{H}_{14}\text{NO}_3\text{PS}$	3*A10+3*A12+3*A1+A3*B3+A82+2*A22*C22 O,O-diethyl O-4-nitrophenyl phosphorothioate 15.72	0	56.55	83.0	15.72	23.1 [215]
$\text{C}_{10}\text{H}_{14}\text{N}_4\text{O}_2$	4*A10+2*A12+2*A1+2*A2+A50+A79 8-propyltheophylline 33.3	0	62.32	67.3	33.3	36.0 [215]
	2*A14+3*A15+2*A125+A118+A121+3*A1+3*A19+2*A2					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{10}H_{14}N_4O_2$	8-isopropyltheophylline					
569.3	34.4	0	60.43	54.3	34.4	30.9
	$2*A14+3*A15+2*A125+A118+A121+2*A1+3*A19+2*A1+A3$					[215]
$C_{10}H_{14}N_6O$	1-(2-hydroxyethylmethylamino)-3,5-bis(dimethylamino)-s-triazine					
373.3	17.32	0	46.4	68.5	17.32	25.6
	$5*A1+2*A2+A30*F30+3*A43+3*A41+3*A12$					[242]
$C_{10}H_{14}O$	4- <i>tert</i> -butylphenol					
373.2	14.52	0	38.9	50.7	14.52	18.9
	$3*A1+A4+4*A10+A11+A12+A31$					[101]
$C_{10}H_{14}O$	thymol					
324.2	22.01	0	67.88	52.2	22.01	16.9
	$3*A10+2*A11+A12+3*A1+A3+A31$					[220]
$C_{10}H_{14}O_2$	4-propylbenzoic acid					
301	3.4	11.3				
422	23.3	55.21	66.51	57.8	26.7	24.4
	$A1+2*A2+4*A10+A11+A12+A36$					[177]
$C_{10}H_{14}O_3$	D-camphoric anhydride					
406	29	71.43				
495	8.7	17.58	89	45.3	37.7	22.4
	$2*A14+2*A15+A17+A17+A16+3*A1+A117$					[45]
$C_{10}H_{14}O_3$	DL-camphoric anhydride					
375	24	64				
495	8.7	17.58	81.58	45.3	32.7	17.0
	$2*A14+2*A15+A17+A17+A16+3*A1+A117$					[45]
$C_{10}H_{14}O_8$	(<i>dl</i>) dimethyl diacetyltartrate					
355.2	25.94	0	73.03	81.4	25.94	28.9
	$4*A1+2*A3*B3+4*A38$					[226]
$C_{10}H_{14}O_8$	(<i>d</i>) dimethyl diacetyltartrate					
377.2	29.29	0	77.65	81.4	29.29	30.7
	$4*A1+2*A3*B3+4*A38$					[226]
$C_{10}H_{14}Si$	1-phenyl-1-methyl-1-silacyclobutane					
210.1	12.28	0	58.45	47.5	12.28	10.0
	$5*A10+A11+A14+A15+A139+A1$					[216]
$C_{10}H_{14}Si$	vinyl dimethylphenylsilane					
190.7	12.26	0	64.28	58.9	12.26	11.2
	$2*A1+A5+A6*B6+5*A10+A12+A109$					[216]
$C_{10}H_{15}Br$	1-bromoadamantane					
279	0.88	3.15				
310.5	6.93	22.32				
396.5	3.83	9.66	35.13	42.6	11.64	16.9
	$3*A14+A15+3*A16+A17+A21$					[146]
$C_{10}H_{15}Cl$	1-chloroadamantane					
244.2	6.01	24.61				
442.5	4.87	11.01	35.62	35.9	10.88	15.9
	$3*A14+A15+3*A16+A17+A22$					[146]
$C_{10}H_{15}I$	1-iodoadamantane					
211	2.14	10.7				
347	10.22	51.1	61.8	44.5	12.36	15.4
	$3*A14+A15+3*A16+A17+A29$					[146]
$C_{10}H_{15}N_5$	6,9-dimethyl-8-propyladenine					
411.9	30.2	0	73.32	68.7	30.2	28.3
	$A14+2*A15+3*A19+3*A1+A118+A119+2*A41+A10+A12+A44+2*A2$					[240]
$C_{10}H_{15}NO$	(<i>L</i>) carvoxime					
346.5	22.72	0	65.57	58.9	22.72	20.4
	$A14+3*A15+2*A1+A16+A19+A18+A19+A5+A7+A53$					[226]
$C_{10}H_{15}NO$	(<i>DL</i>) carvoxime					
365.1	17.03	0	46.64	58.9	17.03	21.5
	$A14+3*A15+2*A1+A16+A19+A18+A19+A5+A7+A53$					[226]
$C_{10}H_{16}$	tricyclo[5.2.1.0 ^{2,6}]decane					
352	2.95	0	8.38	0	2.95	0
	No prediction made					[216]
$C_{10}H_{16}$	adamantane					
208.6	3.38	16.18				
541.2	10.9	20.14	36.32	44.9	14.28	24.3
	$3*A14+A15+4*A16$					[216, 189,192]
$C_{10}H_{16}Cl_3NOS$	S-2,3,3-trichloroallyl diisopropylthiocarbamate					
306	27.11	0	88.48	88.6	27.11	27.1

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{pce}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{pce}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{pce}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{pce}}$ (calcd)
$\text{C}_{10}\text{H}_{16}\text{NO}_4\text{PS}$	A5+A7+A4*B4+4*A1+2*A3*B3+3*A22*C22+A93 O-[4-(dimethylamino)sulfonyl]phenyl O,O-dimethyl phosphorothionate	26.5	0	81.08	74.8	26.5
326.8						24.5
$\text{C}_{10}\text{H}_{16}\text{N}_2$	A94+4*A1+A79+4*A10+2*A12 ethyl(1,1-dimethylpropyl)malononitrile	19.25	0	62.59	62.4	19.25
307.5						19.2
$\text{C}_{10}\text{H}_{16}\text{N}_4\text{O}_2\text{S}$	4*A1+2*A2+A4+A4*B4+2*A56 3-(5-(1,1-dimethylethyl)-1,3,4-thiadiazol-2-yl)-4-hydroxy-1-methyl-2-imidazolidinone	25.46	0	62.26	62.2	25.46
408.9						25.5
$\text{C}_{10}\text{H}_{16}\text{O}$	2*A14+4*A15+A127+A16+4*A1+A4+2*A19+A131+2*A118+A30*E30 (D) camphor	16.0	66.1			
242		0.23	0.62			
374		5.3	11.7	78.5	38.0	21.5
452						17.2
$\text{C}_{10}\text{H}_{16}\text{O}_2$	A114+2*A14+A15+A17+A17+A16+3*A1 1,6-cyclodecanedione	29.58	0	79.5	56.6	29.58
372.2						21.1
$\text{C}_{10}\text{H}_{16}\text{O}_4$	A14+7*A15+2*A114 1,4-cyclohexanedione bis ethylene ketal	25.77	0	72.97	54.4	25.77
353.2						19.2
$\text{C}_{10}\text{H}_{17}\text{NO}$	3*A14+5*A15+4*A112+2*A17 D camphor oxime	13.3	34.73			
383		1.8	4.63	39.35	40.6	15.1
389						15.8
$\text{C}_{10}\text{H}_{17}\text{NO}$	2*A14+A15+A16+A17+A17+3*A1+A53+A19 DL camphor oxime	3	8			
375		11.2	29.47			
380		1.2	3.09	40.57	40.6	15.4
388						15.8
$\text{C}_{10}\text{H}_{17}\text{N}_5\text{O}$	2*A14+A15+A16+A17+A17+3*A1+A53+A19 6-methoxy-N,N'-bis(1-methylethyl)-1,3,5-triazine-2,4-diamine	21.18	0	58.26	73.0	21.18
363.5						26.5
$\text{C}_{10}\text{H}_{18}$	5*A1+2*A3*B3+A32+3*A41+3*A12+2*A44 cis-decalin	14.43	0	59.45	52.1	14.43
242.8						12.65
$\text{C}_{10}\text{H}_{18}$	2*A14+4*A15+2*A16 trans-decalin	2.13	9.87			
216.1		9.49	41.22	51.1	52.1	11.62
230.2						11.99
$\text{C}_{10}\text{H}_{18}\text{N}_5\text{S}$	2*A14+4*A15+2*A16 N-(1,1-dimethylethyl)-N'-ethyl-6-(methylthio)-1,3,5-triazine-2,4-diamine	21.42	0	56.99	74.0	21.42
375.9						27.8
$\text{C}_{10}\text{H}_{18}\text{N}_6\text{O}_2$	5*A1+A2+A4*B4+2*A44+3*A41+3*A12+A84 1-(sarcosine)-3,5-bis(dimethylamino)-s-triazine	29.83	0	69.33	68.6	29.83
431						29.6
$\text{C}_{10}\text{H}_{18}\text{O}_4$	5*A1+3*A41+3*A12+3*A43+A36*F36+A2 sebacic acid	40.8	0	101.0	106.9	40.8
404						43.2
$\text{C}_{10}\text{H}_{18}\text{Si}$	8*A2*B2+2*A36*B36 5-trimethylsilyl-2-norbornene	6.84	0	33.93	78.4	6.84
201.6						15.81
$\text{C}_{10}\text{H}_{20}$	2*A14+A14+3*A16+2*A18+3*A1+A109 n-butylcyclohexane	14.14	0	71.28	68.7	14.14
198.4						13.6
$\text{C}_{10}\text{H}_{20}$	A14+A1+A16+3*A2+3*A15 1-decene	7.95	40.09			
198.3		13.81	66.73	106.8	105.5	21.76
206.9						21.8
$\text{C}_{10}\text{H}_{20}$	A1+7*A2*B2+A5+A6 2,2,5,5-tetramethylhex-3-ene	1.21	5.13			
235.8		4.33	17.78			
243.5		10.25	38.12	61.03	46.3	15.79
268.9						12.5
$\text{C}_{10}\text{H}_{20}\text{N}_6$	6*A1+2*A4+2*A6 1-(ethylmethylamino)-3,5-bis(dimethylamino)-s-triazine	21.3	0	55.46	56.3	21.3
384						21.6
$\text{C}_{10}\text{H}_{20}\text{O}$	3*A41+3*A12+2*A43+6*A1+A44+A3*B3 (dl) menthol					
						21.5

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{10}H_{20}O$	301.2	10.25	0	34.03	38.4	10.25	11.6 [226]
		$3*A1+A14+3*A15+2*A16+A30+A3+A16$ (l) menthol					
$C_{10}H_{20}O$	316.2	11.88	0	37.58	38.4	11.88	12.1 [226]
		$3*A1+A14+3*A15+2*A16+A30+A3+A16$ decanal					
$C_{10}H_{20}O_2$	268.2	30.6	0	114.1	113.8	30.6	30.5 [93]
		$8*A2*B2+A1+A34$ decanoic acid					
$C_{10}H_{22}$	304.5	27.99	0	91.28	105.6	27.82	32.2 [216]
		$8*A2*B2+A1+A36$ <i>n</i> -decane					
$C_{10}H_{22}$	243.5	28.7	0	117.99	109.8	28.7	26.7 [216]
		$2*A1+8*A2*B2$ 5-methylnonane					
$C_{10}H_{22}$	186.7	16.65	0	89.19	79.2	16.65	14.8 [216]
		$3*A1+6*A2+A3$ (DL) 4-methylnonane					
$C_{10}H_{22}$	174.7	15.19	0	86.94	79.2	15.19	13.8 [216]
		$3*A1+6*A2+A3$ (DL) 3-methylnonane					
$C_{10}H_{22}$	188.5	18.7	0	99.22	92.4	18.7	17.4 [216]
		$3*A1+6*A2*B2+A3$ 2-methylnonane					
$C_{10}H_{22}$	198.8	17.49	0	87.97	92.4	17.49	18.4 [216]
		$3*A1+6*A2*B2+A3$ 1,10-decanediol					
$C_{10}H_{22}O_2$	345.5	41.7	0	120.69	129.5	41.7	44.8 [215]
		$10*A2*B2+2*A30*B30$ 3(<i>n</i> -heptylthio)-1,2-propanediol					
$C_{10}H_{22}O_2S$	289.5	27.3	94.3				
	292.5	1.7	5.81	100.11	114.3	29.0	33.4 [217]
$C_{10}H_{22}O_3$		$A1+6*A2*B2+A84+2*A30*C30+A3*B3+2*A2$ 3(<i>n</i> -heptyloxy)-1,2-propanediol					
	288	28.8	100				
$C_{10}H_{22}O_3$	246.2	1	4.06	104.06	116.9	29.8	28.8 [217]
		$A1+6*A2*B2+A32+2*A30*C30+A3*B3+2*A2$ 1-decanethiol					
$C_{10}H_{22}S$	247.9	33.3	0	134.31	124.6	33.3	30.9 [216]
		$9*A2*B2+A1+A86$ 3(<i>n</i> -heptylamino)-1,2-propanediol					
$C_{10}H_{23}NO_2$	324.9	28.8	0	88.64	106.9	28.8	34.7 [217]
		$A1+6*A2*B2+A44+2*A30*C30+A3*B3+2*A2$ 1,1,3,3-tetraethyl-5,5-dimethylcyclotrisiloxane					
$C_{10}H_{26}O_3Si_3$	195	0.13	0.67				
	260	9.52	36.62	37.29	78.1	9.65	20.3 [216]
$C_{11}H_8O_2$		$6*A1+4*A2+3*A112+3*A139+A14+3*A15$ 1-naphthoic acid					
	435.2	19.89	0	45.7	42.8	19.89	18.6 [215]
$C_{11}H_8O_2$		$7*A10+2*A12+A36+A12$ 2-naphthoic acid					
	460.2	23.54	0	51.15	42.8	23.54	19.7 [215]
$C_{11}H_9Cl_2NO_2$		$7*A10+2*A12+A36+A12$ 4-chlorobut-2-ynyl 3-chlorophenylcarbamate					
	344.1	26.91	0	78.21	66.4	26.91	22.9 [221]
$C_{11}H_{10}$		$4*A10+2*A12+2*A2+2*A9+2*A22*C22+A69$ 1-methylnaphthalene					
	240.7	4.98	20.69				
$C_{11}H_{10}$	242.7	6.95	28.62	49.3	44.9	11.92	10.9 [216]
		$A1+7*A10+A11+2*A12$ 2-methylnaphthalene					
$C_{11}H_{10}O_2$	288.5	5.61	19.43				
	307.4	12.13	39.43	58.87	44.9	17.74	13.8 [216]
$C_{11}H_{10}O_2$		$A1+7*A10+A11+2*A12$ 2-acetyl-1-naphthol					
	371.8	22.52	0	60.57	57.0	22.52	21.2 [215]
		$6*A10+4*A12+A31+A35+A1$					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{11}H_{10}O_2$	1-acetyl-2-naphthol					
	21.34	0	63.32	57.0	21.34	19.2
$C_{11}H_{10}O_4$	6*A10+4*A12+A31+A35+A1					
	<i>p</i> -methacryloyloxybenzoic acid					
	34	0	74.73	62.7	34	28.5
$C_{11}H_{11}Cl_3O_3$	4*A10+2*A12+A36*B36+A38+A5+A7+A1					
	methyl 2-(2,4,5-trichlorophenoxy)butyrate					
	28.87	0	91.22	78.3	28.87	24.8
$C_{11}H_{12}NO_3PS$	3*A22*E22+2*A1+4*A12+2*A10+A38+A32+A2+A3*B3					
	O,O-dimethyl S-phthalimidomethyl phosphorodithioate					
	26.96	0	78.56	79.7	26.96	27.3
$C_{11}H_{12}Cl_2O_3$	A14+2*A15+A128+2*A19+4*A10+2*A1+A2+A80					
	methyl 4-(2,4-dichlorophenoxy)butyrate					
	32.64	0	105.41	83.5	32.64	25.9
$C_{11}H_{13}F_3N_2O_3S$	2*A22*D22+A1+3*A12+3*A10+A38+A32+3*A2					
	5'-(trifluoromethanesulphonamide)acet-2',4'-xylylide					
	37.66	0	82.35	61.6	37.66	28.1
$C_{11}H_{13}ClO_3$	3*A1+2*A10+2*A11+2*A12+A60+A4*B4+3*A25+A95					
	4-(4-chloro-2-methylphenoxy)butanoic acid					
	32.02	0	85.73	87.6	32.02	32.7
$C_{11}H_{13}NO_4$	3*A10+2*A12+A11+A1+3*A2+A36*C36+A22*C22+A32					
	2,3-diisopropylidenedioxyphenyl-N-methylcarbamate					
	29.45	0	73.14	62.2	29.45	25.1
$C_{11}H_{13}F_3N_4O_4$	A14+2*A15+2*A19+2*A112+A17+3*A1+A69+3*A10+A12					
	N(3),N(3)-diethyl-2,4-dinitro-6-(trifluoromethyl)-1,3-benzenediamine					
	29.13	0	78.29	69.1	29.13	25.7
$C_{11}H_{14}$	4*A12+A10+A11+3*A25+A4*B4+2*A50+A45+A43+2*A1+2*A2					
	pentacyclo[5.4.0.0[2,6].0[3,10].0[5,9]]undecane					
	4.86	29.57				
	6.38	13.41	42.98	34.3	11.24	16.3
$C_{11}H_{14}$	5*A14-4*A15+8*A16					
	1,1-dimethylindan					
	11.99	0	52.73	46.5	11.99	10.6
$C_{11}H_{14}$	A14+2*A15+A17+A1*2+2*A19+4*A10					
	4,6-dimethylindan					
	12.88	0	50.21	47.0	12.88	12.1
$C_{11}H_{14}$	A14+2*A15+2*A19+2*A11+2*A10+2*A1					
	4,7-dimethylindan					
	13.52	0	49.58	47.0	13.52	12.8
$C_{11}H_{14}ClNO$	A14+2*A15+2*A19+2*A11+2*A10+2*A1					
	2-chloro-N-isopropyl N-phenylacetamide					
	26.05	0	74.13	67.2	26.05	23.6
$C_{11}H_{14}O_2$	5*A10+A12+2*A1+A3*B3+A2+A22*B22+A59					
	4-tert-butylbenzoic acid					
	17.91	0	40.7	43.8	17.91	19.3
$C_{11}H_{14}O_3$	4*A10+A11+A12+A36+3*A1+A4					
	(dl) 3-phenyl-3-hydroxy-2,2-dimethylpropanoic acid					
	37.24	0	91.49	64.3	37.24	26.1
$C_{11}H_{14}O_3$	5*A10+A11+A3*B3+A4*B4+2*A1+B30*A30+A36*B36					
	(d) 3-phenyl-3-hydroxy-2,2-dimethylpropanoic acid					
	39.75	0	92.22	64.3	39.45	27.7
$C_{11}H_{14}O_3$	5*A10+A11+A3*B3+A4*B4+2*A1+B30*A30+A36*B36					
	4-methoxyphenylbutyric acid					
	25.3	0	76.46	72.4	25.3	24.0
$C_{11}H_{14}O_3$	4*A10+A11+A12+3*A2+A1+A36*B36+A32					
	(dl) 3-hydroxy-3-phenylvaleric acid					
	35.15	0	89.2	70.7	35.15	27.9
$C_{11}H_{14}O_3$	5*A10+A11+A1+2*A2+A4*B4+B36*A36+B30*A30					
	(d) 3-hydroxy-3-phenylvaleric acid					
	30.96	0	81.69	70.7	30.96	26.8
$C_{11}H_{15}N$	5*A10+A11+A1+2*A2+A4*B4+A36*B36+B30*A30					
	1-cyanoadamantane					
	5.5	19.64				
	15	32.75	52.39	42.8	20.5	19.6
$C_{11}H_{15}NO_2$	3*A14+A15+3*A16+A17+A56					
	4-trans-cyanocyclohexyl (E) 2-butenate					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)	
366.2	24.4	0	66.63	67.2	24.4	24.6	
$C_{11}H_{15}NO_2$	A14+3*A15+2*A16+A56+A38+A1+A6*B6+A6 butyl 4-aminobenzoate						[140]
331.1	20.46	0	61.79	82.7	20.46	27.4	
$C_{11}H_{15}NO_2$	4*A10+2*A12+A1+A45+A38+3*A2 2-(1-methyllethyl)phenyl methylcarbamate						[215]
369.3	26.14	0	70.78	59.6	26.14	22.0	
$C_{11}H_{15}NO_2$	3*A1+A3+4*A10+A11+A12+A69 4-methylthio-3,5-xylyl methylcarbamate						[221]
393.8	30.36	0	77.11	63.7	30.36	25.1	
$C_{11}H_{15}NO_3$	4*A1+2*A11+2*A12+A69+A84+2*A10 1,2-dihydro-6-neopentyl-2-oxonicotinic acid						[215]
469.2	19.33	0	41.2	60.6	19.33	28.4	
$C_{11}H_{15}N_3O_2$	A14+3*A15+A124+2*A18+2*A19+A36+3*A1+A2+A4 N-caproyl-pyrazinamide						[164]
351.7	35.95	0	102.22	90.4	35.95	31.8	
$C_{11}H_{16}$	A1+4*A2+3*A10+A12+2*A41+A71 pentamethylbenzene						[9]
296.8	1.98	6.67					
328.2	10.67	32.51	39.33	47.3	12.65	15.5	
$C_{11}H_{16}N_4O_2$	5*A1+A10+5*A11 8-butyltheophylline						[216]
509.2	32.3	0	63.43	74.5	32.3	37.9	
$C_{11}H_{16}N_4O_2$	2*A14+3*A15+2*A125+A118+A121+3*A1+3*A19+3*A2 8-tert-butyltheophylline						[236]
402.3	48.2	0	119.81	53.4	48.2	21.5	
$C_{11}H_{16}Si$	2*A14+3*A15+2*A125+A118+A121+2*A1+3*A19+3*A1+A4 vinyl dimethylbenzylsilane						[236]
204.1	11.6	0	56.83	65.3	11.6	13.3	
$C_{11}H_{17}N_5$	2*A1+A2+A109+5*A10+A11+A5+A6 6,9-dimethyl-8-butyladenine						[216]
409.2	36	0	87.98	75.8	36	31.0	
$C_{11}H_{18}$	A14+2*A15+3*A19+3*A1+A118+A119+2*A41+A10+A12+A44+3*A2 1-methyladamantane						[240]
169.5	1.91	11.27					
211.5	1.47	6.95					
392	3.71	9.46	27.68	42.7	7.09	16.7	
$C_{11}H_{19}NO_3$	3*A14+A15+3*A16+A17+A1 2-isopropoxyphenyl N-methylcarbamate						[146]
363.1	22.96	0	63.23	72.9	22.96	26.5	
$C_{11}H_{19}NS$	3*A1+A69+A3*B3+4*A10+2*A12+A32 2,4-di-tert-butylthiazole						[215]
258.2	10.5	0	40.67	52.4	10.5	13.5	
$C_{11}H_{19}N_3O$	A14+2*A15+2*A19+A18*B18+6*A1+2*A4+A131+A118 5-butyl-2-ethylamino-6-methylpyrimidin-4-ol						[61]
432.5	20.32	0	46.98	83.9	20.32	36.3	
$C_{11}H_{19}N_3S$	2*A11+2*A12+3*A1+3*A2+A44+A31+2*A41+A2 6-ethylthio-N,N'-bis(1-methylethyl)1,3,5-triazine-2,4-diamine						[215]
377.7	23.94	0	63.38	77.2	23.94	29.2	
$C_{11}H_{20}N_6$	3*A41+3*A12+5*A1+2*A3*B3+A2+A84+2*A44 1-pyrrolidinyl-3,5-bis(dimethylamino)-s-triazine						[215]
403.1	25.61	0	63.53	57.6	25.61	23.2	
$C_{11}H_{20}N_6O$	A14+2*A15+A119+4*A1+2*A43+3*A12+3*A41 1-morpholinyl-3,5-bis(dimethylamino)-s-triazine						[215]
397.4	24.69	0	62.13	62.5	24.69	24.8	
$C_{11}H_{20}N_7S$	A14+3*A15+A119+A112+4*A1+2*A43+3*A41+3*A12 1-(thiomorpholinyl)-3,5-bis(dimethylamino)-s-triazine						[215]
391.2	29.08	0	74.33	64.2	29.08	25.1	
$C_{11}H_{20}O_2$	4*A1+2*A43+3*A41+3*A12+A14+3*A15+A119+A131 undecanolactone						[215]
250.2	3.36	13.43					
275.3	12.61	45.8	59.23	69.8	15.97	19.2	
$C_{11}H_{20}O_4$	A14+9*A15+A115 undecanedioic acid						[216]
385	39.65	0	102.99	116.2	39.65	44.7	
$C_{11}H_{21}N_5S$	9*A2*B2+2*A36*B36 6-(ethylthio)-N,N'-bis(1-methylethyl)-1,3,5-triazine-2,4-diamine						[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}S_{tpcc}}$ (expt)	$\Delta_0^{T_{fus}S_{tpcc}}$ (calcd)	$\Delta_0^{T_{fus}H_{tpcc}}$ (expt)	$\Delta_0^{T_{fus}H_{tpcc}}$ (calcd)	
377.7	23.94	0	63.39	77.2	23.94	29.2	
$C_{11}H_{21}N_7$	5*A1+A2+2*A3*B3+3*A41+2*A44+A84+3*A12 1-(piperiziny)-3,5-bis(dimethylamino)-s-triazine						[221]
382	23.01	0	60.24	63.5	23.01	24.3	
$C_{11}H_{22}$	4*A1+2*A43+3*A41+3*A12+A14+3*A15+A119+A121 1-undecene						[215]
217.3	9.2	42.36				25.7	
224	16.99	75.84	118.2	114.8	26.19	[216]	
$C_{11}H_{22}O$	A1+8*A2*B2+A5+A6 2-undecanone						
290.5	28.78	0	99.07	114.4	28.78	33.2	
$C_{11}H_{22}O_2$	2*A1+8*A2*B2+A35 undecanoic acid						[21]
290	8.13	28.03				34.7	
301.6	25.98	86.15	114.22	114.9	34.11	[216]	
$C_{11}H_{24}$	9*A2*B2+A1+A36 <i>n</i> -undecane						
236.6	6.86	29				29.5	
247.6	22.18	89.6	118.6	119.1	29.03	[216]	
$C_{11}H_{24}$	2*A1+9*A2*B2 2-methyldecane						
224.3	25.06	0	111.73	101.7	25.06	22.8	
$C_{11}H_{24}O$	3*A1+7*A2*B2+A3 methyl <i>n</i> -decyl ether						[216]
243.5	31.71	0	130.12	123.8	31.71	30.2	
$C_{11}H_{24}O_2S$	2*A1+9*A2*B2+A32 3(<i>n</i> -octylthio)-1,2-propanediol						[216]
306.5	39.8	0	129.85	123.4	39.8	37.8	
$C_{11}H_{24}O_3$	A1+7*A2*B2+A84+2*A30*C30+A3*B3+2*A2 3(<i>n</i> -octyloxy)-1,2-propanediol						[217]
296.1	33.4	0	112.8	126.0	33.4	37.3	
$C_{11}H_{25}NO_2$	A1+7*A2*B2+A32+2*A30*C30+A3*B3+2*A2 3(<i>n</i> -octylamino)-1,2-propanediol						[217]
335.9	45.1	0	134.27	116.0	45.1	39.0	
$C_{12}Cl_{10}$	A1+7*A2*B2+A44+2*A30*C30+A3*B3+2*A2 decachlorobiphenyl						[217]
577.7	39.34	0	68.1	72.1	39.34	41.6	
$C_{12}F_{26}$	12*A12+10*A22*G22 perfluorododecane						[215]
170.2	6.9	40.54				46.4	
348.5	38.16	109.5	150.0	133.3	45.06	[67]	
$C_{12}HCl_9$	12*A4*B4+6*A25+20*A26 2,2',3,3',4,5,5',6,6'-nonachlorobiphenyl						
455.8	22.6	0	49.58	70.8	22.6	32.3	
$C_{12}H_2Cl_8$	11*A12+9*A22*G22+A10 2,2',3,3',5,5',6,6'-octachlorobiphenyl						[215]
433.8	22.8	0	52.56	69.5	22.8	30.1	
$C_{12}H_3Cl_7$	10*A12+8*A22*G22+2*A10 2,2',3,3',5,5',6-heptachlorobiphenyl						[215]
395.4	20.3	0	51.34	68.2	20.3	27.0	
$C_{12}H_4Cl_6$	9*A12+7*A22*G22+3*A10 2,2',3,3',5,5'-hexachlorobiphenyl						[215]
424.9	29.2	0	68.72	66.9	29.2	28.4	
$C_{12}H_4Cl_6$	6*A22*F22+4*A10+8*A12 2,2',3,3',6,6'-hexachlorobiphenyl						[215]
385.2	21.1	0	54.78	66.9	21.1	25.8	
$C_{12}H_4Cl_6$	6*A22*F22+4*A10+8*A12 2,2',4,4',6,6'-hexachlorobiphenyl						[215]
386.7	17.5	0	45.25	66.9	17.5	25.9	
$C_{12}H_5Cl_5O_2$	6*A22*F22+4*A10+8*A12 1,3,7-trichlorodibenzodioxin						[215]
421.7	30.8	0	73.04	61.0	30.8	25.7	
$C_{12}H_5Cl_5$	A14+3*A15+2*A112+3*A22*E22+4*A19+5*A10+3*A12 2,2',4,5,5'-pentachlorobiphenyl						[20]
350.1	18.8	0	53.7	65.6	18.8	23.0	
$C_{12}H_5Cl_5$	5*A22*E22+5*A10+7*A12 2,3,4,5,6-pentachlorobiphenyl						[215]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
397.6	21.8	0	54.83	65.6	21.8	26.1
$C_{12}H_6Cl_2O$	5*A22*E22+5*A10+7*A12 3,6-dichlorodibenzofuran					[215]
461.2	32.4	0	70.25	54.8	32.4	25.3
$C_{12}H_6Cl_4$	A14+2*A15+2*A19+2*A19+A112+2*A22*C22+6*A10+2*A12 2,2',4',5-tetrachlorobiphenyl					[20]
339.1	23.4	0	69.01	64.3	23.4	21.8
$C_{12}H_6Cl_4$	6*A12+6*A10+4*A22*D22 2,3,4,5-tetrachlorobiphenyl					[215]
363.9	25.2	0	69.25	64.3	25.2	23.4
$C_{12}H_6Cl_4O_2S$	6*A12+6*A10+4*A22*D22 1,2,4-trichloro-5-((4-chlorophenyl)sulfonyl)benzene					[215]
419.9	28.94	71.21	68.92	64.6	28.94	27.1
$C_{12}H_6O_3$	6*A12+6*A10+4*A22*E22+A88 1-8-naphthalic anhydride					[215]
542.3	23.32	0	43	47.0	23.32	22.5
$C_{12}H_6S$	A14+3*A15+3*A19+A117+6*A10+A12 dibenzothiophene					[221]
373.2	21.6	0	57.74	53.9	21.6	20.1
$C_{12}H_7ClO_2$	8*A10+A131+A14+2*A15+4*A19 1-chlorodibenzodioxin					[283]
378.2	23.2	0	61.34	58.5	23.2	22.1
$C_{12}H_7ClO_2$	A14+3*A15+2*A112+4*A19+A12+7*A10+A22*C22 2-chlorodibenzodioxin					[20]
362.2	23.1	0	63.78	58.5	23.1	21.2
$C_{12}H_7Cl_2NO_3$	A14+3*A15+2*A112+4*A19+A12+7*A10+A22*C22 2,4-dichlorophenyl 4-nitrophenyl ether					[20]
342	22.96	0	67.13	69.3	22.96	23.7
$C_{12}H_7Cl_3$	7*A10+5*A12+2*A22*D22+A50+A32 2,4,6-trichlorobiphenyl					[221]
334.3	16.5	0	49.36	63.0	16.5	21.1
$C_{12}H_7Cl_3$	5*A12+7*A10+3*A22*C22 2,4,5-trichlorobiphenyl					[215]
349.5	22.8	0	65.24	63.0	22.8	22.0
$C_{12}H_8$	5*A12+7*A10+3*A22*C22 acenaphthylene					[215]
116.6	1.4	12.12				
362.6	6.95	19.15	31.27	37.8	8.36	13.7
362.0	10.96	30.28	42.40	37.8	12.36	13.7
$C_{12}H_8Br_2$	A14+2*A15+3*A19+6*A10+A12+2*A16 (dl) 1,2-dibromoacenaphthene					[216,154]
397	25.1	0	63.22	46.4	25.1	18.4
$C_{12}H_8Br_2$	6*A10+A14+2*A15+2*A21+2*A16+A12+3*A19 (d) 1,2-dibromoacenaphthene					[273]
416	26.36	0	63.35	46.4	26.36	19.3
$C_{12}H_8Cl_2$	6*A10+A14+2*A15+2*A21+2*A16+A12+3*A19 (dl) 1,2-dichloroacenaphthene					[273]
339	20.5	0	60.46	43.9	20.5	14.9
$C_{12}H_8Cl_2$	2*A15+A14+6*A10+2*A22*B22+2*A16+3*A19 (d) 1,2-dichloroacenaphthene					[273]
375	21.34	0	56.9	43.9	21.34	16.5
$C_{12}H_8Cl_2$	2*A15+A14+6*A10+2*A22*B22+2*A16+3*A19 2,6-dichlorobiphenyl					[273]
307.9	12.6	0	40.92	61.8	12.6	19.0
$C_{12}H_8Cl_2O_2S$	4*A12+8*A10+2*A22*B22 4,4'-dichlorodiphenylsulphone					[215]
422	24.4	0	57.82	62.0	24.4	26.2
$C_{12}H_8Cl_2O_3S$	8*A10+4*A12+2*A22*C22+A88 4-chlorophenyl 4-chlorobenzenesulfonate					[216]
360.0	23.63	0	65.64	69.7	23.63	25.1
$C_{12}H_8Cl_6O$	8*A10+4*A12+2*A22*C22+A89 1,2,3,4,10,10-hexachloro-6,7-epoxy-1,4,4 α ,5,6,7,8,8 α -octahydro-1,4-endo, exo-5,8-dimethanonaphthalene (Dieldrin)					[215]
405.6	19.33	47.66				
452.9	3.04	6.71	54.37	41.2	22.37	18.7
$C_{12}H_8Cl_6O$	5*A14-2*A15+6*A22*G22+3*A17+2*A19+4*A16+2*A16+A112 1,2,3,4,10,10-hexachloro-6,7-epoxy-1,4,4 α ,5,6,7,8,8 α -octahydro-1,4-endo,					[222]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{fus} S_{tpcc}$ (expt)	$\Delta_0^{fus} S_{tpcc}$ (calcd)	$\Delta_0^{fus} H_{tpcc}$ (expt)	$\Delta_0^{fus} H_{tpcc}$ (calcd)
	endo-5,8-dimethanonaphthalene (Endrin)					
383.7	16.59	43.24				
562.4	4.15	7.38	50.62	41.2	20.74	23.2
$C_{12}H_8N_2$	5*A14-2*A15+6*A22*G22+3*A17+2*A19+6*A16+A112					
	phenazine					
450.2	20.92	0	46.47	51.2	20.92	23.1
$C_{12}H_8N_2$	8*A10+4*A12+2*A41					
	benzo[c]cinnoline					
432.2	20.92	0	48.4	51.2	20.92	22.1
$C_{12}H_8N_2O_5$	8*A10+4*A12+2*A41					
	4,4'-dinitrodiphenyl ether					
418.2	10.29	0	24.61	69.6	10.29	29.1
$C_{12}H_8O$	8*A10+4*A12+2*A50+A32					
	dibenzofuran					
355.7	18.6	0	52.29	52.3	18.6	18.6
$C_{12}H_8OS$	A14+2*A15+A112+2*A19+8*A10+2*A19					
	phenoxathiin					
328.8	20.27	0	61.63	58.9	20.27	19.4
$C_{12}H_8OS_2$	3*A15+A14+4*A19+8*A10+A131+A112					
	diphenylene-2,2'-disulfide S-oxide					
407	17.99	0	44.2	56.6	17.99	23.1
$C_{12}H_8O_2$	8*A10+A14+3*A15+4*A19+A133					
	dibenzodioxin					
395.7	23.2	0	58.63	57.2	23.2	22.6
$C_{12}H_8S$	A14+3*A15+4*A19+8*A10+2*A112					
	dibenzothiophene					
371	21.58	0	58.17	53.9	21.58	20.0
$C_{12}H_8S_2$	A14+2*A15+2*A19+2*A19+A131+8*A10					
	dibenzo[c,e][1,2]dithiin					
386.2	19.3	0	49.97	48.3	19.3	18.7
$C_{12}H_8S_2$	A14+3*A15+2*A19+2*A19+8*A10+A132					
	thianthrene					
429.6	27.55	0	64.13	60.5	27.55	26.0
$C_{12}H_9Cl$	3*A15+A14+8*A10+4*A19+2*A131					
	2-chlorobiphenyl					
304.9	14.54	0	47.7	55.1	14.54	16.8
$C_{12}H_9Cl$	3*A12+9*A10+A22					
	4-chlorobiphenyl					
348.6	13.32	0	38.2	55.1	13.32	19.2
$C_{12}H_9ClN_2$	2*A12+9*A10+A22+A12					
	4-chloroazobenzene					
361.2	27.2	0	75.3	57.0	27.2	20.6
$C_{12}H_8Cl_2O_3S$	9*A10+3*A12+2*A42+A22*B22					
	4-chlorophenylbenzenesulfonate					
332.2	21.44	0	64.53	68.4	21.44	22.7
$C_{12}H_9Cl_3Si$	9*A10+3*A12+A22*B22+A89					
	o-trichlorosilylbiphenyl					
289.5	0.06	0.2				
339.2	20.72	61.09	61.28	63.6	20.78	21.6
$C_{12}H_9Cl_3Si$	9*A10+2*A12+A11+3*A22*D22+A109					
	p-trichlorosilylbiphenyl					
372.9	18.57	0	49.8	63.6	18.57	23.7
$C_{12}H_9Cl_3Si$	9*A10+2*A12+A11+3*A22*D22+A109					
	4-trichlorosilylbiphenyl					
372.9	18.57	0	49.8	65.8	18.57	24.5
$C_{12}H_9N$	9*A10+3*A12+A109+3*A22*D22					
	carbazole					
521	27.2	0	52.2	53.2	27.2	27.5
$C_{12}H_9NS$	A14+2*A15+2*A19+2*A19+8*A10+A121					
	10H-phenothiazine					
438.2	26.92	0	58.75	59.8	26.92	27.4
$C_{12}H_{10}$	A14+3*A15+4*A19+8*A10+A121+A131					
	acenaphthene					
366.6	21.46	0	58.55	41.09	21.46	15.0
$C_{12}H_{10}$	6*A10+A14+2*A15+3*A19+A12					
	biphenyl					
341.5	18.66	0	54.81	59.2	18.66	20.2

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
	10*A10+2*A12					[216]
$\text{C}_{12}\text{H}_{10}\text{N}_2$	<i>trans</i> -azobenzene 22.53	0	66.06	55.7	22.53	19.0
	10*A10+2*A12+2*A42					[216]
$\text{C}_{12}\text{H}_{10}\text{N}_2\text{O}$	azoxybenzene 17.93	0	57.99	64.3	17.93	19.9
	10*A10+2*A12+A54+A42					[215]
$\text{C}_{12}\text{H}_{10}\text{N}_2\text{O}$	4-hydroxyazobenzene 32.99	0	77.59	61.0	32.99	26.0
	9*A10+3*A12+2*A42+A31					[13]
$\text{C}_{12}\text{H}_{10}\text{N}_4\text{O}_2$	4-(4-nitrophenylazo)aniline 31.88	0	65.3	65.0	31.88	31.7
	8*A10+4*A12+2*A42+A50+A45					[13]
$\text{C}_{12}\text{H}_{10}\text{O}$	<i>o</i> -hydroxybiphenyl 16.21	0	48.12	64.6	16.21	21.3
	9*A10+A31+3*A12					[63]
$\text{C}_{12}\text{H}_{10}\text{O}$	diphenyl ether 17.21	0	57.32	63.9	17.21	19.2
	10*A10+2*A12+A32					[216]
$\text{C}_{12}\text{H}_{10}\text{O}_2$	1-naphthaleneacetic acid 22.26	0	54.92	47.8	22.26	19.4
	7*A10+2*A12+A11+A2+A36					[215]
$\text{C}_{12}\text{H}_{10}\text{O}_3$	2-carbomethoxynaphthalene 27.1	0	77.4	54.7	27.1	19.2
	7*A10+A1+A38+3*A12					[247]
$\text{C}_{12}\text{H}_{10}\text{O}_2$	1-acetoxynaphthalene 20.21	0	63.31	54.7	20.21	17.5
	7*A10+3*A12+A1+A38					[118]
$\text{C}_{12}\text{H}_{10}\text{O}_2$	2-acetoxynaphthalene 20.05	0	58.59	54.7	20.05	18.7
	7*A10+3*A12+A1+A38					[118]
$\text{C}_{12}\text{H}_{11}\text{Cl}_2\text{NO}$	3,5-dichloro-N-(1,1-dimethyl-2-propynyl)benzamide 28.68	0	66.94	58.1	28.68	24.9
	3*A12+3*A10+A60+2*A22+C22+2*A1+A4*B4+A8+A9					[221]
$\text{C}_{12}\text{H}_{11}\text{N}$	diphenylamine 17.86	0	54.75	53.9	17.86	17.6
	10*A10+2*A12+A44					[215]
$\text{C}_{12}\text{H}_{11}\text{N}$	2-aminobiphenyl 13.99	0	43.4	65.7	13.99	21.2
	9*A10+3*A12+A45					[205]
$\text{C}_{12}\text{H}_{11}\text{NO}$	1-naphthaleneacetamide 32.82	0	72.05	62.4	32.82	28.4
	7*A10+2*A12+A11+A2+A61					[221]
$\text{C}_{12}\text{H}_{11}\text{NO}_2$	1-naphthyl methylcarbamate 24.51	0	58.88	57.6	24.51	24.0
	7*A10+3*A12+A1+A69					[221]
$\text{C}_{12}\text{H}_{11}\text{N}_3$	<i>p</i> -phenylazoaniline 21.7	0	54.5	62.1	21.7	24.7
	9*A10+3*A12+2*A42+A45					[13]
$\text{C}_{12}\text{H}_{12}$	1,8-dimethylnaphthalene 15.77	0	46.9	45.5	15.77	15.3
	2*A1+2*A11+6*A10+2*A12					[215]
$\text{C}_{12}\text{H}_{12}$	2,6-dimethylnaphthalene 25.06	0	65.39	45.5	25.06	17.4
	2*A1+2*A11+6*A10+2*A12					[215]
$\text{C}_{12}\text{H}_{12}$	2,7-dimethylnaphthalene 23.35	0	63.3	45.5	23.35	16.8
	2*A1+2*A11+6*A10+2*A12					[215]
$\text{C}_{12}\text{H}_{12}$	1,4-dimethylnaphthalene 10.6	0	37.87	45.5	10.6	12.7
	2*A1+2*A11+2*A12+6*A10					[215]
$\text{C}_{12}\text{H}_{12}$	2,3-dimethylnaphthalene 15.9	0	42.06	45.5	15.9	17.2
	2*A1+2*A11+2*A12+6*A10					[215]
$\text{C}_{12}\text{H}_{12}\text{Ge}$	diphenylgermane 11.91	0	49.58	44.5	11.91	10.7

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
						[133]
$C_{12}H_{12}N_2$	10*A10+2*A12+A103 hydrazobenzene					
407.2	17.65	0	43.34	48.6	17.65	19.8
						[215]
$C_{12}H_{12}N_2O$	10*A10+2*A12+2*A44 4'4'-diaminodiphenyl ether					
465.4	7.74	0	16.63	76.8	7.74	35.8
						[216]
$C_{12}H_{12}O_4$	8*A10+4*A12+2*A45+A32 1,4-dimethylcubane dicarboxylate					
437.8	41	0	93.65	34.0	41	14.9
						[340]
$C_{12}H_{12}O_6$	5*A14-7*A15+6*A16+2*A17+2*A1+2*A38 1,2,3-tricarboxymethoxybenzene					
375.7	32.7	0	87.04	75.6	32.7	28.4
						[217]
$C_{12}H_{13}ClF_3N_3O_4$	3*A1+3*A38+3*A12+3*A10 N-(2-chloroethyl)-2,6-dinitro-N-propyl-4-(trifluoromethyl)benzeneamine					
318.4	23.08	0	72.49	75.5	23.08	24.0
						[221]
$C_{12}H_{13}NO_2$	2*A10+3*A112+A11+4*A2+A1+A4*B4+3*A25+2*A50+A43+A22*G22 4-methyl-7-dimethylaminocoumarin					
416.1	23.92	0	57.47	53.0	23.92	22.0
						[216]
$C_{12}H_{13}NO_3S$	A14+3*A15+A115+2*A19+A19+A18*B18+3*A1+3*A10+A12+A43 2,3-dihydro-6-methyl-5-phenylcarbamoyl-1,4-oxathiin-4,4-dioxide					
401.5	26.66	0	66.39	59.4	26.66	23.8
						[221]
$C_{12}H_{14}N_2O_5$	A14+3*A15+A112+A1+2*A19+A134+A60+5*A10+A12 2-cyclohexyl-4,6-dinitrophenol					
378.7	28.03	0	74.02	68.3	28.03	25.9
						[232]
$C_{12}H_{14}N_4O$	A14+3*A15+A16+2*A10+A11+3*A12+A31+2*A50 3,3',4,4'-tetraaminodiphenyl ether					
402.6	25.3	0	62.84	89.8	25.3	36.2
						[227]
$C_{12}H_{14}O_4$	4*A45+6*A12+6*A10+A32 diethyl <i>o</i> -phthalate					
269.9	17.99	0	66.66	79.5	17.99	21.5
						[216]
$C_{12}H_{14}O_4$	4*A10+2*A38+2*A12+2*A2+2*A1 diethyl terephthalate					
317.2	24.69	0	77.82	79.5	24.69	25.2
						[216]
$C_{12}H_{15}ClNO_3S_2$	4*A10+2*A38+2*A12+2*A2+2*A1 S 6 chloro 2,3 dihydro 2 oxobenzoxazol 3 ylmethyl O,O diethyl phosphorodithioate					
320.0	30.03	0	93.86	89.0	30.03	28.5
						[221]
$C_{12}H_{15}NO_2$	A14+2*A15+A126+2*A19+3*A10+A12+A22*C22+3*A2+2*A1+A80 phenylaminoethyl methacrylate					
297.5	25.47	0	85.6	70.4	25.47	21.0
						[216]
$C_{12}H_{15}NO_3$	5*A10+A12+A44+2*A2+A1+A5+A7+A38 2,3-dihydro-2,2-dimethylbenzofuran-7-yl methylcarbamate					
426.24	30.33	0	71.17	61.0	30.33	26.0
						[221]
$C_{12}H_{15}N_2O_3PS$	A14+2*A15+A17+7*A19+A112+3*A10+A12+3*A11+A69 O,O-diethyl O-quinaxalin-2-yl phosphothioate					
304.1	25.4	0	83.5	87.0	25.4	26.5
						[221]
$C_{12}H_{15}N_2O_4$	5*A10+3*A12+2*A41+2*A1+2*A2+A79 9-[(2-acetoxyethoxy)methyl]-2-acetylamino-9H-purine					
407.2	42.33	0	104.0	86.5	42.33	35.2
						[203]
$C_{12}H_{15}N_2O_5$	A14+2*A15+2*A19+A18*B18+A118+A119+ 2*A41+A10+A12+A60+2*A1+3*A2+A38+A32 9-[(2-acetoxyethoxy)methyl]-2-acetylamino-1,9-dihydro-6H-purin-6-one					
477.2	47.37	0	99.27	92.8	47.37	44.3
						[203]
$C_{12}H_{16}$	2*A14+3*A15+3*A19+A18*B18+2*A118+A119+A124+A60+2*A1+3*A2+A38+A32 cyclohexylbenzene					
280.5	15.3	0	54.55	57.2	15.3	16.1
						[216]
$C_{12}H_{16}Cl_2N_2O$	A14+3*A15+5*A10+A11+A16 N-butyl-N'-(3,4-dichlorophenyl)-N-methylurea					
374.3	27.23	0	72.75	86.2	27.23	32.3
						[221]
$C_{12}H_{16}NO_2$	2*A1+3*A2+3*A10+3*A12+2*A22*C22+A64*B64 5-isopropyl- <i>m</i> -tolyl methylcarbamate					
361.3	23.04	0	63.77	60.1	23.04	21.7
						[221]
$C_{12}H_{16}N_2O_2$	4*A1+A3+2*A11+A12+3*A10+A69 4-dimethylamino-3,5-xylyl methylcarbamate					
361.7	18.37	0	50.78	56.9	18.37	20.6
						[221]
	5*A1+2*A10+2*A11+2*A12+A69+A43					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{12}H_{16}N_3O_6S$	424.3	28.05	0	66.1	28.05	33.8
						[221]
$C_{12}H_{16}N_3O_3PS_2$	322.2	25.22	0	78.26	25.22	24.4
						[221]
$C_{12}H_{17}NO_2$	325.1	23.93	0	73.61	23.93	29.2
						[215]
$C_{12}H_{18}$	383.7	1.76	4.58			
	438.7	20.63	47.02	51.6	22.38	21.0
						[216]
$C_{12}H_{18}N_2O$	430.5	33.87	0	78.68	33.87	28.7
						[215]
$C_{12}H_{18}N_2O_2$	361.7	18.37	0	50.79	18.37	20.6
						[215]
$C_{12}H_{18}N_4O_2$	498.4	35.1	0	70.43	35.1	40.7
						[215]
$C_{12}H_{18}N_4O_6S$	414.8	38.48	0	92.78	38.48	37.4
						[221]
$C_{12}H_{18}O_2$	341.5	19.04	0	55.75	19.04	31.2
						[215]
$C_{12}H_{19}ClNO_3$	332.0	21.98	0	66.19	21.98	22.0
						[221]
$C_{12}H_{20}$	221	7.36	33.3			
	245	0.92	3.76	37.06	8.28	9.9
						[146]
$C_{12}H_{20}N_4O_2$	389.6	20.36	0	52.26	20.36	19.3
						[221]
$C_{12}H_{20}O_2$	405.2	15.77	0	38.93	15.77	25.9
						[114]
$C_{12}H_{20}O_3$	344.2	18.83	0	54.7	18.83	20.33
						[109]
$C_{12}H_{20}O_4$	296.2	18.03	0	60.88	18.03	18.3
						[114]
$C_{12}H_{22}$	256.1	1.54	6.01			
	267.5	0.74	2.77			
	273.5	7.08	25.89			
	277.2	6.78	24.46	59.13	15.4	16.5
						[216]
$C_{12}H_{22}N_2O_2$	517.4	13.6	26.29			
	617.8	49.3	79.8	106.1	62.9	49.1
						[248]
$C_{12}H_{22}N_6$	361.5	23.22	0	64.23	23.22	22.2
						[215]
$C_{12}H_{22}O_2$	230.3	24.9	0	104.6	24.9	26.4
						[227]
$C_{12}H_{22}O_2$	236.5	23.36	0	98.78	23.36	28.7
						[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (calcd)
$\text{C}_{12}\text{H}_{22}\text{O}_4$	402.5	dodecanedioic acid 50.57	0	125.64	50.6	50.53 [216]
$\text{C}_{12}\text{H}_{22}\text{O}_4$	244.1	$10^*A2^*B2 + 2^*A36^*B36$ di- <i>n</i> -butyl succinate 29.21	0	119.65	29.21	26.3 [216]
$\text{C}_{12}\text{H}_{23}\text{N}_7$	354.2	$2^*A1 + 8^*A2 + 2^*A38$ 1-(4'-methylpiperiziny)-3,5-bis(dimethylamino)- <i>s</i> -triazine 20.42	0	57.65	20.42	21.1 [242]
$\text{C}_{12}\text{H}_{24}$	199 333.8	$5^*A1 + 2^*A43 + 3^*A41 + 3^*A12 + A14 + 3^*A15 + 2^*A119$ cyclododecane 0.6 14.8	3.02 44.34	47.35	15.4	22.3 [181]
$\text{C}_{12}\text{H}_{24}$	212.9 237.9	$A14 + 9^*A15$ 1-dodecene 4.55 19.87	21.38 83.54	104.92	24.43	29.5 [216]
$\text{C}_{12}\text{H}_{24}\text{N}_2\text{O}_2$	452	$A1 + 9^*A2^*B2 + A5 + A6$ N,N' -di- <i>n</i> -propyladipamide 36.11	0	79.91	36.11	43.0 [282]
$\text{C}_{12}\text{H}_{24}\text{O}_2$	316.9	$4^*A2 + 2^*A1 + 2^*A60 + 4^*A2$ dodecanoic acid 36.65	0	115.7	36.7	39.4 [216]
$\text{C}_{12}\text{H}_{24}\text{O}_4$	383.0	$10^*A2^*B2 + A1 + A36$ 2,2,8,8-tetramethyl-1,3,7,9-tetraoxacyclododecane 23.4	0	61.1	23.4	27.9 [47]
$\text{C}_{12}\text{H}_{24}\text{O}_4$	332.0	$A14 + 9^*A15 + 4^*A112 + 4^*A1 + 2^*A17$ 1,3,9,11-tetraoxacyclohexadecane 35.56	0	107.1	35.56	28.7 [117]
$\text{C}_{12}\text{H}_{24}\text{O}_6$	312.2	$A14 + 13^*A15 + 4^*A112$ 1,4,7,10,13,16-hexaoxacyclooctadecane 34	0	108.9	34	30.0 [120]
$\text{C}_{12}\text{H}_{25}\text{NO}_3$	387.6	$A14 + 15^*A15 + 6^*A112$ N-decylglycine 42.2	0	108.9	42.2	46.3 [249]
$\text{C}_{26}\text{H}_{26}$	263.6	$A19 + 9^*A2^*B2 + A44 + A36^*B36 + A2$ dodecane 36.82	0	139.75	36.82	33.9 [216]
$\text{C}_{12}\text{H}_{26}\text{O}$	300.2	$2^*A1 + 10^*A2^*B2$ 1-dodecanol 40.17	0	133.76	40.17	36.6 [217]
$\text{C}_{12}\text{H}_{26}\text{O}_3$	297.2	$11^*A2^*B2 + A1 + A30$ 3-(<i>n</i> -nonyloxy)-1,2-propanediol 29.5	0	99.26	29.5	40.3 [217]
$\text{C}_{12}\text{H}_{27}\text{ClSn}$	260.2	$A1 + 8^*A2^*B2 + A32 + 2^*A30^*C30 + A3^*B3 + 2^*A2$ tri- <i>n</i> -butyltin chloride 11.43	0	43.93	11.43	28.3 [130]
$\text{C}_{12}\text{H}_{30}\text{O}_3\text{Si}_3$	160 242.3 280.2	$3^*A1 + 9^*A2 + A22^*B22 + A110$ 1,1,3,3,5,5-hexaethylcyclotrisiloxane 0.46 11.82 11.42	2.89 48.8 40.77	92.46	23.71	25.9 [227]
$\text{C}_{12}\text{H}_{30}\text{Si}_6$	352.4 528.8	$6^*A1 + 6^*A2 + 3^*A112 + 3^*A139 + A14 + 3^*A15$ cyclododecamethylhexasilane 16.7 4.2	47.39 7.94	55.33	20.9	25.0 [175]
$\text{C}_{13}\text{H}_5\text{N}_3\text{O}_7$	430.2 449.2	$A14 + 3^*A15 + 6^*A139 + 12^*A1$ 2,4,7-trinitrofluoren-9-one 2.9 23.5	6.74 52.32	59.06	26.4	22.8 [198]
$\text{C}_{13}\text{H}_6\text{Cl}_6\text{O}_2$	437.6	$A14 + 2^*A15 + 2^*A19 + 2^*A19 + 4^*A10 + 3^*A12 + 3^*A50 + A114$ 2,2'-methylenebis(3,4,6-trichlorophenol) 33.26	0	76.01	33.26	35.3 [215]
$\text{C}_{13}\text{H}_7\text{F}_3\text{N}_2\text{O}_5$	364.6	$2^*A10 + 8^*A12 + 2^*A11 + 6^*A22^*G22 + A2 + 2^*A31$ 2-nitro-1-(4-nitrophenoxy)-4-(trifluoromethyl)benzene 18.44	0	50.58	18.44	25.3 [215]
$\text{C}_{13}\text{H}_8\text{Br}_3\text{NO}_2$		$7^*A10 + 4^*A12 + A11 + A4^*B4 + 3^*A25 + 2^*A50 + A32$ 3,5-dibromo-N-(4-bromophenyl)-2-hydroxybenzamide				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{fus}S_{tpcc}$ (expt)	$\Delta_0^{fus}S_{tpcc}$ (calcd)	$\Delta_0^{fus}H_{tpcc}$ (expt)	$\Delta_0^{fus}H_{tpcc}$ (calcd)
497.7	28.67	0	57.6	74.0	28.67	36.8
$C_{13}H_8Cl_2O$	6*A10+6*A12+3*A21+A31+A60 <i>p, p'</i> -dichlorobenzophenone					[221]
420	30.12	0	71.71	66.3	30.12	27.9
$C_{13}H_8O$	2*A22*C22+A35+8*A10+4*A12 xanthene					[215]
373.7	19.2	0	51.38	56.0	19.2	20.9
$C_{13}H_8O$	A14+3*A15+2*A19+8*A10+A112+2*A19 9-fluorenone					[215]
356.4	18.12	0	50.84	49.7	18.12	17.7
$C_{13}H_8OS$	8*A10+A14+2*A15+2*A19+A114+2*A19 thioxanthone					[215]
487.9	35.5	0	72.76	56.3	35.5	27.5
$C_{13}H_8O_2$	A14+3*A15+4*A19+A114+A131+8*A10 xanthone					[160]
449.7	26.12	0	58.08	54.6	26.12	24.6
$C_{13}H_9Cl_3N_2O$	A14+3*A15+4*A19+A112+A114+8*A10 benzoic acid, 2,4,6-trichlorophenyl hydrazide					[216]
439.7	32.71	0	74.4	59.2	32.71	26.0
$C_{13}H_9F_3N_2O_2$	7*A10+5*A12+3*A22*D22+A60+A44 2-[3-(trifluoromethyl)phenyl]amino]-3-pyridinecarboxylic acid					[221]
476	38	0	79.83	72.8	38	34.6
$C_{13}H_9N$	3*A25+A4*B4+7*A10+3*A12+A11+A44+A41+A36 acridine					[85]
383.2	18.58	0	48.48	47.7	18.58	18.3
$C_{13}H_9N$	9*A10+2*A12+A41+2*A12 7,8-benzoquinoline					[284]
324.1	14.1	0	43.51	47.7	14.1	15.5
$C_{13}H_9N$	9*A10+4*A12+A41 phenanthridine					[216]
354	0.02	0.06				
379.7	22.83	60.12	60.18	47.7	22.85	18.1
$C_{13}H_9N_2$	9*A10+4*A12+A41 2-phenylbenzimidazole					[216]
572.2	22.18	0	38.75	65.9	22.18	37.7
$C_{13}H_{10}$	9*A10+A118+A121+A14+2*A15+3*A19+A12 fluorene					
387.9	19.58	0	50.48	51.0	19.58	19.8
$C_{13}H_{10}BrCl_2O_2PS$	8*A10+A14+2*A15+4*A19 O-(4-bromo-2,5-dichlorophenyl)O-methyl phenylphosphonothioate					[216]
345.6	31.35	0	90.73	87.2	31.35	30.1
$C_{13}H_{10}Cl_2S$	7*A10+5*A12+A1+2*A22*D22+A21+A81 <i>p</i> -chlorobenzyl <i>p</i> -chlorophenyl sulfide					[221]
343.8	32.22	0	93.71	68.9	32.22	23.7
$C_{13}H_{10}N_2$	8*A10+3*A12+A11+2*A22*D22+A2+A84 diphenylcarbodiimide					[232]
287.4	18.55	0	64.54	52.9	18.55	15.2
$C_{13}H_{10}N_2O$	10*A10+2*A12+2*A42+A9 1,3-diphenylurea					[227]
512	34.6	0	67.58	60.7	34.6	31.1
$C_{13}H_{10}O$	10*A10+2*A12+A66 benzophenone					[215]
321.0	18.19	0	56.67	63.8	18.19	20.5
$C_{13}H_{10}S$	10*A10+2*A12+A35 thioxanthone					[80]
401.8	26.1	0	64.96	57.6	26.1	23.2
$C_{13}H_{11}N$	A14+3*A15+2*A19+2*A19+A131+8*A10 N-methylcarbazole					[215]
362.5	17.15	0	47.32	49.3	17.15	17.9
$C_{13}H_{11}NO$	A14+2*A15+2*A19+2*A19+A119+A1+8*A10 benzanilide					[216]
436.5	29.61	0	67.84	60.6	29.61	26.5
$C_{13}H_{12}$	10*A10+2*A12+A60 diphenylmethane					[216]
298.3	18.58	0	62.34	62.1	18.58	18.5
	10*A10+A2+2*A11					[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{fus} S_{pcc}$ (expt)	$\Delta_0^{fus} S_{pcc}$ (calcd)	$\Delta_0^{fus} H_{pcc}$ (expt)	$\Delta_0^{fus} H_{pcc}$ (calcd)
$C_{13}H_{12}NO$	1,3-diphenylurea 34.62	0	67.6	60.7	34.62	31.1 [216]
$C_{13}H_{12}O$	10*A10+2*A12+A66 diphenylcarbinol 23	0	67.93	46.9	23	15.9 [216]
$C_{13}H_{13}BrS$	10*A10+2*A11+A3*B3+A30 2-n-propyl-5-(4-bromophenyl)thiophene 15.7	0	43.56	58.7	15.7	21.2 [251]
$C_{13}H_{13}N$	A14+2*A15+A131+2*A19+2*A19+A1+2+4*A10+2*A12+A21 N-benzylaniline 16.76	0	54.84	85.6	16.76	26.2 [215]
$C_{13}H_{13}NO_2$	10*A10+A12+A11+A45+A2 (dl) 2-(1-naphthoxy)propionamide 37.66	0	84.62	69.8	37.66	31.1 [273]
$C_{17}H_{13}NO_2$	7*A10+3*A12+A32+A3*B3+A1+A61 (d) 2-(1-naphthoxy)propionamide 38.07	0	80.16	69.8	38.07	33.2 [273]
$C_{13}H_{14}N_2$	7*A10+2*A12+A32+A3*B3+A1+A61+A12 bis-(4-aminophenyl)methane 9.23	0	25.36	75.0	9.23	27.3 [216]
$C_{13}H_{15}N$	8*A10+2*A11+2*A12+2*A45+A2 1,2,3,4-tetrahydro-9-methylcarbazole 0.08	0.5	45.8	39.2	14.75	12.7 [15]
$C_{13}H_{15}NO_2$	162 323.8 2*A14+3*A15+2*A19+2*A19+4*A10+A119 3,4-dihydro-6-methyl-2H-pyran-5-carboxanilide 19.21	0	50.4	69.8	19.21	26.6 [221]
$C_{13}H_{15}N_3O_2$	A14+3*A15+A112+2*A19+A60+5*A10+A12+A1 3-methyl-1-phenyl-1H-pyrazol-5-yl dimethylcarbamate 21.39	0	65.96	65.9	21.39	21.4 [221]
$C_{13}H_{16}F_3N_3O_4$	A14+2*A15+3*A1+5*A10+A12+A68+3*A19+A12+A118 2,6-dinitro-N,N-dipropyl-4-(trifluoromethyl)benzenamine 22.32	0	69.45	76.6	22.32	24.6 [215]
$C_{13}H_{16}F_3N_3O_4$	2*A10+A11+3*A12+2*A50+A43+3*A25 +A4*B4+2*A1+4*A2 N-butyl-N-ethyl-2,6-dinitro-4-trifluoromethylaniline 36.5	0	107.83	76.6	36.5	25.9 [215]
$C_{13}H_{18}$	2*A10+A11+3*A12+2*A50+A43+3*A25+A4*B4+2*A1+4*A2 1,1,4,6-tetramethylindane 15.74	0	57.53	47.6	15.74	13.0 [215]
$C_{13}H_{18}$	4*A1+2*A10+2*A11+A14+2*A15+A17+2*A19 1,1,4,7-tetramethylindane 11.28	0	45.93	47.6	11.28	11.7 [215]
$C_{13}H_{18}ClNO$	4*A1+2*A10+2*A11+A14+2*A15+A17+2*A19 N-(4-chlorophenyl)-2,2-dimethylpentanamide 23.31	0	64.71	76.5	23.31	27.6 [221]
$C_{13}H_{18}ClNO$	3*A1+2*A2+A4*B4+4*A10+2*A12+A22*B22+A60 N-(3-chloro-4-methylphenyl)-2-methylpentanamide 16.35	0	46.28	72.6	16.35	25.6 [221]
$C_{13}H_{18}N_2O_2$	3*A1+2*A2+2*A12+A11+A3*B3+3*A10+A60+A22*B22 3-cyclohexyl-6,7-dihydro-1H-cyclopentapyrimidine-2,4-(3H,5H)-dione 42.31	0	72.41	64.0	42.31	37.4 [221]
$C_{13}H_{18}O_5S$	3*A14+6*A15+A16+A124+A125+2*A19 dl-2-ethoxy-2,3-dihydro-3,3-dimethyl-5-benzofuranyl methanesulfonate 26.25	0	76.28	68.3	26.25	23.5 [221]
$C_{13}H_{19}NO_2$	A14+2*A15+A19+A17+A16+A122+4*A1+A2+3*A10+A12+A89 hexyl N-phenylcarbamate 32.76	0	100	93.4	32.76	30.7 [102]
$C_{13}H_{19}N_3O_4$	5*A10+A12+A1+5*A2+A69 N-(1-ethylpropyl)-2,6-dinitro-3,4-xylidine 25.19	0	76.92	70.7	25.19	23.1 [215]
$C_{13}H_{21}N_2O$	4*A1+2*A2+A3*B3+A44+2*A50+3*A12+2*A11+A10 N,N-dimethyl-N'-(octahydro-4,7-methano-1H-inden-5-yl)urea 21.74	0	49.81	65.5	21.74	28.6 [221]
$C_{13}H_{22}$	3*A14+A15+5*A16+2*A1+A64 1,3,5-trimethyladamantane 6.3	27.61				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$C_{13}H_{22}O_3$	253.6	1.73	6.82	34.43	38.2	8.03	9.7 [146]
		3*A14+A15+A16+3*A17+3*A1					
$C_{13}H_{24}N_6$	396.2	20.5	0	51.75	62.8	20.5	24.9 [109]
		A14+7*A15+4*A1+2*A17+A117					
$C_{13}H_{24}O_2$	335.8	16.32	0	48.6	65.0	16.32	21.8 [242]
		A14*4*A15+A119+4*A1+2*A43+3*A12+3*A41					
$C_{13}H_{24}O_4$	290.6	18.16	62.37				
	300.4	9.08	30.21	92.55	77.2	27.24	23.2 [216]
$C_{13}H_{26}$	387.5	45.3	0	116.9	134.9	45.3	52.3 [216]
		11*A2*B2+2*A36*B36					
$C_{13}H_{26}O_2$	285.6	0.9	3.15				
	297.6	7.4	24.87	28.02	70.4	8.3	20.9 [181]
$C_{13}H_{26}O_2$	232.8	22.22	0	95.43	90.1	22.22	21.0 [216]
		A14+A16+A1+6*A2+3*A15					
$C_{13}H_{26}O_2Si_3$	307.1	8.72	28.41				
	315.0	33.74	107.11	135.52	133.6	42.47	42.1 [216]
$C_{13}H_{26}O_2Si_3$	226.8	18.29	0	80.64	80.8	18.29	18.3 [216]
		11*A2*B2+A1+A36					
$C_{13}H_{27}NO_2$	343.2	53.2	0	155.01	125.5	53.2	43.1 [217]
		7*A1+5*A10+A12+2*A32+3*A109					
$C_{13}H_{28}$	255	7.66	30.04				
	267.8	28.49	106.27	136.31	137.8	36.15	36.9 [216]
$C_{13}H_{28}O$	302	7.2	23.84				
	390	3.43	8.379	32.64	32.6	10.63	12.7 [216]
$C_{13}H_{28}O$	304.6	45.1		148.11	131.3	45.02	40.3 [224]
	304.5	41.42	138.9				
$C_{13}H_{28}O_2S$	303.5	23.3	76.99				
	301.6	3.6	12.13				
$C_{13}H_{28}O_2S$	305.8	22.09	72.38				
	306.6	18.74	61.09	148.11	131.3	45.02	40.3 [224]
$C_{13}H_{28}O_2S$	291.9	17.3	59.27				
	311.9	17.3	55.47	114.73	142.3	34.6	44.4 [217]
$C_{13}H_{28}O_2$	311	38.9	0	125.08	144.9	38.9	45.1 [217]
		A1+9*A2*B2+A32+2*A30*C30+A3*B3+2*A2					
$C_{13}H_{29}NO_2$	346.6	54.8	0	158.11	134.9	54.8	46.8 [217]
		A1+9*A2*B2+A44+2*A30*C30+2*A2+A3*B3					
$C_{14}H_7ClF_3NO_5$	436.6	37.67	0	86.27	83.0	37.67	36.2 [221]
		5-[2-chloro-4-(trifluoromethyl)phenoxy]-2-nitrobenzoic acid					
$C_{14}H_7ClO_2$	483.0	39	0	80.74	53.3	39	25.7 [216]
		A14+3*A15+4*A19+2*A114+A22*C22+A12+7*A10					
$C_{14}H_8Cl_4$	349.8	23.84	0	68.17	72.7	23.84	25.4 [221]
		8*A10+4*A12+A7+A7+4*A22*D22					
$C_{14}H_8Cl_4$	360.4	23.55	0	65.33	72.7	23.55	26.2 [221]
		1,1-dichloro-2,2-bis(4-chlorophenyl)ethylene					
		8*A10+4*A12+A7+A7+4*A22*D22					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (calcd)
$C_{14}H_8O_2$	555	anthraquinone 32.57	0	58.7	32.57	29.0
		$3*A15+A14+8*A10+4*A19+2*A114$				[216]
$C_{14}H_9ClF_2N_2O_2$	499.5	N-[(4-chlorophenylamino)carbonyl]-2,6-difluorobenzamide 55.99	0	112.08	55.99	33.4
		$7*A10+5*A12+2*A24+A22*E22+2*A60$				[221]
$C_{14}H_9Cl_2NO_3$	358.3	methyl 5-(2,4-dichlorophenoxy)-2-nitrobenzoate 26.31	0	73.44	26.31	28.5
		$6*A10+6*A12+2*A22*E22+A50+A32+A38+A1$				[215]
$C_{14}H_9Cl_3$	337.9	1-chloro-2,2-(bis-(4-chlorophenyl)ethylene) 25.52	0	75.53	25.52	24.0
		$8*A10+4*A12+A6*B6+A7+3*A22*C22$				[232]
$C_{14}H_9Cl_5$	382.1	1,1-(2,2,2-trichloroethylidene)bis(4-chlorobenzene) 26.28	0	68.78	26.28	25.5
		$5*A22*E22+A4*B4+2*A12+2*A11+8*A10+A3$				[215]
$C_{14}H_9Cl_5$	345.8	1-chloro-2-(2,2,2-trichloro-1-(4-chlorophenyl)ethyl)benzene 23.09	0	66.78	23.09	23.1
		$8*A10+2*A12+2*A11+5*A22*E22+A3+A4*B4$				[221]
$C_{14}H_9Cl_5O$	396.3	2-chloro- α -(4-chlorophenyl)- α -(trichloromethyl)benzenemethanol 25.2	0	63.61	25.2	32.9
		$8*A10+2*A11+2*A12+2*A4*B4+5*A22*F22+A30*F30$				[215]
$C_{14}H_9Cl_5O$	347.2	4-chloro- α -(4-chlorophenyl)- α -(trichloromethyl)benzenemethanol 19.56	0	56.35	19.56	28.9
		$8*A10+2*A11+2*A12+2*A4*B4+5*A22*F22+A30*F30$				[221]
$C_{14}H_9NO_4PS$	308.2	O-ethyl O-(4-nitrophenyl)phenylphosphonothioate 25.05	0	81.29	25.05	28.3
		$9*A10+3*A12+A50+A1+A2+A81$				[221]
$C_{14}H_9NO_2$	524.2	1-aminoanthraquinone 28.78	0	54.9	28.78	30.7
		$A14+3*A15+2*A114+4*A19+7*A10+A45+A12$				[13]
$C_{14}H_{10}$	347.5	phenanthrene 0.22	0.63			
	372.4	16.46	44.21	44.83	16.68	16.5
		$10*A10+4*A12$				[216]
$C_{14}H_{10}$	488.9	anthracene 29.37	0	60.08	29.37	21.6
		$10*A10+4*A12$				[216]
$C_{14}H_{10}$	334	diphenylacetylene 20.5	0	61.4	20.5	17.9
		$10*A10+2*A9+2*A12$				
$C_{14}H_{10}Cl_2O_2$	440.2	bis(4-chlorophenyl)acetic acid 31.66	0	71.92	31.66	34.2
		$8*A10+2*A12+2*A11+A3*B3+2*A22*C22+A36*C36$				[215]
$C_{14}H_{10}Cl_4$	382.1	1,1'-(2,2-dichloroethylidene)bis(4-chlorobenzene) 27.31	0	71.48	27.31	24.4
		$8*A10+2*A11+2*A12+A3*B3+4*A22*D22$				[215]
$C_{14}H_{10}N_2O_2$	484.2	1,4-diaminoanthraquinone 24.2	0	49.98	24.2	31.5
		$A14*3*A15+2*A114+4*A19+6*A10+2*A45+2*A12$				[13]
$C_{14}H_{10}O$	429	anthrone (some decomposition upon melting) 26.8	0	62.47	26.8	22.9
		$A14+3*A15+2*A19+2*A19+8*A10+A114$				[82]
$C_{14}H_{10}O_2$	84	benzil 0.04	0.5			
	368	23.56	64.02	64.52	23.6	25.2
		$10*A10+2*A12+2*A35$				[216]
$C_{14}H_{10}O_3$	313.2	benzoic anhydride 17.15	0	54.77	17.15	21.7
		$10*A10+2*A12+A39$				[287]
$C_{14}H_{11}Cl_2NO_2$	438.2	3-(3,5-dichlorophenyl)-1,5-dimethyl-3-azabicyclo[3.1.0]hexanedione 30.09	0	68.67	30.09	29.0
		$2*A14+A128+2*A17+3*A12+2*A22*C22+2*A1$				[221]
$C_{14}H_{11}NO_3$	464	N-salicylidene- <i>m</i> -aminobenzoic acid 33.11	0	71.36	33.11	36.2
		$8*A10+4*A12+A36*C36+A31+A42+A6*B6$				[216]
$C_{14}H_{12}$		9,10-dihydrophenanthrene				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (calcd)
306.5	12.8	0	41.77	54.7	12.8	16.8
$C_{14}H_{12}$	8*A10+A14+3*A15+4*A19 <i>trans</i> -stilbene					[216]
398.2	27.4	0	68.81	69.7	27.4	27.8
$C_{14}H_{12}$	10*A10+2*A12+2*A6 9-methylfluorene					[215]
319.2	16.32	0	51.13	53.9	16.32	17.2
$C_{14}H_{12}F_3NO_4S_2$	A14+2*A15+4*A19+A16+A1+8*A10 1,1,1-trifluoro- <i>n</i> -[2-methyl-4-(phenylsulphonyl)phenyl]methane sulfonamide					[252]
418.4	31.79	0	75.97	68.7	31.79	36.0
$C_{14}H_{12}O_2$	8*A10+A11+3*A12+A88+A95+A4*B4+3*A25+A1 diphenylacetic acid					[221]
420.4	31.25	0	74.34	58.7	31.25	24.7
$C_{14}H_{12}O_2$	10*A10+2*A11+A3*B3+A36 benzyl benzoate					[216]
293.1	20.44	0	69.76	71.9	20.44	21.1
$C_{14}H_{12}O_4$	10*A10+A11+A12+A2+A38 1,2-dicarbomethoxynaphthalene					[221]
358.2	27.6	0	77.05	65.1	27.6	23.3
$C_{14}H_{12}O_4$	6*A10+4*A12+2*A1+2*A38 1,3-dicarbomethoxynaphthalene					[217]
378.7	30.5	0	80.54	65.1	30.5	24.64
$C_{14}H_{12}O_4$	6*A10+2*A12+2*A1+2*A38+2*A12 1,4-dicarbomethoxynaphthalene					[217]
340.2	20.4	0	59.96	65.1	20.4	22.13
$C_{14}H_{12}O_4$	6*A10+2*A12+2*A1+2*A38+2*A12 1,5-dicarbomethoxynaphthalene					[217]
392	26.4	0	67.35	65.1	26.4	25.5
$C_{14}H_{12}O_4$	6*A10+2*A12+2*A1+2*A38+2*A12 1,6-dicarbomethoxynaphthalene					[217]
371.8	22.1	0	59.44	65.1	22.1	24.2
$C_{14}H_{12}O_4$	6*A10+2*A12+2*A1+2*A38+2*A12 1,7-dicarbomethoxynaphthalene					[217]
363.2	20	0	55.07	65.1	20	23.6
$C_{14}H_{12}O_4$	6*A10+2*A12+2*A1+2*A38+2*A12 2,3-dicarbomethoxynaphthalene					[217]
324.2	20.2	0	62.31	65.1	20.2	21.1
$C_{14}H_{12}O_4$	6*A10+2*A12+2*A1+2*A38+2*A12 2,7-dicarbomethoxynaphthalene					[217]
410.2	26.6	0	64.85	65.1	26.6	26.7
$C_{14}H_{14}$	6*A10+2*A12+2*A1+2*A38+2*A12 1,2,3,4-tetrahydrophenanthrene					[217]
302.6	11.17	36.91				
285	5.83	20.44				
298	1.77	5.92	63.28	49.5	18.76	14.7
$C_{14}H_{14}$	A14+3*A15+2*A19+2*A12+6*A10 phenyl- <i>o</i> -tolylmethane					[31]
279.8	19.24	0	68.78	62.6	19.24	17.5
$C_{14}H_{14}^*$	9*A10+3*A11+A1+A2 2,2'-dimethylbiphenyl					[216]
293.1	2.28	0	7.78	0	2.28	0
$C_{14}H_{14}$	Prediction not made					[216]
273.2	1,2-diphenylethane					
324.3	2.25	8.23				
	22.73	70.09	78.32	69.2	24.98	22.4
$C_{14}H_{14}^*$	10*A10+2*A11+2*A2 2-ethylbiphenyl					[216]
267.1	2.07	0	7.74	0	2.07	0
$C_{14}H_{14}$	Prediction not made					[216]
373.3	1,2,3,4-tetrahydroanthracene					
388	19.16	51.33				
	2.92	7.53	58.85	49.5	22.08	19.2
$C_{14}H_{14}Cl_2N_2O$	A14+3*A15+2*A19+6*A10+2*A12 1-[2-(2,4-dichlorophenyl)-2-(propenyloxy)ethyl]-1H-imidazole					[216]
322.6	30.5	0	94.55	74.5	30.5	24.0
	A14+2*A15+3*A18*B18+A119+A118+2*A2+A3*B3+A5+A6 +3*A10+2*A12+A11+2*A22*E22+A32					[221]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{pcc}$ (expt)	$\Delta_0^{T_{fus}} H_{pcc}$ (calcd)
$C_{14}H_{14}NO_4PS$	O-ethyl O-(4-nitrophenyl)phenylphosphonothioate					
308.2	25.05	0	81.28	74.2	25.05	22.9
	$9*A10+3*A12+A1+A2+A81$					[221]
$C_{14}H_{14}O_2$	(dl) 1,2-diphenyl-1,2-dihydroxyethane					
393	31.38	0	79.85	71.9	31.38	28.2
	$10*A10+2*A11+2*A3*B3+2*A30*B30$					[273]
$C_{14}H_{14}O_2$	(d) 1,2-diphenyl-1,2-dihydroxyethane					
420.5	34.31	0	81.59	71.9	34.31	30.2
	$10*A10+2*A11+2*A3*B3+2*A30*B30$					[273]
$C_{14}H_{14}O_3$	2-(6-methoxy-2-naphthyl)propionic acid					
439.2	29.41	0	66.96	58.6	29.41	25.7
	$6*A10+3*A12+A11+2*A1+A3*B3+A36*B36+A32$					[33]
$C_{14}H_{14}O_3$	2-pivaloylindan-1,3-dione					
381.5	25.99	0	68.12	62.9	25.99	24.0
	$A14+2*A15+2*A19+2*A114+4*A10+3*A1+A4*B4+A35+A16$					[215]
$C_{14}H_{15}N_3$	N,N-dimethyl-4-phenylazoaniline					
389.2	23.08	0	59.3	53.7	23.08	20.9
	$2*A1+A43+9*A10+3*A12+2*A42$					[13]
$C_{14}H_{16}$	heptacyclo[6.6.0[2,6].0[3,13].0[4,11].0[5,9].0[8,1].0[10,14]]tetradecane					
355	14.67	41.32				
440	5.57	12.66	53.98	31.0	20.24	13.6
	$7*A14-7*A15+12*A16$					[127]
$C_{14}H_{16}ClN_3O_2$	1-(4-chlorophenoxy)-3,3-dimethyl-(1H,1,2,4-triazol-1-yl)-2-butanone					
351.4	22.87	0	65.06	76.6	22.87	26.9
	$4*A10+2*A12+3*A1+A4*B4+A3*B3+A14+2*A15+2*A18*B18$					[221]
	$+2*A118+A22*F22+A119+A35+A32$					
$C_{14}H_{16}F_3N_3O_4$	N-(cyclopropylmethyl)-2,6-dinitro- <i>n</i> -propyl-4-(trifluoromethyl)benzenamine					
305.8	22.51	0	73.61	70.9	22.51	21.7
	$3*A12+2*A10+A11+A4*B4+3*A25+2*A50+A43+A1+3*A2+A14+A16$					[221]
$C_{14}H_{16}O_8$	1,2,3,4-tetracarboxymethoxybenzene					
404.7	40.4	0	99.79	85.9	40.4	34.8
	$4*A1+4*A38+4*A12+2*A10$					[217]
$C_{14}H_{16}O_8$	1,2,3,5-tetracarboxymethoxybenzene					
389.2	32.6	0	83.89	85.9	32.6	33.4
	$4*A1+4*A38+4*A12+2*A10$					[217]
$C_{14}H_{16}O_8$	1,2,4,5-tetracarboxymethoxybenzene					
416.7	35.7	0	85.4	85.9	35.7	35.8
	$4*A1+4*A38+4*A12+2*A10$					[217]
$C_{14}H_{17}ClNPO_4S_2$	S-[2-chloro-1-(1,3-dihydro-1,3-dioxo-2H-isoindol-2-yl)ethyl]O,O-diethylphosphorodithioate					
340.0	25.27	0	74.33	101.1	25.27	34.4
	$A14+2*A15+2*A19+4*A10+A128+4*A2+2*A1+A80$					[221]
$C_{14}H_{17}NO_2$	4-methyl-7-diethylaminocoumarin					
343.8	17.88	0	52.02	67.2	17.88	23.1
	$A14+3*A15+2*A19+A19+A18*B18+A115+3*A10+A12+3*A1+2*A2+A43$					[216]
$C_{14}H_{18}$	1,2,3,4,5,6,7,8-octahydroanthracene					
331.4	2.51	7.59				
345.4	18.34	53.1	60.69	54.7	20.86	18.9
	$2*A14+6*A15+4*A19+2*A10$					[216]
$C_{14}H_{18}ClN_3O_2$	β (4-chlorophenoxy)- α -(1,1-dimethylethyl)-1H-1,2,4-triazole-1-ethanol					
377.8	24.47	0	64.77	73.0	24.47	27.6
	$A14+2*A15+2*A118+A119+2*A18*B18+2*A3*B3$					[221]
	$+A32+A22*F22+4*A10+2*A12+3*A1+A4+A30*F30$					
$C_{14}H_{19}Cl_2NO_2$	4[<i>p</i>]-[bis(2-chloroethyl)amino]benzene]butanoic acid					
338.9	29.18	0	86.1	102.7	29.18	34.8
	$3*A2+4*A2+4*A10+A11+A12+A36*D36+2*A22*D22+A4$					[221]
$C_{14}H_{19}NO$	2-(dimethylamino)-1,2-diphenylethanone					
334.2	22.38	0	66.97	64.7	22.38	21.6
	$10*A10+A11+A12+A35*B35+A43+2*A1+A3*B3$					[253]
$C_{14}H_{20}$	diamantane					
407.2	4.44	10.89				
440.4	8.95	20.33				
517.9	8.66	16.72	47.95	45.4	22.05	23.5
	$5*A14-A15+8*A16$					[216]
$C_{14}H_{20}$	1,8-cyclotetradecadiyne					
370	22.6	0	61.06	55.3	22.6	20.4
	$A14+11*A15+4*A20$					[108]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{14}H_{20}O$	1-diamantanol					
	408	4.9	12.01			
	395	18	45.57			
	573	9.6	16.75	74.33	27.3	32.5
$C_{14}H_{20}O$	5*A14-A15+7*A16+A17+A30					[144]
	4-diamantanol					
	448	9.77	21.81			
	484	16.4	33.88	55.69	27.3	26.17
$C_{14}H_{20}ClNO_2$	5*A14-A15+7*A16+A17+A30					[144]
	2-chloro-N-(2,6-diethylphenyl)-N-(methoxymethyl)acetamide					
	315.9	25.31	0	80.13	86.5	25.31
						27.3
$C_{14}H_{20}N_3O_5PS$	3*A1+4*A2+3*A10+2*A11+A12+A32+A22*C22+A59					[221]
	O-6-ethoxycarbonyl-5-methylpyrazolo[1,5-a]pyrimidin-2-yl O,O-diethyl phosphorothioate					
	324.4	27.32	0	84.22	97.6	27.32
						31.66
$C_{14}H_{21}N_3O_4$	A14+2*A15+A18+2*A19+A118+A119+A10+A11+A12+A41+A38+4*A1+3*A2+A79					[221]
	4-(1,1-dimethylethyl)-n-(1-methylpropyl)-2,6-dinitrobenzeneamine					
	338.8	20.84	0	61.52	63.5	20.84
						21.5
$C_{14}H_{22}$	2*A10+A11+3*A12+5*A1+A4+A2+A3*B3+2*A50+A44					[221]
	n-octylbenzene					
	234.2	29.96	0	127.91	110.4	29.96
						25.9
$C_{14}H_{22}N_4O_2$	A1+7*A2*B2+5*A10+A11					[216]
	8-heptyltheophylline					
	472.7	33	0	69.81	95.9	33
						45.3
$C_{14}H_{22}N_4O_6S$	2*A14+3*A15+2*A125+A118+A121+3*A1+3*A19+6*A2					[216]
	4-(dipropylamino)-N,N-dimethyl-3,5-dinitrobenzenesulfonamide					
	413.6	32.57	0	78.75	85.7	32.57
						35.4
$C_{14}H_{22}O$	4*A1+4*A2+A43+2*A50+A94+2*A10+4*A12					[221]
	2,6-di-tert-butylphenol					
	310.7	16.57	0	53.33	51.6	16.57
						16.0
$C_{14}H_{24}$	6*A1+2*A4+2*A11+A12+3*A10+A31					[101]
	1,3,5,7-tetramethyladamantane					
	183.3	0.23	1.25			
	337.2	9.82	29.12	30.38	35.9	10.05
$C_{14}H_{24}$	3*A14+A15+4*A17+4*A1					[146]
	cis-anti-trans-perhydrophenanthrene					
	313	11.16	0	35.64	59.7	11.16
						18.7
$C_{14}H_{24}$	3*A14+5*A15+4*A16					[216]
	cis-syn-trans-perhydrophenanthrene					
	273	10.48	0	38.39	59.7	10.48
						16.3
$C_{14}H_{24}$	3*A14+5*A15+4*A16					[216]
	trans-anti-trans-perhydrophenanthrene					
	283	11.83	0	41.81	59.7	11.83
						16.9
$C_{14}H_{24}O_2$	3*A14+5*A15+4*A16					[216]
	1,8-cyclotetradecanedione					
	417.2	27.53	0	65.99	71.4	27.53
						29.8
$C_{14}H_{24}NO_4PS_3$	A14+11*A15+2*A114					[114]
	O,O-diisopropyl S-2-phenylsulfonlaminoethyl phosphorodithioate					
	310.4	30.61	0	98.63	91.2	30.61
						28.3
$C_{14}H_{24}O_4$	5*A10+A12+2*A2+4*A1+2*A3*B3+A95+A80					[221]
	1,6-cyclodecanedione bis ethylene ketal					
	450.2	32.68	0	72.58	69.2	32.68
						31.2
$C_{14}H_{26}O$	3*A14+9*A15+4*A112+2*A17					[114]
	4,4,8,8-tetramethylcyclodecanone					
	378.2	16.32	0	43.15	59.1	16.32
						22.4
$C_{14}H_{26}O_2$	A14+7*A15+4*A1+2*A17+A114					[111]
	decyl methacrylate					
	250.7	30.55	0	121.85	133.4	30.55
						33.5
$C_{14}H_{28}$	2*A1+9*A2*B2+A5+A7+A38					[216]
	cyclotetradecane					
	328	28.7	0	87.51	74.1	28.7
						24.3
$C_{14}H_{28}O$	11*A15+A14					[119]
	2-tetradecanone					
	306.7	49.12	0	160.16	142.4	49.12
						43.7
$C_{14}H_{28}O_2$	2*A1+A35+11*A2*B2					[216]
	ethyl dodecanoate					
	271.5	9.31	0	34.3	0	9.31
	Prediction not made					[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{pcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{pcc}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{pcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{pcc}}}$ (calcd)
$C_{14}H_{28}O_2$	tetradecanoic acid					
	45.1	0	137.92	142.9	45.1	46.7
$C_{14}H_{28}O_4$	$12*A2*B2+A1+A36$					[216]
	2,2,9,9-tetramethyl-1,3,8,10-tetraoxacyclotetradecane					
$C_{14}H_{29}NO_3$	30.5	0	74.5	80.1	30.5	32.8
	$A14+11*A15+4*A112+4*A1+2*A17$					[27]
$C_{14}H_{29}NO_3$	N-dodecylglycine					
	48.4	0	123.12	138.2	48.4	54.3
$C_{14}H_{29}NO_3$	$A1+11*A2*B2+A2+A36*B36+A44$					[249]
	N-octyl-L-leucine					
$C_{14}H_{29}NO_3$	7.6	21.28				
	29.3	73.6	94.88	110.0	36.9	43.8
$C_{14}H_{29}NO_3$	$3*A1+7*A2*B2+A3+A3*B3+A36*B36+A44+A2$					[249]
	N-octyl-DL-leucine					
$C_{14}H_{29}NO_3$	6.8	19.23				
	27.2	74.09	93.33	110.0	34.0	40.4
$C_{14}H_{30}$	$3*A1+7*A2*B2+A3+A3*B3+A36*B36+A44+A2$					[249]
	tetradecane					
$C_{14}H_{30}$	45.07	0	161.54	147.1	45.07	41.1
	$2*A1+12*A2*B2$					[216]
$C_{14}H_{30}O$	1-tetradecanol					
	47.01	151.04				
$C_{14}H_{30}O$	25.1	80.75				
	1.8	5.86				
$C_{14}H_{30}O$	306	76.57				
	311	22.01	70.71			
$C_{14}H_{30}O$	311.6	49.37	158.75	140.6	49.37	43.7
	311					[224]
$C_{14}H_{30}O_2S$	$A1+13*A2*B2+A30$					
	3(<i>n</i> -undecylthio)-1,2-propanediol					
$C_{14}H_{30}O_2S$	2.5	8.92				
	4.9	16.95				
$C_{14}H_{30}O_2S$	280.2	4.6	15.58			
	289.1	18.3	57.66	99.11	151.6	30.3
$C_{14}H_{30}O_2S$	295.2					48.1
	317.4					[217]
$C_{14}H_{30}O_3$	$A1+10*A2*B2+A84+2*A30*C30+2*A2+A3*B3$					
	3(<i>n</i> -undecyloxy)-1,2-propanediol					
$C_{14}H_{30}O_3$	43.1	0	138.27	154.2	43.1	48.1
	$A1+10*A2*B2+A32+2*A30*C30+2*A2+A3*B3$					[217]
$C_{14}H_{31}NO_2$	3(<i>n</i> -undecylamino)-1,2-propanediol					
	58.2	0	166.86	144.2	58.2	50.3
$C_{14}H_{31}NO_2$	$A1+10*A2*B2+A44+2*A30*C30+2*A2+A3*B3$					[217]
	4,4'-diphenylmethane diisocyanate					
$C_{15}H_{10}N_2O_2$	27.3	0	87.06	78.5	27.3	24.6
	$8*A10+2*A12+2*A11+2*A58+A2$					[216, 104]
$C_{15}H_{11}ClF_3NO_4$	2-chloro-1-(3-ethoxy-4-nitrophenoxy)-4-(trifluoromethyl)benzene					
	30.07	0	83.8	82.4	30.07	29.6
$C_{15}H_{11}ClN_2O$	$6*A10+A11+5*A12+A1+A2+A4*B4+3*A25+A22*G22+A50+2*A32$					[215]
	7-chloro-1,3-dihydro-5-phenyl-2H-1,4-benzodiazepin-2-one					
$C_{15}H_{11}ClN_2O$	34	0	156.9	69.7	34	15.1
	$A14+4*A15+2*A121+A22*C22+A16+A19+A18+5*A10+A12$					[216]
$C_{15}H_{12}$	4-methylphenanthrene					
	0.02	0.12				
$C_{15}H_{12}$	295	0.11				
	324.9	14.04	43.21	43.44	44.8	14.6
$C_{15}H_{12}N_2O_3$	$A1+9*A10+4*A12+A11$					[216]
	1,4-diamino-2-methoxyanthraquinone					
$C_{15}H_{12}N_2O_3$	35.29	0	68.5	72.4	32.29	37.3
	$A14+3*A15+2*A114+4*A19+5*A10+2*A45+3*A12+A1+A32$					[13]
$C_{15}H_{13}Cl_2NO_2$	1,1-(di- <i>p</i> -chlorophenyl)-2-nitropropane					
	21.39	0	60.38	66.8	21.39	23.7
$C_{15}H_{14}O$	$8*A10+2*A11+2*A12+A3+A3*B3+A1+2*A22*C22+A50$					[221]
	1,3-diphenylacetone					
$C_{15}H_{14}O$	20.2	0	65.77	73.8	20.2	22.7
	$10*A10+2*A11+2*A2+A35$					[217]
$C_{15}H_{15}ClN_2O_2$	3-[4-[4-chlorophenoxy]phenyl]-1,1-dimethylurea					
	34.87	0	81.88	82.9	34.87	35.3
$C_{15}H_{15}N$	$8*A10+4*A12+2*A1+A22*C22+A32*C32+A64*B64$					[215]
	N-isopropylcarbazole					
$C_{15}H_{15}N$	0.64	4.64				
	180	0.38	2.09			

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

	$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
	395.2	17.73	44.86	51.6	57.2	18.75	22.6
$C_{15}H_{15}NO$		A14+2*A15+2*A19+8*A10+2*A1+A3*B3+A119+2*A19					
		N-methyldiphenylacetamide					
	439.8	30.23	0	68.73	64.3	30.23	28.3
$C_{15}H_{16}N_2O_2$		10*A10+2*A11+A3*B3+A1+A60					
		<i>a</i> -cyclopropyl- <i>a</i> -(4-methoxyphenyl)-5-pyrimidinemethanol					
	383.1	26.63	0	69.51	88.0	26.63	33.7
$C_{15}H_{16}O$		A14+7*A10+A1+A4*B4+2*A11+A12+2*A41+A30*D30+A32+A16					
		<i>p</i> - α -cumylphenol					
	346.4	21.68	0	62.58	66.3	21.68	23.0
$C_{15}H_{16}O_2$		8*A10+3*A12+A11+2*A1+A3+A31					
		4,4'-dihydroxydiphenyl-2,2-propane					
	433	30.1	0	69.52	66.0	30.1	28.6
$C_{15}H_{17}Br_2NO_2$		2*A1+A4+8*A10+2*A12+2*A11+2*A31					
		3,5-dibromo-4-hydroxybenzoxazole octanoyl ester					
	318.3	26.49	0	83.23	105.8	26.49	33.7
$C_{15}H_{18}Cl_2N_2O_3$		4*A12+2*A10+2*A21+A56+A38+A1+6*A2					
		3-[2,4-dichloro-5-(1-methylethoxy)phenyl]-5-(1,1-dimethylethyl)-1,3,4-oxadiazol-2(3H)-one					
	360.6	26.39	0	73.19	89.3	26.39	32.2
$C_{15}H_{18}N_2O_6$		A14+2*A15+A19+A126+A118+5*A1					
		+A3*B3+A4+4*A12+2*A10+2*A22*E22+A32					
	341.3	18.89	0	55.37	80.2	18.89	27.4
$C_{15}H_{21}NO$		4*A1+A2+A3+2*A10+3*A12+A11+2*A50+A38+A7+A6*B6					
		2-methyl-1-phenyl-2-(N-piperidinyl)-1-propanone					
	310.2	16.74	0	53.97	71.7	16.74	22.2
$C_{15}H_{21}NO_4$		5*A10+A12+A4*B4+2*A1+A14+3*A15+A119+A35					
		methyl N-(2-methoxyacetyl)-N-(2,6-xylyl)-dl-alaninate					
	345.5	26.46	0	76.58	82.1	26.46	28.4
$C_{15}H_{23}N_3O_2$		5*A1+A2+A3*B3+3*A10+2*A11+A12+A38+A32+A59					
		N-capryl-pyrazinamide					
	360.5	50.58	0	140.31	104.7	50.58	37.7
$C_{15}H_{24}O$		A1+6*A2+3*A10+A12+2*A41+A71					
		2,6-di- <i>tert</i> -butyl-4-methylphenol					
	343.7	23.85	0	69.39	52.2	23.85	17.9
$C_{15}H_{24}O_2$		7*A1+2*A10+3*A11+A12+2*A4+A31					
		2,6-di- <i>tert</i> -butyl-4-methoxyphenol					
	374.4	26.9	0	71.86	59.0	26.9	22.1
$C_{15}H_{28}O_2$		7*A1+2*A4+2*A11+2*A12+2*A10+A31+A32					
		pentadecanolactone					
	283	27.3	96.47				
	308.5	6.99	22.65	119.12	84.6	34.29	26.1
$C_{15}H_{30}$		A14+13*A15+A115					
		cyclopentadecane					
	210.1	8.5	40.46				
	336.6	8.5	25.25	65.71	77.8	17	26.2
$C_{15}H_{30}$		A14+12*A15					
		<i>n</i> -decylcyclopentane					
	251.0	33.14	0	132.01	127.6	33.14	32.0
$C_{15}H_{30}O$		A14+A16+A1+9*A2*B2+2*A15					
		2-pentadecanone					
	312.2	54.57	0	174.8	151.7	54.39	47.4
$C_{15}H_{30}O_2$		2*A1+A35+12*A2*B2					
		pentadecanoic acid					
	318.7	8.12	25.48				
	325.7	41.52	127.49	152.97	152.3	49.64	49.6
$C_{15}H_{30}O_2$		13*A2*B2+A1+A36					
		methyl myristate					
	291.6	50.21	0	172.17	154.8	50.21	45.1
$C_{15}H_{31}NO_3$		2*A1+12*A2*B2+A38					
		N-decyl-L-valine					
	378.1	21.3	56.33				
	380.6	15.4	40.46	96.8	121.5	36.7	46.3
$C_{15}H_{31}NO_3$		3*A1+A3+A3*B3+9*A2*B2+A44+A36*B36					
		N-decyl-DL-valine					
	358.1	63.1	0	176.21	121.5	63.1	43.5

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{pcc}$ (expt)	$\Delta_0^{T_{fus}} H_{pcc}$ (calcd)
$C_{15}H_{31}NO_3$	356.1					[249]
$C_{15}H_{32}$	270.9					
	283.1					
$C_{15}H_{32}O$	316					
	316.6					
$C_{15}H_{32}O_2S$	299					
	325.5					
$C_{15}H_{32}O_3$	323					
$C_{15}H_{33}NO_2$	351.9					
$C_{16}F_{34}$	176.5					
	177.7					
	186.7					
	402.2					
$C_{16}H_{10}$	120.8					
	423.8					
$C_{16}H_{10}$	383.4					
$C_{16}H_{11}F_3O$	356.8					
$C_{16}H_{12}F_2$	301.2					
$C_{16}H_{12}F_2O$	343.4					
$C_{16}H_{13}FO$	354.4					
$C_{16}H_{12}Ge$	320					
$C_{16}H_{12}Si$	316.2					
$C_{16}H_{14}$	319.9					
	385.1					
	412.8					
$C_{16}H_{14}Cl_2O_3$	310.4					
$C_{16}H_{14}Cl_2O_4$	314.4					
$C_{16}H_{14}O_2$	187					
	418.6					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}S_{\text{tpcc}}}$ (calcd)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (expt)	$\Delta_0^{T_{\text{fus}}H_{\text{tpcc}}}$ (calcd)
$C_{16}H_{14}O_6$	1,2,3-tricarbomethoxynaphthalene 23.7	0	65.34	75.4	23.7	27.4 [217]
362.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,2,4-tricarbomethoxynaphthalene 32.1	0	81.53	75.4	32.1	29.7 [217]
393.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,2,5-tricarbomethoxynaphthalene 25.5	0	70.25	75.4	25.5	27.4 [217]
363	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,2,6-tricarbomethoxynaphthalene 35.9	0	86.15	75.4	35.9	31.4 [217]
416.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,2,7-tricarbomethoxynaphthalene 36.1	0	84.5	75.4	36.1	32.2 [217]
427.2	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,2,8-tricarbomethoxynaphthalene 24.8	0	67.63	75.4	24.8	27.7 [217]
366.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,3,5-tricarbomethoxynaphthalene 25.9	0	64.35	75.4	25.9	30.4 [217]
402.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,3,7-tricarbomethoxynaphthalene 37.2	0	83.39	75.4	37.2	33.7 [217]
446.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,3,8-tricarbomethoxynaphthalene 27.7	0	71.46	75.4	27.7	29.3 [217]
388.2	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,4,5-tricarbomethoxynaphthalene 26.5	0	65.77	75.4	26.5	30.3 [217]
402.2	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	1,4,6-tricarbomethoxynaphthalene 30.2	0	73.6	75.4	30.2	30.9 [217]
409.2	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	2,3,5-tricarbomethoxynaphthalene 41	0	101.96	75.4	41	30.3 [217]
401.7	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{14}O_6$	2,3,6-tricarbomethoxynaphthalene 34.4	0	86.27	75.4	34.4	30.1 [217]
399.2	$3*A1+3*A38+2*A12+5*A10+3*A12$					
$C_{16}H_{15}Cl_2NO_2$	1,1-bis(4-chlorophenyl)-2-nitrobutane 15.41	0	46.65	73.8	15.41	24.4 [221]
330.3	$8*A10+2*A12+2*A11+A3+A3*B3+A1+A2+2*A22*C22+A50$					
$C_{16}H_{15}Cl_3O_2$	1-methoxy-2-(2,2,2-trichloro-1(4-methoxyphenyl)ethyl)benzene 22.45	0	64.58	85.6	22.45	29.8 [221]
347.6	$8*A10+2*A12+2*A11+2*A1+A4*B4+2*A32+3*A22*E22+A3*B3$					
$C_{16}H_{15}Cl_3O_2$	1,1'-(2,2,2-trichloroethylidene-bis(4-methoxy)benzene 27.48	76.14	76.21	85.6	27.48	30.9 [221]
360.6	$8*A10+2*A12+2*A11+2*A1+A4*B4+2*A32+3*A22*E22+A3*B3$					
$C_{16}H_{15}N$	4'-propylbiphenyl-4-carbonitrile 22.7	0	67.01	76.8	22.7	26.0 [216]
338.8	$A1+2*A2+8*A10+A11+3*A12+A56$					
$C_{16}H_{16}$	1,2,3,6,7,8-hexahydropyrene 5.02	13.32				
377						
407.7	18.09	44.37	57.69	45.0	23.11	18.4 [18]
	$2*A14+6*A15+6*A19+4*A10$					
$C_{16}H_{16}N_2O_2^*$	anisaldazine 29.75	0	67.31	0	29.75	0 [216]
442	No prediction made					
$C_{16}H_{16}N_2O_4$	ethyl [3-[(phenylamino)carbonyl]oxy]phenylcarbamate 32.75	0	83.09	90.2	32.75	35.6 [221]
394.1	$9*A10+3*A12+2*A69+A1+A2$					
$C_{16}H_{16}N_2O_4$	methyl 3- <i>m</i> -tolylcarbamoyloxphenylcarbamate 39.62	0	93.49	83.6	39.62	35.4 [221]
423.8	$2*A1+3*A12+A11+8*A10+2*A69$					
$C_{16}H_{16}O_2$	(<i>d</i>) 2-(<i>p</i> -methoxyphenyl)propiophenone					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{fus} S_{tpcc}$ (expt)	$\Delta_0^{fus} S_{tpcc}$ (calcd)	$\Delta_0^{fus} H_{tpcc}$ (expt)	$\Delta_0^{fus} H_{tpcc}$ (calcd)
326	21.76	0	66.74	76.8	21.76	25.0
$C_{16}H_{16}O_2$	9*A10+2*A12+A3*B3+2*A1+A32+A11+A35 (dl) 2-(<i>p</i> -methoxyphenyl)propiophenone	26.36	0	74.67	26.36	27.1
353	26.36	0	74.67	76.8	26.36	27.1
$C_{16}H_{16}O_3$	9*A10+2*A12+A3*B3+2*A1+A32+A11+A35 2,2-dimethoxy-1,2-diphenylethanone	20.86	0	61.63	20.86	28.2
338.5	20.86	0	61.63	83.4	20.86	28.2
$C_{16}H_{17}NO$	10*A10+A11+A12+2*A1+2*A32*C32+A35+A4*B4 N,N-dimethyl-2,2-diphenylacetamide	25.43	67.55	62.47	25.43	28.2
407.1	25.43	67.55	62.47	69.3	25.43	28.2
$C_{16}H_{18}FN_3O_3$	2*A1+10*A10+2*A11+A3*B3+A59 1-ethyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)-3-quinolinecarboxylic acid	32.97	0	65.91	32.97	41.3
500.2	32.97	0	65.91	82.5	32.97	41.3
$C_{16}H_{18}N_2O^*$	2*A14+6*A15+2*A119+A121+A114+3*A19 +A18*B18+2*A12+2*A10+A24+A36+A1+A2 4- <i>n</i> -butyl-4'-hydroxyazobenzene	5.25	0	14.93	5.25	0
351.6	5.25	0	14.93	0	5.25	0
$C_{16}H_{18}N_4O_4$	No prediction made N,N-(2-hydroxyethyl)-4-(4-nitrophenyl)azoaniline	32.43	0	66.97	32.43	44.6
484.2	32.43	0	66.97	92.1	32.43	44.6
$C_{16}H_{19}BrO_2$	8*A10+4*A12+4*A2+2*A30*E30+2*A42+A43+A50 4- <i>trans</i> -(4-bromophenyl)cyclohexyl (E)-2-butenate	28.4	0	73.16	28.4	30.9
388.2	28.4	0	73.16	79.6	28.4	30.9
$C_{16}H_{19}ClO_2$	A14+3*A15+2*A16+4*A10+A11+A12+A21+A38 +A1+A6*B6+A6 4- <i>trans</i> -(4-chlorophenyl)cyclohexyl (E)-2-butenate	30.2	0	78.2	30.2	30.2
386.2	30.2	0	78.2	78.2	30.2	30.2
$C_{16}H_{19}FO_2$	A14+3*A15+2*A16+4*A10+A11+A12+A38 +A1+A6*B6+A6+A22*B22 4- <i>trans</i> -(4-fluorophenyl)cyclohexyl (E)-2-butenate	25.1	0	70.86	25.1	27.9
354.2	25.1	0	70.86	78.6	25.1	27.9
$C_{16}H_{19}N_3O_2$	A14+3*A15+2*A16+4*A10+A11+A12+A38+A1+A6*B6+A6+A24 N,N-(2-hydroxyethyl)-4-phenylazoaniline	29.96	0	73.61	29.96	36.3
407	29.96	0	73.61	89.3	29.96	36.3
$C_{16}H_{20}N_2$	9*A10+3*A12+4*A2+2*A30*E30+2*A42+A43 tetracyclopropylsuccinonitrile	22.3	0	57.18	22.3	25.1
390	22.3	0	57.18	64.4	22.3	25.1
$C_{16}H_{20}O_6P_2S_3$	4*A14+4*A16+2*A4*B4+2*A56 O,O',O'-tetramethyl O,O'-thiodi- <i>p</i> -phenylene bis(phosphorothioate)	33.03	0	108.94	33.03	31.5
303.2	33.03	0	108.94	104.0	33.03	31.5
$C_{16}H_{22}NClO_3$	8*A10+4*A12+A84+2*A79+4*A1 N-(chloroacetyl)- <i>n</i> -(2,6-diethylphenyl)glycine ethyl ester	23.84	0	74.97	23.84	30.7
318	23.84	0	74.97	96.6	23.84	30.7
$C_{16}H_{22}O_3Si_3$	3*A1+5*A2+3*A10+2*A11+A12+A22*C22+A59+A38 1,1,3,3-tetramethyl-5,5-diphenylcyclotrisiloxane	22.19	0	65.66	22.19	23.4
338.0	22.19	0	65.66	69.3	22.19	23.4
$C_{16}H_{24}N_6$	4*A1+A14+3*A15+3*A112+3*A139+10*A10+2*A11 1-(methylphenethylamino)-3,5-bis(dimethylamino)- <i>s</i> -triazine	20.04	0	59.96	20.04	24.5
334.2	20.04	0	59.96	73.2	20.04	24.5
$C_{16}H_{25}NO_2$	5*A1+2*A2+A11+5*A10+3*A41+3*A12+3*A43 nonyl phenylcarbamate	28.07	0	85.77	28.07	43.3
327	28.07	0	85.77	132.4	28.07	43.3
$C_{16}H_{28}O_2$	5*A10+A12+A1+8*A2*B2+A69 1,9-cylohexadecanedione	17.95	59.59	82.47	17.95	27.7
301.2	17.95	59.59	82.47	78.8	17.95	27.7
351.2	8.03	22.87	82.47	78.8	25.98	27.7
$C_{16}H_{28}O_4$	A14+13*A15+2*A114 1,7-cyclododecanedione bis ethylene ketal	36.94	0	77.26	36.94	36.6
478.2	36.94	0	77.26	76.6	36.94	36.6
$C_{16}H_{32}$	3*A14+11*A15+2*A17+4*A112 cyclohexadecane	18.83	69.42	86.45	18.83	27.1
271.2	18.83	69.42	86.45	81.5	24.27	27.1
283.2	1.26	4.43	86.45	81.5	24.27	27.1
332.2	4.18	12.59	86.45	81.5	24.27	27.1
$C_{16}H_{32}$	A14+13*A15 <i>n</i> -decylcyclohexane	38.62	0	142.29	38.62	35.6
271.4	38.62	0	142.29	131.3	38.62	35.6

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (calcd)	
	9*A2*B2+A1+A14+3*A15+A16						[215]
$C_{16}H_{32}$	1-hexadecene						
249.2	3.87	15.53					
277.5	30.21	108.86	124.39	161.5	34.08	44.8	
	A5+A6+13*A2*B2+A1						[165]
$C_{16}H_{32}O_2$	hexadecanoic acid						
335.7	54.81	0	163.27	161.6	54.81	54.2	
	14*A2*B2+A1+A36						[216]
$C_{16}H_{32}O_4$	6,6,14,14-tetramethyl-1,3,9,11-tetraoxacyclohexadecane						
358.6	29.71	0	82.84	87.5	29.71	31.4	
	A14+13*A15+4*A112+4*A1+2*A17						[117]
$C_{16}H_{32}O_4$	2,2,10,10-tetramethyl-1,3,9,11-tetraoxacyclohexadecane						
371.3	25.94	0	69.87	87.5	25.94	32.5	
	A14+13*A15+4*A112+4*A1+2*A17						[117]
$C_{16}H_{32}O_8$	1,4,7,10,13,16,19,22-octaoxacyclotetracosane						
292.2	34.5	0	118.07	118.4	34.5	34.6	
	A14+21*A15+6*A112						[120]
$C_{16}H_{33}NO$	N-hexyl decanamide						
301	6	19.93					
311	31	99.68	119.61	157.9	37	49.1	
	2*A1+13*A2*B2+A60						[127]
$C_{16}H_{33}NO$	N-butyl dodecanamide						
322.1	39	0	121.08	151.0	39.0	48.7	
	2*A1+10*A2*B2+A60+3*A2						[127]
$C_{16}H_{33}NO_3$	N-tetradecylglycine						
379.6	6.8	17.91					
396.6	47.4	119.52	137.43	156.8	54.2	62.2	
	13*A2*B2+A1+A44+A36*B36+A2						[249]
$C_{16}H_{33}NO_3$	N-decyl-L-leucine						
343.1	1.2	3.5					
383.1	27.5	71.78	75.28	128.7	28.7	49.3	
	3*A1+9*A2*B2+A3+A3*B3+A36*B36+A44+A2						[249]
$C_{16}H_{33}NO_3$	N-decyl-DL-leucine						
357.1	28.9	0	71.78	128.7	28.9	45.9	
	3*A1+9*A2*B2+A3+A3*B3+A36*B36+A44+A2						[249]
$C_{16}H_{34}$	hexadecane						
291.3	53.35	183.13					
291.1	51.46	176.79	176.79	165.8	53.35	48.3	
	2*A1+14*A2*B2						[216]
$C_{16}H_{34}O$	1-hexadecanol						
322.3	33.6	104.18					
322.2	23.72	73.22					
322.2	58.41	181.17	181.17	159.3	58.41	51.3	
	A1+15*A2*B2+A30						[224]
$C_{16}H_{34}O_2S$	3(<i>n</i> -tridecylthio)-1,2-propanediol						
296.9	11.3	38.06					
330.6	22.7	68.66	106.72	170.3	34	56.3	
	A1+12*A2*B2+A84+2*A30*C30+2*A2+A3*B3						[217]
$C_{16}H_{34}O_3$	3(<i>n</i> -tridecyloxy)-1,2-propanediol						
342.2	51.4	0	158.54	172.9	51.4	56.0	
	A1+12*A2*B2+A32+2*A30*C30+2*A2+A3*B3						[217]
$C_{16}H_{35}NO_2$	3(<i>n</i> -tridecylamino)-1,2-propanediol						
354.9	68.7	0	193.58	162.9	68.7	57.8	
	A1+12*A2*B2+A44+2*A30*C30+2*A2+A3*B3						[217]
$C_{16}H_{36}Ge$	tetrabutylgermane						
198.6	19.1	0	96.17	120.8	19.1	24.0	
	4*A1+12*A2+A102						[53]
$C_{16}H_{40}O_4Si_4$	octaethylcyclotetrasiloxane						
208.2	12.22	58.7					
213.4	13.71	64.24	122.94	115.7	25.92	24.7	
	8*A1+8*A2+4*A112+4*A139+A14+5*A15						[227]
$C_{17}H_{12}$	1,2-benzofluorene						
399.9	3.8	9.5					
462.8	18.4	39.76	49.26	50.9	22.2	23.6	
	A14+2*A15+4*A19+10*A10+2*A12						[216]
$C_{17}H_{12}$	2,3-benzofluorene						
489.7	23.4	0	47.78	50.9	23.4	24.9	
	A14+2*A15+4*A19+10*A10+2*A12						[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{17}H_{12}O$	4-ethynyl-1-[(4-ethynylphenyl)methoxy]benzene					
371.2	21.2	0	57.11	63.4	21.2	23.5 [216]
	$2*A9+2*A8+3*A12+A11+A2+A32+8*A10$					
$C_{17}H_{12}O_2$	4-benzoyl-1-naphthol					
440.6	28.64	0	65	69.0	28.64	30.4 [215]
	$11*A10+5*A12+A35+A31$					
$C_{17}H_{12}O_2$	1-benzoyl-2-naphthol					
414.1	31.35	0	75.71	69.0	31.35	28.6 [215]
	$11*A10+5*A12+A35+A31$					
$C_{17}H_{12}O_2$	2-benzoyl-1-naphthol					
343.9	20.18	0	58.68	69.0	20.18	23.7 [215]
	$11*A10+5*A12+A35+A31$					
$C_{17}H_{12}O_2$	1-naphthyl benzoate					
329.2	16.98	0	51.58	66.7	16.98	22.0 [118]
	$12*A10+4*A12+A38$					
$C_{17}H_{12}O_2$	2-naphthyl benzoate					
381.2	26.23	0	68.81	66.7	26.23	25.4 [118]
	$12*A10+4*A12+A38$					
$C_{17}H_{13}F_3O$	4- <i>n</i> -propoxy-2',3',4'-trifluorodiphenylacetylene					
327.3	26.1	0	79.74	80.5	26.1	26.4 [196]
	$6*A10+6*A12+3*A24+2*A2+A1+2*A9+A32$					
$C_{17}H_{14}F_2$	4- <i>n</i> -propyl-3',4'-difluorodiphenylacetylene					
311	20.2	0	64.95	72.0	20.2	22.4 [196]
	$7*A10+A11+1*A12+2*A24+2*A2+A1+2*A9$					
$C_{17}H_{14}F_2O$	4- <i>n</i> -propoxy-2',4'-difluorodiphenylacetylene					
326.9	25.2	0	77.09	78.8	25.2	25.8 [196]
	$7*A10+5*A12+2*A24+2*A2+A1+2*A9+A32$					
$C_{17}H_{14}N_2O_2$	2,2-bis-(4-cyanatophenyl)propane					
355.8	26.69	0	75.02	71.7	26.69	25.51 [216]
	$2*A1+A4+2*A11+2*A12+2*A58+8*A10$					
$C_{17}H_{14}O_5$	3-[1-(2-furanyl)-3-oxobutyl]-4-hydroxy-2H-1-benzopyran-2-one					
391.8	33.88	0	86.49	87.5	33.88	34.3 [221]
	$2*A14+5*A15+5*A19+2*A18+A18*B18+4*A10+A1$ $+A2+A3+A35+A115+A112+A30*D30$					
$C_{17}H_{15}F$	4- <i>n</i> -propyl-4'-fluorodiphenylacetylene					
324	24.1	0	74.38	70.2	24.1	22.8 [196]
	$8*A10+A11+3*A12+A24+2*A2+A1+2*A9$					
$C_{17}H_{15}FO$	4- <i>n</i> -propoxy-4'-fluorodiphenylacetylene					
356.8	27.1	0	75.95	77.0	27.1	27.5 [196]
	$8*A10+4*A12+A24+2*A2+A1+2*A9+A32$					
$C_{17}H_{16}Br_2O_3$	isopropyl 4,4'-dibromobenzilate					
348.1	24.55	0	70.53	93.5	24.55	32.5 [216]
	$8*A10+2*A11+2*A12+A30*D30+A38+2*A1+A3*B3+2*A21+A4*B4$					
$C_{17}H_{18}FN_3O_3$	1-cyclopropyl-6-fluoro-1,4-dihydro-4-oxo-7-(1-piperazinyl)-3-quinolinecarboxylic acid					
541.5	64.48	0	119.08	91.2	64.48	49.4 [36]
	$3*A14+6*A15+3*A19+A18*B18+A114+2*A119$ $+A121+2*A10+2*A12+A36*F36+A24$					
$C_{17}H_{19}FNO_2$	4- <i>trans</i> -(3-fluoro-4-cyanophenyl)cyclohexyl (E)-but-2-enoate					
393.2	21.1	0	53.66	81.5	21.1	32.0 [140]
	$3*A10+2*A12+A11+A24+A56+A14+3*A15+2*A16+A38+A6+A6*B6+A1$					
$C_{17}H_{19}F_3O_3$	4- <i>trans</i> -(trifluoromethoxyphenyl)cyclohexyl (E)-but-2-enoate					
340.2	21.6	0	63.49	83.6	21.6	28.4 [140]
	$4*A10+A12+A11+3*A25+A14+3*A15+2*A16$ $+A38+A6+A6*B6+A1+A32+A4*B4$					
$C_{17}H_{21}NO_2$	N,N-diethyl-2-(1-naphthoxy)propionamide					
345.3	24.57	0	71.16	80.3	24.57	27.7 [221]
	$3*A1+2*A2+B3*A3+7*A10+3*A12+A32+A59$					
$C_{17}H_{34}O$	9-heptadecanone					
323.9	66.68	0	205.87	170.4	66.68	55.2 [21]
	$2*A1+A35+A14*A2*B2$					
$C_{17}H_{34}O_2$	heptadecanoic acid					
329.2	7.44	22.59				
334.3	51.33	153.55	176.15	170.9	58.77	57.1 [216]
	$15*A2*B2+A1+A36$					
$C_{17}H_{34}O_2$	methyl palmitate					
307.2	68.16	0	221.84	173.5	68.16	53.3
305.2	55.35		181.4		55.35	

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{17}H_{35}NO_3$	2*A1+14*A2*B2+A38 N-tetradecyl-L-alanine 52.3	0	142.47	157.6	52.3	[217, 391]
$C_{17}H_{35}NO_3$	367.1					57.9 [249]
$C_{17}H_{35}NO_3$	2*A1+13*A2*B2+A3*B3+A44+A36*B36 N-dodecyl-L-valine 33.1	0	87.08	140.2	33.1	53.3 [249]
$C_{17}H_{35}NO_3$	380.1					
$C_{17}H_{35}NO_3$	3*A1+A3+A3*B3+11*A2*B2+A44+A36*B26 N-dodecyl-DL-valine 64.4	0	176.63	140.2	64.4	51.1 [249]
$C_{17}H_{36}$	364.6					
$C_{17}H_{36}$	3*A1+A3+A3*B3+11*A2*B2+A44+A36*B36 <i>n</i> -heptadecane 10.96	38.56				
$C_{17}H_{36}$	284.3					
$C_{17}H_{36}$	295.1	40.17	136.11	174.67	175.1	51.13 [216]
$C_{17}H_{36}O_2S$	2*A1+15*A2*B2 3(<i>n</i> -tetradecylthio)-1,2-propanediol 16.3	53.88				
$C_{17}H_{36}O_2S$	302.5					
$C_{17}H_{36}O_2S$	336.4	26.8	79.67	133.55	179.6	43.1 60.4 [217]
$C_{17}H_{36}O_3$	A1+13*A2*B2+A84+2*A30*C30+2*A2+A3*B3 3(<i>n</i> -tetradecyloxy)-1,2-propanediol 62.1	0	187.44	182.2	62.1	60.4 [217]
$C_{17}H_{37}NO_2$	331.3					
$C_{17}H_{37}NO_2$	A1+13*A2*B2+A32+2*A30*C30+2*A2+A3*B3 3(<i>n</i> -tetradecylamino)-1,2-propanediol 64.9	0	182.2	172.2	64.9	61.3 [217]
$C_{17}H_{37}NO_2$	356.2					
$C_{18}H_{10}$	A1+13*A2*B2+A32+2*A30*C30+2*A2+A3*B3 benzofluoranthene 5.35	13.28				
$C_{18}H_{10}$	402.8					
$C_{18}H_{10}$	402.1	0.88	2.19			
$C_{18}H_{10}$	352.7	0.44	1.23			
$C_{18}H_{10}$	424	11.8	27.83	44.53	36.1	18.47 15.3 [264]
$C_{18}H_{12}$	10*A10+5*A12+3*A13 triphenylene 24.74	0	52.53	44.1	24.74	20.8 [216]
$C_{18}H_{12}$	471					
$C_{18}H_{12}$	12*A10+6*A12 chrysene 3.22	6.29				
$C_{18}H_{12}$	512.2					
$C_{18}H_{12}$	531.4	26.15	49.21	55.5	44.1	29.37 23.4 [255]
$C_{18}H_{12}$	12*A10+6*A12 1,2-benzanthracene 21.38	0	49.23	44.1	21.38	19.1 [215]
$C_{18}H_{12}$	434.3					
$C_{18}H_{12}$	12*A10+6*A12 3,4-benzophenanthrene 16.32	0	48.75	44.1	16.32	14.8 [215]
$C_{18}H_{12}$	334.7					
$C_{18}H_{13}FO$	12*A10+6*A12 4-ethoxy-4'-fluorodiphenylacetylene 33.9	0	84.71	64.4	33.9	25.8 [195]
$C_{18}H_{14}$	400.2					
$C_{18}H_{14}$	A1+A2+4*A12+8*A10+4*A9+A24*B24+A32 <i>m</i> -terphenyl 22.59	0	62.76	73.9	22.59	26.6 [256]
$C_{18}H_{14}$	360					
$C_{18}H_{14}$	14*A10+4*A12 <i>p</i> -terphenyl 0.3	1.6				
$C_{18}H_{14}$	193.6					
$C_{18}H_{14}$	487	35.3	72.5	74.1	73.9	35.6 35.9 [38,155]
$C_{18}H_{14}$	14*A10+4*A12 <i>o</i> -terphenyl 17.2	0	52.3	73.9	17.2	24.3 [91]
$C_{18}H_{14}O_3$	329.4					
$C_{18}H_{14}O_3$	14*10+4*A12 cinnamic anhydride 32.77	0	102.02	87.6	32.77	28.1 [215]
$C_{18}H_{15}F_3O$	321.2					
$C_{18}H_{15}F_3O$	10*A10+2*A12+A39+2*A6+2*A6*B6 4- <i>n</i> -butoxy-1',3',4'-trifluorodiphenylacetylene 36	0	104.53	87.7	36	30.2 [196]
$C_{18}H_{15}ClSi$	344.4					
$C_{18}H_{15}ClSi$	6*A10+6*A12+3*A24+A32+3*A2+A1+2*A9 triphenylchlorosilane 26.88	0	72.53	77.9	26.88	28.9 [216]
$C_{18}H_{15}N$	370.6					
$C_{18}H_{15}N$	15*A10+3*A12+A109+A22*B22 triphenylamine 24.89	0	62.21	66.5	24.89	26.6 [217]
$C_{18}H_{15}N$	400.2					
$C_{18}H_{15}N$	15*A10+3*A12+A43					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (expt)	$\Delta_0^{T_{fus}} S_{pcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{pcc}$ (expt)	$\Delta_0^{T_{fus}} H_{pcc}$ (calcd)	
$C_{18}H_{15}OP$	431.9	triphenylphosphine oxide 24.22	0	56.08	56.1	24.22	24.2 [246]
$C_{18}H_{15}O_4P$	322.5	15*A10+3*A12+A73 triphenyl phosphate 29.61	0	91.81	78.8	29.61	25.4 [215]
$C_{18}H_{15}P$	354.4	15*A10+3*A12+A74 triphenylphosphine 19.69	0	55.56	68.0	19.69	24.1 [246]
$C_{18}H_{16}F_2$	323.5	15*A10+3*A12+A72 4- <i>n</i> -butyl-3',4'-difluorodiphenylacetylene 25.3	0	78.21	79.1	25.3	25.6 [196]
$C_{18}H_{16}O_3$	371.2	7*A10+A11+4*A12+2*A24+3*A2+ A1+2*A9 1,8-diphenyl-2,3,5-trioxabicyclo[4.3.0]non-7-ene 21.7	0	58.46	45.3	21.7	16.8 [257]
$C_{18}H_{16}O_8$	423.7	10*A10+A11+A12+2*A14+3*A15+A19+A18+6*A16 +A17+A112+A113 1,2,3,4-tetracarboxymethoxynaphthalene 35.9	0	84.47	85.8	35.9	36.3 [217]
$C_{18}H_{16}O_8$	438.2	4*A1+4*A38+4*A10+6*A12 1,2,4,5-tetracarboxymethoxynaphthalene 36.4	0	82.89	85.8	36.4	37.6 [217]
$C_{18}H_{16}O_8$	470.2	4*A1+4*A38+4*A10+6*A12 1,2,5,6-tetracarboxymethoxynaphthalene 42.1	0	89.37	85.8	42.1	40.3 [217]
$C_{18}H_{16}O_8$	407.2	4*A1+4*A38+4*A10+6*A12 1,2,6,7-tetracarboxymethoxynaphthalene 34.2	0	83.76	85.8	34.2	34.9 [217]
$C_{18}H_{16}O_8$	458.2	4*A1+4*A38+4*A10+6*A12 2,3,6,7-tetracarboxymethoxynaphthalene 42.2	0	91.88	85.8	42.2	39.3 [217]
$C_{18}H_{16}O_8$	477.2	4*A1+4*A38+4*A10+6*A12 1,4,5,8-tetracarboxymethoxynaphthalene 36.1	0	75.69	85.8	36.1	40.9 [217]
$C_{18}H_{17}Cl_2NO_3$	341.7	4*A1+4*A38+4*A10+6*A12 ethyl <i>N</i> -benzoyl- <i>n</i> -(3,4-dichlorophenyl)- <i>dl</i> -alaninate 27.06	0	79.19	90.9	27.06	31.1 [221]
$C_{18}H_{17}F$	329.9	8*A10+4*A12+2*A1+A2+A3*B3+A38+2*A22*D22+A59 4- <i>n</i> -butyl-4'-fluorodiphenylacetylene 18.5	0	56.08	77.4	18.5	25.5 [196]
$C_{18}H_{17}FO$	346.7	8*A10+A11+3*A12+A24+3*A2+A1+2*A9 4- <i>n</i> -butoxy-4'-fluorodiphenylacetylene 25.4	0	73.26	84.2	25.4	29.4 [196]
$C_{18}H_{18}$	369	8*A10+4*A12+A24+3*A2+A1+2*A9+A32 1-methyl-7-isopropylphenanthrene 18.03	0	48.87	46.6	18.03	17.2 [216]
$C_{18}H_{18}ClNS$	370.3	8*A10+3*A1+2*A11+A3+4*A12 2-chloro-9-(3-dimethylaminopropylidene)-10-thioxanthene 27.82	0	75.13	77.8	27.82	28.8 [216]
$C_{18}H_{18}N_2O_2$	521.2	A14+3*A15+2*A19+38*A19+A131+7*A10+A12+A22*C22+A6*B6 +2*A2+2*A1+A43 N,N'-(2-hydroxyethyl)-1,4-diaminoanthraquinone 32.34	0	62.05	86.0	32.34	44.8 [13]
$C_{18}H_{18}O_2$	387.2	3-diphenylmethyl-2,4-pentanedione 27.02	0	69.78	73.3	27.02	28.4 [259]
$C_{18}H_{18}O_3$	343.9	2*A1+A3*B3+2*A35+A3+10*A10+2*A11 butyl 9-hydroxy-9H-fluorene-9-carboxylate 25.56	0	74.31	81.2	25.56	27.9 [215]
$C_{18}H_{20}Cl_2$	331.6	A14+2*A15+8*A10+4*A19+A17+A30*B30+A38+A1+3*A2 1,1'-(2,2-dichloroethylidene)bis(4 ethylbenzene) 23.34	0	70.38	76.7	23.34	25.4 [215]
$C_{18}H_{20}O_2$	443.8 441.8	8*A10+4*A11+A3+A3*B3+2*A1+2*A2+2*A22*B22 diethylstilbestrol 31.76	0	71.57	97.8	31.76	43.5 43.5
		8*A10+4*A12+2*A31+2*A7+2*1+2*A2	0	65.1	97.8	28.8	43.5 [221, 394]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{\text{fus}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{\text{fus}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{\text{fus}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{\text{fus}} H_{\text{tpcc}}$ (calcd)
$\text{C}_{18}\text{H}_{22}$	2,3-dimethyl-2,3-diphenylbutane 25.52	0	65.11	55.6	25.52	21.8 [289]
$\text{C}_{18}\text{H}_{22}\text{N}_4$	$10^*A10+4^*A1+2^*A11+2^*A4$ 2,3-dimethyl-2,3-bis(phenylazo)butane 21.09	0	61.6	76.8	21.09	26.3 [258]
$\text{C}_{18}\text{H}_{22}\text{O}_2$	$4^*A1+2^*A4^*B4+10^*A10+2^*A12+4^*A42$ (dl) anisylidenecamphor 26.36	0	70.95	67.9	26.36	25.2 [273]
$\text{C}_{18}\text{H}_{22}\text{O}_2$	$2^*A14+A15+4^*A1+A114+A17+A16+A6+4^*A10+2^*A12+A32+A19+A17$ (d) anisylidenecamphor 30.12	0	75.41	67.9	30.12	27.1 [273]
$\text{C}_{18}\text{H}_{22}\text{O}_2$	$2^*A14+A15+4^*A1+A114+A17+A16+A6+4^*A10+2^*A12+A32+A19+A17$ di- α -cumyl peroxide 28.14	0	90.08	90.1	28.14	28.2 [216]
$\text{C}_{18}\text{H}_{23}\text{FO}_2$	$4^*A1+2^*A4^*B4+2^*A11+10^*A10+A33$ 4- <i>trans</i> -(4-fluorophenylethyl)cyclohexyl (E)-butenoate 25	0	74.58	92.9	25	31.1 [140]
$\text{C}_{18}\text{H}_{28}\text{Si}_4\text{O}_4$	$4^*A10+A11+A12+2^*A2+A14+3^*A15+A16+A16+A38+A24+A1+A6+A6^*B6$ 1,1,3,3,5,5-hexamethyl-7,7-diphenylcyclotetrasiloxane 42.73	0	140.12	78.4	42.73	23.9 [216]
$\text{C}_{18}\text{H}_{30}\text{O}$	$6^*A1+10^*A10+2^*A11+4^*A112+4^*A139+A14+5^*A15$ 2,4,6-tri- <i>tert</i> -butylphenol 19.46	0	48.01	52.5	19.46	21.3 [220]
$\text{C}_{18}\text{H}_{30}\text{O}_4$	$3^*A11+2^*A10+9^*A1+3^*A4+A31+A12$ <i>p</i> -diacetylbenzene diethyl ketal 1.31	7.76				
	23.5	72.05	79.8	117.6	24.81	38.4 [216]
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$4^*A10+2^*A11+2^*A4^*B4+6^*A1+4^*A2+4^*A32$ 1,10-cyclooctadecanedione 11.84	32.96				
	27.03	72.81	105.78	86.2	38.87	32.0 [114]
$\text{C}_{18}\text{H}_{32}\text{O}_4$	$A14+15^*A15+2^*A114$ 1,8-cyclotetradecanedione bis ethylene ketal 30.67	0	67.08	84	30.67	38.4 [114]
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$3^*A14+13^*A15+2^*A17+4^*A112$ <i>trans</i> -9-octadecenoic acid (elaidic acid) 61.55	0	193.8	172.1	61.55	54.7 [216]
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1+14^*A2^*B2+2^*A6+A36$ <i>cis</i> -9-octadecenoic acid 39.6	0	138.24	172.1	39.6	49.3 [216]
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1+14^*A2^*B2+2^*A6+A36$ <i>cis</i> -6-octadecenoic acid 47.5	0	156.43	172.1	47.5	52.3 [216]
$\text{C}_{18}\text{H}_{36}$	$A1+14^*A2^*B2+2^*A6+A36$ <i>n</i> -dodecylcyclohexane 45.84	0	177.11	150.0	45.84	38.8 [216]
$\text{C}_{18}\text{H}_{36}$	$A14+A16+A1+11^*A2^*B2+3^*A15$ cyclooctadecane 29.29	98.22				
	9.87	28.52	126.74	88.9	39.16	30.8 [110]
$\text{C}_{18}\text{H}_{36}$	$15^*A15+A14$ 1,1-dimethylcyclohexadecane 1.26	5.81				
	0.42	1.89				
	14.23	49.02	56.72	82.1	15.9	23.8 [112]
$\text{C}_{18}\text{H}_{36}\text{N}_2\text{O}_2^*$	$A14+13^*A15+2^*A1+A17$ N,N'-di- <i>n</i> -hexyladipamide 40.79	0	94.56	168.7	40.79	72.9 [216]
$\text{C}_{18}\text{H}_{36}\text{O}_2$	$14^*A2^*B2+2^*A1+2^*A60$ octadecanoic acid 61.21	0	178.66	180.2	61.21	61.7 [216]
$\text{C}_{18}\text{H}_{36}\text{O}_2$	$16^*A2^*B2+A1+A36$ ethyl hexadecanoate 15.09	0	50.93	180.6	15.09	53.5 [216]
	$2^*A1+A38+14^*A2^*B2+A2$					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{18}H_{36}O_4$ 373.0	2,2,11,11-tetramethyl-1,3,10,12-tetraoxacyclooctadecane					
	35.1	0	94.1	94.9	35.1	35.4
	A14+15*A15+4*A112+2*A17+4*A1					
$C_{18}H_{37}NO$ 336.1	N-butyl tetradecanamide					
	45	0	133.89	170.0	45	57.1
	2*A1+3*A2+A60+12*A2*B2					
$C_{18}H_{37}NO$ 377.2	octadecanamide					
	59.91	0	158.84	194.8	59.91	73.5
	16*A2*B2+A1+A61					
$C_{18}H_{37}NO_3$ 384.6	N-hexadecylglycine					
	4.5	11.7				
	366.1	5.6				
	393.1	56.5	143.73	170.73	66.6	69.0
	15*A2*B2+A1+A44+A36*B36+A2					
$C_{18}H_{37}NO_3$ 383.1	N-dodecyl-L-leucine					
	33.5	0	87.44	147.3	33.5	56.4
	3*A1+11*A2*B2+A3+A33+A36*B36+A44+A2					
$C_{18}H_{37}NO_3$ 341.1	N-dodecyl-DL-leucine					
	28.9	84.73				
	356.6	31	86.93	171.66	59.9	52.5
	3*A1+11*A2*B2+A3+A33*B3+A36*B36+A44+A2					
$C_{18}H_{38}$ 301.3	octadecane					
	61.5	0	204.6	184.5	61.5	55.6
	2*A1+16*A2*B2					
$C_{18}H_{38}O$ 334.2	octadecanol					
	70.08	0	209.7	178.0	70.08	59.5
	17*A2*B2+A1+A30					
$C_{18}H_{48}Si_6$ 226.3	1,2,3,4,5,6-hexamethyl-1,2,3,4,5,6-hexaethylcyclohexasilane					
	3.8	16.79				
	439.2	1.8	4.1	20.89	5.6	39.6
	A14+3*A15+6*A139+12*A1+6*A2					
$C_{19}H_{13}F_3O$ 424.9	4-ethoxy-4'-trifluoromethyldiphenylacetylene					
	32.73	0	77.03	62.9	32.73	26.7
	A1+A2+A11+3*A12+8*A10+4*A9+3*A25+A4*B4+A32					
$C_{19}H_{14}F_2$ 343.7	4-n-propyl-3',4'-difluorodiphenylacetylene					
	22.03	0	64.1	66.4	22.03	22.8
	A1+2*A2+A11+4*A12+7*A10+4*A9+2*A24					
$C_{19}H_{15}Cl$ 376.8	triphenylchloromethane					
	27.9	0	74.04	70.3	27.9	26.5
	15*A10+3*A11+A22+A4*B4					
$C_{19}H_{16}$ 365.3	triphenylmethane					
	21.97	0	60.13	66.0	21.97	24.1
	15*A10+A3+3*A11					
$C_{19}H_{16}O_2$ 395.2	2-fluorenyl-2-methyl-1,3-cyclopentanedione					
	24.6	0	62.25	57.4	24.6	22.7
	2*A14+4*A15+2*A114+A17+A1+A16+4*A19+8*A10					
$C_{19}H_{17}F_3O$ 315.8	4-n-pentoxy-2',3',4'-trifluorodiphenylacetylene					
	33.1	0	104.81	94.8	33.1	29.9
	6*A10+6*A12+3*A24+A32+4*A2+A1+2*A9					
$C_{19}H_{18}F_2$ 323.1	4-n-pentyl-3',4'-difluorodiphenylacetylene					
	22.1	0	68.4	86.2	22.1	27.9
	7*A10+A11+4*A12+2*A24+4*A2+A1+2*A9					
$C_{19}H_{18}O_2$ 394.2	2-methyl-2-diphenylmethyl-1,3-cyclopentanedione					
	34.3	0	87.01	59.7	34.3	23.5
	A14+2*A15+2*A114+A17+A1+A3+2*A11+10*A10					
$C_{19}H_{19}F$ 337.4	4-n-pentyl-4'-fluorodiphenylacetylene					
	25.6	0	75.87	84.5	25.6	28.5
	8*A10+A11+3*A12+A24+4*A2+A1+2*A9					
$C_{19}H_{19}FO$ 330.9	4-n-pentoxy-4'-fluorodiphenylacetylene					
	27.2	0	82.2	91.3	27.2	30.2
	8*A10+4*A12+A24+4*A2+A1+2*A9+A32					
$C_{19}H_{20}F_3N_3O_3$ 350	2-[3-(trifluoromethyl)-phenyl]amino-3-pyridinecarboxylic acid β -morpholino ethyl ester					
	34.5	0	98.57	91.1	34.5	31.9
	A14+3*A15+A112+A119+2*A2+A38+7*A10+3*A12+A41+A44+A11+A4*B4+3*A25					
$C_{19}H_{20}O_2$ 352.2	3-methyl-3-diphenylmethyl-2,4-pentanedione					
	25.1	0	71.27	77.6	25.1	27.3

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)	
$C_{19}H_{21}NO$	349.2	$3^*A1 + A4^*B4 + 2^*A35 + A3 + 10^*A10 + 2^*A11$ 1,2-diphenyl-2-(N-piperidinyl)-1-ethanone 33.93	0	97.16	77.0	33.93	[259]
$C_{19}H_{23}NO$	73.41 321.6	$10^*A10 + A11 + A12 + A35 + A14 + 3^*A15 + A119 + A3^*B3$ <i>p-n</i> -hexyloxybenzylideneaniline 0.19	2.62	98.72	104.4	31.1	[253]
$C_{19}H_{24}O$	337.7	$9^*A10 + 3^*A12 + A42 + A32 + 5^*A2 + A1 + A6^*B6$ 2- <i>tert</i> -butyl-4-methyl-6- α -methylbenzylphenol 31.38	0	92.92	63.0	31.4	[216]
$C_{19}H_{26}O_2$	428	$5^*A1 + A4 + 4^*A11 + A12 + 7^*A10 + A31 + A3$ testosterone 29.45	0	68.81	60.2	29.45	[101]
$C_{19}H_{28}N_2$	326.2	$4^*A14 + 5^*A15 + 2^*A17 + 4^*A16 + 2^*A1 + A30^*B30 + A19 + A18^*B18 + A114$ 4-(4- <i>n</i> -heptyl-1-piperidinyl)benzoxonitrile 29.01	0	88.95	103.2	29.1	[219]
$C_{19}H_{30}O_2$	455.5	$A14 + 3^*A15 + A119 + A16 + 6^*A2 + A1 + 4^*A10 + 2^*A12 + A56$ 5 α -androstande-3-one-17 β -ol 27.15	0	59.6	60.9	27.15	[26]
$C_{19}H_{38}O$	328	$4^*A14 + 5^*A15 + 2^*A17 + 2^*A1 + 5^*A16 + A30^*B30 + A114$ 2-nonadecanone 68.65	0	209.3	189.0	68.65	[216]
$C_{19}H_{38}O$	330	$A35 + 2^*A1 + 16^*A2^*B2$ 10-nonadecanone 66.67	0	202.04	189.0	66.67	[21]
$C_{19}H_{38}O_2$	338 341.2	$A35 + 2^*A1 + 16^*A2^*B2$ nonadecanoic acid 9.76	28.87	195.93	189.6	67.38	[216]
$C_{19}H_{38}O_2$	310.9	$17^*A2^*B2 + A1 + A36$ methyl octadecanoate 64.4	0	205.8	192.1	64.4	[391]
$C_{19}H_{39}NO_3$	374.1	$2^*A1 + A38 + 16^*A2^*B2$ N-hexadecyl-L-alanine 65.3	0	174.55	176.3	65.3	[249]
$C_{19}H_{39}NO_3$	334.6 365.1	$15^*A2^*B2 + 2^*A1 + A3^*B3 + A44 + A36^*B36$ N-tetradecyl-L-valine 14.9	44.53	100.95	158.8	35.5	[249]
$C_{19}H_{39}NO_3$	370.1	$3^*A1 + A3 + A3^*B3 + 13^*A2^*B2 + A44 + A36^*B36$ N-tetradecyl-DL-valine 68.1	0	184	158.8	68.1	[249]
$C_{19}H_{40}$	296.0 304	$3^*A1 + A3 + A3^*B3 + 13^*A2^*B2 + A44 + A36^*B36$ nonadecane 13.67	46.2	202.1	193.8	61.07	[216]
$C_{20}F_{42}$	149.5 202.9 437.9	$2^*A1 + 17^*A2^*B2$ perfluoroicosane 0.67	4.48	243.37	211.6	92.25	[67]
$C_{20}H_{12}$	551.0	$20^*A4^*B4 + 36^*A26 + 6^*A25$ perylene 31.88	0	57.87	43.7	31.88	[216, 217]
$C_{20}H_{12}$	426.2 454.4	$12^*A10 + 6^*A12 + 2^*A13$ 1,2-benzopyrene 2.51	5.89	42.35	43.7	19.08	[215]
$C_{20}H_{12}$	390.2 454.2	$12^*A10 + 6^*A12 + 2^*A13$ 3,4-benzopyrene 8.49	21.77	59.9	43.7	25.81	[215]
$C_{20}H_{14}$	527.2	$12^*A10 + 6^*A12 + 2^*A13$ triptycene 30.29	0	57.46	60.1	30.29	[216]
$C_{20}H_{14}$		$2^*A14 + 2^*A15 + 2^*A16 + 6^*A19 + 12^*A10$ β,β -binaphthyl	0				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
461.2	38.9	0	84.35	58.9	38.9	27.2
$C_{20}H_{14}O_4$	14*A10+6*A12 phenolphthalein					[216]
534	51.05	0	95.59	114.34	51.05	61.06
$C_{20}H_{15}F_3O$	A14+2*A15+A115+2*A19+A17+2*A31+12*A10 4-propoxy-4'-trifluoromethyldiphenyl diacetylene					[216]
315.9	18.81	0	59.54	70.0	18.81	22.1
$C_{20}H_{16}F_2$	A1+2*A2+A11+3*A12+8*A10+4*A9+3*A25+A4*B4+A32 4-n-butyl-3',4'-difluorodiphenyl diacetylene		71.39	73.6	24.33	25.1
340.8	24.33	0				[195]
$C_{20}H_{18}O_2$	A1+3*A2+A11+4*A12+7*A10+4*A9+2*A24 2-fluoroenyl-2-methyl-1,3-cyclohexanedione		79.65	61.1	35.7	27.4
448.2	35.7	0				[259]
$C_{20}H_{18}O_2Sn$	2*A14+5*A15+2*A114+A17+A1+A16+4*A19+8*A10 (acetyloxy)triphenylstannane		105.44	89.8	41.92	35.7
397.6	41.92	0				[221]
$C_{20}H_{19}F_3O$	15*A10+3*A12+A1+A38+A110 4-n-hexyloxy-2',3',4'-trifluorodiphenylacetylene		95.65	85.3	30.8	27.5
322	30.8	0				[196]
$C_{20}H_{20}F_2$	A1+5*A2+6*A12+6*A10+2*A9+2*A24+A32 4-n-hexyl-3',4'-difluorodiphenylacetylene		77.17	78.5	24.3	24.7
314.9	24.3	0				[196]
$C_{20}H_{20}F_2O$	A1+5*A2+A11+5*A12+6*A10+2*A9+2*A24 4-n-hexyloxy-3',4'-difluorodiphenylacetylene		102.29	85.3	33.1	27.6
323.6	33.1	0				[196]
$C_{20}H_{20}F_2O$	A1+5*A2+6*A12+6*A10+2*A9+2*A24+A32 4-n-hexyloxy-2',4'-difluorodiphenylacetylene		106.26	85.3	34.1	27.4
320.9	34.1	0				[196]
$C_{20}H_{20}O_2$	A1+5*A2+6*A12+6*A10+2*A9+2*A24+A32 2-ethyl-2-diphenylmethyl-1,3-cyclopentanedione		73.78	66.8	28.2	25.5
382.2	28.2	0				[259]
$C_{20}H_{20}O_3$	A14+2*A15+2*A114+A17+A1+A3+2*A11+10*A10+A2 4,4-dimethyl-1,8-diphenyl-2,3,5-trioxabicyclo[4.3.0]non-7-ene		59.86	45.9	22.1	16.9
369.2	22.1	0				[257]
$C_{20}H_{22}O_2$	10*A10+A11+A12+2*A14+3*A15+A19+A18+A162+A17+A112+A113+2*A1 3-ethyl-3-diphenylmethyl-2,4-pentanedione		89.39	84.8	34.7	32.9
388.2	34.7	0				[259]
$C_{20}H_{26}O_2$	3*A1+A4*B4+2*A35+A3+10*A10+2*A11+A2 19-nor-17 α -ethylnyltestosterone		82.67	55.1	39.6	26.4
479	39.6	0				[216]
$C_{20}H_{26}O_2$	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114 +A30*B30+A1+A8+A9 2-tert-butyl-4-methoxymethyl-6- α -methylbenzylphenol		79.1	74.8	29.4	27.8
371.7	29.4	0				[101]
$C_{20}H_{26}O_3$	5*A1+A4+A2+A3+4*A11+A12+7*A10+A31+A32 testosterone formate		66.22	66.2	26.36	26.4
398	26.36	0				26.4
398.2	18.12	0	66.2	66.2	18.1	[219, 396]
$C_{20}H_{28}O_2$	4*A14+5*A15+2*A17+4*A16+2*A1+A37+A19+A18*B18+A114 1-[3,5-di-tert-butyl-4-hydroxyphenyl]-5-hexyn-1-one		73.47	74.9	25.14	25.6
342.2	25.14	0				[39]
$C_{20}H_{30}O_3Si_3$	6*A1+2*A4+2*A10+2*A11+2*A12+3*A2+A8+A9+A31+A35 1,1,3,3-tetraethyl-5,5-diphenylcyclotrisiloxane		65.84	97.9	18.37	27.3
279.1	18.37	0				[216]
$C_{20}H_{32}$	4*A1+4*A2+10*A10+2*A11+3*A112+3*A139+A14+3*A15 10,10,13,13-tetramethylcyclohexadeca-1,5-diyne		58.25	63.8	18.83	20.6
323.2	18.83	0				[113]
$C_{20}H_{36}O_2$	4*A1+A14+13*A15+2*A17+4*A20 1,10-cycloicosanedione		168.28	98.7	55.06	32.3
327.2	55.06	0				[114]
$C_{20}H_{36}O_4$	A14+17*A15+2*A112 1,9-cyclohexadecanedione bis ethylene ketal		104.24	91.4	42.13	36.9
404.2	42.13	0				[114]
$C_{20}H_{40}$	3*A14+15*A15+2*A17+4*A112 1,1,9,9-tetramethylcyclohexadecane		68.93	82.6	25.1	30.1
364.2	25.1	0				[112]
	4*A1+A14+13*A15+2*A17					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
$C_{20}H_{40}$	1,1,4,4-tetramethylcyclohexadecane					
303.2	25.1	0	82.8	82.6	25.1	25.1
	4*A1+A14+13*A15+2*A17					
$C_{20}H_{40}$	1,1-dimethylcyclooctadecane					
283.2	23.85	0	84.21	89.5	23.85	25.3
	A14+15*A15+2*A1+A17					
$C_{20}H_{40}O_2$	eicosanoic acid					
348.2	69.2	0	198.7	198.9	69.2	69.3
	18*A2*B2+A1+A36					
$C_{20}H_{40}O_4$	2,2,6,6,10,10,14,14-octamethyl-1,3,9,11-tetraoxacyclohexadecane					
406.9	24.69	0	60.67	88.7	24.69	36.1
	8*A1+2*A17+2*A17+A14+13*A15+4*A112					
$C_{20}H_{41}NO$	N-hexyl tetradecanamide					
310	8	25.81				
328	7	21.34				
334	35	104.79	151.94	195.2	50	65.2
	2*A1+17*A2*B2+A60					
$C_{20}H_{41}NO_3$	N-tetradecyl-L-leucine					
377.5	32.4	0	85.83	166.0	32.4	62.7
	3*A1+13*A2*B2+A3+A3*B3+A2+A44+A36*B36					
$C_{20}H_{41}NO_3$	N-tetradecyl-DL-leucine					
320.1	1.8	5.62				
349.6	54.8	156.75	162.37	166.0	56.6	58.0
	3*A1+13*A2*B2+A3+A3*B3+A2+A44+A36*B36					
$C_{20}H_{42}$	<i>n</i> -eicosane					
308.8	67.8	0	219.6	203.1	67.8	62.7
	2*A1+18*A2*B2					
$C_{21}H_{16}$	1,2'-dinaphthylmethane					
369.6	30.54	0	82.64	61.8	30.54	22.8
	14*A10+2*A11+A2+4*A12					
$C_{21}H_{17}F_3O$	4- <i>n</i> -butoxy-4'-trifluoromethyldiphenyldiacetylene					
414.3	25.37	0	61.24	77.1	25.37	32.0
	A1+3*A2+A11+3*A12+8*A10+4*A9+3*A25+A4*B4+A32					
$C_{21}H_{18}F_2$	4- <i>n</i> -pentyl-3',4'-difluorodiphenyldiacetylene					
355.1	30.86	0	86.91	80.7	30.86	28.7
	A1+4*A2+A11+4*A12+7*A10+4*A9+2*A24					
$C_{21}H_{21}NO$	N,N-dimethyl-2,2-diphenylbenzeneacetamide					
402	25.43	0	63.26	83.5	25.43	33.6
	15*A10+3*A11+A4*B4+2*A1+A59					
$C_{21}H_{24}O_2$	3-propyl-3-diphenylmethyl-2,4-pentanedione					
349.2	27.1	0	77.61	91.9	27.1	32.1
	3*A1+A4*B4+2*A35+A3+10*A10+2*A11+2*A2					
$C_{21}H_{24}O_3Si_3$	<i>cis</i> -1,3,5-trimethyl-1,3,5-triphenylcyclotrisiloxane					
374.3	43.07	0	115.07	79.2	43.07	29.7
	3*A1+15*A10+3*A11+3*A112+3*A139+A14+3*A15					
$C_{21}H_{24}O_3Si_3$	<i>trans</i> -1,3,5-trimethyl-1,3,5-triphenylcyclotrisiloxane					
320.9	43.66	0	136.07	79.2	43.66	25.4
	3*A1+15*A10+3*A11+3*A112+3*A139+A14+3*A15					
$C_{21}H_{28}O_3$	testosterone acetate					
413	27.88	0	67.51	67.7	27.88	27.9
413.2	22.5	0	54.5	67.7	22.5	27.9
	4*A14+5*A15+2*A17+4*A16+3*A1+A19+A18*B18+A114+A38					
$C_{21}H_{28}O_5$	prednisolone					
513	38.86	0	75.75	78.2	38.86	40.1
	4*A14+5*A15+2*A17+A17+4*A16+2*A1					
	+3*A30*E30+A19+2*A18*B18+A18+A114+A35+A2					
$C_{21}H_{28}O_5$	cortisone					
495	36.86	0	74.46	74.1	36.86	37.2
	4*A14+5*A15+A17+2*A17+3*A16+2*A1+2*A30*E30					
	+A19+A18*B18+2*A114+A35+A2					
$C_{21}H_{30}O_2$	progesterone					
404	26.99	0	66.8	64.6	26.99	26.1
	4*A14+5*A15+2*A17+4*A16+3*A1+A19+A18*B18+A114+A35					
$C_{21}H_{30}O_3$	deoxycorticosterone					
414	27.98	0	67.59	69.4	27.98	28.9
	4*A14+5*A15+2*A17+4*A16+2*A1+A30*C30					
	+A19+A18*B18+2*A114+A35+A2					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$C_{21}H_{30}O_4$ 454	corticosterone 33.32	0	73.39	84.1	33.32	38.7 [219]
	$4*A14+5*A15+2*A17+5*A16+2*A1+2*A30*D30+A19+A18*B18+A114+A35+A2$					
$C_{21}H_{30}O_5$ 486	hydrocortisone 35.84	0	73.75	82.9	35.84	40.3 [219]
	$4*A14+5*A15+3*A17+4*A16+2*A1+3*A30*E30+A19+A18*B18+A114+A35+A2$					
$C_{21}H_{35}N_3N_2$ 362.7	N-palmitoyl-pyrazinamide 51.82	0	142.87	192.5	51.82	69.8 [261]
	$A1+14*A2*B2+3*A10+A12+2*A41+A71$					
$C_{21}H_{42}O$ 336.7	11-heneicosanone 76.2	0	226.31	207.7	76.2	69.9 [262]
	$2*A1+A35+18*A2*B2$					
$C_{21}H_{42}O$ 333.9	2-heneicosanone 77.65	0	232.55	207.7	77.65	69.4 [263]
	$2*A1+A35+18*A2*B2$					
$C_{21}H_{43}NO_3$ 349.1	N-hexadecyl-L-valine 29.1	83.36		177.5	83.9	65.1 [249]
366.6	54.8	149.48	232.84			
	$3*A1+A3+A3*B3+15*A2*B2+A44+A36*B36$					
$C_{21}H_{43}NO_3$ 375.1	N-hexadecyl-DL-valine 80.5	0	214.61	177.5	80.5	66.6 [249]
	$3*A1+A3+A3*B3+15*A2*B2+A44+A36*B36$					
$C_{21}H_{44}$ 305.7	n-heneicosane 15.48	50.65				
313.7	47.7	152.06	202.71	212.4	63.18	66.6 [216]
	$19*A2*B2+2*A1$					
$C_{22}H_{12}$ 554.2	1,12-benzoperylene 17.37	0	31.34	43.2	17.37	24.0 [215]
	$12*A10+4*A13+6*A12$					
$C_{22}H_{12}$ 435.2	o-phenylenepyrene 21.51	0	49.41	36.0	21.51	15.7 [264]
	$A14+2*A15+5*A19+12*A10+2*A13+3*A12$					
$C_{22}H_{14}$ 637.2	picene 35.19	0	55.22	44.0	35.19	28.0 [264]
	$14*A10+8*A12$					
$C_{22}H_{14}$ 553.5	1,2:3,4-dibenzanthracene 25.82	0	46.65	44.0	25.82	24.3 [215]
	$14*A10+8*A12$					
$C_{22}H_{14}$ 544.2	1,2:5,6-dibenzanthracene 31.16	0	57.26	44.0	31.16	23.9 [215]
	$14*A10+8*A12$					
$C_{22}H_{14}O_4$ 425.1	1,4-bis(phenylglyoxaloyl)benzene 32.3	0	75.98	92.2	32.3	39.2 [216]
	$14*A10+4*A12+4*A35$					
$C_{22}H_{18}F_2O$ 370	4-(6-hexenyloxy)-3',4'-difluorodiphenyldiacetylene 37.45	0	101.22	92.5	37.45	34.2 [195]
	$A5+A6+4*A2+5*A12+7*A10+4*A9+2*A24+A32$					
$C_{22}H_{18}F_2O$ 364.4	4-(cis-4-hexenyloxy)-3',4'-difluorodiphenyldiacetylene 35.32	0	96.93	90.9	35.32	33.1 [195]
	$A1+2*A6+3*A2+5*A12+7*A10+4*A9+2*A24+A32$					
$C_{22}H_{18}F_2O$ 364.6	4-(cis-3-hexenyloxy)-3',4'-difluorodiphenyldiacetylene 30.97	0	84.94	91.0	30.97	33.1 [195]
	$A1+2*A6+3*A2+5*A12+7*A10+4*A9+2*A24+A32$					
$C_{22}H_{19}Br_2NO_3$ 372.9	(S)- α -cyano-3-phenoxybenzyl (1R)-cis-3-(2,2-dibromovinyl)-2,2-dimethylcyclopropanecarboxylate 40.71	0	109.18	96.6	40.71	36.0 [221]
	$A14+A17+2*A16+2*A1+A6+A7+2*A21+A38+A3*B3+A56+2*A12+A11+9*A10+A32$					
$C_{22}H_{24}O_3$ 351.2	4-methyl-4-propyl-1,8-diphenyl-2,3,5-trioxabicyclo[4.3.0]non-7-ene 16.6	0	47.27	60.1	16.6	21.1 [257]
	$10*A10+A11+A12+2*A14+3*A15+A19+A18+A16+2*A17+A112+A113+2*A1+2*A2$					
$C_{22}H_{28}$ 414	1,1'-diphenyl-1,1'-bicyclopentyl 31.38	0	75.77	67.4	31.38	27.9 [289]
	$2*A14+4*A15+2*A17+2*A11+10*A10$					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{22}H_{28}N_2O_2$	394	(4R*,5'R*,5'R*)-5,5-diphenyl-3,3',4,4'-tetramethyl-2,2'-bioxazolidine 31.9	0	80.96	31.9	32.4 [185]
$C_{22}H_{28}N_2O_2$	368.8	2*A14+4*A15+6*A16+2*A119+10*A10+2*A11+4*A1+2*A112 (2R*,3R*,6R*,7R*)-2,6-diphenyl-3,4,7,8-tetramethyl- <i>cis</i> -perhydro-[1,4]-oxazino- [3,2- <i>b</i>][1,4]-oxazine 20.9	0	54.03	20.9	31.8 [185]
$C_{22}H_{28}N_2O_2$	379.4	2*14+4*A15+6*A16+2*A119+10*A10+2*A11+4*A1+2*A112 (2R*,3S*,6R*,7S*)-2,6-diphenyl-3,4,7,8-tetramethyl- <i>cis</i> -perhydro- [1,4]-oxazino-[3,2- <i>b</i>][1,4]-oxazine 18.4	0	48.5	18.4	31.2 [185]
$C_{22}H_{28}O_3$	480	2*A14+4*A15+6*A16+2*A119+10*A10+2*A11+4*A1+2*A112 19-nor-17 α -ethynyl-17 β -acetoxy-4-androsten-3-one 27.3	0	56.88	27.3	29.9 [216]
$C_{22}H_{29}FO_5$	539	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114+A38+2*A1+A8+A9 dexamethasone 42.02	0	77.97	42.02	42.5 [219]
$C_{22}H_{30}O_3$	393	4*A14+5*A15+2*A17+2*A17+4*A16+3*A1+3*A30*F30+A19 +2*A18*B18+A18+A114+A35+A2+A27 testosterone propionate 25.64	0	65.24	25.64	29.4
	393.2	22.13	0	56.3	22.1	29.4 [219, 396]
$C_{22}H_{40}O_4$	378.2	4*A14+5*A15+2*A17+4*A16+3*A1+A19+A18*B18+A114+A38+A2 1,10-cyclooctadecanedione <i>bis</i> ethylene ketal 33.56	0	88.72	33.56	37.4 [114]
$C_{22}H_{44}$	359.2	3*A14+17*A15+2*A17+4*A112 1,1,10,10-tetramethylcyclooctadecane 39.58	0	110.19	39.58	32.3 [110]
$C_{22}H_{44}N_2O_2$	415	A14+15*A15+4*A1+2*A17 N,N'- <i>di-n</i> -hexylsebacamide 53.56	0	129.29	53.56	85.5 [216]
$C_{22}H_{44}O_2$	288.4	18*A2*B2+2*A1+2*A60 butyl octadecanoate 2.22	7.7			
	299.7	37.48	125.05	132.75	39.7	66.0 [79]
$C_{22}H_{44}O_2$	313.5	2*A1+A38+19*A2*B2 ethyl eicosanoate 15.58	49.68			
	396.7	7.78	19.62	69.3	23.36	87.3 [216]
$C_{22}H_{44}O_4$	374	2*A1+A38+19*A2*B2 2,2,13,13-tetramethyl-1,3,10,12-tetraoxacyclodocosane 61.9	0	165.51	61.9	41.0 [47]
$C_{22}H_{45}Br$	303.8	A14+19*A15+4*A112+2*A17+4*A1 1-bromodocosane 23.14	76.15			
	317.2	44.98	141.8	217.95	231.1	73.3 [265]
$C_{22}H_{45}NO$	343.1	A1+21*A2*B2+A21 N-hexyl hexadecanamide 57	0	166.13	57	73.4 [217]
$C_{22}H_{45}NO_3$	367.1	2*A1+19*A2*B2+A60 N hexadecyl-L-leucine 46.1	0	125.58	46.1	68.6 [249]
$C_{22}H_{45}NO_3$	333.1	3*A1+15*A2*B2+A3+A3*B3+A2*B2+A44+A36*B36 N-hexadecyl-DL-leucine 4.3	12.91			
	355.1	60.6	170.66	183.57	64.9	66.3 [249]
$C_{22}H_{46}$	315.2	3*A1+15*A2*B2+A3+A3*B3+A2*B2+A44+A36*B36 <i>n</i> -docosane 29.51	93.62			
	316.1	47.84	151.36	245.0	77.3	70.3 [227]
$C_{23}H_{15}ClO_3$	416.5	2*A1+20*A2*B2 2-[(4-chlorophenyl)phenylacetyl]-1H-indene-13(2H)-dione 34.54	0	82.94	34.54	33.2 [221]
$C_{23}H_{21}F_3O$	394.8	A14+2*A15+2*A19+A16+13*A10+2*A114+A22*D22+A35+A12+ 2*A11+A3*B3 4- <i>n</i> -hexyloxy-4'-trifluoromethyldiphenylidiacetylene 33.98	0	86.07	33.98	35.9 [195]
$C_{23}H_{22}O_6$		A1+5*A2+A11+3*A12+8*A10+4*A9+3*A25+A4*B4+A32 [2R-(2a,6aa,12aa)]-1,2,12,12a-tetrahydro-8,9-dimethoxy-2-(1-methylethenyl)- [1]benzopyrano[3,4- <i>b</i>]furo[2,3- <i>h</i>][1]benzopyran-6(6aH)-one (Rotenone)				

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
437.9	35.64	0	81.39	90.3	35.64	39.6
$C_{23}H_{30}O_6$	3* A_{14} +6* A_{15} +2* A_{19} +4* A_{19} +3* A_{112} +3* A_{16} + A_{114} +4* A_{10} +2* A_{12} +3* A_1 +2* A_{32} + A_5 + A_7 prednisolone acetate	0				[221]
511	38.67	0	75.67	81.4	38.67	41.6
$C_{23}H_{30}O_6$	4* A_{14} +5* A_{15} +3* A_{17} +4* A_{16} +3* A_1 +2* A_{30} * E_{30} + A_{19} + 2* A_{18} * B_{18} + A_{18} + A_{114} + A_{35} + A_2 + A_{38} cortisone acetate	0				[219]
509	38.43	0	75.5	78.8	38.43	40.1
$C_{23}H_{32}O_2$	4* A_{14} +5* A_{15} +3* A_{17} +3* A_{16} +3* A_1 +2* A_{30} * E_{30} + A_{19} + 2* A_{18} * B_{18} +2* A_{114} + A_{35} + A_2 + A_{38} 3,3'-di- <i>tert</i> -butyl-5,5'-dimethyl-2,2-dihydroxydiphenylmethane	0	72.65	75.7	29.33	30.6
$C_{23}H_{32}O_3$	8* A_1 +2* A_4 +6* A_{11} +2* A_{12} + A_2 +2* A_{31} +4* A_{10} estra-1,3,5(10)-triene-3- α -17 β pentanoate	0	70	96.2	29.45	40.5
$C_{23}H_{32}O_3$	3* A_{14} +4* A_{15} +2* A_{19} +4* A_{16} + A_{17} + A_{38} + A_{31} +3* A_{10} + A_{12} +2* A_1 +3* A_2 testosterone butyrate	0				[137]
382	24.75	0	64.8	81.9	24.75	31.3
382.2	25.3	0	66.2	81.9	25.3	31.3
$C_{23}H_{32}O_4$	4* A_{14} +5* A_{15} +2* A_{17} +4* A_{16} +3* A_1 + A_{19} + A_{18} * B_{18} + A_{114} + A_{38} +2* A_2 deoxycorticosterone acetate	0	68.98	79.4	29.66	34.1
430	29.66	0	68.98	79.4	29.66	34.1
$C_{23}H_{32}O_6$	4* A_{14} +5* A_{15} +2* A_{17} +4* A_{16} +3* A_1 + A_{38} + A_{19} + A_{18} * B_{18} + A_{114} + A_{35} + A_2 hydrocortisone acetate	0	74.49	87.0	36.95	43.2
496	36.95	0	74.49	87.0	36.95	43.2
$C_{23}H_{44}O_5$	4* A_{14} +5* A_{15} +3* A_{17} +4* A_{16} +3* A_1 +2* A_{30} * E_{30} + A_{19} + A_{18} * B_{18} + A_{114} + A_{35} + A_2 + A_{38} 1-aceto-3-stearin	0	130.31	222.5	41.69	71.2
319.9	41.69	0	130.31	222.5	41.69	71.2
$C_{23}H_{46}O$	2* A_1 +16* A_2 * B_2 + A_3 * B_3 +2* A_{38} + A_{30} * B_{30} +2* A_2 12-tricosanone	0	228.04	226.4	78.03	77.5
342.2	78.03	0	228.04	226.4	78.03	77.5
$C_{23}H_{46}O_2$	2* A_1 + A_{35} +20* A_2 * B_2 methyl docosanoate	0	251.7	228.9	82.3	74.6
325.0	82.3	0	251.7	228.9	82.3	74.6
$C_{23}H_{48}$	2* A_1 + A_{38} +20* A_2 * B_2 <i>n</i> -tricosane	69.37	237.69	231.1	75.73	74.1
313.7	21.76	69.37	237.69	231.1	75.73	74.1
320.7	53.97	168.33	237.69	231.1	75.73	74.1
$C_{24}F_{50}$	2* A_1 +21* A_2 * B_2 perfluorodocosane	19.19	235.9	251.1	104.7	116.8
202.7	3.89	19.19	235.9	251.1	104.7	116.8
465.2	100.8	216.7	235.9	251.1	104.7	116.8
$C_{24}H_{12}$	44* A_{26} +24* A_4 * B_4 +6* A_{25} coronene	0	27.02	42.8	19.2	30.4
710.5	19.2	0	27.02	42.8	19.2	30.4
$C_{24}H_{14}$	12* A_{10} +6* A_{12} +6* A_{13} 1,2:4,5-dibenzopyrene	0	58.63	43.5	30.5	22.6
520.2	30.5	0	58.63	43.5	30.5	22.6
$C_{24}H_{14}$	14* A_{10} +8* A_{12} +2* A_{13} 3,4:9,10-dibenzopyrene	0	50.05	42.5	27.87	24.2
556.8	27.87	0	50.05	42.5	27.87	24.2
$C_{24}H_{14}$	14* A_{10} +8* A_{12} +2* A_{13} 1,2:3,4-dibenzopyrene.	0	49.24	42.5	24.68	21.8
501.2	24.68	0	49.24	42.5	24.68	21.8
$C_{24}H_{18}$	14* A_{10} +8* A_{12} +2* A_{13} 1, 3, 5-triphenylbenzene	0	74.89	88.6	33.4	39.5
446	33.4	0	74.89	88.6	33.4	39.5
$C_{24}H_{18}$	18* A_{10} +6* A_{12} <i>p</i> -quaterphenyl	1.78	66.15	88.6	38.2	52.0
233.0	0.41	1.78	66.15	88.6	38.2	52.0
587.2	37.8	64.37	66.15	88.6	38.2	52.0
$C_{24}H_{28}O_2Si_3$	18* A_{10} +6* A_{12} 1,1,1,5,5,5-hexamethyl-3,3-diphenyltrisiloxane	0	84.12	88.6	22.75	24.0
270.5	22.75	0	84.12	88.6	22.75	24.0
$C_{24}H_{31}FO_5$	6* A_1 +2* A_{32} +3* A_{109} +10* A_{10} +2* A_{11} triamcinolone acetonide	0				[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
566	45.29	0	80.02	78.6	45.29	44.5 [219]
$C_{24}H_{31}FO_6$	5*A14+5*A15+5*A17+4*A16+4*A1+2*A30*F30 +A19+2*A18*B18+A18+A114+A35+A2+A28+2*A112 dexamethasone acetate					
503	37.72	0	75	85.1	37.72	42.8 [219]
$C_{24}H_{32}$	4*A14+5*A15+4*A17+4*A16+4*A1+2*A30*F30 +A19+2*A18*B18+A18+A114+A35+A2+A28+A38 1,1'-diphenyl-1,1'-bicyclohexyl					
455	29.71	0	65.27	74.8	29.71	34.0 [289]
$C_{24}H_{34}$	2*A14+6*A15+2*A17+2*A11+10*A10 1,1-diphenyldecane					
191	1.92	10.08				
281.4	38.83	137.98	148.11	127.5	40.75	35.9 [216]
$C_{24}H_{34}O_3$	10*A10+10*A2+A1+2*A11+A3 testosterone valerate					
380	24.57	0	64.66	89.1	24.57	33.8
380.2	30.96	0	81.45	89.1	31.0	33.8 [219, 396]
$C_{24}H_{40}$	4*A14+5*A15+2*A17+4*A16+3*A1+A19+A18*B18+A114+A38+3*A2 1-cyclohexyl-1-phenyldecane					
275.8	35.17	0	127.58	129.8	35.17	35.8 [216]
$C_{24}H_{44}O_4$	A14+3*A15+5*A10+A3+10*A2+A16+A1+A11 1,11-cycloicosanedione-bis-ethylene ketal					
362.2	43.72	0	120.71	106.2	43.72	38.5 [114]
$C_{24}H_{46}$	3*A14+19*A15+4*A112+2*A17 2,11-dicyclohexyldecane					
300.6	43.93	0	146.15	119	43.93	35.77 [216]
$C_{24}H_{46}$	2*A14+2*A1+2*A16+8*A2+6*A15+2*A3 1,1-dicyclohexyldecane					
300.6	44.35	0	147.54	132.1	44.35	39.7 [216]
$C_{24}H_{48}$	2*A14+2*A16+A1+10*A2+6*A15+A3 cyclotetraeicosane					
297	38	127.95				
322	10.8	33.54	161.49	111.1	48.8	35.8 [181]
$C_{24}H_{48}O_2$	21*A15+A14 ethyl docosanoate					
312.3	9.58	30.68				
321.0	19.16	59.69	90.37	236.6	28.74	75.9 [216]
$C_{24}H_{50}$	2*A1+A38+20*A2*B2+A2 n-tetracosane					
321.3	31.3	97.42				
324.1	54.89	169.37	266.79	240.4	86.19	77.9 [216]
$C_{25}H_{31}FO_8$	22*A2*B2+2*A1 triamcinolone					
543	42.56	0	78.39	86.6	42.56	47.0 [219]
$C_{25}H_{34}O_3$	4*A14+5*A15+4*A17+4*A16+2*A1+4*A30*F30 +A19+2*A18*B18+A18+A114+A35+A2+A28 19-nor-17 α -ethynyl-17 β -(2,2-dimethylpropionyloxy-4-androsten-3-one)					
500	37.8	0	75.6	74.5	37.8	37.3 [216]
$C_{25}H_{40}O_2Si_2$	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114 +A38+4*A1+A8+A9+A4*B4 norethindrone pentamethyldisiloxy ether					
355	22.9	0	64.51	80.1	22.9	28.4 [216]
$C_{25}H_{46}O_6$	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114 +6*A1+A8+A9+2*A32+2*A109 1,2-diaceto-3-sterin					
208.3	45.56	0	218.72	229.7	45.56	47.8 [216]
$C_{25}H_{52}$	3*A1+2*A2+3*A38+A3*B3+16*A2*B2 n-pentacosane					
320.2	26.07	81.42				
326.7	57.74	176.76	258.18	249.8	83.81	81.6 [216]
$C_{25}H_{52}$	2*A1+23*A2*B2 5,5-bis(3,3-dimethylbutyl)-2,2,8,8-tetramethylnonane					
472.7	48.53	0	102.67	93.8	48.53	44.4 [266]
$C_{26}H_{14}$	4*(3*A1+A4+2*A2)+A4 1,12-phenyleneperylene					
541.5	17.28	0	31.91	43.1	17.28	23.3 [215]
$C_{26}H_{18}$	14*A10+8*A12+4*A13 9,9'-bifluorenyl					
519.2	36.9	0	71.07	72.6	36.9	37.7

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}}H_{\text{tpcc}}$ (calcd)
$C_{26}H_{26}OSi_2$	2*A14+4*A15+8*A19+2*A16+16*A10 1,3-dimethyl-1,1,3,3-tetraphenyldisiloxane					[252]
322.0	26.58	0	82.55	95.6	26.58	30.8
$C_{26}H_{26}O_3Si_3$	20*A10+4*A11+2*A1+A32*C32+2*A109 dimethyltetraphenylcyclotrisiloxane					[216]
361.1	28.2	0	78.1	89.1	28.2	32.2
$C_{26}H_{38}$	A14+3*A15+3*A112+3*A139+2*A1+20*A10+4*A11 2,3-dimethyl-2,3-bis(4- <i>tert</i> -butylphenyl)butane					[216]
493	43.93	0	89.11	57.4	43.93	28.3
$C_{26}H_{46}$	4*A11+10*A1+4*A4+8*A10 11-phenyleicosane					[289]
294.3	64.77	0	220.08	187.3	64.77	55.1
$C_{26}H_{52}$	2*A1+17*A2*B2+A3+5*A10+A11 11-cyclohexyleicosane					[290]
269.9	48.7	0	180.44	189.6	48.7	51.2
$C_{26}H_{52}$	A14+A16+2*A1+9*A2*B2+3*A15+A3+8*A2 1,1,4,4,10,10,13,13-octamethylcyclooctadecane					[290]
427.2	6.74	15.77				
438.2	20.17	46.02	61.79	91.2	26.9	40.0
$C_{26}H_{52}O_2$	A14+15*A15+8*A1+4*A17 ethyl tetracosanate					[107, 116]
317.7	11.2	35.27				
327.4	22.94	70.07	105.34	255.2	34.14	83.6
$C_{26}H_{53}NO$	2*A1+22*A2*B2+A38+A2 N-decyl hexadecanamide					[216]
333	5	15.02				
347	63	181.56	196.57	251.2	68	87.2
$C_{26}H_{54}$	2*A1+23*A2*B2+A60 <i>n</i> -hexacosane					[260]
326.5	32.2	98.7				
329.5	59.5	180.6	289.3	259.1	95.3	85.3
$C_{27}H_{30}O_3$	2A1+24*A2*B2 19-nor-17 α -ethynyl-17 β -(benzoyloxy-4-androsten-3-one)					[216]
531	41.5	0	78.15	74.2	41.5	39.4
$C_{27}H_{32}O_3$	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114+A38+A1+ 5*A10+A12+A8+A9 spiro[8.5.0(3,7)]-3,5-diphenyl-1,2,8-trioxa-10,12-tetramethyltetradec-5-ene					[216]
389.2	15	0	38.54	52.7	15	20.5
$C_{27}H_{38}O_3$	3*A14+5*A15+4*A17+10*A10+A11+A12+4*A1+A19+A18+A16+A113+A112 19-nor-17 α -ethynyl-17 β -(heptanoyloxy-4-androsten-3-one)					[257]
340	21.6	0	63.53	97.8	21.6	33.3
$C_{27}H_{46}O$	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114 +A38+2*A1+5*A2+A8+A9 cholesterol					[216]
304.8	2.5	8.2				
420.2	27.41	65.22	73.42	73.7	29.91	31.0
$C_{27}H_{54}N_6$	4*A14+5*A15+5*A16+2*A17+5*A1+3*A2+2*A3+A30+A19+A18 <i>tris</i> N,N-diisobutylamino-1,3,5-triazine					[216, 388]
372.6	35.81	0	96.11	99.1	35.81	36.9
$C_{27}H_{56}$	12*A1+6*A3+6*A2+3*A43+3*A12+3*A41 <i>n</i> -heptacosane					[267]
318	2.26	7.11				
325.4	26.28	80.75				
332.1	59.05	177.82	265.68	268.4	87.59	89.2
$C_{28}H_{16}$	2*A1+25*A2*B2 1,2,4,5,8,9-tribenzopyrene					[268]
608	28.8	0	47.37	43.4	28.8	26.4
$C_{28}H_{28}O_2P_2$	16*A10+10*A12+2*A13 1,4-bis(diphenylphosphino)butane					[264]
405.9	45.3	0	111.6	105.4	45.3	42.8
$C_{28}H_{32}O_4Si_4$	2*A72+20*A10+4*A12+4*A2 1,1,3,3-tetramethyl-5,5,7,7-tetraphenylcyclotetrasiloxane					[269]
186.5	0.24	1.3				
271.5	1.05	3.85				
346.2	27.05	78.13	83.29	98.2	28.34	34.0
	4*A1+20*A10+4*A11+4*A139+4*A112+A14+5*A15					[216]

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{fus} S_{pcc}$ (expt)	$\Delta_0^{fus} S_{pcc}$ (calcd)	$\Delta_0^{fus} H_{pcc}$ (expt)	$\Delta_0^{fus} H_{pcc}$ (calcd)	
$C_{28}H_{32}O_4Si_4$	373.4	1,1,5',7'-tetramethyl-1',3',5,7-tetraphenylcyclotetrasiloxane 24.62	0	65.93	98.2	24.62	36.7 [216]
$C_{28}H_{40}$	432	4*A1+20*A10+4*A11+4*A139+4*A112+A14+5*A15 1,1'-diphenyl-1,1'-bicyclooctyl 35.98	0	83.26	89.6	35.98	38.7 [289]
$C_{28}H_{36}$	439.2	2*A14+10*A15+2*A17+2*A11+10*A10 1,1,5,5,11,11,15,15-octamethylcycloeoicosane 47.7	0	108.6	98.6	47.7	43.3 [107]
$C_{28}H_{56}O_2$	322.7 322.7	A14+17*A15+8*A1+4*A17 ethyl hexacosanoate 13.22 27.05	40.98 83.81	124.79	273.9	40.27	88.4 [216]
$C_{28}H_{58}$	331.3 334.5	2*A1+A38+24*A2*B2+A2 <i>n</i> -octacosane 35.44 64.64	106.98 193.28	300.26	277.8	100.08	92.9 [216]
$C_{29}H_{35}FO_{10}$	508	26*A2*B2+2*A1 triamcinolone diacetate 38.31	0	75.42	98.7	38.31	50.1 [219]
$C_{29}H_{44}O_2$	447.7	4*A14+5*A15+4*A17+4*A16+6*A1+A19 +2*A18*B18+A18+A114+A35+A2+A28+4*A38 3,3',5,5'-tetra- <i>tert</i> -butyldiphenylmethane-4,4'-diol 42.97	0	95.98	76.4	42.97	34.2 [101]
$C_{29}H_{60}$	331.4 336.6	12*A1+4*A4+6*A11+2*A12+A2+4*A10+2*A31 <i>n</i> -nonacosane 29.71 66.11	89.65 196.43	286.08	287.1	95.81	96.6 [216]
$C_{30}H_{46}$	400	2*A1+27*A2*B2 3,4-diethyl-3,4-bis(4- <i>tert</i> -butylphenyl)hexane 29.71	0	74.27	85.9	29.71	34.4 [289]
$C_{30}H_{46}O_2S$	417.2	10*A1+4*A4+4*A11+8*A10+4*A2 <i>bis</i> -[3,5-di- <i>tert</i> -butyl-4-hydroxybenzyl]sulfide 43.1	0	103.3	85.6	43.1	35.7 [101]
$C_{30}H_{60}O_4$	406.4	12*A1+4*A4+6*A11+2*A12+2*A2+2*A31+A84+4*A10 2,2,6,6,9,9,13,13,17,17,20,20-dodecamethyl-1,3,12,14-tetraoxacyclodocosane 57.3	0	141	112.1	57.3	45.5 [47]
$C_{30}H_{60}$	411.2	A14+19*A15*+4*A112+6*A17+12*A1 1,1,4,4,12,12,15,15-octamethylcyclodocosane 58.58	0	142.45	106.0	58.58	43.6 [107]
$C_{30}H_{61}Br$	313.2 339.6	A14+19*A15+4*A17+8*A1 1-bromotricosane 23.85 79.5	76.15 234.09	310.24	305.7	103.34	103.8 [265]
$C_{30}H_{62}$	335.3 338.7	A1+29*A2*B2+A21 triacontane 37.49 68.83	111.82 203.24	315.06	296.4	106.32	100.4 [216]
$C_{31}H_{44}O_2$	400.7	2*A1+28*A2*B2 3,3'- <i>bis</i> -(1-cyclohexylethyl)-5,5'-dimethyldiphenylmethane-2,2'-diol 29.29	0	73.09	101.8	29.29	40.8 [101]
$C_{31}H_{64}$	282.3	6*A11+2*A12+4*A10+2*A31+4*A1+2*A3+2*A14 +6*A15+2*A16+A2 11- <i>n</i> -decylheneicosane 71.13	0	251.96	288.3	71.13	81.4 [216]
$C_{32}H_{14}$	729 770.1	3*A1+27*A2*B2+A3 ovalene 8.08 17.4	11.08 22.59	33.68	41.4	25.48	32.2 [264]
$C_{32}H_{39}ClO_2$	413	14*A10+8*A12+10*A13 norethindrone-6-(4-chlorophenyl)hexanoate 28.8	0	69.73	109.1	28.8	45.0 [216]
$C_{32}H_{64}O_2$	334.7 341.5	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114+ A38+A1+4*A10+A11+A12+A8+A9+5*A2+A22*B22 ethyl triacontanoate 16.2 36.07	48.4 105.65	154.04	311.2	52.27	106.3 [216]
$C_{32}H_{64}O_4$	342.5	2*A1+A38+28*A2*B2+A2 2,2,6,6,10,10,14,14,18,18,22,22-dodecamethyl-1,3,13,15-tetraoxacyclotetracosane 39.7	0	115.9	119.5	39.7	40.9

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
	A14+21*A15+4*A112+6*A17+12*A1					
$C_{32}H_{66}$	dotriacontane					
338.7	41.38	122.18				[47]
343.5	76.57	222.93	345.11	315.1	117.94	108.2
	2*A1+30*A2*B2					
$C_{33}H_{34}O_3$	norethindrone-biphenyl-4-carboxylate					
462	31.6	0	68.4	88.8	31.6	41.0
	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114+A38+A1+9*A10+3*A12+A8+A9					
$C_{33}H_{40}O_3$	norethindrone-4-cyclohexylbenzoate					
482	38.6	0	80.08	87.0	38.6	41.9
	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114+A38					
$C_{33}H_{48}O_3$	norethindrone- <i>trans</i> -3-(4-butylcyclohexyl)propionate					
374	22.5	0	60.16	112.9	22.5	42.2
	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114+A38					
$C_{33}H_{48}O_3^*$	norethindrone- <i>trans</i> -4-hexylcyclohexylcarboxylate					
398	22.6	0	56.78	112.9	22.6	44.9
	4*A14+5*A15+4*A16+A19+2*A17+A18*B18+A114					
$C_{33}H_{68}$	triacontane					
344	105.0	0	305.2	324.4	105.0	111.6
	2*A1+31*A2*B2					
$C_{34}H_{54}$	4,5-dipropyl-4,5-bis(4- <i>tert</i> -butylphenyl)octane					
419	40.58	0	96.86	114.4	40.58	47.9
	4*A4+4*A11+10*A1+8*A2+8*A10					
C_3H_7O	tetracontane					
341.5	29.3	85.8				
345.9	79.96	231.1	316.96	333.7	109.24	115.4
	2*A1+32*A2*B2					
$C_{35}H_{72}$	<i>n</i> -pentatriacontane					
344.7	41.09	119.2				
347.2	86.4	248.85	368.04	343.1	127.49	119.1
	2*A1+33*A2*B2					
$C_{36}H_{18}$	decacyclene					
666	25.4	0	38.14	49.4	25.4	32.9
	3*A14+6*A15+15*A19+3*A12+18*A10					
$C_{36}H_{24}$	1,3,5-tri- α -naphthylbenzene					
472	42.26	0	89.53	88.2	42.26	41.6
	24*A10+12*A12					
$C_{36}H_{46}O_4$	4',4'-didecanoyloxydiphenyl diacetylene					
308	44.9	145.78				
403	42.2	104.71	250.49	183.0	42.2	73.7
	8*A10+4*A12+4*A9+2*A38+16*A2+2*A1					
$C_{36}H_{74}$	hexatriacontane					
345.4	9.92	28.71				
347.1	30.54	88.01				
349.2	88.83	254.41	371.13	352.4	129.29	123.04
	2*A1+34*A2*B2					
$C_{38}H_{50}O_4$	4,4'-diundecanoyloxydiphenyl diacetylene					
339	18.1	53.39				
358	7.59	21.14				
399	36.2	90.73	165.26	197.2	61.59	78.7
	8*A10+4*A12+4*A9+2*A38+18*A2+2*A1					
$C_{38}H_{62}$	5,6-dibutyl-5,6-bis(4- <i>tert</i> -butylphenyl)decane					
386	43.1	0	111.65	143.0	43.1	55.2
	4*A4+4*A11+10*A1+12*A2+8*A10					
$C_{39}H_{74}O_6$	glyceryl trilaurate					
319.5	123.51	0	386.57	360.3	123.51	115.1
	3*A1+30*A2*B2+2*A2+A3*B3+3*A38					
$C_{40}H_{54}O_4$	4,4'-didodecanoyloxydiphenyl diacetylene					
374	50.2	134.22				
401	44	109.73	243.95	255.4	94.2	102.4
	8*A10+4*A12+4*A9+2*A38+20*A2*B2+2*A1					
$C_{40}H_{82}$	tetracontane					
345.4	14.02	40.58				
353.2	133.44	377.82	418.4	389.7	147.46	137.7
	2*A1+38*A2*B2					

Table 5. Experimental and calculated total phase change enthalpy and entropy of database—Continued

$T(K)$	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$C_{42}H_{66}O_{12}$	benzene-hexa- <i>n</i> -hexanoate					
251.6	25.67	102.02				
291.5	12.27	42.11				
348.3	162.59	466.85				
368.7	33.5	90.85	701.82	277.8	234.03	102.5 [216]
$C_{45}H_{86}O_6$	6*A12+6*A38+6*A1+24*A2 glyceryl trimyristate					
330.2	152.2	0	460.92	416.3	152.2	137.5 [216]
$C_{51}H_{98}O_6$	3*A1+36*A2*B2+2*A2+A3*B3+3*A38 glyceryl tripalmitate					
338.9	179.37	0	529.27	472.3	179.37	160.1 [216]
$C_{51}H_{100}ClN_5$	3*A1+42*A2*B2+2*A2+A3*B3+3*A38 2,4-bis-N,N-didodecylamino-6-chloro-1,3,5-triazine					
307.5	34.25	0	111.4	441.0	34.25	135.6 [267]
$C_{51}H_{102}N_6$	4*A1+44*A2*B2*+2*A43+A22*F22+3*A12+A41 tris N,N-dioctylamino-1,3,5-triazine					
312.7	74.25	0	237.45	440.9	74.25	137.9 [267]
$C_{57}H_{110}O_6$	6*A1+42*A2*B2+3*A41+3*A43+3*A12 glyceryl tristearate					
345.7	203.26	0	587.97	532.6	203.26	184.1 [216]
$C_{60}H_{78}Sn_2O$	3*A1+48*A2*B2+2*A2*B2+A3*B3+3*A38 hexakis(2-methyl-2-phenylpropyl)distanoxane					
417.7	71.81	0	171.92	165.8	71.81	69.3 [221]
$C_{60}H_{122}$	12*A1+6*A2+6*A4+30*A10+6*A11+2*A110+A32 hexacontane					
373.2	186.8	0	500.41	576.4	186.8	215.1 [268]
$C_{63}H_{126}N_6$	2*A1+58*A2*B2 tris N,N-didodecylamino-1,3,5-triazine					
314.4	87.68	0	278.88	434.2	87.68	136.5 [267]
$C_{75}H_{150}N_6$	6*A1+54*A2+3*A41+3*A43+3*A12 tris N,N-didodecylamino-1,3,5-triazine					
320.3	119.19	0	372.12	519.8	119.19	166.5 [267]
$C_{78}H_{108}$	6*A1+66*A2+3*A41+3*A43+3*A12 2,3,6,7,10,11-hexakis(1-decynyl)triphenylene					
314.2	63	0	200.54	326.6	63	102.6 [216]
	6*7*A2+6*A1+6*2*A9+6*A10+12*A12					

^aUnits for $\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ and $\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ are $J \cdot mol^{-1} \cdot K^{-1}$ and $kJ \cdot mol^{-1}$, respectively; compounds with molecular formulas characterized with an asterisk (*) were not included in generating the statistics. As noted in the text, some of these compounds exhibit liquid crystal behavior, others display amphiphilic behavior, group values for some are not currently available or the error between experimental and calculated total phase change entropy exceeded 3 standard deviations.

TABLE 6. Experimental and calculated total phase change enthalpies and entropies of fusion of polymers^a

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
CF ₂ *	297	850	2.86			
	605	4100	6.78	9.64	9.83	5.95
		A4*B4 + 2*A26				[389]
CH ₂ *	414.6	4.11	0	9.91	8.4	3.86
		A2*B2				[389]
CH ₂ O*	457.5	9.79	0	21.4	14.0	6.41
		A2*B2 + A32				[389]
C ₂ ClF ₃ *	493	5.02	0	10.2	15.8	7.8
		2*A4*B4 + 2*A26 + A27 + A22*F22				[389]
C ₃ HF ₃ *	495.2	5.44	0	11.0	12.4	6.3
		2*A4*B4 + 2*A26 + A27				[389]
C ₂ H ₂ F ₂ *	483	6.70	0	13.9	16.9	8.18
		A2 + A4*B4 + 2*A26				[389]
C ₂ H ₂ O ₂ *	501	9.74	0	19.4	17.0	7.4
		A2*B2 + A38				[389]
C ₂ H ₃ Cl*	546	11.0	0	20.1	13.5	7.3
		A2 + A3*B3 + A22*C22				[389]
C ₂ H ₃ F*	473	7.54	0	15.0	10.0	5.0
		A2 + A3*B3 + A27				[389]
C ₂ H ₄ O*	342	8.66	0	25.3	23.3	8.0
		2*A2*B2 + A32				[389]
C ₂ H ₄ O*	538	7.11	0	13.2	13.8	7.4
		A2 + A3*B3 + A30*C30				[389]
C ₂ H ₄ O ₂ *	501	11.75	0	23.5	17.0	8.5
		A3*B3 + A38				[216]
C ₃ H ₃ N*	590	5.0	0	8.5	15.0	8.8
		A2 + A3*B3 + A56				[390]
C ₃ H ₄ O ₂ *	366	10.9	0	29.8	26.3	9.6
		2*A2*B2 + A38				[389]
C ₃ H ₆ *	460.7	8.70	0	18.9	8.3	3.8
		A1 + A2 + A3				[389]
C ₃ H ₆ O*	348	8.40	0	24.1	19.6	6.8
		A1 + A2 + A3*B3 + A32				[389]
C ₃ H ₆ O*	308	9.44	0	30.6	32.6	10.0
		3*A2*B2 + A32				[389]
C ₃ H ₆ O ₂ *	328	16.7	0	50.9	35.1	11.5
		2*A2*B2 + A2 + 2*A32				[389]
C ₄ H ₃ F ₃ O ₂ *	448	7.5	0	16.7	31.2	14.0
		A2 + A3*B3 + A4*B4 + A38 + 3*A25				[390]
C ₄ H ₄ O ₄ *						

TABLE 6. Experimental and calculated total phase change enthalpies and entropies of fusion of polymers—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
450	23.0 2A2*B2 + 2*A38 poly(γ -butyrolactone)	0	51.1	34.0	23.0	15.3 [389]
C ₄ H ₄ O ₂ *	337.5	14.0 3*A2*B2 + A38 <i>trans</i> -poly(chloroprene)	0	41.5	35.6	14.0 [389]
C ₄ H ₅ Cl*	368	8.4 2*A2 + A6 + A7 + A22*C22 <i>cis</i> -poly(1,4-butadiene)	0	22.8	25.0	8.4 [390]
C ₄ H ₆ *	284.7	9.2 2*A2 + 2*A6 <i>trans</i> -poly(1,4-butadiene)	0	32.3	29.2	9.2 [389]
C ₄ H ₆ *	356	7.8	21.9			
	437	3.73 2*A2 + 2*A6 poly(1-butene)	8.5	30.5	29.2	11.5 [389]
C ₄ H*	411	7.0 A1 + 2*A2 + A3 poly(isobutylene)	0	17.0	15.4	7.0 [389]
C ₄ H ₈ *	317	12.0 2A1 + A2 + A4 poly(oxytetramethylene)	0	37.9	7.5	12.0 [389]
C ₄ H ₈ O*	330	14.4 4*A2*B2 + A32 <i>cis</i> -poly(isoprene)	0	43.6	41.9	14.4 [389]
C ₅ H ₈ *	301.2	4.35 A1 + 2*A2 + A6 + A7 <i>trans</i> -poly(isoprene)	0	14.4	26.4	4.35 [389]
C ₅ H ₈ *	347	12.8 A1 + 2*A2 + A6 + A7 poly(2,2-bis(chloromethyl)trimethylene-3-oxide)	0	36.9	26.4	12.8 [389]
C ₅ H ₈ Cl ₂ O*	463	32.0 4*A2 + A4 + 2*A22*D22 + A32 poly(methyl methacrylate)	0	69.1	30.7	32.0 [389]
C ₅ H ₈ O ₂ *	450	9.5 2*A1 + A2 + A4*B4 + A38 poly(2,2-dimethylpropiolactone)	0	21.1	27.0	9.5 [389]
C ₅ H ₈ O ₂ *	513	14.9 2*A1 + A2 + A4*B4 + A38 poly(1-pentene)	0	29.0	18.4	14.9 [389]
C ₅ H ₁₀ *	403.2	6.3 A1 + 2*A2*B2 + A3 poly(oxyethyleneoxytetramethylene)	0	15.6	26.9	6.3 [389]
C ₅ H ₁₀ O ₂ *	296	14.3 A2 + 2*A2*B2 + 2*A32 poly(oxy 1,4 phenylene)	0	48.3	53.7	14.3 [389]
C ₆ H ₄ O*	535	7.82 4*A10 + 2*A12 + A32 poly(thio-1,4-phenylene)	0	14.6	19.3	7.8 [389]
C ₆ H ₄ S*	593	8.65 4*A10 + 2*A12 + A84 poly(ϵ -caprolactone)	0	14.6	15.4	8.65 [389]
C ₆ H ₁₀ O ₂ *	342.2	17.9 4*A2*B2 + A38 poly(isopropyl acrylate)	0	52.3	54.2	17.9 [389]
C ₆ H ₁₀ O ₂ *	421	5.9 2*A1 + 2*A3*B3 + A2 + A38 nylon-6 (poly(6-aminohexanoic acid))	0	14.0	30.3	5.9 [389]
C ₆ H ₁₁ NO*	533	26.0 5*A2*B2 + A61 poly(4-methyl-1-pentene)	0	48.8	48.0	26.0 [389]
C ₆ H ₁₂ *	523	10.0 2A1 + 2*A2 + A3	0	19.0	16.6	10.0 [389]

TABLE 6. Experimental and calculated total phase change enthalpies and entropies of fusion of polymers—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)		
C_8H_8^*	516.2	poly(styrene)	10.0	0	19.4	18.1	10.0	9.3
		$A2+A3+5*A10+A11$						
C_8H_8^*	504	poly(xylylene)	5.0	9.9				
			1.5	2.7				
			10.0	14.0	26.6	24.6	16.5	17.5
$\text{C}_8\text{H}_8\text{O}^*$	580	$4*A10+2*A11+2*A2$						[389]
		poly(2,6-dimethyl-1,4-diphenylene oxide)	5.95	0	10.3	20.5	5.95	11.9
$\text{C}_8\text{H}_{12}\text{O}_4^*$	320	$2*A1+2*A10+2*A11+2*A12+A32$						[389]
		polyester-2,6 (poly(ethylene adipate))	15.9	0	49.7	15.9		
$\text{C}_8\text{H}_{12}\text{O}_4^*$	338		21.0	0	62.1	66.8	21.0	22.6
		$4*A2*B2+2*A2+2*A38$						
$\text{C}_8\text{H}_{16}\text{O}^*$	347	poly(oxyoctamethylene)	29.3	0	84.4	79.1	29.3	27.5
		$8*A2*B2+A32$						
$\text{C}_{10}\text{H}_8\text{O}_4^*$	553	poly(ethylene terephthalate)	26.9	0	48.6	44.2	26.9	24.4
		$2*A2+4*A10+2*A12+2*A38$						
$\text{C}_{11}\text{H}_{20}\text{O}_2^*$	365	poly(undecanolactone)	39.5	0	108.2	100.7	39.7	36.8
		$10*A2*B2+A38$						
$\text{C}_{11}\text{H}_{21}\text{NO}^*$	493	nylon-11 (poly(11-aminoundecanoic acid))	44.7	0	90.7	94.5	44.7	46.6
		$10*A2*B2+A61$						
$\text{C}_{12}\text{H}_{12}\text{O}_4^*$	518	poly(tetramethylene terephthalate)	32.0	0	61.8	58.4	32.0	30.3
		$4*A2+4*A10+2*A12+2*A38$						
$\text{C}_{12}\text{H}_{20}\text{O}_4^*$	356.2	polyester-2,10 (poly(ethylene decanedioate))	31.9	0	89.6	104.0	31.9	37.0
		$8*A2*B2+2*A2+2*A38$						
$\text{C}_{12}\text{H}_{22}\text{N}_2\text{O}_2^*$	574	nylon-6,6, α (poly(hexamethylene hexanediamide))	57.8	0	100.6	96.0	57.8	55.1
		$10*A2*B2+A61$						
$\text{C}_{12}\text{H}_{23}\text{NO}^*$	500	poly(12-aminododecanoic acid)	48.4	0	96.8	103.8	48.4	51.9
		$11*A2*B2+A61$						
$\text{C}_{13}\text{H}_{24}\text{O}_2^*$	368	poly(tridecanolactone)	50.6	0	137.5	119.3	50.6	43.9
		$12*A2*B2+A38$						
$\text{C}_{14}\text{H}_{10}\text{O}_4^*$	610	poly(ethylene naphthalene-2,6-dicarboxylate)	25.0	0	41.0	44.0	25.0	26.8
		$2*A2+6*A10+2*A12+2*A38$						
$\text{C}_{14}\text{H}_{16}\text{O}_4^*$	434	poly(hexamethylene terephthalate)	35.0	0	80.6	72.6	35.0	31.5
		$6*A2+4*A10+2*A12+2*A38$						
$\text{C}_{15}\text{H}_{28}\text{N}_2\text{O}_2^*$	500	nylon-6,9 poly(hexamethylene nonanediamide)	69.0	0	138.0	123.9	69.0	62.0
		$14*A2*B2+2*A61$						
$\text{C}_{15}\text{H}_{28}\text{O}_2^*$	370.5	poly(pentadecanolactone)	63.4	0	171.1	139.7	63.4	51.1
		$14*A2*B2+A38$						
$\text{C}_{16}\text{H}_{28}\text{O}_4^*$	343	poly(decamethylene adipate)	42.7	0	124.4	145.6	42.7	49.9
		$14*A2*B2+2*A38$						
$\text{C}_{16}\text{H}_{30}\text{N}_2\text{O}_2^*$	506	nylon-6,10 (poly(hexamethylene decanediamide))	71.7	0	141.6	133.2	71.7	67.4
		$14*A2*B2+2*A61$						
$\text{C}_{18}\text{H}_{12}\text{O}^*$	753	poly(oxy-2,6-diphenyl-1,4-diphenylene)	12.2	0	16.2	48.5	12.2	36.5
			87.	0	113.0	48.4	87.0	37.4

TABLE 6. Experimental and calculated total phase change enthalpies and entropies of fusion of polymers—Continued

T (K)		ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$ (calcd)
	$12^*A10+6^*A12+A32$						[389, 390]
$C_{18}H_{32}O_4^*$ 338		43.2	0	127.8	164.2	43.2	55.5
	$16^*A2^*B2+2^*A38$						[389]
$C_{18}H_{24}O_4^*$ 411		46.1	0	112.1	123.0	46.1	50.6
	$10A2^*B2+4^*A10+2^*A38$						[389]
$C_{18}H_{32}N_2O_2^*$ 520		80.1	0	154.0	151.8	80.1	78.9
	$16^*A2^*B2+2^*A61$						[389]
$C_{19}H_{12}O_3^*$ 668.2		37.4	0	56.0	57.8	37.4	38.6
	$12^*A10+6^*A12+2^*A32+A35$						[389]
$C_{20}H_{36}O_4^*$ 353		50.2	0	142.2	182.8	50.2	64.5
	$18^*A2^*B2+2^*A38$						[389]

^aThese compounds were not included in generating the statistics.

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids^a

T (K)	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (expt)	$\Delta_0^{T_{fus}} S_{tpce}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpce}$ (expt)	$\Delta_0^{T_{fus}} H_{tpce}$ (calcd)
CBr ₂ Cl ₂	dibromodichloromethane					
	258.8	5.4	21.0			
	294.4	2.3	7.9	28.8	44.4	13.1
CHBr ₃	tribromomethane					
	281.5	11.1		39.4	42.7	12.0
CH ₂ O*	formic acid					
	281.4	12.68		45.1	12.7	
CH ₂ F ₂	Group value not available					
	136.4	4.4		32.0	39.9	5.4
CH ₂ I ₂	Group value not available					
	279.2	12.1		43.2	45.9	12.8
CH ₃ Br	bromomethane					
	179.5	6.0		33.3	35.1	6.3
CH ₃ I	iodomethane					
	206.8	9.1		44.1	37.0	7.7
CH ₄ *	methane					
	90.7	0.94		10.4	0.94	
CH ₄ N ₂ O*	Group value not available					
	408.1	12.9		31.7	12.9	
	406.5	14.8		36.4	14.8	
CH ₄ N ₂ S*	Group value not available					
	452.2	12.6		27.8	12.6	
CH ₆ BrN	methylammonium bromide					
	397.7	1.6	4.0			
	488.4	3.51	7.2			
COS*	Group value not available					
	134.3	4.73		35.2	4.73	
CS ₂ *	Group value not available					
	161.1	4.39		27.2	4.39	
CSe ₂ *	Group value not available					
	229.5	6.36		27.7	6.36	
C ₂ Cl ₃ F ₃ *	Group value not available					
	287.5	4.11		14.3	4.11	
C ₂ Cl ₄ F ₂ *	Group value not available					
	314.2	4.0		12.7	51.7	16.2
C ₂ F ₄ O*	Group value not available					
	113.7	4.87		42.8	4.7	
C ₂ F ₂ O ₂ *	Group value not available					
	260.7	13.4		51.4	13.4	
C ₂ HF ₅	Group value not available					
	172.6	2.25		13.0	39.9	6.8

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)	
	A4*B4+A3*B3+3*A25+2*A26						[306]
C ₂ H ₂	142.7	2.54	17.8				
	192.4	3.76	19.5	37.3	29.8	6.3	
	acetylene						
	2*A8						[322]
C ₂ H ₂ AsCl ₃	270.7	trans-β-(chlorovinyl)dichloroarsine		63.2	50.1	17.1	
		17.1					
	A98+2*A6*B6+3*A22*D22						[369]
C ₂ H ₃ N ₃ O ₃	311.7	1,1,1-trinitroethane					
	329.2	4.6	14.77	50.4	47.7	16.3	
		11.7	35.60				
	A1+3*A50+A4*B4						[329, 352]
C ₂ H ₃ N ₃ O ₇	312.5	2,2,2-trinitroethanol					
	344.9	18.0	57.6	65.5	59.5	20.7	
		2.7	7.9				
	A2+3*A50+A30*D30+A4*B4						[352]
C ₂ H ₆ Cd*	254.4	dimethylcadmium					
	270.5	1.52	5.98	35.0	33.2	9.36	
		7.84	29.0				
	2*A1+A114						[328]
C ₂ H ₆ BF ₄ N*	283.5	dimethylammonium tetrafluoroborate					
	375	7.5	26.5	35.8		11.0	
		3.5	9.3				
	Group value not available						[216]
C ₂ H ₆ BrN*	369.9	ethylammonium bromide					
	439.5	12.1	32.6	52.0		20.6	
		8.5	19.4				
	Group value not available						[216]
C ₂ N ₂ *	245.3	cyanogen					
		8.11		33.1		8.11	
	Group value not available						[216]
C ₃ H ₄ N ₄ O ₆ *	375.5	1,3,3-trinitroazetidene					
		30.3		80.7		30.3	
	[303, 332]						
C ₃ H ₆ N ₂ O ₅	281.7	2,2-dinitropropanol					
	366.7	15.06	53.5	61.2	53.7	17.9	
		2.85	7.8				
	A1+A2+A4*B4+2*A50+A30*C30						[352]
C ₃ H ₆ N ₂ O ₇ *	341.2	2,2-dinitro-1,3-propanediol					
		21.34		62.5	71.7	21.34	
	2*A2+2*A50+2*A30*D30+A4*B4 (Decomposes before melting)						[352]
C ₃ H ₉ In*	358.7	trimethylindium					
		14.3		39.9	33.5	14.3	
	3*A1+A105						[304]
C ₃ H ₉ Tl*	311.2	trimethylthallium					
		16.74		53.8	53.8	16.7	
	3*A1+A143						[370]
C ₃ H ₁₀ BrN*	464.6	propylammonium bromide					
		13.3		28.7		13.3	
	Group value not available						[216]
C ₄ F ₈ S*	146.0	octafluorotetrahydrothiophene					
	266.7	10.88	74.5	82.4	45.3	13.0	
		2.09	7.8				
	A14+8*A28+4*A17+2*A15+A131						[317]
C ₄ H ₃ Cl ₃ OS*	286.25	methyl trichlorothioacrylate					
		20.4		71.2		20.4	
	Group value not available						[216]
C ₄ H ₄ Se*	122.7	selenophene					
	192.8	0.3	2.5				
240.2	1.11	5.73					
		4.58	19.1	27.3		6.0	
	Group value not available						[216]
C ₄ H ₆ O ₆ *	445.1	D-tartaric acid					
		32.3		72.6	85.2	32.3	
	2*A3*B3+2*A30*D30+2*A36*D36						[392]
C ₄ H ₆ O ₂	180.6	vinyl acetate					
		8.46		46.8	47.9	8.5	
						8.7	

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_4H_6O_3$	A1+A5+A6+A38 <i>p</i> -dioxanone 16.14		53.5	48.8	16.1	[353] 14.7
$C_4H_6O^*$	A14+3*A15+A115+A112 cyclobutanone 10.8		48.8	35.7	10.8	[338] 7.9
$C_4H_{10}Te^*$	A14+A15+A114 diethyl telluride 7.62		47.2	47.2	7.6	[393] 7.7
$C_4H_{11}AsO_2^*$	2*A1+2*A2+A140 diethylarsinic acid 19.9		48.3	25.4	19.9	[299] 10.4
C_5F_{10}	2*A1+2*A2+A142 perfluorocyclopentane 5.0	41.9				
	3.0	12.6	54.5	42.8	8.0	12.1
C_5F_{12}	A14+2*A15+10*A28+5*A17 perfluoro- <i>n</i> -pentane 6.8		46.0	63.4	6.8	[335] 9.3
$C_5H_7NS^*$	6*A25+6*A26+5*A4*B4 2,4-dimethylthiazole 2.90		13.0	51.7	2.9	[335] 11.5
$C_5H_8N_2O_2$	A14+2*A15+2*A19+A18*B18+2A1+A131+A118 3-methyl-2,5-piperazinedione 30.62		56.3	52.8	30.6	[61] 28.7
$C_5H_8O^*$	A14+3*A15+2*A124+A16+A1 cyclopentanone 11.4		51.3	39.4	11.4	[375] 8.7
$C_5H_{10}N_2O_2$	A14+2*A15+A114 N-acetylsarcosinamide 27.4	0	66.4	59.0	27.4	[393] 24.2
$C_5H_{11}Br$	2*A1+A2+A59+A61 3-bromopentane 8.40		50.2	57.1	8.4	[354] 9.6
$C_5H_{12}O$	2*A1+2*A2+A3*B3+A21 2-pentanol 8.48		42.4	41.3	8.5	[312] 8.3
$C_5H_{12}O$	2*A1+2*A2+A3*B3+A30 3-pentanol 9.08		44.47	41.3	9.1	[361] 8.4
$C_6H_3ClN_2O_4$	2*A1+2*A2+A3*B3+A30 2,4-dinitrochlorobenzene 20.2		62.0	51.3	20.2	[361] 16.7
$C_6H_3ClN_2O_4$	3*A10+3*A12+2*A50+A22*C22 2,6-dinitrochlorobenzene 18.95		52.5	51.3	19.0	[334] 18.5
$C_6H_5ClO_3S$	3*A10+3*A12+2*A50+A22*C22 4-chlorobenzene sulfonic acid 10.6		31.8	31.8	10.6	[334] 10.6
C_6H_6	4*A10+2*A12+A22*B22+A145 2,4-hexadiyne 1.00		8.48	24	1.00	[348] 2.8
$C_6H_7NO_2$	2*A1+4*A9 ethyl- α -cyanoacrylate 12.86		52.9	56.7	12.9	[152] 13.8
$C_6H_8N_2O_2S$	A1+A2+A5+A7+A38+A56 <i>p</i> -aminobenzene sulfonamide 1.63	4.0				[350] 28.3
	24.0	54.7	58.7	64.4	25.7	28.3
	23.0	52.4	56.4	64.4	24.7	28.3
$C_6H_8O_4$	4*A10+2*A12+A45+A96 1,6-anhydro-2-deoxy- β -D-arabino-hexopyranose 12.55		38.9	56.9	12.6	[305,395] 18.4
$C_6H_8O_6^*$	2*A14+2*A15+4*A16+2*A30*D30+2*A112 L-ascorbic acid					[376]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{type}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{type}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{type}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{type}}$ (calcd)	
466.15	37.04		79.5	84.2	79.5	39.2	
C ₆ H ₉ NO	A14+2*A15+4*A30*E30+2*A19+A16+A3*B3+A2+A115 N-vinylpyrrolidone						[392]
	286.2	15.28		53.4	40.4	15.3	11.6
C ₆ H ₁₀ B ₂ N ₄ *	A14+2*A15+A6*B6+A5+A114+A119 pyrazabole						[368]
	354.3	11.83		33.4		11.8	
C ₆ H ₁₀ N ₂ O ₂	Group value unavailable						[123]
	556.1	30.6		55.1	55.7	30.6	31.0
C ₆ H ₁₀ N ₂ O ₂	A14+3*A15+2*A124+2*A16+2*A1 1,4-dimethyl-2,5-piperazinedione						[375]
	418.2	22.0		52.7	36.3	22.0	15.2
C ₆ H ₁₀ O*	A14+3*A15+2*A125+2*A1 cyclohexanone						[375]
	221	8.51	38.5				
C ₆ H ₁₀ O ₅	242.6	1.25	5.2	43.7	43.1	9.8	10.5
	A14+3*A15+A114 1,6-anhydro-β-D-glucopyranose						[393]
C ₆ H ₁₀ O ₅	385	24.9		64.8	59.4	24.9	22.9
	2*A14+2*A15+5*A16+3*A30*E30+2*A112 1,6-anhydro-β-D-glucopyranose						[376]
C ₆ H ₁₀ O ₅	404	24.5		60.6	59.4	24.5	24.0
	2*A14+2*A15+5*A16+3*A30*E30+2*A112 1,6-anhydro-β-D-galactopyranose						[376]
C ₆ H ₁₀ O ₅	401	22.8		56.9	59.4	22.8	23.8
	2*A14+2*A15+5*A16+3*A30*E30+2*A112 1,6-anhydro-β-D-altropyranose						[376]
C ₆ H ₁₀ O ₅	375	2.43	6.5				
	388	18.0	46.4	52.8	59.4	23.1	20.4
C ₆ H ₁₀ O ₅	2*A14+2*A15+5*A16+3*A30*E30+2*A112 1,6-anhydro-β-D-mannopyranose						[376]
	364	18.3		50.2	59.4	18.3	21.6
C ₆ H ₁₁ Cl	2*A14+2*A15+5*A16+3*A30*E30+2*A112 1-chloro-1-methylcyclopentane						[376]
	164.2	1.3	7.8				
C ₆ H ₁₁ FO ₅	178.8	5.7	31.9				
	189.1	0.7	3.9	43.6	46.2	8.74	7.7
C ₆ H ₁₁ FO ₅	A14+A15*2+A1+A4*B4+A22 2-deoxy-2-fluoro-D-glucopyranose						[377]
	427.2	38.2		89.4	79.0	38.2	33.8
C ₆ H ₁₁ FO ₅	A14+3*A15+A2+A112+4*A30*E30+5*A16+A28 3-deoxy-3-fluoro-D-glucopyranose						[336]
	378.2	18.3		48.4	79.0	18.3	29.9
C ₆ H ₁₁ FO ₅	A14+3*A15+A2+A112+4*A30*E30+5*A16+A28 6-deoxy-6-fluoro-D-glucopyranose						[336]
	412.2	27.2		66.0	74.3	27.2	30.6
C ₆ H ₁₁ NO ₂ *	A14+3*A15+A2+A112+4*A30*E30+5*A16+A27 5,5-dimethylperhydro-1,3-oxazine-2-one						[336]
	440.1	28.50		64.8	64.8	28.5	28.5
C ₆ H ₁₂	2*A1+A14+3*A15+A17+A125 4-methylpent-1-ene						[297]
	118.9	4.93		41.5	48.5	4.9	5.8
C ₆ H ₁₂ O ₅	2*A1+A2+A3+A5+A6 1-deoxy-D-glucopyranose						[326]
	403.2	27.4		68.0	76.3	27.4	30.8
C ₆ H ₁₂ O ₅	A14+3*A15+A2+A112+4*A30*E30+4*A16 2-deoxy-D-glucopyranose						[336]
	398.7	34.5		86.5	76.3	34.5	30.4
C ₆ H ₁₂ O ₅	A14+3*A15+A2+A112+4*A30*E30+4*A16 3-deoxy-D-glucopyranose						[336]
	387.2	32.6		84.2	76.3	32.6	29.5
C ₆ H ₁₂ O ₅	A14+3*A15+A2+A112+4*A30*E30+4*A16 6-deoxy-D-glucopyranose						[336]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
409.2	22.7		55.5	72.1	22.7	29.6
$\text{C}_6\text{H}_{12}\text{O}_6$	$A14+3*A15+A1+A112+4*A30*E30+5*A16$ α -D-glucopyranose (α -D-glucose)					[336]
423.2	34.3		81.1	90.3	34.3	39.2
$\text{C}_6\text{H}_{12}\text{O}_6$	$A14+3*A15+5*A30*F30+A2+5*A16+A112$ D-mannopyranose					[336]
391.2	24.7		63.1	90.3	24.7	35.3
$\text{C}_6\text{H}_{12}\text{Si}$	$A14+3*A15+5*A30*F30+A2+5*A16+A112$ 1-trimethylsilyl-1-propyne					[378]
204.5	10.96		53.6	37.7	11.0	7.7
$\text{C}_6\text{H}_{14}\text{O}_6$	$4*A1+A9+A109$ L-Iditol					[208,309]
352.8	30.90		87.6	108.0	30.9	38.1
$\text{C}_6\text{H}_{13}\text{AsO}_2^*$	$2*A2+4*A3*B3+6*A30*F30$ dipropylarsinic acid					[325]
408	22.1		54	39.6	16.2	22.1
$\text{C}_7\text{H}_5\text{F}_4\text{NO}_2$	$2*A1+4*A2+A142$ 1,1,3-trihydrotetrafluoropropyl α -cyanoacrylate					[381]
154.0	0.30	2.04				
287.4	19.95	69.4	71.5	71.9	20.3	20.7
$\text{C}_7\text{H}_5\text{N}_3\text{O}_6$	$A5+A7+A56+A38+A2+4*A26+A3*B3+A4*B4$ 2,4,6-trinitrotoluene					[337]
352.2	23.4		66.5	53.4	23.4	18.8
$\text{C}_7\text{H}_8\text{O}_2$	$A1+3*A50+3*A12+A11+2*A10$ 4-methoxyphenol					[217]
328.4	18.30		55.7	57.2	18.3	18.8
$\text{C}_7\text{H}_{10}\text{N}_2\text{O}_2$	$4*A10+2*A12+A1+A31+A32$ 1,3,5-trimethyluracil					[301]
428.7	16.1		37.6	38.5	16.1	16.5
$\text{C}_7\text{H}_{12}\text{N}_2\text{O}_2$	$A14+A15*3+2*A125+3*A1+A18*B18+A19$ N-acetyl-L-prolinamide					[379]
417.5	29.3		70.3	66.3	29.3	27.7
$\text{C}_7\text{H}_{12}\text{O}^*$	$A14+3*A15+A61+A16+A1+A146$ cycloheptanone					[380]
227	12.4	54.6				
232.6	0.43	1.8				
259.3	1.39	5.4	61.8	46.8	14.2	12.1
$\text{C}_7\text{H}_{14}\text{N}_2\text{O}_2$	$A14+3*A15+A114$ N-acetyl-L-valinamide					[393]
509	39.1		76.8	56.0	39.1	28.5
$\text{C}_7\text{H}_{14}\text{O}$	$3*A1+A3*B3+A3+A61+A60$ 2-heptanone					[380]
237.7	19.71		82.9	77.0	19.7	18.3
$\text{C}_7\text{H}_{14}\text{O}$	$2*A1+4*A2*B2+A35$ 3-heptanone					[214]
236.0	17.53		74.3	74.8	17.5	17.7
$\text{C}_7\text{H}_{14}\text{O}$	$2*A1+3*A2*B2+A2+A35$ 4-heptanone					[214]
240.2	16.16		67.3	68.2	16.2	16.4
$\text{C}_7\text{H}_{14}\text{O}_6$	$2*A1+4*A2+A35$ 1-methoxy- α -D-glucopyranoside					[214]
424.2	37.6		88.6	83.9	37.6	35.6
$\text{C}_7\text{H}_{14}\text{O}_6$	$A14+3*A15+A1+A2+A32+A112+4*A30*E30+5*A16$ 3-methoxy- α -D-glucopyranoside					[336]
425.6	41.3		97.0	83.9	41.3	35.7
$\text{C}_7\text{H}_{14}\text{O}_6$	$A14+3*A15+A1+A2+A32+A112+4*A30*E30+5*A16$ methyl α -D-mannopyranoside					[336]
455.2	44.7		98.2	83.9	44.7	38.2
$\text{C}_7\text{H}_{13}\text{Br}$	$A14+3*A15+A1+A2+A32+A112+4*A30*E30+5*A16$ 1-bromoheptane					[336]
214.4	21.76		101.5	90.9	21.8	19.5
$\text{C}_8\text{BrF}_{17}$	$A1+6*A2*B2+A21$ 1-bromoperfluorooctane					[333]
146.4	1.60	10.93				
278.9	12.13	43.49	54.4	103.0	13.7	28.7
	$8*A4*B4+3*A25+14*A26+A21$					[310]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{\text{fus}} S_{\text{pcc}}$ (expt)	$\Delta_0^{\text{fus}} S_{\text{pcc}}$ (calcd)	$\Delta_0^{\text{fus}} H_{\text{pcc}}$ (expt)	$\Delta_0^{\text{fus}} H_{\text{pcc}}$ (calcd)
$\text{C}_8\text{H}_5\text{Br}_3$	2,4,5-tribromostyrene					
340.3	25.10		73.8	59.9	25.1	20.4
	$2^*A10+4^*A12+A5+A6+3^*A21$					[295]
$\text{C}_8\text{H}_8\text{O}_3$	5,6-dioxycarbonyl[2.2.1]bicyclohept-2-ene					
323.6	0.90				2.78	
342.4	8.70	25.41				
388.4	3.60	9.27	37.5	43.2	13.2	16.8
	$3^*A14+A15+A116+4^*A16+2^*A18$					[359]
$\text{C}_8\text{H}_{10}\text{O}_4$	<i>trans, trans</i> -2,6-octadiene-1,8-dioic acid					
439.0	11.04	25.13				
541.0	27.77	51.32	76.5	65.2	38.8	36.3
	$2^*A6+2^*A36^*B36+2^*A2$					[46]
$\text{C}_8\text{H}_{10}\text{O}_4$	<i>trans, cis</i> -2,6-octadiene-1,8-dioic acid					
380.0	22.78		60.0	65.2	22.8	24.8
	$2^*A6+2^*A36^*B36+2^*A2$					[46]
$\text{C}_8\text{H}_{12}\text{B}_2\text{Cl}_6\text{O}_5^*$	1,3-diethyl-1,3-bis(trichloroacetoxy)-1,3-diboroxane					
327.2	24.22		74.0		24.2	
	Group value unavailable					[186]
$\text{C}_8\text{H}_{12}\text{N}_2\text{O}_2$	1,3-dimethyl-5-cethyluracil					
354.4	19.4		54.8	45.6	19.4	16.2
	$A14+3^*A15+2^*A125+3^*A1+A2+A18^*B18+A19$					[379]
$\text{C}_8\text{H}_{14}\text{O}^*$	cyclooctanone					
174	2.51	14.4				
226.8	13.8	60.8				
318.7	2.87	9.0	84.3	50.5	19.2	16.1
	$A14+5^*A15+A114$					[393]
$\text{C}_8\text{H}_{16}\text{B}_2\text{O}_5^*$	1,3-diacetoxy-1,3-diethyl-1,3-diboroxane					
377.2	21.60		57.3		21.6	
	Group value unavailable					[186]
$\text{C}_8\text{H}_{16}\text{N}_2\text{O}_2$	N-acetyl-L-isoleucine amide					
529.6	41.8		78.9	63.1	41.8	33.4
	$3^*A1+A2+A3+A3^*B3+A60+A61$					[354]
$\text{C}_8\text{H}_{16}\text{N}_2\text{O}_2$	N-acetyl-L-leucine amide					
382	0.3	0.8				
404	16.55	41.0	41.8	63.1	16.9	25.5
	$3^*A1+A2+A3+A3^*B3+A60+A61$					[380]
$\text{C}_8\text{H}_{16}\text{O}^*$	cyclooctanol					
261.3	2.12	8.11				
295.0	2.06	6.98	15.1	38.9	4.2	11.5
	$A14+5^*A15+A16+A30$					[365]
$\text{C}_8\text{H}_{16}\text{O}_4^*$	12-crown-4					
290.7	22.46		77.3	71.5	22.5	20.8
	$A14+9^*A15+4^*A112$					[398]
$\text{C}_8\text{H}_{17}\text{Br}$	1-bromooctane					
218.2	24.69		113.2	100.2	24.7	21.9
	$A1+7^*A2^*B2+A21$					[333]
$\text{C}_8\text{H}_{18}\text{O}_4$	1,2,7,8-tetrahydroxyoctane					
352.2	36.7		104.2	112.0	36.7	39.4
	$4^*A30^*D30+6^*A2+2^*A3^*B3$					[346,347]
$\text{C}_8\text{H}_{18}\text{Zn}^*$	di- <i>tert</i> -butyl zinc					
300.0	45.30		151.0	70.8	45.3	21.2
	$6^*A1+2^*A4^*B4+A111$					[294]
$\text{C}_8\text{H}_{18}\text{AsO}_2^*$	dibutylarsinic acid					
412	29.5		71.5	67.0	29.5	27.6
	$2^*A1+6^*A2^*B2+A142$					[381]
$\text{C}_9\text{H}_8\text{O}_3$	<i>endo</i> -5-norbornene-2,3-dicarboxylic anhydride					
367.2	15.7	42.8				
437.2	3.71	8.49	51.3	44.2	19.4	19.3
	$3^*A14+A15+2^*A18+4^*A16+A117$					[318]
$\text{C}_9\text{H}_8\text{O}_3$	<i>exo</i> -5-norbornene-2,3-dicarboxylic anhydride					
416.2	21.77		52.3	44.2	21.8	18.4
	$3^*A14+A15+2^*A18+4^*A16+A117$					[318]
$\text{C}_9\text{H}_8\text{O}_4$	4-acetoxybenzoic acid					
467.2	26.35		56.4	56.1	26.4	26.4
	$A1+A38+A36^*B36+4^*A10+2^*A12$					[307]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{pcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{pcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{pcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{pcc}}$ (calcd)
C ₉ H ₁₀ O ₂	417.6	2,3-dimethylbenzoic acid		43.8	18.30	18.3
		18.30				
C ₉ H ₁₀ O ₂	442.9	2*A1+3*A10+2*A11+A12+A36		51.0	22.60	19.5
		22.60				
C ₉ H ₁₂ O	292.8	3,5-dimethylbenzoic acid		50.0	14.64	15.6
		14.64				
C ₉ H ₁₄ O	300.5	2,6-diisopropylphenol		47.0	14.11	14.1
		14.11				
C ₉ H ₁₄ N ₂ O ₂	355.0	4*A1+3*A10+2*A3+A31+2*A11+A12		74.2	26.3	18.7
		26.3				
C ₉ H ₁₄ N ₂ O ₂	354.7	bicyclo[3.3.1]nonan 9-one		63.0	22.4	14.1
		22.4				
C ₉ H ₁₅ N ₃ O ₃ *	450.6	2*A14+3*A15+2*A16+A114		71.4	59.0	34.7
		59.0				
C ₉ H ₁₅ N ₃ O ₃ *	457.8	1,3-dimethyl-5-propyluracil		74.2	27.0	32.6
		27.0				
C ₉ H ₁₅ N ₃ O ₃ *	434.1	A14+3*A15+2*A125+3*A1+2*A2+A18*B18+A19		74.2	32.2	32.9
		32.2				
C ₉ H ₁₆ O*	247	N-acetylglucyl-L-prolinamide		64.9	5.4	16.2
		5.4				
C ₉ H ₁₆ O*	298	cyclononane		123.4	16.3	16.2
		16.3				
C ₉ H ₁₉ Br	243.2	1-bromononane		58	30.1	26.63
		30.1				
C ₁₀ H ₁₀ N ₄ O ₂ S	538.7	A1+8*A2*B2+A21		78.8	31.2	42.5
		31.2				
C ₁₀ H ₁₂	189.8	sulfadiazine		37.5	7.11	9.2
		7.11				
C ₁₀ H ₁₂	216.1	7*A10+3*A12+2*A41+A95+A45		46.2	48.3	9.8
		48.3				
C ₁₀ H ₁₂	304.7	3*A14+A15+4*A18+4*A19		48.3	1.79	14.7
		1.79				
C ₁₀ H ₁₂ O ₂ *	333.2	endo-dicyclopentadiene		4.08	1.36	20.5
		1.36				
C ₁₀ H ₁₂ O ₂ *	333.6	2-acetyl-3,5-dimethylphenol		75.5	25.2	17.9
		25.2				
C ₁₀ H ₁₂ O ₃ *	394.2	A14+2*A15+2*A112+A17+A1+5*A10+A11		65.9	27.2	28.7
		27.2				
C ₁₀ H ₁₂ O ₃ *	419.7	4-n-propoxybenzoic acid		67.3	5.88	28.7
		5.88				
C ₁₀ H ₁₂ O ₃ *	426.7	4*A10+2*A12+2*A2+A1+A36*B36+A32		67.3	27.2	28.7
		27.2				
C ₁₀ H ₁₂ O ₄	459.3	Forms liquid crystal		62.5	28.7	32.0
		28.7				
C ₁₀ H ₁₄ O	247.7	2,5-diethoxy-1,4-benzoquinone		46.6	11.55	13.6
		11.55				
C ₁₀ H ₁₅ F*	221.6	2*A1+2*A2+A14+3*A15+2*A18+2*A19+2*A32+2*A114		6.77	1.50	8.4
		1.50				

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{\text{fus}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{\text{fus}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{\text{fus}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{\text{fus}} H_{\text{tpcc}}$ (calcd)
						[145]
$\text{C}_{10}\text{H}_{16}$	$3*A14+A15+3*A16+A17+A27$ <i>d</i> -limonene 11.38		57.1	57.7	11.4	11.5
$\text{C}_{10}\text{H}_{16}\text{N}_2\text{O}_2$	$2*A1+A5+A7+A16+A18+A19+A14+3*A15$ 1,3-dimethyl-5-butyluracil 22.0		70.5	59.8	22.0	[213,293]
$\text{C}_{10}\text{H}_{16}\text{O}^*$	$A14+A15*3+2*A125+3*A1+3*A2+A18*B18+A19$ 1-hydroxyadamantane 2.50	6.8		26.9	2.5	9.9
$\text{C}_{10}\text{H}_{16}\text{O}^*$	$3*A14+A15+3*A16+A17+A30$ 2-hydroxyadamantane 0.30	0.92				[172]
	3.74	9.56	10.5	32.1	4.0	12.6
$\text{C}_{10}\text{H}_{18}\text{O}^*$	$3*A14+A15+5*A16+A30$ cyclodecanone 24.3		82.3	57.9	24.3	[172]
$\text{C}_{10}\text{H}_{22}\text{O}$	$A14+7*A15+A114$ 1-decanol 37.66		134.5	103.0	37.7	28.9
$\text{C}_{10}\text{H}_{23}\text{AsO}_2^*$	$A1+9*A2*B2+A30$ dipentylarsinic 36.0		88.8	85.6	36.0	[321]
$\text{C}_{10}\text{H}_{30}\text{Si}_5\text{O}_5$	$2*A1+8*A2*B2+A142$ decamethylcyclopentasiloxane 20.37		90.1	67.8	20.4	[381]
$\text{C}_{11}\text{H}_8\text{N}_2$	$10*A1+5*A112+5*A139+A14+7*A15$ 9H-pyrido[3,4- <i>b</i>]indole 25.50		54.0	56.5	25.5	[121]
$\text{C}_{11}\text{H}_{12}\text{N}_4\text{O}_2\text{S}$	$A14+2*A15+2*A19+2*A19+7*A10+A121+A41$ sulfamerazine 31.6		61.2	79.4	40.9	[323]
$\text{C}_{11}\text{H}_{12}\text{N}_4\text{O}_3\text{S}$	$6*A10+3*A12+A11+A1+2*A41+A95+A45$ 4-amino-N-(6-methoxy-3-pyridazinyl)benzenesulfonamide (sulphamethoxypyridazine) 22.30		49.2	86.2	22.3	[382]
$\text{C}_{11}\text{H}_{12}\text{N}_2\text{O}^*$	$6*A10+4*A12+A95+A45+2*A41+A1+A32$ antipyrene 24.52		63.6	49.1	25.4	[194]
$\text{C}_{11}\text{H}_{13}\text{N}_3\text{O}_3\text{S}$	$A14+2*A15+2*A1+A119+A125+5*A10+A12+A19+A18*B18$ sulfisoxazole 29.2		62.5	83.1	29.2	[395]
$\text{C}_{11}\text{H}_{14}\text{O}_2^*$	$4*A10+2*A12+A45+A95+A14+2*A15+3*A19+2*A1+A112+A118$ 2-acetyl-3,5-dimethylanisole 0.99		3.06	63.4	1.0	[382]
	$4*A1+2*A11+2*A12+2*A10+A32+A38$ Reported enthalpy of fusion is too small, and the published enthalpy and entropy of fusion data are internally inconsistent					[11]
$\text{C}_{11}\text{H}_{14}\text{O}_3^*$	4- <i>n</i> -butoxybenzoic acid 18.83	44.8				
	2.93	6.78	51.5	74.4	21.8	32.2
	$4*A10+2*A12+3*A2+A1+A36*B36+A32$ Forms liquid crystal					[178]
$\text{C}_{11}\text{H}_{16}\text{O}_2^*$	1-adamantanecarboxylic acid 2.25		4.29	38.6	2.3	20.2
	$3*A14+A15+3*A16+A17+A36$ Reported entropy is too small					[149]
$\text{C}_{11}\text{H}_{20}\text{O}^*$	cycloundecanone 23.0		80.5	61.6	23.0	17.7
	$A14+8*A15+A114$					[393]
$\text{C}_{11}\text{H}_{23}\text{Br}$	1-bromoundecane 33.47		127.1	128.1	33.5	33.7
	$A1+10*A2*B2+A21$					[333]
$\text{C}_{12}\text{HF}_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorododecane 23.00		66.76	137.9	23.00	47.5
	$11*A4*B4+3*A25+22*A26+A3*B3$					[68]
$\text{C}_{12}\text{H}_8\text{O}_2\text{S}$	dibenzothiophene sulfone 23.72		46.7	40.4	23.7	20.5

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$C_{12}H_{10}N_2$	A14+2*A15+2*A19+2*A19+8*A10+A134 1-methyl-9H-pyrido[3,4- <i>b</i>]indole 27.20		53.3	57.1	27.2	29.1 [327]
$C_{12}H_{10}O_2$	A1+A14+2*A15+2*A9+2*A19+6*A10+A11+A121+A41 α -naphthyl acetate 20.21		63.3	54.6	20.2	17.4 [323]
$C_{12}H_{10}O_2$	7*A10+3*A12+A1+A38 β -naphthyl acetate 20.05		58.6	54.6	20.1	18.7 [118]
$C_{12}H_{10}O_2S$	7*A10+3*A12+A1+A38 diphenyl sulfone 21.78		54.7	59.3	21.8	23.6 [327]
$C_{12}H_{10}S$	10*A10+2*12+A88 diphenylsulfide 13.98		54.19	61.10	13.98	15.76 [207]
$C_{12}H_{10}Te^*$	10*A10+2*A12+84 diphenyl telluride 15.35		57.2	57.0	15.4	15.3 [300]
$C_{12}H_{12}N_2$	10*A10+2*A12+A140 benzidine 19.10		47.7	72.0	19.1	28.8 [4]
$C_{12}H_{14}N_4O_2S$	4*A12+8*A10+2*A45 4-amino-N-[2,6-dimethyl-4-pyrimidinyl]benzene sulfonamide 45.11		87.5	58.6	45.1	30.2 [358]
$C_{12}H_{14}N_4O_2S$	2*A1+5*A10+3*A12+2*A11+2*A41+A95 sulfisomidine 42.7		81.5	80.0	42.7	41.9 [382]
$C_{12}H_{14}N_4O_2S$	5*A10+2*A41+A95+A45+3*A12+2*A11+2*A1 sulfamethazine 31.1		66.0	80.0	31.1	37.7 [382]
$C_{12}H_{16}O_2^*$	5*A10+3*A12+2*A11+2*A1+2*A41+A95+A45 4- <i>n</i> -pentylbenzoic acid 2.60	10.32				
		9.90				
		1.50	41.47	57.7	14.0	22.8 [177]
$C_{12}H_{16}O_3^*$	A1+2*A2+4*A10+A11+A12+A36 Forms liquid crystal 4- <i>n</i> -pentoxybenzoic acid 21.8	54.7				
		2.1	59.7	81.5	23.9	34.4 [178]
$C_{12}H_{16}O_4$	4*A10+2*A12+4*A2+A1+A36*B36+A32 Forms liquid crystal 2,5-dipropoxy-1,4-benzoquinone 8.60	24.09				
		33.6	97.0	84.0	42.2	38.7 [342]
$C_{12}H_{16}O_6$	2*A1+4*A2+A14+3*A15+2*A18*B18+2*A19+2*A32+2*A114 α -phenoxy- α -D-glucopyranoside 39.0	90.9		95.8	39.0	41.1 [384]
$C_{12}H_{18}N_2O_3S^*$	A14+3*A15+A2+A32+A112+4*A30*E30+5*A16+5*A10+A12 3-(<i>p</i> -tolyl-4-sulfonyl)-1-butyl urea 25.6		63.3		25.6	
$C_{12}H_{18}O$	Group value not available 2-(1'-cyclohexenyl)cyclohexanone 17.26		61.9	59.0	17.3	16.5 [314]
$C_{12}H_{18}O$	2*A14+6*A15+A18+A19+A16+A114 3,5-diisopropylphenol 12.13		37.2	53.4	12.1	17.4 [330]
$C_{12}H_{18}O_6$	4*A1+3*A10+2*A3+A31+2*A11+A12 R,R,R-4,8,12-trimethyl-1,5,9-trioxacyclododeca-2,6,10-trione 21.5	56.6		84.7	21.5	32.2 [206]
$C_{12}H_{20}O$	3*A1+A14+9*A15+3*A115+3*A16 2-cyclohexylcyclohexanone 18.00		65.0	58.2	18.0	16.1 [314]
$C_{12}H_{22}O$	2*A14+6*A15+2*A16+A114 cyclododecanone 16.85		50.2	65.3	16.9	21.9
$C_{12}H_{22}O$	A14+A114+9*A15 <i>trans</i> -2-cyclohexylcyclohexanol		50		16.6	

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)	
325.8	14.52		44.6	46.6	14.5	15.2	
$C_{12}H_{31}AsO_2^*$	2*A14+6*A15+3*A16+A30 dihexylarsinic acid					[313]	
393	16.4	41.8					
405	24.35	60.1	101.9	104.2	40.7	42.2	
$C_{12}H_{36}O_6Si_6$	2*A1+10*A2*B2+A142 dodecamethylcyclohexasiloxane					[381]	
269.0	28.58		106.3	76.9	28.6	20.7	
$C_{13}H_8O_2$	12*A1+6*A112+6*A139+A14+9*A15 S-(+)-4-isobutyl- α -methylphenyl acetic acid					[121]	
325.5	18.70		57.5	57.5	18.7	18.7	
$C_{13}H_{10}N_2O_2^*$	3*A1+A2+A3+A3*B3+4*A10+2*A11+A36 N-phenyl 4-nitrobenzaldehyde imine					[319]	
347.15	24.56		70.7	64.0	24.6	22.2	
$C_{13}H_{10}O_2^*$	9*A10+3*A12+A6*B6+A42+A50 (2-hydroxyphenyl)phenylmethanone					[397]	
308.2	0.67		2.17		0.67		
	No prediction made: Reported enthalpy of fusion is too small, and the published enthalpy and entropy of fusion data are internally inconsistent						[11]
$C_{13}H_{11}N^*$	N-phenylbenzaldehyde imine						
329.65	20.42		61.9	61.2	20.4	20.2	
$C_{13}H_{12}N_2O^*$	10*A10+2*A12+A6*B6+A42 7-methoxy-1-methyl-9H-pyrido[3,4- <i>b</i>]indole					[397]	
536.6	48.80		90.9	64.5	48.8	42.6	
$C_{13}H_{17}N_3O^*$	2*A1+A14+2*A15+2*A19+2*A19+5*A10+A11+A121+A41+A32 aminopyrine					[323]	
380	27.17		71.5	52.9	27.2	20.1	
$C_{13}H_{18}O_2^*$	A14+2*A15+4*A1+A43+A119+A125+5*A10+A12+2*A19 4- <i>n</i> -hexylbenzoic acid						
371.0	17.40	46.90					
380.0	2.40	6.31	53.21	65.1	19.80	24.7	
	A1+2*A2+5*A10+A11+A12+A36 Forms liquid crystal						[177]
$C_{13}H_{18}O_2$	benzaldehyde 2,2-dimethylpropylene glycol acetal						
307.6	18.6		60.5	60.2	18.6	18.5	
$C_{14}H_5F_{25}^*$	5*A10+A11+A14+3*A15+A17+A16+2*A1+2*A112 1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorotetradecane					[385]	
344.2	20.80		60.43	149.5	20.80	51.5	
	12*A4*B4+3*A25+22*A26+A1+A2 Amphiphilic compound						[68]
$C_{14}H_9F_{17}O_2^*$	perfluorooctylethylene methacrylate						
210	5.0	23.8					
253	9.0	35.6	59.4		14.0		
	Amphiphilic compound						[16]
$C_{14}H_9F_{21}O^*$	ω -perfluorodecyl-1-butanol						
360	21.30		59.2		21.3		
	Amphiphilic compound						[17]
$C_{14}H_{12}N_2O_2$	4-nitro-4'-methylbenzylidene aniline						
402.0	27.30		67.9	64.6	27.3	25.9	
$C_{14}H_{12}O_2^*$	A1+8*A10+A11+3*A12+A42+A6*B6+A50 (2-methoxyphenyl)phenylmethanone					[302]	
350.2	0.68		1.94		0.7		
	No prediction made: Reported enthalpy of fusion is too small, and the published enthalpy and entropy of fusion data are internally inconsistent						[11]
$C_{14}H_{14}O_3^*$	Naproxen						
428.5	31.5		73.5	58.6	31.5	25.1	
$C_{14}H_{20}O_4$	2*A1+A3*B3+6*A10+A11+2*A12+A32+A36*B36 2,5-dibutoxy-1,4-benzoquinone					[394]	
328.3	4.70	14.32					
364.5	2.30	6.31					
473.3	31.5	66.55	87.2	98.2	38.5	46.5	
$C_{14}H_{22}$	2*A1+6*A2+A14+3*A15+2*A18*B18+2*A19+2*A32+2*A114 1,4-di- <i>tert</i> -butylbenzene					[342]	
341.5	22.48		65.8	46.4	22.5	15.9	
$C_{14}H_{23}NO_2$	6*A1+4*A10+2*A11+2*A4 <i>n</i> -decyl- α -cyanoacrylate					[362]	
294.5	41.80		142.0	133.3	41.8	39.3	
	A1+9*A2*B2+A5+A7+A38+A56						[351]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$C_{14}H_{28}$	<i>trans</i> -1,4-di- <i>tert</i> -butylcyclohexane					
363.2	17.15		47.2	51.1	17.2	18.6
	6*A1+A14+2*A15+2*A4+2*A16					[41]
$C_{14}H_{28}$	<i>cis</i> -1,4-di- <i>tert</i> -butylcyclohexane					
293.2	8.79		30.0	51.1	8.8	15.0
	6*A1+A14+2*A15+2*A4+2*A16					[41]
$C_{14}H_{26}B_2N_4^*$	4,4,8,8-tetraethylpyrazobole					
342.4	28.61	83.61				
379.2	3.22	8.49	92.1		31.8	
	Group value unavailable					[123]
$C_{14}H_{31}AsO_2^*$	diheptylarsinic acid					
299.0	30.1	100.7				
389.0	20.3	52.3	153.0	122.8	50.4	47.7
	2*A1+12*A2*B2+A142					[381]
$C_{14}H_{42}O_7Si_7$	tetradecamethylcycloheptasiloxane					
237.7	20.88		87.8	86.0	20.9	20.4
	14*A1+7*A112+7*A139+A14+11*A15					[121]
$C_{15}H_{11}NO_2$	1-(methylamino)-9,10-anthracenedione					
443.2	28.81		65.0	49.1	28.8	21.8
	A14+3*A15+2*A114+4*A19+7*A10+A1+A44+A12					[315]
$C_{15}H_{12}ClN_5O_4$	5-[(4-chloro-2-nitrophenylazo)-1-ethyl-1,2-dihydro-6-hydroxy-4-methyl-2-oxo-3-pyridinecarbonitrile					
500.2	35.16		70.3	85.8	35.2	42.9
	3*A10+3*A12+A50+A22*F22+2*A42+A30*F30+2*A1+A2+A56+A14+3*A15+A125+4*A19					[315]
$C_{15}H_{12}N_2O_2$	5,5-diphenylhydantoin					
574.0	36.29		63.2	66.4	36.3	38.1
	A14+2*A15+2*A124+A17+10*A10+2*A11					[395]
$C_{15}H_{14}O_2^*$	(2-hydroxy-4,6-dimethylphenyl)phenylmethanone					
405.2	0.67		1.65		0.67	
	No prediction made: Reported enthalpy of fusion is too small, and the published enthalpy and entropy of fusion data are internally inconsistent					[11]
$C_{15}H_{16}S_2$	2,2-bis(phenylthio)propane					
329.0	24.4		74.2	75.4	24.4	24.8
	2*A1+10*A10+2*A12+A4*B4+2*A84					[363]
$C_{15}H_{20}N_2O_4S^*$	4-acetyl-N-[(cyclohexylamino)carbonyl]benzene sulfonamide					
457.0	41.08		89.9		41.1	
	Group value not available					[358]
$C_{15}H_{21}NO_2$	1-methyl-4-phenylpiperidine-4-carboxylic acid ethyl ester (meperidine)					
308.2	24.60		79.8	68.0	24.6	21.0
	2*A1+5*A10+A11+A38+A14+3*A15+A2+A17+A119					[296]
$C_{16}H_9F_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosafuoroheptadecane					
147	0.70	4.76				
314	1.40	4.46				
349	21.0	60.17	69.39	163.7	23.1	57.2
	3*A2+A1+12*A4*B4+3*A25+22*A26					[17]
$C_{16}H_{14}O_3$	Amphiphilic compound					
	(±) α -(3-benzoylphenyl)propionic acid					
367.4	28.23		76.8	70.6	28.2	25.9
	9*A10+A11+2*A12+A35+A36*B36+A1+A3*B3					[209]
$C_{16}H_{16}$	2,2-metacyclophane					
404.0	21.42		53.0	51.3	21.4	20.7
	A14+7*A15+4*A19+2*A18+6*A10					[316]
$C_{16}H_{16}$	2,2-metaparacyclophane					
315.0	0.98	3.11				
354.0	12.76	36.05	39.2	46.0	13.7	16.3
	A14+8*A15+4*A19+3A18+5*A10					[316]
$C_{16}H_{16}^*$	2,2-paracyclophane					
323.2	0.21		0.65	40.7	0.2	13.2
	A14+9*A15+4*A19+4*A18+4*A10					[360]
$C_{16}H_{16}O_2^*$	(2-hydroxyphenyl)-2,4,6-trimethylphenylmethanone					
353.2	0.49		1.39		0.49	
	No prediction made. Reported enthalpy of fusion is too small, and the published enthalpy and entropy of fusion data are internally inconsistent					[11]
$C_{16}H_{17}ClN_4O_4^*$	2,2'-[[3-chloro-4-[(4-nitrophenyl)azo]phenyl]imino]bis-ethanol					
463.2	29.78		64.3		29.8	
	Group value not available					[13]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)
$C_{16}H_{17}Cl_2N_5O_4^*$	371.2	1-[[2-chloro-4-[(2-chloro-4-nitrophenyl)azo]-5-(methylamino)phenyl]amino]- 2-propanol N-oxide 30.62		82.5	82.5	30.6
		Group value not available				30.6
$C_{16}H_{17}F_{15}O^*$	285.8	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7-pentadecafluoro-8-hexadecanone 34.20		119.7	147.5	34.2
		7*A4*B4 + 3*A25 + 12*A26 + A1 + 7*A2 + A35				42.2
		Amphiphillic compound				[23]
$C_{16}H_{20}O_3$	387.6	3-benzoyl-1,2,2-trimethylcyclopentanecarboxylic acid 20.35		52.5	60.0	20.4
		3*A1 + A14 + 2*A15 + 2*A17 + A16 + 2*A10 + A12 + A36*B36 + A35				23.3
$C_{16}H_{23}N^*$	339.4	N-cyclohexyl(2,4,6-trimethyl)benzaldehyde imine 25.61		75.5	92.1	25.6
		A14 + 3*A15 + A16 + 3*A1 + 2*A10 + A6*B6 + A42 + A12				31.2
$C_{16}H_{24}O_4$	333.7	2,5-dipentoxy-1,4-benzoquinone 9.0		26.97		
	414.6	36.5		88.04	112.5	45.5
		2*A1 + 8*A2 + A14 + 3*A15 + 2*A18*B18 + 2*A19 + 2*A32 + 2*A114		115.0		46.6
$C_{16}H_{35}AsO_2^*$	379	dioctylarsinic acid 20.7		54.6		
	402	35.8		89	143.6	56.5
		2*A1 + 14*A2*B2 + A142			141.4	56.6
$C_{17}H_{16}ClN_5O_3$	428.2	3-[[4-[(2-chloro-4-nitrophenyl)azo]phenyl](2-hydroxyethyl)amino]propanenitrile 26.29		61.4	90.7	26.3
		4*A2 + 7*A10 + 5*A12 + A30*F30 + A22*F22 + A50 + A56 + 2*A42 + A43				38.8
$C_{17}H_{17}ClO_6^*$	495.2	([2S]-trans-7-chloro-,2',4,6-trimethoxy-6'-methylspiro[benzofuran- 2(3H),1'-(2)cyclohexene]-3,4'-dione 39.39		79.6	77.8	39.4
		4*A1 + A10 + 3*A12 + 2*A14 + 4*A15 + A112 + 2*A114 + 3*A32 + A22*F22 + A17 + A18 + 3*A19 + A16				38.5
$C_{17}H_{17}Cl_2N_5O_4$	471.2	N-[4-chloro-2-[(2-chloro-4-nitrophenyl)azo]-5-[(2-hydroxypropyl)amino]phenyl] acetamide 38.87		82.5	81.9	38.9
		5*A10 + 7*A12 + 2*A22*F22 + 2*A1 + A2 + A3*B3 + A30*F30 + A50 + 2*A42 + A60 + A44				38.6
$C_{17}H_{19}NO_3$	528.2	7,8-didehydro-4-5-epoxy-17-methylmorphinan-3,6-diol (morphine) 28.87		54.7	73.9	28.9
		4*A14 + 3*A15 + 3*A16 + A17 + A119 + A1 + 2*A18 + A30*D30 + 3*A19 + 2*A10 + A12 + A112 + A31 + A114				39.0
$C_{17}H_{19}NO_3$	539.2	4,5-epoxy-3-hydroxy-17-methylmorphinan-6-one (hydromorphone) 35.61		66.0	54.8	35.6
		2*A10 + 4*A14 + 3*A15 + 3*A19 + A31 + A112 + A1 + A119 + 3*A16 + A12 + A17 + A114				29.5
$C_{17}H_{21}ClO_4$	440.2	3-(3-chloro-4-methoxybenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 29.10		66.1	82.7	29.1
		4*A1 + A14 + 2*A15 + 2*A17 + A16 + 3*A10 + 3*A12 + A36*D36 + A35 + A32 + A22*D22				36.4
$C_{17}H_{21}F_{15}^*$	220.0	1,1,1,2,3,3,4,4,5,5,6,6-dodecafluoro-2-(trifluoromethyl)hexadecane 3.00		13.64		
	261.0	18.00		68.96	82.6	21.0
		7*A4*B4 + 6*A25 + A27 + 8*A26 + A1 + 9*A2			143.8	37.5
		Amphiphillic compound				[22]
$C_{17}H_{21}NO_6$	426.9	3-(3-nitro-4-methoxybenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 32.36		75.8	84.2	32.4
		4*A1 + A14 + 2*A15 + 2*A17 + A16 + 3*A10 + 3*A12 + A36*D36 + A35 + A32 + A50				35.9
$C_{17}H_{21}N_3O_2$	384.2	2,2'-[[3-methyl-4-4(phenylazo)phenyl]imino]bis-ethanol 31.90		83.0	88.4	31.9
		8*A10 + 3*A12 + A11 + 2*A42 + A1 + 4*A2 + 2*A30*E30 + A43				34.0
$C_{17}H_{22}O_3$	468.2	3-(4-methylbenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 30.07		64.2	60.6	30.1
		4*A1 + A14 + 2*A15 + 2*A17 + A16 + 4*A10 + A11 + A12 + A36*B36 + A35				28.4
$C_{17}H_{23}NO_3^*$	422.0	3-[(hydroxyimino)phenylmethyl]-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester 33.80		80.1	67.4	33.8
		4*A1 + A14 + 2*A15 + A12 + 5*A10 + 2*A17 + A7 + A53 + A38 + A16				28.4
$C_{17}H_{23}NO_4^*$	498.6	3-(4-methoxy-3-aminobenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 41.32		82.9	87.9	41.3
		4*A1 + A14 + 2*A15 + 2*A17 + A16 + 3*A10 + 3*A12 + A36*D36 + A35 + A32 + A45				43.8

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{fus}} S_{pce}$ (expt)	$\Delta_0^{T_{fus}} S_{pce}$ (calcd)	$\Delta_0^{T_{fus}} H_{pce}$ (expt)	$\Delta_0^{T_{fus}} H_{pce}$ (calcd)
$C_{18}H_{15}Ge^*$ 405.0	tris(pentafluorophenyl)germane 34.90		86.2	86.2	34.9	34.9
	18*A12+15*A24+A141					[308]
$C_{18}H_{11}NO_3$ 539.2	2-(3-hydroxy-2-quinolyl)-1H-indene-1,3(2H)-dione 30.89		57.3	80.3	30.9	43.3
	9*A10+4*A12+A14+2*A15+4*A19+A114+A30*D30+A41					[315]
$C_{18}H_{13}F_{25}^*$ 317.2 352.2	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorooctadecane 3.30 21.80	10.4 61.70	72.10	177.9	25.10	62.7
	12*A4*B4+3*A25+22*A26+A1+5*A2 Amphiphilic compound					[68]
$C_{18}H_{20}$ 274.5	6-(4-biphenyl)-1-hexene 15.10		55.0	93.0	15.1	25.5
	9*A10+2*A12+A11+4*A2+A5+A6					[97]
$C_{18}H_{20}$ 332.0 351.0 377.0	3,3-paracyclophane 7.36 0.46 11.76	22.17 1.31 31.19	54.7	48.1	19.6	18.1
	A14+11*A15+4*A19+4*A18+4*A10					[316]
$C_{18}H_{20}O_2^*$ 380.2	(2-hydroxyl-4,6-dimethylphenyl)-2,4,6-trimethylphenylmethanone 0.84		2.21		0.84	
	Reported enthalpy of fusion is too small, and the published enthalpy and entropy of fusion data are internally inconsistent					[11]
$C_{18}H_{21}N^*$ 339.6	N-benzyl-pivalophenone imine 27.86		82.05	84.2	27.9	28.6
	10*A10+A11+A12+3*A1+A4+A2+A6*B6+A42					[397]
$C_{18}H_{21}NO_3$ 430.3	7,8-didehydro-4,-5-epoxy-3-methoxy-17-ethylmorphanan-6-ol (codeine) 23.81		55.3	62.6	23.8	38.5
	2*A10+4*A14+3*A15+3*A19+A112+2*A1+A119+4*A16+A12+A17+A32+A30*D30+2*A18					[374]
$C_{18}H_{22}O_4$ 409.5 412.4	4,4'-di-(2-methoxyethoxy)biphenyl 17.53 22.67	42.81 54.97	97.8	111.6	40.2	40.3
	2*A1+4*A2+8*A10+4*A12+4*A32					[345]
$C_{18}H_{24}N_2O_5^*$ 433.0	3-[(hydroxyimino)(4-methoxy-3-nitrophenyl)methyl]-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester 31.99		73.9	77.6	32.0	33.6
	5*A1+A14+2*A15+3*A12+3*A10+2*A17+A16+A38+A32 +A50+A7+A53					[373]
$C_{18}H_{24}O_3$ 387.6	3-(4-ethylbenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 22.54		58.2	67.7	22.5	26.3
	4*A1+A14+2*A15+2*A17+A16+4*A10+A2+A11+A12+A36*B36+A35					[366]
$C_{18}H_{24}O_3$ 460.6	3-(3,4-dimethylbenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 32.31		70.2	61.2	32.3	28.2
	5*A1+A14+2*A15+2*A17+A16+3*A10+2*A11+A12+A36*B36+A35					[366]
$C_{18}H_{24}O_3$ 386.8	3-(2,4-dimethylbenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 18.81		48.6	61.2	18.8	23.7
	5*A1+A14+2*A15+2*A17+A16+3*A10+2*A11+A12+A36*B36+A35					[366]
$C_{18}H_{24}O_4$ 394.6	3-(4-ethoxybenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid 22.05		55.9	88.5	22.1	34.9
	4*A1+A14+2*A15+2*A17+A16+4*A10+A2+2*A12+A36*C36+A35+A32					[366]
$C_{18}H_{25}NO_3$ 445.0	3-[4-(dimethylamino)benzoyl]-1,2,2-trimethylcyclopentanecarboxylic acid 25.03		56.3	72.1	25.0	32.1
	5*A1+A14+2*A15+2*A17+A16+4*A10+2*A12+A36*C36+A35+A43					[366]
$C_{18}H_{25}NO_4^*$ 433.0	3-[(hydroxyimino)(4-methoxyphenyl)methyl]-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester 36.99		85.4	74.8	37.0	32.4
	5*A1+A14+2*A15+2*A12+4*A10+2*A17+A16+A38+A32+A53+A7					[373]
$C_{18}H_{28}O_4$ 332.3 412.1	2,5-di-n-hexyloxy-1,4-benzoquinone 5.3 38.9	15.95 94.39	110.3	126.6	44.2	52.2
	2*A1+10*A2+A14+3*A15+2*A18*B18+2*A19+2*A32+2*A114					[342]
$C_{18}H_{32}O_2$ 303	linoelaidic acid 47.70		157.4	163.8	47.7	49.6
	A1+12*A2*B2+4*A6+A36					[331]
$C_{18}H_{32}O_2$	4-octadecynoic acid					

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
348	57.94		166.5	151.2	57.9	52.6
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 12*A2*B2 + 2*A9 + A36 + 2*A2$ 5-octadecynoic acid					[331]
325	54.41		167.4	149.0	54.4	48.4
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 11*A2*B2 + 2*A9 + A36 + 3*A2$ 6-octadecynoic acid					[331]
324	54.92		169.5	155.6	54.9	50.4
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 7-octadecynoic acid					[331]
322	53.61		166.5	155.6	53.6	50.1
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 8-octadecynoic acid					[331]
320	55.30		172.8	155.6	55.3	49.8
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 9-octadecynoic acid					[331]
319	54.87		172.0	155.6	54.9	49.6
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 10-octadecynoic acid					[331]
319	52.23		164.0	155.6	52.3	49.6
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 11-octadecynoic acid					[331]
320	55.97		174.9	155.6	56.0	49.8
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 12-octadecynoic acid					[331]
320	49.79		155.6	155.6	49.8	49.8
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 13-octadecynoic acid					[331]
322	55.51		172.4	155.6	55.5	50.1
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 14-octadecynoic acid					[331]
337	52.74		156.5	151.2	52.7	51.0
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 12*A2*B2 + 2*A9 + A36 + 2*A2$ 16-octadecynoic acid					[331]
347	60.10		173.2	155.6	60.1	54.0
$\text{C}_{18}\text{H}_{32}\text{O}_2$	$A1 + 14*A2*B2 + 2*A9 + A36$ 17-octadecynoic acid					[331]
340	54.20		159.4	157.7	54.2	52.9
$\text{C}_{18}\text{H}_{34}\text{B}_4\text{N}_4^*$	$15*A2*B2 + A9 + A36 + A8$ 4,4,8,8-tetrapropylpyrazabole					[331]
382.2	33.00		86.3		33.0	
	Group value unavailable					[123]
$\text{C}_{18}\text{H}_{34}\text{O}_2$	<i>trans</i> -3-octadecenoic acid					
334	57.15		171.1	169.6	57.2	56.7
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 13*A2*B2 + 2*A6 + A36 + A2$ <i>trans</i> -4-octadecenoic acid					[331]
333	55.88		167.8	167.4	55.9	55.7
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 12*A2*B2 + 2*A6 + A36 + 2*A2$ <i>trans</i> -5-octadecenoic acid					[331]
319	45.11		141.3	165.2	45.1	52.7
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 11*A2*B2 + 2*A6 + A36 + 3*A2$ <i>trans</i> -6-octadecenoic acid					[331]
326	60.15		184.5	171.8	60.2	56.0
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 14*A2*B2 + 2*A6 + A36$ <i>trans</i> -10-octadecenoic acid					[331]
326	58.52		179.5	171.8	58.5	56.0
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 14*A2*B2 + 2*A6 + A36$ <i>trans</i> -11-octadecenoic acid					[331]
317	58.49		184.5	171.8	58.5	54.5
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 14*A2*B2 + 2*A6 + A36$ <i>trans</i> -12-octadecenoic acid					[331]
325	56.71		174.5	171.8	56.7	55.8
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 14*A2*B2 + 2*A6 + A36$ <i>trans</i> -13-octadecenoic acid					[331]
318	55.62		174.9	165.2	55.6	52.5
$\text{C}_{18}\text{H}_{34}\text{O}_2$	$A1 + 11*A2*B2 + 2*A6 + A36 + 3*A2$					[331]

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
$\text{C}_{18}\text{H}_{34}\text{O}_2$	<i>trans</i> -14-octadecenoic acid					
327	57.06		174.5	167.4	57.1	54.7
	$A1 + 12^*A2^*B2 + 2^*A6 + A36 + 2^*A2$					[331]
$\text{C}_{18}\text{H}_{34}\text{O}_2$	<i>trans</i> -15-octadecenoic acid					
331	58.98		178.2	169.6	59.0	56.1
	$A1 + 13^*A2^*B2 + 2^*A6 + A36 + A2$					[331]
$\text{C}_{18}\text{H}_{36}$	<i>cis, cis</i> -1,3,5-tri- <i>tert</i> -butylcyclohexane					
393.2	26.78		68.1	54.4	26.8	21.4
	$9^*A1 + A14 + 3^*A15 + 3^*A4 + 3^*A16$					[41]
$\text{C}_{18}\text{H}_{36}$	<i>cis, trans</i> -1,3,5-tri- <i>tert</i> -butylcyclohexane					
338.2	17.99		53.2	54.4	18.0	18.4
	$9^*A1 + A14 + 3^*A15 + 3^*A4 + 3^*A16$					[41]
$\text{C}_{18}\text{H}_{38}\text{O}_2$	2-(hexadecyloxy)ethanol					
311.7	14.94	47.93				
318.5	37.32	117.2	165.1	193.7	52.3	61.7
	$A1 + 15^*A2^*B2 + A32 + A30^*B30 + 2A2$					[88]
$\text{C}_{18}\text{H}_{38}\text{O}_9$	1, ω -dimethoxyocta(oxyethylene)					
276.2	60.1		217.6	226.4	60.1	62.5
	$2^*A1 + 16^*A2^*B2 + 9^*A32$					[386]
$\text{C}_{18}\text{H}_{39}\text{AsO}_2$	di- <i>n</i> -nonylarsinic acid					
383	24.3	63.5				
399	38.1	95.5	159	160	62.4	63.8
	$2^*A1 + 16^*A2^*B2 + A142$					[381]
$\text{C}_{18}\text{H}_{54}\text{O}_9\text{Si}_9$	octadecamethylcyclononasiloxane					
246.2	25.64		104.1	104.2	25.6	25.7
	$18^*A12 + 9^*A139 + 9^*A112 + A14 + 15^*A15$					[121]
$\text{C}_{19}\text{H}_{21}\text{F}_{19}^*$	1,1,1,2,3,3,4,4,5,5,6,6,7,7,8,8-hexadecafluoro-2-(trifluoromethyl)octadecane					
274.0	1.00	3.65				
298.0	25.00	83.89	87.54	370.2	26.00	110.3
	$9^*A4^*B4 + 6^*A25 + A27 + 12^*A26 + A1 + 9^*A2$					[22]
$\text{C}_{19}\text{H}_{15}\text{N}^*$	Amphiphillic compound					
	N-phenyl benzophenone imine					
392.3	29.14		74.3	76.0	29.14	29.8
	$15^*A10 + 3^*A12 + A7 + A42$					[397]
$\text{C}_{19}\text{H}_{24}\text{O}_3^*$	3-[(2,3-dihydro-1H-inden-5-yl)carbonyl]-1,2,2-trimethylcyclopentanecarboxylic acid					
404.3	22.50		55.7	61.4	22.5	24.8
	$3^*A1 + 2^*A14 + 4^*A15 + 2^*A17 + A16 + 2^*A19 + 3^*A10 + A12 + A36^*B36 + A35$					[366]
$\text{C}_{19}\text{H}_{26}\text{O}_4$	3-(4-methoxy-2,6-dimethylbenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid					
416.7	28.31		67.9	82.6	28.3	34.4
	$6^*A1 + A14 + 2^*A15 + 2^*A17 + A16 + 2^*A10 + 2^*A12 + 2^*A11 + A36^*C36 + A35 + A32$					[366]
$\text{C}_{19}\text{H}_{26}\text{O}_6$	1,2,2-trimethyl-3-(2,4,6-trimethoxybenzoyl)cyclopentanecarboxylic acid					
432.2	29.68		68.7	96.2	29.7	41.6
	$6^*A1 + A14 + 2^*A15 + 2^*A17 + A16 + 2^*A10 + 4^*A12 + A36^*E36 + A35 + 3^*A32$					[366]
$\text{C}_{19}\text{H}_{27}\text{NO}_3^*$	2-[(3,4-dimethylphenyl)(hydroxyimino)methyl]-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester					
426.0	39.14		91.9	68.6	39.1	29.2
	$6^*A1 + A14 + 2^*A15 + 2^*A11 + A12 + 3^*A10 + 2^*A17 + A16 + A38 + A53 + A7$					[373]
$\text{C}_{19}\text{H}_{27}\text{NO}_4^*$	3-[(hydroxyimino)(4-ethoxyphenyl)methyl]-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester					
401.0	36.75		91.65	81.9	36.8	32.8
	$5^*A1 + A14 + 2^*A15 + 2^*A12 + 4^*A10 + 2^*A17 + A16 + A38 + A32 + A2 + A7 + A53$					[373]
$\text{C}_{19}\text{H}_{27}\text{NO}_5^*$	3-[(hydroxyimino)(3,4-dimethoxyphenyl)methyl]-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester					
393.0	36.20		92.1	82.2	36.2	32.3
	$6^*A1 + A14 + 2^*A15 + 3^*A12 + 3^*A10 + 2^*A17 + A16 + A38 + 2^*A32 + A7 + A53$					[373]
$\text{C}_{19}\text{H}_{36}\text{O}_2^*$	ethyl margarate (ethyl heptadecanoate)					
291.2	16.57	55.5				
298.4	36.2	115.7	171.2	189.5	52.8	56.6
	$2^*A1 + A2 + 15A2^*B2 + A38$					[391]
$\text{C}_{20}\text{H}_{13}\text{NO}_4$	1-amino-4-hydroxy-2-phenoxy-9,10-anthracenedione					
458.2	30.79		67.2	82.9	30.8	38.0
	$A14 + 3^*A15 + 10^*A10 + 4^*A12 + A31 + A45 + A32 + 2^*A114 + 4^*A19$					[315]
$\text{C}_{20}\text{H}_{17}\text{F}_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosafuoroicosane					
192	2.4	12.5				
329	6.1	19.45				
361	23.7	65.65	97.6	467.7	32.5	166.1

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{fus} S_{tpcc}$ (expt)	$\Delta_0^{fus} S_{tpcc}$ (calcd)	$\Delta_0^{fus} H_{tpcc}$ (expt)	$\Delta_0^{fus} H_{tpcc}$ (calcd)	
324.2	5.60	17.27				[17]	
355.2	21.90	61.66	78.93	467.7	27.50	166.1	
	12*A4*B4+3*A25+22*A26+A1+7*A2						[68]
	Amphiphillic compound (values represent two sets of independent measurements)						
$C_{20}H_{17}N_3O_4^*$	4,11-diamino-2-butyl-1H-naphth[2,3- <i>f</i>]isoindole-1,3,5,10(2H)-tetraone						
490.2	24.85		50.69		24.85		
	No prediction made (reporting authors express concern that the enthalpy is too small)						[315]
$C_{20}H_{19}BrS^*$	2- <i>n</i> -butyl-5-(4-bromobiphenyl-4-yl)thiophene						
501.4	21.40		42.7	101.5	21.4	50.9	
	A1+3*A2+8*A10+4*A12+A21+A14+2*A15+A131+2*A19+2*A18						[14]
$C_{20}H_{21}F_{21}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10-heneicosafuoroicosane						
317	4.0		12.62				
337	24.4	72.38	85.0	186.6	28.4	62.8	
							[17]
306.5	2.20	7.18					
336.7	26.70	79.30	86.5	186.6	28.9	62.8	
	10*A4*B4+3*A25+18*A26+A1+9*A2						[24]
	(Values represent two sets of independent measurements) Amphiphillic compound						
$C_{20}H_{21}F_{19}O^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9-nonadecafluoro-10-eicosanone						
317.9	53.17		167.3	181.4	53.2	57.7	
	9*A4*B4+3*A25+16*A26+A35+A1+9*A2						[21]
	Amphiphillic compound						
$C_{20}H_{23}F_{19}O^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9-nonadecafluoro-10-eicosanol						
346.2	3.60		10.40				
356.0	33.50	94.10	104.5	184.6	37.1	69.2	
	9*A4*B4+3*A25+16*A26+A1+9*A2+A30*B30+A3*B3						[23]
	Amphiphillic compound						
$C_{20}H_{24}$	8-(4-biphenyl)-1-octene						
291.5	21.00		72.0	107.2	21.0	31.3	
	9*A20+2*A12+A11+6*A2+A5+A6						[97]
$C_{20}H_{24}O_6$	dibenzo[18-crown-6]						
435.75	57.46		131.9	106.1	57.5	44.1	
	A14+15*A15+6*A112+4*A19+8*A10						[398]
$C_{20}H_{26}O_3$	1,2,2-trimethyl-3-[(5,6,7,8-tetrahydro-2-naphthalenyl)carbonyl]cyclopentanecarboxylic acid						
421.3	22.94		54.5	65.1	22.9	27.4	
	3*A1+2*A14+5*A15+2*A17+A16+2*A19+3*A10+A12+A36*B36+A35						[366]
$C_{20}H_{28}O_5$	3-(3,4-diethoxybenzoyl)-1,2,2-trimethylcyclopentanecarboxylic acid						
389.3	29.07		74.7	103.0	29.1	40.1	
	5*A1+A14+2*A15+2*A17+A16+4*A10+2*A2+2*A12+A36*C36+A35+2*A32						[366]
$C_{20}H_{32}O_4$	2,5-di- <i>n</i> -heptyloxy-1,4-benzoquinone						
275.8	3.6	13.05					
372.5	17.3	46.44					
406.2	38.4	94.53	154.0	140.8	59.3	57.2	
	2*A1+12*A2+A14+3*A15+2*A18*B18+2*A19+2*A32+2*A114						[342]
$C_{20}H_{40}O_2^*$	methyl nonadecanoate						
304.2	19.4	63.7					
313.2	42.8	136.8	200.5	189.5	62.2	56.6	
	2*A1+17*A2*B2+A38						[391]
$C_{20}H_0O_4$	2,2,12,12-tetramethyl-1,3,11,13-tetraoxycycloeicosane						
369.5	45.60		123.4	102.3	45.6	37.8	
	4*A1+A14+17*A15+2*A17+4*A112						[47]
$C_{20}H_{42}O_{10}$	1, ω -dimethoxynona(oxyethylene)						
289.2	73.9		255.6	249.7	73.9	62.5	
	2*A1+18*A2*B2+10*A32						[386]
$C_{20}H_{43}AsO_2^*$	di- <i>n</i> -decylarsinic acid						
380	24.5	64.4					
400	42.3	105.9	170.2	178.6	66.8	71.4	
	2*A1+18*A2*B2+A142						[381]
$C_{20}H_{50}Si_5$	decaethylcyclopentasilane						
254.8	16.3	63.97					
440.1	1.40	3.18	67.2	114.3	17.7	50.3	
	10*A1+10*A2+A14+2*A15+5*A139						[175]
$C_{20}H_{60}O_{10}Si_{10}$	eicosanomethylcyclodecasiloxane						
265.8	39.76		149.6	113.3	39.8	30.1	
	20*A1+10*A139+10*A112+A14+17*A15						[121]
$C_{21}H_{20}BrN_7O_6$	N-[2-[(2-bromo-4,6-dinitrophenyl)azo]-[(2-cyanoethyl)-2-propenylamino]-						

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
	4-methoxyphenyl] acetamide					
465.2	59.08		127.0	99.7	59.1	46.4
$\text{C}_{21}\text{H}_{20}\text{N}_4\text{O}_3^*$	4*A10+8*A12+3*A2+2*A1+A5+A6+2*A50+A21+2*A42+A32+A60+A56+A43					
403.9	28.43		70.4	101	28.4	40.8
	4-methoxy-N,N-bis(3-pyridinylmethyl)-1,3-benzenedicarboxamide					
$\text{C}_{21}\text{H}_{25}\text{F}_{19}^*$	11*A10+2*A11+3*A12+2*A41+2*A2+2*A60+A1+A32					
310.1	34.00		109.6	177.7	34.0	55.1
	1,1,1,2,3,3,4,4,5,5,6,6,7,7,8,8-hexadecafluoro-(trifluoromethyl)eicosane					
$\text{C}_{21}\text{H}_{29}\text{NO}_3^*$	9*A4*B4+6*A25+A27+12*A26+A1+11*A2					
425.0	38.37		90.3	72.5	38.4	30.8
	3-[(hydroxyimino)(5,6,7,8-tetrahydro-2-naphthalenyl)methyl-1,2,2-trimethylcyclopentanecarboxylic acid methyl ester					
$\text{C}_{21}\text{H}_{30}\text{O}$	4*A1+2*A14+5*A15+3*A10+2*A19+2*A17+A12+A16+A38+A7+A53					
404.7	5.90	14.57				
470.0	15.70	33.40	48.0	55.0	21.6	25.9
	1,1'-diadamantyl ketone					
$\text{C}_{21}\text{H}_{40}$	6*A14+2*A15+6*A16+2*A17+A35					
295.3	31.80		107.7	121.9	31.8	36.0
	<i>trans</i> -2-heptyl-6-butyldecalin					
$\text{C}_{21}\text{H}_{40}$	2*A14+4*A15+4*A16+2*A1+9*A2					
308.8	41.00		133.0	121.9	41.0	37.6
	<i>trans</i> -2-propyl-6-octyldecalin					
$\text{C}_{21}\text{H}_{42}\text{O}_2^*$	2*A14+4*A15+4*A16+2*A1+9*A2					
300.2	18.49	61.6				
309.2	43.18	139.7	201.3	189.5	61.7	56.6
	ethyl nonadecanoate					
$\text{C}_{21}\text{H}_{42}\text{O}_2^*$	2*A1+A2+17*A2*B2+A38					
319.2	73.7		231	210	73.7	61.7
	methyl eicosanoate					
$\text{C}_{21}\text{H}_{43}\text{NO}$	2*A1+18*A2*B2+A38					
348.0	16.02	46.03				
354.0	50.04	141.4	187.4	199.7	66.1	70.7
	N-propylstearamide					
$\text{C}_{21}\text{H}_{43}\text{NO}$	2*A1+16*A2*B2+A60+2A2					
316.0	6.54	20.70				
343.0	49.02	142.9	163.6	204.1	55.6	70.0
	N-heptylmyristamide					
$\text{C}_{21}\text{H}_{43}\text{NO}$	2*A1+18*A2*B2+A60					
337.0	0.07	0.21				
344.0	42.45	123.4	123.6	204.1	42.5	70.2
	N-decylundecanamide					
$\text{C}_{21}\text{H}_{43}\text{NO}$	2*A1+18*A2*B2+A60					
328.0	0.17	0.52				
341.0	66.91	196.2	196.7	204.1	67.1	69.6
	N-laurylnonanamide					
$\text{C}_{21}\text{H}_{43}\text{NO}$	2*A1+18*A2*B2+A60					
313.0	2.08	6.65				
334.0	52.68	157.7	164.4	204.1	54.8	68.2
	N-myristylheptanamide					
$\text{C}_{21}\text{H}_{43}\text{NO}$	2*A1+18*A2*B2+A60					
337.0	1.84	5.45				
350.0	56.03	160.1	165.6	201.9	57.9	70.7
	N-stearylpropanamide					
$\text{C}_{22}\text{H}_{21}\text{F}_{25}^*$	2*A1+17*A2*B2+A60+A2					
207	1.0	4.83				
342	9.5	27.78				
365	25.8	70.68	103.3	206.3	36.3	75.3
	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosafuorodocosane					
339.2	7.50	22.11				
357.2	22.20	62.15	84.26	206.3	29.7	73.7
	12*A4*B4+3*A25+22*A26+A1+9*A2					
$\text{C}_{22}\text{H}_{24}\text{O}_3^*$	Amphiphilic compound (values represent two sets of independent measurements)					
444.2	27.69	62.3	74.6	74.6	27.7	33.1
	3-[[1,1-biphenyl]-4-ylcarbonyl]-1,2,2-trimethylcyclopentanecarboxylic acid					
$\text{C}_{22}\text{H}_{25}\text{F}_{21}^*$	3*A1+A14+2*A15+2*A17+A16+9*A10+3*A12+A36*B36+A35					
334.1	6.00	17.96				
338.1	27.00	79.86	97.8	200.8	33.0	67.9
	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10-heneicosofluorodocosane					

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)
						[22]
	10*A4*B4 + 18*A26 + 3*A25 + A1 + 11*A2 Amphiphillic compound					
C ₂₂ H ₂₆ N ₂ O ₂						
394.0	(4R, 4'R, 5R, 5'R)-5,5-diphenyl-3,3',4,4'-tetramethyl-2,2'-bioxazolidine	31.9	81.0	82.4	31.9	32.5
						[387]
	2*A14 + 4*A15 + 2*A112 + 2*A119 + 6*A16 + 4*A1 + 10*A10 + 2*A11 (2R, 3R, 6R, 7R)-2,6-diphenyl-3,4,7,8-tetramethyl-cis-perhydro-[1,4]-oxazino-[3,2- <i>b</i>]-[1,4]-oxazine					
C ₂₂ H ₂₆ N ₂ O ₂						
379.4		18.4	48.5	82.4	18.4	31.3
						[387]
	2*A14 + 4*A15 + 2*A112 + 2*A119 + 6*A16 + 4*A1 + 10*A10 + 2*A11 N-phenyl-N-[1-(2-phenylethyl)-4-piperidinyl]propanamide (fentanyl)					
C ₂₂ H ₂₈ N ₂ O						
357.2		22.51	63.0	84.6	22.5	30.2
						[296]
	A14 + 3*A15 + A16 + A119 + 3*A2 + 10*A10 + A11 + A12 + A1 + A125 4- <i>n</i> -octyloxy-N-(4-methoxybenzylidene)aniline					
C ₂₂ H ₂₉ NO ₂						
377.3		42.29	112.1	127.0	42.3	47.9
						[141]
	2*A1 + 7*A2 + 2*A32 + 8*A10 + 4*A12 + A6 + A42 N-[4-(methoxymethyl)-1-[2-(2-thienyl)ethyl]-4-piperidinyl]-N-phenylpropanamide (sufentanil)					
C ₂₂ H ₃₀ N ₂ O ₂ S						
370.2		23.85	64.4	102.3	23.9	37.9
						[296]
	2*A14 + 5*A15 + A17 + A119 + A59 + 5*A10 + A12 + 2*A1 + 4*A2 + A32 + 2*A18 + A18*B18 + A19 + A131 2,5-di- <i>n</i> -octyloxy-1,4-benzoquinone					
C ₂₂ H ₃₆ O ₄						
358.2		9.4	26.24			
405.8		43.0	106.0	132.2	52.4	62.9
						[342]
	2*A1 + 14*A2 + A14 + 3*A15 + 2*A18*B18 + 2*A19 + 2*A32 + 2*A114 3,3,6,6,10,10,13,13-octamethylcyclotetradecane-1,8-dione					
C ₂₂ H ₄₀ O ₂						
492.2		24.7	50.2	73.7	24.7	36.3
						[115]
	8*A1 + 4*A17 + 2*A114 + A14 + 11*A15 1-docosanol					
C ₂₂ H ₄₆ O						
333.9		17.24	50.72			
345.2		46.57	134.9	185.6	63.8	74.08
						[88]
	A1 + 21*A2*B2 + A30 di- <i>n</i> -undecylarsinic acid					
C ₂₂ H ₄₇ AsO ₂ *						
384		30.0	78.2			
396		45.1	113.9	192.1	75.1	78.1
						[381]
	2*A1 + 20*A2*B2 + A142 docosamethylcycloundecasiloxane					
C ₂₂ H ₆₆ O ₁₁ Si ₁₁						
216.2		17.73		82.0	122.4	17.7
						[121]
	22*A1 + 11*A139 + 11*A112 + A14 + 19*A15 2-[[4-[(2-acetoxy)ethyl]butylamino]-2-methylphenyl]azo]-5-nitro-1,3-benzenedicarbonitrile					
C ₂₃ H ₂₄ N ₆ O ₄						
424.2		37.88	89.3	105.7	37.9	44.8
						[315]
	5*A10 + 6*A12 + A11 + 2*A56 + A50 + 3*A1 + 5*A2 + 2*A42 + A43 + A38 N-[5-[bis[(2-acetyloxy)ethyl]amino]-2-[(2-bromo-4,6-dinitrophenyl)azo]-4-methoxyphenyl]acetamide					
C ₂₃ H ₂₅ BrN ₆ O ₁₀						
421.2		57.28	136.0	117.1	57.3	49.3
						[315]
	4*A10 + 8*A12 + 4*A1 + 4*A2 + 2*A38 + A21 + 2*A50 + 2*A42 + A32 + A60 + A43 4- <i>n</i> -octyloxy-N-(3,5-dimethylbenzylidene)aniline					
C ₂₃ H ₃₁ NO						
324.7		37.7	116.2	120.8	37.7	39.2
						[141]
	3*A1 + 7*A2 + A32 + 7*A10 + 2*A11 + 3*A12 + A6 + A42 4- <i>n</i> -octyloxy-N-(3,5-dimethoxybenzylidene)aniline					
C ₂₃ H ₃₁ NO ₃						
316.3		35.3	111.6	134.4	35.3	42.5
						[141]
	3*A1 + 7*A2 + 3*A32 + 7*A10 + 5*A12 + A6 + A42 <i>trans</i> -2-heptyl-6-hexyldecalin					
C ₂₃ H ₄₄						
312.2		38.9	124.6	136.1	38.9	42.5
						[40]
	2*A1 + 4*A15 + 4*A16 + 2*A1 + 11*A2 <i>trans</i> -2-pentyl-6-octyldecalin					
C ₂₃ H ₄₄						
314.2		43.5	138.5	136.1	43.5	42.8
						[40]
	2*A1 + 4*A15 + 4*A16 + 2*A1 + 11*A2 methyl behenate (methyl docosanoate)					
C ₂₃ H ₄₆ O ₂ *						
327.2		82.3	231	210	82.3	67.1
						[391]
	2*A1 + 20*A2*B2 + A38 4,4'-bis-(2-thienylmethylidenamino)- <i>trans</i> -stilbene					
C ₂₄ H ₁₈ N ₂ S ₂ *						
567.2		44.90	79.16			
580.2		0.20	0.34	79.5	45.1	
						[86]
	No prediction made (forms liquid crystal)					
	1,2-bis-[5-(β -azastyryl)-2-thienyl]- <i>trans</i> -ethylene					
C ₂₄ H ₁₈ N ₂ S ₂ *						
501.2		45.90	91.6	108.4	45.9	54.3
						[86]
	10*A10 + 2*A12 + 4*A6 + 2*A42 + 2*A14 + 4*A15 + 4*A18 + 4*A19 + 2*A131 1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorotetracosane					
C ₂₄ H ₂₅ F ₂₅ *						

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} S_{tpcc}$ (calcd)	$\Delta_0^{T_{fus}} H_{tpcc}$ (expt)	$\Delta_0^{T_{fus}} H_{tpcc}$ (calcd)	
352.1	10.00	28.40					
364.1	26.00	71.41	99.8	220.5	36.0	80.3	
	12*A4*B4+3*A25+22*26+A1+11*A2 Amphiphillic compound						[22]
$C_{24}H_{25}F_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluoro-14-methyltricosane						
220.0	9.00	34.61					
347.1	25.00	72.02	106.6	207.5	34.0	72.0	
	12*A4*B4+3*A25+22*A26+2*A1+A3+9*A2 Amphiphillic compound						[22]
$C_{24}H_{30}O_4$	2,2'-diphenyl-bi(5,5-dimethyl-1,3-dioxan-2-yl)						
507.1	49.8		98.2	80.6	49.8	40.9	
	2*A14+6*A15+4*A17+4*A112+10*A10+2*A11+4*A1 8-[4-(4'-n-butylbiphenyl)]-1-octene						[385]
$C_{24}H_{32}^*$	2.20	8.85					
248.6	9.60	30.4	39.3	129.1	11.8	40.7	
315.6						[97]	
	8*A10+2*A12+2*A11+A5+A6+A1+9*A2 Forms liquid crystal						
$C_{24}H_{40}O_4$	2,5-di-n-nonyloxy-1,4-benzoquinone						
352.6	8.0	22.69					
383.8	24.2	63.05					
402.7	47.1	117.0	202.7	169.2	79.3	68.1	
	2*A1+16*A2+A14+3*A15+2*A18*B18+2*A19+2*A32+2*A114 dibenzo[24-crown-8]						[342]
$C_{24}H_{40}O_8$	16.6	46.9					
354.1	52.25	139.2	186.1	130.1	68.85	49.1	
375.4						[398, 399]	
	A14+15*A15+6*A112+4*A19+8*A10 <i>trans</i> -2,6-diheptyldecalin						
$C_{24}H_{44}$	40.17		123.0	143.2	40.2	46.8	
326.7						[40]	
	2*A1+4*A15+4*A16+2*A1+11*A2 3,3,7,7,11,11,15,15-octamethylcyclohexadecane-1,9-dione						
$C_{24}H_{44}O_2$	34.30		81.0	81.1	34.3	34.3	
423.2						[115]	
	8*A1+4*A17+2*A114+A14+13*A15 2-(docosanoxy)ethanol						
$C_{24}H_{50}O_2^*$	12.92	40.73					
317.2	43.93	130.8	171.5	249.5	56.9	83.8	
335.9						[88]	
	A1+21*A2*B2+A32+A30*B30 di-n-dodecylarsinic acid						
$C_{24}H_{51}AsO_2^*$	31.4	81.5					
385	49.4	124.1	205.7	215.8	80.8	85.9	
398						[381]	
	2*A1+22*A2*B2+A142 tetracosamethylcyclododecasiloxane						
$C_{24}H_{72}O_{12}Si_{12}$	15.45		65.97	131.5	15.5	30.8	
234.2						[121]	
	24*A1+12*A139+12*A112+A14+21*A15 1,2-bis-[5-(4-methoxy- β -azastyryl)]-2-thienyl]- <i>trans</i> -ethylene						
$C_{26}H_{22}N_2O_2S_2^*$	63.50	118.0					
538.2	0.80	1.41	119.4	123.2	64.30	69.88	
567.2							
	2*A1+8*A10+4*A12+4*A6+2*A42+2*A14+4*A15+4*A18+4*A19+2*A131+2*A32 Forms liquid crystal						[86]
$C_{26}H_{29}F_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosaffluorohexacosane						
363	16.3	44.90					
366	26.1	71.31	116.2		42.4		
						[17]	
359.2	26.0	72.38			26.0		
	Amphiphillic compound (values represent two sets of independent measurements) <i>trans</i> -1-(4-heptanoylphenyl)-4-heptylcyclohexane						[68]
$C_{26}H_{42}O^*$	16.49	48.05					
343.2	7.71	22.37	70.4	145.5	24.2	50.2	
344.7						[25]	
	A14+3*A15+2*A16+4*A10+A11+A12+2*A1+11*A2+A35 Forms liquid crystal						
$C_{26}H_{48}O_2$	4,4,7,7,13,13,16,16-octamethylcyclooctadecane-1,10-dione						
492.2	50.60	102.8		88.5	50.6	43.6	
						[115]	
	8*A1+4*A17+2*A114+A14+15*A15 1-hexacosanol						
$C_{26}H_{54}O$	16.74	50.39					
332.2	67.78	192.7	243.1	251.8	84.5	88.6	
351.7						[78]	
	A1+25*A2*B2+A30						

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{\text{fus}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{\text{fus}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{\text{fus}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{\text{fus}} H_{\text{tpcc}}$ (calcd)
$\text{C}_{26}\text{H}_{55}\text{AsO}_2^*$	di- <i>n</i> -tridecylarsinic acid					
388	36.5	94.0				
396	52.7	133.1	227.2	234.4	89.2	92.8
	2*A1+24*A2*B2+A142					
$\text{C}_{27}\text{H}_{42}\text{Cl}_2\text{N}_2\text{O}_6^*$	chloramphenicol palmitate polymorph A					
367.3	51.04	0	139	188.6	51.04	69.2
	chloramphenicol palmitate polymorph B					
360.8	41.3	0	112.5	188.6	41.3	69.2
	4*A10+A11+A12+A50+A30*F30+2*A22*F22+A60+A38+A2+3*A3*B3+A1+14*A2					
$\text{C}_{28}\text{H}_{22}\text{N}_2\text{O}_2$	1,4-[bis[(4-methylphenyl)amino]-9,10-anthracenedione					
491.2	36.59	74.5		71.5	36.6	35.1
	3*A15+A14+14*A10+4*A19+2*A114+4*A12+2*A1+2*A11+2*A44					
$\text{C}_{28}\text{H}_{31}\text{F}_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorooctacosane					
263.2	43.10		163.8	248.9	43.10	65.5
	12*A4*B4+3*A25+22*A26+A1+15*A2					
$\text{C}_{28}\text{H}_{48}\text{O}^*$	Amphiphilic compound					
	<i>trans</i> -1-heptyl-4-(4-nonanoylphenyl)cyclohexane					
343.4	20.8	60.6				
353.3	11.32	32.1	92.6	159.7	32.1	56.4
	A14+3*A15+2*A16+4*A10+A11+A12+2*A1+13*A2+A35					
$\text{C}_{28}\text{H}_{48}\text{O}_4$	Forms liquid crystal					
	2,5-di- <i>n</i> -undecyloxy-1,4-benzoquinone					
367.4	12.9	35.11				
390.0	28.4	72.8				
397.2	52.1	131.2	239.1	241.6	93.4	96.0
	2*A1+20*A2*B2+A14+3*A15+2*A18*B18+2*A19+2*A32+2*A114					
$\text{C}_{28}\text{H}_{52}\text{O}_2$	4,4,8,8,14,14,18,18-octamethylcycloeicosane-1,11-dione					
418.2	36.80		88.0	95.9	36.8	40.1
	8*A1+4*A17+2*A114+A14+17*A15					
$\text{C}_{28}\text{H}_{59}\text{AsO}_2^*$	di- <i>n</i> -tetradecylarsinic acid					
390	39.3	100.6				
397	58.2	146.6	247.2	253.0	97.5	100.5
	2*A1+26*A2*B2+A142					
$\text{C}_{29}\text{H}_{41}\text{NO}_4$	17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-6,14-ethenomorphinan-7-methanol					
491.3	26.80	54.55		68.6	26.80	33.7
	6*A14+2*A15+5*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+A31+A32+3*A19+A12+2*A10+A2					
$\text{C}_{30}\text{H}_{37}\text{F}_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorotricontane					
365.2	47.80		130.9	263.1	47.80	96.1
	12*A4*B4+3*A25+22*A26+A1+17*A2					
$\text{C}_{30}\text{H}_{56}\text{O}_2$	5,5,8,8,16,16,19,19-octamethylcyclohexacosane-1,12-dione					
442.2	47.70		107.9	95.9	47.7	42.4
	8*A1+4*A17+2*A114+A14+17*A15					
$\text{C}_{30}\text{H}_{60}\text{O}_{15}$	45-crown-15					
311.2	70.6		227	206.8	70.6	64.3
	A14+42*A15+15*A112					
$\text{C}_{30}\text{H}_{63}\text{AsO}_2^*$	di- <i>n</i> -pentadecylarsinic acid					
390	46.4	119				
396	63.6	160.5	279.6	271.6	110.0	107.6
	2*A1+28*A2*B2+A142					
$\text{C}_{31}\text{H}_{43}\text{NO}_5$	3-(acetyloxy)-17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-6,14-ethenomorphinan-7-methanol					
440.3	22.40	50.9		68.9	22.4	30.3
	6*A14+2*A15+6*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+3*A19+A12+2*A10+A2+A38					
$\text{C}_{32}\text{H}_{34}$	1,8-bis-(4-biphenyl)octane					
415.2	56.00		134.9	140.8	56.0	58.5
	18*A10+4*A12+2*A11+8*A2					
$\text{C}_{32}\text{H}_{34}^*$	1,8-bis[4(4'-ethylbiphenyl)]butane					
454.2	46.00		101.3	127.8	46.0	58.1
	2*A1+6*A2+16*A10+4*A12+4*A11					
$\text{C}_{32}\text{H}_{41}\text{F}_{25}^*$	1,1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-pentacosofluorodotriacontane					
369.2	43.40		117.6	277.3	43.4	102.3
	12*A4*B4+3*A25+22*A26+A1+19*A2					
$\text{C}_{32}\text{H}_{64}\text{I}_6$	48-crown-16					
312.2	59.1		189.4	219.1	59.1	68.4
	A14+45*A15+16*A112					

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

<i>T</i> (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)	
C ₃₂ H ₄₅ NO ₅	410.2	17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-(1-oxopropoxy)-6,14-ethenomorphinan-7-methanol		27.10	76.0	27.10	31.2
		6*A14+2*A15+6*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+3*A19+A12+2*A10+2*A2+A38		66.07			
C ₃₂ H ₆₀ O ₂	380.2	5,5,9,9,17,17,21,21-octamethylcyclotetrasiloxane-1,13-dione		32.60	110.7	32.6	42.1
		8*A1+4*A17+2*A114+A14+21*A15		85.7			
C ₃₂ H ₆₇ AsO ₂ *	389 395	di- <i>n</i> -hexadecylarsinic acid		47.4	121.9		
		2*A1+30*A2*B2+A142		169.2	291	290.2	114.2
C ₃₃ H ₄₇ NO ₅	422.1	17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-(1-oxobutoxy)-6,14-ethenomorphinan-7-methanol		32.40	83.1	32.40	35.1
		6*A14+2*A15+6*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+3*A19+A12+2*A10+3*A2+A38		76.76			
C ₃₄ H ₃₁ CIN ₂ O ₃ *	442.2	spiro[isobenzofuran-1(3H),9'(9H)-7'-chloro-6'-(methylcyclohexylamino)-3'-methyl-2'-anilinoxanthene]-3-one		49.0	110.8	49.0	40.5
		3*A14+8*A15+2*A1+13*A10+4*A12+A11+6*A19+A16+A17+A112+A115+A43+A44+A22*E22		83.8	91.5		
C ₃₄ H ₃₂ N ₂ O ₃	476.2	spiro[isobenzofuran-1(3H),9'(9H)-6'-(methylcyclohexylamino)-3'-methyl-2'-anilinoxanthene]-3-one		39.9	90.2	39.9	43.0
		3*A14+8*A15+2*A1+14*A10+3*A12+A11+6*A19+A16+A17+A112+A115+A43+A44		83.8	90.2		
C ₃₄ H ₃₈ *	393.2 422.2	1,6-bis-[4-(4'-ethylbiphenyl)]hexane		3.90	9.92		
		2*A1+8*A2+16*A10+4*A11+4*A12		82.90	92.83	142.0	38.9
C ₃₄ H ₄₉ NO ₅	379.1	17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-(1-oxopentoxo)-6,14-ethenomorphinan-7-methanol		24.00	63.31	24.0	34.2
		6*A14+2*A15+6*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+3*A19+A12+2*A10+4*A2+A38		63.31	90.2		
C ₃₄ H ₆₈ O ₁₇	301.2	51-crown-17		66.6	221.1	66.6	69.7
		A14+48*A15+17*A112			231.4		
C ₃₄ H ₇₁ AsO ₂ *	390 393	di- <i>n</i> -octadecylarsinic acid		50.9	130.6		
		2*A1+32*A2*B2+A142		174.5	305.1	308.8	119.5
C ₃₅ H ₅₁ NO ₅	352.6	17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-(1-oxohexyloxy)-6,14-ethenomorphinan-7-methanol		22.60	64.1	22.6	35.0
		6*A14+2*A15+6*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+3*A19+A12+2*A10+5*A2+A38		64.1	97.3		
C ₃₆ H ₄₂ *	402.2 413.2	1,8-bis-[4-(4'-ethylbiphenyl)]octane		8.40	20.89		
		2*A1+16*A10+4*A12+4*A11+10*A2		101.6	122.5	156.2	50.4
C ₃₆ H ₄₂ *	404.2 464.2	1,4-bis-[4-(4'- <i>n</i> -butylbiphenyl)]butane		12.00	29.68		
		2*A1+16*A10+4*A12+4*A11+10*A2		51.70	81.3	142.0	36.0
C ₃₆ H ₅₃ NO ₅	360.0	17-(cyclopropylmethyl)- α -(1,1-dimethylethyl)-4,5-epoxy-18,19-dihydro-6-methoxy- α -methyl-(1-oxoheptyloxy)-6,14-ethenomorphinan-7-methanol		19.30	53.61	19.30	37.6
		6*A14+2*A15+6*A1+2*A4+A30*E30+4*A16+3*A17+A119+A112+3*A19+A12+2*A10+6*A2+A38		53.61	104.4		
C ₃₆ H ₆₄ O ₄	385.9 393.5	2,5-di- <i>n</i> -pentadecyloxy-1,4-benzoquinone		21.7	56.23		
		2*A1+28*A2*B2+A14+3*A15+2*A18+2*A19+2*A32+2*A14		258.5	314.7	316.0	123.4
C ₃₆ H ₇₄ O ₁₆	317.2	54-crown-18		81.6	257.2	81.6	77.3
		Forms liquid crystal			243.7		

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pcc} (expt)	ΔS_{pcc} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpcc}}$ (calcd)	
$C_{36}H_{74}O_{18}$	301.2	A14+51*A15+18*A112 1, ω -dimethoxyheptadeca(oxyethylene)		453.5	436.2	136.6	
		136.6					131.4
$C_{36}H_{75}AsO_2^*$	394	2*A1+34*A2*B2+18*A32 di- <i>n</i> -nonadecylarsinic acid		327.2	327.4	128.9	
		128.9					129.0
$C_{38}H_{68}O_4$	357.7 370.9 389.0 394.2	2*A1+34*A2*B2+A142 2,5-di- <i>n</i> -hexadecyloxy-1,4-benzoquinone		316.1	334.6	122.9	
		6.8	18.73				133.0
		14.1	38.02				
		19.0	48.84				
$C_{38}H_{78}O_{19}$	305.2	2*A1+30*A2*B2+A14+3*A15+2*A18+2*A19+2*A32+2*A114 1, ω -dimethoxyoctadeca(oxyethylene)		513.5	459.5	156.7	
		156.7					140.2
$C_{40}H_{40}N_2O_4^*$	432.4 513.8 563.3	2*A1+36*A2*B2+19*A32 3,3',4,4'-biphenyltetracarboxy- <i>N,N'</i> -bis-(4- <i>n</i> -hexylphenyl)diimide		113.9		55.60	
		19.9	46.02				[386]
		26.2	50.99				
		9.50	16.86				
$C_{40}H_{50}^*$	398.2 414.2	No prediction made. Forms liquid crystal 1,8-bis[4(4'- <i>n</i> -butylbiphenyl)]octane		97.2	184.6	40.0	
		13.0	32.65				76.5
$C_{40}H_{72}O_4$	383.6 395.3	2*A1+16*A10+4*A12+4*A11+14*A2 Forms liquid crystal 2,5-di- <i>n</i> -heptadecyloxy-1,4-benzoquinone		339.7	353.1	133.9	
		13.0	33.89				139.6
		120.9	305.8				
		2*A1+32*A2*B2+A14+3*A15+2*A18+2*A19+2*A32+2*A114 3,3',4,4'-biphenyltetracarboxy- <i>N,N'</i> -bis-(4- <i>n</i> -heptylphenyl)diimide					
$C_{42}H_{44}N_2O_4^*$	411.0 504.9 560.8	18.80 24.70 11.10		114.5		54.6	
		45.74					[87]
		48.92					
		19.79					
$C_{44}H_{48}N_2O_4^*$	428.5 499.2 553.5	Forms liquid crystal 3,3',4,4'-biphenyltetracarboxy- <i>N,N'</i> -bis-(4- <i>n</i> -octylphenyl)diimide		142.3		65.9	
		36.10	84.25				[87]
		21.30	42.67				
		8.50	15.36				
$C_{44}H_{80}O_4$	385.5 396.2	Forms liquid crystal 2,5-di- <i>n</i> -nonadecyloxy-1,4-benzoquinone		380.2	390.5	150.2	
		16.2	42.0				154.7
$C_{44}H_{90}$	360.9	2*A1+36*A2*B2+A14+3*A15+2*A18+2*A19+2*A32+2*A114 <i>n</i> -tetratetracontane		403.1	425.8	145.5	
		145.5					153.7
$C_{50}H_{102}$	366.9	2*A1+42*A2*B2 <i>n</i> -pentacontane		504.2	481.6	185.0	
		185.0					176.7
$C_{52}H_{106}O_{26}$	316.2	2*A1+48*A2*B2 1, ω -dimethoxypentacos(oxyethylene)		663.3	622.7	209.7	
		209.7					196.9
$C_{54}H_{108}O_{27}$	314.2	2*A1+50*A2*B2+26*A32 81-crown-27		495.4	354.4	155.6	
		155.6					111.3
$C_{56}H_{114}O_{28}$	315.2	A14+78*A15+27*A112 1, ω -dimethoxyheptacos(oxyethylene)		712.6	669.3	224.6	
		224.6					210.9
$C_{92}H_{186}O_{46}$	324.2	2*A1+54*A2*B2+28*A32 1, ω -dimethoxypentatetracos(oxyethylene)		1156.3	1089.0	374.8	
		374.8					353.0
$C_{100}H_{202}$	365.5 388.5	2*A1+90*A2*B2+46*A32 <i>n</i> -hectane		1004.0	946.7	386.8	
		54.8	149.9				[386]
		331.8	854.1				
$C_{192}H_{386}$		2*A1+98*A2*B2 <i>n</i> -dononacontahectane				368.8	

TABLE 7. Calculated and experimental phase change enthalpies and entropies of test solids—Continued

T (K)	ΔH_{pce} (expt)	ΔS_{pce} (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (expt)	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (calcd)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$ (expt)	$\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$ (calcd)
399.1	698.9		1751.2	1802.4	698.9	719.3
	2*A1 + 190*A2*B2 (Author's noted a small premelting transition)					[344]

^aUnits for $\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ and $\Delta_0^{T_{\text{fus}}} H_{\text{tpce}}$ are $\text{J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$ and $\text{kJ}\cdot\text{mol}^{-1}$, respectively; compounds with molecular formulas characterized with an asterisk(*) were not included in generating the statistics. As noted in the table, some of these compounds exhibit liquid crystal behavior, others display amphiphilic behavior, group values for some are not currently available, the error between experimental and calculated total phase change entropy exceeded three standard deviations or some may have been added at a later date.

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