

UNITED STATES DEPARTMENT OF COMMERCE • MAURICE H. STANS, *Secretary*

NATIONAL BUREAU OF STANDARDS • LEWIS M. BRANSCOMB, *Director*

**Ionization Potentials and
Ionization Limits Derived from
the Analyses of Optical Spectra**

Charlotte E. Moore

Office of Standard Reference Data
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FOREWORD

The National Standard Reference Data System provides effective access to the quantitative data of physical science, critically evaluated and compiled for convenience, and readily accessible through a variety of distribution channels. The System was established in 1963 by action of the President's Office of Science and Technology and the Federal Council for Science and Technology, with responsibility to administer it assigned to the National Bureau of Standards.

The System now comprises a complex of data centers and other activities, carried on in academic institutions and other laboratories both in and out of government. The independent operational status of existing critical data projects is maintained and encouraged. Data centers that are components of the NSRDS produce compilations of critically evaluated data, critical reviews of the state of quantitative knowledge in specialized areas, and computations of useful functions derived from standard reference data. In addition, the centers and projects establish criteria for evaluation and compilation of data and make recommendations on needed improvements in experimental techniques. They are normally closely associated with active research in the relevant field.

The technical scope of the NSRDS is indicated by the principal categories of data compilation projects now active or being planned: nuclear properties, atomic and molecular properties, solid state properties, thermodynamic and transport properties, chemical kinetics, and colloid and surface properties.

The NSRDS receives advice and planning assistance from the National Research Council of the National Academy of Sciences-National Academy of Engineering. An overall Review Committee considers the program as a whole and makes recommendations on policy, long-term planning, and international collaboration. Advisory Panels, each concerned with a single technical area, meet regularly to examine major portions of the program, assign relative priorities, and identify specific key problems in need of further attention. For selected specific topics, the Advisory Panels sponsor subpanels which make detailed studies of users' needs, the present state of knowledge, and existing data resources as a basis for recommending one or more data compilation activities. This assembly of advisory services contributes greatly to the guidance of NSRDS activities.

The NSRDS-NBS series of publications is intended primarily to include evaluated reference data and critical reviews of long-term interest to the scientific and technical community.

LEWIS M. BRANSCOMB, *Director*

Contents

	Page
Foreword.....	III
Introduction.....	1
Table 1. Ionizational potentials.....	2
Table 2. Ionization limits and lowest terms.....	6a-6i
Table 3. Bibliography.....	6

Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra

Charlotte E. Moore

A current table of ionization potentials expressed in electron volts and a detailed table giving the limits from which they have been derived are presented. For each spectrum the ground term is given, with the limit as the ground state. The energy levels of terms of the lowest configuration determined from ground state zero, are also included for selected spectra. The literature references used for each spectrum are indicated by number and listed in a bibliography with some 200 entries.

The latest recommended conversion factor (cm^{-1} to eV) 0.000123981 corresponding to $1 \text{ eV} = 8065.73 \text{ cm}^{-1}$ has been used throughout.

Key words: Atomic spectra, ground terms; ground terms, atomic spectra; ionization limits; ionization potentials.

The data in the Volumes on "Atomic Energy Levels" (AEL) [135], [136], [137], include the ionization limits known for individual spectra. The latest table of ionization potentials calculated from these limits was published as Table 34 in Volume III (1958). Much work has been done since then and there has been a steady demand for a revision of this Table.

A fairly comprehensive general bibliography has recently been published [194] which lists for each spectrum the literature references on analyses of atomic spectra dating from the entries in the respective Volume of "AEL" (1949), (1952), (1958), well into 1968. The present compendium is based largely on the references in this Bibliography, with some, but probably not all, later material.

The reliability of the data recorded in the literature is often difficult to appraise. In cases where long series are known in the various spectra, the ionization potentials are well determined. With these as key points, good values can be derived by extrapolation or interpolation along isoelectronic sequences, or by comparison along the rows in the Periodic Chart for spectra of similar stages of ionization. Frequently, however, authors give values of ionization potentials without stating the conversion factor used and without describing clearly how the quoted value was obtained.

For this reason, the present paper includes not only the ionization potentials in eV, but also, the limits in cm^{-1} from which these have been derived. Table 1 gives the ionization potentials in eV for each spectrum.

The conversion factor taken from [195] was used for Table 1, since it is the value currently recommended by the National Academy of Sciences-National Research Council. However, recent measurements [200] suggest that this value may be in error by about 30 parts per million. Therefore, it should be understood that all of the significant figures included in Table 1 may not be meaningful

in an absolute sense. This applies particularly to entries with magnitudes greater than 100 eV.

All limits have been multiplied by the factor 0.000123981 to obtain the entries in Table 1, i.e., $1 \text{ eV} = 8065.73 \text{ cm}^{-1}$. The factor used in "AEL" was 0.00012395 and has been superseded. As a result, in the present table there are systematic differences from the 1958 Table, caused by the change in the conversion factor, as well as the differences caused by improved values of the limits.

Italics denote ionization potentials derived from limits that are bracketed in Table 2.

In compiling Table 1 the author has attempted to indicate roughly the various degrees of accuracy of the limits. Those based on well-established series deserve the greatest weight. When the ionization potential is given to three places, it is felt that the third place is meaningful. The two- and one-place entries are less well defined, but it is hoped that they have some significance. The limits of error assigned by the various investigators provide a general criterion, but these are given for comparatively few spectra. Users should, therefore, consult the limits given in Table 2 and the references in order to evaluate the data for individual spectra.

Table 2 contains the basic data for each spectrum. As in Table 1, the successive stages of ionization are indicated at the heading of each column: I, denoting first spectra (neutral atoms); II, second spectra (singly ionized atoms), etc. The elements are arranged in order of increasing atomic number, Z . The ground state is indicated for each spectrum, together with the ionization limit in cm^{-1} . In every case this limit refers to the ground state of the ion in the next higher stage of ionization. The limits of error are quoted from the original authors. Although not specifically defined, these afford a general guide as to the reliability of the limit.

Although all limits are based on data derived from the analyses of optical spectra, they are determined in various ways, since reliable series are

TABLE I. Ionization potentials*

Z	Element	Spectrum									
		I	II	III	IV	V	VI	VII	VIII	IX	X
1	H	13.598									
2	He	24.587	54.416								
3	Li	5.392	75.638	122.451							
4	Be	9.322	18.211	153.893	217.713						
5	B	8.298	25.154	37.930	259.368	340.217					
6	C	11.260	24.383	47.887	64.492	392.077	489.981				
7	N	14.534	29.601	47.448	77.472	97.888	552.057	667.029			
8	O	13.618	35.116	54.934	77.412	113.896	138.116	739.315	871.387		
9	F	17.422	34.970	62.707	87.138	114.240	157.161	185.182	953.886	1103.089	
10	Ne	21.564	40.962	63.45	97.11	126.21	157.93	207.27	239.09	1195.797	1362.164
11	Na	5.139	47.286	71.64	98.91	138.39	172.15	208.47	264.18	299.87	1465.091
12	Mg	7.646	15.035	80.143	109.24	141.26	186.50	224.94	265.90	327.95	367.53
13	Al	5.986	18.828	28.447	119.99	153.71	190.47	241.43	284.59	330.21	398.57
14	Si	8.151	16.345	33.492	45.141	166.77	205.05	246.52	303.17	351.10	401.43
15	P	10.486	19.725	30.18	51.37	65.023	220.43	263.22	309.41	371.73	424.50
16	S	10.360	23.33	34.83	47.30	72.68	88.049	280.93	328.23	379.10	447.09
17	Cl	12.967	23.81	39.61	53.46	67.8	97.03	114.193	348.28	400.05	455.62
18	Ar	15.759	27.629	40.74	59.81	75.02	91.007	124.319	143.456	422.44	478.68
19	K	4.341	31.625	45.72	60.91	82.66	100.0	117.56	154.86	175.814	503.44
20	Ca	6.113	11.871	50.908	67.10	84.41	108.78	127.7	147.24	188.54	211.270
21	Sc	6.54	12.80	24.76	73.47	91.66	111.1	138.0	158.7	180.02	225.32
22	Ti	6.82	13.58	27.491	43.266	99.22	119.36	140.8	168.5	193.2	215.91
23	V	6.74	14.65	29.310	46.707	65.23	128.12	150.17	173.7	205.8	230.5
24	Cr	6.766	16.50	30.96	49.1	69.3	90.56	161.1	184.7	209.3	244.4
25	Mn	7.435	15.640	33.667	51.2	72.4	95	119.27	196.46	221.8	243.3
26	Fe	7.870	16.18	30.651	54.8	75.0	99	125	151.06	235.04	262.1
27	Co	7.86	17.06	33.50	51.3	79.5	102	129	157	186.13	276
28	Ni	7.635	18.168	35.17	54.9	75.5	108	133	162	193	224.5
29	Cu	7.726	20.292	36.83	55.2	79.9	103	139	166	199	232
30	Zn	9.394	17.964	39.722	59.4	82.6	108	134	174	203	238
31	Ga	5.999	20.51	30.71	64						
32	Ge	7.899	15.934	34.22	45.71	93.5					
33	As	9.81	18.633	28.351	50.13	62.63	127.6				
34	Se	9.752	21.19	30.820	42.944	68.3	81.70	155.4			
35	Br	11.814	21.8	36	47.3	59.7	88.6	103.0	192.8		
36	Kr	13.999	24.359	36.95	52.5	64.7	78.5	111.0	126	230.9	
37	Rb	4.177	27.28	40	52.6	71.0	84.4	99.2	136	150	277.1
38	Sr	5.695	11.030	43.6	57	71.6	90.8	106	122.3	162	177
39	Y	6.38	12.24	20.52	61.8	77.0	93.0	116	129	146.2	191
40	Zr	6.84	13.13	22.99	34.34	81.5					
41	Nb	6.88	14.32	25.04	38.3	50.55	102.6	125			
42	Mo	7.099	16.15	27.16	46.4	61.2	68	126.8	153		

TABLE I. Ionization potentials* – Continued

Spectrum – Continued											Z
XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX	XX	XXI	
											1
											2
											3
											4
											5
											6
											7
											8
											9
											10
1648.659											11
1761.802	1962.613										12
442.07	2085.983	2304.080									13
476.06	523.50	2437.676	2673.108								14
479.57	560.41	611.85	2816.943	3069.762							15
504.78	564.65	651.63	707.14	3223.836	3494.099						16
529.26	591.97	656.69	749.74	809.39	3658.425	3946.193					17
538.95	618.24	686.09	755.73	854.75	918	4120.778	4426.114				18
564.13	629.09	714.02	787.13	861.77	968	1034	4610.955	4933.931			19
591.25	656.39	726.03	816.61	895.12	974	1087	1157	5129.045	5469.738		20
249.832	685.89	755.47	829.79	926.00							21
265.23	291.497	787.33	861.33	940.36							22
255.04	308.25	336.267	895.58	974.02							23
270.8	298.0	355	384.30	1010.64							24
286.0	314.4	343.6	404	435.3	1136.2						25
290.4	330.8	361.0	392.2	457	489.5	1266.1					26
305	336	379	411	444	512	546.8	1403.0				27
321.2	352	384	430	464	499	571	607.2	1547			28
266	368.8	401	435	484	520	557	633	671	1698		29
274	310.8	419.7	454	490	542	579	619	698	738	1856	30
											31
											32
											33
											34
											35
											36
											37
324.1											38
206	374.0										39
											40
											41
											42

TABLE I. *Ionization potentials** – Continued

Z	Element	Spectrum									
		I	II	III	IV	V	VI	VII	VIII	IX	X
43	Tc	7.28	15.26	29.54							
44	Ru	7.37	16.76	28.47							
45	Rh	7.46	18.08	31.06							
46	Pd	8.34	19.43	32.93							
47	Ag	7.576	21.49	34.83							
48	Cd	8.993	16.908	37.48							
49	In	5.786	18.869	28.03	54						
50	Sn	7.344	14.632	30.502	40.734	72.28					
51	Sb	8.641	16.53	25.3	44.2	56	108				
52	Te	9.009	18.6	27.96	37.41	58.75	70.7	137			
53	I	10.451	19.131	33							
54	Xe	12.130	21.21	32.1							
55	Cs	3.894	25.1								
56	Ba	5.212	10.004								
57	La	5.577	11.06	19.175							
58	Ce	5.47	10.85	20.20	36.72						
59	Pr	5.42	10.55	21.62	38.95	57.45					
60	Nd	5.49	10.72								
61	Pm	5.55	10.90								
62	Sm	5.63	11.07								
63	Eu	5.67	11.25								
64	Gd	6.14	12.1								
65	Tb	5.85	11.52								
66	Dy	5.93	11.67								
67	Ho	6.02	11.80								
68	Er	6.10	11.93								
69	Tm	6.18	12.05	23.71							
70	Yb	6.254	12.17	25.2							
71	Lu	5.426	13.9								
72	Hf	7.0	14.9	23.3	33.3						
73	Ta	7.89									
74	W	7.98									
75	Re	7.88									
76	Os	8.7									
77	Ir	9.1									
78	Pt	9.0	18.563								
79	Au	9.225	20.5								
80	Hg	10.437	18.756	34.2							
81	Tl	6.108	20.428	29.83							
82	Pb	7.416	15.032	31.937	42.32	68.8					
83	Bi	7.289	16.69	25.56	45.3	56.0	88.3				

TABLE I. Ionization potentials*—Continued

Z	Element	Spectrum				
		I	II	III	IV	V
84	Po	8.42				
86	At					
86	Rn	10.748				
87	Fr					
88	Ra	5.279	10.147			
89	Ac	6.9	12.1			
90	Th		11.5	20.0	28.8	
91	Pa					
92	U					
93	Np					
94	Pu	5.8				
95	Am	6.0				

* $1\text{cm}^{-1}=0.000123981\text{ eV}$.

known for only a limited number of spectra. For the H I and He I isoelectronic sequences, the theoretical values quoted here are well determined. Edlén, [44], [45], [46], [47], has made a detailed study of formulae for extrapolating ionization limits along sequences of the lighter elements. His values are extensively quoted in Table 2.

Catalán and his associates, [22 to 27], have interpolated values for spectra of neighboring elements in the same stage of ionization. These have been used for spectra in which series are not known. Russell, [166], Sugar and Reader, [156], [181] and others, have described similar general relationships between spectra, that can be used to derive fairly reliable limits.

In Table 2 all ionization limits were recorded that were derived from observed series, from extrapolation or interpolation as described above (Edlén, Catalán, etc.), or from theoretical calculations such as those of the H I and He I series. When all available data from these sources had been entered, if gaps still remained for spectra of a given element in successive stages of ionization, the intervening limits were entered in brackets, as for Ti VIII and Ti IX. These limits, in brackets, represent calculated values interpolated or extrapolated from observed data, and reported in two general tables of ionization potentials in which different methods have been used. For scattered spectra of the elements S V through Zn XIX, the table of Lotz, [116], has been quoted. For larger atomic numbers, the entries in brackets are from the table of Finkelnburg and Humbach, [65]. No attempt has been made, however, to quote *all* such calculated values.

The need for higher ionization limits within a given spectrum increases as laboratory research on absorption series in the vacuum ultraviolet, on series produced with synchrotron radiation as a

source, and the like, advances. At the request of workers in these fields, all components of the ground term, and in selected cases, all levels from the ground configuration, are entered in Table 2. All levels above the ground state are relative to the ground state zero. For example, in the format of "AEL," the lowest levels of O I are as follows:

Desig.	AEL	Table 2
$2p^4\ ^3P_2$	0.000	109837.02 = Limit
$\ ^3P_1$	158.265	158.265
$\ ^3P_0$	226.977	226.977
$\ ^1D_2$	15867.862	15867.862
$\ ^1S_0$	33792.583	33792.583

In compiling Table 2, the energy levels of *only* the ground term have been included for complex spectra, particularly with increasing Z . It is well known that in rare-earth spectra low configurations and low terms overlap in many cases. Consequently, many more low energy levels may be known than those of the ground term. Users are urged to recognize this limitation of the Table and to consult the literature references for further details concerning the low levels that have been reported for individual spectra.

As in "AEL" estimated values of energy levels are given in brackets. Similarly, "x" denotes that the energy level is not connected by observation with the others.

In Table 2, under the term designations for each spectrum, the numbers in italics at the lower left, refer to Table 3. This table is a Bibliography which contains the literature references used for each spectrum to obtain the limits and terms quoted in Table 2.

The importance of stating, clearly, how a limit or an ionization potential has been derived cannot be overemphasized. It is hoped that the present tables will enable each user to judge the quality of the available data used to compile Table 1.

Although the foregoing results are limited to optical spectra, it should be recognized that experimental values of ionization energies have, also, been published. A surface ionization method has been used to obtain ionization potentials for first spectra of rare earths, [196 to 198]. In general, the agreement is satisfactory between the values obtained by the different methods.

Estimates of ionization potentials of third spectra of the lanthanons have been calculated recently "by applying the Born-Haber cycle to the group 3A oxides and arsenides." [199].

After the work on the present publication had been started, the author learned that extensive revisions of the data on the spectra of lighter elements were being prepared by B. Edlén, J. O. Ekberg, and L. Å. Svensson, in Lund. They have most generously furnished much valuable material, in advance of publication, for inclusion here. The author is deeply indebted to these colleagues whose expert judgment and advice greatly enhance the value of the present publication. She is equally grateful to all others who have so willingly contributed their unpublished material.

Washington, D.C.
April 22, 1970

Table 3. Bibliography

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TABLE 2. Ionization limits and lowest terms

Z	Element	Spectrum																		
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII							
1	H	1s 68	¹ S ₀ , 109678.764																	
2	He	1s ² 68	¹ S ₀ , 198310.76 ±0.01																	
3	Li	1s 68	¹ S ₀ , 43906.85																	
4	Be	1s 68	¹ S ₀ , 987660.1																	
5	B	1s 68	¹ S ₀ , 1241259.4																	
6	C	1s 68	¹ S ₀ , 146882.86																	
7	N	1s 68	¹ S ₀ , 2092801.4																	
8	O	1s 68	¹ S ₀ , 3952061.4																	
9	F	1s 68	¹ S ₀ , 3162395 ±30																	
10	Ne	1s 68	¹ S ₀ , 520178.4 ±1.5																	
11	Na	1s 68	¹ S ₀ , 624866 ±3																	
12	Mg	1s 68	¹ S ₀ , 79837.2 ±3.0																	
13	Al	1s 68	¹ S ₀ , 918657 ±4																	
14	Si	1s 68	¹ S ₀ , 1114008 ±10																	
15	P	1s 68	¹ S ₀ , 1493629																	
16	S	1s 68	¹ S ₀ , 1671792																	
17	Cl	1s 68	¹ S ₀ , 1273800 ±310																	
18	Ar	1s 68	¹ S ₀ , 144646																	
19	K	1s 68	¹ S ₀ , 1386500 ±68																	
20	Ca	1s 68	¹ S ₀ , 1858 ±1858																	
21	Sc	1s 68	¹ S ₀ , 1858 ±1858																	
22	Ti	1s 68	¹ S ₀ , 1858 ±1858																	
23	V	1s 68	¹ S ₀ , 1858 ±1858																	
24	Cr	1s 68	¹ S ₀ , 1858 ±1858																	
25	Mn	1s 68	¹ S ₀ , 1858 ±1858																	
26	Fe	1s 68	¹ S ₀ , 1858 ±1858																	
27	Co	1s 68	¹ S ₀ , 1858 ±1858																	
28	Ni	1s 68	¹ S ₀ , 1858 ±1858																	
29	Cu	1s 68	¹ S ₀ , 1858 ±1858																	
30	Zn	1s 68	¹ S ₀ , 1858 ±1858																	
31	Ga	1s 68	¹ S ₀ , 1858 ±1858																	
32	Ge	1s 68	¹ S ₀ , 1858 ±1858																	
33	As	1s 68	¹ S ₀ , 1858 ±1858																	
34	Se	1s 68	¹ S ₀ , 1858 ±1858																	
35	Br	1s 68	¹ S ₀ , 1858 ±1858																	
36	Kr	1s 68	¹ S ₀ , 1858 ±1858																	
37	Rb	1s 68	¹ S ₀ , 1858 ±1858																	
38	Sr	1s 68	¹ S ₀ , 1858 ±1858																	
39	Y	1s 68	¹ S ₀ , 1858 ±1858																	
40	Zr	1s 68	¹ S ₀ , 1858 ±1858																	
41	Nb	1s 68	¹ S ₀ , 1858 ±1858																	
42	Mo	1s 68	¹ S ₀ , 1858 ±1858																	
43	Tc	1s 68	¹ S ₀ , 1858 ±1858																	
44	Ru	1s 68	¹ S ₀ , 1858 ±1858																	
45	Rh	1s 68	¹ S ₀ , 1858 ±1858																	
46	Pd	1s 68	¹ S ₀ , 1858 ±1858																	
47	Ag	1s 68	¹ S ₀ , 1858 ±1858																	
48	Cd	1s 68	¹ S ₀ , 1858 ±1858																	
49	In	1s 68	¹ S ₀ , 1858 ±1858																	
50	Sn	1s 68	¹ S ₀ , 1858 ±1858																	
51	Pb	1s 68	¹ S ₀ , 1858 ±1858																	
52	Bi	1s 68	¹ S ₀ , 1858 ±1858																	
53	Po	1s 68	¹ S ₀ , 1858 ±1858																	
54	At	1s 68	¹ S ₀ , 1858 ±1858																	
55	Rn	1s 68	¹ S ₀ , 1858 ±1858																	

TABLE 2. Ionization limits and lowest terms - continued

IX		X		XI		XII		XIII		XIV		XV		XVI		XVII		XVIII		XIX		XX		XXI		
Spectrum																										
2s ² S _g 46	2645200	2s ² S _g 46	2964400	1s ² S _g 129	14210261	1s ² S _g 49, 68	15829951	1s ² S _g 68	16584138	1s ² S _g 129	21560630	1s ² S _g 49, 68	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
2p ² P _g 46, 48	2662400	2p ² S _g 46	3214800	2s ² S _g 46	3565600	1s ² S _g 129	16825022	1s ² S _g 68	16584138	1s ² S _g 129	21560630	1s ² S _g 49, 68	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
2p ² P _g 46, 48	2831900	2p ² S _g 46, 48	3278800	2s ² S _g 46	3839800	2s ² S _g 46	4222400	1s ² S _g 129	19661693	1s ² S _g 129	21560630	1s ² S _g 49, 68	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
2p ² P _g 46, 48	2998300	2p ² S _g 46, 48	3423000	2p ² P _g 46, 48	3868100	2s ² S _g 46	4520100	2s ² S _g 46	4935900	2s ² S _g 46	5272076	1s ² S _g 129	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
2p ² P _g 46, 48	3057700	2p ² S _g 46, 48	3666100	2p ² P _g 46, 48	4071400	2p ² P _g 46, 48	4543000	2s ² S _g 46	5255900	2s ² S _g 46	5703600	1s ² S _g 129	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
2p ² P _g 46, 48	3226700	2p ² S _g 46, 48	3674000	2p ² P _g 46, 48	4268900	2p ² P _g 46, 48	4773700	2s ² S _g 46	5296700	2s ² S _g 46	6047200	1s ² S _g 129	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
2p ² P _g 46, 48	3407300	2p ² S _g 46, 48	3860900	2p ² P _g 46, 48	4347000	2p ² P _g 46, 48	4986600	2s ² S _g 46	5338800	2s ² S _g 46	6095500	1s ² S _g 129	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
3s ² S _g 52	1418070	2p ² S _g 46	4006000	2p ² P _g 46, 48	4501000	2p ² P _g 46, 48	5074100	2s ² S _g 46	5759100	2s ² S _g 46	6348800	1s ² S _g 129	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68
3p ² S _g 52	1520700	3s ² S _g 52	1704050	2p ² S _g 46	4768900	2p ² P _g 46, 48	5294300	2s ² S _g 46	5856000	2s ² S _g 46	6586600	1s ² S _g 129	24759942	1s ² S _g 68	29507950	1s ² S _g 49, 68	31829012	1s ² S _g 68	35699936	1s ² S _g 68	37190818	1s ² S _g 49, 68	41369608	1s ² S _g 49, 68	44117547	1s ² S _g 49, 68

Table 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum											
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII
21	Sc	3d 4s 27, 135	3d 4s 103290 177.63 2540.97	3d 4s 199700 197.5	3s 4s 592600	3s 4s 179300 4318	3s 4s 896000	3s 4s 29565 30247	3p 4s 142200 5101	3s 4s 1817400	3s 4s 2015080	3s 4s 2015080	3s 4s 553
22	Ti	3d 4s 27, 73	3d 4s 109506 93.94 225.47 386.874	3d 4s 22185 184.9 423.4 194.7 10538.4 10603.6 10721.2	3d 4s 348773 382.1	3s 4s 800300	3s 4s 967000 3825	3s 4s 113690 32168 7291 28557	3s 4s 159000 1569000	3s 4s 1701500 7542	3s 4s 2193900	3s 4s 2193900	3s 4s 235
23	V	3d 4s 27, 135	3d 4s 54400 48 323.42 553.02	3d 4s 118200 36.05 106.63 208.89 553.02	3d 4s 236 145.5 341.5 583.8	3d 4s 1039400	3d 4s 1211200 7600	3d 4s 1401000 4310 9540	3d 4s 1659000	3d 4s 1817400	3d 4s 2057100 9692	3d 4s 2057100 9692	3d 4s 24
24	Cr	3d 4s 103	3d 4s 54570	3d 4s 249700 59.9 185.8 575.0	3d 4s 390000 561 956	3d 4s 1508000	3d 4s 790400 920	3d 4s 1490000 9900	3d 4s 1680000	3d 4s 1971000	3d 4s 2184000	3d 4s 2184000	3d 4s 24
25	Mn	3d 4s 21	3d 4s 59970	3d 4s 271550	3d 4s 413000 96.4 286.8 552.7 885.4	3d 4s 584000	3d 4s 766000 746 1669 15537	3d 4s 1584600	3d 4s 1789000 12530	3d 4s 2003000	3d 4s 2307000	3d 4s 2307000	3d 4s 25
26	Fe	3d 4s 27, 43	3d 4s 63480 415.932 704.003 888.132 978.076	3d 4s 247221 436.2 735.9 1027.3	3d 4s 442000	3d 4s 604000	3d 4s 798500 1189 2002	3d 4s 1238000 1850	3d 4s 1789000	3d 4s 214000	3d 4s 2242000 12657.9 14440	3d 4s 2242000 12657.9 14440	3d 4s 26
27	Co	3d 4s 27, 136	3d 4s 63430 816.00 864.64 1809.33	3d 4s 270200 184.2 1451.3 1866.8	3d 4s 413800	3d 4s 641200	3d 4s 823000 208 586 1129 1789	3d 4s 1501300 2440	3d 4s 220000	3d 4s 2462000 19420	3d 4s 2462000 19420	3d 4s 27	
28	Ni	3d 4s 136	3d 4s 61570 1332.153 2216.519	3d 4s 253700 1360.7 2269.6	3d 4s 442800	3d 4s 669000	3d 4s 1073000 698 802 1520 2379	3d 4s 1501300 2440	3d 4s 220000	3d 4s 2462000 19420	3d 4s 2462000 19420	3d 4s 28	
29	Cu	3d 4s 136	3d 4s 62817.2	3d 4s 297100 2071.8	3d 4s 445124 1660 1862.5 3078.5	3d 4s 6445200	3d 4s 831000	3d 4s 1605000 4110	3d 4s 220000	3d 4s 2462000 19420	3d 4s 2462000 19420	3d 4s 29	

TABLE 2. Ionization Limits and lowest terms—continued

Spectrum		VIII	IX	X	XI	XII	XIII	XIV	XV	XVI	XVII	XVIII	XIX	XX	XXI
ϵ	1290000	$3p$ $^2P_{1/2}^{o,1}$ [1452000]	$3d$ 1S_0 [1817400]	$3s$ $^2S_{1/2}$ [2015680]	$2p$ 1S_0 [5352200]	$2p$ $^2P_{1/2}^{o,1}$ [6093400]	$2p$ $^2P_{3/2}^{o,1}$ [6093400]	$2p$ $^2P_{1/2}^{o,1}$ [6692900]	$2p$ $^2S_{1/2}$ [7468900]						
ζ	5565	$3p$ $^2P_{3/2}^{o,1}$ [5761]													
η	25035														
θ	1359000	$3p$ $^2P_{1/2}^{o,1}$ [1559000]	$3p$ $^2P_{3/2}^{o,1}$ [1741500]	$3s$ $^2S_{1/2}$ [2139300]	$3s$ $^2S_{1/2}$ [2351140]	$2p$ $^2S_{1/2}$ [6350400]	$2p$ $^2P_{1/2}^{o,1}$ [6947300]	$2p$ $^2P_{3/2}^{o,1}$ [6947300]	$2p$ $^2P_{1/2}^{o,1}$ [7584700]						
ι	32168	$3p$ $^2P_{3/2}^{o,1}$ [3125]	$3p$ $^2P_{1/2}^{o,1}$ [7542]												
κ	33239	$3p$ $^2P_{1/2}^{o,1}$ [2857]													
λ	516, 116, 163														
μ	1401000	$3p$ $^2S_{1/2}$ [1659900]	$3p$ $^2P_{1/2}^{o,1}$ [1892900]	$3p$ $^2P_{3/2}^{o,1}$ [2057100]	$3s$ $^2S_{1/2}$ [2486300]	$3s$ $^2S_{1/2}$ [2712250]	$2p$ $^2S_{1/2}$ [7233500]	$2p$ $^2P_{1/2}^{o,1}$ [7856200]							
ν	6000	$3p$ $^2P_{3/2}^{o,1}$ [4310]													
ξ	7580	$3p$ $^2P_{1/2}^{o,1}$ [9540]													
\omicron	27120														
π	1490000	$3p$ $^2P_{1/2}^{o,1}$ [1689000]	$3p$ $^2P_{3/2}^{o,1}$ [1971000]	$3p$ $^2P_{1/2}^{o,1}$ [2184000]	$3p$ $^2P_{3/2}^{o,1}$ [2403600]	$3s$ $^2S_{1/2}$ [2862000]	$2p$ $^2S_{1/2}$ [8151600]								
ρ	9900	$3p$ $^2P_{3/2}^{o,1}$ [7860]													
σ	1596400	$3p$ $^2P_{1/2}^{o,1}$ [1789000]	$3p$ $^2P_{3/2}^{o,1}$ [2093900]	$3p$ $^2S_{1/2}$ [2307000]	$3p$ $^2P_{1/2}^{o,1}$ [2533900]	$3s$ $^2S_{1/2}$ [3200000]	$2p$ $^2S_{1/2}$ [9164300]								
τ	1850	$3p$ $^2P_{3/2}^{o,1}$ [12530]	$3p$ $^2P_{1/2}^{o,1}$ [11700]												
υ	1218400	$3p$ $^2P_{1/2}^{o,1}$ [1893800]	$3p$ $^2P_{3/2}^{o,1}$ [2114000]	$3p$ $^2P_{1/2}^{o,1}$ [2342000]	$3p$ $^2S_{1/2}$ [2668000]	$3p$ $^2P_{1/2}^{o,1}$ [2912000]	$3s$ $^2S_{1/2}$ [3694000]								
ϕ	1850	$3p$ $^2P_{3/2}^{o,1}$ [15682.9]	$3p$ $^2P_{1/2}^{o,1}$ [14440]												
χ	1266000	$3d$ $^1D_{1/2}$ [1501300]	$3p$ $^2P_{1/2}^{o,1}$ [2230000]	$3p$ $^2P_{3/2}^{o,1}$ [2463000]	$3p$ $^2P_{1/2}^{o,1}$ [2710000]	$3p$ $^2S_{1/2}$ [3057000]	$3p$ $^2P_{3/2}^{o,1}$ [3315900]								
ψ	1830	$3d$ $^1D_{3/2}$ [2440]													
ω	3140														
\omicron	1387000	$3d$ $^3P_{2/2}$ [1557000]	$3d$ $^3D_{1/2}$ [1811000]	$3p$ $^2S_{1/2}$ [2591000]	$3p$ $^2P_{1/2}^{o,1}$ [2839000]	$3p$ $^2P_{3/2}^{o,1}$ [3097000]	$3p$ $^2P_{1/2}^{o,1}$ [3742500]								
π	1890	$3p$ $^2P_{3/2}^{o,1}$ [410]	$3p$ $^2P_{1/2}^{o,1}$ [3150]												
ρ	410														
σ	1695000	$3d$ $^3P_{2/2}$ [1695000]	$3d$ $^3D_{1/2}$ [1871000]	$3d$ $^3D_{3/2}$ [2146000]	$3p$ $^2S_{1/2}$ [2975000]	$3p$ $^2P_{1/2}^{o,1}$ [3254000]	$3p$ $^2P_{3/2}^{o,1}$ [3904600]								
τ	5665	$3p$ $^2P_{1/2}^{o,1}$ [5665]													
υ	58, 116														
ω	13695000	$2p$ $^2S_{1/2}$ [13695000]													
ξ	63														

TABLE 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum											
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	XII
30	Zn	4s 95	4s 122	3d ¹⁰ S ₀ 14492.6 ±2	3p 37, 116	3d ⁹ P _{1,2,3} 47910.0 2758.8	3d ⁸ P _{1,2,3} 57100.0	3d ⁷ D _{1,2,3} 100100.0	3d ⁶ S ₀ 140300.0	3d ⁵ D _{1,2,3} 163700.0	3d ⁴ P _{1,2,3} 192000.0	3d ³ P _{1,2,3} 221000.0	3d ² S ₀ 250700.0
31	Ca	4s 96	4s 136	4s S ₀ 10545.8	3s S ₀ 51760.0	3d ⁹ P _{1,2,3} 4837.63 826.19	3d ⁸ P _{1,2,3} 129531.3	3d ⁷ D _{1,2,3} 1767.356	3d ⁶ S ₀ 247700.0	3d ⁵ D _{1,2,3} 116	3d ⁴ P _{1,2,3} 116	3d ³ P _{1,2,3} 116	3d ² S ₀ 116
32	Ce	4f 97, 100	4f 97, 100	4f ⁷ S ₀ 150290.0	4f ⁶ S ₀ 276036.0	4f ⁶ P _{1,2,3} 1469.9609 7125.2989	4f ⁵ P _{1,2,3} 1687.3332	4f ⁴ S ₀ 753800.0	4f ³ S ₀ 102800.0	4f ² S ₀ 102800.0	4f ¹ S ₀ 753800.0	4f ⁰ S ₀ 753800.0	4f ⁰ S ₀ 753800.0
33	As	4p 136	4p 136	4p ³ S ₀ 79165.0	4p ² S ₀ 404369.0	4p ² P _{1,2,3} 1061.0 2538.0 1097.7 2260.2	4p ¹ S ₀ 79165.0	4p ⁰ S ₀ 404369.0	4p ⁰ S ₀ 102800.0	4p ⁰ S ₀ 102800.0	4p ⁰ S ₀ 753800.0	4p ⁰ S ₀ 753800.0	4p ⁰ S ₀ 753800.0
34	Se	4p 136	4p 136	4p ³ S ₀ 17090.0	4p ² S ₀ 28553.0	4p ² P _{1,2,3} 7668.22 1594.42 2534.35 5746.08 22446.03	4p ¹ S ₀ 17090.0	4p ⁰ S ₀ 28553.0	4p ⁰ S ₀ 125300.0	4p ⁰ S ₀ 125300.0	4p ⁰ S ₀ 753800.0	4p ⁰ S ₀ 753800.0	4p ⁰ S ₀ 753800.0
35	Br	4p 136	4p 136	4p ³ S ₀ 173670.0	4p ² S ₀ 3136.4	4p ² P _{1,2,3} 3837.5 1289.1 2767.1	4p ¹ S ₀ 173670.0	4p ⁰ S ₀ 3136.4	4p ⁰ S ₀ 1554700.0	4p ⁰ S ₀ 1554700.0	4p ⁰ S ₀ 753800.0	4p ⁰ S ₀ 753800.0	4p ⁰ S ₀ 753800.0
36	Kr	4p 136, 147	4p 136, 147	4p ³ S ₀ 112914.5	4p ² S ₀ 29820.0	4p ² P _{1,2,3} 196474.8 3571.00	4p ¹ S ₀ 112914.5	4p ⁰ S ₀ 29820.0	4p ⁰ S ₀ 1016500.0	4p ⁰ S ₀ 1016500.0	4p ⁰ S ₀ 2235100.0	4p ⁰ S ₀ 2235100.0	4p ⁰ S ₀ 2235100.0
37	Rb	5s 91	5s 91	5s S ₀ 33690.81 ±0.01	5p ² S ₀ 22043.0	5p ² P _{1,2,3} 22043.0 7380.0	5p ¹ S ₀ 33690.81	5p ⁰ S ₀ 22043.0	5p ⁰ S ₀ 1098000.0	5p ⁰ S ₀ 1098000.0	5p ⁰ S ₀ 2235100.0	5p ⁰ S ₀ 2235100.0	5p ⁰ S ₀ 2235100.0
38	Sr	5s 70	5s 70	5s S ₀ 4932.0 ±0.2	5p ² S ₀ 88064.0	5p ² P _{1,2,3} 88064.0	5p ¹ S ₀ 4932.0	5p ⁰ S ₀ 88064.0	5p ⁰ S ₀ 986700.0	5p ⁰ S ₀ 986700.0	5p ⁰ S ₀ 1428000.0	5p ⁰ S ₀ 1428000.0	5p ⁰ S ₀ 1428000.0
39	Y	4d 5s 22, 136	4d 5s 22, 136	4d ⁴ S ₀ 51447.0	4d ³ S ₀ 165500.0	4d ³ P _{1,2,3} 185400.0 570.41 783.44 1083.7 24.104	4d ² S ₀ 51447.0	4d ¹ S ₀ 165500.0	4d ¹ S ₀ 1041000.0	4d ¹ S ₀ 1041000.0	4d ¹ S ₀ 1351000.0	4d ¹ S ₀ 1351000.0	4d ¹ S ₀ 1351000.0
40	Zr	4d 5s 22, 136	4d 5s 22, 136	4d ⁴ S ₀ 55145.0	4d ³ S ₀ 185400.0	4d ³ P _{1,2,3} 185400.0 570.41 783.44 1083.7 24.104	4d ² S ₀ 55145.0	4d ¹ S ₀ 185400.0	4d ¹ S ₀ 1041000.0	4d ¹ S ₀ 1041000.0	4d ¹ S ₀ 1351000.0	4d ¹ S ₀ 1351000.0	4d ¹ S ₀ 1351000.0

TABLE 2. Ionization limits and lowest terms - continued

Z	Element	Spectrum											
		I	II	III	IV	V	VI	VII	VIII	IX	X	XI	
21	Sc	3d 4s 27, 135	3d 4s 103240 67.68 3d 4s 177.63 3d 4s 2540.97	3d 4s 199700 197.5	3p 4s 592600 52, 182	3p 4s 793000 4318 52, 182	3p 4s 864000 4438 52, 116	3p 4s 1113100 12565 30247 52, 116	3p 4s 1390000 2272 5505 25035	3p 4s 1452000 5761	3p 4s 1817400	3p 4s 2015000	3p 4s 46
22	Ti	3d 4s 27, 73	3d 4s 109506 93.94 288.89 386.874	3d 4s 221735 145.9 436.5 8476.5 10358.4 10003.6 10721.2	3d 4s 94973 362.1	3d 4s 800300	3p 4s 1194000 92700 4530 5884 24130 135, 182	3p 4s 1359000 32168 33239 33, 52	3p 4s 1558000 3125 791 28557 52, 116, 163	3p 4s 1741500 7542	3p 4s 2139000	3p 4s 52	3p 4s 52
23	V	3d 4s 27, 135	3d 4s 118200 36.05 106.63 288.89 339.21	3d 4s 236 145.5 341.3 8476.5 10358.4 10003.6 10721.2	3d 4s 370730 49.4 234.7 10950.3	3d 4s 1035400	3p 4s 1194000 92700 4530 5884 24130 135, 182	3p 4s 1401000 6000 7580 27120	3p 4s 1659000 4310 9540 60, 116	3p 4s 1859200	3p 4s 2057100 9692	3p 4s 52	3p 4s 52
24	Cr	3d 4s 103	3d 4s 133060	3d 4s 297700 181.9 355.8 575.0	3d 4s 390000 244 561 956	3d 4s 730000 920	3p 4s 1299700	3p 4s 1490000 9900 116, 136	3p 4s 1688000 7860 9600 116, 136	3p 4s 1971000	3p 4s 2184000	3p 4s 116	3p 4s 116
25	Mn	3d 4s 21	3d 4s 126145.0 ±0.6	3d 4s 271550	3d 4s 413000 98.4 256.3 822.4 882.4	3d 4s 766000	3d 4s 962001 1355 15337 17, 116, 136	3p 4s 1384600	3p 4s 1790000 12530 116, 136	3p 4s 2003000 10000 11700 116, 136	3p 4s 2307000	3p 4s 116	3p 4s 116
26	Fe	3d 4s 27, 43 3d 4s 27, 136	3d 4s 130524 394.77 704.083 882.52 978.076	3d 4s 267221 738.9 738.9 932.4 1027.3	3d 4s 442000	3d 4s 798500 1189 2327 17, 116	3p 4s 1008000 1051 2327 17, 116	3p 4s 1218400 1850 3, 32, 34	3p 4s 1893800	3p 4s 2114000 15682.9 89, 136	3p 4s 2342000 14440	3p 4s 116	3p 4s 116
27	Co	3d 4s 27, 136	3d 4s 137572 560.45 1597.20	3d 4s 270200 841.2 1451.3 1866.8	3d 4s 413800	3d 4s 822000	3d 4s 1040000 288 1612 1329 2723 116, 136	3p 4s 1266000 1430 3180 4, 116	3p 4s 1501300 2480 3, 75	3p 4s 2230000	3p 4s 2462000 19420	3p 4s 116	3p 4s 116
28	Ni	3d 4s 136	3d 4s 146541.56 1506.94	3d 4s 283700 1560.7 2269.6	3d 4s 402500	3d 4s 697000	3d 4s 1073000 275 802 1520 2379 116, 136	3p 4s 1307000	3p 4s 1890 4110 58, 116	3p 4s 1811000 3150 3, 67	3p 4s 2591000	3p 4s 116	3p 4s 116
29	Cu	4s 163	4s 165669.2 ±0.5	4s 297100 2071.8	4s 445124 ±1600 1862.5 3078.5	4s 881000	4s 1121000	4s 1339000	4s 1605000	4s 1671000 2485 5465 58, 116	4s 2166000	4s 116	4s 116

Table 2. Ionization limits and lowest terms—continued

Z	Element	Spectrum					
		I	II	III	IV	V	VI
54	Xe	5p ⁶ 137, 147	5p ⁶ 137	5p ⁴ 171068.4 5p ³ 9794.6 5p ² 8131	5p ⁴ 137		
55	Cs	6s II	5p ⁶ 137	5p ⁴ 202263			
56	Ba	6s ² 7I	6s 137	5s _{1/2} 80686.87			
57	La	5d 6s ² 7I, 137	5d ² 6s ² 181	5f ¹ 89200 ±650 5f ² 1970.70	5d 6s ² D _{1/2} 154664 ±110 3D _{3/2} 1603.26		
58	Ce	4f 5d 6s ² C ₁ 120, 156a	4f 5d ² 181	4f ¹ 87500 ±680 4f ² 1873.95 4f ³ 2382.26	4f ² 4f ¹ 162900 ±120 4f ² 113 4f ³ 3127.05	4f ² 4f ¹ 260200 ±150 4f ² 4389.1	
59	Pr	4f ³ 6s ² 156, 193	4f ³ 6s 181	4f ¹ 85100 ±650 4f ² 1640.01 4f ³ 2988.31	4f ³ 4f ² 174420 ±130 4f ³ 1398.34 4f ⁴ 2893.14 4f ⁵ 4453.76	4f ³ 4f ¹ 314200 ±150 4f ³ 4389.1 4f ⁴ 46500 ±600 4f ⁵ 3021.4	
60	Nd	4f ⁴ 6s ² 1128, 655	4f ⁴ 6s 181	4f ¹ 88500 ±650 4f ² 1470.100 4f ³ 2585.460 4f ⁴ 3801.935	4f ⁴ 4f ³ 177.180		
61	Pm	4f ⁵ 6s ² 154, 156	4f ⁵ 6s 181, 190	4f ¹ 87900 ±650 4f ² 446.45 4f ³ 1133.45 4f ⁴ 1983.52 4f ⁵ 2950.31			
62	Sm	4f ⁶ 6s ² 232, 88	4f ⁶ 6s 168	4f ¹ 89300 ±650 4f ² 326.64 4f ³ 888.22 4f ⁴ 1491.6 4f ⁵ 2327.97 4f ⁶ 3052.65 4f ⁷ 3909.62	4f ⁶ 4f ⁵ 89300 ±650 4f ⁶ 326.64 4f ⁷ 888.22 4f ⁸ 1491.6 4f ⁹ 2327.97 4f ¹⁰ 3052.65 4f ¹¹ 3909.62		
63	Eu	4f ⁷ 6s ² 45740 ±80	4f ⁷ 6s 168	4f ¹ 90700			

TABLE 2. Ionization limits and lowest terms - continued

Z	Element	Spectrum					
		I	II	III	IV	V	VI
64	Gd	4f ⁷ 5d 6s ² 4f ⁷ 49530 ± 110 4f ⁷ 5d 6s ² 4f ⁷ 215.13 4f ⁷ 5d 6s ² 4f ⁷ 522.98 4f ⁷ 5d 6s ² 4f ⁷ 596.11 4f ⁷ 5d 6s ² 4f ⁷ 1719.06 156m, 165	4f ⁷ 5d 6s 4f ⁷ 97900 ± 3000 4f ⁷ 5d 6s 4f ⁷ 261.81 4f ⁷ 5d 6s 4f ⁷ 633.27 4f ⁷ 5d 6s 4f ⁷ 1158.94 4f ⁷ 5d 6s 4f ⁷ 1933.36 165, 181				
65	Tb	4f ⁷ 6s ² 4f ⁷ 417200 ± 150 4f ⁷ 6s ² 4f ⁷ 4f ⁷ 6s ² 4f ⁷ 4f ⁷ 6s ² 4f ⁷ 4f ⁷ 6s ² 4f ⁷	4f ⁷ 6s 4f ⁷ 92900 ± 650 4f ⁷ 6s 4f ⁷ 4f ⁷ 6s 4f ⁷ 4f ⁷ 6s 4f ⁷ 4f ⁷ 6s 4f ⁷				
66	Dy	4f ⁷ 6s ² 4f ⁷ 47820 ± 150 4f ⁷ 6s ² 4f ⁷ 4134.24 4f ⁷ 6s ² 4f ⁷ 7050.61 4f ⁷ 6s ² 4f ⁷ 4f ⁷ 6s ² 4f ⁷	4f ⁷ 6s 4f ⁷ 94100 ± 650 4f ⁷ 6s 4f ⁷ 4941.10 4f ⁷ 6s 4f ⁷ 8650.56 4f ⁷ 6s 4f ⁷ 7463.88 4f ⁷ 6s 4f ⁷ 9432.07 4f ⁷ 6s 4f ⁷ 10953.94 31, 181				
67	Ho	4f ⁷ 6s ² 4f ⁷ 48540 ± 150 4f ⁷ 6s ² 4f ⁷ 4f ⁷ 6s ² 4f ⁷ 4f ⁷ 6s ² 4f ⁷	4f ⁷ 6s 4f ⁷ 95200 ± 650 4f ⁷ 6s 4f ⁷ 4f ⁷ 6s 4f ⁷ 4f ⁷ 6s 4f ⁷				
68	Er	4f ⁷ 6s ² 4f ⁷ 49210 ± 150 4f ⁷ 6s ² 4f ⁷ 6958.34 4f ⁷ 6s ² 4f ⁷ 10750.99 117, 156	4f ⁷ 6s 4f ⁷ 96300 ± 650 4f ⁷ 6s 4f ⁷ 7149.7 4f ⁷ 6s 4f ⁷ 11042.8 4f ⁷ 6s 4f ⁷ 10894.1 181				
69	Tm	4f ⁷ 6s ² 4f ⁷ 49840 ± 150 4f ⁷ 6s ² 4f ⁷ 8771.25 126, 156	4f ⁷ 6s 4f ⁷ 97200 ± 500 4f ⁷ 6s 4f ⁷ 236.94 4f ⁷ 6s 4f ⁷ 8774.02 189a				
70	Yb	4f ⁷ 6s ² 4f ⁷ 50441.0 ± 0.2 20a	4f ⁷ 6s 4f ⁷ 98150 4f ⁷ 6s 4f ⁷ 20 4f ⁷ 6s 4f ⁷ 203300				
71	Lu	5d 6s ² 4f ⁷ 45762.39 ± 0.6 4f ⁷ 6s ² 4f ⁷ 1993.92 20b, 110	5s 112000 ± 3000 4f ⁷ 6s ² 4f ⁷ 181				
72	Hf	5d ² 6s ² 4f ⁷ 56000 ± 2356.68 4f ⁷ 6s ² 4f ⁷ 4867.64 127	5d ² 6s ² 4f ⁷ 120000 ± 300 4f ⁷ 6s ² 4f ⁷ 3050.98 137				
73	Ta	5d ² 6s ² 4f ⁷ 63600 ± 2010.10 4f ⁷ 6s ² 4f ⁷ 5963.92 4f ⁷ 6s ² 4f ⁷ 5621.04 137	5d ² 6s ² 4f ⁷ 268500 ± 800 4f ⁷ 6s ² 4f ⁷ 3282.7 4f ⁷ 6s ² 4f ⁷ 6953.1 111				

TABLE 2. Ionization Limits and Lowest Terms - Continued

Z	Element	Spectrum					
		I	II	III	IV	V	VI
90	Th	6d ² 7s ² 2669.200 4961.661 192	4F _{5/2} [93000] 4F _{3/2} 1521.91 4F _{3/2} 4146.57 4F _{3/2} 6213.55 65, 124	6d ² 161000 3922.7 6674.9 109	5f 4F _{5/2} 231900 4F _{3/2} 4325.38		
91	Pa						
92	U						
93	Np						
94	Pu	5f ⁶ 7s ² 47000 2203.55 4399.55 6144.34 7756.85 10268.24					
95	Am	7, 8 5f ⁷ 7s ² *S _{5/2} 48770 66					

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