

Selected Tables of Atomic Spectra

A Atomic Energy Levels - Second Edition

B Multiplet Tables

H I, D, T

Data Derived from the Analyses of Optical Spectra

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**U.S. DEPARTMENT OF COMMERCE, Peter G. Peterson, Secretary
NATIONAL BUREAU OF STANDARDS,**

Issued September 1972

Library of Congress Catalog Card Number: 64-60074

NSRDS-NBS 3, Section 6

Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.), 3, Sec. 6, 36 pages (Sept. 1972)

CODEN: NSRDAP

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For sale by the Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402
(Order by SD Catalog No. C13.48:3/Sec. 6). Price 40 cents.
Stock Number 0303-0998

Abstract

The present publication is the sixth Section of a series being prepared in response to the persistent need for a current revision of two sets of tables containing data on atomic spectra as derived from analyses of optical spectra. As in the previous sections, Part A contains the atomic energy levels and Part B the multiplet tables. The spectra of hydrogen and of the isotopes deuterium and tritium are included. The form of presentation is described in detail in the text to Section 1.

Key words: Atomic energy levels, H I, D, T; hydrogen spectra, H I, D, T; multiplet tables, H I, D, T; spectra H I, D, T; wavelengths, hydrogen spectra H I, D, T.

Foreword

The National Standard Reference Data System provides effective access to the quantitative data of physical science, critically evaluated and compiled for convenience, and readily accessible through a variety of distribution channels. The System was established in 1963 by action of the President's Office of Science and Technology and the Federal Council for Science and Technology, with responsibility to administer it assigned to the National Bureau of Standards.

The System now comprises a complex of data centers and other activities, carried on in academic institutions and other laboratories both in and out of government. The independent operational status of existing critical data projects is maintained and encouraged. Data centers that are components of the NSRDS produce compilations of critically evaluated data, critical reviews of the state of quantitative knowledge in specialized areas, and computations of useful functions derived from standard reference data. In addition, the centers and projects establish criteria for evaluation and compilation of data and make recommendations on needed improvements in experimental techniques. They are normally closely associated with active research in the relevant field.

The technical scope of the NSRDS is indicated by the principal categories of data compilation projects now active or being planned: nuclear properties, atomic and molecular properties, solid state properties, thermodynamic and transport properties, chemical kinetics, and colloid and surface properties and mechanical properties.

The NSRDS receives advice and planning assistance from the National Research Council of the National Academy of Sciences-National Academy of Engineering. An overall Review Committee considers the program as a whole and makes recommendations on policy, long-term planning, and international collaboration. Advisory Panels, each concerned with a single technical area, meet regularly to examine major portions of the program, assign relative priorities, and identify specific key problems in need of further attention. For selected specific topics, the Advisory Panels sponsor subpanels which make detailed studies of users' needs, the present state of knowledge, and existing data resources, as a basis for recommending one or more data compilation activities. This assembly of advisory services contributes greatly to the guidance of NSRDS activities.

The NSRDS-NBS series of publications is intended primarily to include evaluated reference data and critical reviews of long-term interest to the scientific and technical community.

LAWRENCE M. KUSHNER, *Acting Director*

Preface

The present publication is the sixth Section of a series that is being prepared in response to the increasing demand for a current revision of two sets of tables containing data on atomic spectra as derived from analyses of optical spectra.

The first set, Atomic Energy Levels, NBS Circular 467, consists of three Volumes published, respectively, in 1949, 1952, and 1958, and a fourth one on rare-earth spectra, still in course of preparation. This Circular has been reprinted as NSRDS-NBS 35, Volumes I, II, and III.

The second set consists of two Multiplet Tables; one published in 1945 by the Princeton University Observatory containing multiplets having wavelengths longer than 3000 Å; the other, An Ultra-Violet Multiplet Table, NBS Circular 488, appearing in five Sections, the first in 1950, the second in 1952, and the others in 1962. The 1945 Princeton Multiplet Table has been reprinted as NSRDS-NBS 40.

The present series includes both sets of data, the energy levels and multiplet tables, as parts A and B, respectively, for selected spectra contained in Volume I of "Atomic Energy Levels." The Sections are being published at irregular intervals as revised analyses become available. A flexible paging system permits the arrangement of the various Sections by atomic number, regardless of the order in which the separate spectra are published. Section 1 includes three spectra of silicon, Z=14: Si II, Si III, Si IV. Section 2 contains similar data for Si I. Section 3 covers all the spectra of carbon, Z=6: C I, C II, C III, C IV, C V, C VI. Section 4 includes four spectra of nitrogen, Z=7: N IV, N V, N VI, N VII. Section 5 is scheduled to include the remaining spectra of nitrogen: N I, N II, N III. The form of presentation of the data is described in detail in the text of Section 1. All Sections are arranged identically and the same conversion factor, cm^{-1} to eV, 0.000123981 is used throughout.

The manuscript has been prepared by Charlotte E. Moore who has published the earlier tables. She appreciates the cordial cooperation of numerous atomic spectroscopists. She is particularly indebted to J. D. Garcia and the late J. E. Mack, who carried out the extensive calculations on hydrogen-like spectra, especially for inclusion in this Series. W. C. Martin and his colleagues in the Spectroscopy Section of the National Bureau of Standards have provided valuable suggestions regarding details in the text and tables. Special thanks are due Isabel D. Murray for her meticulous care in preparing the tabular data. The splendid work of Barbara N. Somerville in typing the press copy of this difficult tabular material is, also, gratefully acknowledged.

Washington, D.C., March 24, 1972.

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Deuterium		D	B1 D-1
Tritium		T.....	B1 T-1

NSRDS-NBS 3, SECTION 6

HYDROGEN Z = 1

A H I Atomic Energy Levels

B H I Multiplet Table

Atomic Energy Levels

HYDROGEN

Part A

H I

1 electron

$Z = 1$

Ground state $1s^2S_{1/2}$

$1s^2S_{1/2} \text{ } \mathbf{109678.764 \text{ cm}^{-1}, 911.754 \text{ \AA (Vac)}}$

I P 13.598 eV

The data quoted here refer to the light isotope of hydrogen ${}^1\text{H I}$. Deuterium and tritium are being handled separately.

The energy levels of the spectra of the hydrogen isoelectronic sequence H I to Ca XX have been calculated by Garcia and Mack and reported in a paper entitled "Energy Level and Line Tables for One-Electron Atomic Spectra." For H I the level values are derived to six decimal places to $n = 50$, with $R = 109677.576 \text{ cm}^{-1}$.

In the present table the calculated values have been rounded off to three decimals. Intervals are quoted to four places only in cases of resolved levels. For further details users should consult the original paper.

For unresolved groups the term designations in Part B have no real significance. The center of gravity of all levels having a given value of n has been used to derive the quoted wavelength, regardless of the arbitrary designation entered with "etc." throughout.

Since the publication of "Atomic Energy Levels," Humphreys has reported the observation of the first member of the Sixth Series of H I at 12.37μ , the second line of the Pfund Series and the third, fourth and fifth lines of the Brackett Series.

The Lamb shift and both fine and hyperfine structure of H I have been widely discussed in the literature. In 1964 Edlén and Svensson derived accurate "centre-of-gravity" wavelengths of the Lyman lines, $1s - np$, $n = 2$ to 7, and discussed "the various factors pertinent to their use as standards."

Observations of H I in the radiofrequency range have been of far-reaching significance. For example, the transition between the two hyperfine structure levels $F = 0$ and $F = 1$ of the ground term $1s^2S_{1/2}$, 0.0475 cm^{-1} , is well known as the 21-cm line whose observed frequency is 1420 Mc/s. The presence of ground state hydrogen atoms in the interstellar medium was first established by the detection of this line.

A limited bibliography of work on H I was published by the writer in 1968. Many other papers could be quoted, as for example, the recent work by Shyn and his associates on the measurement of the $2s^2S_{1/2} - 2p^2P_{1/2}^o$ energy separation as $9911.250 \pm 0.063 \text{ MHz}$, as determined by an atomic-beam radiofrequency method.

A more recent paper by Hänsch and his co-workers on Laser Saturation Spectroscopy is of far-reaching importance. These authors have succeeded for the first time in resolving the single fine structure components of H α , and have observed the Lamb shift directly in the optical absorption spectrum.

An excellent general summary of the astrophysical importance of "The Spectra of Hydrogen" is contained in the Presidential Address to the Royal Society of Canada given by Herzberg in 1967.

Atomic Energy Levels

H I—Continued

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H I

H I

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
1s	1s ² S	0 $\frac{1}{2}$	0.000		9s, 9p etc.	9s ² S 9p ² P° 9l ² L	0 $\frac{1}{2}$ to 8 $\frac{1}{2}$	108324.713 to .720	0.007
2p	2p ² P°	0 $\frac{1}{2}$	82258.913	0.0353	10s, 10p etc.	10s ² S 10p ² P° 10m ² M°	0 $\frac{1}{2}$ to 9 $\frac{1}{2}$	108581.983 to .988	0.005
2s	2s ² S	0 $\frac{1}{2}$	82258.949	0.3306					
2p	2p ² P°	1 $\frac{1}{2}$	82259.279						
3p	3p ² P°	0 $\frac{1}{2}$	97492.205	0.0105	11s, 11p etc.	11s ² S 11p ² P° 11n ² N	0 $\frac{1}{2}$ to 10 $\frac{1}{2}$	108772.334 to .338	0.004
3s	3s ² S	0 $\frac{1}{2}$	97492.215	0.098					
3p, 3d	3d ² D 3p ² P°	1 $\frac{1}{2}$	97492.313	0.036	12s, 12p etc.	12s ² S 12p ² P°	0 $\frac{1}{2}$ to 11 $\frac{1}{2}$	108917.111 to .114	0.003
3d	3d ² D	2 $\frac{1}{2}$	97492.349						
4p	4p ² P°	0 $\frac{1}{2}$	102823.842	0.0044	13s, 13p etc.	13s ² S 13p ² P°	0 $\frac{1}{2}$ to 12 $\frac{1}{2}$	109029.782 to .784	0.002
4s	4s ² S	0 $\frac{1}{2}$	102823.846	0.041					
4p, 4d	4d ² D 4p ² P°	1 $\frac{1}{2}$	102823.887	0.015					
4d, 4f	4d ² D 4f ² F°	2 $\frac{1}{2}$	102823.902	0.008	14s, 14p etc.	14s ² S 14p ² P°	0 $\frac{1}{2}$ to 13 $\frac{1}{2}$	109119.183 to .185	0.002
4f	4f ² F°	3 $\frac{1}{2}$	102823.910						
5p	5p ² P°	0 $\frac{1}{2}$	105291.621	0.0023	15s, 15p etc.	15s ² S 15p ² P°	0 $\frac{1}{2}$ to 14 $\frac{1}{2}$	109191.307 to .308	0.001
5s	5s ² S	0 $\frac{1}{2}$	105291.624						
5p, 5d	5d ² D 5p ² P°	1 $\frac{1}{2}$	105291.645	0.021					
5d, 5f	5d ² D 5f ² F°	2 $\frac{1}{2}$	105291.653	0.008					
5g	5g ² G	3 $\frac{1}{2}$	105291.656	0.003	16s, 16p etc.	16s ² S 16p ² P°	0 $\frac{1}{2}$ to 15 $\frac{1}{2}$	109250.335 to .336	0.001
5f	5f ² F°	3 $\frac{1}{2}$	105291.657	0.0000					
5g	5g ² G	4 $\frac{1}{2}$	105291.659	0.0023					
6p	6p ² P°	0 $\frac{1}{2}$	106632.141	0.0013	17s, 17p etc.	17s ² S 17p ² P°	0 $\frac{1}{2}$ to 16 $\frac{1}{2}$	109299.256 to .257	0.001
6s	6s ² S	0 $\frac{1}{2}$	106632.143						
6p, 6d	6d ² D 6p ² P°	1 $\frac{1}{2}$	106632.155	0.012	18s, 18p etc.	18s ² S 18p ² P°	0 $\frac{1}{2}$ to 17 $\frac{1}{2}$	109340.252 to .253	0.001
6d, 6f	6d ² D 6f ² F°	2 $\frac{1}{2}$	106632.159	0.004					
6g	6g ² G	3 $\frac{1}{2}$	106632.161	0.002					
6f	6f ² F°	3 $\frac{1}{2}$	106632.162	0.0000	19s, 19p etc.	19s ² S 19p ² P°	0 $\frac{1}{2}$ to 18 $\frac{1}{2}$	109374.947 to .948	0.001
6g, 6h	6g ² G 6h ² H°	4 $\frac{1}{2}$	106632.163	0.001					
6h	6h ² H°	5 $\frac{1}{2}$	106632.164	0.001					
7p	7p ² P°	0 $\frac{1}{2}$	107440.431		20s, 20p etc.	20s ² S 20p ² P°	0 $\frac{1}{2}$ to 19 $\frac{1}{2}$	109404.570 .570	
7s	7s ² S	0 $\frac{1}{2}$	107440.432	0.0008					
etc.	7i ² I	to 6 $\frac{1}{2}$	to .446	0.015	21s, 21p etc.	21s ² S 21p ² P°	0 $\frac{1}{2}$ to 20 $\frac{1}{2}$	109430.062 .062	
8s, 8p	8s ² S 8p ² P°	0 $\frac{1}{2}$	107965.042	0.010	22s, 22p etc.	22s ² S 22p ² P°	0 $\frac{1}{2}$ to 21 $\frac{1}{2}$	109452.157 to .158	0.001
etc.	8k ² K°	to 7 $\frac{1}{2}$	to .052						

Atomic Energy Levels

H I – Continued

H I – Continued

Config.	Desig.	J	Level	Interval	Config.	Desig.	J	Level	Interval
23s, etc.	23s ² S, etc.	0½, etc.	109471.434		38s, etc.	38s ² S, etc.	0½, etc.	109602.810	
24s, etc.	24s ² S, etc.	0½, etc.	109488.351 to .352	0.001	39s, etc.	39s ² S, etc.	0½, etc.	109606.655	
25s, etc.	25s ² S, etc.	0½, etc.	109503.280		40s, etc.	40s ² S, etc.	0½, etc.	109610.216	
26s, etc.	26s ² S, etc.	0½, etc.	109516.519		41s, etc.	41s ² S, etc.	0½, etc.	109613.519	
27s, etc.	27s ² S, etc.	0½, etc.	109528.315		42s, etc.	42s ² S, etc.	0½, etc.	109616.589	
28s, etc.	28s ² S, etc.	0½, etc.	109538.869		43s, etc.	43s ² S, etc.	0½, etc.	109619.447	
29s, etc.	29s ² S, etc.	0½, etc.	109548.351		44s, etc.	44s ² S, etc.	0½, etc.	109622.113	
30s, etc.	30s ² S, etc.	0½, etc.	109556.900		45s, etc.	45s ² S, etc.	0½, etc.	109624.602	
31s, etc.	31s ² S, etc.	0½, etc.	109564.635 to .636	0.001	46s, etc.	46s ² S, etc.	0½, etc.	109626.932	
32s, etc.	32s ² S, etc.	0½, etc.	109571.657		47s, etc.	47s ² S, etc.	0½, etc.	109629.114	
33s, etc.	33s ² S, etc.	0½, etc.	109578.050		48s, etc.	48s ² S, etc.	0½, etc.	109631.161	
34s, etc.	34s ² S, etc.	0½, etc.	109583.887		49s, etc.	49s ² S, etc.	0½, etc.	109633.084	
35s, etc.	35s ² S, etc.	0½, etc.	109589.231		50s, etc.	50s ² S, etc.	0½, etc.	109634.893	
36s, etc.	36s ² S, etc.	0½, etc.	109594.136			
37s, etc.	37s ² S, etc.	0½, etc.	109598.649			Limit	109678.764	

August, 1971.

Multiplet Table

HYDROGEN

Part B

H I (Z = 1)

I P 13.598 eV Limit 109678.764 cm⁻¹ 911.754 Å (Vac)

Anal A List B August 1971

The data given here refer to the light isotope of hydrogen ^1H I.

REFERENCES

- B. Edlén and L. A. Svensson, Ark. Fys. (Stockholm) **28**, No. 36, 427–446 (1964). C L; W L 930.7483 Å to 1215.6701 Å
 C. J. Humphreys, J. Research Nat. Bur. Std. **50**, No. 1, 1–6, RP2380 (1953). C L; W L 4861 Å to 123684 Å
- A Wavelengths calculated from term values derived by J. D. Garcia and J. E. Mack, J. Opt. Soc. Am. **55**, No. 6, 654–685 (1965). I P, T, C L; W L 914.0386 Å to 887313.171 Å. For higher values of n where the terms are unresolved, the wavelength derived from the statistical mean of the components is quoted.
- P Predicted wavelength for series members having $n > 20$, i.e., for series carried further than in reference A. In calculating these wavelengths the following mean values have been used for the lower level:

2p $^2\text{P}^o$ etc. 82259.102 cm⁻¹
 3d ^2D etc. 97492.296 cm⁻¹
 4f $^2\text{F}^o$ etc. 102823.890 cm⁻¹
 5g ^2G etc. 105291.649 cm⁻¹
 6h $^2\text{H}^o$ etc. 106632.159 cm⁻¹

New UV Multiplet Numbers start with UV 18. The Multiplet Numbers in the 1945 Multiplet Table ($\lambda > 3000 \text{ Å}$) have been replaced. The newly-assigned numbers are printed in bold face type through number 14 to distinguish them from the older ones.

#Raie Ultime

H I

H I

I A	Ref	Int	E P		J	Multiplet No.	I A	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Vac 215.6683‡ 215.6737	A A	L α	0.00 0.00	10.20 10.20	0 $\frac{1}{2}$ –1 $\frac{1}{2}$ 0 $\frac{1}{2}$ –0 $\frac{1}{2}$	1s ^2S — 2p $^2\text{P}^o$ UV 1	949.7430 949.7432	A A	L δ	0.00 0.00	13.05 13.05	0 $\frac{1}{2}$ –1 $\frac{1}{2}$ 0 $\frac{1}{2}$ –0 $\frac{1}{2}$	1s ^2S — 5p $^2\text{P}^o$ UV 4
025.7219 025.7230	A A	L β	0.00 0.00	12.09 12.09	0 $\frac{1}{2}$ –1 $\frac{1}{2}$ 0 $\frac{1}{2}$ –0 $\frac{1}{2}$	1s ^2S — 3p $^2\text{P}^o$ UV 2	937.8034 937.8035	A A	L ϵ	0.00 0.00	13.22 13.22	0 $\frac{1}{2}$ –1 $\frac{1}{2}$ 0 $\frac{1}{2}$ –0 $\frac{1}{2}$	1s ^2S — 6p $^2\text{P}^o$ UV 5
972.5367 972.5371	A A	L γ	0.00 0.00	12.75 12.75	0 $\frac{1}{2}$ –1 $\frac{1}{2}$ 0 $\frac{1}{2}$ –0 $\frac{1}{2}$	1s ^2S — 4p $^2\text{P}^o$ UV 3	930.7483 926.2257	A A		0.00 0.00	13.32 13.39	0 $\frac{1}{2}$ – 0 $\frac{1}{2}$ –	1s ^2S — 7p $^2\text{P}^o$ UV 6 UV 7

Multiplet Table

H1—Continued

H1—Continued

IA	Ref	Int	E P		J	Multiplet No.	IA	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Vac							Vac						
923.1504	A		0.00	13.43	0 $\frac{1}{2}$ -	1s 2S — 9p $^2P^o$ UV 8	913.006	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 27p $^2P^o$ UV 26
920.9631	A		0.00	13.46	0 $\frac{1}{2}$ -	1s 2S — 10p $^2P^o$ UV 9	912.918	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 28p $^2P^o$ UV 27
919.3514	A		0.00	13.49	0 $\frac{1}{2}$ -	1s 2S — 11p $^2P^o$ UV 10	912.839	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 29p $^2P^o$ UV 28
918.1294	A		0.00	13.50	0 $\frac{1}{2}$ -	1s 2S — 12p $^2P^o$ UV 11	912.768	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 30p $^2P^o$ UV 29
917.1806	A		0.00	13.52	0 $\frac{1}{2}$ -	1s 2S — 13p $^2P^o$ UV 12	912.703	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 31p $^2P^o$ UV 30
916.4291	A		0.00	13.53	0 $\frac{1}{2}$ -	1s 2S — 14p $^2P^o$ UV 13	912.645	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 32p $^2P^o$ UV 31
915.8238	A		0.00	13.54	0 $\frac{1}{2}$ -	1s 2S — 15p $^2P^o$ UV 14	912.592	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 33p $^2P^o$ UV 32
915.3290	A		0.00	13.54	0 $\frac{1}{2}$ -	1s 2S — 16p $^2P^o$ UV 15	912.543	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 34p $^2P^o$ UV 33
914.9193	A		0.00	13.55	0 $\frac{1}{2}$ -	1s 2S — 17p $^2P^o$ UV 16	912.498	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 35p $^2P^o$ UV 34
914.5763	A		0.00	13.56	0 $\frac{1}{2}$ -	1s 2S — 18p $^2P^o$ UV 17	912.458	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 36p $^2P^o$ UV 35
914.2862	A		0.00	13.56	0 $\frac{1}{2}$ -	1s 2S — 19p $^2P^o$ UV 18	912.420	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 37p $^2P^o$ UV 36
914.0386	A		0.00	13.56	0 $\frac{1}{2}$ -	1s 2S — 20p $^2P^o$ UV 19	912.385	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 38p $^2P^o$ UV 37
913.826	P		0.00	13.57	0 $\frac{1}{2}$ -	1s 2S — 21p $^2P^o$ UV 20	912.353	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 39p $^2P^o$ UV 38
913.641	P		0.00	13.57	0 $\frac{1}{2}$ -	1s 2S — 22p $^2P^o$ UV 21	912.324	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 40p $^2P^o$ UV 39
913.480	P		0.00	13.57	0 $\frac{1}{2}$ -	1s 2S — 23p $^2P^o$ UV 22	912.296	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 41p $^2P^o$ UV 40
913.339	P		0.00	13.57	0 $\frac{1}{2}$ -	1s 2S — 24p $^2P^o$ UV 23	912.271	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 42p $^2P^o$ UV 41
913.215	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 25p $^2P^o$ UV 24	912.247	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 43p $^2P^o$ UV 42
913.104	P		0.00	13.58	0 $\frac{1}{2}$ -	1s 2S — 26p $^2P^o$ UV 25	912.225	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 44p $^2P^o$ UV 43

Multiplet Table

H I—Continued
H I—Continued

I A	Ref	Int	E P		J	Multiplet No.	I A	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Vac						Air							
912.204	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 45p $^2P^o$ UV 44	4340.4340	A		10.20	13.05	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	2s 2S — 5p $^2P^o$ 3.02
912.185	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 46p $^2P^o$ UV 45	4101.7662	A		10.20	13.22	1 $\frac{1}{2}$ -0 $\frac{1}{2}$	2p $^2P^o$ — 6s 2S 4.01
912.166	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 47p $^2P^o$ UV 46	4101.7650	A	H β	10.20	13.22	1 $\frac{1}{2}$ -2 $\frac{1}{2}$	2p $^2P^o$ — 6d 2D 4
912.149	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 48p $^2P^o$ UV 47	4101.7087	A		10.20	13.22	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	2s 2S — 6p $^2P^o$ 4.02
912.133	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 49p $^2P^o$ UV 48							
912.118	P		0.00	13.59	0 $\frac{1}{2}$ -	1s 2S — 50p $^2P^o$ UV 49	3970.072	A	H ϵ	10.20	13.32		2p $^2P^o$ — 7d 2D etc. 5 etc.
							3889.049	A		10.20	13.39		2p $^2P^o$ — 8d 2D etc. 6 etc.
911.754	A		0.00	13.60	0 $\frac{1}{2}$ -	1s 2S — Limit UV 50	3835.384	A		10.20	13.43		2p $^2P^o$ — 9d 2D etc. 7 etc.
Air							3797.898	A		10.20	13.46		2p $^2P^o$ — 10d 2D etc. 8 etc.
6562.9099	A		10.20	12.09	1 $\frac{1}{2}$ -0 $\frac{1}{2}$	2p $^2P^o$ — 3s 2S 1.01	3770.630	A		10.20	13.49		2p $^2P^o$ — 11d 2D etc. 9 etc.
6562.7520	A		10.20	12.09	0 $\frac{1}{2}$ -0 $\frac{1}{2}$								
6562.8520	A	H α	10.20	12.09	1 $\frac{1}{2}$ -2 $\frac{1}{2}$	2p $^2P^o$ — 3d 2D	3750.152	A		10.20	13.50		2p $^2P^o$ — 12d 2D etc. 10 etc.
6562.7101	A		10.20	12.09	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	1							
6562.7256	A		10.20	12.09	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	2s 2S — 3p $^2P^o$	3734.368	A		10.20	13.52		2p $^2P^o$ — 13d 2D etc. 11 etc.
6562.7720	A		10.20	12.09	0 $\frac{1}{2}$ -0 $\frac{1}{2}$	1.02							
4861.3752	A		10.20	12.75	1 $\frac{1}{2}$ -0 $\frac{1}{2}$	2p $^2P^o$ — 4s 2S	3721.938	A		10.20	13.53		2p $^2P^o$ — 14d 2D etc. 12 etc.
4861.2885	A		10.20	12.75	0 $\frac{1}{2}$ -0 $\frac{1}{2}$	2.01							
4861.3620	A	H β	10.20	12.75	1 $\frac{1}{2}$ -2 $\frac{1}{2}$	2p $^2P^o$ — 4d 2D	3711.971	A		10.20	13.54		2p $^2P^o$ — 15d 2D etc. 13 etc.
4861.2789	A		10.20	12.75	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	2							
4861.2873	A		10.20	12.75	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	2s 2S — 4p $^2P^o$	3703.853	A		10.20	13.54		2p $^2P^o$ — 16d 2D etc. 14 etc.
4861.2980	A		10.20	12.75	0 $\frac{1}{2}$ -0 $\frac{1}{2}$	2.02							
4340.5001	A		10.20	13.05	1 $\frac{1}{2}$ -0 $\frac{1}{2}$	2p $^2P^o$ — 5s 2S	3697.152	A		10.20	13.55		2p $^2P^o$ — 17d 2D etc. 15 etc.
4340.4312	A		10.20	13.05	0 $\frac{1}{2}$ -0 $\frac{1}{2}$	3.01							
4340.4946	A	H γ	10.20	13.05	1 $\frac{1}{2}$ -2 $\frac{1}{2}$	2p $^2P^o$ — 5d 2D	3691.555	A		10.20	13.56		2p $^2P^o$ — 18d 2D etc. 16 etc.
4340.4272	A		10.20	13.05	0 $\frac{1}{2}$ -1 $\frac{1}{2}$	3							
							3686.831	A		10.20	13.56		2p $^2P^o$ — 19d 2D etc. 17 etc.

Multiplet Table

HI-Continued

HI-Continued

IA	Ref	Int	E P		J	Multiplet No.	IA	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Air							Air						
3682.808	A		10.20	13.56		2p ² P°—20d ² D etc. 18 etc.	3651.822	P		10.20	13.59		2p ² P°—50d ² D etc. 37 etc.
3679.352	P		10.20	13.57		2p ² P°—21d ² D etc. 19 etc.							
3676.363	P		10.20	13.57		2p ² P°—22d ² D etc. 20 etc.	3645.979	A		10.20	13.60		2p ² P°—Limit etc. 38
3673.758	P		10.20	13.57		2p ² P°—23d ² D etc. 21 etc.							
3671.476	P		10.20	13.57		2p ² P°—24d ² D etc. 22 etc.	18751.210	A		12.09	12.75	1½-0½	3p ² P°—4s ² S 39.01
3669.464	P		10.20	13.58		2p ² P°—25d ² D etc. 23 etc.	18750.830	A		12.09	12.75	0½-0½	3s ² S — 4p ² P° 39.02
3667.682	P		10.20	13.58		2p ² P°—26d ² D etc. 24 etc.	18751.015	A		12.09	12.75		3d ² D — 4f ² F° etc. 39 etc.
3666.095	P		10.20	13.58		2p ² P°—27d ² D etc. 25 etc.	12818.140	A		12.09	13.05	1½-0½	3p ² P°—5s ² S 40.01
3664.677	P		10.20	13.58		2p ² P°—28d ² D etc. 26 etc.	12817.962	A		12.09	13.05	0½-0½	3s ² S — 5p ² P° 40.02
3663.403	P		10.20	13.58		2p ² P°—29d ² D etc. 27 etc.	12817.945	A		12.09	13.05	0½-1½	3d ² D — 5f ² F° etc. 40 etc.
3662.256	P		10.20	13.58		2p ² P°—30d ² D etc. 28 etc.	12818.082	A		12.09	13.05	12.09	3p ² P°—6s ² S 41.01
3661.219	P		10.20	13.58		2p ² P°—31d ² D etc. 29 etc.	10938.126	A		12.09	13.22	1½-0½	3s ² S — 6p ² P° 41.02
3660.277	P		10.20	13.58		2p ² P°—32d ² D etc. 30 etc.	10938.095	A		12.09	13.22		3d ² D — 6f ² F° etc. 41 etc.
3659.420	P		10.20	13.59		2p ² P°—33d ² D etc. 31 etc.	10049.374	A		12.09	13.32		3d ² D — 7f ² F° etc. 42 etc.
3658.639	P		10.20	13.59		2p ² P°—34d ² D etc. 32 etc.	9545.972	A		12.09	13.39		3d ² D — 8f ² F° etc. 43 etc.
3657.923	P		10.20	13.59		2p ² P°—35d ² D etc. 33 etc.	9229.015	A		12.09	13.43		3d ² D — 9f ² F° etc. 44 etc.
3657.267	P		10.20	13.59		2p ² P°—36d ² D etc. 34 etc.	9014.911	A		12.09	13.46		3d ² D — 10f ² F° etc. 45 etc.
3656.663	P		10.20	13.59		2p ² P°—37d ² D etc. 35 etc.	8862.784	A		12.09	13.49		3d ² D — 11f ² F° etc. 46 etc.
3656.107	P		10.20	13.59		2p ² P°—38d ² D etc. 36 etc.	8750.473	A		12.09	13.50		3d ² D — 12f ² F° etc. 47 etc.

Multiplet Table

H I—Continued

H I—Continued

I A	Ref	Int	E P		J	Multiplet No.	I A	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Air						Air							
8665.019	A		12.09	13.52		$3d^2D \rightarrow 13f^2F^o$ etc. 48 etc.	8276.309	P		12.09	13.58		$3d^2D \rightarrow 32f^2F^o$ etc. 67 etc.
8598.392	A		12.09	13.53		$3d^2D \rightarrow 14f^2F^o$ etc. 49 etc.	8271.932	P		12.09	13.59		$3d^2D \rightarrow 33f^2F^o$ etc. 68 etc.
8545.383	A		12.09	13.54		$3d^2D \rightarrow 15f^2F^o$ etc. 50 etc.	8267.938	P		12.09	13.59		$3d^2D \rightarrow 34f^2F^o$ etc. 69 etc.
8502.483	A		12.09	13.54		$3d^2D \rightarrow 16f^2F^o$ etc. 51 etc.	8264.286	P		12.09	13.59		$3d^2D \rightarrow 35f^2F^o$ etc. 70 etc.
8467.254	A		12.09	13.55		$3d^2D \rightarrow 17f^2F^o$ etc. 52 etc.	8260.936	P		12.09	13.59		$3d^2D \rightarrow 36f^2F^o$ etc. 71 etc.
8437.955	A		12.09	13.56		$3d^2D \rightarrow 18f^2F^o$ etc. 53 etc.	8257.856	P		12.09	13.59		$3d^2D \rightarrow 37f^2F^o$ etc. 72 etc.
8413.318	A		12.09	13.56		$3d^2D \rightarrow 19f^2F^o$ etc. 54 etc.	8255.019	P		12.09	13.59		$3d^2D \rightarrow 38f^2F^o$ etc. 73 etc.
8392.397	A		12.09	13.56		$3d^2D \rightarrow 20f^2F^o$ etc. 55 etc.							
8374.476	P		12.09	13.57		$3d^2D \rightarrow 21f^2F^o$ etc. 56 etc.	8233.208	A		12.09	13.59		$3d^2D \rightarrow 50f^2F^o$ etc. 74 etc.
8359.004	P		12.09	13.57		$3d^2D \rightarrow 22f^2F^o$ etc. 57 etc.							
8345.553	P		12.09	13.57		$3d^2D \rightarrow 23f^2F^o$ etc. 58 etc.	8203.568	A		12.09	13.60		$3d^2D \rightarrow \text{Limit}$ 75
8333.784	P		12.09	13.57		$3d^2D \rightarrow 24f^2F^o$ etc. 59 etc.	40511.92	A		12.75	13.05	$1\frac{1}{2}-0\frac{1}{2}$	$4p^2P^o \rightarrow 5s^2S$ 76.01
8323.426	P		12.09	13.58		$3d^2D \rightarrow 25f^2F^o$ etc. 60 etc.	40511.17	A		12.75	13.05	$0\frac{1}{2}-0\frac{1}{2}$	$4s^2S \rightarrow 5p^2P^o$ 76.02
8314.261	P		12.09	13.58		$3d^2D \rightarrow 26f^2F^o$ etc. 61 etc.	40510.87	A		12.75	13.05	$0\frac{1}{2}-1\frac{1}{2}$	$4s^2S \rightarrow 5p^2P^o$ 76.02
8306.112	P		12.09	13.58		$3d^2D \rightarrow 27f^2F^o$ etc. 62 etc.	40511.28	A		12.75	13.05	$0\frac{1}{2}-0\frac{1}{2}$	$4f^2F^o \rightarrow 5g^2G$ 76
8298.836	P		12.09	13.58		$3d^2D \rightarrow 28f^2F^o$ etc. 63 etc.	40511.579	A		12.75	13.05	$1\frac{1}{2}-0\frac{1}{2}$	$4p^2P^o \rightarrow 6s^2S$ 77.01
8292.308	P		12.09	13.58		$3d^2D \rightarrow 29f^2F^o$ etc. 64 etc.	40511.57	A		12.75	13.22	$0\frac{1}{2}-0\frac{1}{2}$	$4s^2S \rightarrow 6p^2P^o$ 77.02
8286.432	P		12.09	13.58		$3d^2D \rightarrow 30f^2F^o$ etc. 65 etc.	40511.27	A		12.75	13.22	$0\frac{1}{2}-0\frac{1}{2}$	$4f^2F^o \rightarrow 6g^2G$ 77
8281.123	P		12.09	13.58		$3d^2D \rightarrow 31f^2F^o$ etc. 66 etc.	40511.31	A		12.75	13.22	$0\frac{1}{2}-0\frac{1}{2}$	$4f^2F^o \rightarrow 7g^2G$ 78
							40511.512	A		12.75	13.22		
							40511.59	A		12.75	13.32		

Multiplet Table

H I – Continued**H I – Continued**

I A	Ref	Int	E P		J	Multiplet No.	I A	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Air							Air						
19445.564	A		12.75	13.39		4f $^2\text{F}^\circ$ —8g ^2G etc. 79 etc.	14888.010	P		12.75	13.58		4f $^2\text{F}^\circ$ —28g ^2G etc. 99 etc.
18174.123	A		12.75	13.43		4f $^2\text{F}^\circ$ —9g ^2G etc. 80 etc.	14867.017	P		12.75	13.58		4f $^2\text{F}^\circ$ —29g ^2G etc. 100 etc.
17362.110	A		12.75	13.46		4f $^2\text{F}^\circ$ —10g ^2G etc. 81 etc.	14848.142	P		12.75	13.58		4f $^2\text{F}^\circ$ —30g ^2G etc. 101 etc.
16806.522	A		12.75	13.49		4f $^2\text{F}^\circ$ —11g ^2G etc. 82 etc.	14678.114	P		12.75	13.59		4f $^2\text{F}^\circ$ —50g ^2G etc. 102 etc.
16407.193	A		12.75	13.50		4f $^2\text{F}^\circ$ —12g ^2G etc. 83 etc.	74578.80	A		13.05	13.22	$\frac{1}{2}-0\frac{1}{2}$	5p $^2\text{P}^\circ$ —6s ^2S 103.01
16109.314	A		12.75	13.52		4f $^2\text{F}^\circ$ —13g ^2G etc. 84 etc.	74578.250	A		13.05	13.22		5g ^2G —6h $^2\text{H}^\circ$ 103
15880.543	A		12.75	13.53		4f $^2\text{F}^\circ$ —14g ^2G etc. 85 etc.	46525.098	A		13.05	13.32		5g ^2G —7h $^2\text{H}^\circ$ etc. 104 etc.
15700.663	A		12.75	13.54		4f $^2\text{F}^\circ$ —15g ^2G etc. 86 etc.	37395.370	A		13.05	13.39		5g ^2G —8h $^2\text{H}^\circ$ etc. 105 etc.
15556.450	A		12.75	13.54		4f $^2\text{F}^\circ$ —16g ^2G etc. 87 etc.	15438.922	A		13.05	13.43		5g ^2G —9h $^2\text{H}^\circ$ etc. 106 etc.
15341.791	A		12.75	13.56		4f $^2\text{F}^\circ$ —17g ^2G etc. 88 etc.	30383.737	A		13.05	13.46		5g ^2G —10h $^2\text{H}^\circ$ etc. 107 etc.
15260.539	A		12.75	13.56		4f $^2\text{F}^\circ$ —18g ^2G etc. 89 etc.	28722.126	A		13.05	13.49		5g ^2G —11h $^2\text{H}^\circ$ etc. 108 etc.
15191.845	A		12.75	13.56		4f $^2\text{F}^\circ$ —19g ^2G etc. 90 etc.	27575.156	A		13.05	13.50		5g ^2G —12h $^2\text{H}^\circ$ etc. 109 etc.
15133.225	P		12.75	13.57		4f $^2\text{F}^\circ$ —21g ^2G etc. 92 etc.	26744.018	A		13.05	13.52		5g ^2G —13h $^2\text{H}^\circ$ etc. 110 etc.
15082.777	P		12.75	13.57		4f $^2\text{F}^\circ$ —22g ^2G etc. 93 etc.	26119.352	A		13.05	13.53		5g ^2G —14h $^2\text{H}^\circ$ etc. 111 etc.
15039.040	P		12.75	13.57		4f $^2\text{F}^\circ$ —23g ^2G etc. 94 etc.	25636.276	A		13.05	13.54		5g ^2G —15h $^2\text{H}^\circ$ etc. 112 etc.
15000.862	P		12.75	13.57		4f $^2\text{F}^\circ$ —24g ^2G etc. 95 etc.	25254.015	A		13.05	13.54		5g ^2G —16h $^2\text{H}^\circ$ etc. 113 etc.
14967.131	P		12.75	13.58		4f $^2\text{F}^\circ$ —25g ^2G etc. 96 etc.	24945.738	A		13.05	13.55		5g ^2G —17h $^2\text{H}^\circ$ etc. 114 etc.
14937.730	P		12.75	13.58		4f $^2\text{F}^\circ$ —26g ^2G etc. 97 etc.	24693.137	A		13.05	13.56		5g ^2G —18h $^2\text{H}^\circ$ etc. 115 etc.
14911.447	P		12.75	13.58		4f $^2\text{F}^\circ$ —27g ^2G etc. 98 etc.							

Multiplet Table

H I—Continued

H I—Continued

IA	Ref	Int	EP		J	Multiplet No.	IA	Ref	Int	EP		J	Multiplet No.
			Low	High						Low	High		
Air							Air						
24483.323	A		13.05	13.56		5g ² G — 19h ² H° etc. 116 etc.	113056.141	A		13.32	13.43		7i ² I — 9k ² K° etc. 135 etc.
24306.989	A		13.05	13.56		5g ² G — 20h ² H° etc. 117 etc.	87576.773	A		13.32	13.46		7i ² I — 10k ² K° etc. 136 etc.
23017.983	P		13.05	13.59		5g ² G — 50h ² H° etc. 118 etc.	75060.591	A		13.32	13.49		7i ² I — 11k ² K° etc. 137 etc.
123685.270	A		13.22	13.32		6h ² H°— 7i ² I etc. 119 etc.	67701.453	A		13.32	13.50		7i ² I — 12k ² K° etc. 138 etc.
75004.488	A		13.22	13.39		6h ² H°— 8i ² I etc. 120 etc.	62902.015	A		13.32	13.52		7i ² I — 13k ² K° etc. 139 etc.
59066.034	A		13.22	13.43		6h ² H°— 9i ² I etc. 121 etc.	57099.058	A		13.32	13.54		7i ² I — 14k ² K° etc. 140 etc.
51972.598	A		13.22	13.46		6h ² H°— 10i ² I etc. 122 etc.	55236.826	A		13.32	13.54		7i ² I — 15k ² K° etc. 141 etc.
46712.353	A		13.22	13.49		6h ² H°— 11i ² I etc. 123 etc.	53783.083	A		13.32	13.55		7i ² I — 16k ² K° etc. 142 etc.
43752.617	A		13.22	13.50		6h ² H°— 12i ² I etc. 124 etc.	52622.492	A		13.32	13.56		7i ² I — 17k ² K° etc. 143 etc.
41696.572	A		13.22	13.52		6h ² H°— 13i ² I etc. 125 etc.	51678.714	A		13.32	13.56		7i ² I — 18k ² K° etc. 144 etc.
40197.716	A		13.22	13.53		6h ² H°— 14i ² I etc. 126 etc.	50899.318	A		13.32	13.56		7i ² I — 19k ² K° etc. 145 etc.
39064.836	A		13.22	13.54		6h ² H°— 15i ² I etc. 127 etc.							7i ² I — 20k ² K° etc. 146 etc.
38184.101	A		13.22	13.54		6h ² H°— 16i ² I etc. 128 etc.	277958.040	A		13.39	13.43		8k ² K°— 9l ² L etc. 147 etc.
37483.714	A		13.22	13.55		6h ² H°— 17i ² I etc. 129 etc.	162046.877	A		13.39	13.46		8k ² K°— 10l ² L etc. 148 etc.
36916.270	A		13.22	13.56		6h ² H°— 18i ² I etc. 130 etc.	123837.927	A		13.39	13.49		8k ² K°— 11l ² L etc. 149 etc.
36449.295	A		13.22	13.56		6h ² H°— 19i ² I etc. 131 etc.	105006.373	A		13.39	13.50		8k ² K°— 12l ² L etc. 150 etc.
36059.849	A		13.22	13.56		6h ² H°— 20i ² I etc. 132 etc.	93894.586	A		13.39	13.52		8k ² K°— 13l ² L etc. 151 etc.
33293.900	P		13.22	13.59		6h ² H°— 50i ² I 133	86621.417	A		13.39	13.53		8k ² K°— 14l ² L etc. 152 etc.
190567.045	A		13.32	13.39		7i ² I — 8k ² K° etc. 134 etc.	81526.684	A		13.39	13.54		8k ² K°— 15l ² L etc. 153 etc.

Multiplet Table

H I—Continued**H I—Continued**

IA	Ref	Int	E P		J	Multiplet No.	IA	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Air							Air						
388592.763	A		13.43	13.46		9l ² L —10m ² M° etc. 154 etc.	141792.199	A		13.43	13.52		9l ² L —13m ² M° etc. 157 etc.
223343.698	A		13.43	13.49		9l ² L —11m ² M° etc. 155 etc.	125836.471	A		13.43	13.53		9l ² L —14m ² M° etc. 158 etc.
168760.287	A		13.43	13.50		9l ² L —12m ² M° etc. 156 etc.	115363.473	A		13.43	13.54		9l ² L —15m ² M° etc. 159 etc.

NSRDS-NBS 3, SECTION 6

DEUTERIUM AND TRITIUM Z = 1

A D and T Atomic Energy Levels

B D Multiplet Table

B T Multiplet Table

Atomic Energy Levels

DEUTERIUM AND TRITIUM

Part A

D AND T

1 electron

$Z = 1$

Ground state $1s\ ^2S_{01/2}$

$1s\ ^2S_{01/2} D(^2H\ 1)$	109708.608 cm $^{-1}$, 911.506 Å (Vac)	I P (D) 13.602 eV
$1s\ ^2S_{01/2} T(^3H\ 1)$	109718.538 cm $^{-1}$, 911.423 Å (Vac)	I P (T) 13.603 eV

The energy levels are quoted from the paper by Garcia and Mack, who calculated "Energy Levels and Line Tables for One-Electron Atomic Spectra" for the spectra of the hydrogen iso-electronic sequence H I to Ca XX. For D and T the level values are determined to six decimal places to $n = 50$, with $R = 109707.420$ (D) and 109717.350 (T), respectively.

In the present table the calculated values have been rounded off to three decimals. Intervals are given to four places only in cases of resolved levels. For further details users should consult the original paper, in which the formulas and constants are fully described.

REFERENCE

J. D. Garcia and J. E. Mack, J. Opt. Soc. Am. **55**, No. 6, 654-685 (1965). I P, T, C L.

D T

D T

Config.	Desig.	J	Level	Level	Interval	Config.	Desig.	J	Level	Level	Interval
1s	$1s\ ^2S$	$0\frac{1}{2}$	0.000	0.000		6p	$6p\ ^2P^o$	$0\frac{1}{2}$	106661.156	106670.810	0.0013
2p	$2p\ ^2P^o$	$0\frac{1}{2}$	82281.296	82288.743	0.0353	6s	$6s\ ^2S$	$0\frac{1}{2}$	106661.157	106670.811	0.0122
2s	$2s\ ^2S$	$0\frac{1}{2}$	82281.331	82288.779	0.3307	6p, 6d	$6d\ ^2D\ 6p\ ^2P^o$	$1\frac{1}{2}$	106661.169	106670.823	0.0045
2p	$2p\ ^2P^o$	$1\frac{1}{2}$	82281.662	82289.109		6d, 6f	$6d\ ^2D\ 6f\ ^2F^o$	$2\frac{1}{2}$	106661.174	106670.828	0.0022
3p	$3p\ ^2P^o$	$0\frac{1}{2}$	97518.732	97527.559	0.0105	6f, 6g	$6g\ ^2G\ 6f\ ^2F^o$	$3\frac{1}{2}$	106661.176	106670.830	0.0014
3s	$3s\ ^2S$	$0\frac{1}{2}$	97518.743	97527.569	0.0978	6g, 6h	$6g\ ^2G\ 6h\ ^2H^o$	$4\frac{1}{2}$	106661.178	106670.831	0.0009
3d	$3d\ ^2D$	$1\frac{1}{2}$	97518.840	97527.667	0.0002	6h	$6h\ ^2H^o$	$5\frac{1}{2}$	106661.178	106670.832	
3p	$3p\ ^2P^o$	$1\frac{1}{2}$	97518.841	97527.667	0.0360	7p	$7p\ ^2P^o$	$0\frac{1}{2}$	107469.666	107479.393	0.0018
3d	$3d\ ^2D$	$2\frac{1}{2}$	97518.877	97527.703		7s	$7s\ ^2S$	$0\frac{1}{2}$	107469.667	107479.394	0.0077
4p	$4p\ ^2P^o$	$0\frac{1}{2}$	102851.820	102861.129	0.0044	7p, 7d	$7d\ ^2D\ 7p\ ^2P^o$	$1\frac{1}{2}$	107469.674	107479.401	0.0028
4s	$4s\ ^2S$	$0\frac{1}{2}$	102851.824	102861.133	0.041	7d, 7f	$7d\ ^2D\ 7f\ ^2F^o$	$2\frac{1}{2}$	107469.677	107479.404	0.0014
4p, 4d	$4d\ ^2D\ 4p\ ^2P^o$	$1\frac{1}{2}$	102851.866	102861.175	0.015	7f, 7g	$7g\ ^2G\ 7f\ ^2F^o$	$3\frac{1}{2}$	107469.679	107479.406	0.0009
4d, 4f	$4d\ ^2D\ 4f\ ^2F^o$	$2\frac{1}{2}$	102851.881	102861.190	0.0076	7g, 7h	$7g\ ^2G\ 7h\ ^2H^o$	$4\frac{1}{2}$	107469.679	107479.406	0.0006
4f	$4f\ ^2F^o$	$3\frac{1}{2}$	102851.888	102861.198		7h, 7i	$7i\ ^2I\ 7h\ ^2H^o$	$5\frac{1}{2}$	107469.680	107479.407	0.0004
5p	$5p\ ^2P^o$	$0\frac{1}{2}$	105320.271	105329.804	0.0023	7i	$7i\ ^2I$	$6\frac{1}{2}$	107469.680	107479.407	
5s	$5s\ ^2S$	$0\frac{1}{2}$	105320.274	105329.806	0.0211	8p	$8p\ ^2P^o$	$0\frac{1}{2}$	107994.419	108004.194	0.0006
5p, 5d	$5d\ ^2D\ 5p\ ^2P^o$	$1\frac{1}{2}$	105320.295	105329.827	0.0078	8s	$8s\ ^2S$	$0\frac{1}{2}$	107994.420	108004.194	0.0052
5d, 5f	$5d\ ^2D\ 5f\ ^2F^o$	$2\frac{1}{2}$	105320.303	105329.835	0.0039	8p, 8d	$8d\ ^2D\ 8p\ ^2P^o$	$1\frac{1}{2}$	107994.425	108004.199	0.0019
5f, 5g	$5g\ ^2G\ 5f\ ^2F^o$	$3\frac{1}{2}$	105320.306	105329.839	0.0023	8d, 8f	$8d\ ^2D\ 8f\ ^2F^o$	$2\frac{1}{2}$	107994.427	108004.201	0.0009
5g	$5g\ ^2G$	$4\frac{1}{2}$	105320.309	105329.841		8f, 8g	$8g\ ^2G\ 8f\ ^2F^o$	$3\frac{1}{2}$	107994.428	108004.202	0.0006
						8g, 8h	$8g\ ^2G\ 8h\ ^2H^o$	$4\frac{1}{2}$	107994.428	108004.203	0.0004
						8h, 8i	$8i\ ^2I\ 8h\ ^2H^o$	$5\frac{1}{2}$	107994.429	108004.203	0.0003
						8i, 8k	$8i\ ^2I\ 8k\ ^2K^o$	$6\frac{1}{2}$	107994.429	108004.203	0.0002
						8k	$8k\ ^2K^o$	$7\frac{1}{2}$	107994.429	108004.204	

Atomic Energy Levels

D and T—Continued

D T

D T

Config.	Desig.	<i>J</i>	Level	Level	Interval	Config.	Desig.	<i>J</i>	Level	Level	Interval
9p	9p ² P°	0½	108354.188	108363.995	0.0004	10p	10p ² P°	0½	108611.528	108621.359	0.0003
9s	9s ² S	0½	108354.188	108363.995	0.0036	10s	10s ² S	0½	108611.529	108621.359	0.0026
9p, 9d	9d ² D 9p ² P°	1½	108354.192	108363.999	0.0013	10p, 10d	10d ² D 10p ² P°	1½	108611.531	108621.362	0.0010
9d, 9f	9d ² D 9f ² F°	2½	108354.193	108364.000	0.0007	10d, 10f	10d ² D 10f ² F°	2½	108611.532	108621.363	0.0005
9f, 9g	9g ² G 9f ² F°	3½	108354.194	108364.001	0.0004	10f, 10g	10g ² G 10f ² F°	3½	108611.533	108621.363	0.0003
9g, 9h	9g ² G 9h ² H°	4½	108354.194	108364.001	0.0003	10g, 10h	10g ² G 10h ² H°	4½	108611.533	108621.363	0.0002
9h, 9i	9i ² I 9h ² H°	5½	108354.195	108364.002	0.0002	10h, 10i	10i ² I 10h ² H°	5½	108611.533	108621.364	0.0001
9i, 9k	9i ² I 9k ² K°	6½	108354.195	108364.002	0.0001	10i, 10k	10i ² I 10k ² K°	6½	108611.533	108621.364	0.0001
9k, 9l	9l ² L 9k ² K°	7½	108354.195	108364.002	0.0001	10k, 10l	10l ² L 10k ² K°	7½	108611.533	108621.364	0.0001
9l	9l ² L	8½	108354.195	108364.002		10l, 10m	10l ² L 10m ² M°	8½	108611.534	108621.364	0.0001
						10m	10m ² M°	9½	108611.534	108621.364	
							Limit		109708.608	109718.538	

February 1972.

Multiplet Table

DEUTERIUM

Part B

D(^2H 1)

I P 13.602 eV Limit 109708.608 cm⁻¹ 911.506 Å (Vac)

Anal A List C February 1972

REFERENCE

A J. D. Garcia and J. E. Mack, J. Opt. Soc. Am. **55**, No. 6, 654–685 (1965). I P, T, C L; W L 925.9737 Å to 123652.691 Å.
For higher values of n where the terms are unresolved, the wavelength derived from the statistical mean of the components is quoted.

New Multiplet Numbers have been assigned.

D

D

IA	Ref	Int	E P		J	Multiplet No.	IA	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Vac 1215.3376 1215.3430	A A		0.00 0.00	10.20 10.20	$0\frac{1}{2}-1\frac{1}{2}$ $0\frac{1}{2}-0\frac{1}{2}$	$1s\ ^2S - 2p\ ^2P^o$ UV 1	Air 6561.0104	A	D α	10.20	12.09		$2p\ ^2P^o - 3d\ ^2D$ etc. 1 etc.
1025.4429 1025.4401	A A		0.00 0.00	12.09 12.09	$0\frac{1}{2}-1\frac{1}{2}$ $0\frac{1}{2}-0\frac{1}{2}$	$1s\ ^2S - 3p\ ^2P^o$ UV 2	4860.0028	A	D β	10.20	12.75		$2p\ ^2P^o - 4d\ ^2D$ etc. 2 etc.
972.2721 972.2725	A A		0.00 0.00	12.75 12.75	$0\frac{1}{2}-1\frac{1}{2}$ $0\frac{1}{2}-0\frac{1}{2}$	$1s\ ^2S - 4p\ ^2P^o$ UV 3	4339.2829	A	D γ	10.20	13.06		$2p\ ^2P^o - 5d\ ^2D$ etc. 3 etc.
949.4846 949.4848	A A		0.00 0.00	13.06 13.06	$0\frac{1}{2}-1\frac{1}{2}$ $0\frac{1}{2}-0\frac{1}{2}$	$1s\ ^2S - 5p\ ^2P^o$ UV 4	4100.6191	A	D δ	10.20	13.22		$2p\ ^2P^o - 6d\ ^2D$ etc. 4 etc.
937.5483 937.5484	A A		0.00 0.00	13.22 13.22	$0\frac{1}{2}-1\frac{1}{2}$ $0\frac{1}{2}-0\frac{1}{2}$	$1s\ ^2S - 6p\ ^2P^o$ UV 5	3968.9922	A	D ϵ	10.20	13.32		$2p\ ^2P^o - 7d\ ^2D$ etc. 5 etc.
930.4951	A		0.00	13.32	$0\frac{1}{2}-$	$1s\ ^2S - 7p\ ^2P^o$ UV 6	3887.9909	A		10.20	13.39		$2p\ ^2P^o - 8d\ ^2D$ etc. 6 etc.
925.9737	A		0.00	13.39	$0\frac{1}{2}-$	$1s\ ^2S - 8p\ ^2P^o$ UV 7	18745.914	A		12.09	12.75		$3d\ ^2D - 4f\ ^2F^o$ etc. 7 etc.
911.5055	A		0.00	13.60	$0\frac{1}{2}-$	$1s\ ^2S - \text{Limit}$ UV 8	12814.595	A		12.09	13.06		$3d\ ^2D - 5f\ ^2F^o$ etc. 8 etc.
							10935.120	A		12.09	13.22		$3d\ ^2D - 6f\ ^2F^o$ etc. 9 etc.
							10046.640	A		12.09	13.32		$3d\ ^2D - 7f\ ^2F^o$ etc. 10 etc.
							9543.375	A		12.09	13.39		$3d\ ^2D - 8f\ ^2F^o$ etc. 11 etc.

Multiplet Table

TRITIUM

Part B

T(^3H I)

I P 13.603 eV Limit 109718.538 cm⁻¹ 911.423 Å (Vac)

Anal A List D February 1972

REFERENCE

A J. D. Garcia and J. E. Mack, J. Opt. Soc. Am. **55**, No. 6, 654–685 (1965). I P, T, C L; W L 925.8900 Å to 123641.500 Å
For higher values of n where the terms are unresolved, the wavelength derived from the statistical mean of the components is quoted.

T

T

IA	Ref	Int	E P		J	Multiplet No.	IA	Ref	Int	E P		J	Multiplet No.
			Low	High						Low	High		
Vac													
1215.2276	A		0.00	10.20	$0\frac{1}{2}-1\frac{1}{2}$	$1s\ ^2S - 2p\ ^2P^o$		Air					
1215.2330	A		0.00	10.20	$0\frac{1}{2}-0\frac{1}{2}$	UV 1	6560.4166	A	T α	10.20	12.09		$2p\ ^2P^o - 3d\ ^2D$
etc.													etc. 1 etc.
1025.3501	A		0.00	12.09	$0\frac{1}{2}-1\frac{1}{2}$	$1s\ ^2S - 3p\ ^2P^o$	4859.5630	A	T β	10.20	12.75		$2p\ ^2P^o - 4d\ ^2D$
1025.3512	A		0.00	12.09	$0\frac{1}{2}-0\frac{1}{2}$	UV 2							etc. 2 etc.
etc.													
972.1841	A		0.00	12.75	$0\frac{1}{2}-1\frac{1}{2}$	$1s\ ^2S - 4p\ ^2P^o$	4338.8902	A	T γ	10.20	13.06		$2p\ ^2P^o - 5d\ ^2D$
972.1845	A		0.00	12.75	$0\frac{1}{2}-0\frac{1}{2}$	UV 3							etc. 3 etc.
etc.													
949.3987	A		0.00	13.06	$0\frac{1}{2}-1\frac{1}{2}$	$1s\ ^2S - 5p\ ^2P^o$	4100.2479	A	T δ	10.20	13.23		$2p\ ^2P^o - 6d\ ^2D$
949.3989	A		0.00	13.06	$0\frac{1}{2}-0\frac{1}{2}$	UV 4							etc. 4 etc.
etc.													
937.4635	A		0.00	13.23	$0\frac{1}{2}-1\frac{1}{2}$	$1s\ ^2S - 6p\ ^2P^o$	3968.6329	A	T ϵ	10.20	13.33		$2p\ ^2P^o - 7d\ ^2D$
937.4636	A		0.00	13.23	$0\frac{1}{2}-0\frac{1}{2}$	UV 5							etc. 5 etc.
etc.													
.....													
911.4230	A		0.00	13.60		$1s\ ^2S - \text{Limit}$							
						UV 6							