



*Coupled Flow and Multicomponent
Biogeochemical Reactive Transport
Modeling: In Situ Biostimulation
at the Rifle Site*

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Field-Scale Modeling: Understand, Predict, and Control

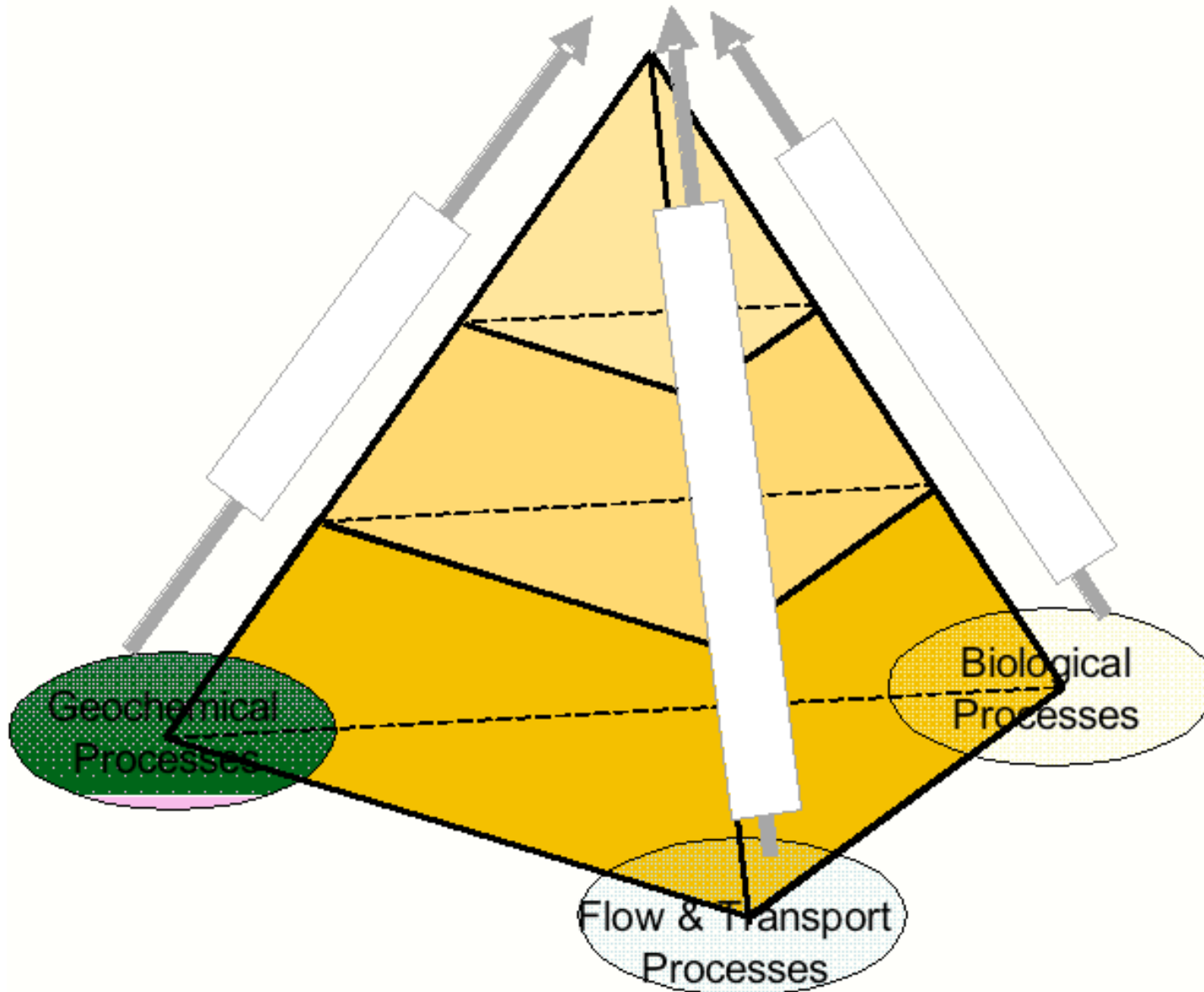
- ▶ Engineering bioremediation for site-specific conditions will require a ***quantitatively predictive understanding*** of the dominant processes and properties controlling contaminant behavior
- ▶ Complex natural environments make the reliable prediction of field scale behavior a scientific challenge
 - Many field-scale issues are difficult to address at the lab scale
 - Many processes and properties are difficult to monitor in the field
 - Need to develop a quantitatively mechanistic understanding of field-scale behaviors by addressing the relevant range of scales and multiple interacting processes
- ▶ In the context of temporally and spatially variable conditions at the site, use ***modeling to develop understanding*** of the interplay between the dominant flow, transport, and biogeochemical processes

Presentation Outline

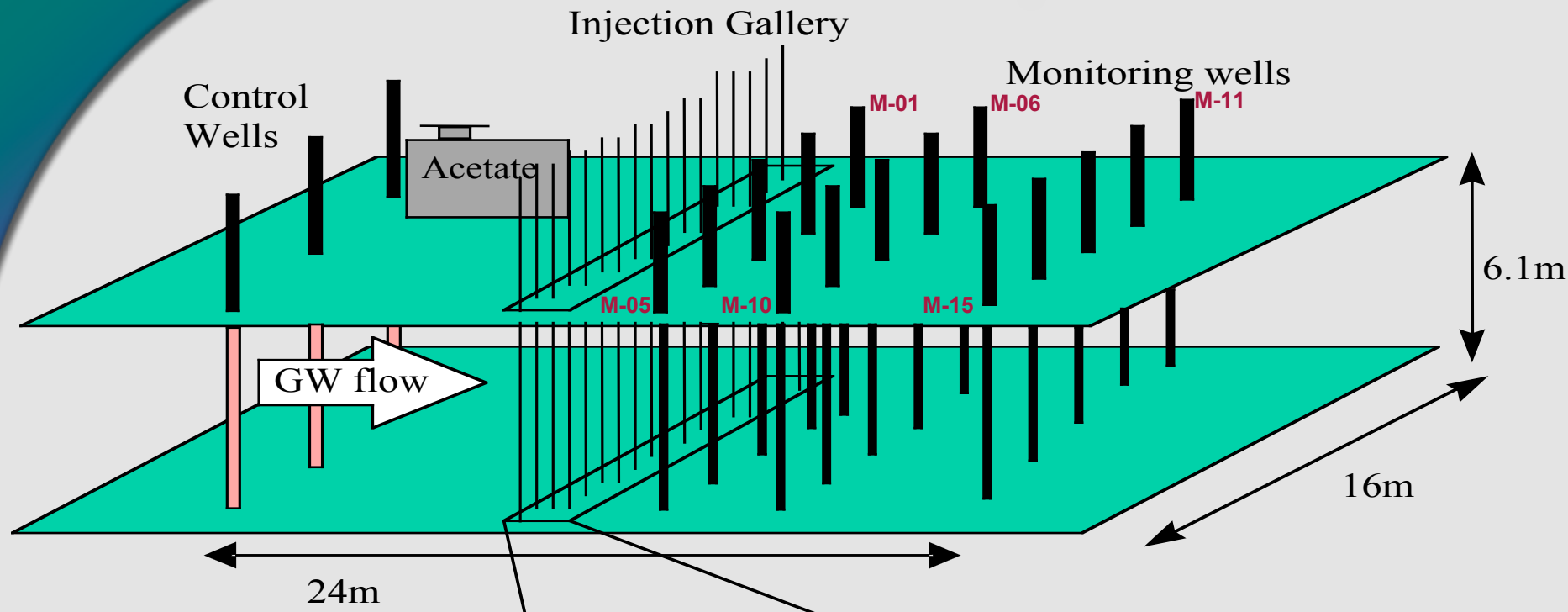
▶ Previous modeling

gaps & issues

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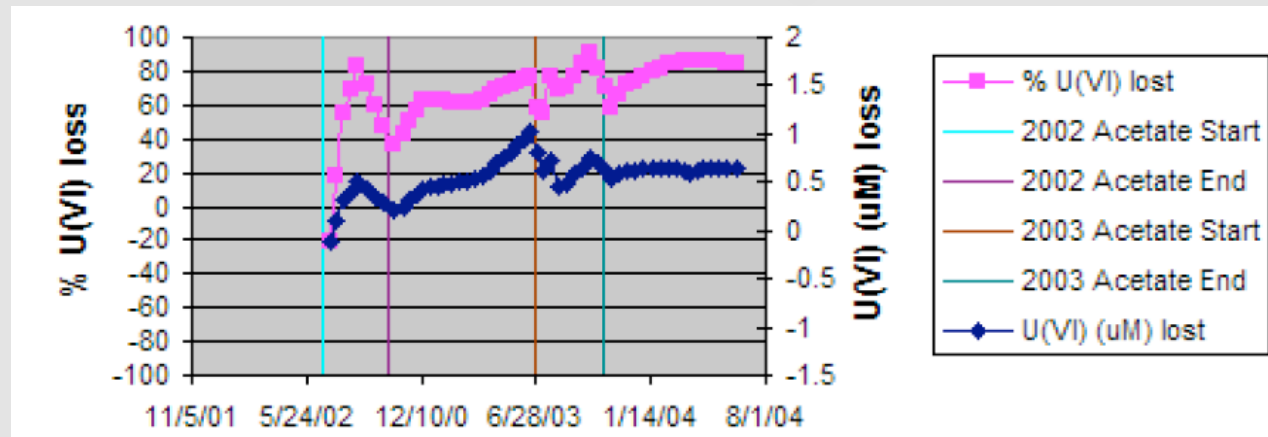


2002 & 2003 Field Experiments



- ▶ Hypothesis: Dissimilatory metal-reducing bacteria grown on acetate can be used to immobilize U(VI) in a field setting
- ▶ Continuous influx of elevated uranium (140 to 350 ug/L)
- ▶ 2002: Initial biostimulation
 - 123-day release period
 - 100 mM acetate, 10 mM bromide
 - average rate ~70 L/d (range 0 to 120 L/d)
- ▶ 2003: 2nd biostimulation
 - 109-day release
 - 300 mM acetate, 10 mM bromide
 - Average rate ~80 L/d (range 0 to 400 L/d)

Observations from Old Rifle Biostimulation Experiments



Proof of Principle

- ▶ Acetate stimulated growth of microbial populations that reduced aqueous U(VI) to U(IV), effectively removing uranium from groundwater through the precipitation of U(IV) mineral
- ▶ Initial bioreduction of aqueous U(VI) was 75 to 85 percent efficient and attributed to iron reducing bacteria (*Geobacter sp.*)
- ▶ Subsequent onset of sulfate reduction, coincided with less efficient U(VI) removal from groundwater

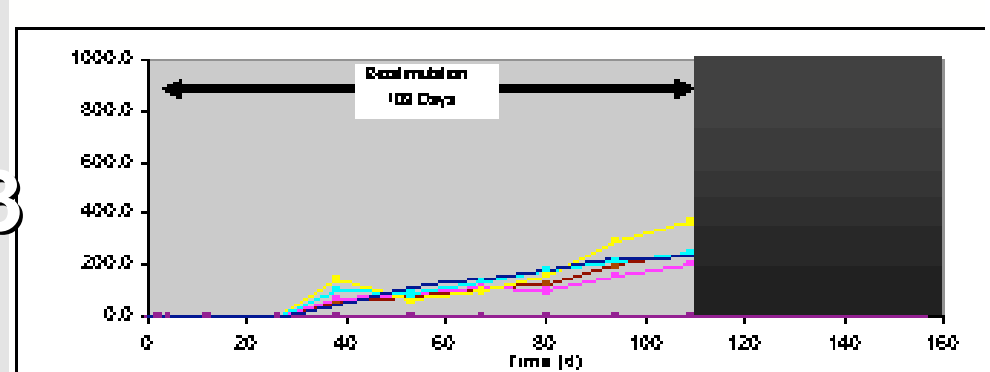
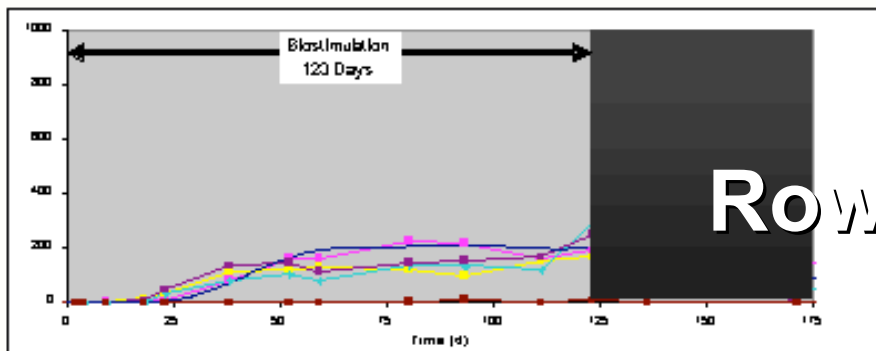
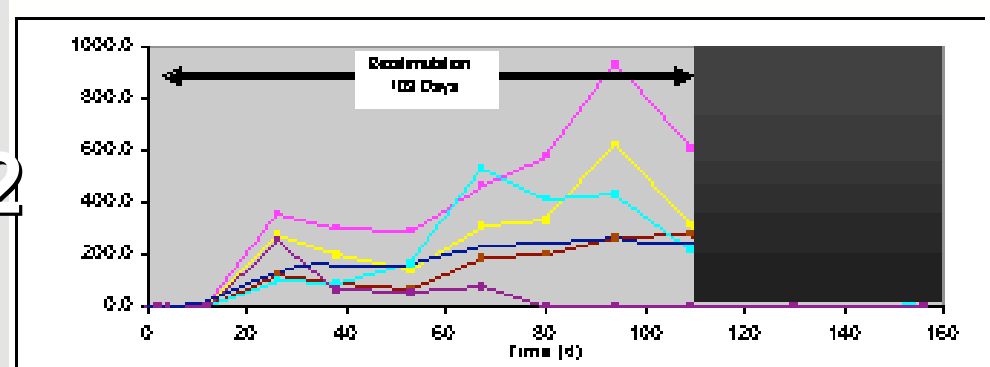
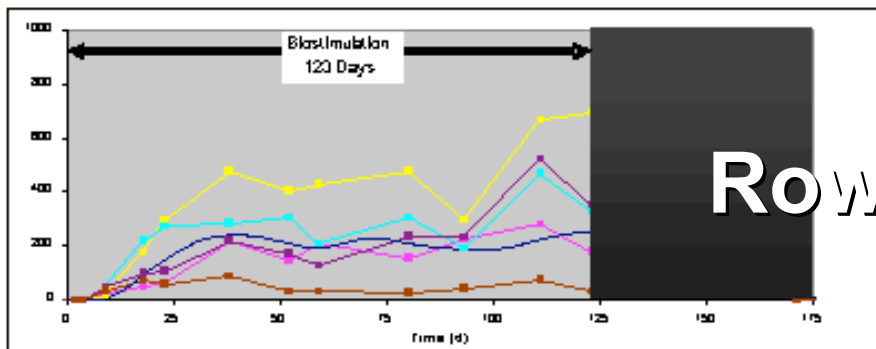
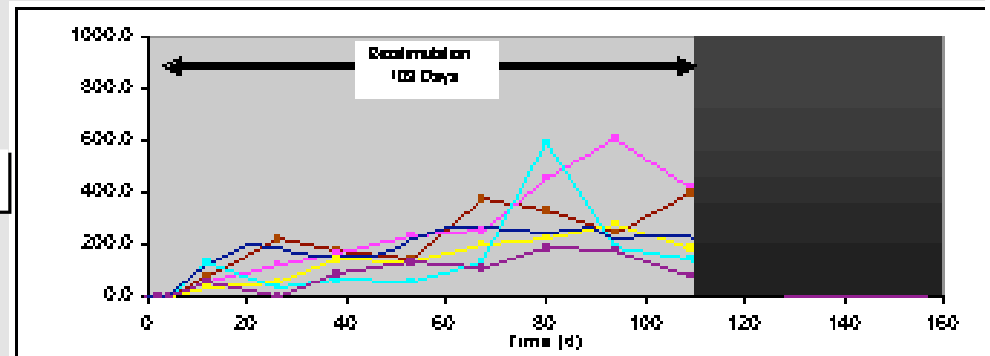
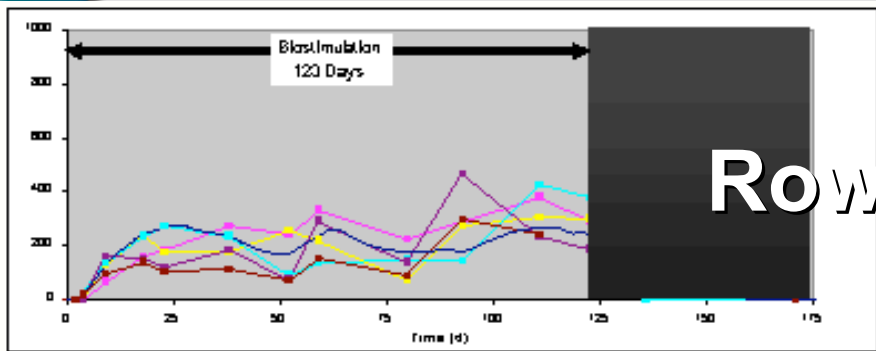
Modeling Approach

- ▶ Use mechanistic coupled process simulators as a systematic framework to
 - gain insight on the dominant processes and properties responsible for observed behaviors in the field
 - identify knowledge gaps that need to be addressed
- ▶ Philosophy
 - Start simple to isolate major behaviors
 - Systematically add process complexity and detail
- ▶ Field-scale flow and biogeochemical reactive transport simulation of biostimulation experiments
 - consistent gradient direction and magnitude: 1-D steady flow
 - iron and sulfate reducers: Fe(III), U(VI), and sulfate TEAPS
 - 2002 field experiment data set: calibrate flow, transport, and biogeochemical reaction parameters
 - 2003 field experiment: strict application of 2002 calibrated parameters

Flow and Transport Modeling

2002

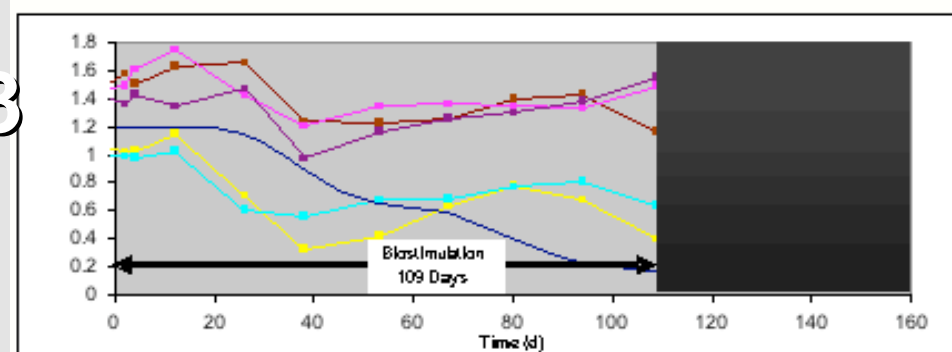
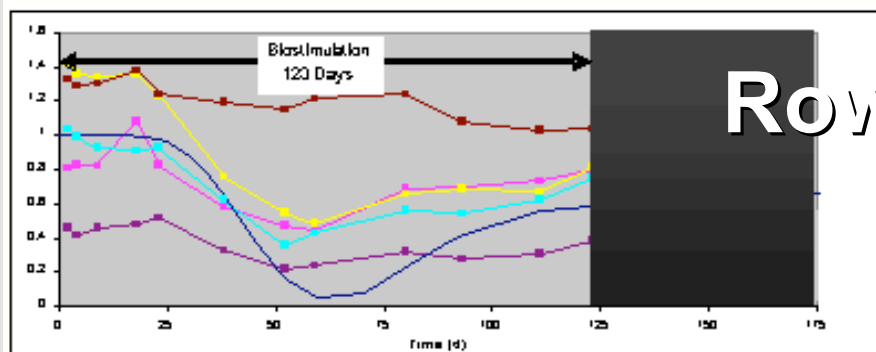
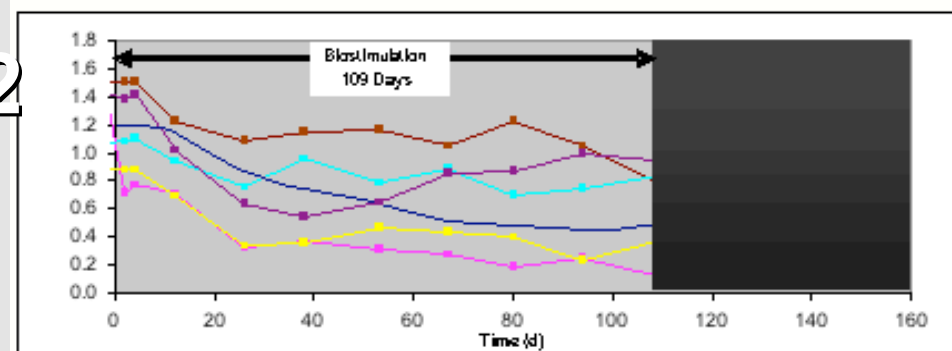
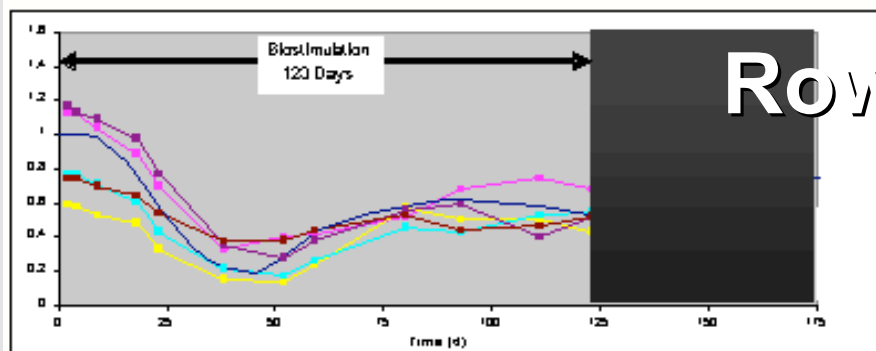
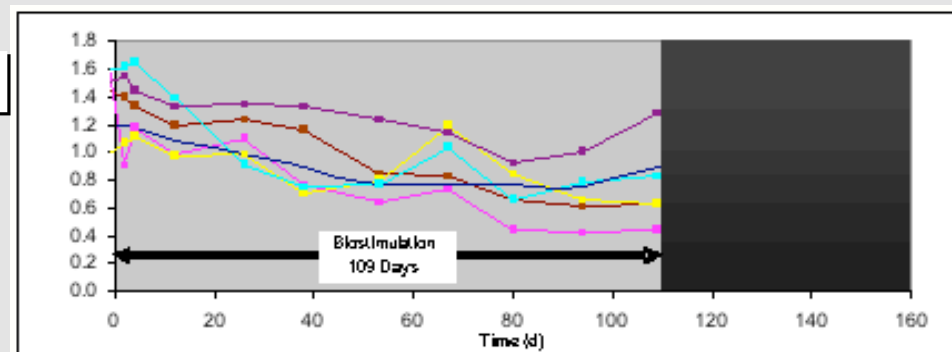
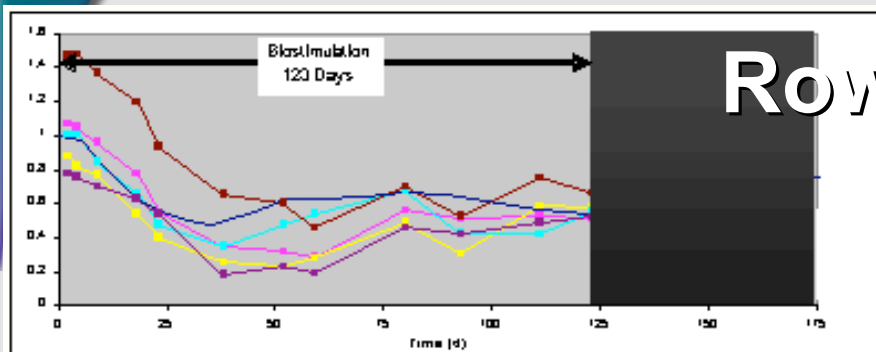
2003



Uranium

2002

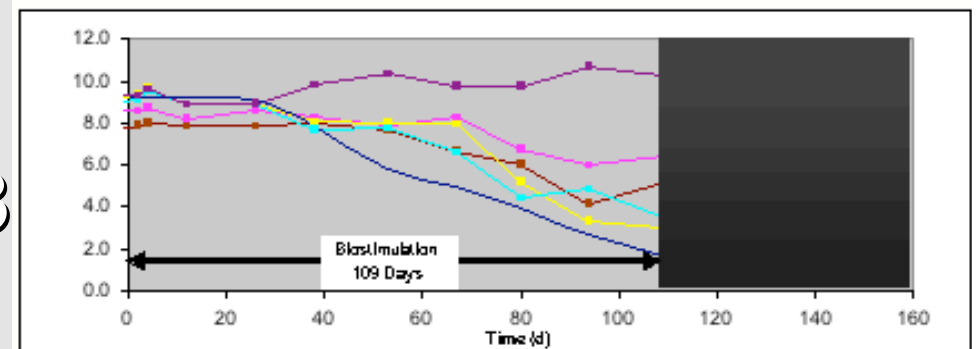
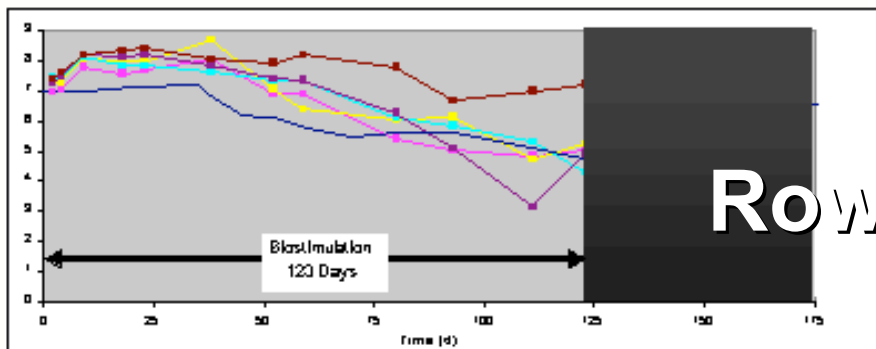
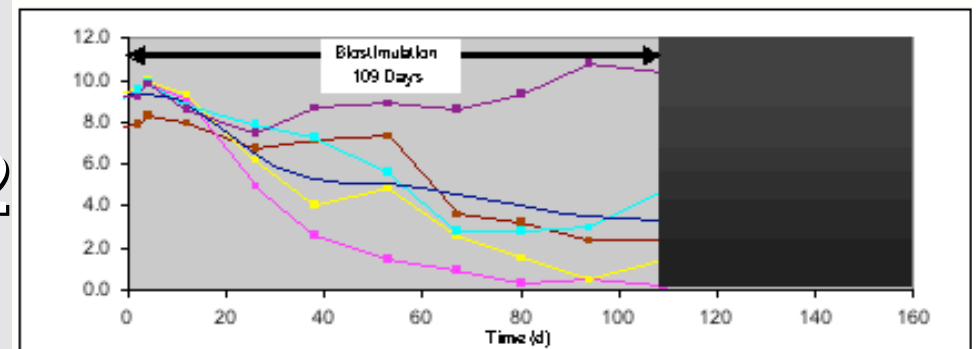
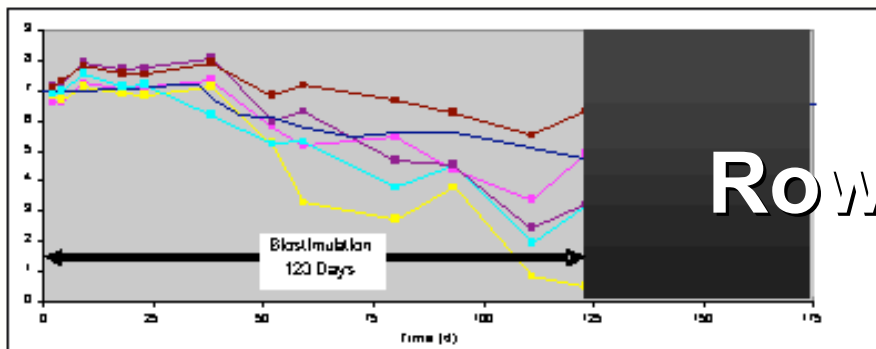
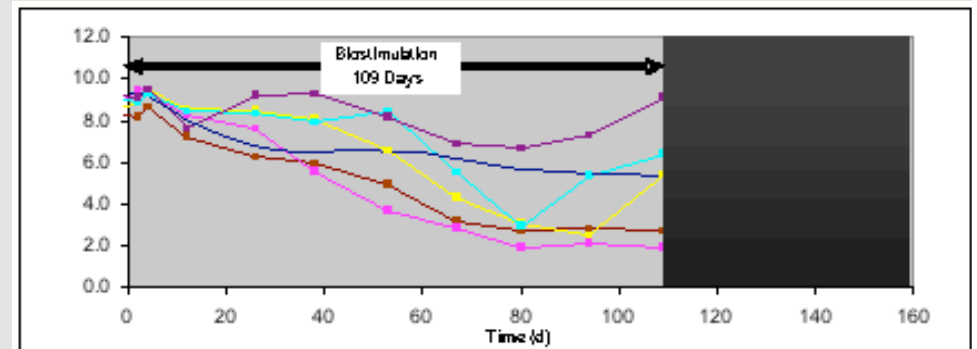
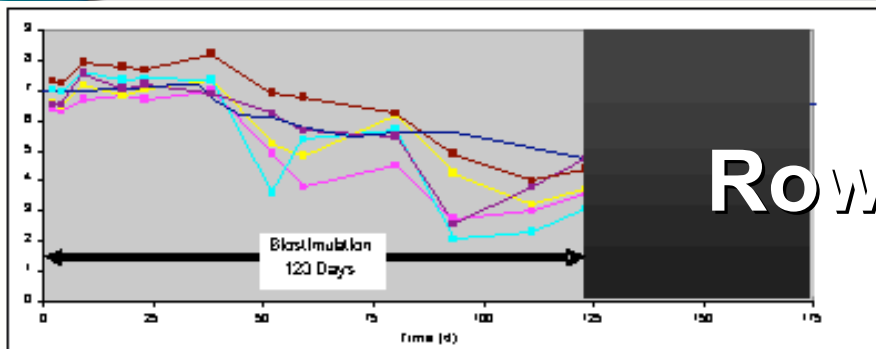
2003



Sulfate

2002

2003



Summary of Results

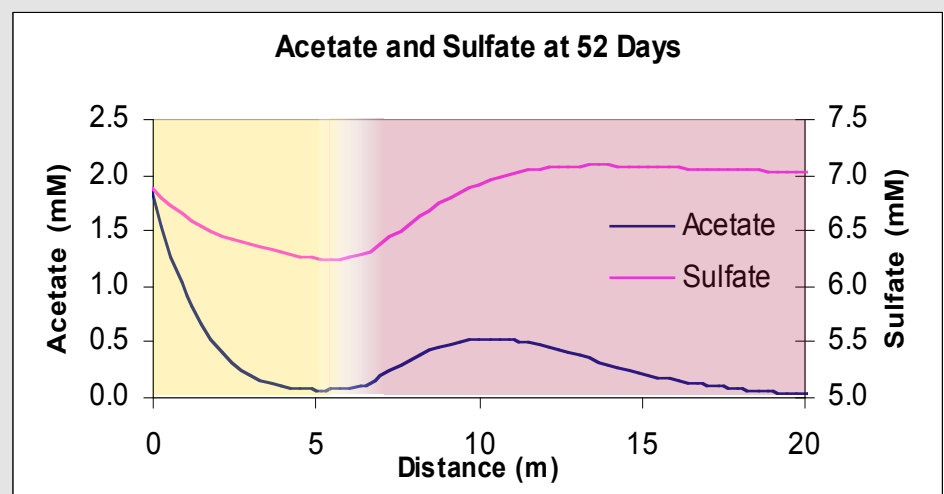
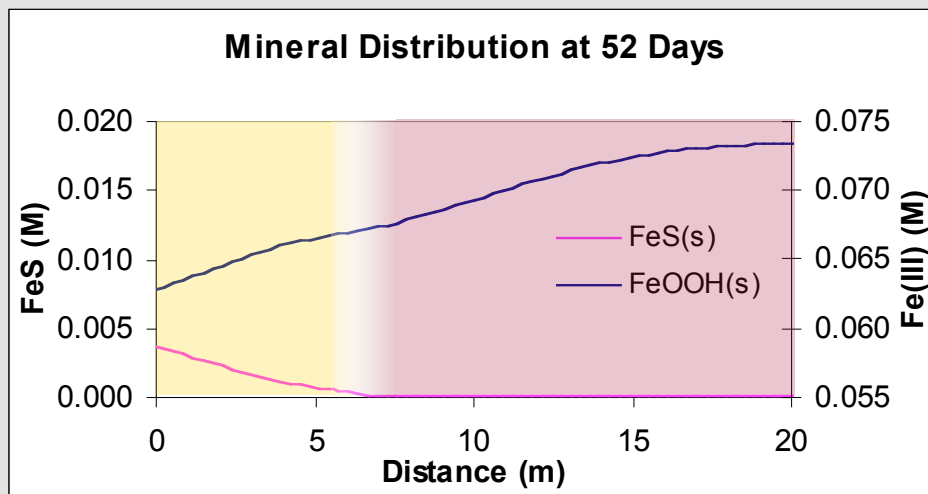
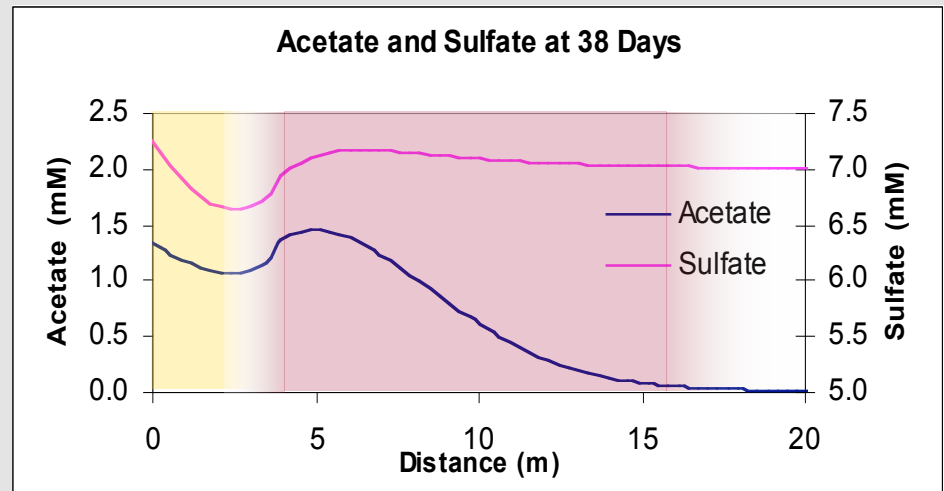
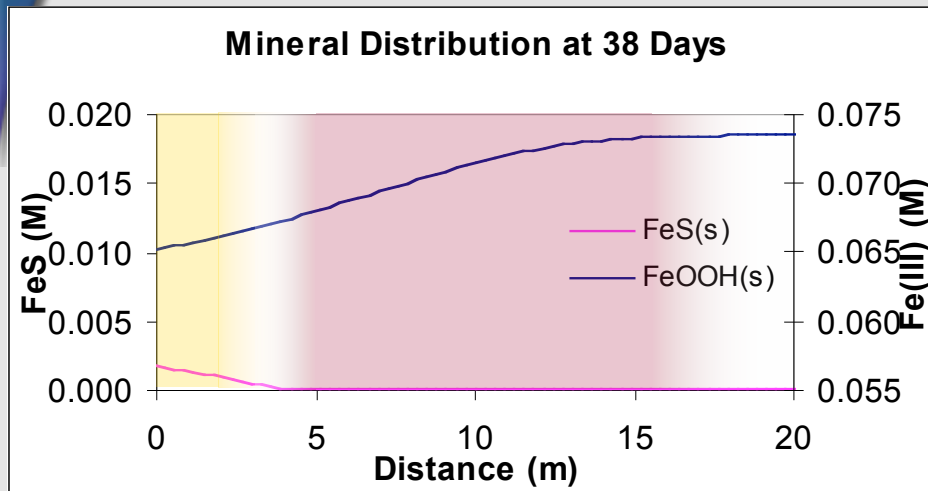
Observations during 2002 & 2003 biostimulation experiments are consistent with:

- 2 dominant microbial populations: iron reducers (i.e., *Geobacter*) and sulfate reducers
- 3 TEAPs: Fe(III) mineral, U(VI), sulfate
- Iron reducers concomitantly responsible for U(VI) reduction
- Onset of sulfate reduction triggered by consumption of threshold amount of Fe(III) mineral by iron reducers
- Lower U(VI) removal rate during sulfate reduction due to competition for acetate

2002 Simulation: Day 38, 52

SOLID PHASE

AQUEOUS PHASE



Principal Knowledge Gaps: Processes and Properties

- ▶ Large gap (“THE GAP”) between fundamental research in geochemistry, microbial ecology and molecular biology, and field-scale reactive transport modeling
 - Need development of detailed biogeochemical reaction network models
 - Need to link new knowledge of cell reactions/metabolism to kinetics and constraints of enzymatic processes
- ▶ Need for 3-D characterization of spatially-variable model parameters
 - controls flow paths, sediment reactivity, rate-limited mass transfer between pore domains, gas entrapment during water table fluctuation
 - simulation will play a key role in characterizing spatially variable parameters
 - testing and linking process models
 - accommodating a variety of data types from different scales
 - joint inverse modeling approaches

Biogeochemical Issues

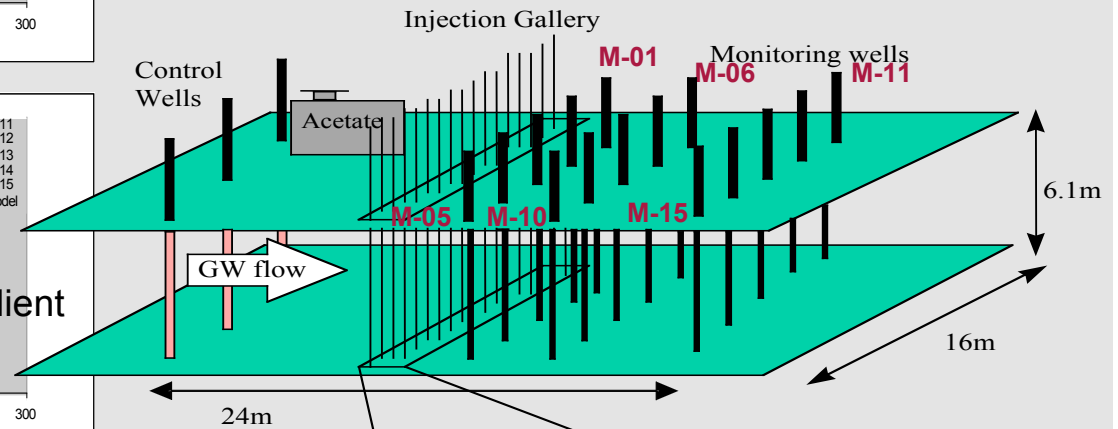
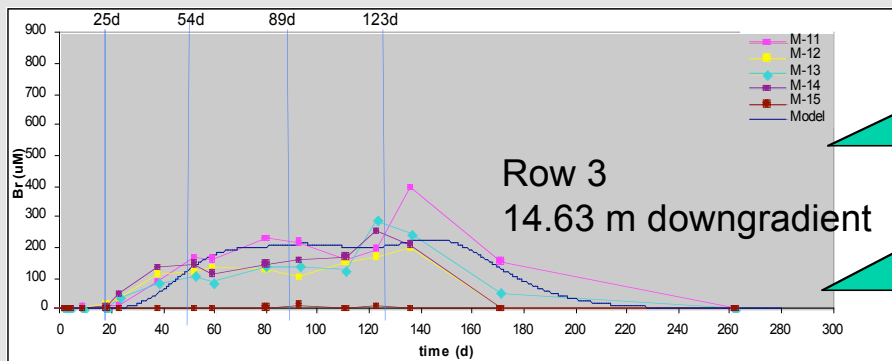
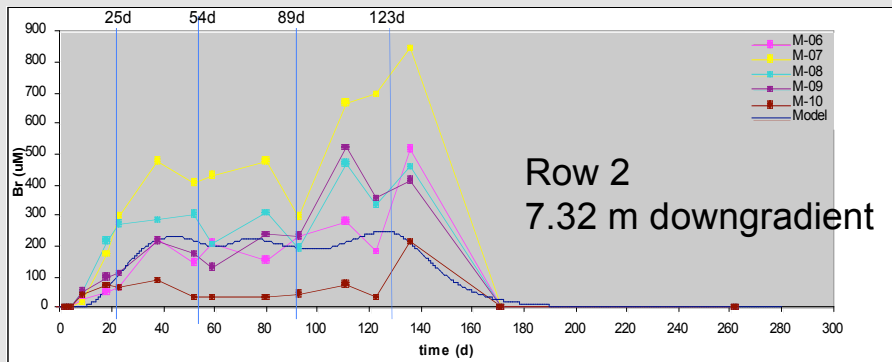
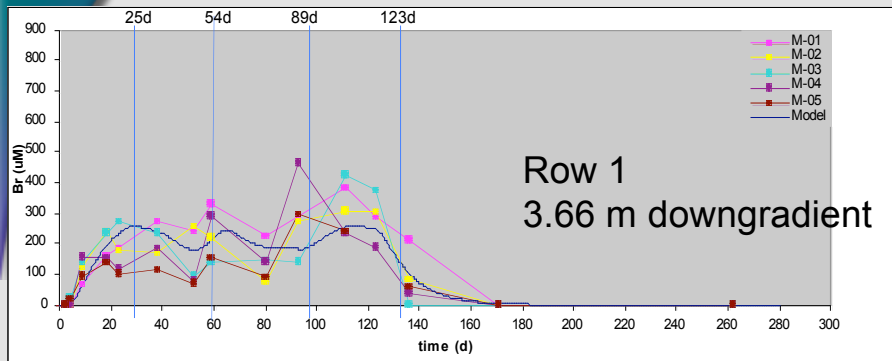
- Are there other microorganisms consuming acetate?
- What factors control the stoichiometry and rates for TEAP reactions?
 - nutrient limitations
 - water chemistry
 - mineral form
- What factors control the onset of sulfate reduction?
 - “bioavailable” iron (e.g., poorly crystalline iron)
 - redox potential
 - metabolic lag
- What is the role of biomass in controlling U mobility?
 - production/consumption/decay
 - attachment/detachment
 - contribution to microbial reaction rates
 - effect on reactivity of mineral surfaces
 - sorption effects
- What is the role of U surface complexation before, during, and after biostimulation?

Preferential Flow and Transport Paths

- ▶ Considerable variability in bromide concentrations in the same row
- ▶ Row 2 has highest concentrations and maximum variability
- ▶ Some wells bypassed
- ▶ Issues

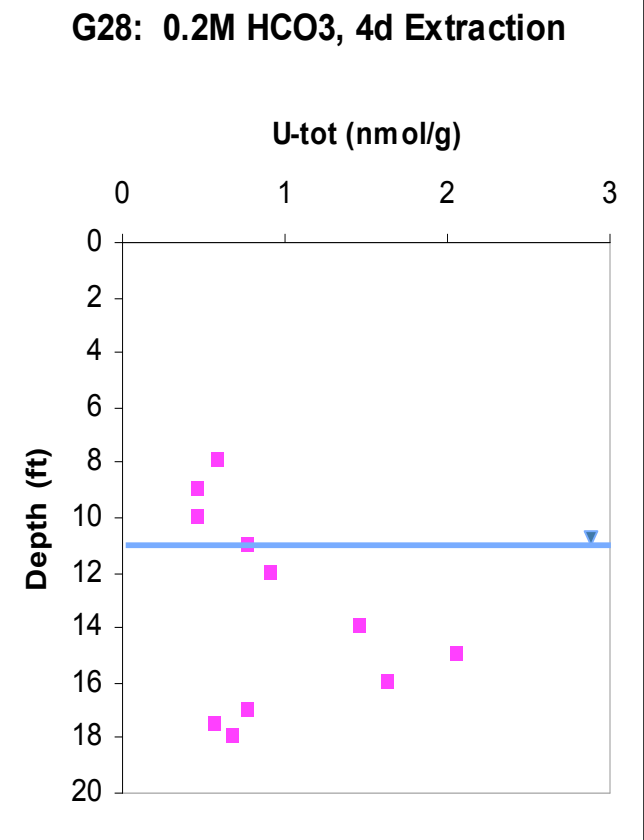
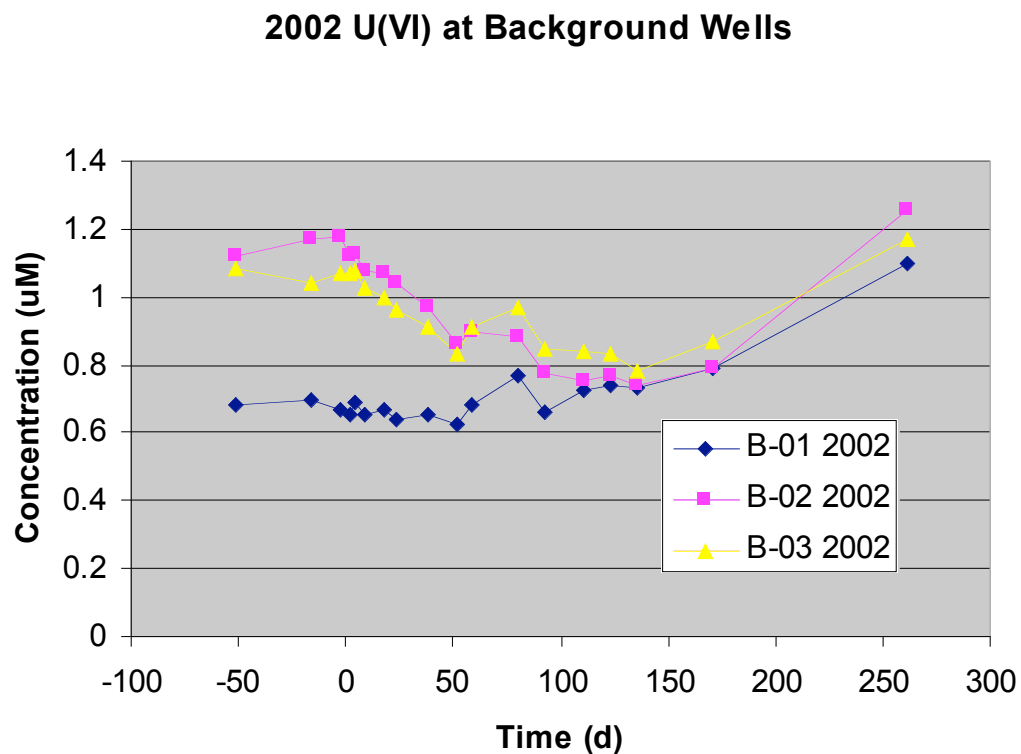
- Heterogeneous sediments
- Nonuniform metering of injectate to gallery wells

Old Rifle Test Plot



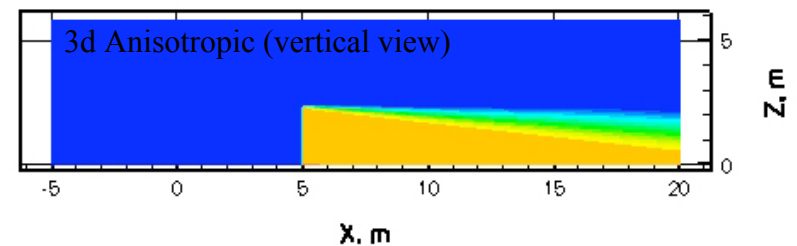
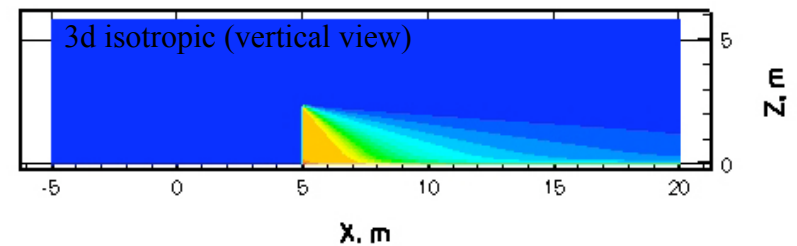
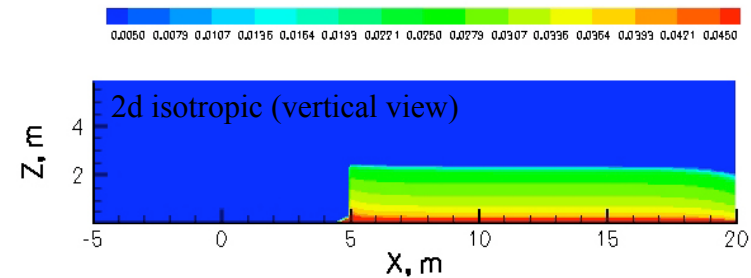
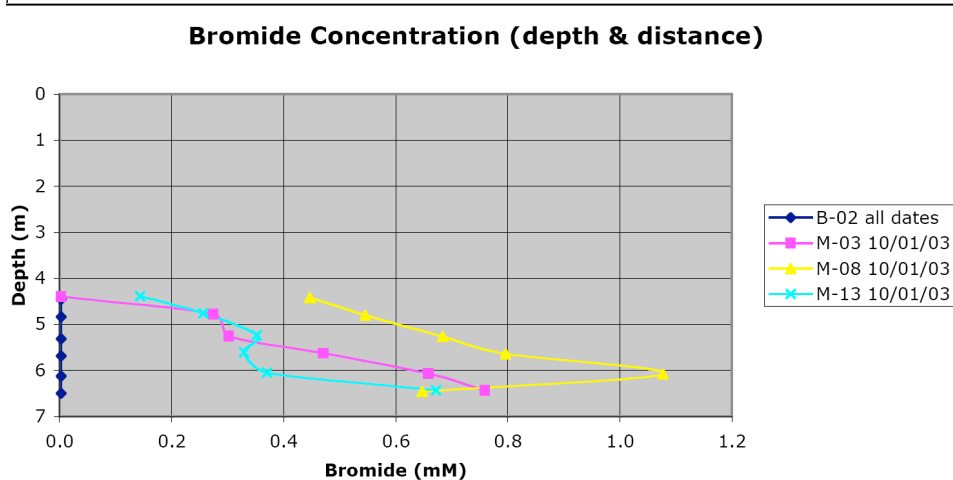
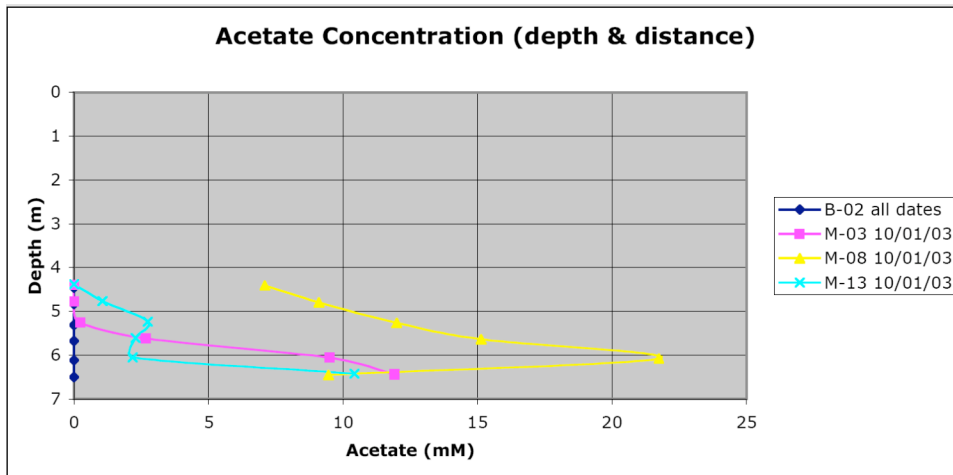
Spatially & Temporally Variable Uranium

- ▶ Spatially variable pre-biostimulation aqueous concentrations in 2002
 - U ranged from 0.3 to 1.5 μM
 - Fe(II) ranged from 18 to 250 μM
- ▶ Initial uranium distribution in G-28 sediments is strongly depth dependent



Density-Dependent Transport

- ▶ Acetate – bromide injectate is denser than GW
- ▶ Multilevel samplers show effect in 2003
- ▶ 3D modeling of individual gallery wells is necessary to accurately represent phenomenon
 - Modeling identifies sensitivity to anisotropy

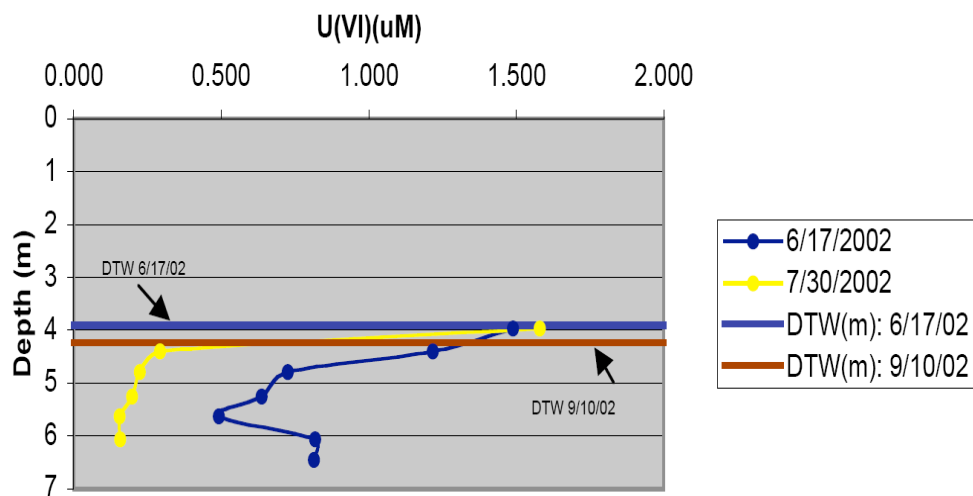


Stratified Water Chemistry

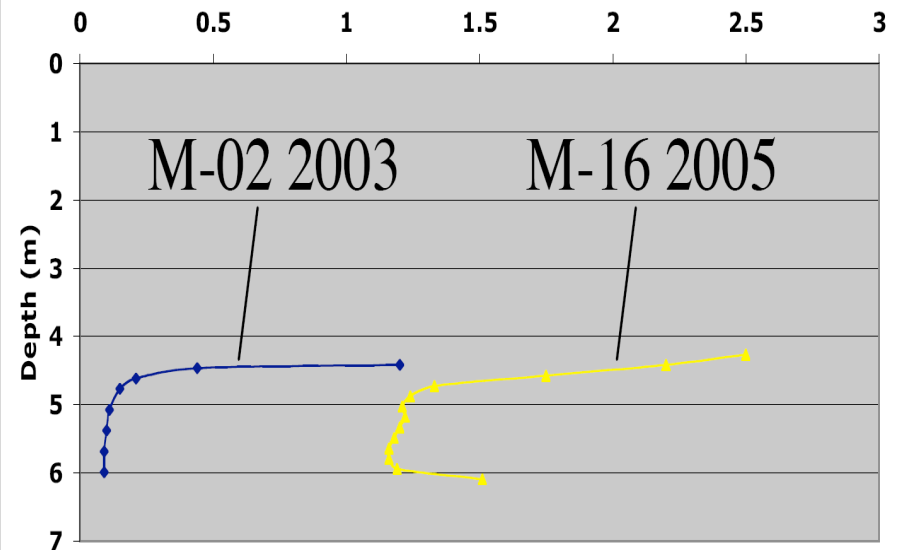
Depth-dependent U(VI) and DO

- ▶ Highest DO and U(VI) near the water table
- ▶ Issues
 - Oxygen diffusion through water table
 - Background utilization of DO
 - Screened interval of wells

M-03 U(VI) vs. Depth (m) 6/17-9/19/02



DO (mg/l)



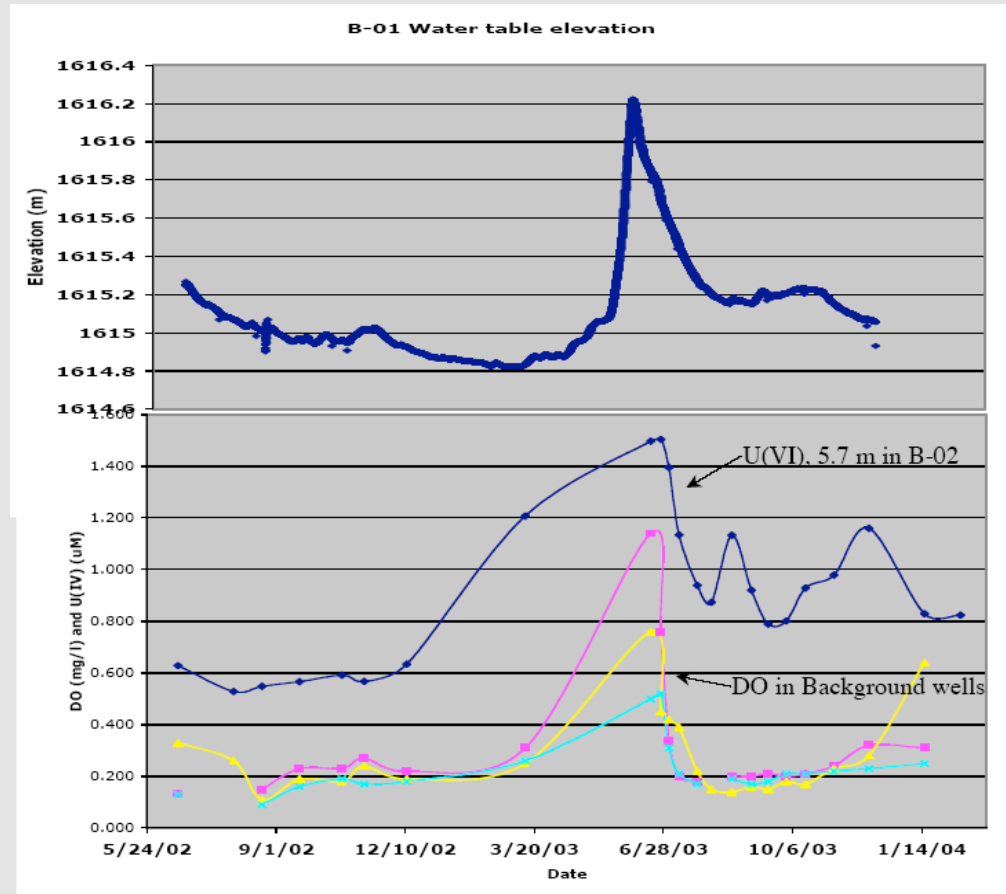
How do seasonal and episodic hydrologic events affect uranium behavior?

Seasonal and event-driven changes

- Velocity field
- Oxidation of zones affected by water table fluctuations

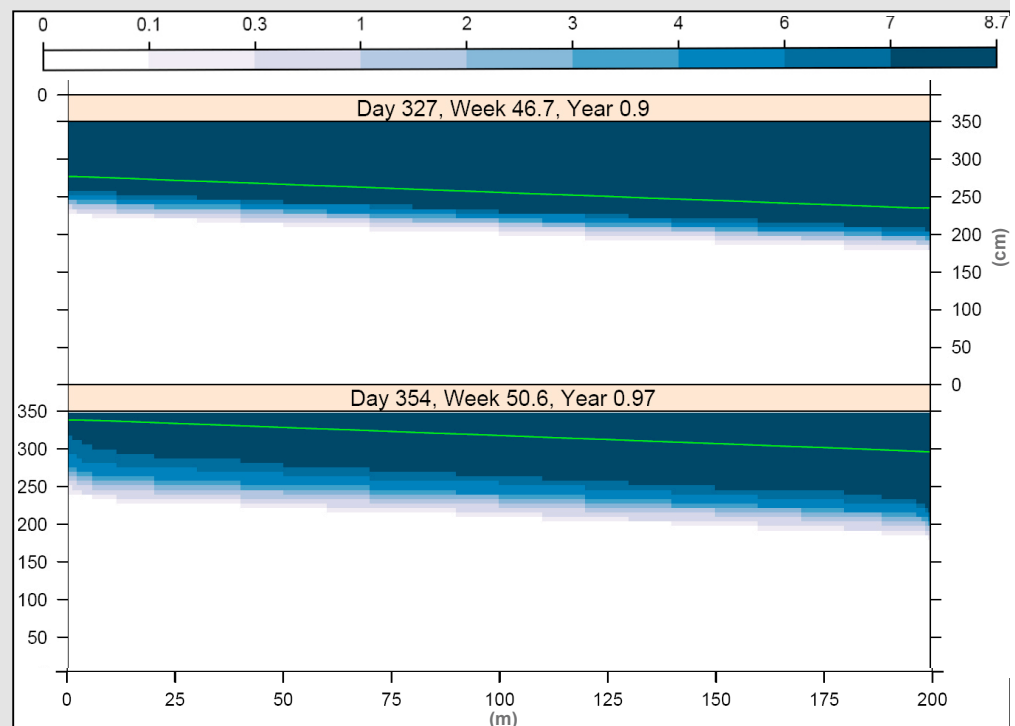
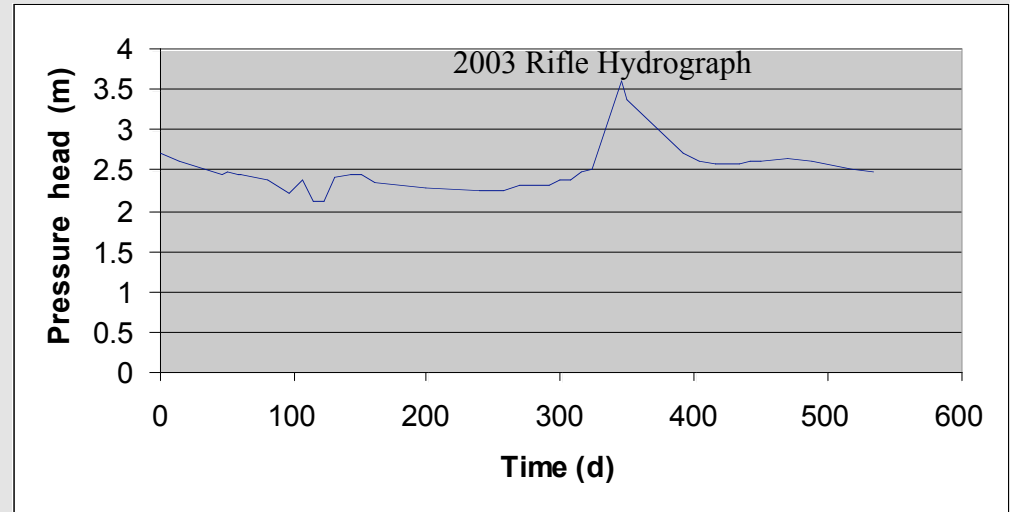
Issues

- Rapid oxidation of zones affected by water table fluctuations
- Highest U concentrations bypassing treatment zones



Oxygen Entrapment during Water Table Fluctuation

- ▶ 2-phase flow model of aquifer-vadose zone system
- ▶ Hysteretic saturation function calculates entrapped gas during 2003 transient
- ▶ Oxygen entrapped during water table rise dissolves into GW
- ▶ Water table recession moves enhanced DO deeper in the aquifer

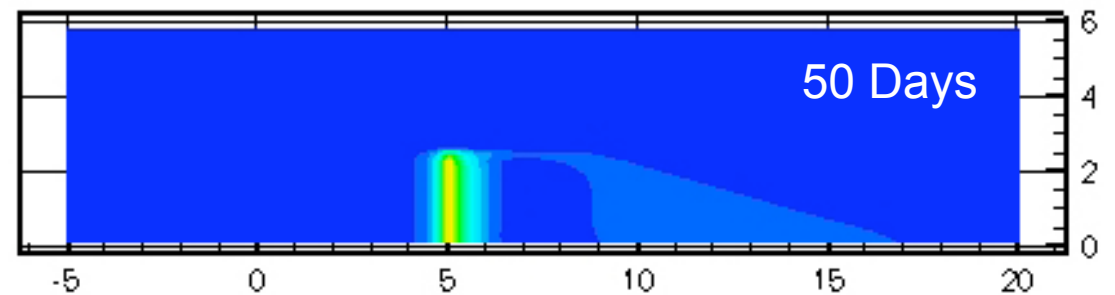
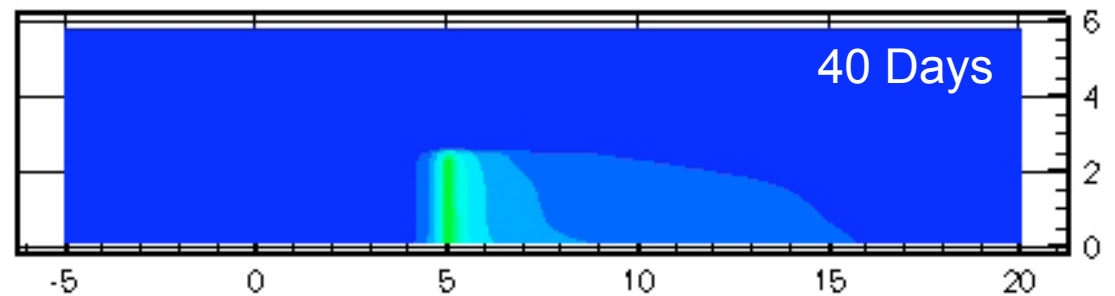
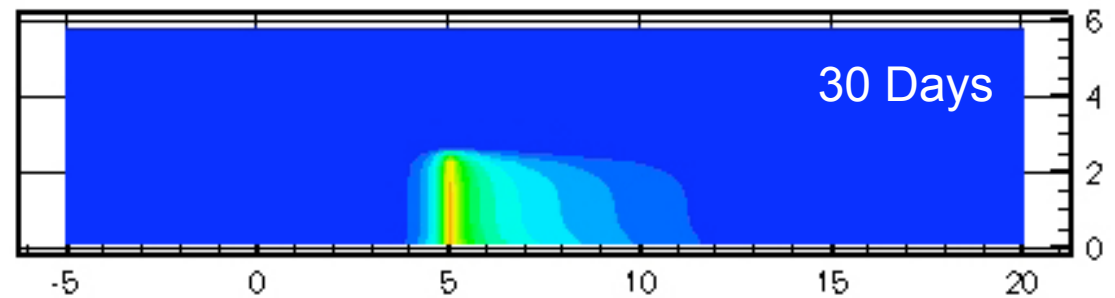


Coupling the Processes

2002 Field Experiment

- ▶ 2-phase 3D flow simulation
- ▶ Oxygen entrapment during water table transients
- ▶ Transient influent water chemistry
- ▶ Transient acetate injection
- ▶ Density effects
- ▶ Biogeochemical reactions

Acetate Concentrations (M)



Path Forward

► Plenty of issues identified by earlier work

○ providing opportunity to begin to holistically address the goal of a quantitative field-scale understanding of uranium bioremediation

Paradigm shift from extension of laboratory studies to field-scale context of site-specific conditions

Address “THE GAP” between fundamental research and field-scale reactive transport modeling

● collaborative modeling expertise

Carl Steefel, biogeochemical reactive transport from Hubbard et al. EMSP project

- Mahadevan/Scheibe, in silico modeling from Lovley et al. ERSP project

