Simulation of Microscale Chemical Separation Processes Using the Lattice-Boltzmann Method

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Introduction

- The efficiency of chemical separation processes is enhanced in microscale systems due to the reduced diffusion length scales.
- Phase interfaces (liquid-liquid, liquid-vapor, fluid-solid) play an important role in many of these processes. Surface tension and wettability can dominate flow behavior at micron length scales.
- The objective of this work is to develop a simulation capability which can model microseparation processes, including interface behavior.



Lattice-Boltzmann (cont)

The density and velocity are calculated by

$$\rho = \sum_{i=0}^{b} f_i$$
 $\rho u = \sum_{i=1}^{b} e_i f_i$
 $\rho_k = \sum_{i=0}^{b} g_i^{(k)}$

The equations of motion are

$$f_i(r+e_i,t+1) = f_i(r,t) - \frac{1}{\tau_{\rho}}(f_i(r,t) - f_i^{eq}(r,t))$$

$$g_i^{(k)}(r+e_i,t+1) = g_i^{(k)}(r,t) - \frac{1}{\tau_k} (g_i^{(k)}(r,t) - g_i^{(k)eq}(r,t))$$

where τ_{P} and τ_{k} are relaxation times.



Equilibrium Distributions

The equilibrium distributions f_i^{eq} and g_i^{(k)eq} are given by

$$\begin{split} f_i^{eq} &= A + \frac{\rho D}{bc^2} e_i \cdot u + \frac{\rho D(D+2)}{2bc^4} u \cdot e_i e_i \cdot u - \frac{\rho D}{2bc} u \cdot u + G_{\alpha\beta} e_{i\alpha} e_{i\beta} \\ f_0^{eq} &= A_0 - \frac{\rho}{c^2} u \cdot u \\ g_i^{(k)eq} &= B^{(k)} + \frac{\rho_k D}{bc^2} e_i \cdot u + J_{\alpha}^{(k)} e_{i\alpha} + \frac{\rho_k D(D+2)}{2bc^4} u \cdot e_i e_i \cdot u - \frac{\rho_k D}{2bc} u \cdot u + H_{\alpha\beta}^{(k)} e_{i\alpha} e_{i\beta} \\ g_0^{(k)eq} &= B_0^{(k)} - \frac{\rho_k}{c^2} u \cdot u \end{split}$$

where D is the dimension, b the number of lattice vectors and c is the lattice constant.



Solution Procedure

At each time step:

- Stream particles from adjacent lattice points.
- Calculate equilibrium particle distribution at each lattice point.
- Collision relax the distribution toward equilibrium values. The rate of relaxation is determined by the time step / viscosity.

Additional steps are required for boundary conditions, body forces, gradient terms, output processing, etc.



Solution Procedure



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Model Features

- For single-phase, single-fluid systems:
 - 2nd order solution to the Navier-Stokes equation
 - Diffusion transport based on gradient of chemical potential
- For multi-phase, multi-component systems:
 - liquid-liquid and liquid-vapor interfaces are represented by a continuous density profiles which freely moves through the lattice grid.
- The wettability and adsorption of individual chemical species on solid surfaces are specified using surface-fluid potential gradients..



Model Equation of State

The free energy functional has the Cahn-Hilliard form

$$\Psi = \int (\Psi(T, \rho, \{\rho_k\} + \kappa(\nabla \rho_1)^2 + \kappa(\nabla \rho_2)^2) dr$$

■ The chemical potential for species k>2 is given by

$$\mu_k = kT \ln(\rho_k) - \alpha_k \frac{\rho_1}{\rho_1 + \rho_2} - \beta_k \frac{\rho_2}{\rho_1 + \rho_2}$$

where α and β control the affinity of the species for the two fluids.



Fluid-fluid Extraction (continuous film)





Fluid-fluid Extraction (continuous film)





Fluid-fluid Extraction (contactor plate)





Pressure Differential (continuous film)





Countercurrent Flow (continuous film)





Pressure Differential (contactor plate)





Breakthrough (contactor plate)





Current Activities

- Simulation of other chemical separations processes
 - Distillation
 - Surface adsorption
- Verification and validation of lattice-Boltzmann simulation method
- Supporting the design of microseparation devices



Conclusions

- A simulation capability has been developed which can model microseparation processes, including convection and diffusion transport.
- In addition, this method is used to model the dynamic interface behavior under different operating conditions. Issues such as interface stability and contactor plate breakthrough can be addressed using this approach.
- The usefulness of this method has been illustrated by simulating the behavior of model separation devices.

