

# PROJECT facts

U.S. DEPARTMENT OF ENERGY  
OFFICE OF FOSSIL ENERGY  
NATIONAL ENERGY TECHNOLOGY LABORATORY



## FEASIBILITY OF LARGE-SCALE CO<sub>2</sub> OCEAN SEQUESTRATION

### Background

The increase of carbon dioxide (CO<sub>2</sub>) and other greenhouse gas emissions, due in large part to the burning of fossil fuels for energy consumption, has spawned debate over global climate change and prompted the emergence of various remediation strategies. The option of ocean sequestration involves capturing and separating CO<sub>2</sub> from the flue gas of fossil fuel-burning power plants and then converting it into a liquid state that is then transported by a submerged pipeline for injection into the deep ocean. Oceans offer enormous potential as carbon storage sinks, but the current level of scientific understanding to support ocean sequestration as a major CO<sub>2</sub> storage option remains uncertain. DOE had ocean sequestration as one of its research pathways in the past but, due to the need for further understanding, concluded projects in its portfolio in this area.

The Monterey Bay Aquarium Research Institute (MBARI) developed a novel set of experimental techniques for the deployment of small quantities of liquid CO<sub>2</sub> in the deep ocean in order to investigate the fundamental concepts of CO<sub>2</sub> ocean sequestration, particularly the behavior of CO<sub>2</sub> hydrates. Using an advanced Remotely Operated Vehicle (ROV) technology, MBARI carried out controlled releases of CO<sub>2</sub> to observe and measure the processes taking place.

### Description

Through a series of experiments conducted offshore California and at the MBARI, researchers investigated the physical, biological, and chemical behavior of CO<sub>2</sub> hydrates on the seafloor at depths up to 3,600 meters. Similar to methane hydrates—the ice-like complexes of water and methane—CO<sub>2</sub> can also form hydrates, and the storage of CO<sub>2</sub> in hydrate pools at the bottom of the ocean were investigated. A number of research cruises using the ROV to study the ocean storage of CO<sub>2</sub> hydrate off Monterey Bay have been completed. The physical chemistry and the biological effects of hydrate formation in the deep ocean were examined by means of small-scale batch experiments.

The research group at Washington University, with MBARI, used in situ Raman spectroscopy to carry out the first direct in situ analysis of CO<sub>2</sub> hydrates on the ocean floor, in the entrained and surrounding fluids, and in the sediments adjacent to the hydrates. Information on hydrate-sediment interaction is essential for the evaluation of ocean sequestration of CO<sub>2</sub>.

### CONTACTS

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## PARTNERS

Monterey Bay Aquarium Research Institute (MBARI)

Washington University at St. Louis

## PERIOD OF PERFORMANCE

09/18/2000 to 03/31/2008

## COST

### Total Project Value

\$1,593,777

### DOE/Non-DOE Share

\$1,270,041 / \$323,736

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## Primary Project Goal

The primary goal of this project is to investigate the chemical, physical, and biological behavior of CO<sub>2</sub> hydrates in the deep ocean. These data are necessary to help evaluate the storage of CO<sub>2</sub> in hydrate pools at the bottom of the ocean, an option that has been considered for long-term CO<sub>2</sub> storage.

## Objectives

- Determine the long-term fate of CO<sub>2</sub> and CO<sub>2</sub> hydrates on the sea floor.
- Study biological responses to the disposed material.
- Examine geochemical interactions with sediments and pore waters.

## Benefits

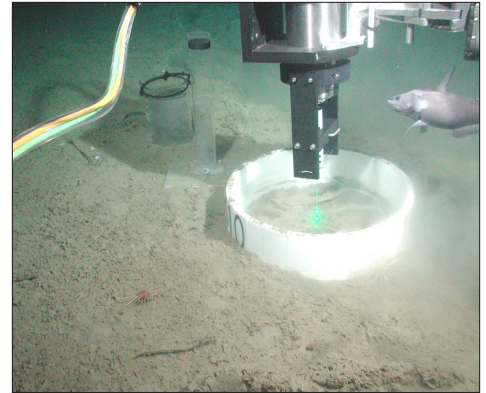
This project will provide further understanding of the behavior of CO<sub>2</sub> within the ocean environment and will afford individuals interested in ocean engineering and the study of CO<sub>2</sub> and methane hydrates with useful information. The research will demonstrate that hydrate pools at the bottom of the ocean have the potential for long-term storage of large quantities of CO<sub>2</sub>.

## Accomplishments

MBARI used a small-scale delivery system with a capacity of 56 liters to study CO<sub>2</sub> interactions in the ocean. Five controlled delivery dives were executed, with the CO<sub>2</sub> delivered to a central corral complex. Results showed a strong tidal periodicity in the water plume of lowered pH and a complex set of biological responses. Below a depth of about 3,000 meters, the density of liquid CO<sub>2</sub> exceeds that of seawater, and the CO<sub>2</sub> is quickly converted into solid hydrate by reaction with the surrounding water. An analysis of the acoustic detection and modeling of a rising deep-sea liquid CO<sub>2</sub> plume was also completed. The results are applicable to detection of leakage of CO<sub>2</sub> from the sea floor, either from natural CO<sub>2</sub> vents or from purposefully disposed CO<sub>2</sub> in sub-sea geologic formations. Research results have been published in peer-reviewed literature and disseminated at scientific meetings, symposia, and conferences.

## Planned Activities

The research team will prepare the final project report, as well as articles for publication in the scientific literature.



*Testing the waters: An experiment to investigate the fundamental science of ocean CO<sub>2</sub> sequestration at a depth of 3,600 meters off the coast of California. A small pool of liquid CO<sub>2</sub> is sensed by the beam of a laser Raman spectrometer to record the chemical status of the material. This sea floor laboratory is controlled by a remotely operated vehicle.*



*Formation of CO<sub>2</sub> hydrate mounds at 3,610 meters.*