# Sequestration

#### 08/2002

U.S. DEPARTMENT OF ENERGY NATIONAL ENERGY TECHNOLOGY LABORATORY

# PHYSICS AND CHEMISTRY OF COAL-SEAM CO<sub>2</sub> Sequestration & Coalbed Methane Production

## Background

Recently, the concept and practice of carbon management via the sequestration of carbon dioxide by coal seams and the concurrent production of coalbed methane (CBM) have increased in potential significance. The injection of  $CO_2$ into deep, unmineable, gassy coal seams may substantially increase  $CH_4$ (methane) production above the level achievable by standard depressurization methods. Water continues to play a key role in CBM production, yet explanations in the coal literature of how water does this on a molecular scale are presently undeveloped. Thus, a fundamental understanding of the mechanism(s) by which sorbed water influences, or can influence, coalbed methane production, with and without  $CO_2$  sequestration is necessary.

Additionally, research is being conducted to obtain information useful for assessing the technical feasibility of  $CO_2$  sequestration in coal-seams. Areas of interest include estimation of the capacity of a coal-seam to adsorb  $CO_2$  (*adsorption isotherm*), the validity of inter-lab comparisons of isotherm data (*inter-lab precision*), and the stability of the  $CO_2$  saturated phase once formed—especially with respect to how it might be affected by changes in the post-sequestration environment (*environmental effects*). The affects of temperature, pressure, and coal rank on the ability of coal to adsorb  $CO_2$  have been investigated.

# **Primary Project Goal**

The goals of the research are to ultimately provide guidelines for drilling of new CBM production wells and enable field engineers to determine if cases of poor  $CO_2$  sequestration and/or low methane productivity can be attributed to non-ideal coalbed temperatures/depths or, perhaps, to other factors.

# **Objectives**

• Determine the temperature dependence of CO<sub>2</sub> sequestration and methane production.

• Determine adsorption isotherms for pure gases in a static system for coals of NETL interest.

• Develop a flow system to generate adsorption isotherms via numerical techniques established for data analysis.

### PRIMARY PARTNERS

National Energy Technology Laboratory Pennsylvania State University University of Pittsburgh University of Oklahoma University of Southern Illinois CSIRO Netherlands Institute of Applied Geoscience TNO Illinois State Geological Survey

### **DOE FUNDING PROFILE**

 Prior FY's
 \$257,000

 FY2002
 \$441,207

 Future FY
 TBA

# TOTAL ESTIMATED COST

\$698,207

DOE

### **CUSTOMER SERVICE**

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### WEBSITE

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# Accomplishments

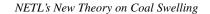
Advanced  $CO_2/CH_4$  Concepts ( $CO_2$  sequestration & CBM production): A method for simultaneously accounting for heats of  $CO_2$  and  $CH_4$  sorption/ desorption, moles of  $CO_2$  and  $CH_4$  sorbed/desorbed, extents of dehydration, and sample temperature was developed and a manuscript was prepared and accepted for presentation at various conferences. Mathematical methods for resolving complex calorimetric thermograms were developed. Accordingly, an apparent correlation between hypothetical extents of coal dehydration and predicted relative viscosities of water in the narrow capillaries, mesopores, and micropores of coal was discovered.

 $CO_2$  Sorption, Transport, & Environmental Chemistry ( $CO_2$  Sequestration): A static system for the measurement of adsorption isotherms was assembled, pressure-tested, and successfully employed to generate data along with a derived equation used to separate the actual surface adsorption from the effects of coal swelling on the isotherm shape. The extent of actual physical adsorption was determined, the heats of adsorption were calculated, and the values were found to agree within 10% of each other. NETL has developed a new theory that allows one to obtain information on coal swelling from the experimentally derived adsorption isotherm.

## **Benefits**

This project will provide guidelines for both efficient sequestration of carbon dioxide in coal seams and enhanced methane production. Through an understanding of the fundamental chemistry involved in the  $CO_2$  adsorption/CH<sub>4</sub> desorption process, it will be possible to select optimum conditions for  $CO_2$ -enhanced coalbed methane production/sequestration. The project has resulted in development of a new theory of coal swelling and how the  $CO_2$  adsorption process affects swelling. The new theory allows one to obtain information on coal swelling from simple adsorption isotherm measurements. The enhanced methane production associated with  $CO_2$  sequestration will help to defray sequestration costs. Additionally, by capturing carbon dioxide and sequestering it, harmful emissions into the atmosphere are prevented that may further increase global warming.

3.0 Adsoprtion Desorption daf 2.5 Langmuir + k (P/z) Pochahontas #3 Adsorbed CO<sub>2</sub>, mmole/g-coal, Langmuir 2.0 - k (P/z) Term -----1.5 1.0 0.5 0.5 1.0 1.5 2.5 5.0 0.0 2.0 3.0 3.5 4.0 4.5 P, MPa Proj186.pmd



Adsorption Isotherms Appear to Be Combinations of a Surface Adsorption Term and a Constant Term

 $n_{\rm exp} = n_{ads} + k(P/z)$