

SANS RESEARCH TOPICS

Boualem Hammouda

National Institute of Standards and Technology
Center for Neutron Research

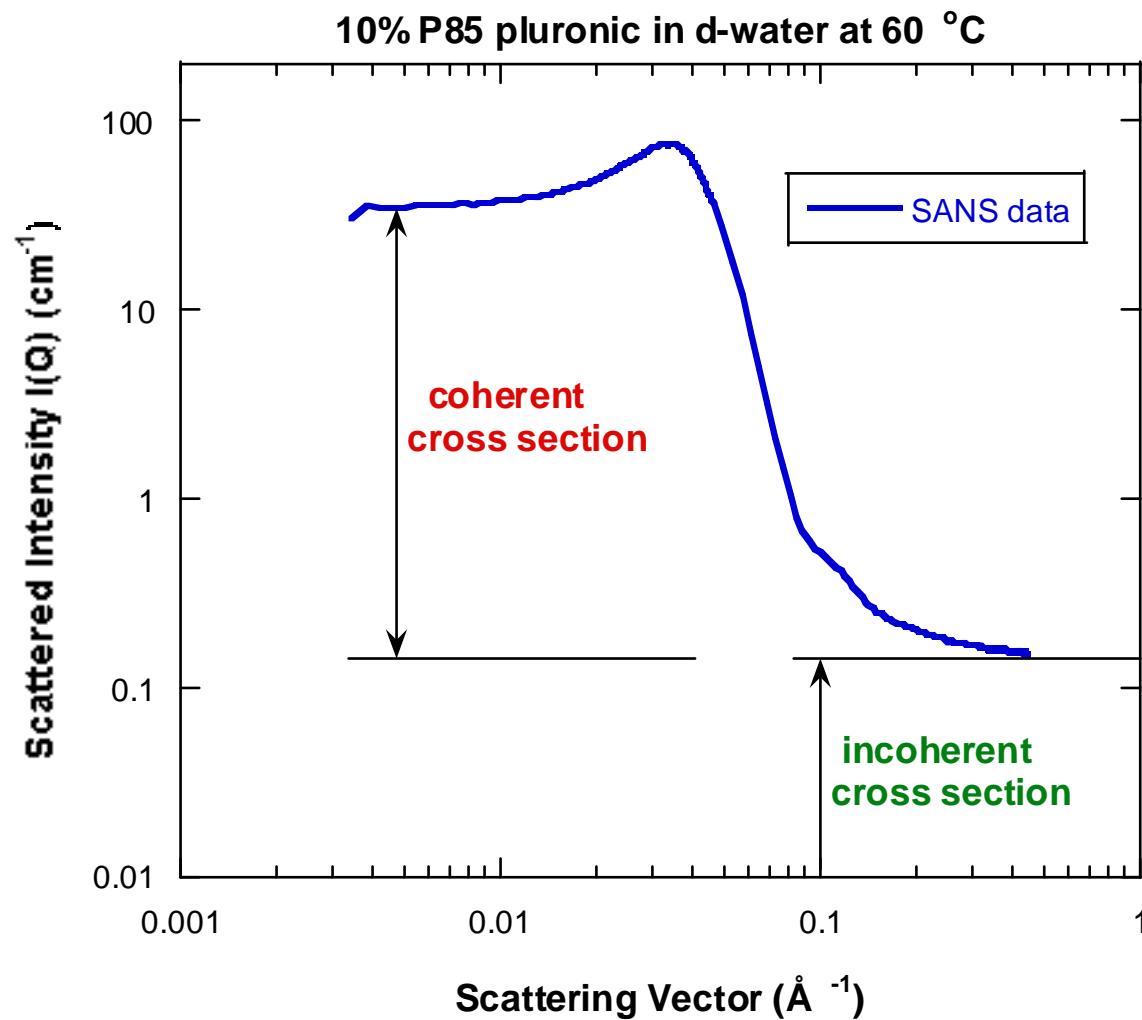
1. SANS from Pluronics
2. Polymer Blend Thermodynamics
3. Helix-to-Coil Transition in DNA

1. SANS FROM PLURONICS

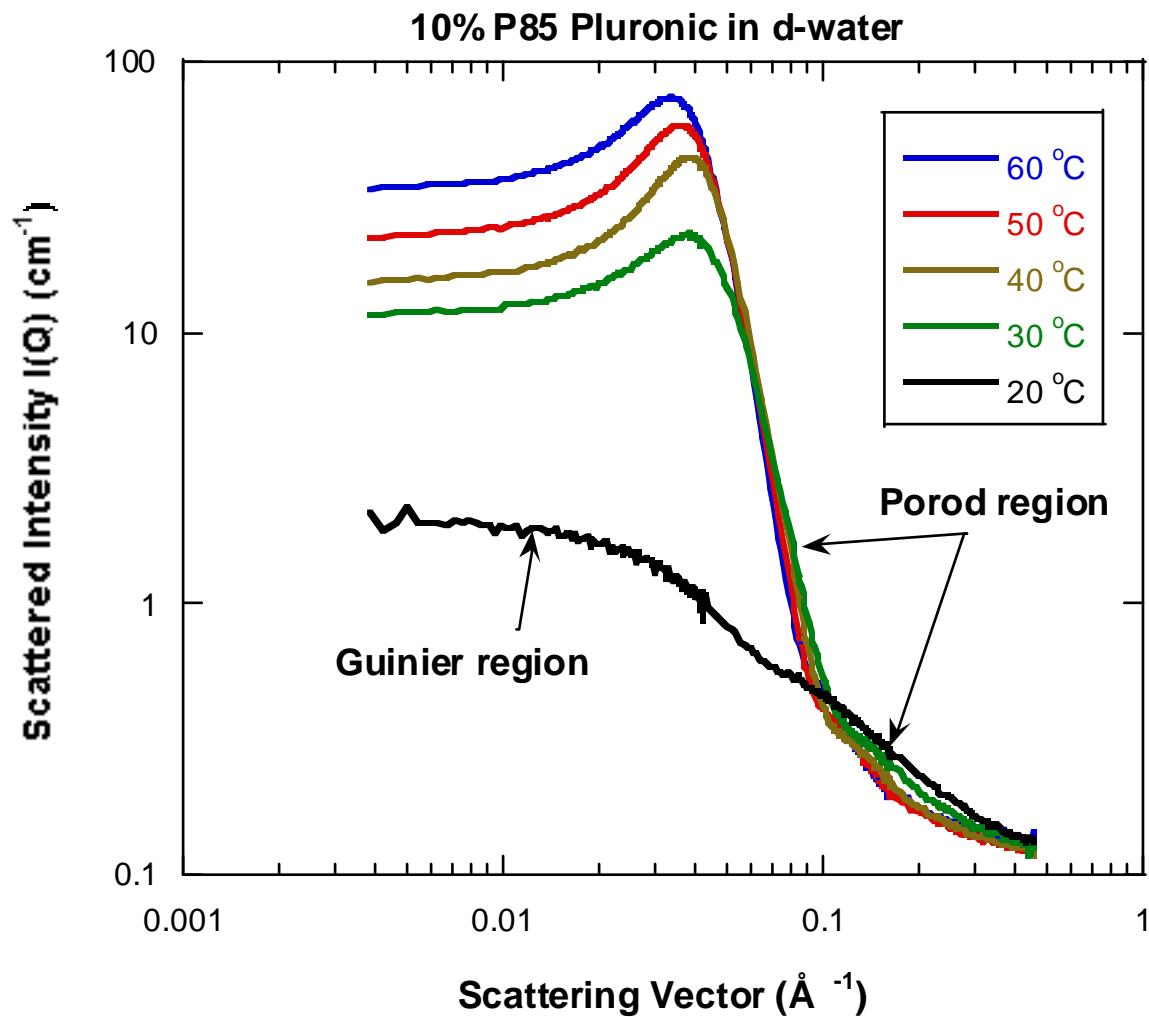
Pluronics are triblock copolymers: PEO-PPO-PEO

PEO: $-\text{CH}_2\text{CH}_2\text{O}-$ is hydrophilic

PPO: $-\text{CH}_2\text{CH}_2(\text{CH}_3)\text{O}-$ is hydrophobic

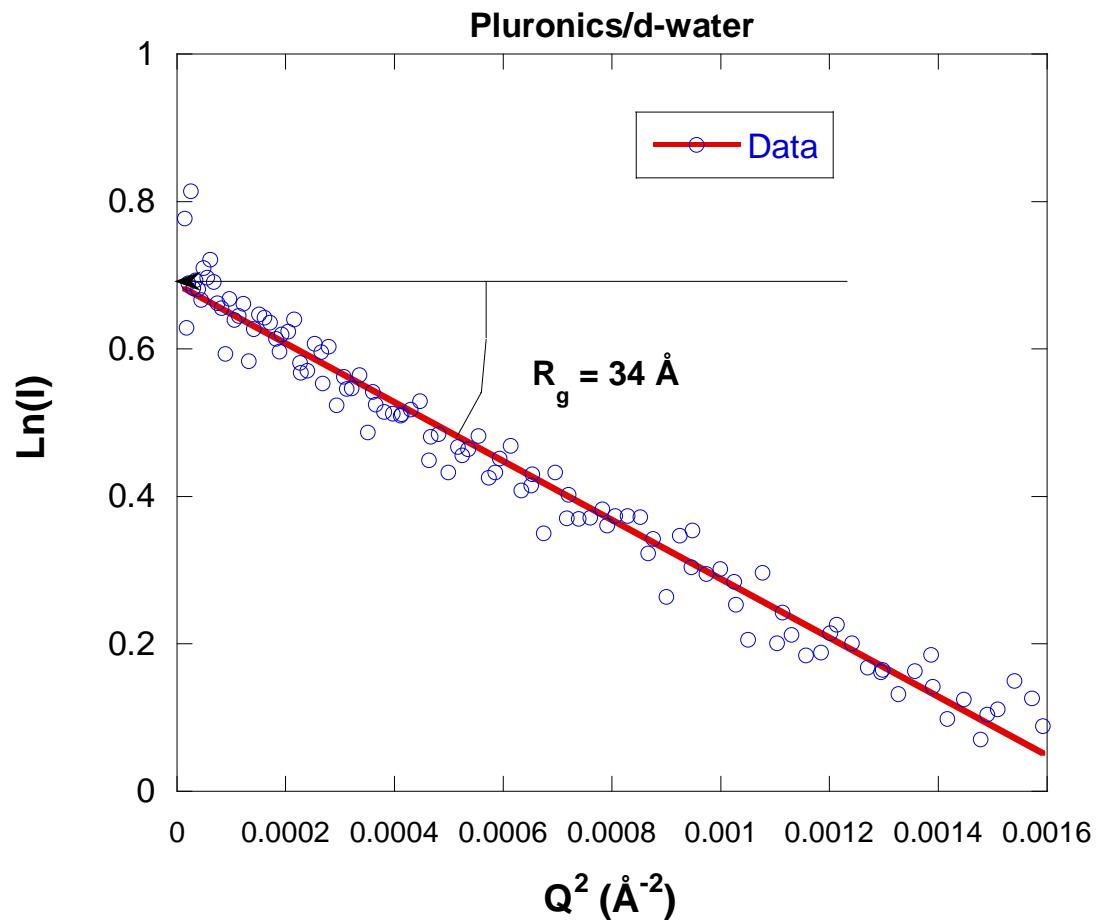
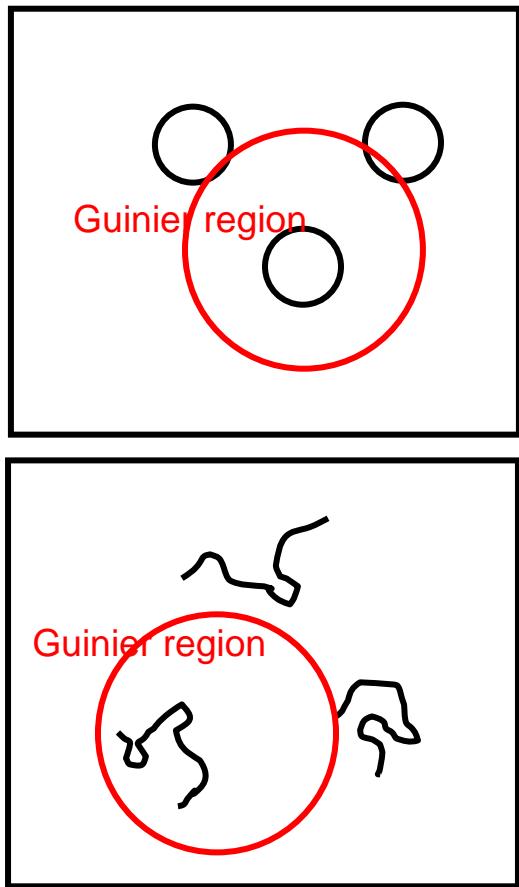


P85 Pluronic forms **micelles** at high temperatures

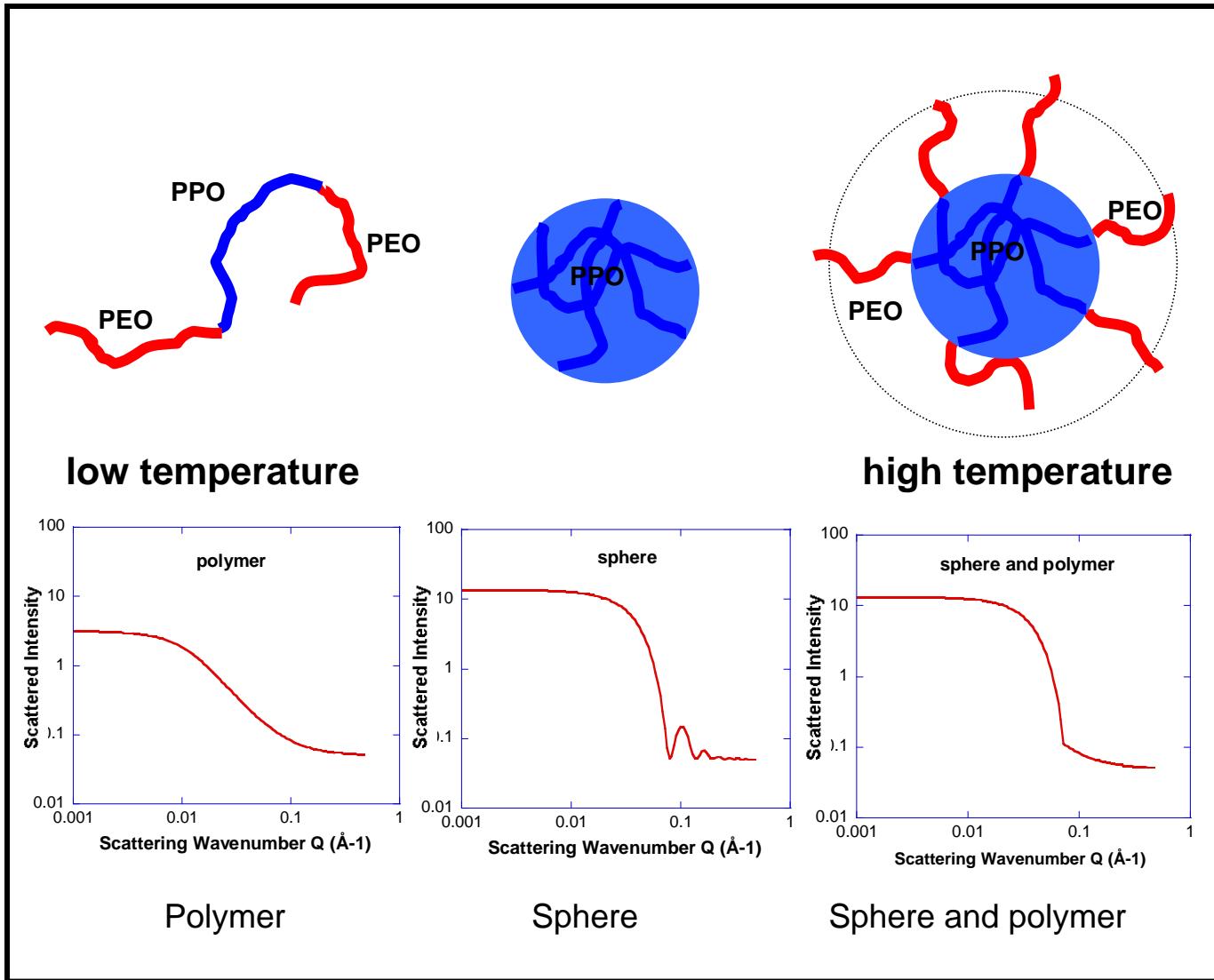


The Guinier Plot

$$I(Q) = I(0) \exp(-Q^2 R_g^2/3)$$



SANS from Pluronics Micelles



Single-Particle Form Factors and Inter-Particle Structure Factors

$$I(Q) = (N_A/V)V_A^2 (b_A/v_A - b_B/v_B)^2 P(Q) S_I(Q)$$

N_A : number of particles, V_A : particle volume, V : sample volume

$(b_A/v_A - b_B/v_B)^2$ = contrast factor

$P(Q)$: single-particle structure factor

$S_I(Q)$: inter-particle structure factor

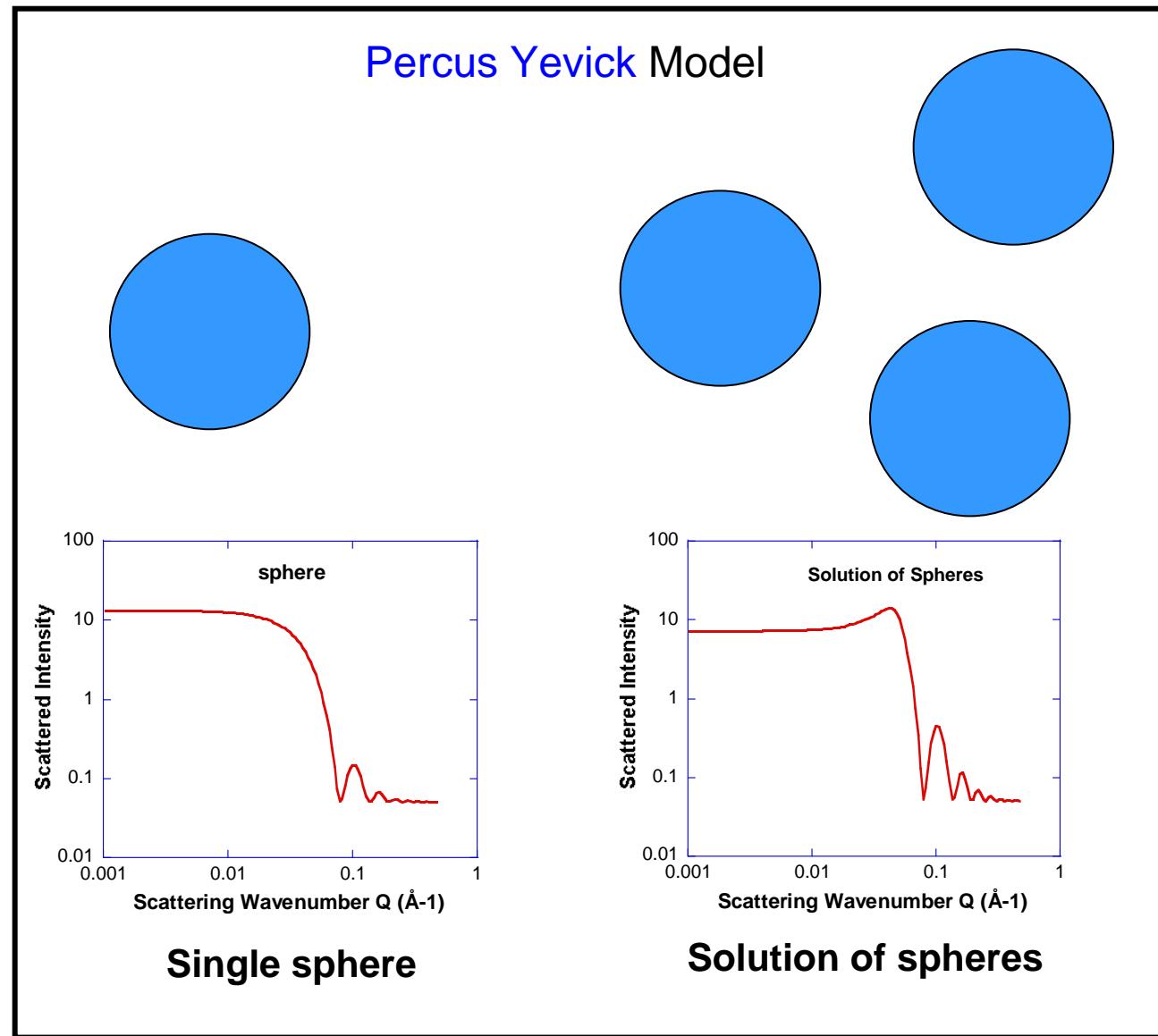
$$P(Q) = \left[\frac{3j_1(QR)}{QR} \right]^2 = \left[\frac{3}{QR} \left(\frac{\sin(QR)}{(QR)^2} - \frac{\cos(QR)}{QR} \right) \right]^2 \text{ for sphere of radius } R.$$

$$P(Q) = \frac{2}{Q^4 R_g^4} \left[\exp(-Q^2 R_g^2) - 1 + Q^2 R_g^2 \right] \text{ for polymer of radius of gyration } R_g.$$

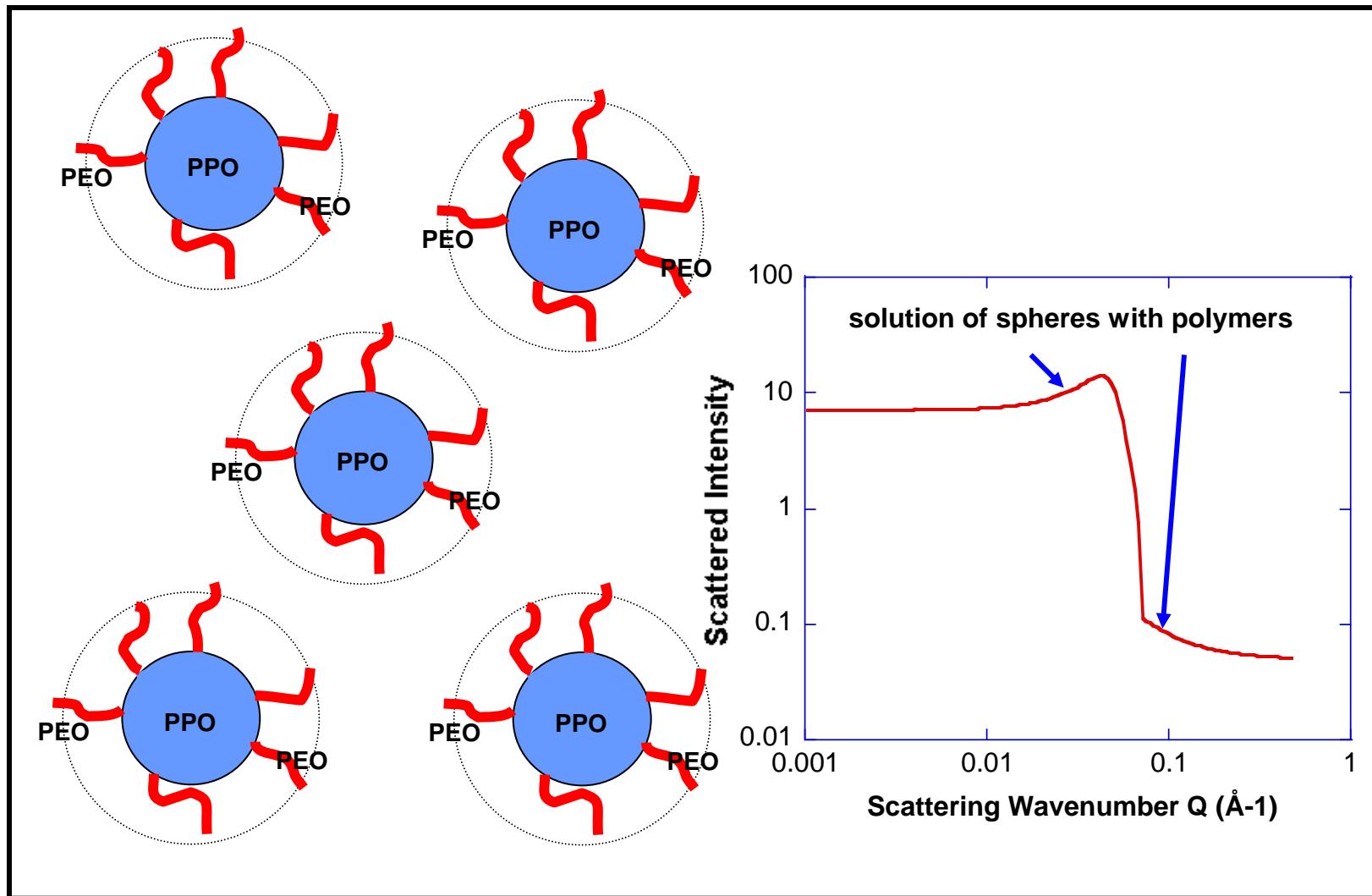
$S_I(Q)$ given by Percus Yevick model for solution of hard spheres.

$S_I(Q)$ given by the Random Phase Approximation model for polymer mixtures.

Solution of Spheres



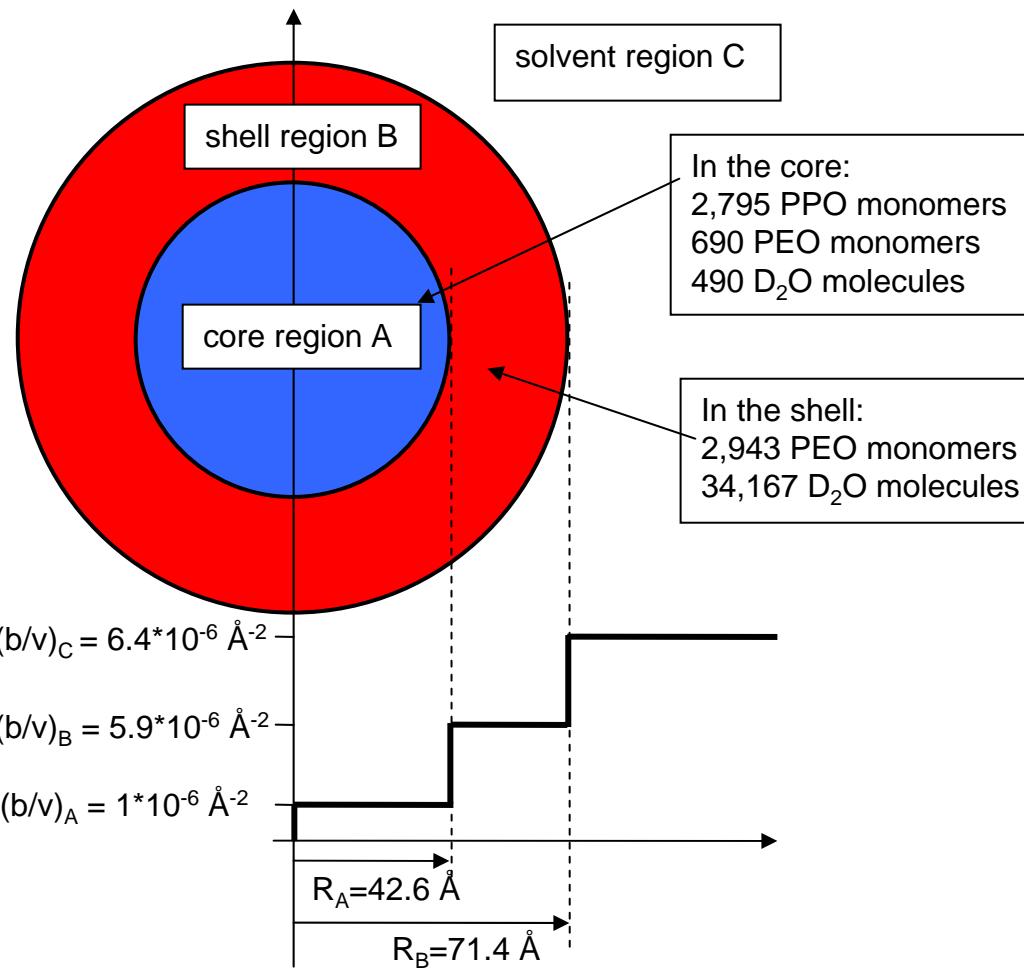
Solution of Spheres with Polymers



Fit SANS Data to a Model of Concentrated Core-Shell Particles

$$\frac{d\Sigma(Q)}{d\Omega} = \frac{N}{V} \left[\left(\frac{b_A}{v_A} - \frac{b_C}{v_C} \right) V_A \frac{3j_1(QR_A)}{QR_A} + \left(\frac{b_B}{v_B} - \frac{b_C}{v_C} \right) \left(V_{A+B} \frac{3j_1(QR_B)}{QR_B} - V_A \frac{3j_1(QR_A)}{QR_A} \right) \right]^2 S_I(Q)$$

10% P85 Pluronic/D₂O, 40 °C



2. POLYMER BLENDS THERMODYNAMICS

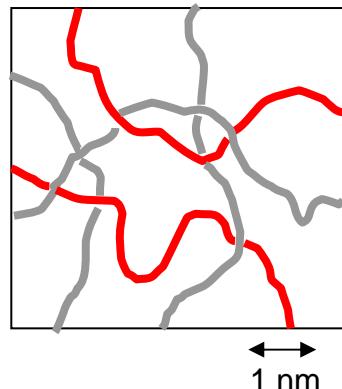
SANS Intensity: $I(Q) = \frac{d\Sigma(Q)}{d\Omega} = \left(\frac{b_1}{v_1} - \frac{b_2}{v_2} \right)^2 S(Q)$

Thermodynamics: $S^{-1}(Q=0) = \frac{1}{k_B T} \frac{\partial^2 G}{d\phi_1^2}$ **Gibbs Free Energy**

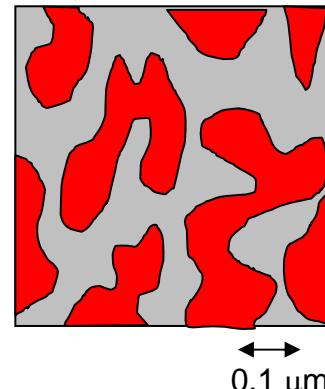
The Random Phase Approximation:

$$S^{-1}(Q) = \frac{1}{n_1 \phi_1 v_1 P_1(Q)} + \frac{1}{n_2 \phi_2 v_2 P_2(Q)} - 2 \frac{\chi_{12}(T)}{v_0}$$

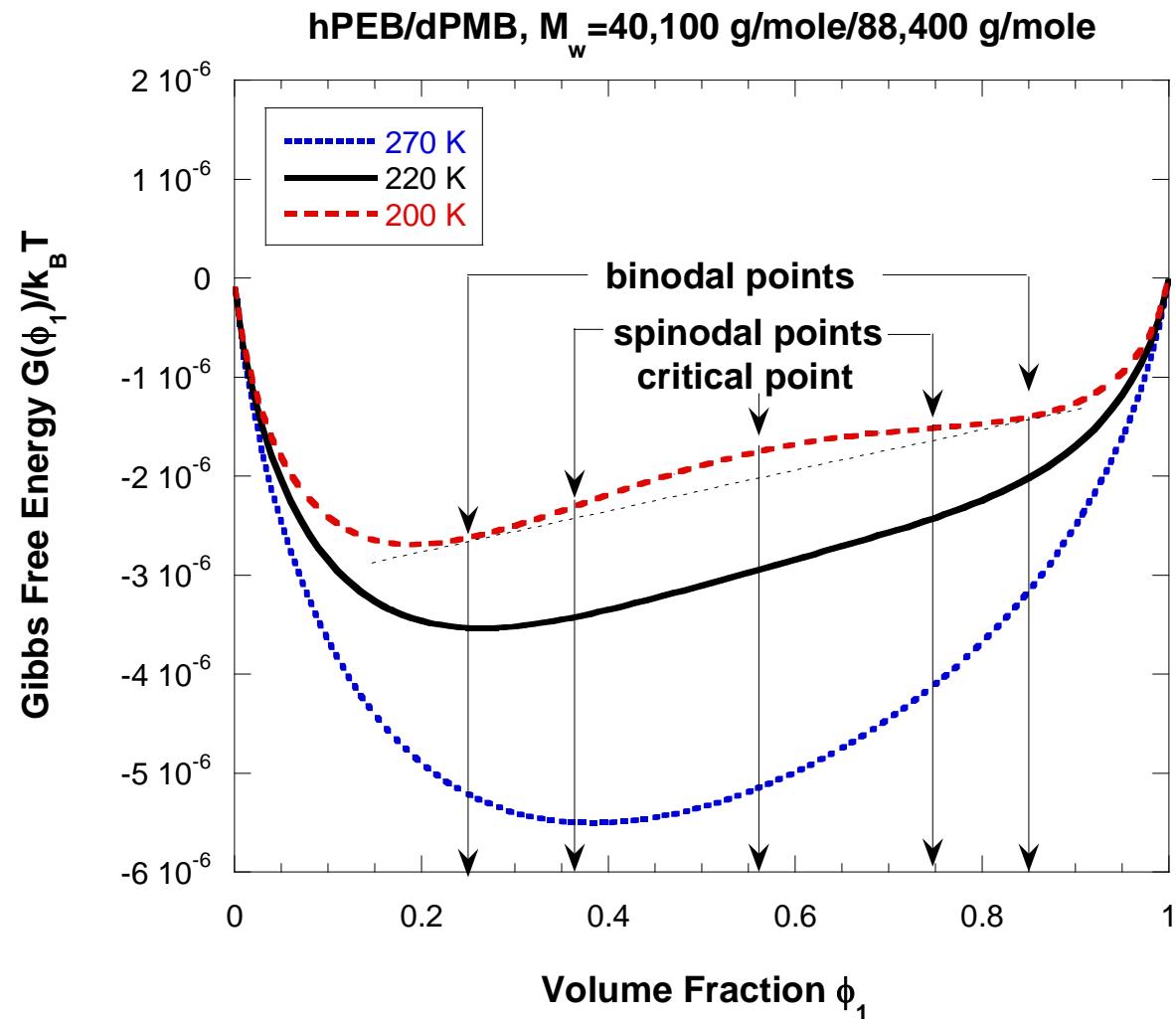
Mixed polymer blend



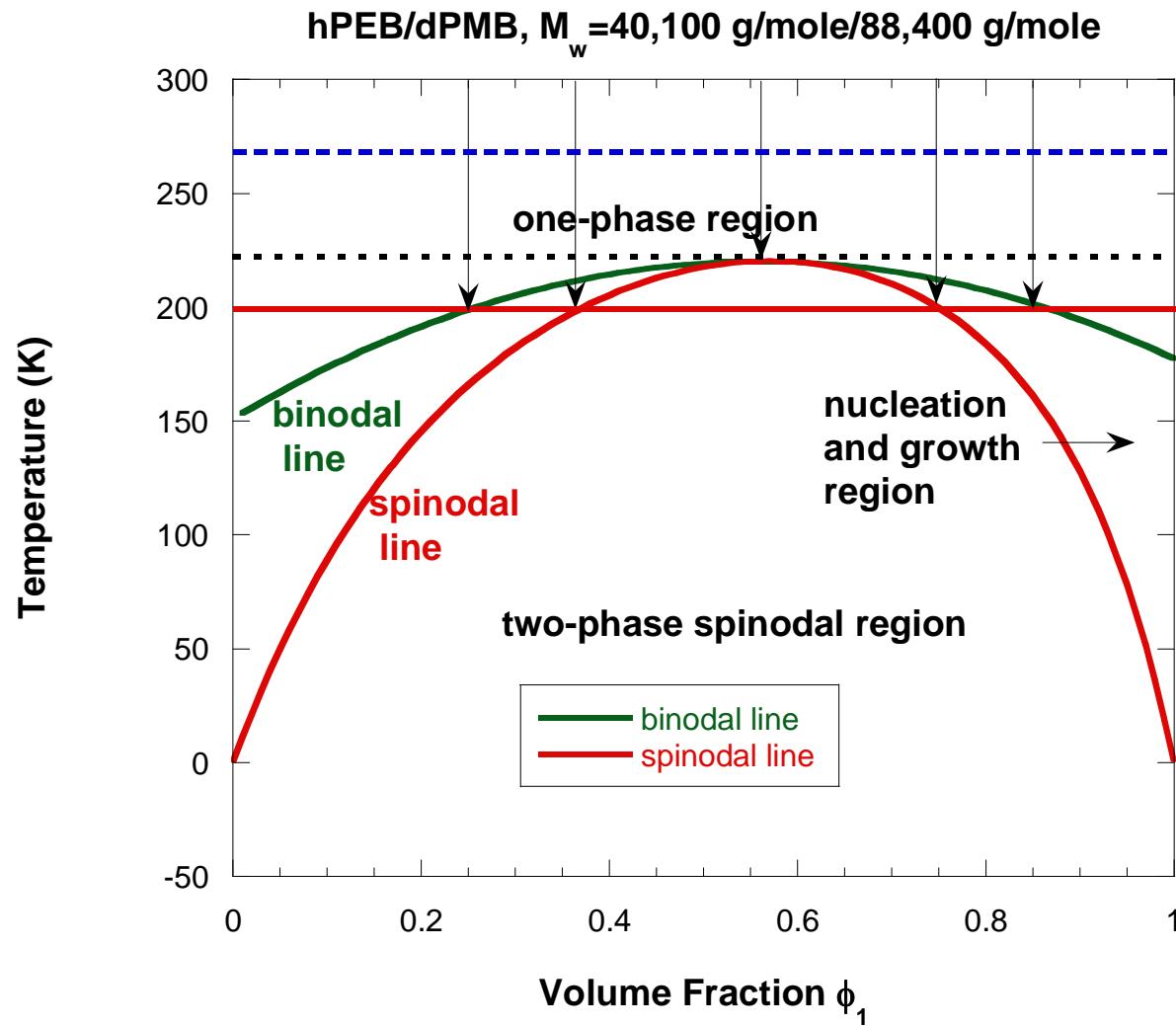
Phase separated blend



Gibbs Free Energy



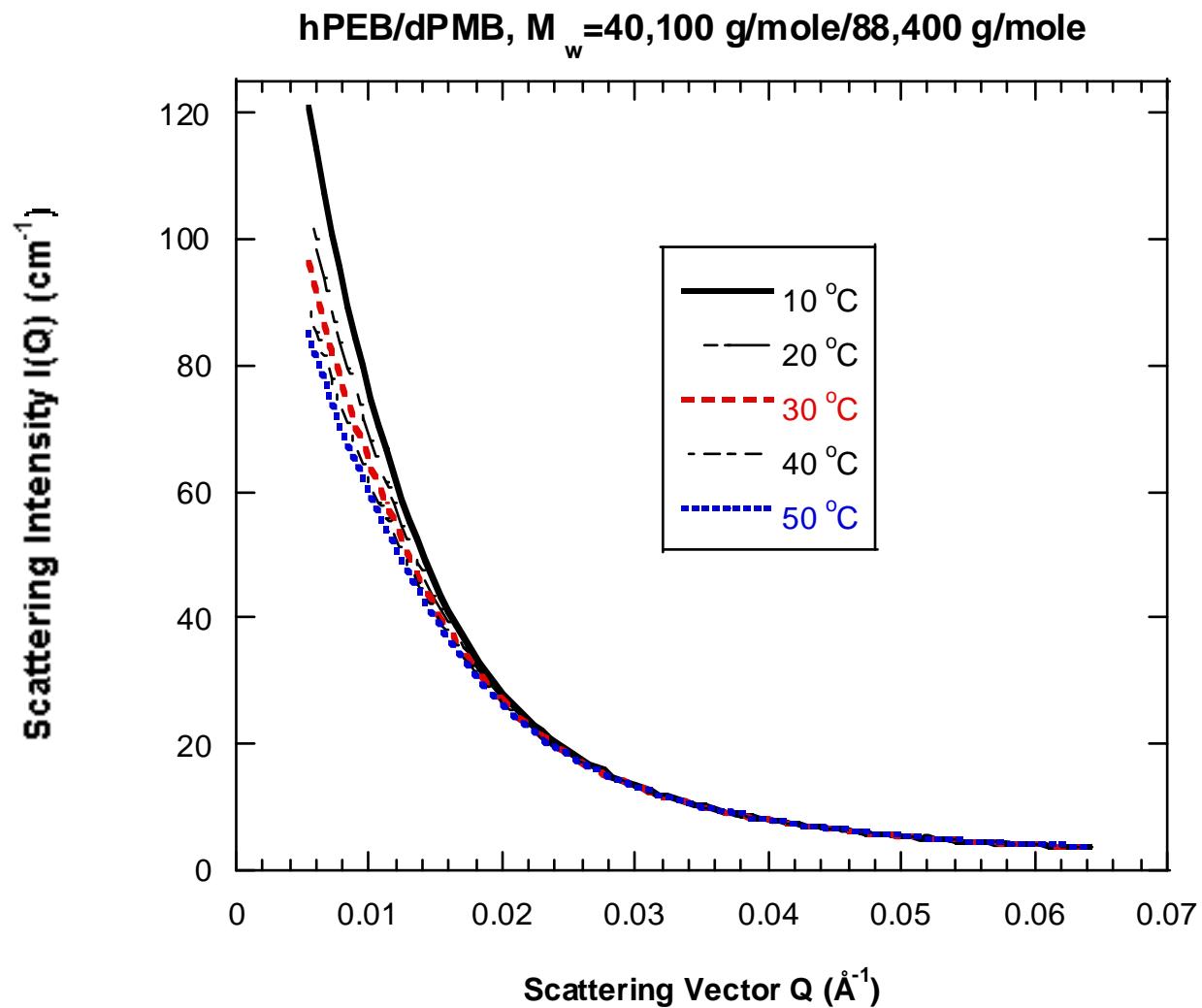
Phase Diagram



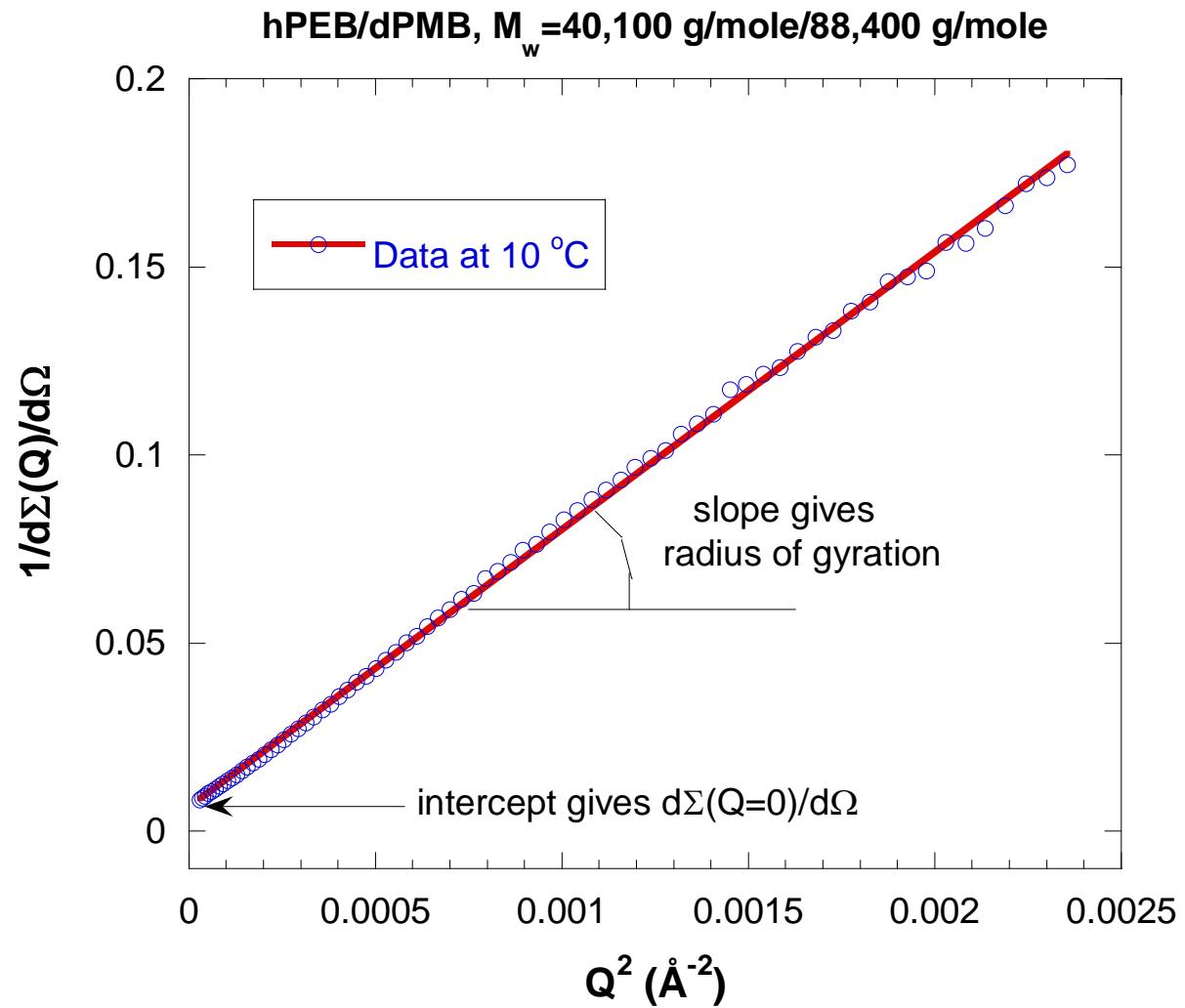
SANS From Polymer Blend Mixtures

Polymers:	Polyethylbutylene / Polymethylbutylene	
	hPEB -(C ₆ H ₁₂)- / dPMB -(C ₅ H ₅ D ₅)-	
Molecular Weights:	M _w =44,100 g/mole	M _w =88,400 g/mole
Volume Fractions:	φ _{hPEB} =0.57	φ _{dPMB} =0.43

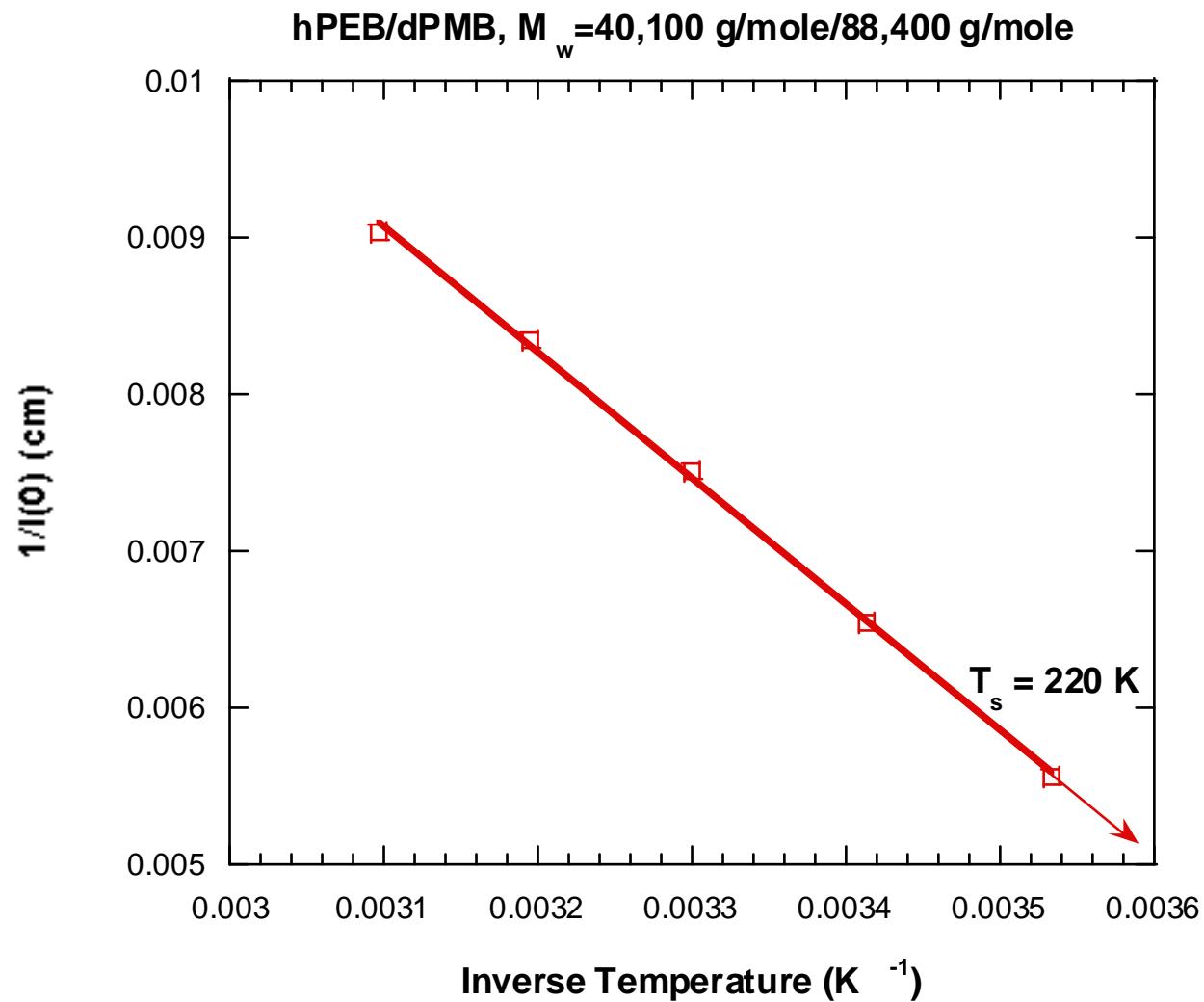
SANS Data



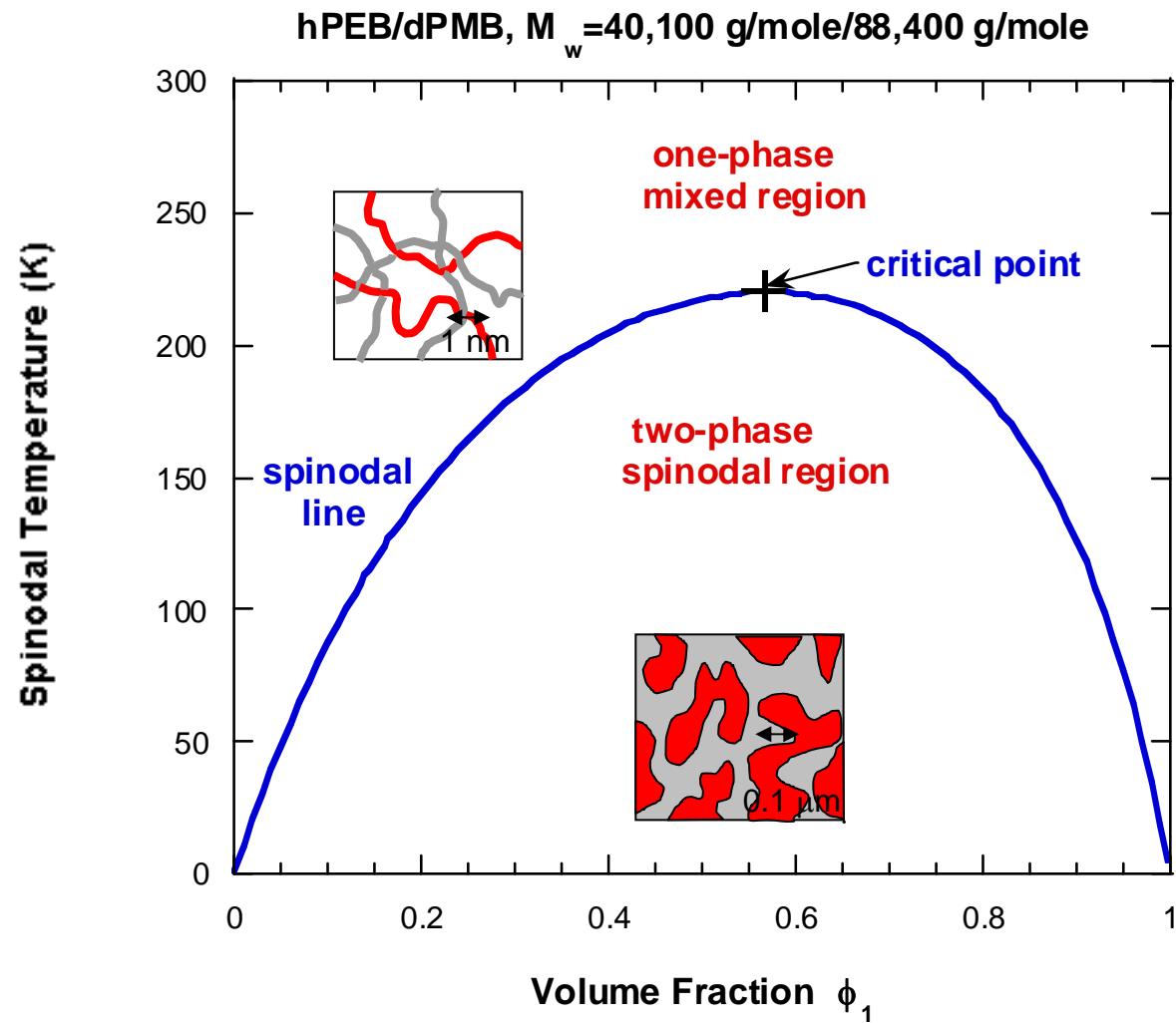
The Zimm Plot



Spinodal Temperature



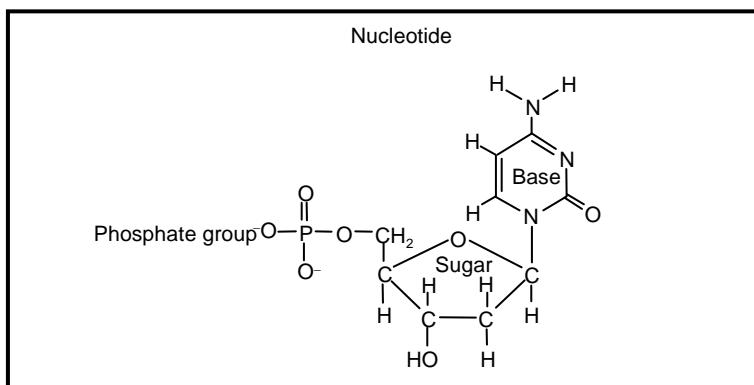
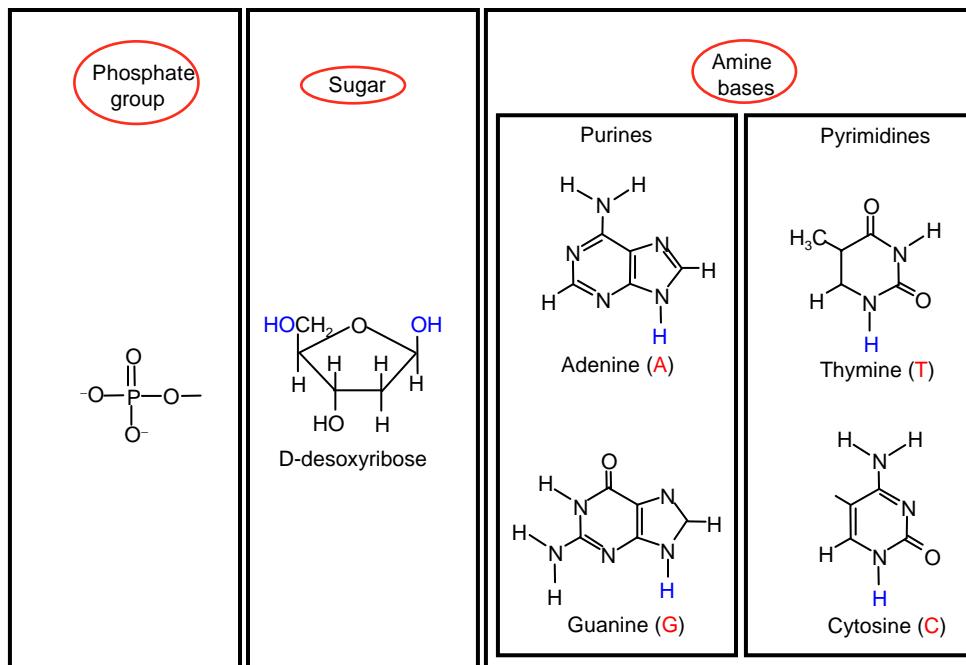
Spinodal Temperature



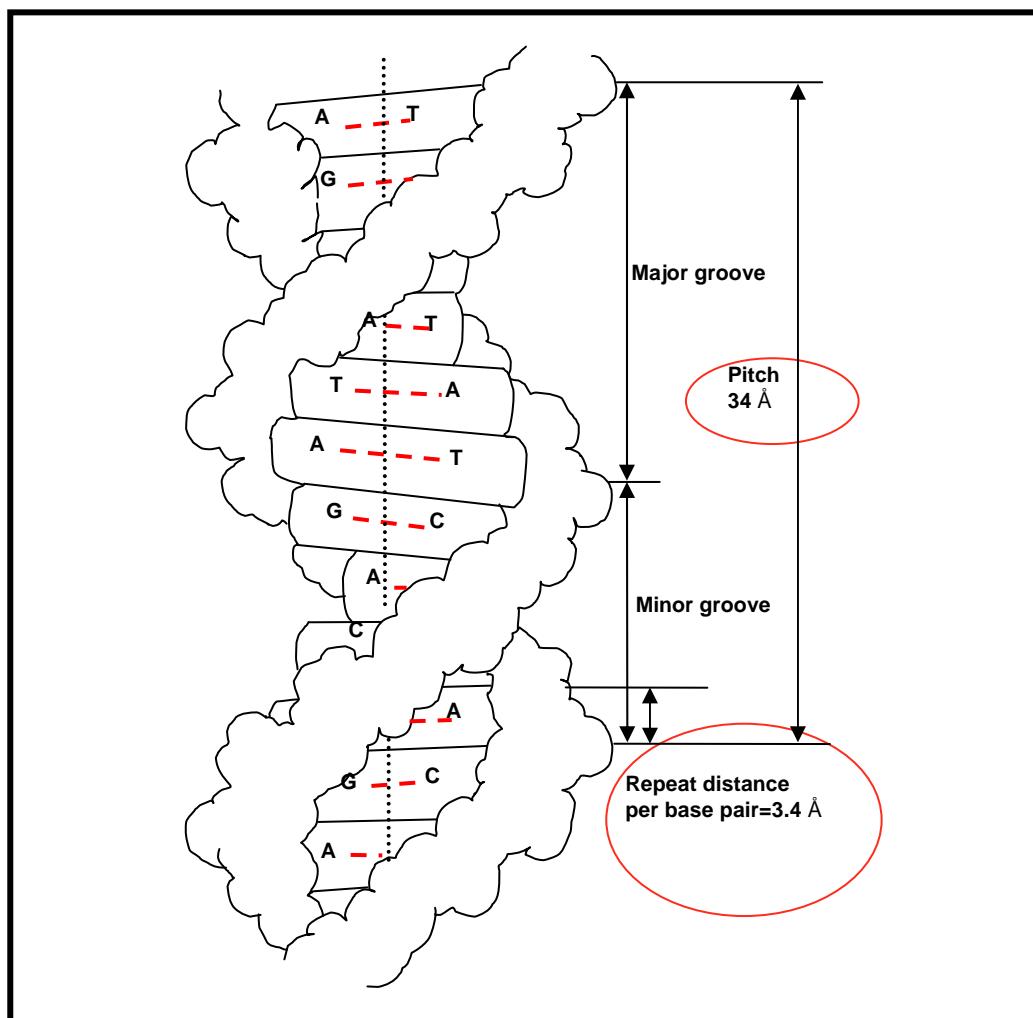
3. HELIX-TO-COIL TRANSITION IN DNA

DNA is the basic building block for life. It encodes for the synthesis of proteins.

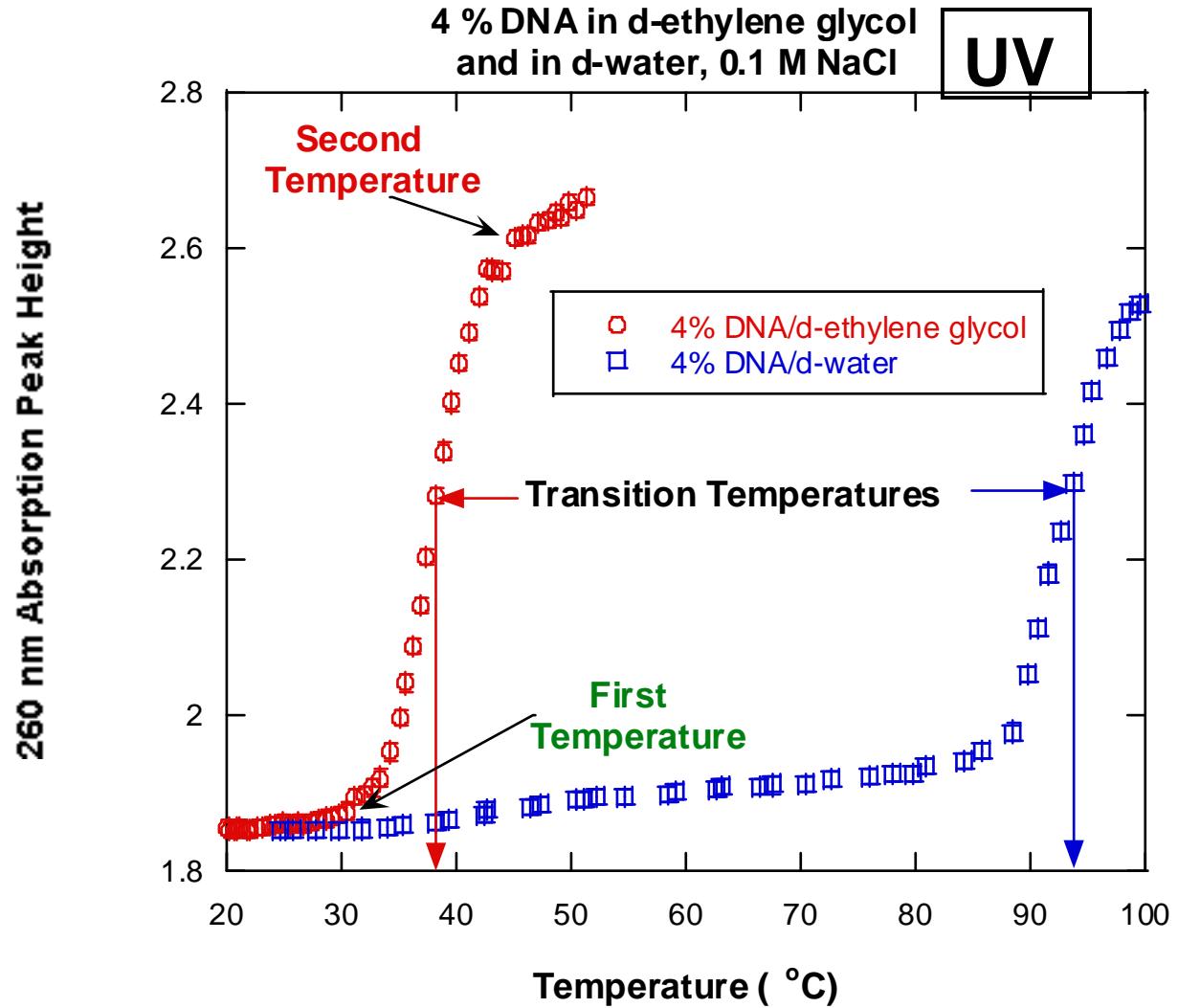
THE DNA MOLECULE



The DNA Helix



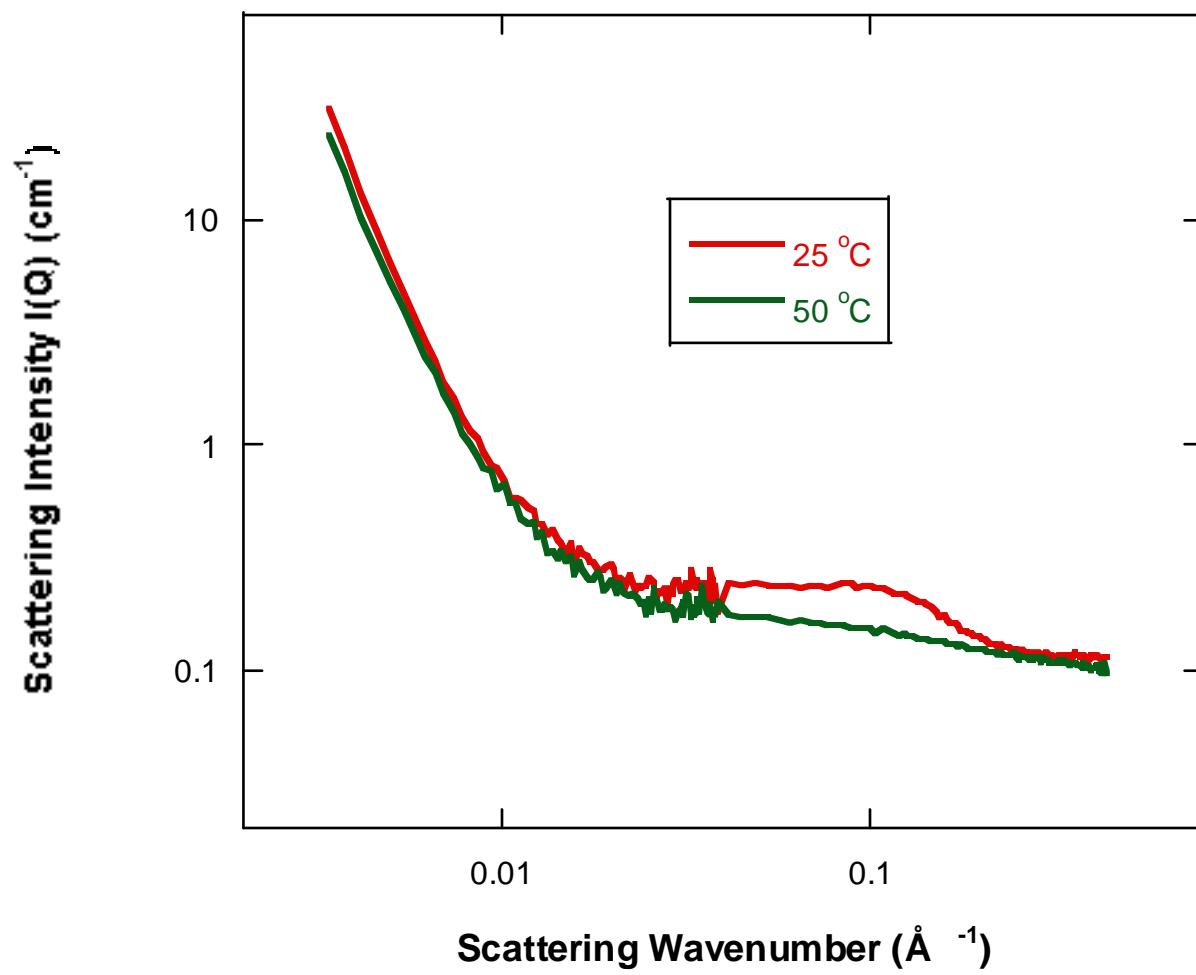
Helix-to-Coil Transition in DNA



SANS Data

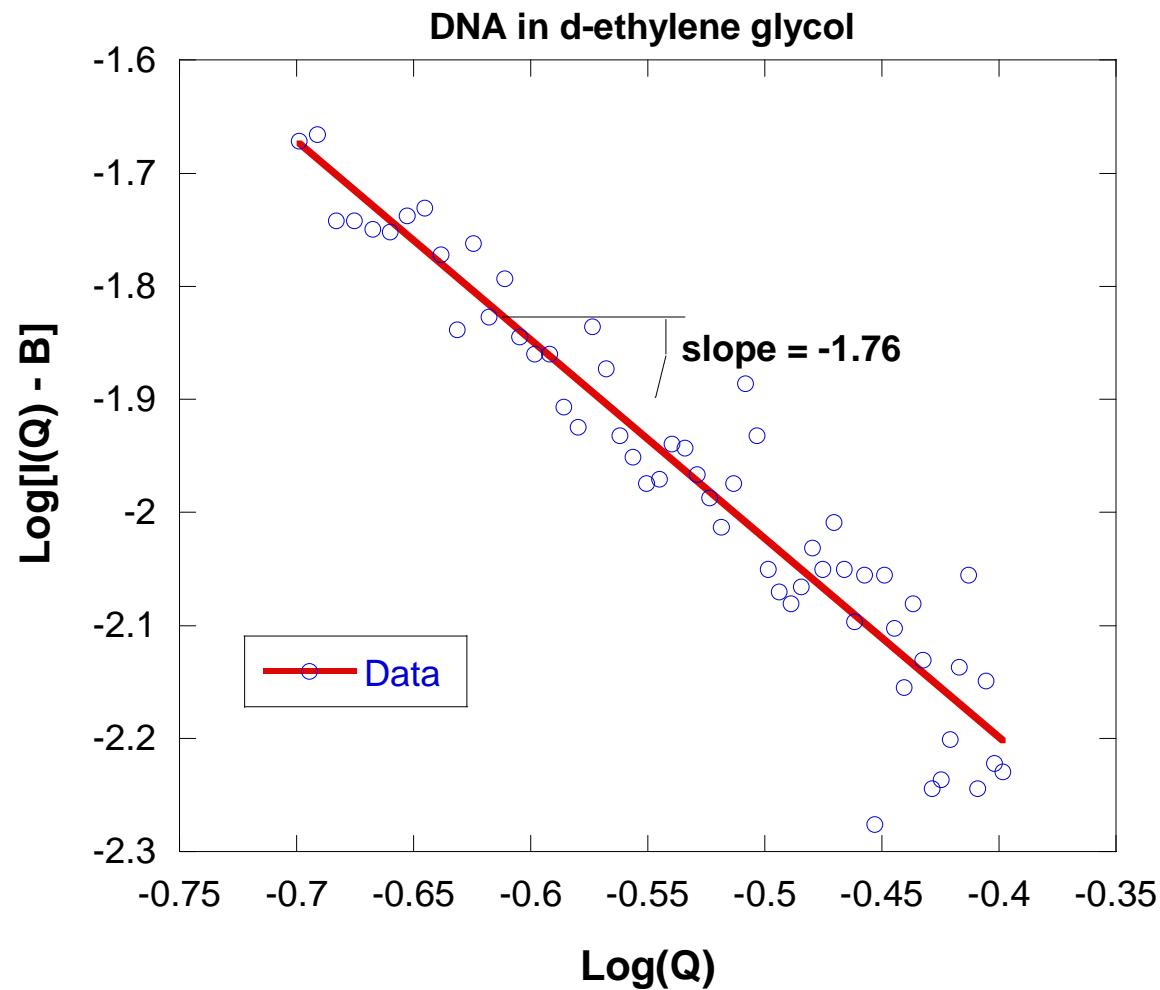
4% DNA in d-ethylene glycol,
0.1M NaCl

SANS



The Porod Plot

$$I(Q) \sim C/Q^m$$



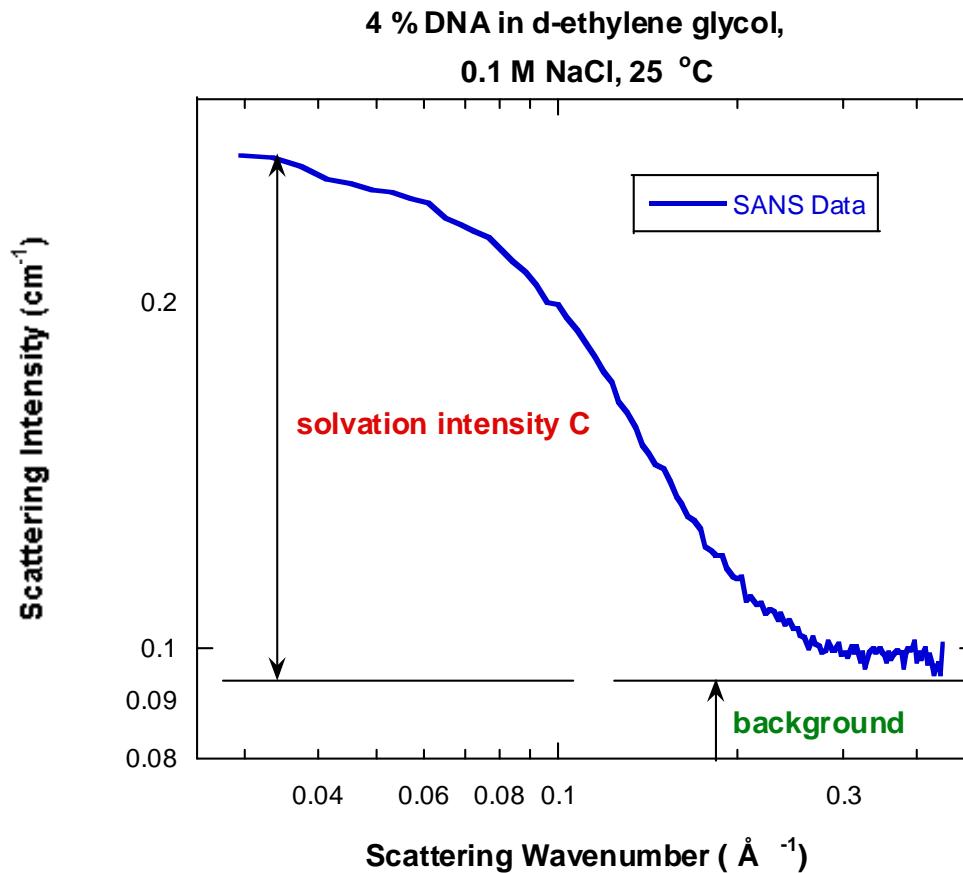
Nonlinear Least-Squares Fit

Functional form: $I(Q) = \text{C}/[1+(QL)^m] + \text{Background}$

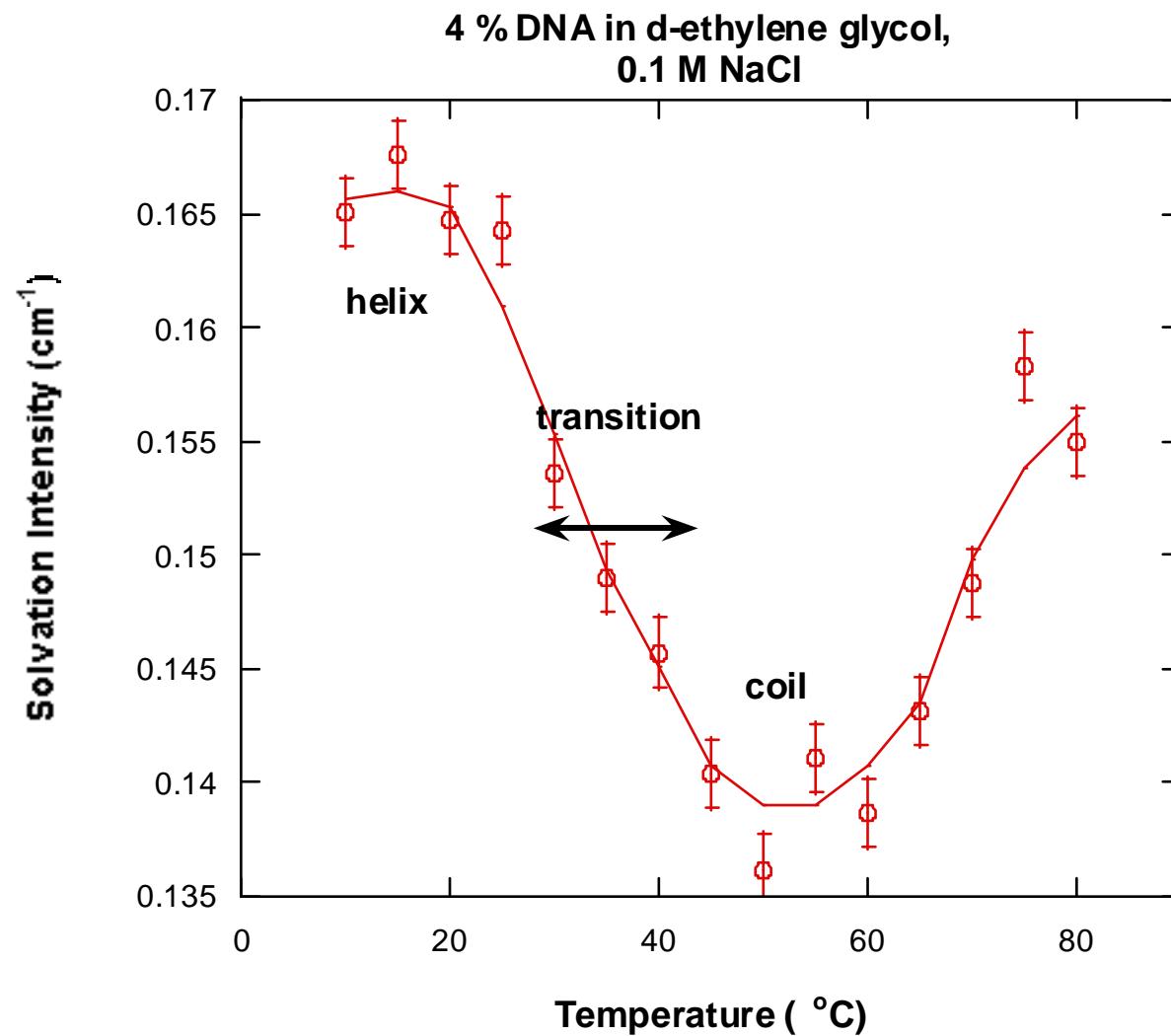
C: solvation intensity

L: correlation length

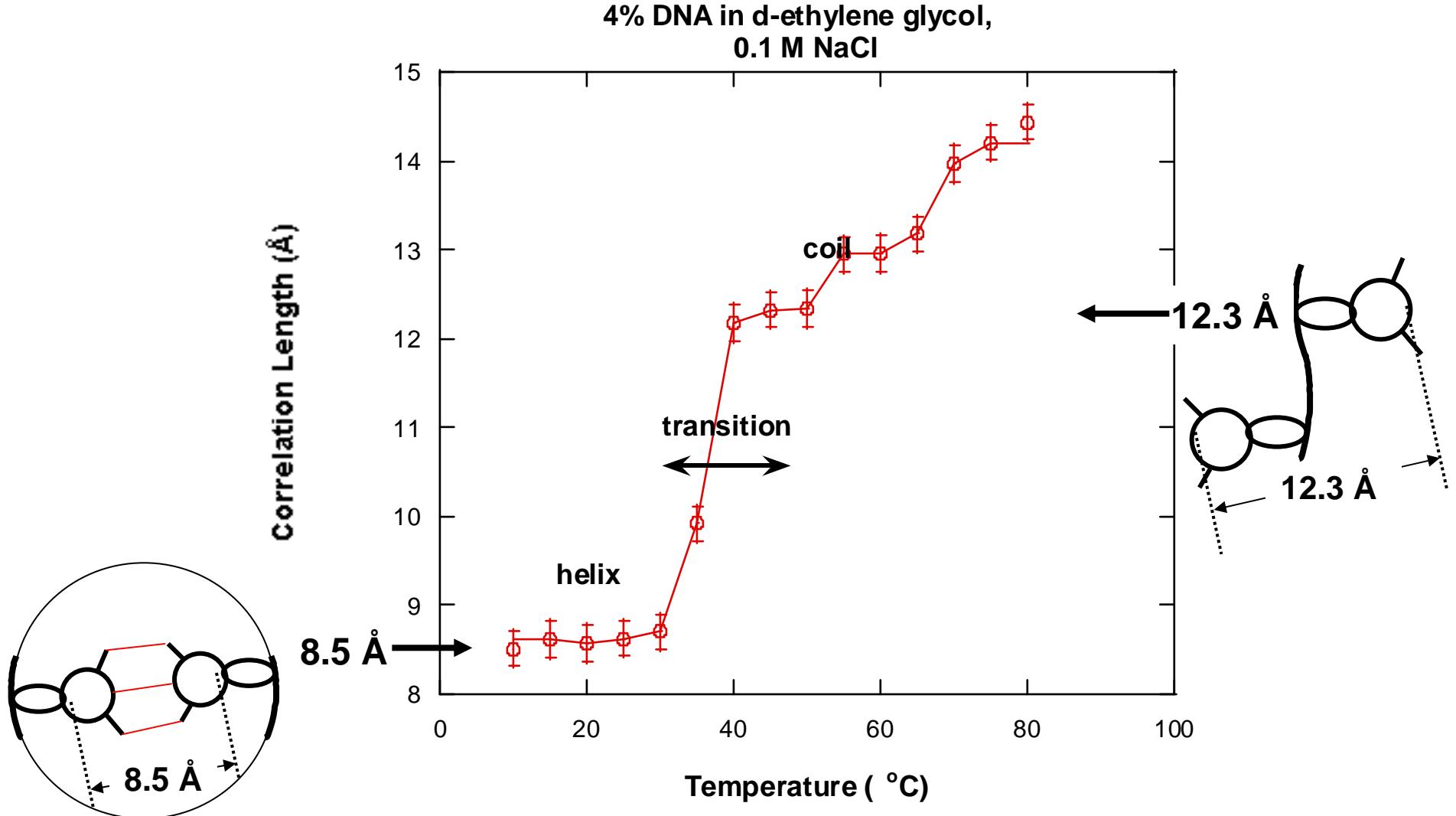
m: Porod exponent



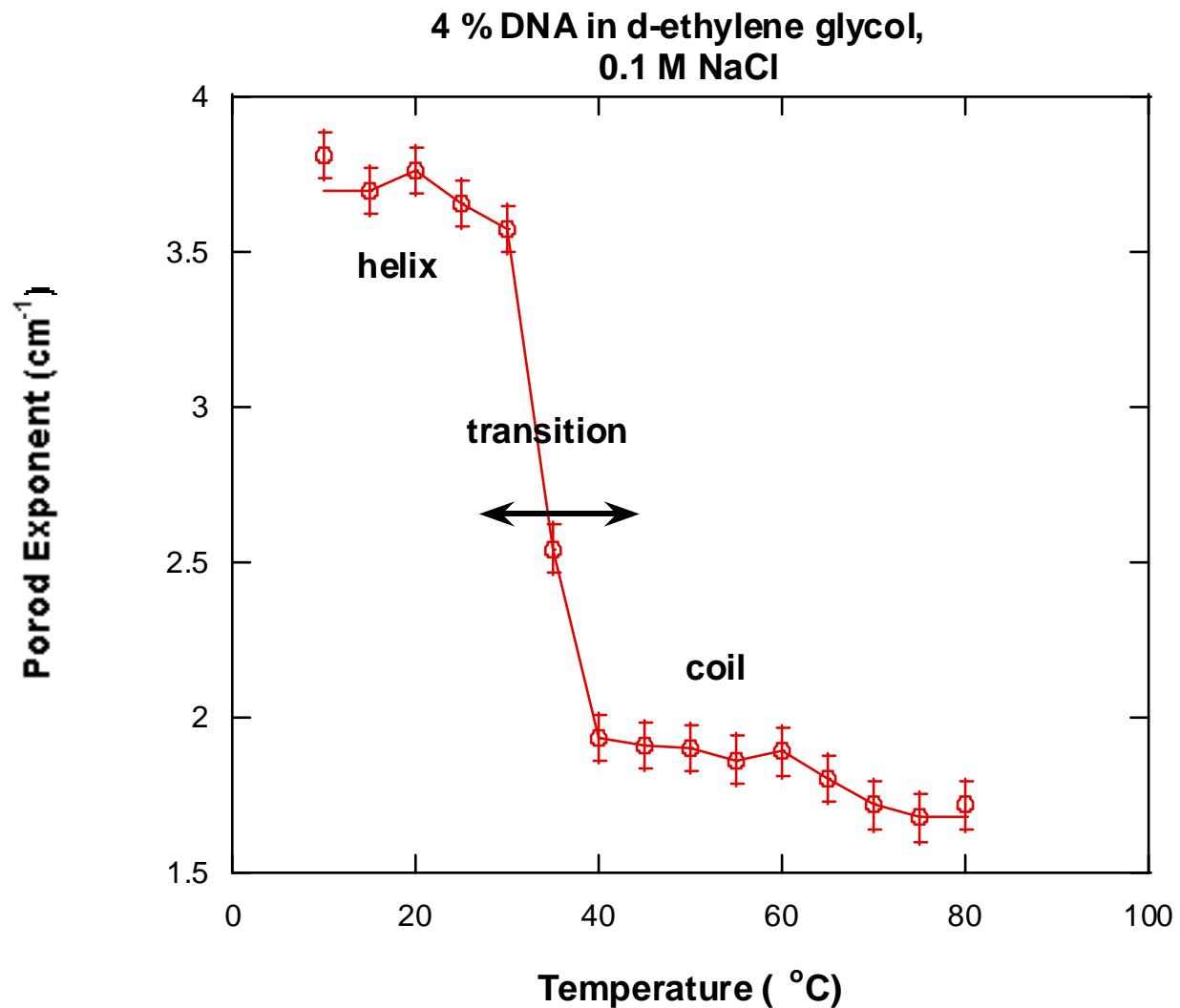
The Solvation Intensity



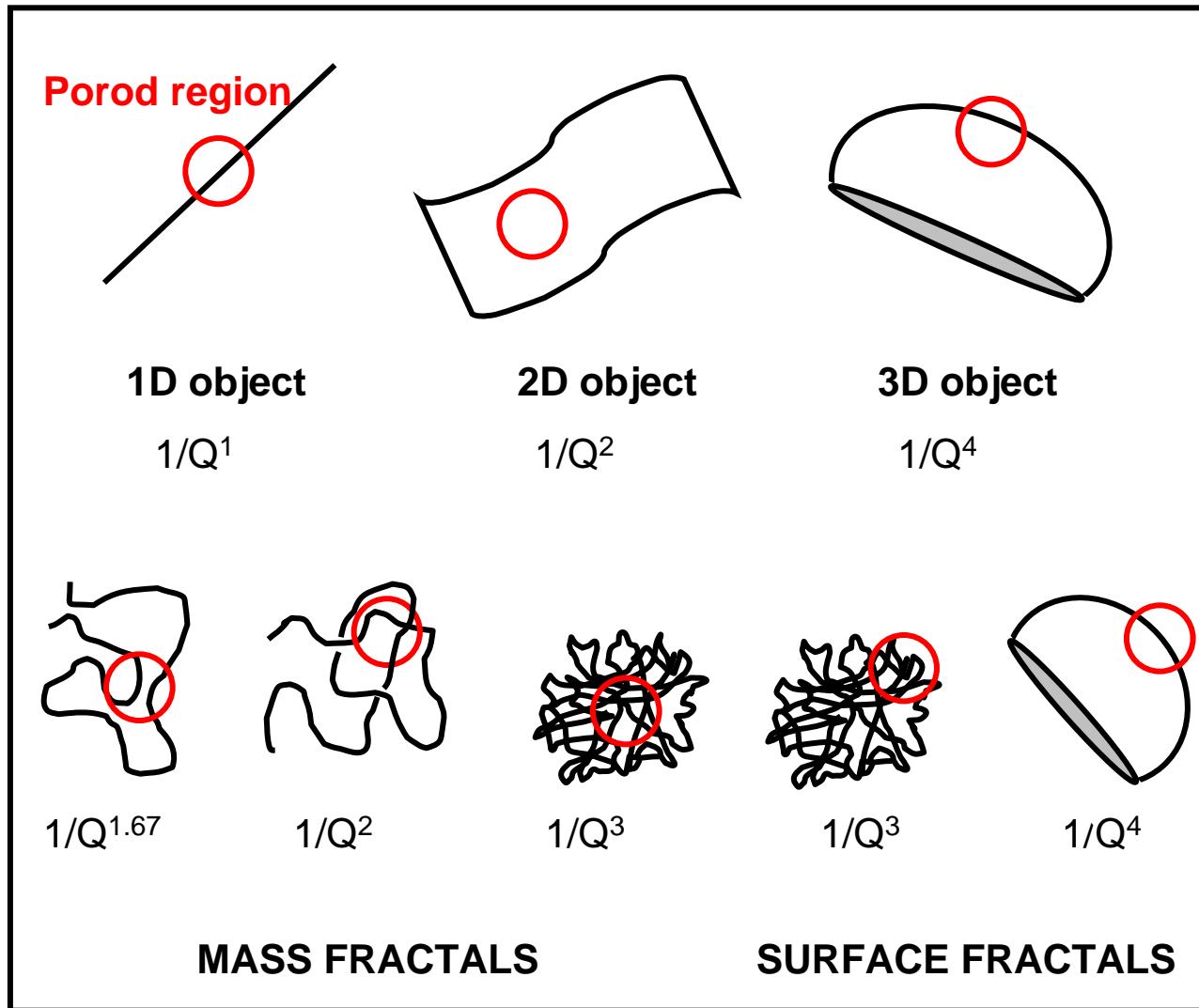
The Correlation Length



The Porod Exponent



POROD EXPONENTS



CONCLUSIONS

- The SANS technique is a valuable characterization method.
- SANS has been effective in **complex fluids, polymers, biology**, etc.
- SANS can determine **structures, phase transitions, and morphology**.
- The NIST SANS gets over 200 users per year, resulting in over 70 publications per year.

ACKNOWLEDGMENTS

NSF-DMR, Steve Kline, Nitash Balsara, David Worcester.

PROBING NANOSCALE STRUCTURES –THE SANS TOOLBOX

http://www.ncnr.nist.gov/staff/hammouda/the_SANS_toolbox.pdf