SCANDIUM1

(Data in kilograms of scandium oxide content, unless otherwise noted)

<u>Domestic Production and Use</u>: Demand for scandium increased slightly in 2003. Although scandium was not mined domestically in 2003, quantities sufficient to meet demand were available in domestic tailings. Principal sources were imports from China, Russia, and Ukraine. Domestic companies with scandium processing capabilities were located in Mead, CO; Urbana, IL; and Knoxville, TN. Capacity to produce ingot and distilled scandium metal was located in Phoenix, AZ; Urbana, IL; and Ames, IA. Scandium used in the United States was essentially derived from foreign sources. Principal uses for scandium in 2003 were aluminum alloys for sporting equipment, metallurgical research, high-intensity metal halide lamps, analytical standards, electronics, and laser research.

Salient Statistics—United States:	<u> 1999</u>	<u>2000</u>	<u>2001</u>	<u>2002</u>	<u>2003^e</u>
Price, yearend, dollars:					
Per kilogram, oxide, 99.0% purity	900	700	700	700	500
Per kilogram, oxide, 99.9% purity	2,000	2,000	2,300	2,000	1,300
Per kilogram, oxide, 99.99% purity	3,000	3,000	2,700	2,500	2,500
Per kilogram, oxide, 99.999% purity	4,000	6,000	4,100	3,200	3,200
Per gram, dendritic, metal ²	270.00	270.00	279.00	178.00	185.00
Per gram, metal, ingot ³	175.00	175.00	198.00	198.00	119.00
Per gram, scandium bromide, 99.99% purity ⁴	91.80	91.80	94.60	94.60	98.40
Per gram, scandium chloride, 99.9% purity ⁴	39.60	39.60	40.80	40.80	42.40
Per gram, scandium fluoride, 99.9% purity ⁴	80.10	80.10	173.00	173.00	180.00
Per gram, scandium iodide, 99.999% purity ⁴	151.00	151.00	156.00	156.00	162.00
Net import reliance ⁵ as a percentage of					
apparent consumption	100	100	100	100	100

Recycling: None.

Import Sources (1999-2002): Not available.

<u>Tariff</u> : Item	Number	Normal Trade Relations 12/31/03	
Mineral substances not elsewhere specified or included:		<u></u>	
Including scandium ores Rare-earth metals, scandium and yttrium,	2530.90.8050	Free.	
whether or not intermixed or interalloyed			
including scandium Mixtures of rare-earth oxides except cerium	2805.30.0000	5.0% ad val.	
oxide, including scandium oxide mixtures	2846.90.2010	Free.	
Rare-earth compounds, including individual rare-earth oxides, hydroxides, nitrates, and other individual compounds,			
including scandium oxide Aluminum alloys, other:	2846.90.8000	3.7% ad val.	
Including scandium-aluminum	7601.20.9090	Free.	

Depletion Allowance: 14% (Domestic and foreign).

Government Stockpile: None.

Events, Trends, and Issues: Nominal prices for domestically produced scandium compounds decreased from the previous year. The supply of domestic and foreign scandium remained strong despite increased demand. Although demand increased in 2003, the total market remained very small. Domestic increases in demand were primarily from recently developed applications in welding wire, scandium-aluminum baseball and softball bats, scandium-aluminum bicycle frames, and high-strength, lightweight handgun frames and cylinders. New demand is expected in future fuel-cell markets and aerospace applications.

SCANDIUM

Scandium's use continued to increase in metal halide lighting. Scandium additions, as the metal or the iodide, mixed with other elements, were added to halide light bulbs to adjust the color to simulate natural sunlight. Demand also continued to increase for scandium-aluminum alloys. Future development of alloys for aerospace and specialty markets, including sports equipment, is expected. Scandium's availability from the Kazakhstan, Russia, and Ukraine increased substantially in 1992, after export controls were relaxed, and sales to the Western World have been increasing. China also continued to supply goods to the U.S. market.

World Mine Production, Reserves, and Reserve Base: Scandium was produced as a byproduct material in China, Kazakhstan, Russia, and Ukraine. Foreign mine production data were not available. No scandium was mined in the United States in 2003. Scandium occurs in many ores in trace amounts, but has not been found in sufficient quantities to be considered as a reserve or reserve base. As a result of its low concentration, scandium has been produced exclusively as a byproduct during processing of various ores or recovered from previously processed tailings or residues.

World Resources: Resources of scandium are abundant, especially when considered in relation to actual and potential demand. Scandium is rarely concentrated in nature due to its lack of affinity to combine with the common ore forming anions. It is widely dispersed in the lithosphere and forms solid solutions in over 100 minerals. In the Earth's crust, scandium is primarily a trace constituent of ferromagnesium minerals. Concentrations in these minerals (amphibole-hornblende, biotite, and pyroxene) typically range from 5 to 100 parts per million equivalent Sc₂O₃. Ferromagnesium minerals commonly occur in the igneous rocks, basalt and gabbro. Enrichment of scandium also occurs in aluminum phosphate minerals, beryl, cassiterite, columbite, garnet, muscovite, rare-earth minerals, and wolframite. Recent domestic production has primarily been from the scandium-yttrium silicate mineral, thortveitite, and from byproduct leach solutions from uranium operations. One of the principal domestic scandium resources is the fluorite tailings from the mined out Crystal Mountain deposit near Darby, MT. Tailings from the mined-out fluorite operations, which were generated from 1952 to 1971, contain thortveitite and associated scandium-enriched minerals. Resources are also contained in the tantalum residues previously processed at Muskogee, OK. Smaller resources are associated with molybdenum, titanium-tungsten, and tungsten minerals from the Climax molybdenum deposit in Colorado and in crandallite, kolbeckite, and varisite at Fairfield, UT. Other lower grade domestic resources are present in ores of aluminum, cobalt, iron, molybdenum, nickel, phosphate, tantalum, tin, titanium, tungsten, zinc, and zirconium. Process residues from tungsten operations in the United States also contain significant amounts of scandium.

Foreign resources are known in Australia, China, Kazakhstan, Madagascar, Norway, Russia, and Ukraine. Resources in Australia are contained in nickel and cobalt deposits in Syerston and Lake Innes, New South Wales. China's resources are in iron, tin, and tungsten deposits in Fujian, Guangdong, Guangxi, Jiangxi, and Zhejian Provinces. Resources in Russia and Kazakhstan are in the Kola Peninsula apatites and in uranium-bearing deposits, respectively. Scandium in Madagascar is contained in pegmatites in the Befanomo area. Resources in Norway are dispersed in the thortveitite-rich pegmatites of the Iveland-Evje Region and a deposit in the northern area of Finnmark. In Ukraine, scandium is recovered as a byproduct of iron ore processing at Zheltye Voda. An occurrence of the mineral thortveitite is reported from Kobe, Japan. Undiscovered scandium resources are thought to be very large.

<u>Substitutes</u>: In applications, such as lighting and lasers, scandium is generally not subject to substitution. In metallurgical applications, titanium and aluminum high-strength alloys and carbon fiber may substitute in sporting goods, especially bicycle frames.

eEstimated.

¹See also Rare Earths.

²Scandium pieces, 99.9% purity, distilled dendritic, 1999-2003 prices converted from 0.5-gram price, from Alfa Aesar, a Johnson Matthey company. ³Scandium, metal lump, sublimed dendritic 99.99% purity, from Alfa Aesar, a Johnson Matthey company, 1999 and 2000. Metal ingot pieces 99.9% purity 2001-3.

⁴Bromide, chloride, and fluoride in crystalline or crystalline aggregate form and scandium iodide as ultradry powder from Alfa Aesar, a Johnson Matthey company.

⁵Defined as imports – exports + adjustments for Government and industry stock changes.

⁶See Appendix C for definitions.