

UNITED STATES DEPARTMENT OF COMMERCE

PETER G. PETERSON, *Secretary*

NATIONAL BUREAU OF STANDARDS • LAWRENCE M. KUSHNER, *Acting Director*

Tables of
Molecular Vibrational Frequencies
Consolidated Volume I

Takehiko Shimanouchi

Department of Chemistry
Faculty of Science
University of Tokyo
Tokyo, Japan



NSRDS-NBS 39

Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.), 39, 164 pages (June 1972)
NSRDAP

© 1972 by the Secretary of Commerce on Behalf of the United States Government

Supersedes and extends the data contained in Tables of Molecular Vibrational Frequencies, NSRDS-NBS-6, Part 1; NSRDS-NBS-11, Part 2; and NSRDS-NBS-17, Part 3.

Issued June 1972

Tables of Molecular Vibrational Frequencies

Consolidated Volume I

T. Shimanouchi

The compilations of fundamental vibrational frequencies of molecules previously published as NSRDS-NBS-6, NSRDS-NBS-11, and NSRDS-NBS-17 have been revised and extended to 52 additional molecules. This consolidated volume includes data on a total of 223 molecules. Selected values of the fundamental vibrational frequencies are given for each molecule, together with observed infrared and Raman spectral data and citations to the original literature. The selection of vibrational fundamentals has been based on careful studies of the spectral data and comprehensive normal-coordinate analyses. An estimate of the accuracy of the selected values is included. The tables provide a convenient source of information for those who require vibrational energy levels and related properties in molecular spectroscopy, thermodynamics, analytical chemistry, and other fields of physics and chemistry.

Key words: Fundamental frequencies; infrared spectra; polyatomic molecules; Raman spectra; vibrational frequencies.

1. Introduction

Establishing the assignment of molecular vibrational frequencies has fundamental importance in elucidating various problems in physics and chemistry. The information concerning the force field and motion of atoms in a molecule can be most directly derived from its vibrational frequencies. If all the vibrational frequencies of a molecule are known, as well as the molecular structure, thermodynamic quantities can be easily computed on the ideal gas model. Thus, the need for a tabulation of evaluated reference data on molecular vibrational frequencies has often been felt by many investiga-

tors. In 1964 a project for producing such tables was initiated at the University of Tokyo in cooperation with the National Standard Reference Data System of the National Bureau of Standards. The evaluated data resulting from this project have been published as Tables of Molecular Vibrational Frequencies, Part 1 (NSRDS-NBS-6), Part 2 (NSRDS-NBS-11) and Part 3 (NSRDS-NBS-17). The present volume consists of the contents of these three publications, after extensive revision in the light of new experimental data, plus tables for 52 additional molecules.

2. Molecules Selected and Their Ordering

The present volume contains tables of fundamental vibrational frequencies for 223 molecules. The molecules were selected from basic organic and inorganic molecules for which the vibrational assignments have been established with little ambiguity. The effort of extending the tables to many other important molecules is continuing in this laboratory. Diatomic molecules and electronically excited species are not included in this volume, since refs. [1] and [2]¹ contain good compilations of data for them. Rotational isomers are treated as independent molecular species, and a separate table is made for each of the isomers. When the gas and liquid state spectra are significantly different from each other, they are tabulated separately.

The molecules are ordered according to the follow-

ing rules:

- (a) Number of carbon atoms.
- (b) Total number of atoms.
- (c) Molecular shape: linear, planar, and non-planar.
- (d) Molecular symmetry, in descending order of the number of symmetry elements. Isotopically substituted molecules directly follow the normal species regardless of their symmetry.
- (e) Atomic number of main atoms.
- (f) Atomic number of the other atoms.

Molecules are first divided into groups by the items (a) and (b) and the ordering of molecules in each group is given by the items (c), (d), (e), and (f). A complete list in the order presented is given at the beginning of the tables. Indices by compound name and empirical formula follow the tables.

¹ Figures in brackets indicate the literature references on page 3.

3. Description of Tables

3.1. Symmetry

The symmetry (point group) of each molecule is given by the Schoenflies notation. Detailed discussions of symmetry properties will be found in refs. [3] and [4].

3.2. Symmetry Number

The symmetry number, σ , is used in the calculation of thermodynamic quantities. It is the number of indistinguishable positions into which the molecule can be transformed by simple rigid rotations. A general discussion and pertinent formulas may be found in ref. [4], page 508.

3.3. Symmetry Species

In the table the normal modes are divided into the symmetry species of the point group to which the molecule belongs. The ordering of species in each point group is given in table I, which is a summary of tables 12–30 of ref. [4]. When a molecule has two or three planes of symmetry, the relationship between the vibrational modes and symmetry species cannot be defined uniquely. In such cases we generally follow the notation adopted in ref. [4].

3.4. Numbering of Frequencies

The numbering is indicated by v_i given in the second column of each table. The normal modes are first grouped into symmetry species, and then those in each species are ordered from higher to lower values of the frequency. However, we always denote the bending vibration of a linear triatomic molecule as v_2 , following the widely accepted tradition. For the C_2X_3 type of molecule we adopt the numbering given in ref. [4], although it is based on D_{3h} symmetry. For some deuterated compounds the frequencies are arranged so that the same v_i numbering is given to the corresponding vibrational modes of deuterated and normal compounds.

3.5. Approximate Type of Mode

The approximate type of mode given in the third column of each table is the local symmetry coordinate which makes the maximum contribution to the normal mode. Local symmetry coordinates are defined for several chemical groups in table II. It should be emphasized that two or more local symmetry coordinates are often coupled strongly in a normal coordinate, and the approximate type of mode given in the table has only limited significance in such a case.

The following abbreviations are used for the type

of mode:

stretch.	stretching
deform.	deformation
rock.	rocking
twist.	twisting
wag.	wagging
scis.	scissors
bend.	bending
sym. or s-	symmetrical
anti. or a-	antisymmetrical
deg. or d-	degenerate
ip-	in-plane
op-	out-of-plane

The plane to which the in-plane and out-of-plane expressions refer is the molecular plane of a planar molecule or the symmetry plane of a general molecule belonging to point group C_s . Local symmetry coordinates of the CX_3 groups attached to a relatively large molecule are designated as s-stretch, s-deform., d-stretch., and d-deform. In such a molecule with low symmetry none of the normal vibrations are genuinely "symmetrical" or "degenerate" with respect to the three-fold symmetry axis of the CX_3 group. However, the notation is retained because it is convenient for indicating the correspondence between similar modes in large and small molecules.

3.6. Selected Value of Frequency

The fundamental frequency ν_i is defined as the difference between the term values $G(v_i = 1, \text{all other } v_j = 0)$ and $G(v_i = 0, \text{and other } v_j = 0)$ expressed in cm^{-1} . Fundamental frequencies rather than harmonic frequencies (ω_i) are listed in the table. Although harmonic frequencies are of greater physical significance, they are accurately known only for a small number of polyatomic molecules. The selected values are rounded to the nearest 1 cm^{-1} .

The letter code, A, B, C, D, or E following the selected value of frequency indicates the evaluator's judgment of the accuracy of the value. The basis for estimating accuracy of an observed frequency is given in table III, together with the range of uncertainty in cm^{-1} for each grade.

Frequencies derived from infrared and Raman measurements in the gaseous state are chosen unless otherwise mentioned. When a detailed analysis of the rotational fine structure of an infrared band is available, the band center ν_0 is chosen as the fundamental frequency and given the uncertainty code A (see below). For a well-analyzed perpendicular band of a symmetric top molecule, the frequency listed contains the nonvibrational part $A' \xi^2$, where A' is the rotational constant of the vibrational level and ξ is the Coriolis coupling constant. This is in accord with the definition of ν_0 given in ref. [4], page 404 and equation (IV, 60).

When the spectra in the gaseous state are not

available, the frequencies observed in the liquid or solid state are listed. When no spectral data have been obtained, the results of normal vibration calculations or of some other methods of estimating frequencies are listed with the grade D or E.

Torsional frequency may be calculated using the barrier height and reduced moment derived from microwave spectroscopy. The value obtained in this way is given as MW (frequency in cm^{-1}) in the "Comments" column or as a footnote for comparison with the value observed or calculated by the normal coordinate treatment. Microwave data are taken from ref. [6] unless otherwise noted.

For many molecules the assignments given in the literature have been checked by normal vibration calculations carried out in this laboratory as part of the project. Revisions in some assignments have been made as a result of these calculations. The details of the normal coordinate treatment and evaluation of force constants may be found in ref. [5].

Thermodynamic quantities may be computed in most cases by employing the harmonic oscillator partition function and by assuming that the harmonic frequencies are not much different from the fundamental frequencies given here. Such an approximation is not adequate, however, for molecules with highly anharmonic motions such as internal rotation, inversion, and ring-puckering. The vibrational partition function should be formed for these molecules by summing the terms due to the individual energy levels.

3.7. Infrared and Raman Spectra

The observed infrared and Raman frequencies are given in the fifth and sixth columns of each table. Rough estimates of relative intensities, band shapes, and polarization characteristics are also given. An additional significant figure is included here when warranted. The abbreviations used here are as follows:

VS	very strong
S	strong
M	medium
W	weak
VW	very weak
ia	inactive
b	broad
vb	very broad
sh	shoulder
p	polarized
dp	depolarized

For some molecules the relative intensities of Raman lines are indicated by the numbers from one to ten in accordance with the tradition widely used. These

estimates of intensity are taken from the original references without any attempt at critical evaluation.

3.8. Comments

In the last column of each table brief comments are added to give special information which is not indicated in the preceding columns. The abbreviations used in this column are as follows:

FR	Fermi resonance with an overtone or a combination tone indicated in the parentheses.
OC	Frequency estimated from an overtone or a combination tone indicated in the parentheses.
CF	Calculated frequency.
SF	Calculation shows that frequency approximately equals that of the vibration indicated in the parentheses.
OV	Overlapped by the band indicated in the parentheses.
MW	Torsional frequency calculated from microwave spectroscopic data.
RP	Frequency determined by the Ritz principle.

3.9. Footnotes and References

The footnote is used to supply other necessary information which cannot be placed simply in the column of Comments. The references accompanying the table are not comprehensive. Only the papers relevant to the present tabulation are cited. The abbreviations IR, R, MW, and Th stand for infrared, Raman, microwave, and theoretical, respectively.

I acknowledge the assistance of the members of my laboratory at the University of Tokyo in carrying out this project. I also express my sincere thanks to many members of the National Bureau of Standards, particularly to C. W. Beckett, D. R. Lide, Jr., E. L. Brady, and S.A. Rossmassler who offered helpful suggestions in the planning of the tables.

References

- [1] Herzberg, G., Spectra of Diatomic Molecules, Second Edition (Van Nostrand, New York, 1950).
- [2] Herzberg, G., Electronic Spectra of Polyatomic Molecules (Van Nostrand, New York, 1966).
- [3] Wilson, E. Bright, Jr., Decius, J. C., Cross, Paul C., Molecular Vibrations (McGraw-Hill, New York, 1955).
- [4] Herzberg, G., Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [5] Shimanouchi, T., Molecular Force Field, in Physical Chemistry, edited by Eyring, Henderson, and Jost. Vol. 4, chap. 6, New York, 1970 (Academic Press).
- [6] Starck, B., Landolt-Börnstein Numerical Data and Functional Relationships in Science and Technology, New Series, Group II, Vol. 4 (Springer-Verlag, Berlin, 1967).

4. Tables of Vibrational Frequencies

C₀-triatomic molecules

		Page		Page	
1	Nitrous oxide, ¹⁴ N ₂ O.....	9	3	Nitrous oxide, ¹⁵ N ₂ O.....	10
2	Nitrous oxide, ¹⁴ N ¹⁵ NO.....	9	4	Water, H ₂ O.....	10
			5	Water-d ₁ , HDO.....	10

Page	C ₀ -eight-atomic molecules	Page	
6 Water-d ₂ , D ₂ O.....	11	55 Diborane, ¹⁰ B ₂ H ₆	36
7 Oxygen difluoride, F ₂ O.....	11	56 Diborane, ¹¹ B ₂ H ₆	37
8 Oxygen dichloride, ³⁵ Cl ₂ ¹⁶ O.....	12	57 Diborane-d ₆ , ¹⁰ B ₂ D ₆	38
9 Hydrogen sulfide, H ₂ S.....	12		
10 Deuterium sulfide, D ₂ S.....	13		
11 Sulfur dioxide, ³² S ¹⁶ O ₂	13		
12 Hydrogen selenide, H ₂ Se.....	14		
13 Hydrogen deuterium selenide, HDSe.....	14		
 C₀-four-atomic molecules			
14 Ammonia, NH ₃	15	58 Carbon dioxide, ¹² C ¹⁶ O ₂	39
15 Ammonia-d ₃ , ND ₃	15	59 Carbon dioxide, ¹³ C ¹⁶ O ₂	39
16 Nitrogen trifluoride, NF ₃	16	60 Carbon disulfide, ¹² C ³² S ₂	40
17 Phosphine, PH ₃	16	61 Carbonyl sulfide, ¹² C ¹⁶ O ³² S.....	40
18 Phosphine-d ₃ , PD ₃	17	62 Hydrogen cyanide, HCN.....	41
19 Phosphorus trifluoride, PF ₃	17	63 Deuterium cyanide, DCN.....	41
20 Phosphorus trichloride, PCl ₃	18	64 Cyanogen chloride, ³⁵ ClCN.....	42
21 Arsine, AsH ₃	18	65 Cyanogen chloride, ³⁷ ClCN.....	42
22 Arsine-d ₃ , AsD ₃	19	66 Cyanogen bromide, ⁷⁹ BrCN.....	43
23 Stibine, SbH ₃	19	67 Cyanogen bromide, ⁸¹ BrCN.....	43
24 Stibine-d ₃ , SbD ₃	20		
 C₀-five-atomic molecules			
25 Silane, SiH ₄	20	68 Formaldehyde, H ₂ CO.....	44
26 Silane-d ₂ , SiH ₂ D ₂	21	69 Formaldehyde-d ₁ , HDCO.....	44
27 Silane-d ₃ , SiHD ₃	21	70 Formaldehyde-d ₂ , D ₂ CO.....	45
28 Silane-d ₄ , SiD ₄	22		
29 Silicon tetrafluoride, SiF ₄	22		
30 Silicon tetrachloride, SiCl ₄	23		
31 Silicon tetrabromide, SiBr ₄	23		
32 Silicon tetraiodide, SiI ₄	24		
33 Germane, GeH ₄	24		
34 Germane-d ₁ , GeH ₃ D.....	25		
35 Germane-d ₂ , GeH ₂ D ₂	25		
36 Germane-d ₃ , GeHD ₃	26		
37 Germane-d ₄ , GeD ₄	26		
38 Germanium tetrachloride, GeCl ₄	27		
39 Germanium tetrabromide, GeBr ₄	27		
40 Tin tetrachloride, SnCl ₄	28		
41 Tin tetrabromide, SnBr ₄	28		
42 Silyl fluoride, SiH ₃ F.....	29		
43 Silyl chloride, SiH ₃ Cl.....	29		
44 Silyl bromide, SiH ₃ Br.....	30		
45 Bromotrichlorosilane, SiBrCl ₃	30		
46 Trichloroiodosilane, SiCl ₃ I.....	31		
47 Tribromoiodosilane, SiBr ₃ Cl.....	31		
48 Chlorotriiodosilane, SiClI ₃	32		
49 Dibromodichlorosilane, SiBr ₂ Cl ₂	32		
 C₀-seven-atomic molecules			
50 Sulfur hexafluoride, SF ₆	33	71 Methane, CH ₄	45
51 Selenium hexafluoride, SeF ₆	33	72 Methane-d ₁ , CH ₃ D.....	46
52 Molybdenum hexafluoride, MoF ₆	34	73 Methane-d ₂ , CH ₂ D ₂	46
53 Tungsten hexafluoride, WF ₆	34	74 Methane-d ₃ , CHD ₃	47
54 Uranium hexafluoride, UF ₆	35	75 Methane-d ₄ , CD ₄	47
		76 Carbon tetrafluoride, CF ₄	48
		77 Carbon tetrachloride, CCl ₄	48
		78 Carbon tetrabromide, CBr ₄	49
		79 Carbon tetraiodide, CI ₄	49
		80 Methyl fluoride, CH ₃ F.....	50
		81 Methyl fluoride-d ₃ , CD ₃ F.....	50
		82 Methyl chloride, CH ₃ Cl.....	51
		83 Methyl chloride-d ₃ , CD ₃ Cl.....	51
		84 Methyl bromide, CH ₃ Br.....	52
		85 Methyl bromide-d ₃ , CD ₃ Br.....	52
		86 Methyl iodide, CH ₃ I.....	53
		87 Methyl iodide-d ₃ , CD ₃ I.....	53
		88 Trifluoromethane, CHF ₃	54
		89 Trichloromethane, CHCl ₃	54
		90 Trichloromethane-d ₁ , CDCl ₃	55
		91 Tribromomethane, CHBr ₃	55
		92 Tribromomethane-d ₁ , CDBr ₃	56
		93 Bromotrichloromethane, CBrCl ₃	56
		94 Tribromochloromethane, CBr ₃ Cl.....	57
		95 Dichloromethane, CH ₂ Cl ₂	57
		96 Dichloromethane-d ₁ , CHDCl ₂	58
		97 Dichloromethane-d ₂ , CD ₂ Cl ₂	58
		98 Dibromomethane, CH ₂ Br ₂	59
		99 Dibromomethane-d ₁ , CHDBr ₂	59
		100 Dibromomethane-d ₂ , CD ₂ Br ₂	60
		101 Dibromodichloromethane, CBr ₂ Cl ₂	60
		102 Bromochloromethane, CH ₂ BrCl.....	61
		103 Bromochloromethane-d ₁ , CHDBrCl.....	61
		104 Bromochloromethane-d ₂ , CD ₂ BrCl.....	62

	Page		Page
105 Formic acid, HCOOH.....	62	149 Ethylene oxide, C ₂ H ₄ O.....	87
106 Formic acid-d ₂ , DCOOD.....	63	150 Ethylene oxide-d ₄ , C ₂ D ₄ O.....	88
C₁-six-atomic molecules			
107 Methanol, CH ₃ OH (Gas).....	63	151 Acetaldehyde, CH ₃ CHO.....	89
108 Methanol, CH ₃ OH (Liquid).....	64	152 Acetaldehyde-d ₁ , CH ₃ CDO.....	90
109 Methanol-d ₁ , CH ₃ OD (Gas).....	64	153 Acetaldehyde-d ₄ , CD ₃ CDO.....	91
110 Methanol-d ₁ , CH ₃ OD (liquid).....	65		
111 Methanol-d ₃ , CD ₃ OII (gas).....	65		
112 Methanol-d ₃ , CD ₃ OH (liquid).....	66		
113 Methanol-d ₄ , CD ₃ OD (gas).....	66		
C₁-seven-atomic molecules			
114 Methylamine, CH ₃ NH ₂	67	154 Ethane, CH ₃ CH ₃	92
115 Methylamine-d ₂ , CH ₃ ND ₂	68	155 Ethane-1,1-d ₃ , CH ₃ CD ₃	93
116 Methylamine-d ₃ , CD ₃ NH ₂	69	156 Ethane-d ₆ , CD ₃ CD ₃	94
117 Methylamine-d ₅ , CD ₃ ND ₂	70	157 Hexafluoroethane, CF ₃ CF ₃	95
C₂-four-atomic molecules			
118 Acetylene, CHCH.....	71	158 Hexachloroethane, CCl ₃ CCl ₃	95
119 Acetylene-d ₁ , CHCD.....	71	159 Hexabromoethane, CBr ₃ CBr ₃	96
120 Acetylene-d ₂ , CD ₂ CD.....	72	160 1,2-Dichloroethane, CH ₂ ClCH ₂ Cl (trans form).....	97
121 Fluoroacetylene, CHCF.....	72	161 1,2-Dichloroethane, CH ₂ ClCH ₂ Cl (gauche form).....	98
122 Chloroacetylene, CHC ₂	73	162 1,2-Dibromoethane, CH ₂ BrCH ₂ Br (trans form).....	99
123 Bromoacetylene, CHCBr.....	73	163 1,2-Dibromoethane, CH ₂ BrCH ₂ Br (gauche form).....	100
C₂-six-atomic molecules			
124 Ethylene, CH ₂ CH ₂	74	164 1-Bromo-2-chloroethane, CH ₂ ClCH ₂ Br (trans form).....	101
125 Ethylene-d ₄ , C ₂ D ₄	75	165 1-Bromo-2-chloroethane, CH ₂ ClCH ₂ Br (gauche form).....	102
126 Tetrafluoroethylene, CF ₂ CF ₂	75	166 Fluoroethane, CH ₃ CH ₂ F.....	103
127 Tetrachloroethylene, CCl ₂ CCl ₂	76	167 Chloroethane, CH ₃ CH ₂ Cl.....	104
128 Tetrabromoethylene, CBr ₂ CBr ₂	76	168 Bromoethane, CH ₃ CH ₂ Br.....	105
129 cis-1,2-Difluoroethylene, CHFCHF.....	77	169 Ethylene imine, C ₂ H ₅ N.....	106
130 cis-1,2-Difluoroethylene-d ₁ , CHFCDF.....	77	170 Methyl formate, HCOOCH ₃	107
131 cis-1,2-Difluoroethylene-d ₂ , CDFCDF.....	78	171 Methyl formate-d ₁ , DCOOCH ₃	108
132 trans-1,2-Dichloroethylene, CHClCHCl.....	78	172 Methyl formate-d ₃ , HCOOC ₂	109
133 trans-1,2-Dichloroethylene-d ₁ , CHClCDCl.....	79	173 Methyl formate-d ₄ , DCOOC ₂	110
134 trans-1,2-Dichloroethylene-d ₂ , CDCICDCl.....	79	174 Acetic acid, CH ₃ COOH.....	111
135 cis-1,2-Dichloroethylene, CHClCHCl.....	80	175 Acetic acid-d ₁ , CH ₃ COOD.....	112
136 cis-1,2-Dichloroethylene-d ₁ , CHClCDCl.....	80		
137 cis-1,2-Dichloroethylene-d ₂ , CDCICDCl.....	81		
138 trans-1,2-Dichloro-1,2-difluoroethylene, CFCICFCI.....	81		
139 1,1-Dichloroethylene, CH ₂ CCl ₂	82		
140 1,1-Dichloroethylene-d ₁ , CHDCCl ₂	82		
141 1,1-Dichloroethylene-d ₂ , CD ₂ CCl ₂	83		
142 1,1-Dichloro-2,2-difluoroethylene, CF ₂ CCl ₂	83		
143 Methyl cyanide, CH ₃ CN.....	84		
144 Methyl cyanide-d ₃ , CD ₃ CN.....	84		
145 Methylisocyanide, CH ₃ NC.....	85		
146 Methylisocyanide-d ₃ , CD ₃ NC.....	85		
C₂-seven-atomic molecules			
147 1,2,5-Oxadiazole, C ₂ H ₂ N ₂ O.....	86	186 Cyclopropane, C ₃ H ₆	120
148 Silylacetylene, SiH ₃ CCH.....	86	187 Cyclopropane-d ₅ , C ₃ D ₆	121
C₂-eight-atomic molecules			
149 Ethylene oxide, C ₂ H ₄ O.....	87	188 Ethylcyanide, CH ₃ CH ₂ CN.....	122
150 Ethylene oxide-d ₄ , C ₂ D ₄ O.....	88		
151 Acetaldehyde, CH ₃ CHO.....	89		
152 Acetaldehyde-d ₁ , CH ₃ CDO.....	90		
153 Acetaldehyde-d ₄ , CD ₃ CDO.....	91		
C₂-nine-atomic molecules			
154 Ethane, CH ₃ CH ₃	92		
155 Ethane-1,1-d ₃ , CH ₃ CD ₃	93		
156 Ethane-d ₆ , CD ₃ CD ₃	94		
157 Hexafluoroethane, CF ₃ CF ₃	95		
158 Hexachloroethane, CCl ₂ CCl ₂	95		
159 Hexabromoethane, CBr ₂ CBr ₂	96		
160 1,2-Dichloroethane, CH ₂ ClCH ₂ Cl (trans form).....	97		
161 1,2-Dichloroethane, CH ₂ ClCH ₂ Cl (gauche form).....	98		
162 1,2-Dibromoethane, CH ₂ BrCH ₂ Br (trans form).....	99		
163 1,2-Dibromoethane, CH ₂ BrCH ₂ Br (gauche form).....	100		
164 1-Bromo-2-chloroethane, CH ₂ ClCH ₂ Br (trans form).....	101		
165 1-Bromo-2-chloroethane, CH ₂ ClCH ₂ Br (gauche form).....	102		
166 Fluoroethane, CH ₃ CH ₂ F.....	103		
167 Chloroethane, CH ₃ CH ₂ Cl.....	104		
168 Bromoethane, CH ₃ CH ₂ Br.....	105		
169 Ethylene imine, C ₂ H ₅ N.....	106		
170 Methyl formate, HCOOCH ₃	107		
171 Methyl formate-d ₁ , DCOOCH ₃	108		
172 Methyl formate-d ₃ , HCOOC ₂	109		
173 Methyl formate-d ₄ , DCOOC ₂	110		
174 Acetic acid, CH ₃ COOH.....	111		
175 Acetic acid-d ₁ , CH ₃ COOD.....	112		
C₂-nine-atomic molecules			
176 Dimethylether, CH ₃ OCH ₃	113		
177 Dimethylether-d ₃ , CH ₃ OCD ₃	114		
C₃-seven-atomic molecules			
178 Allene, CH ₂ CCH ₂	115		
179 Methylacetylene, CH ₃ CCH.....	116		
180 Methylacetylene-d ₁ , CH ₃ CCD.....	116		
181 Methyl-d ₃ -acetylene, CD ₃ CCH.....	117		
182 Methylacetylene-d ₄ , CD ₃ CCD.....	117		
183 Malononitrile, NCCH ₂ CN.....	118		
184 Malononitrile-d ₂ , NCCD ₂ CN.....	118		
C₃-eight-atomic molecules			
185 Propenal, CH ₂ CHCHO.....	119		
C₃-nine-atomic molecules			
186 Cyclopropane, C ₃ H ₆	120		
187 Cyclopropane-d ₅ , C ₃ D ₆	121		
188 Ethylcyanide, CH ₃ CH ₂ CN.....	122		

	Page	Page
C₃-ten-atomic molecules		
189 Acetone, CH ₃ COCH ₃	123	
190 Acetone- α , α -d ₃ , CH ₃ COCD ₃	124	
191 Acetone-d ₆ , CD ₃ COCD ₃	125	
C₃-11-atomic molecules		
192 Propane, CH ₃ CH ₂ CH ₃	126	
193 Propane-2, 2-d ₂ , CH ₃ CD ₂ CH ₃	127	
194 Propane-1, 1, 1-d ₃ , CH ₃ CH ₂ CD ₃	128	
195 Propane-1, 1, 1, 3, 3-d ₆ , CD ₃ CH ₂ CD ₃	129	
196 Propane-d ₈ , CD ₃ CD ₂ CD ₃	130	
197 Methyl acetate, CH ₃ COOCH ₃	131	
198 Methyl acetate-d ₃ , CD ₃ COOCH ₃	132	
199 Methyl-d ₃ -acetate, CH ₃ COOCD ₃	133	
200 Methyl acetate-d ₆ , CD ₃ COOCD ₃	134	
C₄-six-atomic molecules		
201 Butadiyne, HCCCCH.....	135	
C₄-nine-atomic molecules		
202 Furan, C ₄ H ₄ O.....	136	
203 Thiophene, C ₄ H ₄ S.....	137	
204 Thiophene-d ₄ , C ₄ D ₄ S.....	138	
C₄-ten-atomic molecules		
205 1,3-Butadiene, CH ₂ CHCHCH ₂	139	
206 1,3-Butadiene-1-d ₁ (trans), CH ₂ CHCHCHD.....	140	
207 1,3-Butadiene-1, 1, 2-d ₃ , CH ₂ CHCD ₂	141	
208 1,3-Butadiene-1, 1, 4, 4-d ₄ , CD ₂ CHCHCD ₂	142	
C₄-12-atomic molecules		
209 1,3-Butadiene-d ₆ , CD ₂ CDCD ₂	143	
210 2-Butyne, CH ₃ CCCH ₃	144	
C₄-13-atomic molecules		
215 2-Butanone CH ₃ COCH ₂ CH ₃ (trans form).....	149	
C₄-14-atomic molecules		
216 n-Butane, CH ₃ CH ₂ CH ₂ CH ₃ (trans form).....	150	
217 n-Butane, CH ₃ CH ₂ CH ₂ CH ₃ (gauche form).....	151	
C₆-12-atomic molecules		
218 Benzene, C ₆ H ₆	152	
219 Benzene-d ₆ , C ₆ D ₆	153	
C₆-18-atomic molecules		
220 Cyclohexane, C ₆ H ₁₂	154	
221 Cyclohexane-d ₁₂ , C ₆ D ₁₂	155	
Polymer		
222 Poly (methylene), (CH ₂) _n	156	
223 Poly (methylene-d ₂), (CD ₂) _n	157	

TABLE I. Ordering of symmetry species
(In the present volume small letters are used to designate the species of fundamental frequencies)

Point group	Symmetry species	Point group	Symmetry species
C_2	A, B	D_{3h}	$A'_1, A''_1, A'_2, A''_2, E'_1, E''_1$
C_{2v}	A'_1, A''_1	D_{5h}	$A'_1, A''_1, A'_2, A''_2, E'_1, E''_1, E'_g, E''_g$
C_i	A_g, A_u	D_{4h}	$A_{1g}, A_{1u}, A_{2g}, A_{2u}, B_{1g}, B_{1u}, B_{2g}, B_{2u}, E_g, E_u$
C_{2v}	A_1, A_2, B_1, B_2	D_{6h}	$A_{1g}, A_{1u}, A_{2g}, A_{2u}, B_{1g}, B_{1u}, B_{2g}, B_{2u}, E_{1g}, E_{1u}, E_{2g}, E_{2u}$
C_{2h}	A_g, A_u, B_g, B_u	$D_{\infty h}$	$\Sigma_g^+, \Sigma_u^+, \Sigma_g^-, \Sigma_u^-, \pi_g, \pi_u, \Delta_g, \Delta_u, \Phi_g, \Phi_u, \dots$
D_2	A, B_1, B_2, B_3	C_3	A, E
D_{2h}	$A_g, A_u, B_{1g}, B_{1u}, B_{2g}, B_{2u}, B_{3g}, B_{3u}$	C_6	A, B, E_1, E_2
C_{3v}	A_1, A_2, E	S_6	A_g, A_u, E_g, E_u
D_3	A_1, A_2, E	C_{3h}	A', A'', E', E''
C_{5v}	A_1, A_2, E_1, E_2	C_{4h}	$A_g, A_u, B_g, B_u, E_g, E_u$
$C_{\infty v}$	$\Sigma^+, \Sigma^-, \pi, \Delta, \Phi, \dots$	C_{6h}	$A_g, A_u, B_g, B_u, E_{1g}, E_{1u}, E_{2g}, E_{2u}$
C_{4v}, D_4, D_{2d}	A_1, A_2, B_1, B_2, E	T_d, O	A_1, A_2, E, F_1, F_2
C_{6v}, D_6	$A_1, A_2, B_1, B_2, E_1, E_2$	O_h	$A_{1g}, A_{1u}, A_{2g}, A_{2u}, E_g, E_u, F_{1g}, F_{1u}, F_{2g}, F_{2u}$
D_{3d}	$A_{1g}, A_{1u}, A_{2g}, A_{2u}, E_g, E_u$	T	A, E, F
D_{4d}	$A_1, A_2, B_1, B_2, E_1, E_2, E_3$		

TABLE II. Definition of local symmetry coordinates

(a) Local symmetry coordinates for the CH_3 group (see fig. 1a) CH_3 symmetrical stretching: $(\Delta r_1 + \Delta r_2 + \Delta r_3)/\sqrt{3}$ CH_3 degenerate stretching: $(2\Delta r_1 - \Delta r_2 - \Delta r_3)/\sqrt{6}$ $(\Delta r_2 - \Delta r_3)/\sqrt{2}$ CH_3 symmetrical deformation: $(\Delta\alpha_{23} + \Delta\alpha_{31} + \Delta\alpha_{12} - \Delta\beta_1 - \Delta\beta_2 - \Delta\beta_3)/\sqrt{6}$ CH_3 degenerate deformation: $(2\Delta\alpha_{23} - \Delta\alpha_{31} - \Delta\alpha_{12})/\sqrt{6}$ $(\Delta\alpha_{31} - \Delta\alpha_{12})/\sqrt{2}$ CH_3 rocking: $(2\Delta\beta_1 - \Delta\beta_2 - \Delta\beta_3)/\sqrt{6}$ $(\Delta\beta_2 - \Delta\beta_3)/\sqrt{2}$.	(c) Local symmetry coordinates for the CH group (see fig. 1c) CH stretching: Δr_{CH} CH bending: $(2\Delta\beta_{\text{HX}} - \Delta\beta_{\text{HY}} - \Delta\beta_{\text{HZ}})/\sqrt{6}$ $(\Delta\beta_{\text{HY}} - \Delta\beta_{\text{HZ}})/\sqrt{2}$
(b) Local symmetry coordinates for the CH_2 group (see fig. 1b) CH_2 symmetrical stretching: $(\Delta r_1 + \Delta r_2)/\sqrt{2}$ CH_2 antisymmetrical stretching: $(\Delta r_1 - \Delta r_2)/\sqrt{2}$ CH_2 scissors: $(4\Delta\alpha - \Delta\beta_{1X} - \Delta\beta_{2X} - \Delta\beta_{1Y} - \Delta\beta_{2Y})/\sqrt{20}$ CH_2 wagging: $(\Delta\beta_{1X} + \Delta\beta_{2X} - \Delta\beta_{1Y} - \Delta\beta_{2Y})/2$ CH_2 twisting: $(\Delta\beta_{1X} - \Delta\beta_{2X} - \Delta\beta_{1Y} + \Delta\beta_{2Y})/2$ CH_2 rocking: $(\Delta\beta_{1X} - \Delta\beta_{2X} + \Delta\beta_{1Y} - \Delta\beta_{2Y})/2$.	(d) Local symmetry coordinates for the planar CH_2 group (see fig. 1d) CH_2 symmetrical stretching: $(\Delta r_1 + \Delta r_2)/\sqrt{2}$ CH_2 antisymmetrical stretching: $(\Delta r_1 - \Delta r_2)/\sqrt{2}$ CH_2 scissors: $(2\Delta\alpha - \Delta\beta_1 - \Delta\beta_2)/\sqrt{6}$ CH_2 rocking: $(\Delta\beta_1 - \Delta\beta_2)/\sqrt{2}$ CH_2 wagging: $\Delta\theta \cdot \sin \alpha$.
	(e) Local symmetry coordinates for the planar CH group (see fig. 1e) CH stretching: Δr_{CH} $\text{in-plane CH bending: } (\Delta\beta_{\text{HX}} - \Delta\beta_{\text{HY}})/\sqrt{2}$ $\text{out-of-plane CH bending: } \Delta\theta \cdot \sin \gamma_{\text{XY}}$.

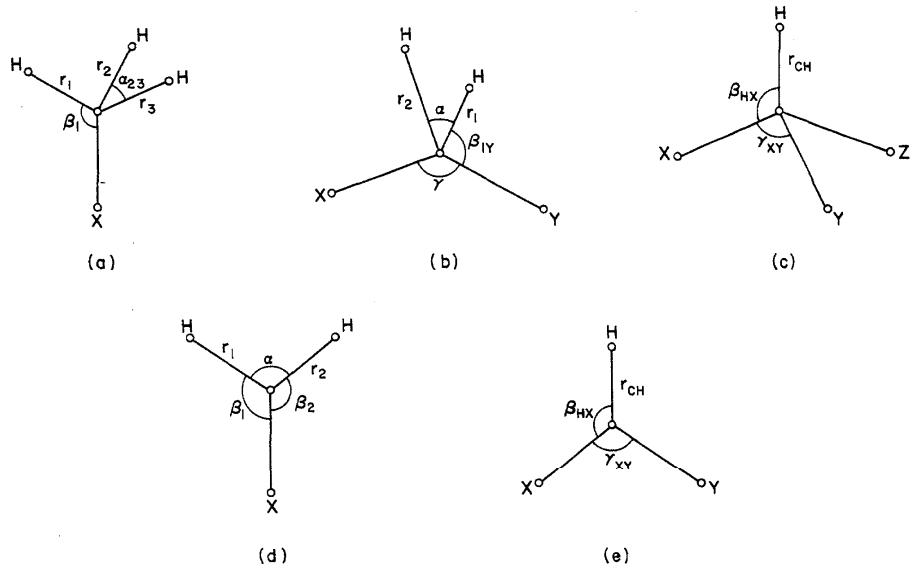


FIGURE 1. Parameters of methyl, methylene, and methin groups.

TABLE III. Uncertainty code for the selected values of frequencies

Notation	Uncertainty	Basis*
A	cm^{-1} 0 ~ 1	(i) Gas, grating spectrometer, rotational fine structure accurately analyzed. (ii) Gas, grating spectrometer, a sharp Q branch.
B	1 ~ 3	(i) Gas, grating spectrometer, rotational fine structure partly analyzed. (ii) Gas, prism spectrometer, fairly high resolution (e.g., $700 \sim 1000 \text{ cm}^{-1}$ for NaCl prism).
C	3 ~ 6	(i) Gas, prism spectrometer, low resolution (e.g., $1000 \sim 2000 \text{ cm}^{-1}$ for NaCl prism). (ii) Solid, liquid or solution, accurate measurement.
D	6 ~ 15	(i) Gas, prism spectrometer, very low resolution (e.g., $>2000 \text{ cm}^{-1}$ for NaCl prism). (ii) Solid, liquid or solution, inaccurate measurement.
E	15 ~ 30	(i) Value estimated from Fermi resonance doublet. (ii) Value estimated from overtone or combination tone. (iii) Calculated frequency.

* The uncertainty assigned here to each method of measurement is a typical value; greater accuracy is often achieved with some of the methods.

Molecule: Nitrous oxide $^{14}\text{N}_2\text{O}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\delta = 1$

No. 1

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	NN stretch.....	2224 A	cm^{-1} (Gas) 2223.7 VW	2224 W	
π	ν_2	Bend.....	589 A	588.7 S	589 W	
σ^+	ν_3	NO stretch.....	1285 A	1284.9 VS	1287 VS	

References

- [1] R. Landolt-Börnstein, "Zahlenwerte und Funktionen aus Physik, Chemie, Astronomie, Geophysik und Technik," 6. Auflage, I. Band, Atom-und Molekularphysik, 2. Teil, Moleküle I. (Springer-Verlag, Berlin, Göttingen, Heidelberg, 1951).
- [2] IR. J. Pliva, J. Mol. Spectrosc. 12, 360 (1964).
- [3] IR. R. P. Grosso and T. K. McCubbin, Jr., J. Mol. Spectrosc. 13, 240 (1964).

Molecule: Nitrous oxide $^{14}\text{N}^{15}\text{NO}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\delta = 1$

No. 2

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	NN stretch.....	2202 A	cm^{-1} (Gas) 2201.6	cm^{-1}	
π	ν_2	Bend.....	585 A	585.3		
σ^+	ν_3	NO stretch.....	1270 A	1269.9		

References

- [1] IR. J. Pliva, J. Mol. Spectrosc. 12, 360 (1964).
- [2] IR. R. P. Grosso and T. K. McCubbin, Jr., J. Mol. Spectrosc. 13, 240 (1964).

Molecule: Nitrous oxide $^{15}\text{N}_2\text{O}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\delta = 1$

No. 3

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	NN stretch.....	2155 Å	cm^{-1} (Gas)	cm^{-1}	
π	ν_2	Bend.....	572 Å	2154.7	571.9	
σ^+	ν_3	NO stretch.....	1265 Å	1265.3		

References

See No. 2.

Molecule: Water H_2O
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 4

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	3657 Å	cm^{-1} (Gas)	cm^{-1} (Gas)	
ν_2		Bend.....	1595 Å	3656.65	3654	
b_1	ν_3	Antisym. stretch.....	3756 Å	1594.59	3755.79	

References

- [1] R. E. F. Barker and W. W. Slater, J. Chem. Phys. 3, 660 (1935).
- [2] IR. W. S. Benedict, N. Gailar and E. K. Plyler, J. Chem. Phys. 24, 1139 (1956).

Molecule: Water-d₁ HDO
 Symmetry C_s Symmetry number $\delta = 1$

No. 5

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	OD stretch.....	2727 Å	cm^{-1} (Gas)	cm^{-1} (Gas)	
ν_2		Bend.....	1402 Å	2726.73	1402.20	
ν_3		OH stretch.....	3707 Å	3707.47	2718	

References

- [1] R. E. F. Barker and W. W. Slater, J. Chem. Phys. 3, 660 (1935).
- [2] IR. W. S. Benedict, N. Gailar, and E. K. Plyler, J. Chem. Phys. 24, 1139 (1956).
- [3] IR. N. Gailar and F. P. Dickey, J. Mol. Spectrosc. 4, 1 (1960).

Molecule: Water-d₂ D₂O
 Symmetry C_{2v} Symmetry number δ = 2

No. 6

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	Sym. stretch.....	2671 A	<i>cm</i> ⁻¹ (Gas)	2671.46	
	<i>v</i> ₂	Bend.....	1178 A		1178.33	
	<i>v</i> ₃	Antisym. stretch.....	2788 A		2788.05	

References

- [1] R. E. F. Barker and W. W. Slater, J. Chem. Phys. 3, 660 (1935).
 [2] IR. W. S. Benedict, N. Gailar, and E. K. Plyler, J. Chem. Phys. 24, 1139 (1956).

Molecule: Oxygen difluoride F₂O
 Symmetry C_{2v} Symmetry number δ = 2

No. 7

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	Sym. stretch.....	928 B	<i>cm</i> ⁻¹ (Gas)	928 S	
	<i>v</i> ₂	Bend.....	461 B		461 S	
	<i>v</i> ₃	Antisym. stretch.....	831 B		831 VS	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
 [2] IR.Th. H. J. Bernstein and J. Powling, J. Chem. Phys. 18, 685 (1950).
 [3] IR.Th. E. A. Jones, J. S. Kirby-Smith, P. J. H. Woltz, and A. H. Nielsen, J. Chem. Phys. 19, 337 (1951).

Molecule: Oxygen dichloride $^{35}\text{Cl}_2^{16}\text{O}$
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 8

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	639 D	cm^{-1} (Gas) 638.6 VS (Ar matrix)	cm^{-1}	
	ν_2	Bend.....	296 C	296.4 W (solid)		
	ν_3	Antisym. stretch.....	686 C	685.9 S		

References

- [1] IR. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR.Th. K. Hedberg, J. Chem. Phys. 19, 509 (1951).
- [3] IR.Th. M. M. Rochkind and G. C. Pimentel, J. Chem. Phys. 42, 1361 (1965).

Molecule: Hydrogen sulfide H_2S
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 9

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	2615 A	cm^{-1} (Gas) 2614.6	cm^{-1}	
	ν_2	Bend.....	1183 A	1182.7		
	ν_3	Antisym. stretch.....	2626 B	2626		

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. J. B. Lohman, F. P. Reding, and D. F. Hornig, J. Chem. Phys. 19, 252 (1951).
- [3] IR. H. C. Allen, Jr., L. R. Blaine, E. K. Plyler, and P. C. Cross, J. Chem. Phys. 24, 35 (1956).
- [4] IR. H. C. Allen, Jr., and E. K. Plyler, J. Chem. Phys. 25, 1132 (1956).

Molecule: Deuterium sulfide D_2S
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 10

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1896 A	cm^{-1} (Gas)	1896.38	
	ν_2	Bend.....			855.45	
	ν_3	Antisym. stretch.....			1999	

References

- [1] IR. A. H. Nielsen and H. H. Nielsen, J. Chem. Phys. 5, 277 (1937).
- [2] R. G. M. Murphy and J. E. Vance, J. Chem. Phys. 6, 426 (1938).
- [3] R. R. E. Miller and D. F. Eggers, Jr., J. Chem. Phys. 45, 3028 (1966).

Molecule Sulfur dioxide $^{32}S^{16}O_2$
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 11

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1151 B	cm^{-1} (Gas)	1151.4 S 517.7 S	1150.5 S, p 524.5 W, p (liquid)
	ν_2	Bend.....				
	ν_3	Antisym. stretch.....				

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. R. D. Shelton, A. H. Nielsen, and W. H. Fletcher, J. Chem. Phys. 21, 2178 (1953).
- [3] IR. S. R. Polo and M. K. Wilson, J. Chem. Phys. 22, 900 (1954).

Molecule: Hydrogen selenide H_2Se
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 12

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	2345 B	cm^{-1} (Gas)	2344.5 S	
b_1	ν_2 ν_3	Bend..... Antisym. stretch.....	1034 A 2358 B	1034.2 S 2357.8 S		

References

- [1] IR. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
 [2] IR.Th. E. D. Palik, J. Mol. Spectrosc. 3, 259 (1959).

Molecule: Hydrogen deuterium selenide $HDSe$
 Symmetry C_s Symmetry number $\delta = 1$

No. 13

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1 ν_2 ν_3	SeD stretch..... Bend..... SeH stretch.....	1691 C 912 C 2352 C	cm^{-1} (Gas) 1691 912 2352	cm^{-1}	

References

- [1] IR.R.Th. D. M. Cameron, W. C. Sears, and H. H. Nielsen, J. Chem. Phys. 7, 994 (1939).
 [2] IR.R. Landolt-Bornstein, "Zahlenwerte und Funktionen aus Physik, Chemie, Astronomie, Geophysik und Technik," 6. Auflage, I. Band, Atom-und Molekularphysik, 2. Teil, Moleküle, I. (Springer-Verlag, Berlin, Göttingen, Heidelberg, 1951).

Molecule: Ammonia NH_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 14

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared ^a	Raman	Comments
a_1	ν_1	Sym. stretch.....	3337 A	cm^{-1} $\left\{ \begin{array}{l} 3336.2s \\ 3337.2a \end{array} \right.$	cm^{-1}	
	ν_2	Sym. deform.....	950 C			
e	ν_3	Deg. stretch.....	3444 A			
	ν_4	Deg. deform.....	1627 A			

^a "s" and "a" refer to symmetric and antisymmetric levels [2].

References

- [1] IR.R. C. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. W. S. Benedict and E. K. Plyler, Can. J. Phys. 35, 1235 (1957).
- [3] IR. J. S. Garing, H. H. Nielsen, and K. N. Rao, J. Mol. Spectrosc. 3, 496 (1959).
- [4] IR. W. S. Benedict, E. K. Plyler, and E. D. Tidwell, J. Chem. Phys. 32, 32 (1960).
- [5] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 523 (1964).

Molecule Ammonia-d₃ ND_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 15

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared ^a	Raman	Comments
a_1	ν_1	Sym. stretch.....	2420 C	cm^{-1} $\left(\text{Gas} \right)$ $\left\{ \begin{array}{l} 2420.1s \\ 2420.6a \end{array} \right.$	cm^{-1}	FR ($2\nu_4$).
	ν_2	Sym. deform.....	748 B			
e	ν_3	Deg. stretch.....	2564 A			
	ν_4	Deg. deform.....	1191 B	2564.0 1191		

^a "s" and "a" refer to symmetric and antisymmetric levels [2].

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. W. S. Benedict and E. K. Plyler, Can. J. Phys. 35, 1235 (1957).
- [3] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 523 (1964).

Molecule: Nitrogen trifluoride NF_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 16

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1032 B	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	Sym. deform.....		1032 S	1050	
	ν_3	Deg. stretch.....		647 B	647 W	
	ν_4	Deg. deform.....		907 C	907 S	
				492 B	492 W	905 515

References

- [1] IR. C. R. Bailey, S. C. Carson, and J. W. Thompson, *J. Chem. Phys.* **5**, 274 (1937).
- [2] IR. M. K. Wilson and S. R. Polo, *J. Chem. Phys.* **20**, 1716 (1952).
- [3] IR.R. E. L. Pace and L. Pierce, *J. Chem. Phys.* **23**, 1248 (1955).
- [4] IR. P. N. Schatz and I. W. Levin, *J. Chem. Phys.* **29**, 475 (1958).
- [5] Th. P. N. Schatz, *J. Chem. Phys.* **29**, 481 (1958).
- [6] IR. I. W. Levin and S. Abramowitz, *J. Chem. Phys.* **44**, 2562 (1966).

Molecule: Phosphine PH_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 17

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	2323 A	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	Sym. deform.....		2322.9	2306	
	ν_3	Deg. stretch.....		992 B	992.1	
	ν_4	Deg. deform.....		2328 B	2327.7	
				1118 A	1118.3	1115

References

- [1] IR. R. Robertson and J. J. Fox, *Proc. Roy. Soc. (London)*, Ser. A, **120**, 161 (1928).
- [2] IR. L. W. Fung and E. F. Barker, *Phys. Rev.* **45**, 238 (1934).
- [3] R. M. Yost and T. F. Anderson, *J. Chem. Phys.* **2**, 624 (1934).
- [4] IR. H. H. Nielsen, *J. Chem. Phys.* **20**, 759 (1952).
- [5] IR. V. M. McConaghie and H. H. Nielsen, *J. Chem. Phys.* **21**, 1836 (1953).
- [6] IR. J. M. Hoffman, H. H. Nielsen, and K. N. Rao, *Z. Elektrochem.* **64**, 606 (1960).
- [7] Th. J. L. Duncan and I. M. Mills, *Spectrochim. Acta* **20**, 523 (1964).

Molecule: Phosphine-d₃ PD₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 18

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1694 B 730 B 1687 D 806 B	cm^{-1} (Gas) 1694 730	cm^{-1}	CF [2].
	ν_2	Sym. deform.....				
	e	Deg. stretch.....				
	ν_4	Deg. deform.....				

References

- [1] IR. E. Lee and C. K. Wu, Trans. Faraday Soc. **35**, 1366 (1939).
 [2] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta **20**, 523 (1964).

Molecule: Phosphorus trifluoride PF₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 19

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	892 B 487 B 860 C 344 B	cm^{-1} (Gas) 892 S 487 M 860 S 344 M	cm^{-1} (Liquid) 890 (10) 486 (3) 840 (10)	
	ν_2	Sym. deform.....				
	e	Deg. stretch.....				
	ν_4	Deg. deform.....				

References

- [1] R. D. M. Yost and T. F. Anderson, J. Chem. Phys. **2**, 624 (1934).
 [2] IR. H. S. Gutowsky and A. D. Liehr, J. Chem. Phys. **20**, 1652 (1952).
 [3] IR. M. K. Wilson and S. R. Polo, J. Chem. Phys. **20**, 1716 (1952).
 [4] Th. A. M. Mirri, F. Scappini, and P. G. Favero, Spectrochim. Acta **21**, 965 (1965).
 [5] IR. I. W. Levin and S. Abramowitz, J. Chem. Phys. **44**, 2562 (1966).

Molecule: Phosphorus trichloride PCl_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 20

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	504 C	cm^{-1} (Gas)	504	cm^{-1} (Liquid)
	ν_2	Sym. deform.....	252 C	252	257 (6) p	
	ν_3	Deg. stretch.....	482 C	482	480 (3) dp	
	ν_4	Deg. deform.....	198 C	198	190 (10) dp	

References

- [1] R. K. W. F. Kohlrausch, Der Smekal-Raman Effekt, Ergänzungsband, 1931–1937, J. Springer, Berlin, 1938.
- [2] IR. P. W. Davais and R. A. Oetjen, J. Mol. Spectrosc. 2, 253 (1958).
- [3] IR. V. Lorenzelli, C. R. 252, 3219 (1961).
- [4] Th. A. M. Mirri, F. Scappini, and P. G. Favero, Spectrochim. Acta 21, 965 (1965).

Molecule Arsine AsH_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 21

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	2116 A	cm^{-1} (Gas)	2116.1	cm^{-1}
	ν_2	Sym. deform.....	906 B	906.0		
	ν_3	Deg. stretch.....	2123 B	2123.0		
	ν_4	Deg. deform.....	1003 B	1003		

References

- [1] IR. R. Robertson and J. J. Fox, Proc. Roy. Soc. (London), Ser. A, 120, 161 (1920).
- [2] IR. E. Lee and C. K. Wu, Trans. Faraday Soc. 35, 1366 (1939).
- [3] IR. V. M. McConaghie and H. H. Nielsen, Phys. Rev. 75, 633 (1949).
- [4] IR. H. H. Nielsen, J. Chem. Phys. 20, 759 (1952).
- [5] IR. H. H. Nielsen, J. Chem. Phys. 20, 1955 (1952).
- [6] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 523 (1964).

Molecule: Arsine-d₃ AsD₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 22

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1523 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	Sym. deform.....	660 C	1523.1	660.0	
	ν_3	Deg. stretch.....	1529 C	1529.3		
	ν_4	Deg. deform.....	714 C	714		

References

- [1] IR. E. Lee and C. K. Wu, Trans. Faraday Soc. 35, 1366 (1939).
- [2] IR. V. M. McConaghie and H. H. Nielsen, Phys. Rev. 75, 633 (1949).
- [3] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 523 (1964).

Molecule: Stibine SbH₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 23

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1891 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	Sym. deform.....	782 C	1890.9	781.5	
	ν_3	Deg. stretch.....	1894 C	1894.2		
	ν_4	Deg. deform.....	831 C	830.9		

References

- [1] IR. H. H. Nielsen, J. Chem. Phys. 20, 759 (1952).
- [2] IR. W. H. Haynie and H. H. Nielsen, J. Chem. Phys. 21, 1839 (1953).
- [3] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 523 (1964).

Molecule: Stibine-d₃ SbD₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 24

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	Sym. stretch.....	1359 B	cm ⁻¹ (Gas) 1358.8		
	<i>v</i> ₂	Sym. deform.....	561 C	561.1		
<i>e</i>	<i>v</i> ₃	Deg. stretch.....	1362 C	1362.0		
	<i>v</i> ₄	Deg. deform.....	593 C	592.5		

References

- [1] IR. W. H. Haynie and H. H. Nielsen, J. Chem. Phys. 21, 1839 (1953).
 [2] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 523 (1964).

Molecule: Silane SiH₄
 Symmetry T_d Symmetry number δ = 12

No. 25

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	Sym. stretch.....	2187 B	cm ⁻¹ (Gas) ia	2187.0 S	
<i>e</i>	<i>v</i> ₂	Deg. deform.....	975 C	ia, ^a 974.6	978 W	
<i>f</i> ₂	<i>v</i> ₃	Deg. stretch.....	2191 A	2190.6		
	<i>v</i> ₄	Deg. deform.....	914 B	914.2		

^a Observed in the infrared through Coriolis interaction with *v*₄.

References

- [1] R. F. B. Stitt and D. M. Yost, J. Chem. Phys. 4, 82 (1936).
 [2] IR. C. H. Tindal, J. W. Straley, and H. H. Nielsen, Phys. Rev. 62, 151 (1942).
 [3] IR.R G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
 [4] IR. D. F. Ball and D. C. McKean, Spectrochim. Acta 18, 1019; 1029 (1962).
 [5] IR. I. W. Levin and W. T. King, J. Chem. Phys. 37, 1375 (1962).

Molecule: Silane-d₂ SiH₂D₂
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 26

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiH ₂ s-stretch.....	2189 C	2189 cm ⁻¹ (Gas)	
	ν_2	SiD ₂ s-stretch.....	1587 C	1587 S	
	ν_3	SiH ₂ scis.....	944 B	944 W	
	ν_4	SiD ₂ scis.....	683 B	682.5 M	
	ν_5	SiH ₂ twist.....	844 E	ia	
	b_1	SiH ₂ a-stretch.....	2183 C	2183 S	CF [1].
	ν_6	SiH ₂ rock.....	743 B	743 S	
	b_2	SiD ₂ a-stretch.....	1601 C	1601 S	
	ν_9	SiH ₂ wag.....	862 B	862 M	

Reference

[1] IR.Th. J. H. Meal and M. K. Wilson, J. Chem. Phys. 24, 385 (1956).

Molecule: Silane-d₃ SiHD₃
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 27

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiH stretch.....	2182 C	2182 cm ⁻¹ (Gas)	
	ν_2	SiD ₃ s-stretch.....	1573 C	1573 S	
	ν_3	SiD ₃ s-deform.....	683 C	683 S	
	ν_4	SiD ₃ d-stretch.....	1598 C	1598 S	SF (ν_6) [1].
	ν_5	SiH bend.....	851 B	851 S	
	ν_6	SiD ₃ d-deform.....	683 C	683 S	SF (ν_3) [1].

References

- [1] IR. J. H. Meal and M. K. Wilson, J. Chem. Phys. 24, 385 (1956).
 [2] IR. I. W. Levin and W. T. King, J. Chem. Phys. 37, 1375 (1962).

Molecule: Silane-d₄ SiD₄
 Symmetry T_d Symmetry number $\delta = 12$

No. 28

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	1558 E	cm^{-1} (Gas)	CF [4].
e	ν_2	Deg. deform.....	700 E	CF [4].
f_2	ν_3	Deg. stretch.....	1597 B	1597 S	
	ν_4	Deg. deform.....	681 C	681 S	

References

- [1] IR. J. H. Meal and M. K. Wilson, J. Chem. Phys. 24, 385 (1956).
- [2] IR. D. F. Ball and D. C. McKean, Spectrochim. Acta 18, 1019; 1029 (1962).
- [3] IR. I. W. Levin and W. T. King, J. Chem. Phys. 37, 1375 (1962).
- [4] Th. T. Shimanouchi, I. Nakagawa, J. Hiraishi, and M. Ishii, J. Mol. Spectrosc. 19, 78 (1966).

Molecule Silicon tetrafluoride SiF₄
 Symmetry T_d Symmetry number $\delta = 12$

No. 29

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	800 C	cm^{-1} (Gas)	ia	800 S
e	ν_2	Deg. deform.....	268 C	ia	268 W	
f_2	ν_3	Deg. stretch.....	1032 B	1031.8 S	1010 W	
	ν_4	Deg. deform.....	389 B	389.35 S	390 W	

References

- [1] IR.R. E. A. Jones, J. S. Kirby-Smith, P. J. H. Woltz, and A. H. Nielsen, J. Chem. Phys. 19, 242 (1951).
- [2] IR. J. Heicklen and V. Knight, Spectrochim. Acta 20, 295 (1964).
- [3] Th. J. L. Duncan and I. M. Mills, Spectrochim. Acta 20, 1089 (1964).
- [4] IR.Th. I. W. Levin and S. Abramowitz, J. Chem. Phys. 44, 2562 (1966).
- [5] IR.Th. I. W. Levin and S. Abramowitz, J. Res. Nat. Bur. Stand. (U.S.), 72A (Phys. and Chem.), No. 3, 247-249 (1968).

Molecule: Silicon tetrachloride SiCl_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 30

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	424 C	cm^{-1} (Gas)	ia	
e	ν_2	Deg. deform.....	150 C	ia	424 (5) p	
f_2	ν_3	Deg. stretch.....	621 C	621 VS	150 (4)	
	ν_4	Deg. deform.....	221 C	610 (2b)	
					221 (4)	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] R. M. L. Delwaalle, J. Phys. Chem. 56, 355 (1952).
- [3] IR.R. A. L. Smith, J. Chem. Phys. 21, 1997 (1953).
- [4] R. D. A. Long, T. V. Spencer, D. N. Waters, and L. A. Woodward, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).
- [5] Th. M. Radhakrishnan, Z. Phys. Chem. (Frankfurt am Main) 41, 197 (1964).

Molecule: Silicon tetrabromide SiBr_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 31

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	249 C	cm^{-1} (Gas)	249 (4) p	
e	ν_2	Deg. deform.....	90 C	ia	90 (3)	
f_2	ν_3	Deg. stretch.....	487 C	487 (1)	
	ν_4	Deg. deform.....	137 C	137 (3)	

References

- [1] R. B. Trumpy, Z. Phys. 68, 675 (1931).
- [2] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [3] R. M. L. Delwaalle, J. Phys. Chem. 56, 355 (1952).
- [4] R. D. A. Long, T. V. Spencer, D. N. Waters, and L. A. Woodward, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).
- [5] Th. M. Radhakrishnan, Z. Phys. Chem. (Frankfurt am Main) 35, 247 (1962).

Molecule: Silicon tetraiodide SiI_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 32

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	168 C	cm^{-1} ia	cm^{-1} (Liquid) 168 S, p	
e	ν_2	Deg. deform.....	63 C	ia	63 M, dp	
f_2	ν_3	Deg. stretch.....	405 C	405 W, dp	
	ν_4	Deg. deform.....	94 C	94 S, dp	

Reference

- [1] R. M. L. Delwaille, J. Phys. Chem. 56, 355 (1952).
 [2] R. M. L. Delwaille and F. François, J. Phys. Radium 15, 206 (1954).

Molecule: Germane GeH_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 33

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	2106 B	cm^{-1} (Gas) ia	cm^{-1} (Gas) 2106 S, p	
e	ν_2	Deg. deform.....	931 D	ia, ^a 930.9	920 W	
f_2	ν_3	Deg. stretch.....	2114 B	2113.6	2106 W (liquid)	
	ν_4	Deg. deform.....	819 B	819.3	816 W (liquid)	

^a Observed in the infrared through Coriolis interaction with ν_4 .

References

- [1] IR. J. W. Straley, C. H. Tindal, and H. H. Nielsen, Phys. Rev. 62, 161 (1942).
 [2] R. K. Schäfer and J. M. Gonzalez Barredo, Z. Phys. Chem. (Leipzig) 193, 334 (1944).
 [3] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
 [4] IR.R. L. P. Lindeman and M. K. Wilson, Z. Phys. Chem. (Frankfurt am Main) 9, 29 (1956).
 [5] IR. A. A. Chalmers and D. C. McKean, Spectrochim. Acta 21, 1941 (1965).

Molecule: Germane-d₁ GeH₃D
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 34

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	GeH ₃ s-stretch.....	2106 C	cm^{-1} (Gas)	
	ν_2	GeD stretch.....	1520 B	1520.4 M		
	ν_3	GeH ₃ s-deform.....	820 C	820 S		
	ν_4	GeH ₃ d-stretch.....	2112 B	2112 S		
	ν_5	GeH ₃ d-deform.....	901 C	901 W		
	ν_6	GeH ₃ rock.....	706 C	706 S		

References

- [1] IR. L. P. Lindeman and M. K. Wilson, J. Chem. Phys. 22, 1723 (1954).
 [2] I.R.R. L. P. Lindeman and M. K. Wilson, Z. Phys. Chem. (Frankfurt am Main) 9, 29 (1956).

Molecule: Germane-d₂ GeH₂D₂
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 35

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	GeH ₂ s-stretch.....	2112 C	cm^{-1} (Gas)	2112	
	ν_2	GeD ₂ s-stretch.....	1512 C		1512	
	ν_3	GeH ₂ scis.....	881 B		881	
	ν_4	GeD ₂ scis.....	620 C		620	
<i>a</i> ₂	ν_5	GeH ₂ twist.....	807 E		807	
	ν_6	GeH ₂ a-stretch.....	2112 C		2112	
	ν_7	GeH ₂ rock.....	657 C		657	
<i>b</i> ₂	ν_8	GeD ₂ a-stretch.....	1522 C		1522	
	ν_9	GeH ₂ wag.....	770 C		770	

Reference

- [1] IR. L. P. Lindeman and M. K. Wilson, Z. Phys. Chem. (Frankfurt am Main) 9, 29 (1956).

Molecule: Germane-d₃ GeHD₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 36

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	GeH stretch.....	2112 B	<i>cm</i> ⁻¹ (Gas) 2112.4		
	<i>v</i> ₂	GeD ₃ s-stretch.....	1504 B			
	<i>v</i> ₃	GeD ₃ s-deform.....	595 C	595		
	<i>v</i> ₄	GeD ₃ d-stretch.....	1522 C	1522		
	<i>v</i> ₅	GeH bend.....	792 B	792.3		
	<i>v</i> ₆	GeD ₃ d-deform.....	625 C	625		

References

- [1] IR. L. P. Lindeman and M. K. Wilson, J. Chem. Phys. 22, 1723 (1954).
 [2] IR.R. L. P. Lindeman and M. K. Wilson, Z. Phys. Chem. (Frankfurt am Main) 9, 29 (1956).

Molecule: Germane-d₄ GeD₄
 Symmetry T_d Symmetry number δ = 12

No. 37

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	Sym. stretch.....	1504 C	<i>cm</i> ⁻¹ (Gas) ia		
	<i>v</i> ₂	Deg. deform.....	665 D	ia, ^a 665 W		
	<i>v</i> ₃	Deg. stretch.....	1522 B	1522.2 S		
	<i>v</i> ₄	Deg. deform.....	596 C	596 S		

^a Observed in the infrared through Coriolis interaction with *v*₄.

Reference

- [1] IR.R. L. P. Lindeman and M. K. Wilson, Z. Phys. Chem. (Frankfurt am Main) 9, 29 (1956).

Molecule: Germanium tetrachloride GeCl_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 38

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	396 C	cm^{-1}	cm^{-1} (Liquid) 396 (10)	
e	ν_2	Deg. deform.....	134 C	134 (6)	
f_2	ν_3	Deg. stretch.....	453 C	453 (1)	
	ν_4	Deg. deform.....	172 C	172 (6)	

References

- [1] R. R. Haun and W. D. Harkins, J. Amer. Chem. Soc. 54, 3917 (1932).
 [2] R. D. A. Long, T. V. Spencer, D. N. Waters, and L. A. Woodward, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).

Molecule: Germanium tetrabromide GeBr_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 39

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	235 C	cm^{-1}	cm^{-1} (Liquid) 235	
e	ν_2	Deg. deform.....	79 C	79	
f_2	ν_3	Deg. stretch.....	327 C	327	
	ν_4	Deg. deform.....	112 C	112	

Reference

- [1] R. D. A. Long, T. V. Spencer, D. N. Waters, and L. A. Woodward, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).

Molecule: Tin tetrachloride SnCl_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 40

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	366 C	cm^{-1} ia	cm^{-1} (Liquid) 366 (10)	
e	ν_2	Deg. deform.....	104 C	ia	104 (5)	
f_2	ν_3	Deg. stretch.....	403 C	403 (6)	
	ν_4	Deg. deform.....	134 C	134 (6)	

References

- [1] R. R. Haun and W. D. Harkins, J. Amer. Chem. Soc. 54, 3917 (1932).
 [2] R. D. A. Long, T. V. Spencer, D. N. Waters, and L. A. Woodward, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).

Molecule: Tin tetrabromide SnBr_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 41

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	220 C	cm^{-1} ia	cm^{-1} (Liquid) 220 (4)	
e	ν_2	Deg. deform.....	64 C	ia	64 (2)	
f_2	ν_3	Deg. stretch.....	279 C	279 (3)	
	ν_4	Deg. deform.....	88 C	88 (4)	

References

- [1] R. B. Trumpy, Z. Phys. 68, 675 (1931).
 [2] R. R. Haun and W. D. Harkins, J. Amer. Chem. Soc. 54, 3917 (1932).
 [3] R. D. A. Long, T. V. Spencer, D. N. Waters, and L. A. Woodward, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).

Molecule: Silyl fluoride SiH_3F
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 42

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiH_3 s-stretch.....	2206 D	cm^{-1} (Gas)	2206
	ν_2	SiH_3 s-deform.....	990 C	990	S	OV (ν_4).
	ν_3	SiF stretch.....	872 B	872	M	
	ν_4	SiH_3 d-stretch.....	2196 C	2196	M	
	ν_5	SiH_3 d-deform.....	956 C	956	M	
	ν_6	SiH_3 rock.....	728 B	728.1 M		

^a The band center was reestimated by Duncan on the basis of the data of Newman et al. [3].

References

- [1] IR. F. A. Andersen and B. Bak, Acta Chem. Scand. **8**, 738 (1954).
- [2] IR. C. Newman, J. K. O'Loane, S. R. Polo, and M. K. Wilson, J. Chem. Phys. **25**, 855 (1956).
- [3] Th. J. L. Duncan, Spectrochim. Acta **20**, 1807 (1964).

Molecule: Silyl chloride SiH_3Cl
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 43

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiH_3 s-stretch.....	2201 D	cm^{-1} (Gas)	2201
	ν_2	SiH_3 s-deform.....	949 D	949	OV (ν_4).
	ν_3	SiCl stretch.....	551 C	551	S	OV (ν_5).
	ν_4	SiH_3 d-stretch.....	2195 B	2195	S	
	ν_5	SiH_3 d-deform.....	954 B	954.4	S	
	ν_6	SiH_3 rock.....	664 B	664.0	M	

References

- [1] IR. A. Monfils, J. Chem. Phys. **19**, 138 (1951).
- [2] IR. A. Monfils, C. R. **236**, 795 (1953).
- [3] IR. C. Newman, J. K. O'Loane, S. R. Polo, and M. K. Wilson, J. Chem. Phys. **25**, 855 (1956).
- [4] Th. J. L. Duncan, Spectrochim. Acta **20**, 1807 (1964).

Molecule: Silyl bromide SiH_3Br
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 44

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiH_3 s-stretch	2200 D	cm^{-1} (Gas) 2200	OV (ν_4).
	ν_2	SiH_3 s-deform	930 C	930 S		
	ν_3	SiBr stretch	430 C	430 M		
	ν_4	SiH_3 d-stretch	2196 C	2196 S		
	ν_5	SiH_3 d-deform	950 B	950.4 S		
	ν_6	SiH_3 rock	633 B	632.6 S		

References

- [1] IR. D. W. Mayo, H. E. Opitz, and J. S. Peake, J. Chem. Phys. **23**, 1344 (1955).
- [2] IR. C. Newman, J. K. O'Loane, S. R. Polo, and M. K. Wilson, J. Chem. Phys. **25**, 855 (1956).
- [3] Th. J. L. Duncan, Spectrochim. Acta **20**, 1807 (1964).

Molecule: Bromotrichlorosilane SiBrCl_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 45

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiCl_3 s-stretch	545 C	cm^{-1} (Liquid) 545 W, p	
	ν_2	SiBr stretch	368 C	368 S, p	
	ν_3	SiCl_3 s-deform	191 C	191 M, p	
	ν_4	SiCl_3 d-stretch	610 C	610 M, dp	
	ν_5	SiCl_3 rock	205 C	205 M, dp	
	ν_6	SiCl_3 d-deform	135 C	135 M, dp	

References

- [1] R. M. L. Delwaille, M. B. Buisset, and M. Delhayé, J. Amer. Chem. Soc. **74**, 5768 (1952).
- [2] R. M. L. Delwaille, J. Phys. Chem. **56**, 355 (1952).
- [3] Th. Y. Kakiuchi, Bull. Chem. Soc. Japan **26**, 260 (1953).

Molecule: Trichloroiodosilane SiCl_3I
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 46

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiCl_3 s-stretch.....	519 C	cm^{-1}	cm^{-1} (Liquid)	
	ν_2	SiI stretch.....	333 C	519 W, p	
	ν_3	SiCl_3 s-deform.....	169 C	333 S, p	
	ν_4	SiCl_3 d-stretch.....	600 C	169 M, p	
	ν_5	SiCl_3 rock.....	197 C	600 W, dp	
	ν_6	SiCl_3 d-deform.....	123 C	197 W, dp	

References

See No. 45.

Molecule: Tribromochlorosilane SiBr_3Cl
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 47

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiCl stretch	579 C	cm^{-1}	cm^{-1} (Liquid)	
	ν_2	SiBr_3 s-stretch.....	288 C	579 W, p	
	ν_3	SiBr_3 s-deform.....	159 C	288 S, p	
	ν_4	SiBr_3 d-stretch.....	498 C	159 M, p	
	ν_5	SiBr_3 d-deform.....	173 C	498 M, dp	
	ν_6	SiCl bend	101 C	173 W, dp	

References

See No. 45.

Molecule: Chlorotriiodosilane SiClI_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 48

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiCl stretch.....	557 C	cm^{-1}	cm^{-1} (Liquid)	
	ν_2	SiI ₃ s-stretch.....	220 C	557 W, p	
	ν_3	SiI ₃ s-deform.....	114 C	220 S, p	
	ν_4	SiI ₃ d-stretch.....	411 C	114 S, p	
	ν_5	SiI ₃ d-deform.....	134 C	411 W, dp	
	ν_6	SiCl bend.....	73 C	134 W, dp	
					73 S, dp	

References

See No. 45.

Molecule: Dibromodichlorosilane SiBr_2Cl_2
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 49

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	SiCl ₂ s-stretch.....	563 C	cm^{-1}	cm^{-1} (Liquid)	
	ν_2	SiBr ₂ s-stretch.....	326 C	563 M, p	
	ν_3	SiCl ₂ scis.....	182 C	326 S, p	
	ν_4	SiBr ₂ scis.....	111 C	182 S, p	
	ν_5	SiCl ₂ twist.....	122 C	111 M, p	
	ν_6	SiCl ₂ a-stretch.....	605 C	122 M, p	
b_1	ν_7	SiCl ₂ rock.....	191 E	605 W, dp	
	ν_8	SiBr ₂ a-stretch.....	508 C	191 VW	
	ν_9	SiBr ₂ rock.....	174 C	508 W, dp	
					174 W, dp	

References

See No. 45.

Molecule: Sulfur hexafluoride SF_6
 Symmetry O_h Symmetry number $\delta = 24$

No. 50

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	Sym. stretch.....	774 B	cm^{-1} (Gas) ia	773.5 VS	
e_g	ν_2	Deg. stretch.....	642 B	ia	641.7 W	
f_{1u}	ν_3	Deg. stretch.....	948 C	947.5	ia	
	ν_4	Deg. deform.....	616 C	615.5	ia	
f_{2g}	ν_5	Deg. deform.....	525 C	ia	525 W	
f_{2u}	ν_6	Deg. deform.....	347 E	ia	ia	OC ($2\nu_6$) [3].

References

- [1] IR. S. Abramowitz and I. W. Levin, J. Chem. Phys. 44, 3353 (1966).
- [2] IR.R. B. Weinstock and G. L. Goodman, Advan. Chem. Phys. 9, 169 (1966), and references cited there.
- [3] R. H. H. Claassen, G. L. Goodman, J. H. Holloway, and H. Selig, J. Chem. Phys. 53, 341 (1970).

Molecule: Selenium hexafluoride SeF_6
 Symmetry O_h Symmetry number $\delta = 24$

No. 51

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	Sym. stretch.....	707 B	cm^{-1} (Gas) ia	706.9 VS	
e_g	ν_2	Deg. stretch.....	659 B	ia	658.7 W	
f_{1u}	ν_3	Deg. stretch.....	780 C	780	ia	
	ν_4	Deg. deform.....	437 C	437	ia	
f_{2g}	ν_5	Deg. deform.....	405 C	ia	405 W	
f_{2u}	ν_6	Deg. deform.....	264 E	ia	ia	OC ($2\nu_6$) [3].

References

- [1] IR.R. B. Weinstock and G. L. Goodman, Advan. Chem. Phys. 9, 169 (1966), and references cited there.
- [2] IR. S. Abramowitz and I. W. Levin, Inorg. Chem. 6, 538 (1967).
- [3] R. H. H. Claassen, G. L. Goodman, J. H. Holloway, and H. Selig, J. Chem. Phys. 53, 341 (1970).

Molecule: Molybdenum hexafluoride MoF_6
 Symmetry O_h Symmetry number $\delta = 24$

No. 52

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	Sym. stretch.....	742 B	cm^{-1} (Gas)	741.5 VS, p	
e_g	ν_2	Deg. stretch.....	652 B	ia	651.6 W, dp	
f_{1u}	ν_3	Deg. stretch.....	741 C	741 VS	ia	
	ν_4	Deg. deform.....	264 C	264 S	ia	
f_{2g}	ν_5	Deg. deform.....	318 C	ia	318 W, dp	
f_{2u}	ν_6	Deg. deform.....	116 E	ia	ia	OC ($2\nu_6$) [3].

References

- [1] IR.R. H. H. Claassen, H. Selig, and J. G. Malm, J. Chem. Phys. 36, 2888 (1962).
- [2] IR.R. B. Weinstock and G. L. Goodman, Advan. Chem. Phys. 9, 169 (1966), and references cited there.
- [3] R. H. H. Claassen, G. L. Goodman, J. H. Holloway, and H. Selig, J. Chem. Phys. 53, 341 (1970).

Molecule: Tungsten hexafluoride WF_6
 Symmetry O_h Symmetry number $\delta = 24$

No. 53

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	Sym. stretch.....	771 B	cm^{-1} (Gas)	771.0 VS, p	
e_g	ν_2	Deg. stretch.....	677 B	ia	677.2 W, dp	
f_{1u}	ν_3	Deg. stretch.....	712 C	712 VS	ia	
	ν_4	Deg. deform.....	258 C	258 S	ia	
f_{2g}	ν_5	Deg. deform.....	320 C	ia	320 W, dp	
f_{2u}	ν_6	Deg. deform.....	127 E	ia	ia	OC ($2\nu_6$) [3].

References

- [1] IR.R. B. Weinstock and G. L. Goodman, Advan. Chem. Phys. 9, 169 (1966), and references cited there.
- [2] IR. S. Abramowitz and I. W. Levin, Inorg. Chem. 6, 538 (1967).
- [3] R. H. H. Claassen and H. Selig, Israel J. Chem. 7, 499 (1969).

Molecule: Uranium hexafluoride UF_6
 Symmetry O_h Symmetry number $\delta = 24$

No. 54

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	Sym. stretch.....	667 B	cm^{-1} (Gas) ia	667.1 VS	
e_g	ν_2	Deg. stretch.....	533 B	ia	532.5 W	
f_{1u}	ν_3	Deg. stretch.....	626 C	626	ia	
	ν_4	Deg. deform.....	186 C	186.2	ia	
f_{2g}	ν_5	Deg. deform.....	202 C	ia	202 W	
f_{2u}	ν_6	Deg. deform.....	142 E	ia	ia	OC ($2\nu_6$) [3].

References

- [1] IR.R. B. Weinstock and G. L. Goodman, *Advan. Chem. Phys.* **9**, 169 (1966), and references cited there.
- [2] IR. B. Frice and H. H. Claassen, *J. Chem. Phys.* **46**, 4603 (1967).
- [3] R. H. H. Claassen, G. L. Goodman, J. H. Halloway, and H. Selig, *J. Chem. Phys.* **53**, 341 (1970).

Molecule: Diborane $^{10}\text{B}_2\text{H}_6$
 Symmetry D_{2h} Symmetry number $\delta = 4$

No. 55

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	BH_2 s-stretch.....	2537 C	cm^{-1} (Gas)	2537 VS	
	ν_2	Ring stretch.....	2110 C	ia	2110 S	
	ν_3	BH_2 scis.....	1186 C	ia	1186 M	
	ν_4	Ring deform.....	816 C	ia	816 S	
a_u	ν_5	BH_2 twist.....	833 C	ia, ^a 833.1 VW	ia	
b_{1g}	ν_6	Ring stretch.....	1768 C	ia	1768 W	
b_{1u}	ν_7	BH_2 wag.....	850 E	ia	OC ($\nu_7 + \nu_{10}$).
	ν_8	BH_2 a-stretch.....	2625 C	2625 VS	ia	CF [9].
b_{2g}	ν_9	BH_2 rock.....	955 E	ia	
	ν_{10}	Ring puckering.....	368 C	368 S	ia	
	ν_{11}	BH_2 a-stretch.....	2640 E	ia	2640 W, b	OC ($\nu_{10} + \nu_{12}$) [6].
	ν_{12}	BH_2 rock.....	930 E	ia	FR ($\nu_9 + \nu_{15}$).
b_{2u}	ν_{13}	Ring stretch.....	1920 E	{ 1882 M (1992 W) }	ia	
b_{3g}	ν_{14}	BH_2 wag.....	977 C	977 S	ia	
	ν_{15}	BH_2 twist.....	1012 E	ia	CF. ^b
	ν_{16}	BH_2 s-stretch.....	2528 C	2528 VS	ia	
	ν_{17}	Ring deform.....	1606 C	1606 VS	ia	
b_{3u}	ν_{18}	BH_2 scis.....	1181 C	1181 VS	ia	

^a Observed very weakly and also confirmed by combination bands.

^b Estimated from ν_{15} of $^{11}\text{B}_2\text{H}_6$.

References

- [1] R. T. F. Anderson and A. B. Burg, J. Chem. Phys. **6**, 586 (1938).
- [2] IR.R. F. Stitt, J. Chem. Phys. **9**, 780 (1941).
- [3] Th. R. P. Bell and H. C. Longuet-Higgins, Proc. Roy. Soc. (London), Ser. A, **183**, 357 (1945).
- [4] IR. W. C. Price, J. Chem. Phys. **16**, 894 (1948).
- [5] IR.R. A. N. Webb, J. T. Neu, and K. S. Pitzer, J. Chem. Phys. **17**, 1007 (1949).
- [6] IR.R. R. C. Lord and E. Nelsen, J. Chem. Phys. **19**, 1 (1951).
- [7] IR. W. J. Lehman, J. F. Ditter, and J. Shapiro, J. Chem. Phys. **29**, 1248 (1958).
- [8] R. R. C. Taylor and A. R. Emery, Spectrochim. Acta **10**, 419 (1958).
- [9] IR. J. T. Kaufman, W. S. Koski, and R. Anacreon, J. Mol. Spectrosc. **11**, 1 (1963).
- [10] Th. T. Ogawa and T. Miyazawa, Spectrochim. Acta **20**, 557 (1964).
- [11] IR. W. L. Smith and I. M. Mills, J. Chem. Phys. **41**, 1479 (1964).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	BH_2 s-stretch.....	2524 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	Ring stretch.....	2104 C	ia	2524 (10) p	
	ν_3	BH_2 scis.....	1180 C	ia	2104 (10) p	
	ν_4	Ring deform.....	794 C	ia	1180 (7)	
	ν_5	BH_2 twist.....	833 C	ia, ^a 833.1 VW	794 (10) p ia	
a_u	ν_6	Ring stretch.....	1768 E	ia	$\left\{ \begin{array}{l} 1788 (1) \\ 1747 (1) \end{array} \right. \text{dp}$	FR ($\nu_5 + \nu_9$).
	ν_7	BH_2 wag.....	850 E	ia	$\left. \begin{array}{l} \\ \end{array} \right\} \text{dp}$	OC ($\nu_7 + \nu_{10}$).
b_{1g}	ν_8	BH_2 a-stretch.....	2612 C	2612 VS	ia	
	ν_9	BH_2 rock.....	950 E	ia	$\left\{ \begin{array}{l} \text{OC} (\nu_5 + \nu_9) \\ \text{OC} (\nu_9 + \nu_{10}) \end{array} \right.$
	ν_{10}	Ring puckering.....	368 C	368 S	ia	
b_{2g}	ν_{11}	BH_2 a-stretch.....	2591 C	ia	2591 (9) dp	
	ν_{12}	BH_2 rock.....	915 E	ia	OC ($\nu_{10} + \nu_{12}$).
b_{2u}	ν_{13}	Ring stretch.....	1915 E	$\left\{ \begin{array}{l} 1887 \text{ M} \\ (1999 \text{ W}) \end{array} \right\}$	ia	
	ν_{14}	BH_2 wag.....	973 C	973 S	ia	
b_{3g}	ν_{15}	BH_2 twist.....	1012 C	ia	1012 (5) dp	
	ν_{16}	BH_2 s-stretch.....	2525 C	2525 VS	ia	
	ν_{17}	Ring deform.....	1602 C	1602 VS	ia	
	ν_{18}	BH_2 scis.....	1177 C	1177 VS	ia	

^a Observed very weakly and also confirmed by combination bands.

References

See No. 55.

Molecule Diborane-d₆ ¹⁰B₂D₆
 Symmetry D_{2h} Symmetry number δ = 4

No. 57

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	ν_1	BD ₂ s-stretch.....	1860 E	cm^{-1} (Gas) ia	cm^{-1} (Liquid) {1880 VS, p 1833 S, p	FR (2 ν_3). OC ($\nu_5 + \nu_7$). OC ($\nu_9 + \nu_{10}$). FR ($\nu_3 + \nu_{18}$).
	ν_2	Ring stretch.....	1511 C	ia	1511 VS, p	
	ν_3	BD ₂ scis.....	929 C	ia	929 p	
	ν_4	Ring deform.....	726 C	ia	726 VS, p	
<i>a_u</i>	ν_5	BD ₂ twist.....	592 D	^a 592 VW	ia	
<i>b_{1g}</i>	ν_6	Ring stretch.....	1273 C	ia	1273 (2) dp	
	ν_7	BD ₂ wag.....	870 E	ia	OC ($\nu_5 + \nu_7$).
<i>b_{1u}</i>	ν_8	BD ₂ a-stretch.....	1999 C	1999 VS	ia	
	ν_9	BD ₂ rock.....	705 E	ia	OC ($\nu_9 + \nu_{10}$).
<i>b_{2g}</i>	ν_{10}	Ring puckering.....	262 C	262 M	ia	
	ν_{11}	BD ₂ a-stretch.....	1980 E	ia	{ 1975 (9) dp (2000 (5))dp	
<i>b_{2u}</i>	ν_{12}	BD ₂ rock.....	740 E	ia	OC ($\nu_{10} + \nu_{12}$). FR ($\nu_5 + \nu_7$).
	ν_{13}	Ring stretch.....	1465 E	{ 1491 M 1459 MS	ia	
<i>b_{3g}</i>	ν_{14}	BD ₂ wag.....	728 C	728 S	ia	
<i>b_{3u}</i>	ν_{15}	BD ₂ twist.....	730 C	ia	730 (4) dp	
	ν_{16}	BD ₂ s-stretch.....	1845 C	{ 1857 VS (1799 S)	ia	FR ($\nu_3 + \nu_{18}$).
	ν_{17}	Ring deform.....	1205 C	1205 VS	ia	
	ν_{18}	BD ₂ scis.....	881 C	881 VS	ia	

^a Observed very weakly and also confirmed by combination bands.

References

- [1] IR. A. N. Webb, J. T. Neu, and K. S. Pitzer, J. Chem. Phys. 17, 1007 (1949).
- [2] IR.R. R. C. Lord and E. Nielsen, J. Chem. Phys. 19, 1 (1951).
- [3] R. R. C. Taylor and A. R. Emery, Spectrochim. Acta 10, 419 (1958).
- [4] Th. T. Ogawa and T. Miyazawa, Spectrochim. Acta 20, 557 (1964).

Molecule: Carbon dioxide $^{12}\text{C}^{16}\text{O}_2$
 Symmetry $D_{\infty h}$ Symmetry number $\delta = 2$

No. 58

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ_g^+	ν_1	Sym. stretch.....	1333 C	cm^{-1} (Gas) ia	cm^{-1} (Gas) $\begin{cases} 1388.15 \\ 1285.40 \end{cases}$	$\begin{cases} \text{FR } (2\nu_2). \\ \end{cases}$
π_u σ_u^+	ν_2 ν_3	Bend..... Antisym. stretch.....	667 A 2349 A	667.38 S 2349.16 VS	ia ia	

References

- [1] IR. E. K. Plyler, L. R. Blaine, and E. D. Tidwell, J. Res. NBS 55, 183 (1955).
- [2] IR. C. P. Courtoy, Can. J. Phys. 35, 608 (1957).
- [3] R. B. P. Stoicheff, Can. J. Phys. 36, 218 (1958).
- [4] IR. C. P. Courtoy, Ann. Sci. Soc. Bruxelles (1) 73, 5 (1959).
- [5] Th. G. A. Amat and M. Pimbert, J. Mol. Spectrosc. 16, 278 (1965).
- [6] IR. H. R. Gordon and T. K. McCubbin, Jr., J. Mol. Spectrosc. 18, 73 (1965); 19, 137 (1966).
- [7] IR. A. Chedin and Z. Cihla, Cah. Phys. 21, 129 (1967).

Molecule: Carbon dioxide $^{13}\text{C}^{16}\text{O}_2$
 Symmetry $D_{\infty h}$ Symmetry number $\delta = 2$

No. 59

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ_g^+	ν_1	Sym. stretch.....	1334 C	cm^{-1} (Gas) ia	cm^{-1} (Gas) $\begin{cases} 1369.90 \\ 1266.03 \end{cases}$	$\begin{cases} \text{FR } (2\nu_2). \\ \end{cases}$
π_u σ_u^+	ν_2 ν_3	Bend..... Antisym. stretch.....	649 A 2283 A	648.91 S 2283.48 VS	ia ia	

References

- [1] R. B. P. Stoicheff, Can. J. Phys. 35, 608 (1957).
- [2] IR. C. P. Courtoy, Ann. Sci. Soc. Bruxelles (1), 73, 5 (1959).
- [3] Th. G. Amat and M. Pimbert, J. Mol. Spectrosc. 16, 278 (1965).
- [4] Th. I. Suzuki, J. Mol. Spectrosc. 25, 479 (1968).

Molecule: Carbon disulfide $^{12}\text{C}^{32}\text{S}_2$
 Symmetry $D_{\infty h}$ Symmetry number $\delta = 2$

No. 60

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ_g^+	ν_1	Sym. stretch.....	658 A	cm^{-1} (Gas) ia	cm^{-1} (Gas) 657.98	
π_u	ν_2	Bend.....	397 B	396.8	ia	
σ_u^+	ν_3	Antisym. stretch.....	1535 B	1535.35	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] R. B. P. Stoicheff, Can. J. Phys. 36, 218 (1958).
- [3] IR. D. Ager, E. K. Plyler, and E. D. Tidwell, J. Res. Nat. Bur. Stand. (U.S.), 66A (Phys. and Chem.) No. 3, 259-264 (1962).

Molecule: Carbonyl sulfide $^{12}\text{C}^{16}\text{O}^{32}\text{S}$
 Symmetry $C_{\infty v}$ Symmetry number $\delta = 1$

No. 61

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CO stretch.....	2062 A	cm^{-1} (Gas) 2062.22	cm^{-1} (Liquid) 2050 W	
π	ν_2	Bend.....	520 A	520.41	521 W dp	
σ^+	ν_3	CS stretch.....	859 B	858.95	858 M p	

References

- [1] R. Landolt-Börnstein, "Zahlenwerte und Funktionen aus Physik, Chemie, Astronomie, Geophysik und Technik," 6. Auflage, I. Band, Atom-und Molekularphysik, 2. Teil, Moleküle I. (Springer-Verlag, Berlin, Göttingen, Heidelberg, 1951).
- [2] IR. A. G. Maki, E. K. Plyler, and E. D. Tidwell, J. Res. Nat. Bur. Stand. (U.S.), 66A, (Phys. and Chem.) No. 2, 163-167 (1962).

Molecule: Hydrogen cyanide HCN
 Symmetry $C_{\infty v}$ Symmetry number $\delta = 1$

No. 62

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CH stretch.....	3311 A	cm^{-1} (Gas)	3311.47 S	
π	ν_2	Bend.....	712 A	711.98 VS	712 W	
σ^+	ν_3	CN stretch.....	2097 A	2096.85 W	2089 S	

References

- [1] R. Landolt-Börnstein, "Zahlenwerte und Funktionen aus Physik, Chemie, Astronomie, Geophysik und Technik," 6. Auflage, I. Band, Atom-und Molekularphysik, 2. Teil, Molekeln I. (Springer-Verlag, Berlin, Göttingen, Heidelberg, 1951).
- [2] IR. H. C. Allen, Jr., E. D. Tidwell, and E. K. Plyler, J. Chem. Phys. 25, 302 (1956).
- [3] IR. A. G. Maki and L. R. Blaine, J. Mol. Spectrosc. 12, 45 (1964).

Molecule: Deuterium cyanide DCN
 Symmetry $C_{\infty v}$ Symmetry number $\delta = 1$

No. 63

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CD stretch.....	2630 A	cm^{-1} (Gas)	2630.30 S	
π	ν_2	Bend.....	569 A	569.04 VS	569	
σ^+	ν_3	CN stretch.....	1925 A	1925.26 W	1906	

References

- [1] R. Landolt-Börnstein, "Zahlenwerte und Funktionen aus Physik, Chemie, Astronomie, Geophysik und Technik," 6. Auflage, I. Band, Atom-und Molekularphysik, 2. Teil, Molekeln I. (Springer-Verlag, Berlin, Göttingen, Heidelberg, 1951).
- [2] IR. H. C. Allen, Jr., E. D. Tidwell, and E. K. Plyler, J. Chem. Phys. 25, 302 (1956).
- [3] IR. A. G. Maki, E. K. Plyler, and R. Thibault, J. Opt. Soc. Amer. 54, 869 (1964).

Molecule: Cyanogen chloride $^{35}\text{ClCN}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\delta = 1$

No. 64

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CN stretch.....	2216 A	cm^{-1} (Gas) 2215.6 VS	2206 (10)	
π	ν_2	Bend.....	378 A	380 S	394 (3)	
σ^+	ν_3	CCl stretch.....	744 C	{ 782.6 S 714.0 S }	730 (5)	RP [2]. FR ($2\nu_2$) [2].

References

- [1] IR.R. W. O. Freitag and E. R. Nixon, J. Chem. Phys. 24, 109 (1956), and references cited there.
 [2] IR. W. J. Lafferty, D. R. Lide, and R. A. Toth, J. Chem. Phys. 43, 2063 (1965).

Molecule: Cyanogen chloride $^{37}\text{ClCN}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\delta = 1$

No. 65

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CN stretch.....	2215 A	cm^{-1} (Gas) 2215.3 VS	2206 (10)	
π	ν_2	Bend.....	378 A	380 S	394 (3)	
σ^+	ν_3	CCl stretch.....	736 C	730 (5)	RP [2]. FR ($2\nu_2$) [2].

References

See No. 64.

Molecule: Cyanogen bromide $^{79}\text{BrCN}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\sigma = 1$

No. 66

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CN stretch.....	2198 A	cm^{-1} (Gas)	cm^{-1} (Liquid)	
π	ν_2	Bend.....	342 A	2198.3	2191	
σ^+	ν_3	CBr stretch.....	575 C	341.5	357	RP [2].
				575	568	

References

- [1] IR.R. W. O. Freitag and E. R. Nixon, J. Chem. Phys. 24, 109 (1956), and references cited there.
- [2] IR. A. G. Maki and C. T. Gott, J. Chem. Phys. 36, 2282 (1962).
- [3] IR. A. G. Maki, J. Chem. Phys. 38, 1261 (1963).

Molecule: Cyanogen bromide $^{81}\text{BrCN}$
 Symmetry $\text{C}_{\infty v}$ Symmetry number $\sigma = 1$

No. 67

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CN stretch.....	2198 A	cm^{-1} (Gas)	cm^{-1} (Liquid)	
π	ν_2	Bend.....	342 A	2198.3	2191	
σ^+	ν_3	CBr stretch.....	575 C	341.5	357	RP [2].
				575	568	

References

See No. 66.

Molecule: Formaldehyde H₂CO
 Symmetry C_{2v} Symmetry number δ = 2

No. 68

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH ₂ s-stretch.....	2783 A	2782.5 S <i>cm</i> ⁻¹ (Gas)	2781.6 S <i>cm</i> ⁻¹ (Gas)	
	<i>v</i> ₂	CO stretch.....	1746 A	1746.1 VS	1742.3 W	
	<i>v</i> ₃	CH ₂ scis.....	1500 A	1500.1 S	1499.7 M	
	<i>b</i> ₁	CH ₂ a-stretch.....	2843 A	2843.1 VS	2866 W	
	<i>v</i> ₄					
	<i>v</i> ₅	CH ₂ rock.....	1249 A	1249.1 S		
<i>b</i> ₂	<i>v</i> ₆	CH ₂ wag.....	1167 A	1167.3 S		

References

- [1] R. D. W. Davidson, B. P. Stoicheff, and H. J. Bernstein, J. Chem. Phys. 22, 289 (1954).
- [2] IR. H. H. Blau, Jr. and H. H. Nielsen, J. Mol. Spectrosc. 1, 124 (1957).
- [3] IR. K. B. Harvey and J. F. Ogilvie, Can. J. Chem. 40, 85 (1962).
- [4] IR.Th. T. Nakagawa, H. Kashiwagi, H. Kurihara, and Y. Morino, J. Mol. Spectrosc., 31, 436 (1969).

Molecule: Formaldehyde-d₁ HDCO
 Symmetry C_s Symmetry number δ = 1

No. 69

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH stretch.....	2844 D	2844.1 S <i>cm</i> ⁻¹ (Gas)	2846.2 S <i>cm</i> ⁻¹ (Gas)	
	<i>v</i> ₂	CD stretch.....	2121 D	2120.7 S	2120.3 S	
	<i>v</i> ₃	CO stretch.....	1723 A	1723.4 VS	1723.2 VS	
	<i>v</i> ₄	CHD scis.....	1400 B	1400.0 S	1397.4 M	
	<i>v</i> ₅	CHD rock.....	1041 D	1041 S		FR (<i>v</i> ₃ + <i>v</i> ₅).
	<i>v</i> ₆	CHD wag.....	1074 C	1074 S		FR (2 <i>v</i> ₆ , 2 <i>v</i> ₅).

Reference

- [1] IR.R. D. W. Davidson, B. P. Stoicheff, and H. J. Bernstein, J. Chem. Phys. 22, 289 (1954).

Molecule: Formaldehyde-d₂ D₂CO
 Symmetry C_{2v} Symmetry number δ = 2

No. 70

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a ₁	ν ₁	CD ₂ s-stretch.....	2056 D	cm ⁻¹ (Gas)	cm ⁻¹	
	ν ₂	CO stretch.....	1700 B	2056.4 S	
	ν ₃	CD ₂ scis.....	1106 C	1700 VS	
b ₁	ν ₄	CD ₂ a-stretch.....	2160 C	1106.0 S	
	ν ₅	CD ₂ rock.....	990 C	2160.3 VS	
b ₂	ν ₆	CD ₂ wag.....	938 E	990.2 S	
				938 S	FR (2ν ₃).

Reference

- [1] IR. E. S. Ebers and H. H. Nielsen, J. Chem. Phys. 6, 311 (1938).

Molecule: Methane CH₄
 Symmetry T_d Symmetry number δ = 12

No. 71

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a ₁	ν ₁	Sym. stretch.....	2917 A	cm ⁻¹ (Gas)	cm ⁻¹ (Gas)	
e	ν ₂	Deg. deform.....	1534 A	ia	2917.0	
f ₂	ν ₃	Deg. stretch.....	3019 A	ia, ^a 1533	1533.6	
	ν ₄	Deg. deform.....	1306 C	3018.9	3019.5	
				1306.2	

^a Observed in the infrared through Coriolis interaction with ν₄ [5].

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
 [2] IR. H. C. Allen, Jr., and E. K. Plyler, J. Chem. Phys. 26, 972 (1957).
 [3] R. M. A. Thomas and H. L. Welsh, Can. J. Phys. 38, 1291 (1960).
 [4] R. J. Herranz and B. P. Stoicheff, J. Mol. Spectrosc. 10, 448 (1963).
 [5] IR. J. Herranz, J. Morcillo, and A. Gómez, J. Mol. Spectrosc. 19, 266 (1966).

Molecule: Methane-d₁ CH₃D
 Symmetry C_{3v} Symmetry number δ = 3

No. 72

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	CH ₃ s-stretch.....	2945 E	cm^{-1} (Gas) 2973 M 2914 M	cm^{-1} (Gas)	FR (2 ν_5).
	ν_2	CD stretch.....	2200 A	2200.0 M		
	ν_3	CH ₃ s-deform.....	1300 C	1300 M	1306	
	ν_4	CH ₃ d-stretch.....	3017 B	3016.9 S		
	ν_5	CH ₃ d-deform	1471 C	1471 W		
	ν_6	CH ₃ rock.....	1155 C	1155 M	1156	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. J. K. Wilmhurst and H. J. Bernstein, Can. J. Chem. 35, 226 (1957).
- [3] IR. H. C. Allen, and E. K. Plyler, J. Res. NBS 63, 145 (1959).
- [4] IR. L. H. Jones, J. Mol. Spectrosc. 4, 86 (1960).

Molecule: Methane-d₂ CH₂D₂
 Symmetry C_{2v} Symmetry number δ = 2

No. 73

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	CH ₂ s-stretch.....	2974 C	cm^{-1} (Gas) 2976 M	cm^{-1} (Gas) 2974	
	ν_2	CD ₂ s-stretch.....	2202 C	2202 W		
	ν_3	CH ₂ scis.....	1436 C	1436 W		
	ν_4	CD ₂ scis.....	1033 C	1033 S	1034	
	ν_5	CH ₂ twist.....	1333 C	ia, ^a 1329 W	1333	
	ν_6	CH ₂ a-stretch.....	3013 C	3013 S		
	ν_7	CH ₂ rock.....	1090 C	1090 S	1090	
	ν_8	CD ₂ a-stretch.....	2234 C	2234 M		
	ν_9	CH ₂ wag.....	1234 C	1234 M		

^a Observed in the infrared through Coriolis interaction with ν_9 .

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. J. K. Wilmhurst and H. J. Bernstein, Can. J. Chem. 35, 226 (1957).

Molecule: Methane-d₃ CHD₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 74

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH stretch.....	2993 C	<i>cm</i> ⁻¹ (Gas)	2993 M	
	<i>v</i> ₂	CD ₃ s-stretch.....	2142 C	2142 M	2141	
	<i>v</i> ₃	CD ₃ s-deform.....	1003 C	1003 M		
	<i>v</i> ₄	CD ₃ d-stretch.....	2263 C	2263 M	2269	
	<i>v</i> ₅	CD ₃ rock.....	1291 C	1291 M	1299	
	<i>v</i> ₆	CD ₃ d-deform.....	1036 C	1036 S	1046	

References

See No. 73.

Molecule: Methane-d₄ CD₄
 Symmetry T_d Symmetry number δ = 12

No. 75

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	Sym. stretch.....	2109 B	<i>cm</i> ⁻¹ (Gas)	ia	2108.9
	<i>v</i> ₂	Deg. deform.....	1092 B	ia,	^a 1092	1091.9
	<i>v</i> ₃	Deg. stretch.....	2259 A		2259.3	2259.3
	<i>v</i> ₄	Deg. deform.....	996 B		996.0	

^a Observed in the infrared through Coriolis interaction with *v*₄ [5].

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. H. M. Kaylor and A. H. Nielsen, J. Chem. Phys. 23, 2139 (1955).
- [3] R. G. C. Shepherd and H. L. Welsh, J. Mol. Spectrosc. 1, 277 (1957).
- [4] R. R. A. Olafson, M. A. Thomas, and H. L. Welsh, Can. J. Phys. 39, 419 (1961).
- [5] IR. H. Herranz, J. Morello, and A. Gómez, J. Mol. Spectrosc. 19, 266 (1966).

Molecule: Carbon tetrafluoride CF_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 76

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	909 B	cm^{-1} (Gas) ia	908.5 S	
e	ν_2	Deg. deform.....	435 B	ia	435.0 S	
f_2	ν_3	Deg. stretch.....	1281 D	1282.6 VS 1260.9 VW	1283.0 W 1263 VW	FR ($2\nu_4$).
	ν_4	Deg. deform.....	632 B	631.73 VS	631.2 S	

References

- [1] R. D. M. Yost, E. N. Lassetre, and S. T. Gross, J. Chem. Phys. 4, 325 (1936).
- [2] IR. C. R. Baeley, J. B. Hale, and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 167, 555 (1938).
- [3] IR.Th. E. K. Plyler and W. S. Benedict, J. Res. NBS 47, 202 (1951), RP 2245.
- [4] IR. P. J. H. Woltz and A. H. Nielsen, J. Chem. Phys. 20, 307 (1952).
- [5] IR.Th. W. F. Edgell and R. E. Moynihan, J. Chem. Phys. 27, 155 (1957).
- [6] R. B. Monostri and A. Weber, J. Chem. Phys. 33, 1867 (1960).
- [7] IR. S. Abramowitz and R. Bauman, Spectrochim. Acta 17, 127 (1961).
- [8] IR. A. Maki, E. K. Plyler, and R. Thibault, J. Chem. Phys. 37, 1899 (1962).
- [9] IR. S. Abramowitz and R. Bauman, J. Chem. Phys. 39, 2757 (1963).
- [10] IR. A. A. Chalmers and D. C. McKean, Spectrochim. Acta 22, 251 (1966).

Molecule: Carbon tetrachloride CCl_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 77

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	Sym. stretch.....	459 C	cm^{-1} (Gas) ia	458.7 (10) p	
e	ν_2	Deg. deform.....	217 C	ia	217.0 (7) dp	
f_2	ν_3	Deg. stretch.....	776 E	{ 789 VS 768 VS	790.4 (4) dp 761.7 (4) dp } FR ($\nu_1 + \nu_4$).	
	ν_4	Deg. deform.....	314 C	309.9 W (liquid)	313.5 (9) dp	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] R. J. P. Zietlow, F. F. Cleveland, and A. G. Meister, J. Chem. Phys. 18, 1076 (1950).
- [3] IR. J. R. Madigan and F. F. Cleveland, J. Chem. Phys. 19, 119 (1951).
- [4] IR. E. K. Plyler and W. S. Benedict, J. Res. NBS 47, 202 (1951), RP2245.
- [5] IR. H. Yoshinaga, J. Chem. Phys. 23, 2206 (1955).
- [6] R. D. A. Long, D. C. Milner, and A. G. Thomas, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).
- [7] R. M. Ito, Spectrochim. Acta 21, 731 (1965).

Molecule: Carbon tetrabromide CBr_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 78

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
				cm^{-1} (Liquid)	cm^{-1} (Benzene soln.)	
a_1	ν_1	Sym. stretch.....	267 C	ia	267 (7) p	
e	ν_2	Deg. deform.....	122 C	ia	122 (10) dp	
f_2	ν_3	Deg. stretch.....	672 C	672 VS	671 (1) dp	
	ν_4	Deg. deform.....	182 C	182 (4) dp	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] R. A. G. Meister, S. E. Rosser, and F. F. Cleveland, J. Chem. Phys. 18, 346 (1950).
- [3] IR.R. E. K. Plyler, W. H. Smith, and N. Acquista, J. Res. NBS 44, 503 (1950), RP2097.
- [4] R. D. A. Long, D. C. Milner, and A. G. Thomas, Proc. Roy. Soc. (London), Ser. A, 240, 499 (1957).

Molecule: Carbon tetraiodide CI_4
 Symmetry T_d Symmetry number $\delta = 12$

No. 79

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
				cm^{-1} (Solid)	cm^{-1} (Solid)	
a_1	ν_1	Sym. stretch.....	178 D	ia	178 (10)	
e	ν_2	Deg. deform.....	90 D	ia	90 (4)	
f_2	ν_3	Deg. stretch.....	555 D	555 VS		
	ν_4	Deg. deform.....	125 E	^a {123 W 127 W}	123 (5)	

^a Crystal field splitting.

Reference

- [1] IR.R. H. Stammreich, Y. Tovares, and D. Bassi, Spectrochim. Acta 17, 661 (1961).

Molecule: Methylfluoride CH_3F
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 80

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch.....	2930 E	cm^{-1} (Gas) 2964 VS 2863 S	cm^{-1}	FR ($2\nu_5$).
	ν_2	CH_3 s-deform.....	1464 A	1464 S		
	ν_3	CF stretch.....	1049 A	1048.6 S		
	ν_4	CH_3 d-stretch.....	3006 A	3005.8 S		
	ν_5	CH_3 d-deform.....	1467 A	1466.5 M		
	ν_6	CH_3 rock.....	1182 A	1182.4 M		

References

- [1] IR. K. P. Yates and H. H. Nielsen, Phys. Rev. 71, 349 (1947).
- [2] IR. J. Pickworth and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 222, 443 (1954).
- [3] IR. F. A. Anderson, B. Bak, and S. Brodersen, J. Chem. Phys. 24, 989 (1956).
- [4] IR. W. L. Smith and I. M. Mills, J. Mol. Spectrosc. 11, 11 (1963).
- [5] IR. E. W. Jones, E. J. L. Popplewell, and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 290, 490 (1966).

Molecule: Methylfluoride-d₃ CD_3F
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 81

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CD_3 s-stretch.....	2110 E	cm^{-1} (Gas) 2090 2150	cm^{-1}	FR ($2\nu_5$).
	ν_2	CD_3 s-deform.....	1136 A	1136		
	ν_3	CF stretch.....	991 A	991		
	ν_4	CD_3 d-stretch.....	2258 A	2258		
	ν_5	CD_3 d-deform.....	1072 A	1072		
	ν_6	CD_3 rock.....	903 A	903		

References

- [1] IR. J. Pickworth and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 222, 443 (1954).
- [2] IR. W. F. Edgell and L. Parts, J. Amer. Chem. Soc. 78, 2358 (1956).
- [3] IR. E. W. Jones, E. J. L. Popplewell, and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 290, 490 (1966).

Molecule: Methylchloride CH_3Cl
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 82

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch	2937 E	cm^{-1} (Gas) 2967.78 M 2879.28 M	cm^{-1} (Liquid) 2955 VS, p 2861 M	FR ($2\nu_5$). FR ($3\nu_6$) [6, 8].
	ν_2	CH_3 s-deform	1355 A	1354.9 S	1370 VW, p	
	ν_3	CCl stretch	732 A	732.1 S	709 VS, p	
	ν_4	CH_3 d-stretch	3039 B	cm^{-1} (Gas) 3039.31 S 3042.75 S	3036 M, dp	
	ν_5	CH_3 d-deform	1452 A	1452.1 M	1446 W, dp	
	ν_6	CH_3 rock	1017 A	1017.3 M	1016 W, dp	

References

- [1] R. J. Wagner, Z. Phys. Chem. B40, 439 (1938).
- [2] IR. J. Pickworth and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 222, 443 (1954).
- [3] IR. W. T. King, I. M. Mills, and B. Crawford, Jr., J. Chem. Phys. 27, 455 (1964).
- [4] IR. Y. Morino and J. Nakamura, Bull. Chem. Soc. Japan 38, 443 (1965).
- [5] IR. E. W. Jones, R. J. L. Popplewell, and H. W. Thompson, Spectrochim. Acta 22, 669 (1966).
- [6] IR. A. G. Maki and R. Thibault, J. Chem. Phys. 48, 2163 (1968).
- [7] IR. M. Morillon-Chapey and G. Graner, J. Mol. Spectrosc. 31, 155 (1969).
- [8] IR. M. Morillon-Chapey, Ph.D. Thesis (University of Paris, 1970).

Molecule: Methylchloride-d₃ CD_3Cl
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 83

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CD_3 s-stretch	2160 A	cm^{-1} (Gas) 2160.28 S	cm^{-1}	
	ν_2	CD_3 s-deform	1029 A	1028.7 S		
	ν_3	CCl stretch	701 A	701.4 S		
	ν_4	CD_3 d-stretch	2283 A	2283.3 S		
	ν_5	CD_3 d-deform	1060 A	1059.9 M		
	ν_6	CD_3 rock	768 A	767.6 M		

References

- [1] IR. J. Pickworth and H. W. Thompson, Proc. Roy. Soc. (London) Ser. A, 222, 443 (1954).
- [2] IR. W. T. King, I. M. Mills, and B. L. Crawford, Jr., J. Chem. Phys. 27, 455 (1964).
- [3] IR. Y. Morino and J. Nakamura, Bull. Chem. Soc. Japan 38, 443 (1965).
- [4] IR. E. W. Jones, R. J. L. Popplewell, and H. W. Thompson, Spectrochim. Acta 22, 659 (1966).
- [5] IR. C. Alamichel, C. R. B268, 483 (1969).

Molecule: Methylbromide CH_3Br
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 84

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch.....	2935 E	cm^{-1} (Gas) 2972 M 2862.1 M	2972 VS 2862 W	FR ($2\nu_5$).
	ν_2	CH_3 s-deform.....	1306 A	1305.9 S	1309 W	
	ν_3	CBr stretch.....	611 A	611.1 S	609 S	
	ν_4	CH_3 d-stretch.....	3056 A	3056.35 S	3068 VS	
	ν_5	CH_3 d-deform.....	1443 A	1442.7 M	1456 M	
	ν_6	CH_3 rock.....	955 A	954.7 M	956 VW	

References

- [1] R. H. L. Welsh, M. F. Crawford, T. R. Thomas, and G. R. Love, Can. J. Phys. 30, 577 (1952).
- [2] IR. H. B. Weissman, R. B. Bernstein, S. E. Rosser, A. G. Meister, and F. F. Cleveland, J. Chem. Phys. 23, 544 (1955).
- [3] IR. Y. Morino and J. Nakamura, Bull. Chem. Soc. Japan 38, 443 (1965).
- [4] IR. Y. Morino, J. Nakamura, and S. Yamamoto, Bull. Chem. Soc. Japan 38, 459 (1965).
- [5] IR. E. W. Jones, R. J. L. Popplewell, and H. W. Thompson, Spectrochim. Acta 22, 647 (1966).
- [6] IR. T. L. Barnett and T. H. Edwards, J. Mol. Spectrosc. 20, 352 (1966).

Molecule: Methylbromide-d₃ CD_3Br
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 85

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CD_3 s-stretch.....	2160 A	cm^{-1} (Gas) 2159.8 VS	cm^{-1}	
	ν_2	CD_3 s-deform.....	992 A	992.0 VS		
	ν_3	CBr stretch.....	577 A	576.7 S		
	ν_4	CD_3 d-stretch.....	2297 A	2297.3 M		
	ν_5	CD_3 d-deform.....	1056 A	1055.6 S		
	ν_6	CD_3 rock.....	713 A	713.0 M		

References

- [1] IR. H. B. Weissman, R. B. Bernstein, S. E. Rosser, A. G. Meister, and F. F. Cleveland, J. Chem. Phys. 23, 544 (1955).
- [2] IR. Y. Morino and J. Nakamura, Bull. Chem. Soc. Japan 38, 443 (1965).
- [3] IR. E. W. Jones, R. J. L. Popplewell, and H. W. Thompson, Spectrochim. Acta 22, 639 (1966).

Molecule: Methyliodide CH_3I
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 86

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch.....	2933 E	cm^{-1} (Gas) 2969.8 M 2861.0 M	cm^{-1}	FR ($2\nu_5$). FR ($\nu_3 + \nu_6$).
	ν_2	CH_3 s-deform.....	1252 A		1251.5 S	
	ν_3	CI stretch.....	533 A	S 532.8 S 3060 A 1435.5 M	532.8 S	
	ν_4	CH_3 d-stretch.....	3060 A		3060.06 S	
	ν_5	CH_3 d-deform.....	1436 C		1435.5 M	
	ν_6	CH_3 rock.....	882 A		882.4 M	

References

- [1] Th. W. T. King, I. M. Mills, and B. Crawford, Jr., J. Chem. Phys. 27, 455 (1957).
- [2] IR. E. W. Jones and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 288, 50 (1965).
- [3] IR. Y. Morino and J. Nakamura, Bull. Chem. Soc. Japan 38, 443 (1965).
- [4] IR. Y. Morino, J. Nakamura, and S. Yamamoto, J. Mol. Spectrosc. 22, 34 (1967).
- [5] IR. T. L. Barnett and T. H. Edwards, J. Mol. Spectrosc. 23, 302 (1967).

Molecule: Methyliodide-d₃ CD_3I
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 87

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CD_3 s-stretch.....	2130 E	cm^{-1} (Gas) 2155.1 2081.0	cm^{-1}	FR ($2\nu_5$).
	ν_2	CD_3 s-deform.....	951 A		950.7	
	ν_3	CI stretch.....	501 A	A 501.4 2298 1049 A	501.4	
	ν_4	CD_3 d-stretch.....	2298 A		2298	
	ν_5	CD_3 d-deform.....	1049 A		1049.3	
	ν_6	CD_3 rock.....	656 A		655.9	

References

- [1] Th. W. T. King, I. M. Mills, and B. Crawford, Jr., J. Chem. Phys. 27, 455 (1957).
- [2] IR. E. W. Jones, R. J. L. Popplewell, and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 288, 39 (1965).
- [3] IR. Y. Morino and J. Nakamura, Bull. Chem. Soc. Japan 38, 443 (1965).
- [4] IR. R. W. Peterson and T. H. Edwards, J. Mol. Spectrosc. 38, 1 (1971).

Molecule: Trifluoromethane CHF_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 88

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH stretch.....	3036 C	cm^{-1} (Gas)	3036 S	cm^{-1} (Liquid)
	ν_2	CF_3 s-stretch.....	1117 C	1117 VS, p	
	ν_3	CF_3 s-deform.....	700 C	700 M	697 S, p	
	ν_4	CH bend.....	1372 C	1372 M	1376 S, dp	
	ν_5	CF_3 d-stretch.....	1152 C	1152 VS	1160 W, dp	
	ν_6	CF_3 d-deform.....	507 C	507 M	508 VS, dp	

References

- [1] IR. H. J. Bernstein and G. Herzberg, J. Chem. Phys. 16, 30 (1948).
- [2] R. D. H. Rank, E. R. Shull, and E. L. Pace, J. Chem. Phys. 18, 885 (1950).
- [3] IR. E. K. Plyler and W. S. Benedict, J. Res. NBS 47, 202 (1951).

Molecule: Trichloromethane CHCl_3
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 89

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH stretch.....	3034 B	cm^{-1} (Gas)	3034.1 M	cm^{-1} (Gas)
	ν_2	CCl_3 s-stretch.....	680 B	680 S	672 S	
	ν_3	CCl_3 s-deform.....	363 C	366 (liquid)	363 M	
	ν_4	CH bend.....	1220 B	1219.7 VS	1217 W	
	ν_5	CCl_3 d-stretch.....	774 B	774.0 VS	760 W	
	ν_6	CCl_3 d-deform.....	261 B	260 (liquid)	261 W	

References

- [1] R. J. R. Nielsen and N. E. Ward, J. Chem. Phys. 10, 81 (1942).
- [2] IR.R. J. R. Madigan and F. F. Cleveland, J. Chem. Phys. 19, 119 (1951).
- [3] IR. T. G. Gibian and D. S. McKinney, J. Amer. Chem. Soc. 73, 1431 (1951).
- [4] IR. A. E. Stanevich and N. G. Yaroslavskii, Opt. Spectrosc. 9, 31 (1961).
- [5] IR. I. Suzuki, unpublished.

Molecule: Trichloromethane-d₁ CDCl₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 90

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD stretch.....	2266 C 659 B 369 C	<i>cm</i> ⁻¹ (Gas) 2266 W 658.5 S 366 W (liquid)	<i>cm</i> ⁻¹ (Liquid) 2255 (2) p 649 (7) p 369 (9) p	
	<i>v</i> ₂	CCl ₃ s-stretch.....				
	<i>v</i> ₃	CCl ₃ s-deform.....				
<i>e</i>	<i>v</i> ₄	CD bend.....	914 B 749 B 262 C	913.9 VS 748.5 VS 262 W (liquid)	908 (1) dp 735 (2) dp 262 (10) dp	
	<i>v</i> ₅	CCl ₃ d-stretch.....				
	<i>v</i> ₆	CCl ₃ d-deform.....				

References

- [1] R. J. P. Zietlow, F. F. Cleveland, and A. G. Meister, J. Chem. Phys. 18, 1076 (1950).
- [2] IR.R. V. R. Madigan, F. F. Cleveland, W. M. Boyer, and R. B. Bernstein, J. Chem. Phys. 18, 1081 (1950).
- [3] IR. R. B. Bernstein, A. G. Gordus, and F. F. Cleveland, J. Chem. Phys. 20, 1979 (1952).
- [4] IR. I. Suzuki, unpublished.

Molecule: Tribromomethane CHBr₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 91

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH stretch.....	3042 B 541 B 222 C	<i>cm</i> ⁻¹ (Gas) 3042 M 541 M 222	<i>cm</i> ⁻¹ (Liquid) 3017 (6) p 540 (4) p 222 (10) p	
	<i>v</i> ₂	CBr ₃ s-stretch.....				
	<i>v</i> ₃	CBr ₃ s-deform.....				
<i>e</i>	<i>v</i> ₄	CH bend.....	1149 B 669 B 155 C	1149 VS 669 VS 155	1143 (2) dp 655 (2) dp 155 (5) dp	
	<i>v</i> ₅	CBr ₃ d-stretch.....				
	<i>v</i> ₆	CBr d-deform.....				

References

- [1] IR.R. A. G. Meister, S. E. Rosser, and F. F. Cleveland, J. Chem. Phys. 18, 346 (1950).
- [2] IR. E. K. Plyler and W. S. Benedict, J. Res. NBS 47, 202 (1951).
- [3] IR. M. T. Forel, J. P. Leicknam, and M. L. Josien, J. Chim. Phys. 57, 1103 (1960).
- [4] IR. L. P. Lindsay and P. N. Schatz, Spectrochim. Acta 20, 1421 (1964).

Molecule: Tribromomethane-d₁ CDBr₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 92

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD stretch.....	2251 C	<i>cm</i> ⁻¹ (Liquid)	2247 (4)	
	<i>v</i> ₂	CBr ₃ s-stretch.....	521 C	521 M	519.3 (7)	
	<i>v</i> ₃	CBr ₃ s-deform.....	222 C	221.6 (10)	
	<i>v</i> ₄	CD bend.....	850 D	{ 858 VS 844 VS	{ 856.5 (3) 840 (3)	
	<i>v</i> ₅	CBr ₃ d-stretch.....	632 C	632 VS	628.5 (5)	FR (<i>v</i> ₃ + <i>v</i> ₅). }
	<i>v</i> ₆	CBr ₃ d-deform.....	153 C	153.4 (8)	

References

- [1] R. A. G. Meister, S. E. Rosser, and F. F. Cleveland, J. Chem. Phys. 18, 346 (1950).
- [2] IR. M. T. Forel, J. P. Leicknam, and M. L. Josien, J. Chim. Phys. 57, 1103 (1960).
- [3] IR. I. Suzuki, unpublished.

Molecule: Bromotrichloromethane CBrCl₃
 Symmetry C_{3v} Symmetry number δ = 3

No. 93

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CCl ₃ s-stretch.....	716 C	<i>cm</i> ⁻¹ (Liquid)	716.3 (2) p	
	<i>v</i> ₂	CBr stretch.....	422 C	420 W	422.3 (10) p	
	<i>v</i> ₃	CCl ₃ s-deform.....	247 C	247.3 (5) p	
	<i>v</i> ₄	CCl ₃ d-stretch.....	775 C	773 VS	775.3 (1) dp	
	<i>v</i> ₅	CBr bend.....	295 C	294 W	295.0 (3) dp	
	<i>v</i> ₆	CCl ₃ d-deform.....	193 C	193.3 (4) dp	

References

- [1] R. J. P. Zietlow, F. F. Cleveland, and A. G. Meister, J. Chem. Phys. 18, 1076 (1950).
- [2] IR. J. R. Madigan and F. F. Cleveland, J. Chem. Phys. 19, 119 (1951).
- [3] IR. E. K. Plyler and W. S. Benedict, J. Res. NBS 47, 202 (1951), RP 2245.

Molecule: Tribromochloromethane CBr_3Cl
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 94

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CCl stretch.....	747 C	cm^{-1} (CS_2 , C_7H_{14} soln.)	cm^{-1} (C_6H_6 , CCl_4 soln.)	
	ν_2	CBr_3 s-stretch	329 C	747 S (CS_2 soln.)	748 (1)	
	ν_3	CBr_3 s-deform.....	210 C	329 W (C_7H_{14} soln.)	326 (10) p	
	ν_4	CBr_3 d-stretch.....	675 C 675 S (CS_2 soln.)	210 (10) p 677 (4) dp	
	ν_5	CCl bend.....	211 E	
	ν_6	CBr_3 d-deform.....	141 C	141 (7) dp	CF [1].

References

- [1] IR.R. A. G. Meister, S. E. Rosser, and F. F. Cleveland, J. Chem. Phys. 18, 346 (1950).
- [2] IR. E. K. Plyler, W. H. Smith, and N. Acquista, J. Res. NBS 44, 503 (1950), RP2097.
- [3] R. M. L. Delwaille, M. B. Buisset, and M. Delhaye, J. Amer. Chem. Soc. 74, 5768 (1952).
- [4] R. R. H. Krupp, S. M. Ferogle, and A. Weber, J. Chem. Phys. 24, 355 (1956).

Molecule: Dichloromethane CH_2Cl_2
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 95

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_2 s-stretch.....	2999 B	cm^{-1} (Gas)	cm^{-1} (Gas)	
	ν_2	CH_2 scis.....	1467 C	2999 M 1467 W	2996 S, p 1430.1 W, p	
	ν_3	CCl_2 s-stretch.....	717 B	717 M	713 S, p	
	ν_4	CCl_2 scis.....	282 B	284 (liquid)	281.5 M, p	
a_2	ν_5	CH_2 twist.....	1153 B	^a ia	1153 VW	
	ν_6	CH_2 a-stretch.....	3040 B	3045 (liquid)	3040 S, dp	
b_2	ν_7	CH_2 rock.....	898 B	897.7 M	893 VW	
	ν_8	CH_2 wag.....	1268 B	1268 S	1265 (liquid)	
	ν_9	CCl_2 a-stretch.....	758 B	758 VS		

^a In the spectrum of liquid CH_2Cl_2 , a weak band is found at 1156 cm^{-1} , which may be assigned to ν_5 .

References

- [1] R. H. L. Welsh, M. F. Crawford, T. R. Thomas, and C. R. Love, Can. J. Phys. 30, 577 (1952).
- [2] IR. T. Shimanouchi and I. Suzuki, J. Mol. Spectrosc. 8, 222 (1962).
- [3] IR.R. F. E. Palma, E. A. Piotrowski, S. Sundaram, and F. F. Cleveland, J. Mol. Spectrosc. 13, 119 (1964).

Molecule: Dichloromethane-d₁ CHDCl₂
 Symmetry C_s Symmetry number δ = 1

No. 96

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH stretch.....	3024 B	<i>cm</i> ⁻¹ (Gas)	3024 (Liquid)	
	<i>v</i> ₂	CD stretch.....	2249 B	2249	2246 M, p	
	<i>v</i> ₃	CH bend.....	1282 B	1282	1276 VW	
	<i>v</i> ₄	CD bend.....	778 C	778 (liquid)	779 W, p	
	<i>v</i> ₅	CCl ₂ s-stretch.....	692 B	692	682 S, p	
	<i>v</i> ₆	CCl ₂ scis.....	283 B	283 M, p	
	<i>v</i> ₇	CH bend.....	1223 A	1222.9	1221 VW	
	<i>v</i> ₈	CD bend.....	890 A	889.8	886 VW	
	<i>v</i> ₉	CCl ₂ a-stretch.....	738 B	738	725 W, dp	

References

- [1] IR. T. Shimanouchi and I. Suzuki, J. Mol. Spectrosc. 8, 222 (1962).
 [2] IR.R. F. E. Palma, E. A. Piotrowski, S. Sundaram, and F. F. Cleveland, J. Mol. Spectrosc. 13, 119 (1964).

Molecule: Dichloromethane-d₂ CD₂Cl₂
 Symmetry C_{2v} Symmetry number δ = 2

No. 97

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD ₂ s-stretch.....	2205 B	<i>cm</i> ⁻¹ (Gas)	2205 W	
	<i>v</i> ₂	CD ₂ scis.....	1052 D	2198 M, p	
	<i>v</i> ₃	CCl ₂ s-stretch.....	687 B	687 M	1052 VW, p	
	<i>v</i> ₄	CCl ₂ scis.....	282 C	677 VS, p	
	<i>a</i> ₂	CD ₂ twist.....	826 C	ia	282 S, p	
	<i>b</i> ₁	CD ₂ a-stretch.....	2304 C	2304 (liquid)	826 VW	
	<i>b</i> ₂	CD ₂ rock.....	^a 712 D	2304 VW	OV (<i>v</i> ₉).
	<i>v</i> ₈	CD ₂ wag.....	957 B	957 VS	716 W	
	<i>v</i> ₉	CCl ₂ a-stretch.....	727 B	727 VS		

^a Calculated from product rule [1].

References

- [1] IR. T. Shimanouchi and I. Suzuki, J. Mol. Spectrosc. 8, 222 (1962).
 [2] IR.R. F. E. Palma, E. A. Piotrowski, S. Sundaram, and F. F. Cleveland, J. Mol. Spectrosc. 13, 119 (1964).

Molecule: Dibromomethane CH_2Br_2
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 98

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_2 s-stretch.....	3009 C	cm^{-1} (Gas)	3009 W	3008 (1)
	ν_2	CH_2 scis.....	1382 C	1382 VW	1402 (0)	
	ν_3	CBr_2 s-stretch.....	588 C	588 M	584 (10)	
	ν_4	CBr_2 scis.....	169 C	169 (10)	
	ν_5	CH_2 twist.....	1095 D	ia	^a 1095	
	ν_6	CH_2 a-stretch.....	3073 B	3073 VW	^a 3064	
	ν_7	CH_2 rock.....	812 B	812 M	^a 813	
	ν_8	CH_2 wag.....	1195 B	1195 VS	^a 1194	
	ν_9	CBr a-stretch.....	653 B	653 VS	640 (0)	

^a Liquid.

References

- [1] R. J. Wagner, Z. Phys. Chem. **B45**, 69 (1939).
- [2] R. M. L. Delwaille and F. Francois, J. Phys. Radium **7**, 15 (1946).
- [3] IR. E. K. Plyler, W. A. Smith, and N. Acquista, J. Res. NBS **44**, 503 (1950) RP2097.
- [4] IR.R. R. S. Dennen, E. A. Piotrowski, and F. F. Cleveland, J. Chem. Phys. **49**, 4385 (1968).

Molecule: Dibromomethane-d₁ CHDBr_2
 Symmetry C_s Symmetry number $\delta = 1$

No. 99

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH stretch.....	3040 C	cm^{-1} (Gas)	3040 W	3028 (Liquid)
	ν_2	CD stretch.....	2249 D	2249 W (liquid)	2245 W, p	
	ν_3	CH bend.....	1220 C	701 W	1239 W, p	
	ν_4	CD bend.....	701 C	1220 W	702 M, p	
	ν_5	CBr_2 s-stretch.....	565 C	565 VW	561 S, p	
	ν_6	CBr_2 scis.....	172 D	172 VS, p	
	ν_7	CH bend.....	1154 B	1154 VS	835 VW, dp	
	ν_8	CD bend.....	838 B	838 VS	623 W, dp	
	ν_9	CBr_2 a-stretch.....	632 B	632 VS		

References

- [1] IR.R. R. S. Dennen, E. A. Piotrowski, and F. F. Cleveland, J. Chem. Phys. **49**, 4385 (1968).

Molecule: Dibromomethane-d₂ CD₂Br₂
 Symmetry C_{2v} Symmetry number δ = 2

No. 100

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD ₂ s-stretch.....	2214 C	cm ⁻¹ (Gas)	2195 M, p	
	<i>v</i> ₂	CD ₂ scis.....	1026 D	2214 W 1026 (liquid)	1028 W, p	
	<i>v</i> ₃	CBr ₂ s-stretch.....	559 C	559 M	551 S, p	
	<i>v</i> ₄	CBr ₂ scis.....	172 D	172 VS, p	
<i>a</i> ₂	<i>v</i> ₅	CD ₂ twist.....	782 D	ia	782 W, p	
<i>b</i> ₁	<i>v</i> ₆	CD ₂ a-stretch.....	2324 C	2324 W	2313 VW, dp	
	<i>v</i> ₇	CD ₂ rock.....	625 B	625 VS	636 VW	
<i>b</i> ₂	<i>v</i> ₈	CD ₂ wag.....	907 B	907 VS	902 W, dp	
	<i>v</i> ₉	CBr ₂ a-stretch.....	608 C	608 (liquid)	612 M, dp	

References

- [1] R. B. Trumpy, Z. Phys. 100, 250 (1936).
 [2] IR.R. R. S. Dennen, E. A. Piotrowski, and F. F. Cleveland, J. Chem. Phys. 49, 4385 (1968).

Molecule: Dibromodichloromethane CBr₂Cl₂
 Symmetry C_{2v} Symmetry number δ = 2

No. 101

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CCl ₂ s-stretch.....	733 C	cm ⁻¹ (Liquid)	734 (1) p	
	<i>v</i> ₂	CBr ₂ s-stretch.....	380 C	377 W	380 (10) p	
	<i>v</i> ₃	CCl ₂ scis.....	242 C	242 (6) p	
	<i>v</i> ₄	CBr ₂ scis.....	154 C	154 (4) p	
<i>a</i> ₂	<i>v</i> ₅	CCl ₂ twist.....	175 C	ia	175 (2) dp	
<i>b</i> ₁	<i>v</i> ₆	CBr ₂ a-stretch.....	683 C	683 VS	684 (3) dp	
	<i>v</i> ₇	CCl ₂ wag.....	229 C	229 (2) dp	
<i>b</i> ₂	<i>v</i> ₈	CCl ₂ a-stretch.....	768 C	768 VS	771 (0) dp	
	<i>v</i> ₉	CCl ₂ rock.....	262 C	262 (1) dp	

References

- [1] IR. E. K. Plyler and W. S. Benedict, J. Res. NBS 47, 202 (1951), RP 2245.
 [2] IR.R. A. Davis, F. F. Cleveland, and A. G. Meister, J. Chem. Phys. 20, 454 (1952).

Molecule: Bromochloromethane CH_2BrCl
 Symmetry C_s Symmetry number $\delta = 1$

No. 102

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_2 s-stretch	3003 A	cm^{-1} (Gas)	2986 M, p	
	ν_2	CH_2 scis	1482 E	^a 1482 M	1410 M, p	
	ν_3	CH_2 wag	1231 B	1231 S	1229 W, p	
	ν_4	CCl stretch	744 B	744 VS	731 M, p	
	ν_5	CBr stretch	614 B	614 S	606 S, p	
	ν_6	CBrCl scis	229 C	229 S, p	
	ν_7	CH_2 a-stretch	3066 B	3066 W	3055 M, dp	
	ν_8	CH_2 twist	1128 C	1128 W (liquid)	1130 W	
	ν_9	CH_2 rock	852 B	852 W	848 W	

^a The corresponding frequency in the liquid state is found at 1407 cm^{-1} . This band may be assigned to the overtone of the CCl stretching vibration.

References

- [1] IR. E. K. Plyler, W. A. Smith, and N. Acquista, J. Res. NBS 44, 503 (1950), RP2097.
- [2] IR.R. A. Weber, A. G. Meister, and F. F. Cleveland, J. Chem. Phys. 21, 930 (1953).
- [3] IR.R. A. N. Tanaka, K. V. Narasimham, A. G. Meister, J. M. Dowling, F. F. Cleveland, S. Sundaram, E. A. Piotrowski, R. B. Bernstein, and S. I. Miller, J. Mol. Spectrosc. 15, 319 (1965).
- [4] IR. I. Suzuki, unpublished.

Molecule: Bromochloromethane-d₁ CHDBrCl
 Symmetry C_1 Symmetry number $\delta = 1$

No. 103

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a	ν_1	CH stretch	3031 C	cm^{-1} (Gas)	3024 S, p	
	ν_2	CD stretch	2252 C	2252 M (liquid)	2246 S, p	
	ν_3	CH bend	1262 C	1262 S (liquid)	1264 W, p	
	ν_4	CH bend	1188 B	1188 M	1179 W, p	
	ν_5	CD bend	868 B	868 M	867 W	
	ν_6	CD bend	746 B	746 W	743 VW	
	ν_7	CCl stretch	711 B	711 S	707 M, p	
	ν_8	CBr stretch	607 C	607 W	586 S, p	
	ν_9	CBrCl scis	228 C	228 S, p	

Reference

- [1] IR.R. A. N. Tanaka, K. V. Narasimham, A. G. Meister, J. M. Dowling, F. F. Cleveland, S. Sundaram, E. A. Piotrowski, R. B. Bernstein, and S. I. Miller, J. Mol. Spectrosc. 15, 319 (1965).

Molecule: Bromochloromethane-d₂ CD₂BrCl
 Symmetry C_s Symmetry number δ = 1

No. 104

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CD ₂ s-stretch.....	2208 B	<i>cm</i> ⁻¹ (Gas)	2208 S	<i>cm</i> ⁻¹ (Liquid)
	<i>v</i> ₂	CD ₂ scis.....	1050 B	1050 W	1042 M, p	
	<i>v</i> ₃	CD ₂ wag.....	936 B	936 S	922 W, p	
	<i>v</i> ₄	CCl stretch.....	717 B	717 S	702 M, p	
	<i>v</i> ₅	CBr stretch.....	582 B	582 S	574 S, p	
	<i>v</i> ₆	CBrCl scis.....	226 C	226 S, p	
	<i>v</i> ₇	CD ₂ a-stretch.....	2305 C	2302 S (liquid)	2305 W, dp	
	<i>v</i> ₈	CD ₂ twist.....	811 B	811 W	809 W, dp	
	<i>v</i> ₉	CD ₂ rock.....	667 C	668 W (liquid)	667 W, dp	

Reference

See No. 103.

Molecule: Formic acid HCOOH
 Symmetry C_s Symmetry number δ = 1

No. 105

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	OH stretch.....	3570 D	<i>cm</i> ⁻¹ (Gas)	3570 M	<i>cm</i> ⁻¹
	<i>v</i> ₂	CH stretch.....	2943 C	2942.8 M		
	<i>v</i> ₃	C=O stretch.....	1770 C	1770 VS		
	<i>v</i> ₄	CH bend.....	1387 C	1387 VW		
	<i>v</i> ₅	OH bend.....	1229 C	1229 W		
	<i>v</i> ₆	C—O stretch.....	1105 C	1105.3 S		
	<i>v</i> ₇	OCO deform.....	625 C	625 M		
	<i>v</i> ₈	CH bend.....	1033 C	1033 W		
	<i>v</i> ₉	Torsion.....	638 C	638 S		

References

- [1] IR. V. Z. Williams, J. Chem. Phys. 15, 232, 243 (1947).
- [2] IR. L. M. Sverdlov, Dokl. Akad. Nauk SSSR 91, 503 (1953).
- [3] IR. W. J. Orville-Thomas, Research 9, S15 (1956).
- [4] IR. J. K. Wilmshurst, J. Chem. Phys. 25, 478 (1956).
- [5] IR.Th. R. C. Millikan and K. S. Pitzer, J. Chem. Phys. 27, 1305 (1957).
- [6] IR.Th. T. Miyazawa and K. S. Pitzer, J. Chem. Phys. 30, 1076 (1959).
- [7] Th. K. Nakamoto and S. Kishida, J. Chem. Phys. 41, 1554 (1964).

Molecule: Formic acid-d₂ DCOOD
 Symmetry C_s Symmetry number δ = 1

No. 106

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	OD stretch.....	2632 C	2632 cm ⁻¹ (Gas)		
	<i>v</i> ₂	CD stretch.....	2232 C	2231.8 M		
	<i>v</i> ₃	C=O stretch.....	1742 C	1742 VS		
	<i>v</i> ₄	CD bend.....	945 C	945 M		
	<i>v</i> ₅	OD bend.....	1040 C	1040 W		
	<i>v</i> ₆	C—O stretch.....	1171 C	1171.3 S		
	<i>v</i> ₇	OCO deform.....	558 C	558 W		
	<i>v</i> ₈	CD bend.....	873 C	873 W		
	<i>v</i> ₉	Torsion.....	491 C	491 W		

References

See No. 105.

Molecule: Methanol CH₃OH (gas)
 Symmetry C_s Symmetry number δ = 1

No. 107

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	OH stretch.....	3681 A	3681 M		
	<i>v</i> ₂	CH ₃ d-stretch.....	3000 C	3000 M		
	<i>v</i> ₃	CH ₃ s-stretch.....	2844 A	2844 S		
	<i>v</i> ₄	CH ₃ d-deform.....	1477 B	1477 M		OV (<i>v</i> ₁₀)
	<i>v</i> ₅	CH ₃ s-deform.....	1455 A	1455 M		
	<i>v</i> ₆	OH bend.....	1345 B	1345 S		
	<i>v</i> ₇	CH ₃ rock.....	1060 D	1060 W		OV (<i>v</i> ₈)
	<i>v</i> ₈	CO stretch.....	1033 A	1033 VS	1032 (2)	
	<i>v</i> ₉	CH ₃ d-stretch.....	2960 C	2960 S	2955 (4)	
	<i>v</i> ₁₀	CH ₃ d-deform.....	1477 B	1477 M		OV (<i>v</i> ₄)
	<i>v</i> ₁₁	CH ₃ rock.....	1165 C		1165 (1) (liquid)	
	<i>v</i> ₁₂	Torsion.....	{ 295 (A) 200 (E) }	80~300		{ MW: ^a 295 (A) 200 (E) }

^a The value of *v*₁₂ is undefined because of the large coupling between internal and overall rotations. The MW values quoted are the calculated separations between the lowest rotational levels (J = K = 0) of the ground and first excited torsional states [2, 5].

References

- [1] R. J. R. Nielsen and N. E. Ward, J. Chem. Phys. 10, 81 (1942).
- [2] IR. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [3] IR. C. Tanaka, K. Kuratani, and S. Mizushima, Spectrochim. Acta 9, 265 (1957).
- [4] IR. M. Van Thiel, E. D. Becker, and G. C. Pimentel, J. Chem. Phys. 27, 95 (1957).
- [5] IR.Th. D. G. Burkhard and D. M. Dennison, J. Mol. Spectrosc. 3, 299 (1959).
- [6] Th. M. Margottin-Maclou, J. Phys. Radium 21, 634 (1960).
- [7] IR. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961) and references cited there.
- [8] Th. G. Zerbi, J. Overend, and B. Crawford, J. Chem. Phys. 38, 122 (1963).
- [9] IR.Th. C. Tanaka and T. Shimanouchi, unpublished.

Molecule: Methanol CH_3OH (liquid)
 Symmetry C_s Symmetry number $\delta = 1$

No. 108

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	OH stretch	3328 D	cm^{-1} (Liquid)	cm^{-1} (Liquid)	
	ν_2	CH_3 d-stretch	2980 C	2980 M	2993 (3)	
	ν_3	CH_3 s-stretch	2834 C	2834 S	2834 (10)	
	ν_4	CH_3 d-deform	1480 C	1480 M	1464 (5b)	OV (ν_{10}).
	ν_5	CH_3 s-deform	1450 C	1450 M		
	ν_6	OH bend	1418 C	1418 M, b		
	ν_7	CH_3 rock	1115 C	1115 M	1107 (2)	
	ν_8	CO stretch	1030 C	1030 VS	1033 (6)	
	ν_9	CH_3 d-stretch	2946 C	2946 S	2940 (9)	
	ν_{10}	CH_3 d-deform	1480 C	1480 M	1464 (5b)	OV (ν_4).
	ν_{11}	CH_3 rock	1165 C		1165 (1)	
	ν_{12}	Torsion	655 D	655 vb		

References

- [1] R. S. Mizushima, Y. Morino, and G. Okamoto, Bull. Chem. Soc. Japan 11, 698 (1936).
- [2] R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules, (Van Nostrand, New York, 1945).
- [3] IR.R. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961), and references cited there.
- [4] IR.Th. C. Tanaka and T. Shimanouchi, unpublished.

Molecule: Methanol-d₁ CH_3OD (gas)
 Symmetry C_s Symmetry number $\delta = 1$

No. 109

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	OD stretch	2718 A	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CH_3 d-stretch	3000 C	3000 M		SF (ν_2 of CH_3OH).
	ν_3	CH_3 s-stretch	2843 A	2843 S		
	ν_4	CH_3 d-deform	1473 B	1473 M		OV (ν_{10}).
	ν_5	CH_3 s-deform	1456 A	1456 M		
	ν_6	OD bend	864 A	864 S		
	ν_7	CH_3 rock	1230 B	1230 W		
	ν_8	CO stretch	1040 A	1040 VS		
	ν_9	CH_3 d-stretch	2960 C	2960 S		SF (ν_9 of CH_3OH). OV (ν_4).
	ν_{10}	CH_3 d-deform	1473 B	1473 M		
	ν_{11}	CH_3 rock	1160 C	1160 VW		
	ν_{12}	Torsion	213 E			CF [5, 6].

References

- [1] IR. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR. C. Tanaka, K. Kuratani, and S. Mizushima, Spectrochim. Acta 9, 265 (1957).
- [3] Th. M. Margottin-Maclou, J. Phys. Radium 21, 634 (1960).
- [4] IR. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961) and references cited there.
- [5] Th. G. Zerbi, J. Overend, and B. Crawford, Jr., J. Chem. Phys. 38, 122 (1963).
- [6] IR.Th. C. Tanaka and T. Shimanouchi, unpublished.

Molecule: Methanol-d₁ CH₃OD (liquid)
 Symmetry C_s Symmetry number δ = 1

No. 110

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	OD stretch	2467 D	cm^{-1} (Liquid)	cm^{-1} (Liquid)	
	ν_2	CH ₃ d-stretch	2978 M	2978 M	2992 (3)	
	ν_3	CH ₃ s-stretch	2838 C	2838 S	2834 (10)	
	ν_4	CH ₃ d-deform	1469 C	1469 M	1463 (5b)	OV (ν_{10}).
	ν_5	CH ₃ s-deform	1449 C	1449 M		
	ν_6	OD bend	940 C	940 M, b	955 (1)	
	ν_7	CH ₃ rock	1231 C	1231 W	1226 (0)	
	ν_8	CO stretch	1038 C	1038 VS	1029 (6)	
	ν_9	CH ₃ d-stretch	2951 C	2951 S	2943 (9)	
	ν_{10}	CH ₃ d-deform	1469 C	1469 M	1463 (5b)	OV (ν_4).
	ν_{11}	CH ₃ rock	1163 C	1163 (1)	
	ν_{12}	Torsion	475 D	475 vb		

References

- [1] R. S. Mizushima, Y. Morino, and G. Okamoto, Bull. Chem. Soc. Japan 11, 698 (1936).
- [2] R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [3] IR.R. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961), and references cited there.
- [4] IR.Th. C. Tanaka and T. Shimanouchi, unpublished.

Molecule: Methanol-d₃ CD₃OH (gas)
 Symmetry C_s Symmetry number δ = 1

No. 111

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	OH stretch	3690 D	cm^{-1} (Gas)	3690 S	
	ν_2	CD ₃ d-stretch	2260 E	2260 M, sh		
	ν_3	CD ₃ s-stretch	2077 C	2077 S		
	ν_4	CD ₃ d-deform	1047 D	1047 W		
	ν_5	CD ₃ s-deform	1134 C	1134 VS		
	ν_6	OH bend	1297 C	1297 VS		
	ν_7	CD ₃ rock	858 C	858 M		
	ν_8	CO stretch	988 C	988 VS		
	ν_9	CD ₃ d-stretch	2235 D	2235 S		
	ν_{10}	CD ₃ d-deform	1075 C	1075 W		
	ν_{11}	CD ₃ rock	877 D	877 M		
	ν_{12}	Torsion	256 E			CF [1, 3].

References

- [1] Th. M. Margottin-Maclou, J. Phys. Radium 21, 634 (1960).
- [2] IR. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961), and references cited there.
- [3] Th. G. Zerbi, J. Overend, and B. Crawford, Jr., J. Chem. Phys. 38, 122 (1963).
- [4] Th. C. Tanaka and T. Shimanouchi, unpublished.

Molecule: Methanol-d₃ CD₃OH (liquid)
 Symmetry C_s Symmetry number $\delta = 1$

No. 112

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	OH stretch.....	3310 D	cm^{-1} (Liquid)	cm^{-1} (Liquid)	
	ν_2	CD ₃ d-stretch.....	2235 D	3310 S, vb	3350 W, vb	
	ν_3	CD ₃ s-stretch.....	2078 C	^a 2235 M	2230 M, dp	
	ν_4	CD ₃ d-deform.....	1069 C	2078 S	2074 VS, p	
	ν_5	CD ₃ s-deform.....	1122 C	1069 W	1072 M, dp	OV (ν_{10}).
	ν_6	OH bend.....	1391 C	1122 VS	1127 M, p	
	ν_7	CD ₃ rock.....	882 C	1391 S, b	1360 VW, vb	
	ν_8	CO stretch.....	982 C	882 M	894 M, dp	OV (ν_{11}).
	ν_9	CD ₃ d-stretch.....	2213 D	^a 2213 M	2213 VW	
	ν_{10}	CD ₃ d-deform.....	1069 C	1069 W	1072 M, dp	OV (ν_4).
	ν_{11}	CD ₃ rock.....	882 D	882 M	894 M, dp	OV (ν_7).
	ν_{12}	Torsion.....	665 D	665 S, vb		

^a The value obtained in the vitreous solid (-180 °C).

References

- [1] IR.R. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961), and references cited there.
 [2] Th. C. Tanaka and T. Shimanouchi, unpublished.

Molecule: Methanol-d₄ CD₃OD (gas)
 Symmetry C_s Symmetry number $\delta = 1$

No. 113

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	OD stretch.....	2724 D	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CD ₃ d-stretch.....	2260 E	2724 S	
	ν_3	CD ₃ s-stretch.....	2080 C	2260	SF (ν_2 of CD ₃ OH).
	ν_4	CD ₃ d-deform.....	1024 D	2080 S	
	ν_5	CD ₃ s-deform.....	1135 C	1024 W	
	ν_6	OD bend.....	1060 D	1135 VS	
	ν_7	CD ₃ rock.....	776 C	1060 W	
	ν_8	CO stretch.....	983 C	776 S	
	ν_9	CD ₃ d-stretch.....	2228 D	983 VS	
	ν_{10}	CD ₃ d-deform.....	1080 C	2228 S	
	ν_{11}	CD ₃ rock.....	892 C	1080 W	
	ν_{12}	Torsion.....	196 E	892 W	CF [1, 3].

References

- [1] Th. M. Margottin-Maclou, J. Phys. Radium 21, 634 (1960).
 [2] IR. M. Falk and E. Whalley, J. Chem. Phys. 34, 1554 (1961).
 [3] Th. C. Tanaka and T. Shimanouchi, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	NH_2 s-stretch.....	3361 B	cm^{-1} (Gas)	3361 W	
	ν_2	CH_3 d-stretch.....	2961 B		2961 VS	
	ν_3	CH_3 s-stretch.....	2820 B		2820 VS	
	ν_4	NH_2 scis.....	1623 B		1623 S	
	ν_5	CH_3 d-deform.....	1473 B		1473 S	1460 M
	ν_6	CH_3 s-deform.....	1430 B		1430 M	
	ν_7	CH_3 rock.....	1130 A		1130 M	
	ν_8	CN stretch.....	1044 A		1044 S	
	ν_9	NH_2 wag.....	780 A		780 VS	781 W
	ν_{10}	NH_2 a-stretch.....	3427 C		3427 W	3470 W
	ν_{11}	CH_3 d-stretch.....	2985 C		2985 VS	
	ν_{12}	CH_3 d-deform.....	1485 D		^a 1485	
	ν_{13}	NH_2 twist.....	1419 D		
	ν_{14}	CH_3 rock.....	1195 D		^a 1195	
a''	ν_{15}	Torsion.....	268 B		268	

^a Estimated from ${}^R\text{Q}$ branch frequency.

References

- [1] R. J. S. Kirby-Smith and L. G. Booner, J. Chem. Phys. 7, 880 (1939).
- [2] IR. A. P. Gray and R. C. Lord, J. Chem. Phys. 26, 690 (1957).
- [3] IR. M. Tsuboi, A. Y. Hirakawa, T. Ino, T. Sasaki, and K. Tamagake, J. Chem. Phys. 41, 2721 (1964).
- [4] IR. M. Tsuboi, A. Y. Hirakawa, and K. Tamagake, J. Mol. Spectrosc. 22, 272 (1967).
- [5] Th. A. Y. Hirakawa, unpublished.
- [6] IR.Th. K. Tamagake, M. Tsuboi, and A. Y. Hirakawa, J. Chem. Phys. 48, 5536 (1968).
- [7] IR.R. J. R. Durig, S. F. Bush, and F. G. Baglin, J. Chem. Phys. 49, 2106 (1968).
- [8] MW. T. Itoh, J. Phys. Soc. Japan 11, 264 (1956).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	ND ₂ s-stretch.....	2479 B	cm^{-1} (Gas)	cm^{-1} (Gas)	
	ν_2	CH ₃ d-stretch.....	2961 B	2961 VS	2969 M	
	ν_3	CH ₃ s-stretch.....	2817 B	2817 S	2824 M	
	ν_4	ND ₂ scis.....	1234 B	1234 S	1214 M	
	ν_5	CH ₃ d-deform.....	1468 B	1468 S	1473 M	
	ν_6	CH ₃ s-deform.....	1430 B	1430 M		
	ν_7	CH ₃ rock.....	1117 A	1117 S		
	ν_8	CN stretch.....	997 A	997 S	995 S	
	ν_9	ND ₂ wag.....	625 A	625 VS		
	ν_{10}	ND ₂ a-stretch.....	2556 B	2556 M	2527 M	
	ν_{11}	CH ₃ d-stretch.....	2985 C	2985 VS		
	ν_{12}	CH ₃ d-deform.....	1485 D	^a 1485		
	ν_{13}	ND ₂ twist.....	1058 E		
	ν_{14}	CH ₃ rock.....	1187 C	1187 M		
	ν_{15}	Torsion.....	228 C	228 S		

^a Estimated from ^RQ branch frequency.

References

- [1] R. J. T. Edsall and H. Schinberg, J. Chem. Phys. **8**, 520 (1940).
- [2] IR. A. P. Gray and R. C. Lord, J. Chem. Phys. **26**, 690 (1957).
- [3] IR. M. Tsuboi, A. Y. Hirakawa, T. Sasaki, and K. Tamagake, J. Chem. Phys. **41**, 2721 (1964).
- [4] IR. M. Tsuboi, A. Y. Hirakawa, and K. Tamagake, J. Mol. Spectrosc. **22**, 272 (1967).
- [5] Th. A. Y. Hirakawa, unpublished.
- [6] IR.Th. K. Tamagake, M. Tsuboi, and A. Y. Hirakawa, J. Chem. Phys. **48**, 5536 (1968).
- [7] IR.R. J. R. Durig, S. F. Bush, and F. G. Baglin, J. Chem. Phys. **49**, 2106 (1968).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	NH ₂ s-stretch.....	3361 B	3361 W cm ⁻¹ (Gas)		
	<i>v</i> ₂	CD ₃ d-stretch.....	2203 B	2203 VS		
	<i>v</i> ₃	CD ₃ s-stretch.....	2077 A	2077 VS		
	<i>v</i> ₄	NH ₂ scis.....	1624 B	1624 S		
	<i>v</i> ₅	CD ₃ d-deform.....	1065 D		CF [3].
	<i>v</i> ₆	CD ₃ s-deform.....	1142 A	1142 S		
	<i>v</i> ₇	CD ₃ rock.....	913 A	913 S		
	<i>v</i> ₈	CN stretch.....	973 B	973 M		
	<i>v</i> ₉	NH ₂ wag.....	740 A	740 VS		
	<i>v</i> ₁₀	NH ₂ a-stretch.....	3427 C	3427 W		
	<i>v</i> ₁₁	CD ₃ d-stretch.....	2236 C	2236 VS		
	<i>v</i> ₁₂	CD ₃ d-deform.....	1077 C	1077 W		
	<i>v</i> ₁₃	NH ₂ twist.....	1416 C	1416 W		
	<i>v</i> ₁₄	CD ₃ rock.....	926 D		CF [3].
	<i>v</i> ₁₅	Torsion.....	247 D		CF [3].

References

- [1] IR. A. P. Gray and R. C. Lord, J. Chem. Phys. 26, 690 (1957).
- [2] IR. M. Tsuboi, A. Y. Hirakawa, and K. Tamagake, J. Mol. Spectrosc. 22, 272 (1967).
- [3] Th. A Y. Hirakawa, unpublished.
- [4] IR.R. J. R. Durig, S. F. Bush, and F. G. Baglin, J. Chem. Phys. 49, 2106 (1968).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>ν</i> ₁	ND ₂ s-stretch.....	2477 B	2477 W	<i>cm</i> ⁻¹ (Gas)	
	<i>ν</i> ₂	CD ₃ d-stretch.....	2202 B	2202 VS		
	<i>ν</i> ₃	CD ₃ s-stretch.....	2073 B	2073 VS		
	<i>ν</i> ₄	ND ₂ scis.....	1227 B	1227 S		
	<i>ν</i> ₅	CD ₃ d-deform.....	1065 D		CF [2.]
	<i>ν</i> ₆	CD ₃ s-deform.....	1123 B	1123 M		
	<i>ν</i> ₇	CD ₃ rock.....	880 B	880 M		
	<i>ν</i> ₈	CN stretch.....	942 A	942 S		
	<i>ν</i> ₉	ND ₂ wag.....	601 A	601 VS		
	<i>ν</i> ₁₀	ND ₂ a-stretch.....	2556 C	2556 W		
	<i>ν</i> ₁₁	CD ₃ d-stretch.....	2238 C	2238 VS		
	<i>ν</i> ₁₂	CD ₃ d-deform.....	1077 C	1077 W		
	<i>ν</i> ₁₃	ND ₂ twist.....	1072 D		CF [2.]
	<i>ν</i> ₁₄	CD ₃ rock.....	910 B	910 M		CF [2].
	<i>ν</i> ₁₅	Torsion.....	201 C		MW: 200 (A). 203 (E).

References

- [1] IR. A. P. Gray and R. C. Lord, J. Chem. Phys. 26, 690 (1957).
- [2] Th. A. Y. Hirakawa, unpublished.
- [3] IR.R. J. R. Durig, S. F. Bush, and F. G. Baglin, J. Chem. Phys. 49, 2106 (1968).
- [4] MW. D. R. Lide, J. Chem. Phys. 27, 343 (1957).

Molecule: Acetylene CHCH
 Symmetry $D_{\infty h}$ Symmetry number $\delta = 2$

No. 118

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ_g^+	ν_1	CH stretch.....	3374 C	cm^{-1} (Gas) ia	3373.7 S	
	ν_2	CC stretch.....	1974 C	ia	1973.8 VS	
σ_u^+	ν_3	CH stretch.....	3289 B	{ 3294.9 S 3281.9 VS }	ia	FR ($\nu_2 + \nu_4 + \nu_5$).
π_g	ν_4	CH bend.....	612 C	ia	611.8 VW	
π_u	ν_5	CH bend.....	730 A	730.3 VS	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. H. C. Allen, Jr., E. D. Tidwell, and E. K. Plyler, J. Res. NBS 57, 213 (1956).
- [3] IR. T. A. Wiggins, E. K. Plyler, and E. D. Tidwell, J. Opt. Soc. Amer. 51, 1219 (1961).
- [4] IR. E. K. Plyler, E. D. Tidwell, and T. A. Wiggins, J. Opt. Soc. Amer. 53, 589 (1963).
- [5] IR. W. J. Lafferty and R. J. Thibault, J. Mol. Spectrosc. 14, 79 (1964).
- [6] IR. J. F. Scott and K. N. Rao, J. Mol. Spectrosc. 16, 15 (1965).
- [7] IR. J. F. Scott and K. N. Rao, J. Mol. Spectrosc. 18, 152 (1965).
- [8] IR. J. F. Scott and K. N. Rao, J. Mol. Spectrosc. 18, 451 (1965).
- [9] IR. J. F. Scott and K. N. Rao, J. Mol. Spectrosc. 20, 438 (1966).

Molecule: Acetylene-d₁ CHCD
 Symmetry $C_{\infty v}$ Symmetry number $\delta = 1$

No. 119

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CH stretch.....	3336 A	cm^{-1} (Gas) 3335.6 S	3335 S	
	ν_2	CC stretch.....	1854 A	1853.8 M	1851 S	
	ν_3	CD stretch.....	2584 A	2583.6 S		
π	ν_4	CH bend.....	518 A	518.38 S		RP [4].
	ν_5	CD bend.....	678 A	677.8 S		RP [4].

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomoc Molecules (Van Nostrand, New York, 1945).
- [2] IR. J. Overend and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 234, 306 (1955).
- [3] IR. H. C. Allen, Jr., E. D. Tidwell, and E. K. Plyler, J. Amer. Chem. Soc. 78, 3034 (1956).
- [4] IR. W. J. Lafferty, E. K. Plyler, and E. D. Tidwell, J. Chem. Phys. 37, 1981 (1962).

Molecule: Acetylene-d₂ CD₂
 Symmetry D_{∞h} Symmetry number δ = 2

No. 120

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ _g ⁺	ν ₁	CD stretch.....	2701 C	cm ⁻¹ (Gas) ia	cm ⁻¹ (Gas) 2700.5 S	
	ν ₂	CC stretch.....	1762 C	ia	1762.4 S	
σ _u ⁺	ν ₃	CD stretch.....	2439 A	2439.24 S	ia	
π _g	ν ₄	CD bend.....	505 C	ia	
π _u	ν ₅	CD bend.....	537 A	536.9 VS	ia	OC (ν ₄ + ν ₅) [1].

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR. J. Overend and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 232, 291 (1955).
- [3] IR. H. C. Allen, Jr., L. R. Blaine, and E. K. Plyler, J. Res. NBS 56, 279 (1956).
- [4] IR. E. D. Tidwell and E. K. Plyler, J. Opt. Soc. Amer. 52, 656 (1962).
- [5] IR. S. Ghergesetti and K. N. Rao, J. Mol. Spectrosc. 28, 27 (1968).

Molecule: Fluoroacetylene CHCF
 Symmetry C_{∞v} Symmetry number δ = 1

No. 121

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ ⁺	ν ₁	CH stretch.....	3355 B	cm ⁻¹ (Gas) 3355 VS	cm ⁻¹	
	ν ₂	CC stretch.....	2255 B	2255 VS		
	ν ₃	CF stretch.....	1055 B	1055 VS		
π	ν ₄	CH bend.....	578 B	578 VS		
	ν ₅	CCF bend.....	367 B	367 M		

References

- [1] IR. W. J. Middleton and W. H. Sharkey, J. Amer. Chem. Soc. 81, 803 (1959).
- [2] IR. W. S. Richardson and J. H. Goldstein, J. Chem. Phys. 18, 1314 (1960).
- [3] IR. G. R. Hund and M. K. Wilson, J. Chem. Phys. 34, 1301 (1961).

Molecule: Chloroacetylene CH₂CCl
 Symmetry C_{∞v} Symmetry number δ = 1

No. 122

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CH stretch.....	3340 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CC stretch.....	2110 B	3340 VS 2110 VS		
	ν_3	CCl stretch.....	756 B	756 VS		
	ν_4	CH bend.....	604 B	604 S		
	ν_5	CCCl bend.....	326 B	326 W		

References

See No. 121.

Molecule: Bromoacetylene CHCBr
 Symmetry C_{∞v} Symmetry number δ = 1

No. 123

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ^+	ν_1	CH stretch.....	3325 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CC stretch.....	2085 B	3325 VS 2085 VS		
	ν_3	CBr stretch.....	618 C	618 VS		
	ν_4	CH bend.....	618 C	618 VS		SF (ν_4).
	ν_5	CCBr bend.....	295 B	295 W		SF (ν_3).

References

See No. 121.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CH_2 s-stretch	3026 B	cm^{-1} (Gas) ia	cm^{-1} (Gas) 3026.4 (10)p	FR ($2\nu_{10}$). OC ($\nu_4 + \nu_6$) [7].
	ν_2	CC stretch	1623 D	ia	1622.6 (8)p	
	ν_3	CH_2 scis	1342 B	ia	1342.2 (10)p	
a_u	ν_4	CH_2 twist	1023 E	ia	ia	OC ($\nu_4 + \nu_6$) [7].
	ν_5	CH_2 a-stretch	3103 B	ia	3102.5 (1)dp	
b_{1g}	ν_6	CH_2 rock	1236 C	ia	1236 (1)dp (liquid)	
b_{1u}	ν_7	CH_2 wag	949 A	949.3 M	ia	OC ($\nu_4 + \nu_6$) [7].
	ν_8	CH_2 wag	943 C	ia	943 (1)dp (liquid)	
b_{2u}	ν_9	CH_2 a-stretch	3106 B	3105.5 S	ia	OC ($\nu_4 + \nu_6$) [7].
	ν_{10}	CH_2 rock	826 A	826.0 W	ia	
b_{3u}	ν_{11}	CH_2 s-stretch	2989 A	2988.66 S	ia	OC ($\nu_4 + \nu_6$) [7].
	ν_{12}	CH_2 scis	1444 B	1443.5 S	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] R. H. Rank, E. R. Shull, and D. W. E. Axford, J. Chem. Phys. 18, 116 (1950).
- [3] IR. R. L. Arnett and B. L. Crawford, Jr., J. Chem. Phys. 18, 118 (1950).
- [4] R. T. Feldman, J. Romanko, and H. L. Welsh, Can. J. Phys. 34, 737 (1956).
- [5] IR. H. C. Allen, Jr., and E. K. Plyler, J. Amer. Chem. Soc. 80, 2673 (1958).
- [6] IR. D. A. Dows, J. Chem. Phys. 36, 2833 (1962).
- [7] IR. W. L. Smith and I. M. Mills, J. Chem. Phys. 40, 2095 (1963).

Molecule: Ethylene-d₄ C₂D₄
 Symmetry D_{2h} Symmetry number δ = 4

No. 125

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	ν_1	CD ₂ s-stretch.....	2251 C	cm^{-1} (Gas)	2251 VS	
	ν_2	CC stretch.....	1515 C	ia	1515 VS	
	ν_3	CD ₂ scis.....	981 C	ia	981 M	
	ν_4	CD ₂ twist.....	728 E	ia	ia	CF [4].
	ν_5	CD ₂ a-stretch.....	2304 C	ia	2304 W	
	ν_6	CD ₂ rock.....	1009 E	ia	OC ($\nu_6 + \nu_{10}$).
	ν_7	CD ₂ wag.....	720 B	720.0 VS	ia	
	ν_8	CD ₂ wag.....	780 C	ia	780 W	
	ν_9	CD a-stretch.....	2345 C	2345 S	ia	
	ν_{10}	CD ₂ rock.....	586 E	ia	
	ν_{11}	CD ₂ a-stretch.....	2200 C	2200.2 S	ia	
	ν_{12}	CD ₂ scis.....	1078 C	1077.9 S	ia	

^a From product rule.

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR. R. L. Arnett and B. L. Crawford, Jr., J. Chem. Phys. 18, 118 (1950).
- [3] Th. B. N. Cyvin and S. J. Cyvin, Acta Chem. Scand. 17, 1831 (1963).
- [4] Th. J. Hiraishi and T. Shimanouchi, unpublished.

Molecule: Tetrafluoroethylene CF₂CF₂
 Symmetry D_{2h} Symmetry number δ = 4

No. 126

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	ν_1	CC stretch.....	1872 C	cm^{-1} (Gas)	1872 M, p	
	ν_2	CF ₂ s-stretch.....	778 C	ia	777.9 S, p	
	ν_3	CF ₂ scis.....	394 C	ia	394 W, p	
	ν_4	CF ₂ twist.....	190 E	ia	ia	CF [3].
	ν_5	CF ₂ a-stretch.....	1340 D	ia	1340 VW	
	ν_6	CF ₂ rock.....	551 D	ia	551 M (liquid)	
	ν_7	CF ₂ wag.....	406 C	406 S	ia	
	ν_8	CF ₂ wag.....	508 D	ia	508 S (liquid)	
	ν_9	CF ₂ a-stretch.....	1337 C	1337 S	ia	
	ν_{10}	CF ₂ rock.....	218 C	218 S	ia	
	ν_{11}	CF ₂ s-stretch.....	1186 C	1186 S	ia	
	ν_{12}	CF ₂ sciss.....	558 C	558 S	ia	

References

- [1] IR.R. J. R. Nielsen, H. H. Claassen, and D. C. Smith, J. Chem. Phys. 18, 812 (1950).
- [2] R. A. Monfils and J. Duchesne, J. Chem. Phys. 18, 1415 (1950).
- [3] IR. D. E. Mann, N. Acquista, and E. K. Plyler, J. Res. NBS 52, 67 (1954), RP2474.

Molecule: Tetrachloroethylene CCl_2CCl_2
 Symmetry D_{2h} Symmetry number $\delta = 4$

No. 127

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CC stretch.....	1571 D	cm^{-1} (Liquid)	cm^{-1} (Liquid)	OC ($2\nu_4$) [2].
	ν_2	CCl_2 s-stretch.....	447 D	ia	1571 (7)p	
	ν_3	CCl_2 scis.....	237 D	ia	447 (10)p	
	ν_4	CCl_2 twist.....	110 E	ia	237 (7)p	
	ν_5	CCl_2 a-stretch.....	1000 D	ia	1000 (0)	
	ν_6	CCl_2 rock.....	347 D	ia	347 (4)dp	
	ν_7	CCl_2 wag.....	288 D	288 M	ia	
	ν_8	CCl_2 wag.....	512 D	ia	512 (4)dp	
	ν_9	CCl_2 a-stretch.....	908 C	908 S (CS_2 soln.)	ia	
	ν_{10}	CCl_2 rock.....	176 C	176 S	ia	
	ν_{11}	CCl_2 s-stretch.....	777 C	777 S (CS_2 soln.)	ia	
	ν_{12}	CCl_2 scis.....	310 C	310 W	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. D. E. Mann, N. Acquista, and E. K. Plyler, J. Res. NBS 52, 67 (1954), RP2474.
- [3] IR. D. E. Mann, J. H. Meal and E. K. Plyler, J. Chem. Phys. 24, 1018 (1956).

Molecule: Tetrabromoethylene CBr_2CBr_2
 Symmetry D_{2h} Symmetry number $\delta = 4$

No. 128

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CC stretch.....	1535 E	cm^{-1} (Liquid)	cm^{-1} (Liquid)	{FR ($2\nu_9$) [1]. CF [2].}
	ν_2	CBr_2 s-stretch.....	265 D	ia	1547 (2)p	
	ν_3	CBr_2 scis.....	144 D	ia	1515 (1)p	
	ν_4	CBr_2 twist.....	66 E	ia	265 (10)p	
	ν_5	CBr_2 a-stretch.....	880 D	ia	144 (1)p	
	ν_6	CBr_2 rock.....	208 D	ia	880 (1)dp	
	ν_7	CBr_2 wag.....	245 C	245 S	208 (2)dp	
	ν_8	CBr_2 wag.....	464 D	ia	ia	
	ν_9	CBr_2 a-stretch.....	766 C	766 S	464 (1)dp	
	ν_{10}	CBr_2 rock.....	119 C	119 M	ia	
	ν_{11}	CBr_2 s-stretch.....	635 C	635 S	ia	
	ν_{12}	CBr_2 scis.....	188 C	188 M	ia	

References

- [1] R. F. E. Malherbe, G. Allen, and H. J. Bernstein, Can. J. Chem. 31, 1223 (1953).
- [2] IR. D. E. Mann, J. H. Meal, and E. K. Plyler, J. Chem. Phys. 24, 1018 (1956).

Molecule: cis-1,2-Difluoroethylene CHFCHF
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 129

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH stretch.....	3135 D	cm^{-1} (Gas)	3135 W	SF (ν_8).
	ν_2	CC stretch.....	1715 C	1715 S		
	ν_3	CH bend.....	1266 C	1266 S		
	ν_4	CF stretch.....	1014 C	1014 S		
	ν_5	CCF deform.....	255 D	255 W		
a_2	ν_6	CH bend.....	866 E	ia		CF. ^a
	ν_7	Torsion.....	482 E	ia		CF. ^b
b_1	ν_8	CH stretch.....	3135 D	3135 W		SF (ν_1).
	ν_9	CH bend.....	1376 C	1376 S		
	ν_{10}	CF stretch.....	1127 C	1127 VS		
b_2	ν_{11}	CCF deform.....	768 B	768 S		
	ν_{12}	CH bend.....	756 B	756 S		

^a From product rule.

^b Calculated by assuming $\frac{\nu_7(cis)}{\nu_7(trans)} = \frac{\nu_{12}(cis-d_1)}{\nu_{12}(trans-d_1)}$.

References

- [1] IR. R. N. Haszeldine and B. R. Steele, J. Chem. Soc. 1957, 2800 (1957).
- [2] IR. H. G. Viche, Chem. Ber. 93, 1697 (1960).
- [3] IR. N. C. Craig and E. A. Entemann, J. Chem. Phys. 36, 243 (1962).

Molecule: cis-1,2-Difluoroethylene-d₁ CHFCDF
 Symmetry C_s Symmetry number $\delta = 1$

No. 130

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH stretch.....	3125 D	cm^{-1} (Gas)	3125 W	
	ν_2	CD stretch.....	2364 D	2364 W		
	ν_3	CC stretch.....	1692 C	1692 S		
	ν_4	CH bend.....	1330 C	1330 S		
	ν_5	CF stretch.....	1167 C	1167 VS		
	ν_6	CF stretch.....	1033 C	1033 VS		
	ν_7	CD bend.....	889 B	889 M		
	ν_8	CCF deform.....	757 B	757 S		
	ν_9	CCF deform.....	255 D	255 W		
	ν_{10}	CH bend.....	801 B	801 M		
a''	ν_{11}	CD bend.....	633 B	633 M		
	ν_{12}	Torsion.....	469 R	469 W		

Reference

- [1] IR. N. C. Craig and E. A. Entemann, J. Chem. Phys. 36, 243 (1962).

Molecule: cis-1,2-Difluoroethylene-d₂ CDFCDF
 Symmetry C_{2v} Symmetry number δ = 2

No. 131

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν ₁	CD stretch.....	2320 D	cm ⁻¹ (Gas)	cm ⁻¹	SF (ν ₈). CF. ^a CF. ^b SF (ν ₁).
	ν ₂	CC stretch.....	1675 C	2320 W	
	ν ₃	CF stretch.....	1054 C	1675 S	
	ν ₄	CD bend.....	847 B	1054 S	
	ν ₅	CCF deform.....	255 D	847 M	
	ν ₆	CD bend.....	656 E	255 W	
	ν ₇	Torsion.....	459 E	ia	
	ν ₈	CD stretch.....	2320 D	2320 W	
	ν ₉	CF stretch.....	1225 C	1225 VS	
	ν ₁₀	CD bend.....	937 B	937 M	
	ν ₁₁	CCF deform.....	748 B	937 S	
	ν ₁₂	CD bend.....	597 B	748 M	

^a From product rule.

^b Calculated by assuming $\frac{\nu_{12}(\text{cis-}d_2)}{\nu_{12}(\text{trans-}d_2)} = \frac{\nu_7(\text{cis-}d_2)}{\nu_7(\text{trans-}d_2)}$.

Reference

See No. 130.

Molecule: trans-1,2-Dichloroethylene CHClCHCl
 Symmetry C_{2h} Symmetry number δ = 2

No. 132

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> _g	ν ₁	CH stretch.....	3073 C	cm ⁻¹ (Gas)	cm ⁻¹ (Liquid)
	ν ₂	CC stretch.....	1578 C	ia	3073 S, p	
	ν ₃	CH bend.....	1274 C	ia	1578 S, p	
	ν ₄	CCl stretch.....	846 C	ia	1274 S, p	
	ν ₅	CCl deform.....	350 C	ia	846 S, p	
	ν ₆	CH bend.....	900 B	899.8 VS	350 S, p	
	ν ₇	Torsion.....	227 C	227 M	ia	
	ν ₈	CH bend.....	763 B	ia	ia	
	ν ₉	CH stretch.....	3090 C	763 M, dp	ia	
	ν ₁₀	CH bend.....	1200 B	3090 S	ia	
	ν ₁₁	CCl stretch.....	828 B	1200 S	ia	
	ν ₁₂	CCl deform.....	250 D	828 VS	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR.R. H. J. Bernstein and D. A. Ramsey, J. Chem. Phys. 17, 556 (1949).
- [3] IR.R. H. J. Bernstein and A. D. E. Pullin, Can. J. Chem. 30, 963 (1952).
- [4] IR.R. K. S. Pitzer and J. L. Hollenberg, J. Amer. Chem. Soc. 76, 1493 (1954).
- J. M. Dowling, J. Chem. Phys. 25, 284 (1956).
- J. M. Dowling and T. Shimanouchi, unpublished.

Molecule: trans-1,2-Dichloroethylene-d₁ CHClCDCl
 Symmetry C_s Symmetry number δ = 1

No. 133

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH stretch.....	3087 C	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Liquid)	
	<i>v</i> ₂	CD stretch.....	2310 C	3087 S	3074 M	
	<i>v</i> ₃	CC stretch.....	1574 D	2310 S	2304 M	
	<i>v</i> ₄	CH bend.....	1241 C	1574 W (liquid)	1574 S	
	<i>v</i> ₅	CD bend.....	963 C	1241 S	1238 S	
	<i>v</i> ₆	CCl stretch.....	823 C	963 VS	957 S	
	<i>v</i> ₇	CCl stretch.....	775 B	825 VS (liquid)	823 W	
	<i>v</i> ₈	CCCl deform.....	348 C	775 VS	775 M	
	<i>v</i> ₉	CCCl deform.....	245 E	348 VS	
	<i>v</i> ₁₀	CH bend.....	830 C	830 VS	834 W	CF [6].
	<i>v</i> ₁₁	CD bend.....	660 B	660 S	659 W	
	<i>v</i> ₁₂	Torsion.....	224 E	CF [5, 6].

References

See No. 132.

Molecule: trans-1,2-Dichloroethylene-d₂ CDClCDCl
 Symmetry C_{2h} Symmetry number δ = 2

No. 134

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	<i>v</i> ₁	CD stretch.....	2325 C	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Liquid)	
	<i>v</i> ₂	CC stretch.....	1570 C	ia	2325 S	
	<i>v</i> ₃	CD bend.....	992 C	ia	1570 S	
	<i>v</i> ₄	CCl stretch.....	765 C	ia	992 S	
	<i>v</i> ₅	CCCl deform.....	346 C	ia	765 M	
	<i>v</i> ₆	CD bend.....	660 B	660 S	346 S	
	<i>v</i> ₇	Torsion.....	221 E	ia	
	<i>v</i> ₈	CD bend.....	657 C	ia	ia	CF [5, 6].
	<i>v</i> ₉	CD stretch.....	2290 C	2290 S	657 M	
	<i>v</i> ₁₀	CD bend.....	916 C	916 VS	ia	
	<i>v</i> ₁₁	CCl stretch.....	791 C	791 VS	ia	
	<i>v</i> ₁₂	CCCl deform.....	240 E	ia	CF [6].

References

See No. 132.

Molecule: cis-1,2-Dichloroethylene CHClCHCl
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 135

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH stretch.....	3077 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CC stretch.....	1587 C	1590 S (liquid)	3077 1587 S, p	
	ν_3	CH bend.....	1179 C	1183 W (liquid)	1179 S, p	
	ν_4	CCl stretch.....	711 C	714 S (liquid)	711 S, p	
a_2	ν_5	CCl deform.....	173 C	173 S, p	
	ν_6	CII bend.....	876 C	ia	876 W, dp	
b_1	ν_7	Torsion.....	406 C	ia	406 S, dp	
	ν_8	CH stretch.....	3072 C	3072		
b_2	ν_9	CH bend.....	1303 C	1303		
	ν_{10}	CCl stretch.....	857 B	857		
	ν_{11}	CCl deform.....	571 B	571		
	ν_{12}	CH bend.....	697 B	697	563 M, dp	

References

See No. 132.

Molecule: cis-1,2-Dichloroethylene-d₁ CHClCDCl
 Symmetry C_s Symmetry number $\delta = 1$

No. 136

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH stretch.....	3076 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CD stretch.....	2306 C	3076 S (liquid)	3078 VS	
	ν_3	CC stretch.....	1562 C	2306 S 1562 S	2299 VS 1553 S	
	ν_4	CH bend.....	1253 C	1562 S 1253 VS	1245 M	
	ν_5	CD bend.....	957 C	1253 VS 957 VS	950 M	
	ν_6	CCl stretch.....	788 B	957 VS 788 VS	781 W	
	ν_7	CCl stretch.....	711 C	788 VS 711 VS	703 VS	
	ν_8	CCl bend.....	558 C	711 VS 558 S	561 S	
	ν_9	CCl bend.....	175 D	558 S 175 VS	175 VS	
	ν_{10}	CH bend.....	822 C	175 VS 822 VS	817 W	
	ν_{11}	CD bend.....	589 C	822 VS 589 VS	590 W	
	ν_{12}	Torsion.....	387 C	589 VS 387 S	387 S	

References

See No. 132.

Molecule: cis-1,2-Dichloroethylene-d₂ CDCICDCI
 Symmetry C_{2v} Symmetry number δ = 2

No. 137

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD stretch.....	2325 C	cm ⁻¹ (Gas)	cm ⁻¹ (Liquid)	
	<i>v</i> ₂	CC stretch.....	1575 C	1575 S	1570 S	
	<i>v</i> ₃	CD bend.....	850 C	850 M	
	<i>v</i> ₄	CCl stretch.....	700 C	700 S	689 S	
	<i>v</i> ₅	CCCl deform.....	171 C	171 S	
	<i>a</i> ₂	CD bend.....	686 E	ia	CF [6].
	<i>v</i> ₆	Torsion.....	368 C	ia	368 M	
	<i>b</i> ₁	CD stretch.....	2280 B	2280	2280 S	
	<i>v</i> ₈	CD bend.....	1051 B	1051	1040 VS	
	<i>v</i> ₉	CCl stretch.....	766 B	766	761 VS	
	<i>v</i> ₁₀	CCCl deform.....	558 C	558 S	
	<i>b</i> ₂	CD bend.....	540 C	540 S	

References

See No. 132.

Molecule: trans-1,2-Dichloro-1,2-difluoroethylene
 Symmetry C_{2h} Symmetry number δ = 2

CClFCClF

No. 138

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> _g	<i>v</i> ₁	CC stretch.....	1707 C	cm ⁻¹ (Gas)	cm ⁻¹ (Liquid)	
	<i>v</i> ₂	CF stretch.....	1186 C	ia	1707 VS, p	
	<i>v</i> ₃	CCl stretch.....	632 C	ia	1186 W, p	
	<i>v</i> ₄	CF bend.....	425 C	ia	632 M, p	
	<i>v</i> ₅	CCl bend.....	288 C	ia	425 M, p	
	<i>a</i> _u	CFCl wag.....	333 C	333 M	288 M, p	
	<i>v</i> ₆	CFCl wag.....	140 D	ia	CF [2].
	<i>b</i> _g	CFCI wag.....	529 C	ia	ia	
	<i>b</i> _u	CF stretch.....	1190 E	{ 1214 VS 1167 VS }	529 M, dp	
	<i>v</i> ₁₀	CCl stretch.....	892 B	892 VS	ia	
	<i>v</i> ₁₁	CF bend.....	426 C	426 M	ia	
	<i>v</i> ₁₂	CCl bend.....	175 C	175 M	ia	

References

- [1] IR.R. D. E. Mann and E. K. Plyler, J. Chem. Phys. 26, 773 (1957).
 [2] Th. D. E. Mann, L. Fano, J. H. Meal, and T. Shimanouchi, J. Chem. Phys. 27, 51 (1957).

Molecule: 1,1-Dichloroethylene CH_2CCl_2
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 139

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_2 s-stretch.....	3035 D	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CC stretch.....	1627 C	3035 W	3035 VS, p	
	ν_3	CH_2 scis.....	1400 C	1627 VS	1616 VS, p	
	ν_4	CCl_2 s-stretch.....	603 C	1400 M	1391 M, p	
	ν_5	CCl_2 scis.....	299 C	603 VS	601 VS, p	
	ν_6	Torsion	686 D	299 W	299 S, p	
	ν_7	CH_2 a-stretch.....	3130 D	ia	686 M, dp	
	ν_8	CH_2 rock.....	1095 C	^a 3130 W	3130 S, dp	
	ν_9	CCl_2 a-stretch.....	800 B	1095 VS	1088 VW	
	ν_{10}	CCl_2 rock.....	372 C	800 VS	788 M, dp	
	ν_{11}	CH_2 wag.....	875 B	372 M	375 S, dp	
	ν_{12}	CCl_2 wag.....	460 B	875 S	874 W	
				460 S	458 M, dp	

^a CCl_4 solution.

References

- [1] IR. H. W. Thompson and P. Torkington, Proc. Roy. Soc. (London), Ser. A, 184, 21 (1945).
- [2] R. P. Joyner and G. Glocker, J. Chem. Phys. 20, 302 (1952).
- [3] IR.Th. T. Shimanouchi and S. Shimizu, unpublished.

Molecule: 1,1-Dichloroethylene-d₁ CHDCCl_2
 Symmetry C_s Symmetry number $\delta = 1$

No. 140

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH stretch.....	3082 D	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CD stretch.....	2288 D	^a 3082 W	^a 2288 W	
	ν_3	CC stretch	1585 C	2288 S	1585 S	
	ν_4	CHD scis.....	1280 C	1585 M	1280 M	
	ν_5	CHD rock.....	999 C	1280 S	999 VS	
	ν_6	CCl_2 a-stretch.....	741 C	999 W	741 S	
	ν_7	CCl_2 s-stretch.....	590 C	741 VS	590 S	
	ν_8	CCl_2 rock.....	348 C	590 W	348 S	
	ν_9	CCl_2 scis.....	306 E	348 M	306 S	
	ν_{10}	CHD wag.....	819 B	306 W	819 S	
	ν_{11}	Torsion	555 C	819 M	555 W	
	ν_{12}	CCl_2 wag.....	444 B	555 S	444 M	CF [1].

^a CCl_4 solution.

Reference

- [1] IR.Th. T. Shimanouchi and S. Shimizu, unpublished.

Molecule: 1,1-Dichloroethylene-d₂ CD₂CCl₂
 Symmetry C_{2v} Symmetry number δ = 2

No. 141

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν ₁	CD ₂ s-stretch.....	2262 D	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹	
	ν ₂	CC stretch.....	1565 C	^a 2262 W 1565 VS		
	ν ₃	CD ₂ scis.....	1039 E			CF [1].
	ν ₄	CCl ₂ s-stretch.....	580 C	580 VS		
	ν ₅	CCl ₂ scis.....	305 E			CF [1].
	ν ₆	Torsion.....	488 E	ia		CF [1].
	<i>b</i> ₁	CD ₂ a-stretch.....	2380 D	^a 2380 W		
	ν ₇	CD ₂ rock.....	998 C	998 VS		
	ν ₈	CCl ₂ a-stretch.....	697 C	697 S		
	ν ₉	CCl ₂ rock.....	327 C	327 M		SF (ν ₁₁). SF (ν ₉).
	ν ₁₀	CD ₂ wag.....	697 C	697 S		
	ν ₁₁	CCl ₂ wag.....	439 B	439 S		

^a CCl₄ solution.

Reference

- [1] IR.Th. T. Shimanouchi and S. Shimizu, unpublished.

Molecule: 1,1-Dichloro 2,2-difluoroethylene CF₂CCl₂
 Symmetry C_{2v} Symmetry number δ = 2

No. 142

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν ₁	CC stretch.....	1749 B	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Liquid)	
	ν ₂	CF stretch.....	1032 B	1749 VS 1032 VS	1738.8 S 1027.6 M	
	ν ₃	CCl stretch.....	622 C	622 M	623.0 S	
	ν ₄	CF ₂ scis.....	434 C	434	433.8 VS	
	ν ₅	CCl ₂ scis.....	258 C	258 S	258 VS	
	ν ₆	Torsion.....	167 D		167 VW	
	<i>b</i> ₁	CF stretch.....	1327 B	1327 VS	1313 VW	
	ν ₇	CCl stretch.....	989 B	989 VS	986 VW	
	ν ₈	CF ₂ rock.....	459 C	459 VW	454 W	
	ν ₉	CCl ₂ rock.....	192 C	192	187.8 W	
	ν ₁₀	CF ₂ wag.....	564 C	564 S	560.8 VS	
	ν ₁₁	CCl ₂ wag.....	323 C	323 W		

References

- [1] IR.R. J. R. Nielsen and H. H. Claassen, J. Chem. Phys. 18, 485 (1950).
 [2] IR. D. E. Mann and E. K. Plyler, J. Chem. Phys. 23, 1989 (1955).
 [3] Th. D. E. Mann, L. Fano, J. H. Meal, and T. Shimanouchi, J. Chem. Phys. 27, 51 (1957).

Molecule: Methylcyanide CH_3CN
 Symmetry $\text{C}_{\delta v}$ Symmetry number $\delta = 3$

No. 143

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch.....	2954 A	cm^{-1} (Gas)	2954.1 M	2942 VS
	ν_2	CN stretch.....	2267 A	2266.5 M	2249 S	
	ν_3	CH_3 s-deform.....	1385 C		1376 M	
	ν_4	CC stretch.....	920 A	920.2 S	918 S	
	ν_5	CH_3 d-stretch.....	3009 A	3009.2 S	2999 S	
	ν_6	CH_3 d-deform.....	1448 D	1447.9 S		
	ν_7	CH_3 rock.....	1041 A	1040.8 M	1440 M, b	FR ($\nu_7 + \nu_8$).
	ν_8	CCN bend.....	362 B	362 S	380 S	

References

- [1] R. A. Dadien, Monatsh. Chem. 57, 437 (1931).
- [2] R. A. W. Reitz and R. Skrabel, Monatsh. Chem. 70, 398 (1937).
- [3] IR.R. P. Venkateswarlu, J. Chem. Phys. 19, 293 (1951).
- [4] IR.R. H. W. Thompson and R. L. Williams, Trans Faraday Soc. 48, 502 (1952).
- [5] IR. F. W. Parker, A. H. Nielsen, and W. H. Fletcher, J. Mol. Spectrosc. 1, 107 (1957).
- [6] IR. I. Nakagawa and T. Shimanouchi, Spectrochim. Acta 18, 513 (1962).
- [7] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).
- [8] Th. G. Amat and H. H. Nielsen, Molecular Orbitals in Chemistry, Physics and Biology p. 293 (Academic Press, New York, 1964).
- [9] IR. G. Gauffre and G. Amat, Symposium on Molecular Structure and Spectroscopy, B9, Columbus, Ohio, 1970.
- [10] IR. H. Matsuura, Bull. Chem. Soc. Japan 44, 2379 (1971).

Molecule: Methylcyanide-d₃ CD_3CN
 Symmetry $\text{C}_{\delta v}$ Symmetry number $\delta = 3$

No. 144

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CD_3 s-stretch.....	2126 A	cm^{-1} (Gas)	2125.6	2112 S
	ν_2	CN stretch.....	2278 A	2277.6	2258 S	
	ν_3	CD_3 s-deform.....	1110 B	1110	1103 W	
	ν_4	CC stretch.....	831 A	831.3	834 W	
	ν_5	CD_3 d-stretch.....	2257 A	2256.6	2258 S	
	ν_6	CD_3 d-deform.....	1046 A	1046.4	1041 W	
	ν_7	CD_3 rock.....	847 A	846.6		
	ν_8	CCN bend.....	331 B	331.2	348 M	

References

- [1] IR.R. J. C. Evans and H. J. Bernstein, Can. J. Chem. 33, 1746 (1955).
- [2] IR. W. H. Fletcher and C. S. Shoup, Proceedings of International Symposium on Molecular Structure and Spectroscopy C 204 (Tokyo, 1962).
- [3] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).

Molecule: Methyl isocyanide CH_3NC
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 145

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch.....	2966 B	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	NC stretch.....	2166 B	2951 S	2161 S	
	ν_3	CH_3 s-deform.....	1429 D	1429	1414 M	
	ν_4	CN stretch.....	945 B	944.6 M	928 M	
	ν_5	CH_3 d-stretch.....	3014 B	3014.3 S	3002 W	
	ν_6	CH_3 d-deform.....	1467 B	1466.9 S	1456 W	
	ν_7	CH_3 rock.....	1129 B	1129.3 S		
	ν_8	CNC bend.....	263 C	263 W	290 S	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR. H. W. Thompson and R. L. Williams, Trans. Faraday Soc. 48, 502 (1952).
- [3] IR. R. L. Williams, J. Chem. Phys. 25, 656 (1956).
- [4] Th. W. H. Fletcher and C. S. Shoup, Proceedings of the International Symposium on Molecular Structure and Spectroscopy, C204 (Tokyo, 1962).
- [5] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).

Molecule: Methyl isocyanide-d₃ CD_3NC
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 146

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CD_3 s-stretch.....	2251 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	NC stretch.....	2165 B	2250.6 W		
	ν_3	CD_3 s-deform.....	1117 B	2165.0 W		
	ν_4	CN stretch.....	877 B	1117.4 W		
	ν_5	CD_3 d-stretch.....	2263 B	876.7 M		
	ν_6	CD_3 d-deform.....	1058 B	2262.9 S		
	ν_7	CD_3 rock.....	900 B	1058.2 S		
	ν_8	CNC bend.....	249 C	900.1 S		OC ($\nu_2 + \nu_8$)

References

- [1] IR. J. G. Mottern and W. H. Fletcher, Spectrochim. Acta 18, 995 (1962).
- [2] IR.Th. W. H. Fletcher and C. S. Shoup, Proceedings of the International Symposium on Molecular Structure and Spectroscopy, C204 (Tokyo, 1962).
- [3] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).

Molecule: 1,2,5-Oxadiazole $C_2H_2N_2O$
 Symmetry C_{2v} Symmetry number $\delta = 2$

No. 147

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CII stretch.....	3140 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	ip-Ring II.....	1418 B	3140 VW	3144 VS, p	
	ν_3	ip-Ring III.....	1316 B	1418 S	1422 VS, p	
	ν_4	CH ip-bend.....	1038 D	1316 M	1315 VS, p	
	ν_5	ip-Ring IV.....	1006 B	1039 sh	1038 W, p	
	ν_6	ip-Ring VII.....	872 C	1006 S	998 M, p	
	ν_7	CH op-bend.....	824 D	872 S ia, 824 sh (liquid)	864 M, p 824 VW, dp	
a_2	ν_8	op-Ring I	635 E	ia	OC ($2\nu_8, \nu_4 + \nu_8, \nu_8 + \nu_{12}$).
	ν_9	CH stretch.....	3133 D	3133 sh (liquid)	
	ν_{10}	ip-Ring I.....	1546 D	1546 VW (liquid)	
	ν_{11}	CH ip-bend.....	1177 B	1177 M	1172 VW, dp	
	ν_{12}	ip-Ring V.....	952 B	952 S	951 W, dp	
	ν_{13}	ip-Ring VI.....	889 B	889 S	
	ν_{14}	CH op-bend.....	839 B	839 VS	
b_1	ν_{15}	op-Ring II.....	631 B	631 W	626 VW, dp	

Reference

- [1] IR.R. G. Sbrana, M. Ginanneschi, and M. P. Marzocchi, Spectrochim. Acta 23A, 1757 (1967).

Molecule: Silylacetylene SiH_3CCH
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 148

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH stretch.....	3311 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	SiH_3 s-stretch.....	2192 B	3311.4 M	2192.4 VS	
	ν_3	CC stretch.....	2055 B	2192.4 VS	2054.9 S	
	ν_4	SiH_3 s-deform.....	935 B	2054.9 S	935.3 VS	
	ν_5	SiC stretch.....	659 D	935.3 VS	659 S	
	ν_6	SiH_3 d-stretch.....	2193 A	659 S	2192.9 VS	
	ν_7	SiH_3 d-deform.....	946 D	2192.9 VS	946.4 VS	
e	ν_8	SiH_3 rock.....	685 D	946.4 VS	685.4 VS	
	ν_9	CH bend.....	668 D	685.4 VS	668 VS	
	ν_{10}	SiCC deform.....	220 E	668 VS	220	

^a These frequencies are taken from ref. 1. The band centers of ν_5 , ν_7 , ν_8 , and ν_9 given in ref. 2 are different from the values listed in this table by $10\text{--}20\text{ cm}^{-1}$, due to the different assignment of the vibration-rotation lines.

References

- [1] IR. E. A. V. Ebsworth, S. G. Frankiss, and W. J. Jones, J. Mol. Spectrosc. 13, 9 (1964).
 [2] IR. R. B. Reeves, R. E. Wilde, and D. W. Robinson, J. Chem. Phys. 40, 125 (1964).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH ₂ s-stretch.....	3006 C	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Liquid)	
	<i>v</i> ₂	CH ₂ scis.....	1498 B	3005 S, p 1498 W	3005 S, p 1490 W, p	
	<i>v</i> ₃	Ring stretch.....	1271 B	1271 S	1266 S, p	
	<i>v</i> ₄	CH ₂ wag.....	1120 D	1118 W (CS ₂ soln.)	1120 M, p	
<i>a</i> ₂	<i>v</i> ₅	Ring deform.....	877 B	877 VS	867 M, dp	
	<i>v</i> ₆	CH ₂ a-stretch.....	3063 D	ia	3063 W, dp	OV (<i>v</i> ₁₃).
	<i>v</i> ₇	CH ₂ twist.....	1300 E	ia		
	<i>v</i> ₈	CH ₂ rock.....	860 E	ia		
<i>b</i> ₁	<i>v</i> ₉	CH ₂ s-stretch.....	3006 C	3006 S	3005 S, p	OV (<i>v</i> ₁).
	<i>v</i> ₁₀	CH ₂ scis.....	1472 B	1472 W		
<i>b</i> ₂	<i>v</i> ₁₁	CH ₂ wag.....	1151 D	1151 M	1150 W, dp	
	<i>v</i> ₁₂	Ring deform.....	892 D	892 VS		
	<i>v</i> ₁₃	CH ₂ a-stretch.....	3065 B	3065 S	3063 W, dp	
	<i>v</i> ₁₄	CH ₂ twist.....	1142 D	1142 M	1150 W, dp	
	<i>v</i> ₁₅	CH ₂ rock.....	822 B	822 M	807 M, dp	

References

- [1] I.R.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR.Th. Hs. H. Günthard, B. Messikommer, and M. Kohler, Helv. Chim. Acta 33, 1809 (1950).
- [3] I.R.R. H. W. Thompson and W. T. Cave, Trans. Faraday Soc. 47, 946 (1951).
- [4] I.R. R. C. Lord and B. Nolin, J. Chem. Phys. 24, 656 (1956).
- [5] I.R.R. W. J. Potts, Spectrochim. Acta 21, 511 (1965).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD ₂ s-stretch.....	2204 C	cm ⁻¹ (Gas)	cm ⁻¹ (Liquid)
	<i>v</i> ₂	CD ₂ scis.....	1311 C	1311 M	2204 S	1301 VS
	<i>v</i> ₃	Ring stretch.....	1013 C	1014 W	1013 S	
	<i>v</i> ₄	CD ₂ wag.....	970 C	970 VS	952 M	
	<i>v</i> ₅	Ring deform.....	755 C	755 VS	755 M	
<i>a</i> ₂	<i>v</i> ₆	CD ₂ a-stretch.....	2250 D	ia	2250 W	
	<i>v</i> ₇	CD ₂ twist.....	1083 D	ia	1083 VW	
	<i>v</i> ₈	CD ₂ rock.....	581 D	ia	581 W	
<i>b</i> ₁	<i>v</i> ₉	CD ₂ s-stretch.....	2174 C	2174 VS	2157 M	
	<i>v</i> ₁₀	CD ₂ scis.....	1145 D	1145 VW		
<i>b</i> ₂	<i>v</i> ₁₁	CD ₂ wag.....	952 D	952 M	
	<i>v</i> ₁₂	Ring deform.....	809 C	809 S	786 M	
	<i>v</i> ₁₃	CD ₂ a-stretch.....	2317 C	2317 VS	2319 S	
	<i>v</i> ₁₄	CD ₂ twist.....	896 C	896 S	896 W	
	<i>v</i> ₁₅	CD ₂ rock.....	577 C	577 W		

References

- [1] IR. C. W. Arnold and F. A. Matsen, Bull. Amer. Phys. Soc. **29**, 8 (1954).
- [2] IR.R. R. C. Lord and B. Nolin, J. Chem. Phys. **24**, 656 (1956).

Molecule: Acetaldehyde CH_3CHO
 Symmetry C_s Symmetry number $\delta = 1$

No. 151

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_3 d-stretch.....	3005 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CH_3 s-stretch.....	2917 D	3005 M	3001 W	
	ν_3	CH stretch.....	2822 C	2822 M	2843 W, p	
	ν_4	CO stretch.....	1743 C	1743 VS	1714 S, p	
	ν_5	CH_3 d-deform.....	1441 C	1441 S	1426 S	
	ν_6	CH bend.....	1400 C	1400 S	1391 S	
	ν_7	CH_3 s-deform.....	1352 C	1352 S	1342 M	
	ν_8	CC stretch.....	1113 C	1113 S	1109 M, p	
	ν_9	CH_3 rock.....	919 C	919 M	911 M	
	ν_{10}	CCO deform.....	509 C	509 S	512 S, p	
	ν_{11}	CH_3 d-stretch.....	2967 C	2967 M	2964 W	
	ν_{12}	CH_3 d-deform.....	1420 C	1420 S	1426 S, dp	
	ν_{13}	CH_3 rock.....	867 C	867 M	885 M	
	ν_{14}	CH bend.....	763 C	763 W	767 M, dp	
	ν_{15}	Torsion	150 C	150 W	MW: 150 (A), 148 (E) [2].

References

- [1] IR.R. J. C. Evans and H. J. Bernstein, Can. J. Chem. 34, 1084 (1956).
- [2] MW. D. R. Herschbach, J. Chem. Phys. 31, 91 (1959).
- [3] IR. W. G. Fateley and F. A. Miller, Spectrochim. Acta 19, 389 (1963).
- [4] IR.R.Th. P. Cossee and J. H. Schachtschneider, J. Chem. Phys. 44, 97 (1966).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₃ d-stretch.....	3028 C	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Liquid)	
	<i>v</i> ₂	CH ₃ s-stretch.....	2917 D	3028 M	2998 W	
	<i>v</i> ₃	CD stretch.....	2071 C	2071 W	2917 S, p	
	<i>v</i> ₄	CO stretch.....	1743 C	1743 VS	2097 W, p	
	<i>v</i> ₅	CH ₃ d-deform.....	1442 C	1442 S	1702 S, p	
	<i>v</i> ₆	CD bend.....	1109 C	1109 S	1426 S	
	<i>v</i> ₇	CD ₃ s-deform.....	1353 C	1353 S	1111 S, p	
	<i>v</i> ₈	CC stretch.....	1043 C	1043 W	1343 M	
	<i>v</i> ₉	CH ₃ rock.....	849 C	849 M	1080 W	
	<i>v</i> ₁₀	CCO deform.....	500 C	500 S	858 M	
	<i>v</i> ₁₁	CH ₃ d-stretch.....	2970 C	2970 M	505 M, p	
	<i>v</i> ₁₂	CH ₃ d-deform.....	1420 C	1420 S	2965 W	
	<i>v</i> ₁₃	CH ₃ rock.....	802 C	802 W	1426 S	
	<i>v</i> ₁₄	CD bend.....	668 C	668 W	820 W, sh	
	<i>v</i> ₁₅	Torsion.....	145 D	145	674 W, dp	

References

- [1] IR.R. J. C. Evans and H. J. Bernstein, Can. J. Chem. 34, 1084 (1956).
 [2] IR.R.Th. P. Cossee and J. H. Schachtschneider, J. Chem. Phys. 44, 97 (1966).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	CD ₃ d-stretch.....	2265 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CD ₃ s-stretch.....	2130 C	2130 W	2128	
	ν_3	CD stretch.....	2060 C	2060 M	2072	
	ν_4	CO stretch.....	1737 C	1737 VS	1706	
	ν_5	CD ₃ d-deform.....	1045 C	1045 M	1090	
	ν_6	CD bend.....	938 C	938 M		
	ν_7	CD ₃ s-deform.....	1028 C	1028 M	1024	SF (ν_{12}).
	ν_8	CC stretch.....	1151 C	1151 S	1153	
	ν_9	CD ₃ rock.....	747 C	747 W	762	
	ν_{10}	CCO deform.....	436 C	436 S	422.4	
	ν_{11}	CD ₃ d-stretch.....	2225 C	2225 W		
	ν_{12}	CD ₃ d-deform.....	1028 C	1028 M	1024	SF (ν_7).
	ν_{13}	CD ₃ rock.....	573 C	573 W		
	ν_{14}	CD bend.....	670 D	CF [5].
	ν_{15}	Torsion.....	116 C	116 W	MW [3].

References

- [1] R. W. Wood, J. Chem. Phys. 4, 536 (1936).
- [2] IR.R. J. Carrell Morris, J. Chem. Phys. 11, 230 (1943).
- [3] MW. D. R. Herschbach, J. Chem. Phys. 31, 91 (1959).
- [4] IR. W. G. Fateley and F. A. Miller, Spectrochim. Acta 19, 389 (1963).
- [5] IR.R.Th. P. Cossee and J. H. Schachtschneider, J. Chem. Phys. 44, 97 (1966).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	CH_3 s-stretch.....	2954 B	cm^{-1} (Gas) ia	cm^{-1} (Gas) 2953.7	
	ν_2	CH_3 s-deform.....	1388 B	ia	1388.4	
	ν_3	CC stretch.....	995 A	ia	994.8	
	ν_4	Torsion.....	289 B	289	ia	
	ν_5	CH_3 s-stretch.....	2896 B	2895.8	ia	
	ν_6	CH_3 s-deform.....	1379 A	1379.2	ia	
	ν_7	CH_3 d-stretch.....	2969 A	ia	2968.7	
	ν_8	CH_3 d-deform.....	1468 A	ia	1468.1	
	ν_9	CH_3 rock.....	1190 E	ia	OC [2, 3].
	ν_{10}	CH_3 d-stretch.....	2985 A	2985.4	ia	
	ν_{11}	CH_3 d-deform.....	1469 C	1469	ia	
	ν_{12}	CH_3 rock.....	822 A	821.6	ia	FR ($\nu_4 + \nu_{12}$).

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. L. G. Smith, J. Chem. Phys. 17, 139 (1949).
- [3] IR. G. E. Hansen and D. M. Dennison, J. Chem. Phys. 20, 313 (1952).
- [4] R. J. Romanko, T. Feldman, and H. L. Welsh, Can. J. Phys. 33, 588 (1955).
- [5] IR.R. R. Van Riet, Bull. Roy. Acad. Belg. 1955, 188.
- [6] R. D. W. Lepard, D. E. Shaw, and H. L. Welsh, Can. J. Phys. 44, 2353 (1966).
- [7] IR. S. Weiss and G. E. Leroy, J. Chem. Phys. 48, 962 (1968).
- [8] IR. A. R. H. Cole, W. J. Lafferty, and R. J. Thibault, J. Mol. Spectrosc. 29, 365 (1969).
- [9] IR. I. Nakagawa and T. Shimanouchi, J. Mol. Spectrosc. 39, 255 (1971).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a₁</i>	ν_1	CH ₃ s-stretch.....	2912 E	cm^{-1} (Gas) 2955.1 S 2897.4 S	2955.5	{FR (2 ν_9).
	ν_2	CD ₃ s-stretch.....	2098 E	2139.6 S 2089.7 S	2898.2	FR (2 ν_{11}).
	ν_3	CH ₃ s-deform.....	1387 B	1386.6 W		
	ν_4	CD ₃ s-deform.....	1122 B	1122.0 W		
	ν_5	CC stretch.....	904 A	903.8 VW	904.7	
	ν_6	Torsion.....	253 B	253 VW		
	ν_7	CH ₃ d-stretch.....	2977 D	2976.5 S	2976.6	
	ν_8	CD ₃ d-stretch.....	2240 E	2240 S		
	ν_9	CH ₃ d-deform.....	1471 A	1471.1 M		
	ν_{10}	CH ₃ rock.....	1115 B	1115.0 W		
	ν_{11}	CD ₃ d-deform.....	1066 B	1065.7 M	1062.6	
	ν_{12}	CD ₃ rock.....	678 A	678.4 M		

References

- [1] IR. R. Van Riet, Ann. Soc. Sci. Bruxelles, Ser. I, 71, 102 (1957).
- [2] R. D. E. Shaw and H. L. Welsh, Can. J. Phys. 45, 3823 (1967).
- [3] IR. S. Weiss and G. E. Leroi, J. Chem. Phys. 48, 962 (1968).
- [4] IR.Th. I. Nakagawa and T. Shimanouchi, J. Mol. Spectrosc. 39, 255 (1971).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> _{1g}	<i>v</i> ₁	CD ₃ s-stretch.....	2083 B	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Gas)	
	<i>v</i> ₂	CD ₃ s-deform.....	1155 A	ia	2083.0	
	<i>v</i> ₃	CC stretch.....	843 A	ia	1154.5	
<i>a</i> _{1u}	<i>v</i> ₄	Torsion.....	208 B	208	ia	
	<i>v</i> ₅	CD ₃ s-stretch.....	2087 B	2087.4	ia	
<i>a</i> _{2u}	<i>v</i> ₆	CD ₃ s-deform.....	1077 B	1077.1	ia	
	<i>v</i> ₇	CD ₃ d-stretch.....	2226 A	ia	2225.6	
	<i>v</i> ₈	CD ₃ d-deform.....	1041 B	ia	1041	
<i>e</i> _g	<i>v</i> ₉	CD ₃ rock.....	970 C	ia	970 (liquid)	
	<i>v</i> ₁₀	CD ₃ d-stretch.....	2235 B	2235	ia	
	<i>v</i> ₁₁	CD ₃ d-deform.....	1081 B	1080.9	ia	
<i>e</i> _u	<i>v</i> ₁₂	CD ₃ rock.....	594 A	594.4	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR. G. E. Hansen and D. M. Dennison, J. Chem. Phys. **20**, 313 (1952).
- [3] IR.R. R. Van Riet, Bull. Roy. Acad. Belg. 1955, 188.
- [4] R. D. W. Lepard, D. M. C. Sweeny, and H. L. Welsh, Can. J. Phys. **40**, 1567 (1962).
- [5] IR. S. Weiss and G. E. Leroy, J. Chem. Phys. **48**, 962 (1968).
- [6] IR. I. Nakagawa and T. Shimanouchi, J. Mol. Spectrosc. **39**, 255 (1971).

Molecule: Hexafluoroethane CF_3CF_3
 Symmetry D_{3d} Symmetry number $\delta = 6$

No. 157

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	CC stretch.....	1228 D	cm^{-1} (Gas)	1228	OC. ^a
	ν_2	CF_3 s-stretch.....	807 C	ia	807 VS, p	
	ν_3	CF_3 s-deform.....	348 C	ia	348 W, p	
	ν_4	Torsion.....	68 D	ia	ia	
	ν_5	CF_3 s-stretch.....	1117 B	1117 VS	ia	
	ν_6	CF_3 s-deform.....	714 B	714 VS	ia	
	e_g	CF_3 d-stretch.....	1250 C	ia	1250 VW, dp	
	ν_7	CF_3 d-deform.....	619 C	ia	619 W, dp	
	ν_8	CF_3 rock.....	372 C	ia	372 W, dp	
	e_u	CF_3 d-stretch.....	1251 B	1251 VS	ia	
	ν_{10}	CF_3 d-deform.....	520 C	520 S	ia	
	ν_{12}	CF_3 rock.....	220 C	220 S	ia	

^a Mean value of frequencies obtained from six combination bands [2].

References

- [1] IR. D. E. Mann and E. K. Plyler, J. Chem. Phys. 21, 1116 (1953).
- [2] IR.R. R. A. Carney, Y. A. Piotrowski, A. G. Meister, J. H. Braun, and F. F. Cleveland, J. Mol. Spectrosc. 7, 209 (1961).
- [3] Th. Fujiyama and T. Shimanouchi, unpublished.

Molecule: Hexachloroethane CCl_3CCl_3
 Symmetry D_{3d} Symmetry number $\delta = 6$

No. 158

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	CC stretch.....	975 C	cm^{-1} (Solid)	975 VW, p	
	ν_2	CCl_3 s-stretch.....	431 C	ia	431 VS, p	
	ν_3	CCl_3 s-deform.....	170 C	ia	170 W	
	ν_4	Torsion.....	61 D	ia	ia	
	a_{2u}	CCl_3 s-stretch.....	675 C	675 S	ia	
	ν_5	CCl_3 s-deform.....	372 C	372 S	ia	
	e_g	CCl_3 d-stretch.....	859 C	ia	859 W	
	ν_7	CCl_3 d-deform.....	340 C	ia	340 M	
	ν_8	CCl_3 rock.....	223 C	ia	223 S	
	e_u	CCl_3 d-stretch.....	778 C	778 VS	ia	
	ν_{10}	CCl_3 d-deform.....	271 C	271 S	ia	
	ν_{12}	CCl_3 rock.....	114 C	114 W	ia	

References

- [1] IR.R. R. A. Carney, E. A. Piotrowski, A. G. Meister, J. H. Braun, and F. F. Cleveland, J. Mol. Spectrosc. 7, 209 (1961).
- [2] IR.R.Th. T. Fujiyama and T. Shimanouchi, unpublished.

Molecule: Hexabromoethane CBr_3CBr_3
 Symmetry D_{3d} Symmetry number $\delta = 6$

No. 159

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CC stretch.....	940 C	cm^{-1} (Solid)	cm^{-1} (Solid)	
	ν_2	CBr_3 s-stretch.....	255 C	ia	940 M	
	ν_3	CBr_3 s-deform.....	120 C	ia	255 VS, p	
	ν_4	Torsion.....	51 D	ia	120 W	
	a_{1u}	CBr_3 s-stretch.....	559 C	559 S	ia	
	a_{2u}	CBr_3 s-deform.....	254 C	254 S	ia	
	e_g	CBr_3 d-stretch.....	768 C	ia	768 M, dp	
	ν_7	CBr_3 d-deform.....	204 C	ia	204 S, dp	
	ν_8	CBr_3 rock.....	139 C	ia	139 M	
	e_u	CBr_3 d-stretch.....	656 C	656 VS	ia	
	ν_{10}	CBr_3 d-deform.....	168 C	168 S	ia	
	ν_{12}	CBr_3 rock.....	82 C	82 M	ia	

References

See No. 158.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CH_2 s-stretch.....	2957 D	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CH_2 scis.....	1445 C	ia	1445 (4b) dp	
	ν_3	CH_2 wag.....	1304 C	ia	1304 (6) p	
	ν_4	CC stretch.....	1052 C	ia	1052 (4) p	
	ν_5	CCl stretch.....	754 C	ia	754 (10b) p	
	ν_6	CCl deform.....	300 C	ia	300 (8) p	
a_u	ν_7	CH_2 a-stretch.....	3005 D	3005 W (liquid)	ia	SF (gauche ν_1 , gauche ν_{11}).
	ν_8	CH_2 twist.....	1123 B	1122.5 W	ia	
b_g	ν_9	CH_2 rock.....	773 B	772.5 M	ia	
	ν_{10}	Torsion.....	123 C	123 M	ia	
	ν_{11}	CH_2 a-stretch.....	3005 D	ia	3005 (8b) dp	
b_u	ν_{12}	CH_2 twist.....	1264 C	ia	1264 (3) dp	
	ν_{13}	CH_2 rock.....	989 C	ia	989 (2) p	
	ν_{14}	CH_2 s-stretch.....	2983 C	2983.3 M	ia	
b_u	ν_{15}	CH_2 scis.....	1461 A	1460.6 S	ia	
	ν_{16}	CH_2 wag.....	1232 B	1232.3 S	ia	
	ν_{17}	CCl stretch.....	728 C	728.3 VS	ia	
	ν_{18}	CCl deform.....	222 C	222.3 W	ia	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR.R. S. Mizushima, Y. Morino, I. Watanabe, T. Shimanouchi, and S. Yamaguchi, J. Chem. Phys. 17, 591 (1949).
- [3] IR.R. John T. Neu and Wm. D. Gwinn, J. Chem. Phys. 18, 1642 (1950).
- [4] IR.R. J. K. Brown and N. Sheppard, Trans. Faraday Soc. 48, 128 (1952).
- [5] Th. I. Nakagawa and S. Mizushima, J. Chem. Phys. 21, 2195 (1953).
- [6] IR. I. Ichishima, H. Kamiyama, T. Shimanouchi, and S. Mizushima, J. Chem. Phys. 29, 1190 (1958).
- [7] IR. S. Mizushima, T. Shimanouchi, I. Ichishima, and H. Kamiyama, Revue Universelle des Mines 15, 447 (1959).

Molecule: 1,2-Dichloroethane $\text{CH}_2\text{ClCH}_2\text{Cl}$ (gauche form)
 Symmetry C_2 Symmetry number $\delta = 2$

No. 161

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i>	ν_1	CH_2 a-stretch.....	3005 D	cm^{-1} (Gas) (liquid)	3005 (8b) dp	SF (ν_{11} , trans ν_7).
	ν_2	CH_2 s-stretch.....	2957 D	2957 M (liquid)	2957 (10) p	SF (trans ν_1 , trans ν_{14}).
	ν_3	CH_2 scis.....	1433 C	1433 M (liquid)	1429 (6) dp	OV (ν_{13}).
	ν_4	CH_2 wag.....	1315 C	1315 W	1304 (6)	
	ν_5	CH_2 twist.....	1207 C		1207 (5) p	
	ν_6	CC stretch.....	1027 D	1027 W	1031 (2) dp	
	ν_7	CH_2 rock.....	948 B	947.7 M	943 (5) p	
	ν_8	CCl stretch.....	669 C	669 M	654 (8) p	
	ν_9	CCl deform.....	272 D	272 VW (liquid)	265 (5) p	
<i>b</i>	ν_{10}	Torsion.....			125 (5b)	
	ν_{11}	CH_2 a-stretch.....	3005 D	3005 W	3005 (8b) dp	SF (ν_1 , trans ν_7).
	ν_{12}	CH_2 s-stretch.....	2957 C	2957.2 W		
	ν_{13}	CH_2 scis.....	1436 B	1436.3 W		
	ν_{14}	CH_2 wag.....	1292 B	1292.1 S		
	ν_{15}	CH_2 twist.....	1146 D	1146 VW	1145 (3) dp	
	ν_{16}	CH_2 rock.....	890 B	890.3 M	881 (4) dp	
	ν_{17}	CCl stretch.....	693 B	692.5 W	677 (6b) dp	
	ν_{18}	CCl deform.....	410 C	409.6 M	411 (5) dp	

References

See No. 160.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CH_2 s-stretch.....	2972 D	cm^{-1} (Liquid)	cm^{-1} (Liquid)	
	ν_2	CH_2 scis.....	1440 C	ia	2972 (10) p	
	ν_3	CH_2 wag.....	1255 C	ia	1440 (5) dp	
	ν_4	CC stretch.....	1053 C	ia	1255 (10b) p	SF (ν_{12}).
	ν_5	CBr stretch.....	660 C	ia	1053 (9) dp	
	ν_6	CCBr deform.....	190 C	ia	660 (10b) p	
	ν_7	CH_2 a-stretch.....	3037 D	3037 S	190 (10) p	
	ν_8	CH_2 twist.....	1087 C	1087 M	ia	
	ν_9	CH_2 rock.....	753 C	753 S	ia	
	ν_{10}	Torsion.....	118 D	118 (gas)	132 (0)	
b_g	ν_{11}	CH_2 a-stretch.....	3013 D	ia	3013 (4b) dp	
	ν_{12}	CH_2 twist.....	1255 C	ia	1255 (10b) p	
	ν_{13}	CH_2 rock.....	933 C	ia	933 (2) p	SF (ν_3).
	ν_{14}	CH_2 s-stretch.....	2974 D	2974 S	ia	
b_u	ν_{15}	CH_2 scis.....	1441 D	1441 M	ia	
	ν_{16}	CH_2 wag.....	1186 C	1186 VS	1186 (0)	
	ν_{17}	CBr stretch.....	589 C	589 S	ia	
	ν_{18}	CCBr deform.....	193 D	193	ia	

References

- [1] IR.R. S. Mizushima, Y. Morino, I. Watanabe, T. Shimanouchi, and S. Yamaguchi, J. Chem. Phys. 17, 591 (1949).
- [2] IR.R. John T. Neu and Wm. D. Gwinn, J. Chem. Phys. 18, 1642 (1950).
- [3] IR.R. J. K. Brown and N. Sheppard, Trans. Faraday Soc. 48, 128 (1952).
- [4] IR. I. Ichishima, H. Kamiyama, T. Shimanouchi, and S. Mizushima, J. Chem. Phys. 29, 1190 (1958).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i>	ν_1	CH_2 a-stretch.....	3005 D	cm^{-1} (Liquid)	3005 (5)	SF (ν_{11}).
	ν_2	CH_2 s-stretch.....	2953 D	2953 VS	2953 (8) p	SF (ν_{12}).
	ν_3	CH_2 scis.....	1420 C	1420 M	1419 (3) dp	SF (ν_{13}).
	ν_4	CH_2 wag.....	1278 C	1278 M	1276 (3)	
	ν_5	CH_2 twist.....	1104 C	1104 M	1104 (1) dp	SF (ν_{15}).
	ν_6	CC stretch.....	1019 C	1019 M	1019 (1)	
	ν_7	CH_2 rock.....	898 C	898 M	899 (3) p	
	ν_8	CBr stretch.....	550 C	550 M	551 (8)	
	ν_9	CCBr deform.....	231 C	231 (3) p	
	ν_{10}	Torsion.....	91 D	91 (2b) dp	
<i>b</i>	ν_{11}	CH_2 a-stretch.....	3005 D	3005 (5)	SF (ν_1).
	ν_{12}	CH_2 s-stretch.....	2953 D	2953 VS	2953 (8) p	SF (ν_2).
	ν_{13}	CH_2 scis.....	1420 C	1420 M	1419 (3) dp	SF (ν_3).
	ν_{14}	CH_2 wag.....	1245 C	1245 S	1243 (1)	
	ν_{15}	CH_2 twist.....	1104 C	1104 W	1104 (1) dp	SF (ν_5).
	ν_{16}	CH_2 rock.....	836 C	836 S	836 (2) dp	
	ν_{17}	CBr stretch.....	589 C	589 S	583 (6b) dp	
	ν_{18}	CCBr deform.....	355 C	355	355 (5) dp	

References

See No. 162.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_2 s-stretch.....	2960 D	cm^{-1} (Solid)	cm^{-1} (Liquid)	SF (ν_2 , gauche ν_3 , gauche ν_4). SF (ν_1 , gauche ν_3 , gauche ν_4).
	ν_2	CH_2 s-stretch.....	2960 D	2960 (10vb)	
	ν_3	CH_2 scis.....	1446 D	1446 S		
	ν_4	CH_2 scis.....	1444 C	1444 (3b)	
	ν_5	CH_2 wag.....	1284 C	1284 M	1284 (7) p	
	ν_6	CH_2 wag.....	1203 C	1203 S	1203 (3)	
	ν_7	CC stretch.....	1052 C	1056 M	1052 (4) dp	
	ν_8	CCl_4 stretch.....	726 C	722 S	726 (10b) p	
	ν_9	CBr stretch.....	630 C	630 S	630 (9)	
	ν_{10}	CCl_4 deform.....	251 C	251 (10) p	SF (gauche ν_{17}).
	ν_{11}	CCBr deform.....	202 C	202.0 (CCl_4 soln.)	210 (2b)	
	ν_{12}	CH_2 a-stretch.....	3010 D	3010 (3vb)	SF (ν_{13} , gauche ν_1 , gauche ν_2).
	ν_{13}	CH_2 a-stretch.....	3010 D	3010 (3vb)	SF (ν_{12} , gauche ν_1 , gauche ν_2).
a''	ν_{14}	CH_2 twist.....	1259 C	1258 VW	1259 (3)	
	ν_{15}	CH_2 twist.....	1111 D	1111 M		
	ν_{16}	CH_2 rock.....	961 C	961 VW	961 (1b)	
	ν_{17}	CH_2 rock.....	763 D	763 M		
	ν_{18}	Torsion.....	123 C	123 (CCl_4 soln.)		

References

- [1] R. S. Mizushima, Y. Morino, Y. Miyahara, and M. Tomura, Sci. Pap. Inst. Phys. Chem. Res. Japan 39, 387 (1942).
- [2] IR.R. J. K. Brown and N. Sheppard, Trans. Faraday Soc. 48, 128 (1952).
- [3] IR. L. R. Blaine, J. Res. NBS 67C (Engr. and Instr.) No. 3, 207 (1963).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i>	ν_1	CH ₂ a-stretch.....	3010 D	cm^{-1} (Liquid)	cm^{-1} (Liquid) 3010 (3vb)	SF (ν_2 , trans ν_{12} , trans ν_{13}).
	ν_2	CH ₂ a-stretch.....	3010 D	3010 (3vb)	SF (ν_1 , trans ν_{12} , trans ν_{13}).
	ν_3	CH ₂ s-stretch.....	2960 D	2960 (10vb)	SF (ν_4 , trans ν_1 , trans ν_2).
	ν_4	CH ₂ s-stretch.....	2960 D	2960 (10vb)	SF (ν_3 , trans ν_1 , trans ν_2).
	ν_5	CH ₂ scis.....	1428 D	1428 S	1421 (3b)	OV (ν_6).
	ν_6	CH ₂ scis.....	1428 D	1428 S	1421 (3b)	OV (ν_5).
	ν_7	CH ₂ wag.....	1299 C	1299 S	1299 (1)	
	ν_8	CH ₂ wag.....	1260 C	1260 S	1259 (3)	
	ν_9	CH ₂ twist.....	1190 D	1190 M	1189 (2) p	
	ν_{10}	CH ₂ twist.....	1127 C	1127 M	1128 (1) dp	
	ν_{11}	CC stretch.....	1025 C	1025 M	1023 (1)	
	ν_{12}	CH ₂ rock.....	923 C	923 S	919 (3) p	
	ν_{13}	CH ₂ rock.....	856 C	856 S	852 (2)	
	ν_{14}	CCl stretch.....	664 C	664 S	665 (6)	
	ν_{15}	CBr stretch.....	571 C	571 S	568 (9) p	
	ν_{16}	CCCl deform.....	385 C	385 (3) dp	
	ν_{17}	CCBr deform.....	251 D	251 (10)	
	ν_{18}	Torsion.....	107 D	107 (2b)	SF (trans ν_{10}).

References

See No. 164.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_3 d-stretch.....	3003 D	cm^{-1} (Gas)	3003 VS	2986 VS, dp
	ν_2	CH_2 s-stretch.....	2941 D		2941	2941 VS, p
	ν_3	CH_3 s-stretch.....	2915 D		2915 S	2921 M
	ν_4	CH_2 scis.....	1479 C		1479 M	1480 W, b, dp
	ν_5	CH_3 d-deform.....	1449 D		1449 S	1458 M, b, dp
	ν_6	CH_3 s-deform.....	1395 C		1395 S	1393 W, p
	ν_7	CH_2 wag.....	1365 D		1365 M (liquid)	1365 VW
	ν_8	CH_3 rock.....	1108 C		1108 VS	1103 S, p
	ν_9	CC stretch.....	1048 D		1048 VS	1041 M, b, dp
	ν_{10}	CF stretch.....	880 B		880 VS	873 VS, p
	ν_{11}	CCF deform.....	415 C		415	419 W, p
	ν_{12}	CH_2 a-stretch.....	3003 D		3003 VS	2986 VS, dp
	ν_{13}	CH_3 d-stretch.....	3003 D		3003 VS	2986 VS, dp
	ν_{14}	CH_3 d-deform.....	1449 D		1449 S	1458 M, b, dp
	ν_{15}	CH_2 twist.....	1277 C		1277	1276 W, b, dp
	ν_{16}	CH_3 rock.....	1048 D		1048 VS	1041 M, b, dp
	ν_{17}	CH_2 rock.....	810 C		810 W	815 WV
	ν_{18}	Torsion.....	243 B		243	

References

- [1] IR.R. D. C. Smith, R. A. Saunders, J. Rud Nielsen, and E. E. Ferguson, J. Chem. Phys. 20, 847 (1952).
- [2] IR. E. Catalano and K. S. Pitzer, J. Phys. Chem. 62, 873 (1958).
- [3] IR. G. Sage and W. Klemperer, J. Chem. Phys. 39, 371 (1963).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_2 s-stretch.....	2967 D	cm^{-1} (Gas) 2977 M (solid)	cm^{-1} (Liquid) 2967 M, p	
	ν_2	CH_3 d-stretch.....	2946 C	2946 S	2934 M, p	
	ν_3	CH_3 s-stretch.....	2881 C	2881 S	2883 W, p	
	ν_4	CH_3 d-deform.....	1463 D	1463 S (solid)		
	ν_5	CH_2 scis.....	1448 D	1448 S	1453 M, dp	OV (ν_{14}).
	ν_6	CH_3 s-deform.....	1385 C	1385 S	1383 W, dp	
	ν_7	CH_2 wag.....	1289 C	1289 VS	1283 W, p	
	ν_8	CH_3 rock.....	1081 D	1081 VW	1072 M, p	
	ν_9	CC stretch.....	974 D	974 VS	969 W, dp	OV (ν_{16}).
	ν_{10}	CCl stretch.....	677 C	677 VS	659 VS, p	
	ν_{11}	CCl deform.....	336 C	336 M	337 S, p	
	ν_{12}	CH_2 a-stretch.....	3014 D	3014 VS	3013 W	
	ν_{13}	CH_3 d-stretch.....	2986 D	2986 VS	2978 W	
	ν_{14}	CH_3 d-deform.....	1448 D	1448 S	1453 M, dp	OV (ν_5).
	ν_{15}	CH_2 twist.....	1251 D	1251 VW	1248 W, dp	
	ν_{16}	CH_3 rock.....	974 D	974 VS	969 W, dp	OV (ν_9).
	ν_{17}	CH_2 rock.....	786 B	786 M		
	ν_{18}	Torsion.....	251 B	251 W		MW: 251 [4].

References

- [1] IR.R. L. W. Daasch, C. Y. Liang, and J. Rud Nielsen, *J. Chem. Phys.* **22**, 1293 (1954).
- [2] R. G. Allen and H. J. Bernstein, *Can. J. Chem.* **32**, 1124 (1954).
- [3] IR. R. N. Dixon, *Spectrochim. Acta* **9**, 59 (1957).
- [4] MW. D. R. Lide, *J. Chem. Phys.* **30**, 37 (1959).
- [5] IR. W. G. Fateley and F. A. Miller, *Spectrochim. Acta* **17**, 857 (1961).
- [6] IR. N. T. McDevitt, A. L. Rozek, F. F. Bentley, and A. D. Davidson, *J. Chem. Phys.* **42**, 1173 (1965).
- [7] IR. F. A. Miller and F. E. Kiviat, *Spectrochim. Acta* **25A**, 1363 (1969).
- [8] Th. A. B. Dempster and G. Zerbi, *J. Mol. Spectrosc.* **39**, 1 (1971).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_3 d-stretch.....	2988 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CH_2 s-stretch.....	2937 B	2936.5 S	2971 (2b) P	OV (ν_{13}).
	ν_3	CH_3 s-stretch.....	2880 B	2879.8 S	2924 (2) P	
	ν_4	CH_2 scis.....	1451 D	1451 M	1442 (2b) dp	OV (ν_5, ν_{14}).
	ν_5	CH_3 d-deform.....	1451 D	1451 M	1442 (2b) dp	OV (ν_4, ν_{14}).
	ν_6	CH_3 s-deform.....	1386 B	1386 M		
	ν_7	CH_2 wag.....	1252 E	{ 1258 VS 1247 VS }	1248 (2b) p	FR ($\nu_9 + \nu_{11}$).
	ν_8	CH_3 rock.....	1061 D	1061 VW	1069 (1) p	
	ν_9	CC stretch.....	964 B	964 S	960 (1b) dp	OV (ν_{15}).
	ν_{10}	CBr stretch.....	583 B	583 VS	560 (10) p	
	ν_{11}	CCBr deform.....	290 B	290 S	292 (3) p	
	ν_{12}	CH_2 a-stretch.....	3018 B	3018 S		
	ν_{13}	CH_3 d-stretch.....	2988 C	2988 S	2971 (2b) p	OV (ν_1).
	ν_{14}	CH_3 d-deform.....	1451 D	1451 M	1442 (2b) dp	OV (ν_4, ν_6).
	ν_{15}	CH_2 twist.....	1248 E			CF [7].
a''	ν_{16}	CH_3 rock.....	964 D	964 S	960 (1b) dp	OV (ν_9).
	ν_{17}	CH_2 rock.....	770 B	770 M		
	ν_{18}	Torsion.....	247 C	247		MW: 247 [5, 6].

References

- [1] R. J. Wagner, Z. Phys. Chem. B40, 439 (1938).
- [2] R. J. Wagner, Z. Phys. Chem. B45, 69 (1939).
- [3] R. J. K. Brown and N. Sheppard, Trans. Faraday Soc. 50, 535 (1954).
- [4] IR. R. N. Dixon, Spectrochim. Acta 9, 59 (1957).
- [5] MW. D. R. Lide, J. Chem. Phys. 30, 37 (1959).
- [6] MW. C. Flanagan and L. Pierce, J. Chem. Phys. 38, 2963 (1963).
- [7] IR.Th. T. Shimanouchi and S. Takayama, unpublished.
- [8] IR. N. T. McDevitt, A. L. Rozek, F. F. Bentley, and A. D. Davidson, J. Chem. Phys. 42, 1173 (1965).
- [9] IR. S. Kinumaki and Y. Musha, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	NH stretch.....	3338 C	cm^{-1} (Gas)	3338 W	3302 M, p
	ν_2	CH ₂ a-stretch.....	3079 D	3079 S	3059 M, dp	OV (ν_{11}).
	ν_3	CH ₂ s-stretch.....	3015 D	3015 S	2999 VS, p	OV (ν_{12}).
	ν_4	CH ₂ scis.....	1482 C	1482 W	1471 W, p	
	ν_5	Ring stretch.....	1211 C	1211 S	1212 VS, p	
	ν_6	CH ₂ twist.....	1095 D	1095 S	1088 W, p	
	ν_7	CH ₂ wag.....	1090 D	1090 S	1088 W, p	
	ν_8	NH bend.....	998 C	998 M	1028 W	
	ν_9	Ring deform.....	856 C	856 VS	855 M, dp	
	ν_{10}	CH ₂ rock.....	773 C	773 S	787 W, dp	
	ν_{11}	CH ₂ a-stretch.....	3079 D	3079 S	3059 M, dp	OV (ν_2).
	ν_{12}	CH ₂ s-stretch.....	3015 D	3015 S	2999 VS, p	OV (ν_3).
	ν_{13}	CH ₂ scis.....	1463 C	1463 W	1452 W, dp	
	ν_{14}	CH ₂ twist.....	1268 C	1268 M	1276 VW	
	ν_{15}	NH bend.....	1237 C	1237 M	1297 W, p	
	ν_{16}	CH ₂ wag.....	1131 C	1131 M	1130 VW	
	ν_{17}	Ring deform.....	904 C	904 S		
	ν_{18}	CH ₂ rock.....	817 D	817 M, dp	

References

- [1] IR.R. H. W. Thompson and W. T. Cave, Trans. Faraday Soc. 47, 951 (1951).
- [2] IR. H. T. Hoffman, Jr., G. E. Evans, and G. Glockler, J. Amer. Chem. Soc. 73, 3028 (1951).
- [3] IR.R. W. J. Potts, Spectrochim. Acta 21, 511 (1965).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₃ d-stretch.....	3045 D	<i>cm</i> ⁻¹ (Gas)	3045 M	3030 (3b)
	<i>v</i> ₂	CH ₃ s-stretch.....	2969 D	2969 S	2955 (10) p	
	<i>v</i> ₃	CH stretch.....	2943 D	2943 S		
	<i>v</i> ₄	C=O stretch.....	1754 C	1754 VS	1717 (5b) p	
	<i>v</i> ₅	CH ₃ d-deform.....	1454 D	1454 W (CCl ₄ soln.)		
	<i>v</i> ₆	CH ₃ s-deform.....	1445 D	1445 M		
	<i>v</i> ₇	CH bend.....	1371 D	1371 W	1379 (4b) p	
	<i>v</i> ₈	C-O stretch.....	1207 C	1207 VS	1207 (0.5b)	
	<i>v</i> ₉	CH ₃ rock.....	1166 D	1166 VS	1157 (1b)	
	<i>v</i> ₁₀	O-CH ₃ stretch.....	925 C	925 S	912 (10) p	
	<i>v</i> ₁₁	OCO deform.....	767 C	767 M	765 (2)	
	<i>v</i> ₁₂	COC deform.....	318 D	318 M		
	<i>v</i> ₁₃	CH ₃ d-stretch.....	3012 D	3012 M		
	<i>v</i> ₁₄	CH ₃ d-deform.....	1443 E	1443 W (CCl ₄ soln.)	1440 (3b)	
	<i>v</i> ₁₅	CH ₃ rock.....	1168 D	1168 M		
	<i>v</i> ₁₆	CH bend.....	1032 C	1032 M	1030 (0.5)	
	<i>v</i> ₁₇	C-O torsion.....	332 D	332 M	332 (3b) p	
	<i>v</i> ₁₈	CH ₃ torsion.....	130 D	130 VW		MW: 132 [3].

References

- [1] R. K. W. F. Kohlrausch, Ramanspektren, p. 263 (Edwards Bros., Inc., Ann Arbor, 1945).
- [2] IR.R. J. K. Wilmsurst, J. Mol. Spectrosc. 1, 201 (1957).
- [3] MW. R. F. Curl, J. Chem. Phys. 30, 1529 (1959).
- [4] IR. W. G. Fateley and F. A. Miller, Spectrochim. Acta 17, 857 (1961).
- [5] IR. H. Susi and T. Zell, Spectrochim. Acta 19, 1933 (1963).
- [6] IR.Th. S. Ichikawa, K. Toriyama, and T. Shimanouchi, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	CH ₃ d-stretch.....	3041 C	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CH ₃ s-stretch.....	2967 C	3041 M 2967 S		
	ν_3	CD stretch.....	2216 C	2216 S		
	ν_4	C=O stretch.....	1739 E	{ 1755 VS 1716 VS		
	ν_5	CH ₃ d-deform.....	1448 E	1448 W (CCl ₄ soln.)		FR (2 ν_{16}).
	ν_6	CH ₃ s-deform.....	1441 D	1441 M		
	ν_7	CD bend.....	1048 D	1048 M		
	ν_8	C-O stretch.....	1213 C	1213 VS		
	ν_9	CH ₃ rock.....	1157 D	1157 VS		
	ν_{10}	O-CH ₃ stretch.....	878 C	878 S		
	ν_{11}	OCO deform.....	762 C	762 M		
	ν_{12}	COC deform.....	315 E	315 M		
	ν_{13}	CH ₃ d-stretch.....	3007 D	3007 S		
	ν_{14}	CH ₃ d-deform.....	1440 E	1440 W (CCl ₄ soln.)		
	ν_{15}	CH ₃ rock.....	1164 E		CF [2], OV (ν_9).
	ν_{16}	CD bend.....	870 E		CF [2], OV (ν_{10}).
	ν_{17}	C-O torsion.....	290 E	290 M		
	ν_{18}	CH ₃ torsion.....	130 E		CF [2].

References

- [1] IR. H. Susi and T. Zell, Spectrochim. Acta 19, 1933 (1963).
 [2] IR.Th. S. Ichikawa, K. Toriyama, and T. Shimanouchi, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CD ₃ d-stretch.....	2284 D	2284 cm ⁻¹ (Gas)	
	<i>v</i> ₂	CD ₃ s-stretch.....	2087 D	2087 M	
	<i>v</i> ₃	CH stretch.....	2931 D	2931 S	
	<i>v</i> ₄	C=O stretch.....	1754 C	1754 VS	
	<i>v</i> ₅	CD ₃ d-deform.....	1060 E	1060 W	OV (<i>v</i> ₁₄).
	<i>v</i> ₆	CD ₃ s-deform.....	1102 E	1102 S	
	<i>v</i> ₇	CH ip-bend.....	1368 D	1368 M	
	<i>v</i> ₈	C-O stretch.....	1210 C	1210 VS	
	<i>v</i> ₉	CD ₃ rock.....	985 D	985 M	
	<i>v</i> ₁₀	O-CD ₃ stretch.....	877 C	877 M	
	<i>v</i> ₁₁	OCO deform.....	714 C	714 M	
	<i>v</i> ₁₂	COC deform.....	297 E	297 M	
	<i>v</i> ₁₃	CD ₃ d-stretch.....	2258 D	2258 M	
	<i>v</i> ₁₄	CD ₃ d-deform.....	1060 E	1060 W	OV (<i>v</i> ₅).
	<i>v</i> ₁₅	CD ₃ rock.....	905 D	905 W	
	<i>v</i> ₁₆	CH op-bend.....	1040 E	1040 W	
	<i>v</i> ₁₇	C-O torsion.....	312 E	312 M	
	<i>v</i> ₁₈	CD ₃ torsion.....	96 E	CF [2].

References

See No. 171.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	CD ₃ d-stretch.....	2291 D	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CD ₃ s-stretch.....	2100 D	2291 M	2100 M	
	ν_3	CD stretch.....	2210 D	2100 M	2210 S	
	ν_4	C=O stretch.....	1739 E	{ 1749 VS 1719 VS	FR ($2\nu_{16}$).
	ν_5	CD ₃ d-deform.....	1060 E	1060 W	
	ν_6	CD ₃ s-deform.....	1107 D	1107 S	OV (ν_{14}).
	ν_7	CD bend.....	1041 E	1041 W	
	ν_8	C-O stretch.....	1203 D	1203 VS	
	ν_9	CD ₃ rock.....	974 D	974 M	
	ν_{10}	O-CD ₃ stretch.....	840 D	840 M	
	ν_{11}	OCO deform.....	708 D	708 M	
	ν_{12}	COC deform.....	295 E	295 M	
	ν_{13}	CD ₃ d-stretch.....	2267 D	2267 M	
	ν_{14}	CD ₃ d-deform.....	1060 D	1060 W	OV (ν_4).
	ν_{15}	CD ₃ rock.....	908 D	908 M	
	ν_{16}	CD op-bend.....	870 D	870 W	
	ν_{17}	C-O torsion.....	280 D	280 M	
	ν_{18}	CD ₃ torsion.....	96 E	CF [1].

Reference

[1] IR.Th. S. Ichikawa, K. Toriyama, and T. Shimanouchi, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	OH stretch.....	3583 B	cm^{-1} (Gas)	cm^{-1}	
	ν_2	CH_3 d-stretch.....	3051 B	3051 VW		
	ν_3	CH_3 s-stretch.....	2944 B	2944 VW		
	ν_4	$\text{C}=\text{O}$ stretch.....	1788 B	1788 VS		
	ν_5	CH_3 d-deform.....	1430 C	1430 sh		
	ν_6	CH_3 s-deform.....	1382 B	1382 M		
	ν_7	OH bend.....	1264 B	1264 M		
	ν_8	C-O stretch.....	1182 B	1182 S		
	ν_9	CH_3 rock.....	989 B	989 M		
	ν_{10}	CC stretch.....	847 B	847 W		
	ν_{11}	OCO deform.....	657 B	657 S		
	ν_{12}	CCO deform.....	581 B	581 M		
	ν_{13}	CH_3 d-stretch.....	2996 B	2996 VW		
	ν_{14}	CH_3 d-deform.....	1430 C	1430 sh		
	ν_{15}	CH_3 rock.....	1048 B	1048 W		
	ν_{16}	$\text{C}=\text{O}$ op-bend.....	642 B	642 S		
	ν_{17}	C-O torsion.....	534 B	534 M		
	ν_{18}	CH_3 torsion.....	93 E	CF [3].
References						
[1] IR.		W. Weltner, J. Amer. Chem. Soc. 77, 3941 (1955).				
[2] IR.		J. K. Wilmsurst, J. Chem. Phys. 25, 1171 (1956).				
[3] IR.		M. Haurie and A. Novak, J. Chim. Phys. 62, 137 (1965).				
[4] IR.		M. Ohara and T. Shimanouchi, unpublished.				

Molecule: Acetic acid-d₁ CH₃COOD
 Symmetry C_s Symmetry number δ = 1

No. 175

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₃ d-stretch.....	3039 B	<i>cm</i> ⁻¹ (Gas)	3039 VW	
	<i>v</i> ₂	CH ₃ s-stretch.....	2952 B		2952 VW	
	<i>v</i> ₃	OD stretch.....	2642 B		2642 M	
	<i>v</i> ₄	C=O stretch.....	1775 B		1775 VS	
	<i>v</i> ₅	CH ₃ d-deform.....	1440 C		1440 sh	
	<i>v</i> ₆	CH ₃ s-deform.....	1383 B		1383 S	
	<i>v</i> ₇	C-O stretch.....	1270 B		1270 S	
	<i>v</i> ₈	CH ₃ rock.....	990 D		990 sh	
	<i>v</i> ₉	OD bend.....	955 B		955 S	
	<i>v</i> ₁₀	CC stretch.....	840 B		840 W	
	<i>v</i> ₁₁	OCO deform.....	609 B		609 M	
	<i>v</i> ₁₂	CCO deform.....	543 B		543 M	
	<i>v</i> ₁₃	CH ₃ d-stretch.....	2997 D		2997 VW	
	<i>v</i> ₁₄	CH ₃ d-deform.....	1440 C		1440 sh	
	<i>v</i> ₁₅	CH ₃ rock.....	1052 B		1052 W	
	<i>v</i> ₁₆	C=O ip-bend.....	603 B		603 M	
	<i>v</i> ₁₇	C-O torsion.....	415 B		415 M	
	<i>v</i> ₁₈	CH ₃ torsion.....	93 E		CF [3].

References

See No. 174.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 d-stretch.....	2996 B	cm^{-1} (Gas)	2996 S	
	ν_2	CH_3 s-stretch.....	2817 B	2817 S	2815 VS, p	
	ν_3	CH_3 d-deform.....	1464 D	1464 M		
	ν_4	CH_3 s-deform.....	1452 D	1452 M	1452 S, dp	
	ν_5	CH_3 rock.....	1244 B	1244 W		
	ν_6	CO s-stretch.....	928 B	928 S	922 S, p	
	ν_7	COC deform.....	418 C	418 M	428 M, p	
	ν_8	CH_3 d-stretch.....	2952 C	ia	2952 S	
	ν_9	CH_3 d-deform.....	1464 D	ia		SF (ν_3).
	ν_{10}	CH_3 rock.....	1150 C	ia	1150 M, d	
b_1	ν_{11}	Torsion.....	203 E	ia		CF [3].
	ν_{12}	CH_3 d-stretch.....	2996 B	2996 S	2989 S	OV (ν_1).
	ν_{13}	CH_3 s-stretch.....	2817 B	2817 S	2815 VS, p	OV (ν_2).
	ν_{14}	CH_3 d-deform.....	1464 D	1464 M		OV (ν_3).
	ν_{15}	CH_3 s-deform.....	1452 D	1452 M	1452 S, dp	OV (ν_4).
b_2	ν_{16}	CH_3 rock.....	1227 C		1227 W	
	ν_{17}	CO a-stretch.....	1102 B	1102 VS	1104 M, dp	
	ν_{18}	CH_3 d-stretch.....	2925 B	2925 S		
	ν_{19}	CH_3 d-deform.....	1464 D	1464 M		
	ν_{20}	CH_3 rock.....	1179 B	1179 VS	1170 sh	OV (ν_3).
	ν_{21}	Torsion.....	242 C	242 W		

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] R. R. C. Taylor and G. L. Vidale, J. Chem. Phys. 26, 122 (1957).
- [3] IR.R. Y. Kanazawa and K. Nukada, Bull. Chem. Soc. Japan 35, 612 (1962).
- [4] IR. W. G. Fateley and F. A. Miller, Spectrochim. Acta 18, 977 (1962).
- [5] IR. J.-P. Perchard, M.-T. Forel, and M.-L. Josien, J. Chim. Phys. 61, 632 (1964).
- [6] Th. T. Shimanouchi and M. Oka, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>ν</i> ₁	CH ₃ d-stretch.....	2992 B	2992 S <i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹	
	<i>ν</i> ₂	CH ₃ s-stretch.....	2819 B	2819 S		
	<i>ν</i> ₃	CD ₃ d-stretch.....	2244 B	2244 S		
	<i>ν</i> ₄	CD ₃ s-stretch.....	2058 B	2058 S		
	<i>ν</i> ₅	CH ₃ d-deform.....	1465 C	1465 M		
	<i>ν</i> ₆	CH ₃ s-deform.....	1453 C	1453 M		
	<i>ν</i> ₇	CH ₃ rock.....	1212 B	1212 M		
	<i>ν</i> ₈	CO a-stretch.....	1156 C	1156 VS		SF (<i>ν</i> ₁₇).
	<i>ν</i> ₉	CD ₃ s-deform.....	1111 B	1111 S		
	<i>ν</i> ₁₀	CD ₃ d-deform.....	1061 C	1061 M		SF (<i>ν</i> ₁₈).
	<i>ν</i> ₁₁	CD ₃ rock.....	947 C	947 W		
	<i>ν</i> ₁₂	CO s-stretch.....	860 C	860 M		
	<i>ν</i> ₁₃	COC deform.....	395 E		CF [2].
	<i>ν</i> ₁₄	CH ₃ d-stretch.....	2932 B	2932 S		
	<i>ν</i> ₁₅	CD ₃ d-stretch.....	2189 B	2189 S		
	<i>ν</i> ₁₆	CH ₃ d-deform.....	1462 D	1462 M		
	<i>ν</i> ₁₇	CH ₃ rock.....	1156 C	1156 VS		SF (<i>ν</i> ₈). SF (<i>ν</i> ₁₀).
	<i>ν</i> ₁₈	CD ₃ d-deform.....	1061 C	1061 M		
	<i>ν</i> ₁₉	CD ₃ rock.....	901 C	901 W		
	<i>ν</i> ₂₀	Torsion.....	227 E		CF [2].
	<i>ν</i> ₂₁	Torsion.....	164 E		CF [2].

References

- [1] IR. J.-P. Perchard, M.-T. Forel, and M.-L. Josien, J. Chim. Phys. **61**, 632 (1964).
 [2] Th. T. Shimanouchi and M. Oka, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CII ₂ s-stretch.....	3015 Å	cm^{-1} (Gas)	3015.0	
	ν_2	CH ₂ scis.....	1443 Å	1442.6	
	ν_3	CC stretch.....	1073 Å	1072.6	
	ν_4	CH ₂ twist.....	865 C	865	865 (liquid)	
b_2	ν_5	CH ₂ s-stretch.....	3007 Å	3006.7		
	ν_6	CC stretch.....	1957 C	1957	1960 (liquid)	
e	ν_7	CH ₂ scis.....	1398 C	1398	1421 (liquid)	
	ν_8	CH ₂ a-stretch.....	3086 Å	3085.5		
	ν_9	CH ₂ rock.....	999 Å	999.1		
	ν_{10}	CH ₂ wag.....	841 Å	840.8		
	ν_{11}	CCC deform.....	355 Å	355.3		

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR. C. H. Miller and H. W. Thompson, Proc. Roy. Soc. (London), Ser. A, 200, 1 (1949).
- [3] IR.R. R. C. Lord and P. Venkateswarlu, J. Chem. Phys. 20, 1237 (1952).
- [4] IR. J. Overend and H. W. Thompson, J. Opt. Soc. Amer. 43, 1065 (1953).
- [5] R. B. P. Stoicheff, Can. J. Phys. 33, 811 (1955).
- [6] IR. K. N. Rao and E. D. Palik, J. Chem. Phys. 23, 2112 (1955).
- [7] IR. J. Overend and H. W. Thompson, Trans. Faraday Soc. 52, 1295 (1956).
- [8] IR. K. N. Rao, A. H. Nielsen, and W. H. Fletcher, J. Chem. Phys. 26, 1572 (1957).
- [9] IR. K. N. Rao and E. D. Palik, J. Mol. Spectrosc. 1, 24 (1957).
- [10] IR. J. Overend and B. L. Crawford, J. Chem. Phys. 29, 1002 (1958).
- [11] R. S. Brodersen and E. H. Richardson, J. Mol. Spectrosc. 4, 439 (1960).
- [12] IR. I. M. Mills, W. L. Smith, and J. L. Duncan, J. Mol. Spectrosc. 16, 349 (1965).
- [13] IR. A. G. Maki and R. A. Toth, J. Mol. Spectrosc. 17, 136 (1965).

Molecule: Methylacetylene CH_3CCH
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 179

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH stretch	3334 C	cm^{-1} (Gas)	3334	
	ν_2	CH_3 s-stretch	2918 E	{ 2941 M 2881	3305 M 2941 VS, p	
	ν_3	$\text{C}\equiv\text{C}$ stretch	2142 A	2142.2 M	2142 VS, p	
	ν_4	CH_3 s-deform	1382 D		1382 S, dp	
	ν_5	C-C stretch	931 C	930.7 W	930 S, p (gas)	
	ν_6	CH_3 d-stretch	3008 A	3008.3 M	2971 M	
	ν_7	CH_3 d-deform	1452 B	1452 M	1448 M	
	ν_8	CH_3 rock	1053 A	1052.5 W	1035 VW	
	ν_9	CH bend	633 C	633 S	643 S, dp	
	ν_{10}	CCC bend	328 C	328 W	336 VS, dp	

References

- [1] R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR. D. R. J. Boyd and H. W. Thompson, Trans. Faraday Soc. 48, 493 (1952).
- [3] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).

Molecule: Methylacetylene-d₁ CH_3CCD
 Symmetry C_{3v} Symmetry number $\delta = 3$

No. 180

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 s-stretch	2920 E	cm^{-1} (Gas) { 2941.0 M 2881.0 M	cm^{-1} {	
	ν_2	CD stretch	2617 B	2616.8 S		
	ν_3	$\text{C}\equiv\text{C}$ stretch	2060 C	2060.3 W		
	ν_4	CH_3 s-deform	1378 E	1378 W	OV (ν_7). CF [1].
	ν_5	C-C stretch	886 E			
	ν_6	CH_3 d-stretch	3009 B	3008.9 M		
	ν_7	CH_3 deform	1454 B	1453.5 M		
	ν_8	CH_3 rock	1051 B	1051.0 W		
	ν_9	CD bend	498 B	497.5 S		
	ν_{10}	CCC bend	314 B	314 M		

References

- [1] IR. R. J. Grisenthwaite and H. W. Thompson, Trans. Faraday Soc. 50, 212 (1954).
- [2] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).

Molecule: Methyl-d₃-acetylene CD₃CCH
 Symmetry C_{3v} Symmetry number δ = 3

No. 181

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	CH stretch.....	3336 A	cm^{-1} (Gas)	3335.8 S	
	ν_2	CD ₃ s-stretch.....	2110 E		{ 2121.0 M 2077.0 M }	
	ν_3	C≡C stretch.....	2142 A		2142.0 M	
	ν_4	CD ₃ s-deform.....	1115 B		1115 M	
	ν_5	C-C stretch.....	830 B		830 W	
	ν_6	CD ₃ d-stretch.....	2235 A		2234.9 M	
	ν_7	CD ₃ d-deform.....	1048 A		1048.2 M	
	ν_8	CD ₃ rock.....	835 A		835.4 W	
	ν_9	CH bend.....	633 B		633 S	
	ν_{10}	CCC bend.....	305 B		304.5 M	

References

[1] IR. M. T. Christensen and H. W. Thompson, Trans. Faraday Soc. 52, 1439 (1956).

[2] Th. J. L. Duncan, Spectrochim. Acta 20, 1197 (1964).

Molecule: Methylacetylene-d₄ CD₃CCD
 Symmetry C_{3v} Symmetry number δ = 3

No. 182

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	CD stretch.....	2616 A	cm^{-1} (Gas)	2616.3 VS	
	ν_2	CD ₃ s-stretch.....	2110 E		{ 2121 M 2077 M }	
	ν_3	C≡C stretch.....	2008 A		2008.4 W	
	ν_4	CD ₃ s-deform.....	1110 A		1110.1 M	
	ν_5	C-C stretch.....	810 E			
	ν_6	CD ₃ d-stretch.....	2235 A		2234.8 M	
	ν_7	CD ₃ d-deform.....	1048 A		1048.2 M	
	ν_8	CD ₃ rock.....	834 A		834.4 W	
	ν_9	CD bend.....	492 B		492 VS	
	ν_{10}	CCC bend.....	294 B		294 M	

References

See No. 181.

Molecule: Malononitrile NCCH₂CN
 Symmetry C_{2v} Symmetry number δ = 2

No. 183

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>ν</i> ₁	CH ₂ s-stretch.....	2935 C	<i>cm</i> ⁻¹ (Liquid)	<i>cm</i> ⁻¹ (Liquid)	
	<i>ν</i> ₂	CN s-stretch.....	2275 C	2935 VS	2929 (5)	
	<i>ν</i> ₃	CH ₂ scis.....	1395 C	2275 M	2263 (7)	
	<i>ν</i> ₄	CC s-stretch.....	890 C	1395 VS	1386 (4)	
	<i>ν</i> ₅	CCC deform.....	582 C	890 S	892 (5)	
	<i>ν</i> ₆	CCN bend.....	167 C	582 M	574 (3b)	
	<i>ν</i> ₇	CH ₂ twist.....	1220 C	167 (10)	
	<i>ν</i> ₈	CCN bend.....	367 C	ia, 1220 VW	1214 (3)	
	<i>ν</i> ₉	CN a-stretch.....	2275 C	ia, 371 M	367 (10)	SF (<i>ν</i> ₁₂).
	<i>ν</i> ₁₀	CH ₂ wag.....	1318 C	2275 M	2263 (7)	SF (<i>ν</i> ₂).
	<i>ν</i> ₁₁	CC a-stretch.....	982 C	1318 W	1310 (2)	
	<i>ν</i> ₁₂	CCN bend.....	366 C	982 S	975 (1)	
	<i>ν</i> ₁₃	CH ₂ a-stretch.....	2968 C	366 S	367 (10)	
	<i>ν</i> ₁₄	CH ₂ rock.....	933 C	2968 VS	367 (10)	SF (<i>ν</i> ₈).
	<i>ν</i> ₁₅	CCN bend.....	337 C	933 M	2960 (1)	
				337 S		

References

- [1] R. K. W. F. Kohlrausch and G. Prinz Ypsilanti, Z. Phys. Chem. B29, 274 (1935).
- [2] IR.Th. F. Halverson and R. J. Francel, J. Chem. Phys. 17, 694 (1949).
- [3] IR.R.Th. T. Fujiyama and T. Shimanouchi, Spectrochim. Acta 20, 829 (1964).

Molecule: Malononitrile-d₂ NCCD₂CN
 Symmetry C_{2v} Symmetry number δ = 2

No. 184

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>ν</i> ₁	CD ₂ s-stretch.....	2146 C	<i>cm</i> ⁻¹ (Liquid)	<i>cm</i> ⁻¹ (D ₂ O soln.)	
	<i>ν</i> ₂	CN s-stretch.....	2272 C	2146 S	2146 (4)	
	<i>ν</i> ₃	CD ₂ scis.....	1037 C	2272 M	2273 (8)	
	<i>ν</i> ₄	CC s-stretch.....	858 C	1037 S	1033 (3)	
	<i>ν</i> ₅	CCC deform.....	577 C	858 M	854 (5)	
	<i>ν</i> ₆	CCN bend.....	163 C	577 M	581 (2)	
	<i>ν</i> ₇	CD ₂ twist.....	892 C	163 (4)	
	<i>ν</i> ₈	CCN bend.....	356 C	ia, 892 VW	892 (1)	
	<i>ν</i> ₉	CN a-stretch.....	2272 C	ia	356 (4)	SF (<i>ν</i> ₁₂).
	<i>ν</i> ₁₀	CD ₂ wag.....	1153 C	2272 M	2273 (8)	SF (<i>ν</i> ₂).
	<i>ν</i> ₁₁	CC a-stretch.....	1165 M	1162 (0.5)	FR (<i>ν</i> ₁₂ + <i>ν</i> ₁₄).	
	<i>ν</i> ₁₂	CCN bend.....	1142 M	1142 M	1130 (0.5)	
	<i>ν</i> ₁₃	CD ₂ a-stretch.....	2230 C	1162 (0.5)		
	<i>ν</i> ₁₄	CD ₂ rock.....	795 C	1130 (0.5)		
	<i>ν</i> ₁₅	CCN bend.....	302 C	795 W	2228 (2)	SF (<i>ν</i> ₈).
				302 (1)	

Reference

- [1] IR.R.Th. T. Fujiyama and T. Shimanouchi, Spectrochim. Acta 20, 829 (1964).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_2 a-stretch.....	3103 C	cm^{-1} (Gas)	cm^{-1}	
	ν_2	$\text{CH}(\beta)$ stretch.....	3028 D	3103 M	3028 M	
	ν_3	CH_2 s-stretch.....	3000 D	3000 M		
	ν_4	$\text{CH}(\alpha)$ stretch.....	2800 C	2800 S		
	ν_5	CO stretch.....	1724 C	1724 VS		
	ν_6	$\text{C}=\text{C}$ stretch.....	1625 C	1625 M		
	ν_7	CH_2 scis.....	1420 C	1420 S		
	ν_8	$\text{CH}(\alpha)$ ip-bend.....	1360 C	1360 M		
	ν_9	$\text{CH}(\beta)$ ip-bend.....	1275 C	1275 W		
	ν_{10}	C-C stretch.....	1158 C	1158 S		
	ν_{11}	CH_2 rock.....	912 C	912 S		
	ν_{12}	CCO deform.....	564 C	564 M		
	ν_{13}	CCC bend.....	327 C	327 M		
	ν_{14}	$\text{CH}(\beta)$ op-bend.....	993 B	993 S		
	ν_{15}	$\text{CH}(\alpha)$ op-bend.....	980 E			
	ν_{16}	CH_2 wag.....	959 B	959 S		
	ν_{17}	CH_2 twist.....	593 C	593 S		
	ν_{18}	CC torsion.....	157 C	157 M		
						CF [1, 2].

^a Numbering of atoms: $\text{C}^{\gamma}\text{H}_2\text{C}^{\beta}\text{HC}^{\alpha}\text{HO}$.

References

- [1] IR. J. C. D. Brand and D. G. Williamson, Disc. Faraday Soc. 35, 184 (1963).
- [2] IR. R. K. Harris, Spectrochim. Acta 20, 1129 (1964).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1'	ν_1	CH ₂ s-stretch.....	3038 C	cm^{-1} (Gas)	cm^{-1} (Gas)	FR ($2\nu_{14}$).
	ν_2	CH ₂ scis.....	1479 D	ia	3038 S, p 1504 W, p 1453 W, p	
a_1''	ν_3	Ring stretch.....	1188 C	ia	1188 S, p	OC ($\nu_5 + \nu_{10}$).
	ν_4	CH ₂ twist.....	1126 D	ia, 1126 VW	ia, 1133	
a_2'	ν_5	CH ₂ wag.....	1070 D	1075 (solid)	ia	
a_2''	ν_6	CH ₂ a-stretch.....	3103 C	3103 S	ia	
	ν_7	CH ₂ rock.....	854 C	854 S	ia	
e'	ν_8	CH ₂ s-stretch.....	3025 C	3025 VS	3020 VS, p	
	ν_9	CH ₂ scis.....	1438 C	1438 M	1442 M, dp	
	ν_{10}	CH ₂ wag.....	1029 C	1029 S	1023 VW (liquid)	
	ν_{11}	Ring deform.....	866 C	866 VS	866 S, dp	
e''	ν_{12}	CH ₂ a-stretch.....	3082 C	ia	3082 S, dp	
	ν_{13}	CH ₂ twist.....	1188 C	ia	1188 M	
	ν_{14}	CH ₂ rock.....	739 C	ia	739 W, dp	

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945), and references cited there.
- [2] IR.R. A. W. Baker and R. C. Lord, J. Chem. Phys. **23**, 1636 (1955).
- [3] IR.Th. Hs. H. Günthard, R. C. Lord, and T. K. McCubbin, Jr., J. Chem. Phys. **25**, 768 (1956).
- [4] R. P. M. Mathai, G. G. Shepherd, and H. L. Welsh, Can. J. Phys. **34**, 1448 (1956).
- [5] IR. C. Brecher, E. Krikorian, J. Blanc, and R. S. Halford, J. Chem. Phys. **35**, 1097 (1961).
- [6] IR. J. L. Duncan and D. C. McKean, J. Mol. Spectrosc. **27**, 117 (1968).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> _{1'}	<i>v</i> ₁	CD ₂ s-stretch.....	2236 C	<i>cm</i> ⁻¹ (Gas) ia	2236 VS, p	
	<i>v</i> ₂	CD ₂ scis.....	1274 D	ia	1274 S, p	
	<i>v</i> ₃	Ring stretch.....	956 C	ia	956 S, p	
	<i>v</i> ₄	CD ₂ twist.....	800 D	ia, 800 VW	ia	CF [2].
<i>a</i> _{2'}	<i>v</i> ₅	CD ₂ wag.....	870 D	ia, 875 (solid)	ia	CF [2], OC (<i>v</i> ₅ + <i>v</i> ₁₁).
<i>a</i> _{2''}	<i>v</i> ₆	CD ₂ a-stretch.....	2336 C	2336 VS	ia	
	<i>v</i> ₇	CD ₂ rock.....	614 C	614 W	ia	
<i>e</i> _'	<i>v</i> ₈	CD ₂ s-stretch.....	2211 C	2211 VS	2204 W, dp	
	<i>v</i> ₉	CD ₂ scis.....	1072 C	1072 S	1068 W, dp	
	<i>v</i> ₁₀	CD ₂ wag.....	885 C	885 M	884 M, dp	
	<i>v</i> ₁₁	Ring deform.....	717 C	717 VS	721 M, dp	
<i>e</i> _{''}	<i>v</i> ₁₂	CD ₂ a-stretch.....	2329 C	ia	2329 S, p	
	<i>v</i> ₁₃	CD ₂ twist.....	940 E	ia	CF [2], OC (2 <i>v</i> ₁₃).
	<i>v</i> ₁₄	CD ₂ rock	528 C	ia	528 W, dp	

References

- [1] IR.R. A. W. Baker and R. C. Lord, J. Chem. Phys. 23, 1636 (1955).
- [2] IR.Th. Hs. H. Günthard, R. C. Lord, and T. K. McCubbin, Jr., J. Chem. Phys. 25, 768 (1956).
- [3] IR. J. L. Duncan and D. C. McKean, J. Mol. Spectrosc. 27, 117 (1968).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	CH_3 d-stretch.....	3001 C	cm^{-1} (Liquid)	cm^{-1} (Liquid)	OV (ν_{14}).
	ν_2	CH_2 s-stretch.....	2955 C	2955 VS	2949 VS, p	
	ν_3	CH_3 s-stretch.....	2900 C	2900 S	2898 S, p	
	ν_4	CN stretch.....	2254 C	2254 VS	2251 VS, p	
	ν_5	CH_3 d-deform.....	1465 C	1465 S	1466 VS, p	SF (ν_{16}).
	ν_6	CH_2 scis.....	1433 C	1433 S	1436 M, p	
	ν_7	CH_3 s-deform.....	1387 C	1387 M	1374 VW, p	
	ν_8	CH_2 wag.....	1319 C	1319 M	1322 W, p	
	ν_9	C-CN stretch.....	1077 C	1077 S	1078 M, p	
	ν_{10}	CC stretch.....	1005 C	1005 M	1010 S, p	
	ν_{11}	CH_3 rock.....	836 C	836 W	838 S, p	
	ν_{12}	CCC deform.....	545 C	545 M	548 M, p	
	ν_{13}	CCN bend.....	226 C	226 M	226 M, p	
a''	ν_{14}	CH_3 d-stretch.....	3001 C	3001 VS	2999 S	OV (ν_1).
	ν_{15}	CH_2 a-stretch.....	2849 C	2849 S	2850 M	SF (ν_5).
	ν_{16}	CH_3 d-deform.....	1465 C	1465 S	1466 VS, dp	
	ν_{17}	CH_2 twist.....	1256 C	1256 VW	1270 VW, dp	
	ν_{18}	CH_3 rock.....	1022 E	CF [2].
	ν_{19}	CH_2 rock.....	786 C	786 M	784 VW, dp	
	ν_{20}	CCN bend.....	378 C	378 M	378 M, dp	
	ν_{21}	Torsion.....	222 C	MW [2].

References

- [1] IR.R. N. E. Duncan and G. J. Janz, J. Chem. Phys. 23, 434 (1955).
- [2] MW. V. W. Laurie, J. Chem. Phys. 31, 1500 (1959).
- [3] IR.R.Th. T. Fujiyama, Bull. Chem. Soc. Japan 44, 89 (1971).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 d-stretch.....	3019 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CH_3 s-stretch.....	2937 D	3018.5 S	3005.5 S	SF (ν_{18}).
	ν_3	CO stretch.....	1731 C	2937 S	2922 VS, p	SF (ν_{14}).
	ν_4	CH_3 d-deform.....	1435 C	1731 VS	1710.5 S, p	
	ν_5	CH_3 s-deform.....	1364 C	1435 S	1430 S	
	ν_6	CH_3 rock.....	1066 C	1363.5 VS	1356 W	SF (ν_{16}).
	ν_7	CC stretch.....	777 C	1066 M, p		
	ν_8	CCC deform.....	385 C	777 W	787 VS, p	
	ν_9	CH_3 d-stretch.....	2963 E	385 W	393 W, dp	
	ν_{10}	CH_3 d-deform.....	1426 E	ia	CF [4].
a_2	ν_{11}	CH_3 rock.....	877 E	ia	CF [4].
	ν_{12}	Torsion.....	105 D	ia	CF [4].
	ν_{13}	CH_3 d-stretch.....	3019 C	3018.5 S	3005.5 S, dp	CF [4]; MW: 102 [1].
	ν_{14}	CH_3 s-stretch.....	2937 D	2937 S	2922 VS	SF (ν_1).
b_1	ν_{15}	CH_3 d-deform.....	1410 C	1410 S	SF (ν_2).
	ν_{16}	CH_3 s-deform.....	1364 C	1363.5 VS	SF (ν_5).
	ν_{17}	CC stretch.....	1216 C	1215.5 VS	1221 M, dp	
	ν_{18}	CH_3 rock.....	891 C	891 M	902.5 W, dp	
	ν_{19}	CO ip-bend.....	530 C	530 S	531 M, dp	
	ν_{20}	CH_3 d-stretch.....	2972 C	2972 S	2967 S	
	ν_{21}	CH_3 d-deform.....	1454 C	1454 S		
	ν_{22}	CH_3 rock.....	1091 C	1090.5 M		
	ν_{23}	CO op-bend.....	484 C	484 W	493 W, dp	
	ν_{24}	Torsion.....	109 D	109	MW: 102. [1].

References

- [1] MW. J. D. Swalen and C. C. Costain, *J. Chem. Phys.* 31, 1562 (1959).
- [2] IR. W. G. Fateley and F. A. Miller, *Spectrochim. Acta* 18, 977 (1962).
- [3] IR.R.Th. P. Cossee and J. H. Schachtschneider, *J. Chem. Phys.* 44, 97 (1966).
- [4] IR.R.Th. G. Dellepiane and J. Overend, *Spectrochim. Acta* 22, 593 (1966).
- [5] IR.R.Th. M. Mikami, Ph.D. Thesis, (University of Tokyo, 1969).
- [6] R. T. Fujiyama and T. Shimanouchi, *Bull. Chem. Soc. Japan*, in press.

Molecule: Acetone- α , α , α -d₃ CH₃COCD₃
 Symmetry C_s Symmetry number $\delta = 1$

No. 190

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	ν_1	CH ₃ d-stretch.....	3018 C	cm^{-1} (Gas)	3017.5 S	3004.5 S
	ν_2	CH ₃ s-stretch.....	2922 C	2921.5 VS, p	
	ν_3	CD ₃ d-stretch.....	2265 C	2265 M	2256 S	
	ν_4	CD ₃ s-stretch.....	2115 E	{ 2150 VVW 2095 VW	2141.5 VS, p 2095.5 S, p	} FR (2 ν_9).
	ν_5	CO stretch.....	1734 C	1734 VS	1706 S	
	ν_6	CH ₃ d-deform.....	1430 C	1430 S	1427.5 M	
	ν_7	CH ₃ s-deform.....	1360 C	1360 VS	1361.5 VW	
	ν_8	CC stretch.....	1225 C	1224.5 VS	1227.5 W	
	ν_9	CD ₃ s-deform.....	1058 C	1057.5 W	
	ν_{10}	CH ₃ rock.....	1021 C	1021 S	1029.5 W	
	ν_{11}	CD ₃ d-deform.....	1003 C	1003 M, p	
	ν_{12}	CD ₃ rock.....	781 C	781 W	780.5 VW	
	ν_{13}	CC stretch.....	740 C	735 W	740 VS, p	
	ν_{14}	CO ip-bend.....	502 C	501.5 S		
	ν_{15}	CCC deform.....	352 C	352 W	356.5 W	
	ν_{16}	CH ₃ d-stretch.....	2968 C	2968 S	2965 S	
	ν_{17}	CD ₃ d-stretch.....	2222 C	2222 M	2217.5 S	
	ν_{18}	CH ₃ d-deform.....	1447 C	1447 S		
	ν_{19}	CH ₃ rock.....	1035 C	1035 S		
	ν_{20}	CD ₃ d-deform.....	999 C	999 S		
	ν_{21}	CD ₃ rock.....	764 D	764 M (solid)		
	ν_{22}	CO op-bend.....	438 C	438	444 W	CF [2].
	ν_{23}	CH ₃ torsion.....	106 E	CF [2].
	ν_{24}	CD ₃ torsion.....	78 E	

References

- [1] IR.R.Th. G. Dellepiane and J. Overend, Spectrochim. Acta 22, 593 (1966).
 [2] IR.R.Th. M. Mikami, Ph.D. Thesis, (University of Tokyo, 1969).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD ₃ d-stretch.....	2264 C	2263.5 S <i>cm</i> ⁻¹ (Gas)	
	<i>v</i> ₂	CD ₃ s-stretch.....	2123 C	2123 W	2108.5 VS, p	SF (<i>v</i> ₁₃). SF (<i>v</i> ₁₄).
	<i>v</i> ₃	CO stretch.....	1732 C	1732 VS	1700.5 S	
	<i>v</i> ₄	CD ₃ s-deform.....	1080 C	1080 M	1088 M, p	
	<i>v</i> ₅	CD ₃ d-deform.....	1035 D	1035 M	1036 M	
	<i>v</i> ₆	CD ₃ rock.....	887 C	887 W	889 M, p	
	<i>v</i> ₇	CC stretch.....	689 C	689 W	695.5 VS, p	
	<i>v</i> ₈	CCC deform.....	321 C	321 W	330 VW, dp	
	<i>v</i> ₉	CD ₃ d-stretch.....	2219 E	ia	CF [3].
<i>a</i> ₂	<i>v</i> ₁₀	CD ₃ d-deform.....	1021 E	ia	CF [3].
	<i>v</i> ₁₁	CD ₃ rock.....	669 E	ia	CF [3].
	<i>v</i> ₁₂	Torsion.....	75 E	ia	CF [3].
<i>b</i> ₁	<i>v</i> ₁₃	CD ₃ d-stretch.....	2264 C	2263.5 S	2256.5 S	
	<i>v</i> ₁₄	CD ₃ s-stretch.....	2123 C	2123 W	SF (<i>v</i> ₁). SF (<i>v</i> ₂).
	<i>v</i> ₁₅	CC stretch.....	1242 C	1241.7 VS	1248.5 VW	
	<i>v</i> ₁₆	CD ₃ s-deform.....	1035 D	1035 M	1036 M	
	<i>v</i> ₁₇	CD ₃ d-deform.....	1004 C	1004 M	1006 sh	
	<i>v</i> ₁₈	CD ₃ rock.....	724 D	724 W (solid)	
<i>b</i> ₂	<i>v</i> ₁₉	CO ip-bend.....	475 C	475 S	478 W, dp	
	<i>v</i> ₂₀	CD ₃ d-stretch.....	2227 C	2226.5 S	2222 S	
	<i>v</i> ₂₁	CD ₃ d-deform.....	1050 C	1050 S	
	<i>v</i> ₂₂	CD ₃ rock.....	960 C	960 M	
	<i>v</i> ₂₃	CO op-bend.....	405 C	405 W	410 VW, dp	
	<i>v</i> ₂₄	Torsion.....	79 E	CF [3].

References

- [1] IR.R.Th. P. Cossee and J. H. Schachtschneider, J. Chem. Phys. 44, 97 (1966).
- [2] IR.R.Th. G. Dellepiane and J. Overend, Spectrochim. Acta 22, 593 (1966).
- [3] IR.R.Th. M. Mikami, Ph.D. Thesis (University of Tokyo, 1969).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_3 d-stretch.....	2977 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	CH_3 s-stretch.....	2962 D	2962		
	ν_3	CH_2 s-stretch.....	2887 C	2887		
	ν_4	CH_3 d-deform.....	1476 C	1476		
	ν_5	CH_2 scis.....	1462 C	1462		
	ν_6	CH_3 s-deform.....	1392 C	1392		
	ν_7	CH_3 rock.....	1158 C	1158	1152 W	
	ν_8	CC stretch.....	869 C	869	867 S	
	ν_9	CCC deform.....	369 C	369	375 W	
	ν_{10}	CH_3 d-stretch.....	2967 C	ia	2967 M	
a_2	ν_{11}	CH_3 d-deform.....	1451 C	ia	1451 S	
	ν_{12}	CH_2 twist.....	1278 C	ia	1278 W	
	ν_{13}	CH_3 rock.....	940 D	ia	940 VW	
	ν_{14}	Torsion.....	^a 216 C	ia	MW [10,11].
b_1	ν_{15}	CH_3 d-stretch.....	2968 C	2968		
	ν_{16}	CH_3 s-stretch.....	2887 C	2887		OV (ν_3).
	ν_{17}	CH_3 d-deform.....	1464 C	1464		
	ν_{18}	CH_3 s-deform.....	1378 C	1378		
	ν_{19}	CH_2 wag.....	1338 C	1338	1338 M	
	ν_{20}	CC stretch.....	1054 C	1054	1054 M	
	ν_{21}	CH_3 rock.....	922 C	922		
b_2	ν_{22}	CH_3 d-stretch.....	2973 C	2973		
	ν_{23}	CH_2 a-stretch.....	2968 C	2968		
	ν_{24}	CH_3 d-deform.....	1472 C	1472		
	ν_{25}	CH_3 rock.....	1192 C	1192		
	ν_{26}	CH_2 rock.....	748 C	748		
	ν_{27}	Torsion.....	^a 268 C	MW [10,11].

^a These values are in agreement with the results of neutron-inelastic scattering experiment (D. M. Grant, R. J. Pugmire, and R. C. Livingston, J. Chem. Phys. 52, 4424 (1970)).

References

- [1] IR. D. M. Gates, J. Chem. Phys. 17, 393 (1949).
- [2] IR. H. L. McMurry, V. Thornton, and F. E. Condon, J. Chem. Phys. 17, 918 (1949).
- [3] IR. H. L. McMurry and V. Thornton, J. Chem. Phys. 19, 1014 (1951).
- [4] IR. J. J. Comeford and J. H. Gould, J. Mol. Spectrosc. 5, 474 (1960).
- [5] Th. H. Takahashi, Nippon Kagaku Zasshi 82, 1304 (1961).
- [6] IR. R. G. Snyder and J. H. Schachtschneider, Spectrochim. Acta 19, 85 (1963).
- [7] Th. J. H. Schachtschneider and R. G. Snyder, Spectrochim. Acta 19, 117 (1963).
- [8] IR. J. N. Gayles, Jr. and W. T. King, Spectrochim. Acta 21, 543 (1965).
- [9] Th. J. N. Gayles, Jr., W. T. King, and J. H. Schachtschneider, Spectrochim. Acta 23A, 703 (1967).
- [10] MW. E. Hirota, C. Matsumura, and Y. Morino, Bull. Chem. Soc. Japan 40, 1124 (1967).
- [11] MW. J. R. Hoyland, J. Chem. Phys. 49, 1908 (1968).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH ₃ d-stretch.....	2974 C	cm ⁻¹ (Gas)	2974	
	<i>v</i> ₂	CH ₃ s-stretch.....	2883 C		2883	
	<i>v</i> ₃	CD ₂ s-stretch.....	2141 C		2141	
	<i>v</i> ₄	CH ₃ d-deform.....	1459 C		1459	
	<i>v</i> ₅	CH ₃ s-deform.....	1392 C		1392	
	<i>v</i> ₆	CH ₃ rock.....	1207 D		1207	
	<i>v</i> ₇	CD ₂ scis.....	1064 C		1064	
	<i>v</i> ₈	CC stretch.....	843 C		843	
	<i>v</i> ₉	CCC deform.....	362 E		CF [5].
<i>a</i> ₂	<i>v</i> ₁₀	CH ₃ d-stretch.....	2956 E	ia	OC (<i>v</i> ₁₀ + <i>v</i> ₁₈) [4].
	<i>v</i> ₁₁	CH ₃ d-deform.....	1453 E	ia	CF [5].
	<i>v</i> ₁₂	CH ₃ rock.....	1083 E	ia	OC (<i>v</i> ₁₂ + <i>v</i> ₁₆) [4].
	<i>v</i> ₁₃	CD ₂ twist.....	777 E	ia	CF [5].
	<i>v</i> ₁₄	Torsion	*208 E	ia	OC (<i>v</i> ₁₈ - <i>v</i> ₁₄) [4].
<i>b</i> ₁	<i>v</i> ₁₅	CH ₃ d-stretch.....	2974 C	2974	SF (<i>v</i> ₁). SF (<i>v</i> ₂).
	<i>v</i> ₁₆	CH ₃ s-stretch.....	2883 C	2883	
	<i>v</i> ₁₇	CH ₃ d-deform.....	1461 C	1461	
	<i>v</i> ₁₈	CH ₃ s-deform.....	1374 C	1374	
	<i>v</i> ₁₉	CC stretch.....	1203 C	1203	
	<i>v</i> ₂₀	CH ₃ rock.....	964 C	964	
	<i>v</i> ₂₁	CD ₂ wag.....	829 C	829	
<i>b</i> ₂	<i>v</i> ₂₂	CH ₃ d-stretch.....	2963 C	2963	
	<i>v</i> ₂₃	CD ₂ a-stretch.....	2182 C	2182	
	<i>v</i> ₂₄	CH ₃ d-deform.....	1476 C	1476	
	<i>v</i> ₂₅	CH ₃ rock.....	1146 C	1146	
	<i>v</i> ₂₆	CD ₂ rock.....	622 C	622	
	<i>v</i> ₂₇	Torsion	*217 E	CF [4].

^a Assigning frequencies higher than these by 10–20 percent may be more reasonable in view of the results for CH₃CH₂CH₃.

References

- [1] IR. H. L. McMurry and V. Thornton, J. Chem. Phys. 18, 1515 (1950).
- [2] IR. H. L. McMurry and V. Thornton, J. Chem. Phys. 19, 1014 (1951).
- [3] Th. H. Takahashi, Nippon Kagaku Zasshi 82, 1304 (1961).
- [4] IR. J. N. Gayles, Jr. and W. T. King, Spectrochim. Acta 21, 543 (1965).
- [5] Th. T. Shimanouchi and T. Ueda, unpublished.
- [6] Th. J. N. Gayles, Jr., W. T. King, and J. H. Schachtschneider, Spectrochim. Acta 23A, 703 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₃ d-stretch.....	2966 C	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹	
	<i>v</i> ₂	CH ₃ s-stretch.....	2934 D			
	<i>v</i> ₃	CH ₂ s-stretch.....	2882 C			
	<i>v</i> ₄	CD ₃ d-stretch.....	2225 C			
	<i>v</i> ₅	CD ₃ s-stretch.....	2075 C			
	<i>v</i> ₆	CH ₂ scis.....	1461 D			
	<i>v</i> ₇	CH ₃ d-deform.....	1460 D			
	<i>v</i> ₈	CH ₃ s-deform.....	1383 C			
	<i>v</i> ₉	CH ₂ wag.....	1332 C			
	<i>v</i> ₁₀	CC stretch.....	1132 C			
	<i>v</i> ₁₁	CH ₃ rock.....	1101 C			
	<i>v</i> ₁₂	CD ₃ d-deform.....	1062 C			
	<i>v</i> ₁₃	CD ₃ s-deform.....	999 D			
	<i>v</i> ₁₄	CC stretch.....	846 C			
	<i>v</i> ₁₅	CD ₃ rock.....	750 C			
	<i>v</i> ₁₆	CCC deform.....	339 E			CF [2].
	<i>v</i> ₁₇	CH ₃ d-stretch.....	2966 C	2966		SF (<i>v</i> ₁).
	<i>v</i> ₁₈	CD ₂ a-stretch.....	2935 C	2935		
	<i>v</i> ₁₉	CD ₃ d-stretch.....	2214 C	2214		
	<i>v</i> ₂₀	CH ₃ d-deform.....	1461 D	1461		SF (<i>v</i> ₆).
	<i>v</i> ₂₁	CH ₂ twist.....	1285 D	1285		
	<i>v</i> ₂₂	CH ₃ rock.....	1129 C	1129		
	<i>v</i> ₂₃	CD ₃ d-deform.....	1063 C	1063		
	<i>v</i> ₂₄	CH ₂ rock.....	831 C	831		
	<i>v</i> ₂₅	CD ₃ rock.....	660 C	660		
	<i>v</i> ₂₆	CH ₃ torsion.....	*216 E			CF [2].
	<i>v</i> ₂₇	CD ₃ torsion.....	*161 E			CF [2].

^a Assigning frequencies higher than these by 10–20 percent may be more reasonable in view of the results for CH₃CH₂CH₃.

References

- [1] IR. J. N. Gayles, Jr. and W. T. King, Spectrochim. Acta. 21, 543 (1965).
- [2] TH. T. Shimanouchi and T. Ueda, unpublished.
- [3] TH. J. N. Gayles, Jr., W. T. King, and J. H. Schachtschneider, Spectrochim. Acta 23A, 703 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH ₂ s-stretch.....	2883 C	cm ⁻¹ (Gas)	cm ⁻¹	
	<i>v</i> ₂	CD ₃ d-stretch.....	2225 C	2225	SF (<i>v</i> ₂₃).
	<i>v</i> ₃	CD ₃ s-stretch.....	2091 C	2091	SF (<i>v</i> ₁₆).
	<i>v</i> ₄	CH ₂ scis.....	1467 C	1467	
	<i>v</i> ₅	CD ₃ s-deform.....	1098 E	CF [4].
	<i>v</i> ₆	CD ₃ d-deform.....	1066 C	1066	SF (<i>v</i> ₁₉).
	<i>v</i> ₇	CD ₃ rock.....	962 E	962	
	<i>v</i> ₈	CC stretch.....	711 D	711	
	<i>v</i> ₉	CCC deform.....	315 E	CF [4].
	<i>v</i> ₁₀	CD ₃ d-stretch.....	2222 E	ia	CF [4].
	<i>v</i> ₁₁	CH ₂ twist.....	1257 E	ia	CF [4].
	<i>v</i> ₁₂	CD ₃ d-deform.....	1052 E	ia	CF [4].
	<i>v</i> ₁₃	CD ₃ rock.....	700 E	ia	CF [4].
	<i>v</i> ₁₄	Torsion.....	*142 E	ia	OC (<i>v</i> ₁₄ + <i>v</i> ₂₁ , <i>v</i> ₂₁ - 2 <i>v</i> ₁₄) [3].
<i>b</i> ₁	<i>v</i> ₁₅	CD ₃ d-stretch.....	2227 C	2227	
	<i>v</i> ₁₆	CD ₃ s-stretch.....	2091 C	2091	SF (<i>v</i> ₃).
	<i>v</i> ₁₇	CH ₂ wag.....	1331 C	1331	
	<i>v</i> ₁₈	CC stretch.....	1131 C	1131	
	<i>v</i> ₁₉	CD ₃ d-deform.....	1066 C	1066	SF (<i>v</i> ₆).
	<i>v</i> ₂₀	CD ₃ s-deform.....	920 E	920	
	<i>v</i> ₂₁	CD ₃ rock.....	725 C	725	
	<i>v</i> ₂₂	CH ₂ a-stretch.....	2929 C	2929	
	<i>v</i> ₂₃	CD ₃ d-stretch.....	2225 C	2225	SF (<i>v</i> ₂).
	<i>v</i> ₂₄	CD ₃ d-deform.....	1087 C	1087	
	<i>v</i> ₂₅	CH ₂ rock.....	1066 D	1066	
	<i>v</i> ₂₆	CD ₃ rock.....	640 C	640	
	<i>v</i> ₂₇	Torsion.....	*173 E	OC (<i>v</i> ₂₁ + <i>v</i> ₂₇ - <i>v</i> ₁₄) [3].

^a Assigning frequencies higher than these by 10-20 percent may be more reasonable in view of the results for CH₃CH₂CH₃.

References

- [1] IR. H. L. McMurry and V. Thornton, J. Chem. Phys. 19, 1014 (1951).
- [2] Th. H. Takahashi, Nippon Kagaku Zasshi 82, 1304 (1961).
- [3] IR. J. N. Gayles, Jr. and W. T. King, Spectrochim. Acta 21, 543 (1965).
- [4] Th. T. Shimanouchi and T. Ueda, unpublished.
- [5] Th. J. N. Gayles, Jr., W. T. King, and J. H. Schachtschneider, Spectrochim. Acta 23A, 703 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD ₃ d-stretch.....	2225 C	cm ⁻¹ (Gas)	cm ⁻¹	
	<i>v</i> ₂	CD ₃ s-stretch.....	2122 C	2225	2122	
	<i>v</i> ₃	CD ₂ s-stretch.....	2081 C	2081		
	<i>v</i> ₄	CD ₃ s-deform.....	1086 D	1086		
	<i>v</i> ₅	CD ₂ scis.....	1064 D	1064		
	<i>v</i> ₆	CD ₃ d-deform.....	1064 D	1064		
	<i>v</i> ₇	CD ₃ rock.....	959 C	959		
	<i>v</i> ₈	CC stretch.....	712 C	712		
	<i>v</i> ₉	CCC deform.....	332 E		
	<i>v</i> ₁₀	CD ₃ d-stretch.....	2221 E	ia	CF [3].
<i>a</i> ₂	<i>v</i> ₁₁	CD ₃ d-deform.....	1064 E	ia	CF [4].
	<i>v</i> ₁₂	CD ₂ twist.....	945 E	ia	CF [4].
	<i>v</i> ₁₃	CD ₃ rock.....	659 E	ia	CF [4].
	<i>v</i> ₁₄	Torsion.....	^a 143 E	ia	OC (<i>v</i> ₁₄ + <i>v</i> ₂₂ , <i>v</i> ₁₄ + <i>v</i> ₂₄) [3].
	<i>v</i> ₁₅	CD ₃ d-stretch.....	2224 C	2224	
<i>b</i> ₁	<i>v</i> ₁₆	CD ₃ s-stretch.....	2081 C	2081	SF (<i>v</i> ₃). SF (<i>v</i> ₄). SF (<i>v</i> ₅ , <i>v</i> ₆). SF (<i>v</i> ₁₅). SF (<i>v</i> ₂₅ + <i>v</i> ₂₇ - <i>v</i> ₂₁) [3.]
	<i>v</i> ₁₇	CC stretch.....	1203 C	1203	
	<i>v</i> ₁₈	CD ₃ d-deform.....	1086 D	1086	
	<i>v</i> ₁₉	CD ₃ s-deform.....	1068 D	1068	
	<i>v</i> ₂₀	CD ₂ wag.....	862 D	862	
	<i>v</i> ₂₁	CD ₃ rock.....	688 C	688	
	<i>v</i> ₂₂	CD ₃ d-stretch.....	2224 C	2224	
	<i>v</i> ₂₃	CD ₂ a-stretch.....	2149 D	2149	
	<i>v</i> ₂₄	CD ₃ d-deform.....	1064 D	1064	
	<i>v</i> ₂₅	CD ₃ rock.....	949 D	949	
	<i>v</i> ₂₆	CD ₂ rock.....	544 D	544	
	<i>v</i> ₂₇	Torsion.....	^a 172 E	

^a Assigning frequencies higher than 10–20 percent may be more reasonable in view of the results for CH₃CH₂CH₃.

References

- [1] IR. H. L. McMurry and V. Thornton, J. Chem. Phys. 19, 1014 (1951).
- [2] Th. H. Takahashi, Nippon Kagaku Zasshi 82, 1304 (1961).
- [3] IR. J. N. Gayles, Jr. and W. T. King, Spectrochim. Acta 21, 543 (1965).
- [4] Th. T. Shimanouchi and T. Ueda, unpublished.
- [5] Th. J. N. Gayles, Jr., W. T. King, and J. H. Schachtschneider, Spectrochim. Acta 23A, 703 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	$\text{CH}_3(\text{O})$ d-stretch.....	3035 D	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	$\text{CH}_3(\text{C})$ d-stretch.....	3031 E	3035 M	3028 (3b)	SF (ν_2 of $\text{CH}_3\text{COOCD}_3$).
	ν_3	$\text{CH}_3(\text{O})$ s-stretch.....	2966 D	2966 S	2954 (3) p	SF (ν_4 of $\text{CH}_3\text{COOCD}_3$).
	ν_4	$\text{CH}_3(\text{C})$ s-stretch.....	2964 E	2942 (7b) p	
	ν_5	$\text{C}=\text{O}$ stretch.....	1771 C	1771 VS	1738 (3b) p	
	ν_6	$\text{CH}_3(\text{O})$ d-deform.....	1460 E	1460 W, sh (CCl_4 soln.)	OV (ν_{20}).
	ν_7	$\text{CH}_3(\text{O})$ s-deform.....	1440 D	1440 M	
	ν_8	$\text{CH}_3(\text{C})$ d-deform.....	1430 E	SF (ν_8 of $\text{CH}_3\text{COOCD}_3$).
	ν_9	$\text{CH}_3(\text{C})$ s-deform.....	1375 D	1375 S	1372 (0.5) p	
	ν_{10}	$\text{C}-\text{O}$ stretch.....	1248 C	1248 VS	1254 (0)	
	ν_{11}	$\text{CH}_3(\text{O})$ rock.....	1159 E	1159 VW (liquid)	
	ν_{12}	$\text{O}-\text{CH}_3$ stretch.....	1060 C	1060 S	1044 (2b)	
	ν_{13}	$\text{CII}_3(\text{C})$ rock.....	980 C	980 W	980 (1b) p	
	ν_{14}	CC stretch.....	844 C	844 M	844 (8) p	
	ν_{15}	$\text{C}=\text{O}$ ip-bend.....	639 C	639 M	640 (7) p	
	ν_{16}	CCO deform.....	429 C	429 M	433 (3) p	
	ν_{17}	COC deform.....	303 D	303 M	303 (1b) p	
	ν_{18}	$\text{CH}_3(\text{O})$ d-stretch.....	3005 D	3005 M	3002 (3b)	
	ν_{19}	$\text{CH}_3(\text{C})$ d-stretch.....	2994 D	2994 W	
	ν_{20}	$\text{CH}_3(\text{O})$ d-deform.....	1460 E	1460 W, sh (CCl_4 soln.)	1449 (4b) dp	OV (ν_6).
	ν_{21}	$\text{CH}_3(\text{C})$ d-deform.....	1430 E	1430 W	
	ν_{22}	$\text{CH}_3(\text{O})$ rock.....	1187 D	1187 W	1187 (0.5b)	
	ν_{23}	$\text{CH}_3(\text{C})$ rock.....	1036 E	1036 W (solid)	
	ν_{24}	$\text{C}=\text{O}$ op-bend.....	607 D	607 M	610 (0) dp	
	ν_{25}	$\text{C}-\text{O}$ torsion.....	187 D	187 W	
	ν_{26}	C-C torsion.....	136 E	136 VW (liquid)	
	ν_{27}	O- CH_3 torsion.....	110 E	110 VW (liquid)	

References

- [1] R. K. W. F. Kohlrausch, *Ramanspektren*, p. 263 (Edwards Bros., Inc., Ann Arbor, 1945).
- [2] IR. B. Nolin and R. N. Jones, *Can. J. Chem.* 34, 1382 (1956).
- [3] IR.R. J. K. Wilmsurst, *J. Mol. Spectrosc.* 1, 201 (1957).
- [4] IR.Th. S. Ichikawa, Y. Udagawa, and T. Shimanouchi, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₃ d-stretch.....	3032 D	3032 M cm ⁻¹ (Gas)		
	<i>v</i> ₂	CD ₃ d-stretch.....	2275 E	2275 W		
	<i>v</i> ₃	CH ₃ s-stretch.....	2967 D	2967 S		
	<i>v</i> ₄	CD ₃ s-stretch.....	2087 E	2087 W		
	<i>v</i> ₅	C=O stretch.....	1768 C	1768 VS		
	<i>v</i> ₆	CH ₃ d-deform.....	1455 E	1455 W, sh (CCl ₄ soln.)		OV (<i>v</i> ₂₀).
	<i>v</i> ₇	CH ₃ s-deform.....	1439 D	1439 M		
	<i>v</i> ₈	CD ₃ d-deform.....	1007 D	1007 M		
	<i>v</i> ₉	CD ₃ s-deform.....	1086 C	1086 S		
	<i>v</i> ₁₀	C-O stretch.....	1268 C	1268 VS		
	<i>v</i> ₁₁	CH ₃ rock.....	1160 D	1160 W		
	<i>v</i> ₁₂	O-CH ₃ stretch.....	1049 D	1049 W		
	<i>v</i> ₁₃	CD ₃ rock.....	780 C	780 M		
	<i>v</i> ₁₄	CC stretch.....	860 C	860 M		
	<i>v</i> ₁₅	C=O ip-bend.....	599 C	599 M		
	<i>v</i> ₁₆	CCO deform.....	390 C	390 M		
	<i>v</i> ₁₇	COC deform.....	298 D	298 M		
	<i>v</i> ₁₈	CH ₃ d-stretch.....	3004 D	3004 M		
	<i>v</i> ₁₉	CD ₃ d-stretch.....	2253 D	2253 W		
	<i>v</i> ₂₀	CH ₃ d-deform.....	1455 E	1455 W, sh (CCl ₄ soln.)		OV (<i>v</i> ₆).
	<i>v</i> ₂₁	CD ₃ d-deform.....	1033 D	1033 W		
	<i>v</i> ₂₂	CH ₃ rock.....	1181 E	1181 W		
	<i>v</i> ₂₃	CD ₃ rock.....	918 C	918 M		
	<i>v</i> ₂₄	C=O op-bend.....	525 D	525 M		
	<i>v</i> ₂₅	C-O torsion.....	178 D	178 M		
	<i>v</i> ₂₆	C-C torsion.....	98 E	CF [2].	
	<i>v</i> ₂₇	O-CH ₃ torsion.....	110 E	CF [2].	

References

- [1] IR. B. Nolin and R. N. Jones, Can. J. Chem. 34, 1382 (1956).
 [2] IR.Th. S. Ichikawa, Y. Udagawa, and T. Shimanouchi, unpublished.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CD ₃ d-stretch.....	2288 D	2288 M <i>cm</i> ⁻¹ (Gas)		
	<i>v</i> ₂	CH ₃ d-stretch.....	3031 D	3031 W		
	<i>v</i> ₃	CD ₃ s-stretch.....	2104 D	2104 M		
	<i>v</i> ₄	CH ₃ s-stretch.....	2964 D	2964 W		
	<i>v</i> ₅	C=O stretch.....	1769 C	1769 VS		
	<i>v</i> ₆	CD ₃ d-deform.....	1050 E	1050 W		OV (<i>v</i> ₂₀).
	<i>v</i> ₇	CD ₃ s-deform.....	1106 C	1106 S		OV (<i>v</i> ₂₁).
	<i>v</i> ₈	CH ₃ d-deform.....	1430 E	1430 W (CCl ₄ soln.)		
	<i>v</i> ₉	CH ₃ s-deform.....	1375 D	1375 S		
	<i>v</i> ₁₀	C—O stretch.....	1268 C	1268 VS		
	<i>v</i> ₁₁	CD ₃ rock.....	985 D	985 W		
	<i>v</i> ₁₂	O—CD ₃ stretch.....	1043 D	1043 M		
	<i>v</i> ₁₃	CH ₃ rock.....	947 C	947 M		
	<i>v</i> ₁₄	CC stretch.....	781 C	781 M		
	<i>v</i> ₁₅	C=O ip-bend.....	619 C	619 M		
	<i>v</i> ₁₆	CCO deform.....	420 C	420 M		
	<i>v</i> ₁₇	COC deform.....	270 D	270 M		
	<i>v</i> ₁₈	CD ₃ d-stretch.....	2263 D	2263 M		
	<i>v</i> ₁₉	CH ₃ d-stretch.....	2994 D	2994 W		
	<i>v</i> ₂₀	CD ₃ d-deform.....	1050 D	1050 W		OV (<i>v</i> ₆).
	<i>v</i> ₂₁	CH ₃ d-deform.....	1430 E	1430 W (CCl ₄ soln.)		OV (<i>v</i> ₈).
	<i>v</i> ₂₂	CD ₃ rock.....	908 E	908 VW		
	<i>v</i> ₂₃	CD ₃ rock.....	1015 E	1015 W, sh		
	<i>v</i> ₂₄	C=O op-bend.....	600 D	600 W, sh		
	<i>v</i> ₂₅	C—O torsion.....	165 D	165 M		
	<i>v</i> ₂₆	C—C torsion.....	136 E		CF [2].
	<i>v</i> ₂₇	O—CD ₃ torsion.....	81 E		CF [2].

References

See No. 198.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CD ₃ (O) d-stretch.....	2285 D	cm ⁻¹ (Gas)	cm ⁻¹	
	<i>v</i> ₂	CD ₃ (C) d-stretch.....	2275 E	2285 M	SF (<i>v</i> ₂ of CD ₃ COOCH ₃).
	<i>v</i> ₃	CD ₃ (O) s-stretch.....	2099 D	2099 M	
	<i>v</i> ₄	CD ₃ (C) s-stretch.....	2087 E	SF (<i>v</i> ₄ of CD ₃ COOCH ₃).
	<i>v</i> ₅	C=O stretch.....	1767 C	1767 VS	
	<i>v</i> ₆	CD ₃ (O) d-deform.....	1059 E	1059 W	OV (<i>v</i> ₂₀).
	<i>v</i> ₇	CD ₃ (O) s-deform.....	1106 C	1106 S	
	<i>v</i> ₈	CD ₃ (C) d-deform.....	1003 E	1003 W	
	<i>v</i> ₉	CD ₃ (C) s-deform.....	1086 E	SF (<i>v</i> ₉ of CD ₃ COOCH ₃).
	<i>v</i> ₁₀	C—O stretch.....	1282 C	1282 VS	
	<i>v</i> ₁₁	CD ₃ (O) rock.....	975 D	975 M	
	<i>v</i> ₁₂	O—CH ₃ stretch.....	1045 E	1045 W	
	<i>v</i> ₁₃	CD ₃ (C) rock.....	828 E	828 W	
	<i>v</i> ₁₄	CC stretch.....	747 C	747 M	
	<i>v</i> ₁₅	C=O ip-bend.....	585 C	585 M	
	<i>v</i> ₁₆	CCO deform.....	334 C	334 M	
	<i>v</i> ₁₇	COC deform.....	266 D	266 M	
	<i>v</i> ₁₈	CD ₃ (O) d-stretch.....	2264 D	2264 M	
	<i>v</i> ₁₉	CD ₃ (C) d-stretch.....	2253 E	SF (<i>v</i> ₁₉ of CD ₃ COOCH ₃).
	<i>v</i> ₂₀	CD ₃ (O) d-deform.....	1059 D	1059 W	OV (<i>v</i> ₆).
	<i>v</i> ₂₁	CD ₃ (C) d-deform.....	1038 E	1038 W	
	<i>v</i> ₂₂	CD ₃ (O) rock.....	908 E	SF (<i>v</i> ₂₂ of CH ₃ COOCD ₃).
<i>a''</i>	<i>v</i> ₂₃	CD ₃ (C) rock.....	925 D	925 M	
	<i>v</i> ₂₄	C=O op-bend.....	522 D	522 M	
	<i>v</i> ₂₅	C—O torsion.....	160 D	160 W	
	<i>v</i> ₂₆	C—C torsion.....	100 E	CF [2].
	<i>v</i> ₂₇	O—CH ₃ torsion.....	80 E	CF [2].

References

See No. 198.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
σ_g^+	ν_1	CH stretch.....	3293 D	cm^{-1} (Gas) ia	cm^{-1} (Gas) 3293 VW (liquid)	
	ν_2	C≡C stretch.....	2184 C	ia	2184 VS	
	ν_3	C-C stretch.....	874 C	ia	874 W	
	ν_4	CH stretch.....	3329 C	3329 VS	ia	
	ν_5	C≡C stretch.....	2020 C	2020 M	ia	
	ν_6	CH bend.....	627 C	ia	627 M	
	ν_7	CCC bend.....	482 C	ia	482 S	
	ν_8	CH bend.....	630 B	630 VS	ia	
	ν_9	CCC bend.....	231 E	ia	231 VW (liquid)	

Reference

[1] I.R.R. A. V. Jones, Proc. Roy. Soc. (London), Ser. A, 211, 285 (1952).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	ν_1	CH stretch.....	3154 D	cm^{-1} (Gas)	3154 VS, p	
	ν_2	CH stretch.....	3140 D	3140 sh		
	ν_3	ip-Ring II.....	1491 C	1491 VS	1483 VS, p	
	ν_4	ip-Ring III.....	1384 C	1384 M	1380 S, p	
	ν_5	ip-Ring IV.....	1140 D	1140 sh (liquid)	1137 VS, p	
	ν_6	CH ip-bend.....	1066 C	1066 S	1061 M, p	
	ν_7	CH ip-bend.....	995 C	995 VS	986 M, p	
	ν_8	ip-Ring VII.....	871 C	871 S		
	ν_9	CH op-bend.....	863 C	ia		OC ($\nu_9 + \nu_{16}$, $\nu_9 + \nu_{17}$, $\nu_9 + \nu_{19}$, $\nu_9 + \nu_2$).
	ν_{10}	CH op-bend.....	728 D	ia	728 W, dp	
<i>b</i> ₁	ν_{11}	op-Ring I.....	613 D	ia	613 VW, dp	
	ν_{12}	CH stretch.....	3161 C	3161 M		
	ν_{13}	CH stretch.....	3129 C	3129 M	3121 S, dp	
	ν_{14}	ip-Ring I.....	1556 C	1556 W		
	ν_{15}	CH ip-bend.....	1267 C	1267 VW	1270 VW, dp	
	ν_{16}	CH ip-bend.....	1180 C	1180 VS	1171 W, dp	
	ν_{17}	ip-Ring V.....	1040 D	1040 sh (liquid)	1034 M, dp	
<i>b</i> ₂	ν_{18}	ip-Ring VI.....	873 D		873 W, dp	
	ν_{19}	CH op-bend.....	838 C	838 VW	839 W, dp	
	ν_{20}	CH op-bend.....	745 C	745 VS		
	ν_{21}	op-Ring II.....	603 C	603 S	601 W, dp	

References

- [1] R. A. W. Reitz, Z. Phys. Chem. B38, 381 (1938).
- [2] IR. H. W. Thompson and R. B. Temple, Trans. Faraday Soc. 41, 27 (1945).
- [3] IR.R. G. B. Guthrie, D. W. Scott, W. N. Hubbard, C. Katz, J. P. McCullough, M. E. Gross, K. D. Williamson, and G. Waddington, J. Amer. Chem. Soc. 74, 4662 (1952).
- [4] IR. B. Bak, S. Brodersen, and L. Hansen, Acta Chem. Scand. 9, 749 (1955).
- [5] IR.R. M. Rico, M. Barrachina, and J. M. Orza, J. Mol. Spectrosc. 24, 133 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH stretch.....	3126 C	<i>cm</i> ⁻¹ (Gas)	3107 (10) p	
	<i>v</i> ₂	CH stretch.....	3098 C	3098 S	3084 (5sh)	
	<i>v</i> ₃	ip-Ring II.....	1409 C	1409 S	1407 (7) p	
	<i>v</i> ₄	ip-Ring III.....	1360 C	1360 VW	1358 (5) p	
	<i>v</i> ₅	CH ip-bend.....	1083 C	1083 S	1081 (5) p	
	<i>v</i> ₆	CH ip-bend.....	1036 C	1036 S	1035	
	<i>v</i> ₇	ip-Ring IV.....	839 C	839 VS	832 (5) p	
	<i>v</i> ₈	ip-Ring VII.....	608 C	608 W	606 (2) p	
	<i>v</i> ₉	CH op-bend.....	903 D	ia, 900 VW (solid)	903 (0) dp	
	<i>v</i> ₁₀	CH op-bend.....	688 D	ia	688 (0) dp	
<i>b</i> ₁	<i>v</i> ₁₁	op-Ring I.....	567 D	ia, 565 VW (liquid)	567 (0) dp	
	<i>v</i> ₁₂	CH stretch.....	^a 3125 E			
	<i>v</i> ₁₃	CH stretch.....	3086 C	3086 S	3076 (sh)	
	<i>v</i> ₁₄	ip-Ring I.....	1504 D	1504 VW	1502 (0) dp	
<i>b</i> ₂	<i>v</i> ₁₅	CH ip-bend.....	1256 C	1256 S	1257 (0)	
	<i>v</i> ₁₆	CH ip-bend.....	^a 1085 E	OV (<i>v</i> ₅).
	<i>v</i> ₁₇	ip-Ring V.....	872 C	872 M	869 (4) dp	
	<i>v</i> ₁₈	ip-Ring VI.....	751 D	763 VW	751 (1) dp	
	<i>v</i> ₁₉	CH op-bend.....	867 E	OC (<i>v</i> ₉ + <i>v</i> ₁₉ , 2 <i>v</i> ₁₉).
	<i>v</i> ₂₀	CH op-bend.....	712 C	712 VS	
	<i>v</i> ₂₁	op-Ring II.....	452 C	452 W	453 (0) dp	

^a These frequencies were estimated from isotopic rule [3].

References

- [1] IR. H. W. Thompson and R. B. Temple, Trans. Faraday Soc. 41, 27 (1945).
- [2] R. K. W. Kohlrausch and H. Schreiner, Acta Phys. Austriaca 1, 373 (1948).
- [3] IR.R. M. Rico, J. M. Orza, and J. Morcillo, Spectrochim. Acta 21, 689 (1965).
- [4] IR.R. V. T. Aleksanyan, Ya. M. Kimelfeld, N. N. Magdesieva, and Yu. K. Yurev, Opt. Spectrosc. 22, 116 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>ν</i> ₁	CD stretch	2343 C	<i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹ (Liquid)	
	<i>ν</i> ₂	CD stretch	2290 C	2343 M	2326 (6)	
	<i>ν</i> ₃	ip-Ring II	1376 C	2290 M		
	<i>ν</i> ₄	ip-Ring III	1248 C	1376 S	1372 (10) p	
	<i>ν</i> ₅	CD ip-bend	896 C	1248 W	1240 (5)	
	<i>ν</i> ₆	CD ip-bend	785 C	896 M	891 (10) p	
	<i>ν</i> ₇	ip-Ring IV	731 C	785 M	780 (3) dp	
	<i>ν</i> ₈	ip-Ring VII	585 D	731 VS	723 (3)	
	<i>ν</i> ₉	CD op-bend	752 E	585 VW ia, 756 (solid)	582 (2) p 752 (3) dp	SF (<i>ν</i> ₁₇).
<i>b</i> ₁	<i>ν</i> ₁₀	CD op-bend	532 D	ia	532 (2) dp	
	<i>ν</i> ₁₁	op-Ring I	488 D	ia	488	
	<i>ν</i> ₁₂	CD stretch	^a 2340 E	2305 C	2305 M	
	<i>ν</i> ₁₃	CD stretch	2305 C	2305 M	2286 (4) dp	
	<i>ν</i> ₁₄	ip-Ring I	1459 C	1459 M		
	<i>ν</i> ₁₅	CD ip-bend	1034 C	1034 S		
	<i>ν</i> ₁₆	CD ip-bend	846 C	846 S	847 (2)	
<i>b</i> ₂	<i>ν</i> ₁₇	ip-Ring V	752 D	756 (solid)	752 (3) dp	
	<i>ν</i> ₁₈	ip-Ring VI	712 C	712 W		
	<i>ν</i> ₁₉	CD op-bend	684 C	684 VW	682 (1)	
	<i>ν</i> ₂₀	CD op-bend	531 C	531 VS		
	<i>ν</i> ₂₁	op-Ring II	414 C	414 VW	411	

^a This frequency was estimated from isotopic rule [2].

References

- [1] R. K. W. Kohlrausch and H. Schreiner, Acta Phys. Austriaca 1, 373 (1948).
- [2] IR.R. M. Rico, J. M. Orza, and J. Morcillo, Spectrochim. Acta 21, 689 (1965).
- [3] IR.R. V. T. Aleksanyan, Ya. M. Kimelfeld, N. N. Magdesieva, and Yu. K. Yurev, Opt. Spectrosc. 22, 116 (1967).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CH_2 a-stretch.....	3087 D	cm^{-1} (Gas)	3087 M	
	ν_2	CH stretch.....	3003 D	ia	3003 M	
	ν_3	CH_2 s-stretch.....	2992 D	ia	2992 S	
	ν_4	C=C stretch.....	1630 D	ia	1630 VS	
	ν_5	CH_2 scis.....	1438 D	ia	1438 S	
	ν_6	CH bend.....	1280 D	ia	1280 S	
	ν_7	C-C stretch.....	1196 D	ia	1196 S	
	ν_8	CH_2 rock.....	894 D	ia	894 W	
	ν_9	CCC deform.....	512 D	ia	512 S	
a_u	ν_{10}	CH bend.....	1013 B	1013.4 VS	ia	
	ν_{11}	CII_2 wag.....	908 B	907.8 VS	ia	
	ν_{12}	CH_2 twist.....	522 B	522.2 M	ia	
b_g	ν_{13}	C-C torsion.....	162 B	162.3 VW	ia	
	ν_{14}	CH bend.....	976 D	ia	976 W	
	ν_{15}	CII_2 wag.....	912 D	ia	912 S	
b_u	ν_{16}	CH_2 twist.....	770 D	ia	770 VW	
	ν_{17}	CH_2 a-stretch.....	3101 B	3100.6 S	ia	
	ν_{18}	CH stretch.....	3055 B	3054.9 S	ia	
	ν_{19}	CH_2 s-stretch.....	2984 B	2984.3 S	ia	
	ν_{20}	C=C stretch.....	1596 B	1596.0 S	ia	
	ν_{21}	CH_2 scis.....	1381 B	1380.7 W	ia	
	ν_{22}	CH bend.....	1294 B	1294.3 W	ia	
	ν_{23}	CH_2 rock.....	990 B	989.7 M	ia	
	ν_{24}	CCC deform.....	301 B	300.6 VW	ia	

References

- [1] R. C. M. Richards and J. R. Nielsen, J. Opt. Soc. Amer. 40, 438 (1950).
- [2] Th. L. M. Sverdlov and N. V. Tarasova, Opt. Spectrosc. 9, 159 (1960).
- [3] IR. R. K. Harris, Spectrochim. Acta 20, 1129 (1964).
- [4] IR.R.Th. K. Abe, Ph.D. Thesis (University of Tokyo, 1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₂ a-stretch.....	3100 C	3100.4 S <i>cm</i> ⁻¹ (Gas)	3090 M <i>cm</i> ⁻¹ (Solid)	
	<i>v</i> ₂	CH stretch.....	3075 D	3075 W	
	<i>v</i> ₃	CH stretch.....	3048 C	3047.9 S		
	<i>v</i> ₄	CH stretch.....	3021 C	3020.5 S		
	<i>v</i> ₅	CH ₂ s-stretch.....	3003 D	3003 M	
	<i>v</i> ₆	CD stretch.....	2286 C	2285.9 M	2276 M	
	<i>v</i> ₇	C=C stretch.....	1631 D	1631 VS	
	<i>v</i> ₈	C=C stretch.....	1580 B	1579.7 S	1572 M	
	<i>v</i> ₉	CH ₂ scis.....	1409 D	1409 VW	
	<i>v</i> ₁₀	CH ip-bend.....	1304 E		CF [1].
	<i>v</i> ₁₁	CH ip-bend.....	1288 D	1288 S	
	<i>v</i> ₁₂	CH ip-bend.....	1270 C	1270 M	1272 VW	
	<i>v</i> ₁₃	C-C stretch.....	1183 D	1185 W	1183 S	
	<i>v</i> ₁₄	CH ₂ rock.....	964 D	964 W (solid)		
<i>a''</i>	<i>v</i> ₁₅	CD ip-bend.....	793 D	793 W	
	<i>v</i> ₁₆	CCC deform.....	511 D	511 M	
	<i>v</i> ₁₇	CCC deform.....	288 C	288 VW		
	<i>v</i> ₁₈	CH op-bend.....	1008 B	1008.0 VS		
	<i>v</i> ₁₉	CH op-bend.....	960 B	959.9 S		
	<i>v</i> ₂₀	CH ₂ wag.....	909 B	908.6 VS	921 M	
	<i>v</i> ₂₁	CH op-bend.....	849 B	849.2 S	862 M	
	<i>v</i> ₂₂	CD op-bend.....	674 C	673.9 VW		
	<i>v</i> ₂₃	CH ₂ twist.....	464 C	464.0 W		
	<i>v</i> ₂₄	C-C torsion.....	161 E		CF [1].

Reference

[1] IR.R.Th. K. Abe, Ph.D. Thesis (University of Tokyo, 1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a'</i>	<i>v</i> ₁	CH ₂ a-stretch.....	3099 C	3099.1 S <i>cm</i> ⁻¹ (Gas)	<i>cm</i> ⁻¹	
	<i>v</i> ₂	CH stretch.....	3016 C	3016.4 S		
	<i>v</i> ₃	CH ₂ s-stretch.....	2995 C	2995.4 S		
	<i>v</i> ₄	CD ₂ a-stretch.....	2342 C	2341.9 S		
	<i>v</i> ₅	CD stretch.....	2268 C	2267.9 S		
	<i>v</i> ₆	CD ₂ s-stretch.....	2217 C	2217.1 S		
	<i>v</i> ₇	C=C stretch.....	1630 C	1630.4 M		
	<i>v</i> ₈	C=C stretch.....	1549 B	1548.5 S		
	<i>v</i> ₉	CH ₂ scis.....	1425 C	1425 M		
	<i>v</i> ₁₀	CH ip-bend.....	1298 C	1298 W		
	<i>v</i> ₁₁	C-C stretch.....	1185 C	1185 W		
	<i>v</i> ₁₂	CD ₂ scis.....	1080 C	1080 W		
	<i>v</i> ₁₃	CD ip-bend.....	992 D	991.8 W (solid)		
	<i>v</i> ₁₄	CH ₂ rock.....	880 D	879.9 M (solid)		
<i>a''</i>	<i>v</i> ₁₅	CD ₂ rock.....	757 E		CF [1].
	<i>v</i> ₁₆	CCC deform.....	476 E		CF [1].
	<i>v</i> ₁₇	CCC deform.....	280 D	280 W		
	<i>v</i> ₁₈	CH op-bend.....	991 B	990.6 VS		
	<i>v</i> ₁₉	CH ₂ wag.....	909 B	909.2 VS		
	<i>v</i> ₂₀	CD op-bend.....	791 B	791.3 W		
	<i>v</i> ₂₁	CD ₂ wag.....	715 E	{ 734.0 S 710.1 VS }		FR (<i>v</i> ₁₇ + <i>v</i> ₂₃). CF [1].
	<i>v</i> ₂₂	CH ₂ twist.....	674 B	673.8 S		
	<i>v</i> ₂₃	CD ₂ twist.....	439 C	439.0 M		
	<i>v</i> ₂₄	C-C torsion.....	153 E		

Reference

[1] IR.Th. K. Abe, Ph.D. Thesis (University of Tokyo, 1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	<i>v</i> ₁	CH stretch.....	3010 D	<i>cm⁻¹</i> (Gas)	<i>cm⁻¹</i> (Solid)	
	<i>v</i> ₂	CD ₂ a-stretch.....	2315 D	ia	3010 M	
	<i>v</i> ₃	CD ₂ s-stretch.....	2212 D	ia	2315 S	
	<i>v</i> ₄	C=C stretch.....	1610 D	ia	2212 S	
	<i>v</i> ₅	CH ip-bend.....	1296 D	ia	1610 VS	
	<i>v</i> ₆	C-C stretch.....	1170 D	ia	1296 S	
	<i>v</i> ₇	CD ₂ scis.....	1040 D	ia	1170 S	
	<i>v</i> ₈	CD ₂ rock.....	740 D	ia	1040 S	
	<i>v</i> ₉	CCC deform.....	457 D	ia	740 W	
	<i>v</i> ₁₀	CH op-bend.....	955 B	955.1 S	457 S	
	<i>v</i> ₁₁	CD ₂ wag.....	728 B	728.0 VS	ia	
	<i>v</i> ₁₂	CD ₂ twist.....	397 C	397 W	ia	
<i>a_u</i>	<i>v</i> ₁₃	C-C torsion.....	149 E	ia	CF [1].
	<i>v</i> ₁₄	CH op-bend.....	948 D	ia	948 M	
	<i>v</i> ₁₅	CD ₂ wag.....	728 D	ia	728 S	
	<i>v</i> ₁₆	CD ₂ twist.....	610 D	ia	610 VW	
<i>b_g</i>	<i>v</i> ₁₇	CH stretch.....	3041 C	3040.8 S	ia	
	<i>v</i> ₁₈	CD ₂ a-stretch.....	2350 D	2350 S	ia	
	<i>v</i> ₁₉	CD ₂ s-stretch.....	2228 C	2228 S	ia	
	<i>v</i> ₂₀	C=C stretch.....	1535 B	1535.0 S	ia	
	<i>v</i> ₂₁	CH ip-bend.....	1335 C	1335.2 M	ia	
	<i>v</i> ₂₂	CD ₂ scis.....	1031 C	1031.3 S	ia	
	<i>v</i> ₂₃	CD ₂ rock.....	817 C	816.5 M	ia	
	<i>v</i> ₂₄	CCC deform.....	258 C	258 W	ia	

Reference

[1] IR.R.Th. K. Abe, Ph.D. Thesis (University of Tokyo, 1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	<i>v</i> ₁	CD ₂ a-stretch.....	2320 D	<i>cm</i> ⁻¹ (Gas)	2320 M	
	<i>v</i> ₂	CD stretch.....	2250 D	ia	2250 M	
	<i>v</i> ₃	CD ₂ s-stretch.....	2210 D	ia	2210 M	
	<i>v</i> ₄	C=C stretch.....	1580 D	ia	1580 VS	
	<i>v</i> ₅	C-C stretch.....	1196 D	ia	1196 M	
	<i>v</i> ₆	CD ₂ scis.....	1045 D	ia	1045 W	
	<i>v</i> ₇	CD ip-bend.....	918 D	ia	918 S	
	<i>v</i> ₈	CD ₂ rock.....	746 D	ia	746 M	
	<i>v</i> ₉	CCC deform.....	439 D	ia	439 S	
	<i>v</i> ₁₀	CD op-bend.....	741 B	741.4 W	ia	
<i>a_u</i>	<i>v</i> ₁₁	CD ₂ wag.....	718 B	718.4 S	ia	
	<i>v</i> ₁₂	CD ₂ twist.....	381 C	381.1 W	ia	
	<i>v</i> ₁₃	C-C torsion.....	140 C	140 VW	ia	
<i>b_g</i>	<i>v</i> ₁₄	CD op-bend.....	799 D	ia	799 S	
	<i>v</i> ₁₅	CD ₂ wag.....	705 D	ia	705 S	
<i>b_u</i>	<i>v</i> ₁₆	CD ₂ twist.....	603 D	ia	603 VW	
	<i>v</i> ₁₇	CD ₂ a-stretch.....	2320 D	2320.3 M	ia	
	<i>v</i> ₁₈	CD stretch.....	2266 C	2265.9 M	ia	
	<i>v</i> ₁₉	CD ₂ s-stretch.....	2218 C	2218.0 M	ia	
	<i>v</i> ₂₀	C=C stretch.....	1520 B	1519.6 S	ia	
	<i>v</i> ₂₁	CD ₂ scis.....	1048 C	1048.0 W	ia	
	<i>v</i> ₂₂	CD ip-bend.....	1005 C	1005.4 M	ia	
	<i>v</i> ₂₃	CD ₂ rock.....	769 D	768.9 W (solid)	ia	
	<i>v</i> ₂₄	CCC deform.....	250 C	250 W	ia	

Reference

[1] IR.R.Th. K. Abe, Ph.D. Thesis (University of Tokyo, 1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'_1	ν_1	CH_3 s-stretch.....	2916 C	cm^{-1} (Gas)	cm^{-1} (Liquid)	
	ν_2	$\text{C}\equiv\text{C}$ stretch.....	2240 E	ia	2916 S p	
	ν_3	CH_3 s-deform.....	1380 C	ia	2310 S p	
	ν_4	C-C stretch.....	725 E	ia	2233 S p	
				ia	1380 S	
a''_1	ν_5	CH_3 torsion ^a		ia	774 M p	
a''_2	ν_6	CH_3 s-stretch.....	2938 B	2938 S	693 M p	
	ν_7	CH_3 e-deform.....	1382 B	1382 M	ia	
	ν_8	C-C stretch.....	1152 B	1152 W	ia	
e'	ν_9	CH_3 d-stretch.....	2973 B	2973 S		
	ν_{10}	CH_3 d-deform.....	1456 B	1456 S		
	ν_{11}	CH_3 rock.....	1054 B	1054 M		
	ν_{12}	CCC deform.....	213 C	213 VW	
e''	ν_{13}	CH_3 d-stretch.....	2966 D	ia	2966 W	
	ν_{14}	CH_3 d-deform.....	1448 C	ia	1448 M dp	
	ν_{15}	CH_3 rock.....	1029 C	ia	1029 M dp	
	ν_{16}	CCC deform.....	371 C	ia	371 S dp	

^a Free rotation [5].

References

- [1] IR.R. G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules (Van Nostrand, New York, 1945).
- [2] IR.R. I. M. Mills and H. W. Thompson, Proc. Roy. Soc. (London) A226, 306 (1954).
- [3] IR.R. I. M. Mills and H. W. Thompson, Proc. Roy. Soc. (London) A228, 287 (1955).
- [4] IR.R. D. J. C. Yates and P. Lucchesi, J. Chem. Phys. 35, 243 (1961).
- [5] Th. H. C. Longuet-Higgins, Mol. Phys. 6, 445 (1963).
- [6] Th. J. T. Hougen, Can. J. Phys. 42, 1920 (1964).
- [7] Th. P. R. Bunker, J. Chem. Phys. 47, 718 (1967).

Molecule: Cyclobutane C_4H_8
 Symmetry D_{2d} Symmetry number $\delta = 4$

No. 211

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH ₂ s-stretch.....	2895 D	cm^{-1} (Gas) ia	cm^{-1} (Liquid) 2916 p 2866 p	{FR ($2\nu_2$, $2\nu_{13}$).}
	ν_2	CH ₂ scis.....	1443 C	ia	1443 p	SF (ν_{18}).
	ν_3	Ring stretch.....	1001 C	ia	1001 p	SF (ν_{14}).
	ν_4	CH ₂ a-stretch.....	2975 E	ia	CF [3].
	ν_5	CH ₂ rock.....	741 C	ia	741 dp	
	ν_6	Ring puckering.....	197 C	ia	197	CF [3].
	ν_7	CH ₂ wag.....	1260 E	ia	ia	CF [3].
a_2	ν_8	CH ₂ twist.....	1257 E	ia	ia	CF [3].
	ν_9	CH ₂ wag.....	1219 C	ia	1219 dp	
b_1	ν_{10}	Ring deform.....	926 C	ia	926 dp	
	ν_{11}	CH ₂ twist.....	1222 E	ia	CF [3].
	ν_{12}	CH ₂ s-stretch.....	2893 E	1443 dp	CF [3].
	ν_{13}	CH ₂ scis.....	1443 C	1001 p	SF (ν_2).
	ν_{14}	Ring deform.....	1001 D	SF (ν_8).
	ν_{15}	CH ₂ a-stretch.....	2987 C	2987 S	
	ν_{16}	CH ₂ rock.....	627 C	627 S	2952	
e	ν_{17}	CH ₂ a-stretch.....	2952 C	1223 W	
	ν_{18}	CH ₂ twist.....	1223 C	749 W	
	ν_{19}	CH ₂ rock.....	749 C	749 W	2897 S	
	ν_{20}	CH ₂ s-stretch.....	2887 D	2878 S	2878 S	
	ν_{21}	CH ₂ scis.....	1447 C	1447 S	
	ν_{22}	CH ₂ wag.....	1257 C	1257 S	
	ν_{23}	Ring deform.....	898 C	898 S	

References

- [1] IR.R. G. W. Rathjens, Jr., N. K. Freeman, W. D. Gwinn, and K. S. Pitzer, J. Amer. Chem. Soc. 75, 5634 (1953).
- [2] IR.R. D. G. Rea, Ph.D. Thesis, Massachusetts Institute of Technology, 1954.
- [3] Th. R. C. Lord and I. Nakagawa, J. Chem. Phys. 39, 2951 (1963).
- [4] IR. T. Ueda and T. Shimanouchi, J. Chem. Phys. 49, 470 (1968).
- [5] IR.R. J. M. R. Stone and I. M. Mills, Mol. Phys. 18, 653 (1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CD ₂ s-stretch	2124 E	<i>cm</i> ⁻¹ (Gas)	ia	
	<i>v</i> ₂	CD ₂ scis	1160 C	ia	1160 p	CF [2].
	<i>v</i> ₃	Ring stretch	882 C	ia	882 p	CF [2].
	<i>v</i> ₄	CD ₂ a-stretch	2224 E	ia	CF [2].
	<i>v</i> ₅	CD ₂ rock	632 E	ia	CF [2].
	<i>v</i> ₆	Ring puckering	158 D	ia	RP [3].
<i>a</i> ₂	<i>v</i> ₇	CD ₂ wag	1010 E	ia	ia	CF [2].
	<i>v</i> ₈	CD ₂ twist	889 E	ia	ia	CF [2].
<i>b</i> ₁	<i>v</i> ₉	CD ₂ wag	1078 C	ia	1078 dp	
	<i>v</i> ₁₀	Ring deform	746 C	ia	746 dp	CF [2].
<i>b</i> ₂	<i>v</i> ₁₁	CD ₂ twist	864 E	ia	CF [2].
	<i>v</i> ₁₂	CD ₂ s-stretch	2115 E	1040 dp	
<i>e</i>	<i>v</i> ₁₃	CD ₂ scis	1040 D	938 dp	SF (<i>v</i> ₁₈). SF (v ₁₄).
	<i>v</i> ₁₄	Ring deform	938 D	938 dp	
	<i>v</i> ₁₅	CD ₂ a-stretch	2242 C	2242 S		
	<i>v</i> ₁₆	CD rock	483 C	483 S		
	<i>v</i> ₁₇	CD ₂ a-stretch	2230 C	2230 dp	
	<i>v</i> ₁₈	CD ₂ twist	938 D	938 dp	SF (v ₁₄). SF (v ₁₈).
	<i>v</i> ₁₉	CD ₂ rock	556 C	556 W		
	<i>v</i> ₂₀	CD ₂ s-stretch	2103 E		
	<i>v</i> ₂₁	CD ₂ scis	1078 C	1078 S		
	<i>v</i> ₂₂	CD ₂ wag	1048 C	1048 S		
	<i>v</i> ₂₃	Ring deform	734 C	734 S		CF [2].

References

- [1] IR.R. D. G. Rea, Ph.D. Thesis (Massachusetts Institute of Technology, 1954).
- [2] Th. R. C. Lord and I. Nakagawa, J. Chem. Phys. 39, 2951 (1963).
- [3] IR.R. J. M. R. Stone and I. M. Mills, Mol. Phys. 18, 653 (1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_1	ν_1	CH_2 s-stretch.....	2989 D	2991 M (solid)	cm^{-1} (Liquid) 2989 S, p	
	ν_2	CH_3 d-stretch.....	2941 C	2940.8	2930 W, p	
	ν_3	CH_3 s-stretch.....	2911 D	2919 W	2911 S, p	
	ν_4	$\text{C}=\text{C}$ stretch.....	1661 C	1661.1 S	1655 S, p	
	ν_5	CH_3 d-deform.....	1470 C	1469.6 S	1462 VW	
	ν_6	CH_2 scis.....	1416 D	1419 W (solid)	1416 S, p	
	ν_7	CH_3 s-deform.....	1366 D	1366 VW, p	
	ν_8	CH_3 rock.....	1064 C	1063.9 S	1058 W, p	
	ν_9	C-C stretch.....	801 C	801 W	803 VS, p	
	ν_{10}	$\text{C}=\text{CC}_2$ ip-deform.....	383 D	384 W (solid)	383 W	
a_2	ν_{11}	CH_3 d-stretch.....	2970 D	ia	2970 W, p	OV (ν_{17}).
	ν_{12}	CH_3 d-deform.....	1459 D	ia	1459 VW	
	ν_{13}	CH_3 rock.....	1076 E	ia	CF [4].
b_1	ν_{14}	CH_2 twist.....	981 E	ia	CF [4].
	ν_{15}	CH_3 torsion.....	193 E	ia	CF [3].
b_1	ν_{16}	CH_2 a-stretch.....	3086 C	3086.0 S	3079 W, dp	
	ν_{17}	CH_3 d-stretch.....	2980 C	2980.4	2970 W, dp	
	ν_{18}	CH_3 s-stretch.....	2893 C	2892.9 W	2892 W, dp	OV (ν_{11}).
	ν_{19}	CH_3 d-deform.....	1458 C	1458.4 S	
	ν_{20}	CH_3 s-deform.....	1381 C	1381.2 S	1386 W	
	ν_{21}	C-C stretch.....	1282 C	1281.9 S	1281 W	
	ν_{22}	CH_3 rock.....	1043 E	CF [4].
	ν_{23}	CH_2 rock.....	974 C	973.7 W	972 VW	
	ν_{24}	$\text{C}=\text{CC}_2$ ip-deform.....	430 D	430 sh (solid)	
	ν_{25}	CH_3 d-stretch.....	2945 C	2944.9 S	
b_2	ν_{26}	CH_3 d-deform.....	1444 C	1443.7 S	1439 VW	
	ν_{27}	CH_3 rock.....	1079 C	1079.0 S	
	ν_{28}	CH_2 wag.....	890 C	889.7 VS	883 W, dp	
	ν_{29}	$\text{C}=\text{CC}_2$ op-deform.....	429 C	429.1 S	431 W, dp	
	ν_{30}	CH_3 torsion.....	196 C	196 VW	

References

- [1] IR. D. R. Smith, B. K. McKenna, and K. D. Möller, J. Chem. Phys. 45, 1904 (1966).
- [2] IR.R. W. Lüttke and S. Braun, Ber. Bunsenges. Phys. Chem. 71, 34 (1967).
- [3] IR.Th. Y. Abe, Ph.D. Thesis (University of Tokyo, 1968).
- [4] IR.R. C. M. Pathak and W. H. Fletcher, J. Mol. Spectrosc. 31, 32 (1969).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i> ₁	<i>v</i> ₁	CH ₂ s-stretch.....	2996 C	2996 M	<i>cm</i> ⁻¹ (Gas)	CF [1].
	<i>v</i> ₂	CD ₃ d-stretch.....	2166 D	2166 W (solid)		
	<i>v</i> ₃	CD ₃ s-stretch.....	2111 C	2111 W		
	<i>v</i> ₄	C=C stretch.....	1650 C	1650 S		
	<i>v</i> ₅	CH ₂ scis.....	1410 C	1410 W		
	<i>v</i> ₆	CD ₃ s-deform.....	1092 D	1092 W (solid)		
	<i>v</i> ₇	CD ₃ d-deform.....	1056 D	1056 M (solid)		
	<i>v</i> ₈	CD ₃ rock.....	850 E		
	<i>v</i> ₉	C-C stretch.....	718 D	718 W (solid)		
<i>a</i> ₂	<i>v</i> ₁₀	C=CC ₂ ip-deform.....	319 C	319 W	CF [1].	CF [1].
	<i>v</i> ₁₁	CD ₃ d-stretch.....	2208 E	ia		
	<i>v</i> ₁₂	CD ₃ d-deform.....	1054 E	ia		
	<i>v</i> ₁₃	CD ₃ rock.....	731 E	ia		
	<i>v</i> ₁₄	CH ₂ twist.....	664 E	ia		
<i>b</i> ₁	<i>v</i> ₁₅	CD ₃ torsion.....	138 E	ia	CF [1].	CF [1].
	<i>v</i> ₁₆	CH ₂ a-stretch.....	3085 C	3085 S		
	<i>v</i> ₁₇	CD ₃ d-stretch.....	2236 C	2236 S		
	<i>v</i> ₁₈	CD ₃ s-stretch.....	2072 C	2072 M		
	<i>v</i> ₁₉	C-C stretch.....	1294 C	1294 M		
	<i>v</i> ₂₀	CD ₃ d-deform.....	1074 C	1074 W		
	<i>v</i> ₂₁	CD ₃ s-deform.....	1052 D	1052 M (solid)		
	<i>v</i> ₂₂	CH ₂ rock.....	880 E		
	<i>v</i> ₂₃	CD ₃ rock.....	745 C	745 W		
	<i>v</i> ₂₄	C=CC ₂ ip-deform.....	400 C	400 M		
<i>b</i> ₂	<i>v</i> ₂₅	CD ₃ d-stretch.....	2204 C	2204 S	CF [1].	CF [1].
	<i>v</i> ₂₆	CD ₃ d-deform.....	1055 C	1055 S		
	<i>v</i> ₂₇	CD ₃ rock.....	923 C	923 M		
	<i>v</i> ₂₈	CH ₂ wag.....	884 C	884 VS		
	<i>v</i> ₂₉	C=CC ₂ op-deform.....	369 C	369 S		
	<i>v</i> ₃₀	CD ₃ torsion.....	143 E		

Reference

[1] IR.Th. Y. Abe, Ph.D. Thesis (University of Tokyo, 1968).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a'	ν_1	$\text{CH}_3(1)$ d-stretch	2983 D	cm^{-1} (Solid) 2983 S (liquid)	cm^{-1} (Liquid) 2983 M	OV ($\nu_2, \nu_{21}, \nu_{22}$).
	ν_2	$\text{CH}_3(4)$ d-stretch	2983 D	2983 S (liquid)	2983 M	OV ($\nu_1, \nu_{21}, \nu_{22}$).
	ν_3	$\text{CH}_3(1)$ s-stretch	2910 D	2910 S (liquid)	2924 S, p	OV (ν_4).
	ν_4	$\text{CH}_3(4)$ s-stretch	2910 D	2910 S (liquid)	2924 S, p	OV (ν_3).
	ν_5	CH_2 s-stretch	2884 D	2884 S (liquid)		
	ν_6	CO stretch	1716 C	1716 S	1715 M, p	
	ν_7	$\text{CH}_3(4)$ d-deform	1460 D	1460 M	1450 M	OV (ν_{24}).
	ν_8	CH_2 scis	1422 C	1422 S	1419 M	
	ν_9	$\text{CH}_3(1)$ d-deform	1413 D	1413 S		OV (ν_{25}).
	ν_{10}	$\text{CH}_3(4)$ s-deform	1373 C	1373 S		
	ν_{11}	$\text{CH}_3(1)$ s-deform	1346 C	1346 S	1345 W	
	ν_{12}	CH_2 wag	1263 D	1263 W	1258 W	OV (ν_{26}).
	ν_{13}	CC(12) stretch	1182 C	1182 S	1169 W	
	ν_{14}	$\text{CH}_3(4)$ rock	1089 C	1089 M	1087 M, p	
	ν_{15}	CC(34) stretch	997 C	997	999 W	
	ν_{16}	$\text{CH}_3(1)$ rock	939 C	939	951 W	
	ν_{17}	CC(23) stretch	760 D	760 S (liquid)	760 M, p	
	ν_{18}	CO ip-bend	590 C	590 S	591 W	
	ν_{19}	CCC(123) deform	413 C	413 S	410 W	
	ν_{20}	CCC(234) deform	260 C	260 S	264 W	
	ν_{21}	$\text{CH}_3(1)$ d-stretch	2983 D	2983 S (liquid)	2983	OV (ν_1, ν_2, ν_{22}).
	ν_{22}	$\text{CH}_3(4)$ d-stretch	2983 D	2983 S (liquid)	2983	OV (ν_1, ν_2, ν_{21}).
	ν_{23}	CH_2 d-stretch	2941 D	2941 S (liquid)		
	ν_{24}	$\text{CH}_3(4)$ d-deform	1460 D	1460 M	1450 M	OV (ν_7).
	ν_{25}	$\text{CH}_3(1)$ d-deform	1413 D	1413 S		OV (ν_9).
	ν_{26}	CH_2 twist	1263 D	1263 W	1258 W	OV (ν_{12}).
	ν_{27}	$\text{CH}_3(4)$ rock	1108 C	1108 W		
	ν_{28}	$\text{CH}_3(1)$ rock	952 C	952 sh	951 W	
	ν_{29}	CH_2 rock	768 D	768 S (liquid)		
	ν_{30}	CO op-bend	460 C	460 VW		
	ν_{31}	CC(34) torsion	201 E			CF [4].
	ν_{32}	CC(12) torsion	106 E			CF [4].
	ν_{33}	CC(23) torsion	87 C	87 W		

References

- [1] IR. A. Pozefsky and N. D. Coggeshall, Anal. Chem. 23, 1611 (1951).
- [2] IR.R. J. E. Katon and F. F. Bentley, Spectrochim. Acta 19, 639 (1963).
- [3] IR.Th. T. Shimanouchi, Y. Abe, and M. Mikami, Spectrochim. Acta, 24A, 1037 (1968).
- [4] IR.R.Th. M. Mikami, Ph.D. Thesis (University of Tokyo, 1969).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CH_3 d-stretch	2965 C	cm^{-1} (Matrix isolation)	2965 (9)	SF (ν_{20}).
	ν_2	CH_3 s-stretch	2872 C	ia	2872 (8)	
	ν_3	CH_2 s-stretch	2853 D	ia	2853 (8)	
	ν_4	CH_3 d-deform	1460 C	ia	1460 (2)	SF (ν_{22}).
	ν_5	CH_2 scis	1442 D	ia	1442 (3)	
	ν_6	CH_3 s-deform	1382 C	ia	CF [9].
	ν_7	CH_2 wag	1361 D	ia	CF [9].
	ν_8	CH_3 rock	1151 C	ia	1151 (4)	
	ν_9	CC stretch	1059 C	ia	1059 (5)	
	ν_{10}	CC stretch	837 C	ia	837 (6)	
	ν_{11}	CCC deform	425 C	ia	425 (4)	
	ν_{12}	CH_3 d-stretch	2968 C	2968 S	ia	SF (ν_{27}).
	ν_{13}	CH_2 a-stretch	2930 C	2930 S	ia	
	ν_{14}	CH_3 d-deform	1461 C	1461 S	ia	SF (ν_{30}), OV (ν_{30}, ν_{31})
	ν_{15}	CH_2 twist	1257 C	1257 W (solid)	ia	
	ν_{16}	CH_3 rock	948 B	948 M	ia	
	ν_{17}	CH_2 rock	731 B	731 S	ia	
a_u	ν_{18}	$\text{CH}_3\text{-CH}_2$ torsion	194 E	ia	CF [9].
	ν_{19}	$\text{CH}_2\text{-CH}_2$ torsion	102 E	ia	CF [9].
	ν_{20}	CH_3 d-stretch	2965 C	ia	2965 (9)	SF (ν_1).
	ν_{21}	CH_2 a-stretch	2912 C	ia	2912 (4)	
	ν_{22}	CH_3 d-deform	1460 C	ia	1460 (2)	SF (ν_4).
	ν_{23}	CH_2 twist	1300 C	ia	1300 (4)	
	ν_{24}	CH_3 rock	1180 D	ia	CF [9].
	ν_{25}	CH_2 rock	803 D	ia	CF [9].
	ν_{26}	$\text{CH}_3\text{-CH}_2$ torsion	225 E	ia	CF [9].
	ν_{27}	CH_3 d-stretch	2968 C	2968 S	ia	SF (ν_{12}).
	ν_{28}	CH_3 s-stretch	2870 C	2870 S	ia	
	ν_{29}	CH_2 s-stretch	2853 E	ia	SF (ν_3). SF (ν_{14}),
	ν_{30}	CH_3 d-deform	1461 C	1461 S	ia	OV (ν_{14}, ν_{31}). OV (ν_{14}, ν_{30}).
	ν_{31}	CH_2 scis	1461 C	1461 S	ia	
	ν_{32}	CH_3 s-deform	1379 B	1379 M	ia	
	ν_{33}	CH_2 wag	1290 B	1290 W	ia	
	ν_{34}	CC stretch	1009 C	1009 W (solid)	ia	
b_g	ν_{35}	CH_3 rock	964 B	964 M	ia	
	ν_{36}	CCC deform	271 E		ia	CF [9].

References

- [1] R. N. Sheppard and G. J. Szasz, J. Chem. Phys. 17, 86 (1949).
- [2] IR. D. W. E. Axford and D. H. Rank, J. Chem. Phys. 17, 430 (1949).
- [3] R. T. Shimanouchi and S. Mizushima, J. Chem. Phys. 17, 1102 (1949).
- [4] R. S. Mizushima and T. Shimanouchi, J. Amer. Chem. Soc. 71, 1320 (1949).
- [5] IR.R. J. K. Brown, N. Sheppard, and D. M. Simpson, Phil. Trans. Roy. Soc. (London), 247A, 35 (1954).
- [6] Th. R. I. Podlovchenko and M. M. Sushchinskii, Opt. Spectrosc. 2, 49 (1957).
- [7] IR. J. J. Comeford and J. H. Gould, J. Mol. Spectrosc. 5, 474 (1960).
- [8] IR. R. G. Snyder and J. H. Schachtschneider, Spectrochim. Acta 19, 85 (1963).
- [9] Th. J. H. Schachtschneider and R. G. Snyder, Spectrochim. Acta 19, 117 (1963).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a</i>	ν_1	CH_3 d-stretch.....	$\text{a}2968$ C	cm^{-1} (Liquid)	cm^{-1} (Liquid)	
	ν_2	CH_3 d-stretch.....	$\text{a}2968$ C			
	ν_3	CH_2 a-stretch.....	$\text{a}2920$ D			
	ν_4	CH_3 s-stretch.....	$\text{a}2870$ C			
	ν_5	CH_2 s-stretch.....	$\text{a}2860$ D			
	ν_6	CH_3 d-deform.....	$\text{a}1460$ C			
	ν_7	CH_3 d-deform.....	$\text{a}1460$ C			
	ν_8	CH_2 scis.....	$\text{a}1450$ D			
	ν_9	CH_3 s-deform.....	$\text{a}1380$ C			
	ν_{10}	CH_2 wag.....	1350 C	1350 W	1281 (0) 1168 (0) 1077 (1) 980 (2) 827 (6) 789 (2) 320 (1)	OV (ν_{32}). CF [5]. CF [5].
	ν_{11}	CH_2 twist.....	1281 C			
	ν_{12}	CH_3 rock.....	1168 D			
	ν_{13}	CC stretch.....	1077 D			
	ν_{14}	CH_3 rock.....	980 D			
	ν_{15}	CC stretch.....	827 D			
	ν_{16}	CH_2 rock.....	788 C	788 M		
	ν_{17}	CCC deform.....	320 C			
	ν_{18}	$\text{CH}_3\text{-CH}_2$ torsion.....	201 E			
	ν_{19}	$\text{CH}_2\text{-CH}_2$ torsion.....	101 E			
<i>b</i>	ν_{20}	CH_3 d-stretch.....	$\text{a}2968$ C	1370 VW	980 (2) 955 (1b)	OV (ν_{14}). CF [5]. CF [5].
	ν_{21}	CH_3 d-stretch.....	$\text{a}2968$ C			
	ν_{22}	CH_2 a-stretch.....	$\text{a}2920$ D			
	ν_{23}	CH_3 s-stretch.....	$\text{a}2870$ C			
	ν_{24}	CH_2 s-stretch.....	$\text{a}2860$ D			
	ν_{25}	CH_3 d-deform.....	$\text{a}1460$ C			
	ν_{26}	CH_3 d-deform.....	$\text{a}1460$ C			
	ν_{27}	CH_2 scis.....	$\text{a}1450$ D			
	ν_{28}	CH_3 s-deform.....	$\text{a}1380$ C			
	ν_{29}	CH_2 wag.....	1370 D			
	ν_{30}	CH_2 twist.....	1233 C	1233 W		
	ν_{31}	CC stretch.....	1133 D	1133 M		
	ν_{32}	CH_3 rock.....	980 D	980 (2)		
	ν_{33}	CH_3 rock.....	955 C	955 (1b)		
	ν_{34}	CH_2 rock.....	747 C	747 S		
	ν_{35}	CCC deform.....	469 D			
	ν_{36}	$\text{CH}_3\text{-CH}_2$ torsion.....	197 E			

^a Deduced from the corresponding frequencies of the trans form.

References

- [1] R. N. Sheppard and G. J. Szasz, J. Chem. Phys. 17, 86 (1949).
- [2] IR. D. W. E. Axford and D. H. Rank, J. Chem. Phys. 17, 430 (1949).
- [3] R. S. Mizushima and T. Shimanouchi, J. Amer. Chem. Soc. 71, 1320 (1949).
- [4] Th. R. I. Podlochenko and M. M. Suschinskii, Opt. Spectrosc. 2, 49 (1957).
- [5] Th. R. G. Snyder and J. H. Schachtschneider, Spectrochim. Acta 21, 169 (1965).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_{1g}	ν_1	CH stretch.....	3062 C	cm^{-1} (Gas) ia	cm^{-1} (Liquid) 3061.9 VS, p	
	ν_2	Ring stretch.....	992 C	ia	991.6 VS, p	
a_{2g}	ν_3	CH bend.....	1326 E	ia	1326 VW	
a_{2u}	ν_4	CH bend.....	673 B	673 S	ia	
b_{1u}	ν_5	CH stretch.....	3068 C	3067.57 VW (solid)	ia	
	ν_6	Ring deform.....	1010 C	1010 W (solid)	ia	
b_{2g}	ν_7	CH bend.....	995 E	ia	ia	
	ν_8	Ring deform.....	703 E	ia	ia	
b_{2u}	ν_9	Ring stretch.....	1310 C	1310 W (liquid)	ia	$OC(\nu_{19} + \nu_7, \nu_{20} + \nu_7).$ $OC(\nu_{19} + \nu_8, \nu_{20} + \nu_8).$
	ν_{10}	CH bend.....	1150 C	1150 W (liquid)	ia	
e_{1g}	ν_{11}	CH bend.....	849 C	ia	848.9 M, dp	
e_{1u}	ν_{12}	CH stretch.....	3063 E	{ 3080 S 3030 S (liquid)	ia	FR ($\nu_{18} + \nu_{16}$).
	ν_{13}	Ring stretch + deform.....	1486 B	1486 S	ia	
e_{2g}	ν_{14}	CH bend.....	1038 B	1038 S	ia	
	ν_{15}	CH stretch.....	3047 C	ia	3046.8 S, dp	
	ν_{16}	Ring stretch.....	1596 E	ia	{ 1606.4 S, dp 1584.6 S, dp	
e_{2u}	ν_{17}	CH bend.....	1178 C	ia	1178.0 S, dp	
	ν_{18}	Ring deform.....	606 C	ia	605.6 S, dp	
	ν_{19}	CH bend.....	975 C	975 W (liquid)	ia	
	ν_{20}	Ring deform.....	410 C	{ 417.7 403.0 (solid)	ia	

References

- [1] IR.R. N. Herzfeld, C. K. Ingold, and H. G. Poole, J. Chem. Soc. 316 (1946).
- [2] IR. R. D. Mair and D. F. Hornig, J. Chem. Phys. 17, 1236 (1949).
- [3] IR. H. Spedding and D. H. Whiffen, Proc. Roy. Soc. (London), Ser. A, 238, 245 (1956).
- [4] Th. S. Califano and B. Crawford, Jr., Spectrochim. Acta 16, 889 (1960).
- [5] Th. J. R. Scherer and J. Overend, Spectrochim. Acta 17, 719 (1961).
- [6] IR. J. L. Hollenberg and D. A. Dows, J. Chem. Phys. 37, 1300 (1962).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_{1g}</i>	<i>ν₁</i>	CD stretch.....	2293 C	<i>cm⁻¹</i> (Gas) ia	2292.6 VS, p	OC (<i>ν₃</i> + <i>ν₁₄</i> , <i>ν₃</i> + <i>ν₁₆</i>).
	<i>ν₂</i>	Ring stretch.....	943 C	ia	943.2 VS, p	
	<i>ν₃</i>	CD bend.....	1037 E	ia	ia	
	<i>ν₄</i>	CD bend.....	497 C	496.5 S (liquid)	ia	
<i>b_{1u}</i>	<i>ν₅</i>	CD stretch.....	2292 E	2292 VW (solid)	ia	OC (<i>ν₇</i> + <i>ν₁₀</i>). OC (<i>ν₈</i> + <i>ν₁₀</i>).
	<i>ν₆</i>	Ring deform.....	969 C	{ 970.48 969.77 966.76 (solid)	ia	
<i>b_{2g}</i>	<i>ν₇</i>	CD bend.....	827 E	ia	ia	OC (<i>ν₇</i> + <i>ν₁₀</i>). OC (<i>ν₈</i> + <i>ν₁₀</i>).
	<i>ν₈</i>	Ring deform.....	601 E	ia	ia	
<i>b_{2u}</i>	<i>ν₉</i>	Ring stretch.....	1286 C	{ 1287.51 1286.41 1285.14 (solid)	ia	OC (<i>ν₇</i> + <i>ν₁₀</i>). OC (<i>ν₈</i> + <i>ν₁₀</i>).
	<i>ν₁₀</i>	CD bend.....	824 C	{ 825.2 822.57 (solid)	ia	
<i>e_{1g}</i>	<i>ν₁₁</i>	CD bend.....	662 C	ia	661.7 M, dp	OC (<i>ν₄</i> + <i>ν₂₀</i> , <i>ν₁₄</i> + <i>ν₂₀</i>).
	<i>ν₁₂</i>	CD stretch.....	2287 C	2287 S	ia	
	<i>ν₁₃</i>	Ring stretch + deform.....	1335 B	1335 M	ia	
<i>e_{2g}</i>	<i>ν₁₄</i>	CD bend.....	814 B	814 S	ia	OC (<i>ν₄</i> + <i>ν₂₀</i> , <i>ν₁₄</i> + <i>ν₂₀</i>).
	<i>ν₁₅</i>	CD stretch.....	2265 C	ia	2264.9 S, dp	
	<i>ν₁₆</i>	Ring stretch.....	1552 C	ia	1551.5 S, dp	
	<i>ν₁₇</i>	CD bend.....	867 C	ia	867.3 S, dp	
	<i>ν₁₈</i>	Ring deform.....	577 C	ia	577.4 M, dp	
<i>e_{2u}</i>	<i>ν₁₉</i>	CD bend.....	795 C	{ 799.91 797.37 794.64 790.9 790.3 (solid)	ia	OC (<i>ν₄</i> + <i>ν₂₀</i> , <i>ν₁₄</i> + <i>ν₂₀</i>).
	<i>ν₂₀</i>	Ring deform.....	352 E	ia	ia	

References

- [1] IR.R. N. Herzfeld, C. K. Ingold, and H. G. Poole, J. Chem. Soc. 316 (1946).
- [2] Th. S. Califano and B. Crawford, Jr., Spectrochim. Acta 16, 889 (1960).
- [3] Th. J. R. Scherer and J. Overend, Spectrochim. Acta 17, 719 (1961).
- [4] IR. D. A. Dows and A. L. Pratt, Spectrochim. Acta 18, 433 (1962).
- [5] IR. J. L. Hollenberg and D. A. Dows, J. Chem. Phys. 37, 1300 (1962).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_{1g}</i>	<i>ν₁</i>	CH ₂ a-stretch.....	2930 E	<i>cm⁻¹</i> (Gas) ia	<i>cm⁻¹</i> (Liquid) 2938 VS, p 2923 VS, p	FR (2 <i>ν₃</i>). SF (<i>ν₂</i> , <i>ν₁₈</i> , <i>ν₂₆</i>). FR (<i>ν₂₃</i> + <i>ν₃₂</i>).
	<i>ν₂</i>	CH ₂ s-stretch.....	2852 C	ia	2852 VS, p	
	<i>ν₃</i>	CH ₂ scis.....	1465 C	ia	1465 M, p	
	<i>ν₄</i>	CH ₂ rock.....	1157 C	ia	1157 S, p	
	<i>ν₅</i>	CC stretch.....	802 C	ia	802 VS, p	
	<i>ν₆</i>	CCC deform + CC torsion.....	383 C	ia	383 M, p	
<i>a_{1u}</i>	<i>ν₇</i>	CH ₂ twist.....	1383 C	^a 1383	ia	SF (<i>ν₂</i> , <i>ν₁₈</i> , <i>ν₂₆</i>). FR (<i>ν₂₃</i> + <i>ν₃₂</i>).
	<i>ν₈</i>	CH ₂ wag.....	1157 C	^a 1157	ia	
	<i>ν₉</i>	CC stretch + CC torsion.....	1057 C	^a 1057	ia	
<i>a_{2g}</i>	<i>ν₁₀</i>	CH ₂ wag.....	1437 C	^a 1437	ia	
<i>a_{2u}</i>	<i>ν₁₁</i>	CH ₂ twist.....	1090 C	^a 1090	ia	
<i>e_g</i>	<i>ν₁₂</i>	CH ₂ a-stretch.....	2915 E	2915 M	ia	SF (<i>ν₁</i> , <i>ν₁₂</i> , <i>ν₂₅</i>). SF (<i>ν₁</i> , <i>ν₁₈</i> , <i>ν₂₅</i>).
	<i>ν₁₃</i>	CH ₂ s-stretch.....	2860 E	ia	
	<i>ν₁₄</i>	CH ₂ scis.....	1437 C	1437 M	ia	
	<i>ν₁₅</i>	CH ₂ rock.....	1030 D	{ 1040 M 1016 M }	ia	
	<i>ν₁₆</i>	CCC deform.....	523 A	523 W	ia	
<i>e_u</i>	<i>ν₁₇</i>	CH ₂ a-stretch.....	2930 E	ia	SF (<i>ν₁</i> , <i>ν₁₂</i> , <i>ν₂₅</i>). SF (<i>ν₁</i> , <i>ν₁₈</i> , <i>ν₂₅</i>).
	<i>ν₁₈</i>	CH ₂ s-stretch.....	2897 E	ia	2897 M, vb	
	<i>ν₁₉</i>	CH ₂ scis.....	1443 C	ia	1443 S, dp	
	<i>ν₂₀</i>	CH ₂ wag.....	1347 C	ia	1347 S, dp	
	<i>ν₂₁</i>	CH ₂ twist.....	1266 C	ia	1266 VS, dp	
	<i>ν₂₂</i>	CC stretch.....	1027 C	ia	1027 VS, dp	
	<i>ν₂₃</i>	CH ₂ rock.....	785 C	^a 785	785 VW, dp	
	<i>ν₂₄</i>	CCC deform + CC torsion.....	426 C	ia	426 S, dp	
	<i>ν₂₅</i>	CH ₂ a-stretch.....	2933 A	2933 VS	ia	
	<i>ν₂₆</i>	CH ₂ s-stretch.....	2863 A	2863 VS	ia	

^a Observed in the crystalline state at about 90 K [8].

References

- [1] R. A. Langseth and B. Bak, J. Chem. Phys. 8, 403 (1940).
- [2] Th. C. W. Beckett, K. S. Pitzer, and R. Spitzer, J. Amer. Chem. Soc. 69, 2488 (1947).
- [3] R. N. I. Peokof'eva, L. M. Sverdlov and M. M. Sushchinskii, Opt. Spectrosc. 15, 250 (1963).
- [4] IR. S. Abramowitz and R. P. Bauman, J. Chem. Phys. 39, 2757 (1963).
- [5] IR.Th. H. Takahashi, T. Shimanouchi, K. Fukushima, and T. Miyazawa, J. Mol. Spectrosc. 13, 43 (1964).
- [6] IR.R. F. A. Miller and H. R. Colob, Spectrochim. Acta 20, 1517 (1964).
- [7] Th. R. G. Snyder and J. H. Schachtschneider, Spectrochim. Acta 21, 169 (1965).
- [8] IR. D. A. Dows, J. Mol. Spectrosc. 16, 302 (1965).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared <i>cm⁻¹</i> (Gas)	Raman <i>cm⁻¹</i> (Liquid)	Comments
<i>a_{1g}</i>	<i>v</i> ₁	CD ₂ a-stretch.....	2152 C	ia	2152 VS, p	
	<i>v</i> ₂	CD ₂ s-stretch.....	2082 C	ia	2082 VS, p	
	<i>v</i> ₃	CD ₂ scis.....	1117 C	ia	1117 M, p	
	<i>v</i> ₄	CD ₂ rock.....	1012 C	ia	1012 W, p	
	<i>v</i> ₅	CC stretch.....	723 C	ia	723 VS, p	
	<i>v</i> ₆	CCC deform. + CC torsion.....	298 C	ia	298 W, p	
<i>a_{1u}</i>	<i>v</i> ₇	CD ₂ twist.....	864 E	ia	ia	CF [4].
	<i>v</i> ₈	CD ₂ wag.....	842 E	ia	ia	CF [4].
	<i>v</i> ₉	CC stretch. + CC torsion.....	1187 E	ia	ia	CF [4].
<i>a_{2g}</i>	<i>v</i> ₁₀	CD ₂ wag.....	1126 E	ia	ia	CF [4].
	<i>v</i> ₁₁	CD ₂ twist.....	778 E	ia	ia	CF [4].
<i>a_{2u}</i>	<i>v</i> ₁₂	CD ₂ a-stretch.....	2206 C	2206 VS	ia	OV (<i>v</i> ₂₅). OV (<i>v</i> ₂₆). OV (<i>v</i> ₁₂). OV (<i>v</i> ₁₃). OV (<i>v</i> ₁₄). OV (<i>v</i> ₁₅). OV (<i>v</i> ₁₆). OV (<i>v</i> ₁₇). OV (<i>v</i> ₁₈). OV (<i>v</i> ₁₉). OV (<i>v</i> ₂₀). OV (<i>v</i> ₂₁). OV (<i>v</i> ₂₂). OV (<i>v</i> ₂₃). OV (<i>v</i> ₂₄). OV (<i>v</i> ₂₅). OV (<i>v</i> ₂₆). OV (<i>v</i> ₂₇). OV (<i>v</i> ₂₈). OV (<i>v</i> ₂₉). OV (<i>v</i> ₃₀). OV (<i>v</i> ₃₁). OV (<i>v</i> ₃₂). CF [4].
	<i>v</i> ₁₃	CD ₂ s-stretch.....	2108 C	2108 VS	ia	
	<i>v</i> ₁₄	CD ₂ scis.....	1091 B	1091 VS	ia	
	<i>v</i> ₁₅	CD ₂ rock.....	917 A	917 VS	ia	
	<i>v</i> ₁₆	CCC deform.....	395 B	395 S	ia	
<i>e_g</i>	<i>v</i> ₁₇	CD ₂ a-stretch.....	2199 C	ia	2199 VS, dp	
	<i>v</i> ₁₈	CD ₂ s-stretch.....	2104 C	ia	2104 VS, dp	
	<i>v</i> ₁₉	CD ₂ scis.....	1071 C	ia	1071 M, dp	
	<i>v</i> ₂₀	CD ₂ wag.....	1212 C	ia	1212 M, dp	
	<i>v</i> ₂₁	CD ₂ twist.....	937 C	ia	937 S, dp	
	<i>v</i> ₂₂	CC stretch.....	795 C	ia	795 S, dp	
	<i>v</i> ₂₃	CD ₂ rock.....	637 C	ia	637 W, dp	
	<i>v</i> ₂₄	CCC deform. + CC torsion.....	373 C	ia	373 M, dp	
<i>e_u</i>	<i>v</i> ₂₅	CD ₂ a-stretch.....	2206 C	2206 VS	ia	OV (<i>v</i> ₁₂). OV (<i>v</i> ₁₃). CF [4].
	<i>v</i> ₂₆	CD ₂ s-stretch.....	2108 C	2108 VS	ia	
	<i>v</i> ₂₇	CD ₂ scis.....	1069 C	1069 M (liquid)	ia	
	<i>v</i> ₂₈	CD ₂ wag.....	1165 A	1165 VS	ia	
	<i>v</i> ₂₉	CD ₂ twist.....	991 A	991 VS	ia	
	<i>v</i> ₃₀	CD ₂ rock.....	687 B	687 S	ia	
	<i>v</i> ₃₁	CC stretch.....	720 A	720 S	ia	
	<i>v</i> ₃₂	CCC deform. + CC torsion.....	203 C	ia	

References

- [1] IR. S. Abramowitz and R. P. Bauman, J. Chem. Phys. 39, 2757 (1963).
- [2] R. N. I. Peokof'eva, L. M. Sverdlov, and M. M. Sushchinskii, Opt. Spectrosc. 15, 250 (1963).
- [3] IR.R. F. A. Miller and H. R. Golob, Spectrochim. Acta 20, 1517 (1964).
- [4] IR.R.Th. H. Takahashi and T. Shimanouchi, to be published.

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
a_g	ν_1	CH_2 s-stretch	2848 C	cm^{-1} (Solid)	2848 S	
	ν_2	CH_2 scis	1440 C	ia	1440 M	
	ν_3	CC stretch	1131 C	ia	1131 M	
a_u	ν_4	CH_2 twist	^a 1050 D	ia, 1050 VW	ia	
	ν_5	CH_2 wag	1370 D	ia	1370 VW	
b_{1g}	ν_6	CC stretch	1061 C	ia	1061 M	
	ν_7	CH_2 a-stretch	2919 C	2919 S	ia	
	ν_8	CH_2 rock	725 C	^b { 731 S 720 S }	ia	
b_{2g}	ν_9	CH_2 twist	1295 C	ia	1295 M	
	ν_{10}	CH_2 s-stretch	2851 C	2851 S	ia	
	ν_{11}	CH_2 scis	1468 C	^b { 1473 S 1463 S }	ia	
b_{3g}	ν_{12}	CH_2 a-stretch	2883 C	ia	2883 S	
	ν_{13}	CH_2 rock	1168 C	ia	1168 W	
b_{3u}	ν_{14}	CH_2 wag	1176 C	1176 VW	ia	

^a 1063 cm^{-1} is given to this mode in ref. 6.

^b Doublet due to the crystal field effect [1, 8].

References

- [1] IR.Th. S. Krimm, C. Y. Liang, and G. B. B. M. Sutherland, J. Chem. Phys. 25, 549 (1956).
- [2] R. J. R. Nielsen and A. H. Woollett, J. Chem. Phys. 26, 1391 (1957).
- [3] IR. R. G. Snyder, J. Mol. Spectrosc. 4, 411 (1960).
- [4] IR. J. R. Nielsen and R. F. Holland, J. Mol. Spectrosc. 4, 488 (1960); 6, 394 (1961).
- [5] Th. M. Tasumi, T. Shimanouchi, and T. Miyazawa, J. Mol. Spectrosc. 9, 261 (1962).
- [6] IR.Th. J. H. Schachtschneider and R. G. Snyder, Spectrochim. Acta 19, 85, 117 (1963).
- [7] R. R. G. Brown, J. Chem. Phys. 38, 221 (1963).
- [8] Th. M. Tasumi and T. Shimanouchi, J. Chem. Phys. 43, 1245 (1965).
- [9] IR.Th. R. G. Snyder, J. Mol. Spectrosc. 23, 224 (1967).
- [10] IR. R. G. Snyder, J. Chem. Phys. 47, 1316 (1967).
- [11] R. R. G. Snyder, J. Mol. Spectrosc. 36, 222 (1970).

Sym. species	No.	Approximate type of mode	Selected value of frequency	Infrared	Raman	Comments
<i>a_g</i>	<i>ν₁</i>	CD ₂ s-stretch.....	2102 C	<i>cm⁻¹</i> (Solid)	2102 S	
	<i>ν₂</i>	CD ₂ scis.....	1146 C	ia	1146 M	
	<i>ν₃</i>	CC stretch.....	966 E	ia	966 VW	
<i>a_u</i>	<i>ν₄</i>	CD ₂ twist.....	743 E	ia	ia	CF [5].
<i>b_{1g}</i>	<i>ν₅</i>	CD ₂ wag.....	1249 C	ia	1249 W	
	<i>ν₆</i>	CC stretch.....	820 E	ia	CF [5].
<i>b_{1u}</i>	<i>ν₇</i>	CD ₂ a-stretch.....	2192 C	2192 S	ia	
	<i>ν₈</i>	CD ₂ rock.....	526 C	^a { 528 M 522 M }	ia	
<i>b_{2g}</i>	<i>ν₉</i>	CD ₂ twist.....	916 C	ia	916 M	
<i>b_{2u}</i>	<i>ν₁₀</i>	CD ₂ s-stretch.....	2088 C	2088 S	ia	
	<i>ν₁₁</i>	CD ₂ scis.....	1090 C	^a { 1092 S 1087 S }	ia	
<i>b_{3g}</i>	<i>ν₁₂</i>	CD ₂ a-stretch.....	2197 C	ia	2197 M	
	<i>ν₁₃</i>	CD ₂ rock.....	991 C	ia	991 M	
<i>b_{3u}</i>	<i>ν₁₄</i>	CD ₂ wag.....	889 E	ia	CF [5].

^a Doublet due to the crystal field effect [5].

References

- [1] IR. N. Sheppard and G. B. B. M. Sutherland, Nature 159, 739 (1947).
- [2] IR. L. C. Leitch, P. E. Gagnon, and A. Cambron, Can. J. Res. B28, 256 (1950).
- [3] R. R. G. Brown, J. Chem. Phys. 38, 221 (1963).
- [4] IR. M. Tasumi, T. Shimanouchi, H. Tanaka, and S. Ikeda, J. Polymer Sci. A2, 1607 (1964).
- [5] IR.Th. M. Tasumi and T. Shimanouchi, J. Chem. Phys. 43, 1245 (1965).

5. Empirical Formula Index

In this index molecules are divided into two groups: (a) those containing no carbon atoms, which are arranged with the elemental symbols of the empirical formulas in alphabetical order and are listed alphabetically, and in ascending order of the empirical formula subscripts; (b) molecules containing carbon, which are ordered in the same way except that carbon is listed first and hydrogen second. No distinction is made for isotopic species in the empirical formula; this deuterium is listed as H.

Compounds Not Containing Carbon

Empirical formula	Name	Molecule No.	Empirical formula	Name	Molecule No.
AsH ₃	Arsine	21	F ₆ W	Tungsten hexa fluoride	53
AsH ₃	Arsine-d ₃	22	GeH ₄	Germane	33
B ₂ H ₆	Diborane- ¹¹ B ₂ H ₆	55	GeH ₄	Germane-d ₁	34
B ₂ H ₆	Diborane- ¹⁰ B ₂ H ₆	56	GeH ₄	Germane-d ₂	35
B ₂ H ₆	Diborane- ¹⁰ B ₂ D ₆	57	GeH ₄	Germane-d ₃	36
BrCl ₃ Si	Bromotrichlorosilane	45	GeH ₄	Germane-d ₄	37
BrH ₃ Si	Silyl bromide	44	H ₂ O	Water	4
Br ₂ Cl ₂ Si	Dibromodichlorosilane	49	H ₂ O	Water-d ₁	5
Br ₃ ClSi	Tribromochlorosilane	47	H ₂ O	Water-d ₂	6
Br ₄ Ge	Germanium tetrabromide	39	H ₂ S	Hydrogen sulfide	9
Br ₄ Si	Silicon tetrabromide	31	H ₂ S	Deuterium sulfide	10
Br ₄ Sn	Tin tetrabromide	41	H ₂ Se	Hydrogen selenide	12
ClH ₃ Si	Silyl chloride	43	H ₂ Se	Hydrogen deuterium selenide	13
ClI ₃ Si	Chlorotriiodosilane	48	H ₃ N	Ammonia	14
Cl ₂ O	Oxygen dichloride	8	H ₃ N	Ammonia-d ₃	15
Cl ₃ ISi	Trichloroiodosilane	46	H ₃ P	Phosphine	17
Cl ₃ P	Phosphorus trichloride	20	H ₃ P	Phosphine-d ₃	18
Cl ₄ Ge	Germanium tetrachloride	38	H ₃ Sb	Stibine	23
Cl ₄ Si	Silicon tetrachloride	30	H ₃ Sb	Stibine-d ₃	24
Cl ₄ Sn	Tin tetrachloride	40	H ₄ Si	Silane	25
FH ₃ Si	Silyl fluoride	42	H ₄ Si	Silane-d ₂	26
F ₂ O	Oxygen difluoride	7	H ₄ Si	Silane-d ₃	27
F ₃ N	Nitrogen trifluoride	16	H ₄ Si	Silane-d ₄	28
F ₃ P	Phosphorus trifluoride	19	I ₄ Si	Silicon tetraiodide	32
F ₄ Si	Silicon tetrafluoride	29	N ₂ O	Nitrous oxide	1
F ₆ Mo	Molybdenum hexafluoride	52	N ₂ O	Nitrous oxide- ¹⁴ N ¹⁵ NO	2
F ₆ S	Sulfur hexafluoride	50	N ₂ O	Nitrous oxide- ¹⁵ N ₂ O	3
F ₆ Se	Selenium hexafluoride	51	N ₂ O	Sulfur dioxide	11
F ₆ U	Uranium hexafluoride	54	O ₂ S		

Compounds Containing Carbon

Empirical formula	Name	Molecule No.	Empirical formula	Name	Molecule No.
CBrCl ₃	Bromotrichloromethane	93	CHBr ₃	Tribromomethane-d ₁	92
CBrN	Cyanogen bromide- ⁷⁹ BrCN	66	CHCl ₃	Trichloromethane	89
CBrN	Cyanogen bromide- ⁸¹ BrCN	67	CHCl ₃	Trichloromethane-d ₁	90
CB ₂ Cl ₂	Dibromodichloromethane	101	CHF ₃	Trifluoromethane	88
CB ₃ Cl	Tribromochloromethane	94	CHN	Hydrogen cyanide	62
CB ₄	Carbon tetrabromide	78	CHN	Deuterium cyanide	63
CCIN	Cyanogen chloride- ³⁵ ClCN	64	CH ₂ BrCl	Bromochloromethane	102
CCIN	Cyanogen chloride- ³⁷ ClCN	65	CH ₂ BrCl	Bromochloromethane-d ₁	103
CCl ₄	Carbon tetrachloride	77	CH ₂ BrCl	Bromochloromethane-d ₂	104
CF ₄	Carbon tetrafluoride	76	CH ₂ Br ₂	Dibromomethane	98
CI ₄	Carbon tetrachloride	79	CH ₂ Br ₂	Dibromomethane-d ₁	99
COS	Carbonyl sulfide	61	CH ₂ Br ₂	Dibromomethane-d ₂	100
CO ₂	Carbon dioxide	58	CH ₂ Cl ₂	Dichloromethane	95
CO ₂	Carbon dioxide- ¹³ CO ₂	59	CH ₂ Cl ₂	Dichloromethane-d ₁	96
CS ₂	Carbon disulfide	60	CH ₂ Cl ₂	Dichloromethane-d ₂	97
CHBr ₃	Tribromomethane	91	CH ₂ O	Formaldehyde	68

Empirical formula	Name	Molecule No.	Empirical formula	Name	Molecule No.
CH ₂ O	Formaldehyde-d ₁	69	C ₂ H ₃ N	Methyl isocyanide-d ₃	146
CH ₃ O	Formaldehyde-d ₂	70	C ₂ H ₄	Ethylene	124
CH ₂ O ₂	Formic acid	105	C ₂ H ₄	Ethylene-d ₄	125
CH ₂ O ₂	Formic acid-d ₂	106	C ₂ H ₄ BrCl	1-Bromo-2-chloroethane, trans form	164
CH ₂ Br	Methyl bromide	84	C ₂ H ₄ BrCl	1-Bromo-2-chloroethane, gauche form	165
CH ₃ Br	Methyl bromide-d ₃	85	C ₂ H ₄ Br ₂	1,2-Dibromoethane, trans form	162
CH ₃ Cl	Methyl chloride	82	C ₂ H ₄ Br ₂	1,2-Dibromoethane, gauche form	163
CH ₃ Cl	Methyl chloride-d ₃	83	C ₂ H ₄ Cl ₂	1,2-Dichloroethane, trans form	160
CH ₃ F	Methyl fluoride	80	C ₂ H ₄ Cl ₂	1,2-Dichloroethane, gauche form	161
CH ₃ F	Methyl fluoride-d ₃	81	C ₂ H ₄ O	Ethylene oxide	149
CH ₃ I	Methyl iodide	86	C ₂ H ₄ O	Ethylene oxide-d ₄	150
CH ₃ I	Methyl iodide-d ₃	87	C ₂ H ₄ O	Acetaldehyde	151
CH ₄	Methane	71	C ₂ H ₄ O	Acetaldehyde-d ₁	152
CH ₄	Methane-d ₁	72	C ₂ H ₄ O	Acetaldehyde-d ₄	153
CH ₄	Methane-d ₂	73	C ₂ H ₄ O ₂	Methyl formate	170
CH ₄	Methane-d ₃	74	C ₂ H ₄ O ₂	Methyl formate-d ₁	171
CH ₄	Methane-d ₄	75	C ₂ H ₄ O ₂	Methyl formate-d ₃	172
CH ₄ O (Gas)	Methanol	107	C ₂ H ₄ O ₂	Methyl formate-d ₄	173
CH ₄ O	Methanol	108	C ₂ H ₄ O ₂	Acetic acid	174
(Liquid)			C ₂ H ₄ O ₂	Acetic acid-d ₁	175
CH ₄ O (Gas)	Methanol-d ₁	109	C ₂ H ₅ Si	Silyl acetylene	148
CH ₄ O	Methanol-d ₁	110	C ₂ H ₅ Br	Bromoethane	168
(Liquid)			C ₂ H ₅ Cl	Chloroethane	167
CH ₄ O (Gas)	Methanol-d ₃	111	C ₂ H ₅ F	Fluoroethane	166
CH ₄ O	Methanol-d ₃	112	C ₂ H ₅ N	Ethylene imine	169
(Liquid)			C ₂ H ₆	Ethane	154
CH ₄ O (Gas)	Methanol-d ₄	113	C ₂ H ₆	Ethane-d ₃	155
CH ₅ N	Methylamine	114	C ₂ H ₆	Ethane-d ₄	156
CH ₅ N	Methylamine-d ₂	115	C ₂ H ₆ O	Dimethyl ether	176
CH ₅ N	Methylamine-d ₃	116	C ₂ H ₆ O	Dimethyl ether-d ₃	177
CH ₅ N	Methylamine-d ₅	117	C ₃ H ₆ N ₂	Malononitrile	183
C ₂ Br ₄	Tetrabromoethylene	128	C ₃ H ₆ N ₂	Malononitrile-d ₂	184
C ₂ Br ₆	Hexabromoethane	159	C ₃ H ₄	Allene	178
C ₂ Cl ₂ F ₂	Trans-1,2-Dichloro-1,2-difluoroethylene	138	C ₃ H ₄	Methylacetylene	179
C ₂ Cl ₂ F ₂	1,1-Dichloro-2,2-difluoroethylene	142	C ₃ H ₄	Methylacetylene-d ₁	180
C ₂ Cl ₄	Tetrachloroethylene	127	C ₃ H ₄	Methylacetylene-d ₃	181
C ₂ Cl ₆	Hexachloroethane	158	C ₃ H ₄ O	Methylacetylene-d ₄	182
C ₂ F ₄	Tetrafluoroethylene	126	C ₃ H ₅ N	Propenal	185
C ₂ F ₆	Hexafluoroethane	157	C ₃ H ₆	Ethyl cyanide	188
C ₂ HBr	Bromoacetylene	123	C ₃ H ₆	Cyclopropane	186
C ₂ HCl	Chloroacetylene	122	C ₃ H ₆ O	Cyclopropane-d ₆	187
C ₂ HF	Fluoroacetylene	121	C ₃ H ₆ O	Acetone	189
C ₂ H ₂	Acetylene	118	C ₃ H ₆ O	Acetone-d ₃	190
C ₂ H ₂	Acetylene-d ₁	119	C ₃ H ₆ O ₂	Acetone-d ₆	191
C ₂ H ₂	Acetylene-d ₂	120	C ₃ H ₆ O ₂	Methyl acetate	197
C ₂ H ₂ Cl ₂	Trans-1,2-Dichloroethylene	132	C ₃ H ₆ O ₂	Methyl-d ₃ -acetate	198
C ₂ H ₂ Cl ₂	Trans-1,2-Dichloroethylene-d ₁	133	C ₃ H ₆ O ₂	Methyl acetate-d ₃	199
C ₂ H ₂ Cl ₂	Trans-1,2-Dichloroethylene-d ₂	134	C ₃ H ₈	Propane	192
C ₂ H ₂ Cl ₂	Cis-1,2-Dichloroethylene	135	C ₃ H ₈	Propane-d ₃	193
C ₂ H ₂ Cl ₂	Cis-1,2-Dichloroethylene-d ₁	136	C ₃ H ₈	Propane-d ₂	194
C ₂ H ₂ Cl ₂	Cis-1,2-Dichloroethylene-d ₂	137	C ₃ H ₈	Propane-d ₆	195
C ₂ H ₂ Cl ₂	1,1-Dichloroethylene	139	C ₃ H ₈	Propane-d ₈	196
C ₂ H ₂ Cl ₂	1,1-Dichloroethylene-d ₁	140	C ₄ H ₂	Butadiyne	201
C ₂ H ₂ Cl ₂	1,1-Dichloroethylene-d ₂	141	C ₄ H ₄ O	Furan	202
C ₂ H ₂ F ₂	Cis-1,2-Difluoroethylene	129	C ₄ H ₄ S	Thiophene	203
C ₂ H ₂ F ₂	Cis-1,2-Difluoroethylene-d ₁	130	C ₄ H ₄ S	Thiophene-d ₄	204
C ₂ H ₂ F ₂	Cis-1,2-Difluoroethylene-d ₂	131	C ₄ H ₆	1,3-Butadiene	205
C ₂ H ₂ N ₂ O	1,2,5-Oxadiazole	147	C ₄ H ₆	1,3-Butadiene-d ₁ , trans	206
C ₂ H ₃ N	Methyl cyanide	143	C ₄ H ₆	1,3-Butadiene-1,1,2-d ₃	207
C ₂ H ₃ N	Methyl cyanide-d ₃	144	C ₄ H ₆	1,3-Butadiene-1,1,4,4-d ₄	208
C ₂ H ₃ N	Methyl isocyanide	145	C ₄ H ₆	1,3-Butadiene-d ₆	209

Empirical formula	Name	Molecule No.	Empirical formula	Name	Molecule No.
C ₄ H ₆	2-Butyne	210	C ₄ H ₁₀	n-Butane, gauche form	127
C ₄ H ₈	Cyclobutane	211	C ₆ H ₆	Benzene	218
C ₄ H ₈	Cyclobutane-d ₈	212	C ₆ H ₆	Benzene-d ₆	219
C ₄ H ₈	2-Methylpropene	213	C ₆ H ₁₂	Cyclohexane	220
C ₄ H ₈	2-Methyl-d ₃ -propene-3,3-d ₃	214	C ₆ H ₁₂	Cyclohexane-d ₁₂	221
C ₄ H ₈ O	2-Butanone, trans form	215	(CH ₂)n	Poly(methylene)	222
C ₄ H ₁₀	n-Butane, trans form	216	(CH ₂)n	Poly(methylene-d ₂)	223