

1 **Chapter 5. Potential for Abrupt Changes in Atmospheric**

2 **Methane**

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9 **Key Findings**

- 10 • The main concerns about abrupt changes in atmospheric methane (CH₄) stem
11 from (1) the large quantity of methane believed to be stored as methane
12 hydrate in the sea floor and permafrost soils and (2) climate-driven changes
13 in emissions from northern high-latitude and tropical wetlands.
- 14 • The size of the hydrate reservoir is uncertain, perhaps by up to a factor of 10.
15 Because the size of the reservoir is directly related to the perceived risks, it is
16 difficult to make certain judgment about those risks.
- 17 • There are a number of suggestions in the scientific literature about the
18 possibility of catastrophic release of methane to the atmosphere based on
19 both the size of the hydrate reservoir and indirect evidence from
20 paleoclimatological studies. However, modeling and detailed studies of ice
21 core methane so far do not support catastrophic methane releases to the
22 atmosphere in the last 650,000 years or the near future. A very large release
23 of methane may have occurred at the Paleocene-Eocene boundary (about 55
24 million years ago) but other explanations for the evidence have been offered.
- 25 • The current network of atmospheric methane monitoring sites is sufficient for
26 capturing large-scale changes in emissions, but it is insufficient for attributing
27 changes in emissions to one specific source sector.

- 1 • Observations show that there have not yet been significant increases in
2 methane emissions from northern high-latitude hydrates and wetlands
3 resulting from increasing Arctic temperatures.
- 4 • While catastrophic release of methane to the atmosphere appears very
5 unlikely, it is very likely that climate change will accelerate the pace of
6 chronic emissions from both hydrate sources and wetlands. The magnitude of
7 these releases is difficult to estimate.

8 **Recommendations**

- 9 • Monitoring the abundance of atmospheric methane and its isotopic
10 composition should be maintained and expanded to allow detection of change
11 in net emissions from northern and tropical wetland regions. Specifically,
12 systematic measurements of CH₄ from tall towers and aircraft in the Arctic
13 and sub-Arctic regions would allow detection of changes in emissions from
14 these sparsely monitored but important regions. For the tropics, expanded
15 surface measurements and continued observations of CH₄ abundances are
16 required.
- 17 • The feasibility of monitoring methane in the ocean water column near marine
18 hydrate deposits, or in the atmosphere near terrestrial hydrate deposits, to
19 detect changes in emissions from those sources, should be investigated, and if
20 feasible this monitoring should be implemented.
- 21 • Efforts should be made to increase certainty in the size of the global methane
22 hydrate reservoirs. The level of concern about catastrophic release of
23 methane to the atmosphere is directly linked to the size of these reservoirs.
- 24 • The size and location of hydrate reservoirs that are most vulnerable to climate
25 change (for example shallow water deposits, shallow sub-surface deposits on
26 land, or regions of potential large submarine landslides) should be identified
27 accurately and their potential impact on future methane concentrations should
28 be evaluated.

- 1 • Improvement in process-based modeling of methane release from marine
2 hydrates is needed. The transport of bubbles is particularly important, as are
3 the migration of gas through the stability zone and the mechanisms
4 controlling methane release from submarine landslides.
- 5 • Modeling efforts should establish the current and future climate-driven
6 acceleration of chronic release of methane from wetlands and terrestrial
7 hydrate deposits. This should include development of improved
8 representations of wetland hydrology and biogeochemistry, and permafrost
9 dynamics, in earth system and global climate models.
- 10 • Further work on the ice core record of atmospheric methane is needed to fully
11 understand the implications of past abrupt changes in atmospheric methane.
12 This work should include high-resolution and high-precision measurements
13 of methane mixing ratios and isotopic ratios, and biogeochemical modeling
14 of past methane emissions and relevant atmospheric chemical cycles. Further
15 development of high-resolution proxies of low-latitude climate and better
16 records of pre-Last Glacial Maximum wetlands are also needed.

17 **1. Background: Why Are Abrupt Changes in Methane a Potential Concern?**

18 **1.1 Introduction**

19 Methane (CH₄) is the most important greenhouse gas that humans directly influence, after
20 CO₂. Concerns about methane's role in abrupt climate change stem primarily from (1) the
21 large quantities of methane stored as solid methane hydrate on the sea floor and to a
22 lesser degree in terrestrial sediments, and the possibility that these reservoirs could be
23 unstable in the face of future global warming, and (2) the possibility of large-scale
24 conversion of frozen soil in the high latitude Northern Hemisphere to methane producing
25 wetland, due to accelerated warming at high latitudes. This chapter summarizes the
26 current state of knowledge about these reservoirs and their potential for forcing abrupt
27 climate change.

28 **1.2 Methane and Climate**

29 A spectral window exists between ~7 and 12 micrometers (μm) where the atmosphere is
30 somewhat transparent to terrestrial infrared (IR) radiation. Increases in the atmospheric

1 abundance of molecules that absorb IR radiation in this spectral region contribute to the
2 greenhouse effect. Methane is a potent greenhouse gas because it strongly absorbs
3 terrestrial IR radiation near 7.66 μm , and its atmospheric abundance has more than
4 doubled since the start of the Industrial Revolution. Radiative forcing (RF) is used to
5 assess the contribution of a perturbation (in this case, the increase in CH_4 since 1750
6 A.D.) to the net irradiance at the top of the tropopause after allowing the stratosphere to
7 adjust to radiative equilibrium. The direct radiative forcing of atmospheric methane
8 determined from an increase in its abundance from its pre-industrial value of 700 parts
9 per billion (ppb) (*MacFarlane-Meure et al., 2006; Etheridge et al., 1998*) to its globally
10 averaged abundance of 1,775 ppb in 2006 is 0.49 ± 0.05 watts per square meter (W m^{-2})
11 (*Hofmann et al., 2006*). Methane oxidation products, stratospheric water (H_2O) vapor,
12 and tropospheric ozone (O_3), contribute indirectly to radiative forcing, increasing
13 methane's total contribution to $\sim 0.7 \text{ W m}^{-2}$ (e.g., *Hansen and Sato, 2001*), nearly half of
14 that for carbon dioxide (CO_2). Increases in methane emissions can also increase the
15 methane lifetime and the lifetimes of other gases oxidized by the hydroxyl radical (OH).
16 Assuming the abundances of all other parameters that affect OH stay the same, the
17 lifetime for an additional pulse of CH_4 (e.g., 1 teragram, Tg; $1 \text{ Tg} = 10^{12} \text{ g} = 0.001 \text{ Gt}$,
18 gigaton) added to the atmosphere would be $\sim 40\%$ larger than the current value.
19 Additionally, CH_4 is oxidized to CO_2 ; CO_2 produced by CH_4 oxidation is equivalent to
20 $\sim 6\%$ of CO_2 emissions from fossil fuel combustion. Over a 100-year time horizon, the
21 direct and indirect effects on RF of emission of 1 kilogram (kg) CH_4 are 25 times greater
22 than for emission of 1 kg CO_2 (*Forster et al., 2007*).

23 The atmospheric abundance of CH_4 increased with human population because of
24 increased demand for energy and food. Beginning in the 1970s, as CH_4 emissions from
25 natural gas venting and flaring at oil production sites declined and rice agriculture
26 stabilized, the growth rate of atmospheric CH_4 decoupled from population growth. Since
27 1999, the global atmospheric CH_4 abundance has been nearly stable; globally averaged
28 CH_4 in 1999 was only 3 ppb less than the 2006 global average of 1775 ppb. Potential
29 contributors to this stability are decreased emissions from the Former Soviet Union after
30 their economy collapsed in 1992 (*Dlugokencky et al., 2003*), decreased emissions from
31 natural wetlands because of widespread drought (*Bousquet et al., 2006*), decreased

1 emissions from rice paddies due to changes in water management (*Li et al., 2002*), and an
2 increase in the chemical sink because of changing climate (*Fiore et al., 2006*). Despite
3 attempts to explain this surprising observation, the exact causes remain unknown, making
4 predictions of future methane levels difficult. *Hansen et al. (2000)* have suggested that,
5 because methane has a relatively short atmospheric lifetime (see below) and reductions in
6 emissions are often cost effective, it is an excellent gas to target to counter increasing RF
7 of CO₂ in the short term.

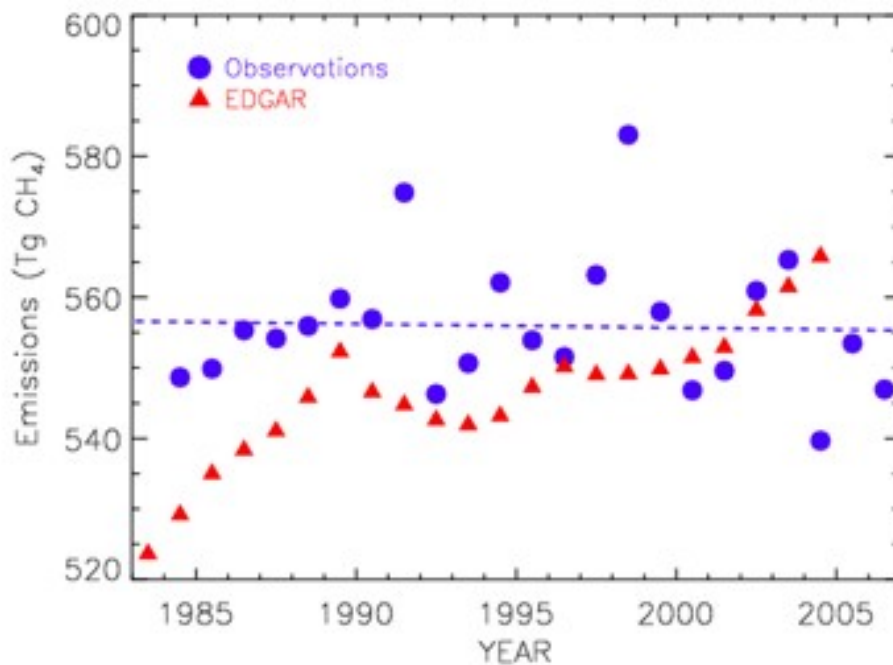
8 **1.3 The Modern Methane Budget**

9 The largest individual term in the global methane budget is removal (removal terms are
10 referred to as “sinks”) from the atmosphere by oxidation of methane initiated by reaction
11 with hydroxyl radical (OH; $\text{OH} + \text{CH}_4 \rightarrow \text{CH}_3 + \text{H}_2\text{O}$) in the troposphere.

12 Approximately 90% of atmospheric CH₄ is removed by this reaction, so estimates of OH
13 concentrations as a function of time can be used to establish how much methane is
14 removed from the atmosphere. When combined with measurements of the current trends
15 in atmospheric methane concentrations, these estimates provide a powerful constraint on
16 the total source. OH is too variable for its large-scale, time-averaged concentration to be
17 determined by direct measurements, so measurements of 1,1,1-trichloroethane (methyl
18 chloroform), an anthropogenic compound with relatively well-known emissions and
19 predominant OH sink, are most commonly used as a proxy. As assessed by the IPCC
20 Fourth Assessment Report (*Forster et al., 2007*), the globally averaged OH concentration
21 is $\sim 10^6$ per cubic centimeter (cm⁻³), and there was no detectable change from 1979 to
22 2004. Reaction with OH is also the major CH₄ loss process in the stratosphere. Smaller
23 atmospheric sinks include oxidation by chlorine in the troposphere and stratosphere and
24 oxidation by electronically excited oxygen atoms [O(1D)] in the stratosphere.

25 Atmospheric CH₄ is also oxidized by bacteria (methanotrophs) in soils, a term which is
26 usually included in budgets as a negative source. These sink terms result in an
27 atmospheric CH₄ lifetime of ~ 9 years ($\pm 10\%$). In other words, at steady state, each year
28 one ninth of the total amount of methane in the atmosphere is removed by oxidation, and
29 replaced by emissions to the atmosphere.

1 When an estimate of the lifetime is combined with global observations in a one-box mass
2 balance model of the atmosphere (that is, considering the entire atmosphere to be a well-
3 mixed uniform box), total global emissions can be calculated with reasonable certainty.
4 Using a lifetime of 8.9 years and NOAA (National Oceanic and Atmospheric
5 Administration) Earth System Research Laboratory (ESRL) global observations of CH₄
6 and its trend gives average emissions of 556±10 teragrams (Tg) CH₄ per year (yr⁻¹), with
7 no significant trend for 1984-2006 ([Figs. 5.1](#) and [5.7](#)). The uncertainty on total emissions
8 is 1 standard deviation (s.d.) of the interannual variability; total uncertainty is on order of
9 ±10%. The total amount of methane in the atmosphere (often referred to as the
10 atmospheric “burden”) is ~5,000 Tg, or 5 Gt CH₄.



11

12 **Figure 5.1.** Methane emissions as function of time calculated with constant lifetime;
13 emissions from EDGAR inventory with constant natural emissions shown as red
14 triangles. EDGAR is Emission Database for Global Atmospheric Research (described in
15 *Olivier and Berdowski, 2001*); 2001 to 2004 emissions are preliminary (source:
16 [http://www.milieuennatuurcompendium.nl/indicatoren/nl0167-Broeikasgasemissies%2C-](http://www.milieuennatuurcompendium.nl/indicatoren/nl0167-Broeikasgasemissies%2C-mondiaal.html?i=9-20)
17 [mondiaal.html?i=9-20](http://www.milieuennatuurcompendium.nl/indicatoren/nl0167-Broeikasgasemissies%2C-mondiaal.html?i=9-20)).

1 Methane is produced by a variety of natural and anthropogenic sources. Estimates of
 2 emissions from individual sources are made using bottom-up and top-down methods.
 3 Bottom-up inventories use emission factors (e.g., average emissions of CH₄ per unit area
 4 for a specific wetland type) and activity levels (e.g., total area of that wetland type) to
 5 calculate emissions. Because the relatively few measurements of emission factors are
 6 typically extrapolated to large spatial scales, uncertainties in emissions estimated with the
 7 bottom-up approach are typically quite large. An example of the top-down method
 8 applied to the global scale using a simple 1-box model is shown in [Fig. 5.1](#) and described
 9 above, but the method can also be applied using a three-dimensional chemical transport
 10 model to optimize emissions from regional to continental scales based on a comparison
 11 between model-derived mixing ratios and observations. Bottom-up inventories are
 12 normally used as initial guesses in this approach. This approach is used to estimate
 13 emissions by source and region. [Table 5.1](#) shows optimized CH₄ emissions calculated
 14 from an inverse modeling study (*Bergamaschi et al., 2007*, scenario 3) that was
 15 constrained by in situ surface observations and satellite-based estimates of column-
 16 averaged CH₄ mixing ratios. It should be noted that optimized emissions from inverse
 17 model studies depend on the a priori estimates of emissions and the observational
 18 constraints, and realistic estimates of uncertainties are still a challenge. For example,
 19 despite the small uncertainties given in the table for termite emissions, emissions from
 20 this sector varied from ~31 to 67 Tg yr⁻¹ over the range of scenarios tested, which is a
 21 larger range than the uncertainties in the table would imply. While total global emissions
 22 are fairly well constrained by this combination of measurements and lifetime, individual
 23 source terms still have relatively large uncertainties.

24 **Table 5.1.** Annual CH₄ emissions for 2003 by source type (from scenario 3 of
 25 *Bergamaschi et al., 2007*); chemical sinks are scaled to total emissions based on *Lelieveld*
 26 *et al. (1998)*. Tg/yr, teragrams per year; 1 Tg = 10¹² g.

Source	Emissions (Tg/yr)	Fraction of total (%)
Coal	35.6±4.4	6.7
Oil and gas	41.8±5.5	7.9
Enteric fermentation	82.0±9.6	15.4
Rice agriculture	48.7±5.1	9.2
Biomass burning	21.9±2.6	4.1
Waste	67.0±10.7	12.6
Wetlands	208.5±7.6	39.2

Wild animals	6.8±2.0	1.3
Termites	42.0±6.7	7.9
Soil	-21.3±5.8	-4.0
Oceans	-1.3±2.9	-0.2
Total	531.6±3.7	
Chemical Sinks	Loss (Tg/yr)	
Troposphere	490±50	92.5
Stratosphere	40±10	7.5
Total	530	

1 The constraint on the total modern source strength is important because any new
 2 proposed source (for example, a larger than previously identified steady state marine
 3 hydrate source) would have to be balanced by a decrease in the estimated magnitude of
 4 another source. The budget presented in [Table 5.1](#) refers to net fluxes to the atmosphere
 5 only. The gross production of methane is very likely to be significantly larger, but
 6 substantial quantities of methane are consumed in soils, oxic freshwater, and the ocean
 7 before reaching the atmosphere (*Reeburgh, 2004*). (The soil sink in [Table 5.1](#) refers only
 8 to removal of atmospheric methane by oxidation in soils).

9 Given the short CH₄ lifetime (~9 yr), short-term changes in methane emissions from
 10 climatically sensitive sources such as biomass burning and wetlands, or in sinks, are seen
 11 immediately in surface observations of atmospheric methane. As implied above, reaction
 12 with methane is one of the major sinks for OH radical (the main methane sink), and
 13 therefore increases in methane levels should cause an increase in the lifetimes of methane
 14 and other long-lived greenhouse gases consumed by OH. Higher methane emissions
 15 therefore mean increased methane lifetimes, which in turn means that the impact of any
 16 short-term increase in methane emissions will last longer.

17 **1.4 Observational Network and Its Current Limitations, Particularly Relative to the** 18 **Hydrate, Permafrost, and Arctic Wetland Sources**

19 The network of air sampling sites where atmospheric methane mixing ratios are measured
 20 can be viewed on the World Meteorological Organization (WMO) World Data Centre for
 21 Greenhouse Gases (WDCGG) Web site (<http://gaw.kishou.go.jp/wdcgg/>) and is
 22 reproduced in [Figure 5.2](#). Methane data have been reported to the WDCGG for ~130
 23 sites. Relatively few measurements are reported for the Arctic, and sites are typically far

1 from potential permafrost, hydrate, and wetland sources. Existing Arctic sites have been
2 used to infer decreased emissions from the fossil-fuel sector of the Former Soviet Union
3 (*Dlugokencky et al., 2003*) and provide boundary conditions for model studies of
4 emissions, but they are too remote from source regions to accurately quantify emissions,
5 so uncertainties on northern emissions will remain large until more continuous
6 measurement sites are added close to sources. The optimal strategy would include
7 continuous measurements from tall towers and vertical profiles collected from aircraft.
8 Measurements from tall towers are influenced by emissions from much larger areas than
9 eddy-correlation flux techniques, which have footprints on the order of 1 square
10 kilometer (km²). When combined with global- or regional-scale models, these
11 measurements can be used to quantify fluxes; the vertical profiles would be used to assess
12 the quality of the model results through the troposphere. To properly constrain CH₄
13 emissions in the tropics, retrievals of CH₄ column-averaged mixing ratios must be
14 continued to complement surface observations.



15
16 **Figure 5.2.** Global network of monitoring sites (blue dots) for long-term observation of
17 atmospheric methane as of this date (<http://gaw.kishou.go.jp/wdcgg/>).

18 **1.5 Abrupt Changes in Atmospheric Methane?**

19 Concern about abrupt changes in atmospheric methane stems largely from the massive
20 amounts of methane present as solid methane hydrate in ocean sediments and terrestrial
21 sediments, which may become unstable in the face of future warming. Methane hydrate is

1 a solid substance that forms at low temperatures and high pressures in the presence of
2 sufficient methane, and is found primarily in marine continental margin sediments and
3 some arctic terrestrial sedimentary deposits (see [Box 5.1](#)). Warming or release of pressure
4 can destabilize methane hydrate, forming free gas that may ultimately be released to the
5 atmosphere. The processes controlling hydrate stability and gas transport are complex,
6 and only partly understood. Estimates of the total amount of methane hydrate vary
7 widely, from 500 to 10,000 gigatons of carbon (GtC) stored as methane in hydrates in
8 marine sediments, and 7.5 to 400 GtC in permafrost (both figures are uncertain, see [Sec.](#)
9 [4](#) below). The total amount of carbon in the modern atmosphere is ~810 GtC, but the total
10 methane content of the atmosphere is only ~4 GtC (*Dlugokencky et al., 1998*). Therefore,
11 even a release of a small portion of the methane hydrate reservoir to the atmosphere
12 could have a substantial impact on radiative forcing.

13 **Box 5.1—Chemistry, Physics, and Occurrence of Methane Hydrate**

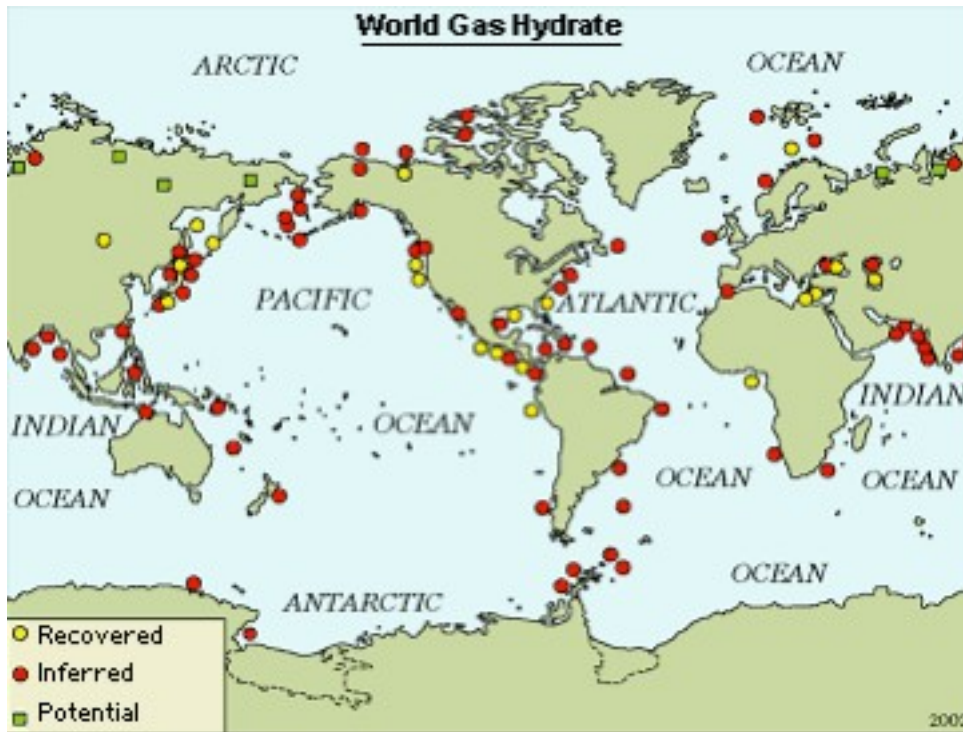
14 A clathrate is a substance in which a chemical lattice or cage of one type of molecule
15 traps another type of molecule. Gas hydrates are substances in which gas molecules
16 are trapped in a lattice of water molecules ([Fig. 5.3](#)). The potential importance of
17 methane hydrate to abrupt climate change results from the fact that large amounts of
18 methane can be stored in a relatively small volume of solid hydrate. For example, 1
19 cubic meter (m³) of methane hydrate is equivalent to 164 m³ of free gas (and 0.8 m³
20 of water) at standard temperature and pressure (*Kvenvolden, 1993*). Naturally
21 occurring gas hydrate on Earth is primarily methane hydrate and forms under high
22 pressure – low temperature conditions in the presence of sufficient methane. These
23 conditions are most often found in relatively shallow marine sediments on continental
24 margins, but also in some high-latitude terrestrial sediments ([Fig. 5.4](#)). Although the
25 amount of methane stored as hydrate in geological reservoirs is not well quantified, it
26 is very likely that very large amounts are sequestered in comparison to the present
27 total atmospheric methane burden.

28 The right combination of pressure and temperature conditions forms what is known as
29 the hydrate stability zone, shown schematically in [Fig. 5.5](#). In marine sediments
30 pressure and temperature both increase with depth, creating a relatively narrow region
31 where methane hydrate is stable. Whether or not methane hydrate forms depends not

1 only on temperature and pressure but also on the amount of methane present. The
2 latter constraint limits methane hydrate formation to locations of significant biogenic
3 or thermogenic methane (*Kvenvolden, 1993*). When ocean bottom water temperatures
4 are near 0°C, hydrates can form at shallow depths, below ~200 m water depth, if
5 sufficient methane is present. The upper limit of the hydrate stability zone can
6 therefore be at the sediment surface, or deeper in the sediment, depending on pressure
7 and temperature. The stability zone thickness increases with water depth in typical
8 ocean sediments. It is important to note, however, that most marine methane hydrates
9 are found in shallow water near continental margins, in areas where the organic
10 carbon content of the sediment is sufficient to fuel methanogenesis. In terrestrial
11 sediments hydrate can form at depths of ~200 m and deeper, in regions where surface
12 temperatures are cold enough that temperatures at 200 m are within the hydrate
13 stability zone.



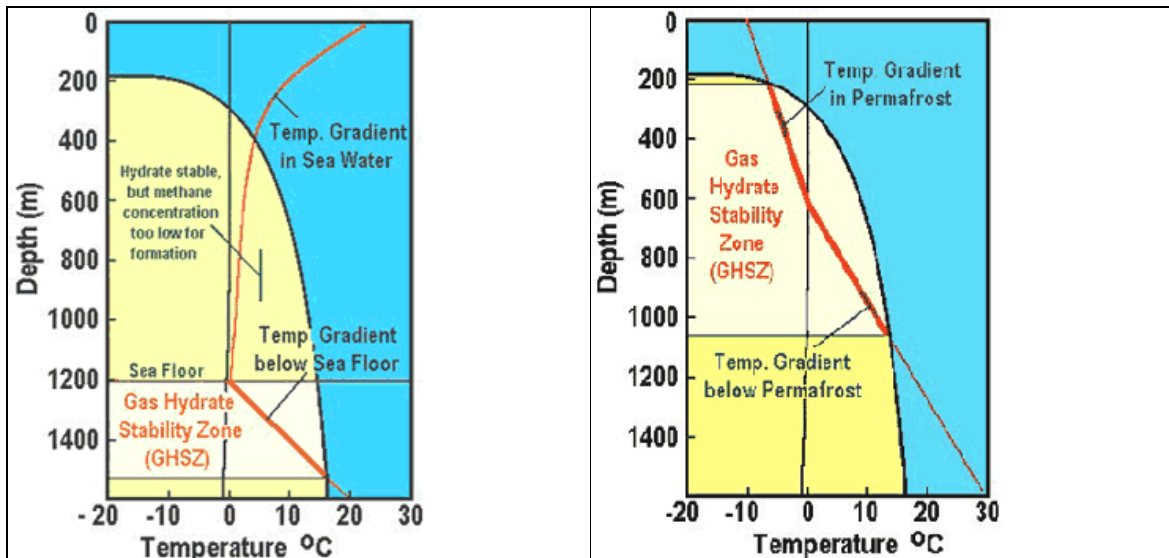
15 **Figure 5.3.** Photographs of methane hydrate as nodules, veins, and laminae in sediment.
16 Source: http://geology.usgs.gov/connections/mms/joint_projects/methane.htm.



1

2 **Figure 5.4.** Map of methane hydrate deposit locations. Source:
 3 http://geology.usgs.gov/connections/mms/joint_projects/methane.htm.

4



5 **Figure 5.5.** Schematic diagram of hydrate stability zone for typical continental margin
 6 (left) and permafrost (right) settings. The red line shows the temperature gradient with
 7 depth. The hydrate stability zone is technically the depth interval where the in situ
 8 temperature is lower than the temperature of the phase transition between hydrate and

1 free gas. In the ocean this can occur above the sea floor, but generally there is not
2 sufficient methane in the water column for methane hydrate to form. For this reason the
3 stability zone in the left figure terminates at the sea floor. From National Energy
4 Technology Laboratory ([http://204.154.137.14/technologies/oil-](http://204.154.137.14/technologies/oil-gas/FutureSupply/MethaneHydrates/about-hydrates/conditions.htm)
5 [gas/FutureSupply/MethaneHydrates/about-hydrates/conditions.htm](http://204.154.137.14/technologies/oil-gas/FutureSupply/MethaneHydrates/about-hydrates/conditions.htm)).

6 Massive releases of methane from marine or terrestrial hydrates have not been observed.
7 Evidence from the ice core record indicates that abrupt shifts in methane concentration
8 have occurred in the past 110,000 years (*Chappellaz et al., 1993a; Brook et al., 1996,*
9 *2000*), although the concentration changes during these events were relatively small.
10 Farther back in geologic time, an abrupt warming at the Paleocene-Eocene boundary
11 about 55 million years ago has been attributed to a large release of methane to the
12 atmosphere, although alternate carbon sources such as oxidation of sedimentary organic
13 carbon or peats have also been proposed (see discussion in [Sec. 4](#)). These past abrupt
14 changes are discussed in detail below, and their existence provides further motivation for
15 considering the potential for future abrupt changes in methane.

16 The large impact of a substantial release of methane hydrates to the atmosphere, if it were
17 to occur, coupled with the potential for a more steady increase in methane production
18 from melting hydrates and from wetlands in a warming climate, motivates several
19 questions this chapter attempts to address:

- 20 1. What is the volume of methane in terrestrial and marine sources and how much of
21 it is likely to be released if climate warms in the near future?
- 22 2. What is the impact on the climate system of the release of varying quantities of
23 methane over varying intervals of time?
- 24 3. What is the evidence in the past for abrupt climate change caused by massive
25 methane release?
- 26 4. What conditions (in terms of sea level rise and warming of bottom waters) would
27 allow methane release from hydrates locked up in sea-floor sediments?
- 28 5. How much methane is likely to be released by warming of northern high-latitude
29 soils, sea level rise, and other climate-driven changes in wetlands?

- 1 6. What are the observational and modeling requirements necessary to understand
2 methane storage and its release under various future scenarios of abrupt climate
3 change?

4 **2. History of Atmospheric Methane**

5 Over the last ~300 years the atmospheric methane mixing ratio increased from ~700-750
6 ppb in 1700 A.D. to a global average of ~1,775 ppb in 2006. Direct atmospheric
7 monitoring has been conducted in a systematic way only since the late 1970s, and data
8 for previous times come primarily from ice cores ([Fig. 5.6](#)). Current levels of methane are
9 anomalous with respect to the long-term ice core record, which now extends back to
10 650,000 years (*Spahni et al., 2005*). There are no direct constraints on methane levels
11 beyond 800,000 years, the age at the bottom of the oldest ice core now available
12 (European Project for Ice Coring in Antarctica (EPICA) Dome C ice core). New
13 international plans to drill at a very low accumulation rate site in Antarctica may in the
14 future extend the record to 1.5 million years (*Brook and Wolff, 2005*).

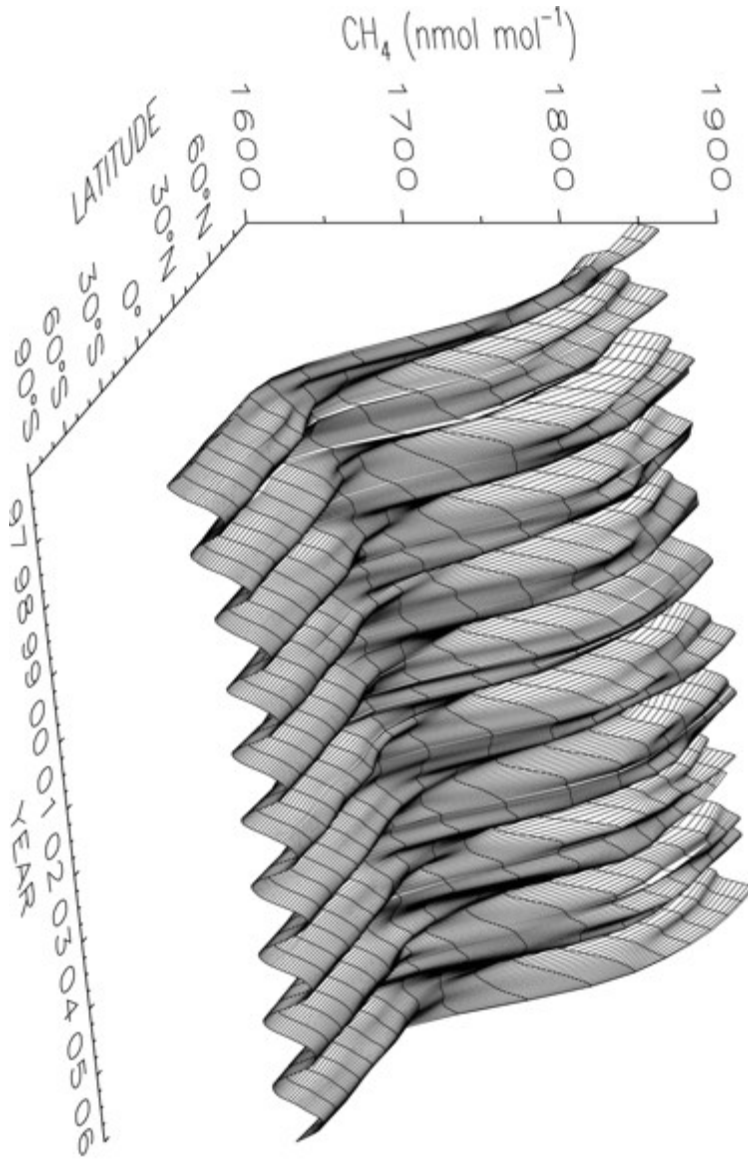
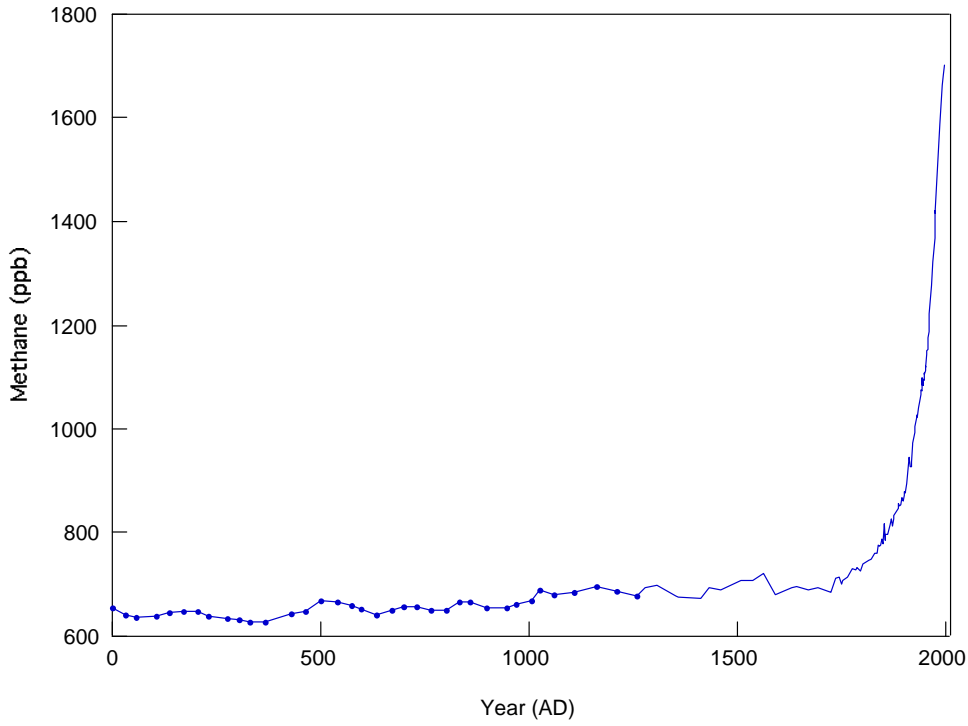
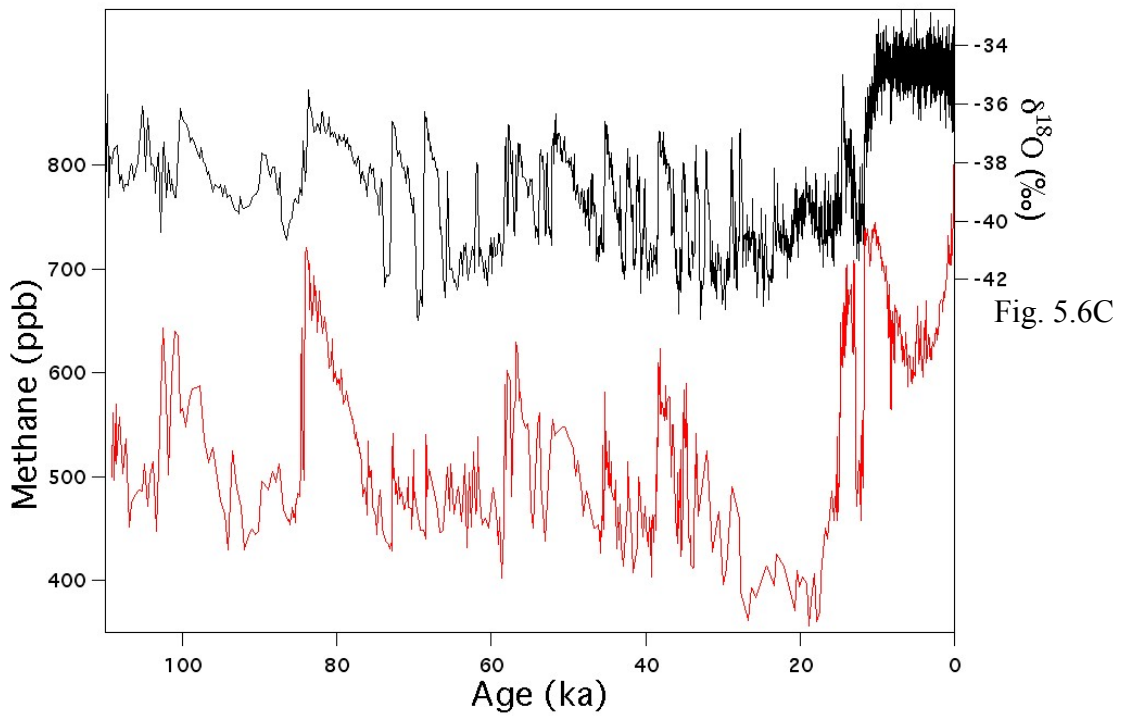


Fig. 5.6A

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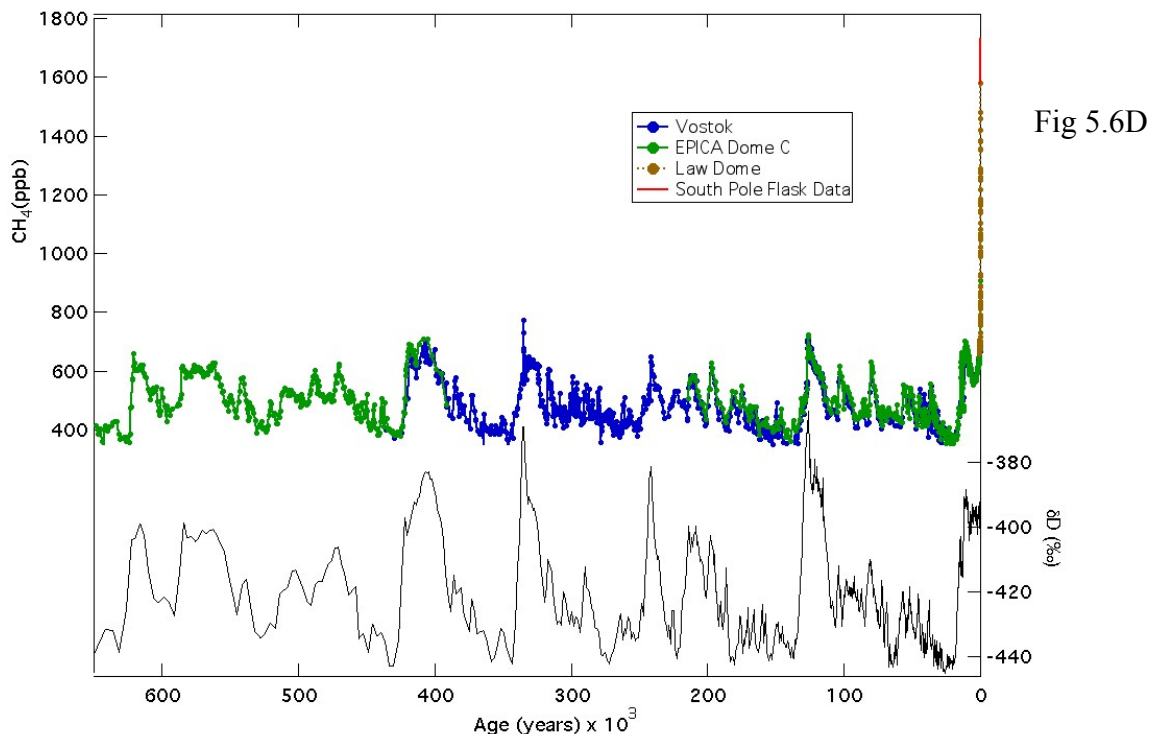


Fig 5.6D

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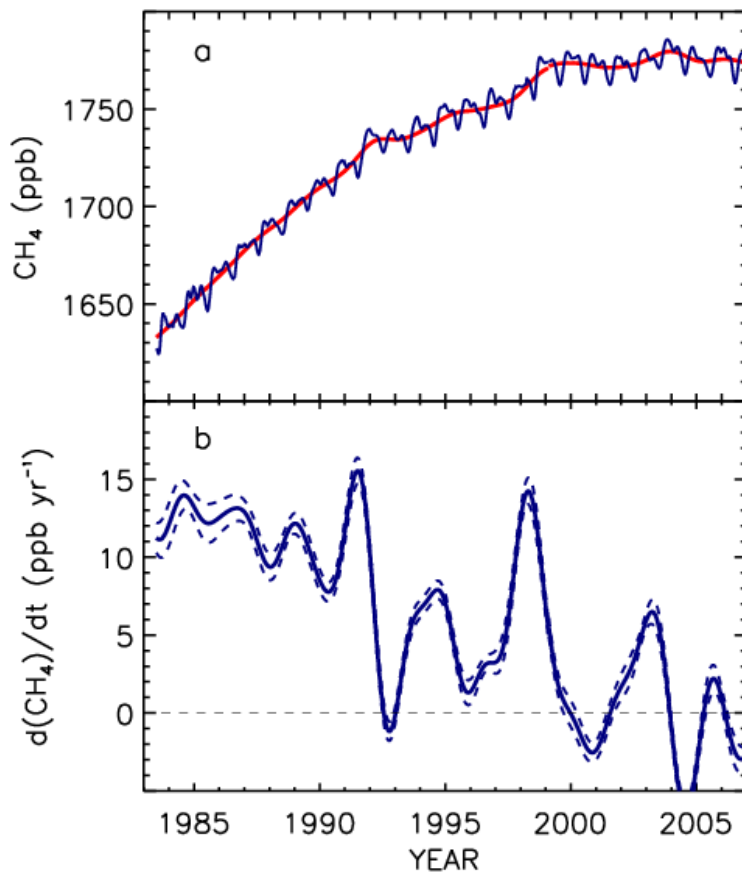
2 **Figure 5.6.** The history of atmospheric methane from ice cores and direct measurements.
 3 A. Zonally averaged representation of seasonal and interannual trends in tropospheric
 4 methane and interhemispheric gradient over the last decade from NOAA Earth System
 5 Research Laboratory (ESRL) data. B. The last 1,000 years from ice cores and direct
 6 measurements (*MacFarling-Meure et al., 2006*) and NOAA ESRL data. C. The last
 7 100,000 years of methane history from the Greenland Ice Sheet Project 2 (GISP2) ice
 8 core in Greenland. $\delta^{18}\text{O}$ is the stable isotope composition of the ice, a proxy for
 9 temperature, with more positive values indicating warmer temperatures. The amplitude of
 10 abrupt methane variations appears positively correlated with Northern Hemisphere
 11 summer insolation (*Brook et al., 1996*). D. A composite of ice core data from the Vostok
 12 and EPICA Dome C ice cores for the last 650,000 years from *Spahni et al. (2005)* with
 13 additional data from *MacFarling-Meure et al. (2006)* and NOAA ESRL. δD is a
 14 temperature proxy. Abbreviations: nmol mol^{-1} , nanomoles per mole; ppb, parts per billion
 15 by mole (same as nanomoles per mole); ‰, per mil.

16 2.1 Direct Observations

17 Early systematic measurements of the global distribution of atmospheric CH_4 established
 18 a rate of increase of $\sim 16 \text{ ppb yr}^{-1}$ in the late 1970s and early 1980s and a strong gradient
 19 between high northern and high southern latitudes of $\sim 150 \text{ ppb}$ (*Blake and Rowland,*
 20 *1988*). By the early 1990s it was clear that the CH_4 growth rate was decreasing (*Steele et*
 21 *al., 1992*) and that, if the CH_4 lifetime were constant, atmospheric CH_4 was approaching
 22 steady state where emissions were approximately constant (*Dlugokencky et al., 1998*).

1 Significant variations are superimposed on this declining growth rate and have been
2 attributed to climate-induced variations in emissions from biomass burning (*van der Werf*
3 *et al.*, 2004) and wetlands (*Walter et al.*, 2001), and changes in the chemical sink after the
4 eruption of Mt. Pinatubo (*Dlugokencky et al.*, 1996). Recent measurements show that the
5 global atmospheric CH₄ burden has been nearly constant since 1999 ([Fig. 5.7](#)). This
6 observation is not well understood, underscoring our lack of understanding of how
7 individual methane sources are changing.

8 Recently published column-averaged CH₄ mixing ratios determined from a satellite
9 sensor greatly enhance the spatial coverage of CH₄ observations (*Frankenberg et al.*,
10 2006). Coverage in the tropics greatly increases measurements there, but coverage in the
11 Arctic remains poor because of the adverse impact of clouds on the retrievals. Use of
12 these satellite data in inverse model studies will reduce uncertainties in emissions
13 estimates, particularly in the tropics.



1

2 **Figure 5.7.** Recent trends in atmospheric methane from global monitoring data (NOAA
 3 ESRL). A, Global average atmospheric methane mixing ratios (blue line) determined
 4 using measurements from the ESRL cooperative air sampling network. The red line
 5 represents the long-term trend. B, Solid line is the instantaneous global average growth
 6 rate for methane; dashed lines are uncertainties (1 standard deviation) calculated with a
 7 Monte Carlo method that assesses uncertainty in the distribution of sampling sites
 8 (*Dlugokencky et al., 2003*).

9 2.2 The Ice Core Record

10 The long term record shows changes in methane on glacial-interglacial time scales of
 11 $\sim 300\text{-}400$ ppb ([Fig. 5.6D](#)), dominated by a strong $\sim 100,000$ year periodicity, with higher
 12 levels during warm interglacial periods and lower levels during ice ages. Periodicity of
 13 $\sim 40,000$ and $20,000$ years is also apparent, associated with Earth's cycles of obliquity and
 14 precession (*Delmotte et al., 2004*). Methane is believed to be a positive feedback to
 15 warming ultimately caused by changes in the Earth's orbital parameters on these time

1 scales. The cyclicity is widely attributed to processes affecting both northern high latitude
2 and tropical wetlands, including growth and decay of Northern Hemisphere ice sheets,
3 and variations in the strength of the monsoon circulation and associated rainfall patterns
4 in Asia, Africa, and South America (*Delmotte et al., 2004; Spahni et al., 2005*).

5 The ice core record also clearly shows another scale of variability, abrupt shifts in
6 methane on millennial time scales that are coincident with abrupt changes in temperature
7 observed in Greenland ice cores ([Fig. 5.6C](#)). These abrupt shifts have been studied in
8 detail in three deep ice cores from Greenland and in several Antarctic ice cores
9 (*Chappellaz et al., 1993a; Brook et al., 1996; Brook et al., 2000; Severinghaus et al.,*
10 *1998; Severinghaus and Brook, 1999; Huber et al., 2006*). Detailed work using nitrogen
11 and argon isotope ratios as gas phase indicators of warming in the ice core record shows
12 clearly that the increase in methane associated with the onset of abrupt warming in
13 Greenland is coincident with, or slightly lags (by a few decades at most), the warming
14 (*Severinghaus et al., 1998; Severinghaus and Brook, 1999; Huber et al., 2006*). Methane
15 closely follows the Greenland ice isotopic record ([Fig. 5.6C](#)), and the amplitude of
16 methane variations associated with abrupt warming in Greenland appears to vary with
17 time. *Brook et al. (1996)* suggested a long-term modulation of the atmospheric methane
18 response to abrupt climate change related to global hydrologic changes on orbital time
19 scales, an issue further quantified by *Flückiger et al. (2004)*.

20 **2.3 What Caused the Abrupt Changes in Methane in the Ice Core Record?**

21 Because the modern natural methane budget is dominated by emissions from wetlands, it
22 is logical to interpret the ice core record in this context. The so-called “wetland
23 hypothesis” postulates that abrupt warming in Greenland is associated with warmer and
24 wetter climate in terrestrial wetland regions, which results in greater emissions of
25 methane from wetlands. Probable sources include tropical wetlands (including regions
26 now below sea level) and high-latitude wetlands in regions that remained ice-free or were
27 south of the major ice sheets. Cave deposits in China, as well as marine and lake
28 sediment records, indicate that enhanced monsoon rainfall in the Northern Hemisphere
29 tropics and subtropics was closely linked to abrupt warming in Greenland (e.g., *Kelly et*
30 *al., 2006; Wang et al., 2004; Yuan et al., 2004; Dykoski et al., 2005; Peterson et al.,*

1 2000). The cave records in particular are important because they are extremely well dated
2 using uranium series isotopic techniques, and high-resolution oxygen isotope records
3 from caves, interpreted as rainfall indicators, convincingly match large parts of the
4 Greenland ice core methane isotopic record.

5 The wetland hypothesis is based on climate-driven changes in methane sources, but it is
6 also possible that changes in methane sinks, primarily the OH radical, played a role in the
7 variations observed in ice cores. Both *Kaplan et al. (2006)* and *Valdes et al. (2005)*
8 proposed that the glacial-interglacial methane change cannot be explained entirely by
9 changes in emissions from wetlands, because in their global climate-biosphere models the
10 difference between Last Glacial Maximum (LGM) and early Holocene methane
11 emissions is not large enough to explain the observed changes in the ice core record. Both
12 studies explain this apparent paradox by invoking increased production of volatile
13 organic carbon (VOC) from the terrestrial biosphere in warmer climates. VOCs compete
14 with methane for reaction with OH, increasing the methane lifetime and the steady-state
15 methane concentration that can be maintained at a given emission rate. Neither of these
16 studies is directly relevant to the abrupt changes in the ice core record, and there are
17 considerable uncertainties in the modeling. Nonetheless, further work on the role of
18 changes in the methane sink on time scales relevant to abrupt methane changes is
19 warranted.

20 The wetland hypothesis has been challenged by authors calling attention to the large
21 marine and terrestrial hydrate reservoirs. The challenge was most extensively developed
22 by *Kennett et al. (2003)*, who postulated that the abrupt shifts in methane in the ice core
23 record were caused by abrupt release of methane from methane hydrates in sea-floor
24 sediments on continental margins. This hypothesis originated from observations of
25 negative carbon isotope excursions in marine sediment records in the Santa Barbara
26 basin, which appear to have coincided with the onset of abrupt warming in Greenland and
27 increases in atmospheric methane in the ice core record. The “clathrate gun hypothesis”
28 postulates that millennial-scale abrupt warming during the last ice age was actually
29 driven by atmospheric methane from hydrate release, and further speculates on a central
30 role for methane in causing late Quaternary climate change (*Kennett et al., 2003*).

1 Some proponents of the clathrate gun hypothesis further maintain that wetlands were not
2 extensive enough during the ice age to be the source of the abrupt variations in methane
3 in the ice core record. For example, *Kennett et al. (2003)* maintain that large
4 accumulations of carbon in wetland ecosystems are a prerequisite for significant
5 methanogenesis and that these established wetlands are exclusively a Holocene
6 phenomenon. Process-based studies of methane emissions from wetlands, on the other
7 hand, emphasize the relationship between annual productivity and emissions (e.g.,
8 *Christensen et al., 1996*). In this view methane production is closely tied to the
9 production of labile carbon (*Schlesinger, 1997*) in the annual productivity cycle (*Kaplan*
10 *et al., 2002; Christensen et al., 1996*). From this perspective it has been postulated that
11 the ice core record reflects changes in rainfall patterns and temperature that could quickly
12 influence the development of anoxic conditions, plant productivity, and methane
13 emissions in regions where the landscape is appropriate for development of water-
14 saturated soil (e.g., *Brook et al., 2000; von Huissteten, 2004*).

15 The hypothesis that there was very little methane emission from wetlands prior to the
16 onset of the Holocene is at odds with models of both wetland distribution and emissions
17 for pre-Holocene times, the latter indicating emissions consistent with, or exceeding,
18 those inferred from the ice core record (e.g., *Valdes et al., 2005; Kaplan et al., 2002;*
19 *2006; Chappellaz et al., 1993b; von Huissteten, 2004*). *von Huissteten (2004)* specifically
20 considered methane emissions during the stadial and interstadial phases of Marine
21 Isotope Stage 3 (~30,000-60,000 years ago), when ice core data indicate that several
22 rapid changes in atmospheric methane occurred ([Fig. 5.6C](#)). *Von Huissteten* describes
23 wetland sedimentary deposits in northern Europe dating from this period and used a
24 process-based model to estimate methane emissions for the cold and warm intervals. The
25 results suggest that emissions from Northern Hemisphere wetlands could be sufficient to
26 cause emissions variations inferred from ice core data. *MacDonald et al. (2006)*
27 presented a compilation of basal peat ages for the circum-Arctic and showed that peat
28 accumulation started early in the deglaciation (at about 16,000 years before present), and
29 therefore emissions of methane from northern hemisphere peat ecosystems very likely
30 played a role in the methane increase at the end of the last ice age. The coincidence of
31 peatland development and the higher Northern Hemisphere summer insolation of late

1 glacial and early Holocene time supports the hypothesis that such wetlands were methane
2 sources at previous times of higher Northern Hemisphere summer insolation (*MacDonald*
3 *et al.*, 2006), for example during insolation and methane peaks in the last ice age or at
4 previous glacial-interglacial transitions (*Brook et al.*, 1996; 2000). In summary, although
5 the sedimentary record of wetlands and the factors controlling methane production in
6 wetlands are imperfectly known, it appears likely that wetlands were important in the pre-
7 Holocene methane budget.

8 The clathrate gun hypothesis is important for understanding the future potential for abrupt
9 changes in methane – concern for the future is warranted if the clathrate reservoir was
10 unstable on the time scale of abrupt late Quaternary climate change. However, as an
11 explanation for late Quaternary methane cycles the clathrate gun hypothesis faces several
12 challenges, elaborated further in [Section 4](#). First, the radiative forcing of the small
13 variations in atmospheric methane burden during the ice age should have been quite
14 small (*Brook et al.*, 2000), although it has been suggested that impacts on stratospheric
15 water vapor may have increased the greenhouse power of these small methane variations
16 (*Kennett et al.*, 2003). Second, the ice core record clearly shows that the abrupt changes
17 in methane lagged the abrupt temperature changes in the Greenland ice core record, albeit
18 by only decades (*Severinghaus et al.*, 1998; *Severinghaus and Brook*, 1999; *Huber et al.*,
19 *2006*; *Grachev et al.*, 2007). These observations imply that methane is a feedback rather
20 than a cause of warming, ruling out one aspect of the clathrate gun hypothesis (hydrates
21 as trigger), but they do not constrain the cause of the abrupt shifts in methane. Third,
22 isotopic studies of ice core methane do not support methane hydrates as a source for
23 abrupt changes in methane (*Sowers*, 2006; *Schaefer et al.*, 2006). The strongest
24 constraints come from hydrogen isotopes (*Sowers*, 2006) and are described further in
25 [Section 4](#).

26 **Box 5.2—The Ice Core Record and Its Fidelity in Capturing Abrupt Events**

27 Around the time of discovery of the abrupt, but small, changes in methane in the late
28 Quaternary ice core records ([Fig. 5.6C](#)) (*Chappellaz et al.*, 1993a some authors
29 suggested that very large releases of methane to the atmosphere might be consistent
30 with the ice core record, given the limits of time resolution of ice core data at that

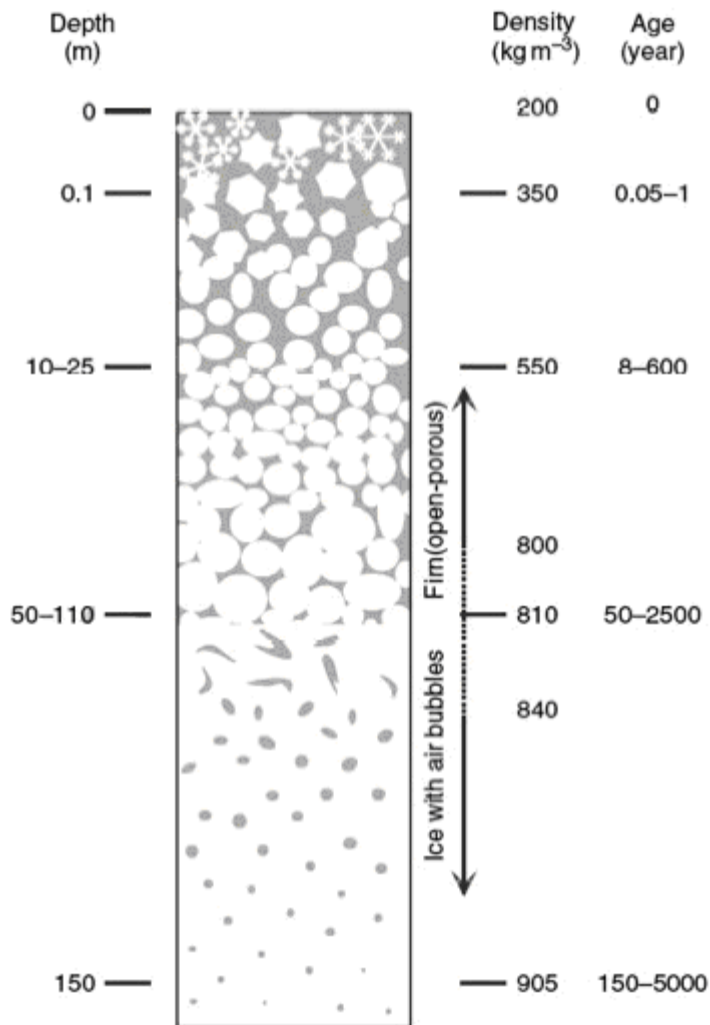
1 time, and the smoothing of atmospheric records due to diffusion in the snowpack
2 (e.g., *Thorpe et al., 1996*). Since that time a large number of abrupt changes in
3 methane in the Greenland ice core record (which extends to ~120,000 years before
4 present) have been sampled in great detail, and no changes greatly exceeding those
5 shown in [Figure 5.6C](#) have been discovered (*Brook et al., 1996; 2000; 2005; Blunier*
6 *and Brook, 2001; Chappellaz et al., 1997; Severinghaus et al. 1998; Severinghaus*
7 *and Brook, 1999; Huber et al. 2006; EPICA Members, 2006*).

8 Could diffusion in the snowpack mask much larger changes? Air is trapped in polar
9 ice at the base of the firn (snowpack) where the weight of the overlying snow
10 transforms snow to ice, and air between the snow grains is trapped in bubbles ([Fig.](#)
11 [5.8](#)). The trapped air is therefore younger than the ice it is trapped in (this offset is
12 referred to as the gas age-ice age difference). It is also mixed by diffusion, such that
13 the air trapped at an individual depth interval is a mixture of air of different ages. In
14 addition, bubbles do not close off all at the same depth, so there is additional mixing
15 of air with different ages due to this variable bubble close-off effect. The overall
16 smoothing depends on the parameters that control firn thickness, densification, and
17 diffusion – primarily temperature and snow accumulation rate.

18 *Spahni et al. (2003)* used the firn model of *Schwander et al. (1993)* to obtain a
19 smoothing function for the Greenland Ice Core Project (GRIP) ice core in Greenland
20 for the late Holocene, which has a Gaussian shape with width at half-height of about
21 20 years. This result is consistent with previous work by *Brook et al. (2000)* for the
22 Greenland Ice Sheet Project (GISP2) ice core. They examined the impact of
23 smoothing on abrupt changes in methane in the Greenland ice core record. (GISP2
24 and GRIP are in similar glaciological environments and their firn characteristics are
25 very similar.) *Brook et al. (2000)* investigated a variety of scenarios for abrupt
26 changes in methane, including those proposed by *Thorpe et al. (1996)*, and compared
27 what the ice core record would record of those events with high-resolution data for
28 several abrupt shifts in methane ([Fig. 5.9](#)).

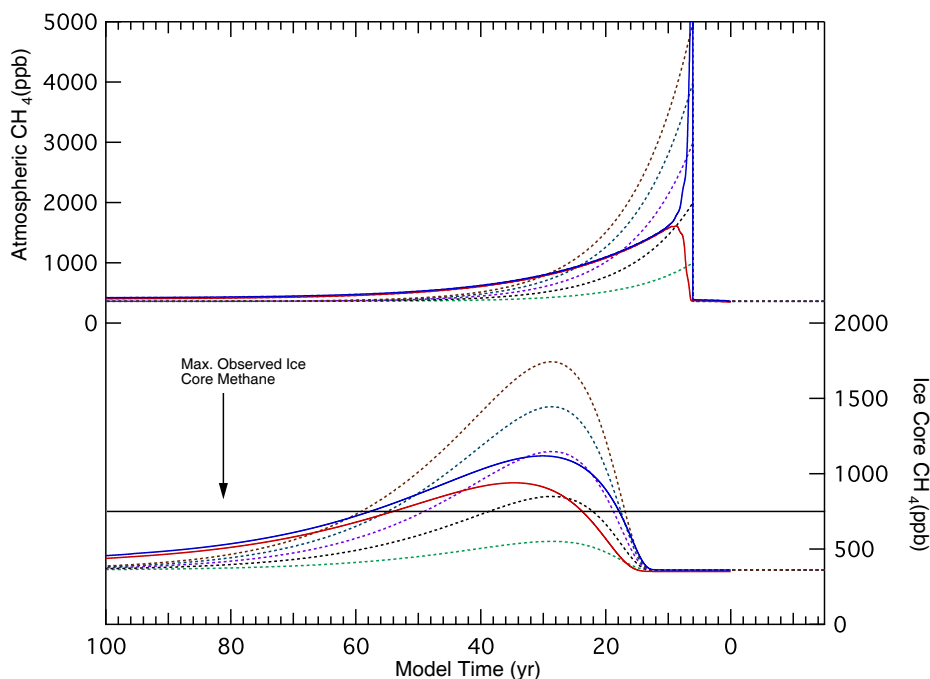
29 Two aspects of the ice core record argue against abrupt, catastrophic releases of
30 methane to the atmosphere as an explanation of the ice core record. First, the abrupt
31 shifts in methane concentration take place on time scales of centuries, whereas

1 essentially instantaneous releases would be recorded in the Greenland ice core record
 2 as more abrupt events (Fig. 5.9). While this observation says nothing about the source
 3 of the methane, it does indicate that the ice core record is not recording an essentially
 4 instantaneous atmospheric change (Brook *et al.*, 2000). Second, the maximum levels
 5 of methane reached in the ice core record are not high enough to indicate extremely
 6 large changes in the atmospheric methane concentration (Fig. 5.9).



7

8 **Figure 5.8.** The firn column of a typical site on a polar ice sheet, from Schwander (2006).
 9 Abbreviations: m, meter; kg m⁻³, kilograms per cubic meter.



1
 2 **Figure 5.9.** Model simulations of smoothing instantaneous release of methane from
 3 clathrates to the atmosphere, and the ice core response to those events. The ice core
 4 response was calculated by convolving the atmospheric histories in the top panel with a
 5 smoothing function appropriate for the GISP2 ice core. The solid lines are the
 6 atmospheric history and smoothed result for the model of a 4,000 teragram release of
 7 methane from *Thorpe et al. (2006)*. The blue solid line represents how an Arctic ice core
 8 would record a release in the Northern Hemisphere, and the red solid line represents how
 9 an Antarctic ice core would record that event (from *Brook et al., 2000*). The dashed lines
 10 represent instantaneous arbitrary increases of atmospheric methane to values of 1,000,
 11 2,000, 3,000, 4,000, or 5,000 ppb (colored dashed lines in top panel) and the ice core
 12 response (bottom panel, same color scheme).

13 **3. Potential Mechanisms for Future Abrupt Changes in Atmospheric Methane**

14 Three categories of mechanism are considered in this chapter as potential causes of
 15 abrupt changes in atmospheric methane in the near future large enough to cause abrupt
 16 climate change. These are outlined briefly in this section, and Sections 4-6 discuss these
 17 mechanisms in more detail.

18 **3.1 Destabilization of Marine Methane Hydrates**

19 This issue is probably the most well known due to extensive research on the occurrence
 20 of methane hydrates in marine sediments, and the large quantities of methane apparently

1 present in this solid phase in continental-margin marine sediments. Destabilization of this
2 solid phase requires mechanisms for warming the deposits and/or reducing pressure on
3 the appropriate time scale, transport of free methane gas to the sediment-water interface,
4 and transport to the atmosphere (see [Box 5.1](#)). There are a number of physical
5 impediments to abrupt release, in addition to the fact that bacterial methanotrophy
6 consumes methane in oxic sediments and the ocean water column. Warming of bottom
7 waters, slope failure, and their interaction are the most commonly discussed mechanisms
8 for abrupt release.

9 **3.2 Destabilization of Permafrost Hydrates**

10 Hydrate deposits at depth in permafrost are known to exist, and although their extent is
11 uncertain, the total amount of methane in permafrost hydrates is very likely much smaller
12 than in marine sediments. Surface warming eventually would increase melting rates of
13 permafrost hydrates. Inundation of some deposits by warmer seawater and lateral
14 invasion of the coastline are also concerns and may be mechanisms for more rapid
15 change.

16 **3.3 Changes in Wetland Extent and Methane Productivity**

17 Although a destabilization of either the marine or terrestrial methane hydrate reservoirs is
18 the most probable pathway for a truly abrupt change in atmospheric methane
19 concentration, the potential exists for a more chronic, but substantial, increase in natural
20 methane emissions in association with projected changes in climate. The most likely
21 region to experience a dramatic change in natural methane emission is the northern high
22 latitudes, where there is increasing evidence for accelerated warming, enhanced
23 precipitation, and widespread permafrost thaw which could lead to an expansion of
24 wetland areas into organic-rich soils that, given the right environmental conditions,
25 would be fertile areas for methane production.

26 In addition, although northern high-latitude wetlands seem particularly sensitive to
27 climate change, the largest natural source of methane to the atmosphere is from tropical
28 wetlands, and methane emissions there may also be sensitive to future changes in

1 temperature and precipitation. Modeling studies addressing this issue are therefore also
2 included in our discussion.

3 **4. Potential for Abrupt Methane Change From Marine Hydrate Sources**

4 **4.1 Impact of Temperature Change on Marine Methane Hydrates**

5 A prominent concern about marine methane hydrates is that warming at the earth surface
6 will ultimately propagate to hydrate deposits and melt them, releasing methane to the
7 ocean-atmosphere system. The likelihood of this type of methane release depends on the
8 propagation of heat through the sea floor, the migration of methane released from hydrate
9 deposits through sediments, and the fate of this methane in the water column,

10 **4.1.1 Propagation of Temperature Change to the Hydrate Stability Zone**

11 The time dependence of changes in the inventory of methane in the hydrate reservoir
12 depends on the time scale of warming and chemical diffusion. There is evidence from
13 paleotracers (*Martin et al., 2005*) and from modeling (*Archer et al., 2004*) that the
14 temperature of the deep sea is sensitive to the climate of the Earth's surface. In general,
15 the time scale for changing the temperature of the ocean increases with depth, reaching a
16 maximum of about 1,000 years for the abyssal ocean. This means that abrupt changes in
17 temperature at the surface ocean would not be transmitted immediately to the deep sea.
18 There are significant regional variations in the ventilation time of the ocean, and in the
19 amount of warming that might be expected in the future. The Arctic is expected to warm
20 particularly strongly, because of the albedo feedback from the melting Arctic ice cap.
21 Temperatures in the North Atlantic appear to be sensitive to changes in ocean circulation
22 such as during rapid climate change during the last ice age (*Dansgaard et al., 1989*).

23 The top of the hydrate stability zone is at 200 to 600 m water depth, depending mainly on
24 the temperature of the water column. Within the sediment column, temperature increases
25 with depth along the geothermal temperature gradient, 30-50°C km⁻¹ (*Harvey and Huang,*
26 *1995*). The shallowest sediments that could contain hydrate only have a thin hydrate
27 stability zone, and the stability zone thickness increases with water depth. A change in
28 the temperature of the deep ocean will act as a change in the upper boundary condition of
29 the sediment temperature profile. Warming of the overlying ocean may not put surface

1 sediments into undersaturation, but the warmer overlying temperature propagates
2 downward until a new profile with the same geothermal temperature gradient can be
3 established. How long this takes is a strong (second order) function of the thickness of the
4 stability zone, but the time scales are in general long. In 1,000 years the temperature
5 signal should have propagated about 180 m in the sediment. In steady state, an increase in
6 ocean temperature will decrease the thickness of the stability zone. *Dickens (2001b)*
7 calculated that the volume of the stability zone ought to decrease by about half with a
8 temperature increase of 5°C.

9 **4.1.2 Impact on Stratigraphic-Type Deposits**

10 After an increase in temperature of the overlying water causes hydrate to melt at the base
11 of the stability zone, the fate of the released methane is difficult to predict. The increase
12 in pore volume and pressure could provoke gas migration through the stability zone or a
13 landslide, or the bubbles could remain enmeshed in the sediment matrix. Hydrate moves
14 down to the base of the stability zone by the accumulation of overlying sediment at the
15 sea floor, so melting of hydrate at the stability zone takes place continuously, not just in
16 association with ocean warming.

17 When hydrate melts, most of the released methane goes into the gas phase to form
18 bubbles, assuming that the porewaters were already saturated in dissolved methane. The
19 fate of the new bubbles could be to remain in place, to migrate, or to diffuse away and
20 react chemically (*Hinrichs et al., 1999; Wakeham et al., 2003*), and it is difficult to
21 predict which will occur. The potential for gas migration through the stability zone is one
22 of the more significant uncertainties in forecasting the ocean hydrate response to
23 anthropogenic warming (*Harvey and Huang, 1995*).

24 In cohesive sediments, bubbles expand by fracturing the sediment matrix, resulting in
25 elongated shapes (*Boudreau et al., 2005*). Bubbles tend to rise because they are less
26 dense than the water they are surrounded by, even at the 200+ atmosphere pressures in
27 sediments of the deep sea. If the pressure in the gas phase exceeds the lithostatic pressure
28 in the sediment, fracture and gas escape can occur (*Flemings et al., 2003*). Modeled and
29 measured (*Dickens et al., 1995*) porewater pressures in the sediment column at Blake

1 Ridge approach lithostatic pressures, indicating that new gas bubbles added to the
2 sediment might be able to escape to the overlying water by this mechanism.

3 There is a differential-pressure mechanism which begins to operate when the bubbles
4 occupy more than about 10% of the volume of the pore spaces (*Hornbach et al., 2004*). If
5 a connected bubble spans a large enough depth range, the pressure of the porewater will
6 be higher at the bottom of the bubble than it is at the top, because of the weight of the
7 porewater over that depth span. The pressure inside the bubble will be more nearly
8 constant over the depth span, because the compressed gas is not as dense as the porewater
9 is. This will result in a pressure gradient at the top and the bottom of the bubble, tending
10 to push the bubble upward. *Hornbach et al. (2004)* postulated that this mechanism might
11 be responsible for allowing methane to escape from the sediment column, and calculated
12 the maximum thickness of an interconnected bubble zone required, before the bubbles
13 would break through the overlying sediment column. In their calculations, and in
14 stratigraphic deposits (they refer to them as “basin settings”), the thickness of the bubble
15 column increases as the stability zone gets thicker. It takes more pressure to break
16 through a thicker stability zone, so a taller column of gas is required. In compressional
17 settings, where the dominant force is directed sideways by tectonics, rather than
18 downward by gravity, the bubble layer is never as thick, reflecting an easier path to
19 methane escape.

20 Multiple lines of evidence indicate that gas can be transported through the hydrate
21 stability zone without freezing into hydrate. Seismic studies at Blake Ridge have
22 observed the presence of bubbles along faults in the sediment matrix (*Taylor et al.,*
23 *2000*). Faults have been correlated with sites of methane gas emission from the sea floor
24 (*Aoki et al., 2000; Zuhlsdorff et al., 2000; Zuhlsdorff and Spiess, 2004*). Seismic studies
25 often show “wipeout zones” where the bubble zone beneath the hydrate stability zone is
26 missing, and all of the layered structure of the sediment column within the stability zone
27 is smoothed out. These are interpreted to be areas where gas has broken through the
28 structure of the sediment to escape to the ocean (*Riedel et al., 2002; Wood et al., 2002;*
29 *Hill et al., 2004*). Bubbles associated with seismic wipeout zones are observed within the
30 depth range which should be within the hydrate stability zone, assuming that the

1 temperature of the sediment column is the steady-state expression of the local average
2 geothermal gradient (*Gorman et al., 2002*). This observation has been explained by
3 assuming that upward migration of the fluid carries with it heat, maintaining a warm
4 channel where gas can be transported through what would otherwise be
5 thermodynamically hostile territory (*Taylor et al., 2000; Wood et al., 2002*).

6 The sediment surface of the world's ocean has holes in it called pockmarks (*Hovland and*
7 *Judd, 1988; Hill et al., 2004*), interpreted to be the result of catastrophic or continuous
8 escape of gas to the ocean. Pockmarks off Norway are accompanied by authigenic
9 carbonate deposits associated with anaerobic oxidation of methane (*Hovland et al.,*
10 *2005*). Pockmarks range in size from meters to kilometers (*Hovland et al., 2005*), with
11 one 700-km² example on the Blake Ridge (*Kvenvolden, 1999*). If the Blake Ridge
12 pockmark is the result of a catastrophic explosion, it might have released less than 1 GtC
13 as methane (assuming a 500-m-thick layer of 4% methane yields 1 GtC). Since each
14 individual pockmark releases a small amount of methane relative to the atmospheric
15 inventory, pockmark methane release could impact climate as part of the ongoing
16 "chronic" methane source to the atmosphere, if the frequency of pockmark eruptions
17 increased. In this sense pockmarks do not represent "catastrophic" methane releases.
18 However, *Kennett et al. (2003)* hypothesized that some apparently inactive pockmark
19 fields may have formed during the last deglaciation and are evidence of active methane
20 discharge at that time.

21 Another mechanism for releasing methane from the sediment column is by submarine
22 landslides. These are a normal, integral part of the ocean sedimentary system (*Hampton*
23 *et al., 1996; Nisbet and Piper, 1998*). Submarine landslides are especially prevalent in
24 river deltas because of the high rate of sediment delivery and because of the presence of
25 submarine canyons. The tendency for slope failure can be amplified if the sediment
26 accumulates more quickly than the excess porosity can be squeezed out. This
27 accumulation can lead to instability of the sediment column, causing periodic Storegga-
28 type landslides off the coast of Norway (see section below on [Storegga Landslide](#)), in the
29 Mediterranean Sea (*Rothwell et al., 2000*), or potentially off the East Coast of the United
30 States (*Dugan and Flemings, 2000*). *Maslin et al. (2004)* find that 70% of the landslides

1 in the North Atlantic over the last 45,000 years (45 kyr) occurred within the time
2 windows of the two meltwater peaks, 15-13 and 11-8 kyr ago. These could have been
3 driven by deglacial sediment loading or warming of the water column triggering hydrate
4 melting.

5 Warming temperatures or sea-level changes may trigger the melting of hydrate deposits,
6 provoking landslides (*Kvenvolden, 1999; Driscoll et al., 2000; Vogt and Jung, 2002*).
7 *Paull et al. (1991)* calculate that landslides can release up to about 1-2 GtC as methane; 1
8 Gt is enough to alter the radiative forcing by about 0.25 watts per square meter (W/m^2).
9 The origin of these estimates is discussed in the section on the [Storegga Landslide](#).

10 **4.1.3 Impact on Structural-Type Hydrate Deposits**

11 In stratigraphic-type hydrate deposits, hydrate concentration is highest near the base of
12 the stability zone, often hundreds of meters below the sea floor. In shallower waters,
13 where the stability zone is thinner, models predict smaller inventories of hydrate.
14 Therefore, most of the hydrates in stratigraphic-type deposits tend to be deep. In contrast
15 with this, in a few parts of the world, transport of presumably gaseous methane through
16 faults or permeable channels results in hydrate deposits that are abundant at shallow
17 depths in the sediment column, closer to the sea floor. These "structural-type" deposits
18 could be vulnerable to temperature-change-driven melting on a faster time scale than the
19 stratigraphic deposits are expected to be.

20 The Gulf of Mexico is basically a leaky oil field (*MacDonald et al., 1994, 2002, 2004;*
21 *Sassen and MacDonald, 1994; Milkov and Sassen, 2000, 2001, 2003; Sassen et al.,*
22 *2001a; Sassen et al., 2003*). Natural oil seeps leave slicks on the sea surface that can be
23 seen from space. Large chunks of methane hydrate have been found on the sea floor in
24 contact with seawater (*Macdonald et al., 1994*). One of the three chunks they saw had
25 vanished when they returned a year later; presumably it had detached and floated away.

26 *Collett and Kuuskraa (1998)* estimate that 500 GtC might reside as hydrates in the Gulf
27 sediments, but *Milkov (2004)* estimates only 5 GtC. In the Community Climate System
28 Model (CCSM) under doubled CO_2 (after 80 years of 1%/year CO_2 increase, from C.
29 Bitz, personal commun., 2007), waters at 500 m depth in the Gulf warm about 0.75°C ,

1 and 0.2° at 1,000 m. In situ temperatures at 500 m are much closer to the hydrate melting
2 temperature, so the relative change in the saturation state is much more significant at 500
3 m than deeper. The equilibrium temperature change in the deep ocean to a large, 5,000
4 GtC fossil fuel release could be 3°C (*Archer et al., 2004*). *Milkov and Sassen (2003)*
5 subjected a two-dimensional model of the hydrate deposits in the Gulf to a 4°C
6 temperature increase and predicted that 2 GtC from hydrate would melt. However, there
7 are no observations to suggest that methane emission rates are currently accelerating.
8 *Sassen et al. (2001b)* find no molecular fractionation of gases in near-surface hydrate
9 deposits that would be indicative of partial dissolution, and suggest that the reservoir may
10 in fact be growing.

11 Other examples of structural deposits include the summit of Hydrate Ridge, off the coast
12 of Oregon, USA (*Torres et al., 2004; Trehu et al., 2004b*) and the Niger Delta (*Brooks et*
13 *al., 2000*). The distribution of hydrate at Hydrate Ridge indicates up-dip flow along sand
14 layers (*Weinberger et al., 2005*). Gas is forced into sandy layers where it accumulates
15 until the gas pressure forces it to vent to the surface (*Trehu et al., 2004a*). *Trehu et al.*
16 (*2004b*) estimate that 30-40% of pore space is occupied by hydrate, while gas fractions
17 are 2-4%. Methane emerges to the sea floor with bubble vents and subsurface flows of 1
18 m s⁻¹, and in regions with bacterial mats and vesicomid clams (*Torres et al., 2002*).
19 Further examples of structural deposits include the Peru Margin (*Pecher et al., 2001*) and
20 Nankai Trough, Japan (*Nouze et al., 2004*).

21 Mud volcanoes are produced by focused upward fluid flow into the ocean and are
22 sometimes associated with hydrate and petroleum deposits. Mud volcanoes often trap
23 methane in hydrate deposits that encircle the channels of fluid flow (*Milkov, 2000;*
24 *Milkov et al., 2004*). The fluid flow channels associated with mud volcanoes are ringed
25 with the seismic images of hydrate deposits, with authigenic carbonates, and with
26 pockmarks (*Dimitrov and Woodside, 2003*) indicative of anoxic methane oxidation.
27 *Milkov (2000)* estimates that mud volcanoes contain at most 0.5 GtC of methane in
28 hydrate; about 100 times his estimate of the annual supply.

1 **4.1.4 Fate of Methane Released as Bubbles**

2 Methane released from sediments in the ocean may reach the atmosphere directly, or it
3 may dissolve in the ocean. Bubbles are not generally a very efficient means of
4 transporting methane through the ocean to the atmosphere. *Rehder et al. (2002)* compared
5 the dissolution kinetics of methane and argon and found enhanced lifetime of methane
6 bubbles below the saturation depth in the ocean, about 500 m, because a hydrate film on
7 the surface of the methane bubbles inhibited gas exchange. Bubbles dissolve more slowly
8 from petroleum seeps, where oily films on the surface of the bubble inhibit gas exchange,
9 also changing the shapes of the bubbles (*Leifer and MacDonald, 2003*). On a larger scale,
10 however, *Leifer et al. (2000)* diagnosed that the rate of bubble dissolution is limited by
11 turbulent transport of methane-rich water out of the bubble stream into the open water
12 column. The magnitude of the surface dissolution inhibition seems small; in the *Rehder et*
13 *al. (2002)* study, a 2-cm bubble dissolves in 30 m above the stability zone, and only 110
14 m below the stability zone. Acoustic imaging of the bubble plume from Hydrate Ridge
15 showed bubbles surviving from 600-700 m water depth where they were released to just
16 above the stability zone at 400 m (*Heeschen et al., 2003*). One could imagine hydrate-
17 film dissolution inhibition as a mechanism to concentrate the release of methane into the
18 upper water column, but not really as a mechanism to get methane through the ocean
19 directly to the atmosphere.

20 Methane can reach the atmosphere if the methane bubbles are released in waters that are
21 only a few tens of meters deep, as in the case of melting the ice complex in Siberia
22 (*Shakhova et al., 2005; Washburn et al., 2005; Xu et al., 2001*) or during time periods of
23 lower sea level (*Luyendyk et al., 2005*). If the rate of methane release is large enough, the
24 rising column of seawater in contact with the bubbles may saturate with methane, or the
25 bubbles can be larger, potentially increasing the escape efficiency to the atmosphere.

26 **4.1.5 Fate of Methane Hydrate in the Water Column**

27 Pure methane hydrate is buoyant in seawater, so floating hydrate is another potential way
28 to deliver methane from the sediment to the atmosphere (*Brewer et al., 2002*). In sandy
29 sediment, the hydrate tends to fill the existing pore structure of the sediment, potentially
30 entraining sufficient sediment to prevent the hydrate/sediment mixture from floating,

1 while in fine-grained sediments, bubble and hydrate grow by fracturing the cohesion of
2 the sediment, resulting in irregular blobs of bubbles (*Gardiner et al., 2003; Boudreau et*
3 *al., 2005*) or pure hydrate. *Brewer et al. (2002)* and *Paull et al. (2003)* stirred surface
4 sediments from Hydrate Ridge using the mechanical arm of a submersible remotely
5 operated vehicle and found that hydrate did manage to shed its sediment load enough to
6 float. Hydrate pieces of 0.1 m survived a 750-m ascent through the water column. *Paull*
7 *et al. (2003)* described a scenario for a submarine landslide in which the hydrates would
8 gradually make their way free of the turbidity current comprised of the sediment and
9 seawater slurry.

10 **4.1.6 Fate of Dissolved Methane in the Water Column**

11 Methane is unstable to bacterial oxidation in oxic seawater. *Rehder et al. (1999)* inferred
12 a methane oxidation lifetime in the high-latitude North Atlantic of 50 years. Methane
13 oxidation is faster in the deep ocean near a particular methane source, where its
14 concentration is higher (turnover time 1.5 years), than in the surface ocean (turnover time
15 of decades) (*Valentine et al., 2001*). Water-column concentration and isotopic
16 measurements indicate complete water-column oxidation of the released methane at
17 Hydrate Ridge (*Grant and Whiticar, 2002; Heeschen et al., 2005*).

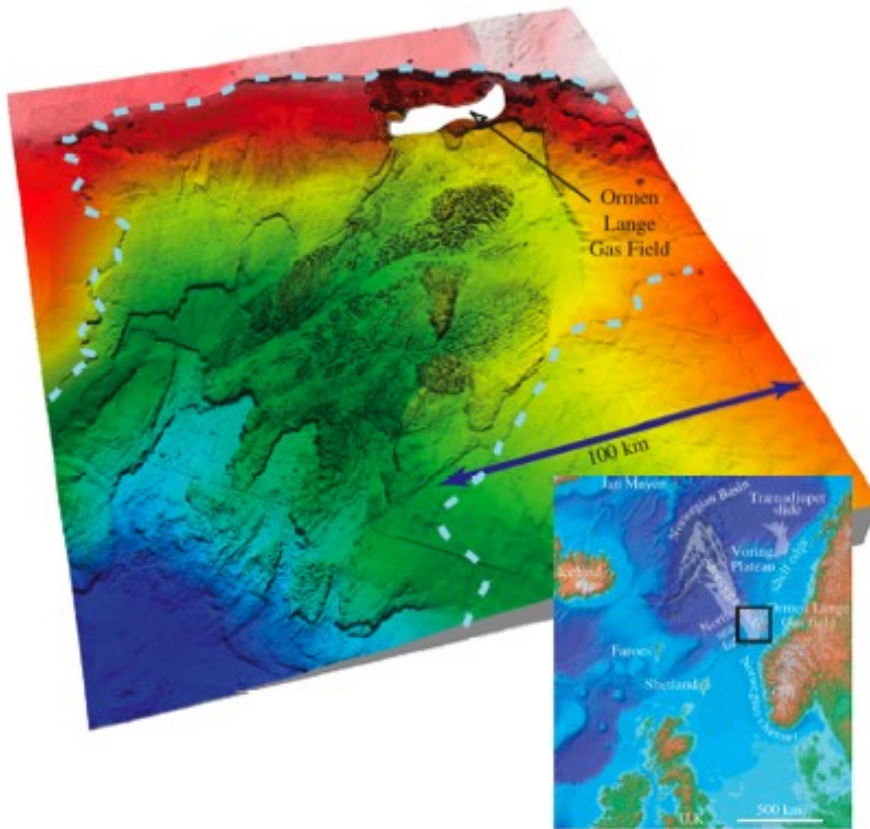
18 An oxidation lifetime of 50 years leaves plenty of time for transport of methane gas to the
19 atmosphere. Typical gas-exchange time scales for gas evasion from the surface ocean
20 would be about 3-5 m per day. A surface mixed layer 100 m deep would approach
21 equilibrium (degas) in about a month. Even a 1,000-m-thick winter mixed layer would
22 degas about 30% during a 3-month winter window. The ventilation time of subsurface
23 waters depends on the depth and the fluid trajectories in the water (*Luyten et al., 1983*),
24 but 50 years is enough time that a significant fraction of the dissolved methane from
25 bubbles might reach the atmosphere before it is oxidized.

26 **4.2 Geologic Data Relevant to Past Hydrate Release**

27 **4.2.1 The Storegga Landslide**

28 One of the largest exposed submarine landslides in the ocean is the Storegga Slide in the
29 Norwegian continental margin (*Mienert et al., 2000, 2005; Bryn et al., 2005*). The slide

1 excavated on average the top 250 m of sediment over a swath hundreds of kilometers
2 wide, stretching halfway from Norway to Greenland ([Fig. 5.10](#)). There have been
3 comparable slides on the Norwegian margin every approximately 100 kyr, roughly
4 synchronous with the glacial cycles (*Solheim et al., 2005*). The last one, Storegga proper,
5 occurred about 8,150 years ago, after deglaciation. It generated a tsunami in what is now
6 the United Kingdom (*D'Hondt et al., 2004; Smith et al., 2004*). The Storegga slide area
7 contains methane hydrate deposits as indicated by a seismic bottom simulating reflector
8 (BSR) (*Bunz and Mienert, 2004; Mienert et al., 2005; Zillmer et al., 2005a, b*)
9 corresponding to the base of the hydrate stability zone 200-300 m, and pockmarks
10 (*Hovland et al., 2005*) indicating gas expulsion from the sediment.



11

12 **Figure 5.10.** Image and map of the Storegga Landslide from *Masson et al. (2006)*. The
13 slide excavated on average the top 250 m of sediment over a swath hundreds of
14 kilometers wide. Colors indicate water depth, with yellow-orange indicating shallow
15 water, and green-blue indicating deeper water.

1 The slide was presumably triggered by an earthquake, but the sediment column must
2 have been destabilized by either or both of two mechanisms. One is the rapid
3 accumulation of glacial sediment shed by the Fennoscandian ice sheet (*Bryn et al., 2005*).
4 As explained above, rapid sediment loading traps porewater in the sediment column
5 faster than it can be expelled by the increasing sediment load. At some point, the
6 sediment column floats in its own porewater (*Dugan and Flemings, 2000*). This
7 mechanism has the capacity to explain why the Norwegian continental margin, of all
8 places in the world, should have landslides synchronous with climate change.

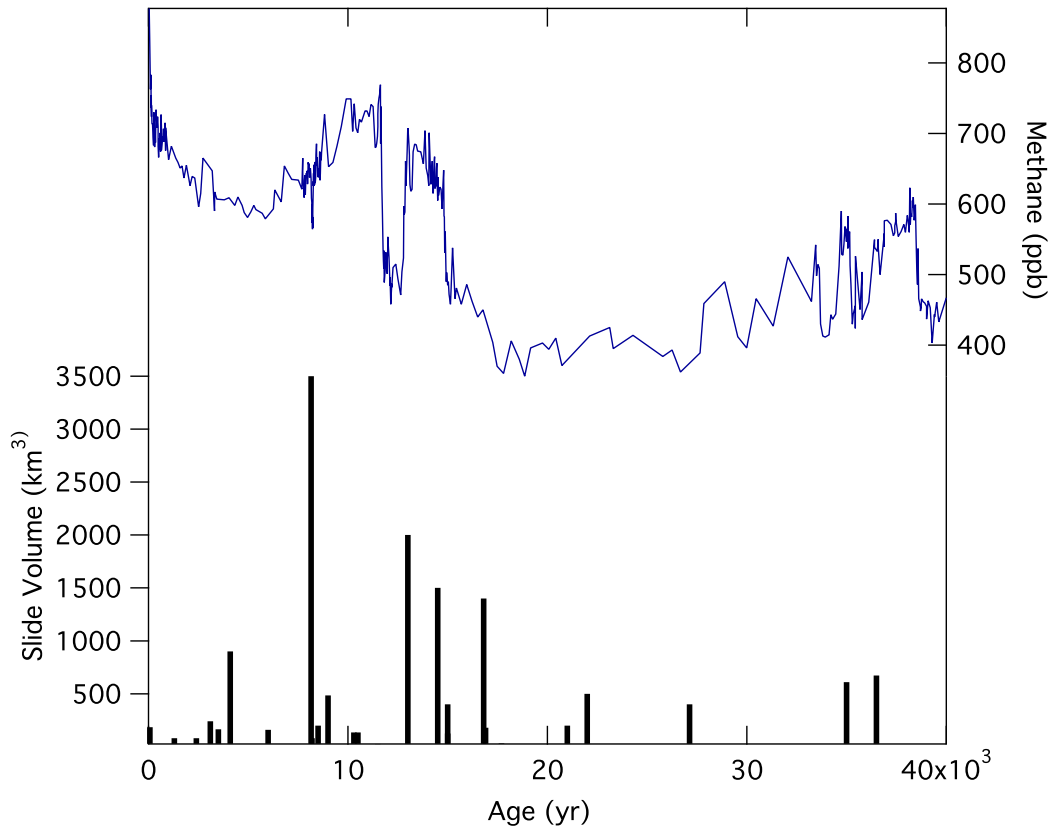
9 The other possibility is the dissociation of methane hydrate deposits by rising ocean
10 temperatures. Rising sea level is also a player in this story, but a smaller one. Rising sea
11 level tends to increase the thickness of the stability zone by increasing the pressure. A
12 model of the stability zone shows this effect dominating deeper in the water column
13 (*Vogt and Jung, 2002*); the stability zone is shown increasing by about 10 m for
14 sediments in water depth below about 750 m. Shallower sediments are impacted more by
15 long-term temperature changes, reconstructions of which show warming of 5-6°C over a
16 thousand years or so, 11-12 kyr ago. The landslide occurred 2-3 kyr after the warming
17 (*Mienert et al., 2005*). The slide started at a few hundred meters water depth, just off the
18 continental slope, just where *Mienert et al. (2005)* calculate the maximum change in
19 HSZ. *Sultan et al. (2004)* predict that warming in the near-surface sediment would
20 provoke hydrate to dissolve by increasing the saturation methane concentration. This
21 form of dissolution differs from heat-driven direct melting, however, in that it produces
22 dissolved methane, rather than methane bubbles. *Sultan et al. (2004)* assert that melting
23 to produce dissolved methane increases the volume, although laboratory analyses of
24 volume changes upon this form of melting are equivocal. In any case, the volume
25 changes are much smaller than for thermal melting that produces bubbles.

26 The amount of methane released by the slide can be estimated from the volume of the
27 slide and the potential hydrate content. Hydrate just outside the slide area has been
28 estimated by seismic methods to fill as much as 10% of the porewater volume, in a layer
29 about 50 m thick near the bottom of the stability zone (*Bunz and Mienert, 2004*). If these

1 results were typical of the entire 10^4 km^2 area of the slide, the slide could have released 1-
2 2 GtC of methane in hydrate (*Paull et al., 1991*).

3 If 1 GtC CH_4 reached the atmosphere all at once, it would raise the atmospheric
4 concentration from today's value of $\sim 1,700$ ppb to ~ 2200 ppb, trapping about 0.25
5 additional W/m^2 of greenhouse heat, or more, considering indirect feedbacks. The
6 methane radiative forcing would subside over a time scale of a decade or so, as the pulse
7 of released methane was oxidized to CO_2 , and the atmospheric methane concentration
8 relaxed toward the long-term steady-state value. The radiative impact of the Storegga
9 Landslide would then be somewhat smaller in magnitude but opposite in sign to the
10 eruption of a large volcano, such as the Mt. Pinatubo eruption (-2 W/m^2), but it would
11 last for longer (10 years for methane and 2 years for a volcano).

12 It is tantalizing to wonder if there could be any connection between the Storegga
13 Landslide and the 8.2 kyr climate event (*Alley and Agustsdottir, 2005*), which may have
14 been triggered by freshwater release to the North Atlantic. However, ice cores
15 record a 75 ppb drop in methane concentration during the 8.2 kyr event (*Kobashi et al.,*
16 *2007*), not a rise. A slowdown of convection in the North Atlantic would have cooled the
17 overlying waters. *Maslin et al. (2004)* suggested that an apparent correlation between the
18 ages of submarine landslides in the North Atlantic region and methane variations during
19 the deglaciation supported the hypothesis that clathrate release by this mechanism
20 influenced atmospheric methane. The lack of response for Storegga, by far the largest
21 landslide known, and a relatively weak association of other large slides with increased
22 methane levels ([Fig. 5.11](#)) suggest that it is unlikely that submarine landslides caused the
23 atmospheric methane variations during this time period.



1

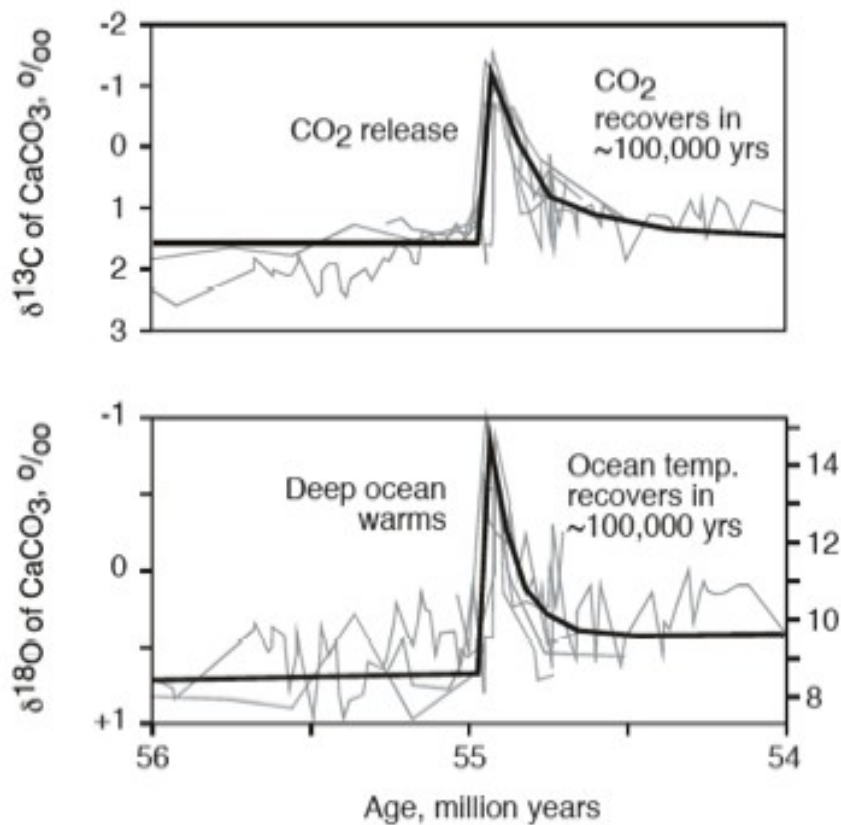
2 **Figure 5.11.** Timing of submarine landslides in the North Atlantic region and pre-
 3 industrial ice core methane variations. Landslide data from *Maslin et al. (2004)*. Methane
 4 data from *Brook et al. (2000)* and *Kobashi et al. (2007)*. Abbreviations: km³, cubic
 5 kilometers; yr, year; ppb, parts per billion.

6 Much of our knowledge of the Storegga Landslide is due to research sponsored by the
 7 Norwegian oil industry, which is interested in tapping the Ormen Lange gas field within
 8 the headlands of the Storegga slide but is concerned about the geophysical hazard of gas
 9 extraction (*Bryn et al., 2005*). Estimates of potential methane emission from the Storegga
 10 slide range from 1 to 5 GtC, which is significant but not apocalyptic. As far as can be
 11 determined, the Storegga Landslide had no impact on climate.

12 **4.2.2 The Paleocene-Eocene Thermal Maximum**

13 About 55 million years ago, the $\delta^{13}\text{C}$ signature of carbon in the ocean and on land
 14 decreased by 2.5-5 per mil (‰) on a time scale of less than 10 kyr, then recovered in
 15 parallel on a time scale of ~120-220 kyr (*Kennett and Stott, 1991; Zachos et al., 2001*).
 16 Associated with this event, commonly called the Paleocene-Eocene Thermal Maximum

1 (PETM), the $\delta^{18}\text{O}$ of CaCO_3 from intermediate depths in the ocean decreased by 2-3‰,
2 indicative of a warming of about 5°C (Fig. 5.12). The timing of the spikes is to a large
3 extent synchronous. Planktonic foraminifera and terrestrial carbon records show a $\delta^{13}\text{C}$
4 perturbation a bit earlier than benthic foraminifera do, suggesting that the carbon spike
5 invaded the deep ocean from the atmosphere (Thomas *et al.*, 2002). Similar events, also
6 associated with transient warmings, although less well documented, have been described
7 from other times in geologic history (Hesselbo *et al.*, 2000; Jenkyns, 2003). The PETM is
8 significant to the present day because it is an analog to the potential fossil fuel carbon
9 release if we burn all the coal reserves.



10

11 **Figure 5.12.** Carbon (top) and oxygen (bottom) isotope record for benthic foraminifera
12 from sites in the south Atlantic and western Pacific Oceans for the Paleocene-Eocene
13 Thermal Maximum (PETM), from Zachos *et al.* (2001), modified by Archer (2007). ‰,
14 per mil.

1 The change in isotopic composition of the carbon in the ocean is attributed to the release
2 of some amount of isotopically light carbon to the atmosphere. However, it is not clear
3 where the carbon came from, or how much of it there was. The magnitude of the carbon
4 shift depends on where it was recorded. The surface change recorded in CaCO₃ in soils
5 (*Koch et al., 1992*) and in some planktonic foraminifera (*Thomas et al., 2002*) is twice as
6 large a change as is reported for the deep sea. Land records may be affected by changes
7 in plant fractionation, driven by changing hydrological cycle (*Bowen et al., 2004*). Ocean
8 records may be affected by CaCO₃ dissolution (*Zachos et al., 2005*) resulting in
9 diagenetic imprints on the remaining CaCO₃, a necessity to use multiple species, or
10 simple inability to find CaCO₃ at all.

11 We can estimate the change in the carbon inventory of the ocean by specifying an
12 atmospheric partial pressure of CO₂ value (pCO₂), a mean ocean temperature, and
13 insisting on equilibrium with CaCO₃ (*Zeebe and Westbroek, 2003*). The ocean was
14 warmer, prior to the PETM event, than it is today. Atmospheric pCO₂ was probably at
15 least 560 ppm at this time (*Huber et al., 2002*). The present-day inventory of CO₂ in the
16 ocean is about 40,000 GtC. According to simple thermodynamics, neglecting changes in
17 the biological pump or circulation of the ocean, the geological steady-state inventory for
18 late Paleocene, pre-PETM time could have been on the order of 50,000 GtC.

19 The lighter the isotopic value of the source, the smaller the amount of carbon that must be
20 released to explain the isotopic shift ([Fig. 5.12](#), top). Candidate sources include methane,
21 which can range in its $\delta^{13}\text{C}$ isotopic composition from -30 to -110‰ . If the ocean $\delta^{13}\text{C}$
22 value is taken at face value, and the source was methane at -60‰ , then 2,000 GtC would
23 be required to explain the isotopic anomaly. If the source were thermogenic methane or
24 organic carbon at $\delta^{13}\text{C}$ of about -25‰ , then 10,000 GtC would be required.

25 *Buffett and Archer (2004)* find that the steady-state hydrate reservoir size in the ocean is
26 extremely sensitive to the temperature of the deep sea. At the temperature of Paleocene
27 time but with everything else as in the present-day ocean, they predict less than a
28 thousand GtC of methane in steady state. As the ocean temperature decreases, the
29 stability zone gets thinner and covers less area. Their model was able to fit 6,000 GtC in

1 the Arctic Ocean, however, using 6°C temperatures from CCSM (*Huber et al., 2002*)
2 (which may be too cold) and assuming that the basin had been anoxic (*Sluijs et al., 2006*).

3 Marine organic matter has an isotopic composition of -20‰, and would require 6,000
4 GtC to explain the isotopic anomaly. *Svensen et al. (2004)* proposed that lava intrusions
5 into organic-rich sediments could have caused the isotopic shift. They cite evidence that
6 the isotopic composition of methane produced from magma intrusion should be -35 to -
7 50‰, requiring therefore 2,500-3,500 GtC to explain the isotope anomaly in the deep
8 ocean. If CO₂ were also released, from metamorphism of CaCO₃, the average isotopic
9 composition of the carbon spike would be lower, and the mass of carbon greater. *Storey*
10 *et al. (2007)* showed that the opening of the North Atlantic Ocean corresponds in time
11 with the PETM. However, volcanic activity continued for hundreds of thousands of
12 years, leaving still unexplained the reason for the fast (<10,000 years) carbon isotope
13 excursion.

14 A comet impact might have played a role in the PETM, and while the isotopic
15 composition of comets is not well constrained, carbon in cometary dust tends to be about
16 -45‰ (*Kent et al., 2003*). *Kent et al. (2003)* calculate that an 11 km comet containing 20-
17 25% organic matter, a rather large icy tarball, could deliver 200 GtC, enough to decrease
18 the δ¹³C of the atmosphere and upper ocean by 0.4‰. It is unlikely that a comet could
19 deliver thousands of GtC, however. An impact strike to a carbonate platform or an
20 organic-rich sediment of some sort could release carbon, but it would take a very large
21 crater to release thousands of gigatons of carbon.

22 Volcanic carbon has an isotopic composition of -7‰, requiring a huge carbon release of
23 ~20,000 GtC to explain the PETM. Excess carbon emissions have been attributed to
24 superplume cycles in the mantle and flood basalt volcanic activity (*Larson, 1991*).
25 However, these events tend to take millions of years to play out (*Dickens et al., 1995*).
26 *Schmitz et al. (2004)* and *Bralower et al. (1997)* find evidence of increased volcanic
27 activity during the PETM interval but view the activity as rearranging ocean circulation,
28 triggering methane release, rather than being a major primary source of carbon itself,
29 presumably because the potential volcanic carbon source is too slow.

1 Acidification of the ocean by invasion of CO₂ drove a shoaling of the depth of CaCO₃
2 preservation in the Atlantic (*Zachos et al., 2005*) although, curiously, the signal is much
3 smaller in the Pacific (*Zachos et al., 2003*). The magnitude of the carbonate
4 compensation depth (CCD) shift in the Atlantic would suggest a large carbon addition, on
5 the order of 5,000 GtC or more (*Archer et al., 1997*).

6 A large carbon release is also supported by the warming inferred from the δ¹⁸O spike.
7 The benthic δ¹⁸O record is clearly interpretable as a temperature change, at a depth of
8 several kilometers in the ocean, from about 8° to about 14°C, in a few thousand years.
9 Warming is also implied by Mg/Ca ratios in CaCO₃ (*Zachos et al., 2003*). The
10 temperature can be altered by both CH₄ and CO₂. *Schmidt and Shindell (2003)* calculated
11 that the steady-state atmospheric CH₄ concentration during the period of excess emission
12 (ranging from 500-20,000 years) would be enough to explain the temperature change.
13 However, the atmospheric-methane concentration anomaly would decay away a few
14 decades after the excess emission ceased. At this point the temperature anomaly would
15 die away also. Hence, as soon as the carbon isotopic composition stopped plunging
16 negatively, the oxygen isotopic composition should recover as the ocean cools. The
17 carbon isotopic composition meanwhile should remain light for hundreds of thousands of
18 years (*Kump and Arthur, 1999*) until the carbon reservoir isotopic composition
19 reapproached a steady-state value. The record shows instead that the oxygen and carbon
20 isotopic anomalies recovered in parallel ([Fig. 5.12](#)). This suggests that CO₂ is the more
21 likely greenhouse warmer rather than CH₄. It could be that the time scale for the pCO₂ to
22 reach steady state might be different than the time scale for the isotopes to equilibrate,
23 analogous to the equilibration of the surface ocean by gas exchange: isotopes take longer.
24 However, in the *Kump and Arthur (1999)* model results, pCO₂ seems to take longer to
25 equilibrate than δ¹³C. The first-order result is that the CO₂ and δ¹³C time scales are much
26 more similar than the CH₄ and δ¹³C time scales would be.

27 A warming of 5°C would require somewhere between one and two doublings of the
28 atmospheric CO₂ concentration, if the climate sensitivity is in the range of IPCC
29 predictions of 2.5 – 4.5°C. Beginning from 600 ppm, we would increase the pCO₂ of the
30 atmosphere to somewhere in the range of 1,200 – 2,400 ppm. The amount of carbon

1 required to achieve this value for hundreds of thousands of years (after equilibration with
2 the ocean and with the CaCO₃ cycle) would be of order 20,000 GtC. This would imply a
3 mean isotopic composition of the spike of mantle isotopic composition, not isotopically
4 light methane. The amount of carbon required to explain the observed $\delta^{18}\text{O}$ would be
5 higher if the initial atmospheric pCO₂ were higher than the assumed 600 ppm. The only
6 way that a biogenic methane source could explain the warming is if the climate
7 sensitivity were much higher in the Paleocene than it seems to be today, which seems
8 unlikely because the ice albedo feedback amplifies the climate sensitivity today (*Pagani*
9 *et al.*, 2006).

10 The bottom line conclusion about the source of the carbon isotopic excursion is that it is
11 still not clear. There is no clear evidence in favor of a small, very isotopically depleted
12 source of carbon. Mechanistically, it is easier to explain a small release than a large one,
13 and this is why methane has been a popular culprit for explaining the $\delta^{13}\text{C}$ shift.
14 Radiative considerations argue for a larger carbon emission, corresponding to a less
15 fractionated source than pure biogenic methane. Thermogenic methane might do, such as
16 the release of somewhat more thermogenic methane than in Gulf of Mexico sediments, if
17 there were a thermogenic deposit that large. Perhaps it was some combination of sources,
18 an initial less-fractionated source such as marine organic matter or a comet, followed by
19 hydrate release.

20 The PETM is significant to the present day because it is a close analog to the potential
21 fossil fuel carbon release if we burn all the coal reserves. There are about 5,000 GtC in
22 coal, while oil and traditional natural gas deposits are hundreds of Gt each (*Rogner*,
23 *1997*). The recovery time scale from the PETM (140 kyr) is comparable to the model
24 predictions, based on the mechanism of the silicate weathering thermostat (400 kyr time
25 scale, *Berner et al.*, 1983).

26 The magnitude of the PETM warming presents an important and currently unanswered
27 problem. A 5,000 GtC fossil fuel release will warm the deep ocean by perhaps 2-4°C,
28 based on paleoclimate records and model results (*Martin et al.*, 2005). The warming
29 during the PETM was 5°C, and this was from an atmospheric CO₂ concentration higher

1 than today (at least 600 ppm), so that a further spike of only 2,000 GtC (based on
2 methane isotopic composition) would have only a tiny radiative impact, not enough to
3 warm the Earth by 5°C. One possibility is that our estimates for the climate sensitivity are
4 too low by a factor of 2 or more. However, as mentioned above, one might expect a
5 decreased climate sensitivity for an ice-free world rather than for the ice-age climate of
6 today.

7 Another possibility is that the carbon release was larger than 2,000 GtC. Perhaps the
8 global average $\delta^{13}\text{C}$ shift was as large as recorded in soils (*Koch et al., 1992*) and some
9 planktonic foraminifera (*Thomas et al., 2002*). The source could have been thermogenic
10 methane, or maybe it was not methane at all but CO_2 , derived from some organic pool
11 such as sedimentary organic carbon (*Svensen et al., 2004*). At present, the PETM serves
12 as a cautionary tale about the long duration of a release of new CO_2 to the atmosphere
13 (*Archer, 2005*). However, our current understanding of the processes responsible for the
14 $\delta^{13}\text{C}$ spike is not strong enough to provide any new constraint to the stability of the
15 methane hydrate reservoir in the immediate future.

16 **4.2.3 Santa Barbara Basin and the Clathrate Gun Hypothesis**

17 *Nisbet (2002)* and *Kennett et al. (2003)* argue that methane from hydrates is responsible
18 for the deglacial rise in the Greenland methane record between 20,000 and 10,000 years
19 ago, and for abrupt changes in methane at other times ([Fig. 5.6C](#)). *Kennett et al. (2000)*
20 found episodic negative $\delta^{13}\text{C}$ excursions in benthic foraminifera in the Santa Barbara
21 Basin, which they interpret as reflecting release of hydrate methane during warm climate
22 intervals. Biomarkers for methanotrophy are found in greater abundance and indicate
23 greater rates of reaction during warm intervals in the Santa Barbara Basin (*Hinrichs et*
24 *al., 2003*) and in the Japanese coastal margin (*Uchida et al., 2004*). *Cannariato and Stott*
25 *(2004)*, however, argued that these results could have arisen from contamination or
26 subsequent diagenetic overprints. *Hill et al. (2006)* measured the abundance of tar in
27 Santa Barbara basin sediments, argued that tar abundance was proportional to methane
28 emissions, and described increases in tar abundance and inferred destabilization of
29 methane hydrates associated with warming during the last glacial-interglacial transition.

1 As discussed in [Section 1](#), there are several arguments against the hypothesis of a
2 clathrate role in controlling atmospheric methane during the last glacial period. Perhaps
3 the most powerful so far is that the isotopic ratio of deuterium to hydrogen (D/H) in ice
4 core methane for several abrupt transitions in methane concentration indicates a
5 freshwater source, rather than a marine source, apparently ruling out much of a role for
6 marine hydrate methane release (*Sowers, 2006*). However, the D/H ratio has not yet been
7 measured for the entire ice core record. The timing of the deglacial methane rise was also
8 more easily explained by wetland emissions than by catastrophic methane release (*Brook*
9 *et al., 2000*). The interhemispheric gradient of methane tells us that the deglacial increase
10 in atmospheric methane arose in part from high northern latitudes (*Dallenbach et al.,*
11 *2000*), although more work is needed to verify this conclusion because constraining the
12 gradient is analytically difficult. The deglacial methane rise could therefore be attributed
13 at least in part to methanogenesis from decomposition of thawing organic matter or from
14 high-latitude wetlands. Regardless of the source of the methane, the climate forcing from
15 the observed methane record ([Fig. 5.6C](#) and [D](#)) is too weak to argue for a dominant role
16 for methane in the glacial cycles (*Brook et al., 2000*).

17 **4.3 Review of Model Results Addressing Past and Future Methane Hydrate**

18 **Destabilization**

19 **4.3.1 Quantity of Methane Potentially Released**

20 Probably the most detailed analysis to date of the potential for methane release from
21 hydrates on a century time scale is the study of *Harvey and Huang (1995)*. Their study
22 calculated the inventory of hydrate and the potential change in that inventory with an
23 ocean warming. They treated as a parameter the fraction of methane in bubbles that could
24 escape the sediment column to reach the ocean, and evaluated the sensitivity of the
25 potential methane release to that escaped fraction. Our picture of methane release
26 mechanisms has been refined since 1995, although it remains difficult to predict the fate
27 of methane from melted hydrates. *Harvey and Huang (1995)* did not treat the invasion of
28 heat into the ocean or into the sediment column. Their conclusion was that the radiative
29 impact from hydrate methane will be much smaller than that of CO₂, or even between
30 different scenarios for CO₂ release. The calculation should be redone, but it is unlikely
31 that an updated calculation would change the bottom-line conclusion.

1 **4.3.2 Climate Impact of Potential Release**

2 *Schmidt and Shindell (2003)* showed that the chronic release of methane from a large
3 hydrate reservoir over thousands of years can have a significant impact on global climate.
4 The accumulating CO₂ from the oxidation of the methane also has a significant climate
5 impact. New CO₂ from methane oxidation accumulates in the atmosphere / ocean /
6 terrestrial biosphere carbon pool and persists to affect climate for hundreds of thousands
7 of years (*Archer, 2005*). If a pool of methane is released over a time scale of thousands of
8 years, the climate impact from the accumulating CO₂ concentration may exceed that from
9 the steady-state increase in the methane concentration (*Harvey and Huang, 1995*;
10 *Dickens, 2001a*; *Schmidt and Shindell, 2003*; *Archer and Buffett, 2005*). After the
11 emission stops, methane drops quickly to a lower steady state, while the CO₂ persists.

12 If hydrates melt in the ocean, much of the methane would probably be oxidized in the
13 ocean rather than reaching the atmosphere directly as methane. This reduces the century-
14 time scale climate impact of melting hydrate, but on time scales of millennia and longer
15 the climate impact is the same regardless of where the methane is oxidized. Methane
16 oxidized to CO₂ in the ocean will equilibrate with the atmosphere within a few hundred
17 years, resulting in the same partitioning of the added CO₂ between the atmosphere and
18 the ocean regardless of its origin. The rate and extent to which methane carbon can
19 escape the sediment column in response to warming is very difficult to constrain at
20 present. It depends on the stability of the sediment slope to sliding, and on the
21 permeability of the sediment and the hydrate stability zone's cold trap to bubble methane
22 fluxes.

23 **4.4 Conclusions About Potential for Abrupt Release of Methane From Marine** 24 **Hydrates**

25 On the time scale of the coming century, it appears likely that most of the marine hydrate
26 reservoir will be insulated from anthropogenic climate change. The exception is in
27 shallow ocean sediments where methane gas is focused by subsurface migration. The
28 most likely response of these deposits to anthropogenic climate change is an increased
29 background rate of chronic methane release, rather than an abrupt release. Methane gas in
30 the atmosphere is a transient species, its loss by oxidation continually replenished by

1 ongoing release. An increase in the rate of methane emission to the atmosphere from
2 melting hydrates would increase the steady-state methane concentration of the
3 atmosphere. The potential rate of methane emission from hydrates is more speculative
4 than the rate from other methane sources such as the decomposition of peat in thawing
5 permafrost deposits, or anthropogenic emission from agricultural, livestock, and fossil
6 fuel industries, but the potential rates appear to be comparable to these sources.

7 **5. Terrestrial Methane Hydrates**

8 There are two sources for methane in hydrates, biogenic production by microbes
9 degrading organic matter in anaerobic environments, and thermogenic production at
10 temperatures above 110°C, typically at depths greater than about 15 km. Terrestrial
11 methane hydrates are primarily biogenic (*Archer, 2007*). They form and are stable under
12 ice sheets (thicker than ~250 m) and within permafrost soils at depths of about 150 to
13 2,000 m below the surface (*Kvenvolden, 1993; Harvey and Huang, 1995*). Their presence
14 is known or inferred from geophysical evidence (e.g., well logs) on Alaska's North
15 Slope, the Mackenzie River delta (Northwest Territories) and Arctic islands of Canada,
16 the Messoyakha Gas Field and two other regions of western Siberia, and two regions of
17 northeastern Siberia (*Kvenvolden and Lorenson, 2001*). Samples of terrestrial methane
18 hydrates have been recovered from 900 to 1,110 m depth in the Mallik core in the
19 Mackenzie River delta (*Kvenvolden and Lorenson, 2001; Uchida et al., 2002*).

20 **5.1 Terrestrial Methane Hydrate Pool Size and Distribution**

21 While most methane hydrates are marine, the size of the contemporary terrestrial
22 methane hydrate pool, although unknown, may be large. Estimates range from less than
23 10 Gt CH₄ (*Meyer, 1981*) to more than 18,000 Gt CH₄ (*Dobrynin et al., 1981*) (both cited
24 in *Harvey and Huang, 1995*). More recent estimates are 400 Gt CH₄ (*MacDonald, 1990*),
25 800 Gt CH₄ (*Harvey and Huang, 1995*), and 4.5-400 GtC; this is a small fraction of the
26 ocean methane hydrate pool size (see [Sec. 4](#)).

27 Terrestrial methane hydrates are a potential fossil energy source. Recovery can come
28 from destabilization of the hydrates by warming, reducing the pressure, or injecting a
29 substance (e.g., methanol) that shifts the stability line (see [Box 5.1](#)). The Messoyakha

1 Gas Field in western Siberia, at least some of which lies in the terrestrial methane hydrate
2 stability zone, began producing gas in 1969, and some production is thought to have
3 come from methane hydrates, though methanol injection made this production very
4 expensive (*Kvenvolden, 1993; Krason, 2000*). A more recent review of the geological
5 evidence for methane production from hydrates at Messoyakha by *Collett and Ginsburg*
6 (*1998*) could not confirm unequivocally that hydrates contributed to the produced gas.
7 Due to low costs of other available energy resources, there had not been significant
8 international industrial interest in hydrate methane extraction during 1970-2000
9 (*Kvenvolden, 2000*), and the fraction of terrestrial methane hydrate that is or will be
10 technically and economically recoverable is not well established. In the U.S., the
11 Methane Hydrate Research and Development Act of 2000 and its subsequent 2005
12 Amendment have fostered the National Methane Hydrates R&D Program, supporting a
13 wide range of laboratory, engineering, and field projects with one focus being on
14 developing the knowledge and technology base to allow commercial production of
15 methane from domestic hydrate deposits by the year 2015, beginning with Alaska's
16 North Slope. Estimates of technically and economically recoverable methane in hydrates
17 are being developed (*Boswell, 2005, 2007*).

18 **5.2 Mechanisms To Destabilize Terrestrial Methane Hydrates**

19 Terrestrial methane hydrates in permafrost are destabilized if the permafrost warms
20 sufficiently or if the permafrost hydrate is exposed through erosion (see [Box 5.3](#)).
21 Destabilization of hydrates in permafrost by global warming is not expected to be
22 significant over the next few centuries (*Nisbet, 2002*; see [Sec. 5.4](#)). *Nisbet (2002)* notes
23 that although a warming pulse will take centuries to reach permafrost hydrates at depths
24 of several hundred meters, once a warming pulse enters the soil/sediment, it continues to
25 propagate downward and will eventually destabilize hydrates, even if the climate has
26 subsequently cooled.

27 Terrestrial methane hydrates under an ice sheet are destabilized if the ice sheet thins or
28 retreats. The only globally significant ice sheets now existing are on Greenland and
29 Antarctica; maps of the global distribution of methane hydrates do not show any hydrates
30 under either ice sheet (*Kvenvolden, 1993*). It is likely, however, that hydrates formed

1 under Pleistocene continental ice sheets (e.g., *Weitemeyer and Buffett, 2006*; see [Sec.](#)
2 [5.3.1](#)).

3 Terrestrial methane hydrates can also be destabilized by thermokarst erosion (a melt-
4 erosion process) of coastal-zone permafrost. Ice complexes in the soil melt where they
5 are exposed to the ocean along the coast, the land collapses into the sea, and more ice is
6 exposed (*Archer, 2007*). The Siberian coast is experiencing very high rates of coastal
7 erosion (*Shakova et al., 2005*). Methane hydrates associated with this permafrost become
8 destabilized through this process, and methane is released into the coastal waters
9 (*Shakova et al., 2005*). Magnitudes of the emissions are discussed below.

10 *De Batist et al. (2002)* analyzed seismic reflection data from Lake Baikal sediments, the
11 only freshwater nonpermafrost basin known to contain gas hydrates, and infer that
12 hydrate destabilization is occurring in this tectonically active lacustrine basin via upward
13 flow of hydrothermal fluids advecting heat to the base of the hydrate stability zone. If
14 occurring, this means of destabilization is very unlikely to be important globally, as the
15 necessary geological setting is rare.

16 Mining terrestrial hydrates for gas production will necessarily destabilize them, but
17 presumably most of this methane will be captured, used, and the carbon emitted to the
18 atmosphere as CO₂.

19 **5.3 Evidence of Past Terrestrial Hydrate Methane Release**

20 No direct evidence has been identified of past terrestrial hydrate methane release in
21 significant quantities. Analyses related to the PETM and clathrate gun hypothesis
22 discussed in [Sec. 4](#) have focused on methane emissions from the larger and more
23 vulnerable marine hydrates. Emissions from terrestrial hydrates may have contributed to
24 changes in methane observed in the ice core record, but there are so far no distinctive
25 isotopic tracers of terrestrial hydrates, as is the case for marine hydrate (*Sowers, 2006*).

26 **5.3.1. Quantity of Methane Released From Terrestrial Hydrates in the Past**

27 *Weitemeyer and Buffett (2006)* modeled the accumulation and release of biogenic
28 methane from terrestrial hydrates below the Laurentide and Cordilleran ice sheets of

1 North America during the last glaciation. Methane was generated under the ice sheet
2 from anaerobic decomposition of buried, near-surface soil organic matter, and hydrates
3 formed if the ice sheet was greater than ~250 m thick. Hydrate destabilization arose from
4 pressure decreases with ice sheet melting/thinning. They simulated total releases for
5 North America of about 40-100 Tg CH₄, with most of the deglacial emissions occurring
6 during periods of glacial retreat during a 500-year interval around 14 kyr before present
7 (BP), and a 2,000-year interval centered on about 10 kyr BP. The highest simulated
8 emission rates (~15-35 Tg CH₄ yr⁻¹) occurred during the dominant period of ice sheet
9 melting around 11-9 kyr BP.

10 *Shakova et al. (2005)* measured supersaturated methane concentrations in northern
11 Siberian coastal waters. This supersaturation is thought to arise from degradation of
12 coastal shelf hydrate, hydrate that had formed in permafrost when the shelf was exposed
13 during low sea level of the last glacial maximum. Methane concentrations in the Laptev
14 and East Siberian Seas were supersaturated up to 800% in 2003 and 2500% in 2004.
15 From this and an empirical model of gas flux between the atmosphere and the ocean, they
16 estimated summertime (i.e., ice-free) fluxes of up to 0.4 Mg CH₄ km⁻² y⁻¹ (or 0.4 g CH₄
17 m⁻² y⁻¹). They assume that the methane flux from the sea floor is of the same order of
18 magnitude, and may reach 1-1.5 g CH₄ m⁻² y⁻¹. These fluxes are low compared to wetland
19 fluxes (typically ~1-100 g CH₄ m⁻² y⁻¹; *Bartlett and Harriss, 1993*), but applied across the
20 total area of shallow Arctic shelf the total annual flux for this region may be as high as 1-
21 5 Tg CH₄ y⁻¹, depending on degree of oxidation in the seawater. (See [Table 5.1](#) above for
22 global methane emissions by source.)

23 **5.3.2 Climate Impact of Past Methane Release From Terrestrial Hydrates**

24 Most studies of climate impacts from possible past methane hydrate releases have
25 considered large releases from marine hydrates (see [Sec. 4](#) above). It is generally not well
26 known what fraction of the methane released from hydrate destabilization is either
27 trapped in overlying sediments or oxidized to carbon dioxide before reaching the
28 atmosphere (*Reeburgh, 2004*), and the same considerations are relevant to release from
29 terrestrial sources.

1 *Weitemeyer and Buffett (2006)* estimated intervals of 500-2,000 years when methane
2 hydrate destabilization from retreat of the North American ice sheet caused increases of
3 atmospheric methane of 10-200 ppb, with the largest perturbation at 11-9 kyr before
4 present. Any effect of methane oxidation before reaching the atmosphere was ignored;
5 this oxidation would have reduced the impact on the atmospheric methane burden. This
6 atmospheric perturbation is equivalent to about 2-25% of pre-industrial Holocene
7 atmospheric methane burdens, and roughly equivalent to a radiative forcing of 0.002 –
8 0.1 W m⁻² (using contemporary values for methane radiative efficiency and indirect
9 effects from *Ramaswamy et al., 2001*).

10 Thermokarst erosion on the Arctic coast of Siberia is thought to cause hydrate
11 destabilization and emissions of methane that are at most 1% of total global methane
12 emissions (*Shakova et al. 2005*), and so this process is very unlikely to be having a large
13 climatic impact..

14 **5.4 Estimates of Future Terrestrial Hydrate Release and Climatic Impact**

15 *Harvey and Huang (1995)* modeled terrestrial methane hydrate release due to global
16 warming (step function temperature increases of 5°C, 10°C, and 15°C, and the
17 propagation of this heat into hydrate-bearing permafrost). Over the first few centuries the
18 methane release is very small, and after 1,000 years, the cumulative methane release is
19 <1%, 2%, and 5% of the total terrestrial methane hydrate pool size, respectively; by 5,000
20 years this cumulative release has increased to 3%, 15%, and 30%, respectively. Even
21 5,000 years after a step function increase in temperature of 15°C, the radiative forcing
22 caused by terrestrial hydrate melting (direct effects of methane plus methane converted to
23 carbon dioxide) was only ~0.3 W/m².

24 Methane release from hydrate destabilization due to decaying ice sheets is unlikely to be
25 substantial unless there are significant hydrate pools under Greenland and/or Antarctica,
26 which does not seem to be the case. Thermoskarst erosion release is the only known
27 present terrestrial hydrate methane source. This process can be expected to continue into
28 the future, and it is very likely that emissions will remain a small fraction of the global
29 methane budget and therefore have a small impact on radiative forcing. However, most

1 recent modeling analyses have focused on marine hydrates (e.g., *Dickens, 2001c; Archer*
2 *and Buffett, 2004*), and more work on the terrestrial hydrate reservoir is clearly needed.

3 **5.5 Conclusions**

4 No mechanisms have been proposed for the abrupt release of significant quantities of
5 methane from terrestrial hydrates (*Archer, 2007*). Slow and perhaps sustained release
6 from permafrost regions may occur over decades to centuries from mining extraction of
7 methane from terrestrial hydrates in the arctic (*Boswell, 2007*), over decades to centuries
8 from continued thermokarst erosion of coastal permafrost in Eurasia (*Shakova et al.,*
9 *2005*), and over centuries to millennia from the propagation of any warming 100-1,000 m
10 down into permafrost hydrates (*Harvey and Huang, 1995*).

11 **6. Changes in Methane Emissions From Natural Wetlands**

12 **6.1 Introduction**

13 Natural wetlands are most extensive at high northern latitudes, where boreal and arctic
14 wetlands have substantial carbon in peat and are frequently associated with permafrost,
15 and in the tropics, often associated with river and lake floodplains. Annual methane
16 emissions from tropical wetlands are roughly twice that from boreal/arctic wetlands.
17 Globally, wetlands are the largest single methane source to the atmosphere, with recent
18 emission estimates ranging from 100 to 231 Tg CH₄ yr⁻¹ (*Denman et al., 2007*),
19 constituting more than 75% of the total estimated natural emissions. Variations in
20 wetland distribution and saturation, in response to long-term variations in climate, are
21 therefore thought to have been main determinants for variation in the atmospheric CH₄
22 concentration in the past (*Chappellaz et al., 1990; Chappellaz et al., 1993a,b; Brook et*
23 *al., 1996, 2000; Delmotte et al., 2004*). Recent interannual variations in methane
24 emissions have been dominated by fluctuations in wetland emissions (*Bousquet et al.,*
25 *2006*), although biomass burning also plays a significant role.

26 Methane emissions from natural wetlands are sensitive to temperature and moisture (see
27 below), and thus to climate variability and change. Emissions can also be influenced by
28 anthropogenic activities that impact wetlands such as pollution loading (e.g., *Gauci et al.,*
29 *2004*), land management (e.g., *Minkinen et al., 1997*), and water management (e.g., *St.*

1 *Louis et al., 2000*). While these anthropogenic impacts can be expected to change in the
2 coming decades, they are unlikely to be a source of abrupt changes in methane emissions
3 from natural wetlands, so this section will focus on climate change impacts.

4 Global climate model projections suggest that the tropics, on average, and the northern
5 high latitudes are likely to become warmer and wetter during the 21st century, with
6 greater changes at high latitudes (*Chapman and Walsh, 2007; Meehl et al., 2007*).

7 Temperatures in the tropics by 2100 are projected to increase by 2-4°C (*Meehl et al.,*
8 *2007*). Precipitation in the tropics is expected to increase in East Africa and Southeast
9 Asia, show little change in West Africa and Amazonia, and decrease in Central America
10 and northern South America (*Meehl et al., 2007*).

11 Warming in the northern high latitudes in recent decades has been stronger than in the
12 rest of the world (*Serreze and Francis, 2006*), and that trend is projected to continue, with
13 multimodel projections indicating that Arctic land areas could warm by between 3.5° and
14 8°C by 2100 (*Meehl et al., 2007*). The northern high latitudes are also expected to see an
15 increase in precipitation by more than 20% in winter and by more than 10% in summer.
16 Climate change of this magnitude is expected to have diverse impacts on the Arctic
17 climate system (*ACIA, 2004*), including the methane cycle. Principal among the projected
18 impacts is that soil temperatures are expected to warm and permafrost, which is prevalent
19 across much of the northern high latitudes, is expected to thaw and degrade. Permafrost
20 thaw may alter the distribution of wetlands and lakes through soil subsidence and
21 changes in local hydrological conditions. Since methane production responds positively
22 to soil moisture and summer soil temperature, the projected strong warming and
23 associated landscape changes expected in the northern high latitudes, coupled with the
24 large carbon source (northern peatlands have ~250 GtC as peat within 1 to a few meters
25 of the atmosphere; *Turunen et al., 2002*), will likely lead to an increase in methane
26 emissions over the coming century.

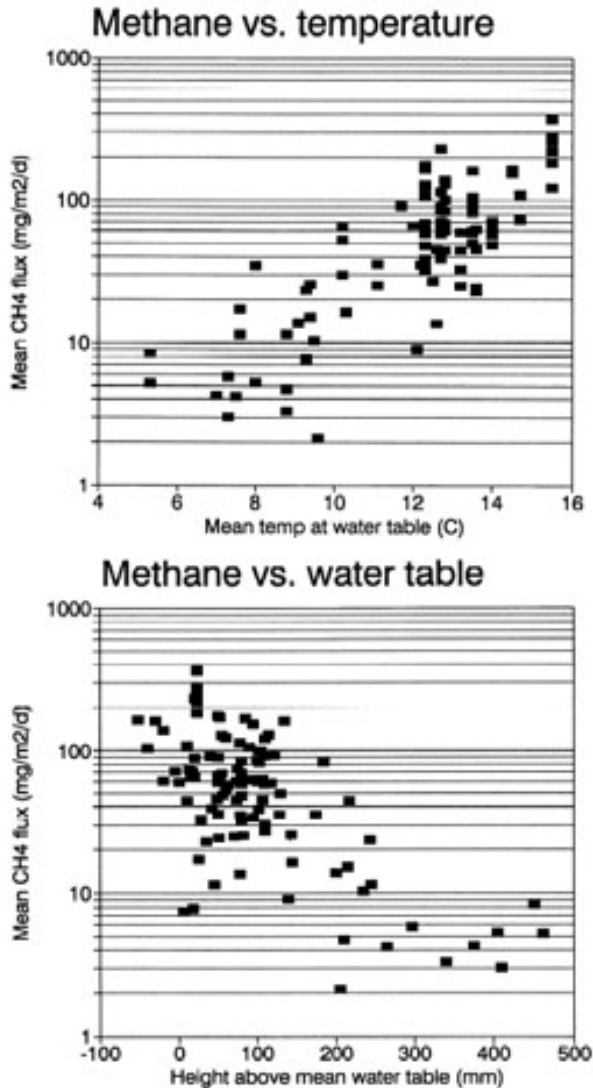
27 **6.2 Factors Controlling Methane Emissions From Natural Wetlands**

28 Methane is produced as a byproduct of microbial decomposition of organic matter under
29 anaerobic conditions that are typical of saturated soils and wetlands. As this methane

1 migrates from the saturated soil to the atmosphere (via molecular diffusion, ebullition
2 (bubbling), or plant-mediated transport), it can be oxidized to carbon dioxide by
3 microbial methanotrophs in oxygenated sediment or soil. In wetlands, a significant
4 fraction of the methane produced is oxidized by methanotrophic bacteria before reaching
5 the atmosphere (*Reeburgh, 2004*). If the rate of methanogenesis is greater than the rate of
6 methanotrophy and pathways for methane to diffuse through the soil are available, then
7 methane is emitted to the atmosphere. Dry systems, where methanotrophy exceeds
8 methanogenesis, can act as weak sinks for atmospheric methane (see [Table 5.1](#)). Methane
9 emissions are extremely variable in space and time, and therefore it is difficult to quantify
10 regional-scale annual emissions (*Bartlett and Harriss, 1993; Melack et al., 2004*). Recent
11 reports of a large source (62-236 Tg CH₄ yr⁻¹) of methane from an aerobic process in
12 plants (*Keppler et al., 2006*) appear to be overstated (*Dueck et al., 2007; Wang et al.,*
13 *2008*).

14 There have not been many field studies measuring methane fluxes from tropical wetlands
15 around the world, but work in the Amazon and Orinoco Basins of South America has
16 shown that methane emissions appear to be most strongly controlled in aquatic habitats
17 by inundation depth and vegetation cover (e.g., flooded forest, floating macrophytes,
18 open water) (*Devol et al., 1990; Bartlett and Harriss, 1993; Smith et al., 2000; Melack et*
19 *al., 2004*). Wet season (high water) fluxes are generally higher than dry season (low
20 water) fluxes (*Bartlett and Harriss, 1993*).

21 At high latitudes, the most important factors influencing methane fluxes are water table
22 depth, soil or peat temperature, substrate type and availability, and vegetation type ([Fig.](#)
23 [5.13](#)). Water table depth determines both the fraction of the wetland soil/peat that is
24 anaerobic and the distance from this zone of methane production to the atmosphere (i.e.,
25 the length of the oxidation zone) and is often the single most important factor controlling
26 emissions (*Bubier et al., 1995; Waddington et al., 1996; MacDonald et al., 1998*). The
27 strong sensitivity of CH₄ emissions to water table position suggests that changing
28 hydrology of northern wetlands under climate change could drive large shifts in
29 associated methane emissions.



1

2 **Figure 5.13.** Relationships between water table height, temperature, and methane
3 emissions for northern wetlands from *Bubier et al. (1995)*. Abbreviations: mg/m²/d,
4 milligrams per square meter per day; mm, millimeters; C, degrees Celsius.

5 Vegetation type controls plant litter tissue quality/decomposability, methanogen substrate
6 input by root exudation (e.g., *King and Reeburgh, 2002*), and the potential for plant-
7 mediated transport of methane to the atmosphere (e.g., *King et al., 1998; Joabsson and*
8 *Christensen, 2001*). Substrate type and quality, generally related to quantity of root
9 exudation and to vegetation litter quality and degree of decomposition, can directly affect
10 potential methane production. Vegetation productivity controls the amount of organic
11 matter available for decomposition.

1 In wetland ecosystems, when the water table is near the surface and substantial methane
2 emissions occur, the remaining controlling factors rise in relevance. *Christensen et al.*
3 (2003) find that temperature and microbial substrate availability together explain almost
4 100% of the variations in mean annual CH₄ emissions across a range of sites across
5 Greenland, Iceland, Scandinavia, and Siberia. *Bubier et al. (1995)* find a similarly strong
6 dependence on soil temperature at a northern peatland complex in Canada. The observed
7 strong relationship between CH₄ emissions and soil temperature reflects the exponential
8 increase in microbial activity as soil temperatures warm. The strong warming expected
9 across the northern high latitudes is likely to be a positive feedback on methane
10 emissions.

11 The presence or absence of permafrost can also have a direct influence on CH₄ emissions.
12 Across the northern high latitudes, permafrost features such as ice wedges, ice lenses,
13 thermokarst, and ice heaving determine the surface microtopography. Small variations in
14 surface topography have a strong bearing on plant community structure and evolution as
15 well as soil hydrologic and nutritional conditions (*Jorgenson et al., 2001, 2006*), all of
16 which are controlling factors for methane emission. Projections of future methane
17 emission are hampered by the difficulty of modeling landscape/watershed hydrology well
18 enough at large scales to realistically represent small changes in wetland water table
19 depth.

20 **6.3 Observed and Projected Changes in Natural Wetlands**

21 **6.3.1 Observed Changes in Arctic Wetlands and Lakes**

22 Increased surface ponding and wetland formation have been observed in warming
23 permafrost regions (*Jorgenson et al., 2001, 2006*). These increases are driven primarily
24 by permafrost-thaw-induced slumping and collapsing terrain features (thermokarst) that
25 subsequently fill with water. For the Tanana Flats region in central Alaska, large-scale
26 degradation of permafrost over the period 1949-95 is associated with substantial losses of
27 birch forest and expansion of wetland fens (*Jorgenson et al., 2001*).

28 In recent decades, lake area and count in discontinuous permafrost regions have
29 decreased in western Siberia (*Smith et al., 2005*) and Alaska (*Riordan et al., 2006*) but

1 have increased in continuous permafrost regions in northwestern Siberia (*Smith et al.*,
2 2005). The differing trends in discontinuous and continuous permafrost zones can be
3 understood if one considers that initial permafrost warming leads to development of
4 thermokarst and lake and wetland expansion as the unfrozen water remains trapped near
5 the surface by the icy soil beneath it. As the permafrost degrades more completely, lake
6 or wetland drainage follows, as water more readily drains through the more ice-free soil
7 to the ground-water system.

8 A strength of the *Smith et al. (2005)* study is that lake abundance is determined via
9 satellite, permitting the study of thousands of lakes and evaluation of the net change
10 across a broad area, which can in turn be attributed to regional driving mechanisms such
11 as climate and permafrost degradation. A similar analysis for wetlands would be useful
12 but is presently intractable because wetlands are not easy to pinpoint from satellite, as
13 inundation, particularly in forested regions, cannot be easily mapped, and wetland-rich
14 landscapes are often very spatially heterogeneous. (*Frey and Smith, 2007*).

15 Present-generation global climate or large-scale hydrologic models do not represent the
16 thermokarst processes that appear likely to dictate large-scale changes in wetland extent
17 over the coming century. However, wetland area can also respond to trends in
18 precipitation minus evaporation (P–E). A positive P–E trend could lead, in the absence of
19 large increases in runoff, to an expansion of wetland area and more saturated soil
20 conditions, thereby increasing the area from which methane emission can occur. Most
21 climate models predict that both Arctic precipitation and evapotranspiration will rise
22 during the 21st century if greenhouse gas concentrations in the atmosphere continue to
23 rise. In at least one model, the NCAR CCSM3, the P–E trend is positive throughout the
24 21st century (*Lawrence and Slater, 2005*).

25 **6.3.2 Observed and Projected Changes in Permafrost Conditions**

26 There is a considerable and growing body of evidence that soil temperatures are
27 warming, active layer thickness (ALT) is increasing, and permafrost is degrading at
28 unprecedented rates (e.g., *Osterkamp and Romanovsky, 1999; Romanovsky et al., 2002,*
29 *Smith et al., 2005; Osterkamp and Jorgenson, 2006*). Continuous permafrost in Alaska,

1 which has been stable over hundreds, or even thousands, of years, has suffered an abrupt
2 increase in degradation since 1982 that “appears beyond normal rates of change in
3 landscape evolution” (*Jorgenson et al., 2006*). Similarly, discontinuous permafrost in
4 Canada has shown a 200-300% increase in the rate of thawing over the 1995-2002 period
5 relative to that of 1941-91 (*Camill, 2005*). *Payette et al. (2004)* present evidence of
6 accelerated thawing of subarctic peatland permafrost over the last 50 years. An example
7 of permafrost degradation and transition to wetlands in the Tanana Flats region of central
8 Alaska is shown in [Figure 5.14](#).



9

10 **Figure 5.14.** Transition from tundra (left, 1978) to wetlands (right, 1998) due to
11 permafrost degradation over a period of 20 years (*Jorgensen et al., 2001*). Photos, taken
12 from the same location in Tanana Flats in central Alaska, obtained from
13 <http://www.arctic.noaa.gov/detect/land-tundra.shtml>.

14 Model projections of soil temperature warming and permafrost degradation in response to
15 the strong anticipated high-latitude warming vary considerably, although virtually all of
16 them indicate that a significant amount of permafrost degradation will occur if the Arctic
17 continues to warm (*Anisimov and Nelson, 1997; Stendel and Christensen, 2002; Zhang et*
18 *al., 2003; Sazonova et al., 2004*). *Buteau et al. (2004)* find downward thawing rates of up
19 to 13 cm yr⁻¹ in ice-rich permafrost for a 5°C warming over 100 years. A collection of
20 process-based models, both global and regional, all with varying degrees of completeness
21 in terms of their representation of permafrost, indicates widespread large-scale
22 degradation of permafrost (and by extension increased thermokarst development), sharply
23 increasing ALTs, and a contraction of the area where permafrost can be found near the

1 Earth's surface during the 21st century (*Lawrence and Slater, 2005; Euskirchen et al.,*
2 *2006; Lawrence et al., 2007; Saito et al., 2007; Zhang et al., 2007*).

3 **Box 5.3—High-Latitude Terrestrial Feedbacks**

4 In recent decades, the Arctic has witnessed startling environmental change. The
5 changes span many facets of the Arctic system including rapidly decreasing sea ice
6 extent, melting glaciers, warming and degrading permafrost, increasing runoff to the
7 Arctic Ocean, expanding shrub cover, and important changes to the carbon balance
8 (*Serreze et al., 2000; ACIA, 2004; Hinzman et al., 2005*). The observed
9 environmental trends are driven largely by temperatures that are increasing across the
10 Arctic at roughly twice the rate of the rest of the world (*Serreze and Francis, 2006*).
11 If the Arctic warming continues and accelerates, as is predicted by all global climate
12 models (*Chapman and Walsh, 2007*), it may invoke a number of feedbacks that have
13 the potential to alter and possibly accelerate Arctic *and* global climate change. If the
14 feedbacks operate constructively, even relatively small changes in the Arctic could
15 conspire to amplify global climate change. Continued environmental change,
16 especially if it occurs rapidly, is likely to have adverse consequences for highly
17 vulnerable Arctic and global ecosystems and negative impacts on human activities,
18 particularly in the Arctic, including costly damage to infrastructure and
19 marginalization of many Arctic communities.

20 The Arctic can influence global climate through both positive and negative feedbacks
21 ([Fig. 5.15](#)). For example, sea-ice retreat reduces surface albedo, enhances absorption
22 of solar radiation, and ultimately leads to greater pan-Arctic warming. Large-scale
23 thawing of permafrost alters soil structural (thermokarst) and hydrologic properties
24 (*Jorgenson et al., 2001*) with additional effects on the spatial extent of lakes and
25 wetlands (*Smith et al., 2005; Riordan et al., 2006*), runoff to the Arctic ocean,
26 ecosystem functioning (*Jorgenson et al., 2001; Payette et al., 2004*), and the surface
27 energy balance. Warming is also expected to enhance decomposition of soil organic
28 matter, releasing carbon to the atmosphere (a positive feedback) (*Zimov et al., 2006*)
29 and also releasing nitrogen which, in nutrient limited Arctic ecosystems, may prompt
30 shrub growth (a negative feedback due to carbon sequestration) (*Sturm et al., 2001*).
31 This greening-of-the-Arctic negative feedback may itself be offset by a positive

1 radiative feedback related to lower summer and especially winter albedos of shrubs
2 and trees relative to tundra (*Chapin et al., 2005*), which promotes an earlier spring
3 snowmelt that among other things affects soil temperature and permafrost (*Sturm et*
4 *al., 2001*).

5 The future of the Arctic as a net sink or source of carbon to the atmosphere depends
6 on the delicate balance between carbon losses through enhanced soil decomposition
7 and carbon gains to the ecosystem related to the greening of the Arctic (*McGuire et*
8 *al., 2006*). Irrespective of the carbon balance, anticipated increases in methane
9 emissions mean that the Arctic is likely to be an effective greenhouse gas source
10 (*Friberg et al., 2003; McGuire et al., 2006*).

11 The Arctic is a complex and interwoven system. On the basis of recent evidence of
12 change, it appears that many of these feedbacks are already operating. Whether or not
13 the positive or negative feedbacks will dominate is a critical question facing climate
14 science. In a recent paper reviewing the integrated regional changes in Arctic climate
15 feedbacks, *McGuire et al. (2006)* conclude that the balance of evidence indicates that
16 the positive feedbacks to global warming will likely dominate over the next century,
17 but whether or not the myriad feedbacks will interact to significantly amplify (or
18 mitigate) global climate change remains difficult to predict, especially since much of
19 the research to date has considered these feedbacks in isolation.

6.4 Observed and Modeled Sensitivity of Wetland Methane Emissions to Climate Change

Field studies indicate that methane emissions do indeed increase in response to soil warming and permafrost thaw. *Christensen et al. (2003)* note that a steady rise in soil temperature will enhance methane production from existing regions of methanogenesis that are characterized by water tables at or near the surface. While this aspect is important, changes in landscape-scale hydrology have the ability to drive a more significant change in methane emissions. For example, at a mire in sub-Arctic Sweden, permafrost degradation and associated vegetation changes have driven a 22-66% increase in landscape-scale CH₄ emissions over the period 1970 to 2000 (*Christensen et al., 2004*). *Bubier et al. (2005)* estimated that in a Canadian boreal landscape with discontinuous permafrost and ~30% wetland coverage, landscape-scale methane fluxes increased by ~60% from a dry year to a wet year, due to changes in wetland water table depth, particularly at the beginning and end of the summer. *Nykänen et al. (2003)* also found higher methane fluxes during a wetter year at a sub-Arctic mire in northern Finland. *Walter et al. (2006)* find that thawing permafrost along the margins of thaw lakes in eastern Siberia accounts for most of the methane released from the lakes. This emission, which occurs primarily through ebullition, is an order of magnitude larger where there has been recent permafrost thaw and thermokarst compared to where there has not. These hotspots have extremely high emission rates but account for only a small fraction of the total lake area. Methane released from these hotspots appears to be Pleistocene age, indicating that climate warming may be releasing old carbon stocks previously stored in permafrost (*Walter et al., 2006*). At smaller scales, there is strong evidence that thermokarst development substantially increases CH₄ emissions from high-latitude ecosystems. Mean CH₄ emission rate increases between permafrost peatlands and collapse wetlands of 13-fold (*Wickland et al., 2006*), 30-fold (*Turetsky et al., 2002*), and up to 19-fold (*Bubier et al., 1995*) have been reported.

A number of groups have attempted to predict changes in natural wetland methane emissions on a global scale. These studies broadly suggest that natural methane emissions from wetlands will rise as the world warms. *Shindell et al. (2004)* incorporate a linear parameterization for methane emissions, based on a detailed process model, into a global

1 climate model and find that overall wetland methane emissions increased by 121 Tg CH₄
2 y⁻¹, 78% higher than their baseline estimate. They project a tripling of northern high-
3 latitude methane emissions, and a 60% increase in tropical wetland methane emissions in
4 a doubled CO₂ simulation. The increase is attributed to a rise in soil temperature in
5 combination with wetland expansion driven by a positive P-E trend predicted by the
6 model. About 80% of the increase was due to enhanced flux rates, and 20% due to
7 expanded wetland area or duration of inundation. The predicted increase in the
8 atmospheric methane burden was 1,000 Tg, ~20% of the current total, equivalent to an
9 increase of ~430 ppb, assuming a methane lifetime of 8.9 years. Utilizing a similar
10 approach but with different climate and emission models, *Gedney et al. (2004)* project
11 that global wetland emissions (including rice paddies) will roughly double, despite a
12 slight reduction in wetland area. The northern wetland methane emissions, in particular,
13 increase by 100% (44 to 84 Tg CH₄ yr⁻¹) in response to increasing soil temperatures and
14 in spite of a simulated 10% reduction in northern wetland areal extent. Using a more
15 process-based ecosystem model, which includes parameterizations for methane
16 production and emission, *Zhuang et al. (2007)* model a doubling of methane emissions
17 over the 21st century in Alaska, once again primarily in response to the soil temperature
18 influence on methanogenesis, and secondarily to an increase in net primary productivity
19 of Alaskan ecosystems. These factors outweigh a negative contribution to methane
20 emissions related to a simulated drop in the water table. It is important to note that these
21 models simulate only the direct impacts of climate change (altered temperature and
22 moisture regimes, and in one case enhanced vegetation productivity) but not indirect
23 impacts, such as changing landscape hydrology with permafrost degradation and
24 changing vegetation distribution. At this time, it is not known whether direct or indirect
25 effects will have a stronger impact on net methane emissions. These models all predict
26 fairly smooth increases in annual wetland emissions, with no abrupt shifts in flux.

27 **6.5 Conclusion About Potential for Abrupt Release of Methane From Wetlands**

28 Tropical wetlands are a stronger methane source than boreal/arctic wetlands and will
29 likely continue to be over the next century, during which fluxes from both regions are
30 expected to increase. However, four factors differentiate northern wetlands from tropical
31 wetlands and make them more likely to experience a larger increase in fluxes: (1) high-

1 latitude amplification of climatic warming will lead to a stronger temperature impact, (2)
2 for regions with permafrost, warming-induced permafrost degradation could make more
3 organic matter available for decomposition and substantially change the system
4 hydrology, (3) the sensitivity of microbial respiration to temperature generally decreases
5 with increasing temperatures (e.g., *Davidson and Janssens, 2006*), and (4) most northern
6 wetlands have substantial carbon as peat. On the other hand, two characteristics of
7 northern peatlands counter this: (1) northern peatlands are complex, adaptive ecosystems,
8 with internal feedbacks and self-organizing structure (*Belyea and Baird, 2007*) that allow
9 them to persist in a relatively stable state for millennia and that may reduce their
10 sensitivity to hydrological change, and (2) much of the organic matter in peat is well-
11 decomposed (e.g., *Frolking et al. 2001*) and may not be good substrate for methanogens.

12 The balance of evidence suggests that anticipated changes to northern wetlands in
13 response to large-scale permafrost degradation, thermokarst development, a positive P-E
14 trend in combination with substantial soil warming, enhanced vegetation productivity,
15 and an abundant source of organic matter will likely conspire to drive a chronic increase
16 in CH₄ emissions from the northern latitudes during the 21st century. Due to the strong
17 interrelationships between temperature, moisture, permafrost, and nutrient and vegetation
18 change, and the fact that negative feedbacks such as the draining/drying of wetlands are
19 also possible, it is difficult to establish how large the increase will be over the coming
20 century. Current models suggest that a doubling of CH₄ emissions could be realized fairly
21 easily. However, since these models do not realistically represent all the processes
22 thought to be relevant to future northern high-latitude CH₄ emissions, much larger (or
23 smaller) increases cannot be discounted.

24 It is worth noting that our understanding of the northern high-latitude methane cycle
25 continues to evolve. For example, a recent field study suggests that prior estimates of
26 methane emissions from northern landscapes may be biased low due to an
27 underestimation of the contribution of ebullition from thermokarst hot spots in Siberian
28 thaw lakes (*Walter et al., 2006*). Another interesting recently discovered phenomena is
29 the cold adaptation of some methanogenic microorganisms that have been found in
30 permafrost deposits in the Lena River basin (*Wagner et al., 2007*). These microbes can

1 produce methane even in the very cold conditions of permafrost, often drawing on old
2 soil organic matter. The activity levels of these cold-adapted methanogens are sensitive to
3 temperature, and even a modest soil warming can lead to an accumulation of methane
4 deposits which, under scenarios where permafrost degradation leads to thermokarst or
5 coastal erosion, could be quickly released to the atmosphere.

6 These recent studies highlight the fact that key uncertainties remain in our understanding
7 of natural methane emissions and their susceptibility to climate change as well as in our
8 ability to predict future emissions. Among the most important uncertainties in our
9 understanding and required improvements to process-based models are (1) the
10 contribution of ebullition and changes in ebullition to total methane emissions; (2) the
11 rate of change in permafrost distribution and active layer thickness and associated
12 changes in distribution of wetlands and lakes as well as, more generally, terrestrial
13 ecosystems; (3) model representation of soil thermal and hydrologic processes and their
14 response to climate change; (4) the contribution that shifts in vegetation and changes in
15 peatland functioning will have on the methane cycle; and (5) representation of the highly
16 variable and regionally specific methane production and emission characteristics. Even
17 with resolution of these issues, all predictions of future methane emissions are based on
18 the accurate simulation and prediction of high-latitude climate. Improvements of many
19 facets critical to the high-latitude climate system are required, including improvements to
20 the treatment of snow, polar clouds, subsoil processes, sub-polar oceans, and sea ice in
21 global climate models.

22 **7. Final Perspectives**

23 Although the prospect of a catastrophic release of methane to the atmosphere as a result
24 of anthropogenic climate change over the next century appears very unlikely based on
25 current knowledge, many of the processes involved are still poorly understood, and
26 developing a better predictive capability requires further work. On a longer time scale,
27 methane release from hydrate reservoir is likely to be a major player in global warming
28 over the next 1,000-100,000 years. Changes in climate, including warmer temperatures
29 and more precipitation in some regions, will likely increase the chronic emissions of
30 methane from both melting hydrates and natural wetlands over the next century. The

1 magnitude of this effect cannot be predicted with great accuracy yet, but is likely to be
2 equivalent to the current magnitude of many anthropogenic sources, which have already
3 more than doubled the levels of methane in the atmosphere since the start of the
4 Industrial Revolution.

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