STODDARD SOLVENT 81

## 3. CHEMICAL AND PHYSICAL INFORMATION

## 3.1 CHEMICAL IDENTITY

Information regarding the chemical identity of Stoddard solvent is located in Table 3-1.

## 3.2 PHYSICAL AND CHEMICAL PROPERTIES

Information regarding the physical and chemical properties of Stoddard solvent is located in Table 3-2.

Stoddard solvent is a petroleum distillate mixture of  $C_7$ - $C_{12}$  hydrocarbons. The mixture consists of three major groups of components: linear and branched alkanes, also known as paraffins (30-50% of the total mixture); cycloalkanes, also called cycloparaffins or naphthenes (not to be confused with naphthalenes which are bicyclic aromatics) (30-40%), and aromatic hydrocarbons (10-20%) (Air Force 1989b; McDermott 1975). Stoddard solvent is a refinery blend of differently treated oil fractions. Its composition varies somewhat, depending on the refinery and the time of production. Table 3-3 lists some of the major components of several Stoddard solvent formulations. Petroleum distillates are often distinguished by boiling or distilling temperatures. Stoddard solvent has a boiling range of 150-200°C (Scott et al. 1959). The 140 flash Stoddard solvent is composed of  $C_5$ - $C_{12}$  hydrocarbons and has a boiling range of 185-207°C (Air Force 1989b). White spirits is a term somewhat synonymous with Stoddard solvent since it has a hydrocarbon range between  $C_7$  and  $C_{11}$ . Six types of white spirits have been identified based on origin. Each type consists of the same components, but the percentages vary (Scheffers et al. 1985). Possible contaminants of Stoddard solvent include lead (<1 ppm) and sulfur (3.5 ppm) (Suntech 1978).

There are a number of related chemical mixtures with components that are different from those of Stoddard solvent. For instance, high-flash aromatic naphtha is a generic term for petroleum distillates primarily consisting of C<sub>9</sub> aromatics (70-80%) with C<sub>8</sub> or C<sub>10</sub> aromatics comprising the rest. Stoddard solvent, in contrast, is only 10-20% aromatic (Clark et al. 1989b; Schreiner et al. 1989). Naphtha is also a general term for petroleum distillates containing predominantly C<sub>5</sub>-C<sub>13</sub> aliphatic hydrocarbons and distilling at 30-238°C (Tenenbein et al. 1984).

**TABLE 3-1. Chemical Identity of Stoddard Solvent** 

Characteristic	Information	Reference  Sax and Lewis 1989  Air Force 1989b;  NIOSH 1989; Sax and Lewis 1989		
Chemical name	Stoddard solvent			
Synonym(s)	Dry cleaning safety solvent, naphtha safety solvent, PD-680, petroleum solvent, spotting naphtha, varnoline, white spirits			
Registered trade name(s)	Texsolve S, Varsol 1	Budavari et al. 1989; Hunter et al. 1992		
Chemical formula	Not applicable <sup>a</sup>			
Chemical structure	Not applicable <sup>a</sup>			
Identification numbers:				
CAS registry	8052-41-3	Sax and Lewis 1989		
NIOSH RTECS	WJ8925000	NIOSH 1990		
EPA hazardous waste	No data			
OHM/TADS	No data			
DOT/UN/NA/IMCO shipping	1268 27	NIOSH 1990		
HSDB	No data			
NCI	No data			

<sup>&</sup>lt;sup>a</sup>Stoddard solvent is a mixture of  $C_7$ – $C_{12}$  hydrocarbons primarily containing straight and branched chain alkanes (30–50%), cycloalkanes (30–40%), and alkyl aromatic hydrocarbons (10–20%) (Air Force 1989b; McDermott 1975). See also Table 3-3.

CAS = Chemical Abstracts Services; DOT/UN/NA/IMCO = Department of Transportation/United Nations/North America/International Maritime Dangerous Goods Code; EPA = Environmental Protection Agency; HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil and Hazardous Materials/Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substances

TABLE 3-2. Physical and Chemical Properties of Stoddard Solvent

Property	Information	Reference		
Molecular weight	144 (mean); 135–145 (range)	Air Force 1989b; Carpenter et al. 1975b		
Color	Clear, colorless	Sax and Lewis 1989		
Physical state	Liquid	Sax and Lewis 1989		
Melting point	No data			
Boiling point	154–202°C	Air Force 1989b		
	160–199°C	Coast Guard 1985		
Density:				
at 20°C	0.78 g/mL	NIOSH 1990		
Odor	Similar to kerosene	NIOSH 1990		
Odor threshold	$0.9 \text{ ppm } (5.1 \text{ mg/m}^3)$	Carpenter et al. 1975b		
	$2 \text{ mg/m}^3$	Hastings et al. 1984		
Solubility:	•	C		
Water	Insoluble	McDermott 1975		
Organic solvent(s)	Absolute alcohol,	Sax and Lewis 1989		
-	benzene, ether,			
	chloroform, carbon			
	tetrachloride, carbon			
	disulfide			
Partition coefficients:				
Log K <sub>ow</sub>	3.16-7.06	Air Force 1989b		
Log K <sub>oc</sub>	2.85-6.74	Air Force 1989b		
Vapor pressure at 25°C	4–4.5 mmHg	McDermott 1975		
Henry's law constant:	•			
at 20°C	$4.4 \times 10^{-4} - 7.4 \times 10^{\circ} \text{ atm} \cdot \text{m}^{3}/\text{mol}$	Air Force 1989b		
Autoignition temperature	232°C	Sax and Lewis 1989		
Flashpoint	37.8–60.0°C	Air Force 1989b		
	38–43°C	Sax and Lewis 1989		
Flammability limits:				
% volume in air at 25°C	0.9–6.0	Carpenter et al. 1975b		
Conversion factors:		•		
at 25°C and 760 mm	1  mg/L = 174.6  ppm;	Carpenter et al. 1975b		
	$1 \text{ ppm} = 5.73 \text{ mg/m}^3$	_		
at 25°C	1 ppm = $5.77 \text{ mg/m}^3$	Air Force 1989b		
Explosive limits	<del>-</del>	McDermott 1975		
Lower limit	0.9%			
Upper limit	6%			

**TABLE 3-3. Possible Formulations of Stoddard Solvent (Percent)** 

Hydrocarbons	White Spirits		White Spirits 3 <sup>a</sup>	Stoddard solvent <sup>b</sup> (regular)	Stoddard solvent <sup>b</sup> (140 flash)	Stoddard solvent <sup>c</sup>	Stoddard solvent <sup>d</sup>	Stoddard solvent <sup>e</sup>
Alkanes (paraffins)	60.0	61.0	62.8	30-50 (48 average)	60.8	34.9	41.6	47.7
n-nonane	11.3	13.3	1.9					
n-decane	7.6	10.0	9.1					
methylnonanes	4.9	7.9						
2,6-dimethyloctane	2.7	4.1						
n-undecane	2.7	2.4	17.5					
dodecanes			11.6					
terdecanes			2.7					
others	30.8	23.3						
Cycloalkanes (cycloparaffins)	39.7	27.3		30-40 (38 average)	35.7		39.5	37.6
monocycloparaffins	16.3	13.7		` 0,	24.5	34.9	27.9	26.0
trimethylcyclohexane	4.7	7.2						
tert-butylcyclohexane	4.5	4.0						
<i>n</i> -butylcyclopentane	5.0	1.3						
<i>n</i> -butylcyclohexane	2.1	1.2						
other cycloparaffins	23.4	13.1						
dicycloparaffins	25.1	20.1			11.2	5.0	11.6	11.6
tricycloparaffins						0.4	0.0	
acenaphthenes						0.4		
acchaphanenes						0.1		
Aromatics	0.3	11.7	17.0	10-20 (14.1 average)			18.9	
alkylbenzenes				14.0	3.03	22.0	17.6	14.1
dimethylethylbenzenes	0	3.0						
n-propylbenzene	0	2.0						
ethyltoluenes	0	1.2						
1,2,4-trimethylbenzene	0	0.9						
other aromatics	0.3	4.6				1.1		
other benzenes				0.1	0.07			0.1
indans/tetralins				<1	0.3	1.8	1.3	0.5
indenes						0.1		
naphthalenes						0.2		
acenaphthalenes						0.3		
tricyclicaromatics	<u> </u>					0.1		

<sup>&</sup>lt;sup>a</sup>Adapted from Verkkala et al. (1984)

<sup>&</sup>lt;sup>b</sup>Adapted from Air Force (1989b)

<sup>&</sup>lt;sup>c</sup>Adapted from American Petroleum Institute (1976)

<sup>&</sup>lt;sup>d</sup>Adapted from Suntech Group (1978); API 1978a

eAdapted from Carpenter et al. (1975b); this paper also includes a mass spectral analysis of components by carbon number within a hydrocarbon class, e.g., C<sub>8</sub> alkanes.

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Benzine and mineral spirits (other associated mixtures) are similar to but not exactly the same as Stoddard solvent. Benzine consists of C<sub>5</sub>-C<sub>9</sub> hydrocarbons (Takeuchi et al. 1975) and boils, on average, at between 154°C and 204°C (Navarte et al. 1989). Benzine and Stoddard solvent distill at about the same temperature range, but their hydrocarbon compositions differ. Mineral spirits have a distillation range of 136-277°C. The distillation range of Stoddard solvent falls within that of mineral spirits (Mehlman and Smart 1982). Therefore, Stoddard solvent may be considered a subset of mineral spirits, but mineral spirits as a whole are not described in this profile.