

### **3. CHEMICAL AND PHYSICAL INFORMATION**

#### **3.1 CHEMICAL IDENTITY**

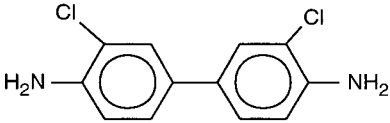
Information regarding the chemical identity of 3,3'-dichlorobenzidine is located in Table 3-1.

#### **3.2 PHYSICAL AND CHEMICAL PROPERTIES**

Information regarding the physical and chemical properties of 3,3'-dichlorobenzidine is located in Table 3-2.

## 3. CHEMICAL AND PHYSICAL INFORMATION

**Table 3-1. Chemical Identity of 3,3'-Dichlorobenzidine**

Characteristic	Information	Reference
Chemical name	3,3'-Dichlorobenzidine	Merck 1989
Synonym(s)	Dichlorobenzidine; 3,3'-dichloro(1,1'-biphenyl)-4,4'-diamine; 3,3'-dichloro-4,4'-biphenyldiamine	Merck 1989
Registered trade name(s)	Curithane	IARC 1982a
Chemical formula	C <sub>12</sub> H <sub>10</sub> Cl <sub>2</sub> N <sub>2</sub>	Merck 1989
Chemical structure		Merck 1989
Identification numbers:		
CAS Registry	91-94-1	Merck 1989
NIOSH RTECS	DD0525000	Chapman & Hall Database 1995
EPA Hazardous Waste	U073	HSDB 1996
OHM/TADS	8100004	HSDB 1996
DOT/UN/NA/IMCO	No data	
HSDB	1632	HSDB 1996
NCI	No data	

CAS = Chemical Abstracts Services; DOT/UN/NA/IMCO = Department of Transportation/United Nations/North America/International Maritime Dangerous Goods Code; EPA = Environmental Protection Agency; HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil and Hazardous Materials/Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substances; IARC = International Agency for Research on Cancer

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**Table 3-2. Physical and Chemical Properties of 3,3'-Dichlorobenzidine**

Property	Information	Reference
Molecular weight	253.13	Merck 1989
Color	Gray to purple	Merck 1989
Physical state	Crystalline solid	Lewis 1993
Melting point °C	132–133	Merck 1989
Boiling point °C	402	HSDB 1996
Density at 19 °C	No data	
Odor	No data	
Odor threshold: Air	No data	
Solubility: water at 20 °C	Almost insoluble 3.1 mg/L 4 mg/L (22 °C)	Merck 1989 DCMA 1989 Banerjee et al. 1978
organic solvent(s)	Soluble in alcohol, benzene, Glacial acetic acid	Merck 1989
Partition coefficients:		
Log K <sub>ow</sub>	3.21 3.5	SRC 1995b Nyman et al. 1997
Log K <sub>oc</sub>	3.2 1.43–2.11 at pH7	Mabey et al. 1982 HSDB 1996
Vapor pressure at 20 °C	4.5 x 10 <sup>-9</sup> torr	DCMA 1989
Henry's law constant: at 25 °C	5.11 x 10 <sup>-11</sup> atm.m <sup>3</sup> /mole	SRC 1994
Degradation half-life in air via reaction with OH radicals	9.7 hours = 39.5704x10 <sup>-12</sup> cm <sup>3</sup> /molecule-sec	SRC 1995a
Dissociation constants		
pK <sub>a,1</sub>	1.6	Nyman et al. 1997
pK <sub>a,2</sub>	3.2	Nyman et al. 1997
Autoignition temperature, °C	No data	
Flashpoint	No data	
Flammability limits at 25 °C	No data	
Conversion factors (25 °C)	ppm = 0.0966 x mg/m <sup>3</sup> mg/m <sup>3</sup> = 10.35 x ppm	IARC 1982a
Explosive limits	No data	

