3. CHEMICAL AND PHYSICAL INFORMATION

3.1 CHEMICAL IDENTITY

The chemical identities of various vanadium compounds are shown in Table 3-1.

3.2 PHYSICAL AND CHEMICAL PROPERTIES

The physical and chemical properties of various vanadium compounds are shown in Table 3-2.

Characteristic	Vanadium .	Vanadium pentoxide	Vanadyl sulfate	Sodium metavanadate
Synonyms	Vanadium-51	Vanadic anhydride; divanadium penta- oxide; divanadium pentoxide; vanadium oxide; vanadic acid; vanadic acid anhydride ^{b,c}	Vanadium oxysulfate; oxysulfato vanadium; vanadium oxide sulfate; vanadium oxosulfate; oxo (sulfat (2-)-0)- vanadium ^{e,d}	Vanadic acid; monosodium salt ^d
Trade names	No data	CI 77938	CI 77940	No data
Chemical formula	v	v ₂ 05 ^c	VOSO4 ^d	NaVO3 ^d
Chemical structure	γf	0 v - o - v 0	0=V ⁺² (SO ₄) ⁻²	$Na^{+1} O^{-} - V > 0$
Identification numbers:				
CAS registry NIOSH RTECS EPA hazardous waste OHM/TADS DOT/UN/NA/IMCO shipping HSDB	7440-62-2 ^g YW1355000 ^d No data No data No data 1022	1314-62-1 ^h YW2450000 F120 7217394 DOT/UN2862 ^b 1024	27774-13-3 ^b YW1925000 ^d No data 721694 ^{55d} NA 9152; UN2931 ^d 1026	13718-26-8 ^d YW1050000 ^d No data No data No data No data
NCI	No data	No data	No data	No data

TABLE 3-1. Chemical Identity of Vanadium and Compounds^a

ω • TABLE 3-1 (Continued)

Characteristic	Sodium orthovanadate	Ammonium metavanadate
Synonyms	Vanadic (II) acid; trisodium salt; sodium vanadate; sodium vanadate oxide; trisodium orthovandate ^d	Ammonium vanadate; vanadic acid; ammonium salt
Trade names	No data	No data
Chemical formula	$Na_{3}VO_{4}^{d}$	NH4VO3
Chemical structure	Na ^{+O-} V ⁰ Na ^{+O-} VO- Na ⁺	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$
Identification numbers:		
CAS registry NIOSH RTECS EPA hazardous waste OHM/TADS DOT/UN/NA/IMCO shipping HSDB NCI	13721-39-6 ^d YW1120000 ^d No data No data No data No data No data	7803-55-6 YW0875000 ^d P119 No data UN 2859 ^d 6310 No data

^PSax 1988 ^{CWindholz 1989 ^dRTECS 1989 ^eHawley 1977 ^fPossible valence states: 1-, 0, 2+, 3+, 4+, 5+ ^gGrayson 1989 ^hNIOSH 1985}

CAS = Chemical Abstracts Services; DOT/UN/NA/IMCO = Department of Transportation/ United Nations/North America/International Maritime Dangerous Goods Code; EPA = Environmental Protection Agency; HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil and Hazardous Materials/ Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substances

Property	Vanadium	Vanadium pentoxide	Vanadyl sulfate	Sodium metavanadate
Molecular weight	50.942 ^b	181.9 ^c	163.0 ^c	121.93 ^b
Color	Light grey or white lustrous ^c	Yellow to rust brown crystals ^c	Blue ^c	Colorless ^b
Physical state	Powder or crystal ^c	Crystals ^c	Crystalline powder ^c	Crystals ^b
Melting point	1,890 ± 10°C; 1,917°C ^{b,c}	690°C ^c	No data	630°C ^b
Boiling point	3380°Cb	1,750°C (decomposes) ^d	No data	No data
Density	6.11g/cm ³ at 18.7°C ^c	3.357 g/cm at 18°C	No data	No data
Odor	No data	Odorless	Odorless ^d	No data
Odor threshold:				
Water	No data	No dat a	No data	No data
Air	No datá	No data	No data	No data
Solubility:				
Water at 20°C	Insoluble ^c	1 g/125 mL ^c	Soluble ^c	211 g/L at 25°C; 388 g/L at 75°C ^f
Organic solvents	No data	Soluble in acetone; insoluble in alcohol ^c	No data	No data
Partition coefficients:				
Log octanol/water K _{ow}	No data	No data	No data	No data
Log K _{oc}	No data	No data	No data	No data
Vapor pressure	No data	0 mmHg ^e	No data	No data
Henry's law constant	No data	No data	No data	No data
Autoignition temperature	No data	Not flammable	No data	No data
Flashpoint	No data	Not flammable	Not flammabled	No data
Flammability limits	No data	Not flammable	Not flammable ^d	No data
Conversion factors	No data	No data	No data	No data
Explosive limits	No data	No data	No data	No data

TABLE 3-2. Physical and Chemical Properties of Vanadium and Compounds*

CHEMICAL AND PHYSICAL INFORMATION

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Property	Sodium orthovanadate	Ammonium metavanadate	
Molecular weight	183.91; 183.94 ^{b,1}	116.98 ^b	
Color	Colorless ^b	White-yellowish or colorless ^b	
Physical state	Hexagonal prisms ^g	Crystals ^b	
Melting point	850°C-856°C ^b	200°C (decomposes) ^b	
Boiling point	No data	No data	
Density	No data	2.326 g/cm ³ at 20°C ^b	
Odor	No data	No data	
Odor threshold:			
Water	No data	No data	
Air	No data	No data	
Solubility:			
Water	Soluble in water ^b	Slightly soluble ^b	
Organic solvent	Insoluble in alcohol ^h	Insoluble in alcohol, ether ^f	
Partition coefficients:			
Log octanol/water K _{ow}	No data	No data	
Log Koc	No data	No data	
Vapor pressure	No data	No data	
Henry's law constant	No data	No data	
Autoignition temperature	No data	Not flammable	
Flashpoint	No data	Not flammable	
Flammability limits	No data	Not flammable	
Conversion factors	No data	No data	
Explosive limits	No data	No data	

TABLE 3-2 (Continued)

*All information obtained from HSDB 1990 unless otherwise noted.
bGrayson 1983
CWindholz 1983
dWeiss 1986
*NIOSH 1985
fStokinger 1981
9Sax 1988
*Weast 1983
'RTECS 1989

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