

## APPENDIX C. NUCLEAR SPECTROSCOPY STANDARDS

## 1. Gamma-ray Energy and Intensity Standards

Table 1 lists some  $\gamma$ -ray energy standards, from the evaluation of Helmer *et al.*<sup>1</sup>, and intensity standards, recommended by the IAEA Co-ordinated Research Programme<sup>2,3</sup> (CRP), for calibration of  $\gamma$ -ray measurements. Most of the isotopes given here have half-lives of more than 30 days, and many are commercially available. The  $\gamma$ -ray energies are based on the *gold standard*, the 411.80205 17 keV transition from <sup>198</sup>Au decay. Uncertainties are intended to represent one standard deviation, and include the 0.3 ppm uncertainty in the definition of the electron volt relative to wavelength. The  $\gamma$ -ray energies reported in Table 1 are from absolute wavelength or curved-crystal spectrometer measurements, which are tied directly to the *gold standard*, and from the measurements of small  $\gamma$ -ray energy differences using Ge detectors. Energies that are rounded to the nearest 0.1-keV and tabulated without uncertainty are not recommended values; however, they have been included because these transitions are useful intensity calibration standards. Other, apparently precise, transition energies and intensities have been tabulated in the *Table of Isotopes*, but the reader should use these values with great caution because of unknown systematic uncertainties which may not have been included. Columns 1 and 2 show the isotope names and half-lives from the *Table of Isotopes*, respectively. Columns 3 and 4 list the  $\gamma$ -ray energies and intensities with their corresponding uncertainties (in italics) in the least significant digit(s).

<sup>1</sup> R.G. Helmer and C. van der Leun, private communication, draft of a paper to be submitted to *Nucl. Instr. Meth.*, 1999.

<sup>2</sup> *X-ray and Gamma-ray Standards for Detector Calibration*, report by the Co-ordinated Research IAEA Programme, IAEA-TECDOC-619 (1991).

<sup>3</sup> R. Vaninbroux, *Emission Probabilities of Selected Gamma Rays for Radionuclides Used as Detector-Calibration Standards*, report presented at the Advisory Group Meeting of the International Atomic Energy Agency (IAEA), Vienna (1985).

**Table 1. Gamma-ray Energies and Absolute Intensities for Some Standard Sources**

Source	Half-life	$E_{\gamma}$ (keV) #	$I_{\gamma}$ (%) &	Source	Half-life	$E_{\gamma}$ (keV) #	$I_{\gamma}$ (%) &
<sup>7</sup> Be	53.12 d	477.6035 <sub>20</sub>	10.45 <sub>10</sub> <sup>†</sup>	<sup>59</sup> Fe	44.503 d	142.651 <sub>2</sub>	
<sup>22</sup> Na	2.6019 y	1274.537 <sub>7</sub>	99.935 <sub>15</sub>			192.349 <sub>5</sub>	
<sup>24</sup> Na	14.9590 h	1368.625 <sub>5</sub>	99.9936 <sub>15</sub>			1099.245 <sub>3</sub>	
		2754.008 <sub>11</sub>	99.855 <sub>5</sub>			1291.590 <sub>6</sub>	
<sup>35</sup> Cl(n, $\gamma$ )		517.1	0.227 <sub>20</sub>	<sup>56</sup> Co	77.27 d	846.7638 <sub>19</sub>	99.933 <sub>7</sub>
		786.3	0.096 <sub>9</sub>			1037.8333 <sub>24</sub>	14.13 <sub>5</sub>
		788.4	0.150 <sub>12</sub>			1175.0878 <sub>22</sub>	2.239 <sub>11</sub>
		1164.9	0.257 <sub>22</sub>			1238.2736 <sub>22</sub>	66.07 <sub>19</sub>
		1600.8	0.034 <sub>3</sub>			1360.196 <sub>4</sub>	4.256 <sub>15</sub>
		1951.1	0.187 <sub>15</sub>			1771.327 <sub>3</sub>	15.49 <sub>5</sub>
		1959.3	0.121 <sub>10</sub>			2015.176 <sub>5</sub>	3.029 <sub>13</sub>
		2863.9	0.060 <sub>5</sub>			2034.752 <sub>5</sub>	7.771 <sub>27</sub>
		3061.7	0.035 <sub>3</sub>			2113.092 <sub>6</sub>	0.366 <sub>6</sub>
		5715.2	0.051 <sub>4</sub>			2212.898 <sub>3</sub>	0.390 <sub>7</sub>
		6110.8	0.197 <sub>16</sub>			2598.438 <sub>4</sub>	16.96 <sub>6</sub>
		6619.4	0.081 <sub>7</sub>			3009.559 <sub>4</sub>	0.995 <sub>21</sub>
		6627.5	0.046 <sub>4</sub>			3201.930 <sub>11</sub>	3.13 <sub>9</sub>
		6977.6	0.0223 <sub>20</sub>			3253.402 <sub>5</sub>	7.62 <sub>24</sub>
		7413.7	0.100 <sub>8</sub>			3272.978 <sub>6</sub>	1.78 <sub>6</sub>
		7790.0	0.086 <sub>7</sub>			3451.119 <sub>4</sub>	0.93 <sub>4</sub>
		8578.2	0.0294 <sub>24</sub>			3548.3	0.178 <sub>9</sub>
<sup>46</sup> Sc	83.79 d	889.271 <sub>2</sub>	99.9844 <sub>16</sub>	<sup>57</sup> Co	271.79 d	14.4130 <sub>4</sub>	9.16 <sub>15</sub>
		1120.537 <sub>3</sub>	99.9874 <sub>11</sub>			122.06065 <sub>12</sub>	85.60 <sub>17</sub>
<sup>44</sup> Ti	63 y	67.8688 <sub>17</sub>				136.47356 <sub>29</sub>	10.68 <sub>8</sub>
		78.3236 <sub>17</sub>		<sup>58</sup> Co	70.86 d	810.7593 <sub>20</sub>	99.45 <sub>1</sub>
<sup>51</sup> Cr	27.7025 d	320.0824 <sub>4</sub>	9.86 <sub>5</sub>			863.951 <sub>6</sub>	0.69 <sub>3</sub>
<sup>54</sup> Mn	312.3 d	834.838 <sub>5</sub>	99.9758 <sub>24</sub>			1674.725 <sub>7</sub>	0.519 <sub>10</sub>
<sup>56</sup> Mn	2.5785 h	846.8	98.87 <sub>3</sub> <sup>†</sup>	<sup>60</sup> Co	5.2714 y	1173.228 <sub>3</sub>	99.857 <sub>22</sub>
		1810.7	27.2 <sub>8</sub> <sup>†</sup>			1332.492 <sub>4</sub>	99.983 <sub>6</sub>
		2113.0	14.3 <sub>4</sub> <sup>†</sup>	<sup>65</sup> Zn	244.26 d	1115.539 <sub>2</sub>	50.60 <sub>24</sub>

Table 1. Gamma-ray Energies and Absolute Intensities (continued)

Source	Half-life	$E_{\gamma}$ (keV) <sup>#</sup>	$I_{\gamma}$ (%) <sup>&amp;</sup>	Source	Half-life	$E_{\gamma}$ (keV) <sup>#</sup>	$I_{\gamma}$ (%) <sup>&amp;</sup>			
<sup>66</sup> Ga	9.49 h	833.5324 <sup>21</sup>	6.03 <sup>23</sup>	<sup>108m</sup> Ag	418 y	433.937 <sup>4</sup>				
		1039.220 <sup>3</sup>	37.9 <sup>12</sup>			614.276 <sup>4</sup>				
		1333.112 <sup>5</sup>	1.23 <sup>5</sup>			722.907 <sup>10</sup>				
		1418.754 <sup>5</sup>					<sup>110m</sup> Ag	249.79 d	446.812 <sup>3</sup>	3.72 <sup>3 †</sup>
		1508.158 <sup>7</sup>				620.3553 <sup>17</sup>				
		1918.329 <sup>5</sup>	2.14 <sup>8</sup>			657.7600 <sup>11</sup>			94.4 <sup>1 †</sup>	
		2189.616 <sup>6</sup>	5.71 <sup>21</sup>			677.6217 <sup>12</sup>			10.40 <sup>8 †</sup>	
		2422.525 <sup>7</sup>	1.96 <sup>7</sup>			687.0091 <sup>18</sup>			6.44 <sup>3 †</sup>	
		2751.835 <sup>5</sup>	23.2 <sup>11</sup>			706.6760 <sup>15</sup>			16.6 <sup>1 †</sup>	
		3228.800 <sup>6</sup>	1.48 <sup>12</sup>			744.2755 <sup>18</sup>			4.70 <sup>4 †</sup>	
		3380.850 <sup>6</sup>	1.40 <sup>12</sup>			763.9424 <sup>17</sup>			22.39 <sup>8 †</sup>	
		3422.040 <sup>8</sup>				818.0244 <sup>18</sup>			7.32 <sup>4 †</sup>	
		3791.009 <sup>6</sup>	1.02 <sup>11</sup>			884.6781 <sup>13</sup>			72.7 <sup>3 †</sup>	
		4085.853 <sup>9</sup>	1.14 <sup>19</sup>			937.485 <sup>3</sup>			34.31 <sup>12 †</sup>	
		4295.7	3.5 <sup>7</sup>			1384.2931 <sup>20</sup>			24.25 <sup>8 †</sup>	
		4461.202 <sup>9</sup>				1475.7792 <sup>23</sup>			3.99 <sup>2 †</sup>	
4806.007 <sup>10</sup>	1.5 <sup>4</sup>	1505.0280 <sup>20</sup>	13.04 <sup>4 †</sup>							
		1562.2940 <sup>18</sup>								
<sup>75</sup> Se	119.779 d	66.0518 <sup>8</sup>	1.10 <sup>2</sup>	<sup>109</sup> Cd	462.6 d	88.03360 <sup>10</sup>			3.63 <sup>2</sup>	
		96.7340 <sup>9</sup>	3.41 <sup>4</sup>			<sup>111</sup> In	2.8047 d	171.3	90.78 <sup>10</sup>	
		121.1155 <sup>11</sup>	17.1 <sup>1</sup>	245.4	94.16 <sup>3</sup>					
		136.0001 <sup>6</sup>	58.8 <sup>3</sup>	<sup>115m</sup> In	4.486 h	336.2	45.9 <sup>2 †</sup>			
		198.6060 <sup>12</sup>	1.49 <sup>1</sup>			<sup>113</sup> Sn	115.09 d	391.698 <sup>3</sup>	64.89 <sup>13</sup>	
		264.6576 <sup>9</sup>	59.0 <sup>2</sup>	<sup>125</sup> Sn	9.64 d			1806.690 <sup>16</sup>		
		279.5422 <sup>10</sup>	25.0 <sup>1</sup>			1889.884 <sup>16</sup>				
		303.9236 <sup>10</sup>	1.31 <sup>1</sup>	2002.134 <sup>12</sup>						
		400.6572 <sup>8</sup>	11.5 <sup>1</sup>	2201.002 <sup>12</sup>						
				2275.748 <sup>10</sup>						
<sup>82</sup> Br	35.30 h	221.4788 <sup>18</sup>		<sup>124</sup> Sb	60.20 d	602.7260 <sup>23</sup>	98.0 <sup>1 †</sup>			
		554.346 <sup>3</sup>				645.8520 <sup>19</sup>	7.3 <sup>1 †</sup>			
		619.104 <sup>3</sup>				713.776 <sup>4</sup>				
		698.368 <sup>3</sup>				722.782 <sup>3</sup>	11.3 <sup>2 †</sup>			
		776.513 <sup>4</sup>				790.706 <sup>7</sup>				
		827.820 <sup>5</sup>				968.195 <sup>4</sup>				
		1043.993 <sup>5</sup>				1045.125 <sup>4</sup>				
		1317.466 <sup>4</sup>				1325.504 <sup>4</sup>				
		1474.874 <sup>5</sup>				1368.157 <sup>5</sup>				
		1650.328 <sup>5</sup>				1436.554 <sup>7</sup>				
						1690.971 <sup>4</sup>	48.5 <sup>3 †</sup>			
<sup>84</sup> Rb	32.77 d	881.6041 <sup>16</sup>		<sup>125</sup> Sb	2.7582 y	2090.930 <sup>7</sup>	5.66 <sup>9 †</sup>			
		1016.158 <sup>11</sup>				176.314 <sup>2</sup>	6.85 <sup>7</sup>			
<sup>85</sup> Sr	64.84 d	514.0048 <sup>22</sup>	98.4 <sup>4</sup>	380.5	1.518 <sup>16</sup>					
<sup>88</sup> Y	106.65 d	898.036 <sup>4</sup>	94.0 <sup>3</sup>	427.874 <sup>4</sup>	29.7 <sup>3</sup>					
		1836.052 <sup>13</sup>	99.36 <sup>3</sup>	463.365 <sup>4</sup>	10.48 <sup>11</sup>					
<sup>95</sup> Zr	64.02 d	724.193 <sup>3</sup>	44.15 <sup>20 †</sup>	600.597 <sup>2</sup>	17.73 <sup>18</sup>					
		756.7	54.50 <sup>25 †</sup>	606.713 <sup>3</sup>	5.00 <sup>5</sup>					
<sup>94</sup> Nb	2.03×10 <sup>4</sup> y	702.639 <sup>4</sup>	99.79 <sup>5</sup>	635.950 <sup>3</sup>	11.21 <sup>12</sup>					
		871.114 <sup>3</sup>	99.86 <sup>5</sup>	671.441 <sup>6</sup>	1.80 <sup>2</sup>					
<sup>95</sup> Nb	34.975 d	765.803 <sup>6</sup>	99.81 <sup>3</sup>	<sup>125</sup> I	59.408 d	35.5	6.58 <sup>8</sup>			
<sup>99</sup> Mo	65.94 h	40.58323 <sup>17</sup>				<sup>132</sup> Cs	6.479 d	667.714 <sup>2</sup>		
		140.511 <sup>1</sup>		1317.918 <sup>6</sup>						
<sup>95m</sup> Tc	61 d	204.1161 <sup>17</sup>		1985.625 <sup>6</sup>						
		582.0775 <sup>21</sup>								
		786.1922 <sup>27</sup>								
		820.622 <sup>7</sup>								
		835.146 <sup>6</sup>								
<sup>99m</sup> Tc	6.01 h	140.511 <sup>1</sup>	89.0 <sup>2 †</sup>							
<sup>106</sup> Ru	373.59 d	511.8534 <sup>23</sup>								

Table 1. Gamma-ray Energies and Absolute Intensities (continued)

Source	Half-life	$E_{\gamma}$ (keV) <sup>#</sup>	$I_{\gamma}$ (%) <sup>&amp;</sup>	Source	Half-life	$E_{\gamma}$ (keV) <sup>#</sup>	$I_{\gamma}$ (%) <sup>&amp;</sup>
<sup>134</sup> Cs	2.0648 y	475.4	1.49 <sup>2</sup>	<sup>160</sup> Tb	72.3 d	86.7877 <sup>3</sup>	
		563.2	8.36 <sup>3</sup>			197.0341 <sup>10</sup>	
		569.3	15.39 <sup>6</sup>			215.6452 <sup>11</sup>	
		604.7	97.63 <sup>6</sup>			298.5783 <sup>17</sup>	
		795.8	85.4 <sup>3</sup>			879.378 <sup>2</sup>	
		801.9	8.69 <sup>3</sup>			962.311 <sup>3</sup>	
		1038.6	0.990 <sup>5</sup>			966.166 <sup>2</sup>	
		1168.0	1.792 <sup>7</sup>			1177.954 <sup>3</sup>	
		1365.2	3.016 <sup>11</sup>			1271.873 <sup>5</sup>	
<sup>137</sup> Cs	30.07 y	661.657 <sup>3</sup>	85.1 <sup>2</sup>	<sup>161</sup> Tb	6.88 d	25.65135 <sup>3</sup>	
<sup>133</sup> Ba	10.51 y	53.1622 <sup>6</sup>				48.91533 <sup>5</sup>	
		79.6142 <sup>12</sup>				57.1917 <sup>3</sup>	
		80.9979 <sup>11</sup>	34.11 <sup>28</sup>	74.56669 <sup>6</sup>			
		160.6120 <sup>16</sup>		84.25474 <sup>8</sup>			
		223.2368 <sup>13</sup>		<sup>170</sup> Tm	128.6 d	63.12044 <sup>4</sup>	
		276.3989 <sup>12</sup>	7.147 <sup>30</sup>			93.61447 <sup>8</sup>	
		302.8508 <sup>5</sup>	18.30 <sup>6</sup>	<sup>169</sup> Yb	32.026 d	109.77924 <sup>4</sup>	
		356.0129 <sup>7</sup>	61.94 <sup>14</sup>			118.18940 <sup>14</sup>	
		383.8485 <sup>12</sup>	8.905 <sup>29</sup>			130.52293 <sup>6</sup>	
<sup>139</sup> Ce	137.640 d	165.857 <sup>3</sup>	79.87 <sup>6</sup>			177.21307 <sup>6</sup>	
<sup>141</sup> Ce	32.501 d	145.4433 <sup>14</sup>	48.6 <sup>4 †</sup>			197.95675 <sup>7</sup>	
<sup>144</sup> Ce	284.893 d	696.505 <sup>4</sup>				261.07712 <sup>9</sup>	
		1489.148 <sup>3</sup>				307.73586 <sup>10</sup>	
<sup>152</sup> Eu	13.537 y	2185.645 <sup>5</sup>		<sup>172</sup> Hf	1.87 y	23.9330 <sup>2</sup>	
		121.7817 <sup>3</sup>	28.37 <sup>13</sup>			78.7422 <sup>6 †</sup>	
		244.6974 <sup>8</sup>	7.53 <sup>4</sup>			81.7509 <sup>5 †</sup>	
		295.9387 <sup>17</sup>		90.6434 <sup>19</sup>			
		344.2785 <sup>12</sup>	26.57 <sup>11</sup>	<sup>182</sup> Ta	114.43 d	65.72215 <sup>15</sup>	
		367.7891 <sup>20</sup>				67.74970 <sup>10</sup>	
		411.1165 <sup>12</sup>	2.238 <sup>10</sup>			84.68024 <sup>26</sup>	
		444.0	3.125 <sup>14</sup>			100.10595 <sup>7</sup>	14.23 <sup>25 †</sup>
		778.9045 <sup>24</sup>	12.97 <sup>6</sup>			113.67170 <sup>22</sup>	
		867.380 <sup>3</sup>	4.214 <sup>25</sup>			116.4179 <sup>6</sup>	
		964.1	14.63 <sup>6</sup>			152.42991 <sup>26</sup>	7.02 <sup>8 †</sup>
		1085.837 <sup>10</sup>	10.13 <sup>5</sup>			156.3864 <sup>3</sup>	
		1089.737 <sup>5</sup>	1.731 <sup>9</sup>			179.39381 <sup>25</sup>	
1112.076 <sup>3</sup>	13.54 <sup>6</sup>			198.35187 <sup>29</sup>			
1212.948 <sup>11</sup>	1.412 <sup>8</sup>			222.1085 <sup>3</sup>	7.57 <sup>8 †</sup>		
1299.142 <sup>8</sup>	1.626 <sup>11</sup>			229.3207 <sup>6</sup>			
1408.013 <sup>3</sup>	20.85 <sup>9</sup>			264.0740 <sup>3</sup>			
1457.643 <sup>11</sup>				1121.290 <sup>3</sup>	35.3 <sup>2 †</sup>		
<sup>154</sup> Eu	8.593 y	123.0706 <sup>9</sup>	41.2 <sup>5</sup>			1157.302 <sup>3</sup>	
		247.9288 <sup>7</sup>	6.95 <sup>9</sup>			1189.040 <sup>3</sup>	16.42 <sup>10 †</sup>
		591.755 <sup>3</sup>	4.99 <sup>6</sup>			1221.395 <sup>3</sup>	27.20 <sup>22 †</sup>
		723.3014 <sup>22</sup>	20.2 <sup>2</sup>			1231.004 <sup>3</sup>	11.57 <sup>8 †</sup>
		756.8020 <sup>23</sup>	4.58 <sup>6</sup>			1257.407 <sup>3</sup>	
		873.1834 <sup>23</sup>	12.24 <sup>15</sup>			1273.719 <sup>3</sup>	
		996.3	10.48 <sup>13</sup>			1289.145 <sup>3</sup>	
		1004.7	18.2 <sup>2</sup>			1373.824 <sup>3</sup>	
		1274.429 <sup>4</sup>	35.0 <sup>4</sup>			1387.390 <sup>3</sup>	
		1494.048 <sup>5</sup>	0.71 <sup>2</sup>			125.3581 <sup>9</sup>	
1596.4804 <sup>28</sup>	1.81 <sup>2</sup>			162.852 <sup>3</sup>			
<sup>153</sup> Gd	240.4 d	69.67300 <sup>13</sup>		<sup>185</sup> Os	93.6 d	234.156 <sup>4</sup>	
		75.42213 <sup>23</sup>				592.0722 <sup>22</sup>	
		83.36717 <sup>21</sup>				646.127 <sup>4</sup>	
		89.48595 <sup>22</sup>				717.4298 <sup>24</sup>	
		97.43100 <sup>21</sup>				874.826 <sup>4</sup>	
		103.18012 <sup>17</sup>				880.2816 <sup>27</sup>	
		172.85307 <sup>19</sup>					

Table 1. Gamma-ray Energies and Absolute Intensities (continued)

Source	Half-life	$E_{\gamma}$ (keV) <sup>#</sup>	$I_{\gamma}$ (%) <sup>&amp;</sup>	Source	Half-life	$E_{\gamma}$ (keV) <sup>#</sup>	$I_{\gamma}$ (%) <sup>&amp;</sup>
<sup>192</sup> Ir	73.831 d	136.34257 <sup>26</sup>		<sup>210</sup> Pb	22.3 y	46.539 <sup>1</sup>	
		205.79430 <sup>9</sup>				<sup>207</sup> Bi	31.55 y
		295.95650 <sup>15</sup>	28.7 <sup>†</sup>	1063.656 <sup>3</sup>	74.5 <sup>2</sup>		
		308.45507 <sup>17</sup>	29.8 <sup>†</sup>	1770.228 <sup>9</sup>	6.87 <sup>4</sup>		
		316.50618 <sup>17</sup>	83.0 <sup>‡</sup>	<sup>228</sup> Th <sup>@</sup>	1.9131 y	84.4	1.22 <sup>2</sup>
		416.4688 <sup>7</sup>				238.6	43.5 <sup>4</sup>
		468.06885 <sup>26</sup>	47.7 <sup>‡</sup>			241.0	4.10 <sup>5</sup>
		484.5751 <sup>4</sup>				277.4	2.30 <sup>3</sup>
		588.5810 <sup>7</sup>	4.49 <sup>‡</sup>			300.1	3.25 <sup>3</sup>
		604.41105 <sup>25</sup>	8.11 <sup>‡</sup>			510.8	8.18 <sup>10</sup>
		612.46215 <sup>26</sup>	5.28 <sup>‡</sup>			583.187 <sup>2</sup>	30.6 <sup>2</sup>
884.5365 <sup>7</sup>		727.3	6.69 <sup>9</sup>				
860.6	4.50 <sup>4</sup>	1620.7	1.49 <sup>5</sup>				
<sup>198</sup> Au	2.69517 d	411.80205 <sup>17</sup>	95.6 <sup>5</sup>	<sup>239</sup> Np	2.3565 d	106.1	26.7 <sup>4</sup>
		675.8836 <sup>7</sup>	0.806 <sup>7</sup>			228.2	11.12 <sup>15</sup>
		1087.6842 <sup>7</sup>	0.159 <sup>3</sup>			277.6	14.31 <sup>20</sup>
<sup>199</sup> Au	3.139 d	49.82635 <sup>12</sup>		<sup>241</sup> Am	432.2 y	26.3446 <sup>2</sup>	2.4 <sup>1</sup>
		158.37851 <sup>10</sup>				59.5409 <sup>2</sup>	36.0 <sup>4</sup>
		208.20481 <sup>12</sup>				43.5	5.94 <sup>11</sup>
<sup>203</sup> Hg	46.612 d	279.1952 <sup>10</sup>	81.48 <sup>8</sup>	<sup>243</sup> Am	7370 y	74.7	67.4 <sup>10</sup>
		<sup>203</sup> Pb	51.873 h			279.1952 <sup>10</sup>	
401.320 <sup>4</sup>							
680.515 <sup>3</sup>							

<sup>#</sup> From reference 1 when listed with uncertainty. Otherwise rounded to the nearest 0.1 keV.

<sup>&</sup> From reference 2, except where indicated.

<sup>†</sup> From reference 3.

<sup>‡</sup> In equilibrium with <sup>172</sup>Lu (6.70 d).

<sup>@</sup> In equilibrium with decay daughter isotopes.